

Simultaneous determination of bisphenol A and its chlorinated derivatives in human plasma: Development, validation and application of a UHPLC–MS/MS method

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1 Simultaneous determination of bisphenol A and its chlorinated derivatives in

2 human plasma: development, validation and application of a UHPLC-MS/MS

3 method

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27 Abstract

28 Bisphenol A (BPA) is a well-known ubiquitous chemical found in polycarbonate, polysulfone and epoxy resins, used in mass production for many consumer products. BPA exhibits 29 30 endocrine disruptor properties that can potentially induce adverse health effects. In aquatic environments, it can react with chlorine to produce chlorinated derivatives (ClxBPAs). 31 32 ClxBPAs exhibit oestrogenic activity 10 to 105 times higher than BPA itself. Assessing human exposure to endocrine disrupting chemicals is mandatory to assess health risk. 33 Blood, as well as urine matrix, are commonly used to perform human biomonitoring. We 34 therefore developed, fully validated and applied a method based on Ultra High Performance 35 36 Liquid Chromatography couples to a Triple Quad Mass Spectrometer to determine BPA, monochlorobisphenol A (MCBPA), dichlorobisphenol A (DCBPA), trichlorobisphenol A 37 38 (TCBPA) and tetrachlorobisphenol A (TTCBPA) in human blood plasma. The European Medicines Agency guidelines for bioanalytical method validation have been applied. 39 40 Precision and trueness of the method were <15% at medium and high levels of quality control and <20% at the limits of quantification (LOQs). The LOQs were settled at 0.1 ng/mL 41 for BPA, 0.02 ng/mL for TTCBPA and 0.005 ng/mL for MCBPA, DCBPA, and TCBPA. The 42 analytical method was applied to ten patients suffering from end stage renal disease. BPA 43 44 was quantified in all ten patients while MCBPA, DCBPA and TTCBPA were determined in three and TCBPA in four. In conclusion, we have successfully developed a highly sensitive 45 46 method to determine BPA and ClxBPAs in human plasma. Thanks to this method, for the first time, we could demonstrate ClxBPAs occurrence in human blood. 47

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Keywords: bisphenol A; chlorinated bisphenol A; endocrine disrupting chemicals; human
plasma biomonitoring; UHPLC-MS/MS;

52 **1. Introduction**

In 2017, the member state committee of the European CHemicals Agency (ECHA) 53 unanimously agreed that bisphenol A (BPA), exhibits endocrine disruptor properties (ECHA, 54 55 2017). Epidemiological studies have demonstrated that BPA exposure was related to diabetes (Kim and Park, 2013), obesity (Carwile and Michels, 2011), cardiovascular diseases 56 57 (Gao and Wang, 2014) and disorders on reproductive systems (Lee et al., 2015). These adverse health effects combined with the fact that BPA is a ubiquitous compound render this 58 problem a priority to human health (Bergman et al., 2013). BPA is produced in one of the 59 highest volumes of chemicals worldwide. BPA is widely used in the production of 60 61 polycarbonate, polysulfone and epoxy resins (ANSES, 2013; FAO/WHO Expert, 2010) that are applied in many common consumer products such as flasks, paints, varnishes, 62 63 manufacturing sheets and thermal paper (Dodson et al., 2012; Vandenberg et al., 2007). It can also be found in aquatic environments such as surface waters (Fromme et al., 2002), 64 65 and tap water (Dupuis et al., 2012). During drinking water treatment, BPA is in contact with chlorine used as a disinfectant to produce tap water. This may result in the addition of 66 chlorine atoms to the phenolic rings leading to production of BPA chlorinated derivatives 67 (ClxBPAs) (Fan et al., 2013). 68

69 Most human biomonitoring research has focused on BPA only, but it is known that humans may also be exposed to ClxBPAs via drinking water (Dupuis et al., 2012) as mentioned 70 above, and via flame retarded products (Kitamura et al., 2002). Furthermore, ClxBPAs 71 exhibit oestrogenic activity 10 to 105 times higher than BPA itself (Hu et al., 2002; Mutou et 72 al., 2006). In vitro studies have revealed the toxicity of ClxBPAs such as genotoxicity, 73 regulation of adipogenesis (Riu et al., 2011), cancer (Rivas et al., 2002) and endocrine 74 disruption (Viñas et al., 2013). Few epidemiological studies have demonstrated that ClxBPAs 75 exposure is related to metabolic disorders such as obesity and type II diabetes (Andra et al., 76 77 2015; Andra and Makris, 2015).

78 Assessing human exposure to environmental pollutants is a key-point for health risk assessment. It is essential to determine human exposure to BPA and ClxBPAs using human 79 80 biomonitoring. Determination of environmental pollutants in human body fluids, most 81 commonly in urine or blood, is considered as the gold standard method (Zidek et al., 2017). Regarding polar or easily metabolizable endocrine disrupting chemicals (EDCs), studies 82 have focused on urine matrix with the advantage of non-invasive sampling method and to the 83 large volume available. However, to our knowledge CIxBPAs toxicokinetics has not yet been 84 85 characterized. Furthermore, urinary data may require creatinine or specific gravity adjustment in order to take into account urine dilution. Moreover, urine collection may be 86 compromised in newborns or anuric patients. In contrast, blood is the major body fluid 87 distributing exogenous compounds throughout the body. Therefore, micropollutants 88 89 determination in blood is highly relevant for human biomonitoring and for health risk assessment. Hence, in this study we have focused on human plasma, a matrix available in 90 all populations and commonly used in large-scale human biomonitoring even if lower 91 92 micropollutants concentrations may be found in this matrix (Calafat et al., 2013).

93 Regarding BPA, most authors have included a deconjugation step in their analytical method 94 leading to determination of "total" BPA without distinguishing conjugated from unconjugated 95 form (La Rocca et al., 2014; Rönn et al., 2014; Savastano et al., 2015). On the other hand, 96 our work has focused on parent compounds, BPA and CIxBPAs, since CIxBPAs metabolism 97 is still unknown. Moreover, BPA glucuronide does not competitively displace [3H]-17betaestradiol from any of the estrogen receptors (Cho et al., 2017; FAO/WHO Expert, 2010; 98 99 Matthews et al., 2001), and unconjugated BPA is consequently assumed to be the form 100 leading to toxic effects.

101 To our knowledge, only one study has described simultaneous quantification of BPA and 102 ClxBPAs in human serum, but the validation of the reported analytical method does not 103 comply with international guidelines (Liao and Kannan, 2012).

- 104 Therefore, the aim of this work was to develop, validate and apply Ultra High Performance
- 105 Liquid Chromatography coupled to tandem Mass Spectrometry (UHPLC-MS/MS) method for
- simultaneous determination of BPA and ClxBPAs in human plasma.

107 **2. Materials and methods**

2.2.

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2.1. Chemicals and standards

Bisphenol A and deuterated BPA (BPA-d₁₆) were purchased from Sigma-Aldrich® (St Louis, USA). Monochlorobisphenol A (MCBPA), 2,2'-dichlorobisphenol A and 2,6'-dichlorobisphenol A (DCBPA), trichlorobisphenol A (TCBPA), tetrachlorobisphenol A (TTCBPA) and deuterated DCBPA standard (DCBPA-d₁₂) were obtained from @rtMolecule® (Poitiers, France). Methanol (MeOH) and water were purchased from Carlo Erba (Val de Reuil, France). All solvents and reagents were of UHPLC grade and free of BPA and ClxBPAs. Polypropylene or glass materials were used to avoid BPA contamination.

116 117 Preparation of stock solutions, working standard solutions and calibration standards

118 Stock solutions were prepared at 10 μ g/mL for BPA and for BPA-d₁₆, and at 2 μ g/mL for 119 ClxBPAs and for DCBPA-d₁₂ in MeOH. These solutions were stored at -20°C. For each experiment, two working solutions were prepared extemporaneously in MeOH: the analytical 120 standard solution (AS) at 32 ng/mL and 6.4 ng/mL, for BPA and ClxBPAs respectively, the 121 internal standard solution (IS) at 16 ng/mL and 3.2 ng/mL for BPA-d₁₆ and for DCBPA-d₁₂, 122 123 respectively. The AS solution was then diluted in MeOH to obtain calibration standards from 32 ng/mL to 1 ng/mL for BPA, from 6.4 to 0.05 ng/mL for MCBPA, DCBPA and TCBPA, and 124 from 6.4 to 0.2 ng/mL for TTCBPA. These solutions were used to prepare calibration 125 standards in plasma. Human plasma, collected from anonymous donors was used for the 126 127 preparation of calibration standards in plasma and quality controls (QCs). Three standard solutions (SS) were carried out for each analyte in MeOH (16, 2 and 1 ng/mL for BPA; 3.2, 128 0.4, 0.05 for MCBPA, DCBPA, TCBPA; 3.2, 0.4 and 0.2 for TTCBPA) to prepare three levels 129 of quality controls. 130

131 2.3. Samples

Human plasma, collected using lithium heparin tubes free of BPA, from patients suffering from end stage renal disease and hospitalized at the University Hospital of Poitiers were used to apply our analytical method. Informed and written consent was obtained from all subjects enrolled in the study. All samples were stored at -20 °C.

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2.4. Extraction and optimization

Each plasma sample (500 µL) was spiked with 50 µL of IS. Then, 50 µL of calibration 137 standard were added to obtain final concentrations ranging from 3.2 ng/mL to 0.1 ng/mL for 138 BPA, from 0.64 to 0.02 ng/mL for TTCBPA, and from 0.64 to 0.005 ng/mL, for MCBPA, 139 DCBPA, and TCBPA. For blank and patient samples, AS was replaced by 50 µL of MeOH. 140 Then, 2.5 mL of water were added. Samples were shaken for one minute before performing 141 142 solid phase extraction (SPE). The SPE procedure for clean-up samples was performed using glass Oasis HLB® Cartridges (200 mg/5 mL Waters®, Milford, USA). Cartridges were 143 conditioned with MeOH and then equilibrated with water. Samples were loaded on the wet 144 cartridges, and then washed (MeOH/water, 30/70, v/v, 4 mL) and dried. Elution was 145 performed with MeOH (4 mL). Extracts were evaporated to dryness at 40°C under a gentle 146 147 nitrogen stream. Residues were dissolved ((MeOH/water 30/70, v/v, 500 µL) and vortexed for 10 s. Finally, 20 µL of extract were injected into the UHPLC–MS/MS apparatus. 148

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2.5. Liquid chromatography

The UHPLC Shimadzu® Nexera X2 system (Kyoto, Japan) was equipped with an Acquity CSH C18 (1.7 μ m particle size, 2.1 x 100 mm, Waters®, Milford, USA) column maintained at 40°C, and a binary mobile phase was delivered in the gradient mode, at a flow rate of 350 μ L.min⁻¹. Water (A) and MeOH (B) were used as mobile phase solvents. Gradient started at 30% of B and linearly increased to 90% of B (0.5 to 7 min) and then to 99% of B (7 to 7.5 min); maintained at 99% of B (7.5 to 10 min); decreased linearly to 30% of B (10 to 10.5 min) and then maintained at 30% of B (10.5 to 13 min) to re-equilibrate the column.

157 2.6. Mass spectrometry

Quantification was performed using an API 6500+ mass spectrometer (ABSciex®, Concord, 158 Canada) equipped with an electrospray ionization (ESI) interface, operating in negative 159 160 ionization mode. Ion source and MS/MS conditions were optimized by direct infusion of 1 μ g/mL solutions of the individual analytes into the mass spectrometer at 10 μ L/min. Ions 161 162 were analyzed in multiple reaction monitoring mode (MRM) using two specific transitions of each precursor ion. The first transition was used for quantification while the second was used 163 for confirmation. Dwell and cycle times were optimized using the Scheduled MRM® 164 algorithm. Data were acquired using Analyst® software, and processed using MultiQuant® 165 166 software.

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2.7. Method validation

Analytical method was validated according to EMA guidelines (European Medecines Agency
Science Medicines Health, 2011). Validation parameters included selectivity, precision,
trueness, linearity, Matrix Effect (ME), and REcovery (RE).

Five calibration curves containing analytes and internal standards and three levels of quality
controls (QC) were prepared as described above. Linear regression was performed with 1/x
weighting factor.

To determine selectivity, for each analyte and deuterated standard we analyzed chromatograms at their corresponding retention time in background plasma samples. High selectivity was defined by the absence of any detectable peak.

Calibration curves ranged from 0.1 to 3.2 (0.1, 0.2, 0.4, 0.8, 1.6, 3.2) ng/mL for BPA, from 0.02 to 0.64 (0.02, 0.04, 0.08, 0.16, 0.32, 0.64) ng/mL for TTCBPA and from 0.005 to 0.64 (0.005, 0.01, 0.02, 0.04, 0.08, 0.16, 0.32, 0.64) ng/mL for MCBPA, DCBPA and TCBPA. In addition, quantification of QC (0.1, 0.2, 1.6 ng/mL for BPA, 0.02, 0.04, 0.32 ng/mL for TTCBPA and 0.005, 0.04, 0.32 ng/mL for MCBPA, DCBPA and TCBPA), was systematically performed during sample analyses. Precision (expressed as relative standard deviation) and

trueness (expressed as percentage of target concentration) were assessed using five replicates of QC assayed the same day (intraday) and over five different days (inter-day).

185 Relative standard deviation (RSD) was supposed to be lower than 15 % except at the limit of 186 quantification (LOQ) of which the value was supposed to be lower than 20 %. Trueness was 187 supposed to range from 85 to 115 % except at the LOQ, of which the value was supposed to 188 range from 80 to 120 % (European Medecines Agency Science Medicines Health, 2011).

The limit of detection (LOD) was defined for each batch of analyses as three times the standard deviation of the mean peak area (n=5) obtained using blank plasma samples ("Network of reference laboratories and related organisations for monitoring and biomonitoring of emerging environmental pollutants (NORMAN) - Protocol for the validation of chemical and biological monitoring methods," 2009). A compound was considered to be detectable in a sample if the peak area was greater than LOD.

In accordance with guidelines, the limit of quantification (LOQ), was set at the lowest level of calibration standard (0.1 ng/mL for BPA, 0.02 ng/mL for TTCBPA, and 0.005 ng/mL for MCBPA, DCBPA and TCBPA), and fully validated using QC prepared with plasma spiked at this concentration.

Linearity of the calibration curve was assessed using correlation coefficients (r²) and contamination was estimated using intercept value. Linearity was demonstrated when the correlation coefficient was greater than 0.99.

To determine ME and RE of the extraction step (Matuszewski et al., 2003), several levels of QCs were prepared according to the following procedures:

- 100 μL of IS and 100 μL of SS were spiked to 800 μL of mobile phase to obtain
 Matrix effect solution (MES) solution. Then, 20 μL were injected into the UHPLC MS/MS system to obtain MS/MS response A, expressed using the AS peak area and
 the ratio AS peak area / IS peak area (area ratio A).

208 - 500 μ L of MES were used to dissolve blank plasma extract obtained following the 209 extraction procedure described above. Then, 20 μ L were injected into the UHPLC-

210 MS/MS system to obtain MS/MS response B, expressed using the AS peak area and 211 the ratio AS peak area / IS peak area (area ratio B).

50 μL of IS and 50 μL of SS were spiked to 400 μL of plasma and then extracted
following the procedure described above. Then, 20 μL were injected into the UHPLCMS/MS system to obtain MS/MS response C, expressed using the ratio AS peak area
/ IS peak area (area ratio C).

The percentages of ME and RE were assessed as follow:

217 ME (%) = (Area B / Area A) x 100

218 IS-corrected ME (%) = (Area ratio B / Area ratio A) x 100

219 RE (%) = (Area ratio C/ Area ratio B) x 100

220 2.8. Application to patients suffering from end stage renal disease

We have randomly chosen ten plasma samples from patients suffering from end stage renal disease and hospitalized at the University Hospital of Poitiers. Indeed, for these anuric patients, urine is not available therefore it is mandatory to determine EDCs in another matrix such as blood. The study was approved by the French Ethics Committee (2016-A00475-48), and patients gave their informed consent after being fully informed. After centrifugation plasma samples were stored at -80°C up to 6 months before analysis. Plasma was collected using lithium heparin tubes free of BPA. 228 **3. Results**

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3.1. Ultra High Performance Liquid Chromatography and MS/MS conditions

The optimal conditions for the electrospray ionization were as follows: ion spray voltage set at -4500V, temperature set at 650°C, curtain gas set at 30.0 psi, collision gas set as high, ion source gas 1 set at 40.0 psi and ion source gas 2 set at 60.0 psi. Ion source and specific MS/MS conditions were optimized and results are listed in Table 1.

Table 1. Descriptive table of retention times, precursor ions, fragment ions and MS/MS optimization conditions (declustering potential, DP; entrance potential, EP; collision energy, CE; collision cell exit potential, CXP) for all analytes. For CE and CXP, first and second values are for quantification and confirmation ions, respectively.

Analyte	Retention	Precursor	Quantification	Confirmation	DP	EP	CE	CXP
	time (min)	ion (m/z)	ion (m/z)	ion (m/z)	(V)	(V)	(V)	(V)
BPA	6.5	226.967	211.800	132.800	-65	-10	-24 /	-19 /
							-34	-11
BPA-dia	6.4	241.078	222.900	141.900	-90	-10	-28 /	-25 /
Britan	0.1						-34	-15
MCBPA	7.0	260.896	181.700	209.700	-120	-10	-38 /	-19 /
	7.0						-30	-21
DCBPA	75	294.869	243.700	215.800	-100	-10	-32 /	-25 /
	7.5						-42	-23
TCBPA	7.9	328.713	249.700	277.700	-85	-10	-42 /	-15 /
							-34	-29
TTCBPA	8.2	364.794	313.700	285.600	-110	-10	-36 /	-33 /
							-44	-29
DCBPA-	7 /	306 988	224 900	252 800	-120	-10	-42 /	-23 /
d 12	7.4	500.900	224.900	232.000	-120	-10	-34	-25

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Chromatographic conditions were optimized and for each target analyte we obtained high
peak resolution (Figure 1). Retention times for these five EDCs are reported in Table 1.

242 Figure 1. Examples of chromatograms obtained for medium quality control (0.2 ng/ml for





For each analyte, in blank plasma samples (AS free), no significant peak of ClxBPAs was detected at the corresponding retention time (Figure 2), in accordance with high selectivity of the method. Due to BPA contamination of human plasma, a non-significant peak of BPA was observed in blank samples (Figure 2).



Figure 2. Examples of chromatograms obtained for blank plasma.

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3.2.2. Precision and trueness

The accuracy of the method was fully demonstrated. Indeed, RSD did not exceed 17% at LOQ, and 12% at the other levels for intra and inter-day analysis. Trueness ranged from 83 to 115 at LOQ, and from 85 to 107 % at the other levels for intra and inter-day analysis (Table 2).

259	Table 2. Intra and inter-day precision and trueness for each analyte at each level of QC
260	(n=5), Matrix Effect (ME) and Recovery (RE) data for high level quality control (n=3) in
261	human plasma for all analytes.

Compound	Concentration of QC (ng/mL)	Precision (%)		Trueness (%)		Matrix	IS- corrected	Recovery
Compound		Intra-day	Inter-day	Intra-day	Inter-day	(%)	Matrix effect (%)	(%)
BPA	0.1 (LOQ)	11	11	94	101	56	86	
	0.2	9	5	107	99			
	1.6	3	4	100	97	54	84	92
MCBPA	0.005 (LOQ)	12	16	92	104	69	92	
	0.04	12	7	104	99			
	0.32	9	12	102	94	56	94	90
DCBPA	0.005 (LOQ)	12	14	92	100	81	110	
	0.04	9	11	100	100			
	0.32	3	7	89	105	61	105	48
TCBPA	0.005 (LOQ)	12	17	92	83	99	133	
	0.04	12	9	101	91			
	0.32	11	6	91	100	79	134	75
TTCBPA	0.02 (LOQ)	11	8	115	108	77	103	
	0.04	10	11	99	98			
	0.32	10	12	85	95	67	115	31

3.2.3. Linearity and intercept

BPA and CIxBPAs calibration curves showed appropriate linearity (r²>0,99) (Table 3). 265 Furthermore, due to contamination of human plasma, an intercept value of 0.320 ± 0.113 266 was obtained for BPA calibration curves (Table 3). Indeed, BPA was detected in plasma from 267 anonymous donors used for standards and quality controls (Figure 2), however it was always 268 at a level markedly lower than the limit of quantification (0.1 ng/mL) and has been taken into 269 account in the calculation method. To summarize, the BPA/IS peak area ratio of blank 270 plasma was daily subtracted from the BPA/IS ratio of each standard used to construct the 271 272 calibration curve.

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Table 3. Mean (+/- SD) correlation coefficient (r²), slope, and y-intercept of calibration curves

	BPA	MCBPA	DCBPA	TCBPA	TTCBPA
Correlation coefficients	0.996 ±	0.997 ±	0.998 ±	0.997 ±	0.990 ±
r² (mean +/- SD)	0.003	0.003	0.001	0.002	0.007
Slope (mean +/ SD)	1.283 ±	1.922 ±	4.975 ±	3.091 ±	0.302 ±
Slope (mean +/- SD)	0.235	0.487	0.353	0.160	0.066
Intercept (mean area	0.320 ±	0.007 ±	0.003 ±	0.005 ±	0.002 ±
ratio +/- SD)	0.113	0.004	0.002	0.001	0.004

275 prepared with plasma (n=5).

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3.2.4. Recovery and matrix effect

Results concerning ME and RE are reported in Table 2. ME were low to moderate from 56 to 99% and from 54 to 79 % for low and high levels of QC, respectively. However, they were satisfactorily corrected with BPA- d_{16} and DCBPA- d_{12} . RE were good to moderate from 92 to 31% (Table 2).

282 3.3. Application to human plasma samples

The method was applied to human blood samples from patients suffering from end stage renal disease. Figure 3 shows a chromatogram obtained from a human plasma sample (n° 10) in which BPA and ClxBPAs were found.

Figure 3 Examples of chromatograms and concentrations obtained for BPA and ClxBPAs in human plasma sample (ng/mL) (detected but not quantified, > LOD < LOQ).



BPA was quantified in all ten patients with concentrations ranging from 0.266 to 86.831 ng/mL while MCBPA, DCBPA and TCBPA were quantified in two, one and one patient, respectively. TTCBPA was detected in three out of the ten patients (Table 4). Some patients exhibit multi exposure to BPA and ClxBPAs while others are only exposed to BPA (Table 4).

Sample	BPA	MCBPA	DCBPA	TCBPA	TTCBPA
1	0.329	ND	ND	ND	> LOD < LOQ
2	0.428	> LOD < LOQ	> LOD < LOQ	> LOD < LOQ	ND
3	0.458	0.040	> LOD < LOQ	0.006	> LOD < LOQ
4	0.266	ND	ND	> LOD < LOQ	ND
5	5.625	ND	ND	ND	ND
6	41.007	ND	ND	ND	ND
7	0.379	ND	ND	ND	ND
8	9.622	ND	ND	ND	ND
9	86.831	ND	ND	ND	ND
10	0.518	0.024	0.006	> LOD < LOQ	> LOD < LOQ

Table 4. Quantification of unconjugated bisphenol A and its chlorinated derivatives in human
patient plasma (ng/mL) (not detected, ND; detected but not quantified, > LOD < LOQ).

297 **4. Discussion**

298 We have developed and validated (according to EMA guidelines (European Medecines Agency Science Medicines Health, 2011)) an original UHPLC-MS/MS method for 299 300 simultaneous determination of BPA and CIxBPAs in human plasma. According to the results the method was ultrasensitive and reliable. Indeed, although moderate matrix effects have 301 302 been observed, using deuterated internal standards, high-accuracy of the method has been demonstrated. Several reliable methods for simultaneous determination of BPA and CIxBPAs 303 have been already reported by our team but due to the specific properties of the plasma 304 matrix we had to develop entirely original extraction and separation procedures (Cariot et al., 305 306 2012; Grignon et al., 2016; Venisse et al., 2019).

So far, the toxicokinetics of CIxBPAs remains unknown, even though adverse health effects 307 308 have been suggested. Moreover, the few published studies have essentially provided data in urine samples (Andra and Makris, 2015; Kalyvas et al., 2014; Vela-Soria et al., 2014; 309 310 Venisse et al., 2014). This may be due to a lack of analytical method, particularly regarding human plasma matrix. On the other hand, our method offers reliable determination of 311 ClxBPAs in human plasma, which is mandatory in toxicokinetic investigative studies. Like 312 most of EDCs, BPA and ClxBPAs are found at a trace level and exhibit their oestrogenic 313 activity at a very low concentration, acting according to non-monotonic dose-response 314 relationships (Bergman et al., 2013; Lagarde et al., 2015). To comply with this requirement, 315 our method provides very low LOQs (0.1 ng/mL for BPA, 0.02 ng/mL for TTCBPA and 0.005 316 ng/mL for MCBPA, DCBPA, and TCBPA). 317

318 Cocktail effects are key points of EDCs toxicology since synergistic or antagonistic 319 mechanisms have been described for these micropollutants (Gaudriault et al., 2017). Hence, 320 using our method, simultaneous determination of five EDCs in human plasma may be 321 performed. This "cocktail-exposure" assessment better reflects real-life situation and 322 provides more relevant data, thereby heightening the statistical power of epidemiologic 323 studies. Future HBM studies for EDCs should take into account their cocktail effects using

analytical methods simultaneously assessing several EDCs as does the method developed
in this work. Finally, simultaneous determination of EDCs is mandatory in order to assess the
relationship between EDCs exposure and health outcomes.

To our knowledge, only Liao *et al.* have reported simultaneous determination of BPA and ClxBPAs in human blood samples (Liao and Kannan, 2012). However, TTCBPA determinations were not included and the method requires two solid phase extractions, thereby increasing duration and analysis cost. In addition, the method developed by Liao *et al.* is ten times less sensitive than our method, since their LOQs have been settled at 0.05 ng/mL for MCBPA DCBPA and TCBPA and therefore did not allow to detect ClxBPAs in human samples.

Another method has been developed to determine ClxBPAs in human plasma but without including BPA. This method is based on gas chromatography technology, requiring a derivatization step (del Olmo et al., 2005). Moreover, the method requires a large blood sample volume (about 10 mL) while providing a poor sensitivity; indeed, their LOQs were 160 to 900 times higher than LOQs obtained using our method. Last but not least, this method has never been applied to determine ClxBPAs in human blood samples.

The method developed in our study is suitable to assess ultratrace levels of BPA and ClxBPAs in human plasma. Indeed, our method enabled us to determine BPA as well as, for the first time, ClxBPAs concentrations, in several human blood samples.

343

344 **5.** Conclusion

345 In conclusion, we have successfully developed a highly sensitive method to determine BPA

346 and ClxBPAs in human plasma. Thanks to this method, for the first time, we could

347 demonstrate ClxBPAs occurrence in human blood. Finally, this method is suitable to perform

348 large-scale human biomonitoring of these EDCs.

349

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