

# A fully automated on-line salting-out assisted liquid-liquid extraction capillary electrophoresis methodology: Application to tyrosine kinase inhibitors in human plasma

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1	A fully automated on-line salting-out assisted liquid-liquid extraction capillary electrophoresis
2	methodology: Application to tyrosine kinase inhibitors in human plasma
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#### Abstract

A fully automated analytical methodology combining salting-out assisted liquid-liquid extraction (SALLE) and capillary electrophoresis (CE) for the analysis of three Tyrosine Kinase Inhibitors (TKIs) in plasma samples is proposed. The automated methodology, called A-SALLE-CE-UV, makes full use of the advantages of both techniques by combining desalting, protein precipitation, automated liquid-liquid extraction, in-line CE stacking and electrophoretic separation of analytes in plasma samples in a fully integrated way. At first, the capillary is used to deliver appropriate micro-volumes of extraction agent solutions (acetonitrile, salt) in the plasma sample. ACN and salting-out agent (NaCI) solutions are added by pressure from outlet vials into the sample vial (inlet) containing human plasma sample spiked with the three tested TKIs. After addition of both ACN and NaCI solutions, mixing is achieved by generating air bubbles leading to a two phases separation and extraction of TKIs in the upper mostly organic phase (ACN). The upper phase containing the TKIs is then injected and analyzed by CE-UV. Due to the presence of ACN, the analytes are stacked in-line and successfully separated in the same capillary. The results obtained in terms of limit of detection (LOD), limit of quantification (LOQ), sensitivity enhancement factor (SEF), repeatability and linearity demonstrate the applicability of the proposed method for possible therapeutic drug monitoring (TDM) of TKIs.

#### 1. Introduction

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Capillary electrophoresis (CE) is a separation technique extensively used for the analysis of a wide range of drugs in biological matrices such as plasma [1]. However, the presence of interfering compounds (e.g. protein, salts) in the sample can affect the analytical performances of CE methodologies causing peak broadening, baseline distortion and a lack of specificity [2]. Therefore, sample preparation techniques such as liquid-liquid extraction (LLE) and solid-phase extraction (SPE) are frequently used prior to CE analysis to remove proteins, desalt, concentrate the analytes of interest and obtain efficient electrophoretic peaks [3,4]. However, these methods are based on the use of complex, long procedure and expensive sample preparation protocols (multi-steps) which are not suitable for routine analysis. In the last decade, several improvements of extraction techniques have been reported to address these problems, such as the development of solid phase microextraction (SPME) and liquid phase microextraction (LPME) techniques [5,6]. More recently, new LLE technique based on salting-out assisted liquid-liquid extraction appeared to be well suited for the determination of drugs in plasma samples as it allows desalting, deproteinization and extraction of analytes in the same procedure [7]. SALLE is based on the addition of salts (e.g. sodium chloride (NaCl)) into a sample solution composed of a miscible mixture of water and an organic solvent (such as acetonitrile (ACN)), inducing a two phases separation (organic phase from bulk aqueous phase) [8]. SALLE can be easily coupled with CE technique (SALLE-CE-UV) since the high content of ACN in the upper phase allows the in-line CE stacking of the analytes in the capillary even for injected sample volumes exceeding 80% of the capillary volume till the detector window. Excellent peak efficiencies, good resolution and high sensitivity were obtained with SALLE-CE-UV methodology for the analysis of drugs in plasma [7]. Since a few years, analytical strategies tend towards automation and on-line combination of sample preparation procedures with separation systems such as liquid chromatography (LC), gas chromatography (GC) and CE [9-11]. Indeed, automation offers many advantages including lab efficiency, reduced human errors and lower operating cost per sample. Many works dealt with the automation of LPME techniques including single drop microextraction [12,13], dispersive liquid-liquid microextraction [9,14-17] and hollow fiber-based microextraction [18-20]. However, only two reports mention the automation of SALLE. The first one is based on the oxidation of diclofenac with potassium ferricyanide in an alkaline medium, was followed by the separation of the ACN phase from the homogeneous sample solution, the simultaneous extraction of the derivative and in-line organic-phase detection by an optical probe [21]. The second work is based on the aspiration of 1-octylamine, urine sample solution containing tetracycline and salting-out agent solution (NaCl) into a mixing chamber of a flow system followed by air bubbles mixing, resulting in isotropic solution formation [11]. After phase separation, the micellar 1-octylamine phase containing analyte is mixed with methanol and analysed by a LC-UV. Both methods require special technical modifications of analytical materials and the use of additional instruments such as external syringe or peristaltic pumps, multi-port selection valves and cone-shaped glass. This include homemade modifications, which are difficult to realize, making them not suitable for routine analysis. Herein, we describe the first demonstration of a fully automated SALLE capillary electrophoresis (CE)-UV methodology, called A-SALLE-CE-UV. The applicability of the proposed A-SALLE-CE-UV methodology was evaluated for the analysis of 3 tyrosine kinase inhibitors (TKIs) namely lapatinib ditosylate, erlotinib hydrochloride and sorafenib in human plasma samples as proof-of-concept analytes. These basic molecules are extensively used for the treatment of malignant tumors such as renal cell carcinoma, lung cancer and breast cancer [22-24]. Different factors including the effects of filling schemes of the sample vial by several solutions (ACN, salt), time/volume of mixing using air bubbles and the type of salting-out agents on the performance of A-SALLE-CE-UV methodology were optimized. Several analytical performances such as the limit of detection (LOD), the limit of quantification (LOQ), sensitivity enhancement factor (SEF), repeatability, and linearity were determined.

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# 2. Experimental:

#### 2.1. Chemicals and reagents:

All reagents were of analytical grade and used as received without further purification. Acetonitrile (ACN, purity  $\geq$  99.9%) was from Honeywell Riedel de Haen (Seelze, Germany). 6-amino caproic acid (purity  $\geq$  99%) was from Acros organics (Geel, Belgium). Ammonium acetate (purity  $\geq$  99%), potassium chloride (KCl, purity  $\geq$  99%), sodium chloride (NaCl, purity  $\geq$  99.5%), sodium hydroxide (NaOH, purity  $\geq$  99%), citric acid (purity  $\geq$  99%), dimethylsulfoxide (DMSO) were purchased from Sigma Aldrich (Saint-Quentin-Fallavie, France). Sorafenib (purity  $\geq$ 99%), erlotinib hydrochloride (purity  $\geq$ 99%) and lapatinib ditosylate (purity  $\geq$ 99%) were purchased from Molekula (Munich, Germany).

#### 2.2. Apparatus and CE software:

All CE separations were carried out using a Beckman P/ACE-MDQ instrument (Fullerton, USA) equipped with a UV detection system. 600 μL eppendorf tubes (Thermoscientific, Pittsburgh, USA) used as mixing chamber between plasma sample, ACN and NaCl stock solution. The eppendorf containing the solutions was placed in a plastic 2 ml vial (P/ACE<sup>TM</sup> MDQ, reference number 970657, France) covered with plastic cap (P/ACE<sup>TM</sup> MDQ, reference number 144656, grey, France). The cap and the crown of the eppendorf tube was cut to be adjusted in the CE plastic vial (see Figure 1S in supplementary information). 2 ml glass vials (P/ACETM MDQ, reference number 358807, France) covered with vial caps (P/ACETM MDQ, reference number 144648, red, France) were filled with other solutions. Data were acquired by 32 Karat software (Version 8.0, Beckman Coulter, Fullerton, USA). Uncoated fused-silica capillaries (50 μm ID) were obtained from Polymicro Technologies (Phoenix, USA). The length of the capillary was 31 cm (21 cm effective length to the detector).

#### 2.3. Preparation of background electrolyte and stock solutions:

The background electrolyte (BGE) was chosen based on our previous studies [7,25] and was a mixture of citric acid (1800 mM) and 6-amino caproic acid (137 mM) at pH 2.0 and ionic strength (IS) 150 mM. BGE composition with the desired IS and pH was calculated with the software "Peakmaster" 5.3 (Prague, Czech Republic, https://web.natur.cuni.cz/~gas/). Stock solutions of erlotinib hydrochloride, sorafenib and lapatinib ditosylate were prepared in DMSO solvent at the concentration of 1.0 g/L due to their low solubility in water. NaCl, ammonium acetate and KCl solutions were prepared at a concentration of 25% (m/v) in water. All stock samples were stored frozen at ~20°C. All other solutions were stored in refrigerator at 4°C.

#### 2.4. Samples preparation

- Standard sample solution: relevant volumes of distilled water and the three TKIs stock solutions were mixed in an eppendorf tube of 600  $\mu$ L to obtain a final volume of 80  $\mu$ L. Then, the eppendorf tube containing the standard solution was placed in the CE plastic vials.
- Plasma sample solution: relevant volumes of plasma solution and the three TKIs stock solutions were mixed in an eppendorf tube of 600  $\mu$ L to obtain a final volume of 80  $\mu$ L. Then, the spiked plasma was mixed during 1 hour at room temperature. Then, the eppendorf tube containing the spiked plasma solution was placed in the CE plastic vials.

#### 2.5. The A-SALLE-CE-UV procedure:

New capillaries were pre-conditioned with 1.0 M NaOH for 3.0 min under 20 psi followed by distilled water for 3.0 min under 20 psi. Then, the capillary was flushed with BGE for 5.0 min under 20 psi. The A-SALLE-CE-UV procedure (Fig. 1) consisted of a commercially available CE apparatus, a silica capillary and a 600  $\mu$ L tube containing the standard or the plasma sample placed in a plastic vial (mixing chamber). At first, the capillary was used as a tool to deliver 150  $\mu$ L of ACN and 20  $\mu$ L of NaCl from outlet vials to the inlet vial holding the 600  $\mu$ L tube filled with 80  $\mu$ L of sample containing TKIs (60 psi pressure, reverse mode) were applied (see video 1 in the supplementary information). This was followed by a mixing step

using air bubbles at 90 psi during 5.0 minutes (see video 2 in the supplementary information) leading to two phases separation and extraction of TKIs in the upper phase. The capillary was rinsed with water (60 psi during 2.0 minutes) between each addition of liquids (sample, NaCl solution and ACN) in the 600 µL tube and dried with air (30 psi during 1.0 minutes), to start each filling of the sample vial from a cleaned and empty capillary. This rinsing procedure was optimized to ensure that NaCl solution doesn't get in contact with ACN to avoid any precipitation of NaCl inside the capillary. Then, the upper phase containing TKIs was hydrodynamically injected (forward mode) in the capillary at 0.8 psi during 94 seconds to obtain an injection volume equivalent to 287 nL (70% of the capillary effective length). All separations were carried out with an applied voltage of 15 kV. The capillary was thermostated at 25°C. The capillary usage life was between 45 to 50 analysis. UV detection was performed at 254 nm. The detailed A-SALLE-CE-UV methodology is mentioned in Table 1S in supplementary information. Corrected peak areas were calculated by dividing each peak area by its corresponding migration time.

#### 3. Results and discussion

Automation and on-line combination of sample preparation with analytical separation techniques fit perfectly with the criteria required for analytical methods dedicated to routine analysis such as therapeutic drug monitoring (TDM): simple, fast and cost-effective. TDM of TKIs is required to evaluate the relation between dose and efficacy/toxicity of TKIs. As mentioned before, SALLE applies the salting-out effect to separate water-miscible organic solvent such as ACN from aqueous/biological sample. The separation will result in two phases: upper mostly organic and lower mostly aqueous. SALLE-CE-UV methodology offers a number of features for the analysis of TKIs in human plasma including high separation ability between TKIs, applicability to saline sample matrix, easy operational procedure, low consumption of organic solvent, and environmental benignity.

#### 3.1. Preliminary Trials

According to our previous work [7], SALLE methodology based on a ratio of 60/40 (v/v) ACN/sample in the presence of 2% (m/v) of NaCl gives the best compromise to obtain an efficient plasma protein precipitation, a good separation of the two phases and a moderate dilution of the samples. Moreover, a BGE composed of a mixture of 6-amino caproic acid and citric acid at pH of 2.0 with ionic strength (IS) of 150 mM allows an efficient electrophoretic separation of the TKIs. Therefore, these SALLE-CE-UV conditions were used for the automation of the overall procedure of the A-SALLE-CE-UV methodology. The first experimental investigations were realized to check if it was possible to 1/ obtain a two phases separation, 2/ successfully extract the TKIs in the upper (mostly ACN) phase and 3/ separate the TKIs in an automated manner. Three TKIs (Fig. 2) were chosen as proof-of-concept analytes. The length of the capillary was adjusted to be immersed only in the upper phase (mostly organic) of the sample after SALLE extraction. The adjustment of the capillary length passed through several steps. At first, two samples were prepared by SALLE (manually) and by A-SALLE. In order to visualize properly the two phases separation, methyl red indicator was added (Fig. 2S, supplementary results). Estimated heights of

the upper phase of both manually and automatically prepared samples was verified to be the same. Accordingly, the length of the capillary inlet was optimized and measured to be immersed only in the upper phase (Ltot = 31 cm, Leff = 21 cm). Both ends of the capillary were set at the same level to avoid siphoning effect that would affect the robustness of the proposed methodology. The filling scheme of the sample vial with the different extraction solutions (ACN, salt stock solution) allowing to automate the extraction procedure can be resumed as following: ACN/NaCl/air. ACN and NaCl stock solution were successively added from outlet vials (hydrodynamic injection, reverse mode) to the sample vial containing 160 µL of TKIs at 5 µg/mL at 80 psi during 8.5 min (300 µL) and at 80 psi during 3.5 min (40 μL) respectively. The final volume was fixed at 500 μL for preliminary trials. Mixing of the different solutions was done by generating air bubbles from an empty outlet vial at 80 psi during 5 min. After this step, a two phases separation was observed in the inlet vial confirming the possibility to automate SALLE technique. The upper mostly organic phase was then injected. An electrophoretic separation between the 3 TKIs was obtained. Maximum upper phase injected volume was 10% of the capillary effective length (Fig. 3). This was less than what was obtained by manual preparation (80 % of the capillary effective length for SALLE-CE-UV methodology [7]). This could be attributed to the difference in composition of the obtained upper phases by automated and by manual methodologies in terms of ACN, NaCl and water. Moreover, extraction efficiencies (%) and sensitivity enhancement factors (SEF) of the 3 TKIs obtained by A-SALLE-CE-UV methodology were four and ten times less than those obtained by manual SALLE-CE-UV methodology (98 – 100 % for extraction efficiencies, 144-265 for SEF), respectively. This lack of performance was mainly attributed to an insufficient mixing of ACN, NaCl and sample solutions in the inlet vial affecting both extraction efficiency (%) and sensitivity. However, similar electrophoretic profiles were obtained from A-SALLE-CE-UV and manual methodology in terms of migration times. Good repeatability (n = 3) were obtained for the 3 TKIs with RSD of less than 0.7% and of less than 3.6% for migration times and corrected peak areas respectively, confirming the feasibility of

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combining SALLE preparation procedure with CE separation in an automated manner. Nevertheless, further optimizations were required to improve the efficiency of the A-SALLE-CE-UV methodology.

3.2. Optimization of the mixing efficiency of standard sample solution with ACN and NaCl

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The first step in the optimization of the automated extraction procedure was to adjust both filling pressure and time to obtain accurate volumes of both ACN/NaCl solution to the standard sample contained in a 600 µL tube (mixing chamber). Five different pressures ranging from 5 to 80 psi were evaluated with a fixed filling time of 6 min. The mean added volume (n=3) of ACN or NaCl (V<sub>x</sub>) to the sample was calculated based on mass differences ( $V_x$  (ml) = ( $M_{filled}$  -  $M_{empty}$ )]/ $\rho$ ), where  $M_{filled}$  is the mass (g) of the 600 µL eppendorf tube filled with ACN solvent or NaCl solution, mempty is the mass (g) of the empty 600 μL eppendorf tube and ρ is the density of the ACN solvent (0.79) or aqueous NaCl solution (1). The mean volumes of both ACN and NaCl increased linearly (correlation coefficient (r<sup>2</sup>) higher than 0.999) from 4 to 317 μL with good repeatability (RSD < 1.53 %, n= 3) for the different tested pressures (Fig. 4A and B). Hydrodynamic pressure of 60 psi was chosen as the optimal pressure as it gave the best compromise between low RSD (less than 0.63 %, n=3) and rapidity of filling the vials. The same procedure was repeated varying the filling time from 1 to 8 minutes with fixed 60 psi pressure. Mean volumes of both ACN and NaCl increased linearly from 8 to 293 nL (r<sup>2</sup> > 0.999) for the different filling times (Fig. 4C and D) with high repeatability (RSD < 0.63 %, n= 3) whatever the injection time applied. The second step to ameliorate the mixing efficiency was to decrease the sample, ACN and NaCl volumes introduced in the 600 μL eppendorf tube. Indeed, more than 83 % (500 μL) of the 600 μL eppendorf tube was filled with solutions. Consequently, a loss of solutions and a lack of mixing efficiency could occur during the mixing step with air bubbles affecting the repeatability and the performance of the A-SALLE-CE-UV methodology. Moreover, decreasing the solution volumes helps to reduce the time needed to fill the inlet vial. Therefore, the volumes of sample, ACN and NaCl were divided by a factor of 2 and the total volume was fixed at 250 µL. NaCl and ACN were added to the standard samples at 60 psi during 2.4 (20 µL) and 4.0 minutes (150 µL), respectively. The pressure applied for the mixing by air bubbles was increased to 90 psi during 5 min without risk of loss of sample. Improvements in the injected sample volume (up to 40%), SEF (up to 125) and extraction efficiencies (up to 45%) for the 3 TKIs were observed confirming the benefit of decreasing ACN, NaCl and sample volumes besides increasing the air pressure used for mixing. The third step in the optimization of the extraction procedure, was to modify the filling scheme of the sample vial by ACN and NaCl. Instead of adding the entire volume of these solutions to the standard sample at once as previously described, NaCl and ACN solutions were added to the sample in 2 steps each to obtain total added volumes of 20 and 150 μL, respectively. So, 75 μl of ACN and 10 μl of NaCl were successively added to the standard sample solution at 60 psi during 2.0 minutes and at 60 psi during 1.2 minutes, respectively. Mixing by air bubbles at 90 psi during 2.0 minutes was performed after the first addition of ACN and salt, and repeated after the next addition of ACN. After the second NaCl solution addition, a final mixing was realized at 90 psi during 5.0 minutes after addition all solutions  $(V_{total} = 250 \mu L)$ . The filling scheme of sample vial can be resumed as following: ACN/NaCl/air/ACN/air/NaCl/air. The injected volume of the upper phase could be successfully increased to 287 nl (corresponding to 70% of the capillary effective length) without peak broadening due to in-line acetonitrile CE stacking increasing the sensitivity of the A-SALLE-CE-UV methodology (Fig. 5). Moreover, TKIs were efficiently extracted (60–62% of extraction efficiency) in the upper phase. The injection was repeated three times. RSD ranging from 0.85 to 1.12 % for migration times and from 1.40 to 2.40 % for peak areas were obtained. SEF factors of 94.1, 134.1 and 196.3 were obtained for lapatinib, erlotinib and sorafenib, respectively. All results indicate an improvement of the analytical performances of the proposed A-SALLE-CE-UV methodology due to better mixing efficiency. However, the values obtained in term of extraction efficiency (%) and SEF were about 1.6 times less important than with SALLE-CE-UV (98

- 100 % for extraction efficiencies, 144-265 for SEF factors). The mixing generated by a stirrer and a

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centrifugator in the case of manual SALLE-CE-UV is more efficient than with air bubbles for A-SALLE-CE-UV. Other developments (time and pressure of mixing, volume and shape of the mixing chamber) were realized without technical modifications at the level of the commercially available CE apparatus but didn't give significant improvements. Even though, the performances of the proposed A-SALLE-CE-UV methodology are sufficient to quantify TKIs in human plasma (see 3.4). Consequently, these conditions were defined as being the optimal A-SALLE-CE-UV procedure (see section 2.5; Table 1S in supplementary information) and were applied in the rest of this study.

## 3.3. Effect of the type of the salting-out agent

Different salts were studied with the aim to choose the optimal salting-out agent. NaCl, potassium chloride (KCl) and ammonium acetate were tested because these salting-out agents gave the best extraction efficiencies in many works [26,27]. It was found that corrected peak area values were 1.3 times more important with KCl than for the two other salts (NaCl and ammonium acetate) suggesting better extraction efficiencies (Fig. 6A). However, KCl lead to the reduction of peak efficiencies compared to the other two salting-out agents (Fig. 6B). As KCl is more soluble in a 100% ACN phase ( $\approx$  24 mg/L) than for the two other salts ( $\approx$  3 mg/L), the amount of salt in the upper phase obtained with KCl was probably higher than that of NaCl and ammonium acetate. The presence of salt in the upper phase allows to extract more easily the 3 TKIs due to better solubility. However, the presence of salts in the injected sample, affects the performance of in-line acetonitrile stacking [28]. NaCl and ammonium acetate nearly lead to the same performance in terms of corrected peak areas of TKIs but, NaCl gave the most efficient peaks. Better repeatability (n=3) for corrected peaks areas was obtained with NaCl with values of less than 2.5 %. Therefore, NaCl was chosen as the salting out agent for the A-SALLE-CE-UV as it gives the best compromise between extraction efficiency, repeatability and peak efficiency.

#### 3.4. Analysis of human plasma spiked with TKIs:

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The performance of the proposed A-SALLE-CE-UV methodology was evaluated on plasma sample spiked with the three TKIs. The pH of the BGE was 2.0 and the ionic strength 150 mM. Protein precipitation was observed in the sample vial after the addition of ACN. Figure 7 shows the electropherograms obtained for blank plasma (no TKIs) and TKIs plasma sample (injected volume equal to 70% of the capillary effective length). No interfering peaks of endogenous proteins were observed at the migration time of the TKIs demonstrating the specificity of the proposed methodology. The maximum injected sample volume without peak broadening was similar to that obtained with a standard sample solution (287 nl corresponding to 70% of the capillary effective length). This result indicates that the in-line stacking performances were similar for standard and plasma solutions and confirmed the robustness of the proposed methodology. The analytical performances of the developed methodology were evaluated and are summarized in table 1. LOQ of lapatinib, erlotinib and sorafenib were determined at 1260, 200 and 640 ng/ml respectively. These LOQ values were slightly higher than that obtained by manual sample preparation since the type of mixing process was different. However, these values are still lower than mean plasma levels observed for patients treated by TKIs, which are estimated at 1570, 1200 and 1440 ng/ml for lapatinib, erlotinib and sorafenib respectively [29–31]. Good repeatability was obtained at the LOQ levels (n = 6) in term of migration time (RSD < 2.88 %) and corrected peak areas (RSD < 2.73%). Good linearity of the calibration curve from 200 to 5000 ng/ml was obtained with r<sup>2</sup> higher than 0.995. Moreover, several advantages are offered by the proposed A-SALLE methodology. The small volume of the sample used (80 µL compared to 2.0 mL of plasma for manual SALLE methodology) which will be more convenient for the patient's quality of life i.e. less blood sample collected. The reduction in the volume of organic solvent used (about 20x less) compared to manual methodology which will effectively lower the cost and toxicity of analysis per sample. In addition, automation makes analyses easier to realize minimizing manual labor and human error. All the above results confirm the possibility to automate SALLE by CE without the need to make any modification at the level of the commercially available CE machine. Moreover, the applicability of the proposed A-SALLE-CE-UV for the routine analysis of these three TKIs was successfully tested against plasma sample spiked with the three TKIs.

#### 4. Conclusion

A novel automated salting-out assisted liquid-liquid microextraction approach for plasma sample before CE analysis was developed. The capillary was used as a tool to deliver appropriate microvolumes of ACN and NaCl to the inlet vial containing the plasma sample. This was followed by mixing using air bubbles leading to a two phases separation and extraction of the TKIs in the upper phase. NaCl was chosen as the salting out agent as it gives the best compromise between high extraction and peaks efficiency. The A-SALLE-CE-UV methodology has the advantages to eliminate simultaneously biological proteins, salts and to extract the molecules of interest from the sample matrix. This new approach, A-SALLE-CE-UV, was successfully applied for the analysis of 3 TKIs in plasma samples. The extracted solution (upper phase, mostly ACN solvent) could be injected up to 70% of the capillary effective length due to the existence of in-line acetonitrile stacking phenomenon inside the capillary. Thus, it can be considered as a rapid, simple, cheap, convenient and well suited for possible TDM of TKIs and for further applications concerning the use of SALLE. This A-SALLE-CE-UV methodology does not require to make any technical modifications at the level of the commercially available CE apparatus.

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#### Figures captions

- 413 Figure 1: A-SALLE-CE-UV procedure. Step (1) Delivering of appropriate micro-volumes of solutions (ACN
- and NaCl stock solution) from the outlet side to the inlet side (sample vial) by hydrodynamic injection,
- 415 followed by mixing with air bubbles injected from an empty outlet vial. Step (2) Separation into 2 phases
- 416 (upper phase: mostly organic phase. Bulk phase: mostly aqueous) after mixing step. Step (3)
- 417 hydrodynamic injection of the upper phase. Step (4) Separation and detection of the three TKIs by CE-
- 418 UV.

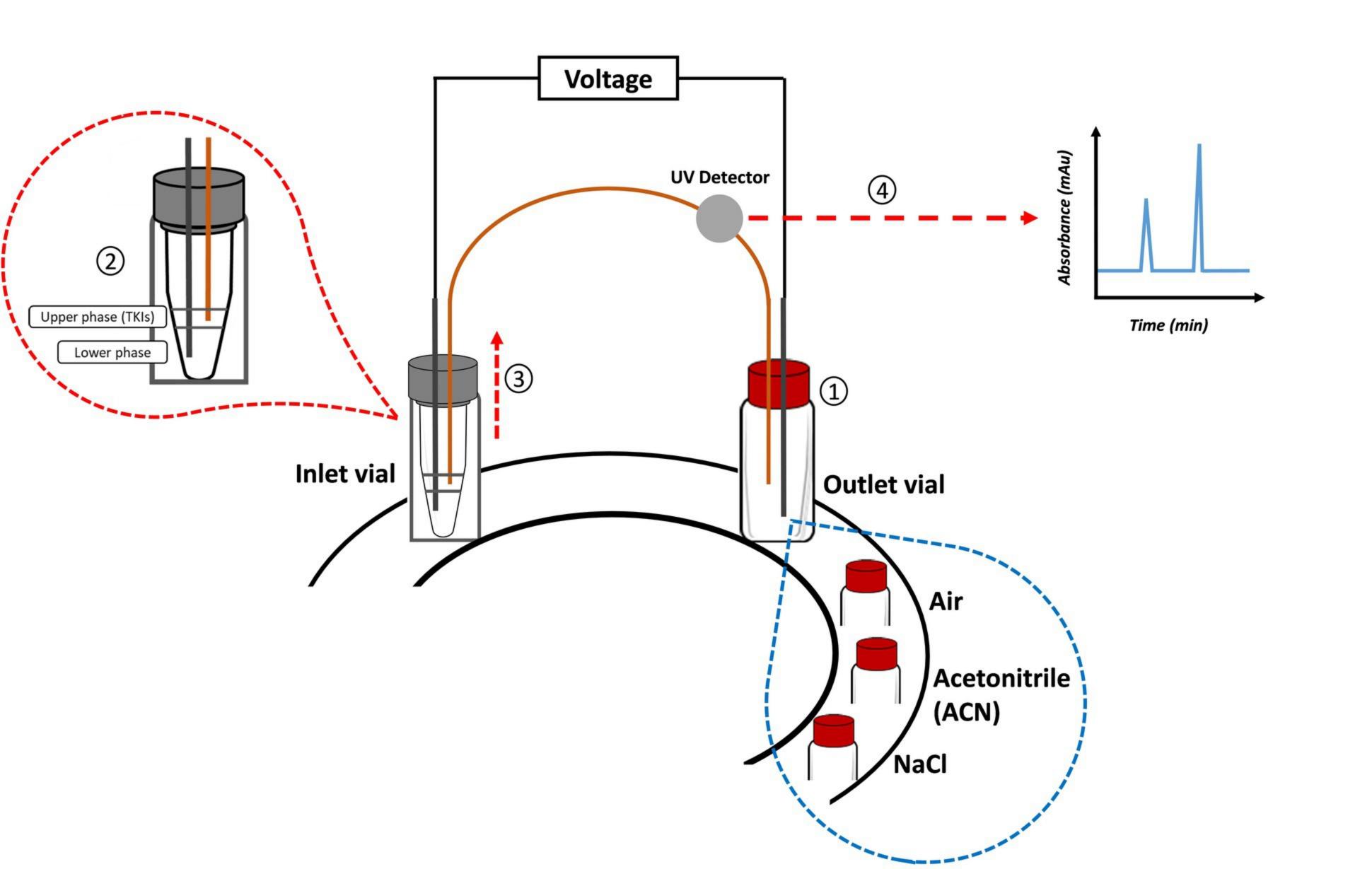
- 419 Figure 2: Molecular structures of the studied TKI: (1) lapatinib ditosylate, (2) erlotinib hydrochloride
- 420 and (3) sorafenib.
- 421 Figure 3: Analysis of lapatinib, erlotinib and sorafenib after injecting the upper phase (mostly ACN)
- 422 obtained by A-SALLE-CE-UV methodology. Analysis conditions: silica capillary 50 μm i.d., total length 31
- 423 cm, effective length 21 cm. BGE: citric acid ε-aminocaproic acid buffer IS 150 mM pH 2.0.
- 424 Temperature: 25°C. Separation voltage: 15 kV. Sample injection volume: 39 nl (corresponding to 10% of
- the capillary effective length). Detection: 254 nm.
- 426 Figure 4: Effect of increasing the hydrodynamic pressure during 6 minutes on the volume obtained of
- 427 (A) ACN and (B) NaCl. Effect of increasing the time at 60 psi on the volume obtained of (C) ACN and (D)
- 428 NaCl.
- 429 Figure 5: Analysis of lapatinib, erlotinib and sorafenib after injecting the upper phase obtained by A-
- 430 SALLE-CE-UV methodology (3 repetitions). Analysis conditions: silica capillary 50 μm i.d., total length 31
- 431 cm, effective length 21 cm. BGE: citric acid  $-\epsilon$ -aminocaproic acid buffer IS 150 mM pH 2.0.
- Temperature: 25°C. Separation voltage: 15 kV. Sample injection volume: 287 nl (corresponding to 70%
- 433 of the capillary effective length). Detection: 254 nm.
- Figure 6: Effect of the type of salting-out agent on (A) corrected peak areas and (B) theoretical plates
- 435 (N) for the 3 studied TKIs.

- **Figure 7**: Electropherograms obtained for blank human plasma sample and human plasma samples spiked with the three TKIs using the A-SALLE-CE-UV procedure. Analysis conditions: silica capillary 50  $\mu$ m i.d., total length 31 cm, effective length 21 cm. BGE: citric acid –  $\epsilon$ -aminocaproic acid buffer IS 150 mM pH 2.0. Temperature: 25°C. Separation voltage: 15 kV. Sample injection volume: 287 nl (corresponding to 70 % of the capillary effective length). Detection: 254 nm.

# 442 Table caption

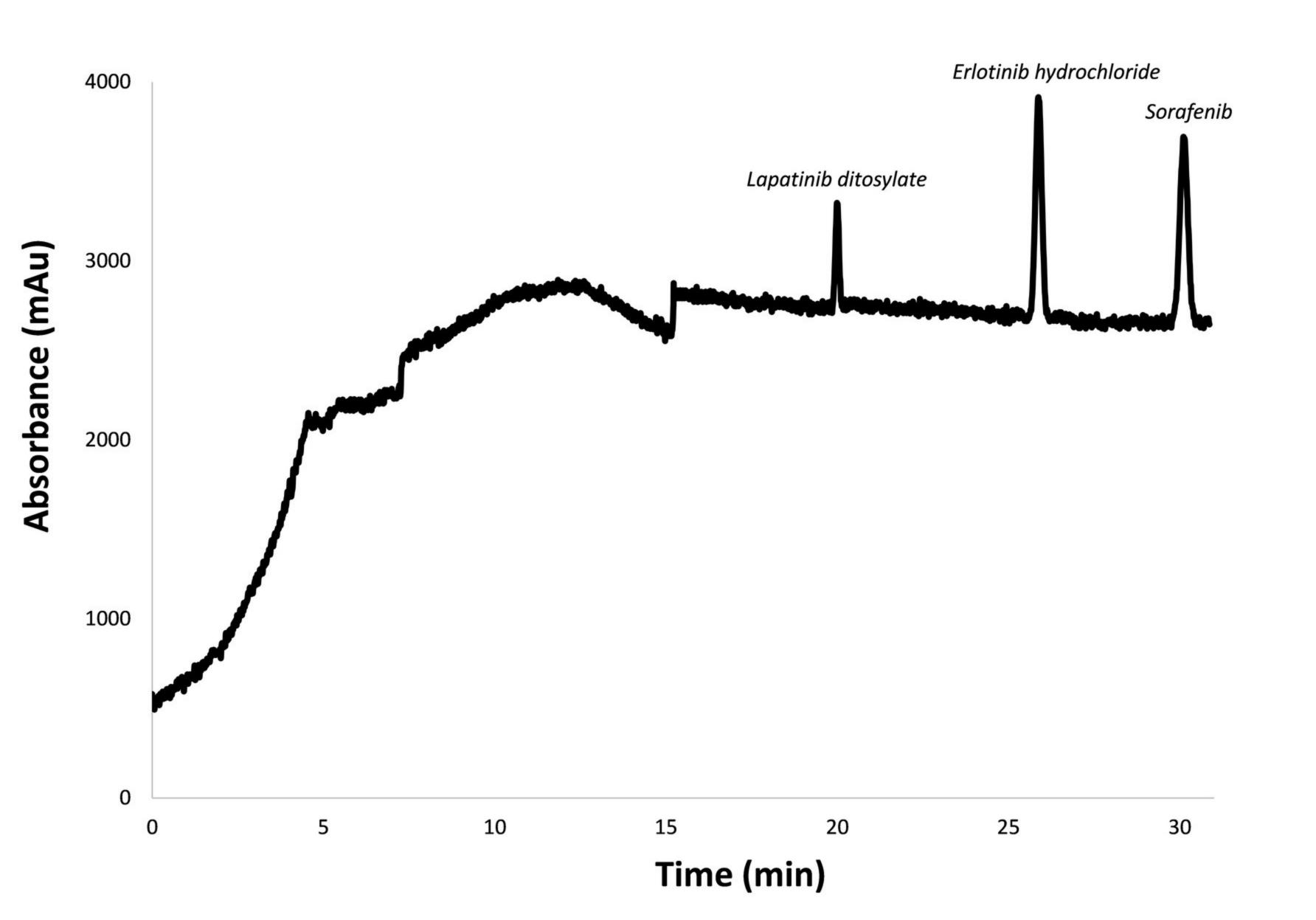
- Table 1: Summary of the A-SALLE-CE-UV quantitative results obtained on human plasma samples spiked with lapatinib, erlotinib and sorafenib. Injected sample volume equals 70 % of the capillary

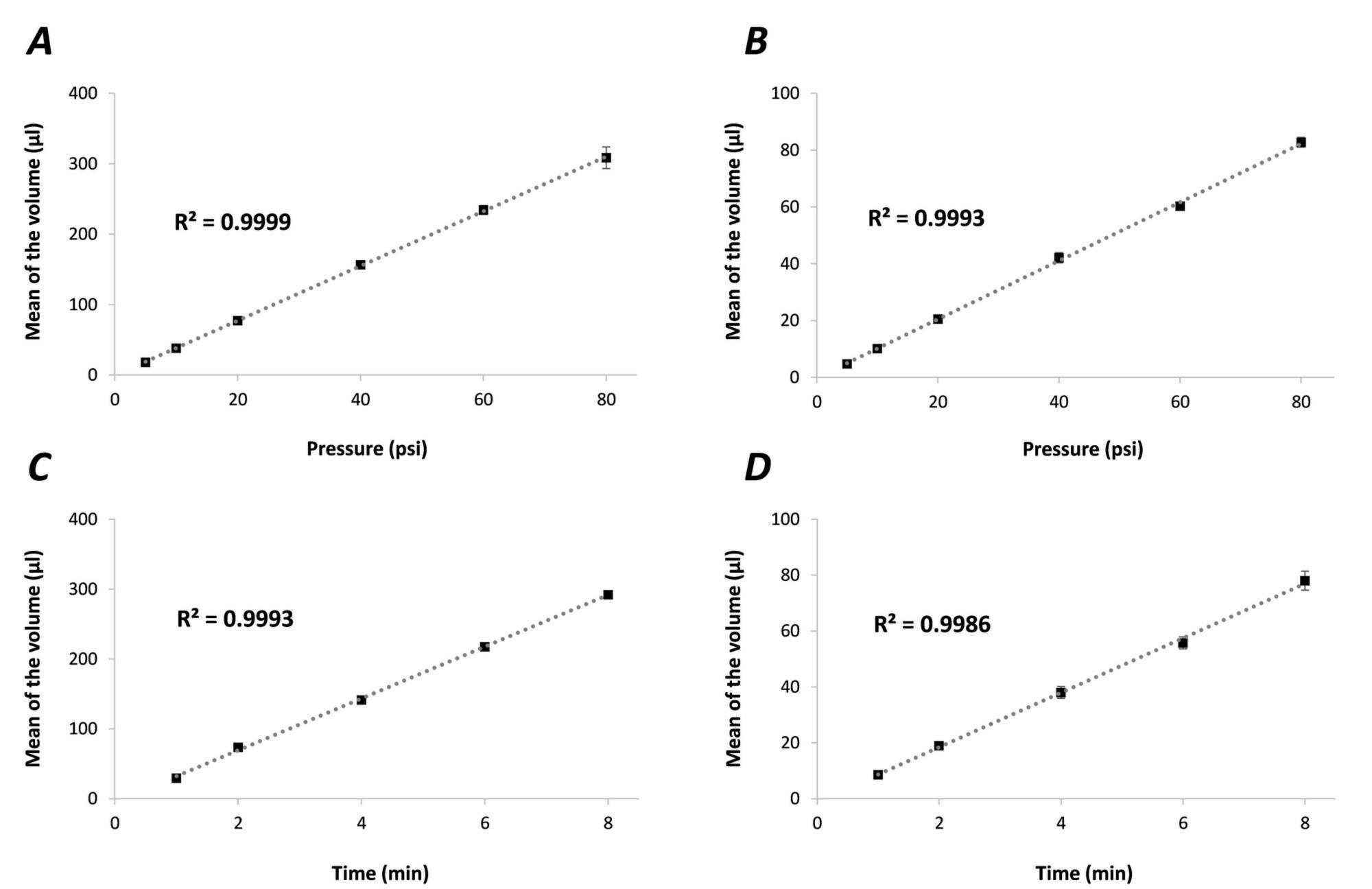
445 effective length.

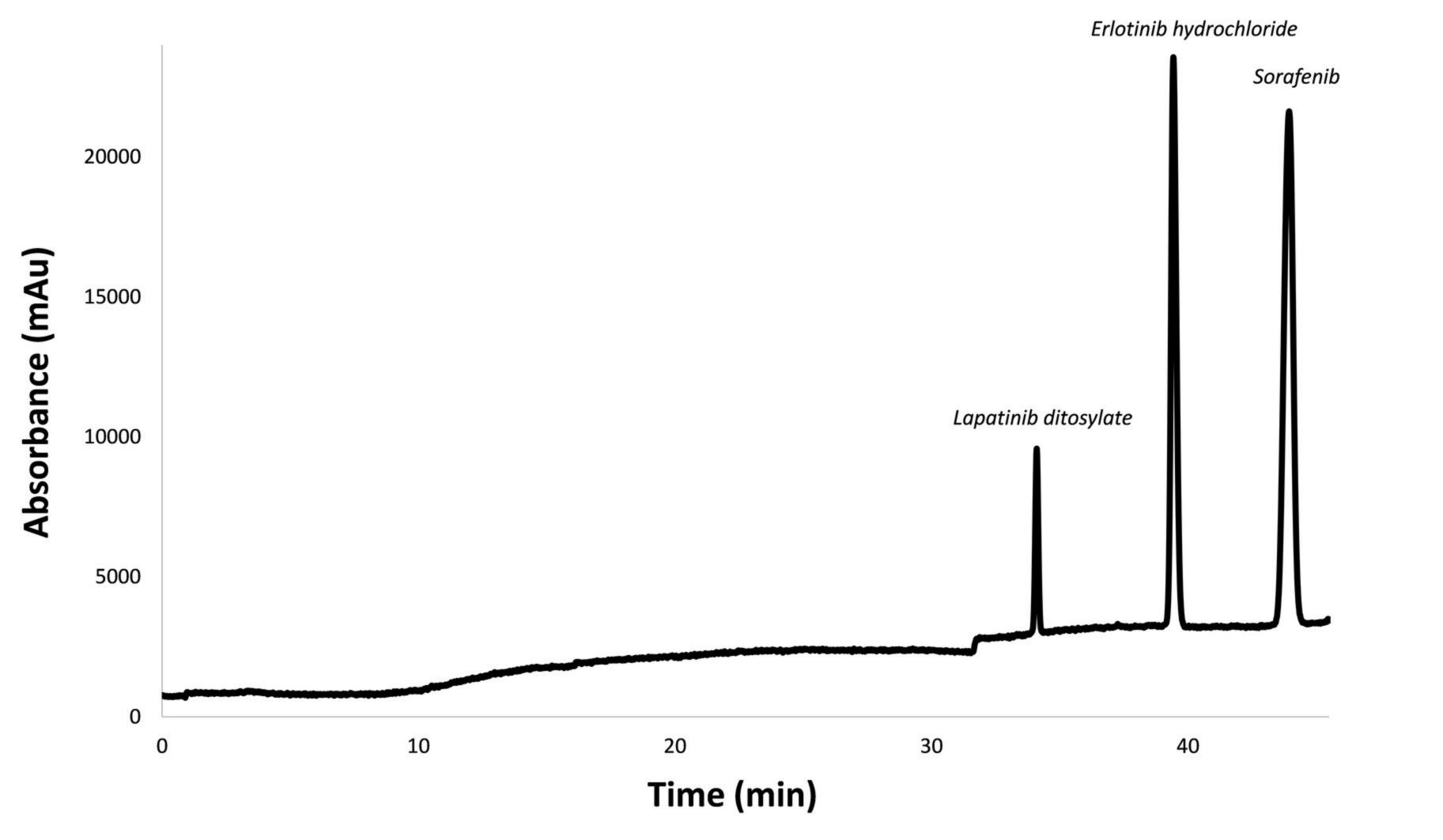


(1) (2) 
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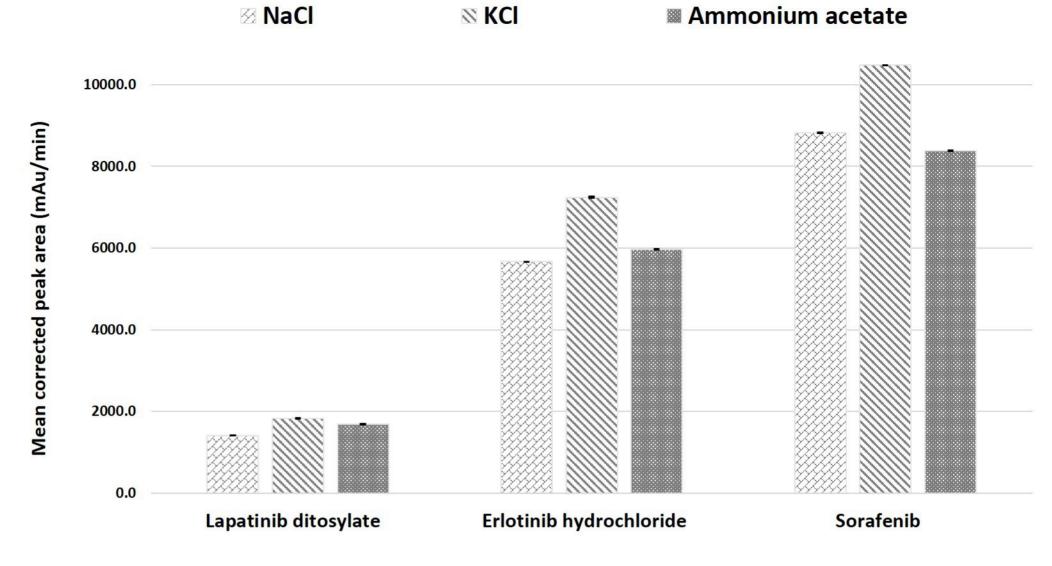
(3)



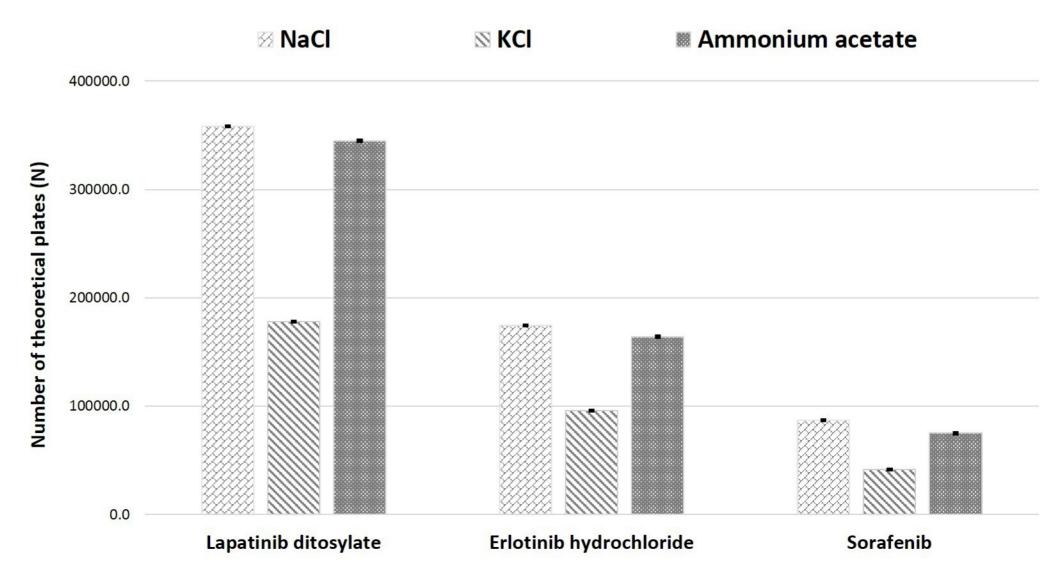












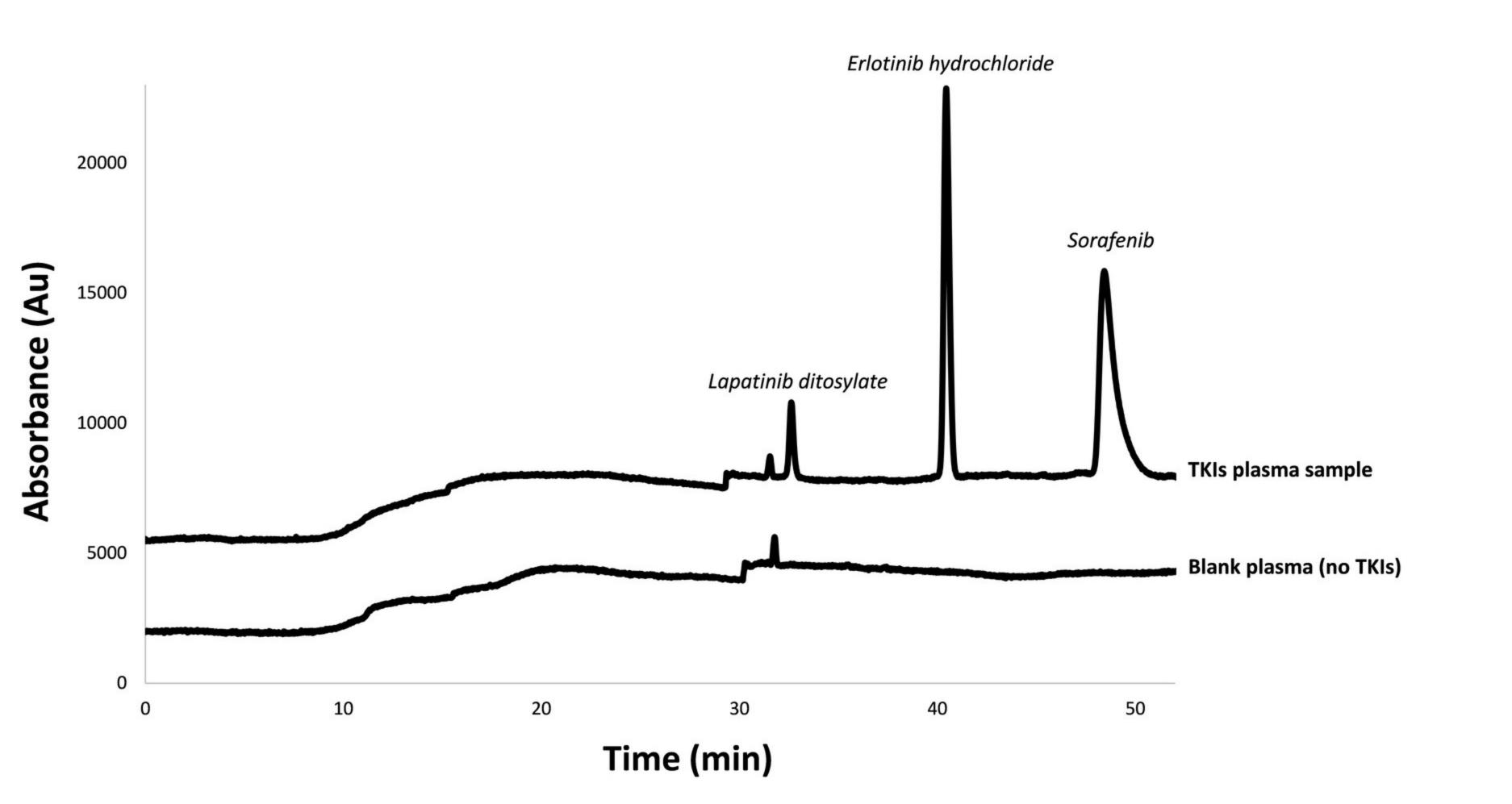


Table 1

A-SALLE-CE-UV	Lapatinib ditosylate	Erlotinib hydrochloride	Sorafenib	
performance parameters	, ,		ŕ	
Extraction efficiency (%)	62	62	60	
Specificity	Confirmed	Confirmed	Confirmed	
Linearity (ng/ml)	1360-5000	200-5000	640-5000	
Correlation coefficient (r²)	0.996	0.998	0.995	
LOD (ng/ml)	1100	80	200	
LOQ (ng/ml)	1360	200	640	
% RSDª (corrected peak area)	2.19	2.73	2.67	
% RSD <sup>a</sup> (migration time)	2.88	1.40	2.03	

<sup>&</sup>lt;sup>a</sup> Calculated for 6 successive runs at the LOQ levels

