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Community primary production and calcification in a NW Mediterranean ecosystem dominated by calcareous macroalgae

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ABSTRACT: The community metabolism of a shallow infralittoral ecosystem dominated by the calcareous macroalgae *Corallina elongata* was investigated in Marseilles (NW Mediterranean), by monitoring hourly changes of seawater pH and total alkalinity over 6 d in February 2000. Fair weather conditions prevailed over the study period as indicated by oceanographic (temperature, salinity, and current velocity and direction) and meteorological variables, which validated the standing water hypothesis. This temperate ecosystem exhibited high community gross primary production (GPP = 519 ± 106 mmol C m⁻² d⁻¹, n = 6) and also supported high rates of community respiration (R). As a result, the system was slightly autotrophic (net community production, NCP = 20 mmol C m⁻² d⁻¹), with a GPP/R ratio of 1.06. NCP exhibited circadian variations with 2- to 3-fold changes in community respiration, both in the light and in the dark. Rates of net community calcification also exhibited circadian variations, with positive rates (up to 24 mmol CaCO₃ m⁻² h⁻¹) for irradiance values >300 W m⁻² (about 1380 μmol photon m⁻² s⁻¹). Below this irradiance threshold, net community dissolution prevailed. Daily net calcification (G) was on average 8 mmol CaCO₃ m⁻² d⁻¹. CO₂ fluxes generated by primary production, respiration, and calcification suggest that the study site was a potential atmospheric CO₂ sink of 15 mmol CO₂ m⁻² d⁻¹ at the time of measurement.

KEY WORDS: Ecosystem metabolism · Primary production · Respiration · Calcification · Calcareous macroalgae · *Corallina elongata* · pH · Total alkalinity · TA

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INTRODUCTION

The coastal zone is characterized by large spatial changes in ecosystem distribution and temporal changes in biological processes, which generate strong gradients of biogeochemical fluxes in space and time. The role of the coastal zone in the carbon cycling, as well as its trophic status, is still a matter of debate (Gattuso et al. 1998, Mackenzie et al. 2004, Duarte et al. 2005), partly due to large differences between proximal and distal parts of the coastal ocean and the lack of a typology, which would enable extension of the data from a local scale to regional and global scales (Gazeau et al. 2004). Marine macrophytes are major contributors to benthic photoautotrophy in many prox-

imal coastal ecosystems, so-called macrophyte-dominated ecosystems, which are among the most productive systems and also account for much of the respiration in coastal zones (Gattuso et al. 1998, Middelburg et al. 2005).

The organic carbon metabolism (photosynthesis and respiration) of seagrass communities has been examined since the introduction of metabolic studies in aquatic ecology, largely through the use of benthic chambers. In contrast to seagrasses—although net primary production data are available for several species of macroalgae (e.g. Littler 1980)—the organic carbon metabolism of macroalgal-dominated systems has only been examined in a relatively limited number of studies. In particular, very few data are available for

temperate climate regions, as pointed out in recent reviews addressing the European (Gazeau et al. 2004) and global (Gattuso et al. 1998, Middelburg et al. 2005) coastal zones.

High levels of gross primary production (GPP) in macroalgal-dominated marine communities were first reported in reef systems by Smith & Marsh (1973). Despite high levels of autotrophic activity, the net ecosystem metabolism (i.e. net community production, NCP), which is the difference between GPP and ecosystem respiration (R), appears nearly balanced in most studies.

Shallow-water ecosystems have also proved to be significant contributors to the global inorganic carbon cycle (Mackenzie et al. 2004, Muller-Karger et al. 2005). Approximately 25% of the global precipitation of calcium carbonate and 50% of CaCO_3 accumulation occur in coastal regions (Milliman 1993, Wollast 1998). It must, however, be noted that estimates of CaCO_3 production may be biased because a relatively large amount of data is available for coral reefs (reviewed by Gattuso et al. 1998), but comparatively little information is available on temperate systems.

Although calcification and photosynthesis both decrease the dissolved inorganic carbon (DIC) concentration, they displace the carbonate equilibrium in opposing directions, increasing and decreasing the partial pressure of CO_2 ($p\text{CO}_2$), respectively. A simple model was derived to predict the effect of the organic and inorganic metabolic activities on air–sea CO_2 fluxes (Gattuso et al. 1995). In addition, the precipitation of 1 mol of CaCO_3 decreases total alkalinity (TA)

by 2 mol. Thus, both net organic carbon metabolism and carbonate dynamics, hereafter collectively referred to as community metabolism, are important to study in order to address the role of coastal ecosystems in the C cycle. Despite a 13-fold increase in the publication rate of coastal biogeochemistry studies since 1971, important gaps in knowledge still exist (Gattuso et al. 2005). Indeed, carbonate dynamics in coastal ecosystems represent a very marginal proportion (<1%) of coastal biogeochemistry studies. Furthermore, ecosystems exhibiting high rates of primary production such as algal beds and seagrass meadows have been subject to comparatively little investigation (8%).

In temperate regions, such as in the Mediterranean Sea, calcareous macroalgae can thrive and dominate in a large range of habitats, from the mediolittoral to the deepest part of the photic layer (Canals & Ballesteros 1997). To the best of our knowledge, simultaneous measurements of organic and inorganic carbon metabolism at the ecosystem level have never been conducted in such an environment. The present study investigated the winter time community metabolism of a Mediterranean marine shallow ecosystem dominated by the *Corallina elongata* community (Ballesteros 1988), employing the pH-TA technique (Smith & Key 1975).

MATERIALS AND METHODS

Study site and community composition. The study site (Anse des Cuivres, NW Mediterranean Sea; Fig. 1) is a shallow cove (surface area: ca. 900 m²; average depth: 3 m) dominated by hard substratum (ca. <30% soft bottoms). It is relatively sheltered from the dominant NW winds and opens to the southern bay of Marseilles. The rocky substratum, including vertical walls, exhibits high surface cover by the *Corallina elongata* community, which is typical of the rocky NW Mediterranean infralittoral (Ballesteros 1988). *C. elongata* biomass ranged from 820 to 2544 g dry weight (DW) m⁻² (El Haïkali et al. 2004). Among the other conspicuous calcareous macroalgae are the articulated species *Jania corniculata* and *Amphiroa rubra* and the encrusting species *Mesophyllum alternans* and *Lithophyllum lichenoides*.

Field measurements. Meteorological and oceanographic measurements were conducted during the period 23–29 February 2000. A Davies Weather Monitor 2 station located on the roof of the laboratory (20 m from the study site) recorded wind speed and direction every 30 min. Surface irradiance (W m^{-2}) was measured using a LI-COR LI-190SA pyranometer sensor (400 to 800 nm), with a LI-185B interface. It was also expressed in

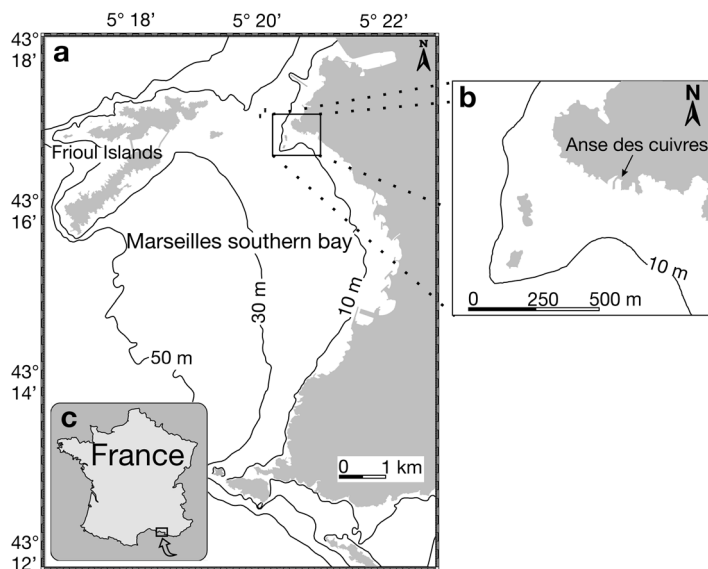


Fig. 1. Location map of the study site: (a) Marseilles southern bay, (b) Endoume coastal area, (c) Marseilles southern bay in relation to France

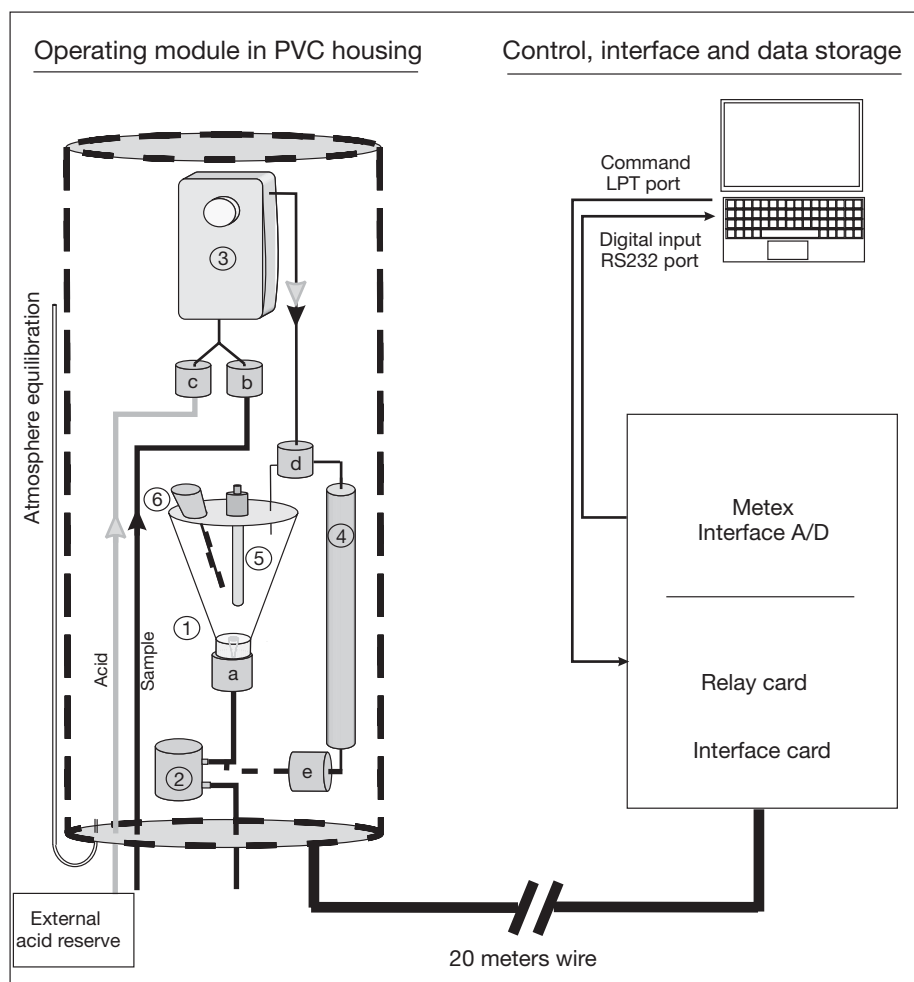


Fig. 2. Schematic representation of the automatic module for alkalinity measurement. See the 'Field measurements' section in 'Materials and methods' for more details

quantum units ($\mu\text{mol m}^{-2} \text{s}^{-1}$) using the approximate conversion factor according to Morel & Smith (1974). Seawater temperature (T) and salinity (S) were measured by a WTW LF197S cell immersed to 1.5 m depth. Irradiance, T , and S data were sampled every 50 s and averaged over 30 min. Current intensity and direction were measured every 5 min and averaged over 30 min until 26 February at 10:00 h, with an upward-directed Sontek ADP (acoustic Doppler profiler) dedicated to shallow-water deployment (20 cm high cells); the ADP was moored at 4.2 m depth at the opening of the Anse des Cuivres (Fig. 1).

The pH and TA were measured every hour from 01:00 to 23:00 h by an automatic module for alkalinity measurement (AMAM; Fig. 2) operating *in situ* in surface water. This unit sampled seawater (100 ml), and pH was first measured. Then, the sample was acidified, and TA was estimated from final-end-point potentiometric titration. The measurement module is con-

nected to the command box (power supply, electronic and relay cards) by a 20 m wire. At the start of each measurement, the polycarbonate conical 250 ml measuring cell (1 in Fig. 2), fitted on the bottom with a pump (2), behind a solenoid valve (a), is first drained of the previous acidified sample then rinsed twice with seawater. A precision electronic piston pump (contiburette $\mu\text{10MO laboCAT}$, 0.1% accuracy) (3) is used for seawater (100 ml, filtered on 125 μm) and acid (30 ml; ionic strength: 0.7) delivery, using 2 upstream independent tubes, each of them fitted with its own solenoid valve (b for seawater, c for acid). Injection is done at the top of the cell by a Teflon tube (1 mm inner diameter). Between the seawater and acid injection phases, all the piston pump tubes are purged and rinsed into an internal reservoir (4) used to collect residual volumes from injection circuit rinsing. An electric rotary stirrer (5) is used to homogenize the solution before pH measurement by a pH probe

(Hanna Instrument with built-in amplification circuits) (6) located at the top of the measuring cell. All liquid junctions are constructed in silicon tubing, and the 5 solenoid valves (a to e) enable or disable fluid circulation for sample or acid injection, cell rinsing and draining, and purging into the internal reservoir or its draining. Electrode voltage outputs are digitized by an 8 bit A/D (analog to digital) converter (Metex M 4650-CR, ± 0.05 mV accuracy, 0.02 mV resolution) and transmitted to a computer. The pH was recorded over 30 s at 1 Hz. The resulting means and SDs were saved.

Before initiation of the experiment, the linearity of the electrode was tested and calibration was performed using N-bromo-succinimide (N.B.S.) buffers (Merck; 4.01, 7.01, and 10.01 at 25°C). The reproducibility of the AMAM was determined on filtered surface seawater ($n = 17$). For pH and TA, uncertainties were ± 0.002 pH units and $\pm 3 \mu\text{mol kg}^{-1}$ (1 SD), respectively. TA values estimated with the AMAM instrument were within $6 \mu\text{mol kg}^{-1}$ of the determinations carried out in the laboratory using final-end-point titration. TA was normalized to a salinity of 38 ($\text{TA}_{38} = \text{TA} \times 38/S$) to remove the changes due to salinity variations.

Community productivity and calcification. Calcification and partitioning of organic and inorganic carbon metabolism were estimated using the alkalinity anomaly (pH-TA; Smith & Key 1975). DIC was calculated from pH and TA with the CO2SYS program (Lewis & Wallace 1998) using the Peng option for dissociation constants on the N.B.S. scale. Since instantaneous air-sea CO_2 fluxes were negligible ($< 0.2 \text{ mmol m}^{-2} \text{ h}^{-1}$) compared to ΔDIC , net primary production and calcification were computed as follows:

$$\text{npc} = -\Delta\text{DIC}\rho z - g \quad (1)$$

$$g = -\frac{1}{2} \Delta\text{TA} \rho z \quad (2)$$

where npc is net community production ($\text{mmol m}^{-2} \text{ h}^{-1}$), g is community calcification ($\text{mmol m}^{-2} \text{ h}^{-1}$), ρ is seawater density (kg m^{-3}), z is water depth (m) and ΔDIC and ΔTA are hourly changes in DIC and total alkalinity ($\text{mmol kg}^{-1} \text{ h}^{-1}$). By convention, lower case letters refer to hourly fluxes and capital letters refer to fluxes integrated over the day, night, or 24 h (Eqs. 1 to 4).

Computation of daily metabolic budgets. Net daily ecosystem primary production was calculated as described by Falter et al. (2001). Briefly, npc was integrated from sunrise to sunset to yield the net ecosystem production that occurs during the daylight hours ($\text{NCP}_{\text{light}}$) and from sunset to sunrise to yield the nighttime respiration in the dark (R_{dark}).

GPP and R were calculated as follows:

$$\text{GPP} = \text{NCP}_{\text{light}} + R_{\text{light}} \quad (3)$$

$$R = R_{\text{light}} + R_{\text{dark}} \quad (4)$$

where R_{light} is the community respiration rate in the light, assuming that hourly rates of light and dark respiration are similar. This neglects the possibility that R_{light} may be higher than R_{dark} , as shown in a large-scale coral reef mesocosm (Langdon et al. 2003).

Daily net calcification rate (G) was calculated by integrating the hourly rates of calcification over 24 h.

RESULTS

Environmental and hydrodynamic conditions

Total irradiance and its distribution during the course of the days were similar throughout the experiment with, for maximum irradiance levels (540 W m^{-2} , about $2485 \mu\text{mol photons m}^{-2} \text{ s}^{-1}$), an SD of $< 2\%$ of the mean ($n = 6$). Low to moderate ($1.6 \pm 1.4 \text{ m s}^{-1}$, $n = 322$) NW winds were dominant, and the wind speed was higher during the day than at night (2.1 vs. 1.2 m s^{-1} , t -test, $p < 0.0001$). A transient peak occurred on 25 February (6.6 m s^{-1}). Current velocity was low ($1.8 \pm 0.6 \text{ cm s}^{-1}$, $n = 3240$), and did not exhibit any circadian or vertical patterns. The third quartile of current intensity distribution was 2.2 cm s^{-1} , and maximum values were $< 4 \text{ cm s}^{-1}$, except in the surface layer during the wind event of 25 February. Temperature ($12.6 \pm 0.6^\circ\text{C}$) and salinity (37.76 ± 0.04) showed typical winter values for the area. As a result of the calm and sunny conditions, seawater temperature exhibited weak diurnal variations (0.7°C) and a significant linear trend against time ($+0.25^\circ\text{C d}^{-1}$, $r = 0.89$, $n = 322$).

CO₂ related parameter dynamics

Large pH and TA changes of, respectively, 0.243 and $126 \mu\text{mol kg}^{-1}$ were recorded over the study period. The pH exhibited regular and synchronous circadian changes from day to day, with minima around 08:00 h and maxima around 17:00 h (Fig. 3A). TA_{38} variations were noisier, but also displayed significant diurnal changes, with a slight increase trend from midnight to sunrise, a marked decrease ($-41 \pm 23 \mu\text{mol kg}^{-1}$, $n = 6$) between late morning (11:00 h) and mid-afternoon (15:00 h), and relatively stable values from 15 h to midnight (Fig. 3B). A significant trend against time was observed for TA_{38} ($-7 \mu\text{mol kg}^{-1} \text{ d}^{-1}$, $r = 0.46$, $n = 152$).

Primary production and calcification

Hourly npc showed diurnal changes (Fig. 4) and reached the highest values ($60 \pm 9 \text{ mmol C m}^{-2} \text{ h}^{-1}$) at solar noon (13:00 h local time). The compensation irra-

diance was lower in the morning than in the afternoon (40 vs. 280 W m^{-2} , i.e. 198 vs. 1293 $\mu\text{mol photons m}^{-2} \text{s}^{-1}$). The rate of respiration in the dark exhibited a consistent negative trend during the course of the night. It was highest 1 h after sunset ($-47 \text{ mmol m}^{-2} \text{h}^{-1}$) and decreased to significantly lower ($-13 \text{ mmol m}^{-2} \text{h}^{-1}$; z-test, $p < 0.0001$) and relatively constant levels during the second part of the night (01:00 to 07:00 h).

Calcification also showed a strong diurnal pattern (Fig. 4). Net calcification was observed with rates up to $24 \pm 10 \text{ mmol CaCO}_3 \text{ m}^{-2} \text{h}^{-1}$ for irradiance values

$>300 \text{ W m}^{-2}$ (about 1380 $\mu\text{mol photons m}^{-2} \text{s}^{-1}$). Net CaCO_3 dissolution prevailed below that threshold (average of $-3 \text{ mmol CaCO}_3 \text{ m}^{-2} \text{h}^{-1}$).

The mean community GPP and R were 519 ± 106 and $500 \pm 136 \text{ mmol C m}^{-2} \text{d}^{-1}$ ($\pm\text{SD}$), respectively. There was therefore a slightly positive NCP ($19 \text{ mmol C m}^{-2} \text{d}^{-1}$). Despite the strong variability of GPP (342 to 657 $\text{mmol C m}^{-2} \text{d}^{-1}$) and R (277 to 671 $\text{mmol C m}^{-2} \text{d}^{-1}$), the GPP: R ratio always remained close to 1 (1.1 ± 0.1 , $n = 6$). G varied greatly from one day to the next and ranged from -33 (net dissolution) to $+39$ (net calcifica-

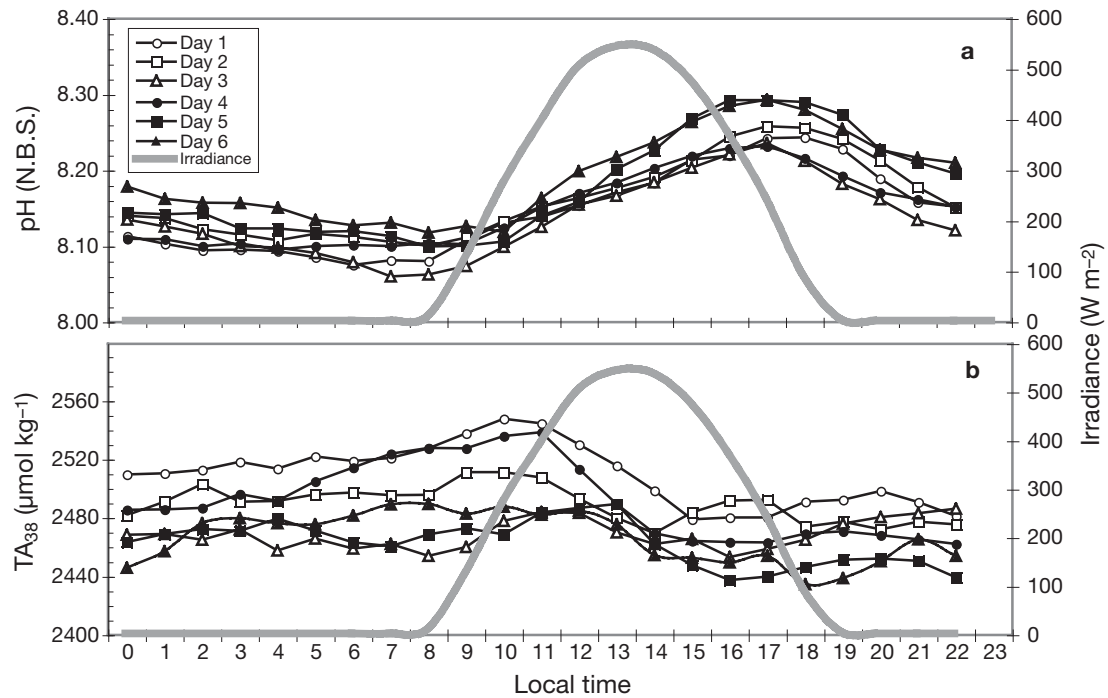


Fig. 3. (a) pH (using N-bromo-succinimide [N.B.S.] buffers 4.01, 7.01, and 10.01 at 25°C) and (b) TA_{38} (total alkalinity normalized to a salinity of 38) time series at the study site. Hourly mean irradiance levels also plotted

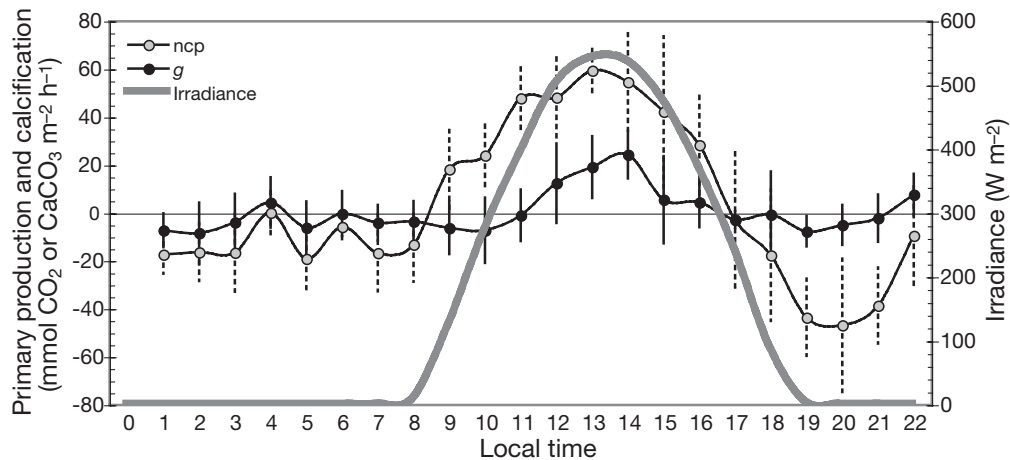


Fig. 4. Mean hourly rates of primary production and calcification over 6 d by the pH-TA technique. Mean hourly irradiance levels also plotted. Error bars: $\pm\text{SD}$

tion) $\text{mmol CaCO}_3 \text{ m}^{-2} \text{ d}^{-1}$. Its mean value for the study period was $8 \pm 32 \text{ mmol CaCO}_3 \text{ m}^{-2} \text{ d}^{-1}$. The mean G:GPP ratio was 0.015.

DISCUSSION

Ecosystem primary production and calcification

The current velocity was mostly $< 2 \text{ cm s}^{-1}$ during the first 3 d of the experiment; hence, the water mass was likely relatively still. No current data are available for the remaining 3 d, but it is reasonable to assume that requirements for the standing water approach were met during the whole study, as wind speed was not significantly greater during the subsequent 3 d (t -test, $p = 0.36$, $n = 144$). The assumption of low water exchange between the cove and the open water is strongly reinforced by a simple budget. The expected decrease in TA in the cove is about $5.2 \mu\text{mol kg}^{-1} \text{ d}^{-1}$ ($8000 \mu\text{mol CaCO}_3 \text{ m}^{-2} \text{ d}^{-1}/\rho z$, with $\rho = 1028.6 \text{ kg m}^{-3}$ and $z = 3 \text{ m}$). This value reflects very well the significant negative trend of $7 \mu\text{mol kg}^{-1} \text{ d}^{-1}$ observed during the experiment.

The hourly time series of net primary production and calcification during 6 successive days provides insight on short-term changes in the diurnal cycle and reveals original dynamics.

The gpp:r ratio decreases during the afternoon, as shown by the high afternoon compensation irradiance (280 W m^{-2} , about $1293 \mu\text{mol photons m}^{-2} \text{ s}^{-1}$; data not shown) and the previous ncp decrease. As the system studied harbors photophilic benthic macroalgae, an increase in R , rather than a decline in photosynthesis resulting from photoinhibition, is likely responsible for the afternoon depression. From a simple graphic approach based on the CO_2 data (Fig. 4), a 2- to 3-fold increase can be inferred between morning and afternoon R rates.

The rates of R_{dark} were also quite variable (Fig. 4), decreasing 3-fold to relatively constant values during the first part of the night. Similar night-time R dynamics were also observed in early water-flow metabolic studies, but neglected in the computation of daily metabolic parameters by the diurnal curve method, since the R baseline was '...drawn through the night time R rates after they have achieved a constant value' (Marsh & Smith 1978). Langdon (1993) also reported a decrease in R rate through the night in large phytoplankton incubation chambers. More recently, during a field enclosure study of macroalgal-dominated coral reef community metabolism, Gattuso et al. (1997) observed considerable scatter (ca. 3-fold) of night-time R rates. Such variability was not observed in a large coral reef mesocosm (Falter et al. 2001, Langdon et al. 2003).

It has been established that R_{light} often exceeds R_{dark} in micro- and macroalgae as well as in zooxanthellae *in hospite*. Employing a new ^{14}C isotope dilution method in the Biosphere 2 coral reef mesocosm largely dominated by macroalgae, Langdon et al. (2003) showed that R_{light} was twice as high as R_{dark} . Differences between R_{dark} and R_{light} rates have direct consequences for the conventional estimate of GPP and R , which might be underestimated by as much as 40% (Langdon et al. 2003), but not for the estimate of NCP.

However, rather than strictly opposing values for daylight and darkness periods, our observations suggest large (up to 3-fold) variations in respiratory activity under both daylight and dark conditions, with significantly higher R rates during the afternoon and the first part of the night. This dynamic relationship supports the hypotheses that a photosynthetically produced pool of dissolved organic matter is oxidized and short-term coupling occurs between photosynthesis and R (Falter et al. 2001).

Community net calcification (g) also exhibited marked diurnal variations, with highest CaCO_3 deposition rates during the period of maximum irradiance. There were numerous CaCO_3 depositing organisms at the study site, conspicuously macroalgae, among which *Corallina elongata* exhibited both the highest biomass (820 to 2544 g DW m^{-2}) and the highest calcification rate, with a light to dark calcification ratio of 3.6 (El Haïkali et al. 2004). Calcification and photosynthesis are directly linked in organisms showing light-enhanced calcification. However, unlike ncp, g was positive (net CaCO_3 deposition) for irradiances $> 300 \text{ W m}^{-2}$ ($1380 \mu\text{mol photons m}^{-2} \text{ s}^{-1}$). Differences in calcification and primary production dynamics or in the light response curve are not surprising at the system level (see for example Barnes & Devereux 1984, Gattuso et al. 1996, 1997), but our observations are in contrast to previous results for coral reefs, where net CaCO_3 deposition is sometimes measured both during the day and at night (Gattuso et al. 1996) and, in a macroalgal-dominated coral reef community in Moorea (French Polynesia), where net CaCO_3 deposition was measured at irradiances $> 500 \mu\text{mol photons m}^{-2} \text{ s}^{-1}$ (Gattuso et al. 1997). The alkalinity anomaly technique provides access to net calcification only (i.e. the CaCO_3 precipitation/dissolution balance). Nevertheless, night-time negative g values (when macroalgae still deposit CaCO_3)—and apparent high irradiance requirements for positive g values—suggest high CaCO_3 dissolution rates within the system, as also indicated by the comparison of our data with direct estimate of *C. elongata* calcification rates (see below).

From a 3 yr study using the O_2 method in the Biosphere 2 mesocosm, Falter et al. (2001) made the sur-

prising observation that the variance of NCP at a time scale of <1 wk was significantly larger than the variance at all other time scales. Our observations suggest that the short-term variability of primary production and calcification can be even greater in natural open systems and support their conclusions on the need of longer time integration than usually conducted in order to estimate community metabolism.

However, due to the time variations in biotic and abiotic factors such as community composition, nutrients, temperature, daylength, and irradiance, more data covering seasonal variations are needed before attempting an annual estimate of community metabolism.

Biologically induced CO₂ fluxes

In a calcifying system, several processes occur during the day, either increasing (calcification and respiration) or decreasing (dissolution of CaCO₃ and photosynthesis) the seawater pCO₂ level. When GPP and *R* are nearly balanced, increased seawater pCO₂ induced by calcification can dominate biologically induced CO₂ fluxes, which explains why most reef flats behave as sources of CO₂ (Gattuso et al. 1999). Applying the model derived by Gattuso et al. (1995) to the average metabolic rates estimated in this work (NCP = 20.8 and *G* = 8.4 mmol m⁻² d⁻¹), the Anse des Cuivres was a potential net sink of atmospheric CO₂ of 15 mmol CO₂ m⁻² d⁻¹ at the time of the experiment. This is the first estimate of potential fluxes for a shallow temperate system dominated by calcareous macroalgae. Further metabolic investigations at the relevant time scales are required to determine the annual metabolic balance and its net influence on the pCO₂ of surrounding waters, which is all the more important when the water volume is low and the water residence time is high. However, despite very large gross metabolic rates, the influence of macrophyte-based ecosystems on air-sea CO₂ fluxes in temperate coastal zones may not be prominent. Indeed, shallow (7 m) *Posidonia oceanica*-vegetated sediments in Palma Bay (western Mediterranean Sea) were found to be a potential net CO₂ sink of only 1 mmol CO₂ m⁻² d⁻¹, as a mean on the annual scale due to slight autotrophy and net CaCO₃ deposition (Barrón et al. 2006). Furthermore, the solubility coefficient of CO₂ is strongly dependant on temperature, which is the driving factor in the seasonal variability of air-sea CO₂ fluxes in the western Mediterranean Sea. As a result, Gazeau et al. (2005) suggest that *P. oceanica* meadows could only decrease the summer-time CO₂ emission from the Mediterranean continental shelf by a modest 2.5%.

Comparison with other macroalgal-dominated ecosystems and significance for the coastal budget

Considering an average microphytobenthic GPP of 16 mmol C m⁻² d⁻¹ for the 0 to 5 m depth range in the Mediterranean Sea (Gazeau et al. 2004) and a planktonic primary production of about 8 mmol C m⁻² d⁻¹ in winter (Lefèvre et al. 1997), macroalgae and epiphytic microalgae must be the main contributors to GPP at the study site. Despite a community structure based on calcareous macrophytes (which are low productivity species according to the functional form model of Littler 1980), the GPP of Anse des Cuivres in winter is slightly greater than the average value for macrophyte-dominated systems (519 vs. 477 mmol C m⁻² d⁻¹; Gattuso et al. 1998) due to high biomass. Reports on GPP and *R* of macrophyte-dominated benthic systems span over >1 order of magnitude (Table 1). This variability is due to the absolute and relative surface area covered by the major communities and the type of substrate, which seems to overwhelm the effect of environmental variables and the dominant species.

R is close to GPP and stands close to the average rate reported by Middelburg et al. (2005) for macroalgal-dominated benthic systems (500 vs. 483 mmol C m⁻² d⁻¹).

In agreement with most reports on the metabolism of macrophyte-dominated benthic systems, the Anse des Cuivres was autotrophic at the time of measurement. The GPP:*R* ratio was nearly balanced, and NCP (19 mmol m⁻² d⁻¹) was ca. 5% of GPP.

Reports on calcification rates primarily arise from coral reef ecosystems, where absolute and relative surface coverage as well as community structure explain strong differences in calcification rates at small spatial scales (Kinsey 1985). High levels of CaCO₃ deposition (66 to 110 mmol m⁻² d⁻¹) were measured in algal pavement zones (Kinsey 1985), while slight net CaCO₃ dissolution (-0.8 mmol m⁻² d⁻¹) prevailed at fringing reefs dominated by macroalgae (Gattuso et al. 1997). Our estimate indicates moderate calcification rates in the system studied (8 mmol CaCO₃ m⁻² d⁻¹), with a *G*:GPP ratio (0.02) 1 order of magnitude lower than at coral reefs (0.2 to 0.3; Langdon et al. 2000). This estimate is in good agreement with the total carbonate production of photophilic algae on the Mallorca-Menorca shelf (NW Mediterranean Sea), estimated to be 8 mmol m⁻² d⁻¹ as the mean value of an annual biometric survey (Canals & Ballesteros 1997), and compares well with a net estimate over Palma Bay (Gazeau et al. 2005), largely dominated by *Posidonia oceanica*.

The results presented above, and their comparison with benthic and pelagic metabolic rates in the western Mediterranean Sea, provide new insights to the potential significance of rocky-shore shallow communities to the coastal C budget.

Table 1. Range of metabolic data for macrophyte (mostly macroalgae)-dominated ecosystems. NWM: northwestern Mediterranean, GPP: gross primary production, *R*: respiration, NCP: net community production, *G*: daily net calcification. Units — GPP, *R* and ncp: mmol CO₂ m⁻² d⁻¹; *G*: mmol CaCO₃ m⁻² d⁻¹

Type or location	GPP	<i>R</i>	NCP	<i>G</i>	Source
Algal turf in coral reefs	83–500				Larkum (1983)
Coral reef algal pavement zone	167–583	42–558		66–110	Kinsey (1985)
Macroalgal-dominated coral reef community	248	240		–0.8	Gattuso et al. (1997)
Macrophyte-dominated ecosystems (review)	74–2762	16–2493	–252 to 890		Gattuso et al. (1998)
Temperate lagoon, seagrasses	56–366	103–381			Ziegler & Benner (1998)
Macroalgal-dominated coral community	25–290	32–272	–112 to 61		Falter et al. (2001)
Macroalgal-dominated communities (review)		16–1271			Middelburg et al. (2005)
Palma Bay, NWM			10–20	10	Gazeau et al. (2005)
<i>Posidonia oceanica</i> -vegetated sediments, NWM	25–173	18–142	–25 to 89	1.5	Barrón et al. (2006)
Calcareous-macrophyte-dominated, NWM	519	500	19	8	Present study

First, their primary production is 1 to 2 orders of magnitude larger than that of soft-bottom microphytobenthic (2.5 to 16 mmol C m⁻² d⁻¹) or pelagic (20 mmol C m⁻² d⁻¹) communities (Gazeau et al. 2004). As a result, their contribution to the C fluxes is comparatively larger in narrow continental shelf areas, which characterizes most of the Mediterranean Sea. Nonetheless, the balance between primary production and *R* in macrophyte-dominated ecosystems (Gattuso et al. 1998, Gazeau et al. 2005, Barrón et al. 2006) results in NCP values that do not differ much from those of other systems. For example, NCP is 16 mmol m⁻² d⁻¹ for the Gulf of Lions (one of the so-called 'preferred LOICZ budget sites' in the Mediterranean Sea). On the other hand, benthic NCP over the 0 to 200 m depth range in the Mediterranean Sea was estimated to be –19.5 mmol C m⁻² d⁻¹ (Gazeau et al. 2004).

The shallow calcifying communities play a potentially significant role in the production of CaCO₃ particles (Canals & Ballesteros 1997). El Haïkali et al. (2004) have estimated that the calcification rate of *Corallina elongata*, the dominant species in the cove investigated in the present study, to be 140 mmol CaCO₃ m⁻² d⁻¹. This value is much higher than the mean rate of *G* estimated in the present study (8 mmol m⁻² d⁻¹). The discrepancy could be due to larger space integration, hence lower areal biomass and calcification rates and higher rates of calcification in seasons other than winter, but also reflects strong CaCO₃ dissolution within the system. However, due to the ubiquitous presence of coralline algae in the Mediterranean Sea, calcification rates, even if modest compared to other systems, have the potential to significantly influence the regional carbon balance.

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