



HAL
open science

PEG-cored phosphorus dendrimers: Synthesis and functionalization

Hanna Dib, Jérémy Rebière, Cyrille Rebout, Omar Alami, Saïd El Kazzouli, Nabil El Brahmi, Regis Laurent, Béatrice Delavaux-Nicot, Anne-Marie Caminade

► **To cite this version:**

Hanna Dib, Jérémy Rebière, Cyrille Rebout, Omar Alami, Saïd El Kazzouli, et al.. PEG-cored phosphorus dendrimers: Synthesis and functionalization. *Results in Chemistry*, 2022, 4, pp.100304. 10.1016/j.rechem.2022.100304 . hal-03578838

HAL Id: hal-03578838

<https://hal.science/hal-03578838>

Submitted on 17 Feb 2022

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.



Distributed under a Creative Commons Attribution - NonCommercial - NoDerivatives 4.0 International License



PEG-cored phosphorus dendrimers: Synthesis and functionalization

Hanna Dib^{a,1}, Jérémy Rebière^{b,c,1}, Cyrille Rebut^{b,c}, Omar Alami^{b,c,d}, Saïd El Kazzouli^d, Nabil El Brahmi^{b,c,d}, Régis Laurent^{b,c}, Béatrice Delavaux-Nicot^{b,c}, Anne-Marie Caminade^{b,c,*}

^a College of Engineering and Technology, American University of the Middle East, Kuwait

^b Laboratoire de Chimie de Coordination (LCC), CNRS UPR8241, 205 Route de Narbonne, 31077 Toulouse Cedex 4, France

^c LCC-CNRS, Université de Toulouse, CNRS, Toulouse, France

^d Euromed Research Center, Euromed Faculty of Pharmacy, Euromed. University of Fez, Route de Meknes, 30000 Fez, Morocco

ARTICLE INFO

Dedicated to Professor Claudio Pettinari on the occasion of his 58th birthday

Keywords:

Dendrimer
Polyethylene glycol (PEG)
Nitrogen ligand
Phosphorhydrazone
1,3,5-Triazine

ABSTRACT

Dendrimers are macromolecular hyperbranched nano-structures, which properties are mainly driven by the type of their terminal functions. This paper reports the synthesis of dendritic structures built from difunctional polyethylene glycols (PEGs) used as core. Three different PEG lengths have been used: PEG₃₀₀ (ca 6 CH₂CH₂-O units), PEG₁₀₀₀ (ca 22 CH₂CH₂-O units), and PEG₂₀₀₀ (ca 45 CH₂CH₂-O units). The first level of branching units is based on a dialdehyde of a dipiperazine triazine derivative. The second and third level of branching units are constituted of phosphorhydrazone linkages. The growing of the dendritic branches was found extremely slow with both PEG₁₀₀₀ and PEG₂₀₀₀ used as core, and was stopped early, whereas this problem was not observed with PEG₃₀₀ used as core. These PEG-cored dendritic structures were finally functionalized on the surface with two types of nitrogen ligands.

Introduction

Dendrimers are monodispersed macromolecular hyperbranched structures, consisting of three basic architectural components, namely a core, an interior of monomer shells (named generations) consisting of repeating branch-cell units, and terminal functional groups which constitute the outer shell or periphery [1,2]. Most properties of dendrimers are governed by the nature of the terminal functional groups, even if the internal structure can play a key role, in particular for biology/nanomedicine [3]. The solubility in water of dendrimers is of utmost importance when considering their biological properties. In many cases, water-solubility is attained when having either positive [4] or negative charges on the surface. The most classical dendrimers such as PAMAM (polyamidoamine) [5] and PPI (polypropyleneimine) dendrimers [6,7] have nitrogen atoms as branching points, which can be easily protonated to increase the solubility in water. Besides charged entities, in the internal structure or on the surface, polyethylene glycol (PEG) is widely used as a neutral polymer, soluble in a large variety of solvents, including in water. A large number of dendrimers have been functionalized on their surface with PEGs of variable length, either on all the terminal functions [8], or on some of the terminal functions

(stochastic grafting) [9]. Another way to increase the solubility of dendrimers in water, based on using a PEG polymer as core, with a dendritic wedge (dendron) grown on each side, has been developed. A non-exhaustive list of references concerning hybrid PEG-cored dendrimers comprises PAMAM dendrons [10–13], 1,3,5-triazine dendrons [14], L-Lysine dendrons [15–17], ether dendrons [18,19], ester dendrons [20,21], small citric acid dendrons [22], and special dendrons synthesized through orthogonal amine/epoxy and thiol-yne chemistries [23], but there is no example to date of PEG-cored phosphorus dendrimers.

However, phosphorus dendrimers [24], and especially phosphorhydrazone dendrimers have outstanding properties for catalysis [25], materials [26], and above all for nanomedicine [27,28]. The synthesis of such dendrimers has been already carried out starting from different cores, in particular P(S)Cl₃ [29], N₃P₃Cl₆ [30,31], which can be specifically functionalized [32], but also an octaldehyde phthalocyanine [33,34], or a fluorescent linker [35]. This paper will present our results concerning the synthesis of the first PEG-cored phosphorhydrazone dendrimers, and representative examples of their surface functionalization with nitrogen ligands. Indeed, such type of ligands is able to complex diverse metals, and was recently found suitable for catalyzing

* Corresponding author at: Laboratoire de Chimie de Coordination (LCC), CNRS UPR8241, 205 Route de Narbonne, 31077 Toulouse Cedex 4, France.

E-mail address: anne-marie.caminade@lcc-toulouse.fr (A.-M. Caminade).

¹ H. Dib and J. Rebière contributed equally to this work.

<https://doi.org/10.1016/j.rechem.2022.100304>

Received 22 December 2021; Accepted 9 February 2022

Available online 12 February 2022

2211-7156/© 2022 The Author(s).

Published by Elsevier B.V. This is an open access article under the CC BY-NC-ND license

(<http://creativecommons.org/licenses/by-nc-nd/4.0/>).

Heck and Sonogashira couplings when complexing PdCl₂ on the periphery of phosphorhydrazone dendrimers [36]. The same family of dendrimers functionalized with nitrogen ligands also displays original anti-cancer properties when complexing CuCl₂ [37], or [AuCl₂⁺] [AuCl₄⁻] [38], but not FeCl₂ [39]. However, none of these complexes is soluble in water, which precludes real practical use in nanomedicine. We report here our efforts to try to increase the solubility in water of dendritic nitrogen ligands by introducing a PEG of variable length at the core.

Results and discussions

Synthesis of PEG-cored phosphorhydrazone dendrimers

Dihydroxy polyethylene glycols (PEGs) with different molar masses were chosen as core (**1a**, **1b**, and **1c**, Scheme 1). Small, medium, and large PEGs, with MW = 300, 1000, and 2000 g mol⁻¹ (in average, 6, 22 and 45 O-CH₂-CH₂- units, respectively) were selected. As the classical synthesis of phosphorus dendrimers necessitates at one step the presence of aldehydes [29], we choose a di-(piperazinoaldehyde) (compound **2**) built from 2,4,6-trichloro-1,3,5-triazine, which was previously used as precursor of a building block suitable for the synthesis of original dendrimers. The presence of such structure (compound **2**) induces special color features, as the dendritic structures in which it is incorporated are either yellow or orange/red, depending on the synthetic step considered, and on the environment [40].

The reaction was carried out with 2 equiv. of the dialdehyde **2**, and one equiv. of the dihydroxy-PEGs **1a-c**, with KOH as base in refluxing dichloromethane, to afford generation zero dendrimers **3a-G'**₀, **3b-G'**₀ and **3c-G'**₀ (Scheme 1). The reaction was monitored by ¹³C {¹H} NMR, showing the disappearance of the signals at 61 and 72.5 ppm corresponding to the carbon atoms in the α and β position of the PEG hydroxyl group, in favour of the appearance of signals at 66.9 and 69.2 ppm. Concurrently, ¹H NMR spectra of the three compounds **3a-G'**₀, **3b-G'**₀ and **3c-G'**₀ showed signals related to the aldehyde functions at ca 9.82 ppm, aromatic protons at ca 6.95 and 7.80 ppm and the signals for the piperazine protons at ca 3.48 and 4.00 ppm. The PEG chain displayed a broad signal between 3.4 and 3.8 ppm, whatever the chain length was. The ¹³C {¹H} NMR showed the presence of two signals for the carbon atoms of the 1,3,5-triazine ring, one at 170.75 ppm for the carbon atom bearing the oxygen atom and the other at 165.6 ppm for the carbon atoms bearing the piperazine cycles. Compounds, **3a-G'**₀, **3b-G'**₀ and **3c-G'**₀ were isolated in 79%, 65% and 87% yield, respectively. Deceptively, dendrimer **3a-G'**₀ with a PEG₃₀₀ as core is not soluble in water, but dendrimers **3b-G'**₀ and **3c-G'**₀ with PEG₁₀₀₀ and PEG₂₀₀₀ as core, respectively, are water-soluble.

As in the "classical" synthesis of phosphorus dendrimers [29], the next step involved the condensation of the phosphorhydrazone **4** on the aldehyde functions of compounds **3a-c-G'**₀, in chloroform. The reaction was monitored with ³¹P {¹H} NMR, by the appearance of a singlet at ca 63 ppm, and with ¹H NMR, by the disappearance of the aldehyde proton signal. The synthesis of the first generation dendrimer **3a-G**₁ was

performed at room temperature for 24 h, whereas the synthesis of the first generation dendrimers **3b-G**₁ and **3c-G**₁ required heating at 40° C for three weeks (Scheme 2). After completion of the reaction and purification, these first generation dendrimers were isolated in 56%, 68% and 71% yield, respectively.

These differences, i.e. the increase in temperature but also time to perform the reaction, could be explained by the fact that increasing the size of PEGs can create different geometries (Fig. 1), as already proposed [14]. It is obvious that the terminal functions should be more easily accessible for the PEG₃₀₀ core than for the PEG₂₀₀₀ core, which could hide the functional groups.

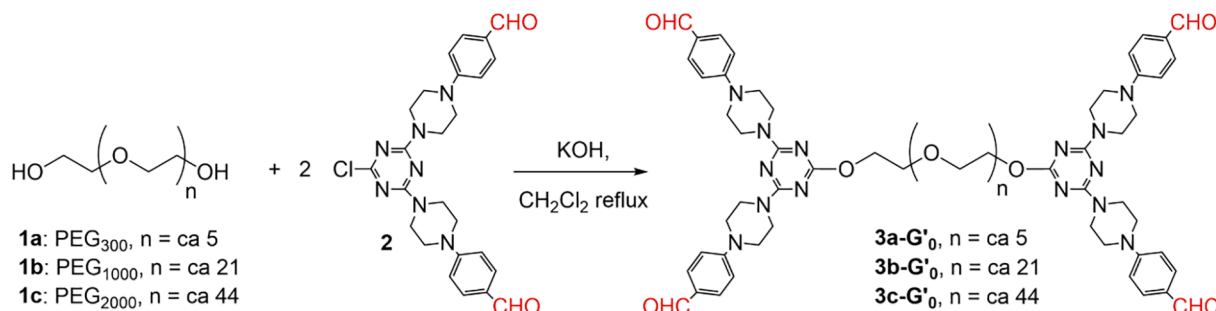
In view of the very long reaction times needed for the synthesis of dendrimers **3b-G**₁ and **3c-G**₁, only **3a-G**₁ was used for the preparation of higher generation dendrimers. The chlorine atoms of **3a-G**₁ were substituted with 8 equiv. of 4-hydroxybenzaldehyde **5**, in the presence of cesium carbonate, at room temperature in THF, to afford dendrimer **3a-G'**₁, functionalized with 8 aldehydes (Scheme 3). The reaction was monitored by ³¹P {¹H} NMR, which showed that the signal corresponding to the phosphorus atoms in the P(S)Cl₂ group at 63.12 ppm disappeared in favor of a more shielded signal at 60.85 ppm. Furthermore, the signal for aldehyde protons was clearly observable in ¹H NMR at 9.95 ppm. Dendrimer **3a-G'**₁ was obtained as yellow-orange solid with 65% yield.

The 2nd generation dendrimer **3a-G**₂ was obtained by condensation reaction of the phosphorhydrazone **4** on the aldehyde functions of **3a-G'**₁, as was done previously for the synthesis of **3a-G**₁ from **3a-G'**₀ (Scheme 4). The reaction was followed by ¹H NMR, via the disappearance of the aldehyde signal. Dendrimer **3a-G**₂ is characterized in particular by the presence of two singlets in ³¹P {¹H} NMR, at 62.04 ppm, corresponding to the internal layer of phosphorus atoms, and at 63.03 ppm corresponding to the terminal P(S)Cl₂ functions. Dendrimer **3a-G**₂ was obtained as a red solid in 86% yield.

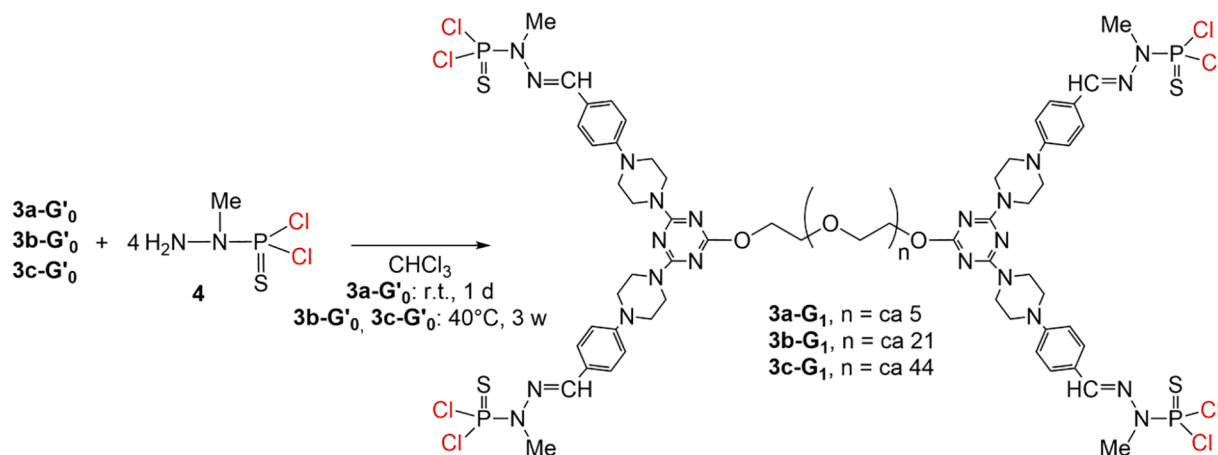
The synthesis of the second generation dendrimer bearing aldehyde terminal functions, **3a-G'**₂ was carried out by substituting the chlorine atoms of **3a-G**₂ by 16 equiv. of 4-hydroxybenzaldehyde **5** in the presence of cesium carbonate (Scheme 5), as done previously for the synthesis of **3a-G'**₁ from **3a-G**₁. The reaction was followed by ³¹P {¹H} NMR spectroscopy, which displayed a shielding of the signal of the phosphorus atoms at the periphery of the dendritic structure, from 63.03 ppm for **3a-G**₂ (P(S)Cl₂) to 60.55 ppm for **3a-G'**₂ (P(S)(OAr)₂). The obtained compound **3a-G'**₂ was isolated as red solid in 62% yield. In view of the facility of the reactions carried out at each step, higher generations could be obtained, but we decided to stop the growing of this hybrid dendrimer/polymer at this step.

Functionalization of the surface of PEG-cored dendrimers with N,N-ligands

We have already shown the usefulness of diverse types of nitrogen ligands for catalysis and for biology [41], as indicated in the introduction. In a first attempt, we have functionalized the PEG-cored dendrimers bearing aldehyde terminal functions with 2-hydrazinopyridine, **6**. The condensation reactions were first carried out with the generation



Scheme 1. Synthesis of generation zero of PEG-cored dendrimers.



Scheme 2. Synthesis of 1st generation PEG-dendrimers.

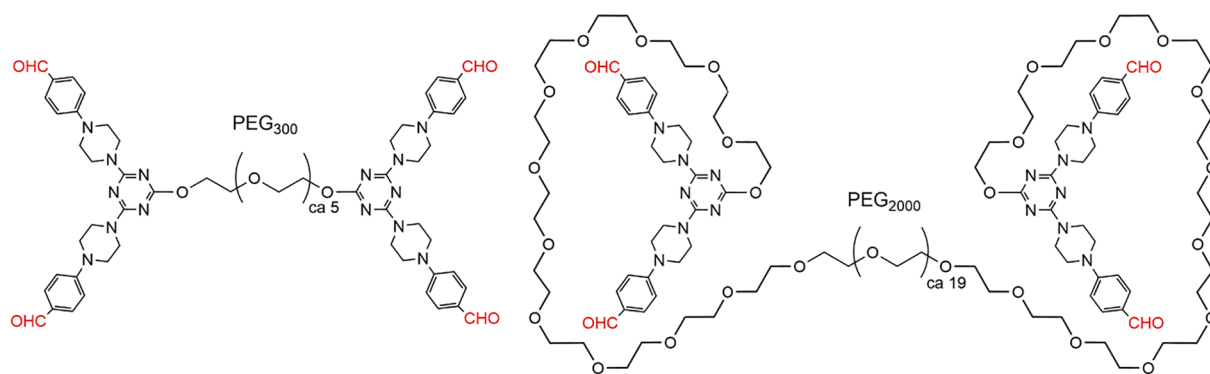
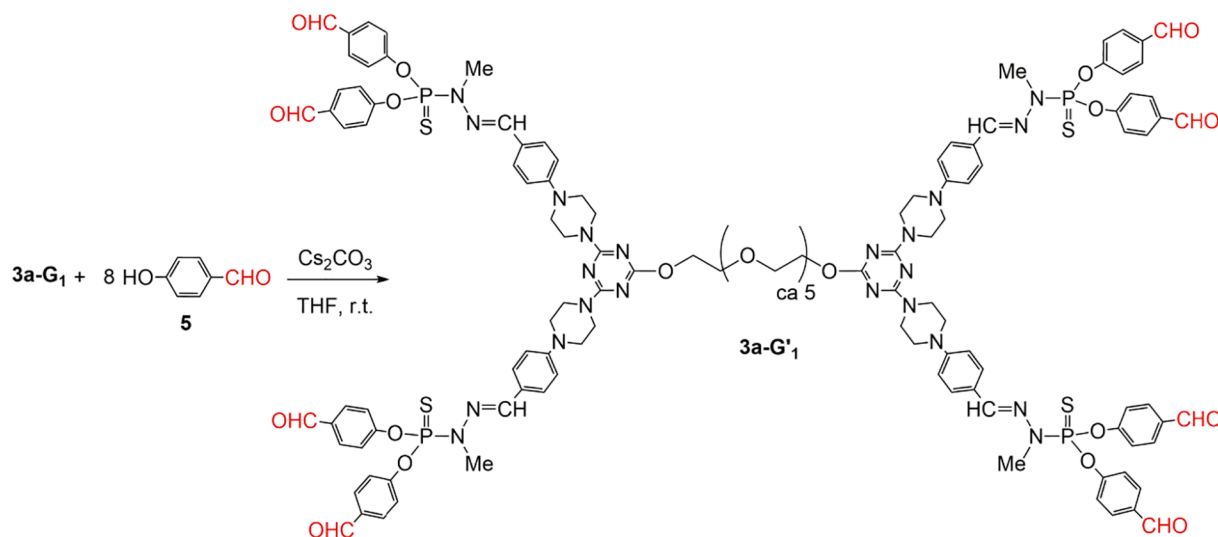
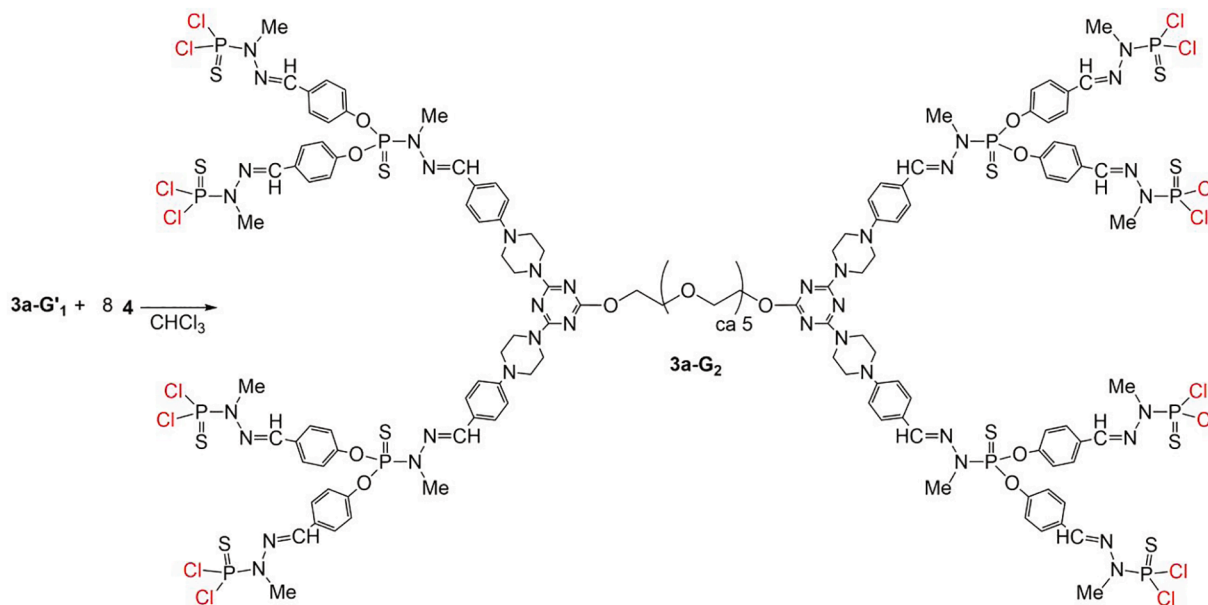
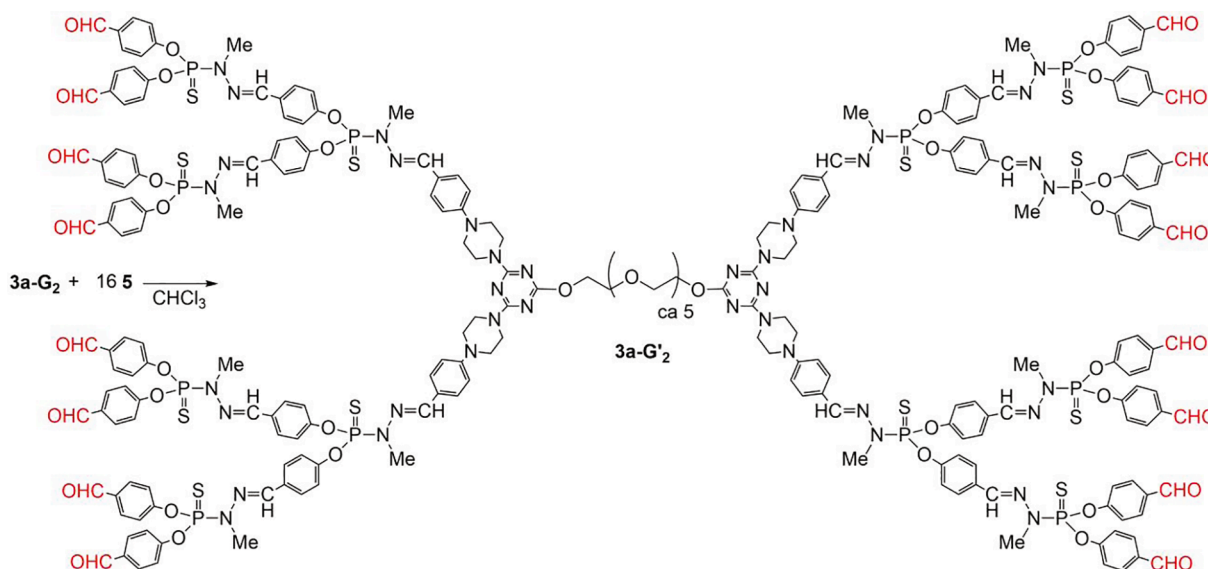


Fig. 1. Possible different geometries of PEG chains, depending on their length, and showing a possible shielding of the functional groups by the long PEG chain.

Scheme 3. Synthesis of dendrimer $3a-G'_1$.

zero dendrimers $3a-c-G'_0$, by heating in THF, to afford compounds $7a-c-G'_0$ (Scheme 6). Under these conditions, the condensation was relatively fast (2 days) with dendrimer $3a-G'_0$, but very slow with dendrimers $3b-G'_0$ which needed 9 days at 40 °C and above all with dendrimer $3c-G'_0$, which needed 9 days at 40 °C, followed by 10 days at reflux, to go to completion. Such behavior is reminiscent to what we have observed for the condensation of the same dendrimers with the phosphorhydrazide 4,

and is certainly due here also to the shielding of the active functions by the long PEG chain. The reaction was monitored with 1H NMR by the disappearance of the characteristic signals of aldehyde functions at 9.82 ppm. Compounds $7a-c-G'_0$ were characterized with $^{13}C \{^1H\}$ NMR by a signal at 126.2 ppm, characteristic of the carbon atom of the $CH = N$ bond formed and a signal at 7.74 ppm in 1H NMR for the corresponding proton. Compounds $7a-G'_0$, $7b-G'_0$, and $7c-G'_0$ were isolated in 75%,

Scheme 4. Synthesis of the second generation dendrimer **3a-G₂**.Scheme 5. Synthesis of dendrimer **3a-G'₂**.

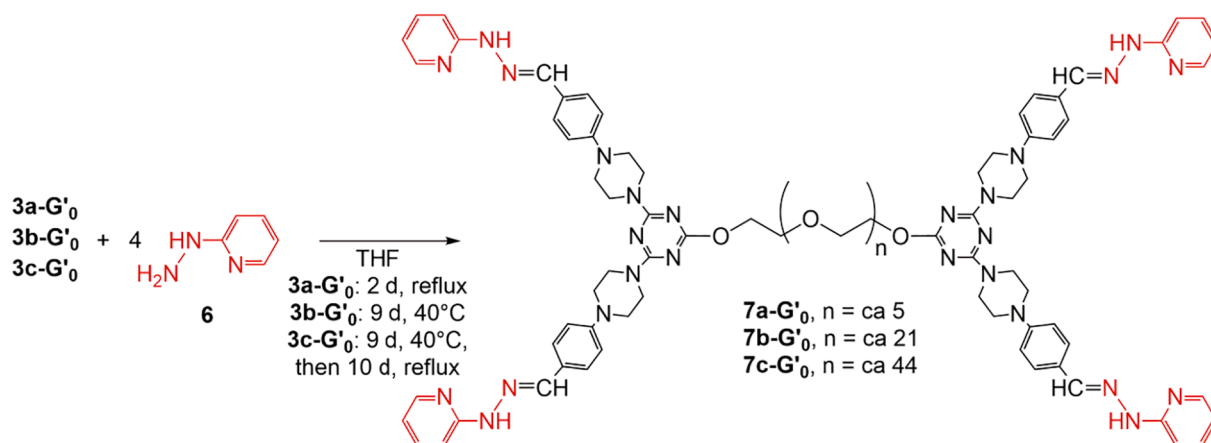
98%, and 99% yield, respectively.

The condensation reaction of 2-hydrazinopyridine with the 1st generation dendrimer **3a-G'₁** was also carried out at room temperature, but it needed 7 days to afford **7a-G'₁** (Scheme 7). The reaction was monitored by $^{31}\text{P}\{^1\text{H}\}$ NMR spectroscopy, which showed a slight signal shielding, from 60.85 ppm to 60.25 ppm. The ^1H NMR spectrum displayed both the disappearance of the signal corresponding to the aldehyde functions, and the appearance of the signal of the $\text{CH}=\text{N}$ at 7.92 ppm. Dendrimer **7a-G'₁** was isolated as a red powder in 93 % yield.

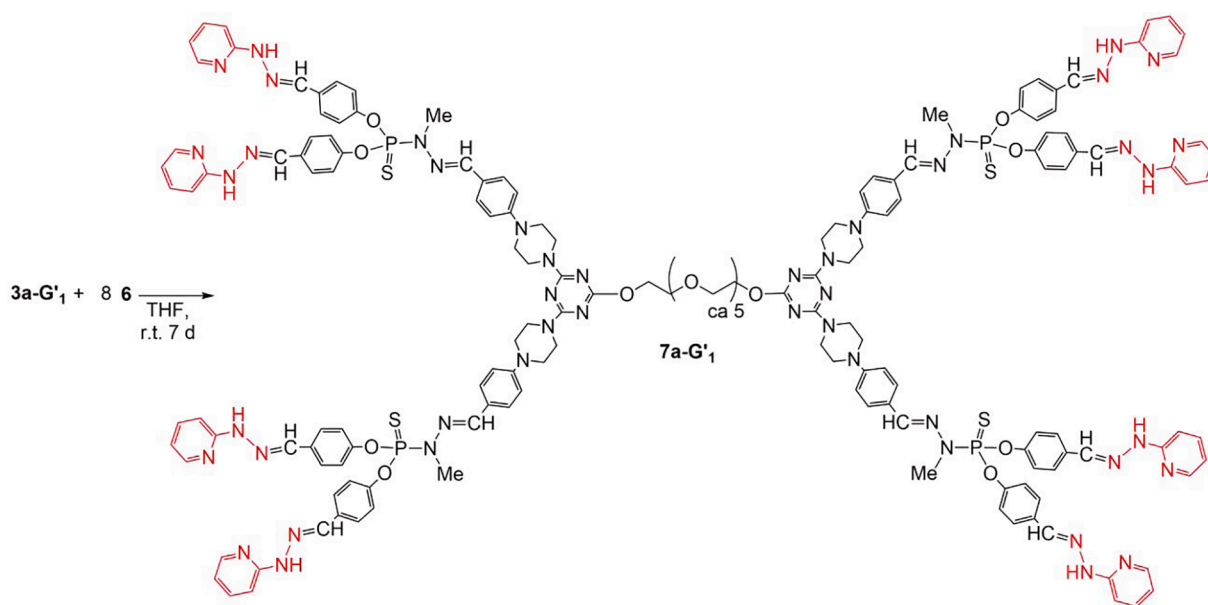
The second type of reaction used for grafting nitrogen ligands on the PEG-cored dendrimers was carried out with the phenol **8**, synthesized by condensation reaction, between tyramine and di-2-pyridyl ketone, as recently published [36]. Such phenol is suitable for the grafting to $\text{P}(\text{S})\text{Cl}_2$ terminal functions of the dendrimers. The reaction was performed with dendrimer **3a-G₁**, in THF, in the presence of cesium carbonate, at room temperature, to afford dendrimer **9a-G₁** in 65% yield (Scheme 8). The reaction was monitored by $^{31}\text{P}\{^1\text{H}\}$ NMR, which showed the

shielding of the signal from 62.56 ppm to 60.45 ppm in CDCl_3 .

Complexation of copper has been attempted with some of the dendrimers bearing the iminopyridine N,N-ligands as terminal functions. Dendrimers **7a-G'₀** and **7a-G'₁** were dissolved in DMF, then an excess of a solution of CuCl_2 was gradually added (8.78 equiv. for **7a-G'₀**, for maximum 8 equiv. needed; 17.37 equiv. for **7a-G'₁**, for maximum 12 equiv. needed), considering that both the pyridine imine, and the piperazine could complex copper. Afterwards, this excess of copper was removed by washing with methanol until a colorless washing solution was obtained. The copper species were obtained as dark brown powders in good yields, 91 and 92% yield, respectively. They are slightly soluble in DMF and DMSO, but not in water or in other organic solvents. Their NMR spectra could not be recorded, as these species are both very poorly soluble as indicated, and paramagnetic, due to the presence of copper (II). Thus, these species have neither been characterized, nor been tested in biology.



Scheme 6. Condensation reaction of 2-hydrazinopyridine with the PEG-cored dendrimers of generation zero.



Scheme 7. Condensation reaction of 2-hydrazinopyridine with the first generation dendrimer $3a-G'_1$.

Conclusion

The first PEG-cored phosphorus dendrimers have been synthesized, characterized, and functionalized with nitrogen ligands. A large difference in the reactivity and in the properties has been observed, depending on the PEG length. Dendritic structures incorporating as core the smallest PEG (PEG₃₀₀) can be easily processed. No difference in reactivity was observed, when compared with “classical” phosphorhydrazone dendrimers built from a N₃P₃ core. However, deceptively PEG₃₀₀ is not large enough to induce solubility in water. The reactions of compounds in which the largest PEGs (medium PEG₁₀₀₀ and large PEG₂₀₀₀) were used as core needed several weeks of heating to go to completion, most probably because the terminal “reactive” functions are buried inside the PEG, and thus shielded. So, despite the fact that the presence of a larger PEG could induce solubility in water, at least for the smallest synthesized dendritic wedges, the lengthy reaction time precludes a really practical use of such PEG as core. Incorporation of PEG moiety -only- as core of phosphorus dendrimers was inadequate to attain sufficient water solubility. Probably, it would be more advantageous to include PEG chain within the branches of dendrimers, which could potentially solve also the shielding of terminal functional groups by the long PEG chain. However, the resulting compounds would be more

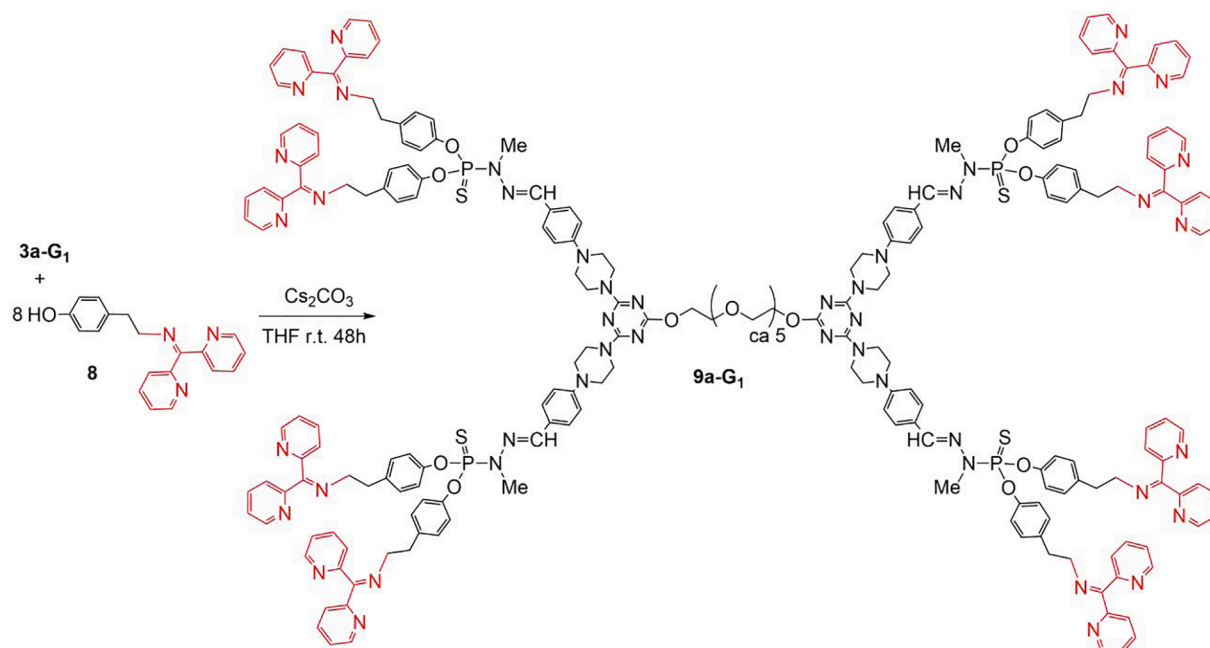
hyperbranched polymers than dendrimers.

On the other hand, due to the very poor solubility of the copper complexes formed with PEG-cored dendrimers, they could not be characterized and exploited. However, this work confirms the usefulness of the 1,3,5-triazine piperazine dialdehyde derivative (compound **2**) to afford colored compounds [40]. Indeed, even if present in relatively small quantities, such as in dendrimer $3a-G'_2$, a red powder could be obtained. In contrast, the synthesis of “normal” phosphorhydrazone dendrimers always leads to white powders. Therefore, in our group work is in progress to develop the design of new dendrimers incorporating synthon **2**, as potential tools for staining cells or for coloring hard surfaces.

Experimental section

General

All reactions were carried out in the absence of air using standard Schlenk techniques and vacuum-line manipulations. Commercial samples were used as received. All solvents were dried and purified with MBraun SPS 800 solvent purification system before use. ¹H, ¹³C, and ³¹P NMR measurements were performed on Bruker AV300 and AV400.

Scheme 8. Synthesis of dendrimer 9a-G₁.

Coupling constants are reported in Hz and chemical shifts in ppm relative to TMS for ^1H and ^{13}C NMR. Chemical shifts for ^{31}P NMR spectra are calibrated with phosphoric acid as an internal reference. The first-order peak patterns are indicated as s (singlet), d (doublet), t (triplet), q (quadruplet), qn (quintuplet). Complex non-first-order signals are indicated as m (multiplet); broad signals are indicated as br. ^{13}C NMR signals were assigned using Jmod, HMQC and HMBC sequences when required. Due to the presence of solvents inside the dendrimers, no reliable elemental analyzes could be obtained. The presence of solvents is necessary to ensure the solubility of the dendrimers in the next synthetic step. If they are removed, the dendrimers could not be solubilized again, even when using ultrasounds or heating. Compound 2 was synthesized as published [40], as well as compound 8 [36]. The numbering used for the assignment of NMR signals is shown in Fig. 2.

Synthesis of dendrimer 3a-G'₀

A solution of PEG₃₀₀, 1.0 g (3.34 mmol) in 10 mL of dichloromethane (DCM) was added dropwise to a solution of 20 mL of DCM containing 3.62 g of 2 (7.35 mmol) and 412 mg of KOH, at $T = 40^\circ\text{C}$. The reaction mixture was heated to reflux for 24 h. After thin layer chromatography (TLC) control, a filtration followed by column purification was performed (DCM/ethyl acetate 3/1 then 10% ethanol, v/v). After evaporation of the solvents under vacuum, 3.17 g of 3a-G'₀ as a reddish solid were obtained (79% yield). ^1H NMR (400 MHz, CDCl_3): δ (ppm) = 3.48 (br s, H-C₆, 16H), 3.66–3.71 (m, H-C_p, 16H), 3.85 (t, H-C₂, $^3J_{\text{HH}} = 5.1$ Hz, 4H), 4.00 (br s, H-C₅, 16H), 4.49 (t, H-C₁, $^3J_{\text{HH}} = 5.7$ Hz, 4H), 6.96 (d, H-C₈, $^3J_{\text{HH}} = 9$ Hz, 8H), 7.81 (d, H-C₉, $^3J_{\text{HH}} = 8.7$ Hz, 8H), 9.82 (s, H-C₁₁, 4H). ^{13}C { ^1H } NMR (100 MHz, CDCl_3): δ (ppm) = 42.8 (s, C₅), 47.0 (s, C₆), 65.7 (s, C₁), 69.2 (s, C₂), 70.6 (s, C_p), 113.7 (s, C₈), 127.5 (s, C₁₀),

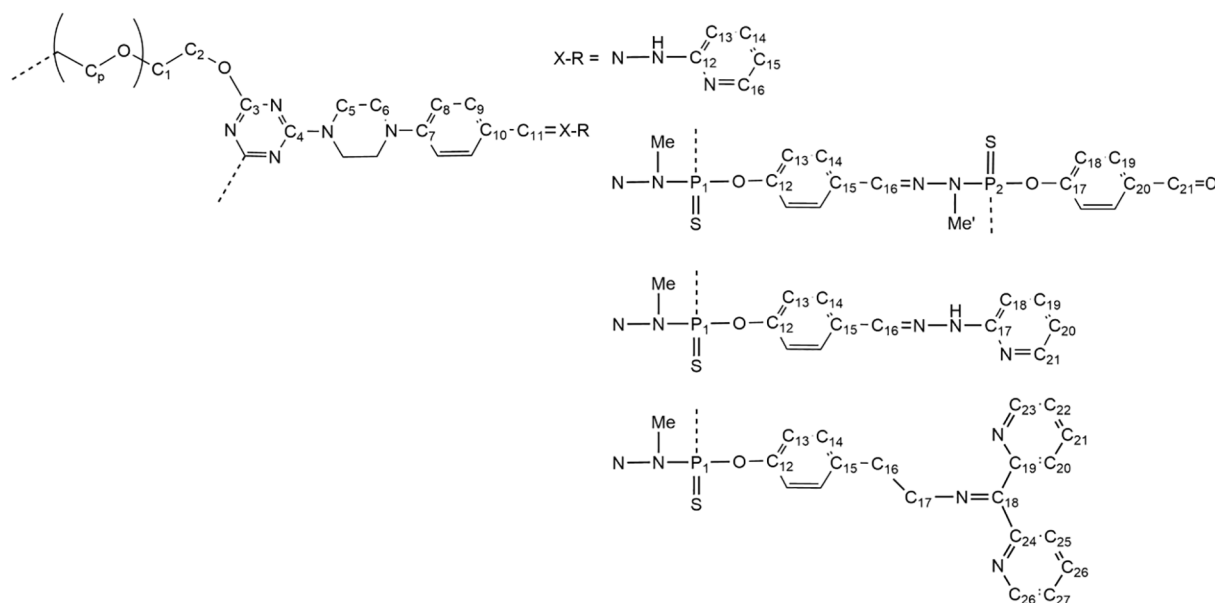


Fig. 2. Numbering used for NMR assignments.

131.9 (s, C₉), 154.8 (s, C₇), 166.0 (s, C₄), 170.9 (s, C₃), 190.4 (s, C₁₁).

Synthesis of dendrimer 3b-G⁰

A solution of PEG₁₀₀₀, 2.0 g (2.0 mmol) in 10 mL of DCM was added dropwise to a solution of 15 mL of DCM containing 2.01 g of **2** (4.1 mmol) and 230 mg of KOH, at T = 40° C. The reaction mixture was refluxed for 48 h. After separating the salts by centrifugation and evaporating the solvent, 10 mL of water were added to precipitate the excess of **2**. After filtration, the aqueous solution was placed in freeze dryer to remove water. 2.44 g of **3b-G⁰** were obtained as a yellow solid (65% yield). ¹H NMR (400 MHz, CDCl₃): δ (ppm) = 3.43 (br s, H-C₆, 16H), 3.60–3.67 (m, H-Cp, 80H), 3.78 (t, H-C₂, ³J_{HH} = 5.1 Hz, 4H), 3.94 (br s, H-C₅, 16H), 4.48 (t, H-C₁, ³J_{HH} = 5.7 Hz, 4H), 6.90 (d, H-C₈, ³J_{HH} = 9 Hz, 8H), 7.73 (d, H-C₉, ³J_{HH} = 8.7 Hz, 8H), 9.74 (s, H-C₁₁, 4H). ¹³C {¹H} NMR (100 MHz, CDCl₃): δ (ppm) = 42.7 (s, C₅), 46.8 (s, C₆), 66.9 (s, C₁), 69.2 (s, C₂), 70.6 (s, C_p), 113.6 (s, C₈), 127.3 (s, C₁₀), 131.9 (s, C₉), 154.8 (s, C₇), 166.2 (s, C₄), 170.8 (s, C₃), 190.4 (s, C₁₁).

Synthesis of dendrimer 3c-G⁰

A solution of PEG₂₀₀₀, 2.0 g (1.0 mmol) in 10 mL of DCM was added dropwise to a solution of 15 mL of DCM containing 1.01 g of **2** (2.05 mmol) and 115.7 mg of KOH, at T = 40° C. The reaction mixture was heated to reflux for 48 h. After separating the salts by centrifugation and evaporating the solvent, 10 mL of water were added to precipitate the excess of **2**. After filtration, the aqueous solution was placed in freeze dryer to remove water. 1.27 g of **3c-G⁰** as a yellow oil was obtained (87% yield). ¹H NMR (400 MHz, CDCl₃): δ (ppm) = 3.50 (br s, H-C₆, 16H), 3.76 (t, H-Cp, 172H), 3.83 (t, H-C₂, ³J_{HH} = 5.1 Hz, 4H), 4.0 (br s, H-C₅, 16H), 4.62 (t, H-C₁, ³J_{HH} = 5.7 Hz, 4H), 6.96 (d, H-C₈, ³J_{HH} = 9 Hz, 8H), 7.81 (d, H-C₉, ³J_{HH} = 8.7 Hz, 8H), 9.83 (s, H-C₁₁, 4H). ¹³C {¹H} NMR (100 MHz, CDCl₃): δ (ppm) = 42.9 (s, C₅), 46.8 (s, C₆), 65.7 (s, C₁), 69.2 (s, C₂), 70.6 (s, C_p), 113.8 (s, C₈), 127.7 (s, C₁₀), 131.9 (s, C₉), 154.6 (s, C₇), 164.5 (s, C₄), 169.8 (s, C₃), 190.4 (s, C₁₁).

Synthesis of dendrimer 3a-G₁

In a Schlenk tube containing activated molecular sieves, were placed 800 mg of **3a-G⁰** (0.66 mmol) in 5 mL of CHCl₃. 15.4 mL of compound **4** (c = 0.215 mol/L) were added dropwise at 0° C. Then the reaction was stirred at room temperature for 24 h. The reaction was monitored by ³¹P {¹H} NMR and ¹H NMR. After filtration to remove the molecular sieves, the filtrate was then poured into 200 mL of pentane under argon, to precipitate **3a-G₁**, recovered by filtration. The solid recovered was washed using 200 mL of pentane/ether solution (1/1, v/v). A final washing in 200 mL pentane/ether (1/1, v/v) was performed. 734 mg of the reddish **3a-G₁** product was then collected and dried under vacuum (yield 56%). ¹H NMR (400 MHz, CDCl₃): δ (ppm) = 3.35 (br s, H-C₆, 16H), 3.51 (d, CH₃, ³J_{HP} = 14.2 Hz, 12H), 3.66–3.73 (m, H-Cp, 20H), 3.85 (t, H-C₁, ³J_{HH} = 5.4 Hz, 4H), 4.00 (br s, H-C₅, 16H), 4.47 (t, H-C₂, ³J_{HH} = 5.7 Hz, 4H), 6.99 (d, H-C₈, ³J_{HH} = 8.4 Hz, 8H), 7.26 (s, H-C₁₁, 4H), 7.40 (d, H-C₉, ³J_{HH} = 8.4 Hz, 8H). ³¹P {¹H} NMR (CDCl₃): δ (ppm) = 63.40 (s, P). ¹³C {¹H} NMR (100 MHz, CDCl₃): δ (ppm) = 31.85 (d, ²J_{CP} = 12.9 Hz, CH₃), 43.0 (s, C₅), 48.3 (s, C₆), 68.0 (s, C₁), 69.2 (s, C₂), 70.6 (s, C_p), 115.6 (s, C₈), 125.5 (s, C₁₀), 128.7 (s, C₉), 142.2 (d, ³J_{CP} = 12.9 Hz, C₁₁), 152.3 (s, C₇), 166.1 (s, C₄), 170.5 (s, C₃).

Synthesis of dendrimer 3b-G₁

In a Schlenk tube containing activated molecular sieves, were placed 1.91 g of **3b-G⁰** (0.99 mmol) in 20 mL of CHCl₃. Then, 34 mL of compound **4** (c = 0.18 mol/L) were added dropwise at 0° C. The reaction was stirred at 40° C for 3 weeks. The reaction was monitored by ³¹P {¹H} NMR and ¹H NMR. Following the disappearance of aldehyde groups, the mixture was filtered, and the filtrate was poured in 200 mL of pentane,

in which **3b-G₁** precipitated, and was recovered by filtration. Then a washing was performed with 200 mL of pentane/ether solution (1/1, v/v). A final washing with 200 mL of pentane/ether (1/2, v/v) was performed. 1.8 g of a yellowish solid **3b-G₁** was then collected and dried under vacuum (68% yield). ¹H NMR (300 MHz, CDCl₃): δ (ppm) = 3.39 (br s, H-C₆, 16H), 3.51 (d, CH₃, ³J_{HP} = 14.3 Hz, 12H), 3.65–3.75 (m, H-Cp, 80H), 3.86 (br s, H-C₁, 4H), 4.14 (br s, H-C₅, 16H), 4.51 (br s, H-C₂, 4H), 7.18 (br s, H-C₈, 8H), 7.72 (m, H-C₁₁ and H-C₉, 12H). ³¹P {¹H} NMR (121.5 MHz, CD₂Cl₂): δ (ppm) = 63.16 (s, P).

Synthesis of dendrimer 3c-G₁

In a Schlenk tube containing activated molecular sieves, were placed 1.05 g of **3c-G⁰** (0.36 mmol) in 10 mL of CHCl₃. Then 12 mL of compound **4** (c = 0.18 mol/L) were added dropwise at 0° C. Stirring was maintained at a temperature of 40° C for 3 weeks. The reaction was monitored by ³¹P {¹H} NMR and ¹H NMR. Following the disappearance of aldehyde groups, the mixture was filtered, and the filtrate was poured in 200 mL of pentane to precipitate **3c-G₁**, which was then recovered by filtration. Then a washing was performed with 200 mL of pentane/ether solution (1/1, v/v). A final washing with 200 mL of pentane/ether (1/2, v/v) was performed. 0.94 g of a yellowish solid **3c-G₁** was then collected and dried under vacuum (71% yield). ¹H NMR (400 MHz, CDCl₃): δ (ppm) = 3.35 (br s, H-C₆, 16H), 3.51 (d, CH₃, ³J_{HP} = 12.7 Hz, 12H), 3.66–3.76 (m, H-Cp, 172H), 3.85 (br s, H-C₁, 4H), 4.04 (br s, H-C₅, 16H), 4.50 (br s, H-C₂, 4H), 7.03 (d, H-C₈, ³J_{HH} = 8.2 Hz, 8H), 7.69 (br s, H-C₁₁ and H-C₉, 12H). ³¹P {¹H} NMR (CDCl₃): δ (ppm) = 63.30 (s, P). ¹³C {¹H} NMR (100 MHz, CDCl₃): δ (ppm) = 31.85 (d, ²J_{CP} = 13.05 Hz, CH₃), 42.9 (s, C₅), 48.7 (s, C₆), 68.0 (s, C₁), 69.15 (s, C₂), 70.6 (s, C_p), 115.9 (s, C₈), 126.0 (s, C₁₀), 128.75 (s, C₉), 142.2 (d, ³J_{CP} = 18.7 Hz, C₁₁), 151.9 (s, C₇), 165.7 (s, C₄), 170.8 (s, C₃).

Synthesis of dendrimer 3a-G¹

494 mg (4.05 mmol) of hydroxybenzaldehyde **5** were placed in a Schlenk tube, with 1.56 g of Cs₂CO₃ in 10 mL of THF, to which were added dropwise 750 mg of **3a-G₁** (0.40 mmol) dissolved in 5 mL of THF. The reaction mixture was left stirring at room temperature for 3 h. After separating the salts by centrifugation, the solution was poured in 200 mL of pentane, to precipitate **3a-G¹**, which was then recovered by filtration. Then a washing was performed with 200 mL of MeOH. A final washing with 200 mL pentane/ether (1/1, v/v) was performed. 0.7 g of **3a-G¹** as a yellow-orange solid was then collected and dried under vacuum (65% yield). ¹H NMR (300 MHz, CDCl₃): δ (ppm) = 3.27 (br s, H-C₆, 16H), 3.51 (d, CH₃, ³J_{HP} = 11.05 Hz, 12H), 3.50–3.72 (br s, H-Cp, 16H), 3.78 (br s, H-C₁, 4H), 3.94 (br s, H-C₅, 16H), 4.46 (br s, H-C₂, 4H), 6.90 (br s, H-C₈, 8H), 7.40 (d, H-C₁₃, ³J_{HH} = 7.4 Hz, 16H), 7.57 (br s, H-C₉, 8H), 7.62 (br s, H-C₁₁, 4H), 7.85 (d, H-C₁₄, ³J_{HH} = 8.3 Hz, 16H), 9.92 (s, H-C₁₆, 8H). ³¹P {¹H} NMR (121.5 MHz, THF D₈): δ (ppm) = 60.5 (s, P). ¹³C {¹H} NMR (100 MHz, CDCl₃): δ (ppm) = 32.9 (d, ²J_{CP} = 13.25 Hz, CH₃), 43.0 (s, C₅), 48.2 (s, C₆), 65.4 (s, C₁), 69.0 (s, C₂), 70.0 (s, C_p), 115.5 (s, C₈), 122.05 (d, ³J_{CP} = 4.8 Hz, C₁₃), 125.5 (s, C₁₀), 128.3 (s, C₉), 131.4 (s, C₁₄), 133.6 (s, C₁₅), 140.9 (d, ³J_{CP} = 13.7 Hz, C₁₁), 152.0 (s, C₇), 155.3 (d, ²J_{CP} = 7.2 Hz, C₁₂), 165.8 (s, C₄), 170.5 (s, C₃), 190.9 (s, C₁₆).

Synthesis of dendrimer 3a-G₂

In a Schlenk tube containing activated molecular sieves, were placed 500 mg of **3a-G¹** (0.20 mmol) in 10 mL of CHCl₃. Then 8.2 mL of compound **4** (c = 0.215 mol/L) were added dropwise at 0° C. The reaction mixture was left stirring at room temperature for 24 h. The reaction was monitored by ³¹P {¹H} NMR and ¹H NMR. After filtration, the filtrate was then poured in 200 mL of pentane, to precipitate **3a-G₂**, which was then recovered by filtration. Then, a washing was performed with 200 mL of pentane/ether solution (1/1, v/v). 0.65 g of **3a-G₂** as a

reddish solid was then collected and dried under vacuum (86% yield). ^1H NMR (400 MHz, CDCl_3): δ (ppm) = 3.35 (br s, H-C₆, 16H), 3.37 (s, CH₃, $^3J_{\text{HP}}$ = 10.6 Hz, 12H), 3.60 (s, CH₃', $^3J_{\text{HP}}$ = 13.75 Hz, 24H), 3.60–3.75 (m, H-C_p, 16H), 3.85 (br s, H-C₁, 4H), 4.00 (s, H-C₅, 16H), 4.50 (br s, H-C₂, 4H), 6.96 (d, H-C₈, $^3J_{\text{HH}}$ = 7.5 Hz, 8H), 7.32 (d, H-C₁₃, $^3J_{\text{HH}}$ = 8.2 Hz, 16H), 7.55–7.80 (m, H-C₉, H-C₁₄, H-C₁₁, H-C₁₆, 36H). ^{31}P { ^1H } NMR (162.0 MHz, CDCl_3): δ (ppm) = 62.04 (s, P₁), 63.03 (s, P₂).

Synthesis of dendrimer 3a-G'2

132 mg (1.08 mmol) of hydroxybenzaldehyde **5** were placed in a Schlenk tube, with 422 mg of Cs_2CO_3 in 10 mL of THF, to which were added dropwise 200 mg of **3a-G**₂ (0.054 mmol) dissolved in 5 mL of THF. The reaction mixture was left stirring at room temperature for 7 days. After separating the salts by centrifugation, the solution was poured in 200 mL of pentane, to precipitate **3a-G**'₂, which was recovered by filtration. Then a washing of the solid with 200 mL pentane/MeOH (4/1, v/v) was performed. 174 mg of **3a-G**'₂ as a reddish solid were then collected and dried under vacuum (62% yield). ^1H NMR (400 MHz, CD_2Cl_2): δ (ppm) = 3.27 (br s, H-C₆, 16H), 3.32–3.47 (m, CH₃ and CH₃', 36H), 3.48–3.68 (m, H-C_p, 16H), 3.82–4.07 (m, H-C₁ and H-C₅, 20H), 4.48 (br s, H-C₂, 4H), 6.88 (br s, H-C₈, 8H), 6.95 (d, H-C₁₃, $^3J_{\text{HH}}$ = 8.7 Hz, 16H), 7.30 (d, H-C₁₃, $^3J_{\text{HH}}$ = 8.3 Hz, 16H), 7.41 (d, H-C₁₈, $^3J_{\text{HH}}$ = 6.9 Hz, 32H), 7.55–7.76 (m, H-C₉, H-C₁₄, H-C₁₆, H-C₁₁, 36H), 7.88 (d, H-C₁₉, $^3J_{\text{HH}}$ = 8.2 Hz, 32H), 9.96 (s, H-C₂₁, 16H). ^{31}P { ^1H } NMR (162.0 MHz, CD_2Cl_2): δ (ppm) = 60.42 (s, P₂), 62.60 (s, P₁).

Synthesis of dendrimer 7a-G'0

In a Schlenk tube containing activated molecular sieves, were placed 600 mg of **3a-G**'₀ (0.496 mmol) in 10 mL of THF. Then a solution of 238 mg of 2-hydrazinopyridine **6** (2.18 mmol) in 5 mL of THF was added dropwise. The reaction mixture was left stirring at reflux for 2 days. The reaction was monitored by ^1H NMR. After filtration, the filtrate was poured in 200 mL of pentane, to precipitate **7a-G**'₀. Then a washing of the solid was performed with 200 mL of pentane/ether solution (5/1, v/v). 586 mg of **7a-G**'₀ as a reddish solid were then collected and dried under vacuum (75% yield). ^1H NMR (300 MHz, CD_2Cl_2): δ (ppm) = 3.31 (br s, H-C₆, 16H), 3.55–3.65 (m, H-C_p, 16H), 3.81 (t, H-C₁, $^3J_{\text{HH}}$ = 7.5 Hz, 4H), 4.00 (br s, H-C₅, 16H), 4.46 (t, H-C₂, $^3J_{\text{HH}}$ = 8.4 Hz, 4H), 6.77 (m, H-C₁₅, 4H), 6.98 (d, $^3J_{\text{HH}}$ = 8.4 Hz, H-C₈, 8H), 7.34 (d, $^3J_{\text{HH}}$ = 8.4 Hz, H-C₁₃, 4H), 7.58–7.68 (m, H-C₉ and H-C₁₄, 12H), 7.74 (s, H-C₁₁, 4H), 8.13 (m, H-C₁₆, 4H), 8.43 (s, NH, 4H). ^{13}C { ^1H } NMR (75 MHz, CD_2Cl_2): δ (ppm) = 43.1 (s, C₅), 48.4 (s, C₆), 65.5 (s, C₂), 69.3 (s, C₁), 70.5 (s, C_p), 106.9 (s, C₁₃), 115.3 (s, C₁₅), 115.6 (s, C₈), 126.2 (s, C₁₀), 127.4 (s, C₉), 137.9 (s, C₁₁), 139.4 (s, C₁₄), 147.7 (s, C₁₆), 151.7 (s, C₁₂), 157.0 (s, C₇), 166.1 (s, C₄), 170.8 (s, C₃).

Synthesis of dendrimer 7b-G'0

In a Schlenk tube containing activated molecular sieves, were placed 530 mg of **3b-G**'₀ (0.277 mmol) in 10 mL of THF. Then a solution of 133 mg of 2-hydrazinopyridine **6** (1.22 mmol) in 5 mL of THF was added dropwise. The reaction mixture was left stirring at 40° C for 9 days. The reaction was monitored by ^1H NMR. After filtration, the filtrate was poured in 200 mL of pentane to precipitate **7b-G**'₀. Then a washing of the solid was performed with 200 mL of pentane/ether solution (1/2, v/v). 617 mg of **7b-G**'₀ as a yellow solid were then collected and dried under vacuum (98 % yield). ^1H NMR (400 MHz, CDCl_3): δ (ppm) = 3.23 (br s, H-C₆, 16H), 3.50–3.65 (m, H-C_p, 80H), 3.80 (t, H-C₁, $^3J_{\text{HH}}$ = 7.5 Hz, 4H), 3.94 (br s, H-C₅, 16H), 4.45 (t, H-C₂, $^3J_{\text{HH}}$ = 8.4 Hz, 4H), 6.69 (m, H-C₁₅, 4H), 6.96 (d, $^3J_{\text{HH}}$ = 8.5 Hz H-C₈, 8H), 7.31 (d, $^3J_{\text{HH}}$ = 8.4 Hz, H-C₁₃, 4H), 7.48–7.60 (m, H-C₉ and H-C₁₄, 12H), 7.71 (s, H-C₁₁, 4H), 8.08 (m, H-C₁₆, 4H), 9.30 (s, NH, 4H). ^{13}C { ^1H } NMR (100 MHz, CDCl_3): δ (ppm) = 43.0 (s, C₅), 48.6 (s, C₆), 61.7 (s, C₂), 69.2 (s, C₁), 70.5 (s, C_p), 107.4 (s, C₁₃), 115.05 (s, C₁₅), 115.8 (s, C₈), 126.7 (s, C₁₀), 127.5 (s, C₉),

138.0 (s, C₁₁), 139.5 (s, C₁₄), 147.2 (s, C₁₆), 151.5 (s, C₁₂), 157.2 (s, C₇), 165.9 (s, C₄), 170.7 (s, C₃).

Synthesis of dendrimer 7c-G'0

In a Schlenk tube containing activated molecular sieves, were placed 490 mg of **3c-G**'₀ (0.168 mmol) in 10 mL of THF. Then a solution of 82 mg of 2-hydrazinopyridine (0.751 mmol) in 5 mL of THF was added dropwise. The reaction mixture was left stirring at 40° C for 9 days then at reflux for 10 days. The reaction was monitored by ^1H NMR. After filtration, the filtrate was poured in 200 mL of pentane to precipitate **7c-G**'₀. Then a washing of the solid was performed with 200 mL of pentane/ether solution (2/1, v/v). 544 mg of **7c-G**'₀ as a yellow solid were then collected and dried under vacuum (99 % yield). ^1H NMR (300 MHz, CDCl_3): δ (ppm) = 3.30 (br s, H-C₆, 16H), 3.66 (m, H-C_p, 176H), 3.85 (t, H-C₁, $^3J_{\text{HH}}$ = 7.5 Hz, 4H), 4.00 (br s, H-C₅, 16H), 4.48 (t, H-C₂, $^3J_{\text{HH}}$ = 8.4 Hz, 4H), 6.73 (m, H-C₁₅, 4H), 6.96 (d, $^3J_{\text{HH}}$ = 8.3 Hz, H-C₈, 8H), 7.34 (d, $^3J_{\text{HH}}$ = 8.4 Hz, H-C₁₃, 4H), 7.55–7.65 (m, H-C₉ and H-C₁₄, 12H), 7.73 (s, H-C₁₁, 4H), 8.12 (br s, H-C₁₆, 4H), 8.40 (br s, NH, 4H). ^{13}C { ^1H } NMR (75 MHz, CDCl_3): δ (ppm) = 43.1 (s, C₅), 48.6 (s, C₆), 65.6 (s, C₂), 69.2 (s, C₁), 70.6 (s, C_p), 107.2 (s, C₁₃), 115.3 (s, C₁₅), 115.8 (s, C₈), 126.5 (s, C₁₀), 127.6 (s, C₉), 138.4 (s, C₁₁), 140.0 (s, C₁₄), 147.6 (s, C₁₆), 151.7 (d, C₁₂), 156.5 (s, C₇), 166.0 (s, C₄), 170.8 (s, C₃).

Synthesis of dendrimer 7a-G'1

In a Schlenk tube containing activated molecular sieves, were placed 199 mg of **3a-G**'₁ (0.070 mmol) in 10 mL of THF. Then a solution of 68 mg of 2-hydrazinopyridine **6** (0.623 mmol) in 5 mL of THF was added dropwise. The reaction mixture was left stirring at room temperature for 7 days. The reaction was monitored by ^1H NMR. After filtration, the filtrate was poured in 200 mL of pentane/ether (9/1, v/v), to precipitate **7a-G**'₁. Then a washing of the solid was performed with 200 mL of pentane/ether solution (6/4, v/v). 212 mg of **7a-G**'₁ as a reddish solid were then collected and dried under vacuum (93 % yield). ^1H NMR (300 MHz, THF, D₈): δ (ppm) = 3.36 (br s, H-C₆, 16H), 3.52 (d, CH₃, $^3J_{\text{HP}}$ = 10.8 Hz, 12H), 3.48–3.55 (m, H-C_p, 16H), 3.64 (t, H-C₁, $^3J_{\text{HH}}$ = 7.8 Hz, 4H), 3.97 (br s, H-C₅, 16H), 4.45 (t, H-C₂, $^3J_{\text{HH}}$ = 5.7 Hz, 4H), 6.68 (d, H-C₂₀, $^3J_{\text{HH}}$ = 8.4 Hz, 8H), 7.01 (d, H-C₁₃, $^3J_{\text{HH}}$ = 7.4 Hz, 16H), 7.30 (d, H-C₁₈, $^3J_{\text{HH}}$ = 8.4 Hz, 8H), 7.54 (d, $^3J_{\text{HH}}$ = 8.4 Hz, H-C₁₉, 8H), 7.65 (s, H-C₉, 8H), 7.68 (d, $^3J_{\text{HH}}$ = 8.4 Hz, H-C₁₄, 16H), 7.86 (s, H-C₁₁, 4H), 7.92 (s, H-C₁₆, 8H), 8.06 (d, H-C₂₁, $^3J_{\text{HH}}$ = 6 Hz, 8H), 9.90 (s, H-N₂, 8H). ^{31}P { ^1H } NMR (121.5 MHz, THF D₈): δ (ppm) = 60.25 (s, P).

Synthesis of dendrimer 9a-G1

180 mg of compound **8** (0.594 mmol) were placed in a Schlenk tube, with 387 mg of Cs_2CO_3 in 10 mL of THF, to which were added dropwise 122 mg of **3a-G**₁ (0.066 mmol) dissolved in 5 mL of THF. The reaction mixture was left stirring at room temperature for 48 h. The reaction was monitored by ^{31}P { ^1H } NMR. After separating the salts by centrifugation, the solution was poured in 200 mL of pentane, to precipitate **9a-G**₁. Then a washing of the solid was carried out in 200 mL pentane/ether (3/1, v/v). 171 mg of **9a-G**₁ as a reddish solid were then collected and dried under vacuum (65% yield). ^1H NMR (400 MHz, CDCl_3): δ (ppm) = 2.83 (t, H-C₁₆, $^3J_{\text{HH}}$ = 7.5 Hz, 16H), 3.49 (br s, H-C₆, 16H), 3.51–3.72 (m, CH₃, H-C₁₇, H-C_p, 44H), 3.84 (t, $^3J_{\text{HH}}$ = 5.1 Hz, H-C₂, 4H), 4.0 (br s, H-C₅, 16H), 4.49 (t, H-C₁, $^3J_{\text{HH}}$ = 5.7 Hz, 4H), 6.64 (d, H-C₂₀, $^3J_{\text{HH}}$ = 8.4 Hz, 8H), 6.93 (d, H-C₈, $^3J_{\text{HH}}$ = 8.4 Hz, 8H), 6.95 (d, H-C₁₃, $^3J_{\text{HH}}$ = 9 Hz, 16H), 7.12 (d, H-C₂₂, $^3J_{\text{HH}}$ = 7.8 Hz, 8H), 7.35–7.44 (m, H-C₂₇, 8H), 7.80–7.91 (m, H-C₉, H-C₁₁, H-C₂₆, H-C₂₁, 28H), 8.17 (d, H-C₂₅, $^3J_{\text{HH}}$ = 7.8 Hz, 8H), 8.43 (d, H-C₂₈, $^3J_{\text{HH}}$ = 4.1 Hz, 8H), 8.63 (d, H-C₂₃, $^3J_{\text{HH}}$ = 4.1 Hz, 8H). ^{31}P { ^1H } NMR (162.5 MHz, CDCl_3): δ (ppm) = 60.45 (s, P). ^{13}C { ^1H } NMR (100 MHz, CDCl_3): δ (ppm) = 32.9 (d, $^2J_{\text{CP}}$ = 13.25 Hz, CH₃), 36.3 (s, C₁₆), 42.3 (s, C₅), 46.9 (s, C₆), 55.6 (s, C₁₇), 65.8 (s, C₁), 69.2 (s, C₂), 70.6 (s, C_p), 114.3 (s, C₈), 115.6 (s, C₁₃), 121.9 (s, C₂₀),

123.6 (s, C₂₅), 124.2 (s, C₂₇), 128.7 (s, C₁₀), 130.7 (s, C₁₅), 131.7 (s, C₉), 136.0 (s, C₂₁), 136.4 (s, C₁₁, C₂₆), 148.4 (s, C₂₃), 149.3 (s, C₂₈), 154.6 (s, C₇), 155.4 (s, C₁₉), 156.0 (s, C₂₄), 156.8 (s, C₁₂), 165.6 (s, C₄), 166.6 (s, C₁₈), 170.5 (s, C₃).

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

Acknowledgments

Thanks are due to the CNRS for financial support, to the ANR DendSwitch, and to TOUBKAL (grant 41405WE to O.A.).

References

- A.-M. Caminade C.-O. Turrin R. Laurent A. Ouali B. Delavaux-Nicot Dendrimers: Towards Catalytic, Material and Biomedical Uses 2011 John Wiley & Sons Ltd. Chichester, UK 10.1002/9781119976530.
- M. Malkoch, S.G. Gallego, Dendrimer Chemistry: Synthetic Approaches Towards Complex Architectures, RSC Publishing, UK (2020), <https://doi.org/10.1039/9781788012904>.
- A.M. Caminade, S. Fruchon, C.O. Turrin, M. Poupot, A. Ouali, A. Maraval, M. Garzoni, M. Maly, V. Furer, V. Kovalenko, J.P. Majoral, G.M. Pavan, R. Poupot, The key role of the scaffold on the efficiency of dendrimer nanodrugs, *Nat. Commun.* 6 (2015) 7722, <https://doi.org/10.1038/ncomms8722>.
- A.M. Caminade, J.P. Majoral, Positively charged phosphorus dendrimers. An overview of their properties, *New J. Chem.* 37 (11) (2013) 3358–3373, <https://doi.org/10.1039/c3nj00583f>.
- D.A. Tomalia, H. Baker, J. Dewald, M. Hall, G. Kallos, S. Martin, J. Roeck, J. Ryder, P. Smith, A new class of polymers – Starburst-dendritic macromolecules, *Polym. J.* 17 (1) (1985) 117–132, <https://doi.org/10.1295/polymj.17.117>.
- C. Wörner, R. Mulhaupt, Poly(nitrile-Functional and Polyamine-Functional Poly(Trimethylene Imine) Dendrimers, *Angewandte Chemie-International Edition in English* 32 (9) (1993) 1306–1308, <https://doi.org/10.1002/anie.199313061>.
- E.M.M. de Brabander-van den Berg, E.W. Meijer, Poly(Propylene Imine) Dendrimers - Large-Scale Synthesis by Heterogeneously Catalyzed Hydrogenations, *Angewandte Chemie-International Edition in English* 32 (9) (1993) 1308–1311, [https://doi.org/10.1002/\(ISSN\)1521-377310.1002/anie.v32:910.1002/anie.199313081](https://doi.org/10.1002/(ISSN)1521-377310.1002/anie.v32:910.1002/anie.199313081).
- A. Hameau, S. Fruchon, C. Bijani, A. Barducci, M. Blanzat, R. Poupot, G.M. Pavan, A.M. Caminade, C.O. Turrin, Theoretical and Experimental Characterization of Amino-PEG-Phosphonate-Terminated Polyphosphorhydrazone Dendrimers: Influence of Size and PEG Capping on Cytotoxicity Profiles, *Journal of Polymer Science Part A-Polymer Chemistry* 53 (6) (2015) 761–774, <https://doi.org/10.1002/pola.27501>.
- S.J. Guillaudeu, M.E. Fox, Y.M. Haidar, E.E. Dy, F.C. Szoka, J.M.J. Frechet, PEGylated dendrimers with core functionality for biological applications, *Bioconjug. Chem.* 19 (2) (2008) 461–469, <https://doi.org/10.1021/bc700264g>.
- T. Kim, H.J. Seo, J.S. Choi, H.S. Jang, J. Baek, K. Kim, J.S. Park, PAMAM-PEG-PAMAM: Novel triblock copolymer as a biocompatible and efficient gene delivery carrier, *Biomacromolecules* 5 (6) (2004) 2487–2492, <https://doi.org/10.1021/bm049563j>.
- T.I. Kim, J.U. Baek, J.K. Yoon, J.S. Choi, K. Kim, J.S. Park, Synthesis and characterization of a novel arginine-grafted dendritic block copolymer for gene delivery and study of its cellular uptake pathway leading to transfection, *Bioconjug. Chem.* 18 (2) (2007) 309–317, <https://doi.org/10.1021/bc0601525>.
- D. Yoyen-Ermis, K. Ozturk, M.A. Kursunel, C. Aydin, D. Ozkazanc, M.U. Gurbuz, A. Uner, M. Tulu, S. Calis, G. Esendagli, Tumor-Induced Myeloid Cells Are Reduced by Gemcitabine-Loaded PAMAM Dendrimers Decorated with Anti-Flt1 Antibody, *Mol. Pharm.* 15 (4) (2018) 1526–1533, <https://doi.org/10.1021/acs.molpharmaceut.7b01075.s001>.
- T. Pio Jude, E. Panchadcharam, K. Masilamani, Computing Zagreb and Randic Indices of PEG-Cored Dendrimers Used for Drug and Gene Delivery, *J. Chem.* 2020 (2020) 2821510, <https://doi.org/10.1155/2020/2821510>.
- H. Namazi, M. Adeli, Solution proprieties of dendritic triazine/poly (ethylene glycol)/dendritic triazine block copolymers, *Journal of Polymer Science Part A-Polymer Chemistry* 43 (1) (2005) 28–41, [https://doi.org/10.1002/\(ISSN\)1099-051810.1002/pola.v43:110.1002/pola.20471](https://doi.org/10.1002/(ISSN)1099-051810.1002/pola.v43:110.1002/pola.20471).
- Y.J. Fu, D.E. Nitecki, D. Maltby, G.H. Simon, K. Berejnoi, H.J. Raatschen, B.M. Yeh, D.M. Shames, Brasch, R. C, Dendritic iodinated contrast agents with PEG-cores for CT imaging: Synthesis and preliminary characterization, *Bioconjug. Chem.* 17 (4) (2006) 1043–1056, <https://doi.org/10.1021/bc060019c>.
- Y.J. Fu, H.J. Raatschen, D.E. Nitecki, M.F. Wendland, V. Novikov, L.S. Fournier, C. Cyran, V. Rogut, D.M. Shames, R.C. Brasch, Cascade polymeric MRI contrast media derived from poly(ethylene glycol) cores: Initial syntheses and characterizations, *Biomacromolecules* 8 (5) (2007) 1519–1529, <https://doi.org/10.1021/bm061141h>.
- C.C. Cyran, Y. Fu, H.-J. Raatschen, V. Rogut, B. Chaopathomkul, D.M. Shames, M. F. Wendland, B.M. Yeh, R.C. Brasch, New macromolecular polymeric MRI contrast agents for application in the differentiation of cancer from benign soft tissues, *J. Magn. Reson. Imaging* 27 (3) (2008) 581–589, [https://doi.org/10.1002/\(ISSN\)1522-258610.1002/jmri.v27:310.1002/jmri.21245](https://doi.org/10.1002/(ISSN)1522-258610.1002/jmri.v27:310.1002/jmri.21245).
- S.-C. Han, S.-H. Jin, J.-W. Lee, Synthesis of Dendrimer with PEG Core by Click Chemistry, *Polymer-Korea* 36 (3) (2012) 295–301, <https://doi.org/10.7317/pk.2012.36.3.295>.
- S. Kumari, K. Achazi, P. Dey, R. Haag, J. Dernecke, Design and Synthesis of PEG-Oligoglycerol Sulfates as Multivalent Inhibitors for the Scavenger Receptor LOX-1, *Biomacromolecules* 20 (3) (2019) 1157–1166, <https://doi.org/10.1021/acs.biomac.8b0141610.1021/acs.biomac.8b01416.s001>.
- I.V. Asharani, D. Thirumalai, Synthesis of Dendrimer-Encapsulated Silver Nanoparticles and Its Catalytic Activity on the Reduction of 4-Nitrophenol, *J. Chin. Chem. Soc.* 59 (11) (2012) 1455–1460, <https://doi.org/10.1002/jccs.v59.1110.1002/jccs.201100734>.
- K.A. Kosakowska, B.K. Casey, J.N.L. Albert, Y. Wang, H.S. Ashbaugh, S.M. Grayson, Synthesis and Self-Assembly of Amphiphilic Star/Linear-Dendritic Polymers: Effect of Core versus Peripheral Branching on Reverse Micelle Aggregation, *Biomacromolecules* 19 (8) (2018) 3177–3189, <https://doi.org/10.1021/acs.biomac.8b00679>.
- M.S. Alavidjeh, I. Haririan, M.R. Khorramzadeh, Z.Z. Ghane, M.S. Ardestani, H. Namazi, Anionic linear-globular dendrimers: biocompatible hybrid materials with potential uses in nanomedicine, *Journal of Materials Science-Materials in Medicine* 21 (4) (2010) 1121–1133, <https://doi.org/10.1007/s10856-009-3978-8>.
- L. Albertazzi, F.M. Mickler, G.M. Pavan, F. Salomone, G. Bardi, M. Panniello, E. Amir, T. Kang, K.L. Killops, C. Brauchle, R.J. Amir, C.J. Hawker, Enhanced Bioactivity of Internally Functionalized Cationic Dendrimers with PEG Cores, *Biomacromolecules* 13 (12) (2012) 4089–4097, <https://doi.org/10.1021/bm301384y>.
- J.P. Majoral, A.M. Caminade, Dendrimers containing heteroatoms (Si, P, B, Ge, or Bi), *Chem. Rev.* 99 (3) (1999) 845–880, <https://doi.org/10.1021/cr970414j>.
- A.-M. Caminade, R. Laurent, Homogeneous catalysis with phosphorus dendrimer complexes, *Coord. Chem. Rev.* 389 (2019) 59–72, <https://doi.org/10.1016/j.ccr.2019.03.007>.
- A.-M. Caminade, A. Beraa, R. Laurent, B. Delavaux-Nicot, M. Hajjaji, Dendrimers and hyper-branched polymers interacting with clays: fruitful associations for functional materials, *J. Mater. Chem. A* 7 (34) (2019) 19634–19650, <https://doi.org/10.1039/c9ta05718h>.
- A.-M. Caminade, Phosphorus dendrimers for nanomedicine, *Chem. Commun.* 53 (71) (2017) 9830–9838, <https://doi.org/10.1039/c7cc04949h>.
- A.-M. Caminade, C.O. Turrin, J.-P. Majoral, Phosphorus Dendrimers in Biology and Nanomedicine, Pan Stanford publishing, Singapore. (2018), <https://doi.org/10.1201/9781315110851>.
- N. Launay, A.-M. Caminade, R. Lahana, J.-P. Majoral, A general synthetic strategy for neutral phosphorus-containing dendrimers, *Angewandte Chemie-International Edition in English* 33 (1516) (1994) 1589–1592, [https://doi.org/10.1002/\(ISSN\)1521-377310.1002/anie.v33:15/1610.1002/anie.199415891](https://doi.org/10.1002/(ISSN)1521-377310.1002/anie.v33:15/1610.1002/anie.199415891).
- M. Slany, M. Bardaji, M.J. Casanova, A.M. Caminade, J.P. Majoral, B. Chaudret, Dendrimer surface-chemistry - Facile route to polyphosphines and their gold complexes, *J. Am. Chem. Soc.* 117 (38) (1995) 9764–9765, <https://doi.org/10.1021/ja00143a023>.
- N. Launay, A.M. Caminade, J.P. Majoral, Synthesis of bowl-shaped dendrimers from generation 1 to generation 8, *J. Organomet. Chem.* 529 (1–2) (1997) 51–58, [https://doi.org/10.1016/s0022-328x\(96\)06293-6](https://doi.org/10.1016/s0022-328x(96)06293-6).
- A.-M. Caminade, A. Hameau, J.-P. Majoral, The specific functionalization of cyclotriphosphazene for the synthesis of smart dendrimers, *Dalton Trans.* 45 (5) (2016) 1810–1822, <https://doi.org/10.1039/c5dt03047a>.
- J. Leclaire, Y. Coppel, A.-M. Caminade, J.-P. Majoral, Nanometric sponges made of water-soluble hydrophobic dendrimers, *Journal of the American Chemical Society* 126 (8) (2004) 2304–2305, <https://doi.org/10.1021/ja039684b.10.1021/ja039684b.s001>.
- J. Leclaire, R. Dagiral, S. Fery-Forgues, Y. Coppel, B. Donnadieu, A.M. Caminade, J. P. Majoral, Octasubstituted metal-free phthalocyanine as core of phosphorus dendrimers: A probe for the properties of the internal structure, *J. Am. Chem. Soc.* 127 (45) (2005) 15762–15770, <https://doi.org/10.1021/ja054797b>.
- T.R. Krishna, M. Parent, M.H.V. Werts, L. Moreaux, S. Gmouh, S. Charpak, A.-M. Caminade, J.-P. Majoral, M. Blanchard-Desce, Water-soluble dendrimeric two-photon tracers for in vivo imaging, *Angewandte Chemie-International Edition* 45 (28) (2006) 4645–4648, [https://doi.org/10.1002/\(ISSN\)1521-377310.1002/anie.v45:2810.1002/anie.200601246](https://doi.org/10.1002/(ISSN)1521-377310.1002/anie.v45:2810.1002/anie.200601246).
- Quentin Vanbellingen Paul Servin Anaïs Coinaud Sonia Mallet-Ladeira Régis Laurent Anne-Marie Caminade 26 8 2333 10.3390/molecules26082333.
- S. Mignani, N. El Brahmi, L. Eloy, J. Poupou, V. Nicolas, A. Steinmetz, S. El Kazzouli, M.M. Bousmina, M. Blanchard-Desce, A.M. Caminade, J.P. Majoral, T. Cresteil, Anticancer copper(II) phosphorus dendrimers are potent proapoptotic Bax activators, *Eur. J. Med. Chem.* 132 (2017) 142–156, <https://doi.org/10.1016/j.ejmech.2017.03.035>.
- S.M. Mignani, N. El Brahmi, S. El Kazzouli, R. Laurent, S. Ladeira, A.-M. Caminade, E. Pedziwiatr-Werbicka, E.M. Szweczyk, M. Bryszewska, M.M. Bousmina, T. Cresteil, J.-P. Majoral, Original Multivalent Gold(III) and Dual Gold(III)-Copper (II) Conjugated Phosphorus Dendrimers as Potent Antitumoral and Antimicrobial Agents, *Mol. Pharm.* 14 (11) (2017) 4087–4097, <https://doi.org/10.1021/acs.molpharmaceut.7b00771.10.1021/acs.molpharmaceut.7b00771.s001>.
- L. Chen, S. Mignani, A. Caminade, J. Majoral, Metal-based phosphorus dendrimers as novel nanotherapeutic strategies to tackle cancers: A concise overview, Wiley

- Interdisciplinary Reviews-Nanomedicine and Nanobiotechnology 11 (6) (2019), <https://doi.org/10.1002/wnan.v11.610.1002/wnan.1577>.
- [40] H. Dib, C. Rebut, R. Laurent, S. Mallet-Ladeira, A. Sournia-Saquet, M.B. Sarosi, E. Hey-Hawkins, J.-P. Majoral, B. Delavaux-Nicot, A.-M. Caminade, Ordered Layered Dendrimers Constructed from Two Known Dendrimer Families: Inheritance and Emergence of Properties, Chemistry-a European Journal 22 (31) (2016) 10736–10742, <https://doi.org/10.1002/chem.201601354>.
- [41] A.-M. Caminade, J.-P. Majoral, Phosphorus dendrimers functionalised with nitrogen ligands, for catalysis and biology, Dalton Trans. 48 (22) (2019) 7483–7493, <https://doi.org/10.1039/c9dt01305a>.