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Self and N₂ broadening coefficients of H₂S probed by submillimeter spectroscopy: comparison with IR measurements and semi-classical calculations.

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ABSTRACT

High-resolution lines of H₂S self-perturbed and perturbed by N₂ have been recorded at room temperature using a frequency modulated THz spectrometer at pressures of H₂S ranging from 0.1 to 1.0 mbar. A non-linear least squares fitting of these spectra has been performed to retrieve self- and N₂-broadening coefficients of pure rotational transitions (mainly in the ^RQ sub-branch) of H₂S in the submillimeter region. We show that the lineshapes are well reproduced by the Speed-Dependent Voigt profile accounting for the speed dependence of relaxation rates which display linear pressure dependencies. Using the Robert and Bonamy formalism, these coefficients were calculated, showing the need to account for all contributions of the intermolecular interaction for N₂- broadening. The calculated coefficients are consistent with the measured data and the average accuracy of the line parameters has been estimated to be about 5 %. The observed J'' and K_a'' rotational dependences have been analyzed using an empirical model and compared to previous rovibrational studies in the ν_2 fundamental band. In our study the self- and the N₂- broadening coefficients decreases monotonically with K_a'' and J'' , respectively.

Key words:

Broadening coefficient, H₂S, submillimeter-wave spectroscopy, speed-dependence effects, semi-classical calculation, rotational dependencies.

1. Introduction

With its characteristic foul odor of rotten eggs, hydrogen sulfide (H_2S) is a colorless, highly toxic, corrosive, and flammable gas, soluble in various liquids including water and alcohol. Most of the atmospheric H_2S has natural origins. It is often produced from the microbial breakdown of organic matter in the absence of oxygen gas via anaerobic digestion. Therefore, H_2S is a good marker of the food spoilage especially fish [1]. Moreover, H_2S is one of the main trace gases released from volcanoes and the emission ratio $\text{H}_2\text{S}:\text{SO}_2$ is an important geochemical probe. Some satellite instruments such as the Infrared Atmospheric Sounding Interferometer (IASI)/MetOp have observed large H_2S plumes during explosive volcanic events [2]. H_2S is detected **not** only in the Earth's atmosphere but also in the atmospheres of Jupiter [3] and of the ice giants Neptune and Uranus with the Gemini-North/NIFS H-band observations [4]. Finally, due to its high toxicity and its very weak value of the low permissible concentration level which is equal to 0.008 mg/m^3 , the monitoring of anthropogenic H_2S in industrial and urban atmospheres is very challenging. Continuous sensing has been performed by Raman lidar [5]. Contrary to SO_2 , IR measurements of H_2S are sparse and difficult [2] and an accurate monitoring of atmospheric H_2S at trace levels requires to fit the H_2S atmospheric spectra with the best set of spectroscopic parameters and the most appropriate line profiles.

From a spectroscopic point of view, H_2S is similar to H_2O and represents a very light asymmetric top with a Ray parameter equal to 0.5234 [6]. Due to its large asymmetry, the rotational spectrum of H_2S shows no coherent line pattern. At room temperature, intense rotational lines are spread out over the complete submillimetre (smm)-wave regions. Numerous investigations have been performed on the pure rotational spectrum of H_2S ; most of them deal with line frequency analyses from the micro-wave (mw) [8] to the far-IR regions [7]. Ground state (GS) smm rotational line profile analyses and broadening parameter measurements are almost non-existent except the study of Helminger and De Lucia **who** determine the self-broadening parameters for only four rotational transitions of H_2S [8]. The temperature dependences of these parameters were studied by **Ball *et al.*** in 1999 [9]. However, the ability of frequency-modulated rotational spectroscopy to perform quantitative analysis of gas phase molecular constituents has been demonstrated [10] and a systematic study of the broadening parameters and their rotational dependences is now possible even for asymmetric top molecules [11]. Most of the broadening parameters of H_2S have been measured in the IR region from rovibrational transitions belonging to fundamental and combination vibrational bands. Tan *et al.* [12] have collected for **HITRAN-2016** all the experimental self and air-broadening parameters of H_2S provided from different experimental IR studies [13-17]. A large variability of the IR data

is observed and the vibrational dependence of the broadening parameters was neglected. Linear functions were used to describe the rotational dependences with a trend of the H₂S self- and air-broadening parameters to decrease with $J''+0.2K_a''$. For all these studies, Voigt line-shapes have been used to fit the measured line profiles. Yet a recent extended analysis of the high resolution FTIR spectra of H₂S in the region of the ν_2 bending fundamental band demonstrates that the line-shape fit could be significantly improved using the Hartman-Tran profile [18] with extra parameters taking into account the speed dependence of the relaxation. The half-widths analysis was made on the basis of the multi-spectrum fit procedure and the self-broadening coefficients were estimated for a few tens of lines for the three most abundant isotopologues [19].

In this study, we report the first measurements of self- and N₂-broadening coefficients of H₂S rotational lines for various values of the rotational quantum numbers J'' and K_a'' in a large frequency range mainly in the Q branch with a frequency-multiplication chain. Owing to the high signal-to-noise ratio (SNR) achieved using the frequency modulation technics, speed-dependence effects were systematically included in the fitting procedure of the observed lines. In addition, semi-classical calculations using the Robert-Bonamy (RB) formalism have been performed to predict the broadening parameters. The experimental smm broadening parameters are compared to previous IR measurements and to the semi-classical calculations. A new insight of the rotational dependence of the self- and N₂-broadening is finally provided.

2. Experimental details and fitting methods

2.1. Experimental setup and experimental conditions

Self and N₂-broadened H₂S lines were recorded in the 160-850 GHz range using a frequency modulated smm-wave spectrometer. This spectrometer is composed by a frequency synthesizer operating in the range of 7-12 GHz and frequency referenced to a GPS timing signal. The microwave synthesizer is used to drive a frequency multiplication chain (Virginia diodes) covering 70 to 900 GHz as described previously [20]. A frequency modulation (FM) was applied to the synthesizer. The FM was fixed at 13 kHz with a modulation depth m selected for each line in the range (0.3-4.5 MHz) according to the expected linewidth, see Fig. 1. Observed lineshapes were recorded using an *InSb* liquid He-cooled bolometer from QMC instrument up to 330 GHz and a zero bias Schottky detector at higher frequencies. Corresponding signals were demodulated thanks to a lock-in amplifier operating at a twice the modulation frequency. The data acquisition and operation of the smm spectrometer was computer controlled using a dedicated program. The

gases H₂S and N₂ were introduced into an absorption cell with Teflon windows presenting a clear diameter of 56 mm. Cell lengths, 28 cm and 113 cm, were employed depending on the observed line strengths. The spectrometer is located in a temperature controlled (295±1 K) room. The H₂S and N₂ gases were purchased from Sigma Aldrich and Praxair respectively with claimed purities better than 99.5%. The gas pressures were measured with a vacuum gauge (Pfeiffer company, model CMR364) operating from 10⁻³ to 1.1 mbar with a reading accuracy of 0.2%.

For self-broadening measurement, H₂S pressures were chosen between 0.09 and 0.8 mbar. In the example spectrum shown in Fig. 1, the targeted frequency was swept using 50 kHz frequency steps of 300 ms duration, with a 100 ms time constant of the lock-in amplifier. Such conditions ensure negligible consequences of a limited detection bandwidth on observed lineshapes [21, 22].

To measure the N₂-broadening coefficients, the rotational lines were recorded using the following procedure: the cell was first filled with pure H₂S at a fixed pressure about 0.1 mbar then about 6-8 lineshapes were recorded with increased N₂ pressures up to about 0.8 mbar.

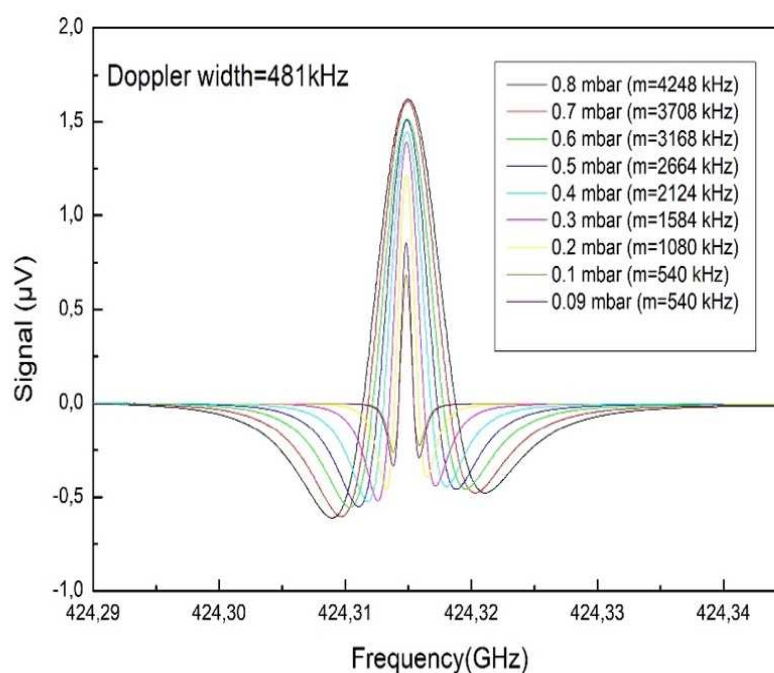


Fig. 1: $4_{4,0} \leftarrow 4_{3,1}$ rotational line of pure H₂S recorded at 424.315 GHz at different pressures. The modulation depth m is adjusted according to the estimated line width.

2.2 . Frequency-modulation submillimeter spectra

In case of a frequency modulated spectrometer with a lock-in detection operating at twice the modulation frequency ν_m , the line profile $F(\nu)$ has a W-shape which is given by the real part of the Fourier transform of [23]:

$$F(\nu) = \Re \left\{ 2 \int_0^{\infty} \Phi(t) \cdot J_2 \left[2\pi m t \cdot \text{sinc}(\pi \nu_m t) \right] \cdot \cos(2\pi \nu_m t) \cdot \exp(-2i\pi \nu t) \cdot dt \right\} \quad (1)$$

where ν is the electromagnetic field frequency, $J_2(x)$ is the 2nd order Bessel function, m is the modulation depth and $\Phi(t)$ is the molecular polarization correlation function, that is the gas pulse response which is null for $t < 0$.

Let us first consider the usual Voigt profil $F_V(\nu)$ which considers molecular displacements and molecular collisions as uncorrelated processes. It is simply given by the convolution of a Gaussian profile, related to the Doppler effect, and of a Lorentzian profile, related to collisional relaxation. So, its correlation function $\Phi_V(t)$ is given by [23]:

$$\Phi_V(t) = \exp \left[2i\pi \nu_0 t - 2\pi \Gamma_{0,V} t - \left(\frac{k \nu_{a0} t}{2} \right)^2 \right] \quad (2)$$

where ν_0 is the absorber line center frequency, $\Gamma_{0,V}$ is the collisional broadening, $k=2\pi\nu_0/c$ is the wavenumber with c the speed of light, and $\nu_{a0} = \sqrt{\frac{2k_B T}{m_a}}$ is the most probable value of the absorber speed defined by the Boltzmann constant k_B , the temperature T and molecular mass m_a of the absorber.

Due to the large intensities of H₂S rotational lines, the Beer-Lambert law had to be considered. This was done in the following way [24, 25]. The transmission of the gas $T(\nu)$ is given by:

$$T(\nu) = \exp[-SP_a l F(\nu)] \quad (3)$$

where S is the line intensity, P_a is the absorbing gas partial pressure, and l the cell length. So, the pulse response $\theta(t)$ of the whole sample is given by:

$$\text{for } t \geq 0: \theta(t) = \int_{-\infty}^{+\infty} T(\nu) \cdot \exp(+2i\pi \nu t) \cdot d\nu \quad (4a)$$

$$\text{for } t < 0: \theta(t) = 0 \quad (4b)$$

In the case of frequency modulation with a $2f$ -detection, the detected signal $D(\nu)$ is simply proportional to the real part of the Fourier transform:

$$D(\nu) \propto \Re \left\{ 2 \int_0^{\infty} \Theta(t) \cdot J_2 \left[2\pi m t \cdot \text{sinc}(\pi \nu_m t) \right] \cdot \cos(2\pi \nu_m t) \cdot \exp(-2i\pi \nu t) \cdot dt \right\} \quad (5)$$

Since such a frequency modulated spectrometer does not allow to measure absolute transmissions, line intensities were fixed at the theoretical values given by HITRAN [26]. Whereas ignoring the Beer-Lambert law leads to strongly erroneous relaxation rates, this approach remains quite reliable since a 25% error in line intensities leads to 1% deviation in retrieved relaxation parameters [25]. Note that the line profile detected by a frequency modulation spectrometer generally exhibits an asymmetry. This artifact may originate from baseline distortion and/or from the gas refractive index through residual standing waves [23]. It was taken into account by introducing an adjustable dispersion term chosen as proportional to the imaginary part of the Fourier transform of Eq. (5).

In a preliminary step to the fitting procedure, the baseline perturbations caused by the instability of the mm-wave optical alignment in the cell are modeled with a 3rd order polynomial function. Next the measured lines with corrected baseline are adjusted using a nonlinear least squares code based on the Eqs. (1-5). Signal amplitude, line center frequency, collisional parameters, as well dispersion term amplitude could be adjusted, whereas Doppler width, line intensities, as well as modulation depth and frequency, were kept fixed at the values corresponding to experimental conditions and tabulated parameters [26]. An example is presented in the Fig. 2 for the $4_{4,0} \leftarrow 4_{3,1}$ rotational line of pure H₂S measured at 295K. Using the Voigt profile and thanks to the good SNR obtained ($\cong 900$), one observes a remarkable difference between observed (orange curve) and fitted (black curve) profiles. Particularly, the observed profile has an amplitude greater than the fitted one and thus is narrower, the amplitude of the residual (black curve) is estimated to 3% of the line amplitude. Such a departure is characteristic of speed dependent and/or of Dicke narrowing effects [27].

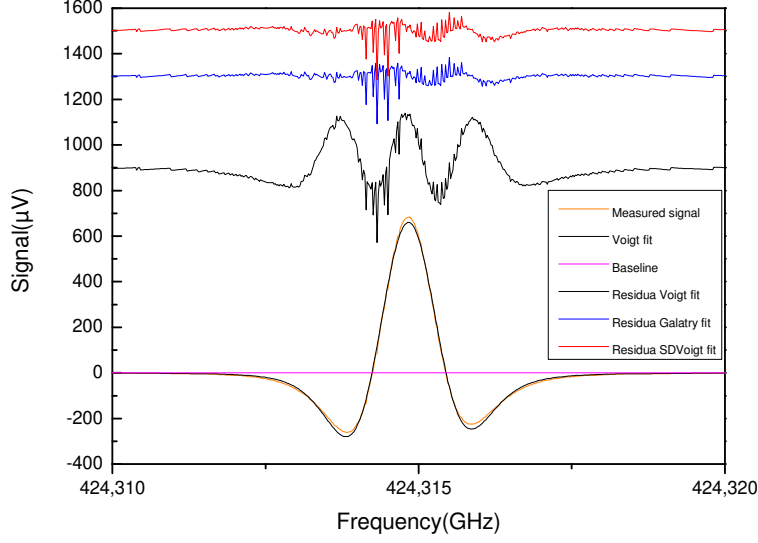


Fig. 2: Adjustments with Voigt, Galatry and SD-Voigt profiles and corresponding residuals enlarged by a factor of 10 for the $4_{4,0} \leftarrow 4_{3,1}$ rotational line of pure H_2S . Experimental conditions: $P=0.1$ mbar, $T=295K$, modulation depth $m = 540$ kHz, modulation frequency $\nu_m = 13$ kHz. Signal to noise ratio (SNR) is about 900. Doppler width= 482 kHz, $\Gamma_{0,V}^{Exp} = 397(20)$ kHz, $\Gamma_{0,SDV}^{Exp} = 469(10)$ kHz, $\Gamma_{2,SDV}^{Exp} = 100(4)$ kHz, $\Gamma_{0,G}^{Exp} = 446(20)$ kHz, $B_G^{Exp} = 315(10)$ kHz.

2.3. Speed-dependence effects

Departures from the Voigt profiles can be related to a combined influence of velocity changing collisions (Dicke narrowing) [28] and of the speed dependence of relaxation rates [29, 30]. To take into account these effects, the speed-dependent Galatry profile [31] has been retained because the corresponding correlation function has an analytical form, that is more convenient in case of frequency modulated absorption lines. Of course, this profile reduces to the usual Galatry profile [32] if SD-effects are negligible, or to the SD-Voigt profile [29, 30] if Dicke effect can be ignored.

Restricting to the velocity changing collisions only, the line narrowing can be modeled via the Galatry profile [32] whose correlation function is given by:

$$\Phi_G(t) = \exp \left\{ 2i\pi\nu_0 t - 2\pi\Gamma_{0,G}t + \frac{1}{2} \left(\frac{kv_{a0}}{2\pi B_G} \right)^2 [1 - 2\pi B_G t - \exp(-2\pi B_G t)] \right\} \quad (6)$$

In this equation B_G is the narrowing parameter that can be considered as the optical diffusion rate related to velocity changing collisions. Since this process influences the Doppler contribution,

the Galatry profile differs significantly from the Voigt profile in the Doppler regime, i.e. at low pressure, and coincide with the Voigt profile in the collisional regime.

Speed dependent effects can be considered separately by using the model first proposed by Berman [29] and Pickett [30] by assuming that collisional interaction is described by the empirical potential $V(r) \propto r^{-q}$, where r is the intermolecular distance and q an empirical parameter. This model leads to an hypergeometric dependence of the relaxation rate $\Gamma_{SDV}(v_a)$ on the absorber speed v_a . However, this model requires rather long numerical calculations, so Rohart *et al.* [33] have proposed the quadratic approximation:

$$\Gamma_{SDV}(v_a) = \Gamma_{0,SDV} + \Gamma_{2,SDV} \cdot \left[\left(\frac{v_a}{v_{a0}} \right)^2 - \frac{3}{2} \right] \quad (7)$$

which leads to a simpler correlation function:

$$\Phi_{SDV}(t) = \frac{\exp\left[2i\pi\nu_0 t - 2\pi\left(\Gamma_{0,SDV} - \frac{3}{2}\Gamma_{2,SDV}\right)t\right]}{(1+2\pi\Gamma_{2,SDV}t)^{3/2}} \cdot \exp\left[-\frac{(k\nu_{a0}t)^2}{4(1+2\pi\Gamma_{2,SDV}t)}\right] \quad (8)$$

where $\Gamma_{0,SDV} = \left\langle \Gamma(v_a) \right\rangle_{v_a}$ is the mean relaxation rate over molecular speeds and $\Gamma_{2,SDV}$ is a pressure dependent parameter describing the speed dependence of relaxation rates. Although approximate, this quadratic model has been retained for the extension of data bases to non Voigt spectral lineshapes [34]. Indeed, this model has been shown to be actually an accurate approximation of the hypergeometric one, at least for SNR lower than several thousand [35]. Moreover, it can be easily connected to the hypergeometric model via [36]

$$\frac{\Gamma_{2,SDV}}{\Gamma_{0,SDV}} = \frac{\lambda\mu}{3(1+\lambda)^{\mu/2}} \cdot M\left[1 - \frac{\mu}{2}; \frac{5}{2}; -\frac{3\lambda}{2}\right] \quad (9)$$

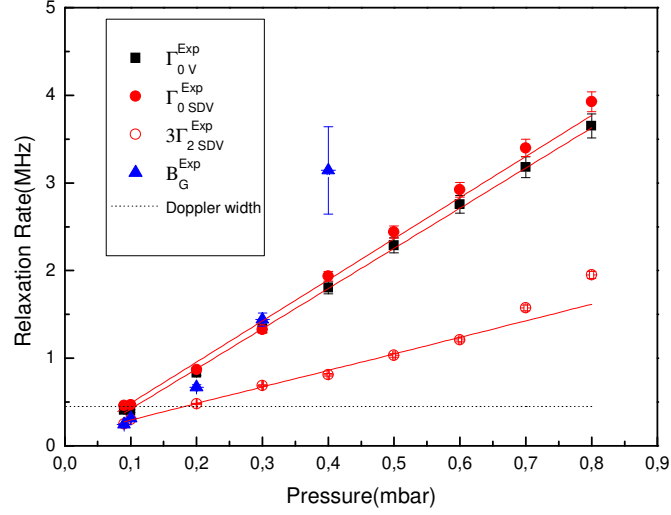
where $\lambda = m_b/m_a$ is the mass ratio of buffer and absorbing molecules, $\mu = (q-3)/(q-1)$, and $M[a;b;x]$ is the hypergeometric confluent function [37].

Let us consider the recorded H₂S self-broadened lineshape displayed in Fig. 2 for a given pressure. One observes the quality of fits is similar using either the Galatry or the SD-Voigt profiles. For both cases and on account of the fairly good SNR obtained, one gets a nearly perfect adjustment with residuals significantly smaller than those obtained with the Voigt profile. Note this lineshape corresponds to the Doppler regime, the collisional broadening rate being lower than the Doppler broadening. However, at much higher pressures corresponding to the collisional

regime, and by contrast with the SD-Voigt profile, it was impossible to perform any fit using the Galatry profile. Let us consider the pressure dependences of the collisional relaxation rates $\Gamma_{0,V}^{Exp}$, $\Gamma_{0,SDV}^{Exp}$, $\Gamma_{2,SDV}^{Exp}$, as well as the Galatry optical diffusion rate B_G^{Exp} , as displayed on Fig. 3a. As expected for the binary collisional regime, Voigt and SD-Voigt relaxation rates behave linearly. On the contrary, the Galatry optical diffusion rate B_G^{Exp} exhibits a nonlinear pressure dependence for collisional broadenings larger than the Doppler width. Moreover, B_G^{Exp} could not be determined for pressures above 0.5 mbar, corresponding fits being impossible. This feature, well explained in Ref. [38], shows that observed departures from the Voigt profile are mainly due to the speed dependence of relaxation rates and that the Dicke effect has a negligible role and can be disregarded in case of self H₂S relaxation. Similar difficulties were also encountered in the case of N₂-induced relaxation of H₂S, as displayed on Fig. 3b. In this case, the non-linear pressure dependence of B_G^{Exp} is also clearly evident, but occurs for pressures higher than in the pure H₂S relaxation case (Fig. 3a), a feature due to the N₂-H₂S relaxation rate about twice smaller than the self H₂S one, as explained in Ref. [38]. As a consequence, the role of velocity changing collisions has been disregarded and only the speed dependence of relaxation rates has been considered in the analysis of H₂S collisional line shape.

The pressure dependence of the relaxation rate is obtained by linear least squares fittings as shown in Fig. 3a and Fig. 3b for the self-broadened $4_{4,0} \leftarrow 4_{3,1}$ and the N₂-broadened $2_{2,1} \leftarrow 2_{1,2}$ transitions. The slopes of the straight lines gives the relaxation coefficients $\gamma_{0,V}^{Exp}$, $\gamma_{0,SDV}^{Exp}$ and $\gamma_{2,SDV}^{Exp}$, expressed in MHz/mbar and defined as $\gamma_{0,V}^{Exp} = \frac{\partial \Gamma_{0,V}^{Exp}}{\partial P}$, $\gamma_{0,SDV}^{Exp} = \frac{\partial \Gamma_{0,SDV}^{Exp}}{\partial P}$, $\gamma_{2,SDV}^{Exp} = \frac{\partial \Gamma_{2,SDV}^{Exp}}{\partial P}$, where $\Gamma_{0,V}^{Exp}$ and $\Gamma_{0,SDV}^{Exp}$ are the pressure broadenings, $\Gamma_{2,SDV}^{Exp}$ is the relaxation speed dependence parameter and P is the gas pressure. The measured values of self-and N₂-broadening coefficients $\gamma_{0,V}^{Exp}$, $\gamma_{0,SDV}^{Exp}$ and $\gamma_{2,SDV}^{Exp}$ with their estimated uncertainties are listed in Tables 2 and 4, respectively.

a



b

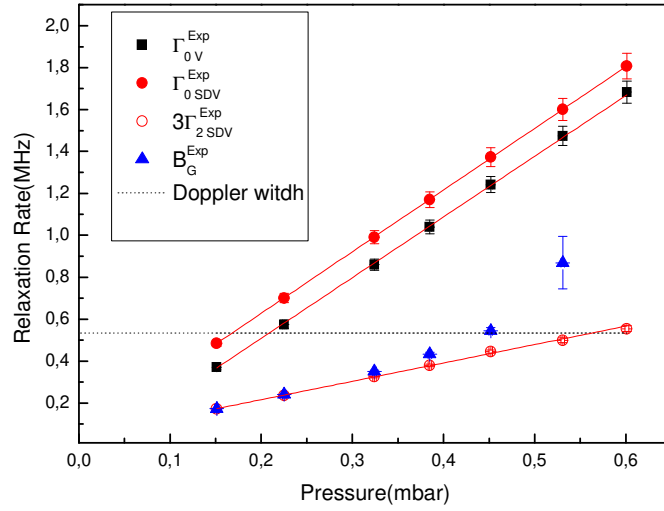


Fig. 3: Pressure dependence of the relaxation rates of H₂S measured at room temperature ($T = 295$ K). The relaxation coefficients are derived from weighted linear least squares:

(a) Self-broadened $4_{4,0} \leftarrow 4_{3,1}$ transition: $\gamma_{0,V}^{Exp} = 4.59(9)$ MHz/mbar, $\gamma_{0,SDV}^{Exp} = 4.74(3)$ MHz/mbar, $\gamma_{2,SDV}^{Exp} = 0.66(3)$ MHz/mbar.

(b) N₂-broadened $2_{2,1} \leftarrow 2_{1,2}$ transition for $P_{H_2S} = 0.01$ mbar: $\gamma_{0,V}^{Exp} = 2.89(2)$ MHz/mbar, $\gamma_{0,SDV}^{Exp} = 2.941(6)$ MHz/mbar, $\gamma_{2,SDV}^{Exp} = 0.292(3)$ MHz/mbar. The error bars correspond to one standard deviation (SD); they take into account the different experimental uncertainties: fitting standard deviation (<1%), cell length (0.7%), temperature (0.34%), pressure (0.4%) and the systematic errors linked to detector and electronic nonlinearity (2%).

3. Broadening coefficients calculations

3.1. Formalism of the theoretical calculations

Since previous calculations of pressure broadening of H₂S were performed in the ν_2 band and for rovibrational lines other than measured in this work, we decided to compute the self- and N₂-broadening coefficients of H₂S in the ground vibrational state in order to reproduce our measurements and to detect an eventual vibrational effect on the pressure broadening phenomena. The computations were performed using the semi-classical model based on the RB formalism. Within this formalism, the pressure broadening half-width Γ_0 in the unit of **Hz** may be expressed as [39,40,41]:

$$\Gamma_0 = \frac{n}{2\pi} \text{Re} \left[\sum_{J_2 K_2} \rho_{J_2 K_2} \int_0^{+\infty} v F_{Max}(v) dv \int_{r_{c0}}^{+\infty} 2\pi r_c \left(\frac{v'_c}{v} \right) S(r_c, v) dr_c \right] \quad (10)$$

where n is the number density of perturbing molecules, v is the relative velocity between the absorber and the perturber, $F_{Max}(v)$ is the Maxwell velocity distribution, $\rho_{J_2 K_2}$ is the relative population of the perturber, r_c and v'_c are position and velocity parameters related to the relative trajectories model and r_{c0} is the distance of closest approach [39]. The differential broadening cross section $S(r_c, v)$ is given by [40]:

$$S(r_c, v) = 1 - \exp \left(\Re e S_{2,i_2} + \Re e S_{2,f_2} + \Re e S_{2,f_2 i_2} \right) \quad (11)$$

where the S_{2,i_2} , S_{2,f_2} and $S_{2,f_2 i_2}$ are the second-order terms of the perturbation development of the differential cross section [39,40,42]. To calculate the self- and N₂-broadening coefficients, we have considered the intermolecular potential involving the electrostatic interactions as well as the induction and the Tipping–Herman potential [43] expressed as:

$$V = V_{\mu_1 \mu_2} + V_{\mu_1 Q_2} + V_{Q_1 \mu_2} + V_{Q_1 Q_2} + 4\epsilon \left[R_1 \left(\frac{\sigma}{r_c} \right)^{13} - A_1 \left(\frac{\sigma}{r_c} \right)^7 \right] P_1(\cos \theta) + 4\epsilon \left[R_2 \left(\frac{\sigma}{r_c} \right)^{12} - A_2 \left(\frac{\sigma}{r_c} \right)^6 \right] P_2(\cos \theta) \quad (12)$$

where P_1 and P_2 are the first- and second-order Legendre polynomials, θ is the angle between the axis of the absorbing molecule and the intermolecular axis, the indices 1 and 2 refer to the absorber and the perturber, ϵ and σ are the Lennard-Jones parameters. In this potential, R_1 , A_1 , R_2 and A_2 were chosen to obtain a reasonable agreement with the experimental values of the pressure broadening coefficients. The term due to the electrostatic interactions is a sum of dipole-dipole, dipole-quadrupole, quadrupole-dipole and quadrupole-quadrupole interactions, where μ and Q are the dipole and quadrupole moments of the molecules.

3.2. Computations of broadening coefficients

The collisional broadening coefficients of H₂S self- and -N₂ perturbed have been calculated using the above semiclassical RB formalism with the bi-resonance functions of Starikov [44]. The computation has been performed at the experimental temperature $T = 296$ K for a complete set of rotational lines. **The rotational constants, the electric multipole moments of H₂S and N₂ and the Lennard-Jones parameters are given in Table 1 with the associated references, as well as the A_1 , R_1 , A_2 and R_2 parameters varied manually until obtaining the best agreement between the measured and calculated broadening coefficients.**

Table 1 Molecular parameters used in the calculations of the self- and N₂-broadening coefficients of H₂S in the GS.

	H ₂ S	N ₂		
M (g.mol ⁻¹)	34.08	28.0134		
A (cm ⁻¹)	10.35995 ^a	-		
B (cm ⁻¹)	9.018496 ^a	1.989622 ^d		
C (cm ⁻¹)	4.730608 ^a	-		
μ (D)	0.974 ^b	0		
Q (DÅ)	0.71 ^b	-1.4 ^e		
ε (K)	309.6 ^c	92.91 ^f		
σ (Å)	3.509 ^c	3.816 ^f		
Ajustable parameters				
Molecular system	A_1	R_1	A_2	R_2
H ₂ S - H ₂ S	0.08	1	0.6	0.6
H ₂ S - N ₂	0.5	5	0.08	0.08

^a Ref. [45], ^b Ref. [46], ^c Ref. [47], ^d Ref. [48], ^e Ref. [49], ^f Ref. [50]

4. Results and discussion

4.1. Self-Broadening coefficients

Table 2 presents the self-broadening coefficients determined from 30 different rotational transitions of H₂S in the 150 – 850 GHz frequency range with $0 < J'' < 12$ and $0 < K_a'' < 8$. Due to the line narrowing, $\gamma_{0,SDV}^{Exp}$ should be larger than $\gamma_{0,V}^{Exp}$. This assumption is systematically verified with the self-broadening coefficients. The averaged relative variation of the self-

broadening coefficient due to the SD effect $\frac{\gamma_{0,SDV}^{Exp} - \gamma_{0,V}^{Exp}}{\gamma_{0,V}^{Exp}}$ is estimated to be in the 2% to 12% range with a mean value of about 7.3%. In Table 2 are also reported the $\frac{\gamma_{2,SDV}^{Exp}}{\gamma_{0,SDV}^{Exp}}$ ratios which give an order of magnitude on speed dependent effects on relaxation. From the mean value of this ratio, about 0.15, and using Eq. (9), one obtains a q value of about 8.0, which corresponds to a rather hard sphere collision model for the Pickett empirical potential [30]. Table 2 compared also experimental with the calculated self-broadening coefficients using the RB formalism. The mean square difference obtained between the calculated coefficients and the experimental ones adjusted with the Voigt profile is evaluated to 0.34 MHz/mbar. This difference, about 7% the mean value of self H₂S broadening coefficients, shows the good quality of our theoretical broadening calculations.

In Fig. 4, the experimental and calculated self-broadening coefficients determined in this study are compared with previous measurements in the IR domain for different vibrational bands (Refs. [13,17,14]), and for microwave rotational transitions by Helminger and De Lucia (Ref. [8]). For all these references, the deviation from Voigt profile and the speed dependence effects were not taken into account. The comparison of the different datasets does not allow detect any vibrational dependence. A significant dispersion of the self-broadening coefficient between the different datasets is observed. Tan *et al.* in Ref. [12] have performed a selection of the most reliable self-broadening and air-broadening coefficients of H₂S for the **HITRAN-2016** edition of the database from different IR datasets [26]. This selection for the self-broadening is also plotted in Fig. 4. From 2016, a first order rotational dependence was considered in the HITRAN database with a linear fit of γ_{self} and γ_{air} versus $J''+0.2 K_a''$ reproducing the slight decreasing of the broadening coefficients observed in Fig. 4 from $J''=2$ to $J''=11$. For the self-broadening, the mean square difference between the $\gamma_{0,SDV}^{Exp}$ determined in this study and the values tabulated in **HITRAN-2016** is evaluated to 0.35 MHz/mbar, which is similar to the previous difference between our measurements and RB calculations.

Table 2. Self-broadening coefficients of H₂S in MHz/mbar determined in this study. Frequencies are given in GHz. J' , K_a' , K_c' , J'' , K_a'' , K_c'' are the upper and lower state quantum numbers. $\gamma_{0,V}^{Exp}$, $\gamma_{0,SDV}^{Exp}$ and $\gamma_{2,SDV}^{Exp}$ are the experimental coefficients fitted with Voigt and SD-Voigt profiles. The ratio $\frac{\gamma_{2,SDV}^{Exp}}{\gamma_{0,SDV}^{Exp}}$ gives the magnitude of the speed dependence. γ_{RB}^{Cal} are the calculated coefficients with the RB formalism. γ_{GS}^{Emp} are the empirical coefficients based on the Starikov model adjusted with the GS self-broadening coefficients determined in this study (see Section 4). Last digit quoted errors in parentheses are equal to one standard deviation (1 SD).

Frequency	J'	K_a'	K_c'	J''	K_a''	K_c''	$\gamma_{0,V}^{Exp}$	$\gamma_{0,SDV}^{Exp}$	$\gamma_{2,SDV}^{Exp}$	$\frac{\gamma_{2,SDV}^{Exp}}{\gamma_{0,SDV}^{Exp}}$	γ_{RB}^{Cal}	γ_{GS}^{Emp}
168.762	1	1	0	1	0	1	4.52(7)	4.84(4)	0.63(2)	0.130(5)	4.61	4.36
216.710	2	2	0	2	1	1	4.91(7)	5.40(2)	0.69(1)	0.128(2)	5.02	5.03
300.505	3	3	0	3	2	1	3.98 (8)	4.34(3)	1.04(2)	0.240(6)	4.80	4.64
369.101	3	2	1	3	1	2	5.39(9)	5.67(3)	1.36(3)	0.240(7)	4.75	5.04
369.127	4	3	1	4	2	2	4.4(1)	4.75(2)	0.74(2)	0.156(5)	4.55	4.66
393.450	2	1	1	2	0	2	5.62(9)	6.08(1)	1.39(6)	0.23(1)	4.99	5.06
407.675	5	4	1	5	3	2	4.08(8)	4.34(2)	0.93(4)	0.21(1)	4.36	4.58
424.314	4	4	0	4	3	1	4.59(9)	4.74(6)	0.66(3)	0.139(8)	4.56	4.47
493.362	6	5	1	6	4	2	4.56(7)	4.74(1)	0.64(2)	0.135(5)	4.17	4.61
505.564	2	2	1	2	1	2	5.09(5)	5.34(1)	1.26(6)	0.24(1)	5.02	5.03
555.254	7	5	2	7	4	3	4.54(9)	4.89(3)	0.84(3)	0.172(7)	4.28	4.54
567.078	6	4	2	6	3	3	4.30 (6)	4.53(2)	0.53(1)	0.117(3)	4.50	4.63
568.049	3	3	1	3	2	2	4.36(7)	4.67(2)	0.93(2)	0.199(5)	4.80	4.64
579.798	5	5	0	5	4	1	4.97(9)	5.15(3)	1.12(3)	0.218(7)	4.33	4.61
593.169	8	6	2	8	5	3	4.17(5)	4.48(3)	0.66(4)	0.15(1)	4.08	4.30
626.473	7	6	1	7	5	2	4.90(9)	5.31(2)	1.19(1)	0.224(3)	3.96	4.46
665.395	4	2	2	4	1	3	5.23(7)	5.60(4)	1.01(5)	0.18(1)	4.96	4.94
687.304	2	0	2	1	1	1	4.01(6)	4.41(3)	0.99(5)	0.23(1)	4.75	4.76
689.121	9	7	2	9	6	3	4.03(9)	4.20(3)	0.52(3)	0.124(8)	3.91	4.05
733.025	10	7	3	10	6	4	4.55(9)	4.90(5)	0.59(3)	0.120(7)	4.03	4.15
746.720	9	6	3	9	5	4	4.34(9)	4.78(2)	0.74(2)	0.155(5)	4.20	4.25
747.301	3	2	2	3	1	3	5.49(8)	5.68 (5)	0.82(3)	0.144(7)	4.75	5.04
748.240	6	6	0	6	5	1	4.44(7)	4.58(3)	0.58(2)	0.127(5)	4.10	4.67
749.433	5	5	1	5	4	2	4.87(6)	5.12(1)	0.71(2)	0.139(4)	4.33	4.61
782.344	11	8	3	11	7	4	4.35(9)	4.79(2)	0.75(4)	0.157(9)	3.91	4.29
794.222	8	7	1	8	6	2	4.54(6)	4.74(4)	0.52(3)	0.110(7)	3.80	4.13
800.850	5	4	2	5	3	3	4.27(5)	4.53(2)	0.68(3)	0.150(7)	4.36	4.58
804.729	8	5	3	8	4	4	4.61(7)	4.80(4)	0.54(2)	0.113(5)	4.39	4.48
839.140	10	8	2	10	7	3	3.64(5)	4.89(3)	0.57(3)	0.117(7)	3.75	3.99
841.280	8	3	6	7	4	3	4.91(6)	5.12(3)	0.83(3)	0.162(7)	4.58	4.51

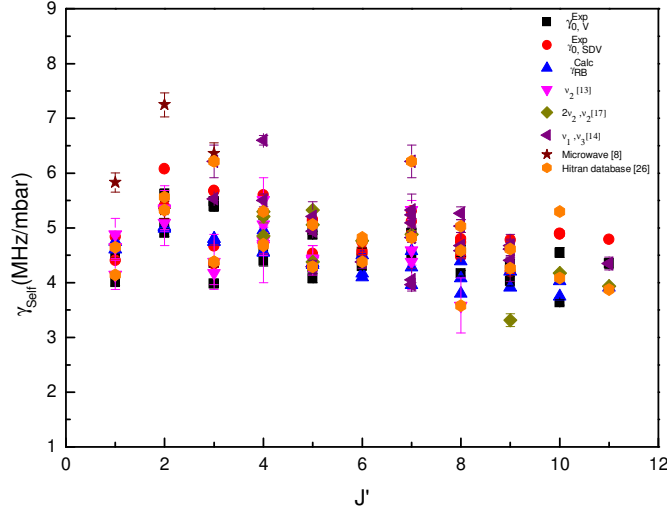


Fig. 4: Self-broadening coefficients γ_{Self} determined in this study compared to the previous IR (Refs. [13, 17, 14, 26]) and mw (Ref. [8]) studies.

In order to analyze more finely the rotational dependence of the H₂S broadening coefficient, the empirical model of Starikov described in Refs. [46,51] for the measured broadening coefficients of C_{2v} asymmetric top molecules such as H₂O and H₂S was adopted. The self-broadening coefficients for the $J'_{K'_a, K'_c} \leftarrow J''_{K''_a, K''_c}$ rotational transitions were fitted using the following empirical equation:

$$\gamma(K''_a, K'_a) = \gamma^{Emp} = x_1 + x_2 \frac{\cosh[x_5(K''_a - K'_a)(K''_a + K'_a)]}{\cosh[x_3(K'_a - x_4)] \times \cosh[x_3(K''_a - x_4)]} \quad (13)$$

where the J rotational dependence is included in an x_k polynomial function (with $k = 1, 2, 3, 4$ and 5) given by:

$$x_k = x_{k0} + x_{k1}(J'' + J') + x_{k2}(J'' + J')^2 \quad (14)$$

The parameters x_{k0} , x_{k1} , and x_{k2} were obtained using a nonlinear least-square method that adjusts the equations (13) and (14) to the experimental values of self-broadening coefficients listed in Table 2 complemented with the calculated values γ_{RB}^{Cal} for the non-measured transitions in the ranges $0 < J'' < 12$ and $0 < K''_a < 8$. These fitted parameters are listed in Table 3 with their uncertainties estimated as the fit standard deviations. These uncertainties do not explicitly reveal the errors from the physical parameters intervening in the experiment, but implicit reflect the accuracy of linewidth measurements. The full set of parameters is included in our GS fit and, contrary to the v_2 fit of Ref. [46], all parameters are determined according to their accuracies.

The γ_{GS}^{Emp} self-broadening coefficients determined in this study from Eq. (13) and Eq. (14) are listed in Table 2. For self-broadening, the comparison between the experimental $\gamma_{0,V}^{Exp}$ and the empirical γ_{GS}^{Emp} provides a mean square difference evaluated to 0.16 MHz/mbar i.e two times smaller than those obtained with γ_{RB}^{Cal} . This result demonstrates the interest of the empirical modeling of Eq.(13) for the interpolation of unmeasured (J,K) lines.

Table 3: Parameters adjusted in the Starikov model using Eqs. 13 and 14 for the empirical determination of the self-broadening coefficients. x_{1k} and x_{2k} are in unit of $\text{cm}^{-1}.\text{atm}^{-1}$, x_{3k} , x_{4k} and x_{5k} are unitless. Last digit quoted errors in parentheses are equal to one standard deviation (1 SD).

Parameter	This study (GS)	Ref. [46] (ν_2)
x_{10}	-5.676(3)	-
x_{11}	0.468(3)	0.00275(5)
x_{12}	-0.00942(1)	-
x_{20}	5.81(4)	0.194(1)
x_{21}	-0.45762(5)	-0.00484(7)
x_{22}	0.00906(2)	-
x_{30}	-0.114(2)	0.107(7)
x_{31}	0.01022(2)	0.00306(4)
x_{32}	-0.000290(2)	-
x_{40}	0.821(4)	5.2(2)
x_{41}	0.04542(3)	-0.50(2)
x_{42}	-0.03610(3)	0.0215(7)
x_{50}	0.0363(4)	0.011(4)
x_{51}	0.00194(2)	-
x_{52}	-0.0001899(4)	-

The rotational dependences of the self-broadening coefficients in K_a'' and in J'' are presented in Fig. 5 which exhibits significant differences especially at low K_a'' and high J values with the empirical coefficients adjusted with our SMM measurements and those adjusted with the IR data (ν_2 band) from Ref. [13]. The increasing of $\gamma_{0,self}$ with J'' at low K_a'' for the RQ sub-branch and the rapid decreasing with K_a'' at high J'' characteristic of the ν_2 rovibrational transitions are not observed in the present study. In the GS, our $\gamma_{0,self}$ (RQ) coefficients monotonically decreases with K_a'' and have a weaker variation with J'' at $K_a''=0$.

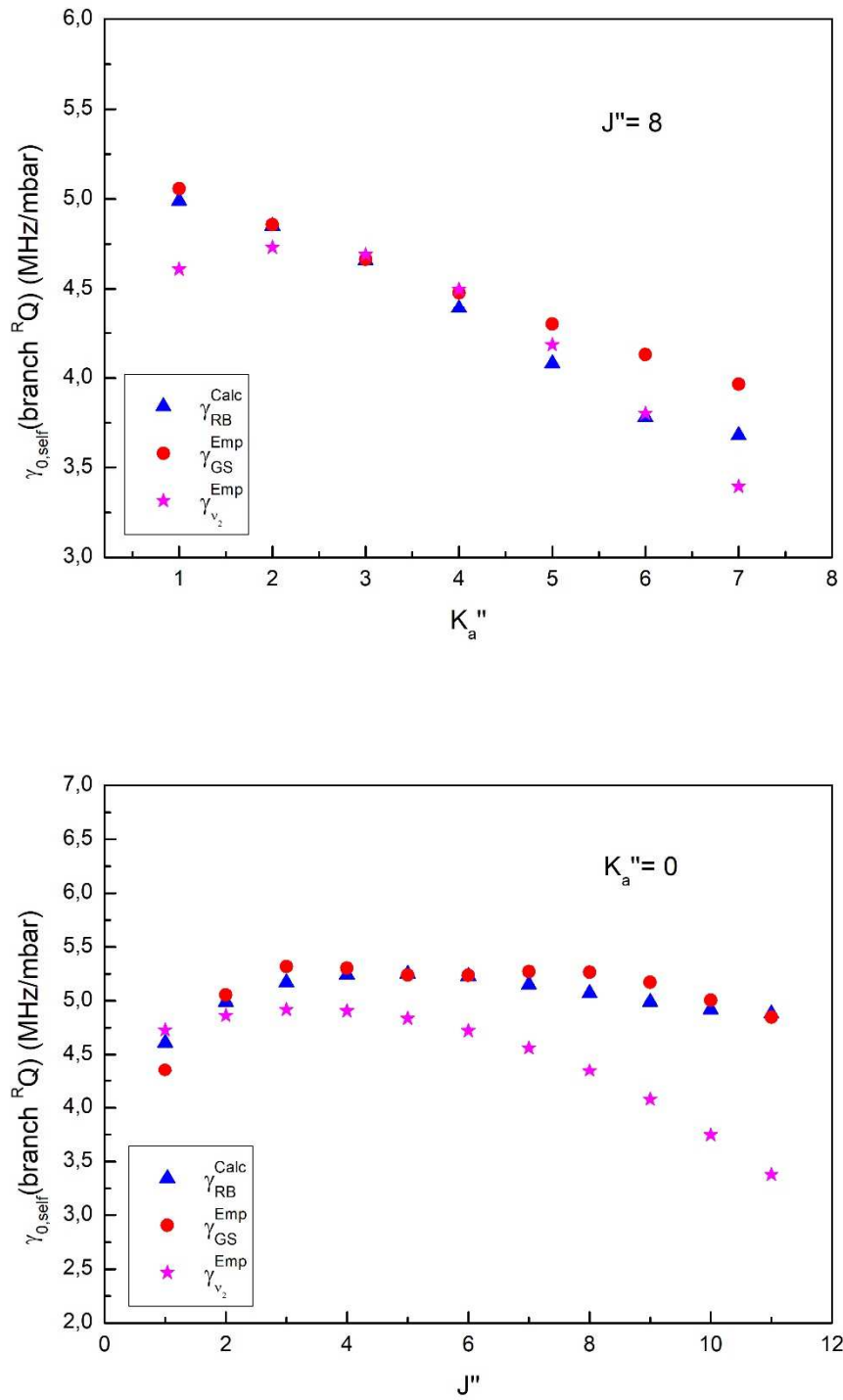


Fig. 5: Examples of K_a'' and J rotational dependences of the H_2S self-broadening coefficients: K_a'' dependence for $J''=8$ (top) and J'' dependence for $K_a''=0$ (bottom). 3 sets of data are compared: the calculated coefficients with the RB formalism γ_{RB}^{Calc} and the empirical coefficients γ_{GS}^{Emp} and $\gamma_{v_2}^{Emp}$ adjusted respectively on the GS rotational transitions (this study) and the v_2 rovibrational transitions from Ref. [13].

4.2. N₂-Broadening coefficients

The N₂-broadening coefficients measured for 27 rotational transitions of H₂S in the GS with $0 < J'' < 12$ and $0 < K_a'' < 8$ using the frequency modulated spectrometer in the frequency range 350 – 810 GHz are listed in Table 4. As for the self-broadening, the SD-Voigt profile provides the best adjustments of the N₂-broadened rotational lines. The $\gamma_{0,SDV}^{Exp}$ N₂-broadening coefficients are greater than the $\gamma_{0,V}^{Exp}$ coefficients for 70% of the measured lines. In Table 4 are also reported the ratios $\frac{\gamma_{2,SDV}^{Exp}}{\gamma_{0,SDV}^{Exp}}$ which give an order of magnitude of speed dependent effects on relaxation. From the mean value of this ratio, about 0.10, and using Eq. (9), one obtains a q value of about 6.0 for the Pickett empirical potential [30], which corresponds approximately to a **Van der Waals** model, revealing lower speed dependent effects than for the self-broadening.

Table 4 present the calculated N₂-broadening coefficients with the Voigt and SD-Voigt profiles, the adjusted ones and the calculated values using the RB formalism. The mean square difference between $\gamma_{0,V}^{Exp}$ and γ_{RB}^{Cal} is evaluated to 0.2 MHz/mbar, that about 8% of the mean value of H₂S-N₂ broadening coefficients. The figure 6 presents the experimental and calculated N₂-broadening coefficients for all rotational transitions studied in this work compared with previous rotational GS (Ref. [9]) and IR rovibrational measurements in the fundamentals bands ν_1 , ν_2 and ν_3 (Refs. [15,52]). In these previous studies, only the Voigt profile was used. All datasets exhibit a monotonic decreasing of the N₂-broadening coefficients with J'' but no vibrational dependence could be unambiguously extracted.

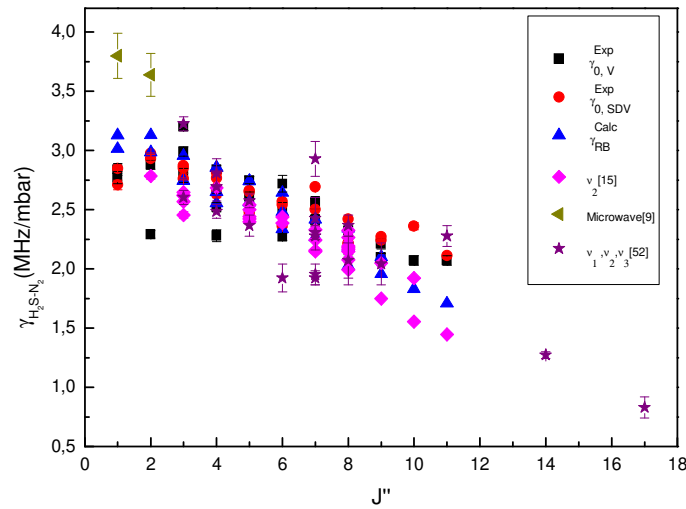


Fig. 6: N₂-broadening coefficients determined in this study compared to the previous IR (Ref. [15, 52]) and MW (Ref. [9]) studies.

Table 4. N₂-broadening coefficients of H₂S in MHz/mbar determined in this study. Frequencies are given in GHz. $\gamma_{0,V}^{Exp}$, $\gamma_{0,SDV}^{Exp}$ and $\gamma_{2,SDV}^{Exp}$ are the experimental coefficients fitted with Voigt and SD-Voigt profiles. The ratio $\frac{\gamma_{2,SDV}^{Exp}}{\gamma_{0,SDV}^{Exp}}$ gives the magnitude of the speed dependence. γ_{RB}^{Cal} are the calculated coefficients with the RB formalism. γ_{GS}^{Emp} are the empirical coefficients based on the Starikov model adjusted with the GS self-broadening coefficients determined in this study (see Section 4). Last digit quoted errors in parentheses are equal to one standard deviation (1 SD).

Frequency	J'	K_a'	K_c'	J''	K_a''	K_c''	$\gamma_{0,V}^{Exp}$	$\gamma_{0,SDV}^{Exp}$	$\gamma_{2,SDV}^{Exp}$	$\frac{\gamma_{2,SDV}^{Exp}}{\gamma_{0,SDV}^{Exp}}$	γ_{RB}^{Cal}	γ_{GS}^{Emp}
369.101	3	2	1	3	1	2	2.99(4)	2.87 (1)	0.24 (1)	0.084(4)	2.75	2.81
369.127	4	3	1	4	2	2	2.81(5)	2.809 (6)	0.266(4)	0.095(2)	2.65	2.66
393.450	2	1	1	2	0	2	3.05(5)	2.972(6)	0.256(3)	0.086(1)	2.99	2.71
407.675	5	4	1	5	3	2	2.75(3)	2.65(2)	0.213(6)	0.080(3)	2.56	2.52
424.314	4	4	0	4	3	1	2.84(3)	2.68(1)	0.197(4)	0.074(2)	2.86	2.74
493.362	6	5	1	6	4	2	2.72(7)	2.57(2)	0.263(9)	0.102(4)	2.48	2.51
505.564	2	2	1	2	1	2	2.89(4)	2.941 (6)	0.292(3)	0.099(1)	3.13	2.79
555.254	7	5	2	7	4	3	2.41(5)	2.50(1)	0.27(2)	0.108(8)	2.27	2.35
567.078	6	4	2	6	3	3	2.50(6)	2.54(1)	0.23(1)	0.091(4)	2.34	2.39
568.049	3	3	1	3	2	2	2.79(6)	2.85(4)	0.25(2)	0.088(8)	2.95	2.89
579.798	5	5	0	5	4	1	2.46(5)	2.45(1)	0.29(2)	0.118(9)	2.74	2.63
593.169	8	6	2	8	5	3	2.32(4)	2.42 (1)	0.267(9)	0.110(4)	2.19	2.22
626.473	7	6	1	7	5	2	2.56(5)	2.690(8)	0.43(4)	0.16(2)	2.42	2.51
650.376	4	4	1	4	3	2	2.52(4)	2.63(4)	0.18(2)	0.068(9)	2.86	2.74
665.395	4	2	2	4	1	3	2.67(3)	2.766(3)	0.237(2)	0.0857(8)	2.56	2.60
687.304	2	0	2	1	1	1	2.77(5)	2.847(7)	0.24(1)	0.084(4)	3.13	2.72
689.121	9	7	2	9	6	3	2.10(3)	2.240(9)	0.19(1)	0.085(5)	2.10	2.07
733.025	10	7	3	10	6	4	2.07(4)	2.36(3)	0.31(3)	0.13(1)	1.83	1.92
736.033	2	1	2	1	0	1	2.84(5)	2.71(4)	0.208(6)	0.077(3)	3.01	2.72
746.72	9	6	3	9	5	4	2.21(4)	2.27(2)	0.246(1)	0.108(1)	1.96	1.96
747.301	3	2	2	3	1	3	3.20(3)	2.760(9)	0.224(1)	0.0812(6)	2.75	2.81
748.240	6	6	0	6	5	1	2.27(3)	2.36(1)	0.162(4)	0.069(2)	2.64	2.67
749.433	5	5	1	5	4	2	2.47(2)	2.478(8)	0.174(5)	0.070(2)	2.74	2.63
782.344	11	8	3	11	7	4	2.07(4)	2.11(2)	0.189(9)	0.090(5)	1.71	1.94
794.222	8	7	1	8	6	2	2.17(5)	2.21(4)	0.139(3)	0.063(3)	2.34	2.39
800.850	5	4	2	5	3	3	2.61(4)	2.66(2)	0.18 (1)	0.068(4)	2.56	2.52
804.729	8	5	3	8	4	4	2.32(5)	2.220(9)	0.158(9)	0.071(4)	2.03	2.10

The N₂-broadening coefficients were also determined using the empirical model of Starikov (Eqs. (12) and (13)) with the same method applied to the self-broadening. The best parameters determined from the fit of our GS N₂-broadening coefficients are listed in Table 5 with the parameters obtained in Ref. [46] with the coefficients measured in the v₂ band [15]. In our study, the full set of x_k parameters was adjusted and correctly determined contrary to those of Ref. [46] where some parameters were not include in the fit (e.g x_{11}) and some parameters were not correctly determined (e.g x_{30} and x_{41}). The empirical N₂-broadening coefficients determined from our x_k set of parameters are shown in Table 4 (column γ_{GS}^{Emp}). For N₂-broadening, the mean square difference between $\gamma_{0,v}^{Exp}$ and γ_{GS}^{Emp} is evaluated to 0.13 MHz/mbar. As for the self-broadening, this difference is significantly reduced compared to those obtained with the γ_{RB}^{Cal} calculated coefficients.

Table 5. Parameters adjusted with the Starikov model (Eqs. (12) and (13)) for the empirical determination of the N₂-broadening coefficients. x_{1k} and x_{2k} are in unit of cm⁻¹.atm⁻¹, x_{3k} , x_{4k} and x_{5k} are unitless. Quoted errors in parentheses are in the last digit given and are equal to one standard deviation (1 SD).

Parameter	This study (GS)	Ref. [46] (v ₂)
x_{10}	-0.10262(2)	-
x_{11}	0.01778(4)	-
x_{12}	-0.000505732(3)	-
x_{20}	0.21667(4)	0.096(2)
x_{21}	-0.02119(2)	-0.0014(2)
x_{22}	0.0005253(2)	-
x_{30}	0.04042(3)	0.1 (4)
x_{31}	-0.00496(2)	-0.0007(2)
x_{32}	0.00014697(3)	-
x_{40}	-6.499(4)	-1.32(8)
x_{41}	9.688(2)	0.3(4)
x_{42}	-1.2698(1)	-
x_{50}	-0.061(2)	-0.016(2)
x_{51}	0.0047421(7)	0.00096(7)
x_{52}	-0.00052909(5)	-

The rotational dependences of the N₂-broadening coefficients on K_a'' and on J'' are presented in Fig. 7. Compared to the self-broadening case (see Fig. 5), the differences between γ_{GS}^{Emp} determined in this study and $\gamma_{v_2}^{Emp}$ are less pronounced especially for the dependence on J'' .

We can notice that for the N₂-broadening, the rotational dependence is more significant on J'' than on K_a'' where a roughly linear decreasing of γ_{0,H_2S-N_2} for the ^RQ sub-branch is observed in agreement with previous studies shown in Fig. 6 and the air-broadening rotational dependence associated with the tabulated HITRAN-2016 parameters [12, 26].

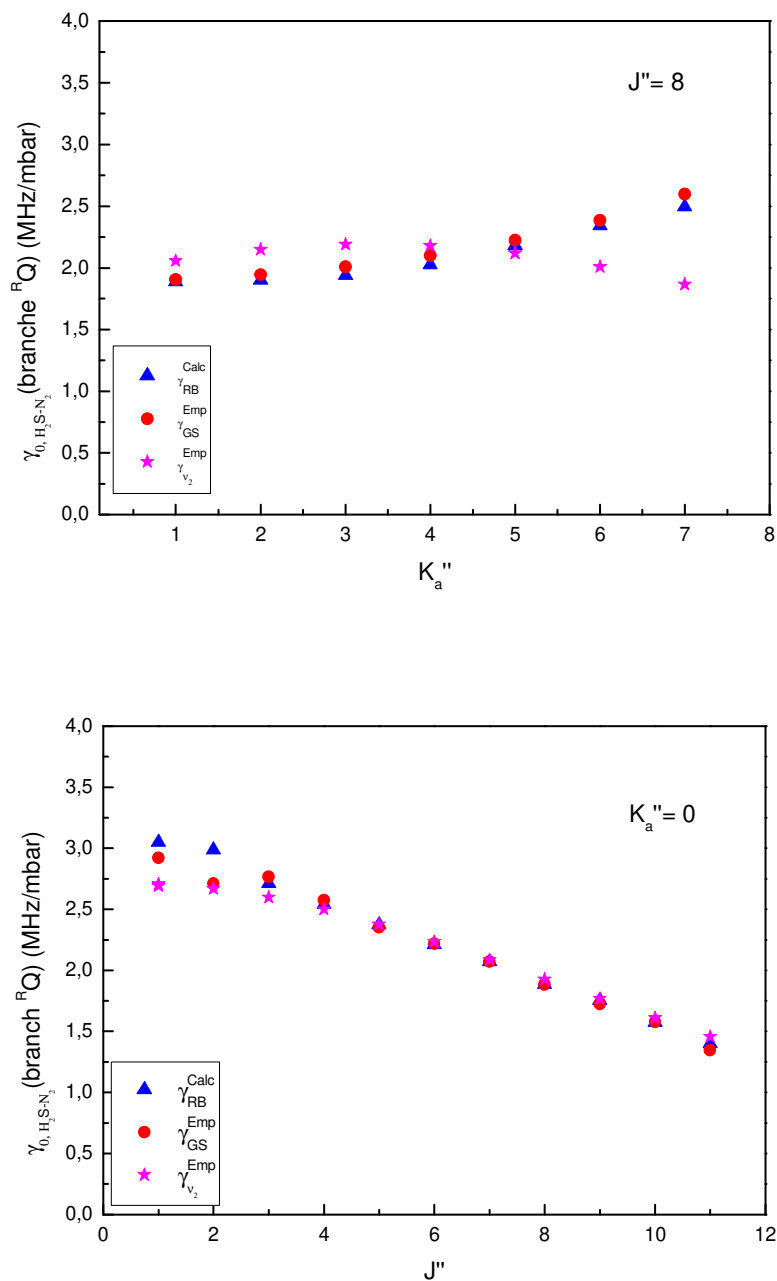


Fig. 7 Examples of K_a'' and J'' rotational dependences of the H₂S N₂-broadening coefficients: K_a'' dependence for $J''=8$ (top) and J'' dependence for $K_a''=0$ (bottom). 3 sets of data are compared: the calculated coefficients with the RB formalism γ_{RB}^{Cal} and the empirical coefficients γ_{GS}^{Emp} and $\gamma_{v_2}^{Emp}$ adjusted respectively on the GS rotational transitions (this study) and on the v_2 rovibrational transitions from Ref. [15].

5. Conclusion

With the use of a frequency modulated THz spectrometer and of a dedicated fitting procedure, we have measured, in the 150 – 850 GHz frequency range, a new set of precise self- and N₂-broadening coefficients at $T = 295$ K for 30 and 27 GS rotational lines (mainly in the ^RQ sub-branch) of H₂S, respectively. Benefiting from good S/N ratios (S/N>500) of the measured smm lines, systematic departures from the usual Voigt model have been stated for all the rotational lines studied in this work. The SD-Voigt profile yields to the best residuals and linear pressure dependences of the broadening and speed dependence parameters are observed in the 0.1 – 0.8 mbar pressure range. In addition, the prediction of self- and N₂-broadening coefficients using the RB formalism with bi-resonance functions leads to a reasonable agreement with the measured values and reproduces the observed rotational dependencies. The experimental broadenings show general consistency with the IR rovibrational measurements performed by other authors, with no evidence of systematic differences between the vibrational states. Using an empirical model developed by Starikov [46, 51] and tested on H₂S broadening coefficients determined in the ν_2 fundamental band, the rotational dependences in J'' and K_a'' were analyzed with the self- and N₂-broadening coefficients measured in this study. Some differences are observed compared to the IR results: (i) our $\gamma_{0,\text{self}}^{\text{RQ}}$ coefficients monotonically decreases with K_a'' and have a weaker variation with J'' at $K_a'' = 0$; (ii) for the N₂-broadening, the rotational dependence is more significant on J'' than on K_a'' with a roughly linear decreasing of γ_{N_2} in the ^RQ branch.

The present release of HITRAN does not account for branch and band type interactions, Dicke narrowing, speed dependence effects or line mixing. Such fine effects may be quantified in pure rotation provided sufficient SNR is available. In the case of line mixing higher pressures measurements are required and a specific baseline treatment. Our results suggest that the existing data base information may be extended with some confidence to the H₂S rotational measurements considered in this work. To go further in this study, it will be interesting to introduce speed-dependent collisional parameters in the semi-classical calculations [53].

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