

Taylor Dispersion Analysis and Atomic Force Microscopy Provide a Quantitative Insight into the Aggregation Kinetics of A β (1–40)/A β (1–42) Amyloid Peptide Mixtures

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- Taylor Dispersion Analysis and Atomic Force
- 2 Microscopy provide quantitative insight on the
- aggregation kinetics of $A\beta(1-40)/A\beta(1-42)$ amyloid
- 4 peptide mixtures
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ABSTRACT

Aggregation of amyloid β peptides is known to be one of the main processes responsible for Alzheimer's disease. The resulting dementia is believed to be due in part to the formation of potentially toxic oligomers. However, the study of such intermediates and the understanding of how they form is very challenging because they are heterogeneous and transient in nature. Unfortunately, few techniques can quantify, in real time, the proportion and the size of the different soluble species during the aggregation process. In a previous work (Deleanu *et al.* Anal. Chem. 2021), we showed the potential of Taylor dispersion analysis (TDA) in amyloid speciation during the aggregation process of $A\beta(1-40)$ and $A\beta(1-42)$. The current work aims at exploring in detail the aggregation of amyloid $A\beta(1-40)$: $A\beta(1-42)$ peptide mixtures with different proportions of each peptide (1:0, 3:1, 1:1, 1:3 and 0:1), using TDA and atomic force microscopy (AFM). TDA allowed for monitoring the kinetics of the amyloid assembly and quantifying the transient intermediates. Complementarily, AFM allowed the formation of insoluble fibrils to be visualized. Together, the two techniques enabled to study the influence of the peptide ratios on the kinetics and the formation of potentially toxic oligomeric species.

KEYWORDS. Taylor dispersion analysis; AFM; peptide aggregation; oligomers; amyloid beta peptides; diffusion coefficient; hydrodynamic radius.

Introduction

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Alzheimer's disease (AD) is the most common primary dementia. It usually presents a progressive course and characteristically affects different cognitive and behavioral functions. Perhaps the cardinal, most frequently observed symptom of disease onset is memory loss¹, which results from initial lesions in the hippocampus (which lies in the medial temporal lobes of the brain and is responsible for long-term memory) further extending to the rest of the brain. At later stages, the degeneration of other cognitive and behavioral areas is observed which will clearly indicate the type of dementia. The brains of Alzheimer's patients present a series of characteristic hallmarks. First, neurofibrillary tangles composed of hyperphosphorylated tau protein are observed in neurons. Then, extracellular fibrillary structures called neuritic (or senile) plaques, that are due to the deposition of amyloid β peptides (A β), are observed². In the 1990's, it was believed that the senile plagues were the most pathogenic forms of the $A\beta^{3,4}$, which resulted from the self-assembly of the two major amyloid peptides $A\beta(1-40)$ and $A\beta(1-42)$. During this self-assembly, species are formed with evolving morphology and size from oligomers, to protofibrils and finally to fibrils and plaques, through a highly complicated process. More recent studies suggested that the main factor of AD pathogenesis was the formation of soluble oligomers of A\beta, which are believed to be more toxic than plaques because they are able to spread across neuronal tissue and bind to membrane receptors, including the prion protein, promoting neurotoxicity and synaptic loss⁵⁻⁷. However, in contrast to fibrils, which are highly stable and can be observed by microscopy, the soluble oligomers are more difficult to detect and to study in real time⁸, since they are metastable, transient⁸ and highly polydisperse in size. Studies have shown that $A\beta(1-40)$ does not quantitatively form small oligomers during the aggregation

process but rather goes from monomers to fibrils following a direct pathway^{9, 10}. On the other hand, A β (1-42) goes through the formation of intermediate species with diverse sizes and shapes ^{11, 12}. Although these two amyloid peptides coexist in vivo^{13, 14}, most of the in vitro studies on Aβ were focused on the study of pure peptide solutions and only a small proportion of the vast AD literature was dedicated to mixtures of these peptides¹⁵⁻¹⁸. Many of the studies dealing with the mixtures were directed toward the kinetics of the aggregation process and more particularly the study of the amyloid fibers 15-17, 19-21 and very few toward the oligomeric structures 18. The aggregation process in such mixtures was already studied by Thioflavin T fluorescence^{17, 21}, sedimentation¹⁵, atomic force microscopy (AFM)^{16, 19}, nuclear magnetic resonance (NMR)^{22, 23} or electron paramagnetic resonance 18 to name few methods. The results from these studies indicated that both A β (1-40) and $A\beta(1-42)$ interact during the aggregation process, with cross seeding between peptides. However, some authors claimed that the fibrils are homomolecular¹⁷ while others stated that heteromolecular 18 fibrils are obtained. In all cases, it was observed that the presence of A β (1-42) accelerated the aggregation of $A\beta(1-40)$ and vice versa. To our knowledge, an in-depth study on the species present during the early stages of the aggregation of amyloid peptide mixtures does not exist to date. For that reason, and to help develop drug candidates targeting the toxic oligomers, new analytical methodologies are required to monitor and size the different species in real time. In this context, Taylor dispersion analysis²⁴⁻²⁶ (TDA) appears as a very promising alternative analytical method. In our previous report²⁷, we showed the ability of TDA to follow the aggregation process of amyloid peptides, using an extensive data treatment that revealed a complete picture of the aggregation process and allowed to size the transient structures. As described elsewhere, TDA allows to determine the molecular diffusion coefficient, D, and hydrodynamic radius, R_h , of a solute, including for mixtures, without any bias in size as compared

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to other sizing methods²⁸, since the small and the large solutes contribute proportionally to their mass abundance in the mixture²⁹. The use of TDA in the field of protein aggregation presents several advantages³⁰⁻³³. It offers low sample consumption (less than 1 μ L for the whole aggregation process), short analysis time allowing a high number of sampling points and real time monitoring of the aggregation, a wide range of sizing (from angstrom to sub-micron) allowing to size the monomers, oligomers and higher size soluble prefibrillar structures, together with a direct analysis without any sample pretreatment or filtration ³⁴⁻³⁶.

In this work, we used TDA to study peptide mixtures of Aβ(1-40) and Aβ(1-42) with the aim of

revealing the size of the transient structures formed during the aggregation process. The results for selected time points were compared with those obtained by AFM to correlate the observations from these two independent techniques. AFM also allowed to study non-soluble fibrils, which cannot be sized or directly observed by TDA.

Materials and methods

Materials

Amyloid beta (1-40) (denoted A β (1-40) in this work) was prepared by fast conventional solid phase peptide synthesis (SPPS) using a *Fmoc* orthogonal strategy as described elsewhere²⁷. Amyloid beta (1-42) (A β (1-42), batch number 100002591, >95%) was purchased from Bachem (Bubendorf, Switzerland). Sodium dihydrogen phosphate, tris(hydroxymethyl)aminomethane, hydrochloric acid fuming 37%, sodium chloride and sodium hydroxide were purchased from Sigma Aldrich (France). The ultrapure water used for all buffers was prepared with a MilliQ system from Millipore (France).

Peptide pretreatment

Both $A\beta(1-40)$ and $A\beta(1-42)$ were first pretreated independently as described elsewhere^{37, 38}. Briefly, $A\beta(1-40)$ and $A\beta(1-42)$ were dissolved in a 0.10 % (m/v) and 0.16 % (m/v) NH₄OH aqueous solution respectively to reach a final peptide concentration of 2 mg/mL. The peptide solutions were then incubated at room temperature for 10 min, separated into several aliquots and freeze-dried. The aliquot volume was calculated in order to obtain 10 nmol of peptide in each Eppendorf tube. The lyophilized peptide aliquots were stored at -20 °C until further use.

Peptide aggregation study by Taylor dispersion analysis

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Briefly, and as described thoroughly elsewhere 25, 29, 30, 39, Taylor dispersion analysis allows for the determination of the molecular diffusion coefficient of a solute which can be obtained from the band broadening resulting from the combination of the Poiseuille parabolic flow and the molecular diffusion by quantifying the temporal variance (σ^2) of the elution profile. For that, a Gaussian function is used to fit the experimental elution peak allowing to obtain the peak variance σ^2 and thus calculate the molecular diffusion coefficient, D, and consequently the hydrodynamic radius, R_h . When more than one size populations are present, a sum of Gaussian functions can be used to fit the experimental trace, or Constrained Regularized Linear Inversion (CRLI)⁴⁰ can be used to get the size distribution. For more details, the reader may refer to the supporting information for the theoretical aspects, equations and more details on the data processing. TDA was performed on an Agilent 7100 (Waldbronn, Germany) capillary electrophoresis system using bare fused silica capillaries (Polymicro technologies, USA) having 40 cm × 50 µm i.d. dimensions and a detection window at 31.5 cm. New capillaries were conditioned with the following flushes: 1 M NaOH for 30 min; ultrapure water for 30 min. Between each analysis, capillaries were rinsed with 20 mM phosphate buffer, pH 7.4 (2 min). Samples were injected hydrodynamically on the inlet end of the capillary (44 mbar, 3 s, injected volume was about 7 nL

corresponding to 1% of the capillary volume to the detection point). Experiments were performed using a mobilization pressure of 100 mbar. The temperature of the capillary cartridge was set at 37 °C. The vial carrousel was thermostated using an external circulating water bath 600F from Julabo (Germany). The solutes were monitored by UV absorbance at 191 nm. The mobile phase was a 20 mM, pH 7.4 phosphate buffer (viscosity at 37 °C is 0.7×10^{-4} Pa.s). To prepare the mixtures, freeze-dried mixed peptide aliquots were prepared so that the final sample would contain 13 nmol of total peptide except for the pure samples where the amount of peptide was of 10 nmol. First, each of the required stock aliquots (see previous section) were dissolved in 100 µL of 0.16% (m/v) NH₄OH to avoid aggregation during this step, and appropriate volumes were used to obtain the desired mixtures. The final aliquots were immediately subjected to freeze-drying and then stored at -20°C until further use. The resulting peptide powders were dissolved in 20 mM phosphate buffer at pH 7.4 to reach a final total peptide concentration of 133 μM : i) 100 % Aβ(1-40) contained 10 nmol of A β (1-40) dissolved in 75 μ L of buffer; ii) 75 % A β (1-40) and 25 % $A\beta(1-42)$ mixture contained 10 nmol of $A\beta(1-40)$ and 3.33 nmol of $A\beta(1-42)$ dissolved in 100 μ L of buffer; iii) 50 % A β (1-40) and 50 % A β (1-42) mixture contained 6.67 nmol of each peptide and was dissolved in 100 μ L of buffer; iv) 25 % A β (1-40) and 75 % A β (1-42) mixture contained 3.33 nmol of A β (1-40) and 10 nmol of A β (1-42) dissolved in 100 μ L of buffer and finally ν) 0 % A β (1-40) contained 10 nmol of A β (1-42) dissolved in 75 μ L of buffer. After dissolution, the mixtures were immediately transferred to a capillary electrophoresis vial and incubated at 37°C in the capillary electrophoresis instrument carrousel. Aggregation was monitored by injecting the sample $(V_{inj} \approx 7 \text{ nL})$ every 7 min in the case of pure A β (1-42) and the A β (1-40): A β (1-42) 1:3 mixture (25) % A β (1-40)), while it was injected every 20 min for the A β (1-40): A β (1-42) 1:1 mixture (50 % $A\beta(1-40)$) and every 30 min in the case of pure $A\beta(1-40)$ and the $A\beta(1-40)$: $A\beta(1-42)$ 3:1 mixture

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(75 % A β (1-40)). During the monitoring of the aggregation process, each sample was injected for 100 to 125 TDA runs, corresponding to a total injected sample volume between 700 nL and 875 nL. To avoid sample evaporation, the vial cap was changed three times a day. The taylorgrams were recorded with the Agilent Chemstation software, then exported to Microsoft Excel for subsequent data processing. In general, the obtained elution profiles were not Gaussian, meaning that the sample was polydisperse in size. All taylorgrams were fitted on the basis of the right-side elution profile (*i.e.* $t > t_0$, with t_0 the peak time) to remove the spikes present on the left side as described elsewhere²⁷.

Atomic force microscopy (AFM)

For atomic force microscopy characterisation, 5 μ L aliquots of the peptide solutions collected at different aggregation times were dried on silicon substrates freshly cleaned with piranha solution, before gently rinsing with ultrapure water and drying in a flow of nitrogen gas. An Agilent 5500 AFM system with MSNL-F cantilevers (f = 110-120 kHz, k = 0.6 N/m, average tip radius of 2-12 nm) was used for topographical imaging in intermittent contact mode. The AFM topography images were levelled, line-corrected and analysed using Gwyddion⁴¹, a free and open-source SPM (scanning probe microscopy) data visualization and analysis program. Maxima analysis was performed using ImageJ⁴².

Results and discussion

Taylor Dispersion Analysis and data processing

One main objective of this work is to show the influence of the relative proportion of $A\beta(1-40)$ with respect to $A\beta(1-42)$ on the aggregation process. The aggregation of the peptides in the different mixtures was followed at 37°C. Figure 1 shows the taylorgrams recorded at selected

incubation times for A β (1-40), A β (1-40):A β (1-42) mixtures (with molar ratio of A β (1-40) in the mixture of 25%, 50% and 75% corresponding to 3:1, 1:1 and 1:3 mixtures), and A β (1-42). The total peptide concentration in each solution was set at 133.3 μ M. All experimental taylorgrams, for all incubations times t_{ag} , are shown in Figures SI.1 to SI.5. Figure 1 allows for a visual comparison of the aggregation kinetics between the different amyloid peptide ratios. The absorbance decrease of the elution profile with t_{ag} , which is due to the decrease in concentration of the soluble species, was faster when increasing the A β (1-42) content in the mixture, as previously observed for pure peptide solutions²⁷.

In order to elucidate the aggregation process in these solutions, an extensive data treatment was realized on all the obtained taylorgrams. As already observed in our previous report²⁷, sharp peaks or spikes, sometimes appear on the left side of the elution peak because of the presence of large

realized on all the obtained taylorgrams. As already observed in our previous report²⁷, sharp peaks or spikes, sometimes appear on the left side of the elution peak because of the presence of large fibrils in suspension which are out of the Taylor regime^{43, 44} and/or due to specific hydrodynamic behavior for suspended large aggregates⁴⁵. The presence of these spikes imposes a data treatment on the right side of the elution profile. Two different approaches were used to treat the experimental elution profiles. First, the fitting with a finite number of Gaussian functions (n = 4 in this work), leads to the classification of the obtained size populations into four categories: (i) small unidentified molecules ($R_h < 0.9$ nm), (ii) monomers and small oligomers (0.9 nm $< R_h < 5$ nm), (iii) higher mass oligomers (5 nm $< R_h < 50$ nm) and finally (iv) soluble protofibrils (50 nm $< R_h < 300$ nm). The second approach is based on the Constrained Regularized Linear Inversion (CRLI) which aims at finding the probability density function $P_D(D)$ that fits the taylorgram without any hypothesis on the number of populations⁴⁰.

Aggregation process of pure and mixed solutions of A\beta peptides by TDA and AFM

In the case of A β (1-40) alone, the first sign of spikes (fibrils) appears at $t_{ag} \sim 12$ h (Figure SI.1). However, the spike intensity remained relatively low and did not increase drastically over the whole aggregation process (see gray data points in Figure 2). In parallel, the monomer peak intensity remained constant until $t_{ag} \sim 20$ h ("lag phase"), and then decreased rapidly to reach a lower plateau after $t_{ag} \sim 48$ h (red squares in Figure 2). This suggests an initial slow aggregation step followed by a rapid fibrillization catalyzed by the formation of large aggregates (seeds) that do not enter the capillary upon injection⁴⁶, in accordance with our previous report where another concentration (100 μ M) of A β (1-40) was used²⁷. On the contrary, $A\beta(1-42)$ alone did not show any lag phase and a fast decrease of this population was observed with a complete disappearance after less than 2 h (red squares in Figure 2). Meanwhile, when mixed together, $A\beta(1-42)$ seemed to increase the kinetics of aggregation, with the monomer population decreasing after 15 h, 8 h and 6 h for the 75%, 50% and 25% AB(1-40) mixtures respectively, without any visible lag phase (red squares in Figure 2). With the aim of verifying hypotheses formulated from TDA, we also performed AFM imaging for key times of the fibrillization process. The findings by TDA correlated well with AFM observations for A β (1-40) alone (Figure 3), which showed the number of fibrils rise only at t_{ag} = 28.10 h. In AFM images, the substrate remained covered with spherical objects in the 10 nm diameter range (including eventual tip convolution effects) that can be attributed to monomers and oligomers. The coverage by these species only decreased significantly, exposing portions of the bare substrate, for $t_{ag} = 28.10$ h, which demonstrates the consumption of these objects by the fibrillization process.

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In the case of A β (1-42) alone, the kinetics of aggregation were much faster than that of A β (1-40), as demonstrated by the taylorgrams (Figure 1) and the monomer consumption (red squares in Figure 2). The first spikes were observed only after $t_{ag} \sim 10$ min (Figure SI 5), in agreement with AFM showing fibrils for the shortest incubation times, and a complete disappearance of the soluble peptide species was observed after $t_{ag} \sim 7$ h (Figure 3). Similarly, only few monomeric/oligomeric species remained visible after 48 min in AFM images, which showed from this time onward mainly the substrate and fibrils. The decrease in the number of fibrils observed by microscopy for the longest incubation times is likely due to the tangling of fibrils and compaction of these aggregates, that have less affinity for the substrate and are more easily removed during the rinsing and drying steps, as observed for other amyloid systems⁴⁷. When mixed together, aggregation/fibrillization observed by both TDA and AFM accelerated with the proportion of A β (1-42). In the case of the 75% A β (1-40) mixture, the first spikes in TDA appeared after t_{ag} ~6.5 h (Figure SI.2), while they were observed after only ~1.5 h and ~1 h for the 50 % and 25 % Aβ(1-40) mixtures respectively (Figures SI.3 and SI.4 respectively). In the case of the mixture with the highest amount of A β (1-40), the spikes intensity increased with time and remained visible even after 60 h of aggregation suggesting that the formed fibrillary structures are smaller in size than those formed with A β (1-40) alone, and can enter more easily in the capillary during the injection step. These observations were also conveyed in the AFM images, with fibers appearing increasingly early during the aggregation process, and the spherical objects attributed to monomers and oligomers being consumed faster upon raising the proportion of A β (1-42). Indeed, these small species remain predominant on the substrates up to 28.10, 8.15, 3.05 and 0.27 h respectively for 0, 25, 75 and 100% $A\beta(1-42)$.

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As mentioned earlier, the Taylorgrams were treated by fitting a sum of Gaussian functions to extract the abundance of size populations present under each elution peak during the aggregation process. Figure 4 shows the hydrodynamic radii of the populations in the studied mixtures as a function of incubation time, while Table 1 shows the average hydrodynamic radii values for the size populations and the different peptide mixtures. As can be seen in Figure 4, the size of each population was relatively constant during the aggregation process. The population called 'monomer' corresponds to the monomers and small oligomers with an average size of about 2 nm in all mixtures. However, the statistical analysis showed that the monomer populations in A β (1-40) and in the 25 % A β (1-40) mixture were significantly smaller at a 95% confidence level than for the other three mixtures, based on the average hydrodynamic radii obtained over the whole aggregation study for each mixture (see Table 2 and Figure SI.6). Further, Aβ(1-40) alone showed the lowest 'monomer' population size average value of 1.81 ± 0.11 nm, while the 25 % A β (1-40) mixture showed the highest average value of 2.42 ± 0.20 nm. While the other three mixtures, 75 % A β (1-40) mixture, 50% A β (1-40) mixture and A β (1-42) alone were not statistically different (see R_h values in Table 2). These results suggest that when the mixture contained an excess of $A\beta(1-42)$ as compared to $A\beta(1-40)$, the formed oligomers were larger in size than those obtained for $A\beta(1-42)$ alone and for the mixtures with a higher amount of $A\beta(1-40)$. To explain this observation, the following hypothesis is proposed. First, it was already shown that A β (1-40) mainly forms fibrils directly from the monomeric state without passing through intermediate species^{9, 10}. The presence of the first fibrils of $A\beta(1-40)$ catalyzed by the presence of $A\beta(1-42)$ may play the role of seeds for the A β (1-42) peptide²¹, and thus, oligomers with a higher molar mass are formed. When the amount of A β (1-42) decreases, the aggregation process tends to follow the pathway characteristic of $A\beta(1-40)$ (direct formation of fibrils), and thus, the smallest oligomeric

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species tend to remain monomeric with a lower average size. For AFM, in the presence of biomolecules, the lateral dimensions are often affected by tip convolution, leading to unpredictable broadening of surface features^{48, 49}. We therefore measured the average heights of the various objects, which gave a good estimation of radii. The results obtained on the different mixtures of the radii (at times enabling to measure isolated objects) for small spherical objects attributed to monomers/oligomers, and the derived half-widths for the fibers are gathered in Table 2. The radii of monomers/oligomers are in good agreement between TDA and AFM, although values measured for monomers, oligomers and fibers with AFM are not significantly different between ratios.

Table 2: Average hydrodynamic radius in nm obtained by TDA over the whole aggregation process of each mixture. Average heights in nm obtained by AFM on the small spherical objects and on the fibrils for the different A β mixtures and at different aggregation times.

		AFM						
	TDA	Small sph		Fibrils				
% Aβ(1-40) in Aβ(1-40):Aβ(1-42) mixtures	Average R_h (nm) (n = number of points)	Radii ± SD (nm)	t _{ag} (h)	Half-width ± SD (nm)	t _{ag} (h)			
0	1.96 ± 0.10 $(n = 121)$	2.15 ± 0.40	1.47	2.89 ± 0.70	0.8			
25	2.42 ± 0.20 $(n = 77)$	2.22 ± 0.44	8.02	2.78 ± 0.72	27.83			
50	2.00 ± 0.25 $(n = 42)$	N.A.	N.A.	N.A.	N.A.			
75	1.98 ± 0.24 $(n = 77)$	2.32 ± 0.44	12.23	3.02 ± 1.17	27.97			
100	1.81 ± 0.11 $(n = 53)$	2.43 ± 0.60	28.10	3.53 ± 0.93	28.10			

As for the higher mass oligomer population, the size remained relatively constant throughout the aggregation process within a given mixture. However, it was observed that with an increase in the $A\beta(1-40)$ proportion, the average size for the oligomer population was higher than in the case of an excess of A β (1-42) (~18 nm as compared to ~9 nm) (Figure SI.6). Moreover, a sudden increase of the oligomer R_h from about 10 nm to ~ 30 nm can be observed when the "monomer" population decreased in area (Figure 2 for the size and Figure 4 for the area) after \sim 2h for A β (1-42), 24 h for the 75% A β (1-40) mixture and 32 h for A β (1-40). This effect was not observed for the 50% and 25% Aβ(1-40) mixtures. Further, Figure 2 shows that this oligomeric population reaches a maximum in concentration at around 30 min for the A β (1-42) sample, while this maximum is shifted to higher times with the decrease in A β (1-42) proportion (1.3 h, 1.6 h, 5.5 h and 16 h for 25%, 50%, 75% and 100 % A β (1-40) mixtures, respectively). Recent studies suggested that amyloid peptides can undergo liquid-liquid phase separation (LLPS) before the formation of amyloid fibrils $^{50-52}$. The "high mass oligomers" population with R_h ranging from 5 to 50 nm found in this work, might correspond to high-density protein condensates. The size increase of the species over time can be explained by Ostwald ripening.

Regarding the protofibril population, the R_h values varied between 80 and 140 nm, for all the mixtures independently of the peptide proportions (Figure 4).

Hydrodynamic radius distributions by CRLI

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To get a deeper insight into the evolution of the species during the aggregation process, CRLI analysis^{27, 40} was applied on the right part of the taylorgrams (i.e. for $t > t_0$). Figure 5 shows the hydrodynamic radii distributions obtained by CRLI on TDA runs for selected aggregation times for all studied samples, while Figures SI.7 to SI.11 show the distributions over the whole

aggregation process for the studied samples. From these distributions, one can note that for A β (1-40) alone, only the monomer and small oligomer populations were observed throughout the aggregation process without the notable appearance of intermediate species. On the contrary, A β (1-42) aggregation led to intermediate oligomers having an average size around 10 nm accompanied with a broadening and disappearance of the monomeric population in accordance with our previous report²⁷. When mixed together, the presence of intermediate species became more noticeable with the increase in A β (1-42) proportion. These results provide an explanation to the increase in A β mixtures toxicity with an increase in the A β (1-42)/A β (1-40) ratio^{19, 23, 53}, supporting the notion that this toxicity correlates with the amount of intermediate oligomeric species.

Kinetic analysis of the aggregation process

One of the major advantages of using TDA combined with advanced data treatment is the possibility to get the size distributions of the different populations with high throughput during the whole aggregation process. These distributions allow gaining insight on the aggregation mechanism, by modelling the data based on the chain of association and dissociation reactions shown in Figure 6⁵⁴.

Monomer (M)
$$\underset{k_{PM}}{\longleftarrow}$$
 Oligomer (O) $\underset{k_{PO}}{\longleftarrow}$ Protofibrils (P) $\underset{k_{FP}}{\longleftarrow}$ Fibrils (F)

Figure 6. Scheme of the association and dissociation reactions of amyloid species used to model the aggregation process (adapted from⁵⁴). The rate constant for each reaction is indicated close to the respective reaction arrow.

The forward and backward reactions going from the monomer (M) population to the oligomers (O), the protofibrils (P), and finally the fibrils (F), as shown in Figure 6, are modelled by the following set of equations:

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$$\frac{dM(t)}{dt} = -k_{MO}M(t) + k_{OM}O(t) + k_{PM}P(t) + k_{FM}F(t)$$
 (1)

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$$\frac{dO(t)}{dt} = -(k_{OP} + k_{OM})O(t) + k_{MO}M(t)$$
 (2)

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$$\frac{dP(t)}{dt} = -(k_{PF} + k_{OM})P(t) + k_{OP}O(t)$$
 (3)

$$\frac{dF(t)}{dt} = k_{PF}P(t) - k_{FM}F(t) \tag{4}$$

where, M(t), O(t), P(t) and F(t) are the concentrations of the monomers, oligomers, protofibrils and fibrils respectively. k_{MO} is the rate for the reaction transforming monomers into oligomers; similar notations are used for the other rate constants. Note that TDA does not allow the fibrils population to be directly measured, so only equations (1-3) were used. We fit Eqs. (1-3) to the temporal evolution of the population distributions obtained by TDA, by assuming that the (integrated) absorbance signal for each species is proportional to its concentration, and that the same proportionality constant applies to all species. Equation (4) was not used since TDA does not allow the direct quantification of the fibril concentrations. The fit is performed using custom software based on the Scipy package⁵⁵, performing a least square minimization of the set of rate constants, solving numerically Eqs. (1-3) at each iteration. The resulting fitting curves are shown as dot-dashed lines in Figure 2 (see also Figures SI.12 to SI.16), while the fitting parameters are reported in Table 1. Note that for most samples the monomer population asymptotically tends for large t to a non-zero concentration value, due to the dissociation of the higher-order species. The 100 % A β (1-40) could not be fitted entirely, because of the presence of a lag phase at early times.

However, data for this sample could be fitted by excluding the monomer population data points for $t < \sim 20$ h. For all other sample mixtures, the fit allowed to extract the kinetics rate constants reported in Table 1. From these values, it can be deduced that the reaction rates tend to decrease with increasing amounts of A β (1-40) in the mixture. For the 0% A β (1-40) sample, the forward reactions were dominant as compared to the backward ones, suggesting that the aggregation is close to an irreversible reaction rather than to an equilibrium one. When a small amount of Aβ(1-40) is added (25%), the reaction rates of forward and backward reactions become similar. These results can explain the higher toxicity of this $A\beta$ ratio observed in the literature¹⁹. Indeed, the backward reactions correspond to the dissociation of higher order species into monomers or low molar mass oligomers, which are known to be toxic. To our knowledge, this is the first time that kinetic rates of the aggregation mechanism of AB peptides could be determined by measuring directly the monomer and oligomer distributions. When combined with models proposed in the literature, mainly based on measurements of the time-dependent aggregate mass ^{54, 56, 57} (e.g. by ThT fluorescence assay), or numerical simulations⁵⁸, TDA data as those presented here will help reaching a comprehensive understanding of the aggregation process of these amyloid species, potentially contributing to assess the effectiveness of drugs targeting the toxic oligomeric species.

Conclusion

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This work demonstrated the interest of Taylor dispersion analysis to assist the study of complex amyloid peptide mixtures and shed more light on the aggregation process in these systems. The comparison of the results obtained in parallel by TDA and AFM showed the complementarity of the two techniques, where TDA is able to quantify and size of small objects while AFM can size the fibrillary structures not accessible by TDA. In addition, the results confirmed that the kinetics

of aggregation strongly depend on the nature of amyloid-forming peptides and their environment. Under our working conditions, the more amyloid-prone A β (1-42) aggregates more rapidly (minutes scale) compared to $A\beta(1-40)$, which aggregates in about 24h, in agreement with previously published works^{27, 59}. When mixing together the two species, the aggregation rate was highly influenced by the ratio of $A\beta(1-40)$: $A\beta(1-42)$. Indeed, $A\beta(1-42)$ was found to accelerate the aggregation rate of $A\beta(1-40)$, probably by a cross-seeding mechanism. For example, the disappearance of the monomeric species decreased from 48 h in the case of 100 % $A\beta(1-40)$ to 12 h when 25 % of A β (1-42) were present in the sample. In addition to the clear observation on the interplay between both Aβ peptides during the aggregation process of mixtures and the influence of the Aβ ratio on the aggregation rate, this work shows that this ratio modulates the formation of potentially toxic oligomers. In fact, when the peptides were mixed together, intermediate oligomeric species were observed and tended to increase in proportion upon increasing the A β (1-42) content. Modifying the A β ratio changed the onset of the oligomeric species appearance and monomeric species disappearance (monomers and small oligomers R_h lower than 5 nm), as well as the aggregation mechanism (direct formation of fibrils or formation of intermediate species). These results support the importance of understanding the mechanism of the aggregation process in the case of Aß mixtures (in better accordance with the *in vivo* conditions), to better direct research toward an AD therapy able to inhibit the formation of intermediate species depending on the Aβ ratio. In this respect, TDA was shown to be a straightforward method able to give with unprecedented detail new insight on the size and distribution of the species formed during the aggregation process. Finally, the combination of TDA with extensive data processing and highly resolved efficient methods such as AFM paves the way for building a comprehensive picture of the speciation and growth processes, as illustrated here for

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- amyloid peptides, with potential applications to a wide range of biological, organic and inorganic
- polymer systems.

Table 1. Dependence of various parameters on the proportion of Aβ(1-40) in the studied mixtures. Populations' size: Average hydrodynamic radii of the monomer, oligomers and protofibrils size populations obtained by the deconvolution of the taylorgrams with a finite number of Gaussian functions. Kinetics parameters: values of the rate constants k obtained by fitting the peak areas corresponding to the various populations with equations (1) to (3). The last two columns show the time at which the oligomer population reaches its maximum, $t_{oligomers,max}$, and the initial concentration of the monomer population, M(t=0).

	Populations' size						Kinetic parameters							
% Aβ(1- 40) in Aβ(1- 40):Aβ(1-	Monomer and small oligomers		Higher mass oligomers		Protofibrils		Forward reactions		Backward reactions					
42) mixtures	< <i>R_h</i> > (nm)	± SD	< <i>R_h</i> > (nm)	± SD	< <i>R_h</i> > (nm)	± SD	k _{MO} (h ⁻¹)	$k_{OP} (h^{-1})$	k_{PF} (h^{-1})	k _{OM} (h ⁻¹)	k_{PM} (h^{-1})	$k_{FM} (h^{-1})$	t _{Oligomers} , max (h)	M(t=0) (mAU.min)
100%	2.0	0.1	13.0	8.5	101.2	57.8	N.A.	N.A.	N.A.	N.A.	N.A.	N.A.	N.A.	N.A.
75%	2.4	0.2	9.3	1.8	140.2	43.7	1.106	0.817	0.943	1.497	0.123	0.050	4.171	3.551
50%	2.0	0.2	13.1	10.2	89.5	39.4	0.182	0.240	0.171	0.179	0	0.018	0.884	4.268
25%	2.0	0.2	17.9	10.4	135.6	56.9	1.457	1.226	2.190	1.386	1.332	0.045	0.722	3.709
0%	1.8	0.1	19.1	9.8	99.5	40.7	2.991	3.409	3.704	0	0.998	0.048	0.309	4.346

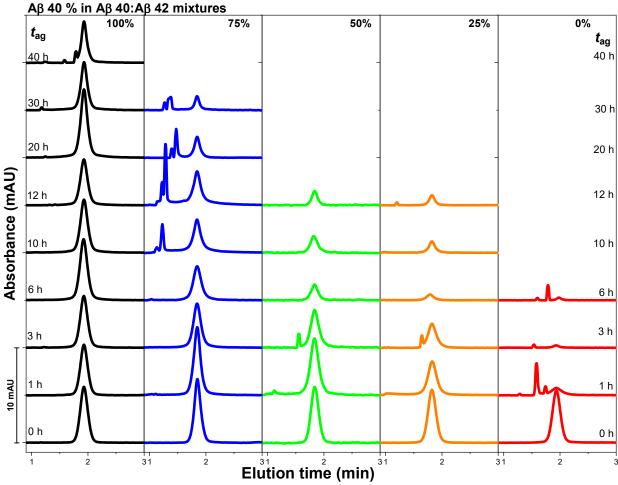


Figure 1. Overview of selected taylorgrams obtained during the aggregation process of Aβ(1-40) :Aβ(1-42) mixtures (Aβ(1-40) % of 100 %; 75%; 50%; 25% and 0%) at different common incubation times. Experimental conditions: Sample: 133 μM of total peptide; 20 mM phosphate buffer, pH 7.4. Incubation: quiescent conditions at 37 °C. Fused silica capillaries: 50 μm i.d. × 40 cm × 31.5 cm. Mobile phase: 20 mM phosphate buffer, pH 7.4. Mobilization pressure: 100 mbar. Injection: 44 mbar for 3 s, $V_i \approx 7$ nL ($V_i / V_d \approx 1$ %). Analyses were performed at 37 °C. UV detection at 191 nm.

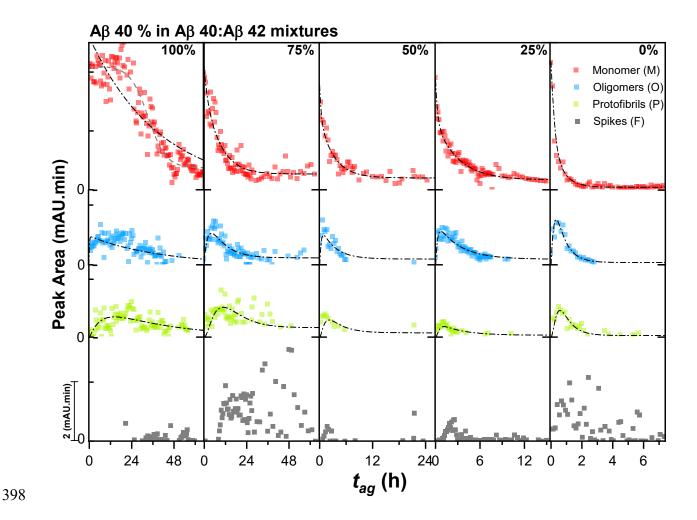


Figure 2. Peak area evolution of the different populations observed during the aggregation process of $A\beta(1-40)$: $A\beta(1-42)$ mixtures and obtained using a fit with finite number of Gaussian functions. Three size populations are represented: monomer and low molar mass oligomers (•), higher molar mass oligomers (•), soluble protofibrils (•) and non-diffusing species ("spikes") (•). Experimental conditions as in Figure 1. Black dashed-dotted lines are the kinetic fits obtained from equations (1) to (3). The grey dashed line in $A\beta(1-40)$ monomer distribution is a guide for the eye. The scale of the vertical axis is the same in all graphs and is indicated on the bottom-left corner of the figure.

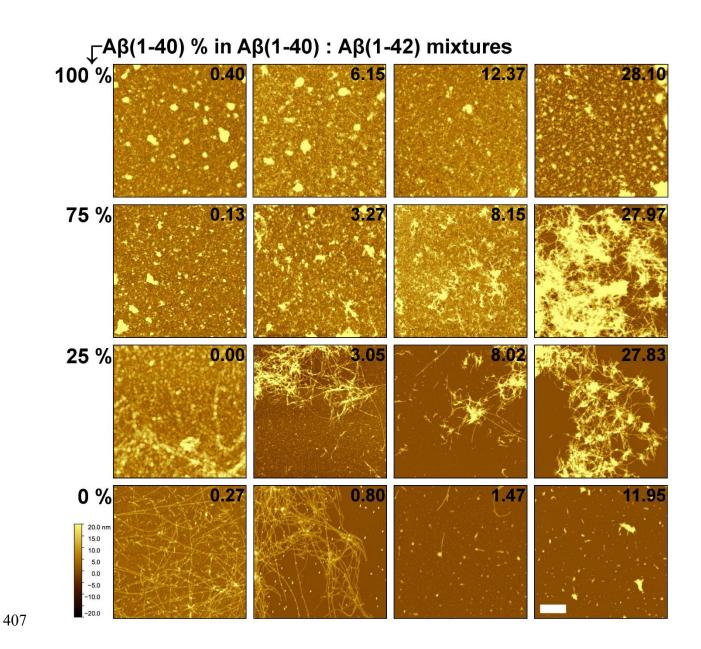


Figure 3: AFM images taken in intermittent contact mode under dry conditions, for different key times of the aggregation process in the presence of different peptide proportions. The $A\beta(1-40)$: $A\beta(1-42)$ ratio is given on the left side of the panel, aggregation times in hours are overlayed on the images. The scale bar on the bottom right corner is valid for all images and equals 1 μ m.

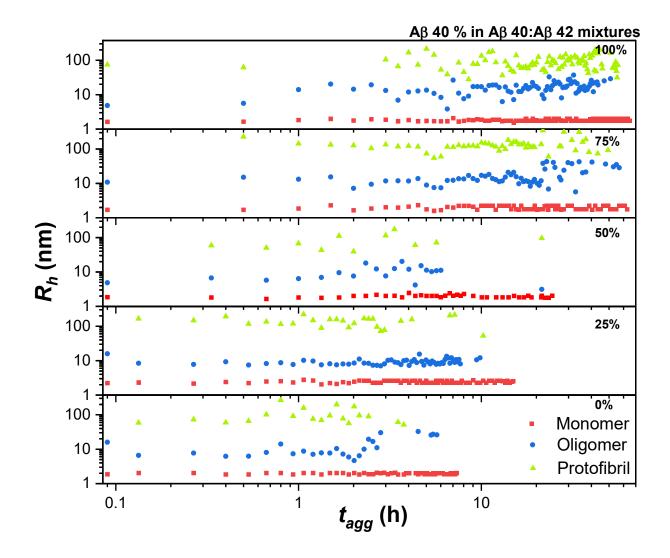


Figure 4. Hydrodynamic radius evolution of the different populations observed during the aggregation process of $A\beta(1-40)$: $A\beta(1-42)$ mixtures and obtained using a fit with finite number of Gaussian functions. Three size populations are represented: monomer and low molar mass oligomers $0.9 < R_h < 5$ nm (\blacksquare), higher molar mass oligomers $5 < R_h < 50$ nm (\blacksquare), and soluble protofibrils $50 < R_h < 300$ nm (\blacksquare). Experimental conditions as in Figure 1.

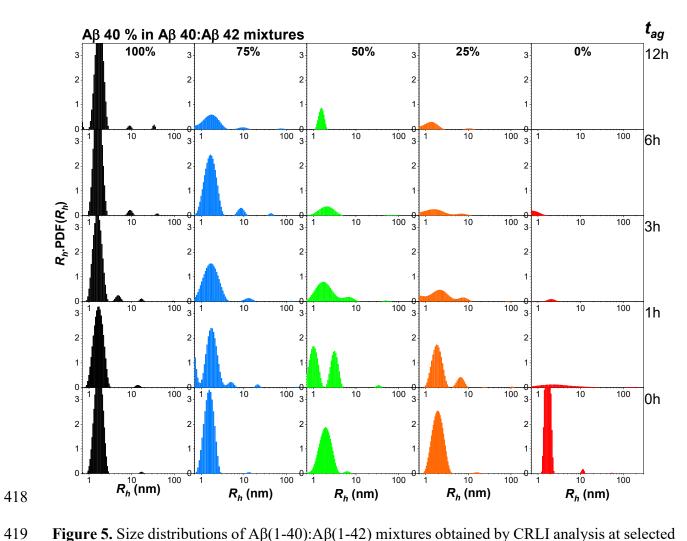


Figure 5. Size distributions of A β (1-40):A β (1-42) mixtures obtained by CRLI analysis at selected incubation times $t_{ag} = 0$ to 12 h. Experimental taylorgrams corresponding to these distributions are shown in Figure 1, the experimental conditions are as in Figure 1.

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AUTHOR CONTRIBUTIONS

Conceptualization – HC and JC designed the experiments. Investigation – MD realized the TDA experiments and treated the data. OD realized the AFM experiments. Methodology – LC and JC fitted the data with the kinetic equations. Software – LC developed the script for the kinetic modelling and the CONTIN TDA data treatment. Supervision – JC and HC supervised MD for TDA experiments. JFH supervised MD for peptide synthesis. Validation – JC and HC validated and interpreted the TDA results. OD and CB validated and interpreted the AFM results. Writing – original draft – JC wrote the first draft and made the figures. OD made figure 3. Writing – review & editing – All authors read and edited the manuscript. All authors have given approval

Supporting information

to the final version of the manuscript.

Theoretical aspects of TDA, experimental elution profiles, average hydrodynamic radii over the

whole aggregation process of each population and for all mixtures; CRLI analysis for all mixtures,

and the kinetic fits.

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