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### ► To cite this version:

Antoine Gobert, Raphaëlle Tourdot-Maréchal, Céline Sparrow, Christophe Morge, Hervé Alexandre. Influence of nitrogen status in wine alcoholic fermentation. Food Microbiology, 2019, 83, pp.71 - 85. 10.1016/j.fm.2019.04.008 . hal-03484453

# HAL Id: hal-03484453 https://hal.science/hal-03484453

Submitted on 20 Dec 2021

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## Influence of nitrogen status in wine alcoholic fermentation

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# Keywords: nitrogen, amino acids, ammonium, alcoholic fermentation, yeasts, wine, volatile compounds.

12 Abstract

13 Nitrogen is an essential nutrient for yeast during alcoholic fermentation. Nitrogen is involved in the biosynthesis of protein, amino acids, nucleotides, and other metabolites, including 14 15 volatile compounds. However, recent studies have called several mechanisms that regulate its 16 role in biosynthesis into question. An initial focus on S. cerevisiae has highlighted that the concept of "preferred" versus "non-preferred" nitrogen sources is extremely variable and 17 18 strain-dependent. Then, the direct involvement of amino acids consumed in the formation of 19 proteins and volatile compounds has recently been reevaluated. Indeed, studies have 20 highlighted the key role of lipids in nitrogen regulation in S. cerevisiae and their involvement 21 in the mechanism of cell death. New winemaking strategies using non-Saccharomyces yeast 22 strains in co- or sequential fermentation improve nitrogen management. Indeed, recent studies

show that non-*Saccharomyces* yeasts have significant and specific needs for nitrogen. Moreover, sluggish fermentation can occur when they are associated with *S. cerevisiae*, necessitating nitrogen addition. In this context, we will present the consequences of nitrogen addition, discussing the sources, time of addition, transcriptome changes, and effect on volatile compound composition.

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#### 51 **1** Introduction

52 The main sources of yeast assimilable nitrogen (YAN) in grape must are ammonium and amino acids. Their concentrations vary depending on geographical location (Rapp and 53 54 Versini, 1995), climate (Ribéreau-Gayon et al., 2006), cultivar or rootstock (Schreiner et al., 55 2017; Stines et al., 2000), and viticulture techniques (Schreiner et al., 2017; Spayd et al., 56 1994). During alcoholic fermentation, yeast take up and metabolize YAN and other nutrients 57 to support growth and produce biomass, as well as volatile compounds (Vilanova et al., 58 2007). YAN deficiency can sometimes lead to sluggish or stuck fermentation (Alexandre and 59 Charpentier, 1998; Bisson, 1999). Under enological conditions, a concentration of 60 approximately 140 mg N/L of YAN is necessary to complete fermentation within a reasonable 61 period of time (Beltran et al., 2005; Bely et al., 1990; Bisson, 1999; Henschke and Jiranek, 62 1993; Jiranek et al., 1995; Kemsawasd et al., 2015), depending on sugar concentration and 63 winemaking practices.

64 During alcoholic fermentation, the consumption of YAN by yeast is regulated by several 65 molecular mechanisms, which have been well described in Saccharomyces cerevisiae. The 66 most recent reviews on the subject date back to 2005 and 2012 (Bell and Henschke, 2005; Ljungdahl and Daignan-Fornier, 2012) and only one review on nitrogen regulation was been 67 recently published (Zhang et al., 2018). The authors examined several studies on the role of 68 69 nitrogen metabolism under enological conditions. Most studies have classified YAN as 70 "preferential or non-preferential sources", depending on the alcoholic fermentation 71 conditions, strains used, and classification method (Beltran et al., 2006; Crépin et al., 2012; 72 Gobert et al., 2017; Jiranek et al., 1995; Kemsawasd et al., 2015; Rollero et al., 2018). The comparison of the studies previously cited highlighted significant differences. Such 73 74 differences could be due to the complexity of nitrogen regulation, which depends on substrate 75 availability, strain phenotype and matrix.

Recent interest in non-Saccharomyces (NS) yeasts in spontaneous fermentation (Combina et 76 77 al., 2005; Cordero-Bueso et al., 2013; Jolly et al., 2014; Liu et al., 2016) and co- or sequential fermentation (Anfang et al., 2009; Ciani et al., 2010; Clemente-Jimenez et al., 2005; Englezos 78 79 et al., 2018; Medina et al., 2013; Padilla et al., 2017; Soden et al., 2000) add a second level of 80 complexity to nitrogen management under enological conditions. Despite the large body of 81 literature concerning the use of NS yeasts to increase the aromatic complexity of wine 82 (Azzolini et al., 2014; Escribano et al., 2018; Lambrechts and Pretorius, 2000; Liu et al., 83 2016; Sadoudi et al., 2017, 2012; Swiegers and Pretorius, 2005), improve ethanol reduction 84 (Canonico et al., 2016; Ciani et al., 2016; Contreras et al., 2015, 2014; Englezos et al., 2016; 85 Gobbi et al., 2014; Röcker et al., 2016; Rolle et al., 2017), or act as bio-protection/control 86 agents in winemaking (Cordero-Bueso et al., 2017; Fernandes Lemos Junior et al., 2016; Qin 87 et al., 2015; Simonin et al., 2018; Wang et al., 2018), little data on nitrogen needs, sources, 88 and preferences are available and have never been reviewed (Andorrà et al., 2010; Gobert et 89 al., 2017; Kemsawasd et al., 2015). In addition, only one study (Englezos et al., 2018) has 90 highlighted specific features of one NS yeast, Starmerrella bacillaris, concerning the 91 management of nitrogen.

92 Aromatic complexity is an essential aspect of wine quality and largely influences consumer 93 acceptance (King et al., 2011; Lattey et al., 2010). Volatile compounds, an essential element 94 of overall wine flavor, are formed during alcoholic fermentation. The relationship between 95 nutrient availability and the production of desirable volatile compounds is one of the main 96 goals in enology and industry. Some YAN sources have been reported to be precursors of 97 volatile compounds in S. cerevisiae (Carrau et al., 2008; Fairbairn et al., 2017; Hazelwood et 98 al., 2008; Ribéreau-Gayon et al., 2006) and non-Saccharomyces (González et al., 2018) 99 principally via the Ehrlich pathway (González et al., 2018; Hazelwood et al., 2008). However, 100 contrary to the generally accepted view, recent studies have shown that the catabolism of 101 consumed branched amino acids (leucine, isoleucine, threonine and valine) plays an indirect 102 role in the formation of some volatile compounds in S.cerevisiae (Crépin et al., 2017; Rollero 103 et al., 2017). Precisely, Crépin et al. (2017) demonstrated the low contribution of the carbon 104 skeletons of consumed amino acids to the production of volatile compounds derived from  $\alpha$ -105 keto acids. Lipid and nitrogen metabolism are interconnected and affect the production of 106 some volatile compounds (Rollero et al., 2017, 2016). The aforementioned studies have 107 shown that there may be a direct relationship between nitrogen sources and the production of 108 volatile compounds and that it may involve more complex mechanisms than those reported to 109 date.

110 Although a large diversity of YAN content and concentration in must can be found during 111 enological processes, ammonium phosphate or ammonium sulfate are commonly added to 112 YAN deficient must. YAN supplementation directly affects biomass production and the 113 performance of alcoholic fermentation (Martínez-Moreno et al., 2012; Varela et al., 2004). 114 However, although it is clear that YAN impact volatile compounds production, the 115 mechanisms implicated seem to be indirect and need to be more investigated. It has been 116 demonstrated that the use of an amino-acid mix for must supplementation increases the rate of 117 alcoholic fermentation more than ammonium phosphate or sulfate addition, with sometimes 118 less production of undesirable volatile compounds (Fairbairn et al., 2017; Kevvai et al., 2016; 119 Martínez-Moreno et al., 2012).

- 120 2 YAN metabolism in yeasts
- 121 2.1 Saccharomyces cerevisiae

#### 122 **2.1.1** Transporters and general regulation in *Saccharomyces cerevisiae*

Although diverse yeast species are used in the first step of winemaking (Barata et al., 2012;
Capozzi et al., 2015; Gilbert et al., 2014), many studies have focused on the nitrogen

metabolism of *S. cerevisiae* (Crépin et al., 2017, 2012; Grenson et al., 1974; Hazelwood et al.,
2008; Jiranek et al., 1995; Mitchell, 1985; Stanbrough and Magasanik, 1995) and many
reviews have focused on its metabolism (Bell and Henschke, 2005; Henschke and Jiranek,
1993; Horák, 1997; Ljungdahl and Daignan-Fornier, 2012; Ramos et al., 2016). Here, we will
only provide a brief general description of *S. cerevisiae* nitrogen metabolism, with a focus on
recent studies. The readers can refer to the above-cited articles for an in-depth description of *Saccharomyces cerevisiae* nitrogen metabolism

In *S. cerevisiae*, YAN is transported into the cell by various specific or non-specific permeases. Ammonium, which represents a significant proportion of nitrogen sources, is transported by three permeases: Mep1p, Mep2p, and Mep3p. The Mep2 protein displays the highest affinity for ammonium, followed closely by Mep1p and finally Mep3p, of which the affinity is much lower (Marini et al., 1997). These transporters consist of uniport systems.

137 Amino acids are assimilated by various, more or less, selective transporters. Amino-acid 138 permeases (AAPs) are active symport systems (Kotyk, 1994; Ramos et al., 2016). Among 139 them, the general amino-acid permease (Gap1) allows the transport of all amino acids. Other AAPs are more selective and transport only one or a group of amino acids (Ramos et al., 140 141 2016). These permeases are regulated by several mechanisms. The first is located on the 142 plasma membrane and forms a complex called Ssy1p-Ptr3p-Ssy5p (SPS). This complex 143 induces an endoproteolytic processing event in response to the extracellular amino-acid status, 144 which activates the transcription of AAP genes (Figure 1) (Andréasson and Ljungdahl, 2004; 145 Ljungdahl, 2009).

146 Nitrogen catabolite repression (NCR) is another regulatory system used by *S. cerevisiae*.
147 Currently, *GAP1*, *CAN1*, *PUT4*, *DIP5*, *UGA4* (amino acids permeases) and *MEP1*, *MEP2*,
148 *MEP3* (ammonium permease) are known to be under the control of this regulatory system

149 (Ljungdahl and Daignan-Fornier, 2012). This system leads the yeast to selectively utilize 150 preferred sources of nitrogen when they are available. Conversely, general de-repression of 151 the genes regulated by the NCR system leads the cell to nonspecifically use other sources of 152 nitrogen in the absence of a preferential nitrogen source. The classification of nitrogen source 153 preferences is not absolute and their repressive effects can vary substantially between yeast 154 strains (Magasanik and Kaiser, 2002). The expression of genes encoding NCR-sensitive 155 AAPs is regulated by complex pathways involving multiple transcription factors with 156 activating or inhibitory effects. The target of these transcription factors is the UAS<sub>NTR</sub> 157 activation sequence, located upstream of the promoter of genes encoding NCR-sensitive 158 AAPs. This element consists mainly of two distinct dodecanucleotide sites with a 159 pentanucleotide consensus sequence 5'-GATATA-3 '. A pair of Dal80/Uga43 proteins and 160 three Gat1, Ure2, Gln3 proteins have been identified and reported to participate in the 161 regulation of NCR-sensitive gene expression across the UAS<sub>NTR</sub> element (Cunningham et al., 162 1996). However, results from a more recent study suggests that Ure2p and GATA 163 transcription factors (Gln3, Gat1, Dal80 and Gzf3 transcription factors) are involved (Georis 164 et al., 2009). Thus, in the study of Cunningham et al. (1996), Gln3p and Gat1p were factors 165 that allowed the activation of NCR-sensitive gene transcription. The factors Dal80/Uga43 are 166 proteins that block the binding site of the Gln3 and Gat1 factors. Ure2 sequesters Gln3 and 167 Gat1 proteins in the cytoplasm. In the study of Georis et al. (2009), Gln3 and Gat1 were 168 shown to be activators of NCR-sensitive gene expression. However, there are differences at 169 the level of the repression systems. Georis et al. (2009) showed that Dal80 and Gzf3, but not 170 Uga43, were blocking proteins of the UAS<sub>NTR</sub> binding site. Ure2 was shown to have the same 171 role in both studies (Figure 2).

In *S. cerevisiae*, the NCR system is controlled by the target of the rapamycin pathway (TOR),
which is comprised of two complexes, TOR complex 1 (TORC1) and TOR complex 2

174 (TORC2). Only TORC1 is involved in the control of the NCR system. This aspect will not be 175 discussed further here, as nitrogen metabolism, including the regulation of sensing, transport, 176 and catabolism was recently extensively reviewed by Zhang et al. (2018). However, a study 177 showed that TORC1-mediated control of NCR is only partial (Fayyad-Kazan et al., 2016). 178 The authors investigated the regulation of the NCR system by glutamate and glutamine, major 179 nitrogen sources for biosynthesis. The results showed a negative role of the anabolic 180 glutamate dehydrogenase (Gdh1) on Gat1 and Gln3 (transcriptional activators of NCR genes) 181 activity under repressive nitrogen conditions. Thus, preferred nitrogen sources may trigger 182 NCR-sensitive gene repression or TORC1 activation, probably through transient glutamine 183 accumulation.

184 More broadly, global gene expression is strongly influenced by the concentration of nitrogen 185 during fermentation. In synthetic must under conditions of low nitrogen, genes mainly 186 associated with protein synthesis and RNA and nucleic acid metabolism are downregulated at 187 the beginning of fermentation (24 to 48 h) (Mendes-Ferreira et al., 2007). In contrast, genes 188 involved in energy generation, carbohydrate metabolism, oxidoreductase activity, respiratory 189 chain phosphorylation, transporter activity, respiration, response to oxidative stress, oxygen, 190 and reactive oxygen species metabolism are upregulated. This expression profile is 191 sustainable, as the genes expressed at the end of fermentation are fundamentally the same as 192 those expressed early in the yeast cell response (Mendes-Ferreira et al., 2007). Under the 193 same conditions, Barbosa et al. (2015) showed that gene expression varies highly among yeast 194 strains, depending on nitrogen availability (low or high concentration), fermentation stage, 195 and the interaction of the two factors. In particular, the high fermenter yeast strain (S. 196 cerevisiae UCD522) used in this study showed high expression of 333 genes involved in 197 transport and phosphate metabolism throughout fermentation under conditions of low 198 nitrogen. In contrast, 246 genes were more highly expressed, including flocculation genes,

199 under conditions of high nitrogen (Barbosa et al., 2015). In addition, transcriptional regulation 200 can be affected by the nitrogen source. In 2006, Boer et al., evaluated the transcriptional 201 response of S. cerevisiae in glucose-limited chemostat culture in synthetic must. 202 Phenylalanine, leucine, methionine, and proline were used as "non-preferred" nitrogen 203 sources and asparagine and ammonium as "preferred" nitrogen sources. A group of 23 genes 204 was upregulated during growth on "non-preferred" sources. Among them, six are involved in 205 the metabolism of the "non-preferred" nitrogen sources allantoin and urea (DAL1, DAL2, 206 DAL5, DUR1, DUR2, DUR3) and five encode transporters for nitrogen-containing 207 compounds (GAP1, PTR2, MEP2, MEP3 and OPT2). This group also included the GATA 208 factor. From these 23 genes, 14 were established as NCR targets (Boer et al., 2006).

#### 209

#### 2.1.2 Classification of "preferred" and "non-preferred" nitrogen sources in

#### 210

#### Saccharomyces cerevisiae

211 The phenotypic diversity of S. cerevisiae concerning YAN preferences correlates with the 212 presence of genetic variants (Cubillos et al., 2017; Gutiérrez et al., 2013a; Salinas et al., 213 2012). Currently, three methods are used to classify YAN as "preferred" or "non-preferred" 214 nitrogen sources. The first, used since the 80's and 90's, considers the consumption rate, 215 specific growth rate, and kinetics of fermentation (Grenson, 1992, 1983; Henschke and 216 Jiranek, 1993; Jiranek et al., 1995). The second is based on the response of the NCR system. 217 YAN that induce the activation of transcription of permeases is considered as "preferred" 218 sources. Crépin et al. (2012) used this method to show that classification was possible according to the order of assimilation of YAN by S. cerevisiae. The last method is based on 219 220 the response of the SPS system (Boer et al., 2006; Ljungdahl, 2009). YAN sources that induce 221 the activation of transcription of permeases under the control of the SPS mechanism are 222 considered to be "preferred", as for the method based on NCR regulation.

223 Studies to classify nitrogen sources preferences have yielded varying results throughout the 224 years. Table 1 reviews all studies that have attempted to rank YAN preferences. Three YAN 225 sources (ammonium, asparagine, and glutamine) have been significantly more often classified 226 as preferred nitrogen sources, regardless of the classification method. In contrast, alanine, 227 GABA, glutamic acid, histidine, and methionine have been the least often classified. Overall, 228 17 YAN sources can theoretically be considered to be "preferred". This comparison highlights 229 the difficulty in comparing and determining a pool of preferential YAN. Indeed, in 230 winemaking, the metabolic activity of S. cerevisiae can be strongly influenced by the matrix. 231 Authors have used different synthetic grape musts or real musts, coming from different grape 232 varieties. YAN assimilation may be influenced by other nutrients, such as lipids or vitamins, 233 not present at the same concentration under all conditions. Without standardization of the 234 matrix, it is difficult to draw definitive conclusions on the YAN that are preferentially 235 assimilated. In addition, multifactorial abiotic factors (discussed later in this review) need to 236 be considered when categorizing YAN. The effect of the strain is also a variable that must be 237 considered. Each strain of S. cerevisiae shows differences in terms of their YAN consumption 238 capacities (Brice et al., 2018; Crépin et al., 2012; Cubillos et al., 2017).

239 Several studies have focused on arginine under fermentation conditions. Although arginine is 240 not considered to be a "preferred" nitrogen source in most studies, it plays a major role in 241 nitrogen metabolism. During the first stage of fermentation, part of the amino-acid content, 242 particularly arginine, localizes to the vacuole. After the complete exhaustion of nitrogen 243 sources at the end of the growth phase, arginine stored in the vacuole is remobilized to 244 maintain the biomass concentration, particularly for low biomass-producing S. cerevisiae 245 strains (Crépin et al., 2014). Cheng et al. (2016) showed a relationship between the response of S. cerevisiae to arginine and ethanol stress. They showed that the addition of 250 mg/L of 246 247 arginine under stressful conditions increased the cell density by 2.3-fold over that of the

control (Cheng et al., 2016). The same observation was made by Noti et al. (2018). They demonstrated that arginine exerted a positive effect (relative to other amino acids) on the growth and fermentation rate of *S. cerevisiae*. They suggested that arginine exerts this beneficial effect on growth not only because of its nutritional role as a nitrogen source, but also because of its effect as an osmoprotectant (Noti et al., 2018).

#### 253 **2.1.3** Nitrogen and volatile compounds metabolism in *Saccharomyces cerevisiae*

254 The metabolic network involved in the production of volatile compounds has been well 255 mapped by Styger et al. (2011). They studied genetic factors that affect the conversion of 256 amino acids to aroma compounds, focusing on regulation of the Ehrlich pathway (Hazelwood 257 et al., 2008). They showed that the expression of a cluster of ten genes could be significantly 258 correlated with the production of specific volatile compounds. This cluster could be divided 259 into three categories encoding five dehydrogenases, three decarboxylases, and two reductases. 260 Among the dehydrogenases, GPD2, ADH3, and OYE2 play an important role in cellular redox 261 maintenance reactions, whereas AAD6 and HOM2 appear to be directly involved in Ehrlich 262 reactions. In particular, HOM2 catalyzes the second step in the common pathway for 263 methionine and threonine biosynthesis. Its expression is regulated by Gcn4 and the general 264 control of amino-acid synthesis. Other genes, including PRO2, which catalyzes the second 265 step of proline biosynthesis, can perform their functions directly on the pathway (Styger et al., 266 2011). The function of the genes was verified by deletion of those that appeared to most 267 strongly affect the Ehrlich pathway. Deletion of HOM2 led to the greatest reduction in the 268 concentration of higher alcohols and volatile acids, suggesting that HOM2 plays a central and 269 direct role in the formation of both the higher alcohols and volatile acids. PRO2 deletion led 270 to an increase in the levels of fatty acids in strains grown on medium supplemented with 271 branched-chain amino acids. The authors suggested that PRO2 plays a different role in the 272 Ehrlich pathway than the other selected genes.

273 Other molecules, such as volatile thiols, also play an important organoleptic role, especially in 274 white wine. The regulation of production of two volatile thiols, 4-methyl-4-sulfanylpentan-2-275 one (4MSP) and 3-sulfanyl- hexan-1-ol (3SH), was investigated by Thibon et al. (2008) in 276 synthetic medium. The authors demonstrated that IRC7, a gene encoding a putative 277 cystathionine  $\beta$ -lyase, was one of the main genes encoding the protein that catalyzes 4MSP 278 and 3SH release under enological conditions. They also demonstrated that transcriptional 279 regulation of *IRC7* is associated with the NCR system (Thibon et al., 2008). The NCR system 280 is also involved in the regulation of the production of another thiol, 3-mercapto-hexanol 281 (3MH), by modulating the activity of Gap1 (Subileau et al., 2008).

282 Although many studies have shown a significant correlation between YAN and volatile 283 compounds production (Carrau et al., 2008; Fairbairn et al., 2017; Hernandez-Orte et al., 284 2006; Hernández-Orte et al., 2006, 2004), recent studies suggest that this link may be much 285 weaker than previously thought (Crépin et al., 2017; Rollero et al., 2017). In a first part, 286 Crépin et al. (2017) showed that the main role of the catabolism of most of the consumed 287 amino acids is to supply nitrogen for the *de novo* synthesis of proteinogenic amino acids in 288 accordance with anabolic requirements. This de novo synthesis (provided approximately 60 to 289 80% of the demand for nearly all amino acids) achieved with the carbon backbones of 290 precursors (mainly  $\alpha$ -keto acids) in addition to ammonium donors. In a second part, <sup>13</sup>C 291 tracer used by authors provided evidence that the carbon intermediates originated primarily 292 from the catabolism of sugars through the central carbon metabolism, while the catabolism of 293 and interconversions between amino acids accounted for a limited portion of the formation of 294 α-keto acids. Furthermore, a low contribution of the carbon skeletons of consumed amino 295 acids to the production of volatile compounds derived from  $\alpha$ -keto acids was demonstrated 296 (Crepineet al., 2017). Rollero et al. (2017) complemented this study by investigating how 297 variations in nitrogen and lipid resources can influence the contributions of both nitrogen and

298 carbon metabolism to the production of fermentative aromas using filiation experiments with 299 <sup>13</sup>C-labelled leucine and valine nitrogen sources. They showed that only a small fraction of 300 higher alcohols was synthetized using the carbon skeletons of amino acids. Moreover, the 301 quantity of YAN available before fermentation had a strong impact on both metabolic protein 302 synthesis and higher alcohol flux. Under conditions of low nitrogen (70 mg/L), the direct 303 incorporation of exogenous leucine into biomass was limited to 30%. This observation is in 304 accordance with the results of Crépin et al. (2017), which show that consumed amino acids 305 provide an intracellular nitrogen pool for *de novo* synthesis. However, under conditions of 306 moderate (250 mg/L) to high nitrogen (425 mg/L), the direct incorporation of consumed 307 leucine into biomass reached 70%. This result appears to contradict that of Crépin et al. 308 (2017). However, in this study, a low available YAN condition was tested (180 mg/L of YAN 309 for 240 g/L of glucose). Variations in the concentration of YAN appeared to influence the 310 percentage of direct incorporation of amino acids into biomass formation. Simultaneously, the 311 total production of higher alcohols changed differently, depending on nitrogen availability. 312 Rollero et al. (2017) showed that an increase of nitrogen from 70 to 250 mg/L increases the 313 synthesis of  $\alpha$ -ketoacids via central carbon metabolism, which results in a marked increase in 314 the flux towards the formation of higher alcohols. However, at high concentration of YAN 315 (425 mg/L), intracellular  $\alpha$ -ketoacids are, to a large extent, directed towards the synthesis of 316 amino acids at the expense of higher alcohol formation, resulting in a decrease in the 317 formation of aromas relative to the previous conditions. In opposite, it is indeed well known 318 that the production of higher alcohols is highly affected by the nitrogen concentration and 319 composition of grape must (Beltran et al., 2005; Carrau et al., 2008; Vilanova et al., 2007). 320 Nitrogen starvation increases the production of higher alcohols, probably due to the higher 321 levels of  $\alpha$ -keto acids produced on those conditions, which cannot be transaminated and are in 322 turn converted into higher alcohols.

Nevertheless, the strain under study, use of synthetic medium, concentration of nitrogen sources, temperature, and anaerobic and stirring conditions are all elements to consider in validating the redistribution of YAN into the cell for the *de novo* synthesis of proteinogenic amino acids and the consequences on the production of higher alcohols.

327 2

#### 2.2 Non-Saccharomyces yeasts

Nitrogen metabolism of NS yeasts is a recent topic of interest in enology. Little data is 328 329 available and published studies have never been reviewed. There is also recent 330 biotechnological interest in NS yeasts in winemaking, including their needs, in particular that 331 of nitrogen. The use of NS yeasts in co- or sequential fermentation generally leads to YAN 332 consumption and competition for nitrogen sources with S. cerevisiae. These types of 333 fermentation can ultimately lead to sluggish or stuck fermentation (Medina et al., 2012). 334 Consequently, a better comprehensive understanding of the metabolic nitrogen requirements 335 of NS yeasts and their regulation will aid in the control of fermentation. The first studies 336 focused on preferential nitrogen sources of NS yeasts (Andorrà et al., 2010; Gobert et al., 337 2017; Kemsawasd et al., 2015). As for S. cerevisiae, the term "preferential" is questionable 338 for NS yeasts. The difficulty in identifying "preferential" sources lies in that it depends on the 339 strain, fermentation method and classification method (Andorrà et al., 2010; Gobert et al., 340 2017; Kemsawasd et al., 2015). Available studies provide a first overview of "preferential" 341 nitrogen sources, mainly according to the capacity of NS yeasts to assimilate nitrogen and its 342 rate of consumption. No molecular classification based on the regulation of nitrogen 343 assimilation, such as the NCR or SPS systems (known in S. cerevisiae) is currently available. 344 Currently, only one study indicated the presence of the NCR mechanism in Hanseniaspora 345 vineae (Lleixà et al., 2019). Table 2 shows the first proposed classification of the various "preferential" nitrogen sources according to NS yeast strain, although such a comparison is 346 not evident. Six NS yeasts have been studied: H. uvarum, L. thermotolerans, M. pulcherrima, 347

348 S. bacillaris, T. delbrueckii, and P. membranifaciens. Most publications have reported 349 glutamine as the preferred nitrogen source for H. uvarum, L. thermotolerans, M. pulcherrima, 350 and T. delbrueckii and leucine for H. uvarum, L. thermotolerans, and M. pulcherrima. 351 However, Gobert et al. (2017) showed a significant effect of fermentation temperature on the 352 assimilation and consumption rates of nitrogen sources. For example, leucine was classified 353 as a "preferred" source at 28°C for M. pulcherrima, whereas it was classified as an 354 "intermediate" source at 20°C. This effect has also been observed for S. cerevisiae (Beltran et 355 al., 2006). No studies are available to explain this phenomenon, but the recent study of 356 Englezos et al. (2018) on NS yeast S. bacillaris provides a first answer. Surprisingly, the 357 strains studied showed poor assimilation of amino acids during alcoholic fermentation relative 358 to that of ammonium, which was entirely consumed. In addition, the production of several 359 amino acids, such as alanine, glutamic acid, glycine, and valine was observed, whereas 360 precedent studies have never reported such production. These results showing the ability of S. 361 bacillaris to preferentially consume ammonium suggest less efficient SPS-dependent 362 regulation (based on the S. cerevisiae model) of amino-acid permeases or an inhibitory 363 mechanism mediated by ammonium. A significant proportion of YAN can be assimilated by 364 NS yeast strains before the predominance of S. cerevisiae in the fermentation process. As a 365 consequence, the growth and fermentation kinetics of S. cerevisiae may be negatively 366 affected (Gobert et al., 2017; Kemsawasd et al., 2015; Medina et al., 2012; Rollero et al., 367 2018; Wang et al., 2015). In particular, Gobert et al. (2017) showed that NS yeasts can 368 consume between 66 and 215 mg/L of YAN, depending on the species. Of course, the rate of 369 consumption can vary, depending on the winemaking methods (spontaneous fermentation, coinoculation, sequential fermentation, temperature, aeration) and matrix (YAN must 370 371 composition). Thus, nitrogen deficiencies need to be prevented by nitrogen supplementation 372 to cover the needs of *S. cerevisiae* in the must for optimal alcoholic fermentation.

#### **373 3 Factors that influence YAN assimilation**

#### 374 **3.1 Sugar - nitrogen balance**

Sluggish or stuck fermentations can be prevented by nitrogen supplementation when using 375 376 deficient must. Several studies have been performed to determine the optimal nitrogen 377 concentration in must to guarantee complete fermentation (Alexandre and Charpentier, 1998; 378 Bely et al., 1990; Bisson, 1999; Henschke and Jiranek, 1993; Jiranek et al., 1995; Mendes-379 Ferreira et al., 2004). It is generally accepted that 120 at 140 mg N/L is sufficient (Bely et al., 380 1990) to complete the fermentation of 200 g/L of sugar. However, the needs may be higher, 381 depending on winemaking practices (Mendes-Ferreira et al., 2004). The ratio of nitrogen to 382 carbon sources during fermentation needs to be balanced to ensure good metabolic activity of 383 the yeast. Several studies have associated nitrogen deficiency with a high turnover rate of 384 sugar transporters in nitrogen-deficient must, resulting in a loss of sugar uptake capacity by 385 the cells (Bisson, 1999; Salmon, 1989). Varela et al. (2004) showed that the rate of carbon 386 uptake in "normal" fermentation was 3.6-fold higher during the exponential phase and 10-fold 387 higher during the late stationary phase than for "sluggish" fermentation. The authors found 388 that nitrogen deficiency affects the glucose transporter turnover rate and the expression of at 389 least one: HXT1 (low-affinity glucose transporter). Surprisingly, under conditions in which 390 sugar uptake is low, they showed that adding biomass from sluggish cultures (from another 391 tank) not only reduced the time to finish a problematic fermentation but was also less likely to 392 affect the quality of the resulting wine. The higher the concentration of biomass, the quicker 393 the fermentation was completed, even if the cells were grown in nitrogen-deficient medium 394 (Maisonnave et al., 2013; Varela et al., 2004). Biomass formation coupled with macro- and 395 micronutrients, including YAN, appear to be essential parameters involved in the mechanism 396 of sluggish fermentation.

#### 397 **3.2** *S. cerevisiae* strain effect

398 In 2012, Crépin et al. showed that the kinetics of YAN consumption were strongly strain-399 dependent. Fourteen S. cerevisiae strains were tested in synthetic medium. The maximum 400 rate of nitrogen consumption was between 16.8 and 27.8 mg N/L/h, which represented 40% of 401 the variability in terms of the kinetics of nitrogen consumption. More recently, Lemos Junior 402 et al. (2017) estimated the YAN needs of seven S. cerevisiae vineyard strains in synthetic 403 must using three different YAN concentrations (70 mg N/L, 150 mg N/L, and 300 mg N/L). 404 The authors showed no significant differences in fermentation kinetics at 150 and 300 mg N/L 405 for all but one strain. However, there was a strong decrease in the rate of fermentation at 70 406 mg N/L. Thus, the fermentation kinetics appears to strongly correlate with the S. cerevisiae 407 strain in conventional winemaking. In addition, although YAN supplementation reduced the 408 duration of fermentation in this study, a concentration of 300 mg N/L appears to be the lower 409 limit required for the maximum rate of fermentation. Brice et al. (2018) evaluated the diverse 410 nitrogen needs of S. cerevisiae by determining the specific quantity of nitrogen consumed in 411 an environment containing excess nitrogen, eliminating the nitrogen limitation/starvation 412 factor that is generally observed during the fermentation processes. The authors demonstrated 413 that the differences in the capacity of the strains to consume YAN were a result of differences 414 in their ability to uptake specific nitrogen sources. The differences between the strains tested 415 in this study to import nitrogen sources could be explained by mutations in the coding 416 sequences that modulate the activities of the permeases or differences in the expression 417 pattern of genes encoding these transporters.

418 **3.3 Importance of lipids** 

Lipids, another nutrient present in must, affect nitrogen metabolism and this topic is reviewed here for the first time. The major fatty acids found in must were variable depending on studies (Gallander and Peng, 1979; Yunoki et al., 2005). However, palmitic, stearic, arachidic, behenic, linoleic and linolenic acids are predominant. Phytosterols are also present in grape 423 berries, predominantly  $\beta$ -systerol. Although lipids (sterols and fatty acids) have been shown to 424 play a key role in maintaining membrane integrity and viability during alcoholic fermentation, 425 no link has yet been reported between lipid and nitrogen assimilation. In 2013, Tesnière et al. 426 evaluated the consequences of nutritional lipid/nitrogen imbalances on wine yeast survival 427 during alcoholic fermentation. The authors showed that yeast cell death during lipid-limited 428 fermentation (using ergosterol as lipid source) was strongly influenced by the nitrogen content 429 of the medium, with high nitrogen availability leading to high rates of cell death. Several 430 amino acids were implicated as toxic precursors: arginine, glutamate, and glutamine. These 431 observations correlated with the presence of the TOR-associated protein Tco89. High viability 432 during lipid-limited fermentation was restored when the gene encoding this protein was 433 deleted. This study highlights a relationship between the TORC1 nitrogen signaling pathway 434 and lipid limitation during alcoholic fermentation. Similarly, a study by Rollero et al. (2016) 435 highlighted the role of lipid management in nitrogen metabolism. The addition of phytosterols 436 to low-nitrogen medium influenced the consumption of various nitrogen sources. In 437 particular, the nitrogen sources were most rapidly depleted at the lowest lipid concentration, 438 especially valine, phenylalanine, and leucine. Consistent with these results, in nitrogen-rich 439 medium, phytosterol addition (8 mg/L) increased the consumption rate of nitrogen sources, in 440 particular tyrosine, tryptophan, glutamine, and ammonium. However, this was not true for all 441 tested strains. In the same study, one strain consumed nitrogen more rapidly when the must 442 contained a low concentration of phytosterols (2 mg/L). This effect may be explained by 443 changes in the plasma membrane due to the incorporation of phytosterols. Indeed, the 444 phospholipid composition of the membrane can be influenced and could affect transporter 445 activity, particularly that of the arginine permease Can1p and Gap1p (Lauwers and André, 446 2006; Malinska et al., 2003).

#### 447 **3.4 Temperature**

Temperature influences the expression of several genes involved in nitrogen metabolism, 448 449 particularly that of several transporters. For example, GAP1 is 5.2-fold down-regulated at 450 12.5°C relative to 25°C (Deed et al., 2015). However, the transcription of several other genes, 451 such as ARO9 and ARO10 (involved in catabolism of phenylalanine and methionine and 452 catabolism of multiple amino acids) is induced (Deed et al., 2015; Romagnoli et al., 2012). In 453 2006, Beltran et al. showed that S. cerevisiae grown at a low temperature (13°C) consumed 454 less nitrogen than at 25°C. Indeed, the consumption of ammonium was 79 mg/L at 25°C and 455 59 mg/L at 13°C and that of amino acids 67 mg/L at 25°C and 59 mg/L at 13°C. The 456 expression of GAP1 and MEP2 was not affected by the difference in temperature during 457 fermentation, except at the end of the fermentation at 13°C, at which point the gene 458 expression levels were higher (Beltran et al., 2006).

#### 459 3.5 Oxygen

460 Under anaerobiosis, yeast growth normally requires oxygen in order to favour the synthesis of sterols and unsaturated fatty acids. However, in such conditions, superfluous oxygen 461 462 consumption by yeast cells is observed. The superfluous oxygen consumed by the yeast cells 463 appeared not to be related to classical respiration, but mainly to the operation of several 464 mitochondrial alternative respiration pathways, which were linked to the cell cytochrome 465 contents (Salmon et al., 1998). It was calculated that approximately 124 µmol oxygen/g 466 biomass were necessary for the synthesis of unsaturated fatty acids (Verduyn et al., 1990). In 467 2000, Blateyron and Sablayrolles showed that the combined additions of oxygen and DAP 468 during the fermentation was more efficient to avoid a sluggish fermentation that a DAP 469 addition alone. Authors justified that nitrogen had an immediate effect on the kinetics of 470 fermentation by reactivating protein synthesis, particularly sugar transporters, while adding 471 oxygen mainly influenced the kinetics at the end of fermentation, by synthesizing unsaturated 472 fatty acids and sterols and thus increasing the yeast's resistance to ethanol. However, the

473 precedent section showed that lipids and nitrogen metabolisms were interconnected (Rollero 474 et al., 2016; Tesnière et al., 2013). Thus, oxygen by participating in lipid synthesis, influence 475 indirectly nitrogen assimilation by the yeast. This observation was confirmed in a recent study 476 where in low lipids condition in a must, the oxygen addition inducing the biosynthesis of 477 ergosterol and unsaturated fatty acid by yeasts promoting nitrogen assimilation (Ochando et 478 al., 2016). Therefore oxygen plays an important role in the regulation of assimilation of 479 nitrogen.

480

### **4 Management of nitrogen addition**

#### 481 **4.1 Ammonium salts and biotin**

During winemaking, ammonium sulfate, ammonium phosphate, or diammonium phosphate are most commonly added. Ammonium salts are generally supplemented with biotin. This vitamin, among others, is required by the urea carboxylase involved in the catabolism of arginine (Cooper, 1982), one of the most abundant amino acids in must (Bell and Henschke, 2005; Boulton et al., 1996; Bouzas-Cid et al., 2017; Rapp and Versini, 1995).

487 Biotin also plays a role as a cofactor for pyruvate carboxylase, which catalyzes the 488 transformation of pyruvate to oxaloacetate. Oxaloacetate is a precursor of both α-ketoglutarate 489 and aspartic acid, the key intermediates of nitrogen assimilation, and the synthesis of other 490 nitrogenous compounds (Cooper, 1982; Ljungdahl and Daignan-Fornier, 2012). Although 491 biotin has been reported to be involved in nitrogen metabolism and is largely used in industry, 492 only one study has investigated the influence of nitrogen and biotin interactions on the 493 performance of S. cerevisiae in alcoholic fermentation (Bohlscheid et al., 2007). Two strains 494 of S. cerevisiae were tested in synthetic must with different biotin (0, 1, and 10 µg/L) and 495 YAN (60 and 250 mg/L) concentrations. Both strains exhibited poor growth and very low 496 fermentation rates without biotin. An increase in nitrogen concentration resulted in higher 497 fermentation rates, whereas adjusting biotin from 1 to 10 µg/L had no effect. Based on this

498 single study, coupled biotin/nitrogen sources appear to be essential for fermentation with 499 nitrogen addition. However, only a low concentration of biotin appears to be necessary (1 500  $\mu$ g/L).

#### 501 **4.2** Consequences of nitrogen addition on fermentation parameters

502 It is generally agreed that at least 120-140 mg N/L YAN is required for satisfactory 503 fermentation kinetics and final product quality (Bely et al., 1990; Henschke and Jiranek, 504 1993). However, several studies have shown that the needs can be higher (Coleman et al., 505 2007; Martínez-Moreno et al., 2012; Mendes-Ferreira et al., 2004) and can reach 267 mg N/L 506 YAN to ferment 200 g/L glucose. Taillandier et al. (2007) showed that there is an optimal 507 requirement that varies from 0.62 to 0.91 mg N/g of sugar, depending on the strain of S. 508 cerevisiae. However, the authors found no correlation between sugar assimilation rates and 509 nitrogen requirement (Taillandier et al., 2007). Thus, in the context of nitrogen addition, the 510 yeast strain, matrix, time of addition and nitrogen sources appear to be critical elements for 511 the optimization of enological parameters. We have collected the main conclusions of the 512 studies concerning nitrogen addition in Table 3. This overview highlights the contradictory 513 nature of the results.

514 Most studies included ammonium salts as the principal supplementary YAN source. In must 515 or synthetic must, ammonium chloride addition generally decreases the fermentation time and 516 increases the fermentation rate (Jiménez-Martí et al., 2007; Martínez-Moreno et al., 2014; 517 Torrea et al., 2011; Vilanova et al., 2007). Nevertheless, this is not always true (Martínez-518 Moreno et al., 2012). No trends are observable concerning biomass production and growth 519 rate (Table 3). The addition of ammonium sulfate to must appeared to decrease the 520 fermentation time and increase the fermentation rate and biomass in all cases (Bely et al., 521 2003; Hernández-Orte et al., 2006; Mendes-Ferreira et al., 2004). Only the growth rate

522 parameter was divergent (Hernandez-Orte et al., 2006; Mendes-Ferreira et al., 2004). More 523 data are available concerning DAP addition to must. In four studies, DAP addition decreased 524 the fermentation time and increased the fermentation rate (Carrau et al., 2008; Miller et al., 525 2007; Seguinot et al., 2018; Ugliano et al., 2008). In contrast, three studies showed no effect 526 on the same parameters (Adams and van Vuuren, 2010; Hernández-Orte et al., 2004; 527 Vilanova et al., 2012). Data concerning biomass production and growth rates are surprisingly 528 limited. Only one study showed an increase in biomass after DAP addition to must (Seguinot 529 et al., 2018), whereas two others showed no significant effect (Carrau et al., 2008; Hernández-530 Orte et al., 2004). Hernàndez-Orte et al. (2004) evaluated the impact of DAP addition to must 531 on yeast growth rate. They observed an increase, but it is impossible to make a generalization 532 based on a single study. All of this data shows that ammonium salts have not the same effect. 533 This might be explained by the diversity of the matrix and S. cerevisiae strains used. Testing 534 the effects of all ammonium salts on different S. cerevisiae strains in a single conventional 535 synthetic medium could be a good approach to evaluate nitrogen addition impact.

536 Amino-acid addition to must as a source of nitrogen supplementation has become increasingly 537 common. The effect varies widely, as for ammonium salts. In general, no impact on 538 fermentation time has been reported (Garde-Cerdán and Ancín-Azpilicueta, 2008; Hernández-539 Orte et al., 2004; Martínez-Moreno et al., 2012) except for Miller et al. (2007). Neutral 540 (Garde-Cerdán and Ancín-Azpilicueta, 2008; Hernández-Orte et al., 2004; Miller et al., 2007), 541 decrease (Martínez-Moreno et al., 2012) or increase (Seguinot et al., 2018) effects on 542 fermentation rate can be observed. The results concerning the effect on biomass production 543 were also contradictory (Hernández-Orte et al., 2004; Martínez-Moreno et al., 2012; Seguinot 544 et al., 2018) as for growth rate (Hernández-Orte et al., 2004; Martínez-Moreno et al., 2012). Coupled with ammonium chloride or DAP, amino acid addition appears to decrease 545 546 fermentation time while increasing fermentation rate (Arias-Gil et al., 2007; Beltran et al.,

547 2005; Torrea et al., 2011) and probably biomass production (Beltran et al., 2005) and growth 548 rate (Torrea et al., 2011). As ammonium salts, the variability of the results might be linked to 549 the diversity of the matrix and *S. cerevisiae* strains used (Fairbairn et al., 2017).

550 The addition of nitrogen during the stationary growth phase does not have the same 551 consequences on enological parameters as its addition to must; it decreases the fermentation 552 time and increases the fermentation rate, regardless of the study and nitrogen source (Bely et 553 al., 2003; Hernández-Orte et al., 2006; Jiménez-Martí et al., 2007; Martínez-Moreno et al., 554 2014; Seguinot et al., 2018), with two exceptions for which there were no significant (Adams 555 and van Vuuren, 2010) or opposite effect (Beltran et al., 2005). Generally, nitrogen addition 556 during the stationary growth phase has no significant effect on growth rate (Hernandez-Orte et 557 al., 2006; Martínez-Moreno et al., 2014), excepted in the study of Hernandez-Orte et al. 558 (2006) in which it increased. Half of the published studies showed no significant effect on 559 biomass production (Beltran et al., 2005; Bely et al., 2003; Hernandez-Orte et al., 2006), 560 whereas the other half showed an increase, regardless of the nitrogen source (Hernandez-Orte 561 et al., 2006; Martínez-Moreno et al., 2014; Seguinot et al., 2018).

562 Overall, it appears to be very complicated to compare the effect of YAN addition on essential 563 fermentation parameters: maximum cell biomass, growth rate, fermentation time and 564 fermentation rate. Indeed, the time of nitrogen addition, sources, the use or not of NS yeasts, 565 the strain of S. cerevisiae used, the initial sugar concentration, temperature and the YAN 566 already present in the must appear to influence the fermentation profile. This has led to 567 several contradictions, depending on the experiments (Table 3). For example, Albers et al. (1996) showed that the addition of an amino-acid mixture to must resulted in greater 568 569 maximum cell biomass than the addition of ammonium sulfate. However, in 2005, Beltran et 570 al. showed that ammonium was the preferred nitrogen source for biomass production. The

timing of addition can also affect biomass production. In 2006, Hernandez-Orte et al. showed 571 572 that ammonium sulfate added at the beginning of fermentation significantly increased the 573 maximum biomass. However, the results were compared to those obtained with nitrogen 574 deficient synthetic medium. Thus, it is difficult to conclude whether the addition of an amino 575 acid mixture to the must has a better potential to increase biomass formation (Albers et al., 576 1996) than ammonium sulfate added during the first part of fermentation (Hernandez-Orte et al., 2006). In addition, Hernandez-Orte et al. (2006) reported that adding ammonium sulfate to 577 578 synthetic must always resulted in lower volatile acidity. Similar results were reported by Bely 579 et al. (2003). However, ammonium sulfate added in excess increased the production of 580 volatile acidity. Although this phenomenon is strain-dependent, it is difficult to draw 581 conclusions on the effect of ammonium sulfate addition on volatile acidity without 582 considering the amount added. The link between sugar and nitrogen metabolism has been 583 established. However, comparing recent to previous studies using different sugar 584 concentrations to compare the effect of nitrogen addition may not be informative. For 585 example, Albert et al. (1996) used 20 g/L of sugar, whereas Taillandier et al. (2007) used 240 586 g/L and Bely et al. (2003) 360 g/L. Comparing the effect of nitrogen addition between these 587 studies without considering the initial sugar concentration may be misleading. A 588 multiparametric approach appears to be essential to resolve these issues. A compilation of 589 similarly represented data would provide a first comprehensive view of the impact of nitrogen 590 addition on fermentation. This would show if there are identifiable trends or if the effect of 591 strain is predominant.

# 592 4.3 Consequences of nitrogen addition on volatile compounds produced by S. 593 cerevisiae

Table 4 provides the first overview of studies concerning the impact of nitrogen addition on
volatile compounds and glycerol produced by *S. cerevisiae*.

596 The addition of ammonium chloride (Jiménez-Martí et al., 2007; Martínez-Moreno et al., 597 2014, 2012; Torrea et al., 2011; Vilanova et al., 2007) or ammonium sulfate (Bely et al., 598 2003; Hernandez-Orte et al., 2006) before fermentation appears to decrease higher alcohol 599 production. The impact on the production of other metabolites is less clear (Hernandez-Orte et 600 al., 2006; Jiménez-Martí et al., 2007; Martínez-Moreno et al., 2014; Torrea et al., 2011; 601 Vilanova et al., 2007). The preceding paragraph underlines that there is more data concerning 602 DAP than for other ammonium salts. No significant trends have been reported concerning 603 higher alcohol production (Carrau et al., 2008; Hernández-Orte et al., 2004; Seguinot et al., 604 2018; Ugliano et al., 2008; Vilanova et al., 2012). For ester production, only Hernandez-Orte 605 et al. (2004) reported no effect, whereas other studies (Carrau et al., 2008; Miller et al., 2007; 606 Seguinot et al., 2018; Ugliano et al., 2008; Vilanova et al., 2012) reported a general increase. 607 The same trend was found for fatty acids, except in the study of Vilanova et al. (2012). 608 Concerning the effect of nitrogen addition on volatile acidity, a decrease (Vilanova et al., 609 2012) or a neutral effect were reported (Ugliano et al., 2008). For glycerol, two studies 610 showed no significant effect (Seguinot et al., 2018; Vilanova et al., 2012), whereas Ugliano et 611 al. (2008) reported an increase. DAP addition also affects hydrogen sulfide production. The 612 addition of DAP to Shiraz must fermented with S. cerevisiae led to greater hydrogen sulfide 613 formation than non-supplemented fermentation. Moreover, DAP addition induced prolonged 614 formation of hydrogen sulfide during the latter stage of fermentation, which was associated 615 with higher hydrogen sulfide content in the final wines (Ugliano et al., 2009). Other volatile 616 sulfur compounds were also present in significantly higher concentrations in the wine, 617 including sulfides, disulfides, mercaptans and mercaptoesters (Ugliano et al., 2009). Once 618 again, the strain effect seems to be involved. The study of the effect of ammonium salts on the 619 production of volatile compounds on a large panel of S. cerevisiae strains could be relevant to 620 identify possible trends.

621 The addition of amino acids before fermentation also leads to significantly different 622 outcomes. Most studies have shown a decrease in higher alcohol production (Hernández-Orte 623 et al., 2004; Seguinot et al., 2018), except one in which an increase was reported (Garde-624 Cerdán and Ancín-Azpilicueta, 2008). Similar contradictions were reported in ester 625 production (Garde-Cerdán and Ancín-Azpilicueta, 2008; Miller et al., 2007; Seguinot et al., 626 2018). All available studies reported no significant difference in fatty-acids (Garde-Cerdán 627 and Ancín-Azpilicueta, 2008; Hernández-Orte et al., 2004) or glycerol production (Martínez-628 Moreno et al., 2012; Seguinot et al., 2018). Only one study reported a decrease in volatile 629 acidity (Garde-Cerdán and Ancín-Azpilicueta, 2008). Coupled with ammonium salts, amino-630 acid addition appears to increase fatty acids production (Arias-Gil et al., 2007; Torrea et al., 631 2011) excepted for Beltran et al. (2005). For other metabolites, only Beltran et al. (2005) and 632 Torrea et al. (2011) reported an increase in ester production and volatile acidity, whereas there 633 was no significant effect on glycerol production and a decreased production of higher 634 alcohols. In addition, the addition of DAP coupled with amino acids appears to drastically 635 reduce the production of hydrogen sulfide (Barbosa et al., 2012). This study, in which the 636 authors evaluated three strains of S. cerevisiae, showed that methionine, in particular, 637 suppressed hydrogen-sulfide production under all conditions.

Less data are available concerning nitrogen addition during the stationary growth phase. A 638 639 decrease in higher alcohol production has been reported, regardless of YAN source, except for 640 amino-acid addition alone (Jiménez-Martí et al., 2007; Seguinot et al., 2018). Ammonium 641 chloride addition during the stationary growth phase appears to generally decrease ester 642 production (Jiménez-Martí et al., 2007; Martínez-Moreno et al., 2014) and increase volatile 643 acidity and glycerol production (Martínez-Moreno et al., 2014). The two studies concerning 644 ammonium sulfate addition during the stationary growth phase reported no significant effect 645 on ester production (Bely et al., 2003; Hernandez-Orte et al., 2006), whereas fatty acid

production was decreased. The results concerning volatile acidity are contradictory (Bely et 646 647 al., 2003; Hernandez-Orte et al., 2006). Only DAP appears to increase ester production, in 648 contrast to other ammonium salts. However, there is no significant effect on glycerol 649 concentration (Seguinot et al., 2018). The results obtained after amino-acid addition during 650 the stationary growth phase are also contradictory about ester concentration (Jiménez-Martí et 651 al., 2007; Seguinot et al., 2018). Finally, only Jimenez-Marti et al. (2007) evaluated the 652 impact of the addition of ammonium chloride coupled with amino acids during the stationary 653 growth phase.

654 The addition of amino acids at the beginning of the stationary phase significantly increases 655 the production of isoamyl acetate relative to the control or DAP addition (Seguinot et al., 656 2018). In the same study, the authors assessed the influence of these parameters on the 657 production of the higher alcohol propanol. Propanol is considered to be a marker of nitrogen 658 availability in must (Mouret et al., 2014). After the complete consumption of YAN, the 659 production of propanol restarted immediately upon the addition of nitrogen during the 660 stationary phase and stopped again when no more YAN remained in the medium. Propanol 661 production is strongly dependent on the nature of the added nitrogen source (inorganic or 662 organic). Mouret et al. (2014) showed that DAP addition induced greater production of 663 propanol than amino-acid addition, in particular after addition during the stationary phase. In 664 this study, final propanol production increased 4.3-fold after DAP addition during the 665 stationary phase versus a two-fold increase after the addition of amino acids. It would be 666 interesting to study the metabolic fluxes impacted by nitrogen addition in detail, especially the 667 production of propanol from threonine, in order to explain the link between production 668 differences observed and nitrogen sources.

The data concerning the impact of YAN addition on volatile compound and glycerol 669 670 production by S. cerevisiae highlight several contradictions, as for enological parameters. The 671 highly variable results cast doubt on the direct link between YAN sources and volatile 672 compounds production. This was confirmed by the recent study of Fairbairn et al. (2017) in 673 which the addition of a single amino acid resulted in the predictable production of volatile 674 compounds. The authors proved a linear correlation between amino-acid concentration and 675 the concentration of aromatic compounds directly derived from this amino acid. However, 676 linear correlations were lost and volatile compounds production became unpredictable with 677 the use of more complex nitrogen sources (mix) (Fairbairn et al., 2017). The diversity of 678 methods used, including different matrices, fermentation condition, and strains leads to the 679 lack of consensus, requiring reconsideration of how to approach the issue.

Such considerations are supported by the study of Crépin et al. (2017) who demonstrated that. YAN is mainly catabolized and involved in the *de novo* synthesis of proteinogenic compounds but not in the formation of volatile compounds. Nonetheless, YAN addition has consistently influenced volatile compounds production. Thus, these data suggest that there may be another indirect mechanism involving YAN sources and sugar metabolism for the production of volatile compounds.

#### 686 4.4 Sensory profile impact

The role of nitrogen during wine fermentation has been extensively studied, particularly the effect of nitrogen on the production of volatile compounds. Surprisingly, few studies evaluated its impact of nitrogen addition on wine sensory profiles wines. In Shiraz wines, where DAP was added to a final YAN concentration of 250 or 400 mg/L in must, the scores for the descriptors "confectionary", "red fruit", and "dark fruit" are higher for the highnitrogen fermentations. These descriptors are strongly positively associated with acetate esters 693 and medium chain fatty acids ethyl esters (Ugliano et al., 2010). Along with DAP became the 694 clear influence on perceived fruitiness, the other major effect of nitrogen addition was a 695 suppression of the perceived intensity of descriptors such as "cheese", "earth", and "yeast" 696 (Ugliano et al., 2010). In the same way, wines from Airen musts supplemented with DAP 697 become more "citric" and less sulphurous, independently of the yeast strain used (Hernández-698 Orte et al., 2004). In another study, a mix of amino acids addition (phenylalanine, aspartic 699 acid, threonine and alanine) in Merlot must affect sulphured, vegetal, fusel, floral, lactic and 700 reduction notes. The least altered notes were spices and sweet fruits (Hernández-Orte et al., 701 2006, 2004). In their study, Torrea et al. (2011) have shown that the addition of either organic 702 or inorganic nitrogen influence differently the formation of yeast volatile compounds. 703 Consequences of ammonium nitrogen addition are an increase in ethyl acetate and acetic acid 704 while a mix of amino acids and ammonium nitrogen led to a higher production of acetate and 705 medium chain fatty acids ester. From a sensorial point of view, addition of nitrogen sources to 706 reach 320 mg N/l in must led to wines rated higher in intensity for most of the fruity and floral 707 attributes. However, the type of nitrogen added has no significant effect on the sensorial 708 profile. Yet, sensory properties of wines from musts supplemented with amino acids seem to 709 depend on the yeast strain. No study describes sensory impact about nitrogen addition coupled 710 with lipids and/or vitamins.

#### 711 **4.5** Transcriptome changes in response to nitrogen addition

The expression of several genes that reflects nitrogen limitation under enological conditions has been shown to predict nitrogen deficiency, including that of *CAR1*, an arginase that catabolizes arginine to ornithine and urea (Carrasco et al., 2003); *ACA1*, which is important for carbon source utilization (Jiménez-Martí et al., 2007); *FSP2*, an alpha-glucosidase; *RGS2*, a negative regulator of glucose-induced cAMP signalling; *AQY1*, a spore-specific water channel; *AGX1*, an alanine glyoxylate aminotransferase (Backhus et al., 2001); *DAL4*, 718 allantoin permease; GAP1, general amino acid permease (Gutiérrez et al., 2013b); and ICY1, 719 which affects the consumption of ammonium during fermentation (Martínez et al., 2014). 720 However, the studies focused mainly on the impact of low nitrogen medium on the 721 transcriptional activity of S. cerevisiae. Little data is currently available concerning the impact 722 of nitrogen addition. Table 5 provides the first comparison between gene families up- and 723 downregulated after YAN addition before fermentation. Under low nitrogen conditions, the 724 addition of DAP to synthetic grape juice medium during fermentation downregulated the 725 expression of genes associated with ribosomal proteins, the small nucleolar ribonucleoprotein 726 complex, the nucleolus, and RNA processing (Mendes-Ferreira et al. 2007). In contrast, genes 727 involved in the functions of the cell wall, glucose metabolism, protein folding, energy 728 pathways, alcohol metabolism, and glycolysis were upregulated (Mendes-Ferreira et al., 729 2007). Similarly, Barbosa et al. (2015), evaluated the consequences of the transcriptomic response of various S. cerevisiae strains to DAP addition to synthetic must before 730 731 fermentation. Surprisingly, genes involved in ribosomal protein and RNA processing were 732 upregulated, which is opposite of the results of Mendes-Ferreira et al. (2007). In addition, 733 genes involved in flocculation were the most highly over-represented category in the 734 annotation of the most highly expressed genes during high-nitrogen fermentation, in 735 particular, FLO10, which participates in the fermentation stress response (FSR). Thus, 736 Barbosa et al. (2015) suggested that such overexpression is an adaptation of the yeast to stress 737 conditions and allows the yeast to conserve a good capacity of fermentation. In contrast to 738 that of Mendes-Ferreira et al. (2007), the study of Barbosa et al. (2015) demonstrated that the 739 addition of DAP before fermentation leads to the upregulation of the expression of genes 740 involved in ribosomal protein and RNA processing.

The addition of ammonium sulfate during fermentation, when yeast are in nitrogen starvation,
leads to global transcriptome changes. Tesnière et al. (2017) showed that 30 minutes after

743 ammonium sulfate replenishment, a set of 1,410 genes enriched for those related to RNA 744 processing, ribosome biogenesis, translation, and amino-acid biosynthesis was upregulated. 745 At the same time, 1,564 genes related to protein catabolic processes (ubiquitin-dependent and 746 vacuolar protein degradation), stress responses, and oxidation-reduction processes were 747 downregulated. Under similar conditions, DAP addition downregulated genes including those 748 involved in RNA processing and ribosomal protein synthesis (Mendes-Ferreira et al., 2007). 749 This contradiction was not noted by Tesnière et al. (2017). Conway et al. (2012) also 750 evaluated the impact of the addition of ammonium sulfate coupled with amino acids. In this 751 study, they obtained contradictory results compared to Mendes-Ferreira et al., (2007) (Table 752 5). Thus, it appears that the transcriptional activity related to RNA processing is affected 753 depending on the ammonium salts. However, it is difficult to compare the two conditions as 754 the strains and medium and fermentation parameters were different.

755 As already discussed, amino acids are another potential supplementary nitrogen source. High 756 nitrogen concentrations, including that from exogenous amino-acids and ammonium, can 757 unbalance the general composition of the must, particularly under conditions of high nitrogen 758 content and low micronutrients. In 2017, Duc et al. showed that, among the seven 759 micronutrients evaluated in this study (ergosterol, oleic acid, pantothenic acid, nicotinic acid, 760 thiamin, biotin, and inositol), four trigger yeast cell death when the yeast are in growth-761 restricting amounts in a nitrogen-rich medium. Two (ergosterol, oleic acid) are lipid growth 762 factors, conditional to anaerobic growth, and the other two vitamins (pantothenic acid and 763 nicotinic acid). They showed that the *TORC1/SCH9* signaling pathway was a key complex in 764 this effect. Indeed, the authors observed that decreasing TORC1/SCH9 signaling restored 765 high cell viability under several micronutrient-limited fermentation conditions. Viability was 766 also restored by SCH9 deletion under conditions of oleic acid, ergosterol, and pantothenic 767 acid starvation. Global gene expression experiments showed that two clusters distinguish the

conditions that preserve viability from those that lead to a high rate of cell death. The first included genes involved in mitochondrial respiration, electron transport processes, the TCA cycle, membrane transport, and detoxification. The second included genes involved in nitrogen catabolism, purine metabolism, and membrane transport, including that of nitrogen substrates. Significant differences in gene expression were specifically observed in these clusters under conditions of high nitrogen content and limited-micronutrients (Duc et al., 2017).

775 The addition of amino acids alone as a nitrogen source to must also affects global gene 776 expression (Liu et al., 2017). L-valine, L-leucine, and L-isoleucine added in a non-limiting 777 nitrogen medium resulted in the up- or downregulation of 25 genes during fermentation. 778 Some genes were upregulated throughout fermentation, including: ACT1, a structural protein 779 involved in cell polarization; PDA1, which catalyzes the direct oxidative decarboxylation of 780 pyruvate to acetyl-CoA; SER1, required for serine and glycine biosynthesis; FOX2, a 781 multifunctional enzyme of the peroxisomal fatty acid beta-oxidation pathway; MEP2, 782 ammonium permease; BAP3, which is involved in the uptake of cysteine, leucine, isoleucine, 783 and valine; THR4, a conserved protein that catalyzes the formation of threonine from O-784 phosphohomoserin; AGX1, which catalyzes the synthesis of glycine from glyoxylate; STL1, a 785 glycerol proton symporter of the plasma membrane; and GAP1, general amino acid 786 permease), whereas others genes were upregulated only during the exponential growth phase, 787 including GIP1, a meiosis-specific regulatory subunit of the Glc7p protein phosphatase; DIT2, 788 N-formyltyrosine oxidase; *RCK1*, a protein kinase involved in the oxidative stress response; 789 SPO1 a meiosis-specific prospore protein; and MEK1, meiosis-specific serine/threonine 790 protein kinase. The addition of this mix of amino acids also up-regulated the expression of 791 SUT1, which positively regulates sterol uptake genes under anaerobic conditions; RRI2, a 792 subunit of the COP9 signalosome complex; and CIN5, a basic leucine zipper transcription factor of the yAP-1 related to the environmental stress response. Surprisingly, the expression of few genes related to the biosynthesis of volatile compounds was also affected. Among 412 genes, only the expression of *AAD3, AAD4, AAD14, ADH2*, and *PDC5*, involved in the biosynthesis of aroma substances, was significantly up-regulated (Liu et al., 2017).

797

#### 5 Concluding remarks

798 From grape to wine, the role of nitrogen in winemaking is complex. This review focuses on 799 fermentation and highlights new considerations concerning the impact of nitrogen on yeast 800 metabolism and how biotic, abiotic, and exogenous addition can influence enological 801 parameters. Here, the review of recent studies showed that YAN preferentially consumed by 802 NS yeast strains during the first stage of fermentation vary and other factors, such as 803 competition can be involved. There are no studies that explain why some YAN sources are preferentially consumed by NS yeasts. Moreover, the effect on the expression of genes 804 805 involved in nitrogen regulation of S. cerevisiae in the presence of NS yeasts is still 806 unexplored. The vast majority of studies have referred to "preferred" or "non-preferred" YAN 807 sources, but a close examination in this review shows that this categorization varies widely, 808 even if the same parameters of comparison were used. An exploration of genetic variation 809 appears to be necessary to obtain a better understanding of the preferential assimilation of 810 YAN. Much genetic data are available for S. cerevisiae. In contrast, current genetic data for 811 NS yeast are too limited to develop significant research on nitrogen regulation. Enological 812 nitrogen addition using amino acids is being increasingly performed, but the impact on S. 813 *cerevisiae* in terms of fermentation parameters and volatile compound production is not clear. 814 Indeed, recent studies show that only 5% of amino acids present in the must are directly 815 involved in the synthesis of volatile compounds. Moreover, transcriptomic analysis has shown 816 that the addition of nitrogen does not lead to the significant overexpression of genes involved 817 in volatile compound production. Thus, the significant modification of the volatile compound 818 profiles in wine caused by the addition of amino acids needs to be further investigated.
819 Deciphering the link between nitrogen metabolism and volatile compound synthesis will
820 provide useful information to monitor the fermentative aroma profile of wine.

821 Future studies that will lead to a better comprehension of YAN dynamics, from grape to wine, 822 interconnected with other pathways (vitamins and lipids), will help to explain the production 823 of volatile compounds and allow better control of fermentation by avoiding sluggish 824 fermentation, providing new tools to winemakers. Most of the studies examining the link 825 between nitrogen and volatile compound production were carried out in synthetic must. Given 826 the complexity of the composition of must it will be necessary to include real must in future 827 studies aiming to understand the influence of nitrogen on fermentation kinetics and aroma 828 production.

#### 829 6 Acknowledgments

This work is supported by the Conseil Régional de Bourgogne through the plan d'actions
régional pour l'innovation (PARI) and the European Union through the PO FEDER-FSE
Bourgogne 2014/2020 programs.



#### 835 7 Figures



Figure 1. (A) The non-induced resting state in the absence of inducing amino acids. In the absence of inducing amino acids, low levels of AAPs are present in the plasma membrane (PM). NM, nuclear membrane. (B) The induced state in the presence of extracellular amino acids. The derepressed AAP gene expression leads to increased levels of AAP in the PM and enhanced rates of amino acid uptake (adapted from Ljungdahl et al. (2009)).



843 Figure 2. Translocation into the nucleus of Gat1p/Gln3p complex depending on nitrogen 844 status. In nitrogen repletion condition, Gat1p and Gln3p are sequestered by Ure2p in the 845 cytoplasm. Dal80p and/or Gzf3p block the UAS<sub>NTR</sub> binding site. In nitrogen starvation 846 condition, Ur2p releasesUre2p release Gat1p and Gln3p and the complex migrates into the nucleus. Dal80p/Uga43p and/or Gzf3p release the UAS<sub>NTR</sub> binding site and Gat1p/Gln3p 847 848 complex activates transcription of NCR the controlled genes.

#### 850 8 Tables

851 **Table 1.** Nitrogen compounds considered to be "preferred" nitrogen sources in *S. cerevisiae*.

Amino acids	References
Alanine	8
Ammonium	1, 2, 3, 6, 7, 8, 11, 13
Arginine	2, 3, 8
Asparagine	1, 3, 4, 5, 6, 7, 8, 9, 11, 12, 13
Aspartate	3, 8
Aspartic acid	4, 12
GABA	8
Glutamine	1, 3, 5, 6, 8, 11, 12, 13
Glutamate	8, 11, 13
Glutamic acid	4
Histidine	2
Isoleucine	3, 12
Leucine	2, 3, 12
Lysine	3, 9, 12
Methionine	3
Serine	2, 3, 9, 10, 12
Threonine	2, 3, 9, 10, 12

852 1: (Grenson, 1992), 2: (Henschke and Jiranek, 1993), 3: (Jiranek et al., 1995), 4: (Albers et al.,

- 853 1996), 5: (Magasanik and Kaiser, 2002), 6: (Görgens et al., 2005), 7: (Boer et al., 2006), 8:
- 854 (Ljungdahl and Daignan-Fornier, 2012), 9: (Crépin et al., 2012), 10: (Pinu et al., 2014), 11:
- 855 (Tesnière et al., 2015), 12: (Gobert et al., 2017), 13: (Brice et al., 2018).

Amino acids	NS yeasts	Condition of fermentation	Ref.
		Grape juice 20°C	2, 3
Alanine	Hu, Mp, Sb, Td	Grape juice 28°C	
	<u> </u>	Synthetic medium 25°C	
Ammonium	Mp, Td	Synthetic medium 25°C	2
Arginine	Lt, Hu, Td	Synthetic medium 25°C	2
Acoracio	Lt Dres Tol	Grape juice 28°C	2, 3
Asparagine	Li, Fill, Tu	Synthetic medium 25°C	
Aspartic acid	Hu, Lt	Synthetic medium 25°C	2
Cysteine	Hu	Must 20°C stirring	1
Clutaria agid	Un It Mp	Synthetic medium 25°C	1, 2
		Must 20°C stirring	
		Grape juice 20°C	1, 2, 3
Clutomino	Un It Mr. Td	Grape juice 28°C	
Olutaliille	IIu, Li, Mp, Iu	Synthetic medium 25°C	
		Must 20°C stirring	
Clusing	Sh Dm	Grape juice 20°C	1, 3
Orychie	50, F III	Must 20°C stirring	
Histidine	Mp, Pm	Grape juice 20°C	3
Isolaucina	It Mp Td	Grape juice 28°C	2, 3
	Li, Mp, Tu	Synthetic medium 25°C	
		Grape juice 28°C	1, 2, 3
Leucine	Hu, Lt, Mp	Synthetic medium 25°C	
		Must 20°C stirring	
Iveine	Mn Sh Pm	Grape juice 20°C	3
	1 <b>v</b> 1p, 50, F III	Grape juice 28°C	
Methionine	Hu Mp	Grape juice 28°C	1,3
	IIu, wip	Must 20°C stirring	
Phenylalanine	Hu, Lt	Synthetic medium 25°C	2
Serine	Lt, Pm	Grape juice 28°C	3, 2
Threonine	Mp, Pm	Grape juice 20°C	3,
Tryptophan	Sb	Must 20°C stirring	1
Tyrosine	Lt	Synthetic medium 25°C	2
Valine	Hu	Synthetic medium 25°C	2

858 Table 2. Nitrogen compounds considered to be "preferred" nitrogen sources in NS yeasts
859 depending on the conditions of fermentation.

860 Abbreviations of NS yeasts: Hu: H. uvarum, Lt: L. thermotolerance, Mp: M. pulcherrima, Sb:

861 S. bacillaris, Pm: P. membranifaciens and Td: T. delbrueckii. References 1: (Andorrà et al.,

862 2010), 2: (Kemsawasd et al., 2015), 3: (Gobert et al., 2017)

863

Time of addition	N Source	Biomass	Growth rate	Fermentation time	Fermentation rate	Ref.
	AC	7	7	И	7	7, 14
	AC	nd	nd	Ы	7	9
	AC	Я	И	<b>→</b>	И	15
	AC	<b>→</b>	<b>→</b>	Ы	7	17
	AS	7	nd	И	7	1
	AS	7	<b>→</b>	Ľ	7	3
	AS	7	7	Ľ	7	5
	DAP	<b>→</b>	7	<b>→</b>	<b>→</b>	2
	DAP	nd	nd	Ľ	7	8, 12
ust	DAP	<b>→</b>	nd	Ľ	7	10
Μ	DAP	nd	nd	<b>→</b>	<b>→</b>	13, 16
	DAP	7	nd	Ľ	7	18
	AA	<b>→</b>	7	<b>→</b>	<b>→</b>	2
	AA	nd	nd	Ľ	<b>→</b>	8
	AA	nd	nd	<b>→</b>	<b>→</b>	11
	AA	Ľ	Ľ	<b>→</b>	Ы	15
	AA	7	nd	Ľ	7	17
	AC + AA	7	7	Ľ	7	14
	AC + AA	7	nd	Ľ	nd	4
	DAP + AA	nd	nd	Ч	7	6
	AC	<b>→</b>	<b>→</b>	Ľ	7	7
ase	AC	7	<b>→</b>	Ľ	7	17
ı phâ	AS	<b>→</b>	nd	И	7	1
owth	AS	7	7	Ľ	7	3, 5
y gr(	DAP	nd	nd	<b>→</b>	<b>→</b>	13
nary	DAP	7	nd	Ľ	7	18
tatio	AA	<b>→</b>	<b>→</b>	Ы	7	7
S	AA	7	nd	И	7	18
	AC + AA	<b>→</b>	<b>→</b>	Ы	7	7
	AC + AA	<b>→</b>	nd	7	nd	4

Table 3. Comparison between the impact of the timing of addition and nitrogen source on
 fermentation.

867 1: Bely et al. (2003), 2: Hernandez-Orte et al. (2004), 3: Mendes-Ferreira et al. (2004), 4: Beltran et al. (2005), 5: Hernandez-Orte et al. (2006), 6: Arias-Gil et al. (2007), 7: Jiménez-868 869 Marti et al. (2007), 8: Miller et al. (2007), 9: Vilanova et al. (2007), 10: Carrau et al. (2008), 870 11: Garde-Cerdan et al. (2008), 12: Ugliano et al. (2008), 13: Adams et al. (2010) 14: Torrea 871 et al. (2011), 15: Martinez-Moreno et al. (2012), 16: Vilanova et al. (2012), 17: Martinez-872 Moreno et al. (2014), 18: Seguinot et al. (2018). nd: no data. Abbreviations: N, nitrogen; AC, 873 ammonium chloride; AS, ammonium sulfate; DAP, diammonium phosphate; AA, amino 874 acids. Upward arrows  $(\mathbf{z})$  indicate an increase of the corresponding parameter relative to 875 control, neutral arrows  $(\rightarrow)$  no significant effect, and downward arrows  $(\triangleleft)$  a decrease. 876 Several studies that concern the impact of nitrogen addition during fermentation are 877 intentionally not mentioned in the table (Albers et al., 1996; Barbosa et al., 2012, 2009; 878 Manginot et al., 1998). In these studies, the impact of increasing the nitrogen concentration 879 was evaluated, but no data for controls without nitrogen addition were available. The 880 experimental conditions in reference to the studies (matrix, initial sugar concentration, initial 881 nitrogen concentration and nature) are described in the table 6.

Time of addition	N Source	Higher alcohols	Esters	Fatty acids	Volatile acidity	Glycerol	Ref.
	AC	R	Я	nd	nd	nd	6
	AC	N	7	→	<b>→</b>	→	8
	AC	R	7	7	7	→	12
	AC	nd	nd	nd	nd	→	13
	AC	nd	Ы	nd	7	7	15
	AS	nd	nd	nd	Я	<b>→</b>	1
	AS	R	<b>→</b>	7	Ŕ	nd	4
	DAP	Ы	<b>→</b>	7	nd	nd	2
	DAP	nd	7	nd	nd	nd	7
t	DAP	R	7	7	nd	nd	9
Mus	DAP	<b>→</b>	7	7	R	7	11
<b>F</b>	DAP	Ы	7	<b>→</b>	<b>→</b>	<b>→</b>	14
	DAP	7	7	nd	nd	<b>→</b>	16
	AA	Ы	<b>→</b>	<b>→</b>	nd	nd	2
	AA	nd	Ы	nd	nd	nd	7
	AA	7	7	<b>→</b>	R	nd	10
	AA	nd	nd	nd	nd	→	13
	AA	Ы	7	nd	nd	→	16
	AC + AA	Ы	7	7	7	<b>→</b>	12
	AC + AA	Ы	7	<b>→</b>	nd	→	3
	DAP + AA	nd	nd	7	nd	nd	5
	AC	Ы	Ы	nd	nd	nd	6
vth	AC	nd	Ľ	nd	7	7	15
rov	AS	nd	nd	nd	7	<b>→</b>	1
.y g ase	AS	Ы	→	Я	Ľ	nd	4
nar ph	DAP	Ы	7	nd	nd	<b>→</b>	16
atio	AA	7	Ľ	nd	nd	nd	6
St	AA	<b>→</b>	7	nd	nd	<b>→</b>	16
	AC + AA	Ы	Ы	nd	nd	nd	6
	AC + AA	7	R	<b>→</b>	nd	R	3

Table 4. Impact of the addition of YAN sources on volatile compounds and glycerol
produced by *S. cerevisiae*.

885 1: Bely et al. (2003), 2: Hernandez-Orte et al. (2004), 3: Beltran et al. (2005), 4: Hernandez886 Orte et al. (2006), 5: Arias-Gil et al. (2007), 6: Jiménez-Marti et al. (2007), 7: Miller et al.

887 (2007), 8: Vilanova et al. (2007) 9: Carrau et al. (2008), 10: Garde-Cerdan et al. (2008), 11: 888 Ugliano et al. (2008), 12: Torrea et al. (2011), 13: Martinez-Moreno et al. (2012), 14: 889 Vilanova et al. (2012), 15: Martinez-Moreno et al. (2014), 16: Seguinot et al. (2018), nd: no 890 data. Abbreviations: N, nitrogen; AC, ammonium chloride; AS, ammonium sulfate; DAP, diammonium phosphate; AA, amino acids. Upward arrows (7) indicate an increase in the 891 892 concentration of the corresponding metabolite relative to control, neutral arrows  $(\rightarrow)$  no 893 significant effect, and downward arrows (**>**) a decrease. Several studies that concern the 894 impact of nitrogen addition during fermentation are intentionally not mentioned in the table 895 (Albers et al., 1996; Barbosa et al., 2012, 2009; Manginot et al., 1998). In these studies, the 896 impact of increasing the nitrogen concentration was evaluated, but no data for controls 897 without nitrogen addition were available. The experimental conditions in reference to the studies (matrix, initial sugar concentration, initial nitrogen concentration and nature) are 898 899 described in the table 6.

# **Table 5.** Influence of YAN status on the transcriptional activity of gene families in *S*.

*cerevisiae* during alcoholic fermentation.

N Source & condition	Gene family upregulated	Gene family downregulated	
AS Supplemented versus non- supplemented	RNA processing Ribosome biogenesis Amino-acid biosynthesis	Protein catabolic processes Stress response Oxidation-reduction processes	4
DAP High versus low level	Cell wall functions Glucose metabolism Protein folding Energy pathways Alcohol metabolism Glycolysis	Ribosomal proteins Small nucleolar ribonucleoprotein complex Nucleolus and RNA processing	1
DAP High versus low level	Flocculation Ribosomal proteins RNA processing Autophagy	nd nd nd nd	2
AA Supplemented versus non- supplemented	Biosynthesis of aroma compounds Yeast growth Environmental stress response	nd nd nd	3

903 1: Mendes-Ferreira et al. (2007), 3: Barbosa et al. (2015), 3: Liu et al. (2017), 4: Tesnière et

904 al. (2017), nd: no data. Abbreviations: AS, ammonium sulfate; DAP, diammonium phosphate;

905 AA, amino acids. The experimental conditions in reference to the studies (matrix, initial sugar

906 concentration, initial nitrogen concentration and nature) are described in the table 6.

References	Table 3	Table 4	Table 5	Matrix	Initial sugar concentration (g/L)	Initial nitrogen concentration (mg/L)	Nature of nitrogen source
Bely et al. (2003)	1	1		Must	320 to 370	92	Naturally present in must
Hernandez-Orte et al. (2004)	2	2		Must	nd	175	Naturally present in must
Mendes-Ferreira et al. (2004)	3			GJM	200	66	AS
Beltran et al. (2005)	4	3		SM	200	60 or 300	AC + AA
Hernandez-Orte et al. (2006)	5	4		SM	210	270	AS + AA
Arias-Gil et al. (2007)	6	5		Must	nd	285	Naturally present in must + DAP
Jiménez-Marti et al. (2007)	7	6		SM	200	60 to 300	AS + AA
Mendes-Ferreira et al. (2007)			1	GJM	200	66 to 267	DAP
Miller et al. (2007)	8	7		Must	213	2095	Naturally present in must
Vilanova et al. (2007)	9	8		CDGJ	225	117	AC + AA
Carrau et al. (2008)	10	9		CDGJ	120	75 to 400	DAP + AA
Garde-Cerdan et al. (2008)	11	10		Musts	nd	285 to 1173	Naturally present in must
Ugliano et al. (2008)	12	11		Musts	215	87 to 312	Naturally present in must
Adams et al. (2010)	13			Must	nd	361	Naturally present in must
Torrea et al. (2011)	14	12		Must	200	160	Naturally present in must
Martinez-Moreno et al. (2012)	15	13		SM	200 to 280	0 to 220	AC or AA or AC + AA
Vilanova et al. (2012)	16	14		Must	nd	250	Naturally present in must
Martinez-Moreno et al. (2014)	17	15		SM	240	100 to 200	AC + AA
Barbosa et al. (2015)			3	GJM	200	67 to 670	DAP
Liu et al. (2017)			4	SM	200	300	AC+AA
Tesnière et al. (2017)			5	YNB	20	0	Ammonium (source unspecified)
Seguinot et al. (2018)	18	16		SM	200	100	AC + AA

**Table 6.** Experimental conditions used in the tables 3, 4 and 5.

911

Abbreviations: GJM, grape must modified; SM, synthetic must; CDGJ, chemically defined grape juice; YNB, yeast nitrogen base.

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