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DFT Simulation of the X-Ray Emission and Absorption Spectra for Different Lithium Compounds

HASSEBI Khalil⁽¹⁾, RIVIDI Nicolas⁽²⁾, VERLAGUET Anne⁽³⁾ and JONNARD Phillipe⁽¹⁾



(1) Laboratoire de Chimie Physique—Matière et Rayonnement, Faculté des Sciences et Ingénierie, Sorbonne Université, UMR CNRS, 4 Place Jussieu, CEDEX 05, 75252 Paris, France

(2) Service Camparis, UMS 7154—CNRS, OSU Ecce Terra, Sorbonne Université, 75005 Paris, France

(3) LSD - Lithosphère, structure et dynamique, iSTeP - Institut des Sciences de la Terre de Paris, UMR CNRS 7193, 75005 Paris, France



Motivation

With the development of **X-ray spectroscopic** techniques more light elements with emissions in the soft x-ray range are detected. Our goal is to extend this to

Results

Absorption & Emission Spectra:

the Li element with an easy accessible spectroscopic technique using either **multilayers** [2] **or grating system** that can be implemented in EPMA or SEM.

Lithium has a single valence electron which participates in the formation of bonds with atoms of other elements leading to a dramatic change of emission probability from one compound to the other. In addition to the matrix effect resulting from the presence of other compounds in the sample making the anticipation and detection of **Li K alpha emission** difficult. This is why we proceed by **DFT simulation** to investigate the changes of **emission and absorption spectra** of lithium compounds focusing on the range of Li K emission.



Reaching the **micro** & **nano** quantitative analysis of Lithium in different compounds using (EPMA)

Obtain information on the
 physical & chemical state
 of lithium atom from the
 shape of Li Kα emission

Introducing the acquiredinformation in the new quantification model **Comparing** the calculated spectra with measured ones to **quantify** Li chemical states in unknown compounds





→ Different **number of peaks** and **shoulders**

Figure: Experimental [3] & simulated Li Metal K emission spectra comparision, a Gaussian of 3eV was used to broaden the experimental spectrum.

 \rightarrow Good Agreement between the calculated

and experimental spectra

Methods

Spectra have been computed using density functional theory with all-electron linearized augmented plan wave (LAPW) in conjunction with the PBE-GGA functional adopting the Wien2k [1] package.Density of states, emission and absorption spectra have been determined.

Spectra are calculated for the dipole allowed transitions, generating matrix elements, which are multiplied with a radial transition probability and the partial densities of states.

All calculations have been performed with Lorentzian broadening of 0.03eV to account for the core-hole lifetime, and a gaussian broadening of 1eV to take into consideration the spectrometer function.



Figure: Illustration of the X-ray emission and absorption processes



Energy (E-Ef) (eV)

Density of States:



Unoccupied and **occupied** electronic structures of the different Lithium compounds have been studied by density of states calculations using the wien2k.





Emission & absorption spectra

Sample Gas Detector or CCD Camera

Figure: Wavelength dispersive spectroscopy (WDS) technique implemented inside the EPMA to acquire emission spectra

The emitted X-rays can be detected using the wavelength dispersive spectroscopy (WDS) as represented in the figure above, where periodic multilayers or grating are used as dispersive elements.

Conclusion:

Lithium presents only one emission band corresponding to a transition between Li 2p valences states and 1s core level. The calculations show that the **shape and position** of the **emission and absorption** bands are **highly sensitive** to the **chemical state** of lithium atom.

References:

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