

Substratum influences uptake of radium-226 by plants

Frédéric Girault, Frédéric Perrier, Jean-Marc Ourcival, Roxane Ferry, Yves Gaudemer, François Bourges, Jean-François Didon-Lescot

▶ To cite this version:

Frédéric Girault, Frédéric Perrier, Jean-Marc Ourcival, Roxane Ferry, Yves Gaudemer, et al.. Substratum influences uptake of radium-226 by plants. Science of the Total Environment, 2021, 766, pp.142655. 10.1016/j.scitotenv.2020.142655 . hal-03562595

HAL Id: hal-03562595 https://hal.science/hal-03562595

Submitted on 13 Feb 2023 $\,$

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers. L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.



Distributed under a Creative Commons Attribution - NonCommercial 4.0 International License

Version of Record: https://www.sciencedirect.com/science/article/pii/S0048969720361842 Manuscript_a9e39ad406765877e608041765d1b800

1 Substratum influences uptake of radium-226 by plants

2

- Frédéric Girault^{1,*}, Frédéric Perrier¹, Jean-Marc Ourcival², Roxane Ferry¹,
- 4 Yves Gaudemer³, François Bourges⁴, Jean-François Didon-Lescot⁵
- 5
- ¹Physics of Natural Sites, Institut de Physique du Globe de Paris, Université de Paris, F-75005
 Paris, France.
- ⁸ ²Centre d'Ecologie Fonctionnelle et Evolutive (CEFE), UMR 5175, CNRS, Université Paul Valéry
- 9 Montpellier 3, EPHE, IRD, 1919 route de Mende, F-34293 Montpellier, Cedex 5, France.
- 10 ³Lithosphere Tectonics and Mechanics, Institut de Physique du Globe de Paris, Université de Paris,
- 11 CNRS UMR 7154, F-75005 Paris, France.
- ⁴Géologie Environnement Conseil, 30 rue de la République, F-09200 Saint-Girons, France.
- ¹³ ⁵Station de Recherches INRA/CNRS laboratoire ESPACE, 390 chemin des Boissières, F-30380
- 14 Saint Christol-lès-Alès, France.
- 15
- ^{*}Corresponding author. Email address: girault@ipgp.fr (F. Girault)
- 17
- 18 September 2020, revised manuscript submitted to *Science of the Total Environment*
- 19

20 Abstract

21

22 Radium-226, an alpha emitter with half-life 1600 years, is ubiquitous in natural environments. Present in rocks and soils, it is also absorbed by vegetation. The efficiency of ²²⁶Ra uptake by plants 23 24 from the soil is important to assess for the study of heavy metals uptake by plants, monitoring of 25 radioactive pollution, and the biogeochemical cycle of radium in the Critical Zone. Using a thoroughly validated measurement method of effective 226 Ra concentration (EC_{Ra}) in the laboratory, 26 we compare EC_{Ra} values of the plant to that of the closest soil, and we infer the ²²⁶Ra soil-to-plant 27 28 transfer ratio, R_{SP} , for a total of 108 plant samples collected in various locations in France. EC_{Ra} 29 values of plants range over five orders of magnitude with mean (min-max) of 1.66 ± 0.03 (0.020-113) Bq kg⁻¹. Inferred R_{SP} values range over four orders of magnitude with mean (min-max) of 30 31 0.0188 ± 0.0004 (0.00069–0.37). The mean R_{SP} value of plants in granitic and metamorphic context 32 $(0.073 \pm 0.002; n = 50)$ is significantly higher $(12 \pm 1 \text{ times})$ than that of plants in calcareous and 33 sedimentary context (0.0058 \pm 0.0002; n = 58). This difference, which cannot be attributed to a 34 systematic difference in emanation coefficient, is likely due to the competition between calcium and radium. In a given substratum context, the compartments of a given plant species show coherent 35 36 and decreasing R_{SP} values in the following order (acropetal gradient): roots > bark > branches and stems \approx leaves. Oak trees (*Quercus* genus) concentrate ²²⁶Ra more than other trees and plants in this 37 set. While this study clearly demonstrates the influence of substratum on the ²²⁶Ra uptake by plants 38 39 in non-contaminated areas, our measurement method appears as a promising practical tool to use for 40 (phyto)remediation and its monitoring in uranium- and radium-contaminated areas.

41

42 Abstract word count: 300

43

44 Manuscript word count: 8114

- **Keywords:** radon emanation; radium concentration; radium uptake; soil-to-plant transfer; pollution;
- 47 substratum

49 Highlights

- Effective ²²⁶Ra concentration (EC_{Ra}) from plants in non-contaminated areas
- *EC*_{Ra} of 108 plant samples and nearby local top soil samples measured
- Mean ²²⁶Ra soil-to-plant transfer ratio (R_{SP}) is 0.0188 ± 0.0004 (about 2%)
- Significant substratum effect with 12 ± 1 higher R_{SP} on granite than on limestone
- Competition between calcium and radium thus revealed in non-contaminated areas

55 **1. Introduction**

56

57 The Critical Zone (CZ), which is the domain between the top of the aquifer and the top of 58 the canopy, is the part of the Earth's surface sustaining life. One fundamental issue currently is to 59 assess the fluxes of energy and matter between the various compartments of the CZ and their 60 vulnerability (Brantley et al., 2007; Lin, 2010). Soil, a key compartment at the boundary between the lithosphere, biosphere, and atmosphere, sensitive to natural and unnatural forcing and essential 61 62 for mankind, often appears as the most polluted part of the CZ (Abrahams, 2002; Banwart, 2011). 63 Various pollutants, harmful to the biosphere, have been detected in soil, including heavy metals and 64 radionuclides (e.g., He and Walling, 1996; Manta et al., 2002; Douay et al., 2009; Li et al., 2014; 65 Girault et al., 2016). Vegetation growing on top of this soil is directly capable of capturing naturally 66 occurring elements as well as pollutants in its roots, tissues, flowers and fruits (e.g., Sheppard and 67 Evenden, 1988b; Carini, 1999; Ehlken and Kirchner, 2002). This uptake of tracer elements can also be a way to constrain the biogeochemical cycle of elements from the pedosphere to the biosphere in 68 69 the CZ.

70 Ubiquitous in the environment, the alkaline earth element radium is present in all CZ 71 compartments including water, rock, soil, and vegetation. The radium-226 isotope belongs to the 72 uranium-238 decay chain and is an alpha emitter with a half-life of 1600 ± 7 years (Duchemin et al., 1994). Several studies have focused on the uptake of ²²⁶Ra by plants, quantifying a transfer factor 73 (or concentration ratio) of ²²⁶Ra from the soil to the plant tissues (Simon and Ibrahim, 1990; 74 75 Sheppard et al., 2006; Vandenhove et al., 2009; Uchida et al., 2009). Predominantly, because of the 76 need to assess the radium content of agricultural products and the risk to the population, researches on the ²²⁶Ra uptake by plants have been carried out in ²²⁶Ra-contaminated areas (e.g., Simon and 77 78 Ibrahim, 1990 and references herein), such as former or operating uranium mining and milling sites 79 (Marple, 1980; Vasconcellos et al., 1987; Bettencourt et al., 1988; Ibrahim and Whicker, 1992;

80 Markose et al., 1993; Madruga et al., 2001; Blanco Rodríguez et al., 2002, 2010; Vera Tomé et al., 81 2002; 2003; Chen et al., 2005; Ryan et al., 2005; Soudek et al., 2007a,b, 2010; Carvalho et al., 82 2009; Černe et al., 2011; Medley et al., 2013; Hu et al., 2014; Medley and Bollhöfer, 2016; Yan and 83 Luo, 2016), phosphate fertilizer processing complexes (Paul and Pillai, 1986; Martínez-Aguirre and 84 Periáñez, 1998), radium salt factories (Bettencourt et al., 1988), depleted uranium ammunition sites 85 (Popovic et al., 2008), and other industrial units (Paul and Pillai, 1986). These studies have been complemented by experiments on artificially ²²⁶Ra-enhanced soils at the laboratory scale in pots or 86 87 at larger scale in lysimeter and field experiments (Gerzabek et al., 1998; Bunzl and 88 Trautmannsheimer, 1999; Vandenhove et al., 2005; Vandenhove and Van Hees, 2007; Nezami et 89 al., 2016), providing valuable insights for phytoremediation of contaminated areas (Thiry and Van 90 Hess, 2008; Vera Tomé et al., 2008, 2009; Abreu et al., 2014). By contrast, fewer studies have focused on ²²⁶Ra uptake by plants in non-contaminated areas or control sites (Sam and Eriksson, 91 92 1995; Ham et al., 2001; Karunakara et al., 2003; Pulhani et al., 2005; Popovic et al., 2008; da 93 Conceição et al., 2009; Lauria et al., 2009; Uchida and Tagami, 2009; Dragović et al., 2010; James 94 et al., 2011; Medley et al., 2013; Asaduzzaman et al., 2014; Al-Hamarneh et al., 2016; Mrdakovic 95 Popic et al., 2020). Despite these efforts, the uptake of ²²⁶Ra by plants thus remains insufficiently 96 well understood.

97 The first mechanism of uptake of elements by plants is passive and results from element concentration in groundwater and evapotranspiration. In this mechanism, ²²⁶Ra accumulates in 98 99 leaves and can also be excreted (e.g., Weis and Weis, 2004). In addition, uptake of elements by a 100 living plant is part of the metabolic cycle. In vascular plants, ²²⁶Ra uptake takes place through the various steps of the biological processes (e.g., Simon and Ibrahim, 1990): mobility of Ra^{2+} ions 101 including release and diffusion from the solid phase to the soil solution, exchange of available Ra²⁺ 102 ions by sorption/desorption onto the surfaces of roots, transport of Ra²⁺ ions across membranes in 103 the roots, and diffusion and translocation of ²²⁶Ra into plant tissues. Among other factors, it has 104

105 been recognized that radium uptake depends on the presence of other alkaline earth elements of 106 smaller ionic radius such as barium, strontium, calcium, and magnesium. This suggests that the 107 radium uptake by plants decreases as the concentration of other alkaline earth elements in soil 108 increases, and that incorporation by roots can saturate (Nathwani and Phillips, 1979; Marple, 1980; 109 Simon and Ibrahim, 1987). Generally, a bottom-to-top decreasing gradient (*i.e.*, acropetal) of 110 radium concentration has been observed in plant tissues, from roots to stems and from stems to 111 shoots (e.g., Simon and Ibrahim, 1990). As a testing hypothesis, we could consider that the variability in ²²⁶Ra uptake by plants may be due to different types of soil (substrate) and substratum, 112 113 in particular in non-contaminated areas. However, surprisingly, only small differences or no change 114 at all have been evidenced (e.g., Simon and Ibrahim, 1987, 1990; Vera Tomé et al., 2003; Pulhani et 115 al., 2005).

116 Several methods, such as gamma-ray spectrometry, alpha-particle spectrometry, liquid scintillation counting, and mass spectrometry, are commonly used to measure high ²²⁶Ra levels of 117 numerous materials. However, for low ²²⁶Ra levels in soil and for plant samples of relatively small 118 119 mass, such methods generally give large analytical uncertainty and data have remained limited. 120 Thus, to study ²²⁶Ra uptake by plants in non-contaminated areas, an alternative technique is desired, able to reach low ²²⁶Ra levels for large amount of samples, in a cost-effective manner. A candidate 121 122 high-sensitivity technique with well-constrained leakage effects and relatively small uncertainty for low ²²⁶Ra levels is available; it is based on radon-222 emanation, as already suggested thirty years 123 124 ago (Simon and Ibrahim, 1990). Radon-222 is a radioactive gas (half-life 3.8 days) produced by the alpha decay of ²²⁶Ra. The probability that a ²²⁶Ra atom decays into a ²²²Rn atom able to escape from 125 126 a medium is the emanation coefficient E (Tanner, 1964; Nazaroff, 1992). We define the 222 Rn emanating power of a given material by the effective 226 Ra concentration (*EC*_{Ra}), *i.e.* the product of 127 E by the bulk ²²⁶Ra concentration (C_{Ra}), expressed in Bq kg⁻¹ (Stoulos et al., 2004). Based on the 128 129 accumulation method, EC_{Ra} has been measured in various materials including soils (e.g.,

Markkanen and Arvela, 1992; Girault et al., 2011; Perrier et al., 2016b), rocks and building materials (*e.g.*, Przylibski, 2000; Righi and Bruzzi, 2006; Hassan et al., 2011; Girault et al., 2012), and more recently plants (Perrier et al., 2018). Lately, this method has been updated with a significantly higher sensitivity, allowing EC_{Ra} measurement of material with small mass (< 5 g) and low ²²⁶Ra levels (<10⁻¹⁴ g g⁻¹) (Girault et al., 2017a; Girault and Perrier, 2019). Measuring EC_{Ra} of plants using this high-sensitivity method is particularly suited in areas characterized by low ²²⁶Ra levels.

In this paper, to test the hypothesis of a possible effect of the substratum on the ²²⁶Ra uptake 137 by plants in non-contaminated areas, we present results of effective 226 Ra concentration (EC_{Ra}) in a 138 139 total of 108 plants collected at several non-contaminated sites in France that belong to two 140 geological subsets: granitic and metamorphic context, and calcareous and sedimentary context. Using measured EC_{Ra} of plants and of the nearby soil as well as representative ²²²Rn emanation 141 142 coefficients for plants and soils, we infer the 226 Ra soil-to-plant transfer ratio (R_{SP}). We show that, by contrast with the results available previously in contaminated areas, $R_{\rm SP}$ values strongly depend 143 144 on the substratum. We then discuss our results in terms of plant type, species, and compartment, and of the ²²⁶Ra concentration of soil-plant pairs. Consequences for the assessment of element fluxes in 145 146 the CZ are discussed in the conclusion.

147

148 **2. Material and method**

149

```
150 2.1. Plant and soil samples
```

151

A total of 108 plant samples and their associated nearby local soil samples were collected at different sites in France (Fig. 1). The plants mainly include deciduous trees (*Quercus robur, Fagus sylvatica, Castanea sativa, Tilia ×europaea, Prunus cerasus, Ficus carica, Fraxinus excelsior,*

155 *Corylus avellana*, *Aesculus hippocastanum*), evergreen trees (*Quercus ilex*, *Pinus pinaster*, *Abies* 156 *alba*), shrubs (*Buxus sempervirens*, *Phillyrea latifolia*, *Spartium junceum*, *Cytisus* 157 *oromediterraneus*), and ferns (*Pteridium aquilinum*). Other samples, such as market vegetables (n = 158, mosses (n = 5), mushrooms (n = 4), and algae (n = 1), were also collected.

159 This sample set is divided into two subsets based on geology: the A subset corresponding to plants growing in granitic and metamorphic context (Fig. 1; in red), and the B subset corresponding 160 161 to plants growing in calcareous and sedimentary context (Fig. 1; in blue). Samples from the A 162 subset (granitic and metamorphic context; n = 50) come from: the calco-alkaline biotite Hercynian 163 granite of the Mont-Lozère mountain in the Sapine watershed, Lozère department (44.356567°, 164 3.808383°, 1174 m); the Cévennes granite, a southward extension of the Mont-Lozère Pluton, near 165 Anduze, Gard department (44.077317°, 3.972301°, 201 m); the Hercynian Limousin (Massif 166 Central) Saint-Sylvestre granite near Ambazac, Haute-Vienne department (45.976100°, 1.392017°, 167 426 m); and the leptynite (laminated orthogneiss) of the Saint-Yrieix arc near Saint-Yrieix-la-Perche, Haute-Vienne department (45.474583°, 1.138317°, 304 m). Samples from the B subset 168 169 (calcareous and sedimentary context; n = 58) come from: the Kimmeridgian limestone near 170 Puéchabon, Hérault department (43.738395°, 3.591345°, 271 m); the Callovo-Oxfordian limestone 171 of the Pech Merle cave near Cabrerets, Lot department (44.507217°, 1.643633°, 292 m); the 172 Triassic gypsum mixed with rocky slope debris from the Oxfordian limestone near Anduze, Gard department (44.058753°, 3.979291°, 159 m); the Stampian Fontainebleau sand in the Paris Basin 173 174 near Vaugrigneuse, Essonne department (48.609549°, 2.104894°, 112 m); and the Burdigalian sand and marls of the Orléans formation near Chambon-la-Forêt, Loiret department (48.024976°, 175 176 2.261318°, 152 m).

177 At the selected sites, the radium concentration in groundwater can be considered negligible 178 and the main source of radium is the soil or the regolith. This hypothesis was confirmed at the sites 179 where 226 Ra concentration in groundwater could be measured (Perrier et al., 2016a): 4.8 ± 0.4 mBq

180 L^{-1} (n = 4) in the Sapine watershed, $4.2 \pm 0.3 \text{ mBq} L^{-1}$ (n = 9) in Anduze, < 1 mBq L^{-1} (n = 1) in 181 Pech Merle, $2.7 \pm 0.6 \text{ mBq} L^{-1}$ (n = 3) in Vaugrigneuse, and $15.4 \pm 1.6 \text{ mBq} L^{-1}$ (n = 2) in 182 Chambon-la-Forêt. These values belong to the lower range of ²²⁶Ra concentration in groundwater 183 (Girault et al., 2018).

184 Plant samples were directly collected in the field using clean scissors or shears and placed in 185 plastic bags. For some plants, several samples were collected from different compartments. Then, 186 plant samples, without washing with water, were gently dried in the laboratory at ambient room 187 temperature and cut in centimeter-size pieces. Top soil samples of mass 100-150 g were collected nearby each plant sample in a systematic manner (plant-soil matched pairs) using a clean shovel 188 189 and placed in plastic bags. At some sites, several plant samples can be associated with one 190 representative soil sample. Only large pieces of stones were removed from soil samples at the time 191 of sampling, and no sieving, milling or homogenization were performed afterward in the laboratory. 192 Soil samples were also dried in the laboratory at ambient room temperature, as oven drying can affect the value of EC_{Ra} (Girault and Perrier, 2011, 2012a), in order to keep them as much as 193 194 possible in the original natural condition.

195

196 2.2. Measurement of water pH of soil samples

197

Soil pH was determined in the laboratory using the common technique (*e.g.*, Thu et al., 2020). About 20 g of soil and 100 mL of distilled water were placed in a beaker and stirred during 200 20 to 30 min with a magnetic stirrer. The pH of the suspension solution obtained was measured at about 25°C using a regularly calibrated pH211 Microprocessor pH-meter (Hanna Instruments, 202 USA) an hour minimum after stirring to ensure equilibrium. The pH value of a total of 18

203 representative soils of the two subsets was measured. Absolute experimental uncertainties204 associated with calibration and reproducibility ranged from 0.01 to 0.02.

205

206 2.3. Measurement of effective 226 Ra concentration (EC_{Ra})

207

218

A mass *m* of gently dried plant or soil sample was placed in a hermetically closed container, 208 209 *i.e.*, a glass pot or bottle closed with a natural rubber stopper (Girault and Perrier, 2012a; Perrier et 210 al., 2018). For all samples, after an accumulation time, t, from 4 to 18 days, the air of the pot was 211 sampled using a 125-mL pre-evacuated scintillation flask (Algade, France). After the 3.5 h needed 212 to reach radioactive equilibrium, a photomultiplier (CALENTM, Algade, France) was used to 213 measure the number of counts in 10 min interval. Subtracting the background count of the flask 214 determined before sampling, applying the conversion factor of the photomultiplier, and taking into 215 account dilution during sampling (Girault and Perrier, 2012a), we infer ²²²Rn activity concentration in the flask, C_{Rn} (in Bq m⁻³), and then the effective ²²⁶Ra concentration (EC_{Ra}) of the material in the 216 217 container using (Girault and Perrier, 2011, 2012a):

$$EC_{\rm Ra} = \frac{V}{m} \frac{C_{\rm Rn}}{\left(1 - e^{-\lambda t}\right)},\tag{1}$$

where *V* is the free air volume of the container (m³) and λ is the ²²²Rn decay constant (2.1×10⁻⁶ s⁻¹). Three accumulation experiments giving three *EC*_{Ra} measurements were performed for each sample at different accumulation times and their values were averaged to get the final value. The measurement uncertainty takes into account the counting statistics and the dilution correction. A systematic uncertainty of about 5%, common to all measurements, is due to the absolute calibration of the flask counting in the photomultipliers.

When ²²⁶Ra level was particularly low, a high-sensitivity method was used (Girault et al., 2017a; Girault and Perrier, 2019). In this method, the same protocol was followed, except that air

227 sampling was conducted after a long accumulation time of more than 21 days in order to reach radioactive equilibrium between ²²²Rn and ²²⁶Ra. The number of counts in the flask was then 228 229 recorded in 10 min interval in long counting session of 24 h. The background count was determined 230 precisely before sampling during a counting session of 3 days minimum. The final signal was 231 calculated by subtracting from the arithmetic average of the flask count distribution the arithmetic average of the background distribution. Similarly, the effective 226 Ra concentration (EC_{Ra}) of the 232 233 material in the container was determined using Eq. (1). Generally, one or two additional high-234 sensitivity measurements were performed for samples having an experimental uncertainty larger 235 than about 25% using the first method. For each sample for which significant EC_{Ra} values were 236 obtained using the two methods, or for which EC_{Ra} values were obtained using different subsamples 237 of the same sample, the final retained EC_{Ra} value is a weighted mean accounting for all the 238 available measurements.

239 In the last 10 years, more than 7300 measurements of EC_{Ra} have been performed in our 240 laboratory on various materials, based on more than 17,000 accumulation experiments, and we have 241 thoroughly tested the overall quality of our EC_{Ra} methodology. Based on the regularly measurement 242 of selected reference samples (Girault and Perrier, 2012a,b; Perrier et al., 2018), and on several 243 successful inter-comparison exercises with other measurement techniques from various laboratories 244 worldwide, such as three types of solid-state nuclear track detectors (KodalphaTM LR115 cellulose nitrate films from Dosirad, France; DPR2[™] dosimeters from Algade, France; CR-39 polycarbonate 245 246 films from Kingston Univ., UK) (Girault and Perrier, 2012a,b), ionization chambers coupled with alpha spectroscopy (AlphaGUARDTM, Bertin Instr., Germany) (Nicolas et al., 2014; Girault et al., 247 248 2017b), liquid scintillation counting (Wrocław Univ. of Technology, Poland) and alpha 249 spectrometry methods (CEA, France) (Perrier et al., 2016a), our EC_{Ra} measurement technique can 250 be considered as sufficiently robust and accurate for the present purpose, with the experimental 251 uncertainties assigned carefully.

252 Using the two variants of our method (see above), we carried out a total of 225 EC_{Ra} 253 measurements of plant samples (Fig. 2), with mean (min-max) sample mass of 74 ± 4 (4.2–340) g 254 (Supplementary Fig. S1). The mean relative experimental uncertainty on plant EC_{Ra} was $21 \pm 1 \%$ 255 (n = 225), on average slightly larger than for all our other measurements that include mainly soil 256 and rock samples (11.0 \pm 0.2 %; n = 6636). The mean relative experimental uncertainty on their nearby soil EC_{Ra} was 2.85 ± 0.01 % (n = 32). For a plant sample mass between 15 and 50 g, the 257 relative experimental uncertainty was 4% for EC_{Ra} of 20 Bq kg⁻¹ and about 30% for EC_{Ra} of 0.5 Bq 258 kg⁻¹ (Fig. 2). Such measurement uncertainties are adequate for the present purpose. 259

The effects that can potentially affect, in principle, the interpretation of accumulation 260 261 experiments with rocks and soils (Girault et al., 2012b) could also be present for plant samples. 262 Temperature and humidity content have a significant effect on EC_{Ra} (e.g., Girault and Perrier, 2011, 263 2012b), but should be a second order effect in the temperature range that can be considered here for plants. Physical surface adsorption of radon is largely unknown in the case of plant material, but 264 doping experiments suggested that it must be a second order effect (Perrier et al., 2018). Grain-size 265 266 may in principle affect EC_{Ra} values for solid materials (e.g., Markkanen and Arvela, 1992), but the 267 effect, if any, is unknown for plant. Considering beech leaves from the Sapine site (Fagus sylvatica; A subset), EC_{Ra} values obtained for entire leaves (8.19 ± 0.50 Bq kg⁻¹) and for leaves cut in small 268 (<2 mm size) pieces (8.44 \pm 0.57 Bq kg⁻¹) were compatible within uncertainty. Consequently, we 269 can consider here that the apparent EC_{Ra} that we measure in our experiments is representative of the 270 271 true EC_{Ra} .

272

273 2.4. Determination of ²²⁶Ra soil-to-plant transfer ratio (R_{SP})

274

First, we compared the measured EC_{Ra} values of the plant sample (EC_{RaP}) and of the nearby local soil (EC_{RaS}). We considered the plant–soil pair only when the nearby soil was not more distant

277 than 100 m from the collected plant. On average, the distance between the plant and soil samples 278 was 6 ± 1 m. More than 80% of our plant samples have a soil sample within 10 m. Dispersion of 279 EC_{Ra} values for soil samples have been studied at several locations at various spatial scales and 280 were found reasonably compatible with the mean EC_{Ra} value of the given data set (Girault et al., 281 2012b; Perrier et al., 2018). Some depth profiles (Supplementary Fig. S2), carried out in the soil 282 underlying the Mont-Lozère calco-alkaline biotite Hercynian granite in the Sapine watershed (A 283 subset) and in the soil underlying the Callovo-Oxfordian limestone of the Pech Merle cave (B subset), show that EC_{Ra} values are relatively similar in the depth range 5 cm to 30 cm at a given 284 285 site. Two horizontal profiles at the Sapine site (A subset) do not show significant difference in EC_{Ra} 286 value of soil at distance of at least 60 m (Supplementary Fig. S3a). In addition, a 600-m-long 287 horizontal profile of EC_{Ra} values of soil carried out at Vaugrigneuse site (B subset), where 724 top 288 soils were sampled and their EC_{Ra} value measured (Perrier et al., 2016b), does not show any 289 significant variation of EC_{Ra} around the location of the soil where plants have been collected 290 (Supplementary Fig. S3b). These observations confirm that, at a given site, whatever the considered 291 subset, EC_{Ra} values of soil are relatively homogeneous, and the relation with plant sample in the 292 reasonable ranges of depth and lateral distance, as considered in our sampling methodology, can be 293 considered as representative of the mean EC_{Ra} values of the soil surrounding a given plant.

294 The obtained ratio (EC_{RaP}/EC_{RaS}) might be an interesting transfer parameter in itself. Indeed, it is meaningful to normalize the effective ²²⁶Ra concentration of the plant sample, which is to a first 295 approximation the ²²⁶Ra concentration of the plant sample, as we will show below, to the effective 296 ²²⁶Ra concentration of the soil, which represents the amount of ²²⁶Ra connected to the pore space, 297 298 thus available to the root system. Nevertheless, in a second approach and in order to obtain values 299 of the soil-to-plant transfer ratio (R_{SP}) , similar to the transfer factor or the concentration ratio 300 commonly used in the literature, we eliminate the mean emanation coefficients from the observed ²²⁶Ra plant-to-soil effective concentration ratios. Indeed, because EC_{Ra} gives an effective ²²⁶Ra 301

302 concentration of a given material, dividing EC_{Ra} of a material by its emanation coefficient, *E*, gives 303 the bulk ²²⁶Ra concentration of the material. Here, we used mean representative *E* values for soils 304 and plants, as inferred from ²²⁶Ra doping experiments carried out in the laboratory (Perrier et al., 305 2018): $\langle E_S \rangle = 0.242 \pm 0.035$ for soils, and $\langle E_P \rangle = 0.862 \pm 0.044$ for plants. The *R*_{SP} value is then 306 calculated using (Perrier et al., 2018):

$$R_{\rm SP} = \frac{E_{\rm S}}{E_{\rm P}} \frac{EC_{\rm RaP}}{EC_{\rm RaS}}.$$
(2)

A total of 108 plant samples with a nearby local soil were collected and their respective R_{SP} value could be inferred. Because it accounts for EC_{Ra} and E experimental uncertainties of both plant and soil samples, uncertainty on R_{SP} was generally large, with mean value of 23 ± 9 %. In the following, averages are geometric means except otherwise stated.

In plant material, trapping of radon cannot occur as in minerals (Nazaroff, 1992; Adler and Perrier, 2009), therefore the emanation coefficient of plants must be close to 1 and cannot vary much from plant to plant. In the case of soils, however, the potential range of variations can be large (Sakoda et al., 2011), and we shall return to the matter in details in the discussion below.

316

317 **3. Results**

318

319 *3.1. Effective* ²²⁶*Ra concentrations in plants and their nearby local soils*

320

Effective ²²⁶Ra concentration (EC_{Ra}) values of plant samples (n = 108) range over about five orders of magnitude (Fig. 3c), from 0.020 ± 0.001 to 113 ± 7 Bq kg⁻¹, and with a mean of 1.66 ± 0.03 Bq kg⁻¹ (Table 1). The two largest values, 113 ± 7 and 44 ± 2 Bq kg⁻¹, are measured for a moss and roots of *Fagus sylvatica*, respectively, both from the granitic area near Ambazac. The two smallest values, 0.020 ± 0.001 and 0.09 ± 0.07 Bq kg⁻¹, are measured for chestnuts of *Aesculus*

326 hippocastanum and branches of Prunus cerasus, respectively, both from the sedimentary area near 327 Anduze. The EC_{Ra} values of the nearby soil samples (n = 24) are less scattered (Fig. 3b) with a mean of 18.9 \pm 0.2 Bq kg⁻¹ (Table 1). While the EC_{Ra} values of plants follow a log-normal 328 329 distribution with two modes around 0.5 and 15 Bq kg⁻¹, similar to our whole data set of plant EC_{Ra} 330 (n = 174) (Fig. 3c), the EC_{Ra} values of soils follow a log-normal distribution with a single mode around the mean (Fig. 3b). Although, to first order, we may expect some relation between EC_{Ra} of 331 332 plant and EC_{Ra} of soil, the mean value and the general shape of the log-normal distribution of plant 333 EC_{Ra} appear relatively similar to those of rock EC_{Ra} (Fig. 3a), suggesting a possible relation 334 between a given plant and its substratum. Alternatively, the large range of plant EC_{Ra} may also 335 suggest an efficient dispersion process during the transfer of ²²⁶Ra from soil to the biosphere.

336 The comparison of plant EC_{Ra} for the two subsets provides an important hint. Indeed, a significant difference is observed between EC_{Ra} values of plants collected in granitic and 337 metamorphic context (A subset; mean: 9.2 \pm 0.1 Bq kg⁻¹; n = 50) and EC_{Ra} values of plants 338 collected in calcareous and sedimentary context (B subset; mean: 0.38 ± 0.01 Bq kg⁻¹; n = 58). 339 340 Separating these two subsets, only few plant samples have similar or even slightly larger EC_{Ra} than 341 their respective soil EC_{Ra} (Fig. 4a). However, more plant samples have EC_{Ra} relatively similar to their respective rock EC_{Ra} (Supplementary Fig. S4). Soil EC_{Ra} values are also different, depending 342 on the substratum: 36.1 ± 0.6 Bq kg⁻¹ for the A subset and 12.9 ± 0.2 Bq kg⁻¹ for the B subset (Table 343 1 and Fig. 4a). The EC_{Ra} of the A subset over EC_{Ra} of the B subset ratio is significantly larger for 344 345 plants (24 ± 8) and rocks (31 ± 1) than for soils (2.8 ± 0.1) .

346

347

3.2. Results on the corrected ²²⁶Ra soil-to-plant transfer ratio

348

Normalizing plant EC_{Ra} by emanation coefficient of plants (E_P) and similarly soil EC_{Ra} by 350 E_S , we observe a slight EC_{RaP}/E_P increase with EC_{RaS}/E_S , from the B subset to the A subset (Fig.

4b). The min–max values of EC_{RaP}/E_P and EC_{RaS}/E_S respectively, corresponding to the bulk ²²⁶Ra concentrations, are 0.6–130 Bq kg⁻¹ and 50–370 Bq kg⁻¹ for the A subset, and 0.02–5 Bq kg⁻¹ and 11–190 Bq kg⁻¹ for the B subset, which are consistent with reported ²²⁶Ra concentration in plants and soils in similar contexts.

355 The inferred values of soil-to-plant transfer ratio (R_{SP}) (n = 108) range over four orders of magnitude (Fig. 5a), from 0.0007 \pm 0.0001 to 0.37 \pm 0.06, and with a mean of 0.0188 \pm 0.0004 356 (Table 1). Our R_{SP} values follow a log-normal distribution with two modes around 0.003 and 0.1 357 (Fig. 5a). Our whole R_{SP} data set can be subdivided into our two substratum subsets (Table 1). 358 359 Indeed, the highest mode of the R_{SP} distribution corresponds to the A subset (Fig. 5b), with R_{SP} 360 values from 0.005 ± 0.001 to 0.37 ± 0.06 and mean of 0.073 ± 0.002 (n = 50), while the smallest 361 mode of the R_{SP} distribution corresponds to the B subset (Fig. 5c), with R_{SP} values from 0.0007 ± 0.0001 to 0.13 ± 0.02 and mean of 0.0058 ± 0.0002 (n = 58). 362

While the discrimination between the two subsets appears significant, some tail-ends of the 363 R_{SP} distribution nevertheless exist for both subsets. About 20% of the A subset has a R_{SP} value 364 365 smaller than 0.02, while about 19% of the B subset has a R_{SP} value higher than 0.02. A better discrimination, however, can be proposed for R_{SP} as a function of the inferred plant or soil ²²⁶Ra 366 concentration (Fig. 4c and Supplementary Fig. S5). In Fig. 4c, defining the line by 367 $R_{\rm SP}=1.24\times10^5 C_{\rm RaS}^{-3.43}$, then 6% of the A sample now falls in the B region, while 5% of the B 368 samples falls in the A region. This suggests that the two subsets can be separated almost 369 completely. Overall, our mean R_{SP} value in granitic and metamorphic context is 12 ± 1 times larger 370 than that in calcareous and sedimentary context. This observation indicates definitely, for the first 371 time, a clear quantifiable effect of the substratum on the ²²⁶Ra transfer from soil to plant in non-372 373 contaminated areas.

374

375 **4. Discussion**

376

377 4.1. Updated status on the understanding of radium uptake by plants

378 Our soil-to-plant transfer ratio (R_{SP}) data, although they are inferred from the measurement of effective 226 Ra concentration (EC_{Ra}) and emanation coefficient (E) of both soil and plant 379 380 samples, give compatible values with the transfer factors deduced from the direct measurement of 381 bulk ²²⁶Ra concentrations as reported in the literature. Indeed, our mean R_{SP} value, 0.0188 ± 0.0004, thus about 2%, is consistent with mean ²²⁶Ra transfer factors reported in the literature for all plants, 382 383 ranging from about 2% (Vandenhove et al., 2009) to about 5% (Sheppard et al., 2006). Using a compiled data set from published ²²⁶Ra transfer factors from the literature (38 articles; n = 870), 384 385 90% of the R_{SP} values are comprised between 0.0015 and 1. R_{SP} values above 1 are generally 386 obtained with vegetables growing in radium-contaminated environments or in laboratory 387 experiments using radium-rich nutrient solutions (Simon and Ibrahim, 1990 and references herein; 388 Sam and Eriksson, 1995; Martínez-Aguirre and Periáñez, 1998; Vanderhove and Van Hees, 2007; 389 Asaduzzaman et al., 2014; Hu et al., 2014). R_{SP} values below 0.0015 are generally obtained with 390 vegetables, fruits, and cereals growing in radium-contaminated environments or, in a lesser extent, 391 in non-contaminated areas (Simon and Ibrahim, 1990 and references herein; Markose et al., 1993; 392 Ham et al., 2001; Ryan et al., 2005; Uchida and Tagami, 2007; Uchida et al., 2009 and references 393 herein; Černe et al., 2011). Considering a reasonable range of R_{SP} values over five orders of magnitude (from 0.00025 to 2; n = 844), these R_{SP} values follow a log-normal distribution with one 394 mode, centered on the mean of 0.0448 \pm 0.0003 (Fig. 5d). In the literature, only few R_{SP} values 395 396 smaller than 0.006 (about 11%) are reported. This small range corresponds to the low R_{SP} values we 397 obtained in calcareous and sedimentary context (B subset). To our knowledge, no study has focused 398 on radium transfer from soil to plants in calcareous and sedimentary context, probably because of 399 the low ²²⁶Ra levels that are beyond the detection limit of the commonly used methods (gamma and 400 alpha spectrometry, liquid scintillation counting, and mass spectrometry). This highlights the

401 usefulness of our high-sensitivity EC_{Ra} measurement method, decreasing the detection limit of ²²⁶Ra 402 levels in soil and plant samples with reasonable experimental uncertainty.

403 Our data and the data from the literature suggest that the transfer of ²²⁶Ra from soil to plant 404 may be log-normal (Fig. 5). While the log-normal distribution of radioactivity data is a subject in 405 itself (Bossew, 2010), it could be that transfer of radionuclides and other trace elements to plants is 406 intrinsically affected by large fluctuations resulting in a log-normal distribution. The standard 407 deviation then is as interesting as the mean value and is an independent parameter. However, the 408 process controlling the standard deviation is not clear at this stage and a larger data set would be 409 required to evaluate the distribution function.

410 Contrary to other studies (e.g., Simon and Ibrahim, 1987, 1990; Vera Tomé et al., 2003; 411 Pulhani et al., 2005), we have found a significant difference in R_{SP} depending on the substratum. 412 Our difference of a factor of 12 ± 1 between R_{SP} obtained in granitic and metamorphic context (A 413 subset) and R_{SP} obtained in calcareous and sedimentary context (B subset) may be explained by the 414 larger calcium concentration in the soil above limestone regolith. In the case of the Sapine 415 watershed (A subset), calcium deficit was reported in the soil (Durant et al., 1991; Didon-Lescot, 416 1996). In contrast, a large calcium concentration may saturate incorporation into the plant roots, hence decreasing the transfer of ²²⁶Ra from soil to plant, as reported in earlier studies (Nathwani 417 418 and Phillips, 1979; Marple, 1980; Simon and Ibrahim, 1990). Here, we reveal this effect for the first time in non-contaminated natural environments. This observation suggests that other effects related 419 to substratum may influence R_{SP} of ²²⁶Ra and other radionuclides or heavy metals as well. 420

421

422 4.2. Radium uptake in the different plant compartments and in other biological samples

423

424 For some plant species, we collected several plant compartments. When considering 425 separately the two subsets of samples in different substratum contexts, the $R_{\rm SP}$ values obtained for

426 the different compartments of a given plant appear relatively compatible (Fig. 6 and Tables 2 and 3). The acropetal gradient, *i.e.*, a decrease of 226 Ra concentration from roots to apex, is generally 427 428 observed for all plants: the root-to-shoot ratio ranges from 0.7 ± 0.3 to 2.3 ± 1.2 , with a mean of 1.8 429 \pm 0.4, relatively consistent with published values (e.g., Gunn and Mistry, 1970; Simon and Ibrahim, 430 1990; Vandenhove et al., 2009; Mrdakovic Popic et al., 2020). The mean root-to-stem and stem-toshoot ratios are 2.7 ± 0.3 and 0.8 ± 0.2 , respectively. The stem-to-shoot ratio is slightly smaller for 431 the B subset $(0.7 \pm 0.1; n = 7)$ than for the A subset $(1.0 \pm 0.2; n = 7)$. ²²⁶Ra concentration is indeed 432 433 larger in leaves than in branches or stems for some plants collected in metamorphic context (Saint-434 Yrieix-la-Perche), and for several plants in carbonaceous and sedimentary context (Pech Merle, 435 Anduze, Vaugrigneuse, and Chambon-la-Forêt). Thus, 226 Ra concentration and R_{SP} values decrease 436 in the following order: roots > bark > branches and stems \approx leaves (> flowers > fruits).

437 Samples from the same plant compartment, but collected on different nearby individuals, show mean R_{SP} dispersion of 32 ± 6 % (Tables 2 and 3). Several nearby top soil samples were 438 439 collected at some sites. While EC_{Ra} values of soil samples are relatively homogeneous, indicating 440 our soil samples are representative of the local soil for each site, as detailed above, dispersion of soil EC_{Ra} values amounts to 26 ± 8 % in the Sapine watershed (A subset; n = 6), 34 ± 11 % in 441 442 Anduze (A subset; n = 6), and 27 ± 7 % in Pech Merle (B subset; n = 8). These dispersion values are close to the scarp-scale dispersion of rock EC_{Ra} observed at given sites (from 17 to 69%; Girault 443 444 and Perrier, 2012b; Girault et al., 2012), suggesting some relation between dispersion of R_{SP} values 445 of nearby individuals and substrate or substratum heterogeneity.

In addition, R_{SP} values are relatively similar in the compartments of different plant species growing at the same site. In the Saint-Yrieix leptynite (A subset), a narrow range of R_{SP} values (3– 13%) is obtained for the deciduous trees *Quercus robur* and *Castanea sativa*, the shrub *Spartium junceum*, and the fern *Pteridium aquilinum* (Table 2 and Fig. 6). In the Kimmeridgian limestone near Puéchabon (B subset), a narrow range of R_{SP} values (0.2–0.8%) is also obtained for the

451 evergreen tree Quercus ilex and the shrubs Buxus sempervirens and Phillyrea latifolia (Table 3 and 452 Fig. 6). These observations indicate a strong control of the substratum–substrate pair on the R_{SP} of plant beyond the influence of the plant species. Some significant differences also emerge. In the 453 454 Cévennes granite near Anduze (A subset), mean R_{SP} value obtained for *Quercus robur* (0.13 ± 0.06) 455 is 8 ± 4 times larger than that for *Pinus pinaster* (0.017 \pm 0.006). Considering separately the Quercus genus (Supplementary Fig. S6), R_{SP} values range from 0.0012 ± 0.0002 to 0.37 ± 0.06, 456 457 with a mean of 0.0134 \pm 0.0005 (n = 40). The Quercus genus yields a large R_{SP} difference of a 458 factor of 23 ± 2 between the A subset (0.122 ± 0.006 ; n = 12) and the B subset (0.0052 ± 0.0003 ; n 459 = 28), suggesting that the substratum control on R_{SP} is preponderant for a given plant genus. These 460 observations indicate that oak trees tend to concentrate radium more than other trees or plants.

461 Other samples were also considered: market vegetables, mosses, mushrooms and algae. The EC_{Ra} values of market vegetables (n = 15), for which we do not have the associated soil sample, 462 463 range from 0.05 \pm 0.02 to 0.82 \pm 0.04 Bq kg⁻¹, with a mean of 0.35 \pm 0.08 Bq kg⁻¹, and the mean (min-max) calculated bulk ²²⁶Ra concentration yields 0.41 ± 0.09 (0.05–0.95) Bq kg⁻¹, consistent 464 465 with other reported values (e.g., Lauria et al., 2009). The EC_{Ra} values of mosses, sampled on the 466 ground and on tree bark, are heterogeneous and depend on the substratum, with larger values for the A subset (mean: 39 ± 25 Bq kg⁻¹; n = 4) than for the B subset (1.5 ± 0.3 Bq kg⁻¹; n = 1). Their 467 468 respective R_{SP} values are also significantly larger for the A subset (mean: 0.26 ± 0.14) than for the 469 B subset (0.017 \pm 0.005). Our R_{SP} values for mosses in the A subset are consistent with reported values in metamorphic context in Serbia (mean: 0.29 ± 0.02 ; n = 42; Dragović et al., 2010) and in 470 471 granitic gneiss context of the Fen Complex, Norway (mean: 0.27 ± 0.14 ; n = 5–14; Mrdakovic 472 Popic et al., 2020), and in the B subset with reported values in silico-carbonate context, Norway 473 (mean: 0.036 ± 0.010 ; n = 5–14; Mrdakovic Popic et al., 2020). However, compared with vascular plants, mosses have a larger efficiency to take elements directly by precipitation and airborne 474 deposition from the surrounding air (e.g., Uğur et al., 2003; Krmar et al., 2013; Čučulović et al., 475

476 2016; Wattanavatee et al., 2017). Mosses thus may absorb ²²⁶Ra contained for example in the soil
477 dust or transported by aerosols, making the contribution of the substratum delicate to assess reliably
478 without additional data.

479 The EC_{Ra} values of mushrooms, all sampled in the Mont-Lozère granite in the Sapine watershed (A subset), are heterogeneous, ranging from 0.5 ± 0.3 to 11.1 ± 0.6 Bq kg⁻¹, with a mean 480 of 5 ± 2 Bq kg⁻¹ (n = 4). Their mean (min-max) R_{SP} value gives 0.12 ± 0.05 (0.005-0.26), which 481 482 appears compatible with published R_{SP} data for fungi (mean: 0.08 ± 0.02; n = 5; Kirchner and 483 Daillant, 1998). Finally, a sample of filamentous algae was collected in flowing water from a lead and zinc mine in Nepal. Using the dissolved 226 Ra concentration in water (20 ± 15 mBq L⁻¹; Perrier 484 et al., 2016a; Girault et al., 2018) and the EC_{Ra} value of the algae sample (12.8 ± 0.9 Bq kg⁻¹), we 485 obtain a large value of 226 Ra absorption factor by the algae (740 ± 560), similar to other reported 486 487 absorption factors of filamentous algae near operating mines in India (Jha et al., 2010).

488

489 4.3. Differences in emanation coefficients cannot explain the substratum effect on R_{SP}

490

In our study, we have considered a normalization of our EC_{Ra} data by reasonably representative ²²²Rn emanation coefficient values for soils (E_S) and plants (E_P) and to calculate the ²²⁶Ra soil-to-plant transfer ratio (R_{SP}), a parameter commonly used in the literature. As clearly stated above, these mean E values were chosen based on the results of thorough laboratory experiments using the same method (Perrier et al., 2018). In the following, we discuss the choice of such representative E_S and E_P values, opening perspectives for future studies.

497 Concerning the ²²²Rn emanation coefficient for soils, E_S , we used a value obtained for sands 498 (0.242 ± 0.035; Perrier et al., 2018), as our soils are actually sandy soils, especially at the Sapine 499 site located in Mont-Lozère area (granitic arena; A subset), but also in the sedimentary regions at 500 the Vaugrigneuse site (Fontainebleau sands near Paris; B subset) and at the Chambon-la-Forêt site

501 (sandy loam near Orléans; B subset). While some systematic differences could be present between 502 soils and sands, the current data do not show any significant difference. Considering the work by 503 Markkanen and Arvela (1992), which is by far the most comprehensive paper on $E_{\rm S}$ values with 504 more than 400 Finnish soils analyzed, no systematic significant differences in E_S was found 505 between different soil types (clay, silt, sand, gravel and till), with a mean varying from 0.17 to 0.24 506 and 50% of data in the range 0.11 to 0.31, hence compatible with our mean reference $E_{\rm S}$ value 507 obtained for sands (0.242 ± 0.035) . In the Sakoda et al. (2011) review and in all other publications 508 with sufficient numbers of experimental data, the range of variation of $E_{\rm S}$ is large. This is indeed 509 expected, since the distribution of such data is log-normal and trace element concentrations are 510 affected by large sample-to-sample dispersion (e.g., Bossew, 2010). However, the full min-max 511 range does not give any information on systematic effects on the mean value, which is the only 512 aspect in which we are interested for our current problem. One important point, nevertheless, as 513 detailed below, concerns whether or not we can expect a systematic difference in $E_{\rm S}$ between our A 514 and B subsets, and whether it may explain the 12 ± 1 times difference in R_{SP} between them.

Concerning the ²²²Rn emanation coefficient for plants, E_P , we used a mean value obtained from different plants and different compartments of a given plant (0.862 ± 0.044; Perrier et al., 2018). To our knowledge, the E_P parameter was completely unknown before this study, and thus would be welcome to be refined in detail for various plant samples and plant compartments. Given the small range of variation of E_P based on our experimental data (0.82–0.95; Perrier et al., 2018), an effect of a systematic difference of E_P that may explain the 12 ± 1 times difference in R_{SP} between the A and B subsets can safely be ruled out.

The soils overlying the two geological contexts A and B considered in our study have different properties and characteristics. Although the effect of the soil type on the value of E_S is poorly known, some results are nonetheless available. In a study in which five types of soils overlying different lithologies (limestone, gray shale, red shale, gneiss, and sandstone) were

526 analyzed for their respective $E_{\rm S}$ values (Luetzelshwab et al., 1989), no significant systematic 527 difference was observed between E_S over gneiss (mean: 0.340 \pm 0.039) and limestone (mean: 0.370 528 \pm 0.036), with also similar range of values. Similarly, no major systematic difference in E_S related 529 to substratum effects emerges from the Sakoda et al. (2011) review. Consequently, to date, no 530 significant difference in $E_{\rm S}$ values has been reported between soils overlying rocks of our A and B subsets, hence ruling out any large systematic effect able to explain the obtained large factor of $12 \pm$ 531 532 1. However, this aspect is under-constrained in the literature and would need dedicated experiments. Another aspect we could develop here is the potential effect of soil pH on the ²²²Rn 533 534 emanation coefficients of our soils. The values of water pH of our soils (Supp. Mat. Fig. S7), 535 ranging from 3.93 ± 0.01 to 6.39 ± 0.01 in our nine investigated sites, suggest acidic soils and are 536 compatible with the published map of French soils (GIS Sol, 2011). For one of the site (the Sapine 537 in Mont-Lozère; A subset), our mean pH value $(4.27 \pm 0.27; n = 3)$ is consistent with the detailed 538 results already published (Didon-Lescot, 1996). The obtained pH values are on average slightly 539 larger for the B subset $(5.82 \pm 0.26; n = 11)$ than for the A subset $(4.71 \pm 0.27; n = 7)$, as expected. 540 In a recently published paper, Thu et al. (2020) present values of $E_{\rm S}$ together with the measured 541 water pH of the soil and suggest, despite a large dispersion, a slight variation of E_s versus pH. 542 Using the fit we inferred from Thu et al. (2020) data ($E_{\rm S}$ =0.07+0.036pH), we get corresponding values of $E_{\rm S} = 0.24 \pm 0.01$ for the A subset and $E_{\rm S} = 0.28 \pm 0.01$ for the B subset. Thus, we can 543 544 expect a difference of $E_{\rm S}$ between the A and B subsets of 14 ± 1 %, which is small compared with 545 the observed difference in R_{SP} of a factor of 12 ± 1 . Consequently, even when this recent work is 546 taken into account, there is so far no evidence for a sufficient systematic difference of the mean 547 value of E_S between the A and B subsets, and the difference in R_{SP} between them cannot be 548 attributed to a systematic difference in the ²²²Rn emanation coefficient of soils or plants.

550 4.4. Radium uptake as a function of the concentration in the soil

551

552 As in the case of uranium (e.g., Sheppard and Everden, 1988a), a power-law relationship has been proposed between the bulk ²²⁶Ra concentration of plant and that of soil, referred to as the 553 'linearity hypothesis', for high ²²⁶Ra levels in contaminated areas (Simon and Ibrahim, 1990; 554 Blanco Rodríguez et al., 2002, 2006). Our data set, although obtained in non-contaminated areas 555 with comparatively low ²²⁶Ra levels, seems to also follow, to first order, a power-law relationship 556 557 (dash-dot curve in Fig. 4b). However, several data clearly depart from the linearity hypothesis (Fig. 558 4b), which may suggest the presence of competition between calcium and radium, for example in 559 calcareous context (B subset), as discussed above. More interestingly, the relationship built on a data set obtained for high ²²⁶Ra levels, and considered to better reproduce the uptake behavior of 560 561 calcium and other alkaline earth elements, appears to account for our whole data set (Simon and 562 Ibrahim, 1987, 1990). Indeed, for the relationship between bulk ²²⁶Ra concentration in soil and that in plant (solid line in Fig. 4b), and R_{SP} values (solid line in Fig. 4c), our data set is consistent with 563 this empirical fit, indicating its validity for low 226 Ra levels. The difference in Fig. 4c between R_{SP} 564 obtained in granitic and metamorphic context (A subset) and R_{SP} obtained in calcareous and 565 566 sedimentary context (B subset) then reflects the fact that R_{SP} follows a universal function of the ²²⁶Ra concentration of the substratum. 567

568 Our study on ²²⁶Ra uptake by plants may be of potential interest to the uptake of other alkali 569 earth elements (Ca, Ba, Sr, and Mg). As an example, Sr, highly toxic to living animals, like Ra, is 570 also in competition with Ca through root uptake, and has been found to affect plant physiology at 571 various levels (Moyen and Roblin, 2010). Thus, our methodological and substratum comparison 572 approach may be a significant contribution not only to soil-to-plant transfer of radium, but also to 573 the radioecology of alkali earth elements.

575 **5.** Conclusion

576

We have shown that measuring effective 226 Ra concentration (EC_{Ra}) can be useful to 577 quantify the ²²⁶Ra uptake by the biosphere from soil or water in various environments. We have 578 579 found that, in non-contaminated areas, $R_{\rm SP}$ values are heterogeneous but reveal an important 580 influence by the substratum, with about an order of magnitude larger R_{SP} values in granitic and 581 metamorphic context than in calcareous and sedimentary context. This difference may likely be due 582 to the competition between calcium and radium, which is being evidenced for the first time in non-583 contaminated natural environments. In addition, some significant differences emerge between tree 584 species. One first consequence is that results obtained in one geological context cannot be 585 transferred without caution to another context. Another consequence is that the radium flux into the 586 biosphere is larger in granitic context. This may also be the case for strontium, barium, or heavy 587 metals, but this would require confirmation by dedicated studies.

588 Before definite conclusions can be drawn for potential applications such as the 589 characterization of present and past pollution and phytoremediation, the differences observed in the 590 radium uptake need further confirmation and the processes need to be clarified. In particular, the $R_{\rm SP}$ distributions suggesting that soil-to-plant transfer of ²²⁶Ra and other radionuclides could be log-591 592 normal, so far not considered in most radioecological models (e.g., Whicker et al., 1999; Tuovinen 593 et al., 2016), would need to be studied in details at control sites. A systematic comparison of our effective ²²⁶Ra concentration approach (giving effective concentrations) with other well-established 594 595 methods such as gamma and alpha spectrometry, liquid scintillation counting, and mass 596 spectrometry (giving total concentrations) should be considered in the future. The EC_{Ra} method may also be used to assess the ²²⁶Ra fraction bioavailable for terrestrial plants, because it might be 597 598 close to the 226 Ra accessible to the pore space; thus EC_{Ra} could qualify as an appropriate proxy the ²²⁶Ra concentration relevant for biosphere uptake and plant ecology, therefore a useful complement 599

to other approaches. This proposal, which can only be considered as a hypothesis at the moment, could be tested using leaching experiments on 226 Ra-rich materials in the laboratory, following already established protocols (*e.g.*, Rihs et al., 2011; Georgiev et al., 2014; Chautard et al., 2020). Such an ambitious program would need to be designed carefully, with an adequate selection of the sites and of the considered plant species. The results obtained in this study provide some initial information that would need to be considerably expanded.

606 Our EC_{Ra} measurement method, cost-effective and easy to implement, may be used at a larger scale to determine R_{SP} value in natural environments characterized by low ²²⁶Ra levels and in 607 608 uranium- and radium-contaminated environments as well (Sheppard et al., 2006; Vandenhove et al., 609 2009). In addition, to clarify the processes, EC_{Ra} measurements can be incorporated in controlled 610 conditions in the laboratory to study the factors influencing $R_{\rm SP}$ in a systematic manner 611 (Vandenhove et al., 2005; Vandenhove and Van Hees, 2007; Nezami et al., 2016). Given the 612 relatively cost-effective method of radon emanation, which allows large number of data in numerous environments, the uptake of radium by plants in various environments can become an 613 614 essential tool to study the matter fluxes in the CZ, and reveal, through the pathways of an efficient 615 tracer, the mechanisms to protect or rehabilitate the most fragile parts of the living Earth.

616 Acknowledgments

617

618 We thank Jean Bouillaguet for his help in plant and soil sampling around Saint-Yrieix-la-619 Perche, Haute-Vienne department, Hélène Bouquerel and Aude Isambert for their help in plant and soil sampling around the Pech Merle cave, Lot department, and Benoit Heumez, Eric Parmentier, 620 621 and Xavier Lalanne for their help in the sampling of soils in the compound of the National 622 Magnetic Observatory at Chambon-La-Forêt, Loiret department. Dominique Genty and Bruno 623 Lartiges are thanked for their constructive comments on an earlier version of figures. Bénédicte 624 Ménez is thanked for the loan of the pH-meter. The original version of the manuscript was 625 improved thanks to the careful evaluation and precise comments of two anonymous reviewers. This 626 is IPGP contribution number xxxx.

627 **References**

- 628 Abrahams, P.W., 2002. Soils: their implications to human health. Sci. Total Environ. 291, 1-32.
- Abreu, M.M., Lopes, J., Santos, E.S., Magalhães, M.C.F., 2014. Ecotoxicity evaluation of an amended soil
 contaminated with uranium and radium using sensitive plants. J. Geochem. Explor. 142, 112-121.
- Adler, P.M., Perrier, F., 2009. Radon emanation in partially saturated porous media. Transport Porous Med.
 78, 149-159.
- Al-Hamarneh, I.F., Alkhomashi, N., Almasoud, F.I., 2016. Study on the radioactivity and soil-to-plant transfer factor of ²²⁶Ra, ²³⁴U and ²³⁸U radionuclides in irrigated farms from the northwestern Saudi Arabia. J. Environ. Radioact. 160, 1-7.
- Asaduzzaman, K., Khandaker, M.U., Amin, Y.M., Bradley, D.A., Mahat, R.H., Nor, R.M., 2014. Soil-to-root
 vegetable transfer factors for ²²⁶Ra, ²³²Th, ⁴⁰K, and ⁸⁸Y in Malaysia. J. Environ. Radioact. 135, 120-127.
- 638 Banwart, S., 2011. Save our soils. Nature 474, 151-152.
- Bettencourt, A.O., Teixeira, M.M.G.R., Elias, M.D.T., Faisca, M.C., 1988. Soil to plant transfer of radium226. J. Environ. Radioact. 6, 49-60.
- Blanco Rodríguez, P., Vera Tomé, F., Lozano, J.C., 2002. About the assumption of linearity in soil-to-plant transfer factors for uranium and thorium isotopes and ²²⁶Ra. Sci. Total Environ. 284, 167-175.
- Blanco Rodríguez, P., Vera Tomé, F., Pérez Fernández, M., Lozano, J.C., 2006. Linearity assumption in soilto-plant transfer factors of natural uranium and radium in *Helianthus annuus* L. Sci. Total Environ. 361,
 1-7.
- Blanco Rodríguez, P., Vera Tomé, F., Lozano, J.C., Pérez Fernández, M.A., 2010. Transfer of ²³⁸U, ²³⁰Th,
 ²²⁶Ra, and ²¹⁰Pb from soils to tree and shrub species in a Mediterranean area. Appl. Radiat. Isotopes 68, 1154-1159.
- 649 Bossew, P., 2010. Radon: Exploring the log-normal mystery. J. Environ. Radioact. 101, 826-834.
- Brantley, S.L., Goldhaber, M.B., Ragnarsdottir, K.V., 2007. Crossing disciplines and scales to understand
 the Critical Zone. Elements 3, 307-314.
- Bunzl, K., Trautmannsheimer, M., 1999. Transfer of ²³⁸U, ²²⁶Ra and ²¹⁰Pb from slag-contaminated soils to vegetables under field conditions. Sci. Total Environ. 231, 91-99.
- 654 Carini, F., 1999. Radionuclides in plants bearing fruit: an overview. J. Environ. Radioact. 46, 77-97.
- Carvalho, F.P., Oliveira, J.M., Neves, M.O., Abreu, M.M., Vicente, E.M., 2009. Soil to plant (*Solanum tuberosum* L.) radionuclide transfer in the vicinity of an old uranium mine. Geochem. Explor. Environ. Anal. 9, 275-278.
- Černe, M., Smodiš, B., Štrok, M., 2011. Uptake of radionuclides by a common reed (Phragmites australis
 (Cav.) Trin. ex Steud.) grown in the vicinity of the former uranium mine at Žirovski vrh. Nucl. Eng. Des.
- Chautard, C., Beaucaire, C., Gerard, M., Roy, R., Savoye, S., Descostes, M., 2020. Geochemical
 characterization of uranium mill tailings (Bois Noirs Limouzat, France) highlighting the U and ²²⁶Ra
 retention. J. Environ. Radioact. 218, 106251.
- 663 Chen, S.B., Zhu, Y.G., Hu, Q.H., 2005. Soil to plant transfer of ²³⁸U, ²²⁶Ra and ²³²Th on a uranium mining-664 impacted soil from southeastern China. J. Environ. Radioact. 82, 223-236.
- 665 Čučulović, A., Čučulović, R., Sabovljević, M., Radenković, M.B., Veselinović, D., 2013. Natural
 666 radionuclide uptake by mosses in eastern Serbia in 2008-2013. Arh. Hig. Rada. Toksiko. 67, 31-37.
- da Conceição, F.T., Bonotto, D.M., Jiménez-Rueda, J.R., Roveda, J.A.F., 2009. Distribution of ²²⁶Ra, ²³²Th
 and ⁴⁰K in soils and sugar cane crops at Corumbataí river basin, São Paulo State, Brazil. Appl. Radiat.
 Isotopes 67, 1114-1120.
- 670 Didon-Lescot, J.-F., 1996. Forêt et développement durable au Mont-Lozère. Impact d'une plantation de
 671 résineux, de sa coupe et de son remplacement, sur l'eau et sur les réserves minérales du sol. (in French)
 672 PhD thesis, Université d'Orléans, Orléans, 210p.
- bouay, F., Pruvot, C., Waterlot, C., Fritsch, C., Fourrier, H., Loriette, A., Bidar, G., Grand, C., de Vaufleury,
 A., Scheifler, R., 2009. Contamination of woody habitat soils around a former lead smelter in the North
 of France. Sci. Total Environ. 407, 5564-5577.
- Dragović, S., Mihailović, N., Gajić, B., 2010. Quantification of transfer of ²³⁸U, ²²⁶Ra, ²³²Th, ⁴⁰K and ¹³⁷Cs in mosses of a semi-natural ecosystem. J. Environ. Radioact. 101, 159-164.

- Duchemin, B., Coursol, N., Bé, M.M., 1994. The re-evaluation of decay data for the U-238 chain. Nucl.
 Instrum. Methods A339, 146-150.
- Durand, P., Neal, C., Lelong, F., Didon-Lescot, J.F., 1991. Hydrochemical variations in spruce, beech and
 grassland areas, Mont Lozere, southern France. J. Hydrol. 129, 57-70.
- Ehlken, S., Kirchner, G., 2002. Environmental processes affecting plant root uptake of radioactive trace
 elements and variability of transfer factor data: a review. J. Environ. Radioact. 58, 97-112.
- 684 Georgiev, P., Groudev, S., Spasova, I., Nicolova, M., 2014. Ecotoxicological characteristic of a soil polluted
 685 by radioactive elements and heavy metals before and after its bioremediation. J. Geochem. Explor. 142,
 686 122-129.
- 687 Gerzabek, M.H., Strebl, F., Temmel, B., 1998. Plant uptake of radionuclides in lysimeter experiments.
 688 Environ. Pollut. 99, 93-103.
- 689 Girault, F., Perrier, F., 2011. Heterogeneous temperature sensitivity of effective radium concentration from
 690 various rock and soil samples. Nat. Hazards Earth Syst. Sci. 11, 1619-1626.
- 691 Girault, F., Perrier, F., 2012a. Measuring effective radium concentration with large numbers of samples. Part
 692 I Experimental method and uncertainties. J. Environ. Radioact. 113, 177-188.
- 693 Girault, F., Perrier, F., 2012b. Measuring effective radium concentration with large numbers of samples. Part
 694 II General properties and representativity. J. Environ. Radioact. 113, 189-202.
- 695 Girault, F., Perrier, F., 2019. Radon emanation from human hair. Sci. Total Environ. 660, 421-428.
- 696 Girault, F., Gajurel, A.P., Perrier, F., Upreti, B.N., Richon, P., 2011. Radon emanation of heterogeneous
 697 basin deposits in Kathmandu Valley, Nepal. J. Asian Earth Sci. 40, 595-610.
- 698 Girault, F., Perrier, F., Gajurel, A.P., Bhattarai, M., Koirala, B.P., Bollinger, L., Fort, M., France-Lanord, C.,
 699 2012. Effective radium concentration across the Main Central Thrust in the Nepal Himalayas. Geochim.
 700 Cosmochim. Acta 98, 203-227.
- Girault, F., Perrier, F., Moreira, M., Zanda, B., Rochette, P., Teitler, Y., 2017a. Effective radium-226
 concentration in meteorites. Geochim. Cosmochim. Acta 208, 198-219.
- Girault, F., Perrier, F., Poitou, C., Isambert, A., Théveniaut, H., Laperche, V., Clozel-Leloup, B., Douay, F.,
 2016. Effective radium concentration in topsoils contaminated by lead and zinc smelters. Sci. Total
 Environ. 566-567, 865-876.
- Girault, F., Perrier, F., Przylibski, T.A., 2018. Radon-222 and radium-226 occurrence in water: A review.
 Geol. Soc. Lond. S. P. 451, 131-154, doi:110.1144/SP1451.1143.
- Girault, F., Schubnel, A., Pili, É., 2017b. Transient radon signals driven by fluid pressure pulse, micro-crack
 closure, and failure during granite deformation experiments. Earth Planet. Sci. Lett. 474, 409-418.
- 710 GIS Sol, 2011. L'état des sols de France. Groupement d'intérêt scientifique sur les sols. 188p. (in French)
- Gunn, K.B., Mistry, K.B., 1970. The effect of chelating agents on the absorption of radium by plants. Plant
 and Soil 33, 7-16.
- Ham, G.J., Wilkins, B.T., Ewers, L.W., 2001. ²¹⁰Pb, ²¹⁰Po, ²²⁶Ra, U and Th in arable crops and ovine liver:
 Variations in concentrations in the United Kingdom and resultant doses. Radiat. Prot. Dosim. 93, 151159.
- Hassan, N.M., Ishikawa, T., Hosoda, M., Iwaoka, K., Sorimachi, A., Sahoo, S.K., Janik, M., Kranrod, C.,
 Yonehara, H., Fukushi, M., Tokonami, S., 2011a. The effect of water content on the radon emanation
 coefficient for some building materials used in Japan. Radiat. Meas. 46, 232-237.
- He, Q., Walling, D.E., 1996. Interpreting particle size effects in the adsorption of ¹³⁷Cs and unsupported
 ²¹⁰Pb by mineral soils and sediments. J. Environ. Radioact. 30, 117-137.
- Hu, N., Ding, D., Li, G., Zheng, J., Li, L., Zhao, W., Wang, Y., 2014. Vegetation composition and ²²⁶Ra uptake by native plant species at a uranium mill tailings impoundment in South China. J. Environ. Radioact. 129, 100-106.
- Ibrahim, S.A., Whicker, F.W., 1992. Comparative plant uptake and environmental behavior of U-series
 radionuclides at a uranium mine-mill. J. Radioanal. Nucl. Chem. 156, 253-267.
- James, J.P., Dileep, B.N., Ravi, P.M., Joshi, R.M., Ajith, T.L., Hegde, A.G., Sarkar, P.K., 2011. Soil to leaf
 transfer factor for the radionuclides ²²⁶Ra, ⁴⁰K, ¹³⁷Cs and ⁹⁰Sr at Kaiga region, India. J. Environ. Radioact.
 102, 1070-1077.

- Jha, V.N., Tripathi, R.M., Sethy, N.K., Sahoo, S.K., Shukla, A.K., Puranik, V.D., 2010. Bioaccumulation of
 ²²⁶Ra by plants growing in fresh water ecosystem around the uranium industry at Jaduguda, India. J.
 Environ. Radioact. 101, 717-722.
- Karunakara, N., Somashekarappa, H.M., Narayana, Y., Avadhani, D.N., Mahesh, H.M., Siddappa, K., 2003.
 ²²⁶Ra, ⁴⁰K and ⁷Be activity concentrations in plants in the environment of Kaiga, India. J. Environ.
 Radioact. 65, 255-266.
- Kirchner, G., Daillant, O., 1998. Accumulation of ²¹⁰Pb, ²²⁶Ra and radioactive cesium by fungi. Sci. Total
 Environ. 222, 63-70.
- Krmar, M., Wattanavatee, K., Radnović, D., Slivka, J., Bhongsuwan, T., Frontasyeva, M.V., Pavlov, S.S.,
 2013. Airborne radionuclides in mosses collected at different latitudes. J. Environ. Radioact. 117, 45-48.
- Lauria, D.C., Ribeiro, F.C.A., Conti, A.A., Loureiro, F.A., 2009. Radium and uranium levels in vegetables
 grown using different farming management systems. J. Environ. Radioact. 100, 176-183.
- Li, Z., Ma, Z., van der Kuijp, T.J., Yuan, Z., Huang, L., 2014. A review of soil heavy metal pollution from
 mines in China: Pollution and health risk assessment. Sci. Total Environ. 468-469, 843-853.
- Lin, H., 2010. Earth's Critical Zone and hydropedology: concepts, characteristics, and advances. Hydrol.
 Earth Syst. Sci. 14, 25-45.
- Luetzelschwab, J.W., Helweick, K.L., Hurst, K.A., 1989. Radon concentrations in five Pennsylvania soils.
 Health Physics 56, 181-188.
- Madruga, M.J., Brogueira, A., Alberto, G., Cardoso, F., 2001. ²²⁶Ra bioavailability to plants at the Urgeiriça uranium mill tailings site. J. Environ. Radioact. 54, 175-188.
- Manta, D.S., Angelone, M., Bellanca, A., Neria, R., Sprovieri, M., 2002. Heavy metals in urban soils: A case
 study from the city of Palermo (Sicily), Italy. Sci. Total Environ. 300, 229-243.
- 751 Markkanen, M., Arvela, H., 1992. Radon emanation from soils. Radiat. Prot. Dosim. 45, 269-272.
- Markose, P.M., Bhat, I.S., Pillai, K.C., 1993. Some characteristics of ²²⁶Ra transfer from soil and uranium
 mill tailings to plants. J. Environ. Radioact. 21, 131-142.
- Marple, M.L., 1980. Ra-226 in vegetation and substrates at inactive uranium mill sites. Rep. LA-8183-T, Los
 Alamos Scientific Laboratory, Los Alamos, NM.
- Martínez-Aguirre, A., Periáñez, R., 1998. Soil to plant transfer of ²²⁶Ra in a marsh area: Modelling
 application. J. Environ. Radioact. 39, 199-213.
- Medley, P., Bollhöfer, A., 2016. Influence of group II metals on Radium-226 concentration ratios in the native green plum (*Buchanania obovata*) from the Alligator Rivers Region, Northern Territory, Australia.
 J. Environ. Radioact. 151, 551-557.
- Medley, P., Bollhöfer, A., Parry, D., Martin, P., 2013. Radium concentration factors in passionfruit
 (*Passiflora foetida*) from the Alligator Rivers Region, Northern Territory, Australia. J. Environ. Radioact.
 126, 137-146.
- Moyen, C., Roblin, G., 2010. Uptake and translocation of strontium in hydroponically grown maize plants,
 and subsequent effects on tissue ion content, growth and chlorophyll a/b ratio: comparison with Ca
 effects. Environ. Exp. Bot. 68, 247-257.
- Mrdakovic Popic, J., Oughton, D.H., Salbu, B., Skipperud, L., 2020. Transfer of naturally occurring
 radionuclides from soil to wild forest flora in an area with enhanced legacy and natural radioactivity in
 Norway. Environ. Sci.: Processes Impacts 22, 350.
- Nathwani, J.S., Phillips, C.R., 1979. Adsorption of ²²⁶Ra by soils in the presence of Ca²⁺ ions. Specific adsorption (II). Chemosphere 5, 293-299.
- 772 Nazaroff, W.W., 1992. Radon transport from soil to air. Rev. Geophys. 30, 137-160.
- Nezami, S., Malakouti, M.J., Samani, A.B., Maragheh, M.G., 2016. Effect of low molecular weight organic
 acids on the uptake of ²²⁶Ra by corn (*Zea mays* L.) in a region of high natural radioactivity in RamsarIran. J. Environ. Radioact. 164, 145-150.
- Nicolas, A., Girault, F., Schubnel, A., Pili, É., Passelègue, F., Fortin, J., Deldicque, D., 2014. Radon
 emanation from brittle fracturing in granites under upper crustal conditions. Geophys. Res. Lett. 41, 54365443.
- Paul, A.C., Pillai, K.C., 1986. Transfer and uptake of radium in a natural and in a technologically modified
 radiation environment. J. Environ. Radioact. 3, 55-73.

- Perrier, F., Aupiais, J., Girault, F., Przylibski, T.A., Bouquerel, H., 2016a. Optimized measurement of
 radium-226 concentration in liquid samples with radon-222 emanation. J. Environ. Radioact. 157, 52-59.
- Perrier, F., Girault, F., Bouquerel, H., 2018. Effective radium concentration in rocks, soils, plants, and bones.
 Geol. Soc. Lond. S. P. 451, 113-129.
- Perrier, F., Girault, F., Bouquerel, H., Bollinger, L., 2016b. Effective radium concentration in agricultural
 versus forest topsoils. J. Environ. Radioact. 160, 123-134.
- Popovic, D., Todorovic, D., Frontasyeva, M., Ajtic, J., Tasic, M., Rajsic, S., 2008. Radionuclides and heavy
 metals in Borovac, Southern Serbia. Environ. Sci. Pollut. Res. 15, 509-520.
- Przylibski, T.A., 2000. Estimating the radon emanation coefficient from crystalline rocks into groundwater.
 Appl. Radiat. Isotopes 53, 473-479.
- Pulhani, V.A., Dafauti, S., Hedge, A.G., Sharma, R.M., Mishra, U.C., 2005. Uptake and distribution of
 natural radioactivity in wheat plants from soil. J. Environ. Radioact. 79, 331-346.
- Rihs, S., Prunier, J., Thien, B., Lemarchand, D., Pierret, M.-C., Chabaux, F., 2011. Using short-lived nuclides of the U- and Th-series to probe the kinetics of colloid migration in forested soils. Geochim. 795
 Cosmochim. Acta 75, 7707-7724.
- Ryan, B., Martin, P., Iles, M., 2005. Uranium-series radionuclides in native fruits and vegetables of northern
 Australia. J. Radioanal. Nucl. Chem. 264, 407-412.
- Sakoda, A., Ishimori, Y., Yamaoka, K., 2011. A comprehensive review of radon emanation measurements
 for mineral, rock, soil, mill tailing and fly ash. Appl. Radiat. Isot. 69, 1422-1435.
- Sam, A.K., Eriksson, Å., 1995. Radium-226 uptake by vegetation grown in Western Sudan. J. Environ.
 Radioact. 29, 27-38.
- Sheppard, S.C., Evenden, W.G., 1988a. The assumption of linearity in soil and plant concentration ratios: an
 experimental evaluation. J. Environ. Radioact. 7, 221-247.
- Sheppard, S.C., Evenden, W.G., 1988b. Critical compilation and review of plant/soil concentration ratios for
 uranium, thorium and lead. J. Environ. Radioact. 8, 255-285.
- Sheppard, S.C., Sheppard, M.I., Tait, J.C., Sanipelli, B.L., 2006. Revision and meta-analysis of selected
 biosphere parameter values for chlorine, iodine, neptunium, radium, radon and uranium. J. Environ.
 Radioact. 89, 115-137.
- Simon, S.L., Ibrahim, S.A., 1987. The plant/soil concentration ratio for calcium, radium, lead, and polonium:
 Evidence for non-linearity with reference to substrate concentration. J. Environ. Radioact. 5, 123-142.
- Simon, S.L., Ibrahim, S.A., 1990. Biological uptake of radium by terrestrial plants. In The Environmental
 Behaviour of Radium, Vol. 1, Technical Reports Series No. 310, IAEA, Vienna, 545-599.
- Soudek, P., Petřík, P., Vágner, M., Tykva, R., Plojhar, V., Petrová, Š., Vaněk, T., 2007a. Botanical survey
 and screening of plant species which accumulate ²²⁶Ra from contaminated soil of uranium waste depot.
 Eur. J. Soil Biol. 43, 251-261.
- Soudek, P., Petrová, Š., Benešová, D., Tykva, R., Vaňková, R., Vaněk, T., 2007b. Comparison of ²²⁶Ra
 nuclide from soil by three woody species *Betula pendula*, *Sambucus nigra* and *Alnus glutinosa* during the
 vegetation period. J. Environ. Radioact. 97, 76-82.
- Soudek, P., Petrová, Š., Benešová, D., Kotyza, J., Vágner, M., Vaňková, R., Vaněk, T., 2010. Study of soil–
 plant transfer of ²²⁶Ra under greenhouse conditions. J. Environ. Radioact. 101, 446-450.
- Stoulos, S., Manolopoulou, M., Papastefanou, C., 2004. Measurement of radon emanation factor from
 granular samples: Effects of additives in cement. Appl. Radiat. Isot. 60, 49-54.
- Tanner, A.B., 1964. Radon migration in the ground: A review, in: Adams, J.A.S., Lowder, W.M. (Eds.), The
 natural radiation environment. Univ. of Chicago Press, Chicago, pp. 161-190.
- Thiry, Y., Van Hees, M., 2008. Evolution of pH, organic matter and ²²⁶radium/calcium partitioning in U mining debris following revegetation with pine trees. Sci. Total Environ. 393, 111-117.
- Thu, H.N.P., Thang, N.V., Hao, L.C., 2020. The effects of some soil characteristics on radon emanation and
 diffusion. J. Environ. Radioact. 216, 106189.
- Tuovinen, T.S., Kolehmainen, M., Roivainen, P., Kumlin, T., Makkonen, S., Holopainen, T., Juutilainen, J.,
 2016. Nonlinear transfer of elements from soil to plants: impact on radioecological modeling. Radiat.
 Environ. Biophys. 55, 393-400.
- Uchida, S., Tagami, K., 2007. Soil-to-crop transfer factors of radium in Japanese agricultural fields. J. Nucl.
 Radiochem. Sci. 8, 137-142.

- Uchida, S., Tagami, K., Shang, Z.R., Choi, Y.H., 2009. Uptake of radionuclides and stable elements from
 paddy soil to rice: a review. J. Environ. Radioact. 100, 739-745.
- Uğur, A., Özden, B., Saç, M.M., Yener, G., 2003. Biomonitoring of ²¹⁰Po and ²¹⁰Pb using lichens and mosses
 around a uraniferous coal-fired power plant in western Turkey. Atmos. Environ. 37, 2237-2245.
- Vandenhove, H., Van Hees, M., 2007. Predicting radium availability and uptake from soil properties.
 Chemosphere 69, 664-674.
- Vandenhove, H., Eyckmans, T., Van Hees, M., 2005. Can barium and strontium be used as tracers for
 radium in soil-plant transfer studies? J. Environ. Radioact. 81, 255-267.
- Vandenhove, H., Olyslaegers, G., Sanzharova, N., Shubina, O., Reed, E., Shang, Z., Velasco, H., 2009.
 Proposal for new best estimates of the soil-to-plant transfer factor of U, Th, Ra, Pb and Po. J. Environ.
 Radioact. 100, 721-732.
- Vasconcellos, L.M.H., Amaral, E.C.S., Vianna, M.E., Penna Franca, E., 1987. Uptake of ²²⁶Ra and ²¹⁰Pb by
 food crops cultivated in a region of high natural radioactivity in Brazil. J. Environ. Radioact. 5, 287-302.
- Vera Tomé, F., Blanco Rodr!1guez, P., Lozano, J.C., 2002. Distribution and mobilization of U, Th and ²²⁶Ra
 in the plant–soil compartments of a mineralized uranium area in south-west Spain. J. Environ. Radioact.
 59, 41-60.
- Vera Tomé, F., Blanco Rodríguez, M.P., Lozano, J.C., 2003. Soil-to-plant transfer factors for natural
 radionuclides and stable elements in a Mediterranean area. J. Environ. Radioact. 65, 161-175.
- Vera Tomé, F., Blanco Rodríguez, P., Lozano, J.C., 2009. The ability of *Helianthus annuus* L. and *Brassica juncea* to uptake and translocate natural uranium and ²²⁶Ra under different milieu conditions.
 Chemosphere 74, 293-300.
- Vera Tomé, F., Rodríguez, P.B., Lozano, J.C., 2008. Elimination of natural uranium and ²²⁶Ra from contaminated waters by rhizofiltration using *Helianthus annuus* L. Sci. Total Environ. 393, 351-357.
- Wattanavatee, K., Krmar, M., Bhongsuwan, T., 2017. A survey of natural terrestrial and airborne
 radionuclides in moss samples from the peninsular Thailand. J. Environ. Radioact. 177, 113-127.
- Weis, J.S., Weis, P., 2004. Metal uptake, transport and release by wetland plants: implications for
 phytoremediation and restoration. Environ. Int. 30, 685-700.
- Whicker, F.W., Shaw, G., Voigt, G., Holm, E., 1999. Radioactive contamination: state of the science and its
 application to predictive models. Environ. Pollut. 100, 133-149.
- Yan, X., Luo, X.G., 2016. Uptake of uranium, thorium, radium and potassium by four kinds of dominant
 plants grown in uranium mill tailing soils from the southern part of China. Radioprotection 51, 141-144.

865 **Tables caption**

866

Table 1. Statistics of effective ²²⁶Ra concentration (EC_{Ra}) of soil and plant samples, and inferred ²²⁶Ra soil-to-plant transfer ratios (R_{SP}). All experimental uncertainties are defined at one standard deviation, 68% confidence level (CL).

870

871 **Table 2.** Statistics of effective ²²⁶Ra concentration (EC_{Ra}) of soil and plant samples, and inferred 872 ²²⁶Ra soil-to-plant transfer ratios (R_{SP}) from different compartments of plant species growing in 873 granitic and metamorphic context (A subset).

874

875 **Table 3.** Statistics of effective radium-226 concentration (EC_{Ra}) of soil and plant samples, and 876 inferred ²²⁶Ra soil-to-plant transfer ratios (R_{SP}) from different compartments of plant species 877 growing in calcareous and sedimentary context (B subset).

878 Figures caption

879

Fig. 1. Location of sampling sites in France with simplified geological domains (modified from
BRGM, France).

882

Fig. 2. Relative experimental uncertainty (in %) as a function of EC_{Ra} (in Bq kg⁻¹) for our plant measurements. Our other measurements are plotted for comparison. The symbol color stands for the sample mass used in the accumulation experiments.

886

887 Fig. 3. Distributions of EC_{Ra} values of (a) rocks (n = 2015), (b) soils and sediments (n = 2070), and 888 (c) plants (n = 174). The data from rocks, including samples from Nepal (1128), France (477), and 889 other locations, were partially presented by Girault and Perrier (2019). The data from soils and 890 sediments, including samples from France (1400), Nepal (432), and other locations, and from 891 plants, were partially presented by Perrier et al. (2018). Cumulated distributions (scale on the right 892 hand side) are shown as solid black curves and geometric mean values as vertical dashed black 893 lines. The dashed gray curves represent the log-normal distribution with mean and RMS of 1.9 Bq kg⁻¹ and 6.0 in (a), and 7.1 Bq kg⁻¹ and 1.8 in (b), respectively. In (b), the distribution of EC_{Ra} 894 895 values of top soils nearby plant samples (n = 24) is shown in red. In (c), the distribution of EC_{Ra} values of plants for which we have a nearby top soil sample (n = 108) is also shown in red. 896

897

Fig. 4. (a) Plant EC_{Ra} as a function of EC_{Ra} of the closest top soil sample. (b) Plant EC_{Ra} divided by the emanation coefficient of plants E_P (*i.e.*, bulk plant C_{Ra}) as a function of the closest top soil sample EC_{Ra} divided by the emanation coefficient of soil E_S (*i.e.*, bulk soil C_{Ra}). (c) ²²⁶Ra soil-toplant transfer ratio (R_{SP}) as a function of the bulk soil C_{Ra} . Our whole data set is plotted separately for plants growing in granitic and metamorphic context (A subset; in red) and in calcareous and

903 sedimentary context (B subset; in blue). The closest soil corresponds to a nearby top soil sample 904 available at a distance of less than 100 m from the given plant sample. The data are represented as 905 diamonds and the means as circles. In (a) and (b), the dashed line represents the radioactive 906 equilibrium between plant and soil samples. In (b), the dash-dot curve corresponds to a power-law 907 regression fit $(EC_{Ra}/E_P = 0.00344(EC_{Ra}/E_S)^{1.37}; R^2 = 0.32)$. In (b) and (c), the solid black curve 908 represents the exponential relation established by Simon and Ibrahim (1990) for high ²²⁶Ra levels in 909 soil: in (b) $EC_{Ra}/E_P = 0.04(1 - \exp(-5.1EC_{Ra}/E_S)) + 0.028EC_{Ra}/E_S$, and in (c) $R_{SP} = 0.04/(EC_{Ra}/E_S)(1 - EC_{Ra}/E_S))$ $exp(-5.1EC_{Ra}/E_S))+0.028$. In (c), the grey line, $R_{SP} = 1.24 \times 10^5 (EC_{Ra}/E_S)^{-3.43}$, separates the A and B 910 911 subsets.

912

Fig. 5. Distributions of ²²⁶Ra soil-to-plant transfer ratio, R_{SP} , for (a) all our plant samples for which we have a nearby top soil sample (n = 108), (b) plant samples in the A subset (granitic and metamorphic context; n = 50), (c) plant samples in the B subset (calcareous and sedimentary context; n = 58), and (d) compiled data from the literature (n = 844). The data shown in (a) were partially presented in Perrier et al. (2018). Geometric mean values are shown as vertical dashed black lines. In (d), the dashed gray curve represents the log-normal distribution with mean of 0.045 and RMS of 5.0.

920

921 **Fig. 6.** ²²⁶Ra soil-to-plant transfer ratio, R_{SP} , expressed in percent, for different compartments of 922 given plant species growing in granitic and metamorphic context (A subset; in red, on the right) and 923 in calcareous and sedimentary context (B subset; in blue, on the left).





Figure 3









B: CALCAREOUS & SEDIMENTARY CONTEXT

A: GRANITIC & METAMORPHIC CONTEXT

Table 1

Plant samples set	All set	A subset: Granitic & metamorphic context	B subset: Calcareous & sedimentary context
~ · · ~ ~ · · · ·			
Soil <i>EC</i> _{Ra} (Bq kg ⁻¹)			
Number of samples	24	9	15
Min–Max	2.7–90	12–90	2.7–45
Arithmetic mean	25.6 ± 4.0	41.1 ± 7.4	16.8 ± 3.2
Geometric mean	18.86 ± 0.21	36.12 ± 0.64	12.85 ± 0.18
Median (at 90 % CL)	23.80 ± 0.36	35.8 ± 5.4	11.3 ± 2.2
Plant EC _{Ra} (Bq kg ⁻¹)			
Number of samples	108	50	58
Min–Max	0.020-113	0.49–113	0.020-4.18
Arithmetic mean	7.6 ± 1.4	15.8 ± 2.5	0.64 ± 0.11
Geometric mean	1.663 ± 0.029	9.21 ± 0.12	0.380 ± 0.012
Median (at 90 % CL)	1.36 ± 0.01	12.83 ± 0.02	0.380 ± 0.001
RSP			
Min–Max	0.00069-0.37	0.0053-0.37	0.00069-0.13
Arithmetic mean	0.0625 ± 0.0085	0.117 ± 0.015	0.0158 ± 0.0038
Geometric mean	0.01877 ± 0.00044	0.0729 ± 0.0019	0.00583 ± 0.00022
Median (at 90 % CL)	0.01928 ± 0.00011	0.0959 ± 0.0015	0.00455 ± 0.00006

Table 2

Sampling site	Plant species and location	Soil <i>EC</i> _{Ra} (Bq kg ⁻¹)	Plant EC _{Ra} (Bq kg ⁻¹)	²²⁶ Ra soil-to-plant transfer ratio <i>R</i> _{SP}	
				Mean ± 1σ	Dispersion $\pm 1\sigma$ (%) (number of samples)
La Sapine,	Fagus sylvatica	29.6 ± 1.6			
Mont Lozère granite,	Dead leaves		15.25 ± 0.70	0.145 ± 0.024	
Lozère, France	Leaves		7.4 ± 1.1	0.071 ± 0.015	
	Stems		14.1 ± 1.1	0.1209 ± 0.0015	$2.5 \pm 1.3 (n = 2)$
	Branches		10.5 ± 1.0	0.0898 ± 0.0026	$5.8 \pm 2.9 (n = 2)$
Anduze,	Quercus robur	25.9 ± 1.4			
Cévennes granite,	Acorns		1.78 ± 0.16	0.0193 ± 0.0036	
Gard, France	Dead leaves		4.53 ± 0.30	0.0491 ± 0.0086	
	Leaves		19.50 ± 0.89	0.211 ± 0.036	
	Branches		23.1 ± 4.1	0.2506 ± 0.0075	$39 \pm 14 \ (n = 5)$
	Pinus pinaster	25.9 ± 1.4			
	Pine cones		0.490 ± 0.079	0.0053 ± 0.0012	
	Pine needles		1.760 ± 0.061	0.01906 ± 0.00066	$6.0 \pm 2.5 \ (n = 3)$
	Stems		2.95 ± 0.22	0.0320 ± 0.0057	
	Bark from trunk		1.03 ± 0.11	0.0112 ± 0.0022	
Ambazac,	Fagus sylvatica	46.6 ± 2.7			
Saint-Sylvestre granite,	Branches		18.87 ± 0.79	0.114 ± 0.019	
Haute-Vienne, France	Roots		44.0 ± 1.8	0.265 ± 0.045	
	Castanea sativa	89.8 ± 3.4			
	Burrs		9.15 ± 0.40	0.0551 ± 0.0094	
	Leaves		38.3 ± 1.5	0.231 ± 0.039	
Saint-Yrieix-la-Perche,	Quercus robur	48.3 ± 2.5			
Saint-Yrieix arc leptynite,	Leaves		18.5 ± 1.2	0.107 ± 0.019	
Haute-Vienne, France	Branches		15.6 ± 1.1	0.091 ± 0.016	
	Castanea sativa	51.9 ± 2.7			
	Flowers		17.3 ± 1.2	0.093 ± 0.016	
	Leaves		23.1 ± 4.7	0.125 ± 0.033	
	Branches		19.2 ± 1.2	0.104 ± 0.018	
	Spartium junceum	50.1 ± 2.6			
	Leaves		15.00 ± 0.96	0.084 ± 0.015	
	Stems		5.18 ± 0.43	0.0290 ± 0.0053	
	Pteridium aquilinum	50.1 ± 2.6			
	Leaves		14.0 ± 2.9	0.078 ± 0.021	
	Stems		1.38 ± 0.14	0.0077 ± 0.0015	

Table 3

Sampling site	Plant species	Soil <i>EC</i> _{Ra}	Plant <i>EC</i> _{Ra}	²²⁶ Ra soil-to-plant transfer ratio R _{SP}	
	and location	(Bq kg ⁻¹)	(Bq kg ⁻¹)	Mean ± 1σ	Dispersion $\pm 1\sigma$ (%)
					(number of samples)
Puéchabon,	Quercus ilex	22.6 ± 1.2			
Kimmeridgian limestone,	Leaves		0.269 ± 0.073	0.00339 ± 0.00096	$49 \pm 24 \ (n = 3)$
Hérault, France	Branches and stems		0.33 ± 0.11	0.0042 ± 0.0015	$60 \pm 32 \ (n = 3)$
	Trunk		0.158 ± 0.032	0.00193 ± 0.00039	$40 \pm 16 \ (n = 4)$
	Bark from trunk		1.18 ± 0.34	0.0145 ± 0.0042	$50 \pm 25 (n = 3)$
	Roots		0.63 ± 0.28	0.0078 ± 0.0035	$77 \pm 47 (n = 3)$
	Phillyrea latifolia	22.6 ± 1.2			
	Leaves		0.262 ± 0.057	0.00328 ± 0.00075	$33 \pm 15 (n = 3)$
	Stems		0.1309 ± 0.0039	0.001628 ± 0.000048	$5.1 \pm 2.1 (n = 3)$
	Roots		0.56 ± 0.12	0.0071 ± 0.0017	$41 \pm 19 (n = 3)$
	Buxus sempervirens	22.6 ± 1.2			
	Leaves		0.179 ± 0.039	0.00223 ± 0.00049	$38 \pm 18 (n = 3)$
	Branches		0.1747 ± 0.0057	0.002114 ± 0.000014	5.8 ± 2.9 (n = 2)
	Wood		0.42 ± 0.10	0.0049 ± 0.0015	
	Roots		0.382 ± 0.051	0.00478 ± 0.00071	$26 \pm 11 (n = 3)$
Pech Merle,	Quercus robur	24.5 ± 1.4			
Callovo-Oxfordian limestone,	Dead leaves		0.381 ± 0.038	0.00436 ± 0.00083	
Lot, France	Leaves		0.71 ± 0.29	0.0081 ± 0.0036	
	Stems		0.160 ± 0.019	0.00184 ± 0.00037	
	Branches		0.187 ± 0.014	0.00214 ± 0.00039	
	Bark from trunk		0.67 ± 0.14	0.0077 ± 0.0021	
	Roots		0.51 ± 0.10	0.0058 ± 0.0015	
Anduze,	Tilia ×europaea	7.10 ± 0.71			
Triassic gypsum & Oxfordian limestone,	Dead leaves		0.66 ± 0.10	0.0262 ± 0.0063	
Gard, France	Leaves		0.750 ± 0.085	0.0297 ± 0.0064	
	Branches		0.20 ± 0.11	0.0079 ± 0.0046	
Vaugrigneuse,	Quercus robur	5.40 ± 0.34			
Stampian Fontainebleau sand,	Leaves		1.10 ± 0.081	0.057 ± 0.010	
Essonne, France	Branches		0.753 ± 0.052	0.0392 ± 0.0070	
Chambon-la-Forêt,	Quercus robur	7.10 ± 0.48			
Burdigalian sand and marls,	Leaves		3.46 ± 0.27	0.137 ± 0.025	
Loiret, France	Stems		4.18 ± 0.23	0.165 ± 0.029	
	Branches		2.50 ± 0.20	0.099 ± 0.018	

Graphical Abstract

