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1 The application of DFRC method for the analysis of carbohydrates in a peat bog:

- 2 validation and comparison with conventional chemical and thermochemical degradation
- 3 techniques
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- 10 Abstract

Traditional acid hydrolysis is the main method used for carbohydrates characterization in 11 12 soils and sediments. It has been used to cleave the glycosidic bond, yielding monomers that compose polysaccharides of the non-cellulosic pool (free, ligno-cellulosic and hemicellulosic 13 carbohydrates). In this study, the widely used chemical and thermochemical degradation 14 15 procedures have been compared with the derivatisation followed by reductive cleavage (DFRC) method. This method presented efficiency in specifically cleaving the aryl ether bonds, yielding 16 for sugar moieties linked to aromatic structures (ligno-cellulosic fraction). Once applied for peat 17 samples, some discrepancies were observed. DFRC method revealed some information about the 18 origin of carbohydrates that was hidden from the traditional chemical and thermochemical 19 degradation methods. Several patterns of carbohydrates have been obtained with DFRC 20 indicating difference in the origin and/or degradation rates between monosaccharides. For 21 xylose, it showed an increase in the catotelm (deepest layer of the peatland), indicating an 22 angiosperm vegetation input at the early stage of the peatland formation. Ribose, commonly 23 known as a microbial indicator, showed the same profile as for hexoses and deoxyhexoses, with 24 25 acid hydrolysis. The difference in the profile yielded in the case of DFRC method, revealed its microbial input. These information were hindered from acid hydrolysis and thermochemolysis, 26 27 as the first showed nearly the same profile for all monosaccharides and the second one didn't have the capacity to analyze arabinose and ribose. 28

Generally, higher similarity was obtained between DFRC and acid hydrolysis than between DFRC and thermochemolysis. This is due to the difference of mechanisms involved in chemolytic and thermochemolytic methods. For the first group, both methods shared acid conditions and milder temperature conditions if compared to the second group. In addition thermochemolysis is performed under alkaline conditions.

34 **1. Introduction**

Northern peatlands are estimated to store up to a third of the terrestrial carbon as peat (Vitt et al., 2000); the continental C as peat is estimated for 20% while covering up only 3% of the Earth's surface (Gorham, 1991). These very complex ecosystems impact both carbon cycle and water at local as well as global scale. Since peat is progenitor of coal, a better understanding of the molecular fingerprint of different pools in peats will improve our insight in the formationof coal.

Carbohydrates are the major pool of terrestrial vegetation and represent the main form of 41 photosynthetically assimilated organic matter (OM) in the biosphere. Understanding 42 carbohydrates dynamics in peatlands will allow a better understanding of coal formation. Given 43 their ubiquitous character, carbohydrates potentially represents a powerful tool for elucidating 44 processes, sources and diagenetic pathways of biologically important OM in sediments (Jia el al., 45 2008). Although in most settings (soils and sediments) carbohydrates composition have shown 46 47 high sensitivity to diagenetic alterations, the OM-rich character of peat deposits and the potential for preservation below the acrotelm makes *ombrotrophic* peatlands a suitable environment for 48 also exploring the potential of carbohydrates as paleovegetation proxies (Bourdon et al., 2000; 49 Comont et al., 2006; Delarue et al., 2011; Estournel-Pelardy et al., 2013; Younes et al., 2017). 50

51 The analysis of OM at the molecular level commonly requires several techniques and each technique targets all patterns of one of the three major pools. Most studies dealing with 52 lignin analysis were conducted using cupric oxide under alkaline conditions (CuO-NaOH) 53 54 (Hedges and Ertel, 1982). For lipid fraction analysis, various organic solvents have been used 55 (Bligh and Dyer, 1959). For carbohydrates analysis in soils and sediments, it is mostly performed with acid hydrolysis (Bourdon et al., 2000). The high heat and concentration of acid used, 56 contributes to the cleavage of all glycosidic bonds (relative to non-cellulosic/cellulosic fractions) 57 (Younes and Grasset, 2018). Derivatisation Followed by Reductive Cleavage (DFRC) method 58 showed a high efficiency for cleaving α and β -aryl ethers and is therefore suitable for ligno-59 cellulose materials. Therefore, all studies conducted for the DFRC method were done for lignin 60 subunits analysis (Del Río et al., 2007; Grasset et al., 2010; Lu and Ralph, 1998; Matsui and 61 62 Ohira, 2013; Peng et al., 1999).

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The only attempt to use this technique in a geochemistry study was done by Grasset el al. (2010). The authors applied this technique to characterize lignin in gymnosperm wood, and in soils and sediments formed from degraded gymnosperm wood, in order to assess the supply of terrestrial OM to marine sediments. The purpose of applying this method is a need for finding efficient and more selective method for lignin α - and β -aryl ether cleavage. It was evident that the overall lignin yield and the compound ratios used as plant source proxies have been found to be considerably different from those yielded from the mostly used lignin degradation techniques.

72 For our knowledge, this is the first attempt to analyze the carbohydrate pool in peat samples with the DFRC method. One of the advantages of using peatland samples to conduct this 73 comparison is the bilayer character of this ecosystem. The upper half is the "living" part where 74 fresh OM input and microbial activity occur. The bottom half is the "dead" part where 75 decomposed OM is preserved making the peats a reference for paleoenvironmental 76 77 interpretations (see Younes et al. (2017)). In this study, the DFRC method was applied for 78 characterizing carbohydrates in a French peatland. In order to reveal the extent of DFRC method, carbohydrates yielded moieties were compared with lignin input from CuO-NaOH oxidation and 79 80 the thermally assisted hydrolysis and methylation (THM) method. In case of discrepancy in carbohydrates profile, this will give more insight into the molecular aspect of OM and the 81

occurrence of a carbohydrate fraction that was hindered from the acid hydrolysis and THM
 method. PCA was used to compare between the employed chemical and thermochemical
 degradation methods.

85

86 **2. Materials and methods**

87

88 2.1. Sampling

Samples were collected from a peatland site in the Limousin, France. Settings of the studied
site was developed in a previous study by Younes et al. (2017). Samples of a 1m peat core were
were divided into 4 cm thick slices and then freeze-dried.

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95

94 2.2. Derivatization followed by reductive cleavage (DFRC)

96 The samples were subjected to DFRC analysis according to the procedure of Lu and Ralph, (1998), as dried peat samples were treated with 5 mL of acetyl bromide/acetic acid (8% v/v), 97 stirred at 50 °C for 2 h, then evaporated under reduced pressure until dry. Zinc powder (500 mg) 98 and 25 mL of dioxane/acetic acid/water (5 :4 :1, v/v/v) were added and stirred for 30 min at 50 99 °C. Saturated aqueous ammonium chloride (50 mL) were added, whereupon the solution was 100 acidified to pH 3 with HCl. The mixture was then extracted with CHCl₃ 2 times (total 50 mL). 101 The chloroform layer was dried over magnesium sulfate, then evaporated. Degradation products 102 were acetylated with acetic anhydride (2 mL), acetic acid (0.1 mL) and 1-methylimidazole (0.1 103 mL) at room temperature for 10 min and analyzed by GC (Shimadzu GC-2010) with an FID and 104 by GC-MS and 50 µL of nanodecanoic acid methyl ester was added as an internal standard. 105

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107 2.3. Methods used for validation and comparison

108 2.3.1. CuO- NaOH oxidation for lignin characterization

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The release of lignin monomers was performed by alkaline CuO-NaOH oxidation. This is the classical chemical degradation method used for lignin characterization. About 10 mg of the dried peat samples was oxidized for 2 h at 170 °C in sealed Teflon vials, following the procedure developed by Hedges and Ertel, (1982). CuO-NaOH oxidation yields a suite of one ring phenols (H, *p*-hydroxyphenol; V, vanillyl; S, syringyl and C, cinnamyl; Table. 1) with their aldehyde, ketone and acid counterparts (for further information see Younes et al. (2017) and Younes and Grasset, (2018)).

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118 2.3.2. Acid hydrolysis

Acid hydrolysis was carried out with HCl for 200-300 mg of the dried peat samples at 100 °C for 3 h, the obtained monosaccharides (check Table. 1 for symbols) was derivatized into alditol acetates following the procedure of Rumpel and Dignac, (2006) and well developed in a previous study on the same site by Younes et al. (2017).

125 2.3.3. Thermally Assisted Hydrolysis and Methylation (THM) method

127 The THM method is based on the offline procedure developed by Grasset and Amblès, 128 (1998). About 50 mg of the dried peat samples were mixed with 4 ml of tetramethylammonium 129 hydroxide (TMAH) and heated at 400 °C (for further information see Younes and Grasset, 130 (2018)). THM method yields a suite of monosaccharides and one ring phenols (check Table. 1 131 for symbols).

- 132 2.4.Gas Chromatography (GC) analysis
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GC measurements were performed with a SHIMADZU GC-2010 Plus gas chromatograph equipped with a flame ionization detector at 300 °C. Separation of the compounds was achieved with a 30 m fused silica capillary column (Supelco Equity 5%, 0.25 mm internal diameter, 0.25 μ m film thickness), and helium as carrier gas. The oven temperature was programmed from 60 to 300 °C at 5 °C min⁻¹ and held for 15 min.

139140 2.5. Data normalization

141 The relative contribution of monosaccharide with depth were scaled and centered 142 following the formula below:

Normalized
$$P_i(\%) = \frac{P_i - P_{min}(\%)}{P_{max}(\%) - P_{min}(\%)}$$

143

144 Where: 145 P_{min} (%) = smallest Pi (%) for each monosaccharide along peatland depth records,

145 P_{min} (%) = smallest Pi (%) for each monosaccharide along peatland depth records, 146 P_{max} (%) = biggest Pi (%) for each monosaccharide along peatland depth records.

- 147
- 148 2.6. PCA

PCA is used in order to reduce dimensionality of dataset by producing linear combinations of original variables. PCA breaks down the matrix of initial data, X, to express them as a least-square model (Geladi et al., 2003):

 $X = A \times F + U$ 152 Where, X is the original data matrix, A is the matrix of loadings of the original 153 variables in the new reduced space, F is the matrix of scores of objects or samples and U is the 154 matrix of residuals. PCA aims to reduce the dimensionality of datasets containing many 155 interrelated variables, while retaining as much as possible of the systematic variation in the 156 original data (Joliffe and Morgan, 1992). In the analysis, the original variables are replaced by a 157 few principal components (PCs) variables. The PCs are totally independent with each other and 158 ordered so that PC₁ contains most of the original variation and PC₂ contains the second largest 159 amounts, etc. PCA was carried out using XLSTAT 2014.5 software. The peat depth samples 160 were considered as the observations and normalized data of monosaccharide and lignin as the 161 variables. 162

163

164 **3. Results and Discussion**

165 *3.1. Validation of the DFRC method for carbohydrates molecular characterization :*

166 In order to validate the DFRC method as an efficient technique for the molecular 167 characterization of carbohydrates, the relative abundance of total carbohydrates yielded with 168 DFRC has been compared to lignin content yielded by CuO-NaOH oxidation and THM method in one hand and to the total carbohydrates yielded with HCl hydrolysis and THM method inanother hand (Fig. 1).

171 3.1.1. Carboydrates vs. Lignin input

Contrary to monosaccharides, lignin biomarkers provide more reliable proxies for vascular 172 plant input (Filley et al., 2006; Wysocki et al., 2008; Mason et al., 2009; Abbott et al., 2013; 173 Younes and Grasset, 2018). The relative abundance of carbohydrates can be examined using the 174 ratio of carbohydrates yield (mg carbohydrate / 1 g OC; Jia et al., 2008) at a given depth in the 175 peat core. Fig. 1 presents bi-dimensional plots of total carbohydrates yielded with DFRC, HCl 176 hydrolysis and THM method as a function of "A" proxies, yielded by CuO-NaOH oxidation and 177 THM method. Lignin proxies were used in Younes and Grasset, (2018) as described by Filley et 178 al. (2006) with all phenols normalized to 1 g OC. Lambda $\Lambda_{THM} = G_4 + G_5 + G_6 + S_4 + S_5 + S_6 + S_6$ 179 $G_{18} + P_{18}$ and $\Lambda_{CuO-NaOH} = S + V + C$, indicates the relative lignin yielded by THM method and 180 CuO-NaOH oxidation, respectively. 181

Because THM primarily cleaves β -O-4 bonds and CuO-NaOH oxidation presumably cleaves β -O-4 as well as some C-C bond, the latter accesses a larger part of the unaltered non-oxidized lignin structure (Filley et al., 2006; Wysocki et al., 2008; Younes and Grasset, 2018).

185 Following Fig. 1, high correlation were obtained between input of carbohydrates yielded by DFRC method and HCl hydrolysis, from one side, and lignin yielded by CuO-NaOH oxidation 186 from the other side (67.45% and 69.57% for DFRC method and HCl hydrolysis, respectively). 187 These trends indicate the high correlation of hemicellulosic and ligno-cellulosic fraction along 188 the peat core. The low correlation between the input of carbohydrates yielded by DFRC and HCl 189 hydrolysis from one side, and lignin yielded by THM from another side (4.56% and 2.43% for 190 DFRC and HCl hydrolysis, respectively) indicates that the ligno-cellulosic fraction of the 191 investigated peat core is bonded to the intact non-oxidized lignin that is most likely revealed by 192 CuO-NaOH oxidation. Surprisingly, even the total carbohydrates yielded with THM method 193 194 showed a low correlation with Λ_{THM} . Its higher correlation with carbohydrates yielded with DFRC indicates that the DFRC method is a more reliable technique for carbohydrates analysis as 195 it reveals both the free and ligno-cellulosic fraction rather than only the free and terminal 196 carbohydrates, in the case of THM method. This is due to a higher efficiency of ether cleavage 197 under acidic conditions (Acetic acid used) in case of DFRC than basic conditions (pkb of TMAH 198 199 = 4.2) in case of THM (Wade, 2012).

200 3.1.2. Relative carbohydrate abundance: DFRC vs. Acid Hydrolysis & THM

The total non-cellulose sugar content varied from 6 mg/g at the top to 1 mg/g at the bottom of 201 the peat core, in the case of DFRC method. For HCl hydrolysis and THM method, it varied from 202 26 mg/g at the top to 4 mg/g at the bottom of the peat core and from 227 mg/g in the acrotelm to 203 29 mg/g at the bottom of the peat core, respectively (see Supplementary Information). The 204 difference in the yielded amounts shows the capacity of HCl hydrolysis to target the whole non-205 cellulose fraction (free sugars, sugars bonded to polyphenols and hemicellulose); as for DFRC 206 method, it targets only a specific pool of free sugars and sugars bonded to polyphenols. For THM 207 method, it has showed the capacity of targeting free and terminal carbohydrates (Younes and 208

Grasset, 2018; Estournel-Pelardy et al., 2011) which is more likely a smaller pool compared to
hemicellulose and ligno-cellulose. The higher abundance yielded for THM method is related to
the semi-quantitative approach used in this case. This was not the case of DFRC method and HCl
hydrolysis, as full quantitative approach is used (response factor of each monosaccharide is
identified with calibration curves).

Fig. 2 shows that depth amounts of the nine investigated monosaccharides was higher in the 214 acrotelm (0 to -20 cm) than the mesotelm (-20 to -50 cm) and catotelm (-50 to -100 cm), for all 215 three methods, indicating degradation of carbohydrates with depth (Jia et al., 2008). High 216 correlation between DFRC method and HCl hydrolysis has been obtained (between 65% and 217 87%, Fig. 2). This shows a high consistency between the two methods and proves the findings of 218 the comparison between carbohydrates amounts and lignin input (Fig. 1) as hemicellulosic and 219 ligno-cellulosic pools occurs in a synchronous fashion in the investigated peat core. Lower 220 correlation has been obtained between DFRC and THM methods (between 32% and 41%, Fig. 221 2). Even know that these methods share the free sugar pools, but this pool is more likely less 222 frequent in a terrigenous sediment as peatland, if compared with the occurrence of ligno-223 224 cellulose and hemicellulose (Clymo, 1991).

225 More detailed investigations revealed several patterns. For acid hydrolysis, two main patterns 226 of variation in concentration of each group of monosaccharides were revealed. The first composed of Ara)_{HCl} and Xyl)_{HCl}, showed a rapid decrease from the first depth of the acrotelm (0 227 to -20 cm), and concentration was approximatively constant down core. The second pattern 228 composed of Deoxyhexoses (Rha + Fuc)_{HCl}), Rib)_{HCl} and hexoses (Glu + Man)_{HCl} and Gal)_{HCl}) 229 showed a comparatively slower decline down core (for further details see Younes et al. (2017)). 230 For THM method, the highest input was observed at -20 cm, indicating a free and/or terminal 231 carbohydrate occurrence. The interface between the mesotelm and the catotelm (around -50 cm) 232 showed an increase in non-cellulosic carbohydrates. This indicates free carbohydrates input at 233 the interface of these ecological layers and a probable microbial deposition during peatland 234 accumulation. In the catotelm, the higher input of Xyl)_{THM} indicates a past vegetation changes 235 and/or past humification processes (Cowie and Hedges, 1984), due to the alkali attack of TMAH 236 (for further details see Younes and Grasset, 2018). 237

For DFRC method, different monosaccharides showed a significant difference and the 238 decreasing trend was skewed by small increases at particular positions. For pentose pattern 239 (Ara)_{DFRC} and Xyl)_{DFRC}), it showed a different decreasing profile, Ara significantly decreases 240 241 along depth as for Xyl it showed a more stable input in the acrotelm and an increase at the 242 interface between mesotelm and catotelm and at -70 cm in the catotelm. These trends indicate some difference in the origin between pentoses and/or degradation rate that was hindered by HCl 243 hydrolysis (same profile for Xyl)_{HCl} and Ara)_{HCl}; Fig. 2) and THM (no detection of Ara)_{THM}; Fig. 244 2). For the acrotelm and mesotelm, the higher decrease of Ara)_{DFRC} from the very first depth 245 records indicates a higher degradation rate, once compared with Xyl)_{DFRC}. In the catotelm, the 246 increase of Xyl)_{DFRC}, at particular depths, indicates the preservation of a ligno-cellulosic fraction 247 downcore and an input of vascular vegetation. Several studies have shown a higher input of Xyl 248 in angiosperms than gymnosperms (Wicks et al., 1991). In our case, the angiosperm input in the 249

first stages of the Sagnes peatland formation was shown by an S/V value of 1 by lignin analysis 250 (Younes et al., 2017). These trends indicate that differential degradation between 251 monosaccharides occurred and suggests that Ara and Xyl are components of polysaccharides that 252 are more degradable than other carbohydrates. Differences in degradation among carbohydrates 253 have been observed in several investigations (Jia el al., 2008). Thus the preferential degradation 254 of Ara and Xyl, compared to deoxyhexoses and hexoses, suggests that these monosaccharides 255 were in a less refractory form. The difference of depth profile, in the case of DFRC method, and 256 257 the absence of increases in the case of Ara indicates that this monosaccharide presented a higher 258 degradation rate, in comparison with Xyl.

For Deoxyhexoses (Rha)_{DFRC} and Fuc)_{DFRC}) and hexoses (Gal)_{DFRC} and Glu + Man)_{DFRC}), it showed approximatively the same profile (Fig. 2). The highest input was around -20 cm at the interface between acrotelm and mesotelm. Unlike the decreasing trend in the case of HCl hydrolysis, monosaccharides of this pattern showed their highest input in the mesotelm. This indicates most likely their belonging to the same origin.

264 *3.2.Geochemistry of Carbohydrates*

265 *3.2.1.* Source identification

266 Carbohydrate moieties can come from different plant and microbial sources. Therefore, their compositional patterns can be useful to differentiate between terrestrial and microbial origin, as 267 well as some specific tissue or type of organisms within each of these categories (Cowie and 268 Hedges, 1984; Popper and Fry, 2003; Comont et al., 2006; Jia et al., 2008; Delarue et al., 2011; 269 Younes et al., 2017; Younes and Grasset, 2018). Carbohydrate monomers have been investigated 270 271 to a lesser extent than lignin moieties. This is due to the rapid degradation of carbohydrates and their incorporation in the so called "humic fraction" (Estournel-Pelardy et al., 2013; Younes and 272 Grasset, 2017; Younes and Grasset, 2018). 273

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Despite the limited number of combinations that can be done with carbohydrate moieties, the 275 variation in the compositions of hemicelluloses from one species to another can provide a plant 276 source signature (Comont et al., 2006). Xyl and Ara are indicators of vascular plant input (Cowie 277 and Hedges, 1984). Popper and Fry (2003) mentioned the occurrence of high proportions of Gal, 278 Rha and Man in the hemicelluloses of bryophyte primary cell walls. Commont et al. (2006) 279 showed that these monosaccharides are indicators of moss in ombrotrophic peat bogs. 280 Furthermore, Fuc and Rha could originate from Sphagnum species, given the work of Jia el al. 281 (2008). Other studies have attributed the contribution of these monosaccharides to microbial 282 biomass (Cowie and Hedges, 1984). In our case, deoxyhexoses (Rha + Fuc)_{DFRC} and hexoses 283 (Gal)_{DFRC} and Glu + Man)_{DFRC}) showed a similar pattern between them and a different one with 284 Rib)_{DFRC} (Fig. 2). The latter is considered to have a microbial origin (Jia el al., 2008). Delarue et 285 al. (2011) identified Rib as marker of bacterial and/or fungal activity in peatlands. Following the 286 present data and given the previous studies, hexoses and deoxyhexoses are considered as a 287 Sphagnum indicator and Rib a microbial indicator. 288

289 3.2.2. Carbohydrate proxies :

In order to assess the origin of carbohydrates, two ratios were commonly employed for monosaccharides (Amelung el al., 1996; Nierop et al., 2001; Rumpel and Dignac, 2006), the C_6/C_5 (Gal + Man / Xyl + Ara) and the Deoxy/C₅ (Rha + Fuc / Xyl + Ara).

Bi-dimensional plots of carbohydrates ratios obtained with DFRC method as a function of 293 their HCl hydrolysis and THM method homologues are presented in Fig. 3. For HCl hydrolysis, 294 both ratios showed roughly the same profile with depth (Fig. 3), as their values increased from 0 295 to -20 cm and decreased slightly at the remainder of the core. This indicates a probable microbial 296 input occurred in the mesotelm (Younes et al., 2017). In the case of DFRC method, smaller 297 298 values of both ratios were obtained (between 0.96 and 0.20 for DFRC against 0.85 and 3.21 for HCl hydrolysis, in the case of C₆/C₅; and between 0.03 and 0.18 for DFRC against 0.08 and 0.47 299 for HCl hydrolysis, in the case of $Deoxy/C_5$; Fig. 3). This difference of magnitude between the 300 two methods, could be explained by the higher concentration of Ara and Xyl yielded with DFRC 301 method, as for the HCl hydrolysis both pentoses and hexoses yielded concentrations of the same 302 magnitude. Both ratios showed two increases, one located in the mesotelm and the other one in 303 the catotelm. For the mesotelm, it could be an indicator of microbial and/or Sphagnum 304 305 dominance in the upper half of the peat core. The latter is more probable as a microbial 306 dominance is discarded in this case. For the catotelm, it could be an indicator of an old microbial deposit that was hidden from the mostly used acid hydrolysis method. Once DFRC and acid 307 hydrolysis were compared, high correlation were obtained for Deoxy/C₅ (67.88 %; Fig. 3) and a 308 low correlation for C₆/C5 (15.2 %; Fig. 3). This indicates the occurrence of a hemicellulosic 309 fraction of hexoses (Gal and Man) that is inaccessible to the DFRC method. Although small 310 correlation was obtained between $C_6/C_5)_{HCl}$ and $Gal/C_5)_{DFRC}$, this comparison have allowed to 311 split the peat core into the three ecological layers. The acrotelm is characterized by small values 312 of both ratios, indicating the input of Xyl and Ara from the Sedges vegetation growing the 313 uppermost depth (Younes et al., 2017). The mesotelm is characterized by low values of 314 $Gal/C_5)_{DFRC}$ and high values of $C_6/C_5)_{HCl}$, indicating the occurrence of a hemicellulosic fraction 315 mainly composed of Gal and Man that was hindered from the DFRC method. The catotelm 316 characterized by high values of both ratios due to the higher rate of degradation of pentoses 317 compared to hexoses (Jia et al., 2008; Younes et al., 2017). 318

Xyl/Glu)_{THM} has been used to distinguish between angiosperm and gymnosperm occurrence 319 (Grasset el al., 2009; Younes and Grasset, 2018). Xyl is an indicator of plant input and it is 320 produced in a more significant way by angiosperms than gymnosperms vegetation (Cowie and 321 Hedges, 1984). Glu is characterized by its ubiquity in the living kingdom and is commonly 322 present at high levels in vascular vegetation (Cowie and Hedges, 1984). Comparison of this ratio 323 with it's DFRC homologue showed a low correlation (5.7 %; Fig. 3), even know both methods 324 showed a decreasing profile of Xyl is the upper half (acrotelm and mesotelm; Fig. 2) and an 325 increase in the bottom half (catotelm; Fig. 2). Minor correlations between DFRC and THM 326 methods ratios is due to the difference of mechanisms occurred between these two methods. 327 DFRC is a chemolytic method in acid conditions yet THM is a high temperature 328 329 thermochemolysis under alkaline conditions. This allow the latter method to access the so-called "Humic Fraction" that will be hindered from molecular degradation methods under acid 330 conditions. 331

332 *3.3. Mechanistic and technical implications*

The DFRC degradation method includes two key steps: (a) solubilization of lignin by bromination and acetylation with AcBr and (b) reductive cleavage catalyzed by solid zinc (Zn; Lu and Ralph, 1998). Since ligno-cellulosic fraction is the matter of interest of this study, adjacent phenolic structures at the α - and β - positions were substituted by monosaccharides (highlighted in blue) (Fig. 4).

For the free monosaccharides, the lone pairs of electrons of the hydroxyl groups will attack 338 the carbonyl function of the acetyl bromide. This attack is enhanced by the fact that bromide is 339 considered as a good leaving group. For the ligno-cellulosic bond, the bromide will perform a 340 nucleophilic substitution (SN) reaction, releasing monosaccharides directly bonded to the 341 aromatic structure. The substitution is of SN₂ since the monosaccharide is bonded to the aryl 342 group by the primary alcohol (Wade, 2012). Once the monosaccharide is released, it will 343 undergo acetylation. The Zn is well known as a reducing agent (Wade, 2012), its function in this 344 case is to reduce the α,β -Carbon bond. This reduction will enhance the elimination of the β -345 glyosidic bond and the yielded monosaccharide will attack the carbonyl group of an acetyl 346 347 bromide, thereby performing acetylation.

Following this mechanism, several assumptions can be made about the efficiency of DFRC as a degradation technique, thereby confirming or infirming its efficiency in the molecular characterization of carbohydrates in sediments:

- 351 (1) An exclusive cleavage to lignin moieties as proposed by Lu and Ralph, (1998).
- 352 (2) Higher yield for DFRC due to the higher corrosive effect and higher affinity of acetic
 acid and acetyl bromide compared with a mineral acid the HCl (since the investigated
 matrix is OM).

355 (3) Lower yield but higher selectivity for DFRC due the milder reaction conditions once
 356 compared with the conditions of HCl hydrolysis (See Materials and Methods).

Assumption (1) is directly discarded since the carbohydrates has been analyzed via DFRC. 357 358 Assumption (2) can also be discarded as the DFRC yielded lower amounts for all monosaccharides. Assumption (3) can be accepted, since a very good correlation is obtained for 359 the relative amount of monosaccharides between the two methods ($52\% < R^2 < 87\%$; Fig. 2). 360 HCl is considered as a strong acid, its capacity to cleave ether and ester type bonds has been 361 proven (Wade, 2012), especially if manipulated at high temperatures (> 100 °C). Therefore, we 362 consider that all non-cellulosic bonds are cleaved with the followed HCl procedure. The highest 363 correlation between DFRC and HCl hydrolysis were yielded for Xyl and Ara (Fig. 2). These 364 monosaccharides are considered as a vascular vegetation indicator. The obtained trends can show 365 the efficiency of AcBr in cleaving the ligno-cellulosic bonds and that pentoses are the major 366 carbohydrate components of ligno-cellulose. Lower correlations for the other monosaccharides 367 can be related to the occurrence of hexoses and deoxyhexoses in the free, hemicellulose and 368 ligno-cellulose compartments. 369

At the technical level, DFRC method is an all-in-one method for the characterization of various classes of soil OM, such as carbohydrates, lignin and lipids. Several studies have used this procedure or similar ones in order to analyse lignins (Lu and Ralph, 1998; Matsui and Ohira, 2013) and aromatic compounds in fossilized sediments (Grasset el al., 2010). In this study, although we were interested in revealing the capacity of this degradation technique to analyze carbohydrates, it yielded aromatic compounds, fatty acids and ∞ -hydroxy fatty acids. These families of molecules will be investigated in upcoming studies.

If one considers the time and temperature it takes for the different methods (2 h and 50 °C for
DFRC method, 3 h and 100 °C, for acid hydrolysis and 30 min and 400 °C for THM method),
DFRC is faster and/or requires milder conditions.

Beyond the yielded molecules, DFRC extract varied in weight from 300 mg/g to 500 mg/g of the freeze-dried weighted peat samples. This reveals, not only the degradation capacity, yet the extraction capacity of this method. A consideration of a double-shot procedure, like DFRC treatment followed by acid/alkaline hydrolysis could be interesting (*study under progress*).

In its mechanistic behavior, the acetyl bromide and acetic acid combination employed in DFRC, has the capacity to degrade, in a specific way, ligno-cellulose fraction and/or free carbohydrates entrapped in aromatic macromolecular structures. The "softer" acidic conditions for DFRC method, compared to acid hydrolysis, and the difference of the temperatures used, will make the cleavage of the strong glycosidic bond less probable, in the case of DFRC method.

389 *3.4.Statistical analysis of method comparison*

390 The application of PCA in geochemistry falls into three categories. (1) Determine OM sources based on biomarker specifity (terrigenous or marine) and trace the fate of OM in the 391 oceanic environment (Hu el al., 2006). (2) Revealing similarities and dissimilarities among 392 samples (Pearson et al., 2008). (3) Uncovering relationships among variables by developping 393 summary statistics (i.e. "degradation indices") (Boheme et al., 2004; Younes et al., 2017). Other 394 studies have, specifically, focused on the application of multidimensional statistical analysis for 395 interpreting molecular fingerprints of peat. Schellekens et al. (2009, 2011, 2012, 2015 a and b) 396 used the factor analysis method to extract parameters that might be used to interpret the 397 vegetation change, anaerobic and aerobic decomposition and fire incidence of several peat bogs. 398 399 In our case, PCA was used in order to compare between samples and between two methods of degradation. 400

Some of the strengths of PCA is its ability to analyze variables with interactive relationships 401 402 as they vary with respect to other variables. This method removes intercorrelation between variables by creating independent ones (PC1, PC2 ...). It also presents an "all in one" data 403 visualization method as one can assess geochemical relationships among different compound 404 classes (Younes and Grasset, 2017). In our analysis, two PCAs has been performed in order to 405 statistically validate the comparison between DFRC and HCl hydrolysis and CuO-NaOH 406 oxidation, from one side, and DFRC and THM methods from another side (Fig. 5). This division 407 is done following the difference of mechanisms between the employed methods. As DFRC, 408 CuO-NaOH oxidation and HCl hydrolysis are qualified as "Chemical Degradation Methods" and 409 THM is qualified as a "Thermochemical Method". 410

411 PC₁ and PC₂ of the chemical methods (left plot; Fig. 5) accounted for 86.70% of the total 412 variance. Lower but a high variance was obtained once DFRC and THM methods data was 413 investigated (PC₁ and PC₂ accounted for 76.31% of the total variance; Fig.5, right plot). These values indicate that the comparison of the employed parameters is statistically meaningful for allmethods and reliable trends can be concluded from this study.

For the chemical methods (left plot, Fig. 5), PCA analysis showed all four ratios ($Deoxy/C_5$ 416 and C_6/C_5) plotted together, indicating a stronger relationship than suggested by simple 417 correlation. Ara and Xyl in DFRC and HCl hydrolysis were closer to $\Lambda_{CuO-NaOH}$ than other 418 monosaccharides. This indicates that the ligno-cellulosic fraction is more likely composed of 419 pentoses than hexoses proving the higher Xyl contribution in the case of DFRC method (Fig. 2). 420 421 For the PCA of DFRC and THM methods (right plot, Fig. 5), lower clustered were yielded than the case of chemical methods. Although carbohydrate moieties of the two methods were 422 clustered together, Λ_{THM} was clustered by itself in the negative side of both PCs. 423

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425 Conclusion

This study presents the application of a recently developed and less used molecular analytical technique for the characterization of the carbohydrate pool in peat samples, the DFRC method. Acetyl bromide method is previously used to digest aromatic macromolecular structures, for analytical and industrial purposes. Following this principle, an application of this degradation technique to the OM matrix of a peatland will allow the access of a specific carbohydrate pool linked to lignin/polyphenols and/or entrapped into macromolecular aromatic structures.

In order to validate the DFRC method, a comparison have been held with lignin input. The high correlation between total carbohydrates yielded with DFRC method and acid hydrolysis from one side and lignin input yielded with CuO-NaOH oxidation, from another side, indicates that hemi-cellulosic and ligno-cellulosic pools of carbohydrates occurs simultaneously along the investigated peat core. It also shows that the ligno-cellulosic fraction is more likely linked to the intact non-oxidized lignin (yielded with CuO-NaOH oxidation).

Lower carbohydrates abundance, in the case of DFRC method, indicates its capacity to only target a portion of the carbohydrate fraction which is the free and ligno-cellulosic pools. For carbohydrates amounts, all three methods (acid hydrolysis and DFRC and THM methods) showed a decreasing trend from acrotelm, mesotelm and catotelm indicating the degradation of carbohydrates with depth.

Unlike acid hydrolysis, the DFRC method showed several patterns. This indicates some difference in the origin of the yielded monosaccharides. DFRC showed some difference in the origin and/or degradation rate between Xyl and Ara. It also allowed the identification of Rib as a microbial indicator. These information were hindered from acid hydrolysis and THM method, as the first showed nearly the same profile for all monosaccharides and the second one didn't have the capacity to analyze Ara and Rib.

Briefly, higher correlation (therefore higher similarity) was obtained between DFRC and acid hydrolysis than between DFRC and THM methods. This is due to the difference of mechanisms involved in chemolytic and thermochemolytic methods. For the first group, both methods shared acid conditions and milder temperature conditions if compared to the second group. In addition thermochemolysis is performed under alkaline conditions. These differences allow the application of chemical degradation methods to more recent less oxidized sediments and thermochemical methods to ancient more oxidized sediments. In order to confirm this scope of

456 applications, additional analyses should be performed on mutiple soils and sediments samples.

457 Therefore, we caution against making direct comparison of carbohydrates data yielded with the

- three methods because of different mechanisms involved in each process. Yet DFRC shows as a
- 459 prominent method for the molecular characterization of carbohydrates in soils and sediments.

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Fig.1. Total Carbohydrates (normalized per 1 g OC) yielded by DFRC method compared to total lignin yielded by CuO-NaOH oxidation and THM method.

Fig. 2. Depth record of relative carbohydrate abundance (mg carbohydrate / 1 g OC) with DFRC method, acid hydrolysis and THM method (mg carbohydrate / 1 g OC). Small boxes in HCl hydrolysis and THM method charts present the relationship between relative carbohydrate abundance of these methods from one side and DFRC method from other side (R^2 is the coefficient of determination).

Fig. 3. DFRC carbohydrate proxies compared to their homologues yielded by acid hydrolysis (HCl) and THM method.

Fig. 4. Comparison of the Lu and Ralph mechanism with the proposed mechanism

Fig. 5. PCA of carbohydrate and lignin parameters analyzed using DFRC and CuO-NaOH and HCl (left plot) and DFRC and THM (right plot).













Fig 4



Lignin monomers				
		CuO-NaOH	Monosaccharides ^b	
	I HIM ^a	oxidation		
G4:	3,4-dimethoxybenzaldehyde	S: Syringyl	Xyl:	Xylose
G5:	1-(3,4-dimethoxyphenyl)ethanone	V: Vanillyl	Ara:	Arabinose
G6:	3,4-dimethoxybenzoic acid methyl ester	C: Cinnamyl	Rib:	Ribose
G ₁₈ :	3-(3,4-dimethoxyphenyl)-2-propenoic acid methyl ester		Rha:	Rhamnose
P ₁₈ :	3-(4-hydroxyphenyl)-2-propenoic acid methyl ester		Fuc:	Fucose
S4:	3,4,5-trimethoxybenzaldehyde		Man:	Mannose
S ₅ :	1-(3,4,5-trimethoxypehnyl)ethanone		Gal:	Galactose
S ₆ :	3,4,5-trimethoxybenzoic acid methyl ester		Glu:	Glucose

Table. 1. Symbols of the investigated lignin and carbohydrate moieties.

^a The symbols used are the same as the ones used in Filley et al. (2006) and Younes and Grasset, (2018)

- ^b Symbols are followed by a subscript indicating the method the monosaccharides have been analyzed with (DFRC, THM or HCl).
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