

## A one-pot synthesis and X-Ray structural characterization of new highly substituted-allyl carbamates

Sondes Ghrairi, Pascal Retailleau, Benoit Crousse, Thouraya Barhoumi-Slimi

### ▶ To cite this version:

Sondes Ghrairi, Pascal Retailleau, Benoit Crousse, Thouraya Barhoumi-Slimi. A one-pot synthesis and X-Ray structural characterization of new highly substituted-allyl carbamates. Journal of Molecular Structure, 2022, 1258, pp.132548. 10.1016/j.molstruc.2022.132548. hal-03796662

## HAL Id: hal-03796662 https://hal.science/hal-03796662v1

Submitted on 5 Oct 2022

**HAL** is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers. L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

A one-pot synthesis and X-Ray structural characterization of new highly substituted-allyl carbamates

Sondes Ghrairi, Pascal Retailleau, Benoit Crousse, Thouraya Barhoumi Slimi

 PII:
 S0022-2860(22)00221-6

 DOI:
 https://doi.org/10.1016/j.molstruc.2022.132548

 Reference:
 MOLSTR 132548



To appear in: Journal of Molecular Structure

Received date:3 September 2021Revised date:13 December 2021Accepted date:3 February 2022

Please cite this article as: Sondes Ghrairi, Pascal Retailleau, Benoit Crousse, Thouraya Barhoumi Slimi, A one-pot synthesis and X-Ray structural characterization of new highly substituted-allyl carbamates, *Journal of Molecular Structure* (2022), doi: https://doi.org/10.1016/j.molstruc.2022.132548

This is a PDF file of an article that has undergone enhancements after acceptance, such as the addition of a cover page and metadata, and formatting for readability, but it is not yet the definitive version of record. This version will undergo additional copyediting, typesetting and review before it is published in its final form, but we are providing this version to give early visibility of the article. Please note that, during the production process, errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.

© 2022 Published by Elsevier B.V.

#### A one-pot synthesis and X-Ray structural characterization of

#### new highly substituted-allyl carbamates

Sondes Ghrairi<sup>a</sup>, Pascal Retailleau<sup>b</sup>, Benoit Crousse<sup>c</sup>, Thouraya Barhoumi-Slimi<sup>\*a,d</sup>

<sup>a</sup> University of Tunis El Manar, Laboratory of Structural (bio)Organic Chemistry and Polymers LR99ES14, Department of Chemistry, Faculty of Sciences of Tunis, 2092, Tunis, Tunisia.

<sup>b</sup> Université Paris-Saclay, CNRS, Institut de Chimie des Substances Naturelles, 91198 Gif-sur-Yvette, France.

<sup>e</sup> UMR 8076, BioCIS, Univ. Paris-Sud, CNRS, Université Paris-Saclay, 92290 Châtenay-Malabry, France.

<sup>d</sup> University of Carthage, High Institute of Environmental Sciences and Technologies, Technopark of Borj-Cedria BP-1003, Hammam-Lif 2050, Tunisia.

Email thouraya.barhoumi@isste.u-carthage.tn

#### **Research highlights**

- One-pot synthesis of new variously substituted allylic carbamates starting from β-chloro-α,β-unsaturated aldehydes in high yields.
- Full characterization of the new compounds and the corresponding intermediates by NMR and IR spectroscopic techniques and HRMS.
- X-ray crystallography structural determination of two allyl carbamates.

#### ABSTRACT

A one-pot synthesis of new variously substituted  $\gamma$ -chloro- $\gamma$ , $\beta$ -allylic carbamates have been carried out starting from  $\beta$ -chloro- $\alpha$ , $\beta$ -unsaturated aldehydes. The new carbamates and the corresponding intermediates, the  $\gamma$ -chloro- $\gamma$ , $\beta$ -allylic alcohols, were fully characterized by (<sup>1</sup>H and <sup>13</sup>C) NMR, IR and HRMS spectroscopic techniques. The crystal structures of two allylic carbamates, were established by single-crystal X-ray diffraction.

Keywords: β-chlorovinylaldehydes; one-pot synthesis; allylic carbamates; X-ray diffraction.

#### 1. Introduction

Carbamates and allylic carbamates in particular, are of considerable interest due to their important role in various areas. Indeed, they belong to structural or functional part of many approved drugs [1-5] and prodrugs [6], which are used for the treatment of various diseases such as Alzheimer's disease [7], cancer [8], hepatitis C [1, 9], HIV infection [10], anxiety [11] and epilepsy [11]. They are found to be important raw materials in the synthesis of polyurethanes [12-14], paints [15], wood preservatives [15], pesticides, herbicides, insecticides and fungicides [16]. They are also used as intermediates in numerous organic synthesis [17, 18] as protecting amino groups in peptide synthesis [19, 20], polymers synthesis [21] and as linkers in combinatorial chemistry [22] as well as in the cyclic [23] and heterocyclic compounds synthesis [24-27]. Therefore continuous efforts have been

made to obtain this type of compounds, particularly the allylic carbamates, through simple and efficient methods. The conventional method for preparation is the reaction of an isocyanate with the corresponding allyl alcohol [28, 29] or through a *trans*-carbamoylation reaction of the hydroxyl group with the phenyl carbamate [30]. Coupling reaction of aryloxyallenes with CO<sub>2</sub> and amines using a silver-catalyzed is also reported [31] as well as the reaction of carbonyl compounds with N,N-dialkyl carbamoyl chloride [32]. However, the structural variation of allyl carbamates have shown serious limitations due to either the limited availability of various allyl alcohols bearing tri- or tetra-substituted double bond [28] or by the use of palladium complexes as catalysts strong bases [33] and strong oxidants [34] which are not suitable for allyl carbamates since they can promote unwanted cyclization reactions [35].

In our previous works, we have explored the reactivity of the  $\beta$ -chlorovinyl aldehydes as precursors of highly substituted  $\beta$ -fluoroalkoxyvinyl aldehydes and corresponding alcohols [36], benzenes [37], allylic alcohols [38], phosphoramidates [38], phosphorates and phosphates [39]. As an extension of our work, we report here the one-pot synthesis of a new series of variously substituted allylic carbamates. The new compounds were obtained in mild conditions with good to excellent yields and were fully confirmed by (<sup>1</sup>H and <sup>13</sup>C) NMR, IR spectroscopies and HRMS. The structures of two allyl carbamates were determined by X-Ray crystallography.

#### 2. Results and discussion

#### 2.1. Synthesis

The starting substrates **1**, **2** and **4-9** were synthesized from the corresponding  $\alpha$ -methylene ketones using the Vilsmeier-Haack reaction according to literature procedures [40]. The synthesis of the new  $\gamma$ -chloro- $\beta$ , $\gamma$ -unsaturated alcohols (**11**, **12** and **14-18**) was performed via carbonyl reduction of the  $\beta$ -chloro- $\alpha$ , $\beta$ -unsaturated aldehydes. The chlorovinylaldehydes and cinnamaldehyde (**1-10**) reacted with sodium borohydride as reducing agent in dry THF and gave the products (**11-20**) in good yields (Table1).

**Table1:** Optimized synthesis of  $\gamma$ -chloro- $\beta$ , $\gamma$ -unsaturated alcoohols (11-20).

		78-91%
1-10		11-20
R <sup>2</sup> <sup>∬</sup> R <sup>3</sup> 1eq	Dry THF/T°(C) time (h)	$R^{2} R^{3}$
	NaBH <sub>4</sub> (0.25eq)	

Substrate	<b>R</b> <sup>1</sup>	<b>R</b> <sup>2</sup>	<b>R</b> <sup>3</sup>	time(h)	T(°C)	Product	Yield <sup>b</sup>	Ratio(Z/E) <sup>c</sup>
$(\mathbb{Z}/\mathbb{E})^{\mathrm{a}}$								
1 (100/0)	-(CH <sub>2</sub> CH <sub>2</sub> )-	-(CH <sub>2</sub> CH <sub>2</sub> )-	Cl	3	Reflux	11	92	(100/0)
<b>2</b> (100/0)	-(CH <sub>2</sub> CH <sub>2</sub> )-	-(CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> )-	Cl	3	Reflux	12	96	(100/0)
<b>3</b> (0/100)	Н	Ph	Н	2	Reflux	13	89	(0/100)
4 (95/5)	Н	Ph	Cl	4	Reflux	14	92	(94/6)
<b>5</b> (5/95)	CH <sub>3</sub>	Ph	Cl	5	Reflux	15	87	(5/95)
<b>6</b> (100/0)	Н	<sup>t</sup> Bu	Cl	4	Reflux	16	84	(100/0)
7 (95/5)	Н	CO <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	Cl	2	Reflux	17	78	(95/5)

<b>8</b> (26/74)	CH <sub>3</sub> CH <sub>2</sub>	CH <sub>3</sub>	Cl	5	Reflux	18	76	(24/76)
<b>9</b> (61/39)	Ph	CH <sub>3</sub>	Cl	5	Reflux	19	75	(64/36)
<b>10</b> (83/17)	CO <sub>2</sub> Et	CF <sub>3</sub>	Cl	12	(-10)	20	74	(83/17)

<sup>a</sup>: Z/E isomer ratios measured from <sup>1</sup>H NMR aldehydic signals <sup>b</sup> yields measured after purification. <sup>c</sup>: Z/E Isomer ratios measured in crude products from <sup>1</sup>H-NMR CH<sub>2</sub>OH signals.

Sodium borohydride reacted regioselectively with the vinylic aldehydes **1-10** in dry THF in good to excellent yields. The obtained pale yellow or colorless oils of allylic alcohols **11-20** are stable at room temperature with Z/E proportions identical to those of the starting aldehydes.

Different reaction conditions were attempted to optimize the synthesis of the target carbamates. When the reaction of N-phenyl isocyanate with the allylic alcohol **12** was carried out without any base in dry THF at 25°C, the desired product **27** was observed after 24 h in moderate conversion percentage (Table 2, entry 1). The use of  $Et_3N$  with slight excess gave total conversion (>99%) with excellent yield (entry 3). In CH<sub>2</sub>Cl<sub>2</sub> or  $Et_2O$ , the reaction led to lower conversion percentages (entries 4 and 5). The various explored conditions are displayed in Table 2.

	1		1	*
Entry	Et <sub>3</sub> N (eq.)	Solvent	Time (h)	Conversion [%] <sup>a</sup>
1	0	THF	24	44
2	1	THF	12	96
3	1.5	THF	6	>99
4	1.5	DCM	12	74
5	1.5	Et <sub>2</sub> O	12	72

Table 2: Optimization of the reaction conditions of the allylic carbamate 27 at 25°C

<sup>a</sup>: The conversion % was monitored by <sup>1</sup>H NMR spectroscopy in crude products.

With the optimal reaction conditions outlined above, the synthesized  $\gamma$ -chloro allylic alcoohls **11-20** were subjected to carbamoylation reaction with commercial N-alkyl and N-aryl-isocyanates in dry THF in mild basic medium at room temperature affording successfully new series of enol carbamates **20-45** in moderate to excellent yields (52-95%). Moreover, both carbonyl reduction and carbamoylation were succesfully realized in one-pot procedure, starting from the aldehydes. The resulting variously substituted carbamates from the cyclic aldehydes **21-28** (Table 3), of the cinnamaldehydes **29-37** (Table 4) or from the acyclic  $\alpha$ , $\beta$ -unsaturated aldehydes (Table 5), were obtained as viscous oils or solids, soluble in both CHCl<sub>3</sub> and DMSO. All the reactions were monitored by TLC and <sup>1</sup>H-NMR. All final products were stable at 0°C and were characterized by IR, (<sup>1</sup>H and <sup>13</sup>C) NMR spectroscopies and HRMS.

**Table 3:** one-pot synthesis of carbamates from  $\sigma$ ,  $\beta$ -unsaturated cyclic aldehydes (21-28)

0

		1 <u>)</u> NaBH <sub>4</sub> (0.25eq), Dry THF, Re 2) Et <sub>3</sub> N (1.5eq), RNCO (1.3eq),	flux,(4h) rt, 12h
	1eq 1,2		21-28
Entry	Substrate	R	Product (Yield%) <sup>a</sup>
1	1	CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub>	21 (69)
2	1	CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub>	<b>22</b> (91)
3	1	Ph	<b>23</b> (93)

4	1	3-ClPh	<b>24</b> (68)
5	2	CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub>	25 (88)
6	2	CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub>	<b>26</b> (87)
7	2	Ph	<b>27</b> (91)
8	2	3-ClPh	<b>28</b> (78)

<sup>a</sup>: yields after purification.

 Table 4: One-pot synthesis of carbamates from cinnamaldehyde derivatives (29-37)



Entry	Substrate	<b>R</b> <sup>3</sup>	Products	Yield <sup>a</sup>	Ratio(Z/E) <sup>b</sup>
1	3	CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub>	29	86	(0/100)
2	3	CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub>	30	96	(0/100)
3	3	Ph	31	93	(0/100)
4	3	3-ClPh	32	52	(0/100)
5	4	CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub>	33	63	(92/8)
6	4	3-ClPh	34	82	(93/7)
7	5	CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub>	35	62	(3/97)
8	5	Ph	36	86	(4/96)
9	5	3-ClPh	37	62	(24/76)

a: yields after purification ; b: Z/E Isomer percentages measured in crude products from<sup>1</sup>H NMR spectra.

**Table 5:** One-pot synthesis of carbamates from alkylated  $\alpha$ ,  $\beta$ - unsaturated aldehydes (38-45)

o		٩
R <sub>1</sub>	1) NaBH₄ (0.25eq), dry THF, T(°C)/t(h)	R <sub>1</sub>
R <sub>2</sub> <sup>M</sup> CI	2) Et <sub>3</sub> N (1.5eq), R <sub>3</sub> NCO (1.3eq), rt, 15 h	R <sub>2</sub> <sup>N</sup> CI
1eq 6.7 and 8		38-45

 $R_3$ 

Entry	Substrate	$\mathbb{R}^3$	T(°C)/t(h	Product	Yield (%) <sup>a</sup>	Ratio(Z/E) <sup>b</sup>
1	6	Ph	Reflux./4	38	65	(100/0)
2	8	CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub>	Reflux./4	39	73	(30/70)
3	8	Ph	Reflux./4	40	50	(23/77)

Journal Pre-proof								
4	8	3-ClPh	Reflux./4	41	59	(23/77)		
5	9	Ph	Reflux./4	42	94	(66/34)		
6	9	3-ClPh	Reflux./4	43	86	(66/34)		
7	10	CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub>	-10/12	44	67	(83/17)		
8	10	Ph	-10/12	45	73	(85/15)		

<sup>a</sup>: yields after purification ; <sup>b</sup>: Z/E Isomer percentages measured in crude products from<sup>1</sup>H NMR spectra.

Analysis of the <sup>1</sup>H-NMR spectra showed the formation of the target carbamates by the presence of signals assigned to both N=CH and CONH protons. However, for the carbamate **42**, these signals appear to be duplicated due to *cis* and *trans* conformers. In particular, the compound **36**, after a long term storing or recording the corresponding spectrum in DMSO as solvent instead of CDCl<sub>3</sub>, the additional signals of the second conformer are observed. In all spectra, The NH-Ph signals had moved substantially from chemical shifts 6.6-7.58 ppm in CDCl<sub>3</sub> to 9.7-9.9 ppm in DMSO without significantly affecting the *cis* and *trans* ratios. This is obviously due to the favored formation of the H-bond between the carbamate and DMSO. This is in accordance with our previous work [41].

Table 6: (  $\delta_{NH}$  ppm) cis/trans-amide conformers and ratios in CDCl3 and DMSO-d<sub>6</sub>

Product	$(\delta_{\rm NH})$ CDCl <sub>3</sub>	$(\delta_{\rm NH})$ DMSO-d <sub>6</sub>
	%(cis/trans) <sup>a</sup>	%(cis/trans) <sup>a</sup>
24	7.54 (0/100)	9.97 (0/100)
31	7.17 (0/100)	9.74 (0/100)
36	6.94/7.58 (53/47)	9.98/9.74 (59/41)
38	6.82 (0/100)	9.74 (0/100)
	Product 24 31 36 38	$\begin{array}{c c} Product & (\delta_{NH}) \ CDCl_3 \\ & \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \$

a: cis/trans-amide ratio measured at room temperature from 'H-NMR NH signals integration in CDCl3 and DMSO-d6.

The same behavior was observed in <sup>13</sup>C-NMR spectra of compounds **36** and **42**, which revealed duplication not only around the CH-N*H* bonds, but also in more remote parts of the molecules, such as the carbonyl and the substituents in  $\gamma$  and  $\beta$  positions of the carbamates. The <sup>13</sup>C-NMR spectra of carbamates **23** and **27** in CDCl<sub>3</sub> show only one set of peaks corresponding to a single rotamer. The 2D-NOESY spectrum exhibited a cross correlation between NH and CH<sub>2</sub>O in compound **23** and **27** only indicating the *trans* conformation of OR group with respect of NH. However, no indication on the *trans* NH-CO conformation was collected from the spectrum.

On the other hand, it has been reported that, in IR spectroscopy of compounds bearing NH-CO group, the NHstretching vibration  $v_{NH}$  at 3200 cm<sup>-1</sup> is assigned to the *cis*-amide whereas the  $v_{NH}$  above 3200 cm<sup>-1</sup> is attributed to the *trans*-amide with in addition, a characteristic band at 1540 cm<sup>-1</sup> [42]. In most cases, the infrared spectra of the compounds **20-45** in pure state (Table 6) showed two  $v_{NH}$  stretching bands; a large vibrational band ranging from 3330 to 3350 cm<sup>-1</sup> related to bound NH and a thin vibrational band ranging from 3420 to 3460 cm<sup>-1</sup> related to free NH. Furthermore, the presence of a thin and intense bending band at 1513-1590 cm<sup>-1</sup> indicate reliably the *trans* conformation of N*H*-CO. All the spectra show also intense bands related to the carbonyl C=O at about 1620–1700 cm<sup>-1</sup> and to the double bond C=C at 1625– 1685 cm<sup>-1</sup>. All the characteristic bands are summarized in Table 7.

**Table 7:** IR data of allylic carbamates **21-45** in cm<sup>-1</sup>

Entry	Product	Vr	лн	δ <sub>NH</sub>	ν с=0	<b>v</b> c=c
1	21	-	3346	1526	1711	1695
2	22	3452	-	1515	1712	1619
3	23	3425	-	1524	1732	1600
4	24	3432	-	1523	1732	1598
5	25	3345	3337	1516	1710	1695
6	26	3452	3334	1516	1765	1704
7	27	3435	-	1524	1732	1602
8	28	3421	3325	1523	1735	1598
9	29	3448	-	1514	1720	1695
10	30	-	3340	1519	1695	-
11	31	3435	-	1524	1728	1601
12	32	3420	3338	1524	1731	1597
13	33	3446	3353	1514	1716	1638
14	34	3420	3336	1524	1732	1597
15	35	3445	3351	1515	1715	1698
16	36	3423	3338	1524	1732	1599
17	37	3406	3320	1524	1708	1594
18	38	3423	3336	1525	1731	1600
19	39	3469	3335	1532	1692	1660
20	40	3328	-	1524	1711	1600
21	41	3432	-	1523	1731	1598
22	42	3431	3301	1525	1727	1600
23	43	3432	-	1523	1733	1598
24	44	3450	-	1518	1735	1651
25	45	3432		1526	1740	1601

#### Crystallographic data collection, structure determination and refinement

The proposed structures of allylic carbamates 24 and 36 were confirmed by single crystal X-ray diffraction (SCXRD) analysis. The suitable crystals for both compounds were prepared from their solution in cyclohexane/diethylether (1:4) by slow evaporation of the solvent. The corresponding intensities datasets were measured at room temperature using a RIGAKU XtaLabPro diffractometer equipped with a Mo microfocus sealed tube MM003 generator coupled to a double-bounce confocal Max-Flux® multilayer optic and a HPAD PILATUS3R 200K detector. CrysAlisPro 1.171.39.46 [43] was employed for the data processing, with SCALE3 ABSPACK scaling algorithm implemented for the empirical absorption correction using spherical harmonics. The structures were readily solved by intrinsic phasing methods (SHELX-T program) [44]: one molecule of 24 was placed in the asymmetric unit (asu) of the orthorhombic unit cell, space group Pna21 and four independent conformers of 36 in that of the monoclinic unit cell, space group P21. All the non-hydrogen atoms were refined with anisotropic displacement parameters by full-matrix least-squares methods on  $F^2$  using SHELX-L. [45] The crystal studied of 36 turned out to be an inversion twin whose component mass ratio refined to ca 0.52:0.48. H atoms could be identified in difference maps and therefore the methyl H atoms borne by C11 were allowed as rigid groups to rotate but not tip, so as to match the electron-density maxima. The other H atoms were positioned geometrically and refined with  $U_{iso}$  set to  $xU_{eq}(C)$ , x=1.2 whereas x = 1.5 for the methyl carbons. The positions of the hydrogen atoms in the amide groups were refined with N-H distance restrained to target 0.87(2) Å in 24 or to be similar in 36, with  $U_{iso}$  set to  $1.2U_{eq}(N)$  (only for 36,  $U_{iso}$  being freely refined in 24). Figure 1 shows the molecular structure of 24 whose correct orientation relative to the polar axis direction from the crystal picked up for data collection was established by means of the Flack x parameter; [46] x = -0.05 (4), calculated [46] using 1334 quotients of the type  $[(I^+) - (I^-)]/[(I^+) + (I^-)]$  and one copy of the highly flexible E-isomer molecular structure of **36**. The conformational differences between the four copies are highlighted in an overlay diagram shown in Figure 2. In its asu, these molecules are elongated along the [010] direction and those relatively oriented in parallel way are linked by N–H ... O hydrogen bonds [graph set  $C^1_1(4)$  commonly found in structures made of simple amides, as well in the **24** one] developing along the [0 0 1] direction molecular chains (Figure 3), stacking on top of each other in antiparallel way in the *a* direction and segregated at the x, <sup>1</sup>/<sub>4</sub>, z and x, <sup>3</sup>/<sub>4</sub>, z interfaces of 2<sub>1</sub> symmetry-related (chloromethyl)benzene moieties through edge-to-face and Cl...Cl contacts ( $d_{Cl-Cl}=3.79\text{Å} \ge \text{van}$  der Waals radii sum). In **24**, the N–H ... O chains propagate infinitely along the [1 0 0 ] direction developing paddlewheel structure motifs centered around x, 0.25, 0 and x, 0.75, 0.5 positions, involving molecules related by the glide plane perpendicular to [0 1 0 ] with glide component 0.5, 0, 0 (Figure 4). No halogen contacts are noticed in the **24** crystal packing. Crystal data, data collection and structure refinement details are summarized in Table 8. CCDC 2084634 (**24**) - 2084224 (**36**) contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from the Cambridge Crystallographic Data Centre via <u>www.ccdc.cam.ac.uk/data\_request/cif</u>.

Compound	24	36	
		CI C	
	(2-chlorocyclohex-1-en-1-	(E)-3-chloro-2-methyl-3-	
	yl)methyl(3-chlorophenyl)carbamate	phenylallyl phenylcarbamate	
Empirical formula	C <sub>14</sub> H <sub>15</sub> Cl <sub>2</sub> N O <sub>2</sub>	C <sub>17</sub> H <sub>16</sub> Cl N O <sub>2</sub>	
Formula weight	300.17	301.76	
Temperature (K)	293(2)	293(2)	
Wavelength (Å)	0.71073	0.71073	
Crystal system,	Orthorhombic,	Monoclinic,	
space group	$P na2_1$	P 2 <sub>1</sub>	
Unit cell dimensions   (Å)	9.9172(7) 10.2917(8) 14.0882(10) 90	9.2313(5) 34.2963(12) 10.0818(5) 90	
	90	90	
Volume (Å <sup>3</sup> )	1437.91(18)	3068.5(3)	
Z, $Calculated density (Mg/m3)$	4,	8,	
Absorption coefficient ( $mm^{-1}$ )	0.448	0.252	
F(000)	1264	1264	
Crystal size (mm)	0.44 x 0.20 x 0.17	0.45 x 0.23 x 0.05	
$\theta$ range for data collection (°)	4.063 to 27.102	3.639 to 25.350	
Limiting indices	$-12 \le h \le 12,$ $-13 \le k \le 13,$ $-18 \le l \le 18$	$-10 \le h \le 11,$ $-41 \le k \le 40,$ $-11 \le 1 \le 12$	
Reflections collected / unique	14153 / 3127 0.0749	22490 / 10718 0.0434	

Table 8: Crystal data and structure refinement for compounds 24 and 36

R(int)					
Completeness to $\theta_{\text{full}}$ (%)	99.6 99.6		99.6		
Absorption correction	Semi-empirical from equivalents				
Max. and min. transmission	1.000 and 0.860 1.000 ar		1.000 and 0.468		
Refinement method	Full-matrix least-squares on $F^2$				
Data / restraints / parameters	3127 / 2 / 176 10716 / 7 / 774				
Goodness-of-fit on $F^2$		1.053	0.996		
Final R indices	R1,	0.0445,	0.0501,		
$[I > 2\sigma(I)]$	wR2	0.1224	0.1179		
R indices	R1,	0.0459,	0.0818,		
(all data)	wR2	0.1241	0.1329		
Largest $\Delta$ peak and hole					
$(e.Å^{-3})$		0.285 and -0.200	0.292 and -0.211		
CCDC deposit number		2084634	2084224		



Figure 1 Ortep view of 24 (left), and one 36 conformer (right) in their asu with their respective atom numbering scheme. Displacement ellipsoids are drawn at the 50% probability level. The H atoms are shown as small spheres of arbitrary radii.



Figure 2 Overlay of the four independent molecules of 36.



**Figure 3.** Two independent hydrogen-bonded (N-H...O) chains (underscored by different colors applied to carbon atoms) of molecules in the central part of the unit cell of **36** running along the *c* axis and framed by  $2_1$  symmetry-related molecules. Cyan dashed lines indicate N-H...O intermolecular hydrogen bonds (see table 9).



**Figure 4.** One hydrogen-bonded (N-H...O) chain of the molecules in the unit cell of **24** running along the *a* axis. Cyan dashed lines indicate N-H...O intermolecular hydrogen bonds (see table 9).

D-HA	d(D-H)	d(HA)	d(DA)	<(DHA)
N1-H1O2 <sup>#1</sup>	0.85(2)	2.11(3)	2.942(3)	169(4)
N1(1)-H1N(1)O2(2) N1(2)-H1N(2)O2(1) <sup>#2</sup> N1(3)-H1N(3)O2(4) N1(4)-H1N(4)O2(3) <sup>#3</sup>	$\begin{array}{c} 0.88(3) \\ 0.87(3) \\ 0.87(3) \\ 0.85(3) \end{array}$	2.21(4) 2.18(3) 2.18(3) 2.19(4)	3.013(5) 3.030(5) 3.032(5) 2.998(5)	152(4) 168(5) 163(5) 160(5)

<b>Table 9:</b> Hydrogen bonds for 24 and 36 [A and
---

Symmetry transformations used to generate equivalent atoms: (24) #1 x+1/2,-y+3/2,z; (36) #2 x,y,z-1 #3 x,y,z+1

#### 3. Conclusion

In summary, new series of  $\gamma$ -chloro- $\gamma$ , $\beta$ -unsaturated alcohols and corresponding enol carbamates have been synthesized from  $\beta$ -chloro- $\alpha$ , $\beta$ -unsaturated aldehydes using a facile and efficient method in most cases. The major advantage of this transformation is the non-use of metal catalysts or toxic reagents. The desired compounds were obtained in high yields and characterized by different spectroscopic techniques and X-Ray crystallography.

#### Experimental Section Instruments and chemicals

All solvents were dried, distilled and stored over molecular sieves (4 Å) using standard techniques. [47] The starting aldehydes were prepared using Vilsmeir–Haack reaction. [40] TLC was performed on silica gel, 60F-250 (0.26 mm thickness) plates. The plates were visualized using a UV lamp light (254 nm) or by use of potassium permanganate and 5 g K<sub>2</sub>CO<sub>3</sub>/100 mL water. NMR spectra were performed on Bruker AVANCE 300 MHz spectrometer at the Faculty of Pharmacy at University Paris-Saclay in France or an Ultrafield AVANCE 300 MHz spectrometer at High Institute of Environmental Sciences and Technology at University of Carthage in Tunisia. All NMR spectra were recorded at ambient temperature (290K) unless otherwise noted. <sup>1</sup>H NMR spectra are reported as follows: chemical shift (multiplicity, coupling constant, integration). The following abbreviations are used to indicate multiplicities: s, singlet; d, doublet; t, triplet; q, quartet; m, multiplet. Chemical shifts ( $\delta$ ) were given in parts per million (ppm) from tetramethylsilane with the solvent resonance as the internal standard in CDCl<sub>3</sub> solution. High resolution masse spectra were obtained using a Bruker Esquire electrospray ionization apparatus at the Faculty of Pharmacy at University Paris-Saclay in France.

FTIR spectra were recorded on Nicolet IR200 spectrometer within the wave number range of 4000-400cm<sup>-1</sup> at 25°C at High Institute of Environmental Sciences and Technology at University of Carthage in Tunisia.

Unless otherwise noted, commercially available NaBH<sub>4</sub>, isocyanates and triethylamine were purchased from various commercial sources (Acros, Aldrich) and used without further purification.

#### General Procedure for preparation of $\gamma$ -chloro- $\gamma$ , $\beta$ -allylic alcohols (11-19).

To a stirred solution of  $\beta$ -chloro- $\alpha$ , $\beta$ -vinylic aldehyde (1 mmol) in dry THF (10 mL) was added NaBH<sub>4</sub> (0.25 mmol) at 0°C. The reaction mixture was stirred at adequate temperature. The progress of the reaction was monitored by TLC until the conversion of  $\beta$ -chloro- $\alpha$ , $\beta$ - unsaturated aldehydes was complete. Then, cold distilled water (100 mL) was added to the mixture. The product was extracted with diethyl ether (3\*30mL) and washed by 10 mL of a saturated solution of NaCl. The solvent (diethyl ether) was evaporated, and the crude product was purified by short-column chromatography using (cyclohexane:diethylether) (8:2) as eluent. The same procedure will be used for the reduction of cinnamaldehyde.

#### (2-chlorocyclohex-1-en-1-yl) methanol (11)

Light Yellow oil, **FTIR: neat (cm<sup>-1</sup>)**: v 3317 (OH), 1663 (C=C); <sup>1</sup>H NMR (**300 MHz, CDCl**<sub>3</sub>) **δ**: 4.16 (s, 2H, C<u>H</u><sub>2</sub>OH), 3.37 (s, 1H, O<u>H</u>), 2.27 (m, 2H, C<u>H</u><sub>2</sub>cyc), 2.18-2.16 (m, 2H, C<u>H</u><sub>2</sub>cyc), 1.66-1.56 (m, 4H, C<u>H</u><sub>2</sub>C<u>H</u><sub>2</sub>cyc); <sup>13</sup>C NMR (**75 MHz, CDCl**<sub>3</sub>) **δ**: 132.52 (<u>C</u>-Cl), 128.35 (C=<u>C</u>CH<sub>2</sub>OH), 62.68 (<u>C</u>H<sub>2</sub>OH), 33.81 (<u>C</u>H<sub>2</sub>cyc), 27.78 (<u>C</u>H<sub>2</sub>cyc), 23.79 (<u>C</u>H<sub>2</sub>cyc), 22.13 (<u>C</u>H<sub>2</sub>cyc).

#### (2-chlorocyclohept-1-en-1-yl) methanol (12)

Light Yellow oil, **FTIR: neat (cm<sup>-1</sup>)**: v 3307 (OH), 1651 (C=C); <sup>1</sup>**H** NMR (**300 MHz, CDCl**<sub>3</sub>) δ: 4.21 (s, 2H, CH<sub>2</sub>OH), 2.58-2.55 (m, 2H, CH<sub>2</sub>cyc), 2.50 (s, 1H, O<u>H</u>), 2.34-2.30 (m, 2H, CH<sub>2</sub>cyc), 1.73-1.48 (m, 6H, CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>cyc); <sup>13</sup>C NMR (**75 MHz, CDCl**<sub>3</sub>) δ: 137.51 (<u>C</u>-Cl), 132.56 (<u>C</u>=CCl), 64.75 (<u>C</u>H<sub>2</sub>OH), 38.87 (<u>C</u>H<sub>2</sub>cyc), 31.24 (<u>C</u>H<sub>2</sub>cyc), 30.37 (<u>C</u>H<sub>2</sub>cyc), 26.10 (<u>C</u>H<sub>2</sub>cyc), 25.23 (<u>C</u>H<sub>2</sub>cyc).

#### 3-Phenylprop-2-en-1-ol (13)

Colorless oil , **FTIR: neat (cm<sup>-1</sup>)**: v 3308 (OH), 1657 (C=C); <sup>1</sup>**H NMR** (**300 MHz, CDCl**<sub>3</sub>)  $\delta$  : 7.42-7.25 (m, 5H, H<sub>arm</sub>), 6.64 (d, <sup>3</sup>J<sub>H-H</sub>= 15.9Hz, 1H, C<u>H</u>Ph=CH), 6.39 (dt, <sup>3</sup>J<sub>H-H</sub>= 15.9Hz, <sup>3</sup>J<sub>H-H</sub>= 5.66Hz, 1H, CHPh=CH), 4.33 (m, 2H, CH<sub>2</sub>O), 2.28 (t, <sup>3</sup>J<sub>H-H</sub>= 5.59Hz, 1H,OH); <sup>13</sup>C **NMR (75 MHz, CDCl**<sub>3</sub>)  $\delta$  : 136.76 (C=<u>C</u>Ph) 131.04-127.66 (C<sub>arm</sub>), 126.48 (H<u>C</u>=CPh), 63.56 (<u>C</u>H<sub>2</sub>OH).

#### 3-chloro-3-phenylprop-2-en-1-ol (14)

Light Yellow oil, **FTIR: neat (cm<sup>-1</sup>)**: v 3307 (OH), 1636 (C=C) ; <sup>1</sup>H NMR (**300** MHz, CDCl3) δ : 7.63-7.35 (m, 5H ,H<sub>arm</sub>), 6.37 (**Z**, t, <sup>3</sup>J<sub>H-H</sub>= 5.90Hz, 1H, C<u>H</u>), 6.22 (**E**, t, <sup>3</sup>J<sub>H-H</sub>= 7.25Hz, 1H, C<u>H</u>), 4.54 (**Z**, d, <sup>3</sup>J<sub>H-H</sub>= 5.90Hz, 2H, C<u>H</u><sub>2</sub>OH), 4.16 (**E**, d, <sup>3</sup>J<sub>H-H</sub>= 7.23Hz, 2H, C<u>H</u><sub>2</sub>OH), 3.45 (**Z**+**E**, s, 1H, CH<sub>2</sub>O<u>H</u>); <sup>13</sup>C NMR (**75** MHz, CDCl<sub>3</sub>) δ: 137.26 (**Z**, <u>C</u>-Cl), 136.33(**E**, <u>C</u>-Cl), 133.70-126.44 (C<sub>arm</sub>), 128.92 (**Z**, H<u>C</u>=CCl), 128.68 (**E**, H<u>C</u>=CCl), 60.50 (**Z**, <u>C</u>H<sub>2</sub>OH), 59.76 (**E**, <u>C</u>H<sub>2</sub>OH).

#### 3-chloro-2-methyl-3-phenylprop-2-en-1-ol (15)

Light Yellow oil, **FTIR: neat (cm<sup>-1</sup>)**: v 3317 (OH), 1644 (C=C); <sup>1</sup>**H NMR (300 MHz, CDCl<sub>3</sub>) δ**: 7.30 (m, 5H, H<sub>arm</sub>), 4.40 (**Z**, s, 2H, C<u>H</u><sub>2</sub>OH), 3.99 (**E**, s, 2H, C<u>H</u><sub>2</sub>OH), 2.04 (**E**, s, 3H, C<u>H</u><sub>3</sub>), 1.79 (**Z**, s,3H, C<u>H</u><sub>3</sub>); <sup>13</sup>**C NMR (75 MHz, CDCl<sub>3</sub>) δ**: 138.24 (**E**, <u>C</u>-Cl), 133.47-128.30 (C<sub>arm</sub>), 128.54 (**E**, C=<u>C</u>-CH<sub>3</sub>), 63.73 (**E**, <u>C</u>H<sub>2</sub>OH), 22.72 (**Z**, <u>C</u>H<sub>3</sub>), 17.93 (**E**, <u>C</u>H<sub>3</sub>).

3-chloro-4,4-dimethylpent-2-en-1-ol (16)

Light Yellow oil, **FTIR: neat (cm<sup>-1</sup>)**: v 3307 (OH), 2969-2872 (CH<sub>3</sub> and CH<sub>2</sub>), 1649 (C=C); <sup>1</sup>H NMR (200 MHz, CDCl<sub>3</sub>) δ: 5.65 (t, <sup>3</sup>J<sub>H-H</sub>= 5.80Hz, 1H, C<u>H</u>), 4.21 (d, <sup>3</sup>J<sub>H-H</sub>= 5.80Hz, 2H, C<u>H</u><sub>2</sub>OH), 3.89 (s, 1H, O<u>H</u>), 1.11 (s, 3H, <sup>t</sup>Bu); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) δ: 145.62 (<u>C</u>-Cl), 121.99 (<u>C</u>=CCl), 60.03 (<u>C</u>H<sub>2</sub>OH), 38.40 (<u>C</u>(CH<sub>3</sub>)<sub>2</sub>), 28.59 (C(<u>C</u>H<sub>3</sub>)<sub>2</sub>).

#### Ethyl 2-chloro-4-hydroxybut-2-enoate (17)

Light Yellow oil; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$ : 7.19 (t, 1H, CH, <sup>3</sup>J<sub>H-H</sub>= 5.45Hz), 4.50 (d, 2H, CH<sub>2</sub>, <sup>3</sup>J<sub>H-H</sub>= 5.45Hz), 4.30 (q, 2H, CH<sub>2</sub>CH<sub>3</sub>, <sup>3</sup>J<sub>H-H</sub>= 7Hz), 2.11 (s, 1H, OH), 1.34 (t, 3H, CH3, <sup>3</sup>J<sub>H-H</sub>= 7Hz); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$ : 161.94 (<u>C</u>=O), 140.84 (<u>C</u>-Cl), 124.10 (C=<u>C</u>H), 62.55 (CH<sub>3</sub><u>C</u>H<sub>2</sub>O), 60.30 (<u>C</u>H<sub>2</sub>OH), 14.09 (<u>C</u>H<sub>3</sub>CH<sub>2</sub>O).

#### 3-chloro-2-methylpent-2-en-1-ol (18)

Yellow oil, **FTIR: neat (cm<sup>-1</sup>)**: v 3408 (OH), 1741(C=C) cm<sup>-1</sup>; <sup>1</sup>**H NMR (300 MHz, CDCl**<sub>3</sub>)δ : 4.22 (Z, s, 2H, C<u>H</u><sub>2</sub>OH), 4.09 (*E*, s, 2H, C<u>H</u><sub>2</sub>OH), 2.92 (Z+*E*, s, 1H, O<u>H</u>), 2.43 (*E*, q, <sup>3</sup>J<sub>H-H</sub>= 7.38Hz, 2H, C<u>H</u><sub>2</sub>CH<sub>3</sub>), 2.37 (*Z*, q, <sup>3</sup>J<sub>H-H</sub>= 7.44Hz, 2H, C<u>H</u><sub>2</sub>CH<sub>3</sub>), 1.87 (*E*, s, 3H, C<u>H</u><sub>3</sub>), 1.79 (Z, s, 3H, C<u>H</u><sub>3</sub>), 1.11-1.05 (Z+*E*, t, 3H, CH<sub>2</sub>C<u>H</u><sub>3</sub>); <sup>13</sup>C **NMR (75 MHz, CDCl**<sub>3</sub>) δ: 135.47 (*E*, <u>C</u>Cl), 132.15 (*Z*, <u>C</u>Cl), 129.82 (*E*, CH<sub>3</sub><u>C</u>=CCl), 129.38 (*E*, CH<sub>3</sub><u>C</u>=CCl), 64.06 (*Z*, <u>C</u>H<sub>2</sub>OH), 62.44 (*E*, <u>C</u>H<sub>2</sub>OH), 28.94 (*Z*, <u>C</u>H<sub>2</sub>CH<sub>3</sub>), 28.58 (*E*, <u>C</u>H<sub>2</sub>CH<sub>3</sub>), 17.85 (*E*, <u>C</u>H<sub>3</sub>), 15.85 (*Z*, <u>C</u>H<sub>3</sub>), 13.13 (*E*, CH<sub>2</sub><u>C</u>H<sub>3</sub>), 12.01 (*Z*, CH<sub>2</sub><u>C</u>H<sub>3</sub>).

, e

#### 3-chloro-2-phenylbut-2-en-1-ol (19)

Yellow oil, **FTIR: neat (cm<sup>-1</sup>)**: v 3326 (OH), 1651(C=C) cm<sup>-1</sup>; <sup>1</sup>**H NMR (300 MHz, CDCl3)** δ: 7.40-7.23 (m, 5H, H<sub>arm</sub>), 4.57 (**Z**, s, 2H, C<u>H</u><sub>2</sub>OH), 4.42 (**E**, s, 2H, C<u>H</u><sub>2</sub>OH), 2.37 (**E**, s, 3H, C<u>H</u><sub>3</sub>), 2.08 (**Z**, s, 3H, C<u>H</u><sub>3</sub>), 2.08 (s, 1H, O<u>H</u>); <sup>13</sup>**C NMR (75 MHz, CDCl3)** δ: 139.28 (**E**, <u>C</u>-Cl), 138.47 (**Z**, <u>C</u>-Cl), 136.65-128.32 (C<sub>arm</sub>), 127.58 (**Z**, <u>C</u>CH<sub>3</sub>), 127.45 (**E**, <u>C</u>CH<sub>3</sub>), 64.39 (**Z**, <u>C</u>H<sub>2</sub>OH), 63.30 (**E**, <u>C</u>H<sub>2</sub>OH), 23.99 (**E**, <u>C</u>H<sub>3</sub>), 22.88 (**Z**, <u>C</u>H<sub>3</sub>).

#### General Procedure for preparation of $\gamma$ -chloro- $\gamma$ , $\beta$ -allylic alcohols (20).

0.25 mmol of NaBH<sub>4</sub> was added to the ethyl 3-chloro-4,4,4-trifluoro-2-formylbut-2-enoate aldehyde (1mmol) stirred at -10°C in absolute ethanol. The progress of the reaction was monitored by TLC (by dissolving the sample in ether) using cyclohexane : diethylether (8:2) until the conversion of the aldehyde was complete. Then, water (100 mL) was added to the mixture. The product was extracted with diethyl ether (3x30mL) and washed 2 times with 10 mL of a saturated solution of NaCl. The solvent (diethylether) was evaporated, and the crude product was purified by column chromatography using (cyclohexane:diethylether) as eluent (8:2).

#### Ethyl 3-chloro-4,4,4-trifluoro-2-(hydroxymethyl)but-2-enoate (20)

coloress oil, **FTIR: neat (cm<sup>-1</sup>)**: v 3442 (OH), 11727(C=O) cm<sup>-1</sup>; 1652(C=C) cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, **CDCl**<sub>3</sub>) δ : 4.59 (Z+E, s, 2H, CH<sub>2</sub>O), 4.42-4.29 (Z+E, m, 2H, CH<sub>2</sub>CH<sub>3</sub>), 2.68 (Z, s, 1H, OH), 2.25 (E, s, 1H, OH), 1.40-1.32 (Z+E, m, 3H, CH<sub>2</sub>CH<sub>3</sub>); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) δ : 165.01 (E, CO<sub>2</sub>Et), 164.44 (Z, CO<sub>2</sub>Et), 139.73 (Z, CCl), 132.69 (E, C=CCl), 119.46 (Z+E, q, CF<sub>3</sub>, <sup>3</sup>J<sub>C-F</sub>=271.68Hz), 62.71 (Z, CH<sub>2</sub>CH<sub>3</sub>), 62.51(E,

## <u>CH</u><sub>2</sub>CH<sub>3</sub>), 61.27 (Z, <u>C</u>H<sub>2</sub>O), 59.05 (*E*, <u>C</u>H<sub>2</sub>O), 13.92 (*E*, CH<sub>2</sub><u>C</u>H<sub>3</sub>), 13.72 (Z, CH<sub>2</sub><u>C</u>H<sub>3</sub>); <sup>19</sup>F NMR (188 MHz, CDCl<sub>3</sub>) $\delta$ : -61.55 (*E*, s, CF<sub>3</sub>), -65.03 (Z, s, CF<sub>3</sub>).

#### General Procedure for preparation of allylic carbamates (21-45).

To a stirred solution of 3-chloro allyl alcohols (1 mmol) in dry THF (10 mL) with triethyle amine (1.5mmol) was added commercial isocyanates (1.3 mmol) at 0°C. The reaction mixture was stirred at room temperature for 2-12 h. the progress of the reaction was monitored by TLC. After, the reaction mixture was evaporated and the crude product was purified by column chromatography on silica gel using cyclohexane: diethyl ether (8:2) as the eluent to give the allylic carbamates products.

#### (2-chlorocyclohex-1-en-1-yl) methylpropylcarbamate (21)

**FTIR: neat (cm<sup>-1</sup>)**: v 3346 (NH), 1711 (C=O), 1695 (C=C), 1043 (C-O), 734 (C-Cl) cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, **CDCl**<sub>3</sub>)  $\delta$  5.05 (s, 1H, N<u>H</u>), 4.65 (s, 2H, C<u>H</u><sub>2</sub>O), 3.08-3.02 (m, 2H, NHC<u>H</u><sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 2.27 (m, 2H, C<u>H</u><sub>2</sub>cyc), 2.07 (m, 2H, C<u>H</u><sub>2</sub>cyc), 1.67-1.50 (m, 4H, C<u>H</u><sub>2</sub>C<u>H</u><sub>2</sub>cyc), 1.48-1.41 (m, 2H, NHCH<sub>2</sub>C<u>H</u><sub>2</sub>CH<sub>3</sub>), 0.88-0.81 (*Cis+Trans*, m, 3H, NHCH<sub>2</sub>CH<sub>2</sub>C<u>H</u><sub>3</sub>); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$ : 156.59 (C=O), 130.17 (C=CCl), 128.99 (C=CCl), 64.46 (CH<sub>2</sub>O), 44.29/42.71 (*cis/trans*, NHCH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 33.79 (CH<sub>2</sub>cyc), 27.68 (CH<sub>2</sub>cyc), 23.56 (CH<sub>2</sub>cyc), 23.08/21.00 (*trans/cis*, NHCH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 21.90 (CH<sub>2</sub>cyc), 11.06/10.93 (*trans/cis*, NHCH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>); HRMS (ESI), m/z: calcd for [C<sub>11</sub>H<sub>18</sub>ClNO<sub>2</sub>Na]<sup>+</sup> 254.0924 [M + Na]+; found 254.0926.

#### (2-chlorocyclohex-1-en-1-yl)methylbutylcarbamate (22)

**FTIR: neat (cm<sup>-1</sup>)**: v 3452 (NH), 1712 (C=O), 1515 (C=C), 1137 (C-O), 667 (C-Cl) cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ : 5.11 (s, 1H, NH), 4.63 (s, 2H, CH<sub>2</sub>O), 3.06 (*Cis+Trans*, q, <sup>3</sup>J<sub>H-H</sub>= 6.41Hz, 2H, NHCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 2.24 (m, 2H, CH<sub>2</sub>cyc), 2.04 (m, 2H, CH<sub>2</sub>cyc), 1.62-1.53 (m, 4H, CH<sub>2</sub>CH<sub>2</sub>cyc), 1.43-1.12 (*Cis+Trans*, q, <sup>3</sup>J<sub>H-H</sub>= 6.33Hz, 4H, NHCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 0.81 (*Cis+trans*, t, <sup>3</sup>J<sub>H-H</sub>= 7.30Hz, 3H, NHCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) δ : 156.57 (C=O), 130.04 (C=Cl), 129.00 (C=CCl), 64.40 (CH<sub>2</sub>O), 40.67 (NHCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 33.76 (CH<sub>2</sub>cyc), 31.93/29.52 (*trans/cis*, NHCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 27.63 (CH<sub>2</sub>cyc), 23.55 (CH<sub>2</sub>cyc), 21.88 (CH<sub>2</sub>cyc), 20.75/19.76 (*cis/trans*, NHCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 14.01/13.55 (*cis/Trans*, NHCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>); **HRMS** (ESI), m/z: calcd for [C<sub>12</sub>H<sub>2</sub>ICINO<sub>2</sub>Na]<sup>+</sup> 268.1080 [M+Na]+; found 268.1077.

#### (2-chlorocyclohex-1-en-1-yl)methyl phenylcarbamate (23)

**FTIR: neat (cm<sup>-1</sup>)**: v 3425 (NH), 1732 (C=O), 1601 (C=C), 1050 (C-O) cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ : 7.40-7.03(m, 5H, Harm), 7.01(s,1H,N<u>H</u>), 4.84 (s, 2H, C<u>H</u><sub>2</sub>O), 2.34-2.32 (m, 2H, C<u>H</u><sub>2</sub>cyc), 2.18- 2.14 (m, 2H, C<u>H</u><sub>2</sub>cyc), 1.70-1.59 (m, 4H, C<u>H</u><sub>2</sub>C<u>H</u><sub>2</sub>cyc); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) δ: 153.67(<u>C</u>=O), 137.90(<u>C</u>-Cl), 128.97 (CCl=<u>C</u>), 131.15-118.72(C<sub>arm</sub>), 65.12(<u>C</u>H<sub>2</sub>O), 33.89(<u>C</u>H<sub>2</sub>cyc), 27.93(<u>C</u>H<sub>2</sub>cyc), 23.58(<u>C</u>H<sub>2</sub>cyc), 21.94(<u>C</u>H<sub>2</sub>cyc); HRMS (ESI), m/z: calcd for [C<sub>14</sub>H<sub>16</sub>ClNO<sub>2</sub>Na]<sup>+</sup> 288.0767 [M+Na]+; found 288.0769.

#### (2-chlorocyclohex-1-en-1-yl)methyl (3-chlorophenyl)carbamate (24)

**FTIR: neat (cm<sup>-1</sup>)**: v 3432 (NH), 1732 (C=O), 1598(C=C), 1049 (C-O), 667 (C-Cl) cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ: 7.54 (s, 1H, N<u>H</u>), 7.26-7.02 (m, 4H, H<sub>arm</sub>), 4.86 (s, 2H, C<u>H</u><sub>2</sub>O), 2.37 (m, 2H, C<u>H</u><sub>2</sub>cyc), 2.18 (m, 2H, C<u>H</u><sub>2</sub>cyc), 1.73-1.62 (m, 4H, C<u>H</u><sub>2</sub>cyc); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) δ: 153.43 (<u>C</u>=O), 139.16 (C=<u>C</u>Cl), 134.69-

118.78 (Carm), 116.72 (<u>C</u>=CCl), 65.39 (<u>C</u>H<sub>2</sub>O), 33.90 (<u>C</u>H<sub>2</sub>cyc), 27.99 (<u>C</u>H<sub>2</sub>cyc), 23.58 (<u>C</u>H<sub>2</sub>cyc), 21.95 (<u>C</u>H<sub>2</sub>cyc); **HRMS** (ESI), m/z: calcd for [C<sub>14</sub>H<sub>15</sub>Cl<sub>2</sub>NO<sub>2</sub>Na]<sup>+</sup> 322.0378 [M + Na]+; found 322.0374.

#### (2-chlorocyclohept-1-en-1-yl)methylpropylcarbamate (25)

**FTIR: neat (cm<sup>-1</sup>)**: v 3345/3337 (NH), 1710 (C=O), 1695 (C=C), 1135 (C-O) cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$ : 4.80 (s, 1H, N<u>H</u>), 4.65(s, 2H, C<u>H</u><sub>2</sub>O), 3.08-3.04 (m, 2H, NHC<u>H</u><sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 2.55-2.52(m, 2H, C<u>H</u><sub>2</sub>cyc), 2.20-2.17(m, 2H, C<u>H</u><sub>2</sub>cyc), 1.68-1.65 (m, 2H, NHCH<sub>2</sub>C<u>H</u><sub>2</sub>CH<sub>3</sub>), 1.56-1.41 (m, 6H, C<u>H</u><sub>2</sub>cycC<u>H</u><sub>2</sub>cycC<u>H</u><sub>2</sub>cyc), 0.87- 0.82 (*Cis*+*Trans*, m, 3H, NHCH<sub>2</sub>CH<sub>2</sub>C<u>H</u><sub>3</sub>); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$ : 156.60(<u>C</u>=O), 134.74(C=<u>C</u>Cl), 133.78(<u>C</u>=CCl), 66.37(<u>C</u>H<sub>2</sub>O), 44.38/42.77 (*cis/trans*, NH<u>C</u>H<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 38.90 (<u>C</u>H<sub>2</sub>cyc), 31.24 (<u>C</u>H<sub>2</sub>cyc), 30.28 (<u>C</u>H<sub>2</sub>cyc), 25.75 (<u>C</u>H<sub>2</sub>cyc), 25.07 (<u>C</u>H<sub>2</sub>cyc), 23.16/21.07 (*Trans/cis*, NHCH<sub>2</sub><u>C</u>H<sub>2</sub>CH<sub>3</sub>), 11.12/11.01 (*Trans/cis*, NHCH<sub>2</sub>CH<sub>2</sub><u>C</u>H<sub>3</sub>); **HRMS** (ESI), m/z: calcd for [C<sub>12</sub>H<sub>20</sub>ClNO<sub>2</sub>Na]<sup>+</sup> 268.1080[M + Na]+; found 268.1080.

#### (2-chlorocyclohept-1-en-1-yl)methyl butylcarbamate (26)

FTIR: neat (cm<sup>-1</sup>): v 3452/3334 (NH), 1704 (C=O), 1516 (C=C), 1135(C-O), 666 (C-Cl) cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ : 5.20 (s, 1H, N<u>H</u>), 4.63 (s, 2H, C<u>H</u><sub>2</sub>O), 3.10-3.05 (m, 2H, NHC<u>H</u><sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 2.54-2.50 (m, 2H, C<u>H</u><sub>2</sub>cyc), 2.19-2.16 (m, 2H, C<u>H</u><sub>2</sub>cyc), 1.66-162 (m, 2H, NHCH<sub>2</sub>C<u>H</u><sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 1.55-1.50 (m, 2H, NHCH<sub>2</sub>CH<sub>2</sub>C<u>H</u><sub>2</sub>CH<sub>3</sub>), 1.47-1.20 (m, 6H, C<u>H</u><sub>2</sub>C<u>H</u><sub>2</sub>C<u>H</u><sub>2</sub>cyc), 0.84 (t, <sup>3</sup>J<sub>H-H</sub>= 7.27Hz, 3H, NHCH<sub>2</sub>C<u>H</u><sub>2</sub>CH<sub>2</sub>C<u>H</u><sub>2</sub>C<u>H</u><sub>2</sub>C), 134.33 (C-Cl), 133.80 (ClC=<u>C</u>), 66.06 (CH<sub>2</sub>O), 40.64 (NH<u>C</u>H<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 38.79 (CH<sub>2</sub>cyc), 31.92 (NHCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 31.16 (CH<sub>2</sub>cyc), 30.05 (CH<sub>2</sub>cyc), 25.65 (CH<sub>2</sub>cyc), 24.98 (CH<sub>2</sub>cyc), 19.75 (NHCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 13.55 (NHCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>); HRMS (ESI), m/z: calcd for [C<sub>13</sub>H<sub>22</sub>ClNO<sub>2</sub>Na]<sup>+</sup> 282.1237 [M + Na]<sup>+</sup>; found 282.1239.

#### (2-chlorocyclohept-1-en-1-yl)methyl phenylcarbamate (27)

**FTIR: neat (cm<sup>-1</sup>)**: v 3435 (NH), 1732 (C=O), 1602 (C=C), 1051 (C-O), 667 (C-Cl) cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, **CDCl**<sub>3</sub>)  $\delta$  : 7.44-7.07 (m, 5H, H<sub>arm</sub>), 6.82 (s, 1H, N<u>H</u>), 4.87 (s, 2H, C<u>H</u><sub>2</sub>O), 2.67-2.63 (m, 2H, C<u>H</u><sub>2</sub>cyc), 2.35-2.31 (m, 2H, C<u>H</u><sub>2</sub>cyc), 1.81-1.53 (m, 6H, C<u>H</u><sub>2</sub>C<u>H</u><sub>2</sub>C<u>H</u><sub>2</sub>cyc); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$ : 153.26 (<u>C</u>=O), 139.08 (<u>C</u>-Cl), 116.64 (ClC=<u>C</u>), 135.94-118.76(<u>C</u>arm), 67.30 (<u>C</u>H<sub>2</sub>O), 38.97 (<u>C</u>H<sub>2</sub>cyc), 31.20 (<u>C</u>H<sub>2</sub>cyc), 30.60 (<u>C</u>H<sub>2</sub>cyc), 25.81 (<u>C</u>H<sub>2</sub>cyc), 25.04 (<u>C</u>H<sub>2</sub>cyc); **HRMS** (ESI), m/z: calcd for [C<sub>15</sub>H<sub>18</sub>ClNO<sub>2</sub>Na]<sup>+</sup> 302.0924 [M+Na]<sup>+</sup>; found 302.0923.

#### (2-chlorocyclohept-1-en-1-yl)methyl (3-chlorophenyl)carbamate (28)

**FTIR: neat (cm<sup>-1</sup>)**: v 3428/3322 (NH), 1711 (C=O), 1596(C=C), 1051 (C-O), 680 (C-Cl) cm<sup>-1</sup>;<sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$ : 7.43 (s, 1H, N<u>H</u>), 7.13-6.78 (m, 4H, H<sub>arm</sub>), 4.75 (s, 2H, C<u>H</u><sub>2</sub>O), 2.56-2.52 (m, 2H, C<u>H</u><sub>2</sub>cyc), 2.23-2.19 (m, 2H, C<u>H</u><sub>2</sub>cyc), 1.68-1.44 (m, 6H, C<u>H</u><sub>2</sub>C<u>H</u><sub>2</sub>C<u>H</u><sub>2</sub>cyc); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$ : 153.32 (<u>C</u>=O), 134.75 (<u>C</u>-Cl), 123.45 (ClC=<u>C</u>), 139.09-116.68 (Carm), 67.29 (<u>C</u>H<sub>2</sub>O), 38.96 (<u>C</u>H<sub>2</sub>cyc), 31.19 (<u>C</u>H<sub>2</sub>cyc), 30.57 (<u>C</u>H<sub>2</sub>cyc), 25.80 (<u>C</u>H<sub>2</sub>cyc), 25.04 (<u>C</u>H<sub>2</sub>cyc); **HRMS** (ESI), m/z: calcd for [C<sub>15</sub>H<sub>17</sub>Cl<sub>2</sub>NO<sub>2</sub>Na]<sup>+</sup> 336.0534 [M+Na]<sup>+</sup>; found 336.0532.

#### Cinnamylpropylcarbamate (29)

**FTIR: neat (cm<sup>-1</sup>)**: v 3448 (NH), 1720 (C=O), 1695 (C=C), 1037 (C-O) cm<sup>-1</sup>; <sup>1</sup>**H NMR (300 MHz, CDCl**<sub>3</sub>) δ 7.40-7.20 (m, 5H, Harm), 6.61 (d, <sup>3</sup>J<sub>H-H</sub>= 15.90Hz, 1H, PhC<u>H</u>=CH), 6.30 (td, <sup>3</sup>J<sub>H-H</sub>= 6Hz, <sup>3</sup>J<sub>H-H</sub>= 15.90Hz, 1H,

 $\begin{array}{l} C\underline{H}=CHPh), \ 5.06 \ (s, 1H, N\underline{H}), \ 4.74 \ (d, \ {}^{3}J_{H-H}= \ 6Hz, \ 2H, \ C\underline{H}_{2}OH), \ 3.20-3.13 \ (m, \ 2H, \ NHC\underline{H}_{2}CH_{2}CH_{3}), \ 1.60-1.48 \\ (m, \ 2H, \ NHCH_{2}C\underline{H}_{2}CH_{3}), \ 0.93 \ (t, \ {}^{3}J_{H-H}= \ 7.37Hz, \ 3H, \ NHCH_{2}CH_{2}C\underline{H}_{3}); \ {}^{13}C \ NMR \ (75 \ MHz, \ CDCl_{3}) \ \delta; \\ 156.4853 \ (\underline{C}=O), \ 136.43 \ (C=\underline{C}Ph), \ 133.41-126.57 \ (Carm), \ 124.19 \ (\underline{C}H=CPh), \ 65.22 \ (\underline{C}H_{2}O), \ 42.81 \\ (NH\underline{C}H_{2}CH_{2}CH_{3}), \ 23.19 \ (NHCH_{2}\underline{C}H_{2}CH_{3}), \ 11.18 \ (NHCH_{2}CH_{2}\underline{C}H_{3}); \ HRMS \ (ESI), \ m/z: \ calcd \ for \\ [C_{13}H_{17}NO_{2}Na]^{+} 242.1157 \ [M + Na]^{+}; \ found \ 242.1157. \end{array}$ 

#### Cinnamyl butylcarbamate (30)

**FTIR: neat (cm<sup>-1</sup>)**: v 3340 (NH), 1695 (C=O), 1519 (C=C), 1137 (C-O) cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$ : 7.26-7.11 (m, 5H, H<sub>arm</sub>), 6.50 (d, <sup>3</sup>J<sub>H-H</sub>=15.91Hz, 1H, PhC<u>H</u>=CH), 6.16 (dt, <sup>3</sup>J<sub>H-H</sub>=15.88Hz, <sup>3</sup>J<sub>H-H</sub>=6.28Hz, 1H,PhCH=C<u>H</u>), 4.98 (s, 1H, NH), 4.60 (d, <sup>3</sup>J<sub>H-H</sub>= 6.03 Hz, 2H, CH<sub>2</sub>O), 3.16-3.03 (*cis+trans*, m, 2H, NHC<u>H</u><sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 1.39-1.18 (m, 4H, NHCH<sub>2</sub>C<u>H</u><sub>2</sub>C<u>H</u><sub>2</sub>CH<sub>3</sub>), 0.80 (*trans*, t, <sup>3</sup>J<sub>H-H</sub>=7.25, 3H, NHCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>C<u>H</u><sub>3</sub>); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$ : 156.48/156.41 (*Trans/cis*, <u>C</u>=O), 133.36-126.54 (<u>C</u>arm), 136.42 (CH=<u>C</u>HPh), 124.18 (PhHC=<u>C</u>H), 65.20 (<u>C</u>H<sub>2</sub>O), 40.79/40.39 (*Trans/Cis*, NH<u>C</u>H<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 32.05/31.76 (*Trans/Cis*, NHCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 20.14/19.88 (*Cis/Trans*, CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 13.70 (*Trans*, CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub><u>C</u>H<sub>3</sub>); **HRMS** (ESI), m/z: calcd for [C<sub>14</sub>H<sub>19</sub>ClNO<sub>2</sub>Na]<sup>+</sup> 256.1313 [M+Na]<sup>+</sup>; found 256.1310. *Cinnamylphenylcarbamate* (*31*)

**FTIR: neat (cm<sup>-1</sup>)**: v 3435 (NH), 1728 (C=O), 1601(C=C), 1050 (C-O) cm<sup>-1</sup>; <sup>1</sup>**H NMR (300MHz, CDCl<sub>3</sub>) δ**: 7.50-7.11 (m, 11H,  $H_{arm}+N\underline{H}$ ), 6.74 (d, <sup>3</sup>J<sub>H-H</sub>= 15.89Hz, 1H, CH), 6.39 (dt, (<sup>3</sup>J<sub>H-H</sub>)doublet= 15.87Hz, (<sup>3</sup>J<sub>H</sub>) H)triplet= 6.38 Hz, 1H, C<u>H</u>=CHPh), 4.90 (d, <sup>3</sup>J<sub>H-H</sub>= 6.35Hz, 2H, CH<sub>2</sub>O); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) **δ**: 153.70 (<u>C</u>=O), 138.05-123.52 (Carm), 134.23 (PhH<u>C</u>=CH), 119.05 (PhHC=<u>C</u>H), 65.85 (<u>C</u>H<sub>2</sub>O); **HRMS** (ESI), m/z: calcd for [C<sub>16</sub>H<sub>15</sub>NO<sub>2</sub>Na]<sup>+</sup> 276.1000 [M+Na]<sup>+</sup>; found 276.1003.

#### Cinnamyl (3-chlorophenyl) carbamate (32)

**FTIR: neat (cm<sup>-1</sup>)**: v 3322 (NH), 1713 (C=O), 1596(C=C), 1048 (C-O), 692 (C-Cl) cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, **CDCl3**) **\delta**: 7.58 (s, 1H, N<u>H</u>), 7.43-7.06 (m, 9H, H<sub>arm</sub>), 6.71 (*d*, <sup>3</sup>J<sub>H-H</sub>= 15.90Hz, 1H, PhC<u>H</u>=CH), 6.34 (td, <sup>3</sup>J<sub>H-H</sub>= 6.40Hz, <sup>3</sup>J<sub>H-H</sub>= 15.90Hz, 1H, PhCH=C<u>H</u>), 4.87 (d, <sup>3</sup>J<sub>H-H</sub>= 6.39Hz, 2H, CH2OH); <sup>13</sup>C NMR (75 MHz, CDCl3)  $\delta$ : 153.65 (<u>C</u>=O), 139.37-117.08 (<u>C</u>arm), 134.46 (Ph<u>C</u>=CH), 126.67 (PhC=<u>C</u>H), 66.16 (<u>C</u>H<sub>2</sub>O); **HRMS** (ESI), m/z: calcd for [C<sub>16</sub>H<sub>14</sub>ClNO<sub>2</sub>Na] + 310.0611 [M+Na]<sup>+</sup>; found 310.0603.

#### 3-Chloro-3-phenylallyl propylcarbamate (33)

**FTIR: neat (cm<sup>-1</sup>)**: v 3346/3353 (NH), 1716 (C=O), - (C=C), 1037 (C-O) cm<sup>-1</sup>; <sup>1</sup>H NMR (**300 MHz, CDCl3**) δ : (**Z**) 7.62-7.34 (m, 5H, H<sub>arm</sub>), 6.30 (t, <sup>3</sup>J<sub>H-H</sub>= 6.04Hz, 1H, ClC=C<u>H</u>), 4.94 (s, 1H, N<u>H</u>), 4.95 (d, <sup>3</sup>J<sub>H-H</sub>= 6.04Hz, 2H, C<u>H</u><sub>2</sub>OH), 3.18-3.16 (m, 2H, NHC<u>H</u><sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 1.61-1.49 (m, 2H, NHCH<sub>2</sub>C<u>H</u><sub>2</sub>CH<sub>3</sub>), 1.74-1.63 (m, 2H, CH<sub>2</sub>C<u>H</u><sub>2</sub>CH<sub>3</sub>), 0.93 (t, <sup>3</sup>J<sub>H-H</sub>= 6.64Hz, 3H, NHCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>C<u>H</u><sub>3</sub>). (*E*): 6.17 (t, <sup>3</sup>J<sub>H-H</sub>= 5.83Hz, 1H, C<u>H</u>), 4.56 (d, <sup>3</sup>J<sub>H-H</sub>= 5.80Hz, 2H, CH<sub>2</sub>OH); <sup>13</sup>C NMR (**75 MHz, CDCl**<sub>3</sub>) δ: 156.33 (*Z*, <u>C</u>=O), 137.13 (*Z*, <u>C</u>-Cl), 135.46-126.48 (<u>C</u>arm), 124.36 (*E*, ClC=<u>C</u>H), 122.47 (*Z*, ClC=<u>C</u>H), 62.31 (*Z*, <u>C</u>H<sub>2</sub>O), 42.85 (NHC<u>H</u><sub>2</sub>CH<sub>2</sub>CH<sub>3</sub><sub>2</sub>), 23.18 (NHC<u>H</u><sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 11.16 (NHCH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>); **HRMS** (ESI), m/z: calcd for [C<sub>13</sub>H<sub>16</sub>ClNO<sub>2</sub>Na]<sup>+</sup>276.076 [M+Na]<sup>+</sup>; found 276.0767.

#### 3-chloro-3-phenylallyl (3-chlorophenyl) carbamate (34)

**FTIR: neat (cm<sup>-1</sup>)**: v 3420/3336 (NH), 1732 (C=O), 1597 (C=C), 1062 (C-O) cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, **CDCl<sub>3</sub>**) **\delta**: 7.64-7.05 (m, 10H, H<sub>arm</sub>), 6.95 (s, 1H, N<u>H</u>), 6.35 (**Z**, t, <sup>3</sup>J<sub>H-H</sub>= 6.23Hz, 1H, C<u>H</u>), 6.20 (**E**, t, <sup>3</sup>J<sub>H-H</sub>= 7.5Hz, 1H, C<u>H</u>), 5.06 (**Z**, d, <sup>3</sup>J<sub>H-H</sub>= 6.23Hz, 2H, C<u>H</u><sub>2</sub>O), 4.69 (**E**, d, <sup>3</sup>J<sub>H-H</sub>= 7.5Hz, 2H, C<u>H</u><sub>2</sub>O); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) **\delta**: 153.14 (<u>C</u>=O), 138.96-116.76 (C<sub>arm</sub>), 136.91 (**Z**, C=<u>C</u>Cl), 121.34 (**Z**, <u>C</u>=CCl), 120.84 (**E**, <u>C</u>=CCl), 62.88 (**Z**, <u>C</u>H<sub>2</sub>O), 62.29 (**E**, <u>C</u>H<sub>2</sub>O); **HRMS** (ESI), m/z: calcd for [C<sub>16</sub>H<sub>13</sub>Cl<sub>2</sub>NO<sub>2</sub>Na] + 344.0221 [M+Na]<sup>+</sup>; found 344.0247.

#### 3-chloro-2-methyl-3-phenylallyl propylcarbamate (35)

**FTIR: neat (cm<sup>-1</sup>):** v 3445/3351 (NH), 1715 (C=O), 1699 (C=C), 1135 (C-O), 667 (C-Cl) cm<sup>-1</sup>; <sup>1</sup>H NMR (**300 MHz, CDCl**<sub>3</sub>)  $\delta$  : 7.34 (m, 5H, H<sub>arm</sub>), 4.92 (s, 1H, N<u>H</u>), 4.54 (*E*, s, 2H, C<u>H</u><sub>2</sub>O), 3.11 (*m*, <sup>3</sup>J<sub>H-H</sub>= 6.69Hz, 2H, NHC<u>H</u><sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 2.05 (*E*, s, 3H, C<u>H</u><sub>3</sub>), 1.78 (*Z*, s, 3H, C<u>H</u><sub>3</sub>), 1.57-1.45 (m, 2H, NHCH<sub>2</sub>C<u>H</u><sub>2</sub>CH<sub>3</sub>), 0.91 (t, <sup>3</sup>J<sub>H-H</sub>= 7.37Hz, 2H, NHCH<sub>2</sub>C<u>H</u><sub>2</sub>CH<sub>3</sub>); <sup>13</sup>C NMR (**75 MHz, CDCl**<sub>3</sub>)  $\delta$ : 156.30 (<u>C</u>=O), 138.46 (*Z*, <u>C</u>Cl), 137.96 (*E*, <u>C</u>Cl), 132.27-128.27 (C<sub>arm</sub>), 128.62 (<u>C</u>=CCl), 66.96 (*Z*, <u>C</u>H<sub>2</sub>O), 65.36 (*E*, <u>C</u>H<sub>2</sub>O), 44.42 (*Z*, NH<u>C</u>H<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 43.42/42.79 (cis/Trans *E*, NH<u>C</u>H<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 23.14/21.12 (*trans/Cis, E*, NHCH<sub>2</sub><u>C</u>H<sub>2</sub>CH<sub>3</sub>), 17.95 (*E*, <u>C</u>H<sub>3</sub>), 17.52 (*Z*, <u>C</u>H<sub>3</sub>), 11.15 (*Cis*+*Trans, E*, NHCH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>); **HRMS** (ESI), m/z: calcd for [C<sub>14</sub>H<sub>18</sub>ClNO<sub>2</sub>Na]<sup>+</sup> 290.0924 [M+Na]+; found 290.0930.

#### 3-chloro-2-methyl-3-phenylallyl phenylcarbamate (36)

**FTIR: neat (cm<sup>-1</sup>)**: v 3423/3338 (NH), 1732 (C=O), 1599 (C=C), 1047 (C-O), 895 (C-Cl) cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ : 7.41-7.08 (m, 10H, H<sub>arm</sub>), 6.73 (s, 1H, N<u>H</u>), 5.10 (Z, s, 2H, C<u>H</u><sub>2</sub>O), 4.66 (E, s, 2H, C<u>H</u><sub>2</sub>O), 2.12 (E, s, 3H, C<u>H</u><sub>3</sub>), 1.85 (Z, s, 3H, C<u>H</u><sub>3</sub>); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) δ: 153.17 (<u>C</u>=O), 137.86 (E, C=<u>C</u>Cl), 137.68(Z, C=<u>C</u>Cl), 133.03-123.61 (C<sub>arm</sub>), 119.79 (Z, <u>C</u>=CCl), 118.75 (E, <u>C</u>=CCl), 65.93 (E, <u>C</u>H<sub>2</sub>O), 22.29 (Z, <u>C</u>H<sub>3</sub>), 18.11 (E, <u>C</u>H<sub>3</sub>); HRMS (ESI), m/z: calcd for [C<sub>17</sub>H<sub>16</sub>ClNO<sub>2</sub>Na]<sup>+</sup> 324.0767 [M+Na]+; found 324.0765. 3-chloro-2-methyl-3-phenylallyl (3-chlorophenyl) carbamate (37)

**FTIR: neat (cm<sup>-1</sup>)**: v 3406/3320 (NH), 1708 (C=O), 1594(C=C), 1043 (C-O), 696 (C-Cl) cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ: 7.52 (s, 1H, N<u>H</u>), 7.38-6.96 (m, 9H, Harm), 5.10 (Z, s, 2H, C<u>H</u><sub>2</sub>O), 4.67 (*E*, s, 2H, C<u>H</u><sub>2</sub>O), 2.11 (*E*, s, 3H, C<u>H</u><sub>3</sub>), 1.90 (Z<sub>2</sub> s, 3H, C<u>H</u><sub>3</sub>); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) δ: 153.07 (C=O), 138.99-116.79 (Carm), 137.80 (*E*, CH<sub>3</sub>C=<u>C</u>Cl), 123.63 (*E*, CH<sub>3</sub><u>C</u>=CCl), 68.34 (*Z*, <u>C</u>H<sub>2</sub>O), 66.17 (*E*, <u>C</u>H<sub>2</sub>O), 18.10 (*E*, <u>C</u>H<sub>3</sub>), 17.82 (*Z*, <u>C</u>H<sub>3</sub>); HRMS (ESI), m/z: calcd for [C<sub>17</sub>H<sub>15</sub>Cl<sub>2</sub>NO<sub>2</sub>Na] + 258.0378 [M+Na]<sup>+</sup>; found 258.0381.

#### (Z)-3-chloro-4,4-dimethylpent-2-en-1-yl phenylcarbamate (38)

**FTIR:** neat (cm<sup>-1</sup>): v 3423/3336 (NH), 1731 (C=O), 1600 (C=C), 1061 (C-O) cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$ : 7.43-7.06 (m, 5H, Harm), 6.62 (s, 1H, N<u>H</u>), 5.76 (t, <sup>3</sup>J<sub>H-H</sub>= 5.92Hz, 1H, ClC=C<u>H</u>), 4.87 (d, 2<sup>3</sup>J<sub>H-H</sub>= 5.92Hz, H, C<u>H</u><sub>2</sub>O), 1.22 (s, 9H, (C<u>H</u><sub>3</sub>)<sub>3</sub>); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$ : 153.60/153.54 (cis/trans, C=O), 148.65 (CH=CCl), 137.83-118.74 (*cis+trans*, Carm), 117.40 (<u>C</u>H=CCl), 62.77 (<u>C</u>H<sub>2</sub>O), 38.81 (<u>C</u>(CH<sub>3</sub>)<sub>3</sub>), 28.65 (C(<u>C</u>H<sub>3</sub>)<sub>3</sub>); **HRMS** (ESI), m/z: calcd for [C<sub>14</sub>H<sub>18</sub>ClNO<sub>2</sub>Na]<sup>+</sup> 290.0924 [M+Na]<sup>+</sup>; found 290.0921.

#### 3-chloro-2-ethylbut-2-en-1-yl propylcarbamate (39)

**FTIR: neat (cm<sup>-1</sup>)**: v 3335 (NH), 1692 (C=O), 1660 (C=C), 1042 (C-O)cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ 4.86 (s, 1H, N<u>H</u>), 4.68 (Z, s, 2H, C<u>H</u><sub>2</sub>O), 4.52 (E, s, 2H, C<u>H</u><sub>2</sub>O), 3.07-3.05 (m, 2H, NHC<u>H</u><sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 2.43 (E, q,  ${}^{3}J_{H-H}$ = 7.38Hz, 2H, CH<sub>2</sub>CH<sub>3</sub>), 2.34 (*E*, q,  ${}^{3}J_{H-H}$ = 7.47Hz, 2H, CH<sub>2</sub>CH<sub>3</sub>), 1.79 (*E*, s, C<u>H</u><sub>3</sub>), 1.69 (*Z*, s, C<u>H</u><sub>3</sub>), 1.51-1.39 (m, 2H, NHCH<sub>2</sub>C<u>H</u><sub>2</sub>CH<sub>3</sub>), 1.04 (*E*, t,  ${}^{3}J_{H-H}$ = 7.4Hz, 3H, CH<sub>2</sub>CH<sub>3</sub>), 1.03(*Z*, t,  ${}^{3}J_{H-H}$ = 7.38Hz, 3H, CH<sub>2</sub>CH<sub>3</sub>), 0.84 (t,  ${}^{3}J_{H-H}$ = 7.39Hz, 3H, NHCH<sub>2</sub>CH<sub>2</sub>C<u>H</u><sub>3</sub>) ; <sup>13</sup>C **NMR (75 MHz, CDCl**<sub>3</sub>) **δ**: 156.58/156.40 (*cis/trans*, <u>C</u>=O), 137.39 (*Z*+*E*, C=<u>C</u>Cl), 126.20 (*E*, <u>C</u>=CCl), 120.82 (*Z*, <u>C</u>=CCl), 65.72 (*Z*, <u>CH</u><sub>2</sub>O), 64.18 (*E*, <u>CH</u><sub>2</sub>O), 44.37 (*Z*, <u>C</u>H<sub>2</sub>CH<sub>3</sub>) ,42.76 (*E*, <u>C</u>H<sub>2</sub>CH<sub>3</sub>), 28.95 (*Z*, <u>C</u>H<sub>3</sub>), 28.82 (*E*, <u>C</u>H<sub>3</sub>), 23.11/21.05 (*Trans/ Cis*, NH<u>C</u>H<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 18.10/15.70 (*Trans Cis/*, NHCH<sub>2</sub><u>C</u>H<sub>2</sub>CH<sub>3</sub>), 13.01 (*E*, CH<sub>2</sub><u>C</u>H<sub>3</sub>), 11.98 (*Z*,CH<sub>2</sub><u>C</u>H<sub>3</sub>), 11.08/10.98 (*Trans/Cis*, NHCH<sub>2</sub>CH<sub>2</sub><u>C</u>H<sub>3</sub>) ; **HRMS** (ESI), m/z: calcd for [C<sub>10</sub>H<sub>18</sub>CINO<sub>2</sub>Na]<sup>+</sup> 242.0924 [M+ Na]<sup>+</sup>; found 242.0917.

#### 3-chloro-2-ethylbut-2-en-1-yl phenylcarbamate (40)

**FTIR: neat (cm<sup>-1</sup>)**: v 3328 (NH), 1711 (C=O), 1600(C=C), 1048 (C-O) cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$ : 7.48-7.07 (m, 6H, H<sub>arm</sub>+N<u>H</u>), 4.94 (Z, s, 2H, C<u>H</u><sub>2</sub>O), 4.75 (E, s, 2H, C<u>H</u><sub>2</sub>O), 2.56 (E, q, <sup>3</sup>J<sub>H-H</sub>= 7.38Hz, 2H, CH<sub>2</sub>CH<sub>3</sub>), 2.46 (Z, s, <sup>3</sup>J<sub>H-H</sub>= 7.43Hz, 3H, CH<sub>2</sub>CH<sub>3</sub>), 1.95 (E, s, 3H, C<u>H</u><sub>3</sub>), 1.84 (Z, s, 3H, C<u>H</u><sub>3</sub>), 1.18 (E, t, 3 <sup>3</sup>J<sub>H-H</sub>= 7.38Hz, H, CH<sub>2</sub>CH<sub>3</sub>), 1.17 (Z, t, <sup>3</sup>J<sub>H-H</sub>= 7.43Hz, 3H, CH<sub>2</sub>CH<sub>3</sub>); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$ : 153.87/153.71 (*cis/trans*, <u>C</u>=O), 138.05 (E, C=<u>C</u>Cl), 137.93 (Z, C=<u>C</u>Cl), 134.50-119.02 (Carm), 123.57 (E, <u>C</u>=CCl), 123.50 (Z, <u>C</u>=CCl), 66.31 (Z, <u>C</u>H<sub>2</sub>O), 64.68 (E, <u>C</u>H<sub>2</sub>O), 29.08 (Z, <u>C</u>H<sub>2</sub>CH<sub>3</sub>), 28.95 (E, <u>C</u>H<sub>2</sub>CH<sub>3</sub>), 18.28 (E, <u>C</u>H<sub>3</sub>), 15.91 (Z, <u>C</u>H<sub>3</sub>), 13.15 (E, CH<sub>2</sub><u>C</u>H<sub>3</sub>), 12.09 (Z, CH<sub>2</sub><u>C</u>H<sub>3</sub>); HRMS (ESI), m/z: calcd for [C<sub>13</sub>H<sub>16</sub>CINO<sub>2</sub>Na]<sup>+</sup> 276.0767 [M+ Na]+; found 276.0769.

#### 3-chloro-2-ethylbut-2-en-1-yl (3-chlorophenyl) carbamate (41)

**FTIR: neat (cm<sup>-1</sup>)**: v 3432 (NH), 1731 (C=O), 1598(C=C), 1050 (C-O), 668 (C-Cl) cm<sup>-1</sup>;<sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ: 7.55/7.52 (cis/trans, s, 1H, N<u>H</u>), 7.23-6.95 (m, 4H, H<sub>arm</sub>), 4.89 (Z, s, 2H, C<u>H</u><sub>2</sub>O), 4.73 (E, s, 2H, C<u>H</u><sub>2</sub>O), 2.54 (E, q, <sup>3</sup>J<sub>H-H</sub>= 7.39Hz, 2H, CH<sub>2</sub>CH<sub>3</sub>), 2.44 (Z, q, <sup>3</sup>J<sub>H-H</sub>= 7.41Hz, 2H, CH<sub>2</sub>CH<sub>3</sub>), 1.92 (E, s, 3H, C<u>H</u><sub>3</sub>), 1.83 (Z, s, 3H, C<u>H</u><sub>3</sub>), 1.15 (E, t, <sup>3</sup>J<sub>H-H</sub>= 7.39Hz, 3H, CH<sub>2</sub>C<u>H</u><sub>3</sub>), 1.14 (Z, t, <sup>3</sup>J<sub>H-H</sub>= 7.41Hz, 3H, CH<sub>2</sub>C<u>H</u><sub>3</sub>); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) δ: 153.17/152.86 (*trans/cis*, <u>C</u>=O), 138.37 (E, <u>C</u>Cl), 138.05 (Z, <u>C</u>Cl), 138.99-116.74 (<u>C</u>arm), 123.56 (E, C=<u>C</u>CH<sub>3</sub>), 123.48 (Z, C=<u>C</u>CH<sub>3</sub>), 66.57 (Z, <u>C</u>H<sub>2</sub>O), 64.93 (E, <u>C</u>H<sub>2</sub>O), 29.06 (Z, <u>C</u>H<sub>2</sub>CH<sub>3</sub>), 28.93(E, <u>C</u>H<sub>2</sub>CH<sub>3</sub>), 18.28 (E, <u>C</u>H<sub>3</sub>), 16.00 (Z, <u>C</u>H<sub>3</sub>), 13.11 (E, CH<sub>2</sub><u>C</u>H<sub>3</sub>), 12.04 (Z, CH<sub>2</sub><u>C</u>H<sub>3</sub>); **HRMS** (ESI), m/z: calcd for [C<sub>13</sub>H<sub>15</sub>Cl2NO<sub>2</sub>Na]<sup>+</sup> 310.0378 [M+ Na]+; found 310.0386.

#### 3-chloro-2-phenylbut-2-en-1-yl phenylcarbamate (42)

**FTIR: neat (cm<sup>-1</sup>)**: v 3431/3301(NH), 1727 (C=O), 1600 (C=C), 1053 (C-O), 692 (C-Cl) cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$ : 7.58-6.85 (m, H<sub>arm</sub>), 6.68/6.61 (*Trans/cis*, s, 1H, NH), 5.20/5.16 (*cis/Trans*, *Z*, s, 2H, CH<sub>2</sub>O), 5.02/4.97 (trans/*Cis E*, s, 2H, CH<sub>2</sub>O), 2.46/2.34 (trans/*Cis E*, s, 3H, CH<sub>3</sub>), 2.12/2.02 (trans/*Cis Z*, s, 3H, CH<sub>3</sub>); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$ : 155.56/153.16 (*trans/Cis*, <u>C</u>=O), 132.54/132.07 (*Trans /cis Z* C=CCl), 131.53/131.16 (*trans/Cis E*, C=CCl), 124.11/124.04 (*Cis/trans E*, <u>C</u>=CCl), 123.59/123.42 (*Cis/trans Z*, <u>C</u>=CCl), 138.00-118.08 (*Cis+Trans Z+E*, <u>C</u>arm), 67.69/66.76 (*Trans/cis E*, <u>C</u>H<sub>2</sub>O), 66.06/65.09 (*Trans/cis Z*, <u>C</u>H<sub>2</sub>O), 24.13/23.94 (*Trans/cis Z*, <u>C</u>H<sub>3</sub>), 23.31/23.10 (*trans/Cis E*, <u>C</u>H<sub>3</sub>); HRMS (*ESI*), m/z: calcd for [C<sub>17</sub>H<sub>16</sub>CINO<sub>2</sub>Na]<sup>+</sup> 324.0767 [M+Na]<sup>+</sup>; found 324.0763.

#### 3-chloro-2-phenylbut-2-en-1-yl (3-chlorophenyl)carbamate (43)

**FTIR: neat (cm<sup>-1</sup>)**: v 3432 (NH), 1733 (C=O), 1598 (C=C), 1214 (C-O), 667 (C-Cl) cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ: 7.35- 6.91 (m, 5H, H<sub>arm</sub>), 6.54 (*E*, s, 1H, N<u>H</u>), 6.47 (*Z*, s, 1H, N<u>H</u>), 5.07 (*Z*, s, 2H, C<u>H</u><sub>2</sub>O), 4.90 (*E*, s, 2H, C<u>H</u><sub>2</sub>O), 2.34 (*E*, s, 3H, C<u>H</u><sub>3</sub>), 1.99 (*Z*, s, 3H, C<u>H</u><sub>3</sub>); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) δ: 152.83 (<u>C</u>=O), 139.10-116.68 (<u>C</u><sub>arm</sub>), 129.99 (*E*, PhC=<u>C</u>Cl), 129.92 (*Z*, PhC=<u>C</u>Cl), 123.64 (*E*, Ph<u>C</u>=CCl), 123.49 (*Z*, Ph<u>C</u>=CCl), 66.28 (*Z*, <u>C</u>H<sub>2</sub>O), 65.30 (*E*, <u>C</u>H<sub>2</sub>O), 24.11 (*Z*, <u>C</u>H<sub>3</sub>), 23.28 (*E*, <u>C</u>H<sub>3</sub>); HRMS (*ESI*), m/z: calcd for [C<sub>17</sub>H<sub>15</sub>Cl<sub>2</sub>NO<sub>2</sub>Na]<sup>+</sup> 358.0378 [M+Na]<sup>+</sup>; found 358.0383.

#### Ethyl 3-chloro-4,4,4-trifluoro-2-(((propylcarbamoyl)oxy)methyl)but-2-enoate (44)

**FTIR: neat (cm<sup>-1</sup>)**: v 3450 (NH), 1735 (C=O), 1651(C=C), 1215 (C-O), 667 (C-Cl) cm<sup>-1</sup>; <sup>1</sup>H NMR (200 MHz, CDCl<sub>3</sub>) δ : 4.97 (Z+E, s, 2H, C<u>H</u><sub>2</sub>O), 4.23 (Z+E, m, 2H, C<u>H</u><sub>2</sub>CH<sub>3</sub>), 3.11 (m, 2H, NHC<u>H</u><sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 1.42 (m, 2H, NHCH<sub>2</sub>C<u>H</u><sub>2</sub>CH<sub>3</sub>), 1.27 (Z+E, m, 3H, CH<sub>2</sub>C<u>H</u><sub>3</sub>), 0.87 (m, 2H, NHCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>O); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) δ : 163.92 (E, CO<sub>2</sub>Et), 163.35 (Z, CO<sub>2</sub>Et), 155.08 (C=O), 137.49 (E, CCl), 136.29 (Z, CCl), 121.21 (Z, C=CCF<sub>3</sub>), 119.40 (Z+E, q, <sup>3</sup>J<sub>C-F</sub>= 271.87Mhz, CF<sub>3</sub>), 62.50 (E, CH<sub>2</sub>O), 62.34 (Z, CH<sub>2</sub>O), 61.71 (Z, CH<sub>2</sub>CH<sub>3</sub>), 59.56 (E, CH<sub>2</sub>CH<sub>3</sub>), 43.39/42.86 (*Cis/trans*, NHCH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 22.93/22.29 (*trans/Cis*, NHCH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 13.81 (E, CH<sub>2</sub>CH<sub>3</sub>), 13.64 (Z, CH<sub>2</sub>CH<sub>3</sub>), 10.92/10.14 (*trans/Cis*, NHCH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>); <sup>19</sup>F NMR (188 MHz, CDCl<sub>3</sub>) δ : - 61.83 (E, s, CF<sub>3</sub>), -65.15 (Z, s, CF<sub>3</sub>); HRMS (*ESI*), m/z: calcd for [C<sub>11</sub>H<sub>15</sub>ClNO<sub>4</sub>F<sub>3</sub>Na]<sup>+</sup> 340.0539 [M+Na]<sup>+</sup>; found 340.0534.

#### Ethyl 3-chloro-4,4,4-trifluoro-2-(((phenylcarbamoyl)oxy)methyl)but-2-enoate (45)

**FTIR: neat (cm<sup>-1</sup>)**: v 3432 (NH), 1740 (C=O), 1601(C=C), 1214 (C-O), 666 (C-Cl) cm<sup>-1</sup>; <sup>1</sup>H NMR (200 MHz, CDCl<sub>3</sub>) δ : 7.35-7.06 (m, 5H, H<sub>arm</sub>), 6.95 (s, 1H, N<u>H</u>), 5.12 (*Z*+*E*, s, 2H, C<u>H</u><sub>2</sub>O), 4.31 (*Z*+*E*, m, 2H, C<u>H</u><sub>2</sub>CH<sub>3</sub>), 1.29 (*Z*+*E*, m, 3H, CH<sub>2</sub>C<u>H</u><sub>3</sub>); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) δ : 164.00 (*E*, CO<sub>2</sub>Et), 163.44 (*Z*, CO<sub>2</sub>Et), 152.22 (C=O), 137.22 (*Z*, CCl), 135.75-117.62 (C<sub>arm</sub>), 119.43 (*Z*, q, <sup>3</sup>J<sub>C-F</sub>= 272.43Hz, CF<sub>3</sub>), 62.82 (*Z*, CH<sub>2</sub>O), 62.65 (*E*, CH<sub>2</sub>O), 62.07 (*Z*, CH<sub>2</sub>CH<sub>3</sub>), 59.61(*E*, CH<sub>2</sub>CH<sub>3</sub>), 13.88 (*E*, CH<sub>2</sub>CH<sub>3</sub>), 13.71 (*Z*, CH<sub>2</sub>CH<sub>3</sub>); <sup>19</sup>F NMR (188 MHz, CDCl<sub>3</sub>) δ : -61.62 (*E*, s, CF<sub>3</sub>), -64.98 (*Z*, s, CF<sub>3</sub>); HRMS (*ESI*), m/z: calcd for [C<sub>14</sub>H<sub>13</sub>ClNO<sub>4</sub>F<sub>3</sub>Na]<sup>+</sup> 374.0383 [M+Na]<sup>+</sup>; found 374.0394.

# CRediT author statement

- S. G. performed and validated the experiments,
- P. R. XRay and structural investigation
- B. C. provided expertise and feedback.
- T. B. S. conceived, supervised, wrote the original manuscript and revised it.

#### conflict of interest

NOne

#### Acknowledgements

We are grateful to the Tunisian Ministry of High Education and Scientific Research for financial support (LR99ES14) of this research.

#### References

[1] A. Matošević, A. Bosak, Carbamate group as structural motif in drugs: a review of carbamate derivatives used as therapeutic agents, Archives of Industrial Hygiene and Toxicology. 71 (**2020**) 285–299. <u>https://doi.org/10.2478/aiht-2020-71-3466</u>

 [2] D. Jornada, G. d. S. Fernandes, D. Chiba, T. d.- Melo, J. d.-Santos, M. Chung, The Prodrug Approach: A Successful Tool for Improving Drug Solubility, *Molecules*. 21 (2015) 42. https://doi.org/10.3390/molecules21010042

[3] A. Dal Corso, V. Borlandelli, C. Corno, P. Perego, L. Belvisi, L. Pignataro, C. Gennari, Fast Cyclization of a Proline-Derived Self-Immolative Spacer Improves the Efficacy of Carbamate Prodrugs, Angew. Chem. Int. Ed. 59 (2020) 4176–4181. <u>https://doi.org/10.1002/anie.201916394</u>

[4] A. K. Ghosh, M. Brindisi, Organic Carbamates in Drug Design and Medicinal Chemistry, J. Med. Chem. 58
 (2015) 2895–2940. <u>https://doi.org/10.1021/jm501371s</u>

[5] S. Ray, D. Chaturvedi, Application of organic carbamates in drug design. Part 1: anticancer agents - recent reports, Drugs of the Future. 29 (**2004**) 343. <u>https://doi.org/10.1358/dof.2004.029.04.787236</u>

[6] (a) A. Mattarei, M. Azzolini, M. Zoratti, L. Biasutto, C. Paradisi, "N-Monosubstituted Methoxy-oligo(ethylene glycol) Carbamate Ester Prodrugs of Resveratrol," Molecules. 20 (2015) 16085–16102. <u>https://doi.org/10.3390/molecules200916085</u>. (b) S. M. Rahmathullah, R. R. Tidwell, S. K. Jones, J. E. Hall, D. W. Boykin, Carbamate prodrugs of N-alkylfuramidines, Eur. J. Med. Chem. 43 (2008) 174–177. <u>https://doi.org/10.1016/j.ejmech.2007.03.009</u>

 [7] U. Košak, N. Strašek, D. Knez, M. Jukič, S. Žakelj, A. Zahirović, S. Gobec, N-alkylpiperidine carbamates as potential anti-Alzheimer's agents, Eur. J. Med. Chem. 197 (2020) 112282. https://doi.org/10.1016/j.ejmech.2020.112282

[8] (a) J. Krzywik, M. Aminpour, J. Janczak, E. Maj, M. Moshari, W. Mozga, J. Wietrzyk, J. A. Tuszyński,
A. Huczyński, "An insight into the anticancer potential of carbamates and thiocarbamates of 10-demethoxy-10methylaminocolchicine," Eur. J. Med. Chem. 215 (2021) 113282.
<u>https://doi.org/10.1016/j.ejmech.2021.113282</u> (b) C. Huxley, M. Wibowo, K. Y.- Lum, S.Gordon, S. D'Hyon,
H.Guan, X. Wang, Y. Chen, M. Si, M. Wang, J. M. White, K. Wahi, Q. Wang, J. Holst, R. A. Davis,
Phytochemistry. 179 (2020) 112478. <u>https://doi.org/10.1016/j.phytochem.2020.112478</u>

[9] J. Krzywik, M. Aminpour, J. Janczak, E. Maj, M. Moshari, W.Mozga, A. Huczyński, -10-methylaminocolchicine, Eur. J. Med. Chem. 215 (2021) 113282. <u>https://doi.org/10.1016/j.ejmech.2021.113282</u>
[10] (a) S. Chander, P. Ashok, Y.-T. Zheng, P. Wang, K. S. Raja, A. Taneja, S. Murugesan, Design, synthesis and in-vitro evaluation of novel tetrahydroquinoline carbamates as HIV-1 RT inhibitor and their antifungal

activity, Bioorg. Chem. 64 (**2016**) 66–73. <u>https://doi.org/10.1016/j.bioorg.2015.12.005.</u> (b) S.-W. Chen,Y.-H. Wang, Y. Jin, X. Tian, Y.-T. Zheng, D.-Q. Luo, Y.-Q.Tu, Synthesis and anti-HIV-1 activities of novel podophyllotoxin derivatives, Bioorganic Med. Chem. Lett. 17 (**2007**) 2091–2095. <u>https://doi.org/10.1016/j.bmcl.2006.11.070</u>

[11] W. Löscher,; G. J. Sills,; H. S. White, The ups and downs of alkyl-carbamates in epilepsy therapy: How does cenobamate differ? .Epilepsia, 62 (**2021**) 596–614. <u>https://doi.org/10.1111/epi.16832</u>

[12] N. Kebir, M. Benoit, F. Burel, Elaboration of AA-BB and AB-<sup>'</sup> Type Non-Isocyanate Polyurethanes (NIPUs) Using a Cross Metathesis Polymerization between Methyl Carbamate and Methyl Carbonate Groups, Eur. Polym. J. 107 (2018) 155–163. <u>https://doi.org/10.1016/j.eurpolymj.2018.07.045</u>

[13] D. T. Sheppard, K. Jin, L. S. Hamachi, W. Dean, D. J. Fortman, C. J. Ellison, W. R. Dichtel, Reprocessing Postconsumer Polyurethane Foam Using Carbamate Exchange Catalysis and Twin-Screw Extrusion, ACS Central Science. 6 (2020) 921–927. <u>https://doi.org/10.1021/acscentsci.0c00083</u>

[14] Z. Qiao, Z. Yang, W. Liu, X. Wang, Y. Gao, Z. Yu, J. Xu, Molecular weight switchable polyurethanes enable melt processing, Chem. Eng. J. 384 (**2019**) 123287. <u>https://doi.org/10.1016/j.cej.2019.123287</u>

[15] S. Badreshia, J. G. Marks, Iodopropynyl Butylcarbamate, Contact Derm. 13 (**2002**) 77–79. https://doi.org/10.1097/01634989-200206000-00008

[16] (a) M. Shahid, S. Manoharadas, H.Chakdar, A. F. Alrefaei, M. F. Albeshr, M. H. Almutairi, Biological toxicity assessment of carbamate pesticides using bacterial and plant bioassays: An in-vitro approach, *Chemosphere*. 278 (2021) 130372. <u>https://doi.org/10.1016/j.chemosphere.2021.130372</u> (b) T. M. B. F. Oliveira, F. W. P. Ribeiro, C. P. Sousa, G. R. Salazar-Banda, P. d.- L.-Neto, A. N. Correia, S. Morais, Current overview and perspectives on carbon-based (bio)sensors for carbamate pesticides electroanalysis, Trac-Trend Anal Chem. 124 (2020) 115779. <u>https://doi.org/10.1016/j.trac.2019.115779</u>

[17] T. Nagai, N. Mimata, Y. Terada, C. Sebe, H. Shigehisa, Catalytic Dealkylative Synthesis of Cyclic Carbamates and Ureas via Hydrogen Atom Transfer and Radical-Polar Crossover, Org. Lett. 22 (**2020**) 5522–5527. <u>https://doi.org/10.1021/acs.orglett.0c01872</u>

[18] J. Li, C. Kornhaaß, L. Ackermann, Ruthenium-catalyzed oxidative C–H alkenylation of aryl carbamates, Chem Comm. 48 (**2012**) 11343. <u>https://doi.org/10.1039/c2cc36196e</u>

[19] Z. S. Al-Taie, J. M. Anderson, L. Bischoff, J. Christensen, S. J. Coles, R. Froom, E. C. Thompson, N-carbamate protected amino acid derived guanidine organocatalysts, Tetrahedron. 89 (2021) 132093. https://doi.org/10.1016/j.tet.2021.132093\_

[20] T. W. Greene, P. G. M. Wuts, Greene's Protective Groups in Organic Synthesis, 4th ed., John, New York,
 2007. <u>https://doi.org/10.1021/ja069836d</u>

[21] S. Shi, C. Yao, J. Cen, L. Li, G. Liu, J. Hu, S. Liu, High-Fidelity End-Functionalization of Poly(ethylene glycol) Using Stable and Potent Carbamate Linkages, *Angew. Chem. Int.* 59 (**2020**) 18172–18178. <u>https://doi.org/10.1002/anie.202006687</u>

[22] B. A. Dressman, L. A. Spangle, S. W. Kaldor, Solid phase synthesis of hydantoins using a carbamate linker and a novel cyclization / cleavage step, Tetrahedron Lett. 37 (**1996**) 937–940. <u>https://doi.org/10.1016/0040-4039(95)02395-x</u>

[23] S. Asano, K. Itto-Nakama, H. Arimoto, Carbamate as an accelerating group in intermolecular Pauson-Khand reaction, *Tetrahedron Lett.* 61 (2020) 151974. <u>https://doi.org/10.1016/j.tetlet.2020.151974</u>

[24] M. Das, A. Rodríguez, P. K. T. Lo, W. J. Moran, Synthesis of Oxazolidinones by a Hypervalent Iodine
 Mediated Cyclization of N-Allylcarbamates, Adv. Synth. Catal. 363 (2021) 1646–
 1650. <u>https://doi.org/10.1002/adsc.202001451</u>

[25] A. C. Bell, A. B. Boomsma, N. E. Flikweert, R. M. Hohlman, S. Zhang, R. L. Blankespoor, M. R. Barbachyn, The Synthesis of Functionalized 3-Aryl- and 3-Heteroaryloxazolidin-2-ones and Tetrahydro-3-aryl-1,3-oxazin-2-ones via the Iodocyclocarbamation Reaction: Access to Privileged Chemical Structures and Scope and Limitations of the Method, J. Org. Chem. 85 (**2020**) 6323–6337. <u>https://doi.org/10.1021/acs.joc.9b03400</u>

[26] P. Szcześniak, M. Pieczykolan, S. Stecko, The Synthesis of α,α-Disubstituted α-Amino Acids via Ichikawa Rearrangement, J. Org. Chem. 81 (2016) 1057–1074. <u>https://doi.org/10.1021/acs.joc.5b02628</u>

[27] H.-S. Yoon, J.-H. Kim, E. J. Kang, H.-Y. Jang, Platinum-Catalyzed Cyclization of N-Allyl Carbamates for the Synthesis of 5-Vinyloxazolidinones, *Chem. Eur. J.* (**2012**) 1901–1905. https://doi.org/10.1002/ejoc.201101799\_

[28] M. Garbacz, S. Stecko, The Synthesis of Chiral Allyl Carbamates via Merger of Photoredox and Nickel Catalysis, *Adv. Synth. Catal.* 362 (**2020**) 3213–3222. <u>https://doi.org/10.1002/adsc.202000404</u>

[29] S. Stecko, Total Synthesis of Lacosamide, J. Org. Chem. 79 (2014) 6342–6346. https://doi.org/10.1021/jo500857t

[30] Y. Ichikawa, Y. Morishita, S. Kusaba, N. Sakiyama, Y. Matsuda, K. Nakano, H. Kotsuki, An Efficient Synthesis of Carbamates by Tin-Catalyzed Transcarbamoylation Reactions of Primary and Secondary Alcohols, Synlett. (2010) 1815–1818. <u>https://doi.org/10.1055/s-0030-1258102</u>

[31] C. Qi, D. Yan, W. Xiong, H. Jiang, Silver-catalyzed regioselective coupling of carbon dioxide, amines and aryloxyallenes leading to O-allyl carbamates, *J. CO2 Util.* 24 (**2018**) 120–127. https://doi.org/10.1016/j.jcou.2017.12.013

[32] J. Seijas, ; M. Vázquez-Tato, ; J. C.-Campo, Microwave-Assisted Solvent-Free Synthesis of Enol Carbamates, *Synlett.* (2007) 2420–2424. https://doi.org/10.1055/s-2007-985607\_

[33] J. Cai, M. Zhang, X.Zhao, Synthesis of Allyl Carbamates through the Regioselective Domino Reaction of -Amines, CO2, and Unsymmetrical Allyl Chlorides under Pd Catalysis, Chem. Eur. J. (2015). 5925–5928. https://doi.org/10.1002/ejoc.201500769\_

[34] D. Saberi, S. Mansoori, E. Ghaderi, K. Niknam, Copper nanoparticles on charcoal: an effective nanocatalyst for the synthesis of enol carbamates and amides via an oxidative coupling route, *Tetrahedron Lett*. 57 (2016) 95–99. <u>https://doi.org/10.1016/j.tetlet.2015.11.072</u>

[35] M. Ogawa, M. Nagashima, H. Sogawa, S. Kuwata, T. Takata, Synthesis and Cavity Size Effect of Pd-Containing Macrocycle Catalyst for Efficient Intramolecular Hydroamination of Allylurethane, *Org. Lett.* 17 (2015) 1664–1667. <u>https://doi.org/10.1021/acs.orglett.5b00378</u>.

[36] S. Ghrairi, K. Essalah, B. Crousse, T. Barhoumi-Slimi, *J. Fluorine Chem.* 248 (**2021**) 109837. https://doi.org/10.1016/j.jfluchem.2021.109837

[37] T. Barhoumi-Slimi, M. Our'evitch,  $\beta$ -Chloro- $\alpha$ , $\beta$ -unsaturated carbonyls as convenient precursors of highly substituted benzenes, *J. Iran Chem Soc.* 15 (**2018**) 629–636. <u>https://doi.org/10.1007/s13738-017-1263-1</u>.

[38] L. Bahri, T.Barhoumi-Slimi, R. Mallek, M. A. K. Sanhoury, B. Crousse, M. T. Ben Dhia, One-pot synthesis of new highly substituted allylic phosphorodiamidates, *J. Fluorine Chem.* 189 (**2016**) 96–101. https://doi.org/10.1016/j.jfluchem.2016.08.002  [39] L. Bahri, T. Barhoumi-Slimi, M. A. K. Sanhoury, M. T. Ben Dhia, Synthesis and Characterization of New Variously Substituted Allylic Phosphates, *Heteroatom Chem.* 27 (2016) 190–195. https://doi.org/10.1002/hc.21316

[40] C. M. Marson,; Reactions of carbonyl compounds with (monohalo) methyleniminium salts (vilsmeier reagents), *Tetrahedron*. 48 (**1992**) 3659–3726, <u>https://doi.org/10.1016/s0040-4020(01)92263-x</u>.

[41] I. Hamzi, T. M.Barhoumi-Slimi, R. Abidi, Synthesis, Characterization, and Conformational Study of Acylhydrazones of  $\alpha,\beta$ -Unsaturated Aldehydes, Heteroat. Chem. 27 (**2016**) 139-148. <u>https://doi.org/10.1002/hc.21310</u>

[42] S. N. Podyachev, N. E. Burmakina, V. V. Syakaev, S. N. Sudakova, R. R. Shagidullin, A. I. Konovalov, Synthesis, IR and NMR characterization and ion extraction properties of tetranonylcalix[4] resorcinol bearing acetylhydrazone groups, *Tetrahedron.* 65 (2009) 408–417. <u>https://doi.org/10.1016/j.tet.2008.10.008</u>

[43] O. D. Rigaku, CrysAlis PRO. Rigaku Oxford Diffraction, Yarnton, Oxfordshire, England, 2015.

[44] G. M. Sheldrick, Acta Crystallogr. A71 (2015) 3-8. https://doi.org/10.1107/s2053273314026370

[45] G. M. Sheldrick, Acta Crystallogr. C71 (2015) 3-8. <u>https://doi.org/10.1107/s2053229614024218</u>

Junaler

[46] S. Parsons, H. D. Flack, T. Wagner, Acta Crystallogr. B69 (**2013**) 249-259. https://doi.org/10.1107/s2052519213010014

[47] W.L.F. Armarego, Purification of Organic Chemicals, in: W.L. F. Armarego (Eds.), Purification of Laboratory Chemicals, *Elsevier*, **2017**, 95–634. <u>https://doi.org/10.1016/B978-0-12-805457-4.50003-3.</u>

22