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CRediT authorship contribution statement

A.F. Saad: Conceptualization, Methodology, Writing - Reviewing, Editing. **M. Fromm:** Writing - Reviewing. **S.A. Abd El Aal** and **Mona H. Ibraheim:** Investigation. **A.A. El-Namrouty** Writing - Original draft preparation. **Aya M. Nwara:** Investigation, Data curate, Data analysis. **S.A. Kandil:** Investigation. **Mahmoud S. Dawood:** Writing - Original draft preparation, Investigation.

Loss of chemical bonds induced by high doses of γ -radiation in a PADC polymer film: the influence of dose and dose rate on radiation chemical yields

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Abstract

The effects of gamma rays on molecular properties of a poly allyl diglycol carbonate (PADC) polymer, commercially named as CR-39, have been studied. PADC films were exposed to gamma rays at doses ranging from 0.5 to 3 MGy. Gamma exposure induces considerable molecular modifications in PADC films. These modifications were analyzed using Fourier transform infrared (FT-IR) spectroscopy. FT-IR absorption spectra of the gamma irradiated samples show a change in the intensity and the shape of the absorption bands of the chemical groups forming the PADC polymer relative to the pristine one. We have determined G-values for the loss of carbonate ester and ether chemical moieties within PADC. By comparing our G-values with those from other studies, we show that dose rate and dose significantly affect the radiation chemical yields and play an important role of molecular bond kinematics.

Key Words: PADC; γ -ray; FT-IR; Molecular bonds; G-value, Dose; Dose rate

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1. Introduction

Since the end of the seventies of the last century, poly allyl diglycol carbonate (PADC) polymer material, CR-39, is used for radiation detection (Cartwright et al., 1978; Cassou and Benton, 1978). PADC is commonly used in various branches of science and technology such as nuclear and particle physics, cosmic rays, neutron dosimetry and radiation biology [Nikezic and Yu, 2004; Price, 2005]. Exposure to ionizing radiation such as gamma and X-rays, alpha particles and heavy ions can play a prominent role in modifying physical, chemical, molecular, structural, optical, electrical and thermal properties of the polymeric materials (Marletta, 1990; Steckenreiter et al., 1997; Saha et al., 2000; Saad et al., 2005; Khan et al., 2005; Singh and Prasher, 2006; Sharma et al., 2007; Kumar et al., 2011; Raghuvanshi et al., 2012; Saad et al., 2014a,b). In general, chain scission of PADC polymer is seen as the major process of the degradation of the main chain of that polymer. Such scission process caused by ionizing radiation (UV radiation, X-rays, gamma rays, swift ions), is known to release carbon mono oxide and carbon dioxide (Yamauchi et al., 2003; Saad et al., 2005) and references therein.

In order to describe molecular and chemical damages formed in PADC and other particle track detector (PTD) polymers by ionizing radiations; an international collaboration that aimed at studying radiation damage using spectroscopic measurements has been set up between Japan and France (T. Yamauchi, T. Kusumoto, R. Barillon, and their co-workers) (Barillon and Yamauchi, 2003; Yamauchi et al., 2005a,; Yamauchi et al., 2008a,b; Mori et al., 2009; Yamauchi, 2012; Mori et al., 2011; Mori et al., 2013; Barillon et al., 2013; Kusumoto et al., 2015; Kusumoto et al., 2016; Yamauchi et al., 2018). This team was able to highlight which are the chemical units of the PADC structure that are involved in the destructive process at the origin of ionizing radiation damage trails as mentioned in a recent review paper (Fromm et al., 2019). That vast study is based on analyses of spectroscopic

experimental data of the loss/appearance of specific chemical functions further to PADC/PTD polymer exposure to various ionizing radiations such as alpha particles, gamma rays and heavy ions. The main achievements of these studies are connected with a quantitative determination of the sensibility of PADC chemical functional groups to ionizing radiations expressed in terms of cross-sections and G-values (radiation chemical yields) for the loss of given chemical groups. As previously mentioned, it was also shown by other groups and further confirmed, that carbon dioxide formation goes together with radiation damage (Rickards et al., 1992; Chong et al., 1997; Saad et al., 2001; Yamauchi et al., 2001; Malek, 2002; Yamauchi et al., 2005b; Saad et al., 2005). In addition, it has also been established that the amount of CO₂ in the detector is closely related not only to the latent track formation mechanism but also to the track detection sensitivity (Saad et al., 2001). Carbon dioxide absorbed in the polymer material plays indeed a positive role in the detector's sensitization (Fujii et al., 1995; Csige, 1997; Fujii et al., 1997; Baradacs et al., 2008; d'Errico et al., 2017).

O'Sullivan et al. (O'Sullivan et al., 1981) measured G value of γ - ray (an average of 1.25 MeV from ⁶⁰Co) scissions in PADC per 100 eV as \approx 10, from the correlation between the minimum value of the primary ionization rate and the G values for various types of polymers based on the observed bulk etching rates. Mori et al., 2009 studied the effect of γ -ray irradiation on the radiation chemical yield, for loss of carbonate ester bonds in PADC films by means of FT-IR spectrometry. They found that the G value is 20 (scissions/100 eV) and is somewhat higher than that from heavy ion irradiations (147 MeV Fe ions) (Yamauchi et al., 2008a; Mori et al., 2011). These authors also evaluated the chemical radiation yield for other kinds of heavy ions and protons to different functional groups in PADC films, for example, ether and ethylene. They also found that the highest G values were observed in PADC films irradiated by protons, namely about 20 scissions/100 eV for both bonds, namely

ether and carbonate ester. The values decreased down to about 5 scissions/100 eV for films irradiated with heavier ions. Studying gamma irradiation of PADC at higher doses can be an effective way to understand the latent track formation, because the local dose in a heavy ion track core exceeds several MGys in heavy ions. Recently it was also shown that a critical dose of ~50-60 kGy exists above which carbonate ester density in PADC start decreasing (Kusumoto, et al., 2020). For ether bonds, such threshold was not observed.

In the current paper, we propose to study the changes of the molecular structure induced by γ -radiation at different high doses over the range 0.5 to 3 MGy. For this purpose, FT-IR spectrometer as a standard technique of spectroscopic analysis of polymeric materials is used.

2. Experimental procedure

Fig. 1 shows the chemical structure of a poly allyl diglycol carbonate (PADC) high polymer which has a length of chain between two adjacent carbonate groups in symmetrical positions. PADC sheets of TASTRAK used in this investigation were manufactured by Track Analysis Systems Ltd. (TASL), Bristol, UK. Polymeric detector samples of a uniform thickness of 750 μm were cut to a size of 2×3 cm². The detector samples were irradiated in atmosphere at room temperature using a commercial gamma irradiator of type MC-20 (Russia) with ⁶⁰Co energy source which provides a dose rate of 1.2 kGy/h. The PADC samples were irradiated with different doses ranging from 0.5 to 3 MGy of gamma rays. The samples were irradiated for various times in order to get variable doses. These irradiations were performed at the Gamma Cell, *Nuclear Research Centre*, Inshas , *Atomic Energy Authority* .Egypt. The FT-IR spectrometer was operated in attenuated total reflection (ATR) mode. The working range of the spectrometer is 4600-400 cm⁻¹. A resolution of 4 cm⁻¹, scan numbers of

16 and apodization of cosine were used. The Fourier transform infrared (FT-IR) spectrometer model of Vertex 70, made in Germany is used. FT-IR spectroscopy was performed in the wave number range 4000-400 cm⁻¹ at room temperature on the pristine and irradiated CR-39 samples.

3. Results

3.1. FT-IR spectral changes due to γ-ray irradiation

FT-IR spectra in ATR mode for pristine and irradiated samples of PADC films are shown in Fig. 2. It is easy to find a significant difference in FT-IR spectra between irradiated and unirradiated samples of PADC because changes in the molecular structure due to our irradiation conditions were high, thus it can be easily detected in ATR mode. Absorption bands were identified. The broad absorption peak around 2920 cm⁻¹ is due to C–H stretching. Two strong absorption peaks at 1734 and 1240 cm⁻¹ are assigned to C=O and C-O-C bonds. respectively, which compose carbonate ester bonds (Gagnadre et al., 1993; Darraud et al., 1994; Lounis-Mokrani wt al., 2003; Saad et al., 2001). Three weak absorption peaks at 1132, 1076 and 1022 cm⁻¹ are assigned to ether bonds. In addition, an absorption peak at 959 cm⁻¹ is assigned to wagging vibration of the double bond C=C. Two other absorption peaks at 878 and 785 cm⁻¹ are assigned to C-H rocking out-of-plane deformation (Gagnadre et al., 1993; Lounis-Mokrani wt al., 2003). Indeed, the OH groups at around 3500 cm⁻¹ are clearly suppressed in FT-IR spectra in ATR mode for the pristine and exposed samples of PADC as shown in Fig. 2. This is due to that the yield of OH group formation is below the detection limit in that mode compared to the case of another transmission mode, as presented in our previous studies (Saad et al., 2001; Saad et al., 2005).

A new broad band (as a shoulder) at wave number 1627 cm⁻¹, which could be identified as C=C bond, is found in the FT-IR spectra of gamma exposed samples of PADC films as also shown in Fig. 2. It can be seen that these shoulder absorption bands are strongly affected by gamma irradiation in their shapes and/or intensities. The band area of C=C at 1627 cm⁻¹ is plotted as a function of irradiation dose, as shown in Fig. 3. The net area of the C=C absorption band, which actually represents the net absorbance, ΔA, is given by Eq.

$$\Delta A = A_{irr} - A_0 \tag{1}$$

where A_{irr} and A_o are the area under the absorption peak of the irradiated and unirradiated PADC samples, respectively. The band area of C=C is quickly enhanced with increasing dose up to 1 MGy, then it somewhat decreases with higher dose at 1.5 MGy. Beyond 1.5 MGy the net absorbance sharply increases up to a maximum at 2.5 MGy, and then sharply decreased at 3 MGy as can be seen in Fig. 3. In fact, the response of PADC film to the gamma dose of 2.5 MGy reveals a differing behavior. This inconsistent response to that dose was also observed by other standard analysis techniques as reported in our recent study (Saad et al., 2018). In general, the PADC film might have undergone a very considerable degree of scission of its chain groups, generating a new broad band of C=C at 1627 cm⁻¹, as a result of the γ -ray exposure particularly for that dose. This will be the subject of a separate study. The band assignments for most important chemical functional groups in a PADC polymer were given in the literature and are enlisted in Table 1.

The relative changes in the bonds IR absorption intensities have been evaluated from the relative increase or decrease in the intensity of the peak associated to the functional groups present in the PADC high polymer. IR absorption bands as obtained from the pristine and the irradiated samples of PADC polymer for different doses ranging from 0.5 to 3 MGy. It is apparent that the absorption bands observed at 1734, 1400, 1240, 1132, 1076, 1022, 959, 878 and 785 cm⁻¹ decrease in height due the γ -ray exposure. It is clear that the absorbance

decreases after irradiation for all the peaks except for the triplet bands at 2960, 2925 and 2853 cm^{-1} which show an opposite trend. However, there is a little shift in the peak positions of some absorption bands after irradiation with high doses of γ -rays. In the next paragraphs, we will pay attention to the two strong bands of the carbonate ester bond and the three weak absorption peaks of the ether bond.

3.2. Loss of chemical bonds in a PADC film

3.2.1. Loss of carbonate ester bonds

The normalized absorbance, A/A₀, was attained for each examined absorbed dose, where A is the absorbance of the band considered after exposure and A₀ is that of the pristine one. The normalized absorbance is equal to the survival fraction, N/N₀, which is the ratio of the number density of the considering chemical bond, N, to that of pristine, N₀. That is, A/A₀ = N/N₀. Fig. 4 shows the changes in the normalized absorbance, A/A₀, of the two peaks for carbonate ester bonds as a function of γ -ray dose over the range 1.5 to 3 MGy (just to focusing on the higher doses). Each peak reduced in height monotonically in the examined regions of dose. This figure shows the results for both C=O and C-O-C bonds, presenting a clear linear decline of the normalized absorbance against the absorbed dose for both bonds up to 3 MGy. Trends for the two peaks that compose the carbonate ester bond are almost similar. We used the following simple formula (Mori et al., 2009):

$$N/N_0 = 1 - k_i \cdot D$$
 (2)

where D is the absorbed dose in a unit of Gy and k_i is an experimental constant for the bond i in Gy⁻¹. The constant k_i corresponds to the slope of the fitting lines in Fig. 4. Normalized absorbances are converted to G values for the loss of C=O and C-O-C bonds per 100 eV, using the following simple relation (Mori et al., 2009):

$$G = 9.64 \times 10^6 \,\mathrm{N_0 \,k_i} \tag{3}$$

where N_0 is in mol/kg. N_0 is 7.30 mol/kg for the carbonate ester bond in PADC. The calculations give G values of 4.7 and 6.3 for C=O and C-O-C bonds, respectively. Finally, the average result, G value for the loss of carbonate ester bonds in PADC is determined to be about 5.5 as tabulated and listed in Table 2.

In the current study, the G value for gamma ray is considerably lower than that obtained in a previous study (Mori et al., 2009). With PADC as a target and ⁶⁰Co as a gamma source in both studies; the dose rate used by Mori et al. was 4.5 kGy/h (1.25 Gy/s) and we used 1.2 kGy/h (0.33 Gy/s), in atmospheric conditions within dose range 0.5 to 3 MGy. An important decrease of the G-value from 20 (Mori et al., (Mori et al., 2009) to about 6 (this work) in conjunction with a decrease of the dose rate (by a factor 3.75) is observed. The decrease of dose rate lowers thus radiation chemical yields for destruction of C=O and C-O-C bonds. The so-called dose rate effects (DRE) in polymer properties under irradiation in an oxygen environment are known to be very significant and connected with chain reactions of oxidizing radiation destruction initiated by radiolysis species of radical type (Briskman et al., 2004). DRE yields either from physical or chemical aspects. At high dose rates diffusion limited oxidation (DLO) effects become evident cause for dose rate effect and at low dose rates more complex chemical reactions determine the magnitude of dose rate effect (Sipilä, 2016). Generally, high dose rates of ionizing radiation (typically > 100 kGy/h) have less impact on the properties of polymers compared to lower dose rates (this work 1-5 kGy/h). It was for example shown that at equal total doses the carbonyls and hydroperoxide yields in LDPE (Low density polyethylene) are greatly enhanced when decreasing the dose rate from 0.69 to 0.04 kGy/h (Buttafava et al., 2007). Herein, dose rates (namely 4.5 kGy/h and 1.2 kGy/h) belong to the low or intermediate dose rates. DRE and thus oxidizing radiation destruction generally follows three steps in polymers, namely: a free radical reacts with

molecular oxygen to generate peroxy radicals which themselves abstract hydrogen, then the formed peroxide decomposes to create carbonyl functions. It is clear from Fig.2 that no carbonyl function is created, therefore conventional DRE in an oxygen environment may not be invoked here as a major pathway. Moreover, a recent study of the radiation degradation of Poly (ethylene-co-vinyl acetate) (EVA), containing a carbonyl moiety, demonstrated that the DRE is negligible in the range 0.4 - 1.5 kGy/h (Fabiani et al., 2015). Additionally, it may be quite unlikely to invoke the commercial origin of the polymers used as the cause of observed variations of G-values. BaryotrackTM used by Mori et al. and TastrakTM (this work) are both PADCs, thus with identical structures but with small differences in their initiator compositions (< 5%).

Coming back to the dose rates and especially to low or intermediate dose rates, it has been shown that in the case of aliphatic compounds; when irradiation is carried out at low dose rates in air, the effects are strongly dependent on the dose rate (Wilski, 1990). It is indeed generally assumed that the rate of radicals generation is proportional to the dose rate (Sarcinelli et al., 1997), hence with a dose rate ratio of 3.75 between Mori et al. experiments and the present study it is quite probable that the lower G-value we obtained (~8 instead of 20 for Mori et al.) is a clear manifestation of dose rate dependence.

3.2.2. Loss of ether bonds

Fig. 5 shows the changes in the normalized absorbance, A/A_0 , for ether bonds as a function of the γ -ray dose. The first two peaks of ether group (peaks number: 6 and 7) at 1132 and 1076 cm⁻¹ have more or less the same trend on the γ -ray dose up to 3 MGy, showing almost a linear decrease that can be expressed by Eq. (2). Beyond 2 MGy, the absorbance of the first peak at 1132 cm⁻¹ completely disappear up to 3 MGy, while the absorbance of the second peak at 1076 cm⁻¹ (peak number: 7) is still observed till the end of the dose range as clearly seen in Fig 2. The third peak of that group at 1022 cm⁻¹ (peak number: 8) behaves

differently when compared to the two other peaks. It is difficult to assess the absorbance for that peak. So, we assess the constant k_i for the ether bonds from peaks with wave numbers 1132 to 1076 cm⁻¹ (peaks number: 6 and 7). Eq. (3) gives G-values for the loss of ether bonds to be 7.3 and 6.2 at absorption bands of 1132 and 1076 cm⁻¹, respectively. The average G value for the loss of the ether bonds in PADC is determined to be about 6.9 (scission/100 eV) as seen in Table 2. Based on the loss measurements for both carbonate ester and ether bonds, it is found that the rate of reduction in the normalized absorbance of ether bonds is barely higher than that of carbonate ester bonds. On the other hand, the peak No. 6 (one of ether groups at 1132 cm⁻¹) has completely disappeared after 2 MGy. This means that it has suffered complete radiation destruction, so its G-value cannot be determined at the higher doses used.

As noticed above, the molecular bonds of ether functions in the PADC polymer have suffered a serious degradation due to γ -irradiation, at different high doses over the range 0.5 to 3 MGy. Overall, the radiation molecular yield, G value, has a dependence on the gamma dose in PADC. At high doses over the range 0.5 to 3 MGy, the yield for the loss of ether bonds is barely higher than that of carbonate ester bonds, confirming that the ether part being generally considered more radio-sensitive than the carbonate ester one (Mori et al., 2011; Mori et al., 2013). Here again, the G-value we obtain is smaller than the one measured by Mori et al., 2013. G-values obtained herein both for ether and carbonate ester function loss are relatively close together (average values of respectively 6.9 and 5.5) as those obtained by Mori et al. are quasi the triple (typically 20 for both chemical functions) (Mori et al., 2011; Mori et al., 2013; Kusumoto et al., 2016). So, we may argue that such difference militates in favor of a dose rate effect. We can observe that when the dose rate of 60 Co gamma rays decreases by a factor 3.75, exposing PADC sheets in air leads to a reduction of radiation chemical yields of a factor 3 – 3.6 for ether and carbonate ester function loss.

4. Discussion and conclusion

Molecular modification due to γ -ray damage in the PADC film has been examined by means of FT-IR spectrometry, which was exposed to γ -radiation at incident energies 1.25 MeV with different doses ranging from 0.5 to 3 MGy. The radiation molecular yield, i.e., G values, have been determined to be 5.5 (scission per 100 eV) for the loss of the carbonate ester bonds in PADC film exposed to gamma ray. Similar analyses have been performed for the ether bonds at the same doses. The G-value for the loss of these bonds was about 6.9. In this study, dose and dose rate dependence was observed, for both ether and carbonate ester function loss. It is quite interesting to note that the dose rate effect observed corresponds to a reduction of the dose of 3.75 in comparison with the study of others (Mori et al., 2009) and that this reduction yields a decrease of experimentally determined G-values by typically 3 – 3.6 folds.

The results presented herein demonstrate for the first time the existence of a dose rate effect in PADC exposed to Co-60 γ -radiation. If this last remark does not constitute in itself a major advance (dose rate effects in polymers are well-known); it indicates whatever that dose rate may be an interesting tool to be used in further studies. In spite of Co-60 gamma ray dose is clearly affects the G-values, variations of dose rates could also be an interesting way to adjust G-values in order to simulate heavy-ion irradiation effects using gamma rays. Actually, the radiation chemical yield for the loss of carbonate ester bonds obtained in the present study is close to those measured by Yamauchi et al., 2008 for 147 MeV Fe ions. The used gamma rays (Co-60) are characterized by an electronic LET of 0.4 keV/ μ m as the Fe ions we have

5200 keV/ μ m. The G-value we measured for the loss of ether moiety is close to the one measured for 46 MeV Ar ions (LET = 3880 keV/ μ m) (Yamauchi et al., 2008).

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Table 1
Assignments of FTIR bands of functional groups in PADC polymer film.

Peak Number	Band position (cm ⁻¹)	Functional Groups	Reference	
1 2926		Symmetric CH ₂ stretching vibration of aliphatic hydrocarbons	(Gagnadre et al., 1993; Saad et al., 2001; Yamauchi et al., 2001)	
2	1734	Ester C=O stretching vibration of the alkyl carbonate -C-CO-C	(Gagnadre et al., 1993; Saad et al., 2001; Yamauchi et al., 2001)	
2*	1627	Hydrocarbon C=C stretching of the vinyl group -CH=CH ₂	Present work	
3	1452	CH ₂ deformation vibration and OCH ₂ wagging vibration of –CO–O–CH ₂	(Lounis-Mokrani et al., 2003)	
4	1400	In–plane –CH ₂ – bending vibration of–CO–O–CH ₂	(Lounis-Mokrani et al., 2003)	
5	1240	Asymmetric stretching vibration of C–O–C of aliphatic ester	(Yamauchi et al., 2001)	
6	1132	Symmetric stretching vibration of the C–O–C group of the saturated aliphatic ether	(Gagnadre et al., 1993; Darraud et al., 1994; Lounis-Mokrani et al., 2003)	
7	1076	Asymmetric stretching vibration of the C–O–C group of the saturated aliphatic ether	(Darraud et al., 1994; Lounis- Mokrani et al., 2003)	
8	1022	Symmetric stretching vibration of the C–O–C group of the saturated aliphatic ether	(Darraud et al., 1994; Lounis- Mokrani et al., 2003)	
9	959	Wagging vibration of the double bond C=C	(Lounis-Mokrani et al., 2003)	
10	878	C–H rocking out-of-plane deformation	(Lounis-Mokrani et al., 2003)	
11	785	Ether group or C–H rocking out-of- plane deformation	(Saad et al., 2001)	

Table 2

Radiation chemical yields for loss of carbonate ester and ether bonds in a PADC polymer film.

Bonds			γ- radiation	
			G-value	Average G-value
Carbonate ester	C-O-C C=O	1240 cm ⁻¹ 1734 cm ⁻¹	6.3 4.7	5.5 ≈ 6
Ether	C-O-C	1132 cm ⁻¹ 1076 cm ⁻¹	7.5 6.2	6.9 ≈ 7

$$CH_{2} - CH_{2} - O - C - O - CH_{2} - CH = CH_{2}$$

$$O$$

$$CH_{2} - CH_{2} - O - C - O - CH_{2} - CH = CH_{2}$$

$$O$$

$$Fig. 1.$$

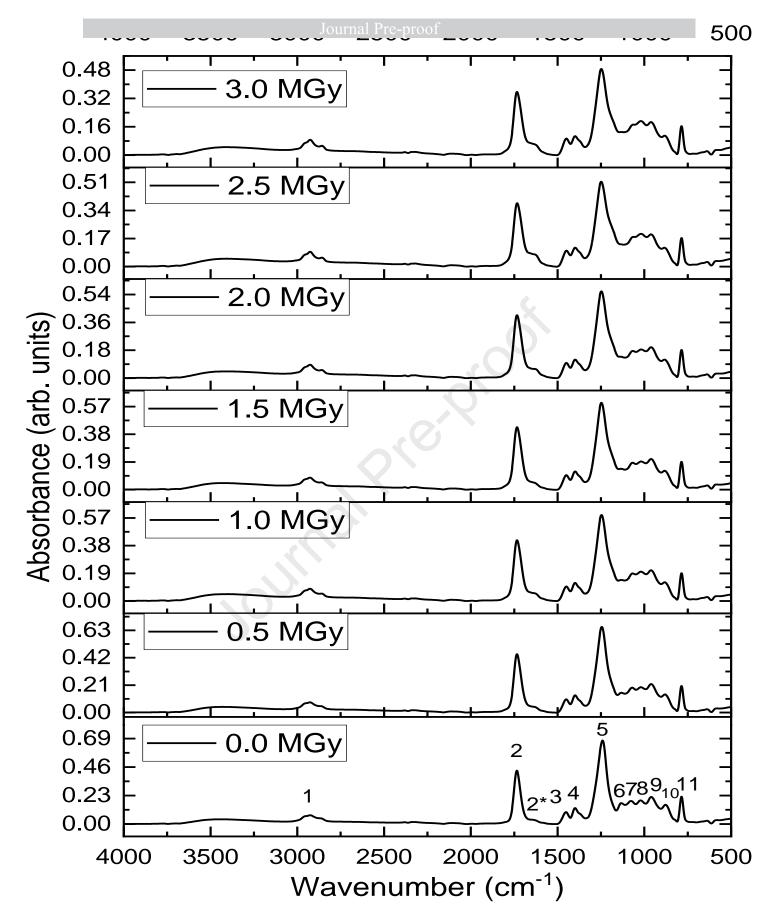


Fig. 2

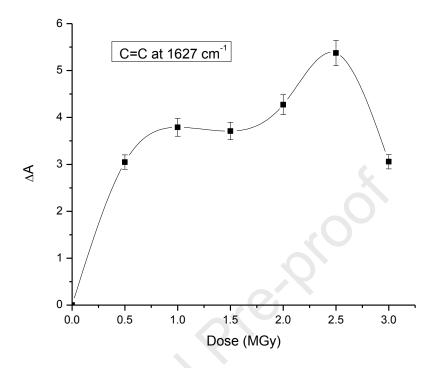


Fig. 3

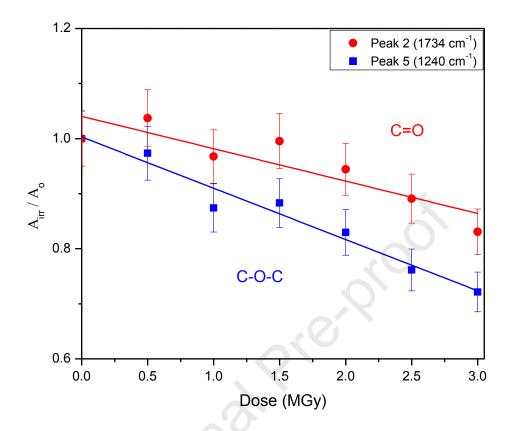
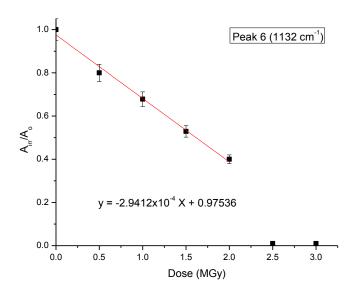
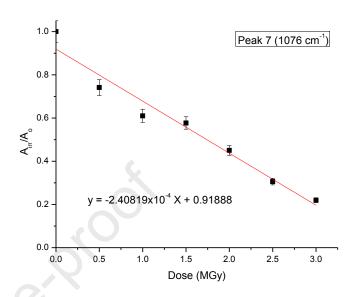
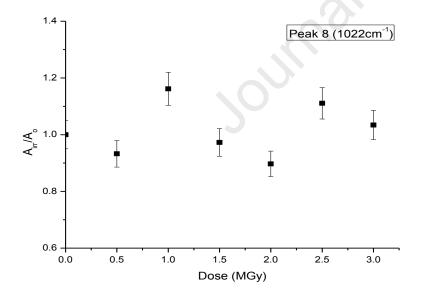


Fig. 4

Fig. 5.







Highlights

- Gamma irradiation has been proposed to study PADC polymer film.
- Molecular properties of pristine and irradiated PADC detectors were investigated.
- Radiation chemical yields, G values, for loss of bonds in PADC have been assessed.
- G values for the loss of carbonate ester and ether bonds are 5.5 and 6.9, respect.
- Radiation chemical yields are influenced by dose and dose rate of gamma rays.

Conflict of Interest Forms

No conflict



Ill[-Declaration of interests

oxtimes The authors declare that they have no known competing f that could have appeared to influence the work reported in the	·
☐The authors declare the following financial interests/person as potential competing interests:	nal relationships which may be considered
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