



Quantifying the influence of dispersion interactions on the elastic properties of crystalline cellulose

Pan Chen, Yoshiharu Nishiyama, Jakob Wohlert

► To cite this version:

Pan Chen, Yoshiharu Nishiyama, Jakob Wohlert. Quantifying the influence of dispersion interactions on the elastic properties of crystalline cellulose. *Cellulose*, 2021, 28 (17), pp.10777-10786. 10.1007/s10570-021-04210-0 . hal-03746715

HAL Id: hal-03746715

<https://hal.science/hal-03746715>

Submitted on 6 Aug 2022

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

Quantifying the Influence of Dispersion Interactions on the Elastic Properties of Crystalline Cellulose

Pan Chen^{1*}, Yoshiharu Nishiyama^{2*}, Jakob Wohlert^{3,4*}

¹Beijing Engineering Research Centre of Cellulose and Its Derivatives, School of Materials Science and Engineering, Beijing Institute of Technology, 100081, Beijing, P.R. China

²Universite Grenoble Alpes, CNRS, CERMAV, 38000 Grenoble, France

³Department of Fibre and Polymer Technology, School of Engineering Sciences in Chemistry, Biotechnology and health, KTH Royal Institute of Technology, 10044 Stockholm, Sweden

⁴Wallenberg Wood Science Center, KTH Royal Institute of Technology, 10044 Stockholm, Sweden

Corresponding authors: panchen@bit.edu.cn, yoshi@cermav.cnrs.fr, jacke@kth.se

Abstract

Dispersion and electrostatic interactions both contribute significantly to the tight assembly of macromolecular chains within crystalline polysaccharides. Using dispersion-corrected density functional theory (DFT) calculation, we estimated the elastic tensor of the four crystalline cellulose allomorphs whose crystal structures that are hitherto available, namely, cellulose I α , I β , II, III_I. Comparison between calculations with and without dispersion correction allows quantification of the exact contribution of dispersion to stiffness at molecular level.

Introduction

Hydrogen bonding, London dispersion, and other electrostatic multipole interactions all play important roles in the spatial organization of molecules in crystalline polysaccharides, such as cellulose, chitin, and chitosan, which are composed of long linear ribbon-like sugar chains where hydroxyl groups are all in equatorial direction. Hydrogen bonds between hydroxyl groups are essentially electrostatic interaction between electronegative oxygen lone-pair electrons and slightly electropositive hydrogen whose electron were pulled by neighbor oxygen and can be considered as interaction between dipoles (Ramos-Cordoba et al. 2011). London dispersion interaction originates from synchronized polarization of atomic nuclei and is always attractive. Atoms in a molecule

have different electronegativity, and thus the sum of all pair-wise electrostatic interactions can be considered as multipole interactions. This is usually considered in modern molecular modeling as interaction between partial charges, which are primarily calculated using DFT method.

Understanding these interactions is of interest in the perspective of the fundamental properties of cellulose materials. One example is the process of dispersing native cellulose fiber aggregates into isolated nanofibrils or into individual polymer chains, that is, dissolution. It has often been stated that it is the “many” and “strong” inter- and intra-molecular hydrogen bonds that hinder defibrillation and dissolution of cellulose (Wang et al. 2016). A parallel can be found in the stability of double stranded DNA, which has often been explained based on the regular hydrogen bonds between opposing nucleobases in a pair. The justification to that is that a typical interaction energy of a base pair is about one order of magnitude higher than dispersion energy from an atom pair (Gooch 2007). However, recent experiments and simulations give more weight to the contribution of dispersion interaction on the stability of double helical structure, to the point that it may even exceed the hydrogen bond interaction (Černý et al. 2008; Kolář et al. 2011; Jahiruddin and Datta 2015; Perumal and Subramanian 2017; Kumar and Patwari 2019; Feng et al. 2019).

For cellulose, the industrially most important crystalline polysaccharide, the interest in how dispersion interaction influences structural stability has been revived, especially in aqueous systems where hydrophobic effects would dominate and hydrogen bonds become insignificant (Bergenstråhle et al. 2010; Medronho et al. 2012; Glasser et al. 2012). Moreover, it has been shown that the dispersion energy contribution to molecular cohesion can be twice the hydrogen bond contribution in the native cellulose crystal (Nishiyama 2018). Therefore, the influence of the different interactions on the structure and properties of cellulose needs to be re-examined.

Another example of where the relative contributions of different molecular interaction has been considered is the intrinsic mechanical properties of cellulose, which has also sometimes been attributed specifically to hydrogen bonds (Eichhorn and Davies 2006). In this case too, a parallel can be made with other biomolecules. Recently, by combining DFT calculations and nanoindentation experiments, both the types and direction of hydrogen bond were found to be correlated to the anisotropic modulus of amino acids (Azuri et al.

2015). It was concluded that the planar hydrogen bonding network co-contributed to the molecular stiffness, resulting in unusually large Young's moduli of amino acid molecular crystals along certain crystal facets. A similar trend for crystalline amino acid hydrogen maleates was reported, where the moduli can differ up to 5-fold between certain facets. (Matveychuk et al. 2018) Furthermore, by comparing DFT calculations with and without dispersion correction, the dispersion interaction was found to induce similar enhancement as hydrogen bonds to the rigidity of diphenylalanine based peptide. (Azuri et al. 2014)

The effect of hydrogen bonds on the elastic moduli of cellulose I β has been studied using molecular mechanics and/or molecular dynamics by switching the explicit hydrogen bond term on and off (Tashiro and Kobayashi 1985; Eichhorn and Davies 2006), in force fields that define such terms. The effect of dispersion interactions has not been studied in isolation, but the contribution from the Lennard-Jones potential, where the dispersion term is one part, was shown to be at least twice that of the contribution from electrostatics (Wohlert et al. 2012; Djahedi et al. 2015), depending on which force field that was used. The elastic tensor has been studied using first principles approach using density functional theory (Dri et al. 2013) and with thermal vibration corrections (Dri et al. 2014), but the contribution of dispersion interaction on mechanical properties has not been calculated. In quantum mechanical calculations, one cannot switch the hydrogen bonding off, but here the dispersion interaction can be simply neglected in the calculation to see its contribution. Hence, we calculated the elastic tensor of all four cellulose allomorphs using DFT to investigate the effect of dispersion interactions on their intrinsic elastic mechanical properties.

Methods

The initial atomic coordinates of crystals are imported from X-ray and neutron studies as well as molecular dynamics simulation and DFT optimization (Langan et al. 1999; Nishiyama et al. 2002, 2003; Wada et al. 2004; Chen et al. 2015). The hydrogen bonding pattern A was used for I α and I β , and the pattern B was used for II and III $_1$. Geometry optimization was performed using the generalized gradient approximation (GGA) functional PBE (Perdew et al. 1996), augmented with pairwise DFT-D2 correction for long-range dispersion (Grimme 2006). Different k-grids are selected according to different unit cells

dimensions: 2 x 3 x 3 for $I\alpha$, 2 x 2 x 2 for $I\beta$ and II , and 4 x 2 x 2 for III . The kinetic energy cutoff was 160 Ry. The convergence threshold of total energy and forces for ionic minimization are 1.0e-6 Ry and 1.0e-5 Ry/bohr respectively. Periodic calculations were carried out using *Quantum Espresso* (version 6.6) (Giannozzi et al. 2009, 2017) in combination with *ElaStic* (Yanchitsky and Timoshevskii 2001; Golesorkhtabar et al. 2013), a universal tool for calculating elastic constants from first principles. All allomorphs except cellulose $I\alpha$ have $P2_1$ symmetry, thus the tensor was determined using the energy-strain method with 13/21 combinations of strain direction to deduce the 13/21 tensor elements (21 elements for triclinic and 13 for monoclinic). For each combination of strain, 11 points were calculated with strain amplitude up to 0.01 and the energy fitted with a parabolic function. To visualize the feature of elastic tensors, the elastic tensors were decomposed into 6 eigenvalues λ_i and 6 pairs of eigentensors of stress and strain equivalents where

$$\sigma_i = \lambda_i \varepsilon_i$$

The stress/strain tensors have 6 independent components and can be visualized in three dimension using the *PAScal* software (Cliffe and Goodwin 2012). The 3D contour surface plots were generated using the online *Anisotropic calculator* from Zuluaga et al. (Zuluaga 2013) and further processed using *Paraview* (Ahrens et al. 2005).

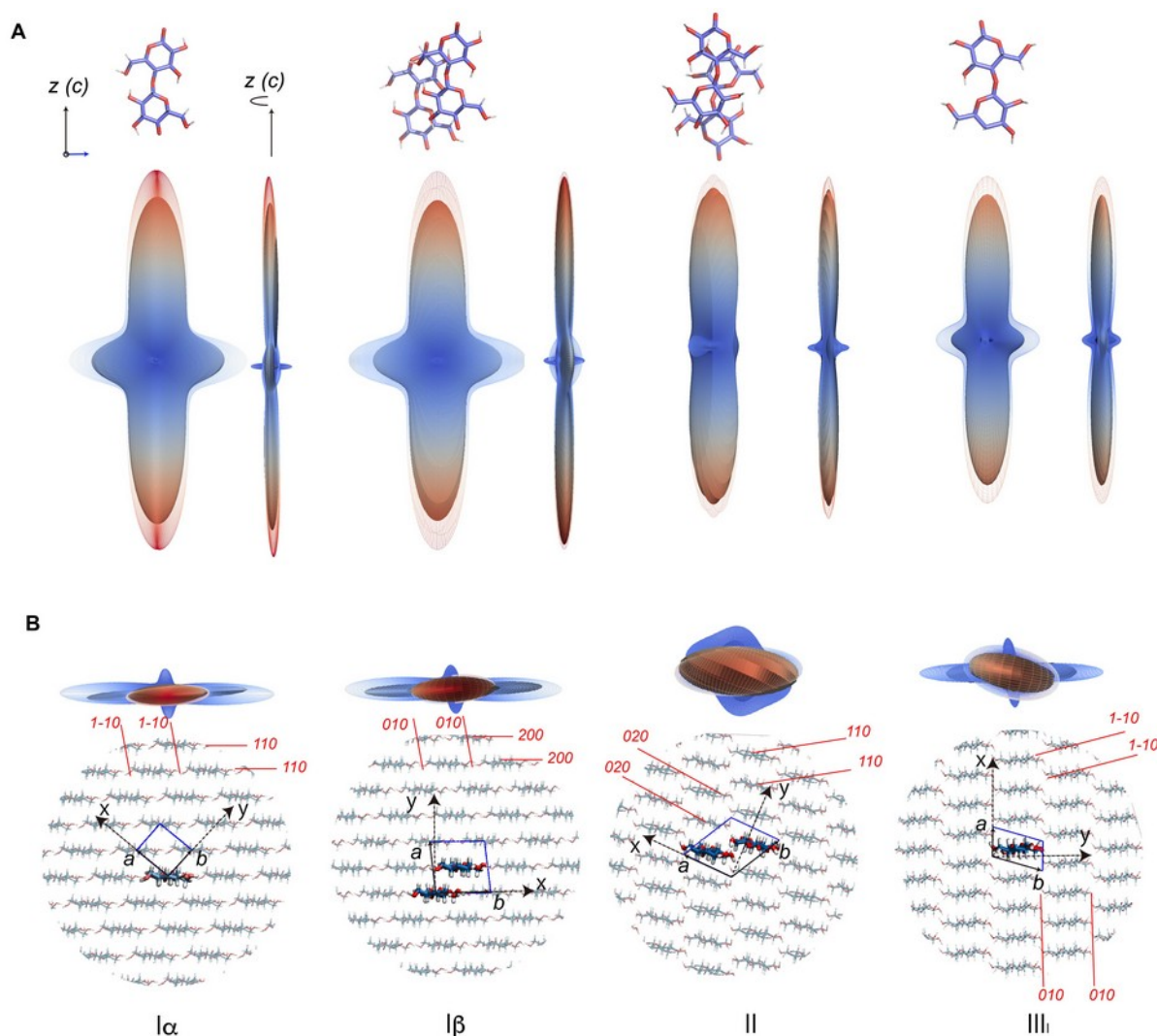
Results and discussion:

Table 1: The unit cell parameters (in Å and degrees), volume (Å³) and *d*-spacing (Å) of crystalline cellulose from experiment (exp) and DFT calculations with dispersion correction (Disp.) or without dispersion correction (No disp.) applied. The difference in percentage is calculated either through (disp. - exp)/exp x 100% or (no disp. - exp)/exp x 100%.

		<i>a</i>	<i>b</i>	<i>C</i>	α	β	γ	<i>Volume</i>	<i>d-spacing</i>		
$I\alpha$									110	010	100
	<i>Exp</i>	10.40	6.717	5.962	80.4	118.1	114.8	333.3	3.908	5.256	6.093
	<i>Disp.</i>	10.40	6.564	5.857	81.7	117.2	114.1	323.9	3.828	5.200	5.984
		0.0%	-2.3%	-1.8%				-2.8%	-2.0%	-1.1%	-1.8%
	<i>No disp.</i>	10.46	6.920	6.920	75.9	115.5	113.2	377.0	4.376	5.666	6.344
		0.6%	3.0%	5.5%				13.1%	12.0%	7.8%	4.1%
$I\beta$									200	110	1-10
	<i>Exp</i> (300K)	7.784	8.201	10.38			96.5	658.3	3.867	5.314	5.959
	<i>Disp.</i>	7.641	8.146	10.40			96.6	636.6	3.762	5.218	5.853
		-1.8%	-0.7%	-0.2%				-2.3%	-1.8%	-1.2%	-1.4%
	<i>No disp.</i>	8.742	8.236	10.47			94.8	750.9	4.356	5.740	6.238
		12.3%	0.4%	0.8%				14.1%	12.6 %	8.0%	4.7%
								020	110	1-10	

II	Exp	8.001	9.030	10.31	117.1	662.9	4.019	4.424	7.203
	Disp.	7.969	8.763	10.38	117.1	644.7	3.899	4.351	7.101
		-0.4%	-3.0%	-0.6%		-2.7%	-3.0%	-1.7%	-1.4%
	No disp.	8.100	9.476	10.44	116.5	718.4	4.246	4.601	7.362
		1.2%	4.9%	1.3%		8.4%	5.6%	4.0%	2.2%
III _I	Exp	4.450	7.850	10.31	105.1	347.7	4.296	7.579	3.379
	Disp.	4.250	7.892	10.39	103.8	338.4	4.126	7.663	3.317
		-4.5%	0.5%	0.8%		-2.7%	-4.0%	-1.1%	-1.8%
	No disp.	4.711	7.868	10.45	104.8	374.3	4.554	7.606	3.530
		5.9%	0.2%	1.3%		7.7%	6.0%	0.4%	4.5%

122



123

124 **Figure 1.** (A): 3D representation of the elastic moduli surface of crystalline cellulose, either
125 with (the grid contour) or without dispersion correction (the solid contour) from two
126 different perspectives. (B): The lateral isosurface in B is multiplied by a factor of two for I α ,
127 I β , and III_I, and by a factor of three for II for better visualization. The snapshots of the four
128 crystalline cellulose allomorphs are labeled with unit cell and deformation vectors. The
129 projections of cross-sections are consistent with the isosurface orientations above. Unit cell
130 parameter *a* is parallel to *x* (except for I β , *b*//*x*), *c* parallel to *z*, and *y* within the *ab* plane.

131

132 The influence of dispersion energy on the predicted unit cell parameters, volume,
133 and *d*-spacing of crystalline cellulose is presented in Table 1. With the dispersion correction
134 applied, DFT generally underestimated the experimental values by up to about 5% but can
135 still be regarded to reproduce the experimental data within a reasonable range. This is
136 acceptable because DFT optimized quantities represent the geometry at 0 K, whereas the
137 experimental measurement refers to that at 300 K. Indeed, crystal structure measurements
138 of I β by neutron diffraction at 15 K showed decreased unit cell dimensions compared to
139 room temperature (Nishiyama et al. 2008), which further lowers the difference between
140 experiment and simulation to less than 1%. In contrast, without dispersion correction, the
141 predicted dimensions were overestimated by up to 10%. In this case, relatively large
142 deviations can be found in the direction normal to the pyranose ring ([1 1 0] of I α , [2 0 0] of
143 I β , [0 2 0] of II, [1 1 0] and [1 -1 0] of III $_1$, as shown in Figure 1B). If also thermal expansion
144 was included, the predicted values would deviate even more from experiment. Thus,
145 dispersion interactions contribute significantly to the tight assembly of cellulose chains in
146 the crystalline state. This is in line with a previous report where a similar strategy was used
147 (Bučko et al. 2011).

148 3D contour plots of the orientation dependent Young's modulus are shown in
149 Figure 1. The full elastic tensor in Voigt representation is given in Table 2 and the
150 corresponding compliance tensor is presented in Table S1. Calculated values for cellulose I β ,
151 especially the diagonal values of stiffness tensor, is consistent with previous DFT results (Dri
152 et al. 2013). The off-diagonal values (C_{ij} , $i \neq j$, Table 2) are also of the same magnitude as in
153 previous work. Cancellation of dispersion interaction resulted in a systematical reduction of
154 the calculated values for all allomorph, varying between -4.0% to -72% for shear moduli
155 (C_{44} , C_{55} , C_{66} in Table 1), and from -10% to -70% for tensile moduli (C_{11} , C_{22} , C_{33} in Table
156 1), respectively. The relative contribution of dispersion interaction on the modulus is
157 anisotropic. It is relatively large (40% to 70 %) in the ring packing direction (010 in I β , 1-10 in
158 I α , 020 in II, and 1-10 in III $_1$, respectively) and small in the longitudinal chain and hydrogen
159 bonding directions. This is expected as the covalent bonds are dominating in this direction.
160 Still, dispersion contributes between to 5 to 17% of the Young's modulus along the chain,

which is similar to the energy decomposition analysis reported (17%) using empirical force field based molecular dynamics simulation (Wohlert et al. 2012) of cellulose I β .

Table 2: The stiffness tensor of crystalline cellulose from DFT calculation. (C11 of I α and C33 of I β , II, and III_I are longitudinal moduli. C44, C55, and C66 are shear moduli. C22 and C33, or C11 and C22, are transverse moduli. The rest are off-diagonal elements.)

Stiffness tensor (GPa)												
Disp. Correction							No disp. Correction					
	C11	C12	C13	C14	C15	C16						
	C12	C22	C23	C24	C25	C26						
	C13	C23	C33	C34	C35	C36						
	C14	C24	C34	C44	C45	C46						
	C15	C25	C35	C45	C55	C56						
	C16	C26	C36	C46	C56	C66						
I α	202.3	7.0	10.3	-1.5	0.8	1.9	167.4	2.2	2.0	-1.2	0.3	0.3
	7.0	47.8	30.7	-22.0	0.3	0.1	2.2	25.8	17.2	-18.1	0.2	-0.8
	10.3	30.7	31.0	-15.8	-1.1	0.6	2.0	17.2	14.7	-12.5	0.1	0.0
	-1.5	-22.0	-15.8	24.3	-0.4	0.0	-1.2	-18.1	-12.5	16.9	-0.1	0.2
	0.8	0.3	-1.1	-0.4	8.3	-7.4	0.3	0.2	0.1	-0.1	7.0	-6.7
	1.9	0.1	0.6	0.0	-7.4	9.4	0.3	-0.8	0.0	0.2	-6.7	9.0
I β	99.2	10.5	11.6	0	0	-0.4	68.2	1.5	3.4	0	0	-0.2
	10.5	17.6	9.2	0	0	-0.1	1.5	5.3	1.0	0	0	0.2
	11.6	9.2	203.6	0	0	-1.5	3.4	1.0	168.3	0	0	0.3
	0	0	0	2.1	-0.9	0	0	0	0	1.3	-0.3	0
	0	0	0	-0.9	15.7	0	0	0	0	-0.3	15.0	0
	-0.4	-0.1	-1.5	0	0	3.6	-0.2	0.2	0.3	0	0	1.0
II	37.4	17.1	3.7	0	0	0.0	41.3	13.9	5.1	0	0	-7.0
	17.1	26.5	5.3	0	0	1.8	13.9	14.4	-1.9	0	0	-2.0
	3.7	5.3	180.3	0	0	2.1	5.1	-1.9	170.9	0	0	2.4
	0	0	0	3.9	-4.0	0	0	0	0	4.8	-3.8	0
	0	0	0	-4.0	10.6	0	0	0	0	-3.8	13.1	0
	0.0	1.8	2.1	0	0	6.0	-7.0	-2.0	-2.4	0	0	8.6
III _I	22.7	20.0	8.5	0	0	-2.2	13.7	10.8	2.8	0	0	0.1
	20.0	61.0	11.9	0	0	6.2	10.8	41.4	6.9	0	0	4.9
	8.5	11.9	173.4	0	0	-1.8	2.8	6.9	152.5	0	0	-0.7
	0	0	0	11.8	-0.4	0	0	0	0	10.7	-0.1	0
	0	0	0	-0.4	3.6	0	0	0	0	-0.1	1.4	0
	-2.2	6.2	-1.8	0	0	3.6	0.1	4.9	-0.7	0	0	3.6

In the transverse directions (C22 and C33 of I α , C11 and C22 of I β , II and III_I) the reduction in stiffness tensor components without dispersion correction was mostly positive,

171 which are 46% and 53% for $I\alpha$, 31% and 70% for $I\beta$, -10% and 46% for II, 40% and 32% for III,
172 respectively.

173 With dispersion correction applied, the C_{11} is 99.2 GPa. '1' in cellulose $I\beta$ is the
174 direction to which both hydrogen bonding interactions and the pyranose rings are almost
175 parallel. This is much higher than the other transverse modulus, which can be ascribed to
176 the assistance of both directional hydrogen bonds and the rigidity of the glucose ring. When
177 no dispersion correction was applied, this elastic modulus drops by 30%, indicating that the
178 dispersion, in this direction, contributes to nearly one third of the total value.

179 In the sugar stacking directions, which is the directions perpendicular to the $[2\ 0\ 0]$
180 direction in $I\beta$, $[1\ 1\ 0]$ in $I\alpha$, $[0\ 2\ 0]$ in II, and $[1\ -1\ 0]$ in III, (Figure 1), the response to
181 switching off the dispersion correction was similar, which was found to contribute 40 to
182 70% of the rigidity.

183 The dispersion correction also has significant influence on the off-diagonal elements.
184 The values of elastic tensor elements depend on the reference frame, and the standard
185 convention is to take the direction 1 along the a -axis, and direction 2 in the ab -plane (as
186 shown in Figure 2). However, the unit cell itself is not necessarily representing the principal
187 axes of physical properties. To compare the different elastic tensors in a more universal
188 way, we decomposed the elastic tensor into eigentensors that are stress/strain tensor pairs
189 that are directly related by a scalar value, the corresponding eigenvalue, which is the
190 measure of stiffness.

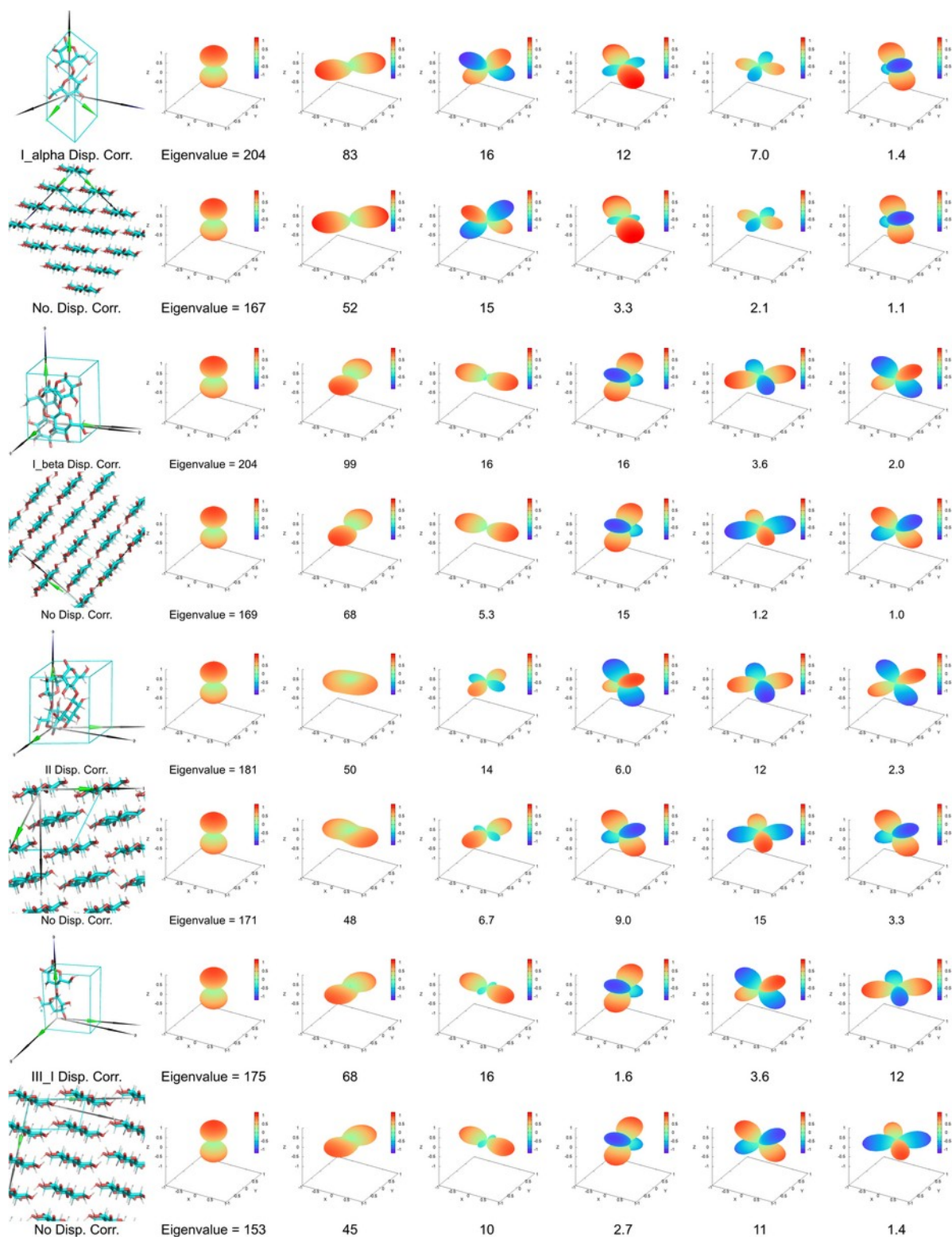


Figure 2. The 3D display of eigenvalue tensor of four cellulose allomorph with or without dispersion correction for I_{α} , I_{β} , II, and III_I, respectively.

The stress/strain tensors have 6 independent elements corresponding to axial (normal) and shear stress/strain. They can be further reduced to three eigenvectors that translates to a pure axial stress/strain. If the corresponding eigenvalues of all eigenvectors have the same sign, they can be represented by ellipsoids, similar to thermal factor in crystallographic representation. However, the stress/strain tensor can imply axial stress/strain of opposite sign. Thus, we represent the tensor using color coding for positive and negative values, with a surface located at a distance proportional to the axial stress/strain in each direction. In general, when the mode is dominated by axial stress/strain, the color is uniform, whereas shear mode shows opposite colors in orthogonal directions.

The estimated Poisson's ratio is shown in Table S2. In comparison to experimental measurements (Nakamura et al. 2004), I β showed slightly larger calculated ν_{31} (0.49 vs. an experimental value of 0.38), and II showed a smaller ν_{32} (0.16 vs. 0.30).

To compare the elastic tensors of different allomorphs and in different conditions (with and without dispersion correction), eigentensors were sorted in descending order of the eigenvalue (stiffest first) of the stiffness tensor calculated with dispersion correction (first row of each allomorph in Figure 2). To find the corresponding eigentensor from the calculations without dispersion correction (shown on the second row), the closest eigentensor was chosen, based on a distance measure. This distance was calculated as an inner product, which is 1 when they match perfectly, and 0 when orthogonal. The corresponding structure is drawn in the same reference frame.

Figure 2 shows that in all cases the highest eigenvalue is dominated by axial deformation along the chain direction, which was already seen from the diagonal elements of the stiffness tensor.

For cellulose I β the eigentensors following the first one have a straightforward interpretation. The second eigentensor is axial deformation almost parallel to the y-axis. This is the direction of inter-molecular hydrogen bonding, parallel to the pyranose plane. The third one is orthogonal to the two, followed by three shear modes: yz, xy, and xz.

The fact that, considering lateral directions only, the structure is stiffest in the direction of inter-molecular hydrogen bonding is often discussed as manifestation of the effect of hydrogen bonding. However, it should be remembered that the density of covalent bonds is

higher in this direction (along unit cell b of $I\beta$ in Figure 1B), while the non-covalent interactions appears only every 8 Å ($d_{010} = 0.82$ nm), compared to every 4 Å ($d_{200} = 0.39$ nm) in the perpendicular direction. Since a typical stable non-bonded separation is of the order of 3 Å, properties perpendicular to the pyranose ring plane are dominated by non-bonded interactions. In the hydrogen bonded plane ($[2\ 0\ 0]$ of $I\beta$), atoms are mostly linked through covalent bonds, contributing to the high rigidity.

For cellulose II and III₁, still only considering lateral directions, the tensile deformation in the pyranose plane is highly coupled to its perpendicular direction and, in the case of cellulose II the second eigentensor almost appears as elongated doughnuts when dispersion correction is on. This tendency can be also verified in the off-diagonal element value that are similar to the diagonal element in x and y directions. The axial deformation in the direction perpendicular to the pyranose plane is coupled to xy shear (3rd tensor). At first sight, one would tend to interpret this feature as due to a zig-zag hydrogen bond pattern leading to a honeycomb-like response, but the fact that the dispersion correction enhances this tendency suggests that the mechanism is more complex and requires further clarification.

The eigentensors of cellulose $I\alpha$ is strikingly different from $I\beta$, except for the stiffest mode. The deformation of a material is dominated by soft components, and thus cellulose $I\alpha$ and $I\beta$ would behave quite differently in complex stress environment in a dense material.

Conclusion:

We have systematically calculated unit cell parameters and elastic moduli of four crystalline cellulose allomorphs using DFT. The influence of the dispersion energy on both crystal structures and elastic constants was investigated by switching the dispersion correction on and off. Our calculations reveal that the dispersion interactions dominate the stacking of cellulose chains, in all crystal forms, especially in the direction perpendicular to the pyranose rings ($[2\ 0\ 0]$ in $I\beta$, $[1\ 1\ 0]$ in $I\alpha$, $[0\ 2\ 0]$ in II, and $[1\ -1\ 0]$ in III₁, respectively). In this direction, dispersion energy contributes more than 50% to the elastic mechanical properties. It further contributes about one third in the hydrogen bonding direction ($[2\ 0\ 0]$ in $I\beta$, $[1\ 1\ 0]$ in $I\alpha$, $[0\ 2\ 0, 1\ 1\ 0]$ in II, and $[1\ -1\ 0, 1\ 1\ 0]$ in III₁, respectively), and 5 to 17% in the chain direction. These findings emphasize that dispersion energy contributes

significantly to the mechanical properties of cellulose, and they also indicate that the contribution from dispersion energy exceeds that of hydrogen bonding during the initial step of defibrillation or dissolution of cellulose.

Acknowledgements:

P.C. thanks the Beijing Natural Science Foundation (2204096) and Beijing Institute of Technology Research Fund for Young Scholars. Partial computational resources were provided by the Swedish National Infrastructure for Computing (SNIC) at the PDC Center for High Performance Computing, KTH Royal Institute of Technology, partially funded by the Swedish Research Council through grant agreement no. 2018-05973.

Declaration:

The authors declare that they have no conflicts of interest. No animal or human trials were undertaken or conducted for this study. All the three authors are aware of the submission and have given their consent to participate. All the authors have read and approved the manuscript before submission.

References:

Ahrens J, Geveci B, Law C (2005) ParaView: An End-User Tool for Large Data Visualization, Visualization Handbook. Elsevier

Azuri I, Adler-Abramovich L, Gazit E, et al (2014) Why Are Diphenylalanine-Based Peptide Nanostructures so Rigid? Insights from First Principles Calculations. *J Am Chem Soc* 136:963–969. <https://doi.org/10.1021/ja408713x>

Azuri I, Meirzadeh E, Ehre D, et al (2015) Unusually Large Young's Moduli of Amino Acid Molecular Crystals. *Angew Chem Int Ed* 54:13566–13570. <https://doi.org/10.1002/anie.201505813>

Bergenstråhle M, Wohler J, Himmel ME, Brady JW (2010) Simulation studies of the insolubility of cellulose. *Carbohydr Res* 345:2060–2066. <https://doi.org/10.1016/j.carres.2010.06.017>

Bučko T, Tunega D, Ángyán JG, Hafner J (2011) Ab Initio Study of Structure and Interconversion of Native Cellulose Phases. *J Phys Chem A* 115:10097–10105. <https://doi.org/10.1021/jp205827y>

292 Černý J, Kabeláč M, Hobza P (2008) Double-Helical → Ladder Structural Transition in the B-
 293 DNA is Induced by a Loss of Dispersion Energy. *J Am Chem Soc* 130:16055–16059.
 294 <https://doi.org/10.1021/ja805428q>

295 Chen P, Ogawa Y, Nishiyama Y, et al (2015) Alternative hydrogen bond models of cellulose II
 296 and III based on molecular force-fields and density functional theory. *Cellulose*
 297 22:1485–1493. <https://doi.org/10.1007/s10570-015-0589-z>

298 Cliffe MJ, Goodwin AL (2012) PASCAL: a principal axis strain calculator for thermal expansion
 299 and compressibility determination. *J Appl Crystallogr* 45:1321–1329.
 300 <https://doi.org/10.1107/S0021889812043026>

301 Djahedi C, Berglund LA, Wohler J (2015) Molecular deformation mechanisms in cellulose
 302 allomorphs and the role of hydrogen bonds. *Carbohydr Polym* 130:175–182.
 303 <https://doi.org/10.1016/j.carbpol.2015.04.073>

304 Dri FL, Hector LG, Moon RJ, Zavattieri PD (2013) Anisotropy of the elastic properties of
 305 crystalline cellulose I β from first principles density functional theory with Van der
 306 Waals interactions. *Cellulose* 20:2703–2718. [https://doi.org/10.1007/s10570-013-](https://doi.org/10.1007/s10570-013-0071-8)
 307 0071-8

308 Dri FL, Shang S, Hector LG, et al (2014) Anisotropy and temperature dependence of
 309 structural, thermodynamic, and elastic properties of crystalline cellulose I β :
 310 a first-principles investigation. *Model Simul Mater Sci Eng* 22:085012.
 311 <https://doi.org/10.1088/0965-0393/22/8/085012>

312 Eichhorn SJ, Davies GR (2006) Modelling the crystalline deformation of native and
 313 regenerated cellulose. *Cellulose* 13:291–307. [https://doi.org/10.1007/s10570-006-](https://doi.org/10.1007/s10570-006-9046-3)
 314 9046-3

315 Feng B, Sosa RP, Mårtensson AKF, et al (2019) Hydrophobic catalysis and a potential
 316 biological role of DNA unstacking induced by environment effects. *Proc Natl Acad Sci*
 317 116:17169–17174. <https://doi.org/10.1073/pnas.1909122116>

318 Giannozzi P, Andreussi O, Brumme T, et al (2017) Advanced capabilities for materials
 319 modelling with Quantum ESPRESSO. *J Phys Condens Matter* 29:465901.
 320 <https://doi.org/10.1088/1361-648X/aa8f79>

321 Giannozzi P, Baroni S, Bonini N, et al (2009) QUANTUM ESPRESSO: a modular and open-
 322 source software project for quantum simulations of materials. *J Phys Condens*
 323 *Matter* 21:395502. <https://doi.org/10.1088/0953-8984/21/39/395502>

324 Glasser WG, Atalla RH, Blackwell J, et al (2012) About the structure of cellulose: debating the
 325 Lindman hypothesis. *Cellulose* 19:589–598. [https://doi.org/10.1007/s10570-012-](https://doi.org/10.1007/s10570-012-9691-7)
 326 9691-7

327 Golezorkhtabar R, Pavone P, Spitaler J, et al (2013) ElastiC: A tool for calculating second-
 328 order elastic constants from first principles. *Comput Phys Commun* 184:1861–1873.
 329 <https://doi.org/10.1016/j.cpc.2013.03.010>

330 Gooch JW (ed) (2007) London dispersion forces (London forces). In: Encyclopedic Dictionary
331 of Polymers. Springer, New York, NY, pp 582–582

332 Grimme S (2006) Semiempirical GGA-type density functional constructed with a long-range
333 dispersion correction. *J Comput Chem* 27:1787–1799.
334 <https://doi.org/10.1002/jcc.20495>

335 Jahiruddin S, Datta A (2015) What Sustains the Unnatural Base Pairs (UBPs) with No
336 Hydrogen Bonds. *J Phys Chem B* 119:5839–5845.
337 <https://doi.org/10.1021/acs.jpcb.5b03293>

338 Kolář M, Kubař T, Hobza P (2011) On the Role of London Dispersion Forces in Biomolecular
339 Structure Determination. *J Phys Chem B* 115:8038–8046.
340 <https://doi.org/10.1021/jp202878d>

341 Kumar A, Patwari GN (2019) Probing the role of dispersion energy on structural
342 transformation of double-stranded xylo- and ribo-nucleic acids. *Phys Chem Chem*
343 *Phys* 21:3842–3848. <https://doi.org/10.1039/C8CP06305B>

344 Langan P, Nishiyama Y, Chanzy H (1999) A Revised Structure and Hydrogen-Bonding System
345 in Cellulose II from a Neutron Fiber Diffraction Analysis. *J Am Chem Soc* 121:9940–
346 9946. <https://doi.org/10.1021/ja9916254>

347 Matveychuk YV, Bartashevich EV, Tsirelson VG (2018) How the H-Bond Layout Determines
348 Mechanical Properties of Crystalline Amino Acid Hydrogen Maleates. *Cryst Growth*
349 *Des* 18:3366–3375. <https://doi.org/10.1021/acs.cgd.8b00067>

350 Medronho B, Romano A, Miguel MG, et al (2012) Rationalizing cellulose (in)solubility:
351 reviewing basic physicochemical aspects and role of hydrophobic interactions.
352 *Cellulose* 19:581–587. <https://doi.org/10.1007/s10570-011-9644-6>

353 Nakamura K, Wada M, Kuga S, Okano T (2004) Poisson's ratio of cellulose I β and cellulose II.
354 *J Polym Sci Part B Polym Phys* 42:1206–1211

355 Nishiyama Y (2018) Molecular interactions in nanocellulose assembly. *Philos Trans R Soc*
356 *Math Phys Eng Sci* 376:20170047. <https://doi.org/10.1098/rsta.2017.0047>

357 Nishiyama Y, Johnson GP, French AD, et al (2008) Neutron Crystallography, Molecular
358 Dynamics, and Quantum Mechanics Studies of the Nature of Hydrogen Bonding in
359 Cellulose I β . *Biomacromolecules* 9:3133–3140. <https://doi.org/10.1021/bm800726v>

360 Nishiyama Y, Langan P, Chanzy H (2002) Crystal Structure and Hydrogen-Bonding System in
361 Cellulose I β from Synchrotron X-ray and Neutron Fiber Diffraction. *J Am Chem Soc*
362 124:9074–9082. <https://doi.org/10.1021/ja0257319>

363 Nishiyama Y, Sugiyama J, Chanzy H, Langan P (2003) Crystal Structure and Hydrogen
364 Bonding System in Cellulose I α from Synchrotron X-ray and Neutron Fiber
365 Diffraction. *J Am Chem Soc* 125:14300–14306. <https://doi.org/10.1021/ja037055w>

366 Perdew JP, Burke K, Ernzerhof M (1996) Generalized Gradient Approximation Made Simple.
 367 Phys Rev Lett 77:3865–3868. <https://doi.org/10.1103/PhysRevLett.77.3865>

368 Perumal SSRR, Subramanian Y (2017) A molecular dynamics calculation of solid phase of
 369 malonic acid: role of hydrogen-bond chains and the elastic constants. J Chem Sci
 370 129:963–974. <https://doi.org/10.1007/s12039-017-1310-6>

371 Ramos-Cordoba E, Lambrecht DS, Head-Gordon M (2011) Charge-transfer and the hydrogen
 372 bond: Spectroscopic and structural implications from electronic structure
 373 calculations. Faraday Discuss 150:345–362. <https://doi.org/10.1039/C1FD00004G>

374 Tashiro K, Kobayashi M (1985) Calculation of crystallite modulus of native cellulose. Polym
 375 Bull 14:213–218. <https://doi.org/10.1007/BF00254940>

376 Wada M, Chanzy H, Nishiyama Y, Langan P (2004) Cellulose III Crystal Structure and
 377 Hydrogen Bonding by Synchrotron X-ray and Neutron Fiber Diffraction.
 378 Macromolecules 37:8548–8555. <https://doi.org/10.1021/ma0485585>

379 Wang S, Lu A, Zhang L (2016) Recent advances in regenerated cellulose materials. Prog
 380 Polym Sci 53:169–206. <https://doi.org/10.1016/j.progpolymsci.2015.07.003>

381 Wohler J, Bergenstr hle-Wohler M, Berglund LA (2012) Deformation of cellulose
 382 nanocrystals: entropy, internal energy and temperature dependence. Cellulose
 383 19:1821–1836. <https://doi.org/10.1007/s10570-012-9774-5>

384 Yanchitsky BZ, Timoshevskii AN (2001) Determination of the space group and unit cell for a
 385 periodic solid☆☆This program can be downloaded from the CPC Program Library
 386 under catalogue identifier: <http://cpc.cs.qub.ac.uk/summaries/ADON>. Comput Phys
 387 Commun 139:235–242. [https://doi.org/10.1016/S0010-4655\(01\)00212-0](https://doi.org/10.1016/S0010-4655(01)00212-0)

388 Zuluaga M (2013) Anisotropy Calculator - 3D Visualization Toolkit.
 389 <https://doi.org/10.4231/D3JD4PQ2D>

390