

UV-induced formation of DNA damage in cells and their mutational consequences

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Thierry Douki, S Adar. UV-induced formation of DNA damage in cells and their mutational consequences. Roberto Improta; Thierry Douki. DNA Photodamage: From Light Absorption to Cellular Responses and Skin Cancer, Royal Society of Chemistry, 2021, Comprehensive Series in Photochemical & Photobiological Sciences, 978-1-83916-196-4. 10.1039/9781839165580-00133 . hal-03689535

HAL Id: hal-03689535 https://hal.science/hal-03689535v1

Submitted on 7 Jun2022

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1	Chapter 7: UV-induced formation of DNA damage in cells and their
2	mutational consequences.
3	
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15 <u>ABSTRACT</u>

16 Ultraviolet radiation induces a wide variety of damage in DNA that can originate from 17 either photochemical reaction triggered by absorption of photons by DNA bases or 18 oxidation processes mostly induced by photosensitization. The nature of DNA damage 19 is dependent of numerous factors such the wavelength, the sequence, the chromatin 20 structure or the cell type. The present chapter provides a summary of the available 21 quantitative information of the yield of UV-induced DNA damage as well as the most 22 recent pieces of information bases on state-of-the-art techniques such as next 23 generation sequencing. The mutational consequences of the DNA photoproducts are 24 briefly discussed.



26 27	Table of Contents	Winduced formation of DNA damage in cells and their mutational
28	consequence	es
29	7.1 Introdu	ction 5
30 31	7.2 UV-ind 7.2.1 Wa	uced pyrimidine dimers in cellular DNA
32	7.2.1.1	CPDs and UVB radiation7
33	7.2.1.2	Distribution of pyrimidine dimers in UVB irradiated cells7
34	7.2.2 U∖	/A-induced CPDs9
35	7.2.2.1	Direct formation of CPD upon exposure to UVA9
36	7.2.2.2	Dark CPDs 11
37	7.2.3 U\	C-induced DNA damage 12
38	7.2.3.1	Interest of UVC in photobiology 12
39	7.2.3.2	DNA damage in the UVC range12
40	7.2.3.3	Photochemistry in bacterial spores 14
41 42	7.3 Oxidati 7.3.1 UV	vely generated DNA damage in cells
43	7.3.2 Ma	ain UVA-induced photooxidation pathways
44	7.3.2.1	One-electron oxidation reactions 16
45	7.2.3.2	Singlet oxygen 16
46	7.2.3.3	Reactive oxygen species arising from the superoxide anion radical 17
47	7.3.3. Ox	idatively generated DNA damage in UVA-irradiated cells
48 49	7.4 Spectra 7.4.1 Fo	al composition and photoproduct distribution
50	7.4.2. U∖	/-induced formation of sensitizers
51	7.4.3. I	mpact of UVA on the repair of UVB-induced dimers



52 53 54	7.5 DN/ 7.6 UV- 7.6.1	A damage at the nucleotide resolution in cellular DNA induced DNA damage in skin Yield of lesions in skin	22 26 27
55	7.6.2	Effect of phototype on the formation of CPDs	28
56 57	7.7 Mut 7.7.1.	agenic consequences of DNA photodamage In vitro data of UV-induced mutagenesis	29 29
58	7.7.2.	Next generation sequencing of human tumours	31
59 60 61 62	Abbreviati Reference	ons : s	33 33



64 **7.1 Introduction**

Information gathered over the years in model systems and isolated DNA made possible 65 66 the identification the most relevant types of DNA damage that could be generated in 67 cells upon exposure to UV radiation. They are either dimeric photoproducts involving 68 adjacent pyrimidine bases or oxidatively generated lesions such as strand breaks and 69 oxidized bases (Fig. 7.1). In the early 1960's, radioactivity-based assays were 70 developed to quantify thymine-containing cyclobutane pyrimidine dimers (CPDs). The 71 subsequent development of immunoassays made possible the collection of large 72 amounts of information in various cellular systems and in skin not only on CPDs but 73 also on pyrimidine (6-4) photoproducts (64-PPs) and their Dewar valence isomers 74 (Dewars)^{1,2}. More specific methods based on molecular biology approaches³ or mass spectrometry detection ⁴ provided data on the individual formation of each dimer at the 75 76 different bipyrimidine sites. In the recent years, the breakthrough made in next 77 generation sequencing provided an unprecedented picture on the formation and repair of DNA in the different region of chromatin ^{5, 6}. Information on oxidatively generated 78 79 lesions was mostly obtained through electrophoretic-based assay such as the comet 80 assay ⁷ or alkaline elution ⁸. These methods were primarily developed for the 81 quantification of strand breaks but could be extended to damaged bases by the use of 82 purified DNA N-glycosylases that recognize well-defined classes of damage. Some 83 specific lesions like 8-oxo-7,8-dihydroguanine (8-oxoGua) were also quantified by 84 chromatographic approaches. The purpose of this chapter is to combine all these 85 pieces of information, with emphasis placed on the most recent results but without 86 neglecting relevant earlier data, in order to provide insight into the distribution of the 87 most frequent UV-induced damage. Factors such as the cell type, the spectrum of the 88 incident light and the specificity of the formation of damage in skin are also discussed.



- 89 Last, a short overview of link between DNA damage and mutagenesis is provided,
- 90 especially in the light of recent next generation sequencing data.



91

Figure 7.1: Chemical structure of the main DNA damage induced by UV radiation.
UVB mostly leads to the formation of pyrimidine dimers while UVA is at the origin of a
strong oxidative stress. However, as discussed in the present chapter, the boundary
between the two wavelengths ranges is less tight that long believed.

96 7.2 UV-induced pyrimidine dimers in cellular DNA

97 UV radiation is an efficient DNA damaging agent but the yield and the nature of the 98 photodamage strongly depend on the wavelength of the incident photons. Major 99 attention has been paid to the UVA (320-400 nm) and UVB ranges (280-320 nm), 100 which represent more than 95% and less than 5% of solar UV, respectively. Information 101 is also available on the DNA damaging effects of UVC light (100-280 nm), which is 102 closer to the UV absorption maximum of DNA at 260 nm.



103 7.2.1 Wavelength-dependent formation of pyrimidine dimers by UVB

104

7.2.1.1 CPDs and UVB radiation

105 The most efficient portion of the solar spectrum for the induction of pyrimidine dimers 106 is UVB. This is easily explained by the fact that the wavelength of the most energetic 107 photons in solar UV is closer to the maximal absorption of DNA (260 nm). 108 Consequently, DNA base may absorb these photons and reach excited states, thereby 109 opening the way to photochemical reaction as addressed by Martinez-Fernandeza and 110 Improta in chapter 2 of this volume. Numerous data are available for CPDs. In cultured cells, the reported yields for broadband UVB lamps varies from 0.01 to 0.06 CPD/10⁵ 111 112 bases per J/m² ⁹⁻¹². The differences between these values can be explained by both 113 the use of different analytical tools, the cell type and the emission spectrum of the UVB 114 sources. The latter point is a major issue because the yield of the pyrimidine dimers 115 varies along the UVB range. Action spectra in cultured human fibroblasts ¹³ show that 116 the yield of CPD is in a 1 / 0.4 / 0.06 ratio at 280, 290 and 300 nm, respectively. In 117 CHO cells, ratio of 1 / 0.22 / 0.01 were determined for the yield of CPD at 290, 300 and 118 310 nm⁸. These values show that the slightest increase in wavelength leads to a 119 significant decrease in the yield of CPDs. They explain the difference in DNA damaging 120 properties of sunlight depending on the hour the day, the latitude and the longitude. 121 When the intensity on the UV radiation decreases as the result of the filter effect of the atmosphere, the ratio between the intensities of UVB and UVA is reduced ^{14, 15}. 122 123 Differences in emission spectra is also of outmost importance when comparing results 124 obtained by different teams using different irradiator.

125

7.2.1.2 Distribution of pyrimidine dimers in UVB irradiated cells



Most of the data discussed above are related to CPDs measured as a whole by 126 127 immunoassays or to the sole TT CPD quantified by specific techniques. Yet, work on model systems and isolated DNA has shown that CPDs are formed at the four 128 bipyrimidine sequences ^{4, 16}. These works also showed that, because of steric 129 130 constraints, only the *cis,syn* diastereoisomer of CPD is detected in double-stranded 131 DNA while other derivatives are produced in dinucleoside monophosphates and single-132 stranded DNA. Similar observations were made in cells with the following decreasing 133 order of frequency for the *cis,syn* CPD: TT > TC > CT > CC ^{10, 17-25}. The ratio between 134 the yields of these CPDs is 10 / 5 / 3 / 1, respectively. 64-PPs are also produced in significant amount by UVB radiation, although in a 3 to 5 lower yield than CPDs 9, 10, 12, 135 136 ²⁶. In UVB-irradiated cultured cells, they are formed in detectable amounts only at TT and TC sites in a 1 to 5 ratio 9, 10, 21, 22, 24, 27. CT and CC 64-PPs ²¹ have been detected 137 138 in small amounts only upon UVC-irradiation of isolated DNA. The ratio between CPD 139 and 64-PP is also dependent on the involved pyrimidine bases, with an approximate 140 ratio of 10 at TT and 1 at TC sites. The ratio between all these different bipyrimidine 141 photoproducts is constant from one mammalian cell type to the other. This reflects that 142 the photochemistry of DNA is hardly modulated by the cellular environment. As detailed 143 below, recent works based on the detection of pyrimidine dimers at the nucleotide 144 resolution by next generation sequencing led to similar results ^{6, 28-31}.

The only parameter that was found to drastically affect the mean distribution of CPDs in cells is the GC content of DNA ³². This mostly concerns bacteria since all eukaryotic cells exhibit similar proportions of G/C and A/T bases pairs. Interestingly, this study of the impact of the GC content of DNA on the formation of pyrimidine dimers led to the conclusion that TC rather than TT dinucleotides are the most reactive sites in DNA. On a local scale, the reactivity of bipyrimidine sites can be modulated by the methylation



of cytosine, an important epigenetic event. Evidence are accumulating for an increased
formation of damage at methylated CpG island ³³, which is associated with increased
mutagenesis ^{17, 23, 34, 35}.

154 It may be mentioned than UVB may also trigger a secondary photoreaction within 64-PPs: the conversion into a Dewar valence isomer of the pyrimidone ring ³⁶⁻³⁹. This has 155 156 been well documented in model systems but is observed in very low yield in cells 157 exposed to biologically relevant doses of UVB. As discussed later in this chapter, formation of Dewar in cells requires both UVB and UVA. Another important secondary 158 159 reaction is the hydrolytic deamination of cytosine moieties in CPDs and CT and CC 160 64-PPs and Dewars⁴⁰⁻⁴². This process leads to the formation of uracil-containing CPDs 161 that play a role in mutagenesis, which is dominated by $C \rightarrow T$ mutations at bipyrimidic 162 sites as discussed later in this chapter. More information on this process can be found 163 in chapter 8 by Taylor and colleagues in this book.

164 **7.2.**

7.2.2 UVA-induced CPDs

165 **7.2.2.1 Direct formation of CPD upon exposure to UVA**

166 Formation of CPDs is not only induced by UVB but also by the less energetic UVA. 167 This photoreaction has long been neglected because UVA radiation was thought not 168 to be absorbed by DNA. Yet, precise spectroscopic studies showed that DNA was actually a chromophore for UVA with an absorption at 350 nm respectively 10⁴ and 10³ 169 170 times lower than at 280 and 300 nm ⁴³. Formation of CPDs in cellular DNA upon 171 exposure to UVA has first been reported in bacteria ⁴⁴. Since then, similar results were obtained in cultured rodent cells^{8, 12, 45, 46} as well as in human fibroblasts⁴⁷, 172 173 keratinocytes ^{22, 48} and melanocytes ⁴⁹. The same observation was made in whole human skin both *in vivo* ⁵⁰ and *ex vivo* ^{22, 24, 51}. The ratio between the yield of CPDs 174



175 after exposure to UVB or UVA is in the same range than the difference in absorption 176 (Table 7.1). This value differs from one work to the other, which can be explained by 177 differences in emission spectra of the UV sources. The importance of this parameter 178 is clearly illustrated in action spectra where values of 60 were found between the ratio 179 at 365 nm compared to either 290 nm or 310 nm in CHO cells⁸. Similarly, a ratio of 6 180 was determined for the yield of CPD in keratinocytes exposed at 302 nm compared to 181 either 365 or 334 nm ⁴⁷. The correlation between the yield of CPDs and the DNA 182 absorption, combined with the observation of CPD in UVA-irradiated isolated DNA, 183 strongly suggests that a direct photochemical rather than a photosensitized process is involved in the UVA-induction of CPD ^{12, 52-54}. A last worth mentioning feature of the 184 UVA photochemistry of DNA is that 64-PPs are at the best minor photoproducts. This 185 186 can be explained by the change in the nature of the excited states generated when the 187 wavelength of the absorbed photons increases from UVB to UVA 55.

Table 7.1: Comparison of the induction of CPDs in various cell types exposed to UVB
and UVA radiations. The reported results are the ratio between the yields of
formation.

Cell type	UV sources*	ratio	ref
CHO cells	290 vs 365 nm 310 vs 365 nm	40000 700	8
CHO cells	bb-UVB vs bb- UVA	700	12
CHO cells	bb-UVB vs bb- UVA	2000	46
Human skin fibroblasts	302 vs 334 nm 302 vs 365 nm	150 1000	47
Human keratinocytes	bb-UVB vs bb- UVA	20000	9
Human skin explants	bb-UVB vs bb- UVA	6500	22
Human skin explants phototype IV	bb-UVB vs bb-	1100	24
Human skin explants phototype II	UVA	1050	

191 *bb: broadband



192 **7.2.2.2**

7.2.2.2 Dark CPDs

193 In addition to the formation of CPDs resulting from the weak but real absorption of UVA 194 photons by DNA, UVA was reported to trigger a delayed formation of CPDs. This phenomenon was first reported in cultured melanocytes ⁵⁶ and further extended to 195 196 keratinocytes ⁵⁷. In both cell types, an oxidative pathway seems to be involved as 197 shown by the inhibiting effects of antioxydants. Mechanistic studies have been 198 performed in melanocytes. In this cell type UVA, and to a lesser extent UVB, were 199 found to induce the formation of dark CPDs in the hours following irradiation of 200 pigmented melanocytes but not of melanocytes originating from albino mice ⁵⁶. 201 Interestingly, not only TT but also the mutagenic TC and CT CPDs were formed by this 202 delayed mechanism. Evidence for a role of oxidation products of melanin and melanin 203 precursor was provided by the replication of this effect on model systems involving 204 oxidizing species, melanin derivatives and isolated DNA. A proposed mechanism is 205 thus the formation of endoperoxide by UVA-induced oxidative stress that would then 206 decompose into excited ketones. Such a mechanism has been previously described 207 for oxidatively generated lesions ⁵⁸. It could induce the formation of CPDs by triplet 208 energy transfer if the energy of the excited ketone is large enough. This pathway 209 remains yet to be completely established in order to explain why C-containing dark 210 CPDs are produced in significant amounts while photosensitized triplet-triplet energy 211 transfer leads to the overwhelming formation of TT-CPD ⁵⁹⁻⁶¹. The biological relevance 212 of this process was assessed by the observation of dark CPDs in human skin in vivo ⁶². Their formation was observed even at the highest UVA wavelengths but not in the 213 214 visible range ⁶³. An interesting phenomenon is the fast decrease in the peak of CPDs 215 following exposure ^{56, 57, 62}. This could be explained by a faster repair of dark than 216 immediate CPDs. Formation of dark CPDs requires molecular contact between



217	endoperoxides and bases. It is therefore favoured in open regions of chromatin where
218	DNA repair is the most efficient. Further work is necessary to confirm this hypothesis
219	that would help quantifying the biological role of dark CPDs.

220

221

7.2.3 UVC-induced DNA damage

7.2.3.1 Interest of UVC in photobiology

222 UVC is absent from the sunlight reaching the surface of Earth and of limited interest in 223 terms of health effects. However, the recent COVID pandemic has led to a regain of 224 interest in the UVC-mediated disinfection technologies ⁶⁴. This strategy has been used 225 for many years, with a first germicidal action spectra published as early as 1946⁶⁵ and 226 confirmed since then by countless works. The maximal efficacy of UVC radiation is 227 observed at 260-265 nm. In practical applications, equipment use the easily available 228 254 nm-emitting high-pressure mercury lamp. Its efficacy at this wavelength is 229 approximately 90% of that at 260 nm. Because 254 nm radiation exhibits severe risks 230 for skin and eyes ⁶⁶, very recent alternatives have been proposed based on higher 231 energy UVC, mostly at 222 nm. Because they exhibit a lower penetration in tissues, 232 these photons are expected to be less damaging than 254 nm. It remains though to 233 determine whether this represents an industrial advantage since all sterilization 234 devices are protected. One novelty would be to use 222 nm light to directly sterilize 235 areas of the human body but much deeper investigation on the harmlessness of these 236 technologies would be necessary. More local application such as surgery site are yet interesting ⁶⁷. 237

238

7.2.3.2 DNA damage in the UVC range

Exposure to 254 nm radiation exhibits its germicidal properties because it induces largeamounts of damage in the genome of bacteria, yeasts and viruses. This is clearly



241 documented by early works on the formation of pyrimidine dimers in DNA ^{16, 68}. This 242 was mostly motivated by the availability of mercury lamps emitting photons with a 243 wavelength close to the maximal absorption of DNA. Today, UVC is mostly used in 244 biological studies as a convenient tool for the fast production of significant amounts of 245 DNA damage. Because it is more efficiently absorbed than UVB, UVC leads to a larger 246 vield of photodamage. The ratio in UVC-induced vield with respect to UVB is around 247 10 in bacteria ¹¹. This trend is different in human skin because UVC is strongly 248 absorbed by the stratum corneum and the upper layer of the cutaneous tissue. The 249 yield of CPDs was found to be similar at 260 and 300 nm in the epidermis and almost 250 1 order of magnitude lower in the dermis ⁶⁹. Interestingly, the same decrease in the 251 action spectrum in the UVC range compared to UVB is observed for skin cancer ⁷⁰. 252 The proportion between the frequencies of the different photoproducts is the same in 253 the UVC range than with UVB²¹. It should be stressed that this analogy holds only for 254 low doses of UVC. Indeed, at this wavelength, secondary photoreactions can take 255 place, in particular for CPDs. The 260 nm absorption of pyrimidine bases is lost upon 256 formation of the cyclobutane dimer but the resulting CPDs keep an absorption band at 257 approximately 230 nm with residual absorption at 254 nm. CPDs are thus able to 258 absorb photons at this wavelength, which triggers a photoreversion reaction into the 259 original bases ^{71, 72}. As a result, the dose-dependent formation is not linear but reaches 260 a plateau ^{21, 73}. This phenomenon does not take place with UVB that is not absorbed 261 by CPDs. Because photoreversion of 64-PPs does not occur, the ratio between CPDs 262 and 64-PPs decreases as the UVC dose increases. An additional issue is that C-263 containing CPDs are more susceptible to photoreversion than TT CPD^{21,72}. Therefore, 264 the ratio between the levels of the four possible CPDs also varies at large UVC dose. 265 Although most data on UVC-induced photodamage in double stranded B-DNA were



obtained at 254 nm, data are available for lower wavelengths for isolated DNA. It was 266 267 thus shown that the yields of CPD at 220 and 240 nm represent 80 and 40% of that at 268 260 nm, respectively ⁷⁴. In the 170-200 nm region, where DNA exhibits a second 269 absorption maximum, the yield of CPD is slightly larger than that determined at 260 nm ⁷⁴. In skin, exposure to 222 nm did not lead to formation of CPDs, in contrast to 270 271 UVB and, to a lesser extent, to 254 nm radiation ⁷⁵. Works carried out on isolated DNA provided information on the effect of higher energy UVC, typically 195 nm. At this 272 273 wavelength, direct ionization of DNA bases is highly efficient and leads to damage of 274 oxidative origin mostly observed at guanine bases. Their yields is a 6-fold larger yield 275 than that of CPDs ^{76, 77}.

276

7.2.3.3 Photochemistry in bacterial spores

277 Like bacteria, yeasts and other microorganisms, bacterial spores are killed upon UVC 278 sterilization as the result of massive induction of DNA damage. A major difference is 279 the nature of the induced photoproducts. In contrast to all other living cells, exposure 280 of spores to UV radiation does not lead to the formation of CPDs and 64-PPs but to a 281 dimer involving two adjacent thymine bases, $5-(\alpha-thyminyl)-thymine$ also known as the 282 spore photoproduct (SP) ⁷⁸. This specific photochemistry is explained by the unusual A-like form of DNA in spores ⁷⁹⁻⁸¹. This conformation is induced by the highly 283 284 dehydrated core environment and the complexation of large amounts of small acid 285 soluble proteins to DNA. In addition, the spore is loaded with large concentrations of dipicolinate that favour the formation of SP^{82,83}. Like CPDs, SP is produced in 286 287 maximum yield at the maximal absorption of DNA⁸⁴. However, it is produced at higher energy (222 nm) in significant yield ⁸⁵. Formation of SP is believed to explain the 288 289 photoresistance of spores since a very efficient radical-SAM (S-adenosyl-methionine) 290 repair enzyme is present in the spore and rapidly reverts SP upon germination ⁸⁶.



291 **7.3 Oxidatively generated DNA damage in cells**

Emphasized has been placed in part 2 on the formation of pyrimidine dimers. Sunlight is also known to induce DNA damage through oxidative pathways. The underlying mechanisms have been documented by many works on isolated DNA. Formation of oxidatively generated lesions was also determined *in vitro*.

296

7.3.1 UVB and oxidative stress

297 Although often discussed for the effect of UVA, oxidatively generated DNA lesions are also relevant in the effects of UVB. The relative yield of this class of damage with 298 299 respect to CPDs is yet very much in favour of the latter. For instance, the difference in 300 yields of formamidopyrimidine glycosylase (Fpg)- and T4 endonuclease V (T4endoV)-301 sensitive sites, corresponding respectively to oxidized guanines and CPDS, is 302 approximately 1000 in CHO cells in the 290-320 nm range ⁸. Similar results were 303 reported in human fibroblasts ⁴⁷. The underlying mechanisms, that may involve 304 excitation of catalase ⁸⁷, cyclooxygenase and reduced nicotinamide adenine 305 dinucléotide phosphate (NADPH)^{88, 89}, remain to be fully elucidated. An interesting 306 recent result, extensively discussed in the chapter 3 by D. Markovitsi in this volume, is 307 the possible direct ionization triggered by absorption of UV photons by DNA bases, even below the energy corresponding to the ionization threshold ⁹⁰⁻⁹². 308

309 **7.3.2 Main UVA-induced photooxidation pathways**

310 UVA is only poorly absorbed by DNA and does not provide enough energy to induce 311 photoionization of DNA components. Therefore, all oxidation reactions induced in 312 UVA-irradiated cells arise from photosensitized reactions. In these processes, UVA 313 photons are absorbed by endogenous cellular components ⁹³ that, upon excitation,



314	trigger oxidation reactions and release of oxidizing species. Numerous works on model
315	systems and isolated DNA has provided valuable mechanistic information.
316	7.3.2.1 One-electron oxidation reactions
317	A first photosensitized process involves electron abstraction from DNA bases and is

known as "type I photosensitization". This photoreaction has been found to be efficient 318 319 with all nucleic bases studied as isolated monomer when the oxidation potential of the 320 excited photosensitizer is large enough. However, when taking place in double 321 stranded DNA, one-electron oxidation leads to the sole formation of damage at 322 guanine bases ⁹⁴⁻⁹⁶. This is explained by the much lower ionization potential of this 323 base compared to other DNA components. Consequently, radical cations produced on 324 T, C or A migrate within DNA to guanine residues where the guanine radical cation 325 gives rise to oxidation products. Those are mostly 8-oxoGua and 4-hydroxy-5-326 formamidopyrimidine (FapyGua). Secondary oxidation products of 8-oxoGua can be 327 produced at large dose but not under biologically relevant condition.

328

7.2.3.2 Singlet oxygen

329 Rather than reacting with DNA components, photosensitizers in their excited state may 330 transfer their energy to dioxygen. Unlike most molecules, the ground state of dioxygen 331 is a triplet state, which limits its direct reactivity with DNA components. However, 332 transfer of energy from the triplet excited state of the sensitizers converts it into the 333 highly reactive singlet oxygen (¹O₂) in a "type II photosensitization reaction". This 334 species, that is not a radical, readily reacts with double bonds in organic molecules 335 leading to the formation of endoperoxides and dioxetanes. It can also lead to the 336 formation of hydroperoxide through the ene-reaction and oxidize sulphur atoms. In



337 DNA, only guanine reacts with ${}^{1}O_{2}$ with formation of an endoperoxide that further leads

to the formation of 8-oxoGua as the only reaction product ^{94, 95, 97}.

339

340

7.2.3.3 Reactive oxygen species arising from the superoxide anion radical

341 Superoxide anion (O₂•·) is produced in large amounts in UV-irradiated cells as the 342 result of enzymatic process or by release from mitochondria ⁹⁸. Superoxide anion is 343 poorly reactive with DNA but is at the origin of more damaging species. Upon 344 dismutation, either catalysed by redox metals or performed by the superoxide 345 dismutase enzymes, O₂• gives rise to hydrogen peroxide H₂O₂. Hydrogen peroxide is also unable to directly damage DNA. However, reduced metal ions like Fe²⁺ or Cu⁺ 346 347 may convert it into the highly reactive hydroxyl radical •OH. In this reaction, metal ions 348 are oxidized into Fe³⁺ or Cu²⁺. The reducing environment of cells, mostly resulting from 349 the large concentration in glutathione, allows the conversion of these ions back to their 350 reduced forms able to react again with H₂O₂. This process is worsened by the release of iron from its storage proteins upon UVA irradiation ^{99, 100}. The high reactivity of •OH 351 352 leads to the formation of a wide series of damage to the four DNA bases ^{99, 100}. Purines 353 are converted into 8-oxo and formamidopyrimidine derivatives upon attack at the C8 354 position. •OH also reacts efficiently with the C5-C6 double bond of pyrimidines, and 355 leads to the formation of 5,6-dihydroxy-5,6-dihydro derivatives (thymine and cytosine 356 glycols) and 5-hydroxycytosine. Oxidation reactions also take place on the methyl 357 group of thymine with formation of 5-hydroxymethyl- and 5-formyl- uracil. •OH can also 358 perform hydrogen atom abstraction from the deoxyribose moiety of the sugarphosphate backbone ^{99, 100}. Consequently, it is one of the few species that can induce 359 360 DNA strand breaks. The superoxide anion not only leads to the formation reactive 361 oxygen species but also of nitrogen oxidizing derivatives. The most documented



362 pathway involves its reaction with NO that leads to the formation of peroxynitrite 363 HOONO and, after subsequent reaction with CO_2 , of nitrosoperoxy carbonate 364 ONOOCO₂⁻. While the former compound damages DNA through its decomposition 365 into •OH ¹⁰¹, the latter gives rise to carbonate radical anion that is a one-electron 366 oxidant ¹⁰².

367

368

7.3.3. Oxidatively generated DNA damage in UVAirradiated cells

369 Information on the formation of oxidation products in the DNA of UVA irradiated cells 370 was mostly inferred from enzymatic assays associated with either electrophoretic 371 analysis or alkaline elution. In these approaches treatment with purified DNA repair 372 enzymes are added to the protocol in order to convert specific classes of DNA base 373 damage into strands breaks that add to those directly produced by the irradiation. The 374 most frequently used enzymes are bacterial Formamidopyrimidine glycosylase and 375 Endonuclease III that recognize oxidized purines and oxidized pyrimidines, 376 respectively. T4endoV allows the detection of CPDs. In all reported studies, oxidized 377 purines, most likely 8-oxoGua, are produced in larger amounts than direct strand breaks over the entire UVA range ^{8, 103}. Additional works showed that oxidized 378 379 pyrimidines are minor lesions ¹⁰³. The distribution of DNA oxidatively generated lesions 380 in UVA-irradiated cells differs from that induced by the hydroxyl radicals produced by 381 γ -rays and from that induced by singlet oxygen produced by a specific sensitizer. The 382 ratio between the relative yield of direct strand breaks, Fpg-sensitive sites and Endo-383 III sensitive sites were 1/0.47/0.16, 0.37/1/0.4 and 1/0.25/0, respectively ¹⁰³. This 384 observation, combined with the fact that type I photosensitization is believed to be only 385 weakly involved in the formation of oxidatively generated DNA lesions ⁹⁷, led to the 386 conclusion that 80% of the DNA damaging oxidizing species are singlet oxygen and



20% hydroxyl radicals ¹⁰³. It should be stressed that the yield of oxidatively generated
lesions is not the same in all cell types. For example, melanocytes are more sensitive
than other cutaneous cells upon UVA irradiation as the result of the presence of
photosensitizing pheomelanin ^{49, 104-107}.

391 8-OxoGua being the most frequent oxidatively generated lesion, it can be used as a 392 probe to compare oxidative stress and CPD formation in the UVA range. Using 393 broadband sources and either enzymatic or chromatographic assays, CPDs have been 394 detected in larger amounts in all cell types including CHO cells and human 395 keratinocytes and melanocytes. The ratio between the amounts of CPD and 8-oxoGua 396 is approximatively 5^{22, 46, 48}, with the exception of melanocytes where the larger yield of 8-oxoGua leads to a ratio of 1.5⁴⁹. This ratio is yet wavelength-dependent as shown 397 398 in CHO cells where the values are 10 at 340 nm, 5.5 at 365 nm and 1 at 390 nm 8. It 399 can thus be concluded from these series of results that oxidatively generated lesions 400 cannot be considered as the most frequent damaging events of the nuclear DNA in 401 UVA-irradiated cells. Of course, this does not rule out that oxidative stress exhibits 402 major effects on other biomolecules and in other cell compartments.

403 **7.4** Spectral composition and photoproduct distribution

404 Most of the data available on the DNA damaging of sunlight were obtained with pure 405 UVB or UVA sources. Some more environmentally relevant works used simulated 406 sunlight that gathered the effects of both spectral ranges. However, only limited 407 amounts of information are available on the possible combined effects of different 408 wavelength ranges. The photobiological responses may not be additive and lead to 409 antagonistic or synergistic effects on the induction of some types of DNA damage. 410 Occurrence of such phenomena would emphasize the need to take into account the 411 emission spectra of the UV source. This is partially done when weighted action spectra



412 are used but this approach relies only on additivity. Evidence are already available in
413 field of DNA damage of such complex spectral effects ¹⁰⁸.

414 **7.4.1** Formation of Dewar valence isomers

Dewar valence isomers ³⁶ are produced when 64-PPs absorb a UV photon and 415 416 undergo an intramolecular electrocyclisation in their pyrimidone ring ³⁷. The latter 417 moiety exhibits a maximum absorption at 325 nm in TT and CT 64-PP and at 315 nm 418 in TC and CC 64-PP ^{38, 39, 109}. Consequently, UVB is expected to efficiently induce this 419 reaction. This is actually the case in model systems like small oligonucleotides but 420 much less in isolated and cellular double-stranded DNA, at least at biologically relevant 421 doses ^{12, 21}. This difference is explained by the overwhelming presence of normal 422 bases that preferentially absorb UVB photons, which are then not available for the 423 isomerization of 64-PPs ¹¹⁰. Interestingly, 64-PPs exhibit a residual absorption in the 424 UVA range. The ratios between the areas under the curve of the absorption spectrum 425 in the UVB (240-320 nm) and UVA (320-380) are 0.76 for TT 64-PP and 0.44 for TC 426 64-PP. By comparison, the corresponding value is only 0.02 for DNA. These figures 427 shows the much lower shielding effect of 64-PP by normal DNA bases in the UVA than 428 the UVB range. This ability of UVA at photoisomerising 64-PPs was put forward already 429 in the first article reporting the structure of Dewar in dinucleoside monophosphate ³⁸. 430 Since then, several works using different quantification tools have shown that Dewars 431 are present in significant amounts in cells exposed to simulated sunlight, namely a 432 combination of UVB and UVA ^{12, 46, 111, 112}. Additional support for the biological 433 relevance of Dewar isomer came from the observation of its formation in cells exposed to natural sunlight ^{113, 114} and from its unambiguous detection in marine 434 435 microorganisms collected in the ocean ¹¹⁵. Recently, the extent of photoisomerisation 436 of 64-PPs in isolated DNA exposed to sunlight was reported to depend on the season



¹¹⁶, following the UVB/UVA ratio. It is difficult to provide a value for the ratio between the amount of 64-PPs and Dewars generated in DNA by exposure to natural or simulated sunlight. Indeed, the proportion of Dewars increases at the expense of that of 64-PPs when the dose increases. All these results could be rationalized in terms of emission spectrum and absorption at the different wavelengths ¹¹⁰.

442

7.4.2. UV-induced formation of sensitizers

443 Pyrimidine dimers are not the only types of photodamage impacted by the effects of 444 two UV wavelengths ranges. Although this process has not been directly documented 445 in cells, test tube experiments strongly suggest that UVB-mediated degradation of 446 various biomolecules leads to the formation of photosensitizers that increase the extent 447 of UVA-mediated oxidative stress. A first example are endogenously produced pterin 448 derivatives ^{117, 118}. Pterin is a well-known photosensitizer studied for its damaging 449 properties of biomolecules and its application in photodynamic therapy ¹¹⁹⁻¹²¹. Another 450 source of photosensitizing photoproducts is tryptophan. Its main photoproduct is N-451 formylkynurenine that is known to exhibit photodynamic properties ^{122, 123}. UVB 452 irradiation of tryptophan also generates 6-formylindolo[3,2-b]carbazole (FICZ) which is 453 also a photosensitizer ¹²⁴. The yield of FICZ is yet very low ¹²⁵ and its role in UVA-454 induced oxidative stress is unlikely. Other combinations of wavelengths may enhance oxidative stress. For example, UVA-induced lipofuscin was reported to photosensitize 455 keratinocytes to visible light ¹²⁶. 456

457

7.4.3. Impact of UVA on the repair of UVB-induced dimers

458 Several works have unambiguously shown that exposure to UVB, especially in a 459 chronic or repetitive pattern, led to the enhancement of DNA repair capacities ¹²⁷⁻¹³¹ 460 and in particular of nucleotide excision repair, which handles pyrimidine dimers.



Evidence are also accumulating that UVA exhibits the opposite effects. Exposure to 461 UVA prior to UVB has been shown to reduce the rate of removal of CPDs, both in 462 463 keratinocytes ⁹ and melanocytes ¹³². *In vitro* studies also showed that UVA, alone or in the presence of photosensitizer, induced oxidative damage to repair proteins ¹³³ like 464 465 RPA ¹³⁴ or PCNA ¹³⁵. The combination of these two opposite impacts of UVB and UVA 466 on DNA repair strongly suggests that the persistence of pyrimidine dimers could 467 depend on the ratio between these two wavelength ranges in the spectrum of the 468 incident light.

469 **7.5 DNA damage at the nucleotide resolution in cellular DNA**

470 The advent of genome sequencing opens new avenues for the analysis of UV-induced pyrimidine dimers. In recent years, several techniques were developed for the 471 472 identification and study of CPDs and 64-PPs at single-nucleotide resolution (Fig. 7.2). 473 UV dimers have been mapped in bacteria ¹³⁶, yeast ^{29, 137, 138}, and human genomes ^{6,} ^{28, 30, 31, 139}. Three major molecular approaches have been applied to pinpoint the 474 475 damage positions. The first relies on damage-specific endonucleases to fragment 476 DNA, and identifies damage sites based on the nucleotides at the fragment ends. CPD-477 seq, adduct-seq, and most recently GLOE-seq rely on the T4endoV and were therefore only applied to CPDs ²⁹⁻³¹. In contrast, excision-seq and UVDE-seq ^{137, 138} use the UV 478 479 damage endonuclease (UVDE) and can map both CPDs and 64-PPs. However, this 480 technique requires very high damage levels and has not been applied to mammalian 481 genomes. The second approach, used by HS-Damage-seq⁶, relies on blocking of DNA polymerases by the pyrimidine dimers. Here, damaged DNA from cells is subjected to 482 483 in vitro primer extension, and the position of the lesion is identified again based on the 484 fragment end. Using anti-CPD and anti-64-PP antibodies allows discrimination and mapping of the two damage types separately. The third and last approach is most 485



486 straightforward, but is so far limited in its application. This approach relies on 487 identification of the damage-sites directly during single-molecule real-time (SMRT) 488 sequencing. Pyrimidine dimers significantly block DNA polymerization and will alter 489 polymerase kinetics. Direct sequencing of dimers was published in 2011 using 490 synthetic oligonucleotides ¹⁴⁰ and has not significantly advanced since, suggesting a difficulty of scaling to genomic sequences. The recent RADAR-seg method ¹³⁶ may 491 492 overcome these hurdles since it by passes the necessity to directly sequence dimers: 493 it uses T4 endonuclease IV to nick at damage site and *in vitro* synthesis to replace the 494 dimer with a stretch of modified nucleotides prior to SMRT sequencing. However, 495 RADAR-seq reduces the resolution of damage detection and has so far only been 496 applied to in vitro irradiated bacterial DNA.

497 Analysis of the specific pyrimidines composing CPDs, performed with different 498 methods, cell types, and UVC or UVB doses all showed a preference for 499 TT>TC>CT>CC ^{6, 28-31}. HS-Damage-seq, excision-seq and UVDE-seq have also been 500 used to map 64-PP in human and yeast genomes, and have found damages to form preferentially in TC>TT>CT>CC ^{6, 137, 138}. These preferences are consistent with the 501 502 previous bulk DNA analyses. While UVDE will excise multiple photoproduct types, in 503 UVDE-seq, UV-irradiated DNA was pre-treated with photolyase to remove CPDs. This 504 allowed the identification of not only 64-PP but also non-canonical TA and AC photoproducts ¹³⁸. In addition to identifying the specific nucleotides composing the 505 506 dimers, sequencing-based methods also provide the immediate sequence context of 507 dimers. Thus, these methods show enrichment of CPD damage formation within T-508 tracks of DNA, and the preference for 64-PP damage formation in TC or TT sequences 509 flanked by a C upstream and an A downstream of the damaged site ^{6, 137}.





510

511 Figure 7.2. The three major approaches to mapping UV induced dimers. A) Mapping 512 UV dimers based on endonuclease cleavage sites. Prior to cleavage free 3'-OH 513 groups are blocked in order to obstruct adapter ligation. Shown is the schematic of 514 CPD-seq. Adduct-seq, GLOE-seq, excision-seq and UVDE-seq are similar to CPD-515 seq but slightly modified. B) Mapping dimers based on the position at which a DNA 516 polymerase is blocked in-vitro (Damage-seg). C) Direct detection of damages 517 through SMRT sequencing. Red triangle: UV dimer. Red X: blocked DNA ends. 518 Yellow circle: biotin. Figure generated with Biorender.com 519 In the large mammalian genomes, obtaining a quantitative high-resolution map of every damageable bipyrimidine site requires extremely high sequencing coverage. 520 521 Most of the studies to date have not produced such high-coverage maps. Instead, in 522 order to analyze the chromosomal context of damage formation, studies average

523 damage densities over aggregated sequences. These include the study of damage



formation over transcription start sites of genes, well-positioned nucleosome dyads ortranscription factor binding sites:

526

7.5.1 Damage formation in genes

527 In general, the major determinant of CPD damage formation is the frequency of 528 pyrimidine dimers, primarily TT sequences. This frequency is not uniform in the 529 genome, and specifically in genes. TT sequences are depleted near the transcription 530 start sites of genes, and asymmetrically distributed between the transcribed and non-531 transcribed strands, resulting in overall lower damage levels on the transcribed strands 532 of genes ¹⁴¹.

533 **7.5.2 Damage formation in nucleosomal DNA**

534 Nucleosome positions are generally associated with a depletion of TT pairs that could 535 lead to lower levels of damage ¹⁴². Analysis of CPD formation after normalizing to the 536 underlying TT frequencies in yeast and human genomes has shown that the rotational 537 setting of the DNA in the nucleosome affects CPD damage formation: a 10bp periodicity in CPDs is observed, with peaks in the outward facing rotational settings^{29,} 538 539 ¹⁴³. Control experiments with genomic DNA irradiated *in vitro* did not produce this effect 540 confirming that this periodicity is indeed a product of nucleosome binding. This 541 increase in damage frequency at specific rotational settings within a nucleosome can 542 be explained by the conformational changes in DNA that result in the adjacent 543 pyrimidine bases assuming a favorable angle for dimer formation. This aspect is 544 discussed by Gillet at al. in chapter 6 of this book.

545

7.5.3 Damage formation at transcription factor binding sites

546 Transcription factor binding sites were identified as sites of elevated UV mutagenesis547 in melanoma. Several studies assessed damage formation at transcription factor



548 binding sites, focusing on bipyrimidines contained within or adjacent to the consensus 549 binding motifs. In yeast, Abf1 and Reb1 binding sites were associated with lower CPD frequencies ²⁹. In humans, some transcription factors were associated with reduced, 550 while others with elevated damage frequencies 6, 144. These altered damage 551 552 frequencies were absent in irradiated naked DNA or at unbound motif-sites in cells, 553 indicating they are a direct result of binding. Like in nucleosomes, binding can alter 554 DNA confirmation and result in favorable or unfavorable angles for dimerization. Three 555 independent studies identified ETS1 binding sites as hotspots for damage formation, that were associated with subsequent melanoma mutation hotspots ^{28, 30, 139}. 556

Higher mutation rates could be a result of higher damage, but could also be the product of lower repair efficiency. Indeed, a recent study suggests that it is primarily the lower repair efficiency at transcription factor- bound sites that promotes mutagenesis ¹⁴⁴. Repair of UV damages is extensively discussed by John J. Wyrick in chapter 12 of this volume. A key factor in untangling the relative contributions of damage formation versus repair will be to understand how damages affect protein binding in chromatin, including nucleosomes and transcription factors, a subject far less studied to date.

564 7.6 UV-induced DNA damage in skin

565 Most of the information discussed above were obtained in *in vitro* models. The amount 566 of information in skin is less abundant, mostly because of limitation in sample size and 567 technical issues. However, data obtained by immunological assays and in particular 568 immunohistochemistry, and chromatographic techniques provided insight on the 569 impact of the organization of the cutaneous tissues on the formation of photodamage. 570 The vast majority of the information available concerns pyrimidine dimers. Their results 571 show that the skin morphology drastically affects the level of DNA damage (Fig. 7.3).



572 It should also be stressed that eyes represent another target of solar UV. The formation
573 and repair of CPDs have only been investigated recently ¹⁴⁵⁻¹⁴⁸.

574

7.6.1 Yield of lesions in skin

575 UVB and UVA induce the formation of CPDs in both keratinocytes and melanocytes in the same yield in skin ⁵⁰. Interestingly, the mean value of the yield of CPDs induced by 576 UVB in DNA of the epidermis is lower than in culture of keratinocytes, which are the 577 578 most frequent cell types in this skin compartment ²². This can be first explained by the 579 protective effect of the stratum corneum. An additional explanation is the shielding effect of the different layers of keratinocytes in the epidermis ¹¹². This effect is 580 581 wavelength-dependent with a decrease in the CPDs yield per cell layer 15 faster at 582 260 nm than at 230. Determination of accurate guantitative data also showed that the yield of CPDs was roughly 2 and 4 orders of magnitude lower at 320 and 365 nm, 583 584 respectively, compared to 290 nm in both dermis and epidermis ⁶⁹. In the same study, 585 64-PPs were found to be produced in a constant ratio from 260 to 310 nm, and then 586 strongly decrease at higher wavelengths. In contrast to UVB, UVA is poorly absorbed 587 by the stratum corneum and penetrates through the keratinocytes layers without 588 significant attenuation. These observations explain why the ratio between the yield of 589 CPD in the UVB and UVA ranges is lower in skin than in cultured cells ^{22, 24, 48}. It should 590 be stressed that immunohistochemistry results showed that the level of UVA-induced 591 CPDs was slightly higher in the basal layer than in layers located above. ¹⁴⁹. This 592 suggested the occurrence of a reflexion of UVA takes place at the dermal-epidermal 593 iunction.





594

595 Figure 7.3: Penetration of UV radiation in human skin. The highly energetic UVC 596 radiation are rapidly absorbed by the stratum corneum. UVB is absorbed in the 597 epidermis and reaches the top of the dermis. In contrast, UVA is weakly attenuated 598 and interacts with the whole dermis. In terms cell types, keratinocytes present in 599 dense layers in the epidermis are much more affected by both UVA and UVB than 600 fibroblasts sparsely spread in the dermal extracellular matrix.

601

602 **7.6.2 Effect of phototype on the formation of CPDs**

603 Phototypes is a convenient way to stratify the different skin types, which mostly reflect 604 ethnic origins. In Caucasian, the risk of burning and the ability to tan upon exposure to 605 sunlight has led to the definition of phototypes I (fair skin) to IV (brown skin). 606 Phototypes V and VI are Asian and dark brown skins, respectively. A high skin 607 phototype is known to be highly protective against skin cancer, as exemplified by studies carried out in areas harbouring fair- and dark-skinned populations ¹⁵⁰⁻¹⁵². On 608 609 the average, protection factors of 20 and 70 against melanoma and carcinomas, 610 respectively, were observed when skin of phototypes VI were compared to phototype 611 II ^{153, 154}. This was only partially explained by the decrease in the mean level of CPD in 612 keratinocytes that was found to be only 10-times lower in skin of phototype VI



compared to II following exposure to simulated sunlight or UVB 62, 155. The same values 613 were observed for both immediate and dark CPDs ⁶². The latter result further suggests 614 615 that photooxidation of melanin is not the only mechanism at the origin of dark CPDs. 616 A better understanding of the impact of the phototype and of the melanin content was 617 provided by experiments assessing the distribution of melanin and of the yield of CPDs. 618 These works unambiguously showed that the best protection against CPDs between 619 dark and fair skins was observed in basal layer, where the largest proportion of melanin 620 is stored in dark skin ¹⁵⁵⁻¹⁵⁷. The basal layer, that harbours the cells at the origin of 621 melanomas and basal cell carcinomas, is much more protected than the rest of the 622 epidermis. It should be stressed that these strong protective effects are no longer 623 observed when less extreme phototypes are concerned. The ratio between the yield of CPDs in phototype II and IV is only approximately 1.5^{24, 158-160}. Another interesting 624 625 data regarding pigmentation is that tanning of Caucasian skin affords only a weak protection of a factor 2 against the formation of CPDs ¹⁶⁰⁻¹⁶². 626

627 7.7 Mutagenic consequences of DNA photodamage

The biological consequences of DNA photodamage in terms of carcinogenesis are mostly driven by their ability to induce mutations. A first requisite for a lesion to be mutagenic is that it escapes DNA repair as observed for pyrimidine dimers. The mutagenicity of a photodamage also depends on its chemical structure and the resulting impact on the coding properties upon replication of DNA. The combination of all the features explain the mutation spectra observed in tumours

634

7.7.1. In vitro data of UV-induced mutagenesis

635 UV-mutagenesis is investigated for years. The first works involved irradiation of 636 plasmids further introduced and replicated in bacteria. These early studies showed that



637 UVB and UVC mostly led to mutation at bipyrimidine sites. The subsequent works took 638 advantage of progress in oligonucleotide synthesis to study the specific mutational 639 properties of individual pyrimidine dimers with purified polymerases or following incorporation in plasmids and replication in cells ¹⁶³⁻¹⁶⁷. It was thus observed that TT 640 641 CPD was poorly mutagenic in contrast to TT 64-PP mostly because of differences in 642 impact of the lesions on the local structure of DNA. In contrast to TT CPD, deaminated 643 TC CPD (TU CPD) were highly mutagenic, leading to $C \rightarrow T$ transitions because U 644 codes like T during replication. The TC 64-PP was less mutagenic than the 645 corresponding TT photoproduct ¹⁶⁸. In contrast, the TT Dewar was more mutagenic 646 than the TC Dewar and less than the TT 64-PP. It is worth mentioning that in mammalian cells, in addition to deamination ^{41, 169} and structural modifications, 647 648 mutagenesis of UV-induced photoproducts is largely explained by the role of translesional polymerases ¹⁷⁰⁻¹⁷², as discussed in the chapter 16 by Menck et al. in this 649 650 book. Some of these enzymes, like polymerase n, are referred to as "error-free" when 651 bypassing pyrimidine dimers. They are thus expected to accurately incorporate A 652 opposite T in TT CPDs, but also opposite U that would arise from C deamination in TC dimers. It should be stressed that experiments involving cells ²⁰ or mice ¹⁷³ transfected 653 654 with specific repair enzymes led to the conclusion that these mutations mostly result from CPDs rather than 64-PPs, which are very efficiently repaired. Another important 655 656 feature is that $C \rightarrow T$ transitions at TC sites are also predominant in cells exposed to UVA ^{45, 174, 175} in line with the predominant formation of CPDs in this wavelength range. 657 658 The bulk of these data helped defining a UV mutational signature where $C \rightarrow T$ transition at TC sites and CC \rightarrow TT tandem mutations were predominant ^{176, 177}. 659

660 In vitro mutational data are also available for oxidatively generated damage. 661 8-OxoGua, following misincorporation of adenine upon replication leads to $G \rightarrow T$



662 transversion ¹⁷⁸. Information were also obtained on the Fapy derivatives that show that 663 FapyGua leads to $G \rightarrow T$ transversion while 4,6-Diamino-5-formamidopyrimidine 664 (FapyAde) is poorly mutagenic ¹⁷⁹. Among thymine oxidation products, thymine glycols 665 are strongly blocking lesions but poorly mutagenic. 5-Formyluracil also leads to a low 666 mutation frequency ¹⁸⁰. In contrast, cytosine oxidation products are mutagenic, mostly following deamination ^{181, 182}. The contribution of these mutations to UVA-mutational 667 668 signature is yet limited. The $G \rightarrow T$ transversion typical of the most frequent oxidatively 669 generated lesions is a minor mutation, with the exception of one work involving 670 mutation analyses in tumours collected in the basal layer ¹⁸³. Whole genome 671 sequencing analysis of the mutation induced by UVA in xeroderma pigmentosum 672 variant human cells failed to detect a significant contribution of oxidative damage but rather the mutation signature of pyrimidine dimers ¹⁸⁴ with frequent mutations at 673 674 bipyrimidine sites. Mutation at adenine is also observed ¹⁸⁵ while adenine oxidation 675 products are not frequent upon UVA irradiation.

676 **7.7.2. Next generation sequencing of human tumours**

677 Sequencing of specific genes in human tumours such as p53 has confirmed the presence of the UV signature ¹⁸⁶⁻¹⁸⁸. The development of modern and fast sequencing 678 679 techniques, which revolutionized the field of genomic, provided a novel and 680 outstanding perspective on the mutations underlying human cancers. Next generation 681 sequencing techniques have been used to determine millions of mutations present in 682 thousands of tumours collected from different types of cancers ^{189, 190}. In the field of 683 skin carcinogenesis, a first interesting result is that, together with lung cancers, 684 melanoma is one of the cancer type accumulating the largest number of mutations per 685 tumour ¹⁹⁰⁻¹⁹². The value is more than one order of magnitude larger than, for example, 686 in leukaemia or medulloblastoma. This can be explained by the fact that skin is



687 constantly exposed to sunlight and accumulates numerous DNA damage and their 688 resulting mutations during the development of the tumour. An interesting output of this 689 sequencing effort is the statistical analysis of the databank that made possible the determination of mutational signatures ^{193, 194} available on-line from the COSMIC 690 691 consortium ¹⁹⁵. Signatures bearing mutations at bipyrimidine sites (Table 7.2) are observed with a high frequency in melanoma ^{190, 196-199}, basal cell carcinoma ^{200, 201}, 692 squamous cell carcinoma ²⁰² and cutaneous T cell lymphoma ²⁰³. They are basically 693 694 absent from other cancer types. In the field of melanoma, the same approach allowed to show that acral and mucosal melanomas do not exhibit the same signature ^{197, 204}. 695 696 Another interesting information provided by these techniques is related to a mutational 697 signature associated with oxidative stress and 8-oxoGua. In the COSMIC database, 698 this signature is observed in one third of the breast, prostate and stomach cancers but 699 not in skin cancer. A last comment regarding melanoma is the high frequency of a $A \rightarrow T$ transversion leading to the BRAF^{V600E} mutation ²⁰⁵. This mutation does not take place 700 701 at a bipyrimidine site and does not correspond to known oxidative lesions. In addition, 702 it is also frequent in a number of internal cancers and may thus not be of 703 photobiological origin. However, recent UVDE-seq results in UVC-irradiated yeast proposed a role for AT dimers ¹³⁸, although these photoproducts have been observed 704 705 only in test tube experiments and in much lower yield than CPDs ^{206, 207}.

706

Table 7.2: Proportion (in %) of different mutation signatures determined in the

- 707 COSMIC* database for skin cancers. Results represents the ratio between the
- number of tumours exhibiting each signature and the number total number of tumour
- 709

of each cancer. BCC: basal cell carcinoma, SCC: squamous cell carcinoma.

signature	7a	7b	7c	7d	tandem1	18
Mutations	C to T at TC	C-to T at TC and CC	T to A and T to C	T to C and C to T	CC to TT	oxydation
Melanoma	84	82	40	55	100	n.d.**
BCC	100	59	23	41	-	n.d.**

SCC	98	100	74	90	-	n.d.**	

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- 710 *: ¹⁹⁵
- 711 **: not detected
- 712

713 Abbreviations :

714 64-PP: Pyrimidine (6-4) pyrimidone photoproduct; 8-oxoGua: 8-oxo-7,8-715 dihydroguanine; BCC: basal cell carcinoma; CHO: Chinese hamster ovary cells; CPD:

716 cyclobutane pyrimidine dimers; Dewar: Dewar valence isomer of 64-PP; FapyAde: 4,6-

717 Diamino-5-formamidopyrimidine; FapyGua: 4-hydroxy-5-formamidopyrimidine; FICZ :

718 6-Formylindolo[3,2-b]carbazole; Fpg: formamidopyrimidine glycosylase; SAM S-

adenosyl-methionine; SCC: squamous cell carcinoma; SMRT: single-molecule real-

- 720 time sequencing; SP: spore photoproduct (5-(a-thyminyl)-thymine); T4endoV:
- 721 endonuclease V of phage T4; UVDE: UV damage endonuclease
- 722

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