

# Synthesis of ribavirin 2'-Me-C-nucleoside analogues

Jacques Uziel, Fanny Cosson, Aline Faroux, Jean-Pierre Baltaze, Jonathan Farjon, Régis Guillot, Nadège Lubin-Germain

## ▶ To cite this version:

Jacques Uziel, Fanny Cosson, Aline Faroux, Jean-Pierre Baltaze, Jonathan Farjon, et al.. Synthesis of ribavirin 2'-Me-C-nucleoside analogues. Beilstein Journal of Organic Chemistry, 2017, 13, pp.755 - 761. 10.3762/bjoc.13.74 . hal-03553300

# HAL Id: hal-03553300 https://hal.science/hal-03553300

Submitted on 2 Feb 2022

**HAL** is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers. L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.



## Synthesis of ribavirin 2'-Me-C-nucleoside analogues

Fanny Cosson<sup>1</sup>, Aline Faroux<sup>1</sup>, Jean-Pierre Baltaze<sup>2</sup>, Jonathan Farjon<sup>3</sup>, Régis Guillot<sup>2</sup>, Jacques Uziel<sup>\*1</sup> and Nadège Lubin-Germain<sup>\*1</sup>

Full Research Paper		Open Access
Address: <sup>1</sup> Laboratoire de Chimie Biologique, University of Cergy-Pontoise, 5 mail Gay-Lussac, Cergy-Pontoise, France, <sup>2</sup> ICMMO, UMR CNRS 8182, University of Paris Sud, 15 rue Georges Clémenceau, Orsay, France and <sup>3</sup> Laboratoire CEISAM, UMR 6230, University of Nantes, 2 rue de la Houssinière, Nantes, France	<i>Beilstein J. Org. Chem.</i> <b>2017,</b> <i>13</i> , 755–761. doi:10.3762/bjoc.13.74 Received: 06 January 2017 Accepted: 04 April 2017 Published: 21 April 2017	
Email: Jacques Uziel <sup>*</sup> - jacques.uziel@u-cergy.fr; Nadège Lubin-Germain <sup>*</sup> - nadege.lubin-germain@u-cergy.fr * Corresponding author Keywords: alkynylation; antiviral; cancer; C-nucleosides; ribavirin	Associate Editor: J. Aubé © 2017 Cosson et al.; licensee Beilstein-Institut. License and terms: see end of document.	

## Abstract

An efficient synthetic pathway leading to two carbonated analogues of ribavirin is described. The key-steps in the synthesis of these ribosyltriazoles bearing a quaternary carbon atom in the 2'-position are an indium-mediated alkynylation and a 1,3-dipolar cyclization.

### Introduction

The triazole nucleoside ribavirin (RBV, Figure 1) is used for the treatment of a number of viral infections and may be promising as an anticancer drug [1-3]. The antiviral activity of ribavirin is ascribed to a combination of different mechanisms [4]. Although RBV causes some side effects [5-7] essentially due to its

accumulation in red blood cells, it is indispensable in the treatment against hepatitis C virus (HCV). The current standard-ofcare for hepatitis C involves taking a combination [8] of an antipolymerase compound (sofosbuvir [9,10]) and an antiprotease compound (simeprevir [11,12]), both associated to



ribavirin. If the presence of interferon is not required for the therapy, ribavirin is mandatory in the combination, due to its particular role.

Recently, we developed an alkynyl glycosylation protocol allowing us to obtain *C*-nucleoside derivatives and we turned our attention to ribavirin *C*-nucleoside analogues. Moreover, recently De Clerq [13] outlined the potential of *C*-nucleosides in the arsenal of antivirals due to their stability in biological fluids and their bioavailability.

Some years ago, we described an approach leading to the *C*-ribosylated analogue **1** of ribavirin (Figure 2) with the keysteps of the synthesis being an indium-mediated alkynylation of a ribose derivative followed by the Huisgen cycloaddition reaction onto the *C*-alkynyl riboside intermediately obtained [14].

Herein we describe the synthesis of two new carbonated analogues 2 and 3 of RBV modified at the 2'-position (Figure 1). In fact, a quaternization in 2'-position of various nucleosides led to a higher efficacy against HCV as in the case of 2'-C-methylcytidine and 2'-C-methyladenosine [15]. The structure-activity studies of 2'-C-methylnucleosides showed that the methyl substituent must be in 2'-position and on the  $\beta$  face for an optimal efficacy that drops when the methyl is on the  $\alpha$  face or in the 3'-position or if a bulkier ethyl group is used [16]. On the other hand, currently 2'-deoxy-2'-C-methyl-2'-C-fluoronucleosides are developed because a fluoro substituent in the 2'-position increases the antiviral activity and specificity due to a higher tolerance of viral polymerases with respect to incorporation of such compounds [17]. In clinical studies (phase I and II), the fluorinated compound mericitabine in combination with PEG-IFN and RBV was better tolerated and more effective in genotype 1 or 4 patients compared to the standard combination of Peg-IFN and RBV [18,19].

Further, the therapy of untreated patients with HCV genotype 1, 2, or 3 infections with a combination of sofosbuvir (Gilead) and ribavirin for 12 weeks is considered as the most effective treatment at the moment [20].

#### **Results and Discussion**

2'-C-Methylnucleoside **2** was synthesized according to a seven step pathway starting from the commercially available 2-*C*methyl-1,2,3,5-tetra-*O*-benzoyl- $\beta$ -D-ribofuranose (**4**) as shown in Scheme 1. Debenzoylation of **4** followed by selective protection led to derivative **5**, which was submitted to the indium-mediated alkynylation reaction affording the alkynyl riboside **6** with the same  $\beta$ -anomeric selectivity as for the non-methylated derivative [21]. Then, the 1,3-dipolar cycloaddition reaction of **6** with benzyl azide in toluene at 70 °C led to a mixture of regioisomeric triazoles **7** in a 42:58 ratio. The removal of all protecting groups was achieved by treatment of compounds **7** with ammonia followed by catalytic hydrogenolysis. The latter reaction simultaneously cleaves the benzyl and isopropylidene groups affording compound **2** as a single isomer [22].

In the case of 5-(2'-deoxy-2'-methyl-2'-fluoro- $\beta$ -D-ribosyl)-1,2,3-triazole-4-carboxamide (**3**) the synthesis was more delicate as it is necessary to differentiate the 2' position.

After the indium-mediated alkynylation, the obtained alkynyl riboside **8** was submitted to a Huisgen cycloaddition reaction with benzyl azide, under the same conditions as in the previous case, affording the mixture of regioisomeric triazoles **9a** and **9b** in a 37:63 ratio [14] (Scheme 2).

The selective protection of the 3' and 5'-positions requires the full deprotection of the ribose. This was performed by the treatment with Dowex 50Wx8 (H<sup>+</sup>) in methanol. However, carrying out this step with the mixture of **9a/9b** the reaction led to only a moderate yield (52%) and with formation of a partially deprotected compound (only acetal deprotected), stemming from **9b**, demonstrating the different hydrolysis rate of each regioisomer. An HPLC analysis of the disappearance of **9a** and **9b** and the formation of **10a** and **10b** showed a ratio of 2:1 in favor of regioisomer **10a**. The hydrolysis was more efficient when performed with the isolated **9a** or **9b** isomers. In this case, the fully deprotected compounds **10a** and **10b** were obtained in 89% (after 36 h) and 61% (after 2 weeks) yields, respectively.





Scheme 1: Synthesis of 5-(2'-C-methyl-β-D-ribofuranosyl)-1,2,3-triazole-4-carboxamide (2).



This rate difference was also observed in the subsequent protection of the 3',5' positions with 1,3-dichloro-1,1,3,3-tetraisopropyldisiloxane (TIPDSCl<sub>2</sub>). While this reaction proceeded with a very poor yield in the case of the mixture 10a/10b (~10% yield), compound 11a was obtained in 79% yield from pure 10a (24% yield for 11b starting from 10b). Thereafter, the oxida-

tion of **11a** to the corresponding ketone with Dess–Martin periodinane afforded **12a** in 87% yield whereas the reaction of the less reactive isomer **11b** led to **12b** in 44% yield.

With the aim to get some explanations for the different reactivities observed for the two isomers **10a** and **10b**, we investigated the structure of the less reactive compound **10b**. As depicted in Figure 3, compound **10b** displayed an S-type conformation and an *anti* arrangement of the atoms  $O(1^{\circ})-C(1^{\circ})-C(1)-N(2)$  according to the dihedral angle of 0° lower than 90° (43°). Moreover, the benzyl group appeared to be present in two different positions covering the furan ring.



For the attempted methylation of the 2'-position different conditions were tested (Table 1). As described for other nucleosides [23], the use of MeLi led to compound **13b** obtained by an attack from the  $\alpha$ -face even if this proceeded with a very low yield (7%). The use of MeMgBr gave an almost 1:1 mixture by  $\alpha$ - and  $\beta$ -attack; this second one can be explained by a magnesium complexation with the base. More interestingly, the methylation proceeded stereoselectively leading to **13b** in 87% yield when trimethylaluminium was used [24].

The stereochemical outcome of this reaction was determined by selective 1D NOESY experiments (Figure 4). First, the hydrogen H3' in **13a** and **13b** was selectively excited. The nOes observed for compound **13a** are in the following order of decreasing intensity:  $CH_3 > H5'_a > H5'_b-H4' > H1'$  confirming that  $CH_3$  and H3' are spatially close. In the case of **13b** the nOe intensities decrease in the order:  $H5'_b-H4' > CH_3 > H1'$ . The selective excitation of H1' in **13a** led to nOes with decreasing intensities in the order:  $H5'_b-H4' > CH_3 > H3'$ , whereas for **13b** the order was  $CH_3 > H5'_b-H4'$ . This second series of nOes confirms that  $CH_3$  is closer to H1' in **13b** than in **13a**.



The fluorination of 13b with DAST was performed at -20 °C and afforded the desired fluorinated derivative 14 (24%) along



with two elimination products, the exocyclic olefin **15** (16%) and the corresponding endocyclic one **16** (20%). As it was impossible to separate **15** from **14** at this stage, the mixture **14/15** was deprotected with tetrabutylammonium fluoride leading to the mixture of diols **17** and **18** in quantitative yield which were easily separated (Scheme 3).

Finally, the aminolysis of compound **17** followed by catalytic hydrogenolysis in the presence of palladium chloride led to the desired compound **3** [22] (Scheme 4).

#### Conclusion

The indium-mediated alkynylation of a ribose derivative followed by a Huisgen cyclization allowed the access to 2'-quaternized carbonated analogues of ribavirin. While the synthesis of compound **2** starting from 2-methylated ribose derivative **4** was quite easy to perform, the preparation of the fluorinated analogue **3** required a more complicated pathway. This included the selective protection of the 3',5' positions, the stereoselective methylation and the fluorination of the 2' position. The synthesized compounds are currently investigated for their antiviral activities.

#### Experimental

Experimental procedures for compounds **2**, **5–7**, and **10–18** are covered by the Ph.D. thesis of Fanny Cosson [22].

**5-(2'-C-Methyl-β-D-ribofuranosyl)-1,2,3-triazole-4-carboxamide (2)** [22]: Through the solution of compound 7 (mixture of 7a and 7b, 0.49 g, 1.1 mmol) in anhydrous methanol (14 mL) was bubbled ammonia gas for 2 h at 0 °C. Then the mixture was stirred for 12 h at rt and concentrated in vacuum. The residue



 $\label{eq:scheme 3: Fluorination of ethyl 1-benzyl-4-(2'-C-methyl-3',5'-O-(tetraisopropyldisiloxane-1,3-diyl)-\beta-D-ribofuranosyl)-1,2,3-triazole-5-carboxylate (13b).$ 



was purified by flash chromatography (EtOAc/cyclohexane, 3:7 to 1:0) affording the mixture of the corresponding 1-benzyl-4-(2',3'-O-isopropylidene-2'-C-methyl-β-D-ribofuranosyl)-1,2,3triazole-5-carboxamide and 1-benzyl-5-(2',3'-O-isopropylidene-2'-C-methyl-β-D-ribofuranosyl)-1,2,3-triazole-4-carboxamide (0.25 g, 60%) in a 27:73 ratio as an oil. IR (cm<sup>-1</sup>)  $v_{max}$ : 3336, 2360, 2342, 1635; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.32–7.25 (m, 5H, Ph), 6.05 (d, J = 14.6 Hz, 1H, CH<sub>2</sub>Ph), 5.59 (d, J = 14.6 Hz, 1H, CH<sub>2</sub>Ph), 5.37 (s, 1H, H1), 4.54 (d, J = 5.0 Hz, 1H, H3), 4.35-4.32 (m, 1H, H4), 3.82-3.75 (m, 2H, H5, H5'), 1.60 (s, 3H, CH<sub>3</sub>), 1.37 (s, 3H, C(CH<sub>3</sub>)<sub>2</sub>), 1.34 (s, 3H,  $C(CH_3)_2$ ) and  $\delta$  7.32-7.25 (m, 5H, Ph), 6.12 (d, J = 14 Hz, 1H, CH<sub>2</sub>Ph), 5.80 (d, *J* = 14 Hz, 1H, CH<sub>2</sub>Ph), 5.33 (s, 1H, H1), 4.84–4.81 (m, 1H, H4), 4.45 (d, J = 2.7 Hz, 1H, H3), 3.91 (dd, J = 3.7 Hz, 11.9 Hz, 1H, H5), 3.82-3.75 (m, 1H, H5'), 1.64 (s, 3H, CH<sub>3</sub>), 1.24 (s, 3H, C(CH<sub>3</sub>)<sub>2</sub>), 1.23 (s, 3H, C(CH<sub>3</sub>)<sub>2</sub>); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 160.3 (CONH<sub>2</sub>), 149.7 (triazole), 129.6 (Cq Ph), 129–128 (Ph), 125.0 (triazole), 112.9 (C(CH<sub>3</sub>)<sub>2</sub>), 90.0 (C2), 87.1 (C3), 85.1 (C4), 84.1 (C1), 62.0 (C5), 53.8 (CH<sub>2</sub>Ph), 26.6 (C(CH<sub>3</sub>)<sub>2</sub>), 26.5 (C(CH<sub>3</sub>)<sub>2</sub>), 14.3 (CH<sub>3</sub>) and δ 159.1 (CONH<sub>2</sub>), 141.7 (triazole), 135.6 (Cq Ph), 135.1 (triazole), 129-128 (Ph), 114.5 (C(CH<sub>3</sub>)<sub>2</sub>), 90.7 (C2), 87.8 (C3), 84.8 (C4), 82.4 (C1), 62.8 (C5), 54.0 (CH<sub>2</sub>Ph), 26.0 (C(CH<sub>3</sub>)<sub>2</sub>), 25.3 (C(CH<sub>3</sub>)<sub>2</sub>); HRMS calcd for C<sub>19</sub>H<sub>22</sub>O<sub>5</sub>, 330.1467; found, 330.1478.

The mixture of carboxamides and palladium chloride (23 mg, 0.13 mmol) in ethanol (10 mL) was hydrogenated with H<sub>2</sub> at 4 bar for 48 h. After filtration over Celite and concentration in vacuum, the crude was purified by flash chromatography (cyclohexane/EtOAc 2:8 to EtOAc/MeOH 1:1) to afford compound **2** as a white solid (0.12 g, 75%) that was lyophilized.  $[\alpha]_D^{24}$  -47 (*c* 1, MeOH); mp 185–190 °C; IR (cm<sup>-1</sup>) v<sub>max</sub>: 3202, 1663, 1604, 1382, 1014; <sup>1</sup>H NMR (400 MHz, MeOD)  $\delta$  5.48 (s, 1H, H1), 4.11–4.04 (m, 1H, H4), 3.99 (d, *J* = 8.3 Hz, 1H, H3), 3.83 (dd, *J* = 2.7 Hz, 12.4 Hz, 1H, H5), 3.64 (dd, *J* = 3.6 Hz, 11.9 Hz, 1H, H5'), 1.22 ppm (s, 3H, CH<sub>3</sub>); <sup>13</sup>C NMR (100 MHz, MeOD)  $\delta$  164.1 (CONH<sub>2</sub>), 137.8 (triazole), 82.4 (C4), 78.5 (C1), 78.1 (C2), 76.2 (C3), 61.6 (C5), 19.9 (CH<sub>3</sub>); HRMS calcd for C<sub>9</sub>H<sub>15</sub>N<sub>4</sub>O<sub>5</sub>, 259.1042; found, 259.1055.

**5-(2'-Deoxy-2'-fluoro-2'-methyl-β-D-ribofuranosyl)-1,2,3triazole-4-carboxamide (3)** [22]: Through the solution of compound **17** (110 mg, 0.29 mmol) in anhydrous methanol (4 mL) was bubbled ammonia for 2 h at 0 °C. After stirring at rt overnight, the solution was concentrated in vacuum affording 1-benzyl-4-(2'-deoxy-2'-fluoro-2'-methyl-β-D-ribofuranosyl)-1,2,3-triazole-5-carboxamide (90 mg, 89%) as a white powder.  $[\alpha]_D^{25}$  –18.6 (*c* 0.9, MeOH); mp 176 °C; IR (cm<sup>-1</sup>) v<sub>max</sub>: 3397, 3304, 2944, 2883, 1668, 1598, 1458, 1218, 1126, 1048; <sup>1</sup>H NMR (400 MHz, MeOD) δ 7.25–7.18 (m, 5H, Ph), 5.85 (d, *J* = 14.6 Hz, 1H, CH<sub>2</sub>Ph), 5.73 (d, J = 14.6 Hz, 1H, CH<sub>2</sub>Ph), 5.42 (d, J = 24.8 Hz, 1H, H1), 4.14 (dd, J = 8.7 Hz, 18.8 Hz, 1H, H3), 3.97–3.93 (m, 1H, H4), 3.90 (dd, J = 2.8 Hz, 12.4 Hz, 1H, H5), 3.72 (dd, J = 5.0 Hz, 12.4 Hz, 1H, H5'), 1.18 (d, J = 22.5 Hz, 3H, CH<sub>3</sub>); <sup>13</sup>C NMR (100 MHz, MeOD)  $\delta$  160.8 (CONH<sub>2</sub>), 143.7 (d, J = 11.4 Hz, triazole), 135.3 (Cq Ph), 130.4 (triazole), 128.6, 128.2, 127.7 (Ph), 100.8 (d, J = 190 Hz, C2), 82.2 (C4), 79.7 (d, J = 39 Hz, C1), 73.9 (d, J = 23 Hz, C3), 61.3 (C5), 53.0 (-CH<sub>2</sub>Ph), 17.2 (d, J = 26 Hz, CH<sub>3</sub>); <sup>19</sup>F NMR (376.2 MHz, MeOD) –158.2 (m); HRMS calcd for C<sub>16</sub>H<sub>20</sub>FN<sub>4</sub>O<sub>4</sub>, 351.1469; found, 351.1468.

Debenzylation proceeded in ethanol (5 mL) with palladium chloride (9.1 mg, 0.05 mmol) under 4 bar pressure of hydrogen for 24 h. After filtration over Celite and concentration in vacuum, the crude was purified by flash chromatography (pure EtOAc to EtOAc/MeOH 1:1) to afford **3** as a white solid (0.03 g, 45%) that was lyophilized.  $[\alpha]_D^{25}$  +18.5 (*c* 0.7, MeOH); mp 52 °C; IR (cm<sup>-1</sup>) v<sub>max</sub> 3357, 2361, 2341, 1635, 1455, 1072, 1021; <sup>1</sup>H NMR (400 MHz, MeOD)  $\delta$  5.80 (d, *J* = 22 Hz, 1H, H1), 4.06–3.98 (m, 3H, H3, H4, H5), 3.84 (dd, *J* = 2.8 Hz, 11.9 Hz, 1H, H5'), 1.13 (d, *J* = 24.8 Hz, 3H, CH<sub>3</sub>); <sup>13</sup>C NMR (100 MHz, MeOD)  $\delta$  163.8 (-CONH<sub>2</sub>), 137.9 (triazole), 101.2 (d, *J* = 174 Hz, C2), 81.5 (C4), 78.5 (d, *J* = 39 Hz, C1), 72.6 (d, *J* = 23 Hz, C3), 60.6 (C5), 16.3 (d, *J* = 26 Hz, CH<sub>3</sub>); <sup>19</sup>F NMR (376.2 MHz, MeOD) –160.8 (m); HRMS calcd for C<sub>9</sub>H<sub>14</sub>FN<sub>4</sub>O<sub>4</sub>, 261.0999; found, 261.1006.

### Supporting Information

Supporting Information File 1

Experimental details, copies of <sup>1</sup>H NMR, <sup>13</sup>C NMR, HRMS and X-ray spectra. [http://www.beilstein-journals.org/bjoc/content/ supplementary/1860-5397-13-74-S1.pdf]

#### Acknowledgements

The authors gratefully acknowledge the generous support provided by the host institution, University of Cergy-Pontoise.

#### References

- Borden, K. L. B.; Culjkovic-Kraljacic, B. Leuk. Lymphoma 2010, 51, 1805–1815. doi:10.3109/10428194.2010.496506
- Zahreddine, H. A.; Culjkovic-Kraljacic, B.; Assouline, S.; Gendron, P.; Romeo, A. A.; Morris, S. J.; Cormack, G.; Jaquith, J. B.; Cerchietti, L.; Cocolakis, E.; Amri, A.; Bergeron, J.; Leber, B.; Becker, M. W.; Pei, S.; Jordan, C. T.; Miller, W. H.; Borden, K. L. B. *Nature* 2014, *511*, 90–93. doi:10.1038/nature13283
- Shelton, J.; Lu, X.; Hollenbaugh, J. A.; Cho, J. H.; Amblard, F.; Schinazi, R. F. *Chem. Rev.* 2016, *116*, 14379–14455. doi:10.1021/acs.chemrev.6b00209

- Paeshuyse, J.; Dallmeier, K.; Neyts, J. Curr. Opin. Virol. 2011, 1, 590–598. doi:10.1016/j.coviro.2011.10.030
- Manns, M. P.; Wedemeyer, H.; Cornberg, M. Gut 2006, 55, 1350–1359. doi:10.1136/gut.2005.076646
- Fried, M. W.; Shiffman, M. L.; Reddy, K. R.; Smith, C.; Marinos, G.; Gonçales, F. L., Jr.; Häussinger, D.; Diago, M.; Carosi, G.; Dhumeaux, D.; Craxi, A.; Lin, A.; Hoffman, J.; Yu, J. *N. Engl. J. Med.* 2002, *347*, 975–982. doi:10.1056/NEJMoa020047
- Manns, M. P.; McHutchison, J. G.; Gordon, S. C.; Rustgi, V. K.; Shiffman, M.; Reindollar, R.; Goodman, Z. D.; Koury, K.; Ling, M.-H.; Albrecht, J. K. *Lancet* 2001, *358*, 958–965. doi:10.1016/S0140-6736(01)06102-5
- Miller, M. H.; Agarwal, K.; Austin, A.; Brown, A.; Barclay, S. T.; Dundas, P.; Dusheiko, G. M.; Foster, G. R.; Fox, R.; Hayes, P. C.; Leen, C.; Millson, C.; Ryder, S. D.; Tait, J.; Ustianowski, A.; Dillon, J. F. *Aliment. Pharmacol. Ther.* **2014**, *39*, 1363–1375. doi:10.1111/apt.12764
- Sofia, M. J.; Bao, D.; Chang, W.; Du, J.; Nagarathnam, D.; Rachakonda, S.; Reddy, P. G.; Ross, B. S.; Wang, P.; Zhang, H.-R.; Bansal, S.; Espiritu, C.; Keilman, M.; Lam, A. M.; Micolochick Steuer, H. M.; Niu, C.; Otto, M. J.; Furman, P. A. *J. Med. Chem.* **2010**, *53*, 7202–7218. doi:10.1021/jm100863x
- Zeuzem, S.; Dusheiko, G. M.; Salupere, R.; Mangia, A.; Flisiak, R.; Hyland, R. H.; Illeperuma, A.; Svarovskaia, E.; Brainard, D. M.; Symonds, W. T.; Subramanian, G. M.; McHutchison, J. G.; Weiland, O.; Reesink, H. W.; Ferenci, P.; Hézode, C.; Esteban, R. *N. Engl. J. Med.* **2014**, *370*, 1993–2001. doi:10.1056/NEJMoa1316145
- Lin, T.-I.; Lenz, O.; Fanning, G.; Verbinnen, T.; Delouvroy, F.; Scholliers, A.; Vermeiren, K.; Rosenquist, Å.; Edlund, M.; Samuelsson, B.; Vrang, L.; de Kock, H.; Wigerinck, P.; Raboisson, P.; Simmen, K. Antimicrob. Agents Chemother. 2009, 53, 1377–1385. doi:10.1128/AAC.01058-08
- Reesink, H. W.; Fanning, G. C.; Farha, K. A.; Weegink, C.; Van Vliet, A.; van 't Klooster, G.; Lenz, O.; Aharchi, F.; Mariën, K.; Van Remoortere, P.; de Kock, H.; Broeckaert, F.; Meyvisch, P.; Van Beirendonck, E.; Simmen, K.; Verloes, R. *Gastroenterology* **2010**, *138*, 913–921. doi:10.1053/j.gastro.2009.10.033
- 13. De Clercq, E. J. Med. Chem. 2016, 59, 2301–2311. doi:10.1021/acs.jmedchem.5b01157
- 14. Aït Youcef, R.; Dos Santos, M.; Roussel, S.; Baltaze, J.-P.; Lubin-Germain, N.; Uziel, J. J. Org. Chem. 2009, 74, 4318–4323. doi:10.1021/jo900594x
- Stuyver, L. J.; McBrayer, T. R.; Tharnish, P. M.; Hassan, A. E. A.; Chu, C. K.; Pankiewicz, K. W.; Watanabe, K. A.; Schinazi, R. F.; Otto, M. J. *J. Virol.* **2003**, *77*, 10689–10694. doi:10.1128/JVI.77.19.10689-10694.2003
- 16. Carroll, S. S.; Olsen, D. B. *Infect. Disord.: Drug Targets* **2006**, *6*, 17–29. doi:10.2174/187152606776056698
- Stuyver, L. J.; McBrayer, T. R.; Whitaker, T.; Tharnish, P. M.; Ramesh, M.; Lostia, S.; Cartee, L.; Shi, J.; Hobbs, A.; Schinazi, R. F.; Watanabe, K. A.; Otto, M. J. *Antimicrob. Agents Chemother.* **2004**, *48*, 651–654. doi:10.1128/AAC.48.2.651-654.2004
- Pockros, P. J.; Jensen, D.; Tsai, N.; Taylor, R.; Ramji, A.; Cooper, C.; Dickson, R.; Tice, A.; Kulkarni, R.; Vierling, J. M.; Munson, M. L.; Chen, Y.-C.; Najera, I.; Thommes, J. *Hepatology* **2013**, *58*, 514–523. doi:10.1002/hep.26275
- Wedemeyer, H.; Jensen, D.; Herring, R., Jr..; Ferenci, P.; Ma, M.-M.; Zeuzem, S.; Rodriguez-Torres, M.; Bzowej, N.; Pockros, P.; Vierling, J.; Ipe, D.; Munson, M. L.; Chen, Y.-C.; Najera, I.; Thommes, J. *Hepatology* **2013**, *58*, 524–537. doi:10.1002/hep.26274

- Gane, E. J.; Stedman, C. A.; Hyland, R. H.; Ding, X.; Svarovskaia, E.; Symonds, W. T.; Hindes, R. G.; Berrey, M. M. N. Engl. J. Med. 2013, 368, 34–44. doi:10.1056/NEJMoa1208953
- Lubin-Germain, N.; Baltaze, J.-P.; Coste, A.; Hallonet, A.; Lauréano, H.; Legrave, G.; Uziel, J.; Augé, J. Org. Lett. 2008, 10, 725–728. doi:10.1021/ol7029257
- Cosson, F. Synthèse d'analogues carbonés de la Ribavirine pour leurs activités antivirals. Ph.D. Thesis, University of Cergy-Pontoise, France, 2014.
- Takenuki, K.; Itoh, H.; Matsuda, A.; Ueda, T. Chem. Pharm. Bull. 1990, 38, 2947–2952. doi:10.1248/cpb.38.2947
- 24. Li, N.-S.; Lu, J.; Piccirilli, J. A. J. Org. Chem. 2009, 74, 2227–2230. doi:10.1021/jo802666k

### License and Terms

This is an Open Access article under the terms of the Creative Commons Attribution License (http://creativecommons.org/licenses/by/4.0), which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

The license is subject to the *Beilstein Journal of Organic Chemistry* terms and conditions: (http://www.beilstein-journals.org/bjoc)

The definitive version of this article is the electronic one which can be found at: doi:10.3762/bjoc.13.74