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Gram-Scale Enantioselective Synthesis of (+)-Lucidumone

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ABSTRACT: The first enantioselective total synthesis of (+)-lucidumone is described through a 13-steps synthetic pathway (longest linear sequence). The key steps involve the formation of a bridged bicyclic lactone by an enantioselective inverse electron demand Diels-Alder cycloaddition, C–O bond formation to assemble two fragments and a one-pot retro-[4+2] / [4+2] cycloaddition cascade. The synthesis is scalable, and more than one gram of natural product was synthesized in one batch.

Meroterpenoids are an important class of natural products, displaying a partially terpenic skeleton, along with a part from different biogenetic origin, often polyketide. Due to their interesting structure and often powerful biological activities, meroterpenoids have attracted the interest of synthetic chemists for decades.^{1,2} Lucidumone (1) is a meroterpenoid isolated in 2019 from the fruiting bodies of Ganoderma lucidum, a mushroom commonly used in traditional medicine in China and Japan.^{3,4} This natural product presents an unprecedented 6/5/6/6/5 caged pentacyclic structure, with a hydroquinone, one secondary alcohol and two ketone motifs (Scheme 1A). Its structural complexity can be highlighted by the 6 contiguous stereogenic centers on the bicyclo[2.2.2]octane subunit. Moreover, preliminary studies indicate that lucidumone selectively inhibits COX-2 by directly binding with Tyr385 and Ser530 residues, providing it with an interesting therapeutic potential for the treatment of inflammation.⁵ Interestingly, although lucidumone is produced as an enantiomeric mixture by Ganoderma lucidum, only one enantiomer displayed the selective COX-2 inhibitory activity. Nevertheless, this natural product was isolated in sub-mg scale (extraction yield of $1.5 \cdot 10^{-8}$, 0.4 mg of (+)-1 isolated), hampering in-depth biological activity study.

The biosynthesis of lucidumone is believed to originate from a precursor bearing two isoprenic units and a hydroquinone motif (Scheme 1A), through a series of oxidations and cyclizations.³ Recently, the team of She reported a synthetic study allowing the construction of a tetracyclic structure similar to lucidumone, without the secondary alcohol and where the hydroquinone was methylated (Scheme 1B).⁶ This approach allowed a racemic synthesis of the skeleton in 18 steps, involving a Cu-mediated C–H oxidation, an intramolecular conjugate addition and an acidmediated cyclization. However, the total synthesis of lucidumone was never achieved to date. Attracted by the structural complexity as well as the therapeutic potential of this natural product, we decided to engage in its total synthesis. In this manuscript, we describe an enantioselective gram-scale synthesis of (+)-lucidumone.

Scheme 1. Biosynthesis and retrosynthetic analysis of lucidumone



Our retrosynthetic analysis allowed us to identify $C^2 - C^3$ and $C^6 - C^7$ as key disconnections, which could be achieved through an intramolecular Diels-Alder (IMDA) cycloaddition, providing that $C^{8'}-C^{9'}$ is a double bond which can be reduced in a late stage (Scheme 1C).7,8,9 This key disconnection could greatly simplify the structure into synthon 2, bearing an indenone and a cyclohexadiene moieties for the key cycloaddition. Given the high instability of the cyclohexadiene, we would mask this function as a bicyclic lactone 3, which could regenerate the desired cyclohexadiene by CO₂ extrusion. Fragment 3 would be obtained by C-O bond formation from two fragments of similar size 4 and 5. To allow an enantioselective synthesis, fragment 5 should be obtained enantioselectively, for which we envisaged an inverse electron demand Diels-Alder (IEDDA) reaction involving a 2-pyrone 6 and an enol ether 7.10,11 The introduction of the methyl ketone moiety can be envisaged either in the early steps of the synthesis or in the endgame. This approach presents 3 key steps / challenges. (i) It is unclear whether the electron-demand in the late-stage IMDA reaction will be favorable or not, given the presence of an electron donating group at C^{7'} (supposed to meet the electrondeficient site of the diene C6') and an electron-withdrawing group at $C^{2'}$ (supposed to meet the electron-enriched site C^{3}). (ii) The C–O bond coupling between fragments 4 and 5 will also be an important challenge, as the formation of β -keto enol ethers typically requires the use of excess hydroxy partner, making it unsuitable for fragment coupling.12 (iii) The challenge of enantioselective IEDDA reaction involving 2-pyrones and acyclic enol ethers for the synthesis of bridged bicyclic lactones was recently tackled

Scheme 2. Total synthesis of Lucidumone^a

by our group through the development of a new synthetic methodology.^{13,14}

We then set out to achieve the synthesis of lucidumone. The synthesis of indandione fragment **4** was achieved starting from 3,6-dihydroxyphthalonitrile **8**. Hydrolysis of the dinitrile followed by dehydration and acetylation led to anhydride **9**, which was converted to indandione **10** in the presence of ethyl 3-oxobutanoate under basic conditions (Scheme 2). The phenolic moieties were protected as TBS ethers, affording **4** in 4 steps and 56% yield. Remarkably, this synthesis was easily scalable, as almost 20.5 g of fragment **4** was obtained starting from 14.4 g of starting material **8**.

Regarding the cyclohexadiene fragment, our first idea was to achieve the cycloaddition with a 2-pyrone derivative already bearing the methylketone functionality in C³. However, all our attempts to prepare the 2-pyrone derivative **6a** proved unsuccessful (see Scheme S1-1, S1-2 and S1-3 in the supporting information). Therefore, we applied the IEDDA reaction developed by our group¹³ with pyrone **6b**¹⁵ and benzyl enol ether **7**. The synthesis of (–)-**12** could be efficiently performed in multigram scale. It should be noted that on such a scale, the minor diastereomer could be observed (dr = 48 : 1), although easily separated by column chromatography after the next step. Deprotection of the primary alcohol afforded (–)-**5b** in excellent yield, with 12 g easily obtained in one batch.



^a DIAD = diisopropyl azodicarboxylate, DMP = Dess-Martin periodinane, TBAF = tetra-*n*-butylammonium fluoride

The C–O bond formation to couple **4** and (–)-**5** proved to be particularly difficult. A selection of attempts using model substrates are represented in Table 1. All our attempts to use a primary alcohol as a nucleophile in the presence of bromo indenone 17 failed to give 22a (entries 1-4). Converting the primary alcohol into a halide 19 to achieve O-alkylation of indandione 4 only led to small amounts of C-alkylation product and no 22b was observed. (entry 5). Using tosylate 20 instead of halide 19 in order to favour O versus C alkylation resulted in decomposition of the indandione (entry 6). A combination of indandione 4 with primary alcohol 18 in the presence of Brønsted or Lewis acid was equally unfruitful (entries 7-8). Finally, Mitsunobu conditions smoothly led to O-alkylation, affording (±)-23 in moderate yield (entry 9). After optimization using substrate **5b** (table S₂), the fragments were efficiently coupled, affording (-)-3 in acceptable yield on decagram scale.

Table 1. Study of the C-O bond formation

		OMe
Br 17 OF	X 18, X = OH 19, X = I 20, X = OTs Conditions	22a, R = OMe 22b, R = OTBS
	+ 0 MeO ₂ C HO OMe TBSO (±)-21	
Entry	Conditions	Result
1	17 , 18 , Cul, 1,10-Phen, Cs ₂ CO ₃ , toluene, 100 °C	No reaction
2	17, 18, CuCl, EtOAc,	Decomposition
	NaH, toluene, 120 °C	
3	17 , 18 , nBuLi, THF, o to 65 °C	Decomposition
4	17 , 18 , K ₂ CO ₃ , DMF, 80 °C	Decomposition
5	4, 19 , K₂CO₃, DMF, 70 °C	C-alkylation
6	4 , 20 , K ₂ CO ₃ , DMF, o °C	Decomposition
7	4, 18 , TsOH, toluene, reflux	No reaction
8	4, 18 , TiCl ₄ , toluene, rt to 60 °C	Decomposition
9	4 , (±)- 21 , PPh ₃ , DEAD, THF, -20 °C to rt	(±)-23 (51%)

From intermediate (-)-3, the key CO₂ extrusion via a retro-[4+2]-cycloaddition and subsequent [4+2] cycloaddition could be achieved in one-pot, by refluxing in chlorobenzene. In the endgame of the synthesis, we only needed to introduce the methyl ketone moiety, reduce the olefin and cleave the protecting groups. Despite all of our attempts, the formation of the methyl ketone could not be achieved without affecting the benzylic ketone. Thioesterification in prevision of a Fukuyama coupling¹⁶ or amide formation (with the idea of performing chemoselective Tf₂O-mediated amide activation)¹⁷ were unsuccessful. Nevertheless, we could introduce the methyl ketone by a sequence involving reduction of both the ester and the benzylic ketone using LiAlH₄, Dess-Martin oxidation, Grignard addition and reoxidation. This sequence, although requiring 4 steps, is high yielding (65% over 4 steps) and can be performed in only two days as the reaction time for each step is short (30 min to 2h). Moreover, the synthesis can be performed on multigram scale, and 4.5 g of intermediate (-)-15 can be easily obtained. From (-)-15 having all the carbon skeleton of lucidumone, reduction of the C8'-C^{9'} olefin and cleavage of the benzyl protecting group could be achieved in one-pot under hydrogenation conditions. It should be noted that although reduction of the alkene was completed in 2h, longer reaction times were required for the benzyl deprotection. Finally, deprotection of the phenol moieties allowed to complete the total synthesis of (+)lucidumone 1, with high optical purity. Remarkably, the synthesis was highly scalable, as 1.61 g of natural product was obtained in one batch.

To conclude, the first enantioselective total synthesis of lucidumone was achieved in 13 steps as the longest linear sequence and 25.8% overall yield from commercially available **11** (15.8% in 12 steps from **8**). The synthesis was enabled by an enantioselective IEDDA reaction developed by our group, a key C–O bond formation for fragment assembly and a one-pot retro-[4+2] / [4+2] cycloaddition cascade. The sequence could be performed on gram-scale, and over 1.6 g of natural product was synthesized in one batch, providing enough material for in-depth biological evaluation.

ASSOCIATED CONTENT

Supporting Information.

Experimental procedures, detailed optimizations and analytical data for all new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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Author Contributions

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