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Propionamide (C₂H₅CONH₂): The Largest Peptide-like Molecule in Space

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Abstract

Peptide bonds, as the molecular bridges that connect amino acids, are crucial to the formation of proteins. Searches and studies of molecules with embedded peptide-like bonds are thus important for the understanding of protein formation in space. Here we report the first tentative detection of propionamide ($C_2H_5CONH_2$), the largest peptidelike molecule detected in space toward Sagittarius B2(N1) at a position called N1E that is slightly offset from the continuum peak. New laboratory measurements of the propionamide spectrum were carried out in the 9-461 GHz range, which provide good opportunity to check directly for the transition frequencies of detected interstellar lines of propionamide. Our observing result indicates that propionamide emission comes from the warm, compact cores in Sagittarius B2, in which massive protostellars are forming. The column density of propionamide toward Sgr B2(N1E) was derived to be 1.5×10^{16} cm⁻², which is three-fifths of that of acetamide, and one-nineteenth of that of formamide. This detection suggests that large peptide-like molecules can form and survive during star-forming process and may form more complex molecules in the interstellar medium (ISM). The detection of propionamide bodes well for the presence of polypeptides, as well as other complex prebiotic molecules in the ISM.

Unified Astronomy Thesaurus concepts: Astrochemistry (75); Spectral line identification (2073); Spectroscopy (1558)

Supporting material: data behind figures, machine-readable tables

1. Introduction

Proteins, as the building blocks of living cells, are an essential component of living systems on Earth. Proteins are polymers of amino acids joined together by the peptide bond, -NHCO-. Due to the high molecular weight and extremely low gas-phase abundance, detection of proteins in the interstellar medium (ISM) at the current stage of development of observation facilities looks like a formidable task. Therefore, molecules with peptide-like bonds (Figure 1) are of particular interest for our understanding of possible routes of protein formation in space (Halfen et al. 2011; Kaiser et al. 2013; Belloche et al. 2017).

The number of peptide-like molecules found so far in space is quite limited. The simplest molecule containing a peptidelike bond, formamide (NH₂CHO), was detected in Sagittarius B2 (Sgr B2) in the 1970s (Rubin et al. 1971), and later in

various other sources, including the solar-type protobinary IRAS 16293-1622 (Coutens et al. 2016), the protostellar shock region L1157-B1 (Mendoza et al. 2014), comet 67P/Churyumov-Gerasimenko (Altwegg et al. 2017), and so on. About 30 yr later, its homologue (homologue means molecules that have the same functional group -NHCO-, but with additional methyl group, CH₃-), acetamide (CH₃CONH₂), was detected in Sgr B2(N) (Hollis et al. 2006; Halfen et al. 2011). N-methylformamide (CH₃NHCHO) and urea (NH₂C(O)NH₂) were also identified in Sgr B2(N) (Belloche et al. 2017, 2019). Urea was also detected in a quiescent molecular cloud in the Sgr B2 complex (Jiménez-Serra et al. 2020). Recently, a number of the amide molecules, including NH₂CHO, CH₃CONH₂, and CH₃NHCHO, were detected in massive cores NGC 6334I (Ligterink et al. 2020). These observations suggest that peptide-like molecules might be widespread in space. Whether larger molecules with a peptide-like bond could



Figure 1. Generalized structures and models of formamide, acetamide, N-methylformamide, urea, and propionamide. (a) NH_2CHO . (b) CH_3CONH_2 . (c) CH_3NHCHO . (d) NH_2CONH_2 . (e) $C_2H_5CONH_2$. The far left panel shows a peptide bond connecting two amino acids, where R1 and R2 are so-called side chains. Peptide bonds are highlighted with red dashed squares. Red arrow denotes hindered internal rotation (torsion) of methyl group; blue arrow stands for skeletal torsion motion around corresponding C–C bond. C, H, N, and O atoms are indicated by gray, small light gray, blue, and red spheres, respectively.

form in the ISM is still an open question. In this context, the search for more complex interstellar peptide-like molecules is a key to understanding the growth of peptides.

Propionamide ($C_2H_5CONH_2$), a complex alkyl amide with 12 atoms (Figure 1) (Marstokk et al. 1996), is the next member of the amide chemical family after formamide and acetamide. Various linear and cyclic aliphatic amides, such as monocarboxylic acid amides, dicarboxylic acid monoamides, hydroxy acid amides, carboxy lactams, lactims, and N-acetyl amino acids, have been found in the Murchison meteorite (Cooper & Cronin 1995). Specifically, propionamide was positively identified in the Murchison meteorite (Cooper & Cronin 1995), suggesting an interstellar origin of this molecule. Halfen et al. (2011) suggested an interstellar search for propionamide. They pointed out that the identification of this species in interstellar gas will be challenging, especially given the high density of rotational levels. However, continuous upgrades to observation facilities open new horizons for such searches.

Sgr B2, the massive star-forming region located close to the Galactic Center, has been found to be one of the best hunting grounds for complex organic molecules in the ISM (Li et al. 2017, 2020). Many prebiotic molecules have been detected in this source, such as the branched molecule $i-C_3H_7CN$ (Belloche et al. 2014) and the chiral molecule CH_3CHCH_2O (McGuire et al. 2016). Sgr B2 contains two main sites of star formation, Sgr B2(N) and Sgr B2(M), both of which host several dense, compact, hot cores (Bonfand et al. 2017).

In this paper, we report the tentative detection of $C_2H_5CONH_2$ toward Sgr B2(N). The new laboratory spectroscopic study of propionamide spectrum is discussed in Section 2. The details of the observations and data reduction are described in Section 3. The observation results are presented in Section 4, along with the derivation of column

densities of other related molecules of interest. The possible formation mechanisms of propionamide are discussed in Section 5. The results are summarized in Section 6.

2. Spectroscopy

As a preparatory step of our ISM search, we performed a new laboratory spectroscopic study of the propionamide microwave spectrum. Previously available data (Marstokk et al. 1996) were limited in the 21.4–39 GHz range, and extrapolations to the 84–114 GHz range, where we performed our astronomical search, cannot be taken as reliable. The new laboratory measurements of the propionamide spectrum were carried out in the 9–461 GHz range, providing a good opportunity to check directly for the transition frequencies of detected interstellar lines of propionamide. The rotational transitions belonging to the ground (v = 0), first excited skeletal torsion (v = 1), and first excited methyl torsion vibrational states were analyzed providing an accurate sets of constants for spectral predictions.

From a spectroscopic point of view, propionamide is a molecule with methyl torsional large amplitude motion (red arrow in Figure 1(e)) hindered by a relatively high potential barrier (~760 cm⁻¹). There are three factors that complicate analysis of its rotational spectrum. First, the barrier is rather high, and even in the first excited methyl torsion state $v_{29} = 1$, A - E splittings due to large amplitude methyl torsion motion are of the order of several MHz only. In the ground state, the A - E splittings are resolved mainly for high K_a transitions. Thus, there is not much information on the methyl torsion motion det methyl torsion approximation of the molecule and the methyl torsional parameters in the Hamiltonian model are highly correlated. Second, there is a low-lying skeletal torsion vibration mode v_{30} in this molecule (blue arrow in Figure 1(e)), which is very likely to interact both with the

ground vibrational state and methyl torsion mode v_{29} . Finally, there are additional hyperfine splittings due to the nonzero electric quadrupole moment of the nitrogen atom, which by order of magnitude are comparable in many cases with methyl torsion A - E splittings in this molecule. The presence of two low-lying vibrational modes (skeletal torsion v_{30} with fundamental frequency of $45(7) \text{ cm}^{-1}$ (Marstokk et al. 1996), and methyl torsion v_{29} with fundamental frequency of 179.6 cm⁻¹) and A-E methyl torsion plus quadrupole splittings result in a rather congested spectrum in the microwave range.

We started our analysis of propionamide spectrum from the results of Marstokk et al. (1996) where the data in the 21.4-39 GHz range were obtained using Stark modulated spectroscopy. In order to expand the covered frequency range, the new measurements of the propionamide spectrum were performed in three laboratories. Measurements in the 9.8 $-25.6 \,\mathrm{GHz}$ range were performed at the National Institute of Standards and Technology (Gaithersburg, USA) using the molecular beam Fourier transform microwave spectrometer (Suenram et al. 1989). Measurements in the 49–149 GHz range were done using the automated millimeter wave spectrometer of the Institute of Radio Astronomy of NASU (Kharkiv, Ukraine) (Alekseev et al. 2012). Measurements between 150 and 460 GHz were done using the submillimeter wave spectrometer of PhLAM Laboratory at the University of Lille (France) (Zakharenko et al. 2015).

Analysis of the spectra was performed using the Rho axis method (RAM; Hougen et al. 1994) and the RAM36hf program (Ilyushin et al. 2010; Belloche et al. 2017). The RAM approach allows simultaneous fitting of rotational transitions of A and Esymmetry species in a number of methyl torsion excited states. The transitions belonging to the ground $(v_{29}, v_{30}) = (0, 0)$ and first excited methyl torsion $(v_{29}, v_{30}) = (1, 0)$ states were analyzed simultaneously as the first step of our study. The transitions of the $(v_{29}, v_{30}) = (1, 0)$ state were included mainly with the purpose of better constraining torsional parameters of our RAM Hamiltonian model. In the ground state $(v_{29},$ $v_{30} = (0, 0)$ (denoted as v = 0 in the text), the rotational transitions up to $K_a = 10$, and J up to 50 were included in the fit. Higher K_a (>10) transitions were excluded from consideration because of quite significant deviation from our model presumably due to intervibrational interactions with low-lying skeletal torsion mode $(v_{29}, v_{30}) = (0, 1)$ in this molecule. For the same reason we limited our fitted data set for the first excited methyl torsion state to $K_a \leq 4$ transitions. Such constraint seems reasonable since lines corresponding to relatively low K_a transitions in the ground vibrational state $(v_{29}, v_{30}) = (0, 0)$ are most likely to be detected in the ISM.

Despite the fact that transitions belonging to the $(v_{29}, v_{30}) = (1, 0)$ state were included in the fit, the main torsional parameters remained highly correlated and we had to fix the *F* parameter in our RAM Hamiltonian. We fixed the *F* parameter at 5.55 cm⁻¹, a value that corresponds to methyl top moment of inertia $I_{\alpha} \approx 3.2$ a.m.u.Å². The final data set for the $(v_{29}, v_{30}) = (0, 0), (1, 0)$ states contains 5495 transitions, which due to blending correspond to 2271 measured line frequencies. This data set includes our new measurements as well as data for $(v_{29}, v_{30}) = (0, 0)$ from Marstokk et al. (1996). We have not included in the fit the data on eight rotational transitions from (Marstokk et al. 1996) that correspond to the $(v_{29}, v_{30}) = (1, 0)$ state, since those lines deviate significantly from our current model. Whereas we cannot rule out completely the presence of some

intervibrational interactions for these lines, at present we think that these lines were misassigned in Marstokk et al. (1996). The RAM Hamiltonian model included 26 parameters (with one fixed parameter F). The final set of molecular parameters is presented in Table 1. The overall rms deviation of the fit was 0.035 MHz (corresponding to weighted rms deviation of 0.86).

When the first positive results were obtained in our search for interstellar propionamide we decided to extend our analysis to the $(v_{29}, v_{30}) = (0, 1)$ state (denoted as v = 1 in the text). This state has rather low excitation energy $(45(7) \text{ cm}^{-1}; \text{ Marstokk})$ et al. 1996), and thus, it was quite probable that rotational transitions belonging to this state may be also detected in the ISM. As for the ground vibrational state, we performed an analysis for the $(v_{29}, v_{30}) = (0, 1)$ state simultaneously with the corresponding state where one quanta of methyl torsion and one quanta of skeletal torsion were excited $(v_{29}, v_{30}) = (1, 1)$. This was done again to provide more information on the methyl torsion part of the problem. It appeared that excitation of the skeletal torsion vibration increases the barrier to methyl torsion motion (see Table 1 for comparison of V₃ values). It is evident from the fact that much smaller A - E splittings are observed for the same rotational transitions in the first excited skeletal torsion state $(v_{29}, v_{30}) = (0, 1)$ in comparison with the ground state $(v_{29}, v_{30}) = (0, 0)$. For example, the A - E splitting for the $4_{4,0} \leftarrow 3_{3,1}$ transitions in the ground vibrational state is about 2 MHz, whereas in the first excited skeletal torsion state this splitting is only 0.210 MHz. Thus many transitions for which A - E splittings were resolved in the ground vibrational state $(v_{29}, v_{30}) = (0, 0)$ appeared to be unresolved in the $(v_{29}, v_{30}) = (0, 0)$ $v_{30} = (0, 1)$ state, and therefore, the amount of information about the methyl torsion motion encoded in the A - E splittings of rotational transitions in the first excited skeletal torsion state was even less than in the case of the ground vibrational state. This forced us to fix in the fit for the first excited skeletal torsion state not only the F parameter, but also the ρ parameter, because if we varied the ρ parameter the fit provided a value that corresponded to an unphysically low value of methyl top moment of inertia I_{α} . The F and ρ parameters thus were fixed at the corresponding values from the ground state fit (see Table 1). The final data set for the $(v_{29}, v_{30}) = (0, 1), (1, 1)$ fit contains 4017 transitions with J up to 30, which due to blending correspond to 1027 measured line frequencies. The data set includes our new measurements in the 49-146 GHz frequency range as well as data for the $(v_{29}, v_{30}) = (0, 1)$ state from Marstokk et al. (1996). The RAM Hamiltonian model included 25 parameters for the first excited skeletal torsion state (two parameters were fixed). This parameter set is also given in Table 1. The overall rms deviation of the fit was 0.046 MHz (corresponding to weighted rms deviation of 1.35).

Using the parameters of our final fits, we calculated a list of propionamide transitions in the ground $(v_{29}, v_{30}) = (0, 0)$ and $(v_{29}, v_{30}) = (0, 1)$ states for astronomical use. We used dipole moment components obtained by Marstokk et al. (1996) via Stark measurements $\mu_a = 0.6359(60)$ D and $\mu_b = 3.496(30)$ D, which were rotated from the principal axis system to the rho axis system of our Hamiltonian model. The list includes information on transition quantum numbers, transition frequencies, calculated uncertainties, lower state energies, and transition strengths. We limit our predictions to the transitions with calculated uncertainties lower than 0.1 MHz. It should be noted that calculated uncertainties for transition frequencies cannot predict possible systematic errors due to intervibrational interactions between

	Table 1		
Molecular Parameters of Pr	opionamide Obtained	with the RAM36hf	Program

ntr ^a	Parameter	Operator ^b	$v = 0^{c}$	$v = 1^{c}$
220	F	p_{α}^2	5.55 ^d	5.55 ^d
220	V_3	$\frac{1}{2}(1-\cos 3\alpha)$	759.46(16)	1042.8(14)
211	ρ	$J_z p_{\alpha}$	0.054475(70)	0.054475 ^d
202	$A_{\rm RAM}$	\overline{J}_z^2	0.317819(34)	0.31562(33)
202	$B_{\rm RAM}$	J_x^2	0.138258(34)	0.13829(31)
202	C_{RAM}	J_y^2	0.0949374576(41)	0.096130(29)
202	D_{ab}	$\{J_z, J_x\}$	- 0.047201(65)	- 0.04556(63)
440	F_m	P_{lpha}^4		$-0.3270(50) \times 10^{-2}$
431	$ ho_m$	$J_z p_{lpha}^3$		$0.499(21) \times 10^{-3}$
422	F_J	$J^2 p_{lpha}^2$		$0.844(30) \times 10^{-4}$
422	V_{3J}	$J^2(1-\cos 3\alpha)$	$-0.59516(20) imes 10^{-3}$	$0.234(10) \times 10^{-1}$
422	V_{3K}	$J_z^2(1-\cos 3\alpha)$	$-0.191946(67) \times 10^{-2}$	$-0.341(28) imes 10^{-2}$
422	V_{3bc}	$(J_x^2 - J_y^2)(1 - \cos 3\alpha)$	$-0.14119(18) \times 10^{-3}$	$-0.1125(15) \times 10^{-3}$
404	Δ_J	$-J^4$	$0.28990(20) \times 10^{-7}$	$0.128(12) \times 10^{-7}$
404	Δ_{JK}	$-J^2 J_z^2$	$0.4182(37) \times 10^{-7}$	$0.299(11) \times 10^{-6}$
404	Δ_K	$-J_z^4$	$0.30071(53) \times 10^{-6}$	$0.4772(67) \times 10^{-6}$
404	δ_J	$-2\{J^2, (J_x^2 - J_y^2)\}$	$0.76905(88) imes 10^{-8}$	$0.5872(90) imes 10^{-8}$
404	δ_K	$-\{J_z^2, (J_x^2 - J_y^2)\}$	$-0.100211(26) imes 10^{-6}$	$-0.8083(73) imes 10^{-7}$
642	D_{6bc}	$\frac{1}{2}\sin 6\alpha \{J_x, J_y\}$		$-0.1767(33) \times 10^{-2}$
642	V_{6J}	$J^2(1-\cos 6\alpha)$		$-0.1135(47) imes 10^{-1}$
624	V_{3JJ}	$J^4(1 - \cos 3\alpha)$	$0.545(31) imes 10^{-9}$	
606	Φ_J	J^6	$0.5310(38) \times 10^{-12}$	
606	Φ_{JK}	$J^4 J_z^2$	$-0.12542(88) \times 10^{-10}$	$0.597(25) \times 10^{-11}$
606	Φ_{KJ}	$J^2 J_z^4$	$0.1464(47) \times 10^{-10}$	$-0.3293(52) \times 10^{-10}$
606	ϕ_J	$2J^4(J_x^2-J_y^2)$	$0.2536(19) \times 10^{-12}$	
606	ϕ_{JK}	$J^{2}\{J^{2}_{z}, (J^{2}_{x} - J^{2}_{y})\}$	$-0.3479(28) imes 10^{-11}$	
606	ϕ_K	$\{J_z^4, (J_x^2 - J_y^2)\}$	$0.453(19) imes 10^{-11}$	$-0.533(33) imes 10^{-11}$
808	L_{KKJ}	$J^{2}J_{z}^{6}$	$-0.2497(57) imes 10^{-13}$	
808	l_K	$\{J_z^6, (J_x^2 - J_y^2)\}$	$0.1454(58) imes 10^{-13}$	
	χ_{aa}		$0.67790(56) imes 10^{-4}$	$0.639(30) imes 10^{-4}$
	χ_{bb}		$0.64887(60) imes 10^{-4}$	$0.644(17) imes 10^{-4}$

Notes.

 $a^n = t + r$, where n is the total order of the operator, t is the order of the torsional part and r is the order of the rotational part, respectively.

^b {*A*, *B*} = *AB* + *BA*. The product of the operator in the third column of a given row and the parameter in the second column of that row gives the term actually used in the torsion-rotation Hamiltonian of the program, except for *F*, ρ and *A*_{RAM}, which occur in the Hamiltonian in the form $F(p_{\alpha} - \rho P_{a})^{2} + A_{RAM}P_{a}^{2}$.

^c All values are in cm⁻¹ (except ρ which is unitless). Statistical uncertainties are shown as one standard uncertainty in the units of the last two digits. v = 0 correspond to $(v_{29}, v_{30}) = (0, 0), (1, 0)$ data set, v = 1 correspond to $(v_{29}, v_{30}) = (0, 1), (1, 1)$ data set.

^d Fixed value.

methyl torsion and skeletal torsion modes in the molecule, since those interactions are not taken explicitly into account by our current Hamiltonian model. That is why we further limit our predictions to the rotational quantum number ranges where we do not see in our laboratory study significant deviations from our model. The predictions are made up to 460 GHz and up to J = 40for the ground vibrational state (v_{29} , v_{30}) = (0, 0) and up to 150 GHz and up to J = 30 for the (v_{29} , v_{30}) = (0, 1) state. For both states only the transitions with $K_a < 10$ are given to minimize possible errors due to intervibrational interactions between the ground state and the stack of skeletal torsion excited states $v_{30} = 1$, 2. To exclude all doubts in rest frequencies of the propionamide transitions detected in the ISM, we have checked against laboratory experimental records all clean and partially blended lines presented in this paper.

Along with transition frequency predictions, we also provide tabulated values of the partition function for propionamide (see

	Table 2	
Rotational and	Vibrational Partition Functions of Propionamide at Various	
	Temperatures	

	0 (7)	O(T)
I (K)	$Q_r(I)$	$Q_{\nu}(I)$
300.000	509,564.2	21.422
225.000	330,971.7	8.915
150.000	180,158.2	3.903
75.000	63,695.5	1.796
37.500	22,519.8	1.218
18.750	7961.9	1.033
9.375	2815.0	1.001

Table 2). The rotational $Q_r(T)$ and vibrational $Q_v(T)$ parts of the partition function were calculated using standard expressions:

$$Q_r(T) = g_i g_N \sqrt{\frac{\pi}{ABC} \left(\frac{kT}{h}\right)^3}$$
(1)



Figure 2. Integrated intensity map of C₂H₅CONH₂ 5_{5,*} - 4_{4,*} transition ($E_u = 12.64$ K) at 92.143 GHz in Sgr B2(N1) (the $J_{K_a,*}$ notation stands for the two levels with the same *J*, K_a but different K_c quantum numbers ($K_c = J - K_a$ and $K_c = J - K_a + 1$) which are degenerate). The integrated intensity map of C₂H₅CONH₂ is shown in color scale. The integrated intensity map of CH₃¹³CN is shown in contours, which represent 1, 3, 5, 7, 9, 11, 15, 19, 23, 27, 31, and 35 Jy beam⁻¹. The yellow circle indicates the position of Sgr B2(N1E). The white ellipse shows the size of the respective synthetic beam. The crosses indicate the position of Sgr B2(N1) and Sgr B2(N1S). The integrated frequency range of C₂H₅CONH₂ ranges from 92,142–92,143.5 MHz.

$$Q_{\nu}(T) = \prod_{i=1}^{3N-6} \frac{1}{1 - e^{-E_i/kT}}.$$
 (2)

The total partition function is thus $Q = Q_r Q_\nu$. In Equation (1), *A*, *B*, and *C* are rotational constants in the principal axes system, g_N and g_i are degeneracy factors taking into account, respectively, the tripling of energy levels due to nuclear quadrupole coupling of nitrogen, and the doubling of energy levels due to hindered methyl top internal rotation. For the calculation of the vibrational partition function, we used the excitation energies E_i of all normal modes of propionamide up to 1000 cm^{-1} . The energies of the two lowest modes ν_{30} and ν_{29} were taken respectively from the study by Marstokk et al. (1996), and from this study as a result of global modeling of the methyl torsion potential function based on the parameters obtained from RAM36hf fit. The excitation energies of all other modes were taken from the study of the vibrational spectrum of propionamide by Kuroda et al. (1972).

The data sets treated in this work as well as the predictions produced are available as electronic supplement with this paper. Part of them are presented in Tables A1, A2, A3, and A4 in the Appendix. Transitions for the ground state $(v_{29}, v_{30}) = (0, 0)$ and the first excited skeletal torsion state $(v_{29}, v_{30}) = (0, 1)$ are given separately.

3. Observations and Data Reduction

3.1. IRAM 30 m Telescope

The observations presented in this paper are part of a mapping observations of molecular lines at 3 and 2 mm wavelength toward



Figure 3. Fluxes of the subtracted continuum emission for the identified lines toward Sgr B2(N1E). The red circles denote fluxes subtracted from the v = 0 clean lines. The blue and yellow circles demote fluxes subtracted from the v = 0 and v = 1 partially blended lines, respectively. The error bars stand for uncertainties of fluxes. The overall flux dependence on the frequencies is well consistent with the spectrum energy distribution that is expected by free–free emission, which justifies our continuum removal results.



Figure 4. Transition of C¹⁸O that is identified in the ALMA spectrum of Sgr B2(N1E). The black line shows continuum-subtracted spectrum observed with the ALMA telescope, while red line shows the synthetic spectra of C¹⁸O. The dashed line denotes the 3σ levels.

Sgr B2 in 2019 May with the Institut Radio Astronomie Millimétrique (IRAM) 30 m telescope on Pica Veleta, Spain (project ID: 170-18, PI: Feng Gao). The observing center is Sgr B2(N) $(\alpha_{J2000} = 17^{h}47^{m}20.0^{s}, \delta_{J2000} = -28^{\circ}22'16'')$. Observations were conducted in position-switching mode. The off position of $(\delta \alpha, \delta \beta) = (-752'', 342'')$ was used (Belloche et al. 2013). We use the 3 mm band of the EMIR receiver and FTS backend, with a broad bandpass (7.8 GHz bandwidth), to cover the frequency range of 82.3-90.1 GHz with a frequency resolution of 0.195 MHz (velocity resolution of 0.641 km s⁻¹ at 3 mm). The 1σ rms T_A^* in the line free channels is 4-8 mK at 3 mm. The system temperature ranges within 90–150 K. The integration time is 98 minutes for Sgr B2(N). The telescope pointing was checked every ~ 2 hr on nearby quasar 1757-240. The telescope focus was optimized on 1757-240 at the beginning of the observation. The data reduction was conducted using the GILDAS software package²⁰, including CLASS, GREG, and Weeds (Maret et al. 2011) (version of 2020 April).

²⁰ http://www.iram.fr/IRAMFR/GILDAS



Figure 5. Clean transitions of $C_2H_5CONH_2 v = 0$ toward Sgr B2(N1E). Continuum-subtracted spectrum observed with ALMA in black, the synthetic spectra of $C_2H_5CONH_2$ in red, and the preliminary model of other identified molecules in cyan. Red star denotes the 87,980 MHz line that has been detected with IRAM 30 m. The dashed lines denote the 3σ levels (σ ranges from 0.12 to 0.4 K). The observed data behind these figure panels are available.

(The data used to create this figure are available.)

Rest Freq.	Transition	E,,	$\log_{10} A_{ul}$	$S_{ii}\mu^2$
(MHz)	$J'_{Ka',Kc'} \leftarrow J''_{Ka'',Kc''}$: Sym	(K)	$(s^{-10})^{m}$	(D^{2})
	Ground (<i>v</i> ₂₉ , <i>v</i> ₃₀	(0, 0) = (0, 0) state		
87,980.402(0.002)	$14_{2,13} - 13_{1,12}$: E	33.546	-4.4914	118.0
87,980.454(0.002)	$14_{2,13} - 13_{1,12}$: A	33.546	-4.4914	118.0
92,141.928(0.002)	$5_{5,1} - 4_{4,1}$: E	12.641	-4.3465	54.4
92,142.666(0.002)	$5_{5,1} - 4_{4,1}$: A	12.640	-4.3457	54.5
92,142.895(0.002)	$5_{5,0} - 4_{4,0}$: <i>E</i>	12.641	-4.3465	54.4
92,143.080(0.002)	$5_{5,0} - 4_{4,0}$: A	12.640	-4.3457	54.5
92,402.509(0.003)	$21_{8,13} - 21_{7,14}$: E	94.544	-4.5244	140.0
92,403.069(0.004)	$21_{8,13} - 21_{7,14}$: A	94.544	-4.5243	140.0
92,407.406(0.004)	$29_{9,20} - 29_{8,21}$: E	166.366	-4.4713	217.0
92,408.136(0.003)	$29_{9,20} - 29_{8,21}$: A	166.366	-4.4713	217.0
102,058.096(0.003)	$31_{9,23} - 31_{8,24}$: <i>E</i>	186.341	-4.3451	230.0
102,058.653(0.003)	$31_{9,23} - 31_{8,24}$: A	186.340	-4.3451	230.0
103,341.999(0.003)	$32_{6,26} - 32_{5,27}$: E	186.750	-4.4031	200.0
103,342.235(0.004)	$32_{6,26} - 32_{5,27}$: A	186.749	-4.4031	200.0

 $\label{eq:Table 3} Table \ 3 \\ Rest \ Frequencies \ and \ Transition \ Parameters \ of \ Clean \ C_2H_5 CONH_2 \ Lines \ Detected \ toward \ Sgr \ B2(N1E) \ with \ ALMA$

Note. Column (1): rest frequencies with calculated uncertainties given in parentheses; Column (2): transition; Column (3): upper level energy; Column (4): base 10 logarithm of the Einstein A_{ul} coefficient; Column (5): line strength $S_{ij}\mu^2$.



Figure 6. Partially blended transitions of $C_2H_5CONH_2 v = 0$ detected with the ALMA telescope toward the Sgr B2(N1E). See caption of Figure 5 for more detail. The observed data behind these figure panels are available.

(The data used to create this figure are available.)



Figure 7. Partially blended transitions of $C_2H_5CONH_2 v = 1$ detected with the ALMA telescope toward the Sgr B2(N1E). See caption of Figure 5 for more detail. The observed data behind these figure panels are available. (The data used to create this figure are available.)

In this paper, the intensity is given as the main beam brightness temperature (T_{mb}) , which is defined as $T_{mb} = T_A^* \times \frac{F_{\text{eff}}}{B_{\text{eff}}}$, in which T_A^* is the antenna temperature, F_{eff} is the forward efficiency and B_{eff} is the beam efficiency. The values used were $F_{\text{eff}} = 0.95$ and $B_{\text{eff}} = 0.81$ for 3 mm. The halfpower beam widths (HPBW) are $\sim 24''$ for observations at 3 mm.

3.2. ALMA Archive Data

The interferometric data used here were acquired from the Atacama Large Millimeter/submillimeter Array (ALMA) Science archive of the ReMoCA survey (Re-exploring Molecular Complexity with ALMA) (Belloche et al. 2019). This survey was performed during Cycle 4 between 2016 and 2017. A total of five spectral setups were used to cover frequencies between 84.1



Figure 8. Transitions of NH₂CHO that are identified in the ALMA spectrum of Sgr B2(N1E). Black lines show continuum-subtracted spectrum observed with the ALMA telescope, while red lines show the synthetic spectra of NH₂CHO. The dashed lines denote the 3σ levels. Obvious discrepancies between synthetic and observed spectra are caused by saturation effects.

and 114.4 GHz continuously. Each spectral setup was observed independently, repeated by four executions, with each execution achieving an on-source time of 47–50 minutes. The single polarization mode was employed in the spectral setups to achieve a simultaneous handling of four spectral windows of 1.875 GHz bandwidth each, with a uniform spectral resolution of 0.488 MHz that corresponds to a velocity resolution of 1.3–1.7 km s⁻¹ across the observing band. The phase center of the observations is between the two hot cores Sgr B2(N1) and (N2), at the coordinates $\alpha_{J2000} = 17^{h}47^{m}19.87^{s}$, $\delta_{J2000} = -28^{\circ}22'16''$. The HPBW of the primary beam, equivalent to the field of view of the observations, is between 69'' and 51'' from the lowest to the highest observed frequencies. Details about the observations were presented in Belloche et al. (2019).

The data were calibrated using the standard ALMA data calibration pipeline with Common Astronomy Software Applications package (CASA) version 4.7.0-1 for the first spectral setup, and with CASA version 4.7.2 for the other four spectral setups. The calibrated data were then imaged with CASA version 5.6.1-8. The bandpass calibration was carried out on the quasar J1924-2914, with the exception of one execution in the second spectral setup that used the quasar J1517-2422 instead. The absolute flux density scale was derived from the quasars J1924-2914 or J1733-1304. The phase and amplitude calibrations were performed on the quasar J1744-3116. The images were produced with the TCLEAN deconvolution algorithm in CASA. The HPBW of the synthesized beam depends on the setup and the sideband but is typically $0.6'' \times 0.5''$. The median rms noise level in the

 Table 4

 Parameters of Our Best-fit LTE Model of C2H5CONH2, CH3CONH2,

 NH2CHO, and C¹⁸O toward Sgr B2(N1E) and Sgr B2(N1) with the ALMA Telescope

		r-			
Molecule	Size (")	T _{rot} (K)	$N \pmod{(\text{cm}^{-2})}$	$\frac{\Delta V}{(\mathrm{km}~\mathrm{s}^{-1})}$	$V_{\rm off}$ (km s ⁻¹)
		Sgr B2(N1E)			
$C_2H_5CONH_2,$ v = 0, v = 1	2.3	150	1.5(16)	$4.2(.5)^{a}$	-1
$CH_3CONH_2,$ v = 0	2.3	150	2.5(16)	$3.6(.6)^{b}$	-1
$NH_2CHO, v = 0$	2.3	150	2.8(17)	6 ^{<i>c</i>}	-1
C ¹⁸ O	2.3	150	2.5(17)	3.5(.3)	1
		Sgr B2(N1)			
$C_2H_5CONH_2,$ v=0	2.3	150	3.3(16)	5^d	-1
$CH_3CONH_2,$ v = 0	2.3	150	1.5(17)	5.4(.4) ^e	-1
NH ₂ CHO, $v = 0$	2.3	150	9.5(17)	7^c	-1

Notes. Column (1): molecule name; Column (2): source diameter (FWHM); Column (3): rotational temperature; Column (4): total column density of the molecule. x(y) means $x \times 10^{y}$; Column (5): line width (FWHM); Column (6): velocity offset with respect to the assumed systemic velocity of Sgr B2(N1), $V_{sys} = 64$ km s⁻¹. ^{*a*}: Gaussian fitting result of 87.980 and 92.402 GHz transitions. ^{*b*}: Gaussian fitting result of 92.338 GHz transition. ^{*c*}: As NH₂CHO is optically thick, the line width obtained with Gaussian fitting is unreliable, thus the WEEDS fitting by visual matching was done. ^{*d*}: As C₂H₅CONH₂ blends with an U-line toward the whole region of Sgr B2(N1), the line width and its uncertainty cannot be derived with Gaussian fitting in CLASS, thus the WEEDS fitting by visual matching was done. ^{*d*}: Gaussian fitting result of 92.343 GHz transition.



Figure 9. Transitions of CH_3CONH_2 that are identified in the ALMA spectrum of Sgr B2(N1E). Black lines show continuum-subtracted spectrum observed with the ALMA telescope, while red lines show the synthetic spectra of CH_3CONH_2 . The dashed lines denote the 3σ levels.

channel maps of the data cubes reached for each selected track varies between 0.4 and 1 mJy beam⁻¹. We did not perform self-calibration for the data.

4.2. Atacama Large Millimeter/submillimeter Array (ALMA) Archive Result

4.2.1. Position Selection

Sgr B2(N1) is a hot core that is known to be rich in compelx organic molecules (COMs; Belloche et al. 2013, 2019). However, the spectrum toward the continuum peak position of Sgr B2(N1) is severely affected by the optically thick free-free emission from the H sc II regions, and the weak COM transitions also suffer from serious line blending from other molecules (Belloche et al. 2019). Our pre-analysis indicates that the 92.143 GHz line appears to be the only line that is clean across the whole N1 region. By mapping the integrated intensity of 92.143 GHz line of propionamide and comparing to that of $CH_{3}^{13}CN$ (Figure 2), we selected a position in which propionamide is relatively strong and other molecular emissions are relatively weak. The selected region is about 1.5'' to the east of the hot core Sgr B2(N1). The coordinate is 17:47:19.99, -28:22:18.72(J2000). The size of selected region is $\sim 0.3''$. This new position, which is studied for the first time, is referred to as Sgr B2(N1E). Thus, as the next step we concentrated on the Sgr B2(N1E) direction to avoid high opacity from strong continuum emission and serious line blending from other molecules.

4.2.2. Removal of Continuum Emission

The identification and removal of the continuum emission is difficult yet critical for the identification of propionamide in Sgr B2(N). As stated in Belloche et al. (2019), because of the

4. Observational Results

4.1. IRAM 30 m Result

For the IRAM data we have considered a number of propionamide emission lines among which two lines at 86,954 and 87,980 MHz looked most promising. Unfortunately our pre-analysis showed that even these two lines suffer from serious line blending from other molecules. As is stated in Section 3.1, the HPBWs are $\sim 24''$ for IRAM observations at 3 mm. Within this range, there are many hot cores, H sc II regions, protostellar outflows, and so on (Bonfand et al. 2017; Belloche et al. 2019). Thus, the IRAM spectrum of propionamide may include contributions from a number of hot cores in Sgr B2(N), and analysis of propionamide emission in all these components of Sgr B2(N) is needed to model the propionamide emission from Sgr B2(N), which is beyond the scope of this paper. That is why as the next step of our search we concentrated on ALMA archive data (see Section 3.2), which provided the HPBW for a synthesized beam of order of $0.6'' \times 0.5''$ thus allowing a search for a position in which propionamide is less contaminated by line blending from other molecules.



Figure 10. Integrated intensity maps of NH₂CHO 103.525 GHz line (left) and CH₃CONH₂ 102.445 GHz line (right) overlaid on integrated intensity map of C₂H₅CONH₂ 92.143 GHz line in Sgr B2(N1). The contour maps show the integrated intensity maps of NH₂CHO and CH₃CONH₂. The contours represent 5, 15, 25, 35, 45, 55, 75, 95, 115, 135, 155, and 175 times of σ , with σ the rms noise level equal to 0.01 Jy beam km s⁻¹. The yellow solid circles indicate the position of Sgr B2(N1E), while the yellow dashed circles indicate the range of the selected main Sgr B2(N1) region. The white ellipses show the size of the respective synthetic beams. The crosses indicate the positions of Sgr B2(N1) and Sgr B2(N1S). The integrated frequency range of NH₂CHO ranges within 103,521–103,527 MHz, while the integrated frequency range of CH₃CONH₂ ranges from 102,440 to 102,448 MHz.

large number of spectral lines, their different systemic velocities, and the velocity gradients, it is impossible to split the line and continuum emission in the Fourier domain. Therefore, we performed the splitting in the image plane. As Sgr B2(N1E) is spatially offset from the hot core position, both the continuum emission and molecular line emission toward Sgr B2(N1E) are much weaker than those toward Sgr B2(N1S). The typical line width of molecular line emission toward N1E is 4 km s^{-1} , which is narrower than the typical line width of molecular line emission toward N1S ($\sim 5 \text{ km s}^{-1}$, see Table 4 in Belloche et al. 2019). As such, the baseline extraction for N1E is easier than that for N1S. For each spectral window, we selected 3-5 groups of channels that seem to be free of line emission. A first-order baseline was fitted to these channels. The subtracted continuum emission of lines that is obviously offset from the spectrum energy distribution of free-free emission, is regarded as being affected by line confusion or absorption. The subtracted continuum emission for the identified lines are presented in Figure 3. The overall flux dependence on the frequencies is well consistent with the spectrum energy distribution that is expected by freefree emission.

4.2.3. H₂ Column Density

We used the C¹⁸O column density to derive the H₂ column density. Figure 4. shows the ALMA spectrum of C¹⁸O toward Sgr B2(N1E). The C¹⁸O column density has been obtained with WEEDS by assuming optically thin emission and local thermodynamic equilibrium (LTE, which means that the excitation temperature of the transitions is equal to the rotational temperature of the molecule and the kinetic temperature of the gas) conditions. An uniform excitation temperature T_{ex} of 150 K (Belloche et al. 2019) was adopted. The column density of C¹⁸O was derived to be 2.5×10^{17} cm⁻². According to the Galactic



Figure 11. The 92.143 GHz transition of $C_2H_5CONH_2$ that is identified in the ALMA spectrum of Sgr B2(N1). The black line shows continuum-subtracted spectrum observed with the ALMA telescope, while the red line shows the synthetic spectra of $C_2H_5CONH_2$. The dashed line denotes the 3σ levels.

gradient of $^{16}\text{O}/^{18}\text{O}(\text{Wilson & Rood 1994})$, the abundance ratio of C¹⁸O is significantly different in the Galactic Center region and the solar neighborhood. We adopt an abundance ratio N(H₂)/N(C¹⁸O) two times of that in Orion KL (7 \times 10⁶ cm⁻², Castets & Langer 1995). The N(H₂) was then calculated to be 3.5 \times 10²⁴ cm⁻².

4.2.4. Detection of Propionamide

We used Weeds to do line identification and spectral modeling. We built our own Weeds database containing the spectroscopic data of propionamide, acetamide (Ilyushin et al. 2004) and acetone (Ilyushin et al. 2019). The Jet Propulsion Laboratory (JPL; Pickett et al. 1998) and the Cologne Database for Molecular Spectroscopy (CDMS; Müller et al. 2005; Endres et al. 2016)



Figure 12. Transitions of NH₂CHO that are identified in the ALMA spectrum of Sgr B2(N1). Black lines show continuum-subtracted spectrum observed with the ALMA telescope, while red lines show the synthetic spectra of NH₂CHO. The dashed lines denote the 3σ levels. Obvious discrepancies between synthetic and observed spectra are caused by saturation effects.

databases are used for other molecules. The 3 mm emission of propionamide toward Sgr B2(N1E) is modeled assuming LTE conditions with five parameters: column density, temperature, source size, velocity offset, and line width. We model each spectral window of each observed setup separately to account for the varying angular resolution. The same set of the five parameters is used to model the $C_2H_5CONH_2$ v = 0 and v = 1 states.

Approximately 110 $C_2H_5CONH_2 v = 0$ lines were expected to be higher than the 3σ noise level. Six clean lines were detected toward Sgr B2(N1E). These lines are presented in Figure 5 and Table 3. Approximately 17 partially blended lines were also detected (Figure 6). The remaining 90 lines that match the considered frequency range and noise level threshold are blended with other species, but the observed results do not contradict the predicted intensities for these lines. About 40 $C_2H_5CONH_2 v = 1$ lines with expected intensities above 3σ noise level were expected. Of these, eight partially blend lines were detected (Figure 7), with the rest blended seriously with other species. The data behind Figures 5 and 6, and 7 are available as an electronic supplement with this paper.

With the derived size, a line-width measurement of 4.2(.5) km s⁻¹ with a Gaussian fitting of 92.408 and 103.342 GHz lines in CLASS, and an assumption of excitation temperature of 150 K (which is very close to what was obtained for similar molecules toward Sgr B2(N1S) by Belloche et al. (2019), we simulated the spectral profiles for all transitions of propiona-mide detected toward Sgr B2(N1E) with Weeds. The emission of the vibrationally excited skeletal torsion state of propiona-mide was modeled with the same parameters as those for the vibrational ground state. A column density of 1.5×10^{16} cm⁻²

was obtained for propionamide (see Table 4). Abundance relative to H₂ was estimated to be 4.3×10^{-9} for C₂H₅CONH₂.

4.2.5. Related Molecules

Figures 8 and 9 present transitions of formamide and acetamide that are identified in the ALMA spectrum of Sgr B2(N1E), respectively. Figure 10 presents integrated intensity maps of NH₂CHO and CH₃CONH₂ overlaid on C₂H₅CONH₂ toward Sgr B2(N1). The integrated intensity maps of CH₃CONH₂ and NH₂CHO indicate that the amide family shows quite similar spatial distributions, implying a consistent formation mechanism in the ISM. To compare the column density derived for C₂H₅CONH₂ and NH₂CHO₄ and NH₂CHO toward Sgr B2(N1E) were also derived using the Weeds package. We derived average abundances relative to H₂ of 7.2×10^{-9} for CH₃CONH₂, and 8.0×10^{-8} for NH₂CHO. The abundance ratio [C₂H₅CONH₂]/[CH₃CONH₂] was 0.6, while [CH₃CONH₂]/[NH₂CHO] was estimated to be ~0.1 toward Sgr B2(N1E).

Figure 11 presents the 92.143 GHz transition of $C_2H_5CONH_2$ that is identified in the ALMA spectrum toward the main region of Sgr B2(N1), while Figures 12 and 13 present transitions of NH₂CHO and CH₃CONH₂ that are identified in Sgr B2(N1), respectively. Due to the high column density and large optical depth of NH₂CHO toward Sgr B2(N1), some lines are saturated, such as the 102,065 MHz, 105,465 MHz, and 106,108 MHz lines. A fit to the 92.143 GHz line of $C_2H_5CONH_2$ toward the main Sgr B2(N1) region was also obtained. A mean column density of 3.3×10^{16} cm⁻² was



Figure 13. Transitions of CH₃CONH₂ that are identified in the ALMA spectrum of Sgr B2(N1). Black lines show continuum-subtracted spectrum observed with the ALMA telescope, while red lines show the synthetic spectra of CH₃CONH₂. The dashed lines denote the 3σ levels.

obtained for C₂H₅CONH₂. The column densities derived for CH₃CONH₂ and NH₂CHO are somewhat lower than those derived for Sgr B2(N1S) by Belloche et al. (2019), as these are the average over the entire CH₃CONH₂ emission region. The abundance ratio [C₂H₅CONH₂]/[CH₃CONH₂] was estimated to be ~0.22, which was lower than that of Sgr B2(N1E). The [CH₃CONH₂]/[NH₂CHO] is ~0.16 toward the main Sgr B2(N1) region. Currently the mechanism for the enhancement of propionamide abundance in comparison with acetamide in Sgr B2(N1E) is uncertain.

We have tried to identify as many molecules as we can in the rest of the spectrum with Weeds. Table 5 lists the identified molecules, as well as the fitted parameters under LTE assumption.

5. Discussion

Based on the six clean lines of propionamide toward Sgr B2(N1E), which are consistent with the LTE synthetic spectrum of propionamide, we reported the tentative detection of propionamide in Sgr B2. Future high sensitivity and resolution observations of protostars with narrower line width might help identify more clean transitions and further confirm the detection of this molecule in the ISM. The results presented here provide strong evidence for warm, compact propionamide with high abundance in the hot cores of the Sgr B2 cloud complex. From a chemical perspective, the presence of propionamide is not surprising. As a substituted acetamide and

the third member of the amide chemical family, propionamide is likely to have formation mechanisms that are similar to or associated with those of the two smaller amides: formamide and acetamide.

The formation routes of amide molecules in space are not yet well understood (Ligterink et al. 2020). Under the conditions of Sgr B2(N), it has been proposed that acetamide might be formed through ice chemistry since pure gas-phase models cannot produce enough acetamide compared with observations (Quan & Herbst 2007; Garrod 2008). Therefore, for the next member of amide family, propionamide, we propose three formation routes, all on the ice mantles on cosmic dust particles.

First, propionamide could form from formamide through reaction with ethane (Ligterink et al. 2018). The well-known interstellar formamide was reported to be detected with a relatively large column density toward Sgr B2 (Rubin et al. 1971), thus it is abundant in the source. Ligterink et al. (2018) investigated the products of ice mixtures of methane and isocyanic acid under UV irradiation, and concluded it can lead to formation of propionamide with the aid of ethane as the latter molecule was also found in the products. Therefore, we propose that propionamide could be formed via the following reaction:

$$CH_3CH_3 + NH_2CHO \rightarrow CH_3CH_2CONH_2 + 2H.$$
 (3)

First, these two reactant molecules, could be dissociated by UV or cosmic ray induced photons to form radicals that are

 Table 5

 Parameters of Preliminary Model of Other Identified Species toward Sgr

 B2(N1E) with the ALMA Telescope

Molecule	Size (")	T _{rot} (K)	N (cm ⁻²)	ΔV (km s ⁻¹)	$V_{\rm off}$ (km s ⁻¹)
CH ₃ COOH, vt = 0	2.3	150	6.0(17)	5	0
$CH_3COOH, vt = 1$	2.3	150	1.0(16)	5	0
i-C ₃ H ₇ CN	2.3	150	2.0(16)	5	-2
$n-C_3H_7CN, v = 0$	2.3	150	2.0(16)	5	-2
$CH_3O^{13}CHO, vt = 0,1$	2.3	150	4.0(17)	4	0
$C_2H_3CN, v = 0$	2.3	150	1.0(17)	7	-1
H ₂ CCHC ¹⁵ N	2.3	150	1.2(15)	5	-1
$H_2CCH^{13}CN, v = 0$	2.3	150	1.0(16)	5	-1
H_2^{13} CCHCN, $v = 0$	2.3	150	4.8(15)	6	-1
$C_2H_5OH, v = 0$	2.3	150	1.5(17)	7	-1
$NH_2CHO, v = 0$	2.3	150	2.4(17)	6	-1
$C_2H_5CN, v = 0$	2.3	150	1.5(17)	7	-1
$CH_3^{13}CH_2CN, v = 0$	2.3	150	7.0(15)	5	$^{-2}$
$C_2H_5CN, v20 = 1-A$	2.3	150	8.0(16)	5	-1
$H_2NCH_2CN, v = 0$	2.3	150	5.0(15)	5	-1
CH ₂ OHCHO	2.3	150	3.0(15)	5	-1
$CH_3CN, v8 = 1$	2.3	150	8(16)	6	0
$CH_3CN, v8 = 2$	2.3	150	5(17)	5	-1
$CH_3NCO, vb = 0$	2.3	150	3.0(16)	5	-1
$CH_3OCH_3, v = 0$	2.3	150	2.0(16)	4	-1
$CH_3SH, vt = 0$	2.3	150	2.5(17)	6	0
$SO_2, v2 = 1$	2.3	150	2.5(18)	5	0
HOCH ₂ CN	2.3	150	5.0(15)	5	0
CH ₃ OCHO	2.3	150	6.0(16)	4	0
CH ₃ ¹³ CHO	2.3	150	1.0(18)	5	-1

Note. Column (1): molecule name; Column (2): source diameter (FWHM); Column (3): rotational temperature; Column (4): total column density of the molecule. x(y) means $x \times 10^{y}$; Column (5): line width (FWHM); Column (6): velocity offset with respect to the assumed systemic velocity of Sgr B2(N1), $V_{sys} = 64$ km s⁻¹.

more reactive. These radicals could then more easily form propionamide, as follows:

$$CH_3CH_3 + h\nu \rightarrow CH_3CH_2 + H.$$
 (4)

$$NH_2CHO + h\nu \rightarrow NH_2CO + H.$$
 (5)

$$CH_3CH_2 + NH_2CHO \rightarrow CH_3CH_2CONH_2 + H.$$
 (6)

$$CH_3CH_2 + NH_2CO \rightarrow CH_3CH_2CONH_2.$$
(7)

Similarly, propionamide could also be formed from the next member in the amide family: acetamide. Acetamide was detected toward Sgr B2(N) with relatively high abundance (Hollis et al. 2006; Halfen et al. 2011). We propose that acetamide could form propionamide by reacting with methyl radical directly or via the route of losing one hydrogen atom to form a radical itself and then reacting with the methyl radical, as shown below:

$$CH_3 + CH_3CONH_2 \rightarrow CH_3CH_2CONH_2 + H.$$
 (8)

$$CH_3CONH_2 + h\nu \rightarrow CH_2CONH_2 + H.$$
 (9)

$$CH_3 + CH_2CONH_2 \rightarrow CH_3CH_2CONH_2.$$
 (10)

The third route to form propionamide on dust grains is from isocyanic acid. Garrod et al. (2008) generated possible acetamide by adding methyl radicals to isocyanic acid followed by a hydrogenation step. We propose a similar route could form propionamide as well:

$$HNCO + CH_3CH_2 \rightarrow CH_3CH_2CONH.$$
(11)

$$CH_3CH_2CONH + H \rightarrow CH_3CH_2CONH_2.$$
 (12)

The enhancement of propionamide abundance in comparison with acetamide toward Sgr B2(N1E) is unclear. The physical or chemical conditions in Sgr B2(N1E) might be favored for production or desorption of propionamide. For example, on the grain surface or ice mantle in Sgr B2(N1E), there might be abundant methyl radicals that could react with acetamide and be released to the gas phase via some desorption mechanisms. Future detailed chemical modeling and laboratory simulation will help investigate this issue, as well as the formation route that dominants the formation of propionamide.

6. Summary

The high sensitivity and high angular resolution of the ReMoCA spectral line survey conducted with ALMA toward Sgr B2(N) provides a good opportunity for the detection of new complex organic molecules in space. In this paper, using the ReMoCA survey, we present the first tentative interstellar detection of propionamide, $C_2H_5CONH_2$, the next member of the interstellar amide chemical family after formamide and acetamide. As a preparatory step, a new laboratory spectroscopic study of the propionamide spectrum was performed in the 9–461 GHz range. The rho axis method and RAM36hf program were used to analyze the propionamide spectrum. By mapping propionamide versus $CH_3^{13}CN$, we selected a position 1.5" to the east of the hot core Sgr B2(N1) (referred to here as Sgr B2(N1E)) where propionamide is relatively strong while other molecular emissions are relatively weak.

We report the tentative detection of propionamide toward Sgr B2(N1E) with six clean lines in the ground vibrational state. More than 20 partially blended lines further support our ISM detection. Our results indicate that the spatial distribution of propionamide is similar to that of formamide and acetamide, suggesting similar formation routes for the amide family. The abundance ratio $[C_2H_5CONH_2]/[CH_3CONH_2]$ toward Sgr B2(N1E) is 0.6, while toward Sgr B2(N1E) is estimated to be ~0.22. The enhancement of propionamide abundance in comparison with acetamide toward Sgr B2(N1E) is unclear.

The detection of propionamide in Sgr B2(N) demonstrates that interstellar chemistry can reach sufficient levels of complexity to form relatively large peptide molecules and shows the possible growth of larger amide molecules from smaller ones in a massive star-forming process. It is probable that propionamide might also exist in massive star-forming regions in the Galactic disk, such as Orion KL and NGC 6334 (Ligterink et al. 2020).

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NAOJ. The single dish data are available in the IRAM archive at https://www.iram-institute.org/EN/content-page-386-7-386-0-0-0.html. The interferometric data are available in the ALMA archive at https://almascience.eso.org/aq/.

Appendix

Part of the data sets treated in this work, as well as the predictions produced, are available in Tables A1, A2, A3, and A4.

Table A1

List of Calculated Positions and Assignments of Rotational Transitions in the Ground Vibrational State (v29, v30) = (0, 0) of Propionamide Up to J = 40 in the 1–460 GHz Frequency Range

S_{v}'	(1) <i>m</i> '	Upper F'	Level	K.,'	Ke'	S _v	(2) m	Lower F	Level J	Ka	Kc	(3) Calculated (MHz)	(4) Unc. (MHz)	$(5) E_{\text{low}}$ (cm^{-1})	(6) $S\mu^2$ (D^2)
A2	0	17.0	17	6	11	A1	0	17.0	17	6	12	1046.6175	0.0004	42.1815	0.147E+01
Е	1	17.0	17	6	11	Е	1	17.0	17	6	12	1046.6486	0.0004	42.1817	0.147E+01
A2	0	18.0	17	6	11	A1	0	18.0	17	6	12	1046.6987	0.0004	42.1815	0.156E+01
A2	Ő	16.0	17	6	11	A1	Õ	16.0	17	6	12	1046.7035	0.0004	42.1815	0.139E+01
E	1	18.0	17	6	11	E	1	18.0	17	6	12	1046 7298	0.0004	42.1817	0.156E+01
E	1	16.0	17	6	11	Ē	1	16.0	17	6	12	1046 7346	0.0004	42.1817	0.139E+01
A1	0	10.0	10	4	6	A2	0	10.0	10	4	7	1057.1069	0.0004	15,7960	0.112E+01
E	1	10.0	10	4	6	E	1	10.0	10	4	7	1057 1240	0.0004	15 7963	0.112E+01
A1	0	11.0	10	4	6	A2	0	11.0	10	4	7	1057 2603	0.0004	15 7960	0.124E+01
A1	Ő	9.0	10	4	6	A2	0	9.0	10	4	7	1057 2758	0.0004	15 7960	0.121E+01 0.102E+01
F	1	11.0	10	4	6	E F	1	11.0	10	4	7	1057.2756	0.0004	15 7963	0.102E+01 0.124E+01
E	1	9.0	10	4	6	E	1	0.0	10	4	7	1057 2020	0.0004	15 7963	0.124E+01 0.102E+01
	0	21.0	21	7	14	12	0	21.0	21	7	15	1310 0231	0.0004	62 5847	0.150E+01
F	1	21.0	21	7	14	E E	1	21.0	21	7	15	1310.0201	0.0005	62 5849	0.159E+01
	0	21.0	21	7	14	12	0	21.0	21	7	15	1310.0005	0.0005	62 5847	0.157E+01
A1	0	22.0	21	7	14	A2	0	22.0	21	7	15	1310.1003	0.0005	62.5847	0.152E + 01
E	1	20.0	21	7	14	А2 Е	1	20.0	21	7	15	1310.1042	0.0005	62.5840	0.152E+01
E	1	22.0	21	7	14	E	1	22.0	21	7	15	1310.1379	0.0005	62.5849	0.107E+01 0.152E+01
L 12	0	20.0	21	2	5		1	20.0	6	1	2	1310.1410	0.0003	02.J049 8 1666	0.152E+01
A2 E	1	6.0	7	2	5	E	1	5.0	6	4	2	1240.0361	0.0021	8.1000	0.404E+01
E A 2	1	0.0	7	3	5		1	5.0	0	4	2	1340.9264	0.0021	8.1009	0.458E+01
AZ E	1	8.0	7	2	5	AI E	1	7.0	0	4	2	1340.9208	0.0021	8.1000	0.022E+01
E	1	8.0	7	3	5	E	1	7.0	0	4	2	1340.9950	0.0021	8.1009	0.013E+01
AZ E	0	7.0	7	3	5	AI	0	6.0	0	4	2	1341.3021	0.0021	8.1000	0.538E+01
	1	7.0	7	3	3	E	1	0.0	0	4	2	1341.4303	0.0021	8.1009	0.530E+01
AI E	0	7.0	7	3	4	A2 E	1	7.0	7	3	5	1348.0582	0.0004	8.2114	0.875E+00
E	1	7.0	7	3	4	E	1	7.0	7	3	5	1348.0682	0.0004	8.2116	0.8/5E+00
AI	0	8.0	7	3	4	A2 E	0	8.0	7	3	5	1348.3441	0.0004	8.2114	0.101E+01
E	1	8.0	/	3	4	E	1	8.0	/	3	5	1348.3541	0.0004	8.2116	0.101E+01
AI	0	6.0	7	3	4	A2	0	6.0	7	3	5	1348.3853	0.0004	8.2114	0.770E+00
E	1	6.0	7	3	4	E	1	6.0	1	3	5	1348.3953	0.0004	8.2116	0.770E+00
E	1	6.0	7	3	5	E	1	5.0	6	4	3	1358.9304	0.0023	8.1663	0.684E-01
E	I	8.0	7	3	5	E	1	7.0	6	4	3	1358.9979	0.0023	8.1663	0.916E-01
E	1	7.0	7	3	5	E	1	6.0	6	4	3	1359.4265	0.0023	8.1663	0.792E-01
A1	0	3.0	4	1	4	A2	0	2.0	3	2	1	1407.3994	0.0014	2.2190	0.214E+01
E	1	3.0	4	1	4	E	1	2.0	3	2	1	1407.5548	0.0014	2.2193	0.214E+01
A1	0	5.0	4	1	4	A2	0	4.0	3	2	1	1407.9512	0.0013	2.2190	0.367E + 01
Е	1	5.0	4	1	4	Е	1	4.0	3	2	1	1408.1065	0.0013	2.2193	0.367E + 01
A1	0	4.0	4	1	4	A2	0	3.0	3	2	1	1409.9772	0.0016	2.2190	0.281E + 01
Е	1	4.0	4	1	4	Е	1	3.0	3	2	1	1410.1326	0.0016	2.2193	0.281E + 01
A1	0	4.0	4	2	2	A2	0	4.0	4	2	3	1463.5393	0.0005	3.0850	0.623E + 00
Е	1	4.0	4	2	2	Е	1	4.0	4	2	3	1463.5418	0.0005	3.0853	0.623E + 00
A1	0	5.0	4	2	2	A2	0	5.0	4	2	3	1464.1122	0.0003	3.0850	0.810E + 00
Е	1	5.0	4	2	2	Е	1	5.0	4	2	3	1464.1147	0.0003	3.0853	0.810E + 00
A1	0	3.0	4	2	2	A2	0	3.0	4	2	3	1464.2595	0.0004	3.0850	0.503E+00
Е	1	3.0	4	2	2	Е	1	3.0	4	2	3	1464.2620	0.0004	3.0853	0.503E + 00
Е	1	34.0	34	8	27	Е	1	35.0	35	5	30	1522.9726	0.0091	148.6103	0.519E+01
A2	0	34.0	34	8	27	A1	0	35.0	35	5	30	1523.6387	0.0091	148.6101	0.519E+01
Е	1	35.0	34	8	27	Е	1	36.0	35	5	30	1523.6874	0.0091	148.6103	0.534E+01
Е	1	33.0	34	8	27	Е	1	34.0	35	5	30	1523.7077	0.0091	148.6103	0.504E+01
A2	0	35.0	34	8	27	A1	0	36.0	35	5	30	1524.3535	0.0091	148.6101	0.534E+01

Note. Column (1): quantum numbers for upper energy levels; Column (2): quantum numbers for lower energy levels; Column (3): calculated frequencies; Column (4): uncertainties for frequencies; Column (5): lower energy levels; Column (6): line strength. Full table is available as electronic supplementary material with this article. (This table is available in its entirety in machine-readable form.)

Table A2List of Calculated Positions and Assignments of Rotational Transitions in the First Excited Skeletal Torsion State (v29, v30) = (0, 1) of Propionamide Up to J = 30 in
the 1–150 GHz Frequency Range

	(1)	T 1	T 1				(2)	T.	T 1				(4) 11	(5) E	(C) G 2
S_y'	$\binom{1}{m'}$	Upper F'	J'	K_a'	K_c'	S_y	(2) m	Lower F	Level J	Ka	K_c	(3) Calculated (MHz)	(4) Unc. (MHz)	$(5) E_{low}$ (cm^{-1})	$(b) S\mu^{2}$ (D^{2})
A2	0	17.0	17	6	11	A1	0	17.0	17	6	12	1029.8309	0.0014	87.1110	0.155E+01
Е	1	17.0	17	6	11	Е	1	17.0	17	6	12	1029.8334	0.0013	87.1110	0.155E+01
A2	0	18.0	17	6	11	A1	0	18.0	17	6	12	1029.9102	0.0012	87.1110	0.165E+01
Е	1	18.0	17	6	11	Е	1	18.0	17	6	12	1029.9127	0.0012	87.1110	0.165E+01
A2	0	16.0	17	6	11	A1	0	16.0	17	6	12	1029.9149	0.0013	87.1110	0.147E+01
Е	1	16.0	17	6	11	Е	1	16.0	17	6	12	1029.9174	0.0012	87.1110	0.147E+01
A1	0	10.0	10	4	6	A2	0	10.0	10	4	7	1039.7338	0.0017	60.7602	0.118E+01
Е	1	10.0	10	4	6	Е	1	10.0	10	4	7	1039.7353	0.0017	60.7602	0.118E+01
A1	0	11.0	10	4	6	A2	0	11.0	10	4	7	1039.8837	0.0012	60.7602	0.130E+01
Е	1	11.0	10	4	6	Е	1	11.0	10	4	7	1039.8851	0.0011	60.7602	0.130E+01
A1	0	9.0	10	4	6	A2	0	9.0	10	4	7	1039.8987	0.0013	60.7602	0.108E + 01
Е	1	9.0	10	4	6	Е	1	9.0	10	4	7	1039.9002	0.0012	60.7602	0.108E+01
Е	1	9.0	9	2	7	Е	1	10.0	10	1	10	1045.0356	0.0254	56.3124	0.978E+00
A2	0	9.0	9	2	7	A1	0	10.0	10	1	10	1045.0603	0.0258	56.3124	0.978E+00
Е	1	10.0	9	2	7	Е	1	11.0	10	1	10	1047.5220	0.0155	56.3124	0.108E+01
A2	0	10.0	9	2	7	A1	0	11.0	10	1	10	1047.5467	0.0161	56.3124	0.108E+01
Е	1	8.0	9	2	7	Е	1	9.0	10	1	10	1047.7753	0.0175	56.3124	0.884E + 00
A2	0	8.0	9	2	7	A1	0	9.0	10	1	10	1047.8000	0.0181	56.3124	0.884E + 00
A1	0	21.0	21	7	14	A2	0	21.0	21	7	15	1289.4397	0.0016	107.4964	0.168E+01
A1	0	21.0	21	7	14	A2	0	21.0	21	7	15	1289.4397	0.0016	107.4964	0.168E+01
A1	0	21.0	21	7	14	A2	0	21.0	21	7	15	1289.4397	0.0016	107.4964	0.168E+01
Е	1	21.0	21	7	14	Е	1	21.0	21	7	15	1289.4433	0.0015	107.4964	0.168E+01
A1	0	22.0	21	7	14	A2	0	22.0	21	7	15	1289.5153	0.0015	107.4964	0.176E+01
Е	1	22.0	21	7	14	Е	1	22.0	21	7	15	1289.5189	0.0014	107.4964	0.176E+01
A1	0	20.0	21	7	14	A2	0	20.0	21	7	15	1289.5190	0.0015	107.4964	0.160E+01
Е	1	20.0	21	7	14	Е	1	20.0	21	7	15	1289.5225	0.0014	107.4964	0.160E+01
A1	0	7.0	7	3	4	A2	0	7.0	7	3	5	1325.7526	0.0028	53.1903	0.922E + 00
Е	1	7.0	7	3	4	Е	1	7.0	7	3	5	1325.7535	0.0028	53.1903	0.922E + 00
A1	0	8.0	7	3	4	A2	0	8.0	7	3	5	1326.0316	0.0014	53.1903	0.107E+01
Е	1	8.0	7	3	4	Е	1	8.0	7	3	5	1326.0325	0.0014	53.1903	0.107E+01
A1	0	6.0	7	3	4	A2	0	6.0	7	3	5	1326.0718	0.0019	53.1903	0.812E+00
Е	1	6.0	7	3	4	Е	1	6.0	7	3	5	1326.0728	0.0019	53.1903	0.812E+00
Е	1	13.0	13	8	6	Е	1	14.0	14	7	7	1380.8726	0.0271	78.9832	0.387E+01
Е	1	14.0	13	8	6	Е	1	15.0	14	7	7	1381.0501	0.0239	78.9832	0.416E+01
Е	1	12.0	13	8	6	Е	1	13.0	14	7	7	1381.0638	0.0243	78.9832	0.360E+01
A1	0	13.0	13	8	6	A2	0	14.0	14	7	7	1381.6452	0.0185	78.9831	0.970E+01
A1	0	14.0	13	8	6	A2	0	15.0	14	7	7	1381.8227	0.0131	78.9831	0.104E+02
A1	0	12.0	13	8	6	A2	0	13.0	14	7	7	1381.8364	0.0139	78.9831	0.903E+01
Е	1	13.0	13	8	5	Е	1	14.0	14	7	7	1382.1497	0.0211	78.9832	0.583E+01
Е	1	14.0	13	8	5	Е	1	15.0	14	7	7	1382.3272	0.0165	78.9832	0.626E+01
Е	1	12.0	13	8	5	Е	1	13.0	14	7	7	1382.3409	0.0172	78.9832	0.543E+01
Е	1	13.0	13	8	6	Е	1	14.0	14	7	8	1385.7711	0.0228	78.9830	0.583E+01
Е	1	14.0	13	8	6	Е	1	15.0	14	7	8	1385.9493	0.0188	78.9830	0.626E+01
Е	1	12.0	13	8	6	Е	1	13.0	14	7	8	1385.9630	0.0193	78.9830	0.543E+01
A2	0	13.0	13	8	5	A1	0	14.0	14	7	8	1386.4487	0.0185	78.9830	0.970E+01
A2	0	14.0	13	8	5	A1	0	15.0	14	7	8	1386.6269	0.0131	78.9830	0.104E + 02
A2	0	12.0	13	8	5	A1	0	13.0	14	7	8	1386.6406	0.0139	78.9830	0.903E+01
Е	1	13.0	13	8	5	Е	1	14.0	14	7	8	1387.0482	0.0251	78.9830	0.387E+01
Е	1	14.0	13	8	5	Е	1	15.0	14	7	8	1387.2264	0.0213	78.9830	0.416E+01
Е	1	12.0	13	8	5	Е	1	13.0	14	7	8	1387.2401	0.0218	78.9830	0.360E+01
A1	0	4.0	4	2	2	A2	0	4.0	4	2	3	1439.2088	0.0054	48.0749	0.656E+00
Е	1	4.0	4	2	2	Е	1	4.0	4	2	3	1439.2090	0.0054	48.0750	0.656E+00

Note. Column (1): quantum numbers for upper energy levels; Column (2): quantum numbers for lower energy levels; Column (3): calculated frequencies; Column (4): uncertainties for frequencies; Column (5): lower energy levels; Column (6): line strength. Full table is available as electronic supplementary material with this article. (This table is available in its entirety in machine-readable form.)

Table A3Assignments, Measured Transition Frequencies, and Residuals from the Fit of the Microwave, Millimeter Wave, and Submillimeter Wave (v29, v30) = (0, 0) and (1, 0) Data for Propionamide

<i>S_y</i> ′	(1) <i>m</i> ′	Upper F'	Level J'	<i>Ka</i> ′	<i>K</i> _c ′	Sy	(2) m	Lower F	Level J	Ka	K _c	(3) Measured (MHz)	(4) Unc. (MHz)	(5) Obscalc. (MHz)	(6) Source of Data
E	1	3.0	3	1	2	Е	1	3.0	3	0	3	9828.3293	0.0040	-0.0012	NIST
A1	0	3.0	3	1	2	A2	0	3.0	3	0	3	9828.3967	0.0040	0.0052	NIST
Е	1	4.0	3	1	2	Е	1	3.0	3	0	3	9828.7609	0.0040	-0.0040	NIST
A1	0	4.0	3	1	2	A2	0	3.0	3	0	3	9828.8307	0.0040	0.0048	NIST
A1	0	2.0	3	1	2	A2	0	3.0	3	0	3	9828.9850	0.0080	0.0071	NIST
Е	1	4.0	3	1	2	Е	1	4.0	3	0	3	9829.7715	0.0040	-0.0007	NIST
A1	0	4.0	3	1	2	A2	0	4.0	3	0	3	9829.8431	0.0040	0.0099	NIST
Е	1	2.0	3	1	2	Е	1	2.0	3	0	3	9830.2738	0.0040	-0.0030	NIST
A1	0	2.0	3	1	2	A2	0	2.0	3	0	3	9830.3444	0.0040	0.0066	NIST
A2	0	4.0	4	1	3	A1	0	3.0	3	2	2	11,341.7526	0.0040	0.0004	NIST
A2	0	4.0	4	1	3	A1	0	4.0	3	2	2	11,341.7526	0.0040	0.0004	NIST
Е	1	4.0	4	1	3	Е	1	3.0	3	2	2	11,341.8796	0.0040	-0.0005	NIST
Е	1	4.0	4	1	3	Е	1	4.0	3	2	2	11,341.8796	0.0040	-0.0005	NIST
A2	0	5.0	4	1	3	A1	0	4.0	3	2	2	11,342.0490	0.0040	0.0002	NIST
Е	1	5.0	4	1	3	Е	1	4.0	3	2	2	11,342.1753	0.0040	-0.0015	NIST
A1	0	1.0	2	1	2	A2	0	1.0	1	1	1	12324.3051	0.0040	0.0000	NIST
Е	1	1.0	2	1	2	Е	1	1.0	1	1	1	12,324.3051	0.0040	0.0026	NIST
A1	0	1.0	2	1	2	A2	0	2.0	1	1	1	12,324.8907	0.0040	0.0006	NIST
Е	1	1.0	2	1	2	Е	1	2.0	1	1	1	12,324.8907	0.0040	0.0032	NIST
A1	0	3.0	2	1	2	A2	0	2.0	1	1	1	12,325.5987	0.0040	-0.0015	NIST
Е	1	3.0	2	1	2	Е	1	2.0	1	1	1	12,325.5987	0.0040	0.0011	NIST
A1	0	1.0	2	1	2	A2	0	0.0	1	1	1	12,325.7607	0.0040	-0.0068	NIST
Е	1	1.0	2	1	2	Е	1	0.0	1	1	1	12,325.7607	0.0040	-0.0041	NIST
A1	0	2.0	2	1	2	A2	0	1.0	1	1	1	12,326.2924	0.0040	-0.0012	NIST
Е	1	2.0	2	1	2	Е	1	1.0	1	1	1	12,326.2924	0.0040	0.0014	NIST

Note. Column (1): quantum numbers for upper energy levels; Column (2): quantum numbers for lower energy levels; Column (3): measured frequencies; Column (4): measurement uncertainties for frequencies; Column (5): residuals from the fit; Column (6): source of the data. Full table is available as electronic supplementary material with this article.

(This table is available in its entirety in machine-readable form.)

Table A4		
Assignments, Measured Transition Frequencies, and Residuals from the Fit of the Microwave, Millimeter Wave, and Submillimeter	Wave (v29, v	(30) = (0, 1) and
(1, 1) Data for Propionamide		

										-					
S_{y}'	(1) <i>m</i> ′	Upper F'	Level J'	K_a'	K_c'	S_y	(2) m	Lower F	Level J	Ka	K_c	(3) Measured (MHz)	(4) Unc. (MHz)	(5) Obscalc. (MHz)	(6) Source of Data
A1	0	-1.0	5	2	4	A2	0	-1.0	5	1	5	26,623.9100	0.2000	-0.0432	OSLO96
Е	1	-1.0	5	2	4	Е	1	-1.0	5	1	5	26,623.9100	0.2000	-0.0209	OSLO96
A1	0	-1.0	7	3	4	A2	0	-1.0	7	2	5	26,904.5800	0.2000	0.1708	OSLO96
Е	1	-1.0	7	3	4	Е	1	-1.0	7	2	5	26,904.5800	0.2000	0.2008	OSLO96
A2	0	-1.0	12	3	9	A1	0	-1.0	12	2	10	27,691.6800	0.2000	-0.2383	OSLO96
Е	1	-1.0	12	3	9	Е	1	-1.0	12	2	10	27,691.6800	0.2000	-0.2216	OSLO96
A2	0	-1.0	6	3	3	A1	0	-1.0	6	2	4	28,557.9000	0.2000	0.1453	OSLO96
Е	1	-1.0	6	3	3	Е	1	-1.0	6	2	4	28,557.9000	0.2000	0.1764	OSLO96
A2	0	-1.0	6	1	5	A1	0	-1.0	5	2	4	28,863.8800	0.2000	0.0671	OSLO96
Е	1	-1.0	6	1	5	Е	1	-1.0	5	2	4	28,863.8800	0.2000	0.0586	OSLO96
A1	0	-1.0	5	3	2	A2	0	-1.0	5	2	3	30,039.3300	0.2000	0.1117	OSLO96
Е	1	-1.0	5	3	2	Е	1	-1.0	5	2	3	30,039.3300	0.2000	0.1431	OSLO96
A2	0	-1.0	8	1	7	A1	0	-1.0	8	0	8	31,232.8600	0.2000	-0.0066	OSLO96
Е	1	-1.0	8	1	7	Е	1	-1.0	8	0	8	31,232.8600	0.2000	0.0101	OSLO96
A1	0	-1.0	13	3	10	A2	0	-1.0	13	2	11	31,288.2100	0.2000	-0.0560	OSLO96
Е	1	-1.0	13	3	10	Е	1	-1.0	13	2	11	31,288.2100	0.2000	-0.0414	OSLO96
A1	0	-1.0	14	4	10	A2	0	-1.0	14	3	11	32,071.8600	0.2000	0.0986	OSLO96
Е	1	-1.0	14	4	10	Е	1	-1.0	14	3	11	32,071.8600	0.2000	0.1290	OSLO96
A2	0	-1.0	13	4	9	A1	0	-1.0	13	3	10	32,660.0100	0.2000	0.1812	OSLO96
Е	1	-1.0	13	4	9	Е	1	-1.0	13	3	10	32,660.0100	0.2000	0.2165	OSLO96
A2	0	-1.0	15	4	11	A1	0	-1.0	15	3	12	32,723.9900	0.2000	-0.0986	OSLO96
Е	1	-1.0	15	4	11	Е	1	-1.0	15	3	12	32,723.9900	0.2000	-0.0733	OSLO96
A2	0	-1.0	5	3	3	A1	0	-1.0	5	2	4	33,023.3400	0.2000	-0.0966	OSLO96
Е	1	-1.0	5	3	3	Е	1	-1.0	5	2	4	33,023.3400	0.2000	-0.0644	OSLO96
A1	0	-1.0	7	2	6	A2	0	-1.0	7	1	7	33,056.5200	0.2000	-0.1523	OSLO96

Note. Column (1): quantum numbers for upper energy levels; Column (2): quantum numbers for lower energy levels. F = -1.0 means that quadrupole hyperfine structure was not resolved for this rotational transition; Column (3): measured frequencies; Column (4): measurement uncertainties for frequencies; Column (5): residuals from the fit; Column (6): source of the data. Full table is available as electronic supplementary material with this article.

(This table is available in its entirety in machine-readable form.)

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