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Microstructure-based discrete simulations of the compaction of refractory powder composites

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Abstract

This work describes an original methodology to simulate the compaction behavior of carbon-bonded alumina refractories ($\text{Al}_2\text{O}_3\text{-C}$) by taking explicitly into account the microstructure of each composite phase. The Discrete Element Method (DEM) is used to model the composite as a mixture of fine and coarse alumina particles and graphite flakes. The binder phase, which typically represents 20% of the volume fraction, is accounted for by a specific contact law which adjusts the contact rigidity according to the indentation between the coated particles. Small indentations lead to a soft response while above a critical indentation, the contact becomes much stiffer when alumina particles are in direct contact. This simple model incorporates the granular behavior of the composite as well as the continuous matrix of the binder. Numerical composites with various compositions are numerically submitted to closed-die compaction. We show that the densification behavior during compaction depends primarily on the binder phase for the volume fractions studied here. The contribution of each phase (alumina, graphite, ...) to the total axial stress is investigated, thus providing new insights into the complex behaviour of these composites.

Keywords: discrete element method, powder compaction, refractory composites, $\text{Al}_2\text{O}_3\text{-C}$ refractories, powder densification behavior, closed-die compaction

1. Introduction

Carbon-bonded alumina refractories ($\text{Al}_2\text{O}_3\text{-C}$) are heterogeneous and complex granular composites made of coarse alumina, graphite flakes, and a bonding matrix (mixture of fine

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4 alumina grains and binder). These composites are used for steel casting applications, e.g., for
5 stoppers, ladle shrouds, or submerged nozzles. These refractory parts are designed to control
6 the flow and velocity of the liquid steel and to prevent oxidation of the steel. The final
7 refractory properties are governed both by the properties of the different composite phases
8 and by the processing stages (mixing, compaction, and firing). Depending on the final shape
9 of the industrial part, the powder composite may be compacted isostatically or uniaxially.
10 The compaction may lead to density gradients along the part geometry resulting in thermo-
11 mechanical properties mismatch. These mechanical property heterogeneities may induce
12 stress concentrations, which, associated with extreme loading conditions, can lead to failure.
13 Hence, understanding the link between the composite microstructure, its evolution during
14 processing and the final thermo-mechanical properties is important for the development of
15 new products and the improvement of the industrial process.

16 Because the powder compaction step is central to the forming of refractory composite
17 materials, it is paramount to understand its effect on their microstructure evolution. In this
18 context, compaction has been investigated in the literature both experimentally and numer-
19 ically. Some experimental studies, for example, focused on investigating the densification
20 behavior of a composite mixture composed of hard and soft particles during closed-die or
21 isostatic compaction [1] [2] [3]. From a numerical point of view, the compaction stage has
22 been studied mostly by the Finite Element Method (FEM) which treats the powder as a
23 continuum and uses appropriate constitutive equations. The powder parameters applied in
24 these equations need to be identified through a complex experimental characterization both
25 for ceramic powders [4, 5], powders used for the pharmaceutical industry [6, 7], metal powders
26 [8, 9], and refractory composites [10].

27 An alternative method to model the compaction behavior of powders is the Discrete
28 Element Method (DEM), which explicitly considers the powder as a collection of discrete
29 particles. The DEM may be used more effectively than FEM to clarify the link between
30 the particulate microstructure and the behavior of the powder prior to and post-compaction.
31 This approach of modeling the behavior of granular matter was pioneered by Cundall and
32 Strack in 1979 [11] for geomaterials and has spread to engineering powder materials since
33 then. Employing DEM [12, 13], or the multi-particle finite element method (MPFEM) [14,

34 15, 16, 17] on model powder materials, the compaction stage has been investigated. Some
35 authors also studied the compaction behavior of model powder materials containing different
36 particle sizes [18, 19] and particle size distributions [20]. In particular, the DEM has proved
37 useful to understand the compaction of composite powders with soft (typically metallic) and
38 hard (typically ceramic) particles mixed together [18].

39 This paper is mainly focused on the use of DEM simulations for modeling the compaction
40 behavior of carbon-bonded alumina composites that are considered as model refractory ma-
41 terials, used in the steel casting process. These composites are made of hard (alumina and
42 graphite) and soft (binder) phases and may be shaped via two routes: isostatic or closed-die
43 compaction, depending on the final application.

44 The major part of the numerical and experimental studies concerning the powder com-
45 paction is related to composites made of metal-ceramic materials for the powder-metallurgy
46 process. DEM simulations of the compaction of homogeneous refractory materials (alumina,
47 magnesia) have been reported with various sizes of particles [20, 21]. However, as far as
48 we know, no study has been reported yet in the literature regarding compaction simulations
49 using DEM to model refractory composites on a mixture of hard particles and a soft binder.
50 Especially, a direct link to the real microstructure is still missing in these works. The aim
51 of this paper is to describe a methodology, based on DEM simulations, to account for the
52 composite microstructure.

53 The present article is organized as follows. First, we present some essential information
54 on the typical composite microstructure to be simulated. Second, the main characteristics
55 of the DEM model, and in particular the contact laws are described. The composite in
56 DEM is represented by a mixture of hard particle clusters (coarse alumina), single particles
57 (fine alumina) and bonded particles (graphite). An original approach is presented to account
58 for the binder as an elastic soft-shell covering alumina particles. The capability of this
59 modeling approach to capture the densification behavior of complex refractory composites
60 is highlighted by analysing the simulations at the scale of each phase. Experimental data
61 on closed-die compaction are used throughout the paper for critical comparison with the
62 simulation results.

63 2. Carbon-bonded alumina composites

64 A schematic representation of the composite refractory microstructure studied in this work
65 is shown in Fig. 1 a). It can be thought of as an overlapping hierarchy, with a mesoscale made
66 of coarse hard grains (white fused alumina with 250 μm average size) and graphite (<400 μm)
67 reinforcing a matrix; itself constituted at microscale of fine grains (calcined alumina <50 μm)
68 embedded into a pyrolytic carbonaceous binder (a mixture of resin and solvent).

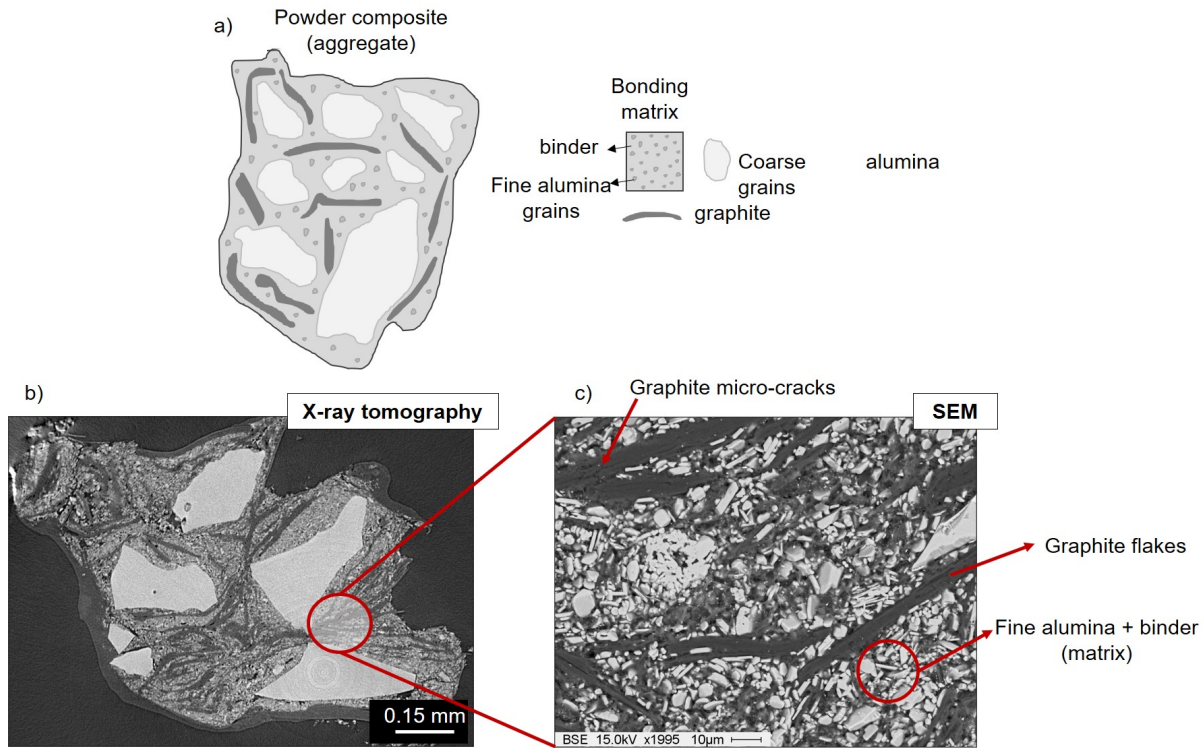


Figure 1: a) Schematic representation of a carbon-bonded alumina composite aggregate, composed of coarse grains and graphite flakes embedded into a continuum matrix. b) 2D slice from a 3D fragment image accessed through X-ray tomography analysis. c) Scanning electron microscope (SEM) image.

69 X-ray computed tomography analysis has been performed on the green mix aggregate
70 sample to examine the overall microstructure arrangement after compaction (Fig. 1 b). The
71 experiment consists of taking X-ray images of a sample tilted by a small angle for each image.
72 The whole volume of the imaged specimen is re-constructed by assembling the digital images.
73 The resulting 3D image is a superimposed projection of a volume in a 2D plane [22]. The
74 observed aggregate has 1.2 mm size and the analysis was performed applying a resolution
75 of 1.0 μm voxel size and 30 keV energy spectrum. Note that the gray levels in a CT slice

76 image correspond to the X-ray attenuation, i.e., reflecting the proportion of X-rays scattered
77 or absorbed as they pass through each voxel. The X-ray attenuation is highly dependent on
78 the X-ray energy, material density, and composition. Typically, alumina grains are denser
79 and absorb more photons by the beam than porosity and graphite.

80 The geometry of coarse grains is dispersed in shape but generally angular, whereas the
81 graphite flakes are mostly planar. The coarse alumina grains are visualized in different
82 dimensions (average size of 250 μm). The graphite flakes exhibit some deformation due to
83 the compaction stage. Three main phases could be distinguished: graphite, a bonding matrix
84 (fine alumina and binder), and coarse alumina.

85 SEM (scanning electron microscope) analysis is also carried out on the final product
86 (fired), displayed in Fig. 1 c). The image highlights mainly the matrix, appearing in grey at
87 low magnification, which is composed of fine calcined alumina grains and a pyrolyzed binder.
88 Fine alumina grains are also dispersed in shape, but some exhibit rounded shapes. Some
89 graphite micro-cracks can be observed that are a consequence of the compaction and of the
90 firing itself. However, the alumina grains remain unaffected.

91 Four model composites are investigated with different compositions (Table 1) to reflect
92 the typical microstructure of industrial composites. Hereafter, the term matrix refers to the
93 group (fine alumina + binder). Mix R is considered as the reference composite. The matrix
94 of Mix 2 is characterized by a lack of binder (and excess of fine alumina grains). Mix 3 has an
95 overall depleted matrix (40 % on Mix 3 against 45 % volume fraction on Mix R, Mix 2, and
96 Mix 4) and an excess of coarse alumina grains and graphite. Mix 4, on the opposite exhibits
97 a matrix with an excess of binder and lack of fine alumina grains. The objective in Mix 2
98 and Mix 3 is to generate a model composite with an excess of hard solid grains. In contrast,
99 Mix 4 presents an excess of soft binder. Mix R can be considered as an intermediate to the
100 other mixes.

101 **3. Modeling particulate and continuous materials with DEM**

102 An in-house DEM code dp3D, dedicated to engineering materials for modelling the com-
103 paction of refractory composites, is used. Spherical discrete elements are generated to model
104 each phase of the composite with specific contact laws that describe the mechanical interac-

Refractory	Coarse alumina	Fine alumina	Graphite	Binder
Mix R (reference)	33	22.5	22	22.5
Mix 2	33	27.5	22	17.5
Mix 3	36	20	24	20
Mix 4	33	20	22	25

Table 1: Model refractory compositions in volume fraction %.

105 tions between them. As classically implemented in DEM, the spherical particles are displaced
106 in the first half time-step according to the imposed increment in strain following the affine
107 solution (homogeneous deformation of the sample). Contact forces are used to compute the
108 total force acting on each particle. Newton’s second law of motion enables the computation
109 of the acceleration and an explicit time-integration scheme is used (velocity-Verlet) to obtain
110 the new position of particles before a new affine strain increment is imposed. New contacts
111 and lost contacts are updated at each time-step.

112 The graphite flakes and coarse alumina are represented as particle clusters (particle as-
113 semblies linked by elastic bonds). The fine alumina particles are modeled as single spheres
114 with no bonds to represent a granular macroscopic behavior, characteristic of green materials
115 (Fig. 2). The clusters (coarse alumina and graphite) are considered fully dense (relative
116 density $RD=1$). This was attained simply by imposing an initial indentation between inter-
117 nal cluster particles such that the macroscopic relative density of the cluster is unity. This
118 indentation is of the order of $0.29R$, where R is the sphere radius.

119 The graphite flakes present an elasto-plastic behavior, and may deform during compaction
120 (particle bonds may deform and break, thus bringing some plasticity at the microscopic and
121 macroscopic scales). The bonds in the coarse alumina clusters cannot fracture, contrary to
122 the graphite. The bond interactions inside the coarse alumina particles are not calculated.
123 In other words, spheres inside large alumina particles serve only to mesh them. Only spheres
124 at the surface of these large alumina particles can interact mechanically with other particles.
125 This type of cluster is generally referred as clump in the literature [23, 24].

126 The external and internal interaction contact models are summarized in Table 2 and
127 schematically described in Fig. 3. The contact models (Hertzian and Bond) are described in

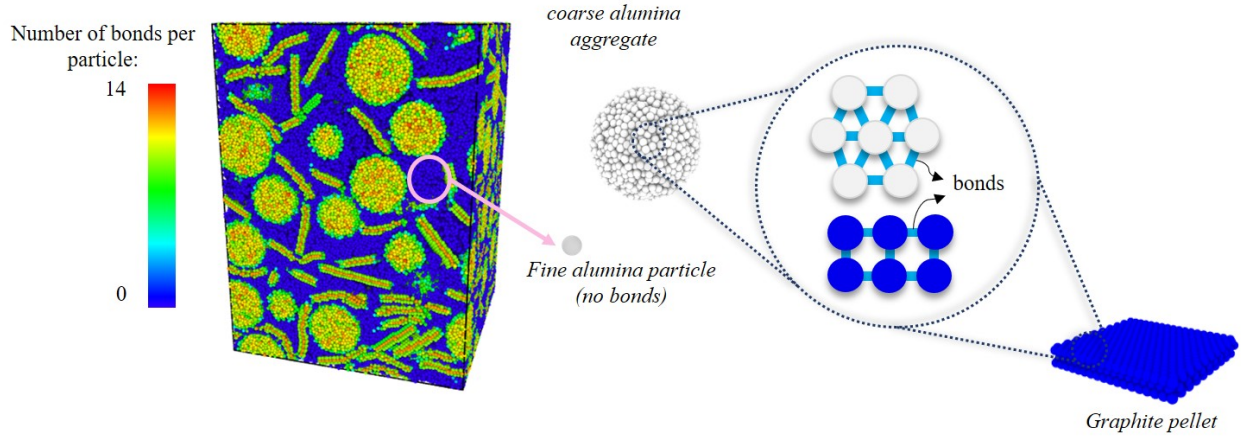


Figure 2: Typical discrete composite after compaction: from blue (no bonds) to red (with bonds). Coarse alumina and graphite are considered as an assembly of particles linked by bonds (cluster), and fine alumina as single non-bonded particles. Note how graphite flakes may deform while coarse alumina particles cannot.

128 the next section.

Composite phase	DEM model	External interactions	Internal interactions
Fine alumina	Single particle	Hertz (Eq. (1))	-
Coarse alumina	Particles cluster	Hertz (Eq. (1))	None
Graphite	Particles cluster	Hertz (Eq. (1))	Bond model (Eqs. (7), (8))

Table 2: Summary of the applied DEM model for each coarse alumina, fine alumina, and graphite phases (if single particle or particle cluster), and the external and internal interaction contact models.

129 3.1. Contact models

130 The Hertzian model is used to compute the normal force at the contact between two elas-
 131 tic particles, together with the Derjaguin Muller and Toporov (DMT) adhesion theory [25],
 132 which adds a tensile force to the standard repulsive Hertzian force. The normal force acting
 133 between two non-bonded particles with radii r_i and r_j , and elastic properties (E_i, ν_i) and
 134 (E_j, ν_j) , respectively, is given by an Hertzian repulsive component (N^{Hertz}) and an adhesive
 135 tensile component (N^{DMT}) (compressive forces are considered as positive) [25, 26]:

136

$$N_e = N^{Hertz} + N^{DMT} = \frac{4}{3}E^* R^{*1/2} \delta_n^{3/2} - 2\pi\omega R^* \quad (1)$$

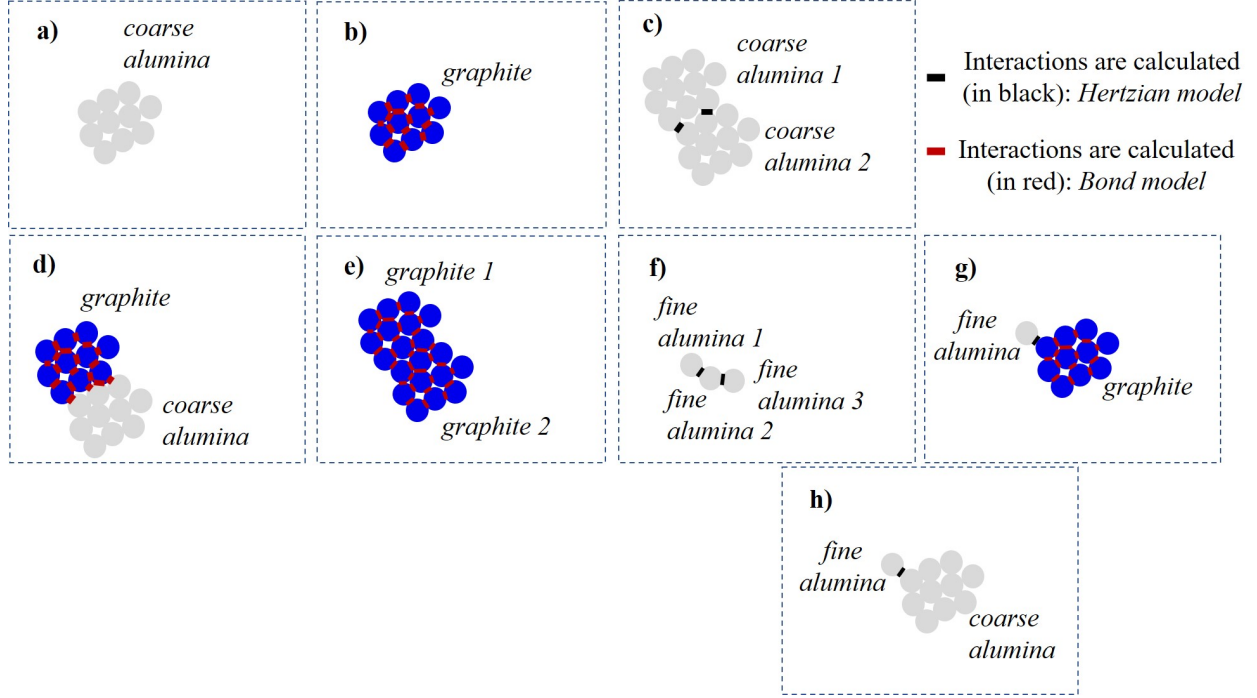


Figure 3: Schematic representation of the different type of contacts between the different discrete phases (fine alumina, coarse alumina and graphite).

137 where the effective Young's modulus between is:

$$E^* = \left(\frac{1 - \nu_i^2}{E_i} + \frac{1 - \nu_j^2}{E_j} \right)^{-1} \quad (2)$$

138 the equivalent radius R^* is:

$$R^* = \left(\frac{1}{r_i} + \frac{1}{r_j} \right)^{-1} \quad (3)$$

139 and where δ_n is the normal indentation, and $\omega = 2\gamma$ is the work of adhesion (with γ the
140 surface energy). Decohesion occurs in the DMT model for a pull-off force $N^{DMT} = 2\pi\omega R^*$.

141 The contact radius a is given by:

$$a^2 = R^* \delta_n \quad (4)$$

142 Contacts may also transmit frictional force (Hertz–Mindlin model) in the sticking mode
143 while the tangential force is limited during sliding by Coulomb friction (friction coefficient
144 μ). The friction law is implemented in incremental form at each time step dt by the friction
145 force vector \mathbf{T} :

$$d\mathbf{T} = -8G^* a \mathbf{v}_t dt \quad |\mathbf{T}| < \mu N^{Hertz} \quad (5)$$

146

$$\mathbf{T} \leftarrow \frac{\mathbf{T}}{|\mathbf{T}|} \mu N^{Hertz} \quad |\mathbf{T}| \geq \mu N^{Hertz} \quad (6)$$

147 where \mathbf{v}_t is the tangential relative velocity at the contact and G^* is the equivalent shear
 148 modulus. Note that the condition for stick or slip ($|\mathbf{T}| < \mu N^{Hertz}$) applies on the repulsive
 149 part of the normal force (N^{Hertz}) and not on the total force N_e . In the slip condition, the
 150 vector $|\mathbf{T}|$ is capped by μN^{Hertz} but its direction is incrementally modified (Eq. (6)).

151 For bonded particles, the contact law is given by the analysis of [27] who studied the elastic
 152 response of bonded contact by the Finite Element Method (FEM). Two spherical particles
 153 (of the same material), connected to each other through a bond of radius a_b , transmit normal
 154 and tangential forces, and resisting moments. The normal contact force (N_b) for a bonded
 155 contact is given by:

$$N_b = \frac{2ER^*}{1 - \nu^2} f_N(a^*, \Psi) a^* u_N \quad (7)$$

156 where E and ν are the Young's modulus and Poisson's ratio of the particles, $f_N(a^*, \Psi)$ is
 157 a function that depends on the relative radius of the bond ($a^* = \frac{a_b}{2R^*}$), Ψ is a geometric
 158 factor which allows for bond interaction to be considered, and u_N is the accumulated normal
 159 displacement at the contact [28]. The normal force can be either in compression ($u_N > 0$) or
 160 in tension ($u_N < 0$). The tangential contact force (T_b) is written as:

$$\mathbf{T}_b = \frac{4ER^*}{(2 - \nu)(1 + \nu)} f_T(a^*) a^* \mathbf{u}_T \quad (8)$$

161 where the function $f_T(a^*)$ depends on the relative bond size a^* , and, \mathbf{u}_T is the accumulated
 162 tangential displacement vector at the contact. f_N and f_T values vary typically in between 1
 163 and 2. More details concerning the form of $f_N(a^*, \Psi)$ and $f_T(a^*)$ used in this work can be
 164 found in earlier works [27, 28, 29, 30].

165 The maximum stress in tension (σ_N) and in shear (σ_T) can be evaluated by using the
 166 Potyondy's model [31]:

$$\sigma_N = \frac{N_b}{\pi a_b^2} - 4 \frac{|M_T|}{\pi a_b^3} \quad (9)$$

167

$$\sigma_T = \frac{T_b}{\pi a_b^2} + 2 \frac{|M_N|}{\pi a_b^3} \quad (10)$$

168 where M_N and M_T are the bonds resisting moments in the normal and tangential directions,

169 respectively:

$$M_N = -\frac{8ER^*{}^3}{(2-\nu)(1+\nu)}f_T(a^*)\theta^N \quad (11)$$

170

$$M_T = -\frac{2ER^*{}^3}{(1-\nu^2)}f_N(a^*, \Psi)\theta^T \quad (12)$$

171 where θ^N and θ^T are the accumulated relative rotations in the normal and tangential direc-
 172 tions, respectively.

173 Regarding the failure criteria applied in our work at a bonded contact, Eqs. (9) and
 174 (10) give at each time step an evaluation of the stress in tension and shear acting on the
 175 bond. If one of these stresses is above the tension strength (Σ_N) or shear strength (Σ_T), the
 176 bond is considered as broken. A fractured bond may transmit a shear force according to a
 177 Hertz-Mindlin friction law (Eq. (6)). Correspondingly, a fractured bond in shear continues
 178 to transmit a resisting moment in the tangential direction but none in the normal direction.

179 3.2. Binder shell model

180 The contact laws described above are classic laws for elastic and bonded interactions.
 181 The very soft binder phase that coats hard particles (fine alumina particles or coarse alumina
 182 particles) introduces a different issue. We simply take the binder into account by stating
 183 that the stiffness of a contact depends on the mutual indentation of particles. For a small
 184 enough indentation, the contact stiffness is dictated by the binder (soft phase) whereas for
 185 an indentation larger than a critical value (linked to the binder thickness on the particles),
 186 the stiffness is dictated by the hard alumina (hard phase). Note that we have observed that
 187 graphite flakes are more difficult to be coated properly with binder. Thus, the binder model
 188 applies only to fine and coarse alumina particles. Still, contacts between an alumina particle
 189 and a graphite flake are affected by the binder but with a smaller thickness.

190 Consider two hard particles with Young's moduli (E_i, E_j) coated by a binder shell with
 191 thickness t_i and t_j . The total radii of the particles (with the binder) are r_i and r_j . We note
 192 E_{binder} , the Young's modulus of the binder. In this model, the effective Young's modulus
 193 (Eq. (2)) used in Eq. (1) is simply multiplied by a factor $\frac{E_{alumina}}{E_{binder}}$ when entering the hard
 194 phase branch. The transition from the soft phase to the hard phase depends on the mutual
 195 indentation and on the thickness of the binder.

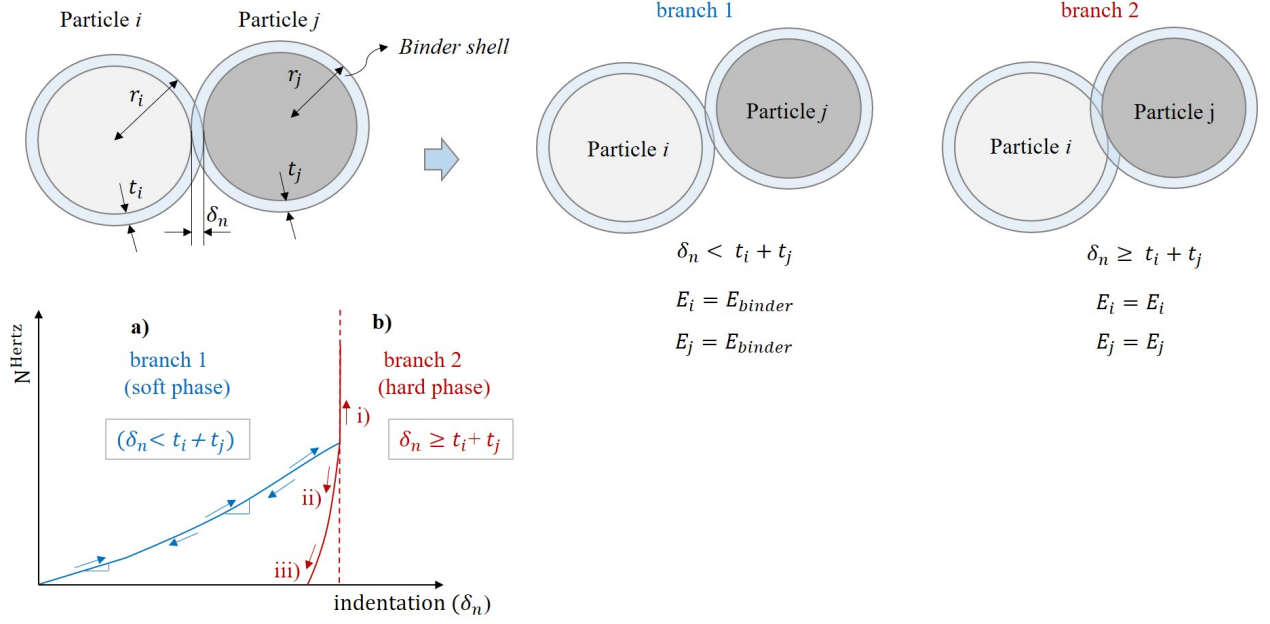


Figure 4: Schematic representation of a contact between two hard particles coated with a binder of thickness t_i and t_j . a) the contact remains within the soft elastic binder phase (branch 1). b) when the indentation reached a critical value i.e., condition $\delta_n \geq t_i + t_j$ applies, the elastic modulus of particles i and j are multiplied by a factor α and the contact enters branch 2. Once in branch 2, the contact may i) continue to load in compression, ii) unload, or iii) fail (if $\delta_n \leq (t_i + t_j) (1 - \frac{1}{\alpha})^{\frac{2}{3}}$).

196 Fig. 4 graphically summarizes the binder model in the case of two coated particles (fine
 197 or coarse alumina). In branch 1 (soft phase) and in the general case of two coated (alumina)
 198 or non-coated (graphite) particles:

$$N^{Hertz} = \frac{4}{3} \left(\frac{1 - \nu_i^2}{\alpha_i E_i} + \frac{1 - \nu_j^2}{\alpha_j E_j} \right)^{-1} R^{*1/2} \delta_n^{3/2} \quad 0 < \delta_n \leq t_i + t_j \quad (13)$$

199 with $\alpha = \frac{E_{binder}}{E_{alumina}}$ for alumina and $\alpha = 1$ for graphite. If the contact has not entered previously
 200 branch 2 (hard phase), any unloading in branch 1 is fully reversible. If the indentation reaches
 201 the critical indentation $t_i + t_j$:

$$N^{Hertz} = \frac{4}{3} \left(\frac{1 - \nu_i^2}{E_i} + \frac{1 - \nu_j^2}{E_j} \right)^{-1} R^{*1/2} \delta_n^{3/2} \quad \delta_n > t_i + t_j \quad (14)$$

202 Two coated alumina particles interact directly through the hard alumina phase in branch
 203 2. Note that in our case the hard phase stiffness is much larger than the binder's stiffness.
 204 Thus, the indention increase is very small in branch 2.

205 If the contact unloads from branch 2, (Fig. 4 ii), the elastic properties in eq. (14) are kept
 206 constant. This is to mimic the extrusion of the binder at the contact between the two hard
 207 particles. The contact may fail (Fig. 4 iii) if the condition: $\delta_n \leq (t_i + t_j) \left(1 - \frac{1}{\alpha}\right)^{\frac{2}{3}}$ applies.
 208 In that case no normal force is transmitted at the contact.

209 Hysteresis is introduced in the model since a contact that has entered branch 2 will not
 210 unload reversibly through branch 1. It should be clear that this model is a very simplistic
 211 idealization of the complex behaviour of hard particles coated with a soft binder. In partic-
 212 ular, the binder plasticity (or viscoplasticity) is oversimplified. Still, our model allows for a
 213 simple representation of particles that behave with a small stiffness when slightly indented
 214 (beginning of the compaction) and exhibit a much larger stiffness when indented above a
 215 critical value (end of the compaction).

216 As stated above, the binder model applies only to fine and coarse alumina particles. For
 217 contacts between alumina particles and graphite flakes, the total binder thickness ($t_i + t_j$) in
 218 Eqs. (13) and (14) is solely given by the alumina binder thickness. Thus, contacts between
 219 alumina and graphite particles are affected by the binder, but with a lesser thickness, which
 220 models very crudely the poorer coating of the graphite flakes.

221 3.3. Binder thickness calculation

222 The binder shell thickness of each composite (Mix R to 4) can be related to its composition,
 223 calculated from the volume fraction of each component (coarse and fine alumina, graphite,
 224 and binder). We use subscripts *ca*, *fa* and *g* for *coarse alumina*, *fine alumina*, and *graphite*,
 225 respectively.

226 Note that all spherical discrete elements have the same diameter (7.8 μm), which corre-
 227 sponds to the mean size of the fine alumina particles. This allows for more efficient CPU
 228 processing of contact detection in the DEM code. The coarse alumina cluster (made approxi-
 229 mately of 4,600 spherical particles) is about 135 μm in size. For simplicity, it is approximated
 230 to be a sphere-like cluster. The graphite flake (made of 1,070 spherical particles) is repre-
 231 sented by a parallelepiped with dimensions of 121 $\mu\text{m} \times 116 \mu\text{m} \times 17 \mu\text{m}$.

232 The binder volume fraction $V_{f(binder)}$ is:

$$V_{f(binder)} = (V_{binder(ca)} + V_{binder(fa)}) / (V_{ca} + V_{fa} + V_g) \quad (15)$$

233 where $V_{binder(ca)}$, $V_{binder(fa)}$ are the total binder volumes surrounding the coarse alumina clus-
 234 ters and fine alumina particles, and V_{ca} , V_{fa} , and V_g are the coarse, fine alumina, and, graphite
 235 volumes, respectively. We assume that the binder thickness on coarse alumina and fine alu-
 236 mina particles is the same, and hereafter noted t . $V_{binder(ca)}$ and $V_{binder(fa)}$ can be simply
 237 computed from the number of coarse and fine alumina particles introduced in the simulation
 238 (n_{ca} , n_{fa}) and their radii (r_{ca} , r_{fa}):

$$V_{binder(ca)} = n_{ca} \frac{4}{3} \pi (r_{ca}^3 - (r_{ca} - t)^3) \quad (16)$$

239

$$V_{binder(fa)} = n_{fa} \frac{4}{3} \pi (r_{fa}^3 - (r_{fa} - t)^3) \quad (17)$$

240 Table 3 lists the typical numbers of discrete elements and particles introduced in simulations
 241 once the number of coarse alumina clusters is fixed for each mix. The total number of discrete
 242 elements is characteristic of the Representative Volume Element chosen for our simulations
 243 (see section 5).

Mix	phase	binder thickness (μm)	number of particles	total number of discrete elements
Mix R	Coarse alumina	0.774	8	117,605
	Fine alumina	0.774	53,695	
	Graphite	-	25	
Mix 2	Coarse alumina	0.569	8	118,499
	Fine alumina	0.569	54,589	
	Graphite	-	25	
Mix 3	Coarse alumina	0.893	8	117,109
	Fine alumina	0.893	53,199	
	Graphite	-	25	
Mix 4	Coarse alumina	0.771	10	132,812
	Fine alumina	0.771	53,192	
	Graphite	-	31	

Table 3: DEM main parameters for each composition.

244 *3.4. Macroscopic stress calculation*

245 The macroscopic stress tensor at the scale of the entire packing is computed by using
 246 the total contact force at each contact (normal and tangential components) and the branch
 247 vector connecting the centers of the two particles p and q in contact ($\mathbf{l}_{p,q}$). The macroscopic
 248 stress tensor is calculated from Love's formulation [32] [33]:

$$\sigma_{ij} = \frac{1}{V} \sum_{contacts} (N_i + T_i) l_{pq,j} \quad (18)$$

249 where the summation is carried out on all contacts transmitting forces in the packing and
 250 where V is the sample volume, N_i and T_i are the i^{th} components of the normal and tangential
 251 contact forces at the contact, respectively. $l_{pq,j}$ is the j^{th} component of the \mathbf{l}_{pq} branch vector.

252 When modeling a composite made of a mixture of distinct phases, it is also useful to
 253 calculate the macroscopic stress contribution of each phase separately. Thus, similarly to Eq.
 254 (18), the stress tensor associated to phase P is defined as:

$$\sigma_{ij,P} = \frac{1}{V} \sum_{contacts,P} (N_i + T_i) (r - \frac{1}{2}\delta_n) n_j \quad (19)$$

255 where n_j is the j^{th} component of the contact normal vector \mathbf{n} . And, r is the radius of the
 256 discrete element of phase P for a given contact. Note that in Eq. (19), the sum is made
 257 on all contacts that include a particle of phase P . The total macroscopic stress tensor (Eq.
 258 (18)) is retrieved by summing all contributions from each phase P from Eq. (19). This is
 259 because the branch vector in Eq. (18) is replaced here by the term $(r - \frac{1}{2}\delta_n)\mathbf{n}$.

260 **4. Generation of numerical microstructures and compaction simulations**

261 Fig. 5 depicts the three stages that are used to characterize a mix from its generation
 262 (Fig. 5 a) to the compaction (Fig. 5 c). The initial composite assembly is obtained by first
 263 randomly locating bonded particle clusters (graphite and coarse alumina) and non-bonded
 264 single particles (fine alumina) in a parallelepipedic periodic cell with (1x1x2) size ratio (Fig.
 265 5 a). At this stage (denominated gas of particles), there is no contact between constituents
 266 (fine alumina, coarse alumina, and graphite flakes) and the relative density is small (0.4). The
 267 number of discrete elements for each phase is chosen to obtain the correct composition for a
 268 given mix (see table 3). Note that by modifying the random seed used to produce a given

269 initial gas of particles, we can produce several packings that exhibit the same macroscopic
 270 properties (essentially the same composition), but with particles located differently. This is
 271 useful to evaluate dispersion in our simulations. We have used 5 different random seeds for
 272 each condition to generate error bars.

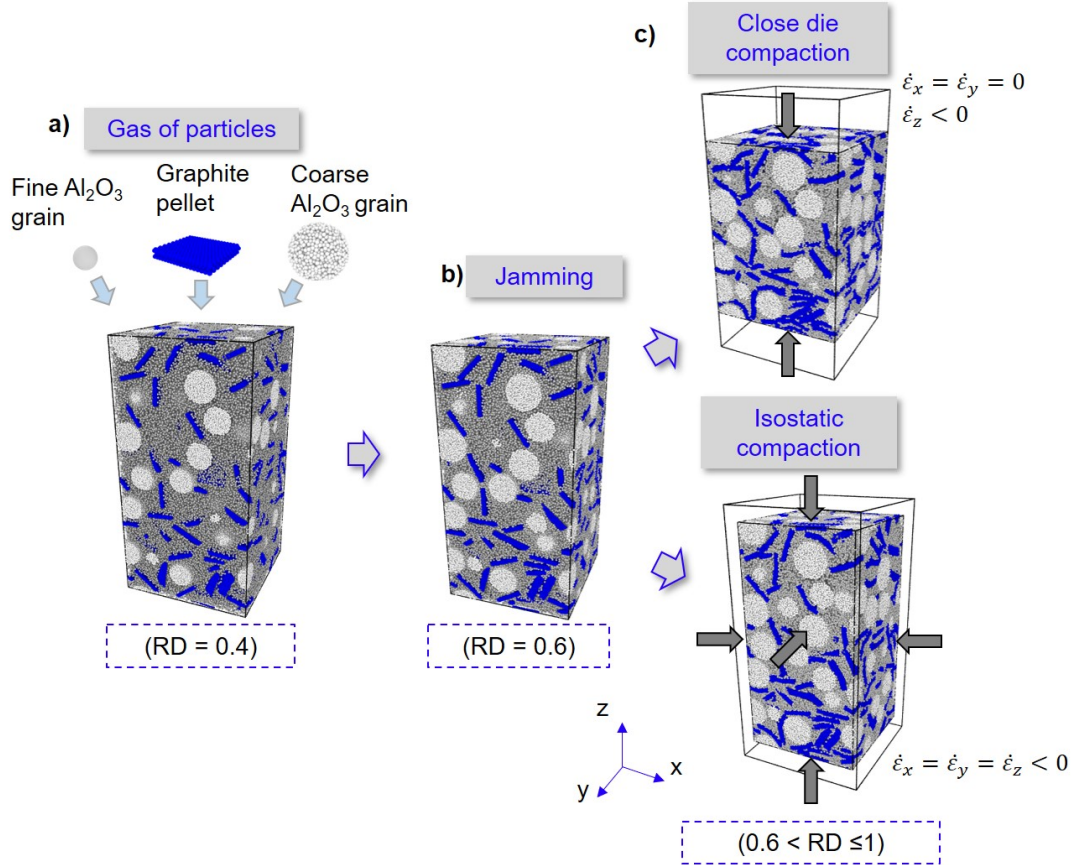


Figure 5: DEM simulations steps performed in this work: a) Particles/aggregates are randomly located inside a box with periodic conditions up to a density $RD = 0.4$. b) the packing is densified until $RD = 0.6$. c) close-die or isostatic compaction simulations are used to investigate the minimum RVE size (section 5).

273 This initial microstructure is further jammed by isostatic densification under a small
 274 macroscopic pressure (0.02 MPa) up to a relative density of 0.6 (Fig. 5 b) [34, 35]. During
 275 this jamming stage, only elastic interactions are considered (Eq. (1)). Friction and adhesion
 276 between particles are set to zero. The jamming stage mimics the gentle rearrangement of
 277 particles in the die without any plasticity or bond breakage. The jammed numerical packing
 278 at $RD=0.6$ represents the mix in the die before compaction. This value of the RD before
 279 compaction has been chosen to be in agreement with experimental observations.

280 From the jammed state, the numerical microstructure can be compacted along two routes:
 281 closed-die or isostatic compaction (Fig. 5 c). In close-die compaction, the packing is com-
 282 pacted by imposing an axial strain-rate ($\dot{\epsilon}_z$) in the z direction and a null strain-rate on
 283 x and y axis. Although we do not simulate the die itself, these conditions (uniaxial com-
 284 paction) should simulate correctly the typical stress conditions encountered by a composite
 285 powder during close-die compaction (far from the die). In isostatic compaction, an identical
 286 compressive strain-rate is imposed on all three axis.

287 The imposed strain-rates are adjusted during simulations to ensure quasi-static conditions
 288 (force equilibrium). The normalized kinetic energy per particle [36] is used as the criterion
 289 for adjusting strain-rates:

$$\tilde{E}_{kin} = \frac{E_{kin}}{n \max(Nr)} \leq 10^{-7} \quad (20)$$

290 where E_{kin} is the total kinetic energy of the particle system, n the total number of particles,
 291 and $\max(Nr)$ is the maximum value for all contacts of the product of the normal force N
 292 and discrete element radius r .

293 5. Representative volume element

294 Figs. 2 and 5 show the typical simulation box with fine alumina, coarse alumina and
 295 graphite flakes to form a composite. In these figures, the total number of particles is approx-
 296 imately 600k. Because the microstructure is quite complex with large size ratio between the
 297 fine alumina and coarse alumina particle, a three-dimensional Representative Volume Ele-
 298 ment (RVE) must be found to ensure that the results do not depend on the particle number
 299 while minimizing the CPU time. We use two criteria to determine the RVE: convergence of
 300 the macroscopic stress (at RD=0.95) and isotropy. The macroscopic response of the com-
 301 posite should be isotropic (due to the random orientation of the graphite flakes). We tested
 302 this through isostatic compaction simulations performed on RVEs containing increasing num-
 303 ber of discrete elements. All RVEs are generated with the composition of mix R (reference
 304 composite).

305 For each RVE, five different packings were generated using five different random seeds to
 306 locate the particles initially in the simulation box. A total of 30 RVEs are used from isostatic

307 compaction simulations to allow deviation measurement. The mean (Σ_m) and deviatoric (Σ_d)
 308 stresses are calculated and plotted in Fig. 6.

309 Fig. 6 shows that too small RVEs with $n \approx 14000$ discrete elements exhibit a large
 310 deviation, a non-isotropic behavior and a too large mean stress as compared to larger RVEs.
 311 For an RVE with $n \geq 380,000$ discrete elements, stresses converge with a much lower standard
 312 deviation. Larger RVEs confirm this trend with a clear convergence of both Σ_m and Σ_d . Fig.
 313 6 shows that an RVE with $n \geq 117,605$ discrete elements is large enough to ensure a good
 314 convergence of Σ_m and Σ_d with reasonable standard deviation.

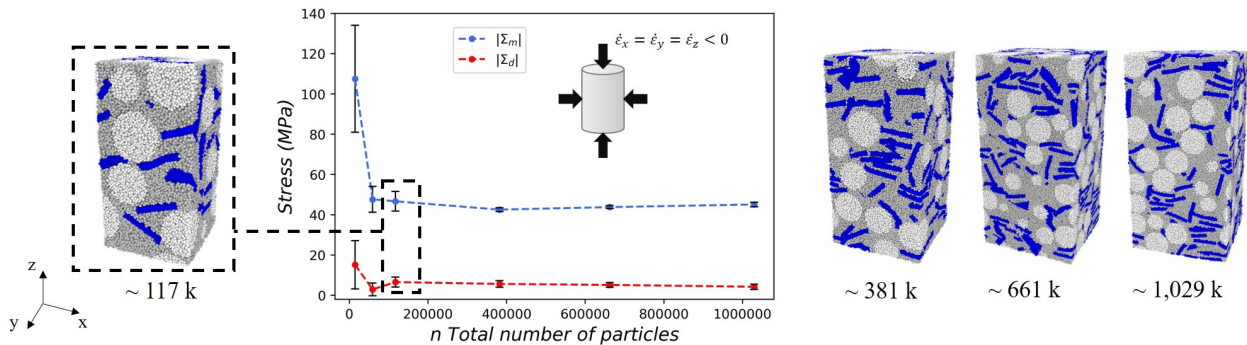


Figure 6: Evolution of the macroscopic deviatoric (Σ_d) and mean (Σ_m) stresses, in module, against the total number of particles, at the end of isostatic compaction simulations (density RD = 0.95).

315 We have repeated the isostatic compaction RVE study for closed-die compaction con-
 316 ditions and observed a similar behavior. In other words, a total number of particles n of
 317 the order of 110,000 is sufficiently large to ensure a good convergence and reasonable CPU
 318 time on both isostatic and closed-die compaction configurations. The typical total clock time
 319 using 8 CPUs (openMP parallelization) is 25 hours for $n = 117,605$ discrete elements and is
 320 approximately linear with n . All results shown hereafter are obtained with n of the order of
 321 100,000 particles (depending on the exact composition of the mix).

322 6. Closed-die compaction simulations

323 The closed-die compaction of the four mixes described in section 2 has been simulated.
 324 These simulations are compared with experimental data obtained from single-action pressing
 325 in a steel die for a maximum axial stress of 35 MPa. Experimental relative density was mea-
 326 sured by recording the piston displacement. The finite rigidity of the set-up was measured by

327 running piston to piston pressing (no powder), and the related displacement was subtracted
328 to the piston displacement to obtain the correct relative density (RD).

329 *6.1. Phase properties*

330 The material parameters at the microscopic scale that enter contact laws (Eqs. (1), (5-
331 12) have been determined to retrieve the macroscopic properties (elasticity and fracture) of
332 alumina and graphite phases. The identification methodology has been described in details
333 elsewhere for elasticity [35] and fracture [37]. Concerning the binder, the value of the elastic
334 modulus has been calibrated using the macroscopic compaction curve of mix R (reference
335 mix). This value (approximately 1 GPa) is much lower than for alumina and graphite phases,
336 as it should for a pyrolytic carbonaceous binder.

337 The material parameters chosen for the simulations are listed in Table 4. Note that the
338 Poisson's ratio has almost no effect and has been chosen as 0.2 for all materials for simplicity.
339 The friction coefficient has also been set to a reasonable value of 0.2 for all contacts. We
340 have observed that its effect is also of second order on the axial macroscopic stress during
341 closed-die compaction.

342 Binder stiffness is the material parameter that most affects the macroscopic compaction
343 behaviour of refractory composites. Fig. 7 shows simulations for increasing E_{binder} from 0.1
344 to 3 GPa for mix R (with $E_{binder} = 1.0\text{GPa}$, the value chosen for standard simulations).
345 The simulation curves are typical of closed-die compaction with an initial flat curve that is
346 characteristic of particle rearrangement followed by an increase of the axial stress when the
347 various phases in the mix deform at the contact (binder coated particles, graphite flakes).
348 The last stage of the compaction is an asymptotic increase of the stress when large RDs are
349 attained. Fig. 7 shows that the relative density attained at the maximum axial stress (35
350 MPa) decreases markedly as the binder rigidity increases. This indicates the importance of
351 the binder phase (with approximately 20% volume fraction of binder) in such composites. The
352 objective of the following sections is to better understand, using the wealth of information
353 provided by DEM simulations, the role of the binder and its interaction with other phases.

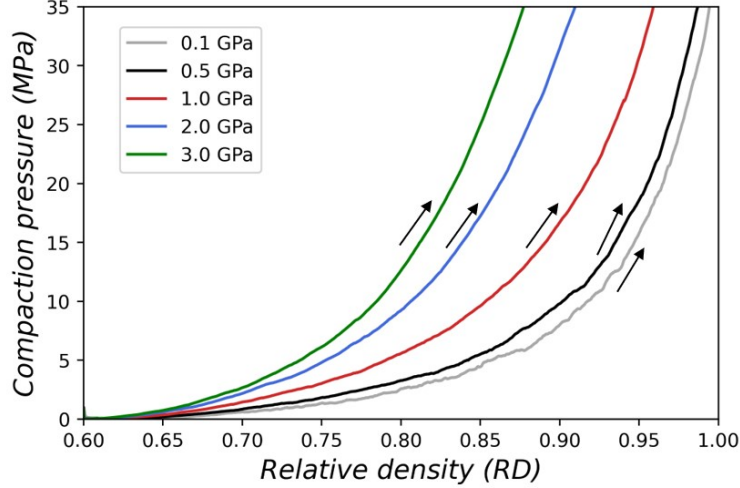


Figure 7: Evolution of the axial stress during closed-die compaction of Mix R with various values of E_{binder} .

Phase	γ (J/m ²)	μ	$\sigma_N = \sigma_T$ (MPa)	E (GPa)	ν
Coarse alumina	28	0.2	-	380	0.2
Fine alumina	28	0.2	-	380	0.2
Graphite	28	0.2	150	35	0.2
Binder	-	0.2	-	1.0	0.2

Table 4: Material parameters of contact laws used for compaction simulations for the four phase of the mixes.

354 6.2. Composites densification behavior

355 Eq. (19) allows to evaluate the contribution of each phase of the mix on the total macro-
 356 scopic stress. The evolution of the axial stresses associated with each phase (fine coated
 357 alumina, coarse coated alumina, and graphite) and the total macroscopic axial stress against
 358 the relative density are shown in Fig. 8 for mix R. The binder phase contribution is not
 359 explicitly included here as it is gathered together with fine coated alumina and coarse coated
 360 alumina. Fig. 8 also shows the experimental data with a good agreement with DEM. This
 361 validates the calibration of the binder stiffness. Note in particular that the curve shape is
 362 well rendered by the DEM simulation.

363 Fig. 8 allows sorting the contribution of each phase. Recall that a contact between two
 364 different phases (for example coarse alumina and fine alumina) is accounted in the contribu-
 365 tion of the two phases (with a 1/2 factor approximately since the branch vector in Eq. (18)

366 is replaced by the term $(r - \frac{1}{2}\delta_n)\mathbf{n}$. Fig. 8 indicates that the fine coated alumina particles
 367 and the graphite flakes have the most important contributions and that the coarse alumina
 368 coated particles have only a minor effect on the total stress. This result is not intuitive in
 369 a continuum mechanics framework since all phases have similar volume fractions and coarse
 370 alumina is the dominant phase in volume fraction (33%). It should be understood by recalling
 371 that the material to be compacted is a particulate material with particles interacting through
 372 their contacts. Because coarse alumina particles are large in size, they generate only a few
 373 contacts that do not percolate. The more numerous contacts between fine alumina particles
 374 have a much greater impact.

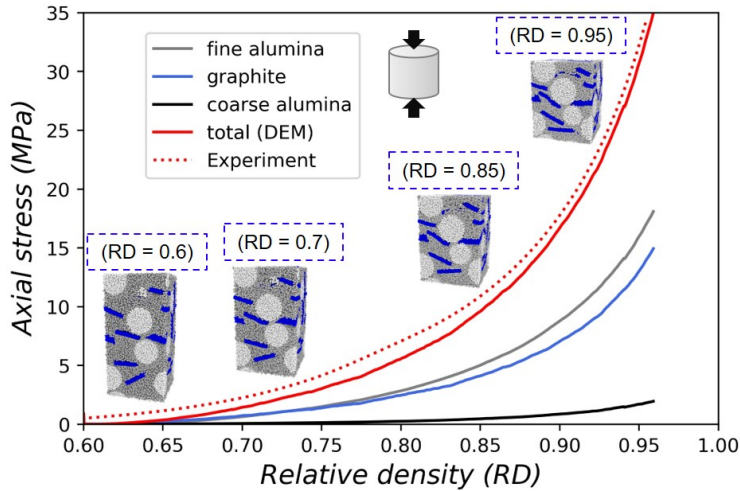


Figure 8: DEM close-die compaction: contribution of the three different discrete phases (fine alumina, coarse alumina, and graphite) to the axial macroscopic stress (final target: $\sigma_z=35$ MPa). Comparison of the total stresses evolution during a DEM closed-die compaction simulation and experiment (Mix R).

375 The above result for mix R, is confirmed for the four mixes by examining the contribution
 376 of each phase. This is carried out in Fig. 9, which gives the axial stress contributions
 377 attained for a target density RD=0.90 for all four mixes. Again, the dominant effect of fine
 378 alumina and graphite phases on the total macroscopic stress is clear for the four mixes. Fig.
 379 9 demonstrates that the binder volume fraction (accounted for in the DEM simulations by
 380 varying the binder layer thickness (see section 3.2)) is the best predictor for the ranking of
 381 the four mixes in terms of their ability to densify. This ranking is in good accordance with
 382 experimental data for the four mixes. Note that DEM material parameters are the same

383 for all four mixes, while the mix composition is solely accounted for by varying the particle
 384 numbers and the binder thickness.

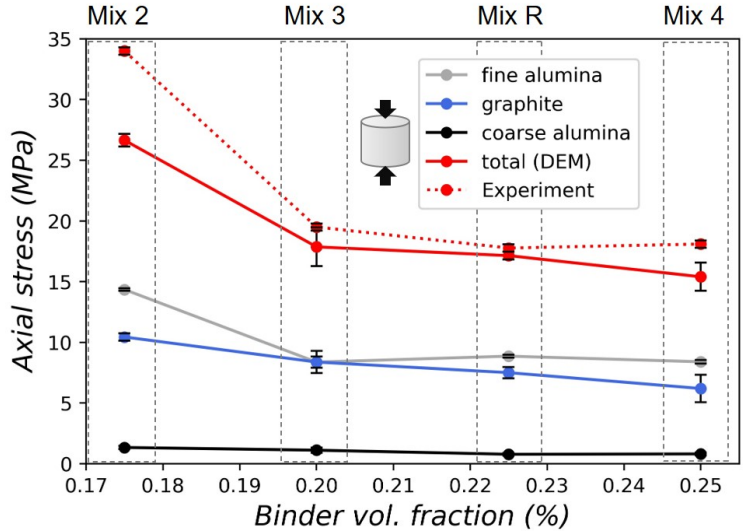


Figure 9: Contribution of the macroscopic axial stress related to each discrete phase (coarse alumina, graphite, and fine alumina) to the total macroscopic axial stress (σ_z) of composites (Mix R to 4), against the binder volume fraction (in %), during close-die compaction simulations (target: density RD=0.90). Comparison with experimental data obtained for the four mixes.

385 6.3. Contact behaviour of coated particles

386 Figs. 8 and 9 give valuable information on the contributions of the hard phases (alumina
 387 and graphite) during compaction. However, they still conflate the effect of the binder with
 388 those phases. To better understand the effect of the binder on the composite densification
 389 during closed-die compaction, it is necessary to analyze the behavior at the particle contact
 390 length scale.

391 For this purpose, the interactions between coated or non-coated particles has been recorded
 392 all along closed-die compaction for Mix R (reference composite) with a target of 35 MPa for
 393 the axial stress. Four typical contacts have been tagged and their normal contact force evo-
 394 lution was examined. The four contacts are coarse alumina - graphite (1), fine alumina -
 395 graphite (2), coarse alumina - fine alumina (3), and fine alumina - fine alumina (4). The
 396 evolution of a contact between two coarse alumina particles is not presented as this type of
 397 contact is very rare.

398 The evolution of the normal contact force (N^{Hertz}) (Eq. (1)), against the particle inden-
 399 tation normalized by the particle radius r , $\tilde{\delta}_n = \delta_n/r$, is shown in Figs. 10 a1) and a2). Note
 400 that the initial normal force is negative (tensile force) as adhesion is introduced only in the
 401 compaction simulation (the jamming stage is conducted with neither friction nor adhesion).
 402 Figs. 10 a1) and a2) illustrate the stiff transition from a binder dominated contact to a hard
 403 phase dominated one, with the material parameters from table 4. The critical value at which
 404 this transition occurs depends on the total binder thickness that coats the two particles (t_i+t_j
 405 in Eqs. (13) and (13)). Since we assume that graphite flakes are poorly coated ($t_2 = 0$), the
 406 transition for contacts involving graphite occurs at approximately half that for contacts with
 407 only alumina particles (≈ 0.2 and 0.4 , respectively).

408 When the transition is reached, the relative indentation $\tilde{\delta}_n$ does not increase anymore
 409 (although the hard phase stiffness is finite) and the contact loads and unloads in the hard-
 410 phase branch. Contact (1) between a coarse alumina particle and a graphite flake exemplifies
 411 such a behaviour. Figs. 10b shows the history of the four contacts as relative density
 412 increases from 0.6 to 0.97. Contact (1) reaches the critical indentation (marked by a star)
 413 soon in the compaction and stays in the hard phase branch afterward. The contact is lost at
 414 approximately 0.87 RD but the memory of its transition to branch 2 is kept in the simulation
 415 and further reloading when the contact is resumed is characteristic of branch 2.

416 Contact (2) exhibits a similar behaviour to contact (1) but is never lost during com-
 417 paction. Contact (3) shows the typical behaviour of a contact between two hard particles.
 418 The transition occurs at a rather large value of $\tilde{\delta}_n$ (≈ 0.4) as the two particles are coated
 419 with binder. Contact (4) illustrates the fact that some contacts never reach branch 2 and
 420 keep a soft behaviour all along the compaction.

421 These four examples represent only a fraction of the hundreds of thousands of contacts
 422 that arise in the simulation. Fig. 11 allows for a more quantitative understanding of how
 423 contacts evolve from the soft to the hard branch. It shows the fraction of contacts in a
 424 given branch for the five types of contact that have non-negligible contributions to the total
 425 macroscopic stress.

426 Below RD=0.85, the fraction of contacts between fine alumina particles corresponds to
 427 approximately 80% of interactions in the soft binder (Fig. 11 a) . Above RD=0.85, a gradual

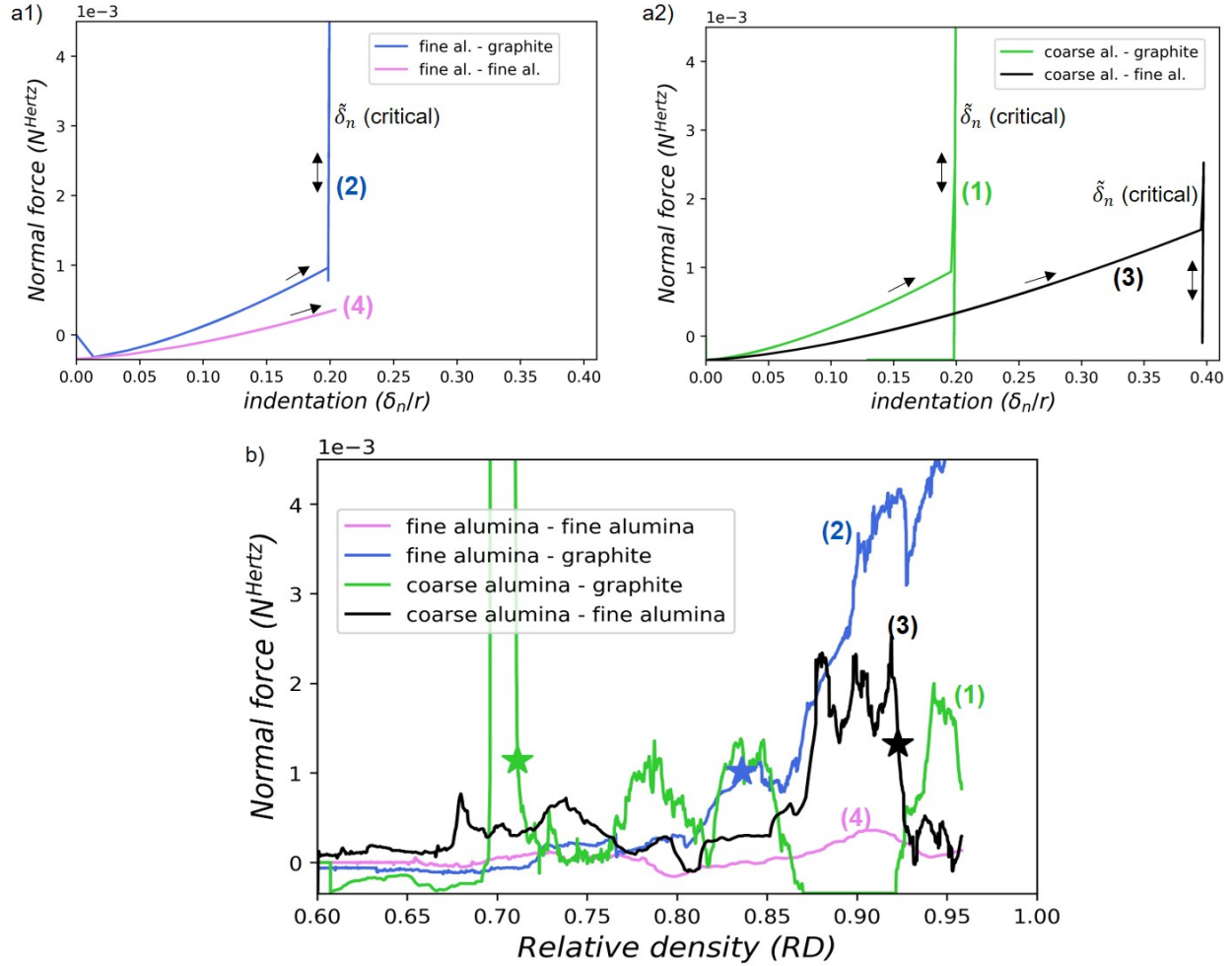


Figure 10: DEM close-die compaction simulation (target $\sigma_z=35$ MPa, Mix R). a1) and a2) Normal Hertzian contact force evolution (N^{Hertz}) versus normalized particle indentation (δ_n/r). Contacts type 1) to 4) represent the particle contact between three components (coarse alumina, fine alumina, and graphite). Once the contact reaches the critical indentation value ($\tilde{\delta}_n$), hard contacts appear. b) N^{Hertz} versus composite relative density (RD). * represents the point where the interactions move towards the hard branch.

428 decrease is observed, associated with an increase of the fraction of direct contacts between
 429 alumina particles (Fig. 11 b)). This increase explains in part the asymptotic behaviour of
 430 the axial stress at large densities (Fig. 8).

431 In any case, Fig. 11 a demonstrates that most contacts involving some binder stay in the
 432 soft branch and that the binder plays a dominant role for a large portion of the compaction.
 433 Recall, that the internal deformation of the graphite flakes also plays an important role on
 434 the total macroscopic stress (Fig. 8), but is not accounted in the contacts involving binder

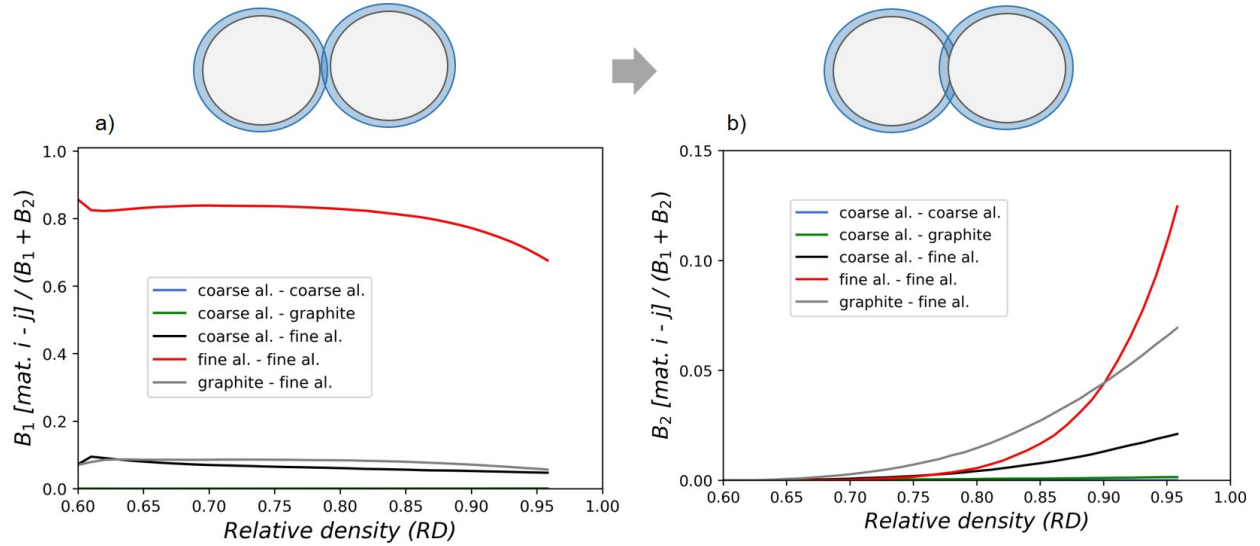


Figure 11: DEM close-die compaction simulation (target $\sigma_z = 35$ MPa, Mix R): fraction of contacts in a) the soft branch, and b) in the hard branch. B_1 and B_2 are the total number of contacts in the soft and hard branch, respectively.

435 shown in Fig. 11.

436 The fraction of contacts in branch 2 depends primarily on the volume fraction of binder
 437 (Fig. 12). A higher volume fraction of binder is associated, for a given stress or relative
 438 density, with a smaller fraction of contacts transitioning to branch 2 (hard phase). This is
 439 expected as the macroscopic deformation required to reach the critical indentation locally
 440 must be greater for a thicker binder coating.

441 We have also observed, that for a given relative density or a given stress, increasing the
 442 binder stiffness results in less contacts in branch 2. This is because reaching the critical
 443 indentation becomes less frequent with stiffer binder. In any case, the binder stiffness has a
 444 significant impact on the macroscopic compaction stress (as indicated by Fig. 7), as a large
 445 majority of contacts remain in branch 1 dominated by the binder stiffness (Fig. 11).

446 7. Conclusions

447 In the present work, an original approach was developed based on a Representative Vol-
 448 ume Element (RVE) to mimic the complex microstructure of model carbon-bonded alumina
 449 refractory composites. The compaction of the numerical microstructure was simulated using
 450 the discrete element method (DEM). Four compositions were studied by varying the volume

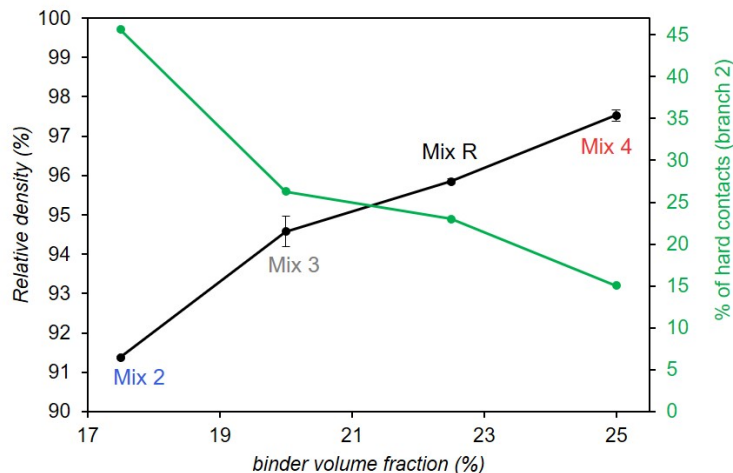


Figure 12: Evolution of the relative density (RD) of each composite, at 35 MPa, and the % of contacts in branch 2 (hard) at the end of the compaction.

451 fraction of each phase, within the typical range that is of interest for industrial applications.
 452 The composite in DEM was modeled by a mixture of single fine particles and bonded particle
 453 clusters (coarse alumina and graphite). For this purpose, a new contact models was devel-
 454 oped to take into account the binder phase within the DEM framework. We have also shown
 455 the importance of carefully choosing the minimum number of discrete elements to ensure
 456 a good compromise between CPU time and model accuracy. When dealing with complex
 457 microstructures, this verification step is crucial as too small an RVE would lead to inaccurate
 458 results.

459 The main conclusion of this work is that for compositions of the order of 20 % volume
 460 fraction of binder, the binder stiffness dominates the mechanical response of the composite
 461 during compaction. Accordingly, varying the binder volume fraction will impact significantly
 462 the compaction behaviour. The DEM framework that we propose allows for a wealth of
 463 information to be obtained, leading to a more subtle picture of how these composites densify
 464 under compaction. In particular, it could lead to microstructure optimization to attain larger
 465 RDs for a given compacting stress.

466 Seeking simplicity, both in the implementation and the post-calculation analysis, we have
 467 chosen to represent coarse alumina particles as monosize spheres. The graphite was modelled
 468 by a rectangular geometry, also with monosize distribution. The real composite microstruc-

469 ture is clearly much more complex. It is characterized by irregular coarse alumina grains
470 and graphite flakes with some dispersion in size and shape. We believe that the main results
471 obtained here should remain valid for more realistic microstructures. Ideally, the angular
472 coarse alumina particles could be modelled by using X-ray tomography images. A number
473 of individual real coarse alumina particles could be included in the simulation as proposed in
474 [38].

475 Coarse and fine alumina hard particles were coated by an elastoplastic soft binder in DEM.
476 This model aimed to represent in a simplified manner the presence of a soft coating on hard
477 particles. The good qualitative and quantitative agreement with experimental compaction
478 curves brings some confidence on the possibility of such a model to represent accurately a
479 continuous matrix that cover most particles. The model includes some plasticity, once the
480 contact has entered the hard-hard branch. It could be further improved by including some
481 plasticity in the soft branch.

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484

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