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Mechanical properties of milimetric silica aerogel particles produced through evaporative drying: a coupled experimental and discrete element approach

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Abstract

An alternative to supercritical drying to process silica aerogels is to apply evaporative drying to obtain particles that can then be used in the preparation of composite products. However, the mechanical behaviour of these particles is still poorly understood. In this work, millimetre-sized particles are studied combining experimental testing and discrete simulations. A statistical study of the compression of individual particle shows that the values of fracture strength are scattered because of the variability in shape, density and initial defects. The individual compression of aerogel particles inside an X-ray tomograph provides valuable information for understanding the influence of density on their elastic and fracture behaviors. Discrete simulations are calibrated using experimental data to reproduce numerically experimental compression observed inside the tomograph allowing the introduction of process induced particle defects. We show that these simulations can be used to determine the Young's modulus and fracture strength of aerogel particles.

Keywords: Silica aerogel, Particles, Uniaxial compression, Discrete Element Method, Fracture, X-ray tomography, Crack propagation

1 1. Introduction

In recent years, silica aerogels have attracted the interest of the thermal insulation industry. 2 These materials indeed offer exceptional thermal performances with thermal conductivities as 3 low as 12 mW.m⁻¹.K⁻¹, coupled with transparency, lightness and soundproofing properties, ow-4 ing to their highly porous nanoscale pearl-necklace network of silica particles. The porosity level 5 is generally over 90% with pores of approximately 10 nm [1]. These materials would be already 6 largely used and produced were it not for their poor mechanical properties and high manufacturing cost, which makes their use very situational for now [2, 3, 4]. The evolution of regulations 8 in the field of thermal insulation over the last years [5] seems to be a perfect context to promote 9 new insulation materials offering enhanced properties for a reduced dedicated volume compared 10 to conventional solutions in today's building, transportation and industry sectors. 11

Silica aerogels are produced by a sol-gel process followed by drying of the solvent used for the gel preparation [6, 7, 8]. The vast majority of studies dealing with silica aerogel mechanical properties concerns monolithic aerogels produced through supercritical drying, which is the privileged way to process high quality silica aerogel monoliths [9, 10, 11]. Very few studies focus

on aerogels produced through evaporative drying [12] and most of them deals with silica aerogel 16 composites reinforced with a second phase during the gel formation [13, 14]. Evaporative dry-17 ing, however, is a potential alternative for manufacturers looking for a lower production cost, as 18 supercritical CO_2 is costly and creates additional constraints in the production process. One of 19 the main drawbacks to aerogel production by evaporative drying is the occurrence of capillary 20 forces as the liquid-vapor interface shape changes with time. These capillary forces cause seri-21 ous problems for the production of large aerogel monoliths as they lead to the structure breakage, 22 which results in particulate aerogels. Presently, the extremely high production cost of bulk silica 23 24 aerogel panels strongly impedes their use in the building industry in this form. However, the use of aerogels in particle form is possible. Some of the typical insulation systems using aerogel 25 particles are: 26

• window panels (raw aerogel particles sandwiched between two transparent panels) [15, 16]

stiff composite panels (mixture of binder and aerogel particles) shaped in any geometry
 [17, 18]

• paintwork, construction sealants and plaster including aerogel particles [19, 20, 21]

30

In addition, it is possible to take advantage of aerogel particles obtained by evaporative dry-31 ing for 3D printing of miniaturized insulators [22]. In all these insulation systems, micrometric 32 and milimetric silica aerogel particles are used and the presence of over-nanometric porosity is 33 not desirable (thermal conductivity of still air is greater than that of aerogel). The mechanical 34 response of products such as stiff composite panels is mainly dependent on the mechanical be-35 haviour of aerogel particles. A proper study of these properties would be very beneficial for the 36 understanding of the failure mechanisms of the composite and the improvement of the current 37 trade-off between strength, thermal performance and cost. Existing works on aerogel monoliths 38 produced through supercritical drying have shown that the mechanical properties of silica aero-39 gels are highly dependent on density [10, 9]. However, manufacturers can only partially rely on 40 41 these studies to design composites, as it is suspected that evaporatively dried and supercritically dried silica aerogels exhibit structural properties that differ from one process to another. Indeed, 42 high stresses induced during evaporative drying can generate defects such as cracks inside par-43 ticles and density variations between particles. These are not (or to a lesser extent) observed in 44 monolithic aerogels produced through supercritical drying. 45

To our best knowledge there is neither experimental nor numerical study available dealing 46 47 with the mechanical behavior of aerogel particles produced through evaporative drying. The purpose of this work is thus to fill this knowledge gap through mechanical characterization of aerogel 48 particles supplemented by modeling. In particular it is sought to understand and quantify the in-49 fluence of pre-existing cracks and density variations on stiffness and strength. The millimetric 50 size and irregular geometry of particles make it impossible to use conventional mechanical tests 51 such as uniaxial compressions on cylindric samples, bending test on bars or brazilian tests. The 52 characterization procedure chosen for this work is uniaxial compression of the aerogel particles 53 themselves. This test is well adapted to the aerogel particles size and shape. More than fifty 54 aerogel particles were compressed in this work, providing enough data for statistical analysis 55 of particles strength. The mechanical tests were paired with density measurements and X-ray 56 57 tomography in order to link the measured properties to the possible density variation and the presence of process induced cracks. The Discrete Element Method (DEM), particularly well-58 suited to simulate damage and crack propagation [23, 24, 25], has already been applied to silica 59

aerogels [26, 27]. Here, it is used to simulate particle compression and contribute to the analysis
 of experimental results.

The article is organized as follows. In section 2, the synthesis process and main properties of the studied silica aerogel are introduced. In section 3, experimental setups are presented. Section 4 presents the DEM model and the numerical procedures to generate aerogel particles made of

65 discrete elements and simulate their compression. Results from both mechanical testing and

⁶⁶ DEM simulation are presented in section 5 and discussed in section 6.

67 2. Silica aerogel particles: synthesis and structure

68 2.1. Synthesis

Hydrophobic aerogels tested in this study were produced by Enersens company (Bourgoin-69 Jallieu, France) using an alkoxyde as a precursor for the sol-gel process: tetraethylorthosilicate 70 (TEOS). The process uses ethanol as solvent. Once the gel is obtained an hydrophobic treat-71 ment adds trimethylsilyl groups on the free surface of the gel. This treatment partially counters 72 the shrinkage of the gel during the following evaporative drying step [28, 29]. The aerogel hy-73 drophobicity is also critical in its end use as a building material for durability reasons, as shown 74 by [30]. Evaporative drying of the gel consists in increasing the temperature above the boiling 75 point of ethanol. As a result of the evaporative drying and the capillary forces induced by the 76 change of curvature of ethanol meniscus [31], the gel block densifies and breaks. The resulting 77 aerogel has the shape of small translucent faceted particles of maximum size 5 mm (see Fig.1). 78 The aerogel particles used for this study were sieved so that their dimension is between 1.00 and 79 1.25 mm. The final size range of the studied particles is typical of the largest particles used in 80 stiff composite panels with bimodal particle size distribution [18]. 81

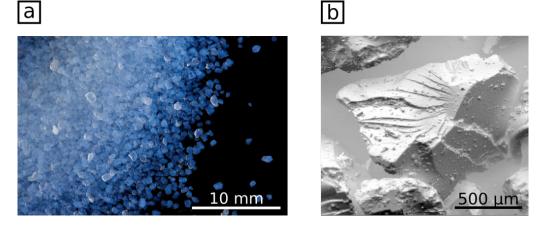


Figure 1: a) Silica aerogel particles produced by Enersens [32]. b) 1 mm silica aerogel particle observed with SEM, topographic contrast (no metallization). The aerogel particle is sprinkled with micrometric particles that are not an original feature of the 1 mm particles.

⁸² 2.2. Structure and general precautions

The silica aerogel studied here has a nanoporous structure with a mean pore size of 10 nm

[1]. The 4 nm silica primary particles are grouped in secondary patterns (agglomerates of primary

silica particles) of approximately 10 nm forming the three dimensional network responsible for
the unique thermal and mechanical properties of the material. The porosity announced by the
manufacturer is 90-98% [33]. All aerogel particles produced following the previous protocol
have proven to be extremely brittle and can potentially break by simple friction while handling.
For this reason, all aerogel particles for mechanical testing were handled by hand only with
minimal transfer and immediately replaced in case of noticeable damage prior to testing. Tests

⁹¹ were carried out at ambient temperature (20 to 23°C) and RH (40 to 60%).

92 **3. Experimental setups**

93 3.1. Density measurement

Density of aerogel particles is critical as it has been shown that elastic and fracture behaviours depend strongly on density [10, 9, 34, 35]. The density of the studied aerogel particles was measured using mercury picnometry on large particle batches (several hundreds of particles). Four particle batches were weighed and analyzed using a Micromeretics Autopore 4 with an increasing pressure from 0.001 to 0.206 MPa. The sample volume is measured at 0.036 MPa. Moreover, individual particle density was obtained by X-ray tomography volume measurements (see section 3.3).

¹⁰¹ 3.2. Particle compression: statistical procedure

Two types of uniaxial compression tests have been carried out on aerogel particles. The first procedure is an experimental study of fracture inducing load under compression. More than 50 individual randomly selected particles were pressed up to failure. This procedure provides information on the scatter of the fracture force. The setup for these compression tests is shown in Fig.2a and was used at two different loading velocities (0.3 and 1.5 mm.mn⁻¹) [9]. The load cell used had a nominal force of 22 N and a sensitivity of 0.01 N. The second procedure is presented hereafter.

¹⁰⁹ 3.3. Particle compression with X-ray tomography observation

Compression of individual aerogel particles within a X-ray tomograph was carried out. The 110 compression device illustrated in Fig.2c was designed to be placed inside a X-ray tomograph 111 (RX-Solution, Annecy, France). Particle compression is performed right after a first tomography 112 scan at 40 kV, 250 μ A. The density of each particle has been calculated before their compres-113 sion from the volume determined from X-ray tomography and the mass measured using a XP26 114 Mettler Toledo microscale. Once a particle has been weighted, it only endures one transfer to 115 the X-ray tomograph for initial scanning and is ready for mechanical testing with no additional 116 manipulation. X-ray tomography is a convenient technique for monitoring aerogel fracture but 117 some cautions are needed as these materials show a low absorption due to their extremely high 118 porosity. Any material too dense compared to silica aerogels placed between the particle and the 119 X-ray source or the CCD camera would absorb too many photons. It could induce a decreasing of 120 the contrast of the scans or create artefacts. For this reason the window and the compression tools 121 of the setup are made of polycarbonate. Low X-ray adsorption of polycarbonate and optimised 122 thickness of material prevent artifacts. Particle compression was carried out at 1.2 mm.mn⁻¹ and 123 stopped at the first sign of fracture (detected by visual monitoring of 2D radiography). Due to 124 some constraints of the setup, the force could not be recorded while the particle was compressed. 125 The compression was paused several times (for a maximum of 20 seconds) once the particle had 126

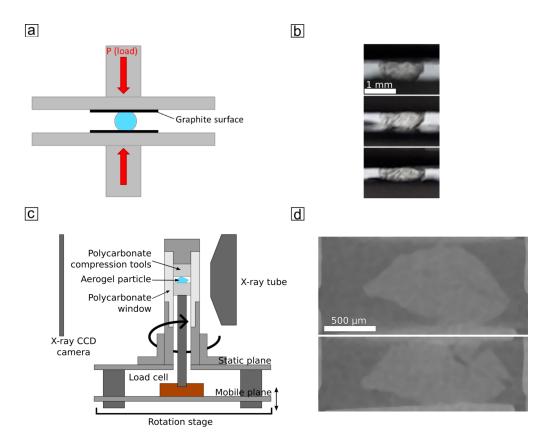


Figure 2: Compression setup for individual particle compression. a) Standard compression setup. b) Images of particles deformation while compressive load increase. c) Compression setup for X-ray tomograph compression. d) Tomographic cross section of a particle during fracture.

been compressed more than 30% of its initial height to record the force. This procedure eased 127 the determination of the particle fracture force F_f by detecting with more accuracy the fracture 128 of the particle. In that case, fracture force F_f and fracture height h_f were estimated as the mean 129 value of two last recorded points with accordingly adjusted error bars. Fracture force and frac-130 ture height were determined for each particle. At least two scans were acquired with a voxel 131 size of 2 μ m, one just before the beginning of the compression and one once crack propagation 132 has been detected in the aerogel particle. Particles were not unloaded for the final scan in order 133 to maintain their integrity. Image processing of the tomography scans allowed the extraction of 134 particles shape by applying a 3D median filter before segmentation of the aerogel phase by sim-135 ple thresholding on graylevels (Fiji software [36]). Process induced cracks within particles were 136 manually segmented as no automated method has given acceptable results for these features. 137

4. Discrete Element Method model

¹³⁹ DEM is particularly well suited to study fracture and crack propagation of a large spec-¹⁴⁰ trum of materials. It was originally designed for the study of granular materials [37] but is also

adapted for continuous materials [23, 38, 39, 40, 24, 41]. Using DEM to study low-stiffness 141 and brittle materials like silica aerogels was successfully carried out in several other studies 142 [26, 27] with discrete elements representing primary silica particles, i.e. the silica particles form-143 ing the nanoscale pearl-necklace network of aerogels. Another simulation method, the hybrid 144 finite-discrete element method (FDEM), is also well adapted to the study of continuous materials 145 fracture as it combines the advantages of the finite element method for the study of continuum 146 mechanical problems and the advantages of the discrete element method for the study of fracture 147 phenomena [42, 43, 44]. FDEM has yet to be tested for aerogel particles fracture and this work 148 will only focus on the use of DEM. 149

150 4.1. DEM framework

DEM simulations are performed with the in-house code dp3D using spherical discrete ele-151 ments. Although silica aerogels are composed at the nanometer scale of discrete entities, their 152 explicit representation within DEM would be computationally too demanding to model volumes 153 as large as a few μm^3 [26]. DEM can nonetheless be used to model efficiently damage and 154 fracture of a continuum media. Within this framework, the material is built as a random pack-155 ing of spheres [41, 24]. Spheres do not represent real particles but are used to mesh a structure 156 while providing a fully discontinuous framework handy for treating fractures. The interactions 157 between spheres are predominantly defined by bonds that transmit normal and tangential forces 158 and resisting moments. For bonds that have broken, if contact is restored, only normal forces in 159 compression are transmitted. The forces acting on bonds have been described by Kumar et al. 160 [41]. An equivalent radius at the bond between two spheres of radii R_p and R_q is defined as: 161

$$R^* = \frac{R_p R_q}{R_p + R_q} \tag{1}$$

¹⁶² The normal and tangential forces acting on the bond are given by simple linear elastic laws:

$$\mathbf{N} = -K_N \delta_N \mathbf{n} = -2\Sigma_N R^* \delta_N \mathbf{n} \tag{2}$$

$$\mathbf{T} = -K_T \delta_T \mathbf{t} = -2\Sigma_T R^* \delta_T \mathbf{t}$$
(3)

where δ_N and δ_T are the normal and tangential relative displacements between the two particle centers, K_N and K_T are local normal and tangential stiffnesses and **n** and **t** are the unit vectors normal and parallel to the contact plane, respectively. K_N and K_T vary with discrete elements dimension. It is thus preferred to use material parameters Σ_N and Σ_T with units of stress, which are independent of discrete element size.

We use the Rankine criterion for fracture, which is well adapted for brittle materials. The Rankine criterion states that a bond breaks when the maximum principal stress $\sigma_{\rm R}$ reaches the critical stress $\sigma_{\rm Rc}$. In DEM, for a bond transmitting normal and tangential stress $\sigma_N = \frac{N}{4\pi R^{*2}}$ and $\sigma_T = \frac{T}{4\pi R^{*2}}$, $\sigma_{\rm R}$ is expressed as [23]:

$$\sigma_{\rm R} = \frac{1}{2}(\sigma_N + \sqrt{\sigma_N^2 + 4\sigma_T^2}) \tag{4}$$

172 4.2. Numerical preparation of aerogel particles

Aerogel particle volumes are extracted from X-ray tomography scans acquired just before 173 compression as discussed in section 3. The X-ray tomography volumes are used to generate 174 imprints of individual aerogel particles. These imprints take the form of 3D binarized images 175 which are then used to cut the correct aerogel particle shape from a random packing of spherical 176 discrete elements. This last step consists in removing all the discrete elements external to the 177 imprint. The generation of the packing has been detailed in Kumar et al. [41]. In short, it 178 consists of jamming a gas of discrete elements to a density of 0.5. This jammed packing is 179 then further densified homothetically to a 0.65 relative density. Bonds are created by defining 180 an interaction range of 1.075 meaning that spheres with centers less than 2×1.075 radius apart 181 are bonded together. Kumar et al. used this protocol to describe how the Young's modulus and 182 Poisson's ratio are related to the density of the numerical packing, average coordination and bond 183 stiffness [41]. 184

Here two different packings were used with respectively 25 μ m and 15 μ m radius discrete 185 elements. These two sizes allow the fracture behaviour of the particles to be observed with two 186 discretization refinements (smaller elements lead to a more refined discrete representation of 187 the particle). Five aerogel particles have been generated following this protocol as illustrated 188 in Fig.3. The process induced cracks observed inside the particles before compression can be 189 introduced in the model by removing discrete elements located directly on the crack and by 190 breaking the remaining bonds crossing the crack between the discrete elements which were not 191 removed by the first step. Two numerical versions of each particle are thus generated, one crack-192 free and one with the observed initial cracks. 193

¹⁹⁴ 4.3. Calibration of numerical aerogel particle compression

The DEM simulated aerogel particles are compressed between two planes. Planes are considered as infinite radius particles and the discrete element-plane contact is a bond transmitting a purely normal force in compression as described by equation 2. The stiffness of the planes is chosen as identical to aerogel bond stiffness.

The DEM model parameters can be calibrated to reproduce the observed mechanical be-199 haviour of particles. In order to calibrate normal and tangential stiffnesses of discrete elements, 200 Σ_N and Σ_T , the data gathered during aerogel particle compression inside the X-ray tomograph 201 are processed. For the same shape and the same orientation of the particles, the calibration is 202 done individually for each aerogel particle by fitting the numerical load vs. displacement curves 203 to the experimental fracture points (F_f, h_f) by tuning Σ_N . The shape of the numerical curves is 204 205 very similar to the ones observed experimentally in uniaxial compression during statistical procedure (see Fig.4 and section 3.2). Kumar et al. [41] have shown that Σ_T / Σ_N is constant for a 206 given Poisson's ratio. No practical solution has been found to measure the Poisson's ratio for the 207 studied aerogels but literature suggests that silica aerogel Poisson's ratio exhibits no discernible 208 variation with density or when using different synthesis processes. We assume that our aerogel 209 Poisson's ratio is 0.2 for all particles analyzed in this work [45, 46]. The resulting Σ_T / Σ_N is 210 0.233 [41]. The fracture of the particle is calibrated by tuning the critical stress σ_{Rc} as defined 211 by the Rankine criterion (eq.(4)). The calibration process is based on the DEM aerogel particles 212 without any initial cracks. 213

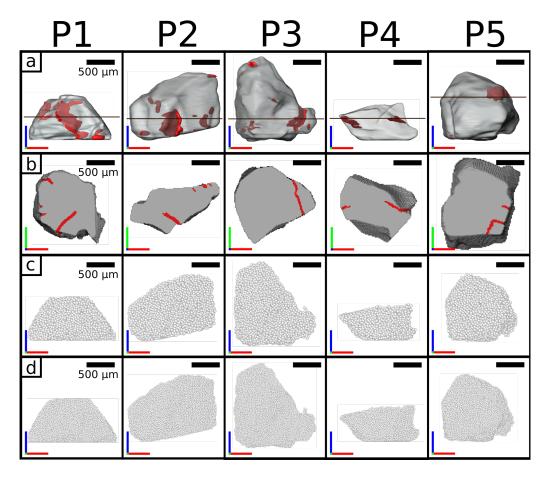


Figure 3: a) Aerogel particles with initial cracks reconstructed from X-ray tomography scans (in red). b) Cross section view within each particles at depth indicated on a). c) DEM aerogel particles prepared using 25 μ m discrete elements. d) DEM aerogel particles prepared using 15 μ m discrete elements. The numbering from P1 to P5 corresponds to a decreasing total crack length.

214 5. Results

215 5.1. Particle compression: statistical procedure

In total, 60 aerogel particles have been deformed at constant speed to initiate crack aperture 216 from a critical defect. Ten measurements have been dismissed: eight because the fracture could 217 not be accurately observed during the compression and two because of a non-satisfactory be-218 haviour (brittle failure from particle directly to dust). Fracture is defined from the detection of a 219 fall or stabilization of the force in the force/displacement curve together with a visual detection 220 of the fracture of the compressed particle. Particle fracture is noticed by a vertical crack (parallel 221 to the loading direction). A representative example of a force/displacement curve is illustrated 222 in Fig.4. In most cases the force is still rising after fracture as the particle generally keeps its 223 integrity, the different parts being held on by adhesive forces. Although somewhat arbitrary, this 224 allows for a simple and consistent fracture criterion. The results obtained using this criterion are 225

summarized in Fig.5 (fracture force vs. normalized fracture height) for the two tested compres-226 sion velocities. A large dispersion of the fracture force F_f and fracture height h_f is observed and 227 no influence of compression velocity (in between 0.3 and 1.5 mm.mn⁻¹) is discernible. Particles 228 compressed following this procedure have error bars of ± 0.01 N for force and ± 0.01 for strain. 229 In an attempt to decorrelate the fracture force from particle size, the normalized force F_f/h_0^2 is 230 calculated and plotted versus the initial height h_0 of the particles in Fig.6. Initial height is an ap-231 proximate indicator of the particles size as the height/width ratio is close to unity for our aerogel 232 particles. Normalizing force by the initial particle height to the square is a basic simplification 233 of aerogel particle geometry to propose some measure of strength. It can not describe fully the 234 complex geometry of particles. A similar simplification has been proposed in soil mechanics for 235 sand particles with similar shapes [47]. 236

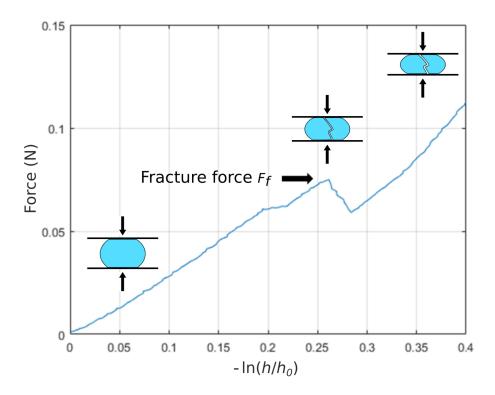


Figure 4: Experimental compression of an individual aerogel particle. h and h_0 are the current and initial sample heights, respectively.

237 5.2. Particle compression with X-ray tomography observation

Five aerogel particles have been compressed inside the X-ray tomograph tomograph giving information on particles shape, density and defects. Particles volume and density are presented in Table 1 as well as fracture force and normalized fracture height. Results from mercury picnometry carried on large batches of aerogel particles are also included in this table (MP). It is worth noting that particle P4, which is significantly smaller than others, departs from other particles

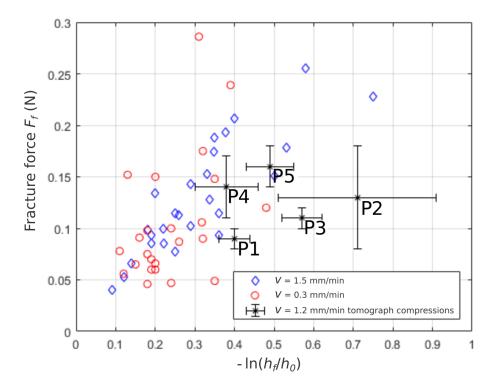


Figure 5: Fracture points (F_f, h_f) of aerogel particles following the two compression procedures : uniaxial particle compression carried out for two velocities. Samples P1 to P5 are those that have been scanned using X-ray tomography. Error bars of particles compressed following the statistical procedure are not shown for more clarity but are around 10% for fracture force and strain.

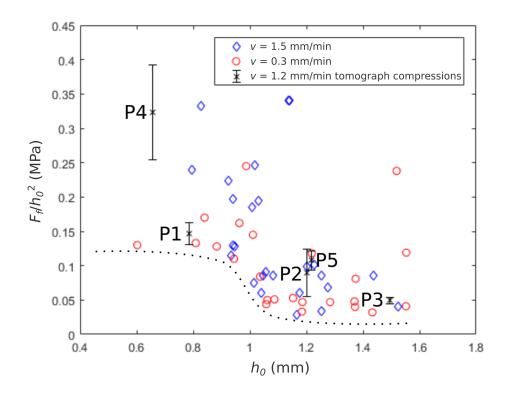


Figure 6: Normalized fracture force of compressed aerogel particles versus particle initial height h_0 .

in terms of its density, which is significantly larger. At a first glance, density is not sufficient to

warrant high fracture force, as P5 as a low density (0.10) and the highest load (0.16 N). The indi-

- vidually measured densities of P2, P3 and P5 are consistent with the mean density from mercury
- ²⁴⁶ picnometry.

Table 1: Experimental results from tomograph particle compression and mercury picnometry (MP). h_f is particle height at fracture. Particle P1 was one of the first aerogel particles compressed inside the tomograph and density was not yet measured at this time.

	Density	Volume	$-\ln(h_{\rm f}/h_0)$	F_{f}
	g/cm ³	mm ³		Ν
P1	n.a.	0.722	0.40 ± 0.04	0.09 ± 0.01
P2	0.13 ± 0.02	0.684	0.71 ± 0.20	0.13 ± 0.05
P3	0.10 ± 0.02	1.076	0.57 ± 0.05	0.11 ± 0.01
P4	0.23 ± 0.04	0.300	0.38 ± 0.08	0.14 ± 0.03
P5	0.11 ± 0.02	0.981	0.49 ± 0.06	0.16 ± 0.02
MP	0.10 ± 0.01			

In Fig.5 the fracture forces of these five particles are illustrated together with the values from 247 the statistical study and appear representative of the larger population tested outside the X-ray 248 tomograph even if the values are slightly lower than the statistical study. This small difference 249 might originate from the difference of friction with the compression tools between the two com-250 pression processes. Indeed, the graphite surface of the tools used during the statistical study 251 induces less friction than the polycarbonate tools used for particles compressed inside the X-ray 252 tomograph. Large error bars are due to the difficulty to detect the fracture under the tomograph 253 setup (see section 3.3). 254

255 5.3. Calibration of DEM particles

The five particles scanned using X-ray tomography are used to recreate numerically com-256 pression tests using the DEM model described in section 4. The DEM model is calibrated as 257 described in subsection 4.3 without introducing the initial cracks observed in X-ray tomography. 258 For each particle the calibrated normal stiffness (Σ_N) and Rankine critical stress of the bonds 259 between discrete elements (σ_{Rc}) are provided in Tables 2 and 3 for the two sizes of discrete el-260 ements used. These values are plotted in Fig.7 as a function of particle density. The differences 261 noted are relatively limited on the calibrated parameters between the two different discretizations. 262 Both Σ_N and σ_{Rc} increase with density, reflecting the stiffer and stronger mechanical response of 263 denser particles. 264

	$\Sigma_N(R = 15 \mu m)$	$\Sigma_N(R=25\mu m)$	relative difference
	MPa	MPa	
P1	2.50	2.27	9.6%
P2	2.08	2.27	8.8%
P3	2.53	2.50	1.2%
P4	8.30	10.00	18.6%
P5	3.00	3.00	0%

Table 2: Calibration of bond normal stiffness Σ_N (Eq.(2)) $\Sigma_N(R = 15\mu m)$ $\Sigma_N(R = 25\mu m)$ relative difference

Table 3: Calibration of critical bond strength σ_{Rc} (Eq.(4))							
	$\sigma_{\rm Rc}(R=15\mu m)$ $\sigma_{\rm Rc}(R=25\mu m)$		relative difference				
	MPa	MPa					
P1	0.15	0.11	30.8%				
P2	0.24	0.23	4.3%				
P3	0.20	0.20	0%				
P4	0.40	0.32	22.2%				
P5	0.25	0.21	17.4%				

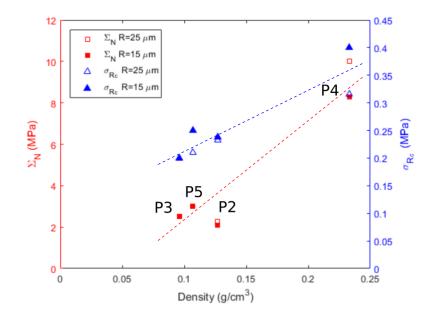


Figure 7: Calibrated DEM parameters Σ_N and σ_{Rc} for the five simulated aerogel particles as a function of particle density. Straight lines are drawn to guide the eye.

²⁶⁵ 5.4. Crack propagation in DEM and influence of initial cracks

In the previous section the five DEM aerogel particles (P1-P5) scanned by X-ray tomogra-266 phy have been calibrated and tested without introducing the initial observed cracks (Fig.3a-b). 267 By introducing these defects in the initial state of the particles before numerical compression, 268 the fracture behaviour could be better described. Particle P1 has been chosen to illustrate the 269 differences between DEM compression with and without initial cracks as illustrated in Fig.8. 270 The initial and final cracks obtained during experimental compression (X-ray tomography) are 271 showed in red on the volume reconstruction of the particle (Fig.8a) while broken bonds between 272 discrete elements are showed on DEM particles (Fig.8b). The color of discrete elements indi-273 cates the chronological order of the fractured bonds from dark red to yellow. The cracks paths 274 obtained in DEM aerogel particles do not follow the path observed with X-ray tomography. 275

The force versus displacement curves for the DEM particle P1 are shown in Fig.9. Fracture 276 forces are measured on DEM curves by identifying the maximum on the force curve preceding 277 the point corresponding to 0.1% broken bonds between discrete elements in the DEM particle. 278 We can observe (Fig.9) a clear drop of the fracture force of DEM aerogel particles with the addi-279 tion of initial cracks in the model. For the sake of completeness the experimental and numerical 280 fracture behavior of the other four aerogel particles is provided in the Appendix. Fracture force 281 values for the five DEM particles with and without initial cracks are summed up in Fig.10. These 282 fracture forces are mean values as each aerogel particle version (with or without initial cracks, us-283 ing either 25 μm or 15 μm discrete elements) was simulated with different discretizations. These 284 five discretizations were simply obtained by providing five different seeds for the random gener-285 ation of the very initial packings of discrete elements. The objective of these different packings 286 is to estimate the scattering of the simulation results due to the discretization of aerogel particles. 287 The resulting uncertainty of DEM fracture force due to the discretization depends slightly on the 288 considered aerogel particle but remains below 7%. 289

290 6. Discussion

291 6.1. Macroscopic behaviour of aerogel particles in compression

As shown in Fig.5, fracture force values are quite scattered whatever the experimental proce-292 dure (outside or inside X-ray tomograph) as fracture depends on particle shape, size, orientation, 293 density and initial defects. However, no clear difference was observed between the three tested 294 velocities. Sensitivity to strain-rate for the tested aerogel particles can thus be considered negli-295 gible for compression rates between 0.3 and 1.5 mm.mn⁻¹. The five particles compressed inside 296 the tomograph present fracture forces slightly lower than the results of the statistical study but 297 still are coherent with the experimental dispersion. Error bars are large for some particles due to 298 the difficulty to visually detect unambiguously the fracture while monitoring the X-ray projec-299 tions of the particles during compression (see section 3.3). Particle P2 exhibits such a problem 300 with thus large error bars. 301

The evolution of the normalized fracture force (units of a stress) with the initial height illustrated in Fig.6 enables to determine if the scatter in particle strength is mainly due to their size or to intrinsic properties (density and defects/cracks) and shape. No clear tendency in the normalized fracture load is observed with the variation of the particle initial height h_0 . The normalized fracture force is still very dispersed, which suggests density variation, initial defects and shape have also a great influence on aerogel particle strength. Interestingly, normalized fracture forces are not observed below 0.1 MPa for small initial particle height h_0 (below 1mm). This

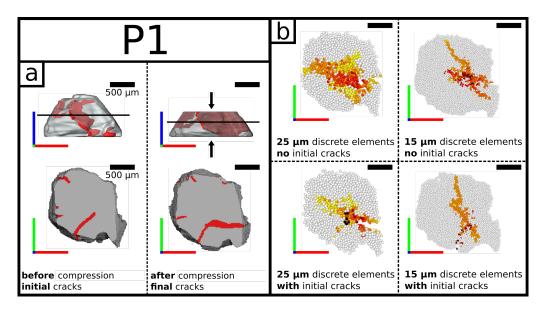


Figure 8: a) P1 particle volumes built from tomography compression test together with a cross section of interest. Top line: 3D representation, bottom line: cross section (location as indicated by a black line in top line) b) P1 particle cross sections at fracture load generated from coarse $(25 \ \mu m)$ and fine $(15 \ \mu m)$ discrete elements. Bond fractures are illustrated chronologically from dark red (earliest broken bonds) to yellow (latest broken bonds) and white (intact bonds).

could be explained by an increasing density and a lower probability to find critical initial de-309 fects for small particle volumes. Density measurements carried out on large particle populations 310 with mercury picnometry and on individual particles for 4 out of 5 of the particles scanned with 311 X-ray tomography tend to illustrate a possible link between size and density. Three particles 312 scanned with X-ray tomography have a density close to the one measured over a large particle 313 population with mercury picnometry $d_{MP} = 0.10 \pm 0.01$ g.cm⁻³ but the smallest particle studied 314 (P4) is much denser ($d_{P4} = 0.23 \pm 0.04$ g.cm⁻³). The three other particles having a density close 315 to the mean density measured with mercury picnometry suggests the total volume occupied by 316 particles as dense as P4 is small enough to have no significant impact on the mean density of a 317 large population of particles. 318

319 6.2. Weibull distribution

The statistical dispersion of fracture load in compression for the large population of individual aerogel particles (52 compressions) has been fitted with a Weibull distribution function. The probability of particle fracture is plotted as a function of the normalized force $\sigma = \frac{F_f}{h_0^2}$ in Fig.11 The fracture probability P_f was fitted with the cumulative distribution function [48]:

$$P_f = 1 - \exp\left(-\left(\frac{\sigma - \sigma_u}{\sigma_0}\right)^m\right) \tag{5}$$

where σ_u , σ_0 and *m* are the translation, scale and shape fitted parameters, respectively. The Weibull parameter *m* gives some information on the dispersion of the normalized fracture force. *m* is low for materials with a large strength dispersion and high for materials with low dispersion.

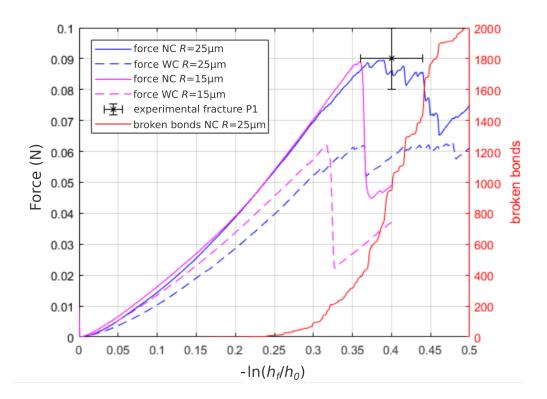


Figure 9: DEM compression of P1 with (WC) and without (NC) initial cracks generated from coarse (25 μ m) and fine (15 μ m) discrete elements. Experimental critical fracture of the particle is highlighted.

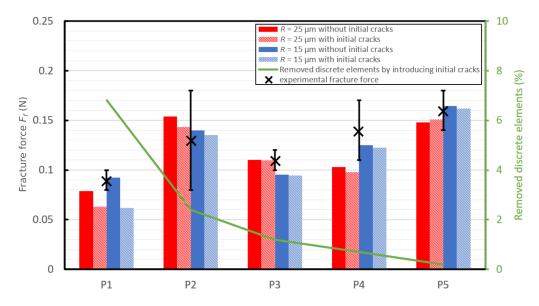


Figure 10: Fracture force for each simulated particle (P1-P5) with and without initial cracks introduced in the coarse (25 μ m) and fine (15 μ m) discrete elements mesh. The fraction of removed discrete element due to initial cracks introduction is plotted on the right hand side axis.

The parameter *m* also called Weibull modulus does not describe in our case a volume effect linked 327 to the probability of presence of a critical defect size in a given volume. Silica aerogels are porous 328 materials with a very high density of defects and do not fulfill the assumptions of Weibull analysis 329 [49, 50]. Weibull distribution law is in our case only used as a fit of the cumulative probability 330 distribution of our statistical study. The fit from Eq. (5) leads to $\sigma_u = 29$ kPa, $\sigma_0 = 92$ kPa and 331 m = 1. The latter value is extremely low. As a reminder, the Weibull modulus of most ceramics, 332 which are already considered as brittle materials with some defects, is generally between 5 and 333 10. Huillca et al. did a Weibull analysis on individual rock aggregates [51]. The rock aggregates 334 are larger than our aerogel particles (5 to 40 mm) but have similar irregular shapes. They found 335 a *m* parameter between 1.56 and 3.05 meaning our aerogel particles have a fracture behaviour 336 more scattered than for brittle rocks with similar aggregate shape. This might be explained by 337 large variations in density, in particle shape and the possible presence of initial defects. Weibull 338 fits were also realized by Woignier et al. [46] on silica aerogel monoliths synthesized using 339 supercritical drying. The modulus m is found between 4 and 7. It is difficult to compare these 340 values with our results as Woignier et al. were not as much dependent on aerogel objects shape 341 as we are with the present aerogel particles. The fact our value for m is much lower is yet 342 not surprising as supercritical drying introduces much less defects and density scattering in the 343 synthesised aerogel objects than evaporative drying. 344

³⁴⁵ 6.3. DEM determination of Young's modulus and tensile strength

As indicated in Tables 2 and 3, the relative difference of calibration parameters Σ_N and σ_{Rc} between the two DEM discretization sizes is generally less than 20%. This low dependency to discretization means that the main factors influencing the calibration of the parameters Σ_N and σ_{Rc} are linked to the nature of the material and not to the numerical effects of the model itself.

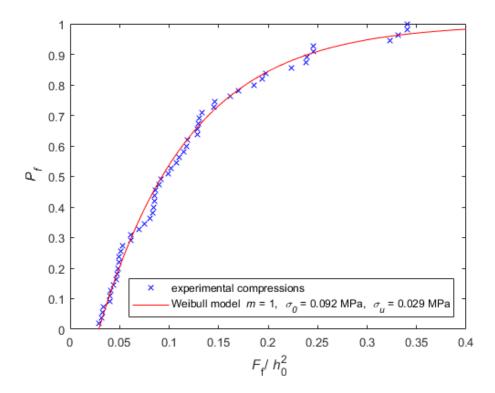


Figure 11: Weibull cumulative distribution fit carried out on the fracture by compression of aerogel particles. The fit leads to $m \approx 1$ in Eq.(5).

³⁵⁰ Σ_N and σ_{Rc} can thus be used with confidence to estimate the macroscopic Young's modulus and ³⁵¹ the fracture strength. The work of Kumar et al. [41] established relationships relating the dis-³⁵² crete element bond stiffness to the macroscopic elastic behaviour of different packings obtained ³⁵³ from various preparation routes including the one that was chosen in this work to model aerogel ³⁵⁴ particles. These equations write:

355

$$E = Z_b D \Sigma_N \frac{\kappa^n}{2\pi} \frac{a_1 + a_2 \alpha}{4 + a_3 \alpha} \tag{6}$$

$$\alpha = \frac{1 - b_2 \nu}{b_1 + b_3 \nu} \tag{7}$$

where $E Z_b$, D and κ are the Young's modulus, the average coordination number, the packing density and the interaction range of the discrete element packing, respectively. The parameters n, a_i and b_i are fitted ($n = 2, a_1 = 0.42, a_2 = 7.45, a_3 = 8.68, b_1 = 0.55, b_2 = 3.02, b_3 = 5.75$) and originate from Kumar's work. The preparation route used here is defined as "weakly jammed" and results in a packing density of D = 0.65 and an average coordination number $Z_b = 6.56$ with an interaction range $\kappa = 1.075$.

Similarly to the Kumar et al. study, Radi et al. [23] established relationships between the macroscopic strength of a packing and the microscopic bond strength. Their equation writes:

$$\sigma_f = \sigma_{Rc}(c_1 Z_b D - c_2) \tag{8}$$

where σ_f is the macroscopic tensile strength, σ_{Rc} the Rankine critical stress of the bonds and c_i are fitted parameters ($c_1 = 0.124$, $c_2 = 0.067$).

These relations allow for the estimation of macroscopic properties from the calibrated mi-366 croscopic parameters of the discrete element packing. For particles P2, P3 and P5 (of similar 367 densities), the resulting estimations are $E = 0.7 \pm 0.1$ MPa and $\sigma_f = 100 \pm 10$ kPa. For particle 368 P4 $E_{P4} = 2.6 \pm 0.3$ MPa and $\sigma_{fP4} = 165 \pm 26$ kPa. It is encouraging that our model and cali-369 bration method lead to a Young's modulus close to 1 MPa as this value is consistent with various 370 studies on the elastic response of silica aerogels as function of their density [9, 10]. As expected, 371 the dense particle P4 has a much higher Young's modulus than other particles. Concerning the 372 estimated fracture strength, the previously cited literature [9, 10] also measured tensile strength 373 (three point bending tests, brazilian tests) for various silica aerogels with various densities. For 374 densities around 0.1 g.cm⁻³ the tensile strength ranges between 10 and 100 kPa. This is con-375 sistent with our value of 100 ± 10 kPa that is calculated from DEM simulations. According to 376 Hiramatsu el al. [52], the fracture force normalized by the initial height to the square of the com-377 pressed particles is a correct estimation of the tensile strength. It is thus consistent to compare 378 the fitted scale parameter σ_0 obtained with the Weibull fit performed previously in section 6.2 379 with the tensile strength obtained with Eq. (8). The two estimations of the tensile strength based 380 on experimentation ($\sigma_0 = 92$ kPa) and simulation ($\sigma_f = 100 \pm 10$ kPa) agree very well. The use 381 of DEM calibrated micro-parameters has the advantage of removing aerogel particle shape and 382 orientation from the possible source of strength scattering. When considering aerogel particles 383 with similar densities, the strength displays a coefficient of variation of 10%. This remaining 384 scattering should originate mainly from the presence of initial cracks. 385

³⁸⁶ 6.4. Influence of process induced cracks on the DEM mechanical behavior of aerogel particles

The introduction of cracks in DEM aerogel particles modifies their fracture behaviour. The objectives of the simulation are to quantify and to explain the influence of the initial cracks on

particles behaviour, as well as to check if the numerical crack propagation matches what was 389 observed in tomography. 390

A positive point for the DEM model is that the final cracks observed for the four configu-391 rations as illustrated for particle P1 in (Fig.8) are located following a clearly identifiable path 392 cutting the particles into two distinct main portions, especially for the thinly meshed particles 393 (15 μ m discrete elements) as the cracks are located in a much thinner zone compared to the 394 $25 \,\mu m$ meshed particle. Particles with and without initial cracks do not fracture by following 395 exactly the same path but the general behaviour is maintained. Still, the main cracks observed 396 in DEM particles do not follow the fracture observed experimentally in tomography, except for 397 the 25 μ m discretized particle for which some similarities can be found. The DEM model fails 398 to reproduce the final crack pattern once the fracture is initiated as it seems to depends too much 399 on the discretization of the model. Also, we believe that the boundary conditions of the experi-400 ment at the onset of crack growth (interaction between the platens and the particle) are difficult 401 to reproduce exactly. The complex crack growth patterns after fracture depend too much on the 402 full strain history to be reproduced accurately in DEM. 403

However the initiation of the cracks is mostly correct with initial broken bonds (shown in 404 black or dark red in Fig.8) localised around or at the extremity of the initial cracks. After its 405 initiation, the crack follows a path imposed by the discrete element packing and so does not 406 describe the real case. Thinly discretized particles (15 μ m) are best suited for the acquisition of 407 precise crack patterns but are four to five times more computationally expensive compared to the 408 $25 \,\mu m$ discretized particles. 409

Although the cracks morphological differences are minor for particle P1 with and without 410 initial cracks, there is a quantitative difference between particles with and without initial cracks 411 with a drop of the fracture force in compression of approximately 30% (Fig.9). Fig.10 indicates 412 that particle P1 is the only simulated particle exhibiting a fracture force drop superior to 15% 413 after initial cracks are introduced in the model. This is consistent with the fact that particle P1 is 414 the particle with the largest relative number of discrete elements removed in the crack modelling 415 process (7% versus 2.5% for P2, the second most impacted particle). Particles P2 to P5 exhibit 416 much smaller variations of the fracture force when initial cracks are introduced. In general, a 417 decrease of the fracture force is almost always observed when initial cracks are introduced but, 418 apart from particle P1, it is low. It is thus safe to conclude that the initial cracks introduced 419 in the DEM aerogel particles following our procedure have rarely a large influence both on the 420 cracks morphology after crack initiation and on the fracture force values. It might be due to the 421 orientation and localisation of the initial crack patterns which rarely weaken the aerogel particles 422 in the uniaxial compression direction. The probability to find a well located and oriented crack 423 increases with the amount of initial damage and only P1 presents a crack parallel to the loading 424 conditions and located in a zone experiencing a high tensile stress (typically close to the center 425 of the particle). It could also explain why tensile strengths lower than 0.1 MPa were not observed 426 for small particles during experimental uniaxial compression (see Fig.6) as the probability to find 427 a well located and oriented defect might increase with particle volume. 428

7. Conclusions 429

Experimental compression tests and DEM simulations were carried out in order to investigate 430 the mechanical behavior of millimetric silica aerogel particles produced by evaporative drying. 431 A first experimental statistical study has shown that the values of fracture force are very 432

much scattered, as quantified by Weibull statistics. This is due to the combination of the vari-433

ability of the particles shape, orientation, density and the presence of initial defects. In order to 434 decorrelate the influence of these parameters, an original methodology based on individual X-ray 435 tomography images of particles coupled with DEM simulations is proposed. Individual density 436 assessment of aerogel particles based on X-ray tomography volumes has shown significant den-437 sity variation (0.1 to 0.23 g/cm³) that explains an important part of the scattering. Particles with 438 similar densities still exhibit important normalized fracture force variations that can be attributed 439 to particles shape and orientation and to the potential presence of initial defects in the particles 440 during compression test. The methodology consists of the numerical replication with DEM of a 441 compression experiment, using the particle morphology and orientation from X-ray tomography 442 images. The calibrated DEM parameters are used to assess the material Young's modulus and 443 strength. This methodology gives encouraging results, with Young's modulus (0.7 ± 0.1 MPa) 444 and strength $(100 \pm 10 \text{ kPa})$ values for a density of around 0.1 g/cm³ all consistent with literature 445 data on supercritically dryed monolithic aerogels. The remaining small scattering in strength (446 coefficient of variation of 10%) is attributed to the methodology uncertainty itself and the inher-447 ent variability due to the presence of initial defects. The relatively modest influence of process 448 induced defects is confirmed by DEM modeling that shows that for particle in compression, the 449 initial cracks within the sample have rarely a highly detrimental effect on strength. It is consis-450 tent with the fact that the obtained value for strength is similar to reported values for monolithic 451 aerogel of similar density. Still, one should keep in mind that for a purely homogeneous tensile 452 loading (which might happen within a composite) the presence of initial cracks in the aerogel 453 particle will probably be more critical. 454

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461 Appendix A. Fracture of DEM aerogel particles P2 to P5

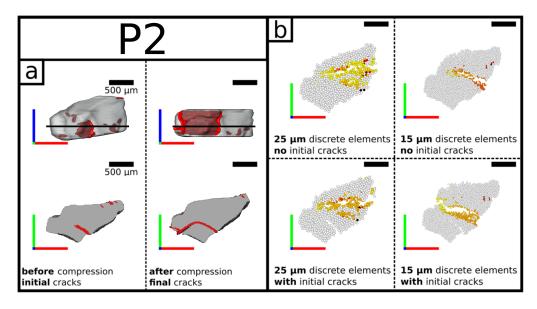


Figure A.12: a) P2 particle volumes built from tomography compression test. b) P2 particle volumes built using 2 types of discrete elements with DEM. The bond fractures are illustrated chronologically from hot to cold colors.

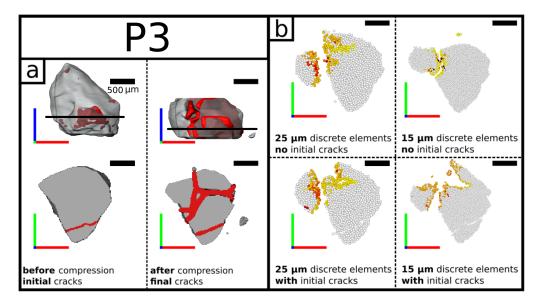


Figure A.13: a) P3 particle volumes built from tomography compression test. b) P3 particle volumes built using 2 types of discrete elements with DEM. The bond fractures are illustrated chronologically from hot to cold colors.

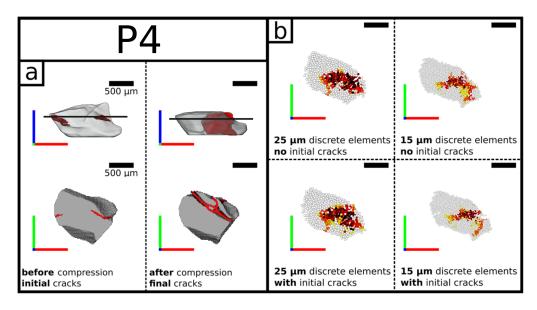


Figure A.14: a) P4 particle volumes built from tomography compression test. b) P4 particle volumes built using 2 types of discrete elements with DEM. The bond fractures are illustrated chronologically from hot to cold colors.

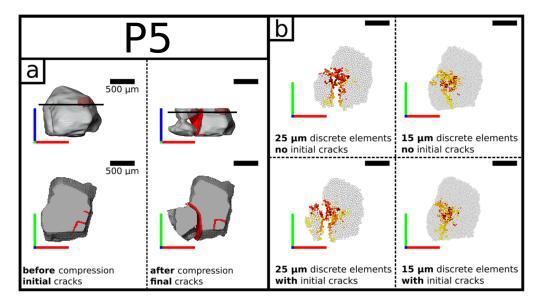


Figure A.15: a) P5 particle volumes built from tomography compression test. b) P5 particle volumes built using 2 types of discrete elements with DEM. The bond fractures are illustrated chronologically from hot to cold colors.

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