

Poly(vinylidene fluoride)-Stabilized Black γ -Phase CsPbI₃ Perovskite for High-Performance Piezoelectric Nanogenerators

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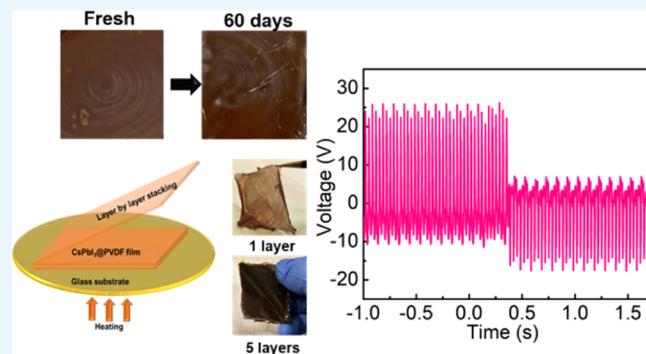
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ABSTRACT: Halide perovskite materials have been recently recognized as promising materials for piezoelectric nanogenerators (PENGs) due to their potentially strong ferroelectricity and piezoelectricity. Here, we report a new method using a poly(vinylidene fluoride) (PVDF) polymer to achieve excellent long-term stable black γ -phase CsPbI₃ and explore the piezoelectric performance on a CsPbI₃@PVDF composite film. The PVDF-stabilized black-phase CsPbI₃ perovskite composite film can be stable under ambient conditions for more than 60 days and over 24 h while heated at 80 °C. Piezoresponse force spectroscopy measurements revealed that the black CsPbI₃/PVDF composite contains well-developed ferroelectric properties with a high piezoelectric charge coefficient (d_{33}) of 28.4 pm/V. The black phase of the CsPbI₃-based PVDF composite exhibited 2 times higher performance than the yellow phase of the CsPbI₃-based composite. A layer-by-layer stacking method was adopted to tune the thickness of the composite film. A five-layer black-phase CsPbI₃@PVDF composite PENG exhibited a voltage output of 26 V and a current density of 1.1 μ A/cm². The output power can reach a peak value of 25 μ W. Moreover, the PENG can be utilized to charge capacitors through a bridge rectifier and display good durability without degradation for over 14 000 cyclic tests. These results reveal the feasibility of the all-inorganic perovskite for the design and development of high-performance piezoelectric nanogenerators.



INTRODUCTION

The piezoelectric effect is considered as a promising solution to remit the energy crisis. Thus, piezoelectric nanogenerators (PENGs) have become a hot topic of research in terms of their potential to convert irregular mechanical energy to electric energy.¹ Various materials have been extensively adopted as piezoelectric materials for mechanical energy harvesting, such as ZnO,^{2,3} GaN,⁴ and a number of piezoelectric ceramics, including BaTiO₃⁵ and lead zirconium titanate (PZT).^{6,7} However, the growth and fabrication of these ceramic materials typically involve a very complex and cost-intensive process.⁵ In addition, the brittle nature of the ceramics makes them fragile under mechanical force.^{5,8} Therefore, it is vital to develop new materials that are easy to be fabricated and possess good performance for PENGs.

Recently, lead halide perovskites have been extensively investigated in the area of optoelectronic devices due to their excellent absorption coefficient, long diffusion length, high photoluminescence quantum efficiency, and superior electrical properties.^{9–12} In addition, the spontaneous polarization of the organic–inorganic lead halide perovskites can be ascribed to the permanent dipoles provided by the molecular cations; in this case, it plays a key role in ferroelectricity.^{13–15} Therefore,

piezoelectric energy harvesting applications beyond photovoltaic solar cells have been explored. For instance, the CH₃NH₃PbI₃ perovskite has been reported with excellent piezoelectric properties, exhibiting an output voltage of 2.7 V and a current density of 140 nA/cm².¹⁶ To improve the performance, the composite-based perovskites have been fabricated with perovskite nanoparticles embedded in polymers such as poly(dimethylsiloxane) (PDMS) and poly(vinylidene fluoride) (PVDF).^{17–20} In comparison to the organic–inorganic halide perovskites, the all-inorganic counterparts are more stable in terms of moisture and heat.²¹ Thus, CsPbBr₃ has been explored as a piezoelectric material with an excellent output voltage of ~16.4 V and an output current of ~604 nA for the 260 nm thick film.²² The polymer-based CsPbBr₃ composites have also been investigated recently and shown improved performance.^{23–25} However, there has been no report

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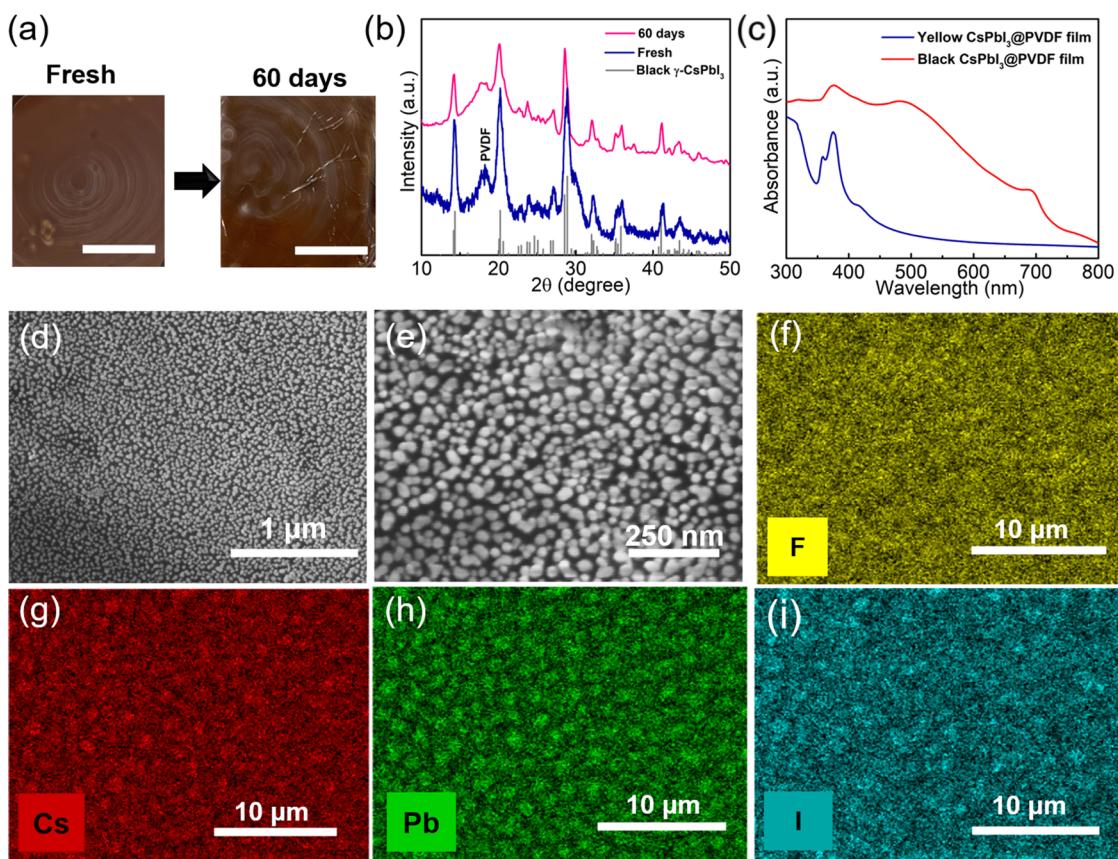


Figure 1. (a) Optical images of the prepared black-phase CsPbI₃@PVDF film and aging for 60 days at ambient conditions. Scale bar: 1 cm. (b) X-ray diffraction (XRD) spectrum of the black-phase CsPbI₃@PVDF film. The gray lines are the calculated black γ -phase CsPbI₃. (c) Ultraviolet–visible (UV–vis) absorption spectra of the prepared yellow and black CsPbI₃@PVDF films. (d–e) Scanning electron microscopic (SEM) images of the black-phase 20 wt % CsPbI₃@PVDF film. (f–i) Corresponding elemental mapping images of F, Cs, Pb, and I in Figure S4b of the black-phase 30 wt % CsPbI₃@PVDF film.

on the piezoelectric energy harvesting performance on the all-inorganic perovskite (CsPbI₃). Since the Cs⁺ cation is small and near the lower limit for lead iodide perovskite formation, the black perovskite phase of CsPbI₃ can typically be stable at temperatures above 300 °C.²⁶ As temperature decreases to room temperature, CsPbI₃ suffers from thermodynamical phase transition to the yellow non-perovskite orthorhombic phase.^{26,27} Although in a recent study it is found that the black perovskite phase of CsPbI₃ with an orthorhombic structure (γ -phase) can be obtained at room temperature through rapid quenching, the black phase is still not stable and degraded into a yellow-phase material while exposed to moisture.²⁸ Limited methods have been developed to sustain the perovskite phase of CsPbI₃ at room temperature.^{29–32} For instance, Li et al.²⁹ have reported that poly(vinylpyrrolidone) (PVP) induces surface passivation on the surface of CsPbI₃ perovskite. The acylamino groups of PVP lead to electron cloud density enhancement on the surface of CsPbI₃, thus lowering surface energy and leading to a stable cubic α -phase CsPbI₃ at room temperature.

In this study, we present a simple but robust route of PVDF-passivated CsPbI₃ to stabilize the black γ -phase of CsPbI₃ perovskite and fabricate the CsPbI₃@PVDF composite for piezoelectric nanogenerators. The PVDF-stabilized black-phase CsPbI₃ perovskite composite film can be stable under ambient conditions for more than 60 days and over 24 h while heated at 80 °C. The PFM measurements of the black CsPbI₃/PVDF

composite display well-developed ferroelectric properties with a high piezoelectric coefficient of 28.4 pm/V. A layer-by-layer stacking method was developed to fabricate CsPbI₃@PVDF composite films with a tunable film thickness. The yellow phase and black phase of the CsPbI₃-based composite films on piezoelectric performance were also compared. The black-phase CsPbI₃-based PVDF composite displays more than 2 times higher voltage and current output. By increasing the film thickness, the voltage output and current density of the black-phase CsPbI₃@PVDF-based PENG with five-layer stacking are obtained at a value of 26 V and 1.1 μ A/cm², respectively, which is more than 6 times higher for voltage and 2 times higher for current output of the two-layer stacked composite PENG. The energy generated by the PENG can be used to charge capacitors. Also, the PENG can generate peak power at 25 μ W. All of these results show that the all-inorganic perovskite has great potential in high-performance piezoelectric energy harvesting applications.

RESULTS AND DISCUSSION

The CsPbI₃@PVDF composite films were prepared via a simple and highly reproducible spin-coating process. Details of the preparation process are outlined in the **Experimental Methods** section. Typically, CsPbI₃/PVDF solution was spin-coated on a glass substrate. At different annealing temperatures, the phases of CsPbI₃ within the composite could be tuned. As shown in Figure S1, a pure yellow color CsPbI₃@PVDF composite film

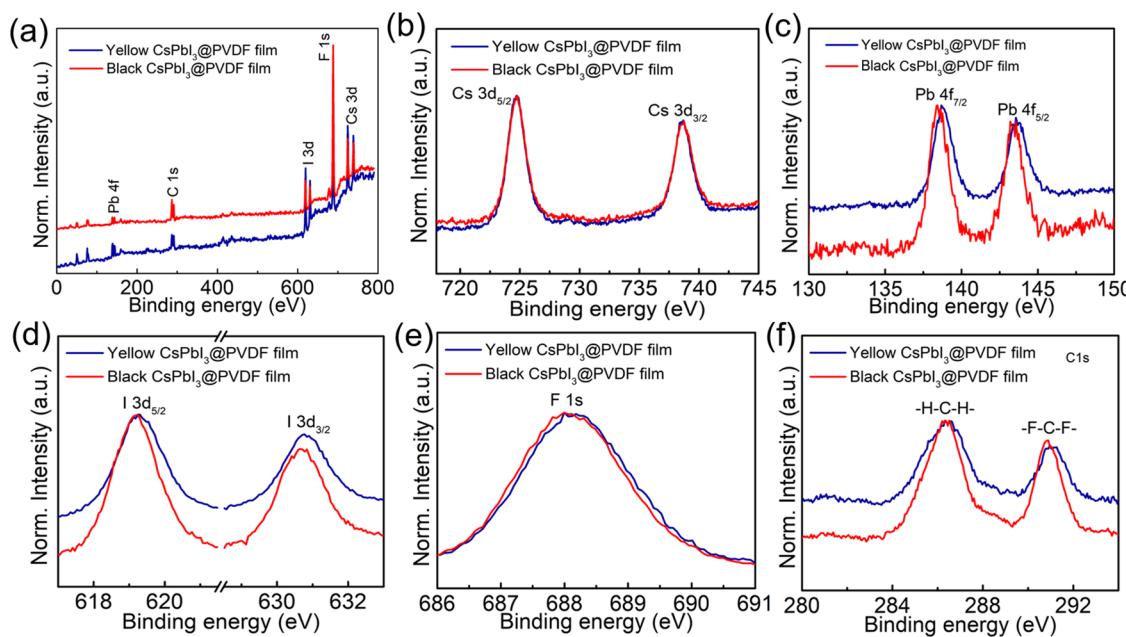


Figure 2. (a) X-ray photoelectron spectroscopy (XPS) survey spectra of yellow and black color CsPbI_3 @PVDF composite films. The high-resolution XPS spectra of (b) Cs 3d, (c) Pb 4f, (d) I 3d, (e) F 1s, and (f) C 1s.

was obtained at 50 °C. By increasing the temperature to 70 °C, a partially brown area could be identified in the composite film, indicating that part of CsPbI_3 formed as black phase. Further increasing the temperature above 130 °C, the color of the CsPbI_3 @PVDF composite film turned to a pure dark-brown or black color (depending on the thickness), suggesting the formation of the pure black-phase CsPbI_3 . The photograph in Figure 1a shows that the black-phase CsPbI_3 @PVDF composite film can be sustained at room temperature and remains stable for more than 60 days. Under ambient conditions and at 80 °C, it also displayed excellent thermal stability without obvious color changes for over 24 h (Figure S2), indicating that the black phase of CsPbI_3 can be well-sustained within PVDF under different conditions. The X-ray diffraction (XRD) patterns of CsPbI_3 @PVDF composite films presented differences in the crystal structures of films obtained at different temperatures. At a low temperature (50 °C), a yellow color CsPbI_3 @PVDF film was obtained and displayed the orthorhombic yellow $\delta\text{-CsPbI}_3$ phase, as shown in Figure S3. As the temperature increased to 70 °C, the interaction between PVDF and CsPbI_3 increased, new XRD peaks started to appear within the obtained composite, which could be assigned to the black γ -phase of CsPbI_3 . In addition, the peak at 20.2° could be assigned to the β -phase of PVDF. By further increasing the temperature, the film obtained at 180 °C exhibited the black $\gamma\text{-CsPbI}_3$ perovskite with an orthorhombic crystal structure, as shown in Figures 1b and S3. The peaks at 17.7, 18.2, and 26.2° could also be identified, which could be ascribed to α -phase PVDF.³³ It was noticed that the (110) peak of α -phase PVDF could not be resolved as it was positioned close to the (112) peak of the black $\gamma\text{-CsPbI}_3$ perovskite at 20.2°. In addition, the XRD pattern obtained after 60 days matches well with that of the freshly prepared sample, indicating the good stability of the black CsPbI_3 perovskite in PVDF (Figure 1b). Figure 1c presents the ultraviolet-visible (UV-vis) absorption spectra of the CsPbI_3 @PVDF film obtained at different temperatures. The yellow CsPbI_3 @PVDF film exhibited a limited visible-light-absorption range of less than 450 nm. In contrast, the black

CsPbI_3 @PVDF film showed an absorption edge above 700 nm, which is consistent with the results from the literature.^{28,30} This indicates the successful formation of the black-phase CsPbI_3 within the composite film.

The surface morphology of the black-phase 20 wt % CsPbI_3 @PVDF film is shown in Figure 1d. It exhibits CsPbI_3 particles uniformly embedded within PVDF. A closer observation of CsPbI_3 particles under high magnifications reveals small CsPbI_3 nanocrystals with a size of ~50 nm, as shown in Figure 1e. As the concentration increased, the obtained 30 wt % CsPbI_3 @PVDF film turned from a dark-brown color to a black color (Figure S4a) and exhibited larger crystals with sizes over 500 nm (Figure S4b). Although due to the fast CsPbI_3 precipitation (within a few seconds) when heated at 180 °C, a few crystal agglomerates on the film surface were observed; there could still be a very thin PVDF film on the crystal surface and passivate the perovskite material, stabilizing the black γ -phase CsPbI_3 . It should be noted that the films prepared with different concentrations of CsPbI_3 are all very stable in air, without obvious degradation for over 6 months (Figure S5). Figure 1f–i shows the EDS elemental mappings of F, Cs, Pb, and I; it demonstrates that CsPbI_3 crystals were dispersed homogeneously in the PVDF polymer and formed the homogeneous CsPbI_3 -PVDF matrix.

To gain more insights into the PVDF stabilization mechanism on the orthorhombic black-phase CsPbI_3 , we conducted X-ray photoelectron spectroscopy (XPS) measurements on both yellow and black CsPbI_3 @PVDF films. The full XPS spectra are shown in Figure 2a. XPS analyses of the high-resolution spectra involving Cs 3d, Pb 4f, I 3d, F 1s, and C 1s were then performed to further clarify their electronic states. Figure 2b shows the typical Cs 3d spectra with no evident peak shifting, in which the strong peaks at 724.83 and 738.68 eV corresponding to Cs 3d_{5/2} and Cs 3d_{3/2}, respectively (Figure 2b). As shown in Figure 2c, the Pb 4f spectrum for the yellow CsPbI_3 @PVDF film was recorded with two contributions 4f_{5/2} and 4f_{7/2} located at 138.68 and 143.56 eV, respectively. As for the black CsPbI_3 @PVDF film, the Pb 4f spectrum shifted to a lower binding energy. Similarly, the two I 3d peaks, corresponding to 3d_{5/2} and 3d_{3/2},

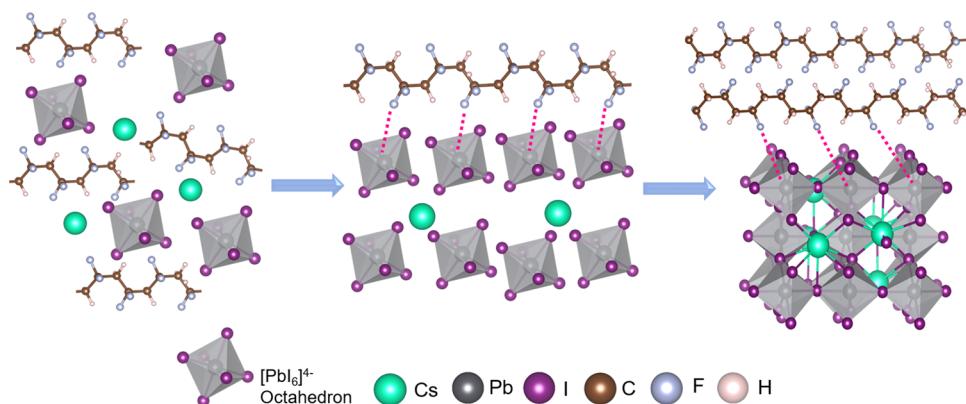


Figure 3. Schematic illustration of the crystal growth and the chemical bonding between $CsPbI_3$ and PVDF.

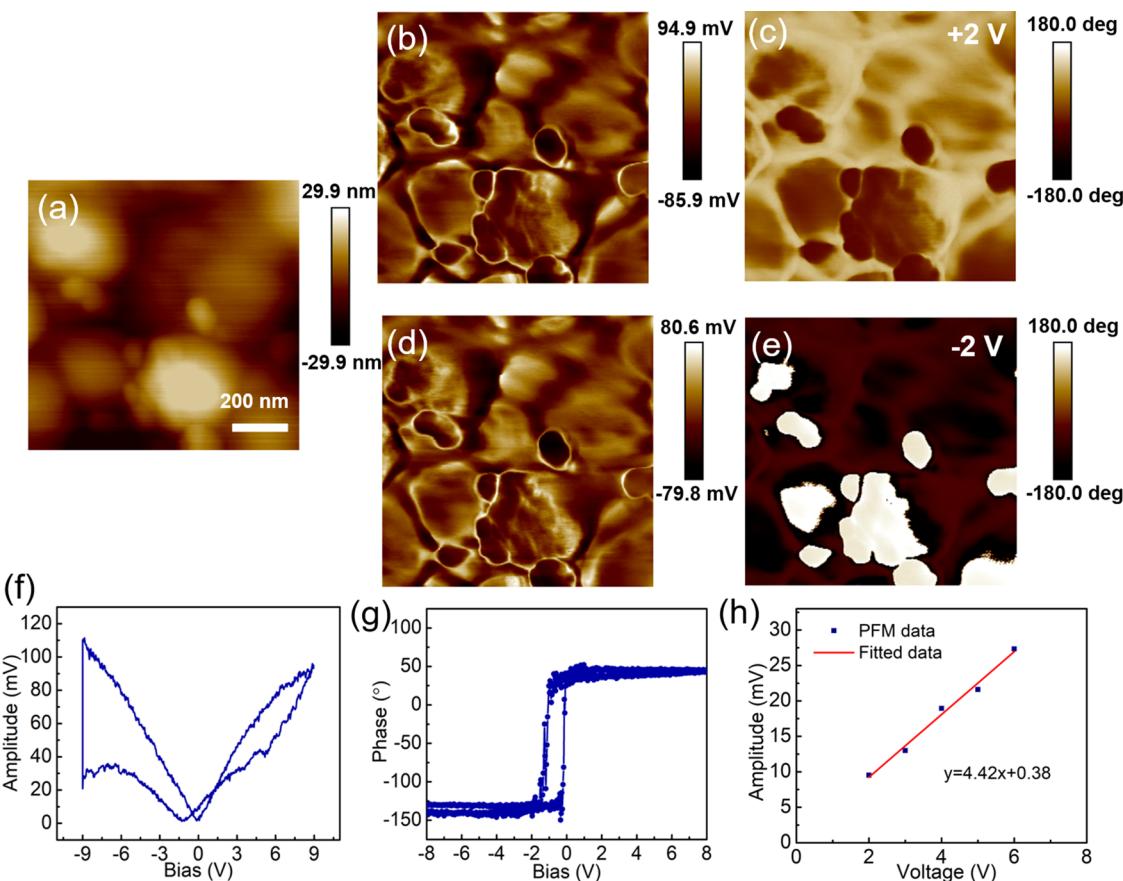


Figure 4. (a) Topography image of a selected region of the black $CsPbI_3$ /PVDF composite. Piezoelectric force microscopy (PFM) measurements of the polarization reversal process. (b) Amplitude and (c) phase images at +2 V DC voltage bias; (d) amplitude and (e) phase images at -2 V DC voltage bias. (f) Amplitude loop, (g) phase hysteresis loop, and (h) PFM amplitude versus the AC voltage of the black $CsPbI_3$ /PVDF composite at 0 V DC bias.

of the black $CsPbI_3$ @PVDF film were located at lower binding energies compared with those of the yellow composite film, as shown in Figure 2d. This indicates the modified chemical environment of the $[PbI_6]^{4-}$ anion and the weaker Pb–I interaction during the formation of the black-phase $CsPbI_3$ crystals in PVDF. As shown in Figure 2e, the F 1s spectrum originating from the PVDF polymer exhibited a lower binding energy shifting as well for the black $CsPbI_3$ @PVDF film. In addition, the C 1s spectrum in Figure 2f shows two peaks corresponding to $-H-C-H-$ and $-F-C-F-$. The $-F-C-F-$ from the black $CsPbI_3$ @PVDF film showed a lower binding

energy than that of the yellow composite film, confirming that the interaction between $CsPbI_3$ and PVDF leads to the stabilization of the black-phase $CsPbI_3$ perovskite. In the FTIR measurements (Figure S6), compared to the pure PVDF, the slight shifts of $-CF_2$ and the $-CF_2$ symmetrical stretching mode of the black $CsPbI_3$ @PVDF film also indicate the interaction between PVDF and $CsPbI_3$.

Based on the above experimental facts, the schematic illustration in Figure 3 summarizes the potential role of PVDF in the crystal growth and phase stabilization of $CsPbI_3$. It is known that the lone pairs from fluorine atoms in the molecule of

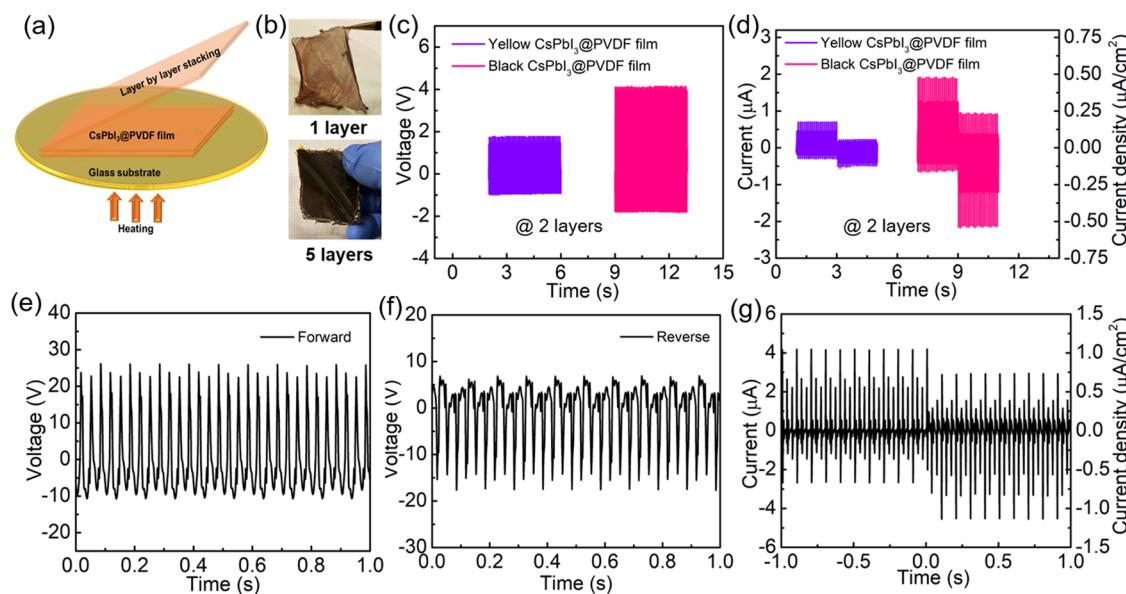


Figure 5. (a) Schematic of fabrication of CsPbI_3 @PVDF composite films with different thicknesses through layer-by-layer stacking. (b) Optical images of one-layer and five-layer CsPbI_3 @PVDF films. Comparison of the output of (c) open-circuit voltage and (d) current between two-layer yellow and black CsPbI_3 @PVDF films. (e) Piezoelectric output voltage of the five-layer black CsPbI_3 @PVDF film-based PENG in the forward connection. (f) Piezoelectric output voltage of the PENG in the reverse connection. (g) Piezoelectric output current of the PENG with obvious current switching.

PVDF determine the conformations of the crystalline PVDF. Each fluorine atom possesses three lone pairs, which offer a large number of coordination centers. At the initial stage, PVDF molecules attract the cations from CsPbI_3 precursors due to the long backbone chain ($-\text{CH}_2\text{--CF}_2-$) and the electronegative $-\text{CF}_2-$ group structure. Compared with the Cs cation, Pb^{2+} has a higher ionic potential,³⁴ which tends to be more easily attracted to the fluorine surface and forms a bond. Then, the positive and negative ions of CsPbI_3 assemble and bond to form the black-phase CsPbI_3 perovskite structure around the $-\text{CF}_2-$ groups. With increasing time, the long-chain PVDF molecule anchored at the surface of CsPbI_3 crystals can further protect the material from air and moisture, increasing the stability of the black-phase CsPbI_3 perovskite. Therefore, the black-phase CsPbI_3 perovskite can still be maintained after 60 days at ambient conditions for the PVDF chemically functionalized CsPbI_3 . While we were preparing the revised manuscript, we noticed a new publication of a yellow δ -phase CsPbI_3 /PVDF-based nanogenerator.³⁵ Based on their first-principles density functional theory (DFT) calculations, it is found that the interaction between CsPbI_3 and PVDF, through the bonding of the F of PVDF with the Cs and Pb within CsPbI_3 , could enhance the polarization and decrease the band gap of CsPbI_3 . Although the authors did not observe the transformation of the yellow-phase CsPbI_3 to the black phase through the interaction between PVDF and CsPbI_3 , this still serves as a strong support for our experimental observation that bondings are formed between PVDF and the CsPbI_3 perovskite. The extra strain induced by PVDF on the CsPbI_3 perovskite prevents its structural change to the yellow δ -phase, leading to the stabilization of the black γ -phase PVDF.

Furthermore, the domain structure and polarization switching behavior of the CsPbI_3 /PVDF composite were investigated using piezoresponse force microscopy (PFM). Figure 4a shows a representative topography image of the black CsPbI_3 /PVDF composite with a square of $1 \times 1 \mu\text{m}^2$. The small grain size in the range of 100–200 nm in the topography image can be attributed

to the compact and uniform distribution of black-phase CsPbI_3 nanoparticles within the PVDF matrix. The amplitude image under a DC voltage bias of 2 V between the tip and the sample is shown in Figure 4b. The PFM signal variation exhibited a well-correlated relation with the grain boundaries, as shown in the topography image (Figure 4a). In addition, the domains coexisting in the same grain also showed distinct amplitude responses (Figure 4b), ruling out the possible artifacts due to grain boundaries or sharp edges. The phase variation image is shown in Figure 4c, showing many downpolarized domains under a 2 V DC bias. As the DC bias was reversed to a negative voltage, the amplitude of the domains did not show a very obvious change (Figure 4d). However, the phase image shows that the domains appeared to be reversed by $\sim 180^\circ$ at -2 V DC bias, indicating an upward polarization. Figure 4f,g show the amplitude loop and hysteresis loop in the phase angle, with a bias voltage of ± 9 V, respectively. The amplitude loop exhibits a characteristic butterfly shape, and the phase angle loop shows a 180° difference, which is consistent with the phase images obtained at reversed DC biases. In addition, amplitude loops and phase hysteresis loops were measured at multiple locations, displaying similar results (Figure S7). The results confirm the presence of ferroelectricity-like behavior in the black CsPbI_3 /PVDF composite thin film. As shown in Figure S8, several characteristic peaks at 613, 763, 976, and 1149 cm^{-1} could be ascribed to the nonpolar α -phase PVDF, which is consistent with the results from the literature.^{33,36} In addition, based on our XRD results, α -phase PVDF could be identified within the black CsPbI_3 @PVDF composite as discussed above (Figure S3). It indicates that the black γ -phase of CsPbI_3 stabilized by PVDF exhibits a ferroelectric nature. Since the amplitude image is proportional to the magnitude of the piezoelectric coefficient (d_{33}), the d_{33} can be calculated by the equation³⁷ $d_{33} = \frac{A}{U} = \frac{V\delta}{U}$, where A is the amplitude, V is the vertical deflection signal of the cantilever (mV, 16 times gain), δ is the tip sensitivity (103 nm/V), and U is the amplitude of the AC voltage. Figure 4h shows the AC voltage versus piezoresponse of the black CsPbI_3 /PVDF

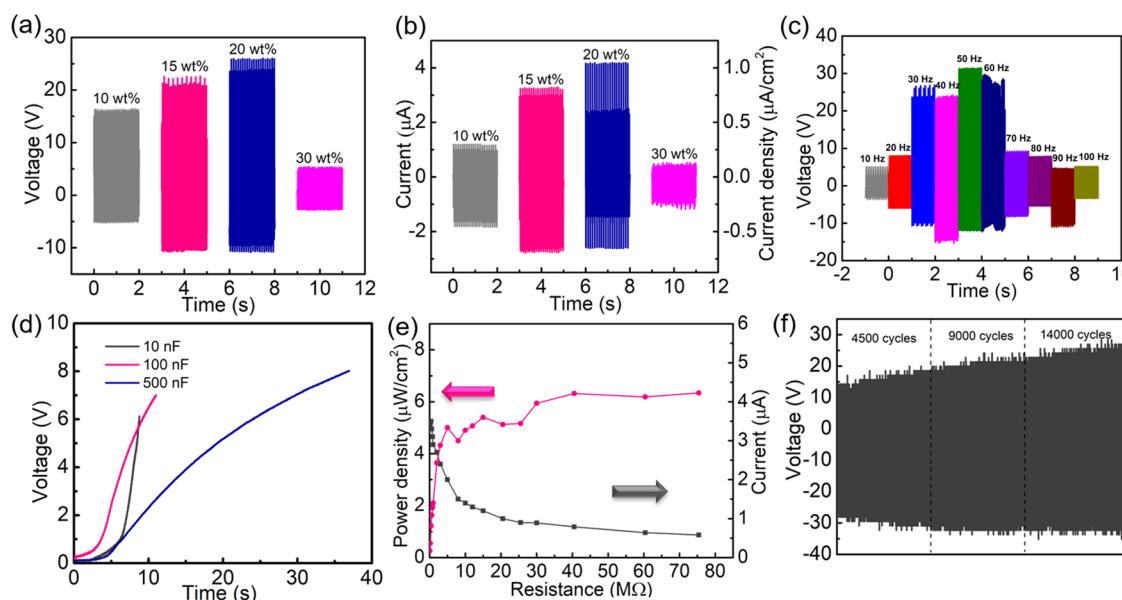


Figure 6. (a) Voltage output and (b) current output of the five-layer black-phase CsPbI₃@PVDF composite film with different mass ratios of CsPbI₃ in PVDF. (c) Output performance of the piezoelectric nanogenerator (five-layer black-phase 20 wt % CsPbI₃@PVDF) as a function of different frequencies. (d) Measured output voltage across the various commercial capacitors of 10, 100, and 500 nF charged by a PENG. (e) Current output and power density at various resistance loadings. (f) Stability of the PENG tested under a frequency of 30 Hz for 14 000 cycles.

composite. The slope was fitted with a linear function, and the value is shown in Figure 4h. The piezoelectric coefficient was estimated to be 28.4 pm/V, which is similar to the value of BaTiO₃ nanoparticles (28 pm/V)³⁸ and comparable to that of the FAPbI₃ nanoparticle.¹⁷ The variation in the dielectric constant with frequency for the black CsPbI₃/PVDF composite was observed in the frequency range 1–100 kHz at room temperature (shown in Figure S9a). It is observed that the dielectric constant decreases as the frequency increases, which is a well-known behavior of dielectric materials.^{39,40} The dielectric constant of the black CsPbI₃/PVDF composite was 17.6 at 1 kHz, which is higher than that of the pure PVDF.⁴¹ The electric field-dependent leak current of the black CsPbI₃/PVDF composite is shown in Figure S9b. The leakage current is less than 10^{-6} A over the electrical field range of ± 100 kV/cm. Therefore, the black CsPbI₃/PVDF composite films with high dielectric constant and low leakage current are beneficial for energy harvesting applications.

To explore the piezoelectric output performance, piezoelectric nanogenerators were fabricated. The thickness of the composite films can be well controlled through a layer-by-layer stacking method, as illustrated in Figure 5a. Figure 5b shows the optical image of a one-layer CsPbI₃@PVDF thin film with semitransparency. In the cross-section SEM image, the thickness was ~ 1.5 μ m, as shown in Figure S10a. However, the five-layer film showed a darker color and a flat surface with no obvious air bubbles, suggesting that the fabricated film has a good quality and interfacial contact. The cross-section SEM images of the five-layer film (Figure S10b) showed a compact film with a thickness of ~ 10 μ m. To explore the piezoelectric performance between the yellow-phase and the black-phase CsPbI₃, two-layer CsPbI₃@PVDF composite films with different CsPbI₃ phases were fabricated as PENGs. It can be clearly seen that the black-phase CsPbI₃@PVDF composite-based PENG exhibits more than 2 times higher output on the voltage and current density, as shown in Figure 5c,d. This suggests that the black-phase CsPbI₃-based PVDF composite exhibits better performance. To further

increase the performance, five-layer black-phase CsPbI₃@PVDF composite film-based PENGs were fabricated. Under a 2.7 N applied force at a frequency of 30 Hz, the voltage output under a forward bias can reach a value of 26 V, which is much better than that of the two-layer black-phase-based PENG (4.1 V), as shown in Figure 5e. Switching-polarity tests were also carried out to verify the generated output signals originating from the piezoelectric phenomenon. A reverse connection was made, and an opposite output signal was measured, as shown in Figure 5f. It is obvious that the electric signals are reversible, indicating that the output signals are generated from the PENG strained by the electrodynamic shaker. In addition, a difference in the voltage peak values was observed, which could be attributed to the difference in the force in the process of applying and releasing force on the piezoelectric nanogenerator. Figure S11 shows the voltage and current output of the black CsPbI₃@PVDF-based nanogenerator at different applied forces. When the applied force decreased from 2.7 to 1.1 N, the piezoelectric outputs of voltage and current decreased gradually. When the applied force was lower than 1 N, the output voltage and current exhibited significant decreases, which indicates the lower sensitivity of the applied force as a force below 1 N. Figure 5g displays the current output with a peak current of 4.5 μ A, corresponding to a 1.1 μ A/cm². The switching output could also be observed while changing the connection, which confirms that the generated output is from the piezoelectric phenomenon.

To optimize the performance, the black-phase CsPbI₃@PVDF composite films with different CsPbI₃ mass ratios were also investigated. As shown in Figure 6a, for CsPbI₃ with mass ratios of 10, 15, 20, and 30 wt %, the voltage outputs of the device were 16, 22, 26, and 5.5 V, respectively, indicating that the PENG with CsPbI₃ with a mass ratio of 20 wt % exhibited the best output. As shown in Figure 6b, a similar trend was observed for current density, with mass ratios of 10, 15, 20, and 30 wt % exhibiting 0.45, 0.8, 1.1, and 0.28 μ A, respectively. The frequency dependence was also investigated within a range from 10 to 100 Hz, as shown in Figure 6c. It is of great

importance to study the relationship between the output performance of the piezoelectric nanogenerator under different frequencies because the mechanical energy from the ambient environment can be varied significantly and irregularly.⁴² The test was performed on the five-layer black-phase 20 wt % CsPbI₃ in PVDF composite-based PENG. As the frequency increased, the piezoelectric output increased, and the output voltage reached a peak value of 31 V at 50 Hz, as shown in Figure 6c. In addition, the voltage output can display high performance in a relatively large frequency range from 30 to 60 Hz without obvious degradation. On further increasing the frequency in the range of 70–100 Hz, the voltage output was much lower and decreased slightly with the increase in frequency. To demonstrate an example of the practical application of the CsPbI₃@PVDF composite-based PENG, the output signals generated by an electrodynamic shaker were directly used for charging capacitors, as shown in Figure 6d. The generated output was rectified through a full-wave bridge rectifier circuit. The capacitors can be gradually charged up with different rates depending on the capacitance. Higher capacitance results in a longer charging time of the capacitor. For a 100 nF capacitor, it only takes 10 s to reach 7 V, while 30 s is needed to charge a 500 nF capacitor to 7 V. Figure 6e shows the output power of the five-layer black-phase 20 wt % CsPbI₃@PVDF composite-based PENG as a function of load resistance. As the resistance increases, the output current gradually decreases. The peak output power was calculated to be 25 μ W at a load of 40.5 M Ω , corresponding to a peak power density of 6.3 μ W/cm². The cycle stability was also measured, as shown in Figure 6f. It was noticed that the output voltage has a tendency to increase with the increase of the number of cycles, which could be caused by the charge accumulation from the incomplete discharge during the cyclic charging and discharging processes. The charge accumulated in the previous process had not fully discharged, and the next cycle had already started, resulting in continuously increasing voltage output.⁴³ After 14 000 cycles, the black-phase CsPbI₃@PVDF composite-based PENG could still maintain a good output without a noticeable decrease, confirming the good stability and durability of the fabricated device.

CONCLUSIONS

We have described a novel approach to fabricate the black γ -phase CsPbI₃ perovskite using PVDF. The polymer not only works as a matrix to form the CsPbI₃@PVDF composite but also passivates the surface of CsPbI₃ through the interaction between the $-CF_2-$ structure and CsPbI₃. The obtained black-phase CsPbI₃@PVDF composite film was stable under ambient conditions for over 60 days and no obvious degradation under 80 °C for over 24 h. Piezoresponse force spectroscopy measurements suggest that the black CsPbI₃/PVDF composite contains well-developed ferroelectric properties with a high piezoelectric charge coefficient (d_{33}) of 28.4 pm/V. The piezoelectric performance between the yellow-phase and black-phase-based CsPbI₃@PVDF composite was also compared. The output signals of the black composite are more than 2 times higher than those from the yellow one. A layer-by-layer stacking method was adopted to fabricate composite films with a tunable thickness. The five-layer black-phase 20 wt % CsPbI₃@PVDF composite-based PENG can generate a voltage and current density output of 26 V and 1.1 μ A/cm², respectively. The generated output signals from the nanogenerator can be used to charge capacitors. The output power can reach a peak value at 25 μ W at a load of 40.5 M Ω . In addition, the fabricated

PENG has excellent durability with no obvious degradation under 14 000 cyclic tests. This PVDF-stabilized CsPbI₃ method could be adopted as a new method for other functional applications. This simple, cost-effective solution process is feasible for the fabrication of large-scale high-performance all-inorganic perovskite composite-based piezoelectric nanogenerators with good stability and durability.

EXPERIMENTAL METHODS

Material Synthesis. *Synthesis of CsPbI₃ NPs.* Cesium iodide (CsI; 259 mg) and 462 mg of lead iodide (PbI₂) were dissolved in 25 mL of dimethylformamide (DMF) with 0.75 mL of *n*-octylamine (OTA) and 5 mL of oleic acid (OA). Then, the mixture was heated at 60 °C for 60 min while stirring. The mixture was dropped into 250 mL of toluene with vigorous stirring for 5 min. After centrifugation and purification, the CsPbI₃ nanocrystals were collected for further use.

Fabrications of CsPbI₃@PVDF Composite Films. CsPbI₃ NPs were homogeneously mixed with a PVDF solution (10 wt % in DMF) in different ratios of 10, 20, and 30 wt % and stirred at room temperature overnight. The as-mixed composite was spin-coated (500 rpm, 30 s) on a glass substrate and dried at different temperatures (50, 70, 130, and 180 °C) to control the final CsPbI₃ phases. The solid CsPbI₃@PVDF composite film was peeled off from the glass substrate for further characterization and device fabrication. The black-phase CsPbI₃@PVDF films were prepared at 180 °C for further characterization unless otherwise specified.

Device Fabrication. To tune the thickness of the CsPbI₃@PVDF composite film, multiple thin films were stacked on each other and annealed at 170 °C under vacuum for 5 min to eliminate air gaps. The composite films were polarized at room temperature with an electric field of ~300 kV/cm for the black CsPbI₃@PVDF composite and 700 kV/cm for the yellow CsPbI₃@PVDF composite for 2 h. Then, the composite film was sandwiched between two patterned copper electrodes on the laminating pouches. The sandwiched structure of copper/CsPbI₃@PVDF/copper was pressed with a commercial thermal laminator to encapsulate the device. The thickness of the CsPbI₃@PVDF composite film was controlled by stacking different layers of thin films.

Measurements and Characterizations. The crystalline structures and phases of the thin films were characterized by X-ray diffraction (D8 Discover, Bruker) and Fourier transform infrared spectroscopy (Nicolet iS50, Thermo), respectively. Scanning electron microscopy (SEM) and energy-dispersive X-ray spectroscopy (EDS) were conducted by a field emission SEM (JSM 7200F). X-ray photoelectron spectroscopy (XPS) was performed with an Mg K α ($h\nu = 1254.6$ eV) X-ray source using a VSW HA100 photoelectron spectrometer. UV-vis spectra were obtained using a UV-2501PC (Shimadzu). Piezoelectric force microscopy measurements (Dimension Icon, Bruker) were taken in the contact and vertical mode with an AC voltage bias applied to the conductive AFM tip and the bottom electrode grounded. The piezoelectric nanogenerator performance was tested using an electrodynamic shaker (Lab Works Inc.). More details on the device measurements could be found in our previous reports.^{19,44,45} The applied mass was 138 g at an acceleration of 2 G. The calculated force (F) was around 2.7 N based on the equation of $F = MA$. If not specified, all of the measurements were tested at a force of 2.7 N. By controlling the mass loading and the acceleration, the electromechanical response along with differ-

ent loadings could be analyzed. The voltage and current were measured with a digital oscilloscope (Tektronix 2004C) and a low-noise current preamplifier (model SR 570, Stanford Research System Inc.), respectively.

■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acsomega.2c00091>.

Optical images and XRD patterns of $\text{CsPbI}_3/\text{PVDF}$ composite films prepared at different temperatures, FTIR measurements of the composite films, SEM images of the black $\text{CsPbI}_3/\text{PVDF}$ composite, piezoelectric nanogenerator performance of the black $\text{CsPbI}_3/\text{PVDF}$ composite at different applied forces ([PDF](#))

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Notes

The authors declare no competing financial interest.

The data that support the findings of this study are available from the corresponding author upon reasonable request.

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