

Nanoparticles-loaded plastic scintillators for fast/thermal neutrons/gamma discrimination: Simulation and results

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1	Nanoparticles-loaded plastic scintillators for fast/thermal
2	neutrons/gamma discrimination: simulation and results
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- 8

9 Abstract

We report herein the application of a new series of plastic scintillators loaded with 10 lithium tetraborate nanoparticles (Li₂B₄O₇) designed for the simultaneous detection of fast 11 neutrons, thermal neutrons and gamma. First, MCNP simulations are performed to highlight 12 13 the potential benefit of Li₂B₄O₇ loading. Then, a nanoparticle-loaded scintillator is prepared and evaluated towards the detection of a partially thermalized neutron/gamma 252-californium 14 source. Several scenarios of radiation/matter interactions are evaluated against the observed 15 results. A detection chamber composed of boron-containing chipboard wood has been 16 specifically designed and a subtraction method was set up to afford the best discrimination 17 18 pattern. For the first time, a triple discrimination is presented with a nanoparticles-loaded plastic scintillator, where both ${}^{6}Li(n,\alpha)$ and ${}^{10}B(n,\alpha)$ signatures are observed. A Figure of Merit 19 of 1.40 was calculated between fast neutrons and gamma rays around 480 keVee. The Figure of 20 Merit between fast + thermal neutrons and gamma rays is slightly worse as it is 1.16 at 380 21 22 keVee.

23

24 Keywords

- 25 Plastic scintillators; nanoparticles; thermal neutrons; fast neutrons; discrimination
- 26

27 1. Introduction

Detection of special nuclear materials (SNMs – mainly ²³³uranium, ²³⁵uranium and 28 ²³⁹plutonium) constitutes a major effort in the field of Homeland Security. For a long time, ³He 29 based detectors were considered as the gold standard in terms of neutron detection that can 30 reveal the trafficking of such SNMs. However, the 2010's have seen a revival in terms of 31 chemical modifications of plastic scintillators generally speaking,¹ and specifically tuned for this 32 application.² The spontaneous fission of materials such as SNMs sees the release of fast 33 neutrons whose kinetic energy is distributed with a Watt spectrum with an average energy 34 around 2 MeV. These neutrons may slow down through multiple scatterings with their 35 36 surroundings, leading to neutrons with much lower energy, falling down to the thermal energy equal to 25 meV at 300 K. Thus, one can admit that a sufficiently shielded SNM with moderating
materials such as polyethylene may only been "seen" as a low-energy neutron emitter. So, the
larger the neutron energy range that is covered, the more versatile and efficient the sensor will
be.

41 Detecting thermal neutrons with plastic scintillators is usually performed by loading the material with neutron absorbing elements. Table 1 gives the comparison between the main 42 elements that are able to create a nuclear reaction of interest with thermal neutrons. ¹⁰B 43 44 loaded plastic scintillators are probably the most studied and they are even commercially available from several suppliers for decades. To the best of our knowledge, the first boron-45 loaded organic scintillator was liquid, and trimethyl borate (B(OCH₃)₃) was the boron source 46 used in the formulation.³ Now, most of the research cited in the literature advocates the use of 47 lithium, boron² or even gadolinium.⁴ Li and B may have both pros and cons in their use, but 48 they do operate in almost the same way, since the created ions (alpha, triton and Li³⁺ particles) 49 represent the thermal neutron signature and display different pulse shapes from gamma 50 interactions. Even at the natural isotopic loading, Gd displays the highest cross section for 51 capturing thermal neutrons (254,000 barns), but the interaction releases gamma-rays that 52 makes it inadequate for pulse shape discrimination. In that case, a compensation or 53 coincidence scheme can be successfully applied.⁵ 54

55

56 Table 1. Typical thermal neutron absorptions.

lsotop e	Thermal neutron capture reaction	25 meV cross section (Barns)	Natural isotopic abundance (%)
³ He	${}_{2}^{3}He + n \rightarrow {}_{1}^{3}H + {}_{1}^{1}p$	5330	0.000137
۴Li	$^{6}Li + n \rightarrow {}^{3}H (2.73 \text{ MeV}) + \alpha (2.05 \text{ MeV})$	940	7.5
¹⁰ B	$^{10}B + n \rightarrow ^{7}Li^{*} + \alpha (1.47 MeV) \rightarrow ^{7}Li + \alpha (1.8 MeV) + \gamma (0.48 MeV)$	3840	19.9
¹¹³ Cd	$^{113}Cd + n \rightarrow ^{114}Cd + \gamma's \ (9 \ MeV)$	20600	12.2
155 Gd		60900	14.7
¹⁵⁷ Gd	$^{155.157}Gd + n \rightarrow ^{156.158}Gd * \rightarrow ^{156.158}Gd + e^{-} + \gamma's (8 MeV)$	254000	15.7

57

Loading plastics with lithium is particularly challenging. Indeed, it is worth mentioning the issues encountered when loading apolar, scintillating polymers (typically polystyrene or poly(vinyltoluene)) with polar elements such as the lithium salt. Among the different available solutions, lithium nanoparticles (NPs) are particularly advantageous. Loading nanoparticles in plastic scintillators is an emerging field, both for gamma spectrometry⁶ and for thermal neutron counting. For the latter purpose, this was already shown by Carturan *et al.* who dispersed oleiccapped lithium fluoride nanocrystals in a polysiloxane matrix.^{7,8,9}

Based on this strategy, we present our results regarding chemical modifications of plastic scintillators, firstly with a high concentration of primary fluorophore for the detection of fast neutrons. It is now well known that materials may present pulse shape discrimination (PSD) properties thanks to the difference between delayed and prompt fluorescence,² and two of them are currently commercially available from Eljen Technology and Amcrys. Secondly, by 70 adding not only lithium NPs, but also a mixture of lithium and boron, namely in the form of 71 lithium tetraborate Li₂B₄O₇, we take advantage of these two neutron absorbers to largely increase thermal neutron capture efficiency. In the field of radiation detection, lithium 72 tetraborate is already known as a tissue equivalent thermoluminescent, gamma-ray 73 dosimeter.¹⁰ However, to the best of our knowledge, scintillators loaded with a mixture of two 74 or more neutron absorbers are poorly studied.^{11,12} From this combination, we expect a counting 75 sensitivity improvement of both fast and thermal neutrons, especially in the case where small 76 sensors are used such as in hand-held devices. 77

Thus, the first section presents the MCNP6.1 simulation, which supports our motivation 78 79 and theory. This simulation allows us to determine the trend of the response of a homogenously loaded plastic scintillator with Li₂B₄O₇ NPs. It shows that for a 10-cm³ size 80 sample, starting with 0.1 weight percent (wt%) loading may already give a relevant capture rate 81 which could be detected. Then the fast and thermal neutron/gamma discrimination is 82 evaluated by exposing the NP-loaded plastic scintillator to a partially thermalized 252-83 californium radioactive source, used as a SNM laboratory surrogate. To confirm all the nuclear 84 processes occurring in the material, several hypotheses were formulated. A boron-containing 85 chipboard wood cage was built and a PSD subtraction method was designed which highlights 86 the signature of up four different interactions: gamma and fast neutrons are easily separated 87 88 by PSD, and two extra features - corresponding to thermal neutron capture by lithium and 89 boron – appear as well.

90

91 2. Experimental section

92 The synthesis of our material is reported elsewhere.¹³ 0.3 wt% of fully enriched ⁶Li₂¹⁰B₄O₇ NPs were added to an argon-saturated solution of monomers containing PPO (20 93 wt%) and POPOP (0.03 wt%). No initiator has been used. 5 cycles of degassing under vacuum 94 were performed then the mixture was poured in a glass jar and sealed under argon atmosphere 95 96 and cured between 60 – 110 °C for 10 to 15 days. After complete polymerization, the mold was shattered and the free piece was cut and polished until obtaining an optical-grade surface. 97 Ultimately, the scintillator was covered with Teflon® tape to optimize optical focalization 98 towards the photomultiplier tube, thereby allowing gamma spectra and n/γ discrimination 99 experiments. The diameter of the scintillator is 32 mm and the thickness 11 mm, weight 9.64 g, 100 density 1.09. 101

102 EJ-200 plastic scintillator was obtained from Eljen Technology.

Gamma-ray spectra were recorded as follows: the sample was optically coupled with 103 RTV141A optical grease to a Hamamatsu R7724-100 photomultiplier. A 387-kBg ⁶⁰Co source 104 was used to excite the material. Two experimental series were performed: one with the 105 106 prepared plastic scintillator and the other with a commercial sample of same size. Pulses 107 coming from the PMT were sorted and treated with a custom-made electronic board. The high voltage was kept constant during both experiments, using a stabilized high voltage module 108 N1470 from CAEN. The Compton edge (CE) was fitted with a Gaussian function and the accurate 109 determination of the Compton edge position was evaluated at 80 % of the decrease of this 110

Gaussian shape. By rule of thumb, the light yield of the prepared plastic scintillator sample could be determined from the commercial data sheet (herein 10,000 ph/MeV for EJ-200 plastic scintillator):

$$R_{sample} = R_{EJ-200} \times V_{CE,sample} / V_{CE,EJ-200}$$

114 115

where *R* is the light yield, and *V* is the channel of the Compton edge position.

In order to assess its neutron/gamma discrimination ability, the scintillator was coupled 116 117 using RTV141A optical grease to a Hamamatsu H11284MOD photomultiplier tube fed with a CAEN N1470 high voltage operating in negative mode. The {scintillator + PMT} system was 118 placed 15 cm away from a 252 Cf source (\approx 1.8 MBq activity), the latter being partially 119 thermalized with a 10 cm thick polyethylene brick. To reduce the gamma-ray incident flux on 120 the sample, a 5 cm denal[®] brick (a tungsten-based alloy) was added between the ²⁵²Cf source 121 122 and the polyethylene brick. The anode signal fed a CAEN DT5730B digitizer, running with the DPP-PSD software. Scintillation pulses were then recorded and post-processed using a charge-123 comparison method.¹⁴ The Figure of Merit was finally calculated; it is a standard factor for the 124 evaluation of the n/γ discrimination power and has already been fully referenced.² Along with 125 this experiment and with the strictly same setup, an energy calibration was performed with a 126 127 530 kBq ¹³⁷Cs source. The quality of the PSD was quantitatively estimated from the calculation of the Figure of Merit (FoM), which has the following formula: 128

129
$$FoM = \frac{|\mu_n - \mu_\gamma|}{2.35(\sigma_n^2 + \sigma_\gamma^2)}$$

where μ_n and μ_{γ} are the mean values of the neutron and the gamma ray peaks, and σ_n and σ_{γ} are the standard deviations of neutron and gamma distributions fitted with Gaussian functions $\mathcal{N}(\mu_n, \sigma_n)$ and $\mathcal{N}(\mu_{\gamma}, \sigma_{\gamma})$.

- 133
- 134 3. Results

135 3.1. Simulation of the spectral signature and capture rate of thermal neutron 136 radiations in Li₂B₄O₇ loaded plastic scintillator

137 The addition of both boron and lithium is firstly motivated by the fact that these 138 elements ultimately release energetic and charged particles after the capture of a slow 139 neutron. The nuclear reactions underlying the emission of these products are:

140 - the ${}^{10}B(n, \alpha)$ absorption reaction: thermal neutrons are absorbed by boron-10 nuclides, 141 which is of natural abundance close to 20%. Following the absorption, the nucleus 142 disintegrates following Eq. (A) with a 6% probability, and following Eq. (B) with a 94% 143 probability, owing that the produced lithium-7 nucleus is found in its fundamental or 144 excited state.

145 ${}^{10}_{5}B + {}^{1}_{0}n \rightarrow {}^{7}_{3}Li + {}^{4}_{2}He$ (A)

146
$${}^{10}_{5}B + {}^{1}_{0}n \rightarrow {}^{7}_{3}Li^* + {}^{4}_{2}He$$
 (B)

147 The associated cross section at $E_{th} = 25 \text{ meV}$ is $\sigma_{E_{th}} = 3840 \text{ b}$ and the Q values are 148 2.31 MeV and 2.79 MeV for Eq. (A) and (B) respectively. The said Q value is shared 149 between a lithium-7 nucleus and an alpha particle, with respective kinetic energies 150 $E_{\frac{7}{3}\text{Li}} = 0.84 \text{ MeV}$, and $E_{\alpha} = 1.47 \text{ MeV}$. From Eq. (B), a 0.48 MeV gamma ray is also 151 released following the de-excitation of lithium-7.

152 - the ⁶Li(n, α) absorption reaction: thermal neutrons are absorbed by lithium-6 nuclides 153 (7.5 % natural abundance), according to Eq. (C).

154
$${}_{3}^{6}\text{Li} + {}_{0}^{1}n \rightarrow {}_{1}^{3}\text{H} + {}_{2}^{4}\text{He}$$
 (C)

155 The associated cross section is $\sigma_{E_{th}} = 940$ b with a Q value of 4.78 MeV, that is shared 156 between a tritium and an alpha particles with respective kinetic energies $E_{^{3}H} =$ 157 2.73 MeV, and $E_{\alpha} = 2.05$ MeV.

Aside the two (n,α) absorption reactions from the embedded nanoparticles, we should mention (n,γ) capture reactions occurring inside the polymer matrix, with microscopic cross sections ranging from the order of 0.1 mb up the order of 100 mb. However, due to the small volume of the scintillator (about 9 cm³), the macroscopic cross section for (n,γ) reactions, as well as the photoelectric and Compton interaction yield of the high-energy gamma rays released after H, C, N, O capture, are expected to be too low for practical use. We shall discuss this assumption in the rest of the analysis.

A modeling study, using the MCNP6.1 particle transport software package¹⁵ was conducted to simulate the response of the plastic scintillator loaded with lithium tetraborate nanoparticles. The aim of this Monte-Carlo study is to provide a description of the response trend of the sensor when confronted to 25 meV, thermalized neutrons. The neutron source is considered homogenous and isotropic, surrounding the sensor. The key parameter appears to be the nanoparticle loading weight.

171 Let us consider a cross-linked plastic scintillator with a suitable composition for fast neutron/gamma discrimination.¹⁶ The density $\rho_m = 1.07 \text{ g. cm}^{-3}$, and the elemental 172 composition (C, H, N, O) are injected into the code. The embedded crystals are of composition 173 Li₂B₄O₇. The scintillator medium was modeled according to the dimensions of the synthetized 174 samples that are experimentally studied in the following section, *i.e.* 1.6-cm-radius by 1.1-cm-175 height in a perfect cylindrical shape. As we used a homogenous spherical neutron source in the 176 model, it thus follows that the plastic thickness in neutron direction is in the order of 1 cm. In a 177 first-order-approximation, we consider their insertion of the elements forming these 178 179 nanocrystals to be homogenous inside the polymer matrix.

¹⁸⁰ The output estimates of interest in view of simulation-based, design study are:

^{181 -} the neutron absorption yield within the scintillator, thereafter noted as τ_a and 182 homogenous to a number of captures per source neutron (n⁻¹);

^{183 -} the energy deposition yield per source neutron inside the scintillator, induced by 184 secondary particles that are released after (n,α) and (n,γ) capture reactions. This yield is

a function of energy *E* and is labeled $\tau_D(E)$. We express this estimate in energy deposition per source neutron, and per energy bin (n⁻¹.keV⁻¹).

The available kinetic energy from the capture reactions $\tau_D(E)$ allows us to separate the respective contributions of boron and lithium with respect to the thermal neutron-induced signal. It is worth mentioning that photoluminescence quenching was not considered in the simulation.

A quantitative analysis of the indirect neutron signature is carried out using the 191 192 following concentrations: 0, 0.3, 0.5, 1, 3 and 5 wt% of nanoparticles. In all cases, the lithium tetraborate content is simulated in its 100 % isotopically enriched lithium and boron version, 193 therefore ⁶Li₂¹⁰B₄O₇. For the sake of explanation, we break down the overall neutron absorption 194 yield τ_a in two terms: $\tau_a(n, \alpha)$ from ${}^{10}B(n, \alpha)$ and ${}^{6}Li(n, \alpha)$ reactions, and $\tau_a(n, \gamma)$ for the 195 abovementioned capture reactions taking place inside the polymer matrix. The evolutions of 196 estimated yields $\tau_a(n, \alpha)$ and $\tau_a(n, \gamma)$ according to the concentration of lithium tetraborate 197 inside the scintillator are reproduced in Figure 1 below. 198





200 Nanoparticle concentration (wt%) 201 Figure 1. Neutron absorption yields $\tau_a(n, \alpha)$ and $\tau_a(n, \gamma)$ as a function of lithium tetraborate concentration within the plastic 202 scintillator.

203

Figure 1 shows that, in terms of the total neutron absorption yield $\tau_a = \tau_a(n, \alpha) + \tau_a(n, \alpha)$ 204 $\tau_a(n,\gamma)$, a gain by a factor of two is observed as soon as 0.1 wt% of nanoparticles are added, 205 206 with a strong growth up to a homogenous concentration of 5 wt%. As far as the study of concept underlying this paper is concerned, we are trying to embed $Li_2B_4O_7$ with the lowest 207 concentration (in order to ensure the highest possible scintillator light yield) that still ensures a 208 significant increase of the neutron absorption yield. Now, the evolution curves show that, when 209 adding 0.3 wt% of lithium tetraborate, a 350 % increase of τ_a compared to the unloaded 210 equivalent scintillator is observed. We thus selected this initial load for the study of concept. 211

Because of the homogenous modeling of the loading, the estimated neutron yield is directly accounted for by sum of the values of $N \cdot \sigma_{E_{th}}$, N being the atomic density of a given neutron absorber, and $\sigma_{E_{th}}$ the above-introduced cross section. Figure 1 illustrates the second motivation behind the choice of lithium-6 and boron-10 as neutron converters: a large microscopic cross section that compensates their relatively low atomic density even with such a low weight fraction as 0.3 wt%.

Now, our model will give accurate expectations of the performance of the lithium 218 tetraborate-loaded scintillator scheme, provided that the charged (n, α) reaction products 219 escape the nanoparticles with as low an energy loss inside Li₂B₄O₇ as possible, so that they 220 remain spectrally identifiable. To assess the appropriateness of the size of the nanoparticles in 221 our application, we estimated the stopping power S of lithium tetraborate, expressed in 222 keV.μm⁻¹, for lithium ions at 0.84 MeV, alpha particles at 1.47 MeV and 2.05 MeV, and tritium 223 (i.e. hydrogen) at 2.73 MeV. The estimation was carried out using the validated software 224 package.¹⁷ The values of S equal 626.9 keV.µm⁻¹ for lithium ions at 0.84 MeV, 372.9 keV.µm⁻¹ 225 and 319.7 keV.µm⁻¹ for alpha particles at 1.47 MeV and 2.05 MeV, and 26.2 keV.µm⁻¹ for tritium 226 227 at 2.73 MeV. Several NP syntheses were performed to afford them the smallest size, and our 228 best method allowed us to isolate Li₂B₄O₇ NPs with an average size around 66 ± 22 nm. This batch was ultimately used for scintillator loading. Considering the most penalizing scenario of a 229 reaction product generated on the surface of the nanoparticle, and having to cross the longest 230 axis of the nanoparticle (herein most probably the NP diameter) before reaching the polymer 231 232 matrix, the maximum range inside the lithium tetraborate is considered equal to 66 nm. It thus 233 follows that a lithium ion of 0.84 MeV will lay down a maximum of 41.3 keV inside the nanoparticle (about 5 % of its initial energy), alpha particles of 1.47 MeV and 2.05 MeV maxima 234 235 of 24.6 keV and 21.1 keV respectively (2 % and 1 %), and a tritium ion a maximum of 1.7 keV (less than 0.1 %). Such differences being far less than the expected energy resolution associated 236 to the scintillator spectral response, we conclude that they will not affect the identification 237 238 procedure, and therefore confirm that the size of the prepared nanoparticle is fully suitable 239 with the application at hand.

The simulated kinetic energy deposition $\tau_D(E)$ inside a 0.3 wt%-loaded scintillator, obtained from the interaction model we described at the beginning of this paragraph is reproduced in Figure 2, where labels (A), (B) and (C) relate to the reaction equations stated at the beginning of this paragraph. The energy deposition spectrum is dominated by the contribution of the reaction products from Eq. (A), (B) and (C), with the full energy expositions of the three Q-values that were listed. We note that the summation of the energy yields at the Q-values:

247
$$\tau_D(2.31 \text{ MeV}) + \tau_D(2.79 \text{ MeV}) + \tau_D(4.78 \text{ MeV}) = 1.42 \cdot 10^{-2} \text{ n}^{-1}$$

equals 99.3 % of the neutron capture yield $\tau_a(n, \alpha)$ at 0.3 wt% that was reported in Figure 1, which indicates that the charged reaction products are fully stopped by the polymer matrix and therefore contribute to the signal of interest. This validates the first motivation for the selection of lithium tetraborate as a dopant that we discussed above. Moreover, we observe that the ratio

253
$$\frac{\tau_D(2.79 \text{ MeV})}{\tau_D(2.31 \text{ MeV})} = \frac{8 \cdot 10^{-4}}{1.23 \cdot 10^{-2}} \sim 7 \%$$

while, as we mentioned, the reaction process of Eq. (B) is 16 times more likely than the one of Eq. (A): this ratio indicates that the gamma ray at 0.48 MeV is very like to escape the scintillator and therefore does not significantly contribute to a distinct signature of ${}^{10}B(n, \alpha)$ reactions from the one carried by the lithium-7 and alpha particle charged products. The same conclusion may be drawn for the electromagnetic products of (n, γ) captures, for which, as expected, the simulated spectrum shows no identifiable signature up to 10 MeV. Eventually, we see that

 $\frac{\tau_D(2.31 \text{ MeV}) + \tau_D(2.79 \text{ MeV})}{\tau_D(2.31 \text{ MeV}) + \tau_D(2.79 \text{ MeV}) + \tau_D(4.78 \text{ MeV})} \sim 92\%$

which corresponds to the ratios between the $N \cdot \sigma_{E_{th}}$ factors for boron-10 and lithium-6. In particular, it follows that boron-10 is expected to be the main contributor to the total alpha particle signature of neutron absorptions within the nanoparticle-loaded scintillator.

265



266 267 Figure 2. Available kinetic from reactions products of thermal neutron absorptions in a 0.3 wt% $Li_2B_4O_7$ -loaded plastic 268 scintillator. (A), (B) and (C) relate to the above-cited equations in the main text.

269

Let us mention, to conclude this paragraph, that the third motivation underlying our design is to use Pulse Shape Discrimination (PSD) as a tool for time-domain-based analysis. Such a processing ensures, in one single treatment, the discrimination of pulses emanating from gamma-rays, fast neutron interactions (explained above as delayed photoluminescence) and thermal neutrons (tritium, helium and lithium ions). In terms of signal characteristics, these particles are sorted from fastest to slowest pulse.¹⁸ Therefore, a PSD method is adequate to sort them into regions of interest inside a bi-parametric diagram, whose ideal pattern, as obtained after pulse classification, is depicted in Figure 3. When the abscissa gives the electronequivalent energy laid down by interaction product, the ordinate represents the fraction between delayed charge against the total charge of a given pulse over a duration. The alpha quenching will be taken into consideration in the next section.



- Figure 3. Idealized bi-parametric response of the NPs-loaded plastic scintillator to a partially moderated neutron source.
- 283

284 3.2. Preparation and characterization of the NP-loaded plastic scintillator

The preparation of the NP-loaded plastic scintillator has been described elsewhere.¹³ It 285 is noteworthy that the NPs are synthesized with the isotopes of interest, which are lithium-6 286 and boron-10. The photoluminescence data are representative of a scintillator with POPOP as 287 288 the secondary fluorophore, giving an emission wavelength maximum close to 425 nm. Li₂B₄O₇ is expected to display a photoluminescence around 360 nm.¹⁹ Under our experimental conditions, 289 we did not notice any photoluminescence emanating from the nanoparticles. Falling under the 290 291 absorption domain of POPOP, our hypothesis is that these photons may be reabsorbed by the secondary fluorophore. The transmission spectrum showed a strong absorption overall the 292 293 visible domain, and as a result only 17 % of the light is transmitted at the maximum of emission of the scintillator. At the same wavelength, a reference EJ-200 plastic scintillator of same size 294 295 and volume showed a 67 % transmission. A picture of the NP-loaded plastic and an EJ-200 sample is given Figure 4. A relative scintillation yield was measured against the same EJ-200 296 sample. ⁶⁰Co pulse height spectra were recorded with the same experimental setup, and the 297 position of the Compton edges was compared according to the method described in the 298 Experimental section. The relative scintillation yield was thus estimated around 3,750 ph/MeV. 299

300



301 302 Figure 4. Picture of EJ-200 (left) and NP-loaded (right) plastic scintillators.

303

304 3.1. Neutron detection

The as-prepared plastic scintillator bearing isotopically enriched ⁶Li₂¹⁰B₄O₇ nanoparticles 305 was exposed to a partially polyethylene-thermalized 252-californium source. Thus, neutrons 306 with various energies can interact within the scintillator, including 25 meV thermal neutrons. 307 Figure 5 represents the energy-calibrated, pulse shape discrimination spectrum of the plastic 308 scintillator. Three different regions of interest appear: the gamma plume at lower Qtail/Qtotal 309 310 ratio, the fast neutron plume at higher Q_{tail}/Q_{total} ratio, and inside this fast neutron region an oval zone centered around 360 keVee that corresponds to thermal neutron capture reactions, 311 as was explained in Figure 3. The calculated Figure of Merit value (FoM) around the Compton 312 edge of ¹³⁷Cs (477 keVee) is 1.40. The FoM was also calculated for six different energy ranges, 313 from low electron equivalent energy (230 keVee) up to 500 keVee, with a focus around the 314 315 position of the alpha oval. It is noteworthy that the range that contains this oval displays a lower FoM, with a value of 1.15, whereas the surrounding FoM values are 1.21 and 1.36. This is 316 explained by the broadening of the neutron-like pseudo-Gaussian shape, thus the increase of 317 σ_n , which counts into the calculation of the FoM. On a side experiment, the NP-loaded plastic 318 319 scintillator was benchmarked with two other plastics: its unloaded counterpart (hereafter 320 designed as "regular") and an EJ-276 commercial scintillator. The three scintillators displayed approximately the same volume within 18%. The calculated FoM values in the same energy 321 range were 1.03, 1.26 and 1.79 for the NP-loaded, regular and EJ-276 plastic scintillators, 322 respectively. 323



325

Figure 5. Neutron/gamma pulse shape discrimination pattern of the ⁶Li₂¹⁰B₄O₇
 nanoparticle-loaded plastic scintillator. A partially thermalized and shielded ²⁵²Cf source was
 used for this experiment.

329

As a result of $Li_2B_4O_7$ loading, this ovoid signature between 300 and 420 keVee may be attributed to various reaction products:

- a) Exclusive contribution of alpha from ⁶Li (2.05 MeV);
- b) Exclusive contribution of alpha from ¹⁰B (1.47 MeV);
- c) Exclusive contribution of triton from ⁶Li (2.73 MeV);
- d) Exclusive contribution of lithium ion from ¹⁰B (0.84 MeV);
- e) Combination of scenarios a) and b);
- f) Exclusive contribution of the particles issued from ⁶Li;
- 338 g) Exclusive contribution of the particles issued from ¹⁰B;
- h) Combination of scenarios f) and g).
- 340
- The quenching factors occurring in plastic scintillators were then estimated. The Bethe formula gives: $(dE/dx) \propto z^2$ where z is the particle charge. The Birks formula gives $E_{ee}(E) \propto$

343 $L(E) \propto \int_0^E \frac{dE}{1+kB\left(\frac{dE}{dx}\right)} \approx \frac{E}{1+kB\varphi}$ with $\varphi \propto z^2$. So with important approximations $QF(E) = \frac{E_{ee}}{E} \propto 1$

344 $\frac{1}{z^2}$, with QF being the quenching factor (in MeVee.MeV⁻¹).

346
$$QF(E) = \frac{E_{ee}}{E} = \frac{0.034 \times E^2 + 0.125 \times E}{E}$$

The Bethe formula being independent from the particle's mass, we can thus consider the quenching factor of tritons somehow equal to that of the proton, so:

349
$$QF_T(2.73) \approx 0.218$$

For z = 2 (alphas), the quenching factor is estimated as

351
$$QF_{\alpha}(1,47) = \left(\frac{0.034 \times 1.47^2 + 0.125 \times 1.47}{1.47}\right)^2 \approx 0.031$$

352
$$QF_{\alpha}(2,05) = \left(\frac{0.034 \times 2.05^2 + 0.125 \times 2.05}{2.05}\right)^2 \approx 0.038$$

Based on the work of Mahl *et al.*,²¹ we can estimate their quenching factor as 60/1470 = 0.041, which agrees with the order of magnitude of the calculated factor (formulation with PSD).

Finally, the same calculation for lithium ions (z = 3) gives a quenching factor

357
$$QF_{Li}(0.84) = \left(\frac{0.034 \times 0.84^2 + 0.125 \times 0.84}{0.84}\right)^3 \approx 0.0036$$

358 If these various quenching factors are reinjected into the different scenarios, we have:

a) A single signal centered around 0.038 × 2050 = 78 keVee;

b) A single signal centered around 0.031 × 1470 = 45 keVee;

361 c) A single signal centered around 0.218 × 2730 = 516 keVee;

d) A single signal centered around 0.0036 × 840 = 3 keVee;

e) A dual response giving an ovoid with two centers at 45 and 78 keVee;

f) A signal centered around 516 + 78 = 594 keVee;

365 g) A signal centered around 45 + 3 = 48 keVee;

h) A signal centered around 594 keVee and a signal around 48 keVee.

In conclusion, as far as orders of magnitude are considered, only scenarios c) and h) remain plausible. Consequently, the ovoid signal between 300 and 420 keVee differs from our rough calculations by a 30-to-50-% relative factor; we shall interpret this feature as being mainly attributable to the interaction between neutron-released tritons and the plastic medium.

Mahl et al. used a subtraction method to emphasize the thermal neutron capture in 372 their design.²¹ Thus, we completed our setup experiment in order to highlight the (n, α) 373 374 reactions. A cage consisting of boron-loaded chipboard wood shielded on five edges the {scintillator + PMT} system. In a first experiment, the ²⁵²Cf source was thermalized with 10 cm 375 high density polyethylene (HDPE) and shielded with 5 cm of denal[®]. The second experiment was 376 performed with the same strict experimental procedure except that HDPE was replaced by 377 378 boron-loaded HDPE, the latter acting as thermal neutron poison, allowing therefore subtracting 379 the thermal neutron captures we are interested in. This double experiment was performed twice at two different threshold values and the difference spectrum between these two 380 experiments was computed. The results given in Figure 6 show the full bidimensional 381 382 acquisition on the top and the neutron-associated region energy spectrum on the bottom. Again, the light ions oval located in the 350 keVee region remains visible but another pattern 383 appears at lower energies, in the range 20 – 60 keVee. These counts are undoubtedly attributed 384 to another neutron capture, but occurring now from the boron atom. 385

The sum of the counts S_{net} was calculated on these two regions (20 – 60 and 300 – 386 387 400 keVee), giving values of 2281 and 771 counts, respectively. The standard deviations Std(S_{net}) in the same regions were also calculated, giving values of 407 and 98, respectively. 388 Thus we have: $S_{net} = 5.6 \times Std(S_{net})$ in the range [20 - 60 keVee] and $S_{net} = 7.9 \times Std(S_{net})$ in the 389 390 range [300 – 400 keVee]. The comparison between S_{net} and Std(S_{net}) reveals that the signal isolated in these areas of interest is significant of the set of physical, neutron-related 391 392 interactions we aimed at emphasizing in this experiment. Our material hence enables the distinct observation of four physical phenomena: gamma interaction, fast neutron interaction, 393 thermal neutron capture by a ¹⁰B nucleus and ultimately thermal neutron capture by a ⁶Li 394 nucleus. 395

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Figure 6. Top: subtracted spectra of pulse shape discrimination between thermalized neutrons
 and thermalized + absorbed neutrons. Bottom: neutron-associated region energy spectrum.

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402 4. Conclusion

403 Lithium tetraborate (Li₂B₄O₇) is a low-Z but dense material that is particularly useful in the thermal neutron detection for its ability to convert 25 meV neutrons to alpha and triton 404 particles. Based on literature precedents, lithium tetraborate nanoparticles were synthesized 405 with a good monodispersity degree and were dispersed into fast n/γ discriminating plastic 406 scintillators. MCNP simulations showed that a beneficial effect of plastic scintillation loading 407 408 could be reached starting at 0.1 wt% of $Li_2B_4O_7$ added to the material. We succeeded in 409 preparing a plastic scintillator with a concentration of 0.3 wt% of nanoparticles, albeit with a moderate light yield. The chemically-modified plastic scintillator then displayed numerous 410 signatures: gamma counting, fast neutrons discrimination and two extra regions emanating 411 from thermal neutrons captures, namely from ⁶Li and ¹⁰B nuclei. 412

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