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LATE PLEISTOCENE VARIABILITY OF THE CARBON ISOTOPIC COMPOSITION OF ORGANIC MATTER IN THE EASTERN MEDITERRANEAN: MONITOR OF CHANGES IN CARBON SOURCES AND ATMOSPHERIC CO₂ CONCENTRATIONS

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Abstract. The organic carbon isotopic record of the sapropels (S1 and S3-S10) and intercalated marl oozes has been determined in a 12-m piston core from the eastern Mediterranean. The $\delta^{13}C_{organic}$ values are systematically lighter (mean = $-21.0 \pm$ 0.82 %) in all sapropels and heavier (mean = -18.8 \pm 1.07 ‰) in the marl oozes. These differences are not due to variable marine and terrestrial organic matter mixtures because all values are heavier than modern plankton in the Mediterranean, there is no relationship between the Corganic/N ratios and the isotopic values, and published information on the abundance and distribution of organic biomarkers shows that terrestrial material constitutes a minor fraction of the total organic matter. Temperature effects on isotope fractionation are also discounted because the change in $\delta^{13}C_{\text{organic}}$ values between glacial and interglacial horizons is in the opposite sense. Diagenesis, which can produce relatively small changes in the carbon isotopic composition of sedimentary organic matter under certain circumstances, is unlikely to have caused the observed differences because this mechanism would cause an enrichment in ¹²C, implying that all values were even heavier originally, and there is no secular trend in the $\delta^{13}C_{organic}$ record. The observed differences in $\delta^{13}C_{\text{organic}}$ between the two lithologies

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Paper number 91PA02674. 0883-8305/92/91PA-02674\$10.00 are probably produced by changes in the isotopic composition and the concentration of dissolved CO₂, First, freshwater flooding during the formation of the sapropels caused the isotopic composition of the dissolved inorganic carbon in the surface waters of the Mediterranean to become lighter because of the ¹³C deficiency in fresh waters. Hence photosynthesis would have produced isotopically lighter organic material. Second, changes in atmospheric pCO₂ between glacial and interglacial periods, as shown by the Vostok ice core, caused marked changes in the concentration of free dissolved CO_2 in the mixed layer; lower values during glacial maxima caused a smaller fractionation of the carbon isotopes by phytoplankton, whereas levels were less limiting during the interglacials. Concentrations of dissolved CO₂ could also have been much higher during the deposition of the sapropels because of the supply of regenerated CO2 to the mixed layer by upwelling, and this could have further lightened the $\delta^{13}C_{organic}$ values in the sapropels themselves. Carbon isotope records may provide an alternative method for estimating atmospheric pCO₂ levels over longer time periods than can be obtained from ice cores.

INTRODUCTION

The occurrence and significance of organic-rich sediments (sapropels) in the Holocene and upper Pleistocene section of the eastern Mediterranean [Kullenberg, 1952] have elicited considerable interest [Olausson, 1961; Ryan, 1972; McCoy, 1974; Ryan and Cita, 1977; Cita et al., 1977; Thunell et al., 1977; Vergnaud-Grazzini et al., 1977; Williams

et al., 1978; Williams and Thunell, 1979; Cita and Grignani, 1982; Rossignol-Strick et al., 1982; Thunell et al., 1983; Calvert, 1983; Rossignol-Strick, 1985]. Such sediments occur intercalated in the nano-fossil and foraminiferal marl oozes that constitute the normal sedimentary facies of this basin. They have been recovered either by piston coring or by drilling; 12 such horizons occur in the late Quaternary section (sapropels S1-S12 of McCoy [1974] and Cita et al. [1977]), whereas an even larger number occur in older strata [Kidd et al., 1978]. They evidently formed mainly during the late stages of deglaciation, or during "warming trends" [Ryan, 1972]. The two prevailing explanations for their origin are (1) increased organic carbon preservation under bottom-water anoxia when deepwater renewal was restricted by the presence of low-salinity surface waters derived from increased run-off, either from the Black Sea [Olausson, 1961; Ryan, 1972; Thunell et al., 1977; Vergnaud-Grazzini et al., 1977; Thunell and Williams, 1982; Jenkins and Williams, 1984] or from the Nile [Rossignol-Strick, 1983, 1985; Rossignol-Strick et al., 1982], and (2) a higher settling flux of organic matter caused by increased primary production when the circulation of the eastern basin was reversed as a consequence of the increased run-off [Berger, 1976; Calvert, 1983; Sutherland et al., 1984].

Whatever mechanism might be responsible for their formation, it is clear that the sapropels contain valuable records of changing oceanographic conditions in the eastern basin of the Mediterranean throughout the late Pleistocene. Knowledge of their geochemistry and stable isotopic characteristics should enable us to obtain information on, among other things, the history of changing hydrographic conditions, circulation, carbon sources, and burial in the basin. Of particular interest here, it should be possible to identify the sources of organic matter in the sapropels from its bulk chemical composition, its molecular characteristics, and its stable isotopic composition. This information will be relevant to the question of the intensity of primary planktonic production, and hence the flux of organic carbon to the seafloor, during sapropel-forming events, the importance of terrestrial versus marine sources of organic materials, and whether different marine planktonic assemblages inhabited the photic zone during the deposition of the sapropel and the marl oozes. Finally, this information will be relevant to the contentious issue of the origin of organic-rich facies in the modern ocean and in the geological record.

Available information on the ultimate sources of organic matter in the sapropels is highly equivocal. On the basis of the abundance of fragments of higher plants, including pollen, high organic C/N ratios, the distribution of n-alkanes, the composition of humic substances and the results of pyrolysis analysis, Deroo et al. [1978] and Sigl et al. [1978] contended

that the organic matter in Pleistocene sapropels recovered by the Deep Sea Drilling Project is largely terrestrial. This would be consistent with the notion that increased terrestrial input was a factor in the formation of these deposits [Rossignol-Strick et al., 1982; Shaw and Evans, 1984]. In contrast, Rossignol-Strick et al. [1982], Shaw and Evans [1984] and Smith et al. [1986] found that the lipid fraction of the organic matter in S1 is predominantly of marine origin, with clearly recognizable inputs from dinoflagellates and coccolithophorids, although a significant terrigenous component is also present. They also pointed out that the marl oozes contain an even higher relative abundance of terrestrial organic material compared with the sapropels. Ten Haven et al. [1987] also identified a wide range of higher plant fragments and pollen in S1 in a profile of cores from the Nile region to the central basin of the eastern Mediterranean but nevertheless concluded that the bulk organic matter comprises a mixture of marine, terrestrial, and bacterial components, with the importance of terrestrial material increasing away from the Nile towards the deep basin.

The carbon isotopic composition of the organic matter in S1 has been interpreted as having a predominantly marine signature [Sutherland et al., 1984; Ten Haven et al., 1987] with δ^{13} C values ranging from -18.5 to -22.6‰. Ten Haven et al. [1987] suggested that the lighter values found in cores from the deep central basin compared with those from the shallow-water area close to the Nile indicate a larger terrestrial component away from the coast, consistent with the organic geochemical

evidence. Sutherland et al. [1984] found that the δ^{13} C values vary from -18.5 to -21.6‰ in S1 at a single site far removed from the influence of the Nile. As we shall discuss below, inadequate knowledge of the carbon isotopic composition of the

organic end-members in the Mediterranean makes it extremely difficult to determine precisely the terrestrial fraction of the organic matter present in the sapropels from this meagre information.

We have attempted to examine this problem by carrying out a more detailed study of the carbon isotopic composition of the organic matter in a sediment sequence from the easternmost Mediterranean containing several sapropel horizons. The original aim of the study was to determine the proportions of marine and terrestrial organic matter in the sapropels and associated sediments, but the results have required a more extensive investigation of the multiple causes of isotopic variability in marine sediment records.

MATERIALS AND METHODS

Core MD 84641 (33°02'N, 32°38'E; 1375 m) was collected during the NOE cruise of the R/V <u>Marion</u> <u>Dufresne</u> in August 1984. It was stored at 4°C and

sampled two weeks after collection. Separate samples were taken for foraminiferal isotopic and for chemical and mineralogical analyses. Sampling was at 10 cm intervals in the marl ooze sections of the core and at 2 cm intervals through the sapropels. Standard methods were used for the separation and cleaning of samples of Globigerinoides ruber; the isotopic composition of these samples was determined using either a VG 602D or a Finnegan 251 mass spectrometer at Gif-sur-Yvette. The results are reported in the standard δ notation relative to the Pee Dee Belemnite (PDB) standard. Organic carbon determinations at Gif-sur-Yvette were made by combustion, followed by coulometric determination of the evolved CO₂, of subsamples that were soaked in 10% HCl overnight and dried to constant weight at 80°C. Absolute precision was better than 0.1%. Samples for chemical and isotopic analyses of the organic fraction at UBC were dried at 60°C to constant weight and then ground to fine powders in a disc mill. Subsamples were used for total carbon and nitrogen determination by combustion/gas chromatography; the precisions determined on replicate subsamples were $\pm 1.25\%$ and $\pm 3.5\%$, respectively. Carbonate carbon was determined by acid digestion followed by coulometric determination of the evolved CO₂; precision was $\pm 2\%$. Organic carbon values were derived by the difference between total and carbonate carbon values, the combined precision being $\pm 2.35\%$. Differences between the two organic carbon data sets were generally less than $\pm 5\%$. The carbon isotopic determinations in the entire core were carried out on decarbonated subsamples using a VG PRISM mass spectrometer with an on-line CHN analyzer as the gas preparation device at UBC. The results are also reported in the δ notation relative to PDB, and the precision of the measurements was $\pm 0.05\%$. A smaller number of samples was also analyzed at Gifsur-Yvette by procedures described by Fontugne [1983], which involved off-line combustion of decarbonated samples in a stream of oxygen and the cleaning and trapping of the CO₂ cryogenically. Determinations were made using a VG 602C mass spectrometer, and the precision was $\pm 0.1\%$. The absolute differences between the two data sets averaged $0.66 \pm 0.59\%$, with no systematic bias.

RESULTS

Core MD 84641 consists of 11.61 metres of pale cream to yellowish brown nanofossil/foraminiferal marl ooze with nine olive green sapropels varying in thickness from 6 to 22 cm. Figure 1a and Table 1 show that the peak organic carbon contents of the sapropels range from 2 to 4.5% by weight. Four of the sapropels (those at 328-341, 374-393, 458-479.5, and 578-596 cm) contain grey mud layers with lower organic carbon contents, producing two maxima in organic carbon content at each level. The



Fig. 1a. Distribution of organic carbon with depth in core MD 84641.

marl oozes have organic carbon contents ranging from 0.1 to 0.3%, with no systematic trends in this lithology down the core.

The $C_{organic}/N$ ratio varies in a systematic manner throughout the core (Figure 1b). It is higher in the sapropels (16.92 ± 1.13), defined as those horizons with >1% organic C, compared with the marl oozes (12.34 ± 4.10). This difference is significant at the <1% level (two-sided t-test).

The chronostratigraphy of the core is based on the $\delta^{18}O_{ruber}$ record (Figure 2a), ¹⁴C dating of planktonic foraminifera in the upper 135 cm of the core and palaeomagnetic reversal stratigraphy. The radiocarbon data show that the uppermost sapropel has age ranges of 6500 and approximately 9000 years B.P. [Fontugne et al, 1989a, b] and is therefore S1 [McCoy, 1974]. The $\delta^{18}O_{ruber}$ record shows a range of -2.3 to +3.2‰, consistent with other foraminiferal records from the Mediterranean that all show a much larger glacial/interglacial range than that in the Atlantic [Vergnaud-Grazzini et al., 1977]. Correlation with the Specmap stack [Martinson et al., 1987] shows that the record in this



Fig. 1b. Distribution of C_{organic}/N ratios with depth in core MD 84641.

TABLE 1. Compositional and isotopic data on core MD 84641

Sample	Internolated	Organic C	Total N	\$130	Sample	Internalated	5180	8130
Depth (cm)	Age, years	percent	nercent	O ⁺⁺ Corganic (%)	Denth (cm)	Age, years	G. ruber (%a)	G. ruber (%m)
Chemistry		£	1	Ny	Forams		0	0
0	0 -	0.47	0.033	-19.90	1	216	0.84	1.84
5	1081	0.31	0.023	-18.81	3	649	0.79	1.95
10	2162	0.31	0.019	-19.99	10	2162	0.04	1.38
15	3243	0.32	0.023	-19.33	13	2811	0.05	1.67
20	4324	0.39	0.030	-20.13	20	4324	-0.57	1.25
25	5405	1.24	0.085	-20.43	22	4757	0.04	0.93
27	5838	1.61	0.108	-20.03	24	5189	-0.31	1.18
30	6486	1.87	0.131	-21.01	26	5622	-0.46	1.00
35	7155	1.93	0.133	-21.30	28	6054	-0.44	0.84
37	8000	1.90	0.124	-21.19	29	62/0	-0.56	0.83
38	4389	1.85	0.124	-21.10	32	7135	-0.60	0.94
40	8777	1.54	0.105	-20.57	36	7784	-0.04	1.01
45	10073	0.53	0.037	-18.38	37	8000	-1.37	1.16
50	11368	0.55	0.024	-17.56	39	8518	-1.28	1.34
55	12663	0.42	0.025	-17.74	40	8777	-0.91	1.02
60	13959	0.27	0.019	-17.79	42	9295	-0.15	1.20
70	16549	0.21	0.024	-18.91	43	9554	-0.04	0.83
80	18619	0.20	0.010	-18.41	45	10073	-0.08	1.04
90	20689	0.16	0.016	-18.57	46	10332	0.16	1.16
100	22581	0.24	0.023	-17.51	48	10850	0.27	1.06
110	24474	0.23	0.013	-16.89	50	11368	1.05	0.78
120	28900	0.19	0.017	-17.09	57	13181	1.24	1.18
140	35320	0.26	0.021	-17.83	60	13959	1.37	1.30
140	38054	0.23	0.020	-17.00	70	15254	0.92	1.14
160	41768	0.27	0.010	-17.25	70	10349	2.47	1.17
170	44582	0.31	0.021	-17.25	85	107/3	3.20	1.51
180	47396	0.29	0.021	-16.69	90	20689	3.01	1.58
190	50210	0.26	0.021	-17.12	95	21635	2.97	1.65
200	53333	0.21	0.019	-16.67	100	22581	2.61	1.48
210	56455	0.24	0.010	-17.30	105	23528	2.63	1.64
220	59578	0.30	0.016	-17.58	110	24474	2.66	1.49
230	62700	0 26	0.010	-16.95	115	25420	2.28	1.45
240	65823	0 29	0.019	-17.04	125	31919	2.75	1.50
250	68945	0.25	0.020	-18 02	130	33326	2.47	1.13
250	72068	0.22	0.010	-18.60	135	34733	2.69	1.47
270	75190	0.23	0.020	-19.30	140	36140	2.27	1.16
278	79313	0.48	0 0 0 3 3	-19.14	150	38954	2.54	111
283	79250	2.70	0.170	-19.92	170	41/08	1.70	0.97
285	80117	2.40	0.147	-20.00	180	44382	1.01	1.47
287	80983	2.47	0.146	-19.43	190	50210	1 26	1.20
290	82283	0.79	0.049	-18 61	200	53333	1.52	1 38
295	84450	0.41	0.032	-17.96	210	56455	1 31	1.24
300	86617	0.38	0.025	-18.31	220	59578	1.23	1.48
310	90950	0.18	0.015	-19.71	230	62700	1.83	1.46
320	92317	0.18	0.017	-19.06	240	65823	1.90	1.61
328	95142	2.20	0.133	-21.34	250	68945	2.24	1.61
330	95848	0.81	0.047	-19.48	260	72068	2.20	1.78
332	96555	0.43	0 027	-18.22	270	75190	0.94	1.51
333	9/614	1.91	0.110	-21.24	280	78313	0.08	0.98
336	98074	2.44	0.144	-21.38	283	79250	-0.69	1.11
340	101667	0.61	0.030	-19.28	263	80117	-0.64	0.75
345	105085	0.01	0.033	-20 17	207	80983	-0.37	0.96
350	110790	0.22	0.025	-19.00	300	86617	1.47	1.33
360	114409	0.15	0.015	-19.95	310	90950	1 67	1 78
370	118029	0.18	0.015	-19 52	320	92317	0.65	1.14
373	119115	3.56	0.208	-20 65	328	95142	0.02	0.81
375	119838	3.84	0.223	-20.62	330	95848	0.40	0.95
377	120562	4.52	0.267	-20.49	332	96555	0.92	1.34
380	121648	3.80	0.218	-20.62	335	97614	0.02	1.11
382	122372	3.27	0.191	-20.69	338	9 8674	0.31	0.57
384	123096	4.47	0.254	-20.60	340	99380	-1.12	0.72
386	123820	3.38	0.201	-20.50	342	101662	0.15	0.87
566	125007	2.18	0.135	-20.47	345	105085	0.49	1.04

				TABLE 1.(c	ontinued)			
Sample	Interpolated	Organic C	Total N	δ ¹³ C _{organic}	Sample	Interpolated	δ ¹⁸ O	δ ¹³ C
Depth (cm)	Age, years	percent	percent	(‱)	Depth (cm)	Age, years	G. ruber (‰)	G. ruber (‰)
Chemistry					Forams			
390	126195	1.80	0.103	-20.10	350	110790	1.63	1.54
392	127382	0.78	0.043	-19.34	360	114409	0.92	1.25
394	128569	0.28	0.021	-19.45	370	118029	-0.15	0.81
396	129757	0.13	0.015	-20.39	373	119115	-0.19	-0.11
398	130944	0.11	0.014	-21.10	375	119838	-0.38	0.16
400	132132	0.26	0.017	-21.26	377	120562	-1.13	0.06
405	135100	0.20	0.020	-20.32	380	121648	-1.02	0.04
410	136536	0.15	0.020	.18 70	382	122372	-0.10	0.11
420	130408	0.15	0.012	-10.77	384	123096	-1.77	0.02
420	142280	0.12	0.015	10.85	386	123820	-1.42	-0.01
430	142200	0.12	0.010	-17.05	199	125007	-1.32	0.15
450	157175	1.07	0.115	21.90	300	126105	-1.12	0.15
460	164673	1.92	0.071	-21.00	302	127382	0.22	1.09
462	166113	0.15	0.017	-18.08	30/	178560	0.34	1 14
402	167602	0.13	0.017	-10.70	394	120757	0.34	1.14
404	160002	1.06	0.132	-22.20	208	130044	0.70	1.20
400	109092	1.90	0.117	-22.13	396	130744	0.40	1.41
405	170381	2.41	0.145	-22.29	400	132132	1.01	1.10
470	1720/1	2.31	0.131	-22.49	405	133100	2.40	1.22
472	1/3560	2.24	0.135	-22.80	410	130336	2.08	1.03
474	175050	2.43	0.151	-20.72	420	139408	0.99	1.14
476	176081	3.15	0.174	-22.45	430	142280	0.51	0.77
478	177112	2.03	0.121	-21.30	440	149728	1.39	1.05
480	178144	0.41	0.031	-20.61	450	157175	0.62	1.18
485	180722	0.23	0.022	-20.67	460	164623	0.43	0.87
490	183300	0.22	0.020	-18.91	474	175050	0.40	1.02
495	184928	0.23	0.019	-18.83	476	176081	0.46	1.11
500	186557	0.22	0.026	-19.31	478	177112	0.71	0.78
505	188185	0.57	0.037	-19.57	480	178144	0.34	1.02
510	189813	0.64	0.039	-19.63	485	180722	1.21	0.97
515	191442	4.35	0.253	-21.19	490	183300	1.53	0.83
517	192093	3.98	0.230	-20.92	495	184928	1.00	0.99
520	193070	4.35	0.244	-20.84	505	188185	0.81	1.20
522	193570	4.31	0.243	-20.89	510	189813	0.69	1.08
524	194070	3.99	0.225	-20.89	515	191442	-1.36	0.87
526	194570	3.77	0.197	-20.55	517	192093	-1.56	0.68
528	195070	3.12	0.170	-20.44	520	193070	-1.81	0.88
530	195570	1.94	0.110	-20.87	522	193570	-1.52	0.69
532	196070	2.16	0.119	-20.78	524	194070	-1.28	0.34
534	196570	1.09	0.059	-19.42	526	194570	-1.66	0.71
536	197070	0.65	0.044	-19.34	528	195070	-1.64	0.95
538	197570	0.63	0.041	-19.12	530	195570	-0.21	0.37
540	198070	0.41	0.031	-18.87	532	196070	-0.49	0.65
550	200570	0.15	0.017	-17.94	534	196570	0.65	0.85
560	206328	0.25	0.019	-17.53	536	197070	0.67	1.25
565	209207	0.23	0.029	-18.82	538	197570	0.71	1.22
567	210358	1.69	0.087	-19.69	540	198070	1.05	1.06
568	210934	1.29	0.072	-20.25	550	200570	1.22	1.14
570	213237	1.29	0.076	-20.25	560	206328	-0.13	1.22
572	213813	0.74	0.045	-19.65	565	209207	-0.10	0.60
574	214389	0.57	0.041	-18.54	567	210358	-0.06	0.41
576	215540	0.78	0.054	-20.57	568	210934	-0.30	0.66
578	216090	0.80	0.053	-20.62	572	213237	-0.46	0.53
580	216640	0.86	0.055	-20.70	574	214389	-1.24	0.17
582	217190	1.32	0.074	-21.92	575	214964	-1.67	1.07
584	217740	1.59	0.091	-22.18	576	215540	-1.70	-0.06
586	218290	2.93	0.153	-21.52	580	216640	-1.60	-0.03
588	218840	2.36	0.141	-23.02	582	217190	2.12	0.22
590	219390	2.52	0.139	-21.81	584	217740	1.73	0.58
592	219940	2.25	0.135	-22.85	586	218290	-1.05	-0.09
594	220490	2.05	0.121	-21.86	590	219390	-0.54	0.57
596	221040	0.39	0.032	-19.49	592	219940	-0.61	0.80
600	222140	0.38	0.028	-19.67	594	220490	-2.33	0.61
610	224890	0.19	0.018	-19.18	596	221040	-0.46	0.66
620	230354	0.21	0.026	-19.27	600	222140	0.24	0.79
630	235818	2.41	0.142	-20.64	610	224890	1.64	0.82
632	236911	2.51	0.136	-20.51	620	230354	0.71	1.33
634	238004	2.13	0.125	-20.57	630	235818	-0.14	0.37
636	239097	0.62	0.050	-18.43	632	236911	0.01	0.27
638	240190	0.50	0.040	-18.17	634	238004	0.50	0.32

TABLE 1.(continued)								
Sample	Interpolated	Organic C	Total N	$\delta^{13}C_{organic}$	Sample	Interpolated	δ180	δ ¹³ C
Depth (cm)	Age, years	percent	percent	(%0)	Depth (cm)	Age, years	G. ruber (‰)	G. ruber (‰)
Chemistry					Forams			
640	240610	0.46	0.040	-18.17	636	239097	0.19	1.09
645	241658	0.43	0.032	-17.60	638	240190	0.76	1.14
650	242707	0.41	0.030	-16.82	640	240610	0.18	1.44
660	244805	0.25	0.022	-17.74	645	241658	-0.27	1.88
670	246902	0.36	0.022	-17.73	650	242707	0.56	1.66
680	249000	0.23	0.023	-18.25	660	244805	0.80	1.14
690	253000	0.22	0.023	-17.94	670	246902	1.44	1.14
700	25/000	0.18	0.023	-18.47	680	249000	1.46	1.00
710	239890	0.19	0.022	-18.70	690	253000	1.23	1.20
730	262780	0.18	0.025	-19.55	700	257000	1.11	0.81
740	270244	0.21	0.021	-19.61	710	25780	1.90	1.25
750	274818	0.15	0.022	-18 20	730	265670	1.75	1.19
760	279392	0.21	0.020	-21.57	740	270244	1.03	1 38
770	283966	0.17	0.029	-19.08	750	274818	0.84	1.39
780	288540	0.19	0.022	-18.59	760	279392	0.86	1.48
790	292027	0.15	0.019	-20.42	770	283966	0.92	1.31
800	295513	0.12	0.022	-19.21	780	288540	0.26	1.75
810	299000	0.12	0.019	-19.35	790	292027	0.76	1.75
820	302667	0.17	0.023	-19.80	800	295513	1.46	1.21
830	306333	0.15	0.017	-19.24	810	299000	1.62	0.91
840	310000	0.19	0.021	-20.33	820	302667	0.70	1.14
850	313333	0.30	0.025	-18.99	830	306333	0.94	1.72
857	31306/	0.35	0.027	-18.81	840	310000	-0.16	1.79
864	310333	0.33	0.025	-18./8	850	313333	1.04	1.21
870	320000	0.29	0.031	-19.55	850	31500/	0.28	1.07
880	321897	0.05	0.033	-19 21	854	318000	1.15	1.01
890	323793	0.14	0.016	-19.21	870	320000	1.13	1.75
900	325690	0.13	0.015	-19.77	880	321897	1.39	1.42
910	327586	0.13	0.019	-21.04	890	323793	0.56	1.29
918	329103	0.35	0.029	-19.95	900	325690	-0.01	1.33
920	329483	1.66	0.098	-20.79	910	327586	0.02	0.75
922	329862	2.10	0.126	-20.74	918	329103	0.30	0.57
924	330241	2.05	0.119	-20.93	922	329862	1.10	0.91
926	330621	2.48	0.148	-21.17	924	330241	0.60	0.63
928	331000	2.56	0.153	-20.81	926	330621	1.54	1.09
930	332121	1.90	0.122	-20.52	928	331000	-0.03	0.31
934	334364	0.52	0.041	-19.34	930	332121	0.42	1.13
940	33/727	0.52	0.040	-19.75	934	334364	0.89	0.74
950	343333	0.30	0.026	-18.90	940	33/72/	-0.34	1.00
970	354545	0.22	0.029	-17.37	950	343333	1.92	0.79
980	360151	0.12	0.014	-10.74	900	348939	0.72	0.65
990	365757	0.13	0.015	-16.62	980	360151	1.15	1.05
1000	371363	0.27	0.019	-17.37	990	365757	1.05	1 40
1010	376969	0.19	0.019	-17.93	1000	371363	1.40	1 35
1020	382575	0.30	0.018	-18.24	1010	376969	1.50	1.48
1030	388181	0.26	0.020	-17.62	1020	382575	1.40	1.41
1040	393787	0.22	0.019	-18.71	1030	388181	1.73	1.31
1050	399393	0.22	0.019	-18.55	1040	393787	1.22	1.22
1060	404999	0.25	0.016	-18.56	1050	399393	1.41	1.36
1070	410606	0.26	0.019	-19.81	1060	404999	0.75	1.40
1080	416212	0.19	0.017	-18.56	1070	410606	0.25	1.35
1090	421818	0.18	0.018	-18.84	1080	416212	0.38	1.36
1110	42/424	0.11	0.019	-18.77	1090	421818	1.19	1.63
1120	433030	0.19	0.025	-19.01	1100	42/424	1.54	1.71
1130	430030	0.19	0.020	-10.08	1120	433030	1.49	1.78
1140	449848	0.19	0.010	-19.19	1130	444747	1.11	1.59
1150	455454	0.21	0.019	-19.73	1140	449848	1.08	1.56
1160	461060	0.27	0.021	-18.40	1150	455454	0.98	1.69
			-	. –	1160	461060	1.39	1.79



Fig. 2a. The $\delta^{18}O_{ruber}$ record in core MD 84641. Stage boundaries were identified on the basis of correlation with the Specmap stack [Martinson et al., 1987].

core appears to extend back to Stage 12 as defined by Emiliani [1955]. The last glacial maximum (Stage 2) has δ^{18} O values of more than 3‰, whereas Stage 8 shows values close to 2‰. The lightest δ^{18} O values are found in Stage 7. The magnetic reversal Blake Event (beginning at 117 Ka) occurs at 367-375 cm depth in the core[Fontugne et al., 1989a; Tucholka et al., 1987], consistent with the isotope stratigraphy (Figure 2a).

The $\delta^{13}C_{ruber}$ record (Figure 2b) shows that the planktonic carbon isotopic composition also varies between wide limits throughout the core. Most noteworthy are the large negative excursions of the $\delta^{13}C$ values in eight of the nine sapropels. The

largest of these excursions occurs in S8, where values of -0.11‰ are found. Sapropel S6 shows only a very small negative excursion from the horizon immediately below S5 to the horizon immediately above S7. Sapropel S10 has a clear δ^{13} C signal in spite of the fact that the $\delta^{18}O_{ruber}$ amplitude at this depth is rather subdued (Figure 2a).



Fig. 2b. The $\delta^{13}C_{\text{ruber}}$ record in core MD 84641.



Fig. 3a. Distribution of organic carbon plotted against age derived from the foraminiferal stratigraphy shown in Figure 2a. The sapropel horizons are labelled S1 and S3-S10 [Cita et al., 1977].

The age model for this core yields sedimentation rates of between 1.34 and 5.28 cm/kyr These are fairly similar to those reported for this part of the eastern Mediterranean by Stanley and Maldonado [1977].

Comparison between the organic carbon (Figure 1a) and $\delta^{18}O_{ruber}$ records (Figure 2a) demonstrates that core MD 84641 contains sapropels S1 and S3-S10 [Cita et al., 1977]. Sapropel S2 appears to be missing (Figure 3a), as is often the case [Rossignol-Strick, 1985] for this rather poorly developed interglacial stage. The magnetic reversal Blake Event (beginning at 117 Ka) occurs between S4 and S5 [Tucholka et al., 1987].

The δ^{13} Corganic record (Figure 3b) shows that there is a very large variation in the isotopic composition of the organic matter in this core, with a total amplitude of more than 6‰. The values are systematically lighter in all sapropel horizons, with values reaching -23‰ in S8, and the heaviest values are confined to the glacial stages; values as heavy as -16‰ are found in stages 2, 8, and 10. In detail, the



Fig. 3b. The δ^{13} Corganic record in core MD 84641.

variations in $\delta^{13}C_{\text{organic}}$ closely follow those of the concentration of organic C; in sapropels S3, S4, S5, and S8 there are minima in organic C between two maxima, and the $\delta^{13}C_{\text{organic}}$ values are seen to change abruptly to heavier values at these same horizons (Figure 3b).

INTERPRETATION

The detailed compositional and isotopic records in core MD 84641 show that the sapropels have higher

 $C_{organic}/N$ ratios and lighter $\delta^{13}C$ values than the intercalated marl oozes and that the $\delta^{13}C_{organic}$ values in the glacial maximum horizons are significantly heavier than in any other part of the core.

The variation in the ${}^{13}C/{}^{12}C$ ratio of organic matter in marine sediments has in the past been ascribed to one of three factors: (1) variable mixtures of carbon derived from marine and terrestrial sources, (2) diagenetic changes in the isotopic composition of the deposited organic material, and (3) variations in the isotopic composition of the marine organic material.

Terrestrial Versus Marine Contributions

In most sedimentary environments, marine planktonic organic carbon is isotopically heavier than that of the carbon delivered to the ocean from terrestrial environments. Solution of the simple mixing equation $\delta^{13}C = F_t * \delta^{13}C_t + F_m * \delta^{13}C_m$, with $F_m + F_t = 1$, where F_t and F_m are the fractions of terrestrial and marine organic matter and $\delta^{13}C_t$ and

 $\delta^{13}C_m$ are the isotopic compositions of the terrestrial and marine organic carbon sources, respectively, provides a method for estimating the proportions of the two types of organic matter present in a given sample [Calder and Parker, 1968]. Making such estimates requires an accurate knowledge of the isotopic compositions of the marine and terrestrial end-members. Published information on the isotopic composition of terrestrial vegetation (summarized by Deines [1980]) shows that plants fall into two categories depending on the initial carbon fixing pathway during photosynthesis. These are C3 plants that use ribulose bisphosphate carboxylaseoxygenase (RUBISCO) as the carboxylating enzyme and which have $\delta^{13}C$ values that cluster around a modal value of -27‰, and C4 plants that use phospoenolpyruvate carboxylase (PEPC) as the initial carboxylating enzyme with $\delta^{13}C$ values close to -14‰. Studies of the input of terrestrial plant debris into the nearshore regions of the ocean [Sackett and Thompson, 1963; Sackett, 1964; Gearing et al., 1977; Tan and Strain, 1979a; 1983; Salomons and Mook, 1981; Fontugne and Jouanneau, 1987] have shown that, with few

exceptions, the terrestrial isotopic signal is that of C_3 plants; C₄ plants (sea grasses, etc.) can contribute significantly to the sediments in some coastal lagoons [Parker, 1964; Shultz and Calder, 1976] and reef tracts [Rasmussen et al., 1990], but the effect of such sources appears to be areally restricted. In order to have an important effect in core MD 84641, a contribution of more than 50% of the organic matter from pure C₄ plants would be required to produce the heavy δ^{13} C values in stages 2, 4, 8, and 10 (Figure 3b). The pollen record from the neighbouring core MD 84642 (32°41'N, 32°35'E 1260 m) shows that gramminea, which are potential C₄ plant contributors, represent less than 20% of pollen content which itself has a low abundance (<150/g) [Cheddadi, 1988]. It is unlikely therefore that the contrast in the δ^{13} C values between the sapropels and the marl oozes is produced by the variable admixture of C3 and C4 plant debris. We do not have measurements of the isotopic composition of the carbon being supplied to the Mediterranean Sea from the surrounding land areas, but we have assumed that its δ^{13} C value is close to -27‰ [Deines, 1980], which is characteristic of C₃ type plants of the temperate zone and similar to that of estuarine particulate organic carbon [Tan and Strain, 1979a, 1979b; Salomons and Mook, 1981; Fontugne and Jouanneau, 1987; Saliot et al., 1988]. Only very small changes in the isotopic composition of terrestrial organic material have been recorded during glacial to interglacial transitions [Fontugne and Duplessy, 1986], so that the isotopic composition of this source material is not likely to have varied significantly.

In contrast to the relatively constant value for the isotopic composition of the terrestrial organic matter supply to offshore marine sediments, some important variations have been reported for marine planktonic organic material [Müller et al., 1983; Fontugne and Duplessy, 1986; Sarnthein et al., 1988; Jasper and Hayes, 1990] so that it is difficult to specify uniquely the isotopic signature of this source. These $\delta^{13}C$ changes have been attributed to (1) sea surface temperature variations [Sackett et al., 1965; Fontugne and Duplessy, 1981], (2) water mass characteristics [Fontugne and Duplessy, 1978], (3) metabolic effects depending on the photosynthetic pathway [Descolas-Gros and Fontugne, 1985, 1990b], 4) phytoplankton species distribution [Fontugne, 1983], and (5) variations in the concentration of dissolved CO₂ in seawater [Rau et al., 1989]. The range of variation is around 10‰, so that this may lead to very large uncertainties in the determination of the terrestrial versus marine carbon flux to the deep ocean.

In the Mediterranean Sea, modern plankton has δ^{13} C values ranging between -24.5‰ in Alboran Sea winter plankton and -22‰ in Eastern Mediterranean winter plankton [Fontugne, 1983], regardless of the

proportion of phytoplankton and zooplankton in the samples. The latter is identical to values of -22% reported by Deuser [1970] in the Black Sea. Hence in the sapropels in core MD 84641, where δ^{13} C has a mean value of $-21.0 \pm 0.8\%$, the contribution of terrestrial organic matter to the bulk organic matter must be very small. Moreover, it must be even

smaller in the marl oozes, where the δ^{13} C values are even heavier (-18.84 ± 1.1‰). The conclusion that we are dealing with a predominantly marine source of organic matter in the sapropels is consistent with the predominantly marine source of the lipids in S1 reported by Smith et al. [1986] and Ten Haven et al. [1987].

Further support for the absence of a significant terrestrial contribution to the sapropels is the absence

of a gradient in δ^{13} C over the whole eastern Mediterranean between the potential source areas of terrestrial supplies, such as the Nile river, and the central parts of the basin (see Table 2). Such a pattern is also encountered in the Black Sea sapropel [Calvert and Fontugne, 1987]. Finally, Ten Haven et al. [1987] have shown that the $\delta^{13}C_{\text{organic}}$ values of sapropel S6 are significantly more negative than those of other sapropels, while the terrigenous fraction, as estimated from the presence of organic biomarkers, is no different from that in other sapropels. Moreover, these authors show that a relative increase in marine biomarkers corresponds with more negative $\delta^{13}C_{\text{organic}}$ values. The organic C/N ratio of organic matter is also

The organic C/N ratio of organic matter is also considered to be a useful indicator of the origin of organic matter in marine sediments [Stevenson and Cheng, 1972]. Values between 6 and 7 by weight reflect the presence of planktonic organic matter [Redfield et al., 1963], whereas typical terrestrial plants exhibit values greater than 20 [Hedges et al., 1986]. In core MD 84641, the C_{organic}/N ratios are higher (mean = 16.9 ± 1.1 ; n = 63) in the sapropels compared with the marl oozes (mean = 12.3 ± 4.1 ; n = 135). Hence, solely on basis of the C/N ratio, one could conclude that both the sapropels and the marl oozes contain a mixture of marine and terrestrial organic matter and that terrestrial organic matter constitutes a larger fraction of the total in the

Core	S1	S3	S4	S5	S6	S7	S8	S9	S10
DED 8708 39°42'N; 13°34'E 2965 m			-22.35	-22.09		-23.24	-21.36	-22.30	-21.64
DI 10103/8K 36°10 N; 20°40'E 2900 m	-21.30 -21.00 -21.60								
TR 83-7 33°39.7'N; 25°59'E 2480 m					-22.70	-22.00 -22.50			
TR 83-30 33°48'N; 28°36.6'E 2810 m	-22.60								
MD 84639 33°40'N; 32°42'E 870 m	-21.18								
MD 84642 32°40'N; 32 34'E 1260 m	-21.20	-19.84 -19.51	-21.07 -20.93 -21.06	-22.47 -21.59	-21.59				
MD 84641 32°40'N; 32°34'E 1375 m	-20.80	-19.80	-20.64	-20.58	-22.44	-20.82	-22.17	-20.57	-20.83
MD 84629 32°13'N; 33°45'E 745 m	-20.88								

TABLE 2. 813Corganic Values in Sapropels From Other Eastern Mediterranean Sites

DED 8708 from M. Paterne (unpublished data, 1990); TR 83-7 and TR 83-30 from Ten Haven et al. (1987); DI 10103/8K from Sutherland et al. (1984)



15

C_{organic}/N_{total}

20

25

30

Fig. 4. Relationship between $\delta^{13}C_{\text{organic}}$ and the C_{organic}/N ratio in core MD84641.

10

5

sapropels. In spite of a clear difference in the Corganic/N ratio between the two lithologies, the degree of correlation between C_{organic}/N and δ^{13} C is quite poor (Figure 4). The lower C/N ratios in the marl oozes are probably partly caused by the presence of a large amount of adsorbed inorganic N relative to total N in these organic-poor sediments [Müller, 1977]. In addition, Suess and Müller [1980] have shown that the organic C/N ratio of sedimentary organic matter in high-productivity environments is much higher than that deposited under oligotrophic conditions. This is due to the preferential removal of nitrogenous organic material during particle settling and prior to burial and the contribution of nitrogen to the sediment by benthic organisms. These processes could contribute to the difference in the C/N ratio of the two facies in core MD 84641 if the marl oozes represent the deposits of an oligotrophic basin, like the present situation, and the sapropels represent the deposits of high-productivity conditions, as argued by Calvert [1983, 1987] and Sutherland et al. [1984]. These complicating factors imply that we cannot use these C/N data to infer organic matter sources in this core.

Effects of Diagenesis

The carbon isotopes are not uniformly distributed among all components of sedimentary organic matter. Differences of up to 7‰ have been reported, for example, between the lignin and cellulose fractions of marine sediments [Benner et al., 1987]. It is also known that the various components of sedimentary organic matter are decomposed at different rates, so that the bulk δ^{13} C value of organic matter can change differentially with diagenesis. Although Sackett [1964], Rogers and Koons [1969], and Dean et al. [1986] discounted the effects of diagenetic alteration of the carbon isotopic composition of sedimentary organic matter, there is

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evidence for some changes in this signal. Spiker and Hatcher [1984] showed that the bulk δ^{13} C values of algal sapropelic sediment accumulating in Mangrove Lake, Bermuda changed by 4‰ over the last 5000 years. They attributed a somewhat irregular change to lighter δ^{13} C values at the base of a marine sapropel horizon to the preferential degradation of the carbohydrate fraction (being relatively enriched in ¹³C) which caused the residual organic material, which was enriched in humin, to become depleted in ¹³C. Dean et al., [1986] pointed out, however, that the environment in this lake has changed from freshwater to marine conditions during this time interval, and the isotopic composition of the dissolved inorganic carbon, as shown by the $\delta^{13}C$ values of carbonate shell remains, has also changed in parallel with the changes in the isotopic composition of the organic matter. They suggested therefore that the observed change may be due to a change in the source of CO₂ (and hence its δ^{13} C value) and not necessarily to diagenesis.

McArthur et al. [1991] found a ¹³C depletion of 2.5‰ in the organic fraction of the upper, oxidized compared with the lower, unoxidized portion of a turbidite from the northeatern Atlantic, and they explained this by the different degrees of alteration of bulk planktonic organic matter which has been supplied to the abyssal seafloor from the neighbouring slope region. Such isotopic change occurs in this situation only after a time interval of several thousand years, because McArthur et al. [1991] were not able to detect a significant change in $\delta^{13}C_{organic}$ in a second turbidite that had been exposed to oxygen for only 200 years.

Glacial to interglacial cyclic variations in δ^{13} Corganic in the sediments of the Pigmy Basin (Gulf of Mexico) could not be explained by diagenesis [Jasper and Gagosian, 1990]. These authors demonstrated, from paired isotopic and N/Corganic ratios, that the composition of the organic matter in this basin is a determined by the mixture of C3-terrigenous and marine components which vary on glacial-interglacial time scales. The conclusion that climatically controlled mixing of terrestrial and marine organic matter accounts for the sedimentary isotopic record in this basin confirms the earlier extensive work of Joyce et al. [1985], Neuman et al. [1973], Sackett [1964], and Sackett and Rankin [1970] on the Gulf of Mexico.

Fontugne and Duplessy [1986] also concluded that diagenesis was not substantially involved in producing the $\delta^{13}C_{\text{organic}}$ record in late Pleistocene cores from the Indian Ocean. They showed that, depending on the location of the cores, Holocene sections of cores had both lighter and heavier $\delta^{13}C$ values compared with late Pleistocene sections, so that there was no consistent secular trend of changing

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isotopic compositions. Moreover, the organic carbon isotopic composition was independent of the total organic carbon content in most of the cores studied. The isotope records in this study were interpreted as a reflection of the balance between marine and terrestrial organic matter supplies to the sediments and the temporal change in the composition of the plankton, whose composition changes with water mass [Fontugne and Duplessy, 1978].

Finally, Hayes et al. [1989] observed a shift in the δ^{13} C values of total organic carbon (TOC) compared with those of the Ni-geoporphyrin fraction of a sequence of calcareous shales and limestones of the Cretaceous Greenhorn Formation in the Interior Seaway of western North America. These authors concluded that the δ^{13} Cporphyrin values closely

reflected the primary planktonic δ^{13} C values (offset by 0.5‰) whereas the δ^{13} C_{TOC} values are further offset because of secondary effects, including foodchain fractionation and sedimentary diagenesis. The inferred isotopic difference between the primary planktonic and the sedimentary organic matter in this rock sequence ranged from 0.6 to 2.8‰, although

the stratigraphic variations in $\delta^{13}C_{TOC}$ closely mirrored those of $\delta^{13}C_{porphyrin}$.

The combined evidence reviewed here suggests that diagenetic changes in the carbon isotopic composition of sedimentary organic matter of up to 2.8 ‰ are possible, depending on the environment of sedimentation and the time elapsed since deposited organic matter has been exposed to oxidative degradation. However, such changes are often very difficult to identify, and they would be expected to produce systematic differences between the isotopic composition of the sedimentary organic matter and the planktonic precursor.

Core MD 84641 covers a time period of approximately 450 kyr during which there are three climatic cycles (Figure 2a) and a series of 9 sapropelic events (Figure 3a). The main features of the record are the presence of consistently lighter

 δ^{13} C values in the sapropels (ranging from -19.4 to -23.0‰), the heaviest values at the glacial maxima (ranging from -16.6 to -21.6‰), and the lack of a secular trend (Figure 3b). Diagenesis would cause the pre-Holocene values to be systematically lighter than they were originally, which would mean that the planktonic organic matter supplied to the sediment during the glacial maxima would have been even more enriched in ¹³C. Moreover, unless the type of organic matter in the sapropels and the marl oozes were significantly different, for which there is no evidence, it is difficult to produce the isotopic contrast between these two lithologies by diagenetic effects alone. The only way that this could happen is by the diagenetic environment in the sediment (oxic in the marl oozes or anoxic in the sapropels) causing the preferential degradation of isotopically lighter

organic matter in the marl oozes and/or heavier organic material in the sapropels. There are currently no data available which test this possibility.

We conclude that diagenesis alone cannot be responsible for the large contrast in isotopic values between the sapropels and the marl oozes in core MD 84641. This is based on the relatively small (2.8‰) changes that can be definitively produced by diagenesis under certain circumstances and/or over long time periods compared with the 6‰ variation that we observe, the lack of a secular trend in the $\delta^{13}C_{\text{organic}}$ record, and the distinct and consistent differences between the two lithologies.

Changes in the Isotopic Composition of Marine Plankton

If contributions of terrestrial organic matter in core MD 84641 are small and diagenesis cannot explain the large and systematic changes in the isotopic composition of the organic carbon, then the

 $\delta^{13}C_{\text{organic}}$ record in core MD 84641 is probably caused by variability of the isotopic composition of the planktonic organic matter produced in the eastern Mediterranean during the time period represented by the core. The isotopic composition of marine planktonic organic matter is known to be variable, and several different factors have been proposed to explain this variability. Descolas-Gros and Fontugne [1990b], Fontugne and Duplessy [1978, 1981] and Sackett et al. [1965] have discussed the temperaturedependent fractionation of carbon isotopes by marine phytoplankton (0.28 to 0.35% / C) which leads to an increase of δ^{13} C values with increasing sea surface temperature (SST). The trend in core MD 84641 is in the opposite sense, since the heaviest δ^{13} C values are found in the glacial maximum horizons when surface water temperatures were roughly 4°C lower than at present [Thiede, 1978; Thunell, 1979]. Hence a temperature effect cannot be involved.

A second possibility is that the isotopic composition of the dissolved inorganic carbon (DIC) pool in the eastern Mediterranean has changed over the last 450 ka. The enclosed continental environment of the Mediterranean Sea suggests that changes in the input of fresh water would have an important effect on the nature of the DIC in surface sea water. Fresh water contains mainly dissolved CO₂ with a low bicarbonate content; the isotopic composition of the ΣCO_2 ranges between -5 to -10%. Seawater, on the other hand, has a high bicarbonate concentration (>90% of the ΣCO_2) and low dissolved CO₂. In the Mediterranean, the isotopic composition of the ΣCO_2 is around 1.5‰ [Duplessy, 1972]. Between these two end-members the δ^{13} C value of Σ CO₂ varies according to the salinity of the water. A large freshwater input, as shown by the large changes in the δ^{18} O values in planktonic for a in the sapropels [Williams et al., 1978; Williams and

Thunell, 1979], will therefore lower the δ^{13} C values of Σ CO₂ which is used during photosynthesis. The

 $\delta^{13}C_{\text{ruber}}$ record in core MD84641 (Figure 2b)

shows large negative excursions in each sapropel, especially in S5 and S8. The full range of individual values is from 1.95 to -0.11‰. These changes are much larger, and the δ^{13} C values are much lighter, than any changes recorded by open ocean planktonic

for aminifera, where the differences between glacial and interglacial values are generally less than 0.2‰. The markedly light δ^{13} C values of <u>G. ruber</u> in the sapropels probably reflect this shift in the isotopic composition of DIC due to mixing of fresh water, and the freshwater contribution could be estimated by

comparing the δ^{13} C values of <u>G. ruber</u> in the Mediterranean Sea with those measured in the open

ocean. Assuming <u>G. ruber</u> has a δ^{13} C value of 1.5‰

at 8000years B.P. (S1) and fresh water has a δ^{13} C value of -10‰, the contribution of fresh water to the surface waters of the eastern Mediterranean at this time is approximately 6%. In S5, where the

 $\delta^{13}C_{ruber}$ excursion reaches -0.09‰, the freshwater contribution amounted to 14%. Because <u>G. ruber</u> lives at around 50 m depth, it does not fully represent the near-surface conditions where phytoplankton grow; these estimates are therefore minima. Such

freshwater contributions would lower the $\delta^{13}C$ of the DIC in the ambient waters to approximately 0.8‰ in S1 and to -0.1‰ in S5, and this would produce

 δ^{13} Corganic values of -20.7 to -21.6‰ assuming the modern plankton is -20‰ and the fractionation involved is -21.5‰ (1.5 to -20‰). The values in the sapropels reach -21:2 and -22.8‰, respectively.

Hence the lighter $\delta^{13}C_{\text{organic}}$ values in the sapropels may be partly produced by the effects of fresh water input into the eastern Mediterranean.

The input of fresh water into the eastern Mediterranean would also increase the concentration of dissolved CO2 in surface waters. As proposed by Rau et al. [1989], this would favour a higher degree of discrimination against ¹³C during photosynthesis, thereby producing ¹³C-depleted organic matter. Such an effect is also consistent with the observations in the Saint Lawrence River [Descolas-Gros and Fontugne, 1990a] showing that the activity of RUBISCO, the enzyme that catalyzes the primary fixation of CO₂ during photosynthesis and which discriminates against ¹³C [Park and Epstein, 1960], is enhanced when salinity is low. Descolas-Gros and Fontugne [1990b] have reviewed the evidence for the occurrence of low $\delta^{13}C_{organic}$ values and high **RUBISCO** activity in natural phytoplankton assemblages. Hence these combined effects could be important in producing the light $\delta^{13}C_{\text{organic}}$ values within the sapropel horizons in core MD 84641.

The much heavier δ^{13} C values in the glacial maximum horizons of core MD 84641 (Figure 3b) also require explanation. Values as heavy as -17‰ are uncommon for open ocean phytoplankton and for marine sediments. As pointed out previously, the reported temperature effect on carbon isotope discrimination [Sackett et al., 1965; Fontugne and Duplessy, 1981; Fontugne, 1983] cannot explain such values because a lower glacial sea surface temperature [Thiede, 1978; Thunell, 1979] would

induce lighter rather than heavier $\delta^{13}C_{\text{organic}}$ values. An alternative metabolic effect which may be involved is the different isotope fractionations induced by different photosynthetic pathways depending on changes in environmental conditions or the assemblages of phytoplankton [Descolas-Gros and Fontugne, 1985; Descolas-Gros and Fontugne, 1990b]. Experimental studies of single species cultures suggest that high activities of RUBISCO are correlated with relatively depleted $\delta^{13}C$ values, whereas more enriched $\delta^{13}C$ values were found with high in vitro PEP carboxylase or PEP carboxykinase

activities which characterize carbon fixation by Bcarboxylation [Descolas-Gros and Fontugne, 1985]. The association between high RUBISCO activity and light δ^{13} C values has also been observed in field phytoplankton assemblages [Descolas-Gros and Fontugne, 1988]. Moreover, Descolas-Gros [1983], Descolas-Gros and de Billy [1983] and Descolas-Gros and Fontugne [1985] have observed a correlation between the abundance of dinoflagellates and heavy δ^{13} C values in natural phytoplankton populations off Portugal. Values as heavy as -16 ‰ have been observed in samples where PEP carboxylase activity was also the highest. Dinoflagellate abundances could have been high during the glacial maxima in the eastern Mediterranean in view of the high salinity and low nutrient status of the basin [Thunell and Williams,

1989], so that the high $\delta^{13}C_{organic}$ values observed in the core would reflect a change in the phytoplankton assemblages present.

Descolas-Gros and Fontugne [1985] found enriched δ^{13} C values in laboratory cultures during the end of the exponential growth phase and associated this with metabolic processes linked to the high cell content within the cultures. Degens et al.

[1968] and Fontugne [1983] also found heavy δ^{13} C values in the Peru upwelling region and attributed them to the ambient high productivity which could lead to free CO₂-limitation as suggested by Deuser [1970] for a diatom bloom in the Black Sea. Descolas-Gros and Fontugne [1990a b] also showed that limitation of free dissolved CO₂ with increasing salinity in the Saint Lawrence River estuary induced increasing β -carboxylase activities in phytoplankton populations leading to the enrichment of ¹³C in the

organic tissue. This effect is unlikely to be involved here, however, because the marl oozes have low organic carbon concentrations and, as in the case of the modern eastern Mediterranean, probably represent the deposits of an oligotrophic basin. The effects of the concentration of dissolved free

CO₂ on the wide variability of the $\delta^{13}C_{organic}$ values of marine plankton have recently been reexamined by Rau et al. [1989], Jasper and Hayes [1990] and Rau et al. [1991]. Accordingly, the marked differences in the carbon isotopic composition of phytoplankton between low and high latitudes [Sackett et al., 1974; Rau et al., 1982; Sackett, 1986a, b] may be linked to the concentration of dissolved $CO_2(aq)$ in the ambient waters, which is largely controlled by sea surface temperature (SST). Thus the greater solubility of CO₂ at low SST produces dissolved CO₂ levels that are 2.5 times higher for a given pCO₂ level at high latitudes (around Antarctica) compared with tropical latitudes [Rau et al., 1989]. Degens et al. [1968] showed that in laboratory experiments the δ^{13} C value of phytoplankton is lighter when higher concentrations of CO_2 were available in the culture medium, and Degens [1969] suggested that the isotopically light organic matter in ancient sediments is probably caused by a higher abundance of CO₂ in the ancient seas. Arthur et al. [1985] and Dean et al. [1986] have likewise suggested that the depleted $\delta^{13}C_{\text{organic}}$ values in Cretaceous strata may be attributed to elevated oceanic, and hence atmospheric, pCO2 levels. McCabe [1985] and Mizutani and Wada [1982] have confirmed the experimental results of Degens et al. [1968], and Degens et al. [1968] and Deuser [1970] have suggested that the heavy $\delta^{13}C_{organic}$ values sometimes observed in plankton blooms are due to CO₂-limitation consequent upon the rapid growth of the phytoplankton.

Jasper and Hayes [1990] have shown that the reconstructed primary $\delta^{13}C_{organic}$ and the for a for a miniferal (G. ruber) δ^{13} C records in a hemipelagic core from the Gulf of Mexico can be calibrated against the atmospheric pCO2 record obtained from the Vostok ice core [Barnola et al., 1987]. They used a relationship proposed by Popp et al. [1989] between the carbon isotopic fractionation associated with photosynthesis and the concentration of dissolved free CO₂ to reconstruct the atmospheric pCO₂ record over the last 90,000 years. They produced a δ^{13} C record of C₃₇ alkadienones [Jasper and Gagosian, 1989] rather than the total organic carbon to represent the phytoplankton biomass, and by calibrating the fractionation of the carbon isotopes at selected time intervals by reference to surface water dissolved CO₂ concentrations derived from the Vostok ice core CO₂ record and assuming equilibrium, obtained atmospheric pCO₂ values of circa. 180 µatm at the last glacial maximum and circa. 272 µatm in the late Holocene. Use of the

 $\delta^{13}C_{organic}$ of the total organic material in the core produced a pCO₂ record that was incongruent with that from Vostok because of the influence of terrestrial organic matter in this particular geological setting.

Rau et al. [1991] used the $\delta^{13}C_{\text{organic}}$ records of the total organic fraction of several deep-sea sediment cores [Froelich et al., 1979; Müller et al., 1983; Fontugne and Duplessy, 1986] to extend this approach. They showed that surface water dissolved CO₂ concentrations derived from the sediment data, using the empirical relationship between planktonic

 δ^{13} C and surface water dissolved CO₂ found earlier by Rau et al. [1989], are consistent with those derived from the atmospheric, and hence oceanic, pCO₂ variations provided by the Vostok record if reasonable assumptions are made about sea surface temperature fluctuations over the last glacialinterglacial cycle. Thus changes in the $\delta^{13}C_{organic}$ of approximately 2‰ across the Holocene/glacial transition in cores from the Indian Ocean [Fontugne and Duplessy, 1986] are of the right magnitude to correspond to a 3 μ M decrease in oceanic CO₂(ag) derived from the ice core record. Such changes in $\delta^{13}C_{\text{organic}}$ are not due to changes in the isotopic composition of the DIC in seawater, since the variations in δ^{13} C values of planktonic foraminifera are much smaller (<0.5%) between recent and glacial periods.

We have attempted to interpret the the $\delta^{13}C_{organic}$ record in core MD 84641 (Figure 3b) in a similar way by using the relationship between planktonic $\delta^{13}C_{organic}$ and measured CO₂(aq) concentrations derived by Rau et al. [1991]. We have made a simple correction to the $\delta^{13}C_{organic}$ values for the variation in the isotopic composition of the near-surface waters by subtracting the difference between the $\delta^{13}C_{ruber}$

values (Figure 2b) and the present δ^{13} C values (1.5%) of the surface water DIC in the eastern Mediterranean [Duplessy, 1972]. Globerigerinoides ruber is known to be a highly variable planktonic species [Emiliani, 1969], and shells of different sizes have significantly different δ^{13} C values [Duplessy et al., 1981]. Since we did not separate the foraminiferal samples into different size fractions nor use distinct morphological types in this study, the correction to the $\delta^{13}C_{\text{organic}}$ values is only approximate. However, the corrected values should yield a more accurate estimate of the CO₂ (aq) of sea surface water. The resulting dissolved CO2 record (Figure 5a) shows that free dissolved CO₂ levels ranged from 4.7 to 11.9 μ M at the site of core MD 84641. The lower values characterize the glacial horizons, especially stages 2-4 and 10, and the higher values are confined to the sapropels, whether they occur in interglacial (S1, S3-5, and S7-10) or



Fig. 5a. Record of surface water dissolved CO₂ concentrations estimated from the core $\delta^{13}C_{organic}$ record using the relationship between CO₂(aq) and planktonic $\delta^{13}C_{organic}$ reported by Rau et al. [1991].

glacial (S6) stages. The high values are slightly higher than what one would expect for SST values of around 20°C (the average modern value in the eastern Mediterranean) at preindustrial pCO2 levels, whereas the low glacial values are a little lower than those estimated for the last glacial maximum when the average SST was approximately 16°C. The higher concentrations derived from the sapropel $\delta^{13}C_{\text{organic}}$ values (Figure 3b) may be due, as suggested earlier, to an additional supply of dissolved CO₂ from below the mixed layer by, for example, upwelling, which may be have been responsible for the formation of the sapropels [Berger, 1976; Calvert, 1983]. It is noteworthy that the estimated dissolved CO₂ values are quite high in Stage 8, where no sapropel occurs, and throughout Stage 9, where a single, thin



Fig. 5b. Surface water pCO₂ record derived from the estimated dissolved CO₂ concentrations in Figure 5a using the solubility data of Weiss [1974] and assuming average modern SST = 20°C and S = 38, average glacial SST = 16°C and S = 40, and average sapropel SST = 20°C and S = 35. The dashed line is a three-point moving average.



Fig. 5c. Surface water pCO₂ record derived from the estimated dissolved CO₂ concentrations in the marl oozes only. The dashed line is a threepoint moving average.Stage boundaries are shown at the top of the figures.

sapropel does occur. These results suggest that because of changes in atmospheric CO₂ and the supply of additional CO₂ by upwelling, the changes in $\delta^{13}C_{\text{organic}}$ observed in the core could be caused almost entirely by variations in the concentrations of dissolved CO₂ in surface waters.

Estimates of the surface water pCO₂ levels recorded by the variations in $\delta^{13}C_{organic}$ have been obtained by taking the estimated values of dissolved CO₂ shown in Figure 5a and applying the solubility constants at appropriate temperatures and salinities from Weiss [1974] and assuming (1) oceanatmosphere equilibrium, which is probably reasonable considering the oligotrophic nature of the eastern Mediterranean, (2) that interglacial and sapropel sea surface temperatures were the same as those of today, (3) that glacial SST was 4°C lower than that of today [Thiede, 1978; Thunell, 1979] and (4) that the sea surface salinities during glacials, interglacials and sapropel events were 40, 38, and 35, respectively [Thunell and Williams, 1989]. The derived values range from 145 to 359 ppmv, with low values in the glacial maxima and high values in interglacials (Figure 5b). The high variability of the estimated pCO₂ levels is produced to a some degree by the high values in each of the sapropels, which, as we have suggested, may be due to additional supplies of dissolved CO₂ by upwelling, which would cause marked disequilibrium of the dissolved CO₂ in the surface waters. A better estimate of surface water pCO₂ values that may be compared directly with atmospheric levels may therefore be obtained by omitting the data from the sapropel horizons. The surface pCO₂ levels estimated using the data from the marl oozes only are shown in Figure 5c; both the range and the variability of the values are reduced, but the main features of the profile are similar that those shown in Figure 5b.

Lowest values occur in stages 2-4, 7 and 10-11, whereas the highest values are found in stages 6 and 8-9. On this time scale, the sediment record shows that surface water pCO_2 was not necessarily at the lowest level during all glacial maxima.

A direct comparison between the surface water pCO₂ values estimated from the $\delta^{13}C_{organic}$ values and the Vostok ice core record, using the revised time scale of Shackleton et al. [1990], is shown in Figure 6. The core top values are very similar to the most recent values recorded in the Vostok core [Barnola et al., 1987], whereas the minimum in MD 84641 in stages 2, 3, and 4 is lower (153 ± 10 ppmv) than that in the Vostok record (203 ± 11). On the other hand, the derived values in stage 5 are similar to those from Vostok, with the values from the two records deviating in stage 6.

The concordance between surface water pCO₂ concentrations derived from sediment records and the atmospheric record from ice cores depends, of course, on the extent of equilibrium between the atmosphere and the sea surface with respect to CO₂. Rau et al. [1991] have pointed out that the available data show major departures from equilibrium, so that direct comparisons should be made with caution. One could, in principle, calibrate the $\delta^{13}C_{\text{organic}}$ record from MD 84641 with the Vostok atmospheric record over the last 140 kyr in a similar manner to that carried out by Jasper and Hayes [1990], but as Rau et al. [1991] have pointed out the adjusted parameter values obtained by these authors (using the





 δ^{13} C values of alkenones) are substantially different from those seen in the modern ocean. We have not attempted this, however, because, in addition to the frequent disequilibrium of CO2 at the sea surface, the inherent variability of the $\delta^{13}C_{\text{ruber}}$ values that we have used to correct the $\delta^{13}C_{organic}$ values from MD 84641, alluded to earlier, the need to interpolate at least one of the records used for calibration and because the error in the Vostok record itself would all contribute to an even higher degree of uncertainty in such a calibration. Nevertheless, the reconstruction of atmospheric pCO₂ records from sediment cores appears to hold considerable promise. Much longer atmospheric pCO₂ records that are linked directly with the marine stratigraphic record [Shackleton, 1990] at minimum cost should be possible using existing piston cores, once the problems of calibration discussed by Rau et al. [1991] are resolved.

SUMMARY AND CONCLUSIONS

The $\delta^{13}C_{\text{organic}}$ record of core MD 84641 from the eastern Mediterranean shows very large variations between glacial maximum horizons and the sapropels. The depleted values observed in the sapropels are similar to those of modern plankton from the same basin, while the heavy values in the glacials represent much larger ¹³C enrichments than are observed in any modern marine plankton. Admixtures of significant quantities of terrestrial organic matter in the sediments, temperature effects on isotope fractionation or diagenesis cannot account for these variations. The most likely cause is variations in the isotopic composition and the concentration of dissolved CO_2 in the mixed layer of the eastern Mediterranean. Extensive freshwater flooding during the sapropels caused the dissolved CO2 to become significantly lighter, so that the phytoplankton became even more ¹³C depleted. Increases in total DIC in the surface waters during the deposition of the sapropels due to upwelling also probably led to larger carbon isotope fractionations because of a high concentration of CO₂(aq). In addition, changes in atmospheric pCO2 caused parallel changes in the concentration of dissolved CO_2 in the surface waters

The dissolved CO₂ record obtained in this study is similar to that obtained by Rau et al. [1991] from two cores spanning the last 160 kyr, one from the northern Atlantic and one from the northern Indian Ocean. In these cores, glacial δ^{13} Corganic values are above -19‰, and this produces dissolved CO₂ concentrations of approximately 7 µM. However, other deep-sea cores have δ^{13} Corganic records that show either no change between the Holocene and the last glacial maximum [Pedersen et al.,1991] or lighter values during glacial maxima [Fontugne, 1983], implying that the explanation given here and by Rau et al. [1991] does not apply globally. This conundrum can be explained, however, by the differences in the oceanographic settings of the cores from which these records were obtained. That is, eutrophic regions, where upwelling supplies both nutrients and dissolved CO2 to the mixed layer, may have had high CO₂ concentrations during the last glacial maximum even though atmospheric CO2 levels were lower [Pedersen et al., 1991]. This would have been even more true if such regions had enhanced upwelling because of higher wind speeds compared to modern conditions, as suggested for the equatorial Pacific by Pedersen [1983]. On the other hand, the oligotrophic regions of the ocean, especially the subtropical gyres and the Mediterranean, may not have changed their states so that nutrient limitation was accompanied by lower dissolved molecular CO2 concentrations because of lower atmospheric CO2 levels. Thus, although longer atmospheric CO2 records could potentially be obtained from deep-sea piston cores than can presently be obtained from ice cores, a judicious selection of the locations of the cores with respect to particular oceanographic settings will be required.

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