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Faculté des Sciences ef

Technologies

Surface states of refractory alloys rich in chromium and strengthened by different types of MC carbides after oxidation at high temperature

Elodie Conrath, Patrice Berthod

□To limit the ↘ of the thermo-mechanical properties of

□Crystallize as pro-eutectic compact HfC [5] and script-like

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□To limit the carbides > and fragmentation [2]

superallovs [3.4]

carbides too [6]

Study of hafnium carbides (HfC):

How reinforcing equiaxed cast Co-based superalloys:

With carbides:

- Generally eutectics and interdendritics
- Thermally very stable (applications at
- very high temperature (T))
- More stable than chromium carbides (Cr is necessarily present in such superalloys)

The most stable : tantalum carbides (TaC)

Often used in Co-based superalloys [1]

*Research of new carbides:

□More stable than TaC:

Elaboration :

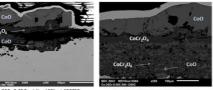
Fusion and solidification under 300mbars Ar in a High Frequency induction furnace (pure elements → ingots of 40g)



Scanning Electron Microscope (SEM) micrographs taken in Back Scattered Electrons mode (BSE) illustrating the oxididized surfaces and deteriorated subsurfaces

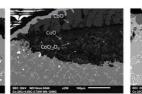
Energy Dispersion Spectrometry (EDS) results for identifying the external and internal oxides and to value the decrease in chromium content in extreme surface

industrial air (50 hours at 1200°C) followed by X-Ray diffraction on external oxides and preparation of embedded and polished crossections The soften as will write Co-25Cr-0.5C after 50h of oxidation at 1200°C



Co-25Cr-0.5C ed 50h at 1200

Oxidation tests in thermo-balance under dry



dized 50h at 1200*

Co-25Cr-0.5C-3.72Hf after 50h of oxidation at 1200°C

In a parallel work [11]:

The purpose of this work:

□3 Co-based quaternary alloys with Hf (Co-25Cr-0.25C-3.72Hf ;

Co-25Cr-0.5C-3.72Hf and Co-25Cr-0.5C-7.44Hf) and 2 Co-based

Better understanding of the differences of kinetic between the

Tested in oxidation in thermo-balance (50h, 1200°C):

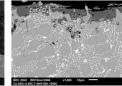
ternary alloys (Co-25Cr-0.25C and Co-25Cr-0.5C): Cast by High Frequency induction melting

• To characterize the oxidation kinetic

HfC-containing alloys and the ternary ones:

Comparison with similar Ni-based alloys

Post-mortem characterization after high T oxidation



General - or at least local - catastrophic oxidation: presence of thick scales, composed of complex oxides 18.1wt.%Cr in the Co-25Cr-0.50C alloy 16.7wt.%Cr in the Co-25Cr-0.25C-3.72Hf alloy

Comparison with the same results obtained on ternary and Hf-rich nickel based alloys elaborated following the same elaboration route and oxidized for 46 hours at 1200°C:

Much better behavior for the Ni-based alloys, ternary or Hf-rich ones: all chromia-forming behavior, higher Cr contents in extreme surface

Ni-25Cr-0.50C after Ni-25Cr-0.50C-3.72H hoursat 1200° roxidation during hours at 1200°C Cr.01 Cr.O

*All the Co-based alloys:

- □Affected by catastrophic oxidation
- Confirmed by the metallographic characterizations
- Contrarily to the Ni-based corresponding alloys (chromia-forming behavior conserved)

*Catastrophic oxidation:

- □Appears early for the quaternary alloys:
 - Fast parabolic kinetics
 - Very high final mass gain
- C. T. Sims, W.C. Hagel, The superalloy, John Wiley & Sons, New York (1972) P. Berthod, S. Michon, L. Aranda, S. Mathieu, J.C. Gachon, Calphad, 27, 353 (2003). M. J. Donachie, S. J. Donachie, Superalloys A technical guide, ASM Int., Materials Park (2002).
- [1] [2] [3] [4] [5] P. Berthod, S. Michon, L. Aranda, P. Steinmetz, proc. of "Matériaux 2006", Dijon, France, Nov. 2006 P. Berthod, Journal of Alloys and Compounds, 481, 746 (2009).
- P. Berthod, Materials Science: An Indian Journal, 9(11), 420 (2013)

Problems with the {m.(dm/dt) = Kp - Kv.m} method: **• To extract correct Kp and Kv values** oBecause not only Cr₂O₃ in the oxide scale

Examples of chromium contents determined in extreme surface by EDS measurements:

Behavior in high T oxidation:

- □ Not good at 1200°C: no great resistance against oxidation at high T
- Worse for the quaternary alloys than for the ternary ones

References

Several solutions:

- □ 7 the Cr content:
- Up to 30 wt.%Cr
- In order to have more than 15 wt.% on extreme surface Adding more Hf:
 - To form not only HfC carbides and to have free Hf atoms in the matrix
 - Hf maybe able to diffuse to the oxidation front:
 - Improve the oxidation behavior
 - Better adherence of the oxide scales
- P. Berthod, Annales de Chimie Science des Matériaux, 34(5), 275 (2009)
- P. Berthod, C. Heil, L. Aranda, Journal of Alloys and Compounds, 50(9), 219 (2001).
 P. Kofstad, High temperature corrosion, Elsevier applied science, London (1988).
- D. J. Young, High temperature oxidation and corrosion of metals, Elsevier, Amsterdam (2008).P. Berthod, E. Conrath, proceedings of the "Eurocorr 2013" congress, Estoril, September 2013.

[7]

[8] [9]

[10] [11]

Better stability than TaC at very high temperature [7, 8] □Hf known to be good for resistance to high T oxidation [9, 10] But, in so high quantities: risks of becoming detrimental for the resistance against the high T oxidation \rightarrow point to verify

