

### A "LAB-ON-A-CHIP" EXPERIMENT FOR ASSESSING MINERAL PRECIPITATION PROCESSES IN FRACTURED POROUS MEDIA

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### **MOTIVATION**

• The understanding of chemical changes in fractured porous media is relevant for various subsurface applications: subsurface nuclear waste disposal, fracking, CO<sub>2</sub> sequestration, etc.



e.g. simulated oxygenated melt water in fractured crystalline porous media in Forsmark site (Sweden) to assess the redox buffering capacity of these host rock formations <u>Trichinero et al. 2018, DOI:10.1007/s11004-017-9718-6</u>

- Reactive Transport Modelling can predict chemical reactions along solute transport pathways in space & time
- Challenge: Description of the coupling between chemical processes and changes in material properties. So far, only empirical relationships and parameters are available and used (e.g. evolution of the reactive surface area of minerals)
- Recent experimental benchmarks have suggested the need to understand the processes occurring at the pore-scale and to develop process-based predictive models and mathematical relationships that account for smallscale heterogeneities

### MICROFLUIDIC EXPERIMENTS



#### Advantages:

- idealized porous media with controlled pore shape/distribution and mineralogy
- feasibility of constant monitoring
- control on chemical system and environmental parameters (e.g. T)
- coupling with Raman Spectroscopy gives information on the mineralogical changes



live monitoring of chemical changes

→ develop an experiment combined with advance pore scale modelling to understand the physics of precipitation/dissolution processes in fractured porous media



### **EXPERIMENTAL SET-UP**

#### Microfluidic combined with raman spectroscopy



• injection of a reacting solution of BaCl<sub>2</sub> at a constant flowrate in the microreactor

 $BaCl_{2(aq)} + SrSO_{4(s)} \rightarrow BaSO_{4(s)} + SrCl_{2(aq)}$ 



 chip manufactured by conventional pdms techniques and closed with glass



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### **COMPACTED POROUS MEDIA**

Porous media generated with this technique can be either homogeneous or heterogeneous

Homogeneous porous media
before reaction
depting area A1
area A1
area A2

Heterogeneous(fractured) porous media
before reaction



- Raman map of selected areas A1, A2, B1 and B2 were collected at regular interval of times
- Raman depth scan and a raman tomograph were collected before and after chemical reactions



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### **MINEROLOGICAL CHANGES**

#### Raman map of the homogeneous porous media after reaction



- Barite precipitated
- along 50µm in the upstream section
- in between the celesine crystals downstream

 A considerable amount of celesine is still present in the system



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### **MINEROLOGICAL CHANGES**

#### Raman map of the heterogeneous porous media after reaction



Area B1



- Barite precipitated mainly across the main flow path
- Barite nucleation appears to start on the step like features of the individual celestine crystals (epitaxial growth)
- Celestine is present in the left side of the reservoir
- Crystals directly adjacent to the fracture undergo dissolution



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### **RAMAN TOMOGRAPHY**

#### 3D representation of the porous media before and after reaction



> 3D tomograph enables the evaluation of porosity, mineral mole fraction, and dissolution rates



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### **RAMAN TOMOGRAPHY**

#### **Evaluation of the mineral amounts in the systems**

Homogeneous case: Porosity increase suggesting that dissoluton is faster than precipitation

Heterogeneous case: Porosity decrease. Complete conversion of celestine to barite

The amount of barite that precipitated in the heterogenous compacted porous media is more than that that precipitated in the homogeneous porous media.

How to explain the differences in minerological changes?



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### **EVALUATION OF VELOCITY FIELDS**

#### **Simulated velocities**

Homogeneous case: velocity in the compacted regions: 9.25×10<sup>-4</sup> ms<sup>-1</sup> Pe numbers 4.65→advection dominates

Heterogeneous case: velocity along the fractures evaluated  $3.5 \times 10^{-3}$ velocity in the compacted regions:  $4 \times 10^{-5}$  ms<sup>-1</sup> Pe number compacted region  $0.25 \rightarrow$  diffusion dominates

Velocity in compacted homogeneous porous media is 19 times higher than in the compacted fractured media





## HOW TO EXPLAIN THE DIFFERENCES IN MINEROLOGICAL CHANGES?

#### Homogeneous case



- High velocities translate to high Pe numbers, and short residence time
- Incoming solution is undersaturated w.r.t to celestine that drive its dissolution
- Insufficient time for barite to nucleate on the surface of celestine
- Dissolution is faster than precipitation→ porosity increase

#### Heterogenous case





- In the compacted regions
- Low velocities translate to low Pe numbers, and longer residence time
- Ample time for barite to nucleate and precipitate as Sr<sup>2+</sup> is released into the pore solution during celestine dissolution
- Along the fracture, velocity is higher, shorter residence time and consequently not enough time for barite to nucleate
- The velocity field has an impact on the residence time and consequently on the precipitation



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### **CONCLUSIONS AND OUTLOOK**

- Establishment of a new flow-through microfluidic reactor coupled with high-resolution imaging that enables in situ and non destructive 3D assessment of mineralogical and microstructural changes with full spatiotemporal resolution on the grain scale
- This methodology allows a systematic study of the chemical and porosity evolution of the system at the pore-scale
- Future work:

The in-situ 3D Raman reconstruction of the pore architecture along with advanced pore-scale modelling will provide a better assessment of the transport properties (e.g. permeability, dispersion tensor) as well as the accessible reactive surface area and reaction rates for describing mineral precipitation/dissolution in porous media.



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