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Dynamics, distribution, and transformations of mercury species from pyrenean highaltitude lakes

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CRediT Author Statements

- Bastien Duval, Alberto de Diego and David Amouroux devised the research.
- Bastien Duval wrote the manuscript.
- Bastien Duval and Emmanuel Tessier performed all mercury measurements.
- Acquisition of the financial support for the project leading to this publication by David Amouroux and
- Alberto de Diego.
- All co-authors helped in the data interpretation and manuscript writing.

Justine Justice Richard

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Abstract

 While mercury (Hg) is a major concern in all aquatic environments because of its methylation and biomagnification pathways, very few studies consider Hg cycling in remote alpine lakes which are sensitive ecosystems. Nineteen high-altitude pristine lakes from Western / Central Pyrenees were investigated on both northern (France) and southern (Spain) slopes (1620 to 2600 m asl.). Subsurface water samples were collected in June 2017/2018/2019 and October 2017/2018 for Hg speciation analysis of inorganic mercury (iHg(II)), monomethylmercury (MMHg), and dissolved gaseous mercury (DGM) to investigate spatial and seasonal variations. In June 2018/2019 and October 2018, more comprehensive studies were performed in four lakes by taking water column depth profiles. Besides, in-situ incubation experiments using isotopically 38 enriched Hg species (iHg(II), 201 MMHg) were conducted to investigate Hg transformation 39 mechanisms in the water column. While iHg(II) (0.08 to 1.10 ng L^{-1} in filtered samples; 0.11 to 1.19 ng L⁻¹ in unfiltered samples) did not show significant seasonal variations in the subsurface 41 water samples, MMHg (<0.03 to 0.035 ng $L⁻¹$ in filtered samples; <0.03 to 0.062 ng $L⁻¹$ in unfiltered samples) was significantly higher in October 2018, mainly because of in-situ methylation. DGM $(0.02 \text{ to } 0.68 \text{ ng } L^{-1})$ varies strongly and can exhibit higher levels in comparison with other pristine areas. Depth profiles and incubation experiments highlighted the importance of in-situ biotic methylation triggered by anoxic conditions in bottom waters. In-situ incubations confirm that significant methylation, demethylation and photoreduction extents are taking place in the water columns. Overall, drastic environmental changes occurring daily and seasonally in alpine lakes are providing conditions that can both promote Hg methylation (stratified anoxic waters) and MMHg photodemethylation (intense UV light). In addition, light induced photoreduction is a major pathway controlling significant gaseous Hg evasion. Global warming and potential eutrophication may thus have direct implications on Hg turnover and MMHg burden in those remote ecosystems. I October 2018, more comprehensive studies were performed blumn depth profiles. Besides, in-situ incubation experiments pecies $(^{199}Hg(II), ^{201}MMHg)$ were conducted to investigate I the water column. While iHg(II) (0.08

Keywords:

Mercury; Biogeochemistry; Alpine lakes; Methylation; Demethylation; Photoreduction

1. Introduction

 Natural sources of mercury (Hg) (volcanic activities and degassing, evasion from aquatic and terrestrial surfaces) and increasing anthropogenic sources (e.g. fossil fuel combustion and gold mining), combined to a high volatility, contribute to the global pool of Hg in the atmosphere 59 and lead to long-range dispersion and deposition away from point sources $1-3$. Hg has also created a scenario of global health concerns due to its conversion into an organometallic compound of elevated neurotoxicity, namely monomethylmercury (MMHg).

 Hg is a biogeochemically driven contaminant because its most harmful organometallic species (MMHg) is naturally produced in aquatic systems thanks to a complex interplay between microbial and chemical processes. MMHg is a contaminant that biomagnifies from one trophic level to another, leading to elevated concentration on top of food chains. The European Water 66 Framework Directive (WFD-2000/60/EG) classifies Hg as one of the 30th most "precarious dangerous pollutants". Human exposure to MMHg is well defined, and mainly associated with fish 68 . consumption ⁴. Yet, most risk assessment studies lack fundamental information on what exactly controls Hg levels in fish. Without detailed knowledge of the Hg biogeochemical cycle and, in 70 particular, the fundamental processes controlling MMHg production $5-7$, it is difficult to precisely 71 estimate Hg's health impacts and socio-economical costs . a biogeochemically driven contaminant because its most harm
a) is naturally produced in aquatic systems thanks to a complex
themical processes. MMHg is a contaminant that biomagnifies
r, leading to elevated concentration o

 The net production of MMHg in aquatic ecosystems is closely related to physical conditions and ecological characteristics, such as temperature, pH, organic matter, redox conditions, ionic strength and either solar radiation or the occurrence of microorganisms, that are able to transform various Hg compounds. Consequently, all these environmental settings have a direct influence on the Hg compounds reactivity or bioavailability, and regulate the extent of 77 methylation, demethylation and reduction pathways ⁹. MMHg bioaccumulation in aquatic food chains is largely dependent on these biotic and abiotic pathways that control its production and degradation or the removal of Hg from the ecosystem (e.g., volatilization, sedimentation). In most surface aquatic systems, a large array of conditions can control these various pathways according to geological, hydroecological or climatic conditions. However, the methylation of inorganic Hg (iHg(II)) in aquatic ecosystems is more specifically due to anaerobic microorganisms such as sulphate-reducing or iron-reducing bacteria, among other groups ^{10,11}.

84 Among the aquatic compartments, lowland lakes have been widely studied regarding Hg 85 pollution, speciation and species transformations ^{12–15}. Several studies highlighted a link between 86 organic matter and Hg in lake ecosystems $13,16$, with evidence of the catchment influence on the 87 Hg levels. Significant photo-reduction and -demethylation were observed in lakes ^{17,18}, exceeding 88 the external inputs of MMHg (rain, snow, runoff) and therefore suggesting MMHg sources from 89 bottom sediments. Nevertheless, the biogeochemical processes that control the speciation and 90 fate of Hg, and especially MMHg levels, remain poorly established 12 . Alpine lakes are better 91 witnesses of Hg past and present contamination in comparison with lowland lakes. They are 92 usually characterized by a low catchment-to-lake-surface-area ratio, as the catchment surface is 93 commonly smaller than lowland lakes. Therefore, atmospheric deposition and catchment 94 weathering are important processes influencing greatly the lake water geochemistry ¹⁹. Thus, 95 mountain lake catchments are viewed and used as excellent proxies of background diffuse 96 contamination 3,20. While alpine lakes, through sediment cores analysis, have been successfully 97 used to reconstruct temporal trends in atmospheric Hg deposition $2,3,21$, Hg behaviour in the 98 aquatic compartment of those pristine areas has been barely investigated. For instance, *Guédron* 99 *et al. ^{22,23}* and *Alanoca et al. ²⁴ have reported Hg cycling features in high-altitude Bolivian lakes.* 100 Despite anthropogenic influences in these urban and mining areas, Hg species cycling is highly 101 dynamic and MMHg production was driven by eutrophication in these lakes, while intense UV 102 light promoted significant demethylation and reduction pathways. Regarding pristine high-altitude 103 lakes, Hg species distribution has been only investigated in one study on four lakes from the 104 **French Alps** 25 . Thus, while Hg transformations in the aquatic compartment have been well 105 described in lower altitude or more accessible aquatic systems $15,24,26-33$, there is a lack of studies 106 in more pristine areas such as alpine lakes, particularly sensitive to global changes and long 107 range transport of pollutants $20,34$. erized by a low catchment-to-lake-surface-area ratio, as the care
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In that sense, *Chételat et al.* ¹² suggested that profound environmental changes may be impacting the cycling and bioaccumulation of Hg. The responsibility of Humans on the dispersion 110 of aerosols and Hg is undeniable $35,36$, yet it is important to take into account the intensification of natural processes occurring through Global Change in the biogeochemical cycle of Hg species. In particular, Climate Change also accelerates many processes having an impact on alpine lakes dynamics: droughts and losses of snow cover that enhance dust production and soil runoff,

 potential decreases of cloud density and ice cover period in lakes, surface waters temperature 115 increase and water column stratification . All these processes will have a direct effect on the aquatic cycling of Hg through enhance microbial or photochemical transformations of Hg or drastic changes in the transport of Hg via inputs to lake waters or sediments and atmospheric exchanges. To better understand the potential role of global change of the Hg cycling in high altitude lakes, we report here for the first time a complete inventory of Hg levels and speciation (iHg(II), MMHg, DGM) as well as in situ measurements of Hg compounds transformation rates in the water column of alpine lakes from the Central - Western Pyrenees. The main objectives of this study were i) to assess the Hg species levels and variability in the aquatic compartment of these ecosystems, and better understand how seasonal or geographical conditions and lake trophic status may control such variability ii) to evaluate the importance of abiotic and biotic pathways affecting the fate of Hg in mountain lakes, and further estimate how selected lakes can be considered as sources of methylmercury and either enhance Hg burial or reemission iii) to highlight potential effects of climatic and biogeochemical changes on these identified pathways and their consequence on the future Hg cycling in remote and alpine aquatic environments. best the Hg species levels and variability in the aquatic comparable better understand how seasonal or geographical conditions to late the importance of abiotic and to late of Hg in mountain lakes, and further estimate how

2. Material and methods

2.1. Site description

 The 19 studied high-altitude lakes (elevation: 1640 – 2600 m asl) are located in three different valleys of the Central-Western Pyrenees (France/Spain): Cauterets (n=7), Panticosa (n=9) and Ayous (n=3) areas (**see Figure 1, Figure A1 and Table A1**). These small lakes show similar physical properties (lake size: 0.37 – 12.82 ha; catchment size: 15 – 3229 ha; maximum depth: 3 – 35 m) but differ from their catchment characteristics and geological background, i.e. mainly granite core (pDe-GR) versus sedimentary rocks (Devonian, De-SR; Permo-Triassic, PT-137 SR; Cretaceous Cr-SR)^{37,38}. Bioclimatic conditions are also substantially different with a decrease 138 in temperature $(2 - 19 \degree C)$, an increase in precipitation and an increase in the duration of the snow 139 cover (50 – 246 days per year; determined according to *Gascoin et al.* ³⁹ using data from Theia Snow collection [\(https://www.theia-land.fr/product/neige/\)](https://www.theia-land.fr/product/neige/) and published elsewhere (*Duval, PhD Thesis* ⁴⁰) at higher altitude. It is worth noting that the winter 2017 – 2018 was colder and/or with

 much higher wet deposition and snow accumulation than the winters 2016 – 2017 and 2018 – 2019.

 The water geochemistry of these lakes has been described elsewhere (*Duval, PhD* 145 *Thesis* ⁴⁰) and the raw data are gathered in **Table A2**. Basically, despite the increase of the primary productivity during the summer (increasing Total Organic Carbon (TOC) associated to 147 decreasing nitrate $(NO₃))$ and the influence of agropastoralism on lakes, all the studied lakes are 148 classified as oligotrophic. The results, in accordance with previous Pyrenean lakes studies ^{19,41-} ⁴³, highlighted the importance of the geological background, the atmospheric inputs, and, although very limited, the Human impact on the aquatic geochemistry of these high-altitude lakes.

Figure 1 : Studied area; circles show position of the lakes, and colors indicate the elevation of the corresponding lakes. Lake acronyms are detailed in **Table A1**. See **Figure A1** for the whole map, including details about the geology of the lakes' catchment.

2.2. Water samples collection and processing

 Subsurface (~0.5 m depth) water samples were collected during all the sampling campaigns (June 2017 [15 lakes, n=81]; October 2017 [10 lakes, n=58]; June 2018 [16 lakes, n=50]; October 2018 [16 lakes, n=40]; June 2019 [4 lakes, n=18]) with triplicate for the first two sampling campaigns to check for the spatial variability in the studied lakes. Additionally, water column profiles were investigated in Lac Gentau (GEN, 5 depths), Lac d'Arratille (ARA, 3 depths), Ibón de Sabocos (SAB, 6 depths) and Ibón Azul Alto (AZU, 3 depths), by sampling at different depths along the day. This study provides for the first time a unique Hg species data set with a total of 319 water samples (**Table A3 and Table A4**).

181 The particles in the water have a key role in the transport and fate of Hg species ¹⁶. Thus, to evaluate their influence on the distribution of Hg species, two kinds of samples were collected and further analysed: filtered (F, dissolved fraction) and unfiltered (UF, total fraction). It is worth noting that considering the general oligotrophic status and elevation of the studied lakes, the particles were scarce and thus it was impossible to collect and analyse them.

 Briefly, all the material needed up to the lake was transported by hiking. An inflatable rubber boat was used to reach the sampling locations. To collect the subsurface water samples, a manually operated ultra-clean sampler (*5L Teflon Coated Go-Flo Water Sampler, General Oceanic*) was deployed using powder-free gloves and avoiding the surface microlayer. For depth sampling, the Go-Flo sampler was deployed using a Kevlar cable and operated at the required depth with a plastic-coated messenger. Back ashore, the water sample was distributed with a clean silicone tubing in dedicated flasks according to the parameter to be further analysed. ne day. This study provides for the first time a unique Hg speciter samples (Table A3 and Table A4).

Hericles in the water have a key role in the transport and fate of H₁

ir influence on the distribution of Hg species,

 First, an appropriate subsampling procedure was applied for dissolved gaseous Hg species (DGM). A 250 mL Teflon container was filled overflow without headspace and somehow following the protocol for the Winkler method (i.e. dissolved oxygen). Then, aliquots for unfiltered and filtered (Sterivex Filter unit, PVDF, 0.22 µm) non-gaseous Hg species (iHg(II), MMHg) were collected in 125 mL Teflon containers and acidified to 0.5 % v/v CH3COOH 99% (Trace metal 198 grade) ³³. The Teflon containers were closed tightly and stored in double PE Zip-lock bags in a 199 portable cooler $(5 - 10 \degree C)$, protected from light.

 Other methods used for the measurement of ancillary parameters (conductivity, pH, temperature, dissolved oxygen saturation, chlorophyll-a, total organic carbon TOC, dissolved

202 inorganic carbon DIC, total alkalinity TA, silicate, sulphate SO_4^2 , chloride CI, nitrate NO₃) are 203 described elsewhere (*Duval, PhD Thesis*⁴⁰).

2.3. Non-gaseous Hg species (iHg(II), MMHg) analysis

 The quantification of non-gaseous Hg species (i.e. iHg(II) and MMHg) was carried out by double species-specific isotopic dilution method (D-SS-IDA) and analysis by capillary GC-ICP-207 MS 33,44. Note that this analytical methodology allows to quantify the iHg(II) and the methylated species (MeHg, i.e., the sum of MMHg and DMHg,). Nevertheless, the purge sample experiment (**see Appendix 1 and Figure A2**) highlights the fact that the measured MeHg are mainly in the 210 form of MMHg. LoD ranged from 9 to 34 pg L^{-1} for iHg(II) and from 3 to 4 pg L^{-1} for MMHg, in good 211 agreement with recent publications 33,44.

 Repeatability of the analytical measurement (n=3) was 2% for both iHg(II) and MMHg. The relative standard deviations (RSD) associated to the samples collected at three different locations of each lake (intra-lake variability, two first sampling campaigns) are 15 and 20% respectively for unfiltered and filtered MMHg, and 17 and 20% respectively for unfiltered and filtered iHg(II) and will be used to describe uncertainties in this paper. If and Figure A2) highlights the fact that the measured MeH_g

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2.4. Gaseous Hg species (DGM) analysis

 The samples for gaseous Hg species (DGM) were processed close to the sampling site within 1 to 4 hours after sampling by stripping out and trapping the volatile Hg species from the 220 water sample into a gold-coated sand trap ^{45,46}. For that purpose, the water sample was gently 221 transferred into a Teflon purge vessel and subsequently purged under an argon flow (500 mL min- $, 1, 20$ min). The moisture in the gas stream was eliminated by a cold trap (- 20 °C) before gaseous Hg collection in the gold-based trap. A second gold trap was always positioned after the sampling trap to prevent atmospheric contamination. Gold traps were sealed with polypropylene Teflon- lined caps and stored in double PE Zip-lock bags, in the dark (< 4°C) until analysis. All this procedure was carried out in an on-field laboratory installed in a mountain hut close to the sampling site.

 The analysis of gaseous Hg species was done by double amalgamation on gold-AFS 229 (DA/Au-AFS) ⁴⁶. Thermodesorption efficiency was controlled by carrying out two consecutive

 analyses of the gold-coated sand trap. Gaseous Hg quantification was done by external calibration of the DA/Au-AFS device using a controlled Hg(0) source. Finally, the analysis of purge blanks allowed us estimating the efficiency of the sample treatment procedure. Considering the 233 five sampling campaigns, the purge efficiency was assessed to reach (95 ± 3) % (n=12). Note that this analytical methodology allows the measurement of the sum of Hg(0) and volatile dimethylmercury (DMHg). Nevertheless, regarding the very small levels of methylated species (MeHg), i.e. the sum of MMHg and DMHg, in comparison with DGM, we can assume that DGM is mainly elemental mercury Hg(0).

238 LoD were quiet constant over the sampling campaigns, ranging from 0.1 to 0.5 pg L^{-1} , 239 which is consistent with the LoD of 0.4 pg L⁻¹ reported by *Bouchet et al.* ⁴⁶.

2.5. Hg species transformation assessments

 In-situ water incubation experiments over 7h (usually from 9 a.m. to 4 p.m. UTC+2) using 242 isotopically enriched Hg species (iHg(II), 201 MMHg) were conducted in the June 2018, October 2018 and June 2019 campaigns at lakes Gentau, Arratille, and Sabocos to investigate Hg species transformation mechanisms in the water column (methylation, demethylation, reduction) 245 according to the procedure published elsewhere $47-49$. ere quiet constant over the sampling campaigns, ranging from
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2.5. Hg species transformation assessments
water incubation experiments over 7h (usuall

 Unfiltered water samples were collected to perform the incubation experiments. Different conditions of incubation were tested depending on sampling depth (subsurface, middle depth, and bottom) or light exposure (Diurnal vs Dark) (**Table A5**).

 After collecting a water sample using Go-Flo sampler, a first set of samples was prepared. Three 125 mL Teflon container (triplicate) per condition were directly filled with the water sample up to the top. Isotopically enriched spikes were added to each of the replicate to obtain 252 concentrations in ¹⁹⁹iHg(II) and ²⁰¹MMHg of respectively 2 and 0.2 ng L⁻¹. It corresponds to about 10 times the natural concentrations observed in the studied ecosystems: high enough to overlap the natural concentrations and low enough to avoid any biotic stress. Using a simple mooring line at the sampling point, incubation bottles, either protected from the light or not, were placed at the corresponding depth. After 7 hours, the Teflon containers were collected, and the incubation processes were stopped by adding high-purity HCl (1 % v/v). Teflon containers were closed tightly 258 and stored in double PE Zip-lock bags in a portable cooler $(5{\text -}10 \text{ °C})$, protected from light, and

259 further transported and stored in the laboratory (5-10 $^{\circ}$ C). This first set of samples allows the determination of methylation, demethylation and MMHg loss rates.

 A second set of samples was prepared to determine Hg reduction rates. After collection of water using the Go-Flo sampler, two 250 mL Teflon containers (duplicate) per condition were 263 filled overflow without headspace and somehow following the protocol for the Winkler method and 264 spiked with 2 and 0.2 ng L^{-1} isotopically enriched ¹⁹⁹iHg(II) and ²⁰¹MMHg, respectively, like in the previous case. They were incubated together with the samples of the first set. The difference in this case is that, at the end of the 7 hours, the elemental Hg (Hg(0)) was immediately recovered 267 in gold-coated sand traps by purging the water samples.

 Quantification was carried out by isotope pattern deconvolution (IPD) isotope dilution analysis (IDA) and analysis by capillary GC-ICP-MS for the water samples following the same 270 protocol used for the determination of organomercury species . The particularity is that the 271 samples were spiked with other enriched isotope solution $(^{198}iHg(II))$ and $^{202}MMHg$). sand traps by purging the water samples.

Cication was carried out by isotope pattern deconvolution (IPI

and analysis by capillary GC-ICP-MS for the water samples for

for the determination of organomercury species ³².

 The gaseous species trapped in the gold-coated sand traps were quantified by IPD and analysed by thermal desorption cryogenic trapping (CT) followed by GC-ICP-MS. Recoveries and mass balance of the added Hg spikes during the experiment were systematically established in 275 order to provide accurate transformation yield as previously described $18,32,48$.

 The Hg species incubation experiment allowed us to calculate methylation (M) (iHg(II) into MMHg), demethylation (D) (MMHg into iHg(II)), MMHg loss (L) (MMHg into iHg(II) and Hg(0)) 278 and reduction (R) (iHg(II) into Hg(0)) potentials (in % day⁻¹) (**Figure A3**) according to *Rodriguez*-279 Gonzales et al. ⁴⁸ with the following formula:

280
$$
Method (M) = \frac{0^{199} M M H g_t - 0^{199} M M H g_{t0}}{0^{199} H g (II)_{t0}} \times \frac{1440}{t} \times 100
$$

281 *Demethylation (D)* =
$$
\frac{0^{201} iHg(II)_t - 0^{201} iHg(II)_{t0}}{0^{201} M M H g_{t0}} \times \frac{1440}{t} \times 100
$$

282
$$
MMHg Loss (L) = \frac{0^{201} M M H g_{t0} - 0^{201} M M H g_t}{0^{201} M M H g_{t0}} \times \frac{1440}{t} \times 100
$$

283
$$
Reduction (R) = \frac{0^{199}Hg(0)_t - 0^{199}Hg(0)_{t0}}{0^{199}Hg(II)_{t0}} \times \frac{1440}{t} \times 100
$$

284 where t is the time of incubation (min), $\frac{2 \times x}{x}$ Hg(II)_{t0}, $\frac{2 \times x}{x}$ MMHg_{t0} and $\frac{2 \times x}{x}$ Hg(0)_{t0} are the 285 concentrations measured at the beginning of the incubation experiment, XXX iHg(II)t, XXX MMHgt and 286 XXXHg(0)_t are the concentrations measured at the end of the incubation experiment.

287 To predict the variations of MMHg concentrations in the lake, the potential net Hg 288 methylation (ng L⁻¹ day⁻¹) was calculated using the methylation and demethylation rates (% day 289 $\frac{1}{2}$ according to the following formula:

290 *Net Methylation* =
$$
\frac{M}{100} \times [iHg(II)]_{ambient} - \frac{D}{100} \times [MMHg]_{ambient}
$$

291 where M is the methylation potential (% day⁻¹), [iHg(II)]_{ambient} is the ambient iHg(II) concentration 292 (ng L⁻¹), D is the demethylation potential (% day⁻¹), [MMHg]_{ambient} is the ambient MMHg 293 concentration (ng L^{-1}).

294 It is worth noting that we chose to use demethylation rates instead of MMHg loss rates to 295 avoid an underestimation of the net methylation since we have only determined MMHg production 296 from iHg(II). Thus, in that case, we specifically consider the most significant reversible 297 methylation/demethylation pathways between iHg(II) and MMHg forms. methylation potential (% day⁻¹), [iHg(II)]_{ambient} is the ambient iH
the demethylation potential (% day⁻¹), [MMHg]_{ambient} is the
ng L⁻¹).
th noting that we chose to use demethylation rates instead of M
estimation

298 **2.6. Hg gaseous fluxes at the air-water interface**

299 According to *Sharif et al.* ³², gaseous Hg fluxes at the air-water interface (volatilization flux 300 densities [FD]) (ng m⁻² h⁻¹) were calculated using the following equation:

301
Volalitzation FD =
$$
k_W \times ([DGM] - \frac{[TGM]}{H})
$$

302 where DGM is the concentration of DGM (i.e. $Hg(0)_{aq}$) measured in the subsurface water (ng m⁻ 303 ³), TGM is the concentration of TGM (i.e. $Hg(0)_{g}$) measured in the atmosphere. As a reference 304 value for atmospheric concentration of TGM we used the measurement performed at the Pic du 305 Midi (2860 m asl.) between February 2012 and January 2013 50 (1.8 ng m⁻³). H is the 306 dimensionless Henry's Law constant corrected for water temperature and salinity 51 and k_W is the Hg(0) gas transfer velocity (m h⁻¹) at the air-water interface. k_W was calculated, according to *Sharif*

308 *et al.* 32 and using a specific model developed for $CO₂$ exchanges in sheltered lakes 52 , as a 309 function of wind speed.

310 The wind speed has been measured during daytime in Lake Gentau ($w = (3.2 \pm 1.4)$ m s⁻¹ in June 311 2018; w = (1.4 ± 0.8) m s⁻¹ in October 2018; w = (2.6 ± 2.0) m s⁻¹ in June 2019), Lake Sabocos 312 (w = (1.5 \pm 1.2) m s⁻¹ in June 2018; w = (2.8 \pm 2.1) m s⁻¹ in October 2018; w = (2.0 \pm 2.0) m s⁻¹ in 313 June 2019), Lake Arratille (w = (1.5 ± 0.7) m s⁻¹ in June 2018) and Lake Azul (w = (0.8 ± 0.8) m 314 s⁻¹ in June 2018). Thus, to estimate the FD in all the sampled lakes, two wind speed ranges will 315 be considered: smooth breeze at 1 m s⁻¹ to stronger wind at 3 m s⁻¹, which corresponds well to sheltered lakes conditions as proposed by *Cole and Caraco*⁵².

317 **3. Results and discussion**

318 **3.1. Total Hg (HgTOT) in all lake waters**

319 The concentrations of iHg(II) and MMHg, both in filtered (F) and unfiltered (UF) samples, 320 and DGM are presented in **Table A3** and their variability is illustrated in **Figure 2.** Total Hg (Hg_{TOT}) 321 is defined here as the sum of concentrations of iHg(II), MMHg and DGM in unfiltered samples 322 (DGM only measured in unfiltered samples). smooth preeze at 1 m s · to stronger wind at 3 m s ·, which conditions as proposed by *Cole and Caraco*⁵².
 SSUITS and discussion
 3.1. Total Hg (Hgror) in all lake waters

incentrations of iHg(II) and MMHg, both in

323 In the following discussions, to establish a proper global assessment of the total Hg 324 (Hg_{TOT}) in the subsurface water of the high-altitude lakes, seven outliers are discarded from the 325 whole database based on to the fact that the concentrations observed were too high compared 326 to the normal distribution of our data and that no valid explanation could be given for such higher 327 values (**see Appendix 2 for more details**). Hg_{TOT} is quite homogeneous along the five sampling 328 campaigns, ranging from 0.17 to 1.37 ng L^{-1} with a median value of 0.48 ng L^{-1} and an average 329 of (0.54 \pm 0.26) ng L⁻¹ (n=66). The homogeneity in the Hg_{TOT} concentrations in the Pyrenean lakes 330 suggests that local inputs through geogenic paths (erosion, lixiviation) are not significant and does 331 not play a key role in the biogeochemistry of Hg as it was found for other trace elements $38,43,53$. 332 This also confirms the fact that Hg inputs in alpine environments like the Pyrenees are mainly due 333 to wet and dry atmospheric deposition. Only two previous studies showed that unfiltered Hg $_{\text{TOT}}$ 334 varies from 2 to 14 ng L^{-1} in the precipitation from the Pyrenees (880 m asl) in 2014 54 and from 335 2 to 170 ng L⁻¹ in the surface snow from the Alps (2448 m asl) in 2008 – 2009 ⁵⁵. In our study, Hg

336 species in meltwater and ice were also measured in Lake Cambalès in June 2018. The Hg $_{\text{TOT}}$ concentrations were much higher in ice (Hg_{TOT} = 19.41 ng L⁻¹) than in meltwater (Hg_{TOT} = 1.05 ng 338 L⁻¹) and subsurface water (Hg_{TOT} = 0.45 ng L⁻¹). These reported concentrations in atmospheric depositions are much higher than the levels observed in the subsurface water from the high- altitude lakes from this study, suggesting a simple dilution effect or more complex exchanges in lake waters as discussed thereafter.

3.2. Mercury compounds distribution in water samples

3.2.1. Variability in subsurface waters of all lakes

Inorganic mercury (iHg(II)) (Figure 2)

345 Filtered and unfiltered $Hg(II)$ levels ($Hg(II)_{F}$ and $Hg(II)_{UF}$) in subsurface water from the 19 sampled lakes over the five sampling campaigns were slightly lower than other worldwide 347 lakes ^{15,22,24,25,28–31} and freshwaters from the same area ^{32,33} (Table 1). iHg(II)_F ranges from 0.08 348 to 1.10 ng L⁻¹ with a median value of 0.23 ng L⁻¹ and iHg(II)_{UF} ranges from 0.11 to 1.19 ng L⁻¹ with 349 a median value of 0.38 ng L^{-1} , similar to concentration levels found in marine waters $56,57$ and typical from a pristine environment. 3.2.1. Variability in subsurface waters of all laterly (iHg(II)) (Figure 2)

and unfiltered iHg(II) levels (iHg(II)_F and iHg(II)_{0F}) in subsurfaces over the five sampling campaigns were slightly lower tha

8-31 and fre

351 The filtered fraction of Hg (iHg(II)_F) represents about 68 ± 16 % (range from 19 to 100 %) 352 of the unfiltered fraction ($iHg(II)_{UF}$), suggesting that most of the mercury in the Pyrenean lakes is found in the dissolved fraction. However, subsurface water samples from Azul in Spring 2017 (39 %), Peyregnets (47 %) and Gentau (19 % to 44 %) in Spring 2018 and Cambalès (41%), Peyregnets (49 %) and Gentau (41 %) in Autumn 2018 are exhibiting higher particulate Hg fraction.

 On **Figure 2**, the homogeneity observed in the iHg(II) concentrations, either filtered or unfiltered, along the five different seasons demonstrates a rather steady state in the Hg mass balance between the investigated periods. Another study reports seasonal variations of iHg(II) in 360 the high altitude subtropical Lake Uru Uru (3686 m asl) . In that case, concentrations were significantly higher during the dry season over the wet season because of the enhanced evaporation occurring at the end of the dry season. In pristine Pyrenean lakes, the concentration

363 of iHg(II) in Autumn (2017 and 2018) is not significantly higher than in early Spring (2017, 2018

364 and 2019) (Kruskal-Wallis test, p-value = 0.79 and 0.54 respectively for iHg(II)_{UF} and iHg(II)_F).

365 **Monomethylmercury (MMHg) (Figure 2)**

366 Filtered and unfiltered MMHg levels (MMHg F and MMHg UF) in subsurface water from the 19 sampled lakes over the five sampling campaigns vary respectively from <0.003 to 0.035 ng L⁻ 1 and from <0.003 to 0.062 ng L⁻¹ (Table 1). The median values, 0.008 for MMHg_F and 0.011 for 369 MMHg_{UF}, are typical from pristine aquatic environments, in the range of what has been found in 370 Lake Superior (MMHg_F = 0.005 ± 0.001 ng L⁻¹ in April 2000; MMHg_F = 0.008 ± 0.002 ng L⁻¹ in 371 August 2000)¹⁵ and in some high-altitude lakes in the Alps (0.002 $-$ 0.005 ng L⁻¹)²⁵. MMHg 372 represents (4 ± 3) % of the filtered non-gaseous Hg and (4 ± 2) % of the unfiltered non-gaseous Hg. 373 From Hg speciation analyses in surface snow from the French Alps (2448 m asl) 374 Marusczak et al. ⁵⁵ concluded that biotic production of MMHg in the snowpack is unlikely, 375 considering the constant ratio MMHg/THg measured throughout the season in their study. The 376 main contributor to Hg in the high-altitude lakes being atmospheric deposition, the higher 377 proportion of MMHg observed in surface waters from high-altitude lakes in the Alps (4 \pm 3 % for 378 MMHg_F and 3 \pm 2 % for MMHg_{UF}) in comparison to surface snow (MMHg vary from non-379 determined to 1.21 % of the total Hg) might be due to in-situ aquatic methylation. ¹⁵ and in some high-altitude lakes in the Alps $(0.002 - 0.005)$
⁹⁵ of the filtered non-gaseous Hg and (4 ± 2) % of the unfiltered
4g speciation analyses in surface snow from the French A
al. ⁵⁵ concluded that bi

380 Regarding the MMHg levels in surface waters of the investigated Pyrenean lakes, the 381 Autumn 2018 sampling campaign stands out from the other ones (**Figure 2**). Median values for 382 MMHguF in Spring 2017, Autumn 2017, Spring 2018, and Spring 2019 were respectively 0.010, 383 0.006, 0.011 and 0.013 ng L⁻¹ while the median value in Autumn 2018 was 0.025 ng L⁻¹. A 384 comparison of MMHg_{UF} levels in Spring 2018 and Autumn 2018 reveals substantial increases in 385 lakes Arratille (+248 %), Gentau (+360 %), Roumassot (+235 %), Bachimaña (+211 %), Coanga 386 (+231 %), Panticosa (+583 %) and Sabocos (+216 %) and more moderate increases in most of 387 the other studied lakes. Spring - summer algal bloom are controlling the whole biological turnover 388 in those high-altitude and oligotrophic lakes 58 , and intense bloom events might be responsible 389 for such increase by promoting methylation processes in the sediments ^{29,59}. Arratille, Gentau, 390 Roumassot, Coanga and Sabocos showed the lowest nitrate (NO₃) concentrations (from below 391 LOD to 0.171 mg L^{-1} in Autumn 2018, which can be an indicator of its removal by higher biological 392 productivity ^{19,20,58,60}. In-situ methylation rates in lakes have already been linked to the trophic

393 status, with increasing eutrophication leading to increasing organic matter loading and subsequent Hg methylation ²².

395 **Dissolved Gaseous Mercury (DGM) (Figure 2)**

396 **Dissolved Gaseous Mercury (DGM) varied strongly, from 0.02 to 0.68 ng L⁻¹ (outliers** 397 excluded), and, overall, was higher than the measured DGM in other pristine areas (**Table 1**). 398 Indeed, DGM in Bolivian lakes vary from 0.001 to 0.017 ng L⁻¹ in Lake Titicaca 22 and from 0.003 399 to 0.125 ng L^{-1} in Lake Uru Uru 24 . In the Lake Superior, in Canada, DGM measured in august 400 2000 was 0.020 ± 0.003 ng L^{-1 15}. Even in the Adour Estuary, downstream to the Pyrenees, DGM 401 content was lower than in the Pyrenean Lakes with values ranging from 0.024 to 0.056 ng L^{-1} 32. 402 Finally, the DGM measured in the present study, with a median value of 0.11 ng L^{-1} , was in the 403 range of the open ocean waters 57 . DGM accounted for 25 ± 11 % of the Hg_{TOT} and sometimes 404 unexpectedly represents up to 55% of the total Hg.

405 The homogeneous iHg(II) concentrations contrasts with the high DGM levels measured 406 in the Pyreneans lakes of our study. Photoreduction is probably important in high-altitude lakes 407 from the Pyrenees. This photoreduction might be triggered by important solar radiation, and the 408 general oligotrophic state of the Pyrenean lakes, in comparison with Bolivian ^{22,24} or Canadian 409 lakes ²⁸, might play a key role in this process. Indeed, organic matter is believed to have an impact 410 on Hg photoreduction, and it is worth noting that DOC concentrations exhibit low to moderate 411 values in the Pyrenean lakes (from 0.62 to 4.63 mg L^{-1}). Higher Hg photoreduction to elemental 412 Hg(0) therefore likely takes place in open water in which low to moderate DOC content has been 413 observed 57 . wer than in the Pyrenean Lakes with values ranging from 0.024
M measured in the present study, with a median value of 0.11
pen ocean waters ⁵⁷. DGM accounted for 25 ± 11 % of the Hg_T
epresents up to 55% of the total

414 Table 1: Comparison of filtered and unfiltered inorganic mercury (iHg(II)), monomethylmercury (MMHg) and Dissolved Gaseous Mercury (DGM) concentrations in the 415 subsurface water samples of the 19 studied lakes with literature data for worldwide pristine areas (oceans, boreal lakes, high altitude lakes) and local areas (freshwaters and 416 (40 total Ha (Ha_{ror} = estuary). %MMHg is calculated as the ratio between MMHg and non-gaseous Hg (MMHg + iHg(II)). %DGM is calculated as the ratio between DGM and total Hg (Hg_{TOT} = 417 MMHg + iHg(II) + DGM). *THg, **Reactive Hg and ***Surface and Depth samples.

Figure 2: Boxplot representations of unfiltered and filtered iHg(II), MMHg and percentage of MMHg (calculated as the ratio between MMHg and non-gaseous Hg (MMHg + $iHg(II)$), and DGM and percentage of DGM (calculated as the ratio between DGM and total Hg (Hg_{TOT} = MMHg + $iHg(II)$ + DGM)) in subsurface water samples of the 19 studied lakes. Bars indicate 10th and 90th percentile, boxes indicate 25th and 75th, marks within each box are medians, and red crosses are mean. Outliers' values especially measured for DGM are included in the graphic presentations (**see Appendix 2**).

 Figure 3 (Lake Gentau) and **Figure 4** (Lake Sabocos) display some physico-chemical parameters, including temperature, dissolved oxygen saturation (not corrected from altitude), 425 chlorophyll-a, TOC, Silicate, SO_4^2 , Cl and NO₃ that might be useful to understand the distribution of mercury species (iHg(II), MMHg and DGM) in the water column of these two different lakes intensively monitored during June 2018, October 2018 and June 2019. The results are also summarised in **Table A4**. Non stratified and shallow Lakes Arratille and Azul depth profiles (**respectively Figure A5 and Figure A6)** are described in **Appendix 3**.

Lake Gentau (Figure 3)

 The water column of Lake Gentau was stratified in the three sampling campaigns (June 2018, October 2018, and June 2019). Indeed, this lake is divided in three various sections: epilimnion, metalimnion, and hypolimnion, all along the year except during the short overturn 435 spring and autumn periods ⁶¹. In addition to the oxygen chemocline (located around 7m depth in June 2018, 10m depth in October 2018 and 8m in June 2019), Gentau also presented a 437 thermocline in the June 2018 sampling campaign (at 10m depth), which was here not matching the depth of the chemocline. Table A4. Non stratified and shallow Lakes Arratille and A
Figure A5 and Figure A6) are described in Appendix 3.
Figure 3)
ter column of Lake Gentau was stratified in the three sampling
2018, and June 2019). Indeed, this l

 The highest iHg(II) concentrations, either unfiltered or filtered, were detected at the 440 deepest sampling point of the lake in June (0.75 and 0.66 ng L^{-1} in unfiltered and filtered samples, 441 respectively) and October 2018 (0.39 and 0.36 ng $L⁻¹$ in unfiltered and filtered samples, respectively). In June 2019, the surface water samples were characterized by the highest iHg(II) 443 concentrations (0.27 and 0.18 ng L⁻¹ in unfiltered and filtered samples, respectively). *Bravo et al.* 29 have shown that the highest total mercury concentrations in river stream systems were associated with important terrestrial DOM (Dissolved Organic Matter) and low nutrient content. In our study, TOC was low and relatively constant all along the water column during the three 447 sampling campaigns (1.1 \pm 0.3 mg L⁻¹ in June 2018, 1.5 \pm 0.2 mg L⁻¹ in October 2018, and 1.2 \pm 448 0.1 mg L^{-1} in June 2019). In addition, nitrate (NO₃), indicator of the biological productivity, 449 decreased with depth in June 2018 (0.18 to 0.08 mg L^{-1}) while it increased with depth in October

450 2018 (<0.06 to 0.14 mg L^{-1}), exhibiting the seasonal changes in primary productivity and water column remineralisation of nitrogen. Also, the important increase observed for iHg(II) in the deepest point of the lake in June and October 2018 is probably due to some inputs of iHg(II) at the water-sediment interface.

454 MMHg concentrations were significantly higher at the deepest point of the lake in 455 comparison with the other four sampled points. Indeed, while unfiltered and filtered MMHg in the 456 water column in June 2018 varied from 0.008 to 0.057 ng $L⁻¹$ and 0.006 to 0.023 ng $L⁻¹$ 457 respectively, MMHg unfiltered and filtered levels were respectively 0.426 and 0.318 ng L^{-1} at 18 458 m depth. In October 2019, the same trend was observed for MMHg: unfiltered and filtered MMHg 459 varied from 0.027 to 0.055 ng L^{-1} and 0.011 to 0.026 ng L^{-1} , respectively, while the deepest 460 sampled point (17m depth) exhibited MMHg levels as high as 0.388 and 0.341 ng L^{-1} for the 461 unfiltered and filtered samples. In June 2019, unfiltered and filtered MMHg concentrations varied 462 from 0.007 to 0.059 ng L⁻¹ and <0.004 to 0.019 ng L⁻¹ respectively, while it reached 0.236 and 463 0.157 ng L⁻¹ at the deepest sampling point (17.5m depth). Moreover, MMHg represents a higher 464 fraction of the non-gaseous Hg species (MMHg and iHg(II)) at the deepest sampled point, from 465 36 to 56% for unfiltered MMHg and from 32 to 60% for filtered MMHg. Anoxic conditions observed 466 in the deepest point of Lake Gentau can host anaerobic microbial activities responsible of in-situ 467 Hg biotic methylation in both water and surface sediments 49 . Such anoxic conditions may not be 468 permanent in those lakes but last for several months and affect MMHg burden in lake waters. 469 Oxygen complete consumption in bottom waters is related to both dark respiration during the ice 470 cover in winter/spring time and also to the "Bowl-shape" - basins (Gentau, Sabocos) in which the 471 organic deposits from the whole lake will rapidly consume bottom waters oxygen between two 472 vertical mixing events. tober 2019, the same trend was observed for MMHg: unfiltered 3
227 to 0.055 ng L⁻¹ and 0.011 to 0.026 ng L⁻¹, respectively, 1
(17m depth) exhibited MMHg levels as high as 0.388 and 0.
iltered samples. In June 2019, un

473 Finally, vertical distribution of DGM in Gentau was quite simple with a constant and 474 progressive decrease from surface waters (10.79 ng L^{-1} in June 2018; 0.06 ng L^{-1} in October 475 2018; 0.10 ng L⁻¹ in June 2019) to the deepest sampled point (0.14 ng L⁻¹ in June 2018 at 18m 476 depth; 0.03 ng L⁻¹ in October 2018 at 17m depth; 0.01 ng L⁻¹ in June 2019 at 17.5m depth). DGM 477 concentrations were also lower in October 2018 than in June 2018 and 2019 all along the water 478 column. This supports the previous assumption regarding the importance of photoreduction 479 processes in those high-altitude lakes, induced by significant solar radiation on surface waters.

506 chemical parameters including mercury speciation obtained in (a) June 2018, (b) October 2018 and (c) June 2019 in Lake Gentau. Red dot points correspond to the LoD. ^Ø correspond to outlier data for DGM in surface water (Grubbs's test; **see Table A3**) **Figure 3:** Depth profiles of temperature, percentage of dissolved oxygen saturation, chlorophyll-a (RFU) and some other **A3**).

Lake Sabocos (Figure 4)

 Lake Sabocos is less stratified than Lake Gentau, but oxygen and temperature varied strongly along the water column with depletion of both physico-chemical parameters with depth, leading to almost anoxic conditions in the bottom part of this lake. Indeed, dissolved oxygen varied from 96 to 1% in June 2018, from 79 to 2% in October 2018 and from 84 to 5% in June 2019, and the chemocline, where dissolved oxygen starts decreasing, was located at 4, 10 and 5m in June 2018, October 2018, and June 2019, respectively. The thermocline was located at the same 517 depths than chemocline and temperature varied from 17 to 5°C in June 2018, 11 to 6°C in October 2018 and 12 to 5°C in June 2019. This lake has also one specificity: it is the only one classified as an alkaline lake with pH values ranging from 7.47 to 7.71 in the surface water (June and October 2018).

521 iHg(II), either unfiltered or filtered, strongly varied all along the water column in the three 522 sampling campaigns (respectively from 0.39 to 1.29 ng L^{-1} and from 0.29 to 0.53 ng L^{-1} in June 523 2018; from 0.53 to 0.90 ng L⁻¹ and from 0.17 to 0.68 ng L⁻¹ in October 2018; from 0.08 to 0.34 ng 524 L⁻¹ and from 0.02 to 0.21 ng L⁻¹ in June 2019). There is no evidence for any specific trend in the iHg(II) distribution in the Lake Sabocos but the strong variations suggest that important Hg species transformations may occur. 5°C in June 2019. This lake has also one specificity: it is the c
lake with pH values ranging from 7.47 to 7.71 in the surface
either unfiltered or filtered, strongly varied all along the water c
aigns (respectively from

 MMHg depth profiles in Lake Sabocos indicate potential in-situ methylation with the highest MMHg levels measured in the deepest part of the lake where the oxygen level was the lowest. MMHg concentrations in unfiltered et filtered waters were respectively 0.060 and 0.025 μ ng L⁻¹ (10 and 7% of the non-gaseous Hg), and 0.052 and 0.021 ng L⁻¹ (25 and 42% of the non- gaseous Hg) in June 2018 and 2019. Increase of MMHg levels was not observed in October 2018. During this sampling campaign, no maximum of chlorophyll was detected in the water column, probably indicating a lower biological production.

 Regarding the DGM concentrations, the high levels measured in surface waters in both 535 June 2018 and 2019 (4.65 and 1.34 ng $L⁻¹$ respectively), in comparison with deepest samples $(0.56$ and 0.06 ng L⁻¹ respectively), is consistent with intense photoreduction enhanced during spring conditions.

Figure 4: Depth profiles of temperature, percentage of dissolved oxygen saturation, chlorophyll-a (RFU) and some other
hard a secondary including magazine assisting aktivited in (a) line 2010, (b) October 2010 and (a) li Sabocos. Red dot points correspond to the LoD. ^Ø correspond to outlier data for DGM in surface water (Grubbs's test; see Table **and a**) chemical parameters including mercury speciation obtained in (a) June 2018, (b) October 2018 and (c) June 2019 in Lake **A3**).

568 **3.3. Mercury species transformations and volatilization in** 569 **the water column of selected alpine lakes**

 Hg species incubation experiments were conducted in lakes Gentau, Sabocos and Arratille as described in **Section 2.5**, to quantify the importance of the Hg species transformations occurring in these high-altitude lakes. The obtained results are summarized in **Table 2,** together with transformation rates obtained in previous studies.

574 *3.3.1. Methylation and Demethylation pathways*

575 **Inorganic mercury Methylation**

576 Methylation potentials under dark and diurnal conditions obtained in unfiltered water 577 samples ranged between <0.03 and 6.97% day⁻¹ in Lake Gentau, between <0.03 and 0.95 % 578 day⁻¹ in Lake Sabocos, and between <0.03 and 0.44 % day⁻¹ in Lake Arratille (**Table 2)**. These 579 methylation potentials are in agreement with data obtained for water samples in Canadian lakes²⁶. 580 high-altitude Bolivian lakes 24 and for marine and coastal waters $32,47$ using similar experimental 581 methods. The highest methylation potentials have been measured under dark conditions in the 582 bottom anoxic zone for Lake Gentau which exhibit 4.34 ± 0.40 % day⁻¹ in June 2018, 6.97 \pm 0.44 583 % day⁻¹ in October 2018 and 1.63 \pm 0.11 % day⁻¹ in June 2019. For Lake Sabocos methylation is 584 0.81 \pm 0.33 % day⁻¹ in June 2018, 0.95 \pm 0.38 % day⁻¹ in October 2018 and 0.63 \pm 0.28 % day⁻¹ 585 in June 2019. No significant difference was observed for methylation potentials for incubation 586 performed under dark or diurnal conditions, confirming that light induced methylation is not 587 significant in those high-altitude pristine lakes. These results also support depth profile 588 measurements and demonstrate that reducing conditions, especially in stratified anoxic waters, 589 promote Hg methylation due to methylating anaerobic bacteria. 3.3.7. Wetrlylation and Demetriyiation patriways

cury Methylation

ation potentials under dark and diurnal conditions obtained if

d between <0.03 and 6.97 % day⁻¹ in Lake Gentau, between

stabocos, and between <0.03 a

590 Lower but measurable methylation extent were determined in the oxic subsurface waters 591 of Lake Gentau with 0.42 \pm 0.12 % day⁻¹ under diurnal conditions and 0.40 \pm 0.17 % day⁻¹ under 592 dark conditions in June 2018, and 0.21 ± 0.17 % day⁻¹ under diurnal conditions in June 2019. In 593 Lake Sabocos, surface water methylation was 0.09 ± 0.05 % day⁻¹ under diurnal conditions and 594 0.26 \pm 0.02 % day⁻¹ under dark conditions in June 2018, and 0.34 \pm 0.25 % day⁻¹ under diurnal 595 conditions and 0.17 \pm 0.07 % day⁻¹ under dark conditions in October 2018. Lake Arratille exhibit

596 methylation potential of 0.44 \pm 0.04 % day⁻¹ under diurnal conditions in June 2018. The formation of MMHg in the oxic freshwater column is not fully understood yet. A recent work conducted in Lake Geneva highlights that particles sinking through oxygenated water column can be 599 responsible for the production of MMHg²⁷. In our study, although the methylation rates measured remain low, it is measurable and suggests that reducing conditions occurring in particles microenvironment within the oxic layer could also play a role.

MMHg Demethylation

 Significant demethylation potentials were measured in the subsurface waters of Lake 605 Gentau (23.8 \pm 4.4 % day⁻¹ in June 2018; 23.9 \pm 2.9 % day⁻¹ in October 2018 and 35.6 \pm 2.2 % 606 day⁻¹ in June 2019) and Lake Sabocos (35.2 \pm 9.7 % day⁻¹ in June 2018; 9.0 \pm 0.5 % day⁻¹ in 607 October 2018 and 12.4 \pm 4.2 % day⁻¹ in June 2019) only at daylight conditions. These important demethylation potentials have been observed in previous studies and were associated to both 609 abiotic and biotic processes $32,49$. In our study, demethylation under dark conditions was detected 610 only in Lake Sabocos within the oxicline in June 2019 (6.8 \pm 4.7 % day⁻¹), while in other lakes and 611 season dark demethylation was always below 4 % day⁻¹. Therefore, our results suggest that in high-altitude pristine lakes, direct light-induced photochemical demethylation in UV exposed surface waters is the most significant pathway that contributes to control the extent of MMHg in 614 the water column . ant demethylation potentials were measured in the subsurface 4.4 % day⁻¹ in June 2018; 23.9 \pm 2.9 % day⁻¹ in October 2018
(019) and Lake Sabocos (35.2 \pm 9.7 % day⁻¹ in June 2018; 9.1
and 12.4 \pm 4.2 % day⁻¹

 Lake Sabocos exhibited an interesting result regarding the middle depth (9m), only studied in June 2019. While in Lake Gentau significant demethylation was measured at the middle 617 depth of the lake (8m) in June 2018 (13.6 \pm 1.2 % day⁻¹) and June 2019 (13.6 \pm 0.4 % day⁻¹) only under diurnal conditions and in lower extent than for the subsurface, Lake Sabocos exhibited 619 significant demethylation at both diurnal and dark conditions in June 2019 (8.2 \pm 3.9 % day⁻¹ and 620 6.8 \pm 4.7 % day⁻¹ respectively). Thus, in both stratified lakes, demethylation occurred within the oxicline and cannot be related to UV photochemical reactions (i.e., no UV light). These results suggest that specific microbial activity are involved due to heterotrophic (Gentau, Sabocos) and/or phototrophic (Sabocos) bacteria specifically developing at the oxycline (middle depth).

Demethylation vs total MMHg loss

 As mentioned previously, demethylation, especially in subsurface waters, is important in Lake Gentau and Lake Sabocos. As a reminder, demethylation potentials calculated here correspond to the transformation of MMHg into iHg(II). In **Table 2**, in addition to the demethylation potentials, total loss of MMHg potential from the experimental water solution has also been determined. Total loss of MMHg correspond to the decrease of MMHg during the incubation and could be connected to both the transformation of MMHg into iHg(II), and the transformation of MMHg into Hg(0) or other precipitated insoluble forms. The linear regression between both MMHg degradation processes (excluding middle depth dark conditions of Lake Sabocos, June 2019) 635 exhibits a very good coefficient of determination ($R^2 = 0.97$; linear trend MMHg Loss = (1.7 \pm 0.1) × Demethylation) (**Figure A7**). The covariation of both calculated potentials (demethylation and loss of MMHg) suggests that similar pathways occur during the different incubation experiments. These results demonstrate that a significant fraction (always below 50%) of the degraded MMHg, not recovered as soluble iHg(II), is converted to Hg(0) (reductive (photo)demethylation) or to insoluble Hg forms as final products. ocesses (excluding middle depth dark conditions of Lake Sabgood coefficient of determination (R² = 0.97; linear trend MMHg
n) (**Figure A7**). The covariation of both calculated potentials (e
suggests that similar pathway

Hg Net methylation assessment

 In anoxic waters, the net methylation assessment (**Table 2**) allows to exhibit a potential 644 significant production of MMHg in the bottom part of Lake Gentau with 3.25 \pm 0.30 ng L⁻¹ day⁻¹ in 645 June 2018, 2.71 \pm 0.17 ng L⁻¹ day⁻¹ in October 2018, and 0.29 \pm 0.02 ng L⁻¹ day⁻¹ in June 2019. 646 While lower ones are determined in Lake Sabocos with 0.45 \pm 0.19 ng L⁻¹ day⁻¹ in June 2018, 647 0.75 \pm 0.30 ng L⁻¹ day⁻¹ in October 2018, and 0.09 \pm 0.04 ng L⁻¹ day⁻¹ in June 2019. On the one hand, at the subsurface, low potential production of MMHg was observed in Lake Arratille with 649 net methylation of 0.07 ± 0.01 ng L⁻¹ day⁻¹ (diurnal condition) and 0.07 ± 0.04 ng L⁻¹ day⁻¹ (dark condition) in June 2018. On the other hand, under diurnal conditions, the subsurface waters of Lake Gentau and Lake Sabocos behave as a sink of MMHg, as net methylation remains negative 652 and varies from -0.01 \pm 0.09 to -0.64 \pm 0.08 ng L⁻¹ day⁻¹.

656 *3.3.2. Reduction and volatilization from alpine lakes*

657 **Reduction and Volatilization Flux Density in selected alpine lakes (Arratille, Gentau and** 658 **Sabocos)**

659 High iHg(II) reduction potentials (**Table 2**) were measured under diurnal conditions in 660 subsurface waters of Lake Gentau (81.2 \pm 0.8 % day⁻¹ in June 2018; 20.8 \pm 0.1 % day⁻¹ in October 661 2018 and 16.9 \pm 0.1 % day⁻¹ in June 2019) and Lake Sabocos (13.2 % day⁻¹ in June 2019), and 662 were less significant in subsurface waters of Lake Arratille (17.3 \pm 2.9 % day⁻¹ in June 2018; 2.8 ± 0.2 % day⁻¹ in October 2018), but still remained among the highest in comparison to previous 664 studies $24,32,47$. These important reduction potentials are consistent with the high DGM 665 concentrations measured, especially in Lakes Gentau and Sabocos. Regarding the dark 666 conditions, reduction potentials at subsurface waters were detected at lower extents only in June 667 2018 (9.1 ± 2.6 % day⁻¹) and October 2018 (1.7 ± 0.1 % day⁻¹) for Lake Gentau, June 2019 (3.5 ± 0.1 % day⁻¹) for Lake Sabocos. Therefore, intense UV light occurring in surface waters of high-669 altitude lakes promotes Hg reduction. n October 2018), but still remained among the highest in comp
These important reduction potentials are consistent with
measured, especially in Lakes Gentau and Sabocos. Re
uction potentials at subsurface waters were detec

 Using the reduction potentials and the non-gaseous Hg measured at the subsurface waters, we can estimate the amount of Hg reduced per day within the first meter depth under diurnal conditions and compare it with the Volatilization Flux Density (FD) calculated as described in **Section 2.6**. All the fluxes calculated are gathered in **Table A3**.

674 **In Lake Gentau, the amount of Hg reduced per day was 473** \pm **5 ng m⁻² day⁻¹ in June** 675 2018, 54 \pm 1 ng m⁻² day⁻¹ in October 2018 and 47 \pm 1 ng m⁻² day⁻¹ in June 2019. Median values 676 of Volatilization FD in Lake Gentau were in the same order of magnitude with 48 ng m⁻² day⁻¹ for 677 1 m s⁻¹ wind speed (from 12 to 143 ng m⁻² day⁻¹) and 72 ng m⁻² day⁻¹ for 3 m s⁻¹ wind speed (from 678 18 to 216 ng $m⁻²$ day⁻¹) (n=10). For Lake Sabocos, the quantity of Hg reduced per day was in the 679 same range than Lake Gentau with 46 ± 1 ng m⁻² day⁻¹ in June 2019, and in the same order of 680 magnitude than the Volatilization FD median values, 121 ng m⁻² day⁻¹ for 1 m s⁻¹ wind speed (from 681 31 to 288 ng m⁻² day⁻¹) and 184 ng m⁻² day⁻¹ for 3 m s⁻¹ wind speed (from 46 to 436 ng m⁻² day⁻¹) 682 (n=4). In these two lakes, even if the Volatilization FD median values are not significantly different 683 from the extent of Hg reduced per day, the overall range remains important. The amount of Hg 684 reduced in Lake Arratille was lower than the other two lakes with 31 ± 5 ng m⁻² day⁻¹ in June 2018 685 and 14 \pm 1 ng m⁻² day⁻¹ in October 2018, which is also similar to the Volatilization FD calculated

686 (from 20 to 178 ng m⁻² day⁻¹ for 1 m s⁻¹ wind speed and from 30 to 270 ng m⁻² day⁻¹ for 3 m s⁻¹ wind speed (n=7)). Overall, the gaseous Hg evasion estimated from the first meter of the water column (subsurface) represents a similar extent than the reduction potential obtained from our incubation experiments. Our results strongly support that UV light induced in-situ reduction is a major pathway controlling significant volatilization of Hg from these alpine lakes.

Volatilization Flux Density (FD) in all the studied lakes and comparison with other fluxes

 Median values of Volatilization FD in all the lakes (except Arratille, Sabocos and Gentau 694 discussed below) were 60 ng m⁻² day⁻¹ for 1 m s⁻¹ wind speed (from 7 to 375 ng m⁻² day⁻¹) and 92 695 ng m⁻² day⁻¹ for 3 m s⁻¹ wind speed (from 11 to 568 ng m⁻² day⁻¹). Volatilization FD did not vary strongly depending on the sampling campaign with median values from 45 to 111 ng m⁻² day⁻¹ (1) 697 m s⁻¹) and from 68 to 168 ng m⁻² day⁻¹ (3 m s⁻¹).

 It is interesting to compare these estimated Volatilization FD with wet Hg deposition fluxes (direct input and snow-ice melt), and Hg accumulation rates in sediments. The latter is commonly used as a probe for past Hg contamination in alpine lakes and atmospheric burden.

 Hg concentration in wet deposition at the peatland of Pinet (Central Pyrenees, 880m asl) 702 was recently measured at 8.0 \pm 4.6 ng L^{-1 54}. Considering that annual precipitation is 1161 mm at 703 Pinet peat (average for the period 2010-2013)⁵⁴ and 2000 mm at lake Marboré (Central Pyrenees, 704 $-$ 2612 m asl) ³, the studied lakes in this work (from 1620 m asl to 2600 m asl) are subjected to 25 to 44 ng m⁻² day⁻¹ Hg inputs from wet deposition. This is well in accordance with model results of Hg deposition in the Pyrenees Region from European Monitoring and Evaluation Programme 707 (EMEP), ranging from 33 to 49 ng m^2 day⁻¹ (Website. http://www.emep.int/). The median 708 Volatilization FD calculated in this work are 60 to 92 ng $m⁻²$ day⁻¹ (1 and 3 m s⁻¹ wind speed, respectively) which is slightly above the maximum wet deposition. Nevertheless, considering the surface of the studied lakes the median Volatilization Flux Density can be estimated to be 2 to 3 mg day⁻¹, while considering the surface of the catchment the wet deposition inputs to Pyrenean 712 high-altitude lakes can be estimated to be 45 to 79 mg day⁻¹. We conclude that Hg evasion does not constitute a major Hg loss at the catchment scale, but it will drastically affect its water column biogeochemistry. w) were 60 ng m⁻² day⁻¹ for 1 m s⁻¹ wind speed (from 7 to 375 ng

3 m s⁻¹ wind speed (from 11 to 568 ng m⁻² day⁻¹). Volatilizatic

ding on the sampling campaign with median values from 45 to

68 to 168 ng m⁻²

715 Lake Marboré has been deeply studied $3,62,63$, and one recent publication is also dedicated 716 to the Hg in sediment cores from this lake (*Duval, PhD Thesis* ⁴⁰). In this study, the most recent 717 sample (2004 AD) exhibits a Hg accumulation rate of 40 μ g m⁻² year⁻¹, which corresponds to an 718 average daily Hg fluxes of 110 ng $m⁻²$ day⁻¹. This flux is in the same range as the median Volatilization Flux Density calculated with the gas exchange model.

 Overall, the evasion fluxes calculated for the alpine lakes are very significant, and due to 721 the high DGM levels measured in these lakes. It is worth noting that these estimations are made using an empirical water-air gas exchange model providing a rough estimate, and that the calculated Volatilization Flux Density corresponds to the free-ice period (negligible during ice cover). However, it is clear that the Hg deposited in bottom lake sediments will be drastically 725 affected by DGM evasion and direct input from snow melt and runoff. Institution Flux Density corresponds to the free-ice period (net or it is clear that the Hg deposited in bottom lake sediments
M evasion and direct input from snow melt and runoff.

Table 2: Methylation (M), Demethylation (D), MMHg Loss (L), Net Methylation (NM) and Reduction (R) potentials (mean ± SD, n=3 for M, D, L and NM, n=2 for R) in unfiltered waters 747 performed under varying light and dark conditions at different depths for Lakes Gentau, Sabocos and Arratille and for sampling campaigns June and October 2018 and June 2019, together with data from the literature. Detection limits are 0.03, 4, 4 and 1 % day-1 for Methylation, Demethylation, MMHg Loss and Reduction yields, respectively. n.d. is not determined.

Implication for Hg cycling in alpine lakes and climate driven perspectives

 Overall, this work focusing on the water column provides a first global picture of the Hg cycling in the high-altitude lake's ecosystems including various transformations rates measured and estimated volatilization fluxes for the first time. As an example, results obtained from Lake Gentau (**Figure 5**) allow us to estimate the fate of Hg in such stratified alpine lakes and to predict the influence of changes in the hydrological cycle, the water column stratification, and the biological productivity due to anthropogenic pressure and/or climate conditions. Indeed, important 765 methylation (290 to 3250 ng m⁻³ day⁻¹) occurs in the deepest and anoxic zone of this lake with 766 possible harmful impacts on the biota, while important (photo)demethylation (10 to 640 ng m⁻³ 767 day⁻¹) and (photo)reduction (47 to 473 ng m⁻³ day⁻¹) leading to Hg evasion (50 to 75 ng m⁻² day⁻¹) take place in the surface of the lake. Future in-depth studies must be conducted on these sentinel ecosystems to better constrain fluxes and transformations and obtain a more accurate mass balance of Hg species. Consequently, further experimental studies are clearly required to assess the effect of lake waters stratification, to quantify atmospheric deposition via direct input and snow/ice melt and to evaluate the eutrophication status and major biogeochemical changes. ul impacts on the biota, while important (photo)demethylation
to)reduction (47 to 473 ng m⁻³ day⁻¹) leading to Hg evasion (50
e surface of the lake. Future in-depth studies must be conducted
better constrain fluxes and

 Direct changes in the alpine ecosystem due to human activities (e.g. tourism, agropastoralism) or more global forcing due to climate change, have a potential effect on major drivers such as water temperature or light incidence. These effects will thus have drastic consequences on the Hg cycle at high altitude lakes that can be anticipated as follow: first, warmer conditions might increase lake water stratification and anoxic conditions in bottom waters that can promote Hg methylation by microorganisms. Such conditions can also increase both Hg (photo)reduction and evasion from surface waters to the atmosphere as we observed during this study. However large uncertainties remain in those evaluations considering that major changes are also expected with atmospheric deposition at local scale and that specific evolution of the lakes depends on their geographic and topographic situations. In this work, we addressed for the first time the importance of photoreduction on the fate of Hg through gaseous Hg evasion flux densities from alpine lakes. In addition, methylation and mainly photodemethylation extents were simultaneously measured in this work, allowing to calculate the net methylation rate to assess the net MMHg balance of these competitive pathways under different seasonal and annual conditions. These preliminary assessments provide here some first conclusions on the potential

effect of global or climate changing drivers on the Hg cycling and impact in those "sentinel"

ecosystems. However more complex features were not taken into account in this work and might

be involved especially during winter ice cover period, very warm summer conditions or water

column mixing events.

Figure 5: Hg transformations (ng m⁻³ day⁻¹) and fluxes (ng m⁻² day⁻¹) in Lake Gentau. Red crosses are pathways not

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Ourham Pre-proof

27 **Appendix 1: Recovery of gaseous and non-volatile Hg species during sample processing**

28 One concern drove our attention regarding the results for Hg species. Indeed, the unfiltered non-29 gaseous mercury, calculated as the sum of $Hg(II)$ and MMHg, ranges from 0.11 to 3.13 ng L⁻¹ in the 30 subsurface water samples with a median value of 0.39 ng L^{-1} , while DGM can be very significant and ranges 31 from 0.02 to 10.79 ng L^{-1} with a median value of 0.11 ng L^{-1} . As a consequence, the recovery of DGM, 32 mainly as Hg(0) (and potentially as DMHg), in collected water samples filtration, acidification and storage is 33 always questionable. To answer properly to that question, in the last sampling campaigns June 2019, we 34 collected in 125 mL Teflon container the water sample that has been previously purged to collect DGM 35 species on gold-coated sand trap. These "purged samples" were acidified adding high-purity HCl (1% v/v) 36 (Trace metal grade) and analysed by GC-ICP-MS following the protocol described in **2. Material and** 37 **Methods**. The **Figure A2** shows a comparison between results obtained in purged and unpurged unfiltered 38 samples for both methylated (MMHg) and inorganic (iHg(II)) Hg. On the one hand, with good coefficient of 39 determination ($R^2 = 0.91$) and a slope not significantly different from 1 (0.94±0.07), the results for iHg(II) 40 analysis confirm the fact that gaseous Hg (DGM) is not a significant contributor to Hg(II) measured in 41 unpurged samples. On the other hand, the slope obtained for the MMHg analysis is lower than 1 ($R^2 = 0.98$, 42 slope = 0.75 ± 0.02), thus suggesting that in unpurged samples some DMHg was also occurring. This was 43 further confirmed by the detection of DMHg (<0.2 pg L^{-1}) in carbotrap columns, used in a parallel study to 44 quantify the volatile selenium species 1.2 . Indeed, in the last depth of Lake Gentau (17.5m, anoxic zone) and 45 in the last two depths of Lake Sabocos (18 and 25m, anoxic zone) and the middle depths of Lake Sabocos 46 (9 and 13m, maximum chlorophyll-a, witness of algae production), DMHg was detected in the carbotrap 47 columns. Regarding the methylated species quantified by GC-ICPMS, removing the samples exhibiting 48 suboxic to anoxic conditions (three last depths from Gentau, 8, 12 and 17.5m; last depth from Sabocos 49 25m), the slope becomes not significantly different from 1 ($R^2 = 0.91$, slope = 0.92±0.09). Overall, in most 50 of the cases, especially regarding subsurface samples, DMHg is negligible in comparison with MMHg, and 51 the measured methylated species are mainly in the form of MMHg. Filon container the water sample that has been previously
d sand trap. These "purged samples" were acidified adding
and analysed by GC-ICP-MS following the protocol descr
A2 shows a comparison between results obtained i

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Appendix 2: Hg measurement outliers for Total Hg (HgTOT) and DGM

 Regarding total Hg (HgTOT), defined here as the sum of concentrations of iHg(II), MMHg and DGM, in unfiltered samples (DGM only measured in unfiltered samples), seven specific samples stood out as 58 outliers (Grubbs's tests, p-value < 0.05 then p-value = $0.063 > 0.05$).

 Major five outliers concern two specific and particular lakes : Lake Gentau (June 2018, 10:50 a.m.) and Lake Sabocos (June 2018, 09:45 a.m., 00:35 p.m. and 3:35 p.m.; June 2019, 00:30 p.m.). This is mainly because Lakes Gentau and Sabocos exhibited important DGM with extremely high values around midday 62 (DGM = 10.79 ng L⁻¹ in June 2018 for Lake Gentau; DGM = 4.65 ng L⁻¹ in June 2018 and DGM = 1.34 ng 63 L⁻¹ in June 2019 for Lake Sabocos) when the light incidence was the highest (Figure A4). The explanation is not clear, but this important DGM might be due to i) accumulation of DGM under the ice cover during the winter or ii) important iHg(II) provided by the snow melt then converted into Hg(0) by reduction pathways. In both cases, it implies that the sampling has been done right after the spring period, where the lakes are frozen and the catchment is covered by snow, which is the case in June 2018. Indeed, a few days before the sampling, more than 90 % of the studied lakes were completely frozen. However, it remains difficult to explain such exceptionally high levels of DGM measured only in these two lakes. Further studies, with intensive monitoring and sampling during several days at the snowmelt period are needed to fully understand the mechanisms occurring in those Pyrenean Lakes. To avoid bias interpretation, these five samples are discarded for the discussion on DGM levels and volatilization in high-altitude lakes. In June 2018 for Lake Gentau; DGM = 4.65 ng L⁻¹ in June 2
Ike Sabocos) when the light incidence was the highest (**Fig**
Ipportant DGM might be due to i) accumulation of DGM unde
Hg(II) provided by the snow melt then conv

 In Lake Coanga in October 2018 at 8:50 a.m., very high levels of iHg(II) were measured in both 74 filtered (iHg(II)_F = 2.68 ng L⁻¹) and unfiltered (iHg(II)_{UF} = 2.88 ng L⁻¹) samples. In this sample, filtered and unfiltered MMHg and DGM does not differ strongly from the other samples suggesting that this high iHg(II), either filtered or unfiltered, might be due punctual iHg(II) inputs or simply to contamination during the sample 77 collection: this sample will be discarded from the discussion on $Hg(II)$, and consequently from Hg_{TOT} . In 78 Lake Panticosa in October 2017 at 00:30 p.m., the relatively high levels of both iHg(II) (iHg(II) $F = 0.64$ ng L⁻ 79 $\frac{1}{1}$; iHg(II)_{UF} = 0.68 ng L⁻¹) and DGM (DGM = 1.21 ng L⁻¹) suggested that this specific sample was enriched in Hg, probably coming either from the catchment (the biggest one, 3229 Ha) or from punctual high Hg inputs.

83 **Appendix 3: Mercury species distribution in the water column of Lake Arratille (Figure A5) and Lake** 84 **Azul (Figure A6)**

85 In the shallow alpine Lake Arratille (2256 m asl.), physico-chemical parameters measured in-situ 86 with the multiparameter probe (temperature, oxygen and chorophyll) did not vary strongly within the water 87 column. Indeed, the average temperature was 5.0 ± 0.4 °C in June 2018 and 8.0 ± 0.9 °C in October 2018, 88 and the average dissolved oxygen was 65 ± 23 % in June 2018 and 71 ± 10 % in October 2018. 89 Nevertheless, it is worth noting that a short decrease in both temperature and oxygen is observed in the last 90 2 meters of the lake in both June and October 2018. Concerning the mercury species distribution, no 91 significant difference was observed for unfiltered and filtered iHg(II) (0.26 \pm 0.10 and 0.18 \pm 0.05 ng L⁻¹, 92 respectively), unfiltered and filtered MMHg $(0.010±0.04$ and $0.008±0.03$ ng L⁻¹, respectively) and DGM 93 $(0.13\pm0.06$ ng L⁻¹) along the water column, suggesting a well-mixed lake with clear water.

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95 In the shallower and more elevated Lake Azul (2420 m asl.), dissolved oxygen (76 \pm 1 % in June 96 2018 and 72 ± 2 % in October 2018) and temperature (4.8 \pm 0.1 °C in June 2018 and 5.1 \pm 0.2 °C in October 97 2018) did not vary at all along the water column. Regarding Hg species, on the one hand, no specific trend 98 was observed in June 2018 for unfiltered and filtered $Hg(II)$ (0.36 \pm 0.05 and 0.30 \pm 0.02 ng L⁻¹, respectively) 99 unfiltered and filtered MMHg (0.006±0.001 and 0.004±0.001 ng L⁻¹, respectively) and DGM (0.12±0.04 ng 100 L⁻¹). On the other hand, a significant increase was observed with depth in October 2018 for all Hg 101 compounds, probably as a result of sediment punctual remobilisation of Hg. in both June and October 2018. Concerning the mercury
was observed for unfiltered and filtered iHg(II) (0.26±0.1¹
d and filtered MMHg (0.010±0.04 and 0.008±0.03 ng L⁻¹,
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 Figure A1: Studied lakes together with geology of their catchments (adapted with permission from *Zaharescu et al.*³); circles show position of the lakes, and colors indicate the elevation of the corresponding lakes. Lake acronyms are detailed in **Table A1**.

(ng L-1) obtained in purged and unfiltered samples for June 2019 sampling campaign. Square blue points correspond to samples collected in well oxygenated depths (oxic water) while triangle red points correspond to samples collected in not well oxygenated depths (anoxic water). Red dashed line is the linear regression using all the samples while blue dashed line included only samples from oxic water.

217 **Figure A5**: Depth profiles of temperature, percentage of dissolved oxygen saturation, chlorophyll-a (RFU) and some other chemical parameters including mercury speciation obtained in (a) June 2018 and (b) October 2018 in Lake Arratille.

Figure A6: Depth profiles of temperature, percentage of dissolved oxygen saturation, chlorophyll-a (RFU) and some other

charming accompating including magazing applicing abtained in (a) line 2018 and (b) Ottaber 2018 in chemical parameters including mercury speciation obtained in (a) June 2018 and (b) October 2018 in Lake Azul.

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Figure A7: Linear relationship between MMHg Demethylation and MMHg Loss. The two black squares, out of the trend, correspond to MMHg demethylation / MMHg loss under dark conditions in the middle depth of Lake Sabocos (June 2019).

Table A1: Some physical characteristics of the sampled lakes. Note that max depth was either measured or estimated from size using allometric relation, and volume of the lakes was also estimated from max depth using a geometric approximation (error can be large). Pre-Devonian Granitic rocks (pDe-GR), Devonian sedimentary rocks (De-SR) include limestone, sandstone and shale. Permo-Triassic sedimentary rocks (PT-SR) include conglomerate, sandstone, lutite, and andesite. Cretaceous sedimentary rocks (Cr-SR) are mainly composed by carbonate rocks.

Table A1 (continued)

Table A2: Main chemical parameters of the studied lakes measured by the multiparametric probe (temperature, conductivity, redox potential), the TOC analyser (TOC as NPOC), the VINDTA 3C Instrument (DIC) and the ionic chromatograph (CI⁻, NO₃⁻, and SO₄²⁻). pH was calculated according to *Kortazar et al.*⁴. Associated error corresponds to standard deviation of three replicated (n=3) points sampled per lake (June 2017 and October 2017).

COA 0.24 0.41 0.35 10.09 7 5.85 169 1.03 108 **BAC** 0.16 0.57 1.68 12.45 25 6.55 129 0.94 290 **SAB** 0.21 0.49 2.33 17.25 115 7.53 207 1.71 1723

Table A3: Filtered and unfiltered inorganic mercury (iHg(II)), monomethylmercury (MMHg) and Dissolved Gaseous Mercury (DGM) concentrations measured in the subsurface water samples of the 19 studied lakes. Volatilization flux densities (FD) are calculated according to the methodology detailed in *Sharif et al.* **⁵** for DGM fluxes and using the specific gas exchange model for sheltered lakes developed by *Cole and Caraco*⁶. Associated error corresponds to standard deviation of three replicated (n=3) points sampled per lake (June 2017 and October 2017). $^{\circ}$ correspond to outlier (Grubbs's test).

Table A4: Filtered and unfiltered inorganic mercury (iHg(II)), monomethylmercury (MMHg) and Dissolved Gaseous Mercury (DGM) concentrations measured in the depth water samples of the 4 intensively studied lakes (Gentau, Sabocos, Arratille and Azul Superior) together with some physicochemical parameters.

Table A5: Operating conditions for the incubation experiments. Note that PFA Teflon bottles (Nalgene) were used to allow in-situ transmission of both UV A and B during incubation (of subsurface waters) **⁵** .

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Highlights

- Hg speciation and cycling was studied in waters of Pyrenean alpine lakes
- Spatial and seasonal variations of Hg levels and speciation remained low
- Hg species transformation rates were measured by in-situ experiments
- Significant Hg methylation was measured in stratified anoxic waters
- Large photochemical reactions and Hg(0) evasion occurred in surface waters
- Climate change and eutrophication can be important drivers of those pathways

OUTING PROCESS

Declaration of interests

 \boxtimes The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

 \Box The authors declare the following financial interests/personal relationships which may be considered as potential competing interests:

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