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Capucine Jourdain de Muizon, Donata Iandolo, Dung Kim Nguyen, Ali Al-Mourabit, Marthe Rousseau. Organic Matrix and Secondary Metabolites in Nacre. *Marine Biotechnology*, 2022, 24, pp.831-842. 10.1007/s10126-022-10145-9 . hal-03795731v2

HAL Id: hal-03795731

<https://hal.science/hal-03795731v2>

Submitted on 4 Oct 2022

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Organic Matrix and Secondary Metabolites in Nacre

Capucine Jourdain de Muizon^{1,2} · Donata Iandolo^{3,4} · Dung Kim Nguyen⁴ · Ali Al-Mourabit¹ · Marthe Rousseau^{3,4} 

Received: 1 March 2022 / Accepted: 12 July 2022

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Abstract

Nacre, also called mother-of-pearl, is a naturally occurring biomineral, largely studied by chemists, structural biologists, and physicists to understand its outstanding and diverse properties. Nacre is constituted of aragonite nanograins surrounded by organic matrix, and it has been established that the organic matrix is responsible for initiating and guiding the biomineralization process. The first challenge to study the organic matrix of nacre lays in its separation from the biomineral. Several extraction methods have been developed so far. They are categorized as either strong (e.g., decalcification) or soft (e.g., water, ethanol) and they allow specific extractions of targeted compounds. The structure of the nacreous organic matrix is complex, and it provides interesting clues to describe the mineralization process. Proteins, sugars, lipids, peptides, and other molecules have been identified and their role in mineralization investigated. Moreover, the organic matrix of nacre has shown interesting properties for human health. Several studies are investigating its activity on bone mineralization and its properties for skin care. In this review, we focus on the organic constituents, as lipids, sugars, and small metabolites which are less studied since present in small quantities.

Keywords Nacre · Biomineralization · Extraction · Organic matrix · Chitin · Glycosaminoglycans · Lipids · Peptides · Bones

Introduction

How is mineral material included in living organisms? How are so many different mineralized structures created with the most common mineral building blocks (calcium carbonate CaCO_3 , silica SiO_2 , and phosphate calcium CaPO_4) and combined with organic material (proteins, carbohydrates, lipids, secondary metabolites)? What is the secret of the diversity of the necessary biominerals produced by living organisms? This is the marvel of biomineralization: a dynamic physiological process by which a living organism synthesizes a mineralized structure (Marin et al. 2007).

Interestingly, the nature of the organic molecules influences or even controls the calcium carbonate crystallization towards one of the possible polymorphs such as calcite, aragonite, or vaterite forms.

In the marine environment, biomineralization is performed by a wide range of living systems going from mollusks, brachiopods, or sponges to corals, sea urchin, coccoliths, and foraminifera (Cusack and Freer 2008). It is used to build a skeleton or a protective structure, as in the case of embryonic eggshells, to perform photosynthesis like coccoliths (Gordon and Du 2001), to detect the earth magnetic field for magnetotactic bacteria (Scheffel et al. 2006), or to protect cells from intense UV radiations (Price et al. 1998). The most common bricks involved in the elaboration of the mineral component are calcium phosphate, calcium carbonate, and silica. The variety of the known biomineral ultrastructures is driven by the diversity of organic matrix. Organic matrix is a minor constituent in terms of weight percentage, but is controlling important structural aspects and allowing the skeletal formation in organisms (Cusack and Freer 2008). In contrast with the high molecular weight molecules of the skeletal part that have been extensively studied, the small ones are poorly investigated.

Mollusks belong to the second-largest phylum in invertebrates (Mollusca) and are of great interest for the study

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of biomineralization process. The Mollusca phylum shows a significant diversity of morphologies and living places (e.g., sea water, fresh water, or earth) (Lowenstam and Weiner 1989). The shell or exoskeleton is the product of a biomineralization process. The structure of shell biominerals is studied applying several approaches including micro and ultrastructural observations by electron microscopy (SEM, TEM), physical measurements (X-ray diffraction), and biochemical characterizations. The results from these studies open to promising perspectives, due to the structure and mechanical properties of molluscan nacre (Mayer 2005) and for the use of bioactive corals- and mollusk-based implants as bone substitutes (Begley et al. 1995; Westbroek and Marin 1998; Zhang et al. 2017).

Among mollusk shell structures, nacre is widely studied for its attractive mechanical properties as it is a tough and strong (Sun and Bhushan 2012), yet light material. Nacre is also worthy of interest for application in the medicinal domain: it has been shown that it is osteogenic, osteoconductive, biocompatible, and biodegradable. A review by Zhang et al. relates the challenges and advances in using nacre as bone substitutes (Zhang et al. 2017). From the structural point of view, nacre is formed by alternating porous organic films and calcium carbonate layers (Grégoire et al. 1954). The organic matrix is studied to understand the mechanism of biomineralization and its structural function as well as for its medicinal properties. To perform these studies, nacre is extracted from the shell of different species of pearl oysters, mainly *Pinctada margaritifera*, *Pinctada maxima*, and *Pinctada fucata* collected in the Pacific Ocean (Polynesia, north Australia, and Japanese coast, respectively) (Southgate and Lucas 2008).

Until now, the study of the organic matrix has as its main objective the identification of the structural proteins composing it. These studies are undertaken either to understand the functions of the matrix and the mechanisms of biomineralization (Addadi et al. 2006) or to develop innovative, biomimetic applications (e.g., nanotechnologies, semiconductors, or natural bioactive factor) (Marin et al. 2007). The organic–inorganic assembly in nacre is also a matter of intensive research to perform the synthesis of artificial nacre from cheap abundant materials (Luz and Mano 2009; Gerhard et al. 2017; Huang et al. 2019). The integration of the organic matrix in nacre plays a key role in its formation, structure, and for its properties, despite being less than 3% of the shell weight.

Nacre is integrated in the mollusk shell structure, composed of several layers. The outer part, or periostracum, is mainly organic and the intermediary layer is calcitic, called prismatic layer. Then there is the nacreous layer covering the inner shell with its aragonitic phase. The mollusk body is directly in contact with the nacre, through an organic layer. The organic matrix is present in this structure at several

levels: either assuring the interface between the layers or inside them. The primary structural components of nacre are pseudohexagonal aragonite tablets as shown in Fig. 1D. The intra-crystalline organic matrix is found inside the tablets while the inter-tabular (nm thick) and inter-lamellar (50–100 nm) matrix are found between individual tablets or layers (Rousseau et al. 2005).

In 1855, Fremy performed the first study on the organic matrix of mollusk shells and designated the insoluble matrix as “conchiolin” (Fremy 1855). Conchiolin was finally biochemically analyzed by Gregoire et al. in 1954 (Grégoire et al. 1954). In the last decades, many scientists have worked on the proteic components of nacre. A large number of shell matrix proteins involved in the nacreous layer formation have been identified, such as Pif (Suzuki et al. 2009), MSI60 (Sudo et al. 1997), lustrin A (Shen et al. 1997), N16/pearlin (Samata et al. 1999), perlucin, perlustrin (Weiss et al. 2000), N14, N66 (Kono et al. 2000), mucoperlin (Marin et al. 2000), AP7, AP24 (Michenfelder et al. 2003), P10, perlwapin (Treccani et al. 2006), perlinhibin (Mann et al. 2007), N19 (Yano et al. 2007, p. 19), N40 (Yan et al. 2007), and blue mussel shell protein (BMSP) (Suzuki et al. 2011). A recent review reports on the advances made in the characterization of shell matrix protein and cellular orchestration in marine molluscan shell biomineralization (Song et al. 2019).

The present review will focus on the other biomolecules extracted from the nacreous organic matrix such as sugars, lipids, peptides, and other small metabolites that are part of the organic matrix and are way less studied than proteins. Methods for their extraction and separation processing will be discussed.

Organic Matrix Structure

Following a consequent increase of calcium in the sea water, some mollusks evolved to grow shells (Marin et al. 1996). Biomineralization remains today an underexplored phenomenon. Many scientists are investigating the process of nacre formation to elucidate the role of the organic matrix. The organic structure is present prior to the formation of minerals as shown in Fig. 2. Proteins are instructing the process of biomineralization and are then incorporated in the mineral, conferring its mechanical properties to the nacre (Launspach et al. 2012).

The nacreous organic matrix have been studied by TEM (transmission electron microscopy), AFM (atomic force microscopy) and SEM (scanning electron microscopy), mass spectrometry (e.g., nanoSIMS—nanoscale secondary ion mass spectrometry), IR (infra-red spectroscopy), or X-ray analysis to determine its physical organization around mineral phases and how the biomolecules are self-organized

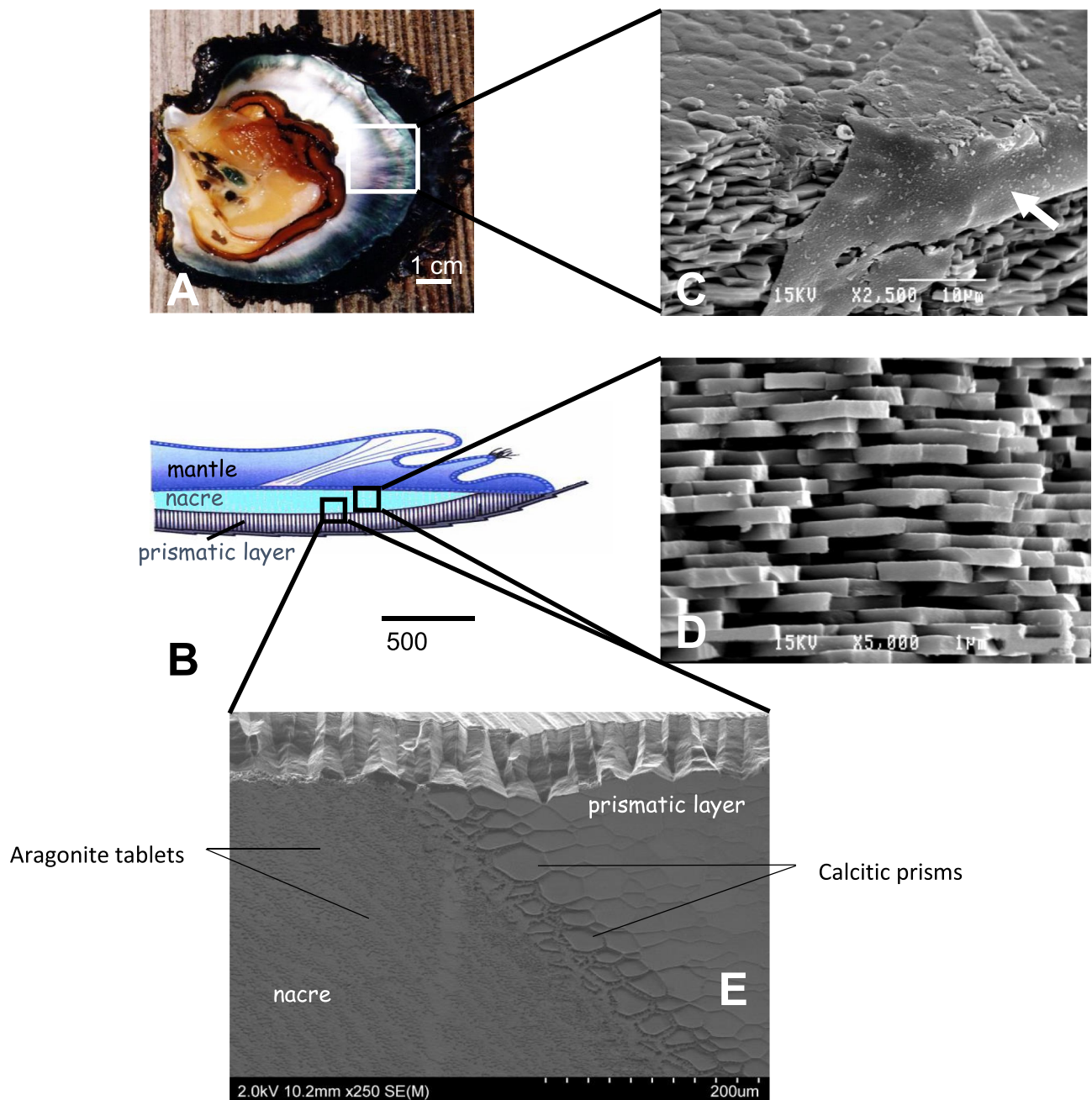


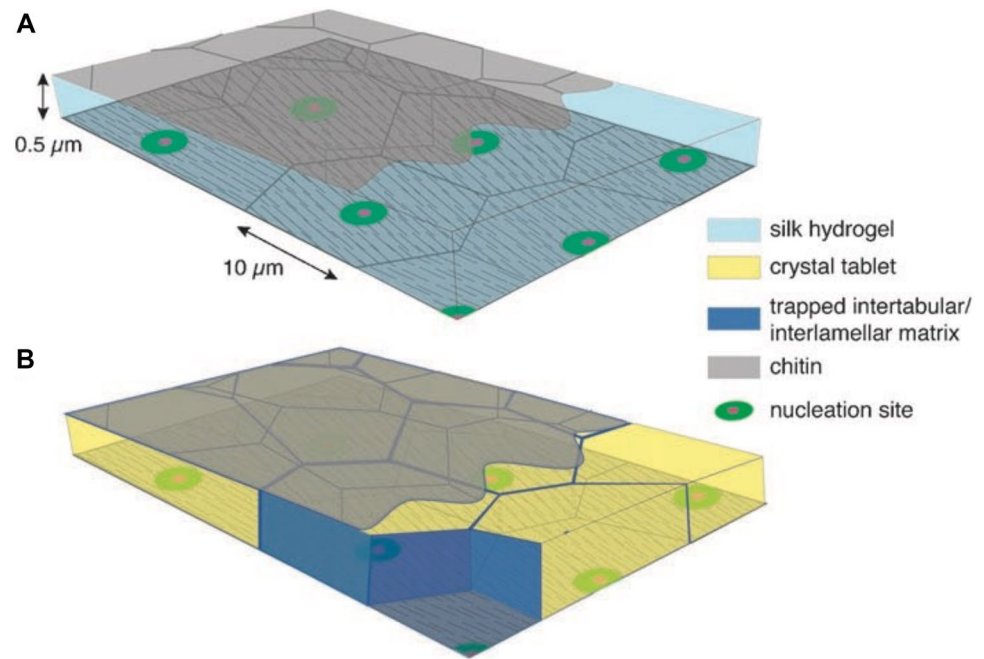
Fig. 1 The structure of the shell of the pearl-producing oyster, *Pinctada margaritifera* (redrawn from Rousseau et al. 2009). **A** An open shell with the soft tissue in place. The brighter nacreous layer of aragonite is clearly visible next to the animal. The nacreous layer coats the darker, prismatic shell of calcite that forms the outer shell. **B** Schematic representation of the shell structure and the overlying mantle of the animal. The mantle is an organic tissue acting as an interface between the organism and nacre tablets, secreting com-

pounds involved in mineralization. **C** Electron microscopy image of the surface membrane covering the forming nacre tablets (white arrow) after mechanical fractures of the shell. **D** Electron microscopy image of the brick-and-mortar structure of the nacreous layer after fracture of the shell. Each tablet is about 0.5–1 µm thick. **E** SEM image of the cross-section of the nacreous to prismatic transition in the shell

(Cartwright and Checa 2007). The mineral part of nacre is composed of aragonite tablets whereas that of the prismatic layer is composed of calcitic prisms, the two most common Ca-carbonate polymorphs in shells. The two scales of

these structures can be clearly seen in Fig. 1E. In nacre, intra-crystalline, inter-tabular, and inter-lamellar matrix were identified. Rousseau et al. studied the composition of the inter-tabular and inter-lamellar matrix. The shell was

Fig. 2 Scheme of a suggested model for the process of nacre formation **A** before mineralization and **B** after mineralization (Addadi et al. 2006)



prepared to access the surface between the mantle and the shell, where nacre is growing from the mantle. This surface was observed by SEM (scanning electron microscope) and nanoSIMS. They concluded that inter-tabular matrix is rich in aragonite nucleating proteins and carboxylates while inter-lamellar matrix is composed of chitin and other organic material (Rousseau et al. 2009). Organic envelopes or membranes are surrounding nacre, as shown in Fig. 1. In the prismatic layer, organic matrix is divided into three categories: interlayer membranes at the calcite-aragonite transition, interprismatic membranes, and intraprismatic membranes (Farre et al. 2011). These three kinds of organic matrix have revealed different compositions under TOF-SIMS (time of flight secondary ion mass spectrometry) analysis. Also, IR spectroscopy has allowed to show that the layers are of irregular thickness (Dauphin et al. 2010).

By studying nacre layers after incomplete decalcification, Weiner and Traub noticed a well-defined spatial relationship between chitin, proteins, and aragonite (Weiner and Traub 1980). These observations confirmed previous results representing the organic matrix as chitin fibrils lined up in a defined direction on a protein surface (Goffinet 1969). The organic matrix is described as a thin layer of β -chitin sandwiched between layers of glycine and alanine-rich proteins (Weiner and Traub 1984). In more recent studies, the way the chitin filament is associated with several proteins has been investigated (Mann et al. 2000, 2007; Weiss et al. 2001; Fu et al. 2005, p. 3). X-ray and electron diffraction pattern of organic matrices and mineral crystals have highlighted that the chitin fibers and the protein polypeptide chains are aligned with the a

and b aragonite crystallographic axes respectively (Fig. 3) (Weiner et al. 1983).

This organization of nacre's organic matrix clearly shows that organic molecules are implied in the process of biomineralization as well. This matrix is mainly composed of proteins and high molecular weight biomolecules, such as sugars or lipopeptides. The specific role of these compounds will be clarified in the following parts.

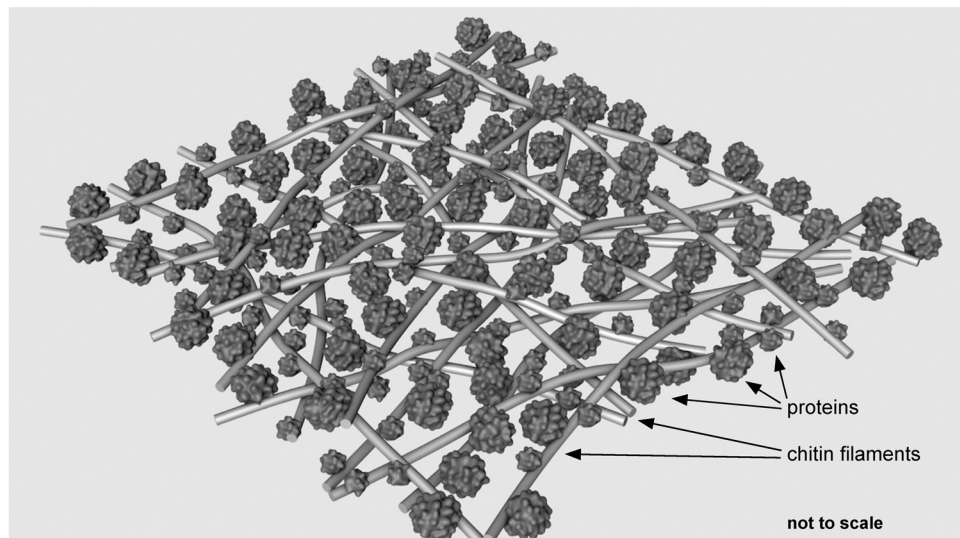
Extraction Methods

As organic matrix is only 2.3% of the nacre (Bourrat et al. 2007), extraction protocols remain a critical point to access the desired organic component. Demineralization is the most obvious method to extract organic compounds. It can be performed by acidification using hydrochloric acid (HCl) or in a softer way by using a chelating agent such as ethylenediaminetetraacetic acid (EDTA). A couple of soft extraction methods, which avoid demineralization, have been developed using either ethanol or water that have been shown to be efficient to solubilize a part of the organic matrix. The organic matrix is divided into two phases, soluble or insoluble in organic solvents.

Demineralization or Total Extraction Methods

Traditionally, the nacre organic matrix was extracted by demineralization. Demineralization by hydrochloric acid is considered to lead to a total extraction of the components of

Fig. 3 Schematic interpretation of protein-decorated chitin matrix (Launspach et al. 2012). Chitin filaments (with a not specified acetylation degree) form a network with pores of varying diameter. The filaments are decorated with different kinds of proteins, which bind to the filaments with different strengths



the organic matrix. This method has been used to study the structure and composition of mollusk shell organic matrix (Marin et al. 2001; Mann et al. 2007). To demineralize the nacreous layer, hydrochloric acid is added to powdered nacre to bring the suspension to pH 4 overnight. This method coupled with chromatography enabled to isolate proteins of the organic matrix. This drastic method certainly implies degradation of some acid-sensitive compounds.

A softer approach to demineralization uses ethylenediaminetetraacetic acid (EDTA). The demineralization can be performed by dialysis against a solution containing EDTA (3 weeks long) (Launspach et al. 2012) or by dissolving powdered nacre, directly into EDTA (stirring 72 h) (Takakura et al. 2008). EDTA-soluble and EDTA-insoluble fractions are thus separated. As proportions are very rarely indicated in the literature, it is difficult to give the yields and the extraction soluble/insoluble ratios.

According to Launspach et al. (2012), several different proteins have been identified to bind to chitin. As chitin is a polysaccharide building a complex intermolecular structure, it stays in the EDTA-insoluble phase after demineralization. Consequently, this demineralization approach does not allow to access the whole panel of proteins and molecules of the nacre organic matrix.

EDTA-based demineralization and further water extraction were compared in a study published by Pereira-Mouriès et al. (2002). Both methods gave around 0.05% (w/w) yield from powdered nacre. The proteins present in the two different extracts were analyzed (Whiteman Alcian blue binding technique, Fourier transformation infrared spectroscopy, electrophoresis, spectrophotometry) and shown to be different. The water-soluble extract seems to achieve a composition close to the EDTA-insoluble matrix. The EDTA-soluble matrix was aspartic-rich. Hydrophilic negatively

charges molecules as aspartic-rich macromolecules are thought to interact specifically from solution to growing calcite crystals, suggesting that they are involved in the crystal formation process (Weiner and Addadi 1997). Importantly, the study of Pereira-Mouriès questions the widely accepted classification between soluble and non-soluble compounds in nacre organic matrix, as some of the EDTA-insoluble compounds are present in the water-soluble matrix (detailed below). In the EDTA-insoluble phase, there might be a crosslinking of proteins mediated by phenoloxidasés, a well-known family of redox enzymes present also in the mollusk shell. This crosslinking would prevent proteins from dissolving in EDTA (Lowenstam et al. 1982).

Partial Extraction Methods

Methods to extract organic compounds from the powdered nacre without demineralizing have also been developed. A soft extraction by pure water of a very fine nacre powder was described for the first time by Almeida et al. in 2000 (Almeida et al. 2000). The powdered nacre was suspended in ultrapure water at room temperature for 20 h under continuous stirring. The water-soluble matrix (WSM) was then separated by size exclusion high-performance liquid chromatography to obtain a first rough separation of the polar compounds which were finally fractionated by anion exchange chromatography. This method seems promising to study molecules that are active on bone formation such as polyanionic polymers involved in mineral growth (Weiner and Addadi 1991).

In another method, organic matrix was extracted from the nacre powder using pure anhydrous ethanol containing 0.1% (v/v) of hydrochloric acid (Zhang et al. 2016). The

polar protic organic solvent ethanol allows to better extract organic compounds from nacre. The ethanol-soluble matrix was then fractionated into cationic and anionic parts. Powdered nacre is suspended in ethanol to extract the molecules at the surface of the fine particles of nacre powder. The ethanol-soluble matrix activity on mineralization was tested in vitro and it was proved in this study that the cationic part of the extract is active in stimulating the mineralization of the osteoblast extracellular matrix. It is interesting to notice that while the anionic part containing polyanionic polymers is the one responsible of calcium carbonate mineral growth according to Weiner and Addadi (1991), it is the cationic part of the ethanol-soluble matrix which is promoting the formation of hydroxyapatite in the extracellular matrix of osteoblasts (Zhang et al. 2016).

A lipophilic extraction method was developed by Rousseau et al. (2006). Lipidic matrix was extracted by the mixture chloroform/methanol (2:1, v/v). This method allowed the identification of different classes of lipids in the nacreous organic matrix.

Finally, a protease-mediated hydrolysis was developed by Sasaki et al. to study proteins and peptides from *Pinctada fucata* nacre (Sasaki et al. 2019; Suzuki et al. 2019; Kintsu et al. 2020). This enzymatic hydrolysis was performed using the protease, nucleicin, and orientase 22 BF derived from *Bacillus subtilis*. Substrate and enzyme were mixed in a buffer solution and were incubated under specific optimal conditions (50 °C, pH 7.0 for nucleicin, and 60 °C, pH 9.2 for orientase). After the enzymatic reaction, the hydrolysate was boiled in water to inactivate the enzyme.

The extraction phase remains a challenge to the study of the nacreous organic matrix. Several methods have been developed to extract different compounds; however, most of these approaches are adapted to extract and study macromolecules and particularly proteins. The only published method adapted to extract specifically other biomolecules and small metabolites of the organic matrix is the lipid extraction method. Nevertheless, this method needs to be improved to selectively extract small molecules.

Biomolecules

Proteins

Proteins are nacre most studied compounds. Several of them have been isolated, identified, and tested for their biological activity. An exhaustive review about proteins in mollusks' shell was recently published (Song et al. 2019), and therefore, they will not be further detailed here. Two older reviews are also available (Marin and Luquet 2004; Suzuki and Nagasawa 2013).

Sugars

Chitin

Chitin represent a huge part of the biomass on earth (Falini and Fermani 2004). It is a long chain polysaccharide, composed of β -(1–4)-linked N-acetyl-D-glucosamine subunits (Fig. 4) (Schönitzer and Weiss 2007; Muzzarelli et al. 2011) that is produced by many different species, such as mollusks, shrimps, insects, or fungi. Three forms of chitin have been identified, known as α -, β -, and γ -chitins. α - and β -chitins are well known. In mollusks shells, only β -chitin is present (Lowenstam and Weiner 1989). This form of chitin is widely studied and a complete crystallographic study is available (Gardner and Blackwell 1975). The chemical and physical properties of chitin are strongly influenced by the degree of polymerization, degree of deacetylation, and covalent or noncovalent protein modification (Weiss and Schönitzer 2006). A review by Falini and Fermani gives a comprehensive overview of chitin structure and role in biomineralization (Falini and Fermani 2004).

Chitin is the major constituent in the larval shell matrix. It seems to play an important function for the guidance of mineral deposition in the mollusk shell (Weiss and Schönitzer 2006). The impact of chitin synthesis and degradation in biomineralization was also highlighted (Weiss et al. 2006; Suzuki et al. 2007; Kintsu et al. 2017). As chitin structure is evolving between the larval and the adult shell, studying this component at the larval stage gives essential information about the role chitin plays in the material properties and development of the shell, by coordinating and guiding the mineralization process (Weiss and Schönitzer 2006). For example, Ps19 has been identified as a chitin binding protein and shows also calcium binding properties (Arroyo-Loranca et al. 2020, p. 19). To support these findings, it has been shown that a partial inhibition of chitin by nikkomycin Z (which imitate UDP-N-acetylglucosamine) leads to shell malformations (Schönitzer and Weiss 2007). Even if it cannot be a sufficient proof that chitin is inducing biomineralization, as nikkomycin Z is a non-selective inhibitor, this study shows that the mollusk chitin synthase plays an

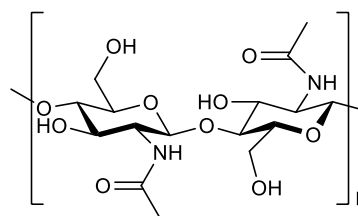


Fig. 4 Structure of the chitin molecule, showing two of the N-acetylglucosamine units that repeat to form the long chains in β -(1–4) linkages

important enzymatic role in the coordinated formation of larval bivalve shells.

A very relevant study on biomineralization has been reported using a β -chitin-silk fibroin complex (Levi-Kalisman et al. 2001). β -Chitin from the pen of the squid *Loligo* and silk from the cocoons of silkworm *Bombyx mori* have been purified and were allowed to interact and stabilized in the β conformation by treatment with methanol. Crystallization was then induced by incubation of the substrate complex in a saturated solution of CaCO_3 . The results of this study showed that mineralization in aragonitic form or calcitic form is induced by the association of chitin and macromolecules. In vitro, chitin was also associated with soluble macromolecules in the absence of silk. In these conditions, calcite is formed if the macromolecules are from calcitic origin, but vaterite spherulites are formed inside the substrate if the macromolecules were from aragonitic origin. This study clearly highlights the central role of chitin in biomineralization. A schematic representation of organic matrix' structure is shown in Fig. 5.

Other Sugars

A few other kinds of sugars have been identified in the nacreous organic matrix from *Pinctada fucata* and *Unio pictorum* such as acidic glycosaminoglycans (GAGs), sulfated sugars, and glycoproteins (Marie et al. 2007; Takakura et al. 2008). Advanced studies of the composition of nacre acid soluble matrix demonstrated that a high ratio of the

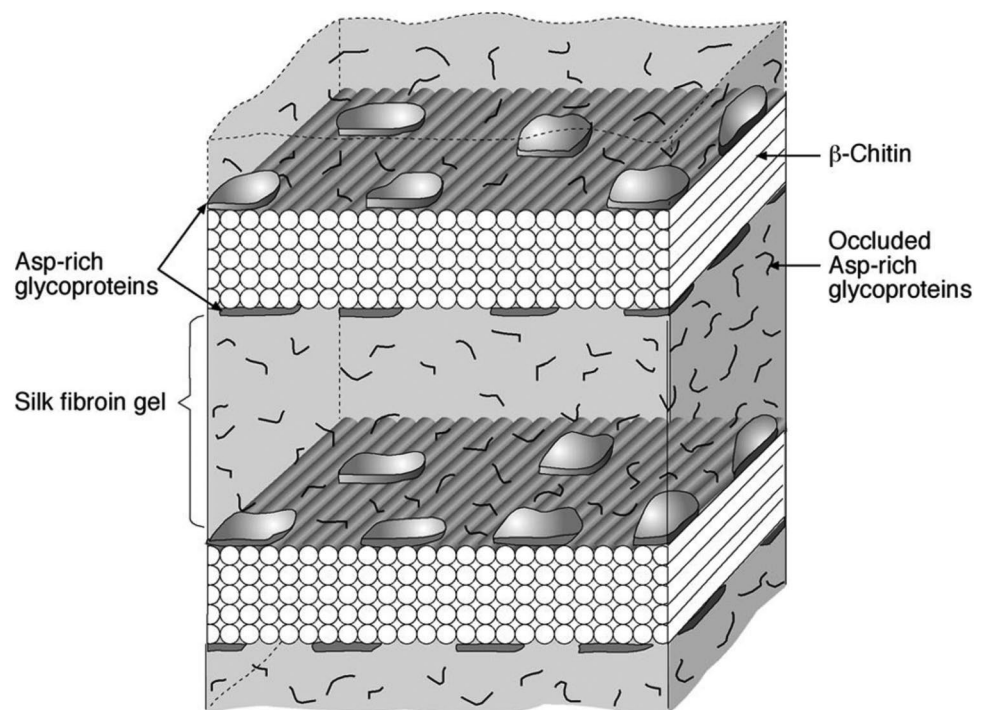
carbohydrates is sulfated and that the major proteins of this soluble matrix are heavily glycosylated (Marie et al. 2007). Marxen et al. proceeded to the isolation by gas chromatography and biochemical identification of the carbohydrates. Glucose, galactose, mannose, N-acetyl-glucosamine, and N-acetyl-galactosamine have been identified after hydrolysis as the major carbohydrates of the soluble and insoluble matrix (Marxen et al. 1998).

Carbohydrate moieties of glycoproteins have also been studied. N-linked oligosaccharides were enzymatically liberated from the nacreous organic matrix of Japanese pearl oyster *Pinctada fucata*. The structure of these carbohydrates covalently linked to the protein nacrein was identified by tandem mass spectrometry (MSMS) analysis (Takakura et al. 2008).

To investigate the role of carbohydrates in biomineralization, carboxylates of Asp-rich proteins and sulfated sugars were mapped on the surfaces of decalcified interlamellar matrices from the nacreous shell layer of the cephalopod *Nautilus pompilius*. This matrix was divided in four zones (Addadi et al. 2006). It had been demonstrated by Crenshaw and Ristedt that the highest concentration of sulfated sugars is in the central zone, where nucleation occurs (Crenshaw and Ristedt 1976; Marxen et al. 1998).

Dauphin and Marin studied cephalopod shell carbohydrates by FTIR (Fourier transform infrared spectrometry) and high-performance anion exchange chromatography. Four species were compared, including *Nautilus macromphalus* and *Nautilus pompilius* which are two nacreous species. They concluded that mannose, fructose, xylose, and ribose are absent from every studied matrix, while deoxyribose is

Fig. 5 A schematic representation of the structure of the demineralized *Atrina* nacreous layer organic matrix (Levi-Kalisman et al. 2001)



specifically present in the *Nautilus* shell. *Nautilus* showed the highest content of glucose and galactose, and the lowest content in glucosamine (Dauphin and Marin 1995).

As for the role of polysaccharides in biomineralization, different hypotheses have been investigated. Sulfated polysaccharides, for example, may have a calcium binding role. Also, it has been shown that complete deglycosylation of the total organic matrix of freshwater mussel *Unio pictorum* leads to a loss of calcium binding ability of the two main matrix proteins and to a modification of the shape of the mineral (Marie et al. 2007). These observations are consistent with Takakura et al. who had concluded that nacrein is supplying Ca^{2+} and HCO_3^- through the carbonic anhydrase domain and N-glycan moieties (Takakura et al. 2008). To support these results, studies were driven on the PmCHST1 gene of *Pinctada fucata martensii*, encoding a protein which participates in the keratan sulfate synthesis. Keratan sulfate possesses considerable amounts of negatively charged sulfonic acid groups and contributes to biomineralization. The repression of this gene induced a decrease of sulfated polysaccharides in extrapallial fluid, the fluid located between the external epithelium of the mantle and the inner face of the shell, and caused disordered growth of nacre layer in shells (Hao et al. 2018). This gene possesses a sulfotransferase domain involved in the transfer of sulfate moieties to chondroitin sulfate disaccharides (Wang et al. 2017). The modification in mineralization due to the extinction of this gene supports the role of sulfated sugar in biomineralization.

Lipids and Terpenes

Even if lipids in nacre are way less studied than sugars or proteins, they are still part of the organic matrix. It is assumed that they represent 0.3 to 0.7% of the total weight of the shell (Rousseau et al. 2006; Saleuddin and Mukai 2017). Only the lipids of the soft part of the oyster have already been studied (Saito 2004). Rousseau et al. published the first results on the extraction of the lipids from the nacre organic matrix (Rousseau et al. 2006, p. 200). The precise lipidic composition of the nacreous organic matrix of two Pteriomorpha mollusks (*Pinctada margaritifera* and *Pinna*

nobilis) was reported by Farre and Dauphin (2009). The organic matrix was studied by FTIR and TLC (thin layer chromatography) to detect lipids. Polar lipids (phospholipids), fatty acids (oleic acid), and waxes were identified as the major lipidic components. Sterols (cholesterol), triglycerides (triolein), and steroids (stearyl oleate) were also detected (Fig. 6). The presence of these compounds is consistent with the common lipids found in shells (Saleuddin and Mukai 2017). It has been shown that the lipidic composition of the two species differs, as these species are living in two very different environments, namely, the Pacific Ocean and the Mediterranean Sea.

It has to be noted that the same combination of lipids is found in several invertebrate skeletons. Lipids and sterols seem to be associated to the biomineralization process, as it was demonstrated that the removal of cholesterol and phospholipids from the shell matrix prevents biomineralization (Rousseau et al. 2006).

When lipids from nacre and flesh of *Pinctada radiata* oysters were studied in vitro (Ben Ammar et al. 2019), it was shown that they induce osteogenic MC3T3-E1 cell differentiation and in this way induce mineralization of the extracellular. These results are suggesting nacre lipids as good candidates in the quest of curing osteoporosis.

Peptides and Other Metabolites

Forty-two peptides from nacre are reported on uniprot.org database (Zhang 2017). Few of them have been studied in detail and some of the reported peptides are the product of protein's hydrolysis as the two that will be detailed herein.

Three n16 proteins have been extracted from water insoluble matrix of *Pinctada fucata*. These three proteins are sharing 89% sequence identity, encoding between 129 and 131 amino acids with the 23 first amino acids apparently comprising a signal peptide (Samata et al. 1999). The 30 amino acid peptide N-terminal half (n16N) of the n16-1 protein was identified as the mineral-binding site (Metzler et al. 2010). This peptide has been recognized to preferentially nucleate aragonite once it is bound to β -chitin. A mutant of this peptide nucleates calcite (Keene et al. 2010). A second

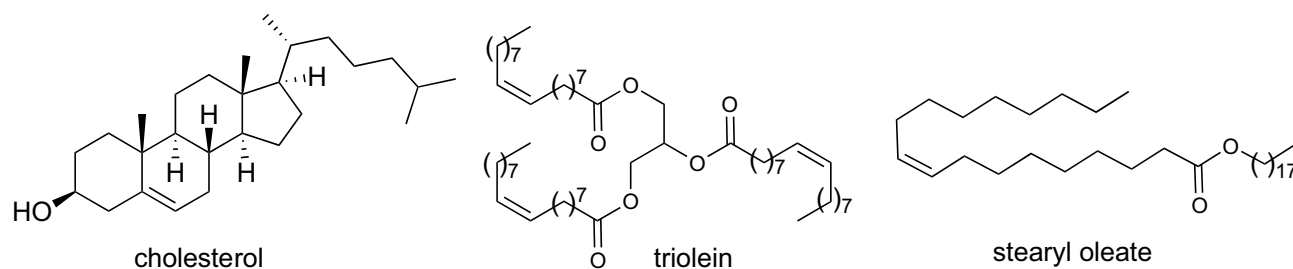


Fig. 6 Cholesterol, triolein, and stearyl oleate are three common lipids found in nacre organic matrix

study noticed that the 15 N-terminal amino acids sequence of n16 protein promotes limited aragonite formation in vitro, while the 15 C-terminal amino acids part promotes calcite formation (Brown et al. 2014).

In another study, the nacreous organic matrix was hydrolyzed using two types of proteases to isolate a six amino acids peptide: GVGSPY. This peptide is an angiotensin I-converting enzyme inhibitory peptide and have shown activity in preventing high blood pressure (Sasaki et al. 2019).

Studies on nacre small molecules have been mostly undertaken on the whole water-soluble matrix mixture extracted from *Pinctada margaritifera* and *Pinctada maxima*. The molecules recovered from this matrix are first separated by size exclusion chromatography based on their molecular mass. It has been shown that between 110 and 300 different molecules ranging from 100 to 700 Da are present in the water-soluble matrix (Duplat et al. 2007; Rousseau et al. 2008). The organic water-soluble matrix presents only a low peptide content (10–12%, w/w).

Fractionation by dialysis, solvent extraction, reversed phase HPLC, or size exclusion chromatography allowed to study the water-soluble matrix in more details. Isolated fractions are then analyzed by mass spectrometry to identify the unknown compounds. Bédouet et al. identified four peptides: H-I/LGGI/L-OH, H-I/LGGGI/L-OH, H-G/KGAGI/L-OH, H-G/KGGGI/L-OH, and K/QGGI/L (Bédouet et al. 2006, 2007). Except for the five peptides mentioned, very few studies have been driven on the low molecular weight molecules extracted from nacre organic matrix.

The studies focusing on the small molecules present in nacre organic matrix aim above all to highlight their biological activity. In their studies, Bédouet et al. separated the low molecular weight components (< 8000 Da) from the water-soluble matrix. This fraction have shown a highly specific inhibitory action for the nacreous water-soluble matrix on both papain and cathepsin B (Bédouet et al. 2007; Duplat et al. 2007).

Earlier mineralization of MC3T3-E1 pre-osteoblasts was observed in the presence of the water-soluble matrix rather than in the presence of melatonin, a molecule known for its activity on cell differentiation (Rousseau 2003). In the same study, it was showed that a nacre fraction containing compounds with a molecular weight between 50 and 300 Da, comprising amino acids and peptides at trace level, was active on cell matrix mineralization. It was postulated that an ensemble of molecules is suggested to promote matrix mineralization, supposedly because the signal molecule of nacre might be the low molecular weight molecules (Rousseau et al. 2008).

These first studies are promising in the light of the suggested biological activities. It would be interesting to carry out additional chemical studies for the isolation of more small molecules from the nacreous organic matrix. Further

chemical studies can turn out important to better understand the process of biomineralization and to potentially identify drug candidates to treat bone diseases.

Conclusion

Biomineralization is an interesting complex phenomenon whose studies are approached from various angles. Nacre is a well-studied biomineral produced by mollusks. An interesting way to investigate how nacre is formed and nacre's mineral structure is to isolate and characterize the organic molecules of nacre. It was shown that the nacreous organic matrix is indeed initializing and guiding biomineralization. Studies were conducted about proteins mostly, but lipids, terpenes, sugars, or peptides are reported as well. Although our understanding of the structure of nacre has improved and the biomineralization processes leading to the formation of nacre have been partially enlightened, much remains to be explored. Understanding biomineralization remains a challenge in fundamental research and is also a point of interest for its applications.

Nacre organic matrix is also studied for its potential use for biomedical applications. Nacre extracts are indeed tested on diverse biological models, including mineralization properties or bones and skin barrier repairing. Small molecules among the secondary metabolites of this matrix remain very poorly studied. New extraction methods have to be developed to target them. The discovery of new small molecules could lead to a better understanding of biomineralization and to the development of new drug candidates.

Author Contribution C.J., A.A., and M.R.: writing—original draft preparation; C.J., D.I., K.N., A.A., and M.R.: writing—review and editing; M.R.: idea and project administration. All authors have read and approved the manuscript.

Funding D.I. acknowledges the financial support by project IDEX-LYON of the University of Lyon in the frame of the “Programme Investissements D’Avenir” (ANR-16-IDEX-0005, IDEX/ELAN/2019/01). The Agence Nationale de la Recherche awards the ANR JCJC 2018 project called NADO for financial support (N°ANR-18-CE18-0003–01).

Declarations

Conflict of Interest M.R. is one of the co-founders of the company Stansea that was created in 2013 and that commercialize nacre.

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