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1 **Microbial Competition in the Subpolar Southern Ocean: an** 2 **Fe-C co-limitation experiment**

3
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25 **1. Abstract**

26 Iron (Fe) is a paradox in the modern ocean—it is central to many life-critical
27 enzymes but is scarce across most surface waters. The high cellular demand and low
28 bioavailability of Fe likely puts selective pressure on marine microorganisms.
29 Previous observations suggest that heterotrophic bacteria are outcompeted by small
30 diatoms for Fe supply in the subantarctic zone of Southern Ocean, thereby
31 challenging the idea of heterotrophic bacteria being more competitive than
32 phytoplankton in the access to this trace metal. To test this hypothesis, incubation
33 experiments were carried out at the Southern Ocean Time Series site (Mar.–Apr.
34 2016). We investigated (a) whether dissolved organic carbon (DOC), dissolved Fe,

1 or both limit the growth of heterotrophic bacteria and, (b) if the presence of potential
2 competitors has consequences on the bacterial Fe acquisition. We observed a
3 pronounced increase in both bulk and cell-specific bacterial production in response to
4 single (+C) and combined (+Fe+C) additions, but no changes in these rates when
5 only Fe was added (+Fe). Moreover, we found that +Fe+C additions promoted
6 increases in cell-specific bacterial Fe uptake rates, and these increases were
7 particularly pronounced (by 13-fold) when phytoplankton were excluded from the
8 incubations. These results suggest that auto- and heterotrophs could compete for Fe
9 when DOC limitation of bacterial growth is alleviated. These interactions between
10 primary producers and nutrient-recyclers are unexpected drivers for the duration and
11 magnitude of phytoplankton blooms in the Southern Ocean.

12 **2. Introduction**

13 Geological timescales have enabled microbes to develop adaptive solutions to an
14 evolving environment, and iron (Fe) had a fundamental role in the metabolic
15 pathways that emerged (Falkowski and de Vargas, 2004; Hunter and Boyd, 2007).
16 As life appeared in an oxygen-free environment, the primordial ocean provided
17 sufficient concentrations of readily available Fe (Hunter and Boyd, 2007). However,
18 the concentration of dissolved Fe (DFe) decreased drastically after two major
19 irreversible oxygenation events (~2.4 billion years and ~542 million years ago, Ilbert
20 and Bonnefoy, 2013) and Fe is found at trace levels in most of today's ocean surface
21 (<0.5nM, Johnson, 1997). Despite these changes, the Fe demand in marine
22 microorganisms remained high. Nature retained this versatile metal, that can have a
23 wide range of oxidation states, as an integral part for a wide range of proteins
24 throughout evolution and many of these proteins are irreplaceable agents for vital
25 cellular metabolic activities (oxygen transport, electron transport, DNA synthesis,
26 etc., Morel and Price, 2003).

27 In phytoplankton, the photosynthetic transport chain is one of the most prominent Fe-
28 dependent processes. One single copy of a photosystem requires 23-24 atoms of Fe,
29 and overall 80% of Fe is allocated to the photosynthetic transport chain in a cell
30 (Behrenfeld et al., 2006; Raven et al., 1999; Strzepek and Harrison, 2004). In
31 heterotrophic bacteria the respiratory chain accumulates more than 90% of the
32 intracellular Fe (Andrews et al., 2003; Tortell et al., 1999). Phytoplankton and
33 bacteria play important roles in the ocean and have direct influence on global

1 biogeochemical cycles. Considering that phytoplankton drive ocean CO₂
2 sequestration via photosynthesis and downward export, while heterotrophic bacteria
3 control much of the oceanic release of CO₂ via respiration, the outcome of a
4 competition for Fe could influence the direction and magnitude of carbon fluxes in
5 the upper ocean.

6 Despite widespread interest in microbial Fe requirements (Blain and Tagliabue,
7 2016; Sarthou et al., 2005; Strzepek et al., 2019; Twining and Baines, 2013), there is
8 no consensus regarding the minimum Fe requirements for phytoplankton or
9 heterotrophic bacteria. There are two reasons to explain why this question has not
10 been resolved. First, there have been few studies on the Fe requirements of
11 heterotrophic bacteria compared with those for phytoplankton. Second, the wide
12 range of Fe content relative to C biomass (Fe:C ratio) that exist for phytoplankton
13 does not favor conclusive comparison (Blain and Tagliabue, 2016, and references
14 herein). What is clear, however, is that DFe in the oceans is overwhelmingly
15 complexed (99%, Rue and Bruland, 1997) by strong organic ligands with evidence of
16 them containing Fe-binding functional groups consistent with biologically produced
17 siderophores (Gledhill and Buck, 2012; Macrellis et al., 2001). Marine microbes
18 have evolved different mechanisms to cope with the diversity of the Fe-binding
19 ligand pool, and the capacity to acquire enough Fe for survival in a ‘highly diffusive’
20 open ocean provides a competitive edge (Desai et al., 2012; Hopkinson and Barbeau,
21 2012; Toulza et al., 2012). The capacity to produce siderophores is generally
22 confined to heterotrophic bacteria (Armstrong et al., 2004), but the ability to take up
23 siderophores may be more widespread than previously thought, and extend to the
24 phytoplanktonic realm (Hogle et al., 2016; Kazamia et al., 2018; McQuaid et al.,
25 2018). Recent studies have revealed that distinct siderophores and strategies are
26 being employed by heterotrophic bacteria (Boiteau et al., 2016, 2019; Bundy et al.,
27 2018; Debeljak et al., 2019). But to date, no eukaryotic phytoplankton have been
28 found to produce or release siderophores. For this reason heterotrophic bacteria are
29 commonly reported as highly efficient competitors, especially in severely Fe-limited
30 environments (Braun and Killmann, 1999).

31 The Southern Ocean is the largest High Nutrient, Low Chlorophyll (HNLC) region in
32 the world ocean, mainly because of Fe limitation. In the Southern Ocean, chlorophyll
33 levels remain low year-round, but phytoplankton blooms occur in areas in the

1 vicinity of land masses (Blain et al., 2007). During austral spring 2011, the KEOPS2
2 project aimed at exploring different phytoplankton blooms east of Kerguelen island.
3 Over the course of the bloom, the release of dissolved organic carbon (DOC) derived
4 from primary production increased the Fe demand of heterotrophic bacteria
5 (Fourquez et al., 2015) which were Fe-limited (Obernosterer et al., 2015). The
6 availability of a labile C source may have led to a higher bacterial Fe demand. These
7 findings have led to the hypothesis that labile organic carbon exacerbated the
8 potential competition between small-sized phytoplankton cells (pico- and
9 nanophytoplankton) and heterotrophic bacteria for Fe (Fourquez et al., 2015). The
10 present work aimed to test the above hypothesis. For this, our experimental design
11 was based on the joint assumption that (1) heterotrophic bacteria are outcompeted for
12 Fe by pico- and nanophytoplankton and (2) that DOC availability to heterotrophic
13 bacteria influences the strength of this relationship.

14 **3. Materials and Methods**

15 **3.1. Site description**

16 This study was carried out as part of the V02-IN2016 voyage of the R.V. Investigator
17 (March 11 to April 17, 2016). During the expedition, we visited on two occasions
18 (March 19 and 29) the Southern Ocean Time Series (SOTS, 47°S, 142°E) site that is
19 located within a low current region in the subantarctic Zone (SAZ) north of the
20 subantarctic Front (SAF) that marks the northern edge of the Antarctic Circumpolar
21 Current (Figure 1). This area represents a large portion of the total area of the
22 Southern Ocean that serves as a strong sink for atmospheric CO₂. Conditions at
23 SOTS are typical and representative of the Indian sector SAZ, from ~90 to 145°E.
24 The absence of Fe is regarded as the primary cause that restricts primary production
25 and constrains the biological pump in the area (Cassar et al., 2011; Sedwick et al.,
26 1997; Trull et al., 2019). The relief of the Fe limitation can occur by aerosol Fe
27 supply in summer in the region that differs in this way from mechanisms of deep
28 mixing and/or sediment input/resuspension that enhances Fe concentrations in
29 surface waters at the vicinity of subantarctic islands such as Kerguelen (Blain et al.,
30 2008; Rembauville et al., 2015).

3.2. Experimental strategy

The study involved two separate sets of incubation experiments: Experiment 1 and 2 (Figure 2 and Suppl. Material Figure 1). Because there is a potentially confounding influence of Fe and C limitation on bacterial processes, the objective of Experiment 1 was to first determine whether Fe, C, or both are limiting or co-limiting factors at SOTS while the objective of Experiment 2 was to determine whether the presence of larger cells (especially pico- and nanoplankton) influences bacterial activities. In addition to the one control that consisted in unamended nutrient seawater (no addition), the following 3 treatments were prepared as triplicates: +Fe, +C, and +Fe+C. Iron (+Fe) was added as FeCl_3 (final concentration of FeCl_3 1 nM), and carbon (+C) was added as trace-metal clean glucose (final concentration of glucose $10 \mu\text{mol L}^{-1}$). Hence, the addition was $16.6 \mu\text{molFe molC}^{-1}$ to attain the bacterial Fe quota observed in Fe-replete bacterial cultures ($16.1 \pm 2.3 \mu\text{molFe molC}^{-1}$, Fourquez et al. 2014). For experiment 1, incubations were performed directly on unfiltered seawater (bacteria were incubated with micro and pico-and nanoplankton communities). Additional incubations were performed on $20\mu\text{m}$ -prefiltered seawater ($<20\mu\text{m}$ condition), and on $1\mu\text{m}$ -prefiltered seawater ($<1\mu\text{m}$ condition) for Experiment 2. We use the term “condition” throughout the manuscript to refer to the different size fractionation treatments (summary in Table 1). The biological response of heterotrophic bacteria was monitored from sub-samples drawn from these incubation bottles, and analysed for several parameters as described in sections below.

3.3. Sampling procedures

Seawater was pumped from the surface ocean ($\sim 5\text{m}$ depth) using a trace metal clean towed fish sampler. Samples were collected directly into a trace metal clean laboratory (clean room, ISO5) where 300 mL of seawater was dispensed into 500 mL acid-washed polycarbonate (PC) bottles under a laminar flow hood (ISO class 5). Overall, 12 (for experiment 1) and 54 (for experiment 2) independent replicates were prepared in bottles capped and sealed with Parafilm. All plastic materials used were acid-washed following GEOTRACES procedures in our home laboratory (GEOTRACES cookbook, (Cutter et al., 2017)). Briefly, PC bottles were soaked for one week in the alkaline detergent Decon 90, then rinsed four times with deionized water and three times with milliQ water. They were subsequently filled with 10%

1 hydrochloric acid (Suprapur, Merck) for one week. After that time, bottles were
2 rinsed 5 times with highly purified water. The PC bottles were dried, and UV
3 sterilized for 15-30 minutes under a laminar flow hood and then stored in triple
4 plastic bag before being used.

5 To minimize risks of potential contamination of samples with metals or dissolved
6 organic matter as an artefact of filtration in preparation for Experiment 2, seawater
7 was filtered at very low pressure (<5 Hg). All incubations were performed at *in situ*
8 temperature (13.5°C). Bottles were placed in a water bath within a controlled
9 temperature room (13.5°C) to avoid temperature fluctuations. For experiment 1,
10 incubations were performed in total darkness. For experiment 2, incubations were
11 conducted under 12:12 light-dark condition, and we employed neutral density
12 screens to attenuate the light intensity. Low levels of light can reduce rates of
13 photosynthesis and the release of DOC associated and may alter Fe uptake rates in
14 photosynthetic cells. Therefore, we opted to use low light intensity (average 4.5
15 $\mu\text{mol photon m}^{-2} \text{ s}^{-1}$, ~1W) in experiment 2 to (a) increase potential competition for
16 Fe stocks between autotrophic pico- and nanoplankton cells and heterotrophic
17 bacteria, and to (b) avoid stimulation of large phytoplankton growth and subsequent
18 organic enrichment artefacts. For subsampling, incubation bottles were transferred to
19 the clean container and opened under a laminar flow hood (ISO class 5). Subsamples
20 from each triplicate for bacterial abundance and production were taken at T0, T1
21 (+24h), T2 (+48h), T3 (+72h) and at T4 (+110h) for Experiment 1 and at T0, T1
22 (+36h), and T2 (+120h) for Experiment 2. The time points for experiment 2 were
23 chosen following results collected during Experiment 1. For both experiments
24 bacterial heterotrophic production, cell abundance, and Fe uptake were measured at
25 several time points. We also measured heterotrophic bacterial respiration at the end
26 of the incubation in Experiment 1 to estimate the bacterial growth efficiency in the
27 different treatments (Figure 2). Methods for each parameter measured are detailed in
28 sections below.

29 **3.4. Cell abundance**

30 For each biological replicate, 4.5 mL subsamples were fixed with glutaraldehyde
31 (0.5% final concentration), kept in the dark at 4°C for 20 min and then shock-frozen
32 in liquid nitrogen. The samples were stored at -80°C until analyses by flow
33 cytometry. Flow cytometry was performed following the protocols in Marie et al.,

1 (1997 and 2005). Frozen samples were rapidly thawed in a water bath at 70 °C for 3
2 min and aliquots taken for autotrophic and/or prokaryote cell counts. Sample aliquots
3 were kept on ice in the dark and promptly analysed on a Becton Dickinson FACScan
4 flow cytometer fitted with a 488 nm laser. MilliQ water was used as sheath fluid for
5 all analyses. Before and after each run, samples were weighed to ± 0.0001 g to
6 determine the volume of sample analysed.

7 Samples for autotrophic cell abundance were prepared by aliquoting 998 μL of
8 sample into a clean 5 mL polycarbonate tube, with 2 μL of PeakFlow Green 2.5 μL
9 beads (Invitrogen) added as an internal fluorescence and size standard. Each sample
10 was run for 5 min at a high flow rate of $\sim 40 \mu\text{L min}^{-1}$. Autotrophic cell populations
11 were separated into regions based on their chlorophyll autofluorescence in red (FL3)
12 versus orange (FL2) bivariate scatter plots. *Synechococcus* cells were determined
13 from their high FL2 and low FL3 fluorescence. Pico- and nano-phytoplankton
14 communities were determined from their relative cell size in side scatter (SSC)
15 versus FL3 fluorescence bivariate scatter plots. Final cell counts in cells L^{-1} were
16 calculated from event counts in the identified regions and analyzed volume.

17 Samples for prokaryote cell abundance were prepared by aliquoting 995 μL of
18 sample to a clean 5 mL polycarbonate tube. Samples with high prokaryote cell
19 counts were diluted to 1:10 with 0.2 μm filtered seawater (FSW) to remove
20 underestimation of cell concentration from coincidence (100 μL sample in 900 μL
21 FSW). Cells were stained for 20 min with 5 μL of SYBR Green I (Invitrogen) at a
22 final dilution of 1:10,000. An additional 2 μL of PeakFlow Green 2.5 μL beads
23 (Invitrogen) was added to the sample as an internal fluorescence and size standard.
24 Each sample was run at a low flow rate of $\sim 12 \mu\text{L min}^{-1}$ for 3 min and prokaryote cell
25 abundance was determined from bivariate scatter plots of SSC versus green (FL1)
26 fluorescence. Final cell counts in cells L^{-1} were calculated from event counts in the
27 identified regions and analyzed volume.

28 **3.5. Heterotrophic bacterial production**

29 Bacterial production was estimated by [^3H] leucine incorporation applying the
30 centrifugation method (Martinez et al., 1996) as described in Obernosterer et al.,
31 (2008). Briefly, 1.5-mL samples were incubated with a mixture of radioactive
32 leucine, L-[3,4,5- ^3H (N)] (PerkinElmer, specific activity $123.8 \text{ mCi.mol}^{-1}$) and

1 nonradioactive leucine at final concentrations of 20 nM. Duplicates plus one “killed
2 sample” were incubated in the dark at the respective *in situ* temperatures for 2–3 h.
3 Linearity of leucine incorporation over this time period was tested in parallel and at
4 two occasions (at the beginning of Experiment 1 and 2). Incubations were terminated
5 by the addition of trichloroacetic acid (TCA; Sigma) to a final concentration of 5%.
6 To facilitate the precipitation of proteins, bovine serum albumin (BSA; Sigma, 100
7 mg L⁻¹, final concentration) was added prior to centrifugation at 16,000 g for 10 min
8 (Van Wambeke et al., 2002). After discarding the supernatant, 1.5 mL of 5% TCA
9 solution was added and the samples were subsequently vigorously shaken on a
10 vortex and centrifuged again. The supernatant was discarded again and 1.5 mL of
11 UltimaGold™ uLLt (PerkinElmer) was finally added. The radioactivity incorporated
12 into bacterial cells was counted in Hidex 300SL Liquid Scintillation Counter. A
13 factor of 1.55 kg C mol leucine⁻¹ was used to convert the incorporation of leucine to
14 carbon equivalents, assuming no isotope dilution (Kirchman et al., 1993). Isotopic
15 dilution ranged from 1.0 to 1.3 as determined on three occasions using a kinetic
16 approach.

17 **3.6. Fe uptake rates**

18 Following the subsampling for bacterial production, Fe uptake experiments were
19 initiated by adding 0.2 nmol L⁻¹ at final concentration of ⁵⁵Fe (as ⁵⁵FeCl₃,
20 PerkinElmer specific activity 2.46 x 10³ Ci mol⁻¹) after 36h and 120h of incubations
21 (independent replicates). After a 24-hour incubation with ⁵⁵Fe, microorganisms were
22 filtered through a stack of nitrocellulose filters (Whatman) of 0.2 and 0.8µm
23 porosity, separated with 20µm mesh filters. These filter porosities were chosen to
24 separate phytoplankton (including *Synechococcus*, >0.8µm) from heterotrophic
25 bacteria (0.2–0.8µm). Before running dry, the filters were rinsed with 0.2-µm filtered
26 Ti(III) citrate EDTA solution (Tovar-Sanchez et al., 2003) for 2 minutes to dissolve
27 any extracellular Fe, followed by three consecutive rinses with 5 mL of 0.2µm
28 filtered seawater for 1 min (Fourquez et al., 2012, 2015). The filters were placed into
29 plastic vials and 10 mL of the scintillation cocktail Filtercount (Perkin Elmer) was
30 finally added. Vials were agitated for 24h before the radioactivity was counted with
31 the Hidex 300SL scintillation counter. Radioactivity on filters was corrected for
32 background (⁵⁵Fe adsorbed into the filter and/or onto particles and not being
33 efficiently washed out by the washing solution) using ⁵⁵Fe-radiotracer medium with

1 dead cells, also called “killed control”. Killed controls were treated the same as above
2 but microorganisms were fixed with 1% of glutaraldehyde (left for 1 hours at 4°C)
3 prior addition of ^{55}Fe . DFe concentration in each incubation bottle was also assessed
4 prior incubation with ^{55}Fe . Subsamples (~40 mL) were measured by flow injection
5 with online preconcentration and chemiluminescence detection (adapted from Obata
6 et al., 1993). An internal acidified seawater standard was measured every day in
7 order to control the stability of the analysis. The detection limit was 40 pmol kg^{-1} and
8 the accuracy of the method was controlled by analyzing the SAFe S (0.110 ± 0.036
9 nmol kg^{-1} ($n = 3$); consensus value $0.093 \pm 0.008 \text{ nmol kg}^{-1}$), and SAFe D1 ($0.66 \pm$
10 $0.06 \text{ nmol kg}^{-1}$ ($n = 4$); consensus value $0.67 \pm 0.04 \text{ nmol kg}^{-1}$) seawater standards.
11 DFe concentration were employed to correct ^{55}Fe uptake estimates from cold DFe
12 present in incubation bottle at the time of the measurement. In incubation bottles, the
13 DFe concentration changed over time also suggesting that there was remineralization
14 taking place. To correct for this, we have multiplied the results from these
15 experiments by the calculated proportion of ^{55}Fe from the total DFe. Calculation
16 details can be found in Fourquez et al. (2015).

17 **3.7. Bacterial respiration**

18 Rates of respiration were determined from dissolved oxygen consumption in 24
19 hours dark incubations at the end (T4, +110h) of Experiment 1 using Winkler
20 titration method. In order to keep the bacteria only for the measurement, all samples
21 were carefully pre-filtered onto $0.8\mu\text{m}$ acid-washed PC filter. Two out of the three
22 biological replicates belonging to each treatment (control, +Fe, +C, and +Fe+C) were
23 used for measurement. Last replicate was employed as a T0 by adding manganese
24 chloride followed by alkaline iodide prior incubation. To estimate the consumption
25 in dioxygen (O_2), the amount of O_2 measured in bottles after 24 hours of incubation
26 was subtracted from T0 measurement. All incubation bottles were opened and gently
27 shaken under flow laminar hood to optimize O_2 level inside and homogeneity
28 between treatments. After these steps, the samples were subsequently transferred into
29 cleaned and acid-washed glass biological oxygen demand stoppered bottle.
30 Incubations lasted for 24 hours in the dark in a temperature-controlled incubator set
31 at 13.5°C (*in situ* temperature). At the end of the incubation time, subsampling (1.8
32 mL) for flow cytometry analysis was taken quickly just prior to add the reactive.
33 Dissolved oxygen concentration was measured based on the whole-bottle modified

1 Winkler titration of Carpenter (1965) plus modifications by Culberson (1991).
2 Bacterial respiration rates were normalized to C biomass by considering the bacterial
3 cell abundance in each incubation bottle.

4 **3.8. Carbon biomass and conversion factor**

5 Direct carbon contents for pico-nanoplankton and microplankton were estimated
6 from particulate organic carbon (POC) measurements. In total, 3 L of seawater
7 sample were first filtered through 20 μ m and subsequently passed through 1.2 μ m
8 (diameter 25mm) Sterlitech silver membrane filters, and dried at 60 °C. The samples
9 were acidified, dried and analyzed by high-temperature combustion (1000°C) to
10 determine POC. The analysis for total nitrogen, carbon and hydrogen was determined
11 using a Thermo Finnigan EA 1112 Series Flash Elemental Analyzer.

12 In parallel, the carbon content was also indirectly estimated using conversion factors.
13 Photosynthetic pico-nanoplankton cell abundance was converted to carbon biomass
14 using constant cell-to-carbon conversion factors based on the literature. Conversion
15 factors used were respectively 255 and 2590 fgC cell⁻¹ for the cyanobacterium
16 *Synechococcus* and for picoeukaryotes (Buitenhuis et al., 2012), and 183 fgC cell⁻¹
17 for nanoeukaryotes (Caron et al., 1994, 2017). We assume the cyanobacterium
18 *Synechococcus* represented the majority of the resident cyanobacteria. For
19 heterotrophic bacteria, the carbon content was estimated using 12.4 fgC cell⁻¹ as
20 reported by Fukuda et al. (1998).

21 **3.9. Statistical analyses**

22 All statistical comparisons were performed using one-way analysis of variance
23 (ANOVA) and a post hoc Tukey test. Differences were considered statistically
24 significant at $p < 0.05$. To evaluate the differences between treatments, statistics were
25 individually performed between nutrient unamended (control) and amended
26 treatments (+Fe, +C, and +Fe+C). We also evaluated statistical differences between
27 conditions by comparing unfiltered (control) to other conditions (<20 μ m or <1 μ m).

28 **4. Results**

29 **4.1. Environmental settings of the study site**

30 As is typical for HNLC regions, at SOTS site the C biomass was dominated by small
31 cells (<20 μ m) which represented 1.51 μ mol L⁻¹ of the 1.90 μ mol L⁻¹ total POC in

1 surface waters. The concentration of DFe was 0.081 ± 0.02 nM at 5m depth.
2 Concentration of dissolved organic carbon (DOC) were not measured during the
3 cruise. Nevertheless, the Southern Ocean surface waters exhibit a DOC concentration
4 range of $\sim 40\text{-}50$ $\mu\text{molC L}^{-1}$ (Hansell et al., 2009). If we consider the upper range of
5 50 $\mu\text{molC L}^{-1}$ to be representative of what can be found at SOTS, the resulting
6 DFe:DOC molar ratio was ~ 1.62 $\mu\text{mol mol}^{-1}$.

7 **4.2. Experiment 1. Responses of bacteria to Fe and C additions**

8 **4.2.1. Cell-specific bacterial production**

9 In Experiment 1, we investigated the responses of bacteria to Fe, C and concomitant
10 Fe and C additions. Sole additions of Fe did not result in any significant
11 enhancement of bulk nor cell-specific BP (Figure 3). In accordance with these
12 results, bacterial abundance in the Fe-amended treatment did not differ from the
13 control (Figure 3a). However, single (+C) and combined (+Fe+C) additions of C
14 significantly stimulated bulk and cell-specific BP. We note a pronounced response in
15 cell-specific BP to single (+C) and combined (+Fe+C) additions of carbon over the
16 time of the experiment (1.3–56-fold and 1.6–26-fold higher than control in +C and
17 +Fe+C, respectively). The results from treatments +Fe+C and +C were not
18 statistically different from each other apart from T3 (+72h, $p=0.012$). The bacterial
19 cell abundance also increased, however, the magnitude of the stimulation was less
20 than that measured for the BP. Cell abundance is a complex function between
21 growth and mortality rates; this decoupling is therefore not surprising. At the end of
22 the experiment, the enhancement of these parameters was still detectable in the +C
23 and +Fe+C treatments but a decrease in cell-specific BP was also observed after 72
24 hours of incubation.

25 **4.2.2. Bacterial respiration and growth efficiency**

26 Bacterial respiration (BR) rates were measured at the end of the Experiment 1. BR
27 varied from 0.39 ± 0.15 and 1.63 ± 0.34 $\text{fmolO}_2 \text{ cell}^{-1} \text{ d}^{-1}$ (standard error, SE; $n=2$). It
28 was intriguing to note that the highest BR rate was measured for +Fe addition alone,
29 which shows also the lowest (3.2%) bacterial growth efficiency (BGE) due to low
30 BP rate (table 2). The highest BGE estimation was measured for the +Fe+C
31 treatment with 57% (Table 2).

1 **4.3. Experiment 2. Responses of bacteria when phytoplankton is absent or** 2 **present**

3 To make the Results section concise and easy to follow, all results presented in
4 figures and text description correspond to the first time point for Experiment 2 (+36h
5 of incubation). Similar conclusions can be formulated from the second time point
6 (+120h of incubation) and data are accessible in the supplementary materials.

7 **4.3.1. Bacterial production in presence and absence of phytoplankton**

8 Cell-specific BP rates across all treatments and conditions are shown in Figure 4a. In
9 the control treatment (no nutrient addition), the cell-specific BP ranged from 4 to 9
10 ($\times 100 \text{ fmolC cell}^{-1} \text{ d}^{-1}$) and no significant difference was found between unfiltered
11 and $<20\mu\text{m}$ nor $<1\mu\text{m}$ conditions, which is evidence that the presence of
12 phytoplankton did not affect cell-specific BP. However, in the nutrient amended
13 treatments two differences were significant: a decrease in BP in the +Fe treatment
14 and an increase in the +Fe+C treatment, respectively. In the +Fe treatment, the cell-
15 specific BP was highest when the whole community was present (unfiltered) and the
16 lowest when bacteria were incubated solely with pico- and nanoplankton ($<20\mu\text{m}$).
17 The difference between these two conditions was significant ($p=0.04$, Figure 4a).
18 Cell-specific BP in +Fe+C treatment were higher than all other treatments, ranged
19 from 19 to 48 ($\times 100 \text{ fmolC cell}^{-1} \text{ d}^{-1}$) and was the highest when bacteria were
20 incubated solo. Cell-specific BP was about 2 times higher in $<1\mu\text{m}$ compared to
21 unfiltered condition, and this difference was highly significant ($p=0.005$, Figure 4a).

22 Significant differences were also found when comparing $<20\mu\text{m}$ and $<1\mu\text{m}$
23 conditions in +Fe and +Fe+C treatments. Indeed, the $<1\mu\text{m}$ condition showed
24 significantly higher rates compared to the $<20\mu\text{m}$ condition with respectively 9 ± 1
25 versus 4 ± 0.4 and 48 ± 6 versus 19 ± 1 ($\times 100 \text{ fmolC cell}^{-1} \text{ d}^{-1}$) for +Fe and +Fe+C
26 treatment. Overall, cell-specific BP was negatively affected by the presence of pico-
27 and nanoplankton cells in the +Fe and +Fe+C treatments while no effect was found
28 in the control (no addition) treatment.

29 **4.3.2. Microbial Fe uptake**

30 To investigate whether heterotrophic bacteria compete for Fe with other members of
31 the microbial community, the bacterial Fe uptake rates were determined for

1 incubations where microplankton (<20 μm condition) or both micro- and pico- and
2 nanoplankton (<1 μm condition) were excluded from incubation. Results were
3 compared to the treatment where all the members of the microbial community were
4 present (unfiltered condition). During Experiment 2, we also compared the
5 contribution of two size-fractions to Fe uptake. Figure 4b shows results for bacteria
6 on a cell-specific basis, and Figure 5 combines data for phytoplankton (<0.8 μm) and
7 bacteria (0.2-0.8 μm) on the volumetric basis for better comparison. Data used to
8 create figure 5 can also be found in detail in Table 3.

9 **4.3.3. Cell-specific Fe uptake by bacteria**

10 The response of bacteria to size-fractionation (condition) and nutrients amendments
11 (treatment) were overall similar for cell-specific Fe uptake (Figure 4b) to those
12 presented for bacterial production (Figure 4a). While no significant differences were
13 found for the control treatment, the uptake of Fe by bacteria was respectively
14 lowered and enhanced in +Fe and +Fe+C treatments. In +Fe treatment, Fe uptake by
15 bacteria is 3 times lower in presence of pico-nanoplankton (<20 μm condition) and
16 decreased from 2.1 ± 0.5 to 0.6 ± 0.1 ($\times 100 \text{ amolFe cell}^{-1} \text{ d}^{-1}$) compared to the
17 unfiltered condition. This difference was significant ($p=0.006$). Across all datasets,
18 the Fe uptake by bacteria was the highest for the +Fe+C treatment and the <1 μm
19 condition. This result makes precise the sense in which the availability of C together
20 with the removal of potential competitors had the greatest effect on the uptake of Fe
21 by bacteria. Interestingly, the concomitant addition of Fe and C did not enhance the
22 bacterial Fe uptake for the other conditions (unfiltered and <20 μm). Values were
23 respectively 4.7 ± 0.3 and 8.5 ± 3.5 ($\times 100 \text{ amolFe cell}^{-1} \text{ d}^{-1}$) for unfiltered and <20 μm
24 conditions while it reached up to 61 ± 21 ($\times 100 \text{ amolFe cell}^{-1} \text{ d}^{-1}$) for the <1 μm
25 condition. Considering all data together, the size fractionation used in the incubation
26 (i.e., presence or absence of phytoplankton) had a greater effect than addition of
27 growth-limiting nutrients.

28 **4.3.4. Fe uptake by phytoplankton**

29 Iron uptake by cells larger than 0.8 μm is presented as the Fe uptake by
30 phytoplankton in Figure 5 and specific Fe uptake by pico-nanoplankton (0.8-20 μm)
31 is given in Table 3. If we consider the unfiltered condition with no nutrient addition
32 to be the closest representation of the natural system, phytoplankton contributed to

1 66±6% of the total Fe uptake at the SOTS site. In terms of percentage contribution,
2 phytoplankton Fe uptake increased to more than 80% when Fe or Fe plus C was
3 added (82±1% and 83±1% for +Fe and +Fe+C treatment, respectively).

4 In the +Fe treatment, the Fe uptake by phytoplankton was slightly lower in <20µm
5 condition compared to unfiltered condition which is explained by the removal of
6 about 20% of the phytoplankton initial biomass. However, the concomitant addition
7 of Fe and C clearly led to the increase of Fe uptake by phytoplankton. As the Fe
8 uptake is higher by 22% in unfiltered conditions compared to <20µm, which is again
9 explained by the removal of larger cells, we consider that the microphytoplankton
10 also benefited from the +Fe+C treatment in some ways (e.g. Fe regenerated by
11 bacteria).

12 **4.4. Carbon biomass and Fe:C ratios**

13 **4.4.1. Contribution to carbon biomass at initial conditions**

14 We examined the carbon biomass partitioning of the pico- and nanoplankton
15 communities across treatments and size fractions, and in relation to the heterotrophic
16 bacteria. First, we compare estimates of C biomass of small photosynthetic cells
17 based on conversion factors and flow cytometry numbers with direct measurements
18 of POC as described before. We found 1.49±0.05 µmolC L⁻¹ (estimate) versus 1.51
19 µmolC L⁻¹ (measure). Given the comparable results we are confident in using
20 conversion factors to investigate variations in carbon biomass in our incubation
21 bottles.

22 Among the pico-and nanoplankton community, picoeukaryotic cells were the most
23 abundant in surface waters at SOTS and represented 71±0.2% (n=9) of carbon
24 biomass while cyanobacteria made up 26±0.3% (n=9) and photosynthetic
25 nanoeukaryotes 2±0.1% (n=9). After 36 hours of incubation, no notable differences
26 in these contributions were found when comparing size fractions and treatments. In
27 the case of total carbon biomass, heterotrophic bacteria were dominant (averaging
28 58%) at the start of the experiment, with values ranging from 1.85 to 2.33 µmol L⁻¹.

29 **4.4.2. Contribution to Fe uptake for bacteria and pico-nanoplankton**

30 In this section, we only consider values measured in <20µm condition as we did not
31 directly measure the carbon content of larger cells (microplankton) so that we cannot

1 evaluate accurately their contribution. Heterotrophic bacteria represented about 52%
2 (control), 56% (+Fe), and 59% (+Fe+C) of the total C biomass (Table 3). However,
3 their contribution to the total uptake of Fe did not reflect this dominance (Figure 5).
4 For the unfiltered condition, heterotrophic bacteria were only responsible for 25, 20
5 and 32% of the total Fe uptake in control, +Fe and +Fe+C treatments, respectively.

6 **4.4.3. Fe:C ratio**

7 We normalized Fe uptake rate to carbon biomass and the resulting Fe:C ratio is
8 presented in Table 3. Comparison of these Fe:C ratios for bacteria among the
9 different treatment and condition indicate that heterotrophic bacteria had a higher Fe
10 content in the +Fe+C treatment. However, this Fe:C ratio also varied from 46 to 591
11 $\mu\text{molFe molC}^{-1}$ (+Fe+C treatment, Table 3) which shows that heterotrophic bacteria
12 can assimilate a substantial amount of Fe when phytoplankton is removed from the
13 experiment. For instance, the Fe:C ratio of heterotrophic bacteria in the $<20\mu\text{m}$
14 condition was nearly twofold higher than in the unfiltered condition, and it was close
15 to 13-fold higher in the $<1\mu\text{m}$ condition. These high Fe:C ratios also indicate that
16 heterotrophic bacteria have rapidly upregulated their Fe acquisition machinery
17 relative to C to acquire more dissolved Fe or that their uptake systems were already
18 activated.

19 The Fe:C ratios were also estimated for photosynthetic pico- and nanoplankton cells
20 (including cyanobacteria). As for heterotrophic bacteria, we observed a pronounced
21 increase of Fe:C ratio in the +Fe+C treatment, but no notable difference between
22 unfiltered and $<20\mu\text{m}$ condition (Table 3). These estimates ranged from 32 to 59
23 $\mu\text{molFe molC}^{-1}$ for both control and +Fe treatments considered versus a range of
24 231-250 $\mu\text{molFe molC}^{-1}$ in +Fe+C treatment.

25 Overall, our calculation of Fe:C ratios show that pico- and nanoplankton constitute a
26 larger fraction of biogenic Fe compared to heterotrophic bacteria in all incubations.
27 However, the highest Fe:C ratio measured in this study was when bacteria were
28 incubated alone (591 ± 208 , $n=3$; +Fe+C treatment) and it was more than 2-fold
29 higher than the maximum we calculated for pico-and nanoplankton (250 ± 50 , $n=3$;
30 +Fe+C treatment for $<20\mu\text{m}$ condition).

31 **5. Discussion**

1 In environmental science, the concept of bioavailability for one or several resources
2 is generally associated with chemical features, in particular, in the case of Fe. This
3 micronutrient is present in multiple chemical forms and redox states (Morel and
4 Price, 2003). However, we show here that biological interactions matter as well. To
5 discuss the results of this study, we first comment on the concept of co-limitation.
6 Next, we discuss the nature of the interspecific interactions that most likely explain
7 our results. Finally, we close this section on implications for future perspectives of
8 research.

9 **5.1. Does carbon availability offset Fe limitation in heterotrophic bacteria?**

10 One particular feature of the Southern Ocean is that both bioavailable Fe and organic
11 carbon can be at growth-limiting concentrations for heterotrophic bacteria in surface
12 waters (Church et al., 2000; Obernosterer et al., 2015). At first sight, our results
13 suggest that heterotrophic bacteria were primarily limited by organic carbon at
14 SOTS. Iron could have a role, however, in affecting BP and bacterial metabolism
15 when the supply of organic carbon is adequate and Fe concentrations are low.
16 Indeed, while these results lead us to the conclusion that heterotrophic bacteria were
17 firstly C-limited, the argument for Fe limitation of heterotrophic bacteria is not so
18 clear. A simple comparison between *in situ* molar DFe:DOC ratio ($2.61 \mu\text{mol mol}^{-1}$)
19 and Fe:C ratios of Fe-limited cultures (e.g. 0.43 ± 0.1 and $7.52 \pm 1.65 \mu\text{molFe molC}^{-1}$
20 for oceanic strains in Fourquez et al., 2014 and Tortell et al., 1996, respectively)
21 would suggest that both nutrients may become limiting. In the present study, Fe
22 alone had no effect on rates of BP rates or cell abundance. But Fe did affect these
23 variables when added together with glucose.

24 Consistent with the hypothesis that bacteria may have been co-limited by Fe and C,
25 we observed that cell-specific BP was positively correlated ($r=0.98$, $n=6$, $p=$
26 0.000373 , Pearson correlation) with bacterial Fe uptake in the +Fe+C treatment. In
27 contrast, there was no significant correlation in the control ($r=-0.33$, $n=6$, $p=0.58$),
28 and in the +Fe treatment ($r=0.29$, $n=6$, $p=0.56$). Such a high correlation in studies
29 using natural communities is uncommon. An explanation for our results is that
30 bacterial growth became Fe-limited, but only after C-limitation was alleviated by the
31 addition of glucose. This can be explained by the increase in the C demand induced
32 by cellular remodeling to support growth and maintenance under Fe stress conditions
33 (Fourquez et al., 2014; Kirchman et al., 2000).

1 There is growing evidence that the expression of alternative pathways is a
2 widespread strategy for heterotrophic bacteria in low Fe environments (Beier et al.,
3 2015; Debeljak et al., 2019; Fourquez et al., 2014; Koedooder et al., 2018). A
4 comparative proteomics approach revealed that Fe limitation leads cells to utilize C
5 through the glyoxylate cycle (Fourquez et al., 2014). This alternative pathway not
6 only bypasses two important Fe-containing enzymes in the Krebs's cycle, but also
7 the two steps where carbon is lost as CO₂. Redirection of glucose into the Entner-
8 Doudoroff pathway also allows Fe-limited cells to supply the Krebs's cycle with
9 substrates while bypassing the first step of glycolysis which is ATP-consuming
10 (Fourquez et al. 2014). Other biomass recycling processes such as amino and organic
11 acid catabolism contribute as well to the regulation of energy production (Fourquez
12 et al. 2014). These underlying mechanisms may explain why BP was stimulated by C
13 addition, strongly stimulated by concomitant Fe and C additions, but not stimulated
14 by the addition of Fe alone in the present study. It also indicates that Fe-C co-
15 limitation for heterotrophic bacterial growth is a predominant feature in the Southern
16 Ocean, but that it can be masked by conventional experimental approaches.

17 **5.2. A minimum Fe:C quota to support bacterial growth?**

18 In our study we explicitly examined the bacterial Fe uptake together with the BP. As
19 we observed there is a good agreement in the trends for each response variable,
20 bacterial growth and Fe uptake rates are likely to be related. However, the link
21 between growth and nutrient uptake is not straightforward in natural communities,
22 and variation in maximum growth rate and minimum cell quota can greatly
23 complicate this relationship. Here we propose a threshold value to reconcile these
24 two variables. Based on the correlation between cell-specific Fe uptake and BP, we
25 derived a minimum Fe:C quota for heterotrophic bacteria of 37 $\mu\text{molFe molC}^{-1}$ that
26 we propose as a threshold limit value to define Fe or C limitation (Suppl. Material
27 Figure 2). Based on this assumption, Fe is the primary limiting element for a cellular
28 quota that is below 37 $\mu\text{molFe molC}^{-1}$, and C is the primary limiting element for a
29 quota above to this value. To understand the boundaries of implication for this
30 threshold limit, it is important to consider how heterotrophic bacteria utilize Fe and
31 organic substrates to gain energy. However, the vastness of biogeochemical
32 gradients—both spatially and temporally—that govern the composition of microbial
33 communities, and the plethora of metabolic strategies among taxa (Hopkinson and

1 Barbeau 2011, Hogle et al.; Debeljak et al. 2019) require similar studies in other
2 ocean regions in order to investigate the spectrum of heterotrophic bacterial Fe:C
3 quotas.

4 **5.3. Carbon availability increases Fe demand: the starting point of** 5 **competition?**

6 When DOC is no longer a limiting resource, the competition for Fe between
7 autotrophic pico- and nanoplankton and heterotrophic bacteria negatively affect the
8 latter (Fourquez et al. 2015, this study). There were also intriguing results in the
9 outcomes of the experiment regarding phytoplankton. In the present study, the uptake
10 of Fe by phytoplankton was similar for the control and the +Fe treatment but
11 increased by nearly five times in the +Fe+C treatment. We have two explanations for
12 this intriguing finding:

13 *1) Phytoplankton had a higher Fe uptake rate in +Fe+C treatment because Fe*
14 *regenerated by bacteria became available*

15 Fe availability influences the growth and abundance of auto- and heterotrophic
16 microorganisms, and heterotrophic bacteria can modify its speciation by the
17 synthesis of organic ligands (Gerringa et al., 2008; Rue and Bruland, 1997). In this
18 context, heterotrophic bacteria could act either as competitors with phytoplankton
19 (Kirchman, 1994; Thingstad, 2000), or on the contrary, facilitate their assimilation of
20 Fe in maintaining Fe solubility within the ecological niche they share (Amin et al.,
21 2009; Hopkinson and Morel, 2009).

22 The high rates of BP in +Fe+C treatment could in part be due to remineralization of
23 Fe during the incubation. The regenerated Fe may become available for
24 phytoplankton. Since the DFe concentration was measured in the incubation bottles
25 at the beginning, after +36h and at the end of the incubation (120 hours, Suppl.
26 Material Figure 3), we were able to directly compare these values with total Fe
27 uptake by microorganisms. The DFe concentrations decreased over time during the
28 incubations (Suppl Material Figure 3). We used a simple approach to (1) verify that
29 remineralization occurred during the incubation and (2) to provide an estimate of the
30 Fe regeneration rate. The regeneration rate of Fe was estimated by subtracting the
31 amount of Fe consumed by the entire microbial community (phytoplankton and
32 bacteria) from the initial Fe concentration as follows:

1
$$DFe_{regenerated} = DFe_{measured} - DFe_{expected}$$

2
$$\text{With } DFe_{expected} = DFe_{initial} - [Total Fe_{uptake} \times t]$$

3 Where DFe initial is the concentration of DFe at the start of the experiment, Total Fe uptake is the
4 amount of Fe consumed by phytoplankton and bacteria during the incubation, and DFe measured is
5 the DFe concentration at the end of the incubation.

6 If we consider the amount of “missing” DFe as the regenerated Fe, we obtain rates of
7 Fe regeneration of 0.48 to 0.92 nmolFe L⁻¹ d⁻¹ (respectively unfiltered and <1µm
8 condition, Suppl. Material Table 1).

9 Iron regeneration within the microbial loop (also termed the “ferrous wheel”;
10 Kirchman, 1996) represents a key term in the Fe budget (Boyd and Ellwood, 2010;
11 Strzepek et al., 2005). For instance, in the SAZ (FeCycle voyage, see Strzepek et al.,
12 2005), it was found that between 30 to 100% of the microbial Fe demand could be
13 met by Fe regeneration mediated by grazers (Boyd et al., 2005). In the naturally Fe
14 fertilized waters off Kerguelen Island, Fe regeneration accounted for roughly 50% of
15 the Fe demand (Sarhou et al., 2008). Heterotrophic bacteria and viruses contribute as
16 much as grazers to Fe recycling (Obernosterer et al., 2008; Poorvin et al., 2004).
17 Many of the metabolites originating from microorganisms can possess Fe-binding
18 properties that can exert strong control on Fe speciation (Boyd et al., 2010; Dalbec
19 and Twining, 2009; Poorvin et al., 2004). Unlike larger cells of phytoplankton (e.g.
20 diatoms), pico- and nanoplankton are equally adept to heterotrophic bacteria at
21 accessing either new or regenerated Fe (Boyd et al. 2012). Ultimately, niche
22 differentiation of bacteria and phytoplankton related to Fe-speciation might act as a
23 selection process (Debeljak et al., 2019; Hogle et al., 2016; Hopkinson and Barbeau,
24 2012).

25 Over the course of a phytoplankton bloom there is a transition from the utilization of
26 new Fe (i.e., winter reserve Fe stocks) to regenerated Fe (Boyd et al., 2012) which
27 maintains primary productivity. During this transition, rapidly growing heterotrophic
28 bacteria may quickly shift to Fe limitation if phytoplankton-derived organic carbon is
29 available, resulting in their enhanced ability to compete for Fe. As the bloom status
30 moves towards senescence and cells exude DOC, competition between pico- and
31 nanoplankton and heterotrophic bacteria may result in different amounts of Fe
32 regenerated. Significantly, In the present study we calculated a Fe regeneration rate

1 nearly two-fold larger for the $<1\mu\text{m}$ ($0.92 \text{ nmolFe L}^{-1} \text{ d}^{-1}$) versus $<20\mu\text{m}$ (0.53
2 $\text{nmolFe L}^{-1} \text{ d}^{-1}$) or in unfiltered seawater ($0.48 \text{ nmolFe L}^{-1} \text{ d}^{-1}$). It is reported that
3 organic Fe complexes are available to few phytoplankton species (Kranzler et al.,
4 2011; Lis et al., 2015; Shaked and Lis, 2012). This could suggest that the bloom
5 duration is primarily set by DOC availability and competition for DFe between
6 heterotrophic bacteria and phytoplankton.

7 2) *Synechococcus like it organic*

8 Many studies make operational distinctions based on size-fractionated samples. In
9 the study of Strzeppek et al. 2005 (FeCycle), flow cytometric analyses revealed that
10 both picophytoplankton and eukaryotic phytoplankton were $>1\mu\text{m}$, and heterotrophic
11 bacteria were submicron in size. We used the same size cutoffs of $1\mu\text{m}$ to separate
12 phytoplankton and heterotrophic bacteria in this study. It remains, however, difficult
13 to draw clear distinctions between the size-fractionation and assigning
14 microorganisms to eukaryotes or prokaryotes. At SOTS, most pico- and
15 nanoplankton (70%) were cyanobacteria (most likely *Synechococcus*).
16 *Synechococcus* is one of the most prominent genera of picoplanktonic marine
17 cyanobacteria (Buitenhuis et al., 2012) that have particularly high Fe demands
18 relative to heterotrophic bacteria and eukaryotic phytoplankton (Lis et al., 2015;
19 Morrissey and Bowler, 2012; Raven, 1990). There is growing evidence of the ability
20 of the genera of *Synechococcus* to assimilate organic nutrients (Yelton et al., 2016),
21 and more broadly there are reports that suggest some of the photosynthetic
22 picoeukaryotes are mixotrophs (Farnelid et al., 2016). Cyanobacteria may be
23 responsible for an important part of the total Fe taken up by the phytoplankton
24 fraction ($>0.8\mu\text{m}$), and that due to their ability to take up organic compounds for
25 their metabolism they may also have benefited from the +Fe+C treatment, as did
26 their heterotrophic counterparts. In our experiment, cyanobacteria, but none of the
27 pico- and nanoeukaryotes had significantly increased in cell abundance in the +Fe+C
28 treatment (Suppl. Material Figure 4). This incubation was performed under very low
29 light intensities which suggests a complementary mechanism such as mixotrophy.
30 This raises the question of whether a mixotrophic capacity can become an
31 advantageous for these microorganisms, and if they could become also competitors
32 for C availability if this is proven.

1 **5.4. The fate and duration of phytoplankton blooms in the Southern Ocean** 2 **driven by interspecific relationships.**

3 Interactions between autotrophic and heterotrophic microbes could affect the
4 dynamics of nutrient-limited phytoplankton blooms. This hypothesis originates from
5 an investigation of what was initially perceived as an isolated event in the vicinity of
6 the Kerguelen plateau (Fourquez et al. 2015). As we reached similar conclusions in
7 the present study, these joint findings raise the issue of whether such interactions are
8 widespread across the Southern Ocean?

9 One stand-out result of our study is the amount of Fe taken up by heterotrophic
10 bacteria in the absence of competition. Our reported bacterial Fe uptake rates are
11 well beyond the range of those previously reported for Southern Ocean microbes. For
12 example, in comparison to the Fe uptake by the entire microbial community, the rate
13 of heterotrophic bacteria alone is by 12-fold higher than that measured during
14 FeCycle (HNLC waters southeast of New Zealand) and more than 100fold higher
15 than that measured above the Kerguelen plateau during summer (KEOPS2; Table 4).
16 The observation of the present study underlines the potential of heterotrophic
17 bacteria to control the decline of the bloom in the absence of competition with pico-
18 and nanoplankton.

19 It is only recently that the potential influence of the composition of phytoplankton
20 community and its interactions with heterotrophic microbes has been taken into
21 consideration (Bunse et al., 2016; Farnelid et al., 2016; Lima-Mendez et al., 2015;
22 Liu et al., 2019; Zhou et al., 2018). For instance, Liu et al. (2019) showed a
23 pronounced association between assemblages of diatoms and heterotrophic microbes
24 at the onset of spring phytoplankton blooms occurring in the region off Kerguelen
25 Island. The quality and quantity of DOC derived from phytoplankton exudates
26 (Landa et al., 2015) and resource competition for Fe (Fourquez et al. 2015) are the
27 two explanations put forward to explain how diatom assemblages shape the habitat
28 type for their heterotrophic counterparts (Liu et al. 2019).

29 Uncertainties remain on the effects of climate change on the composition of
30 phytoplankton assemblage; but there is compelling evidence that all regions of the
31 Southern Ocean will encounter changes in phytoplankton community composition
32 (Deppeler and Davidson, 2017; Hays et al., 2005). Models project that waters of the

1 Southern Ocean will become warmer, and that rising temperatures will cause rates of
2 grazing to increase more rapidly than rates of phytoplankton growth (Behrenfeld,
3 2014; Cael and Follows, 2016; Caron and Hutchins, 2012; Evans et al., 2011;
4 Sarmiento et al., 2010). Increasing temperature is also expected to increase bacterial
5 respiration rates (Vázquez-Domínguez et al., 2007). Thus, phytoplankton standing
6 stocks are likely to decline and the proportion of primary production respired in near-
7 surface waters by heterotrophic bacteria will increase (Cavan et al., 2019; Cavan and
8 Boyd, 2018; Deppeler and Davidson, 2017). The study of Cavan and Boyd (2018),
9 which have predicted an increase in POC-normalised respiration, estimates that the
10 biological pump efficiency (POC export scaled to primary production) would
11 decrease by $17 \pm 7\%$ (SE) by 2100 for the subantarctic site SOTS. Such reports of
12 increased rates in bacterial respiration are enzymatic reactions as the temperature
13 increases, but are also supported by the enhanced release DOC-derived from
14 phytoplankton at higher temperatures to support heterotrophic bacteria (Hutchins et
15 al., 2019).

16 For microbial ecologists, the existence of interactions between primary producers
17 and bacteria that shape the activity and the diversity of both partners is well
18 recognised (Amin et al., 2015), but the mechanisms of such interactions remain
19 mostly unknown. The term “interactive co-limitation” was first proposed by Bertrand
20 et al. (2015) to describe scenarios in which at least two limiting “nutrients cycle are
21 affected by one another through interactions among different microbial functional
22 groups” (Bertrand et al., 2015). The findings of our study appear to be an example of
23 interactive colimitation. Moreover, they expand on the emerging recognition that
24 interaction between microorganisms is an ecological trait to be considered in the
25 study of Fe and C biogeochemistry. As Fe sources — including the inputs, amounts
26 and the nature of Fe — are often considered to be the primary drivers of Southern
27 Ocean productivity, our study teaches us that Fe bioavailability for microorganisms
28 is not a simple matter of chemistry.

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Legends

Figure 1 Study site of R/V Investigator voyage 02. Dot represents location of the SOTS site where experiments were conducted (42°S, 47°S, 51°S, and 54°S at 141°E). SOTS was visited on two occasions on 19 (experiment 1) and 29 March 2016 (experiment 2). SAF-N, SAF-S stands for the northern and southern branches of the subantarctic front, and PF for polar front, respectively. Sea surface height was used to characterize fronts position (AVISO Satellite Data Altimetry). Sea surface Chl a composite image for March 2016 was derived from the Moderate Resolution Imaging Spectroradiometer aboard NASA's Aqua satellite and obtained from the ocean color data distribution site https://oceandata.sci.gsfc.nasa.gov/MODIS-Aqua/Mapped/Monthly/4km/chlor_a/.

Figure 2 Overview of the setup experiment. (1) and (2) respectively refer to Experiment 1 and Experiment 2 with water collected on the same site (SOTS).

Figure 3 Changes in bacterial cell abundance (left panel) and cell-specific bacterial heterotrophic production (right panel) for unamended (control) and amended treatments (+Fe, +C, and both +Fe+C) over time for the experiment 1. Treatments with an asterisk are significantly different from the control (one-way analysis of variance (ANOVA) and post hoc Tukey test; * $p < 0.05$, ** $p < 0.01$, $p < 0.001$).

Figure 4 Cell-specific heterotrophic production (a) and Fe uptake (b) by bacteria (0.2-0.8 μm fraction-size) for the three different treatments (control, +Fe, +Fe+C) and conditions (unfiltered, $< 20\mu\text{m}$, $< 1\mu\text{m}$). Values represent average number of the three biological replicates. Error bars represent the standard deviation of the three biological replicates. Treatments with an asterisk are significantly different from the unfiltered condition (one-way analysis of variance (ANOVA) and post hoc Tukey test; * $p < 0.05$, ** $p < 0.01$).

Figure 5 Contribution of the two size-fraction ($> 0.8\mu\text{m}$ and 0.2-0.8 μm) to Fe uptake in incubation for the three conditions (unfiltered, and prefiltered on $< 20\mu\text{m}$ or $< 1\mu\text{m}$ seawater) and for the three treatments (control (no addition), +Fe, +Fe+C). Bars and errors bars represent respectively the average and the standard deviation of biological triplicates for the Fe uptake measured following 36h (T1) of incubation in conditions described in the materials and methods section. sw, seawater.

Table 1 List of abbreviations used.

<i>Abbreviation</i>	<i>Explanation</i>
Unfiltered	Raw seawater. Microplankton, pico-nanoplankton and heterotrophic bacteria were incubated together.
<20µm	Seawater prefiltered on 20µm mesh-size. Microplankton was excluded from incubation. Pico-nanoplankton and heterotrophic bacteria were incubated together.
<1µm	Seawater prefiltered on 1µm mesh-size. Heterotrophic bacteria were incubated alone.

Table 2 Specific bacterial production, cell specific bacterial respiration and bacterial growth efficiency at the end on the incubation experiment. * Bacterial respiration rates have been converted from O₂ into C units using a RQ value of 1. Samples were prefiltered on 1µm and kept in the dark to measure specifically the consumption of O₂ by heterotrophic bacteria. Means values +/- SE of two biological replicates are given for specific BR and BGE. Means values +/- SD of three biological replicates are given for specific BP.

<i>Treatment</i>	<i>Specific BP</i> (<i>fmol C cell⁻¹ d⁻¹</i>)	<i>Specific BR</i> (<i>fmol C cell⁻¹ d⁻¹</i>)*	<i>BGE</i> (%)
Control	0.04 ± 0.02	0.39 ± 0.15	10 ± 0.1
+Fe	0.05 ± 0.01	1.63 ± 0.34	3.2 ± 1.5
+C	0.77 ± 0.21	0.66 ± 0.21	52 ± 15
+Fe+C	0.54 ± 0.10	0.40 ± 0.09	57 ± 10

Table 3 Iron uptake rates, carbon biomass, and C-normalized Fe uptake rates (Fe:C ratio) of photosynthetic cells (cyanobacteria plus pico- and nanoeukaryotes) and heterotrophic bacteria. Values represent the average \pm standard deviation of the three independent biological replicates for each treatment and condition after 36 hours of incubation.

Condition	Treatment	Fe uptake rate ($\mu\text{molFe L}^{-1} \text{D}^{-1}$)		C biomass ($\mu\text{molC}^{-1} \text{L}^{-1}$)		Fe:C ratio ($\mu\text{molFe molC}^{-1}$)	
		Phot. cells	H. bacteria	Phot. cells	H. bacteria	Photo. cells	H. bacteria
Unfiltered	control	103 \pm 5.4 [‡]	52 \pm 16	1.84 \pm 0.34	1.29 \pm 0.19	51 \pm 0.2 [‡]	41 \pm 15
	+Fe	79 \pm 9.3 [‡]	21 \pm 3.6	1.65 \pm 0.23	1.06 \pm 0.12	49 \pm 10 [‡]	20 \pm 4
	+Fe+C	393 \pm 23 [‡]	101 \pm 5.1	1.79 \pm 0.16	2.20 \pm 0.19	231 \pm 24 [‡]	46 \pm 3
<20 μM	control	114*	38*	1.44 \pm 0.90	1.54 \pm 0.90	59*	15*
	+Fe	60 \pm 10	15 \pm 2.1	1.86 \pm 0.04	2.34 \pm 0.09	32 \pm 6	6 \pm 1
	+Fe+C	383 \pm 47	181 \pm 21	1.56 \pm 0.28	2.37 \pm 0.67	250 \pm 50	83 \pm 34
<1 μM	control		89 \pm 65		2.10 \pm 0.86		39 \pm 20
	+Fe	n/a	40 \pm 3	n/a	2.62 \pm 0.16	n/a	15 \pm 2
	+Fe+C		1258 \pm 136		2.25 \pm 0.55		591 \pm 208

*only one replicate available

[‡] Calculation were done on the basis that photosynthetic cells for size fraction comprised between 0.8-20 μm represented 80% of Fe uptake (as it was measured they represented 80% of POC in unfiltered condition)

Table 4 Total Intracellular Fe uptake rates uptake in this study compared to published studies of natural and Fe-fertilized surface waters in the Southern Ocean.

<i>Experiment location (acronym)</i>	<i>Fe uptake rate (pmolFe L⁻¹ d⁻¹)</i>	<i>References</i>
SAZ waters - SOTS	96-156 847-1258*	This study – unfiltered control This study - <1µm +Fe+C
Kerguelen Plateau (KEOPS)	4.4–6.2	Sarthou et al. (2005)
Kerguelen Plateau (KEOPS2)	19–39.8	Fourquez et al. (2015)
South of Australia (SOIREE)	3.07–11.9	(Bowie et al., 2001)
Southeast of New Zealand (FeCycle)	26.2–101	Strzepek et al. (2005)

* Heterotrophic bacteria only

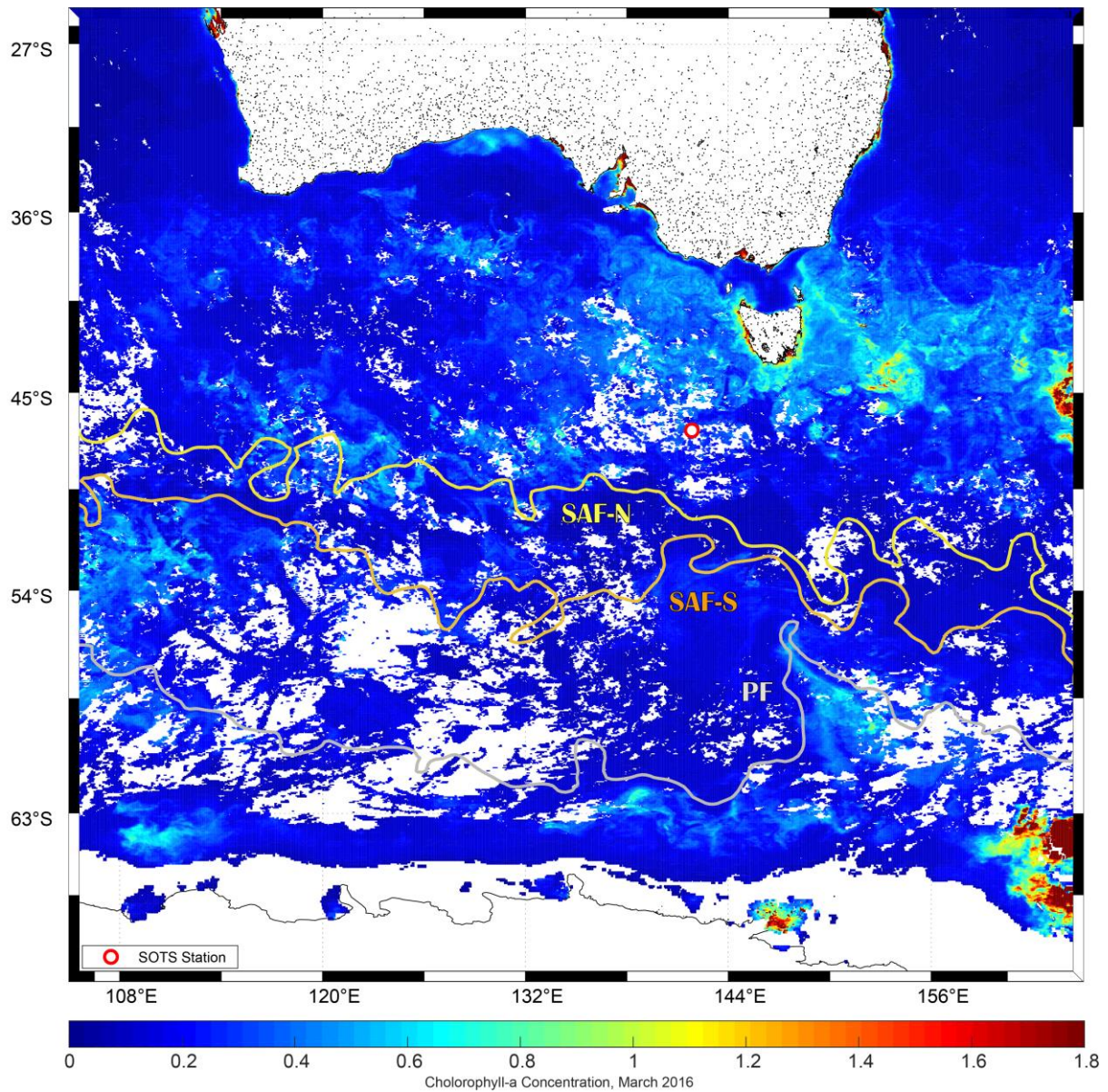


Figure 1

	Conditions	Treatments	Parameters
Experiment (1)	<div style="border: 1px solid black; padding: 2px; display: inline-block;"> Unfiltered </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;"> ● No addition ○ +Fe ○ +C □ +Fe+C </div>	<ul style="list-style-type: none"> • Heterotrophic bacterial production (1) + (2) • Bacterial cell abundance (1) + (2) • Bacterial respiration rate (1) • Bacterial Fe uptake rate (1+ 2) • Cell abundance of small photosynthetic cells (cyano, pico and nanoeukaryotes) (2)
Experiment (2)	<div style="border: 1px solid black; padding: 2px; display: inline-block;"> Unfiltered <20µm <1µm </div>	<div style="border: 1px solid black; padding: 2px; display: inline-block;"> ● No addition ○ +Fe □ +Fe+C </div>	<ul style="list-style-type: none"> • Fe uptake rate for microplankton (>20µm) and pico-nanoplankton (0.8-20µm) (2) • DFe concentration in incubation bottles (2)

Figure 2

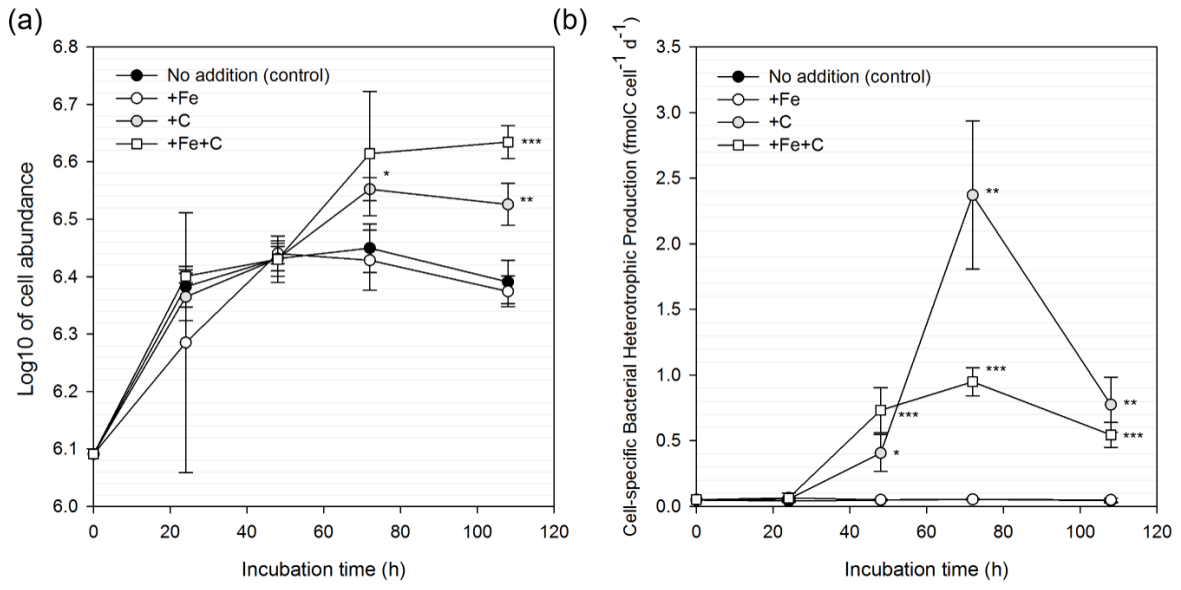


Figure 3

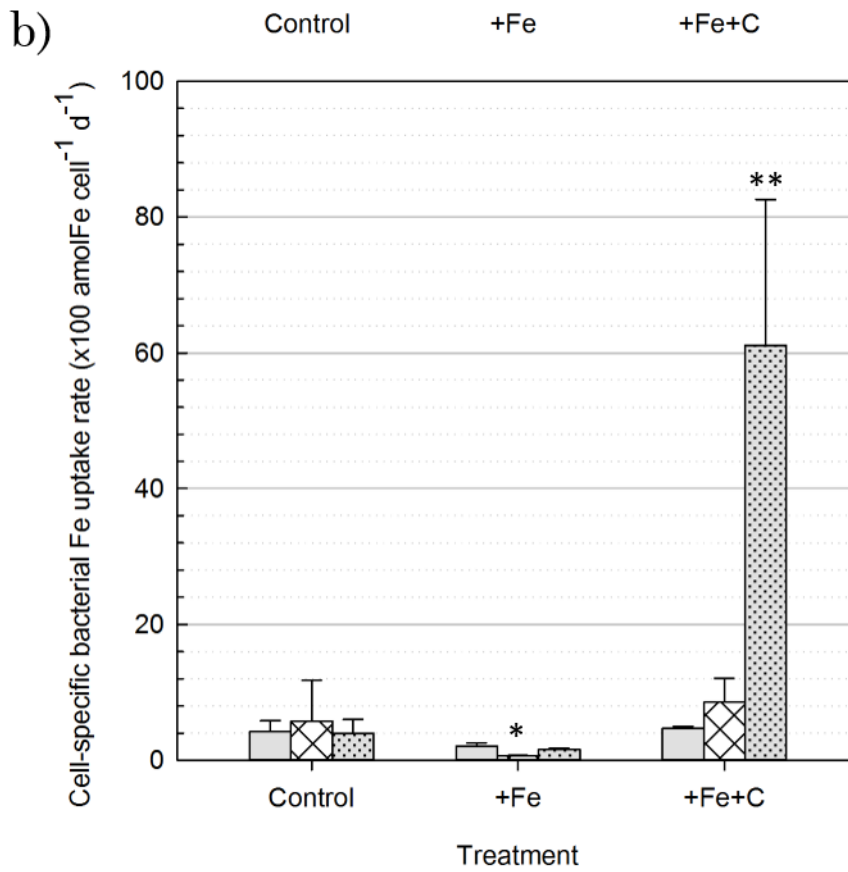
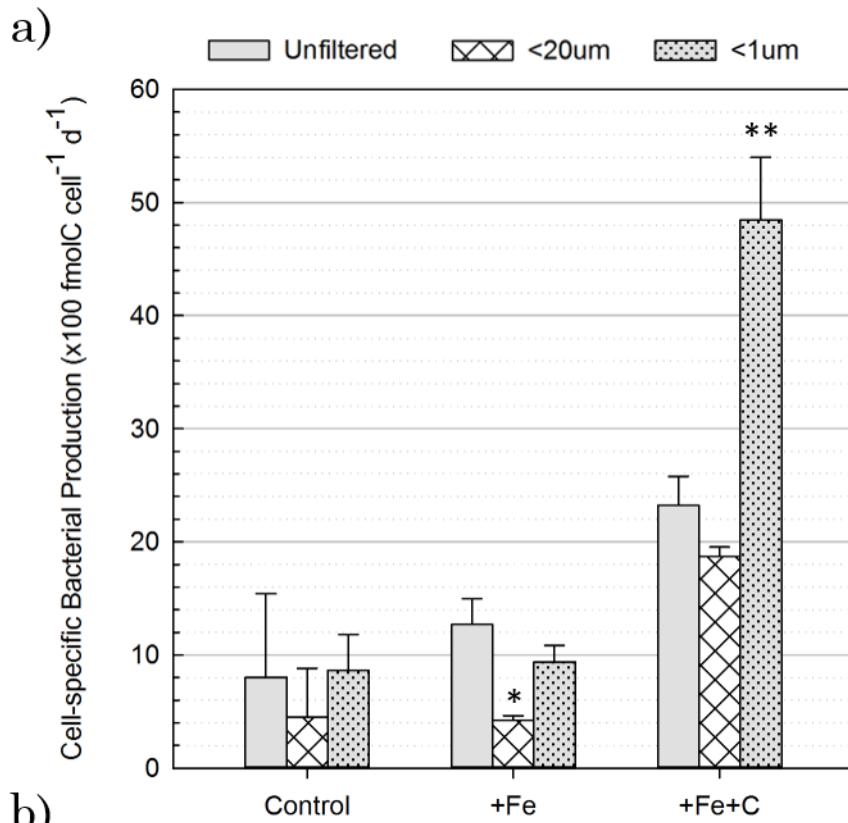


Figure 4

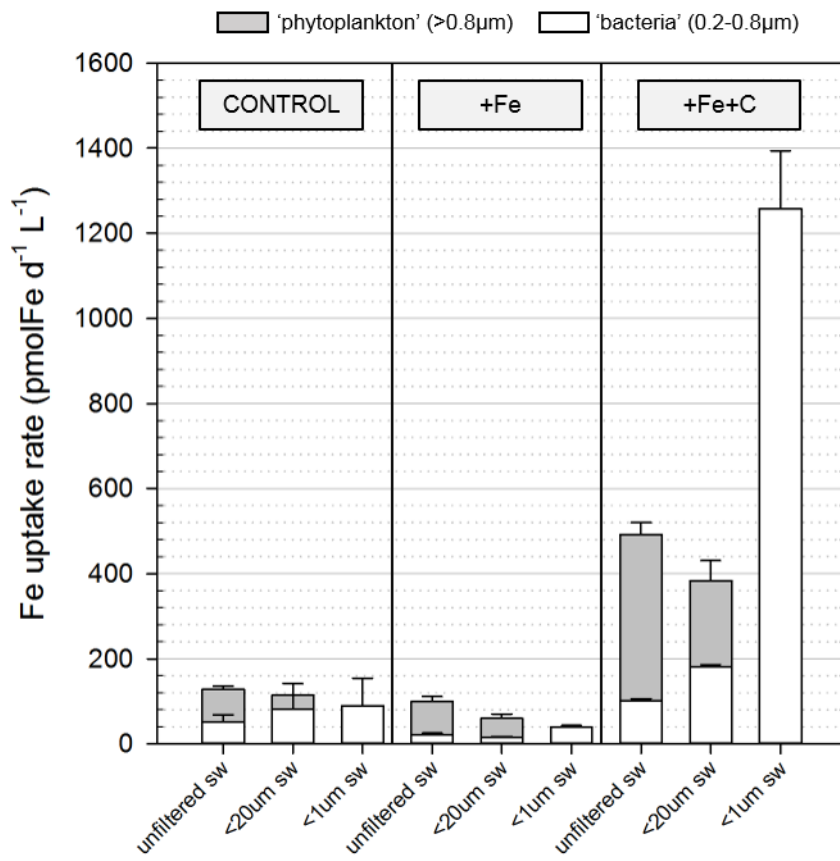
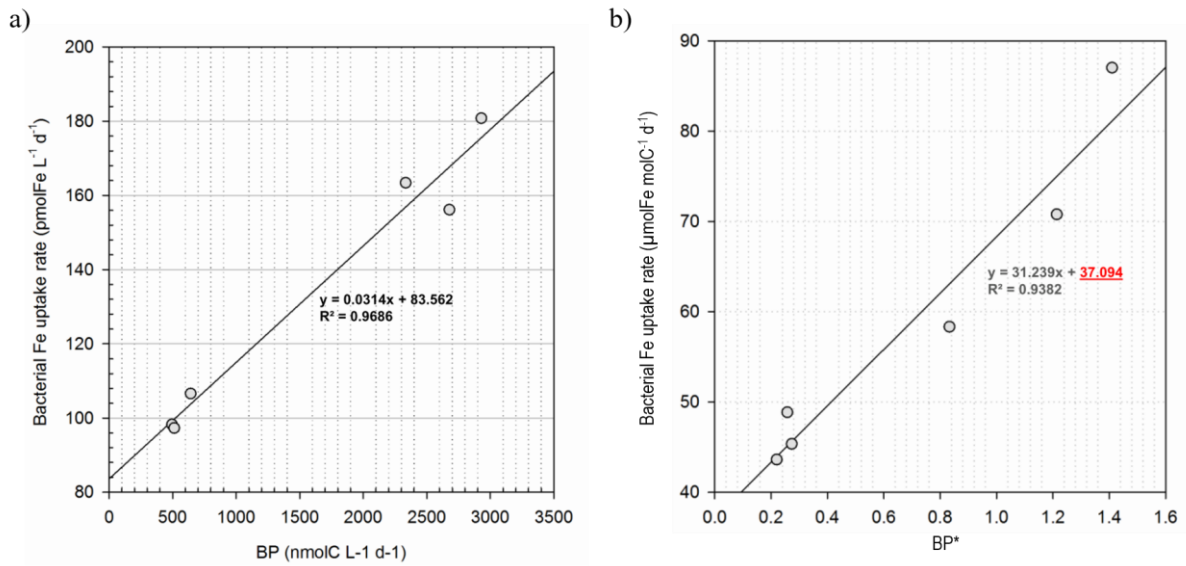
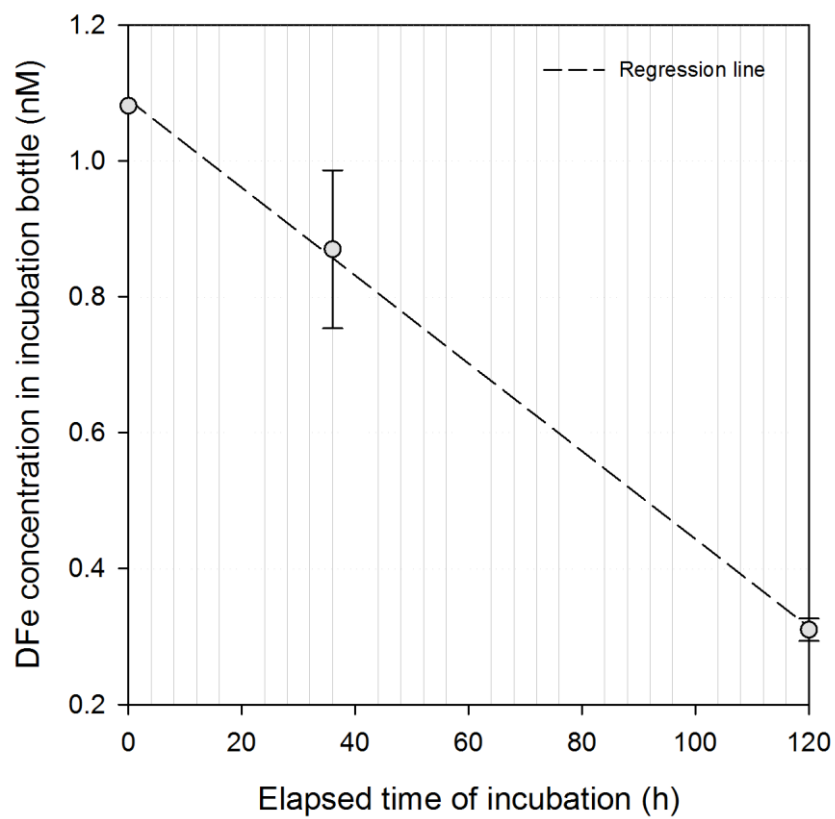


Figure 5



Supplementary Fig 1



Supplementary Fig 2