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GLUTATHIONE CONJUGATES OF OCHRATOXIN A AS BIOMARKERS OF EXPOSURE*

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In the present study the photoreactivity of the fungal carcinogen ochratoxin A (OTA) has been utilised to generate authentic samples of reduced glutathione (GSH) and *N*-acetylcysteine (NAC) conjugates of the parent toxin. These conjugates, along with the nontoxic $OT\alpha$, which is generated through hydrolysis of the amide bond of OTA by carboxypeptidase A, were utilised as biomarkers to study the metabolism of OTA in the liver and kidney of male and female Dark Agouti rats. Male rats are more susceptible than female rats to OTA carcinogenesis with the kidney being the target organ. Our studies show that the distribution of OTA in male and female rat kidney is not significantly different. However, the extent of OTA metabolism was greater in male than female rats. Much higher levels of $OT\alpha$ were detected in the liver compared to the kidney, and formation of $OT\alpha$ is a detoxification pathway for OTA. These findings suggest that differences in metabolism between male and female rats could provide an explanation for the higher sensitivity of male rats to OTA toxicity.

KEY WORDS: *bioactivation, carcinogenicity, DNA adduction, glutathione, metabolism, mutagenicity, N-acetylcysteine*

Ochratoxin A (OTA, Figure 1) is a mycotoxin produced by several fungi of *Aspergillus* and *Penicillium* species (1-5). It consists of a chlorophenolic group containing a dihydroisocoumarin moiety that is amide-linked to phenylalanine. The toxin is a common contaminant of cereals and agricultural products (2, 3), and the International Agency for Research on Cancer (IARC) has classified OTA as a possible human carcinogen (group 2B) (6). OTA is nephrotoxic and is a potent renal carcinogen in rats (7) and in chicks (8). In humans, OTA has been associated with nephropathies in the Balkans (9) and in North African countries (10).

Because OTA is a possible human carcinogen present in food products, tolerable daily intakes (TDIs) are proposed by government agencies to manage the risk of OTA exposure. The TDI for OTA established by the Joint FAO/WHO Expert Committee on Food Additives has been set at ~ 14.28 ng kg⁻¹ b.w. per day (100 ng kg⁻¹ b.w. per week), based on nephrotoxic effects in pigs (11). Health Canada has proposed a TDI of 4 ng kg⁻¹ b.w. per day that considers tumour formation by OTA (5). These differences in TDI stem from a lack of consensus concerning the mechanism

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of action (MOA) for OTA. However, Hibi et al. (12) have recently demonstrated the *in vivo* mutagenicity of OTA in male rat kidney, suggesting involvement of a direct genotoxic mechanism in OTA-mediated renal carcinogenesis. The more stringent TDI proposed for OTA by Health Canada is more in line with carcinogens whose MOA involves direct genotoxicity (5).

Direct genotoxicity by OTA may stem from its oxidative metabolism into electrophiles that are capable of reacting covalently with DNA to generate DNA adducts (13, 14). OTA possesses a chlorophenolic group, and chlorophenols are well known to undergo oxidative dechlorination reactions to afford quinone/ hydroquinone redox couples (15, 16). The hydroquinone metabolite (OTHQ, Figure 1) of OTA has been detected in the urine (17) and kidneys (18) of rats, and in human blood and urine samples (19). Oxidation of OTHQ generates the quinone electrophile OTQ that reacts with DNA to generate covalent DNA adducts, as evidenced using ³²P-postlabeling (20). OTA also reacts photochemically with 2'-deoxyguanosine (dG) to generate a carbon (C)-linked nonchlorinated ochratoxin B (OTB)-dG adduct (21) that has recently been detected in the kidney of male rats using liquid chromatography-mass spectrometry (LC-MS) (22). The C5-Cl atom of OTA plays a critical role in OTBdG formation, as the nonchlorinated OTB analog lacks direct genotoxicity and does not react photochemically with dG to generate the OTB-dG adduct (23).

The C5-Cl atom of OTA also plays a role in its metabolism by carboxypeptidase A into the nontoxic ochratoxin a (OT α , Figure 1) and phenylalanine. The nonchlorinated OTB derivative undergoes hydrolysis by carboxypeptidase A at a faster rate than OTA (24, 25) and this has been invoked as a rationale for the lower toxicity of the OTB analog (26). The nontoxic $OT\alpha$ has previously been detected as the major OTA metabolite in human blood and urine samples (27) and may serve as a sensitive biomarker for OTA exposure (28). Other biomarkers for OTA exposure would include the OTB-dG adduct (22), as OTA is mutagenic (12), and OTB-dG formation suggests that a reactive metabolite of OTA has reached a toxicologically significant target (DNA). However, the presence of OTB-dG in treated rat kidney samples is low, 20 to 70 per 10^9 nucleotides (22). Furthermore, the adduct OTB-dG is unstable and undergoes hydrolysis in the presence of acid, which is commonly used for OTA extraction from biological and food samples (27, 29). These traits (low abundance and poor stability) make OTB-dG unsuitable as a general biomarker for OTA exposure.

Electrophiles generated from the metabolism of OTA also react with reduced glutathione (GSH) to produce GSH-conjugates. The OTB-GSH conjugate (Figure 1) is the major species formed from the photoreaction of OTA with excess GSH (23) and corresponds structurally with OTB-dG. Nucleophilic



Figure 1 Chemical structures of OTA and OTA metabolites.

attachment of GSH to the quinone electrophile OTQ affords OTHQ-GSH (Figure 1) that has been characterised previously using NMR spectroscopy and is generated by treatment of OTA with rat liver microsomes in the presence of excess GSH (30). Unlike the DNA adduct OTB-dG, OTB-GSH and OTHQ-GSH are stable to acid treatment and can be extracted from biological samples using methodology established for extraction of the parent toxin OTA (27, 29).

In the present study authentic samples of OTB-GSH, OTHQ-GSH, and the corresponding *N*-acetylcysteine (NAC) conjugates of OTA have been used as biomarkers to examine OTA metabolism in the liver and kidney of male and female rats. Our goal was to use the conjugates to determine sex and organ differences in OTA metabolism, given that the male rat is particularly susceptible to OTA-mediated carcinogenesis, with the kidney being the target organ (7). Our studies provide new evidence for OTB-GSH and OTHQ-GSH formation in liver and kidney of both sexes. Interestingly, clear differences in the extent of OTB-GSH versus OTHQ-GSH formation was observed in male versus female and liver versus kidney, with the male kidney providing preferential OTB-GSH formation. These findings may help shed light on the susceptibility of male rats to OTAmediated carcinogenesis.

MATERIALS AND METHODS

Materials

OTA (benzene free, CAS# 303-47-9) was purchased from Sigma (Saint Quentin Fallavier, France). A synthetic sample of OTHQ (30) was available in the laboratory at Guelph, Canada, and was used as a mixture of diastereomers [3(R, S)], (Figure 1). Stock solutions of OTA and OTHQ in aqueous 50 mmol L^{-1} phosphate buffer pH 7.4 were prepared as outlined previously (31, 32). Glutathione (GSH) and Nacetylcysteine (NAC) were purchased from Sigma-Aldrich (Oakville, ON, Canada). All HPLC solvents were of chromatography grade and water used for buffers was obtained from a MilliQ filtration system (18.2 MW).

Synthesis of GSH and NAC Conjugates

Authentic samples of GSH and NAC conjugates of OTA were generated from the photoreaction of OTA or OTHQ with GSH or NAC. Reaction mixtures (2 mL total volume) of 1 mmol L⁻¹ OTA or OTHQ in the presence of 15 molar equivalent GSH or NAC were irradiated at 350 nm using a Rayonet Chamber Reactor, model RPR-200, as outlined for the synthesis of OTB-GSH and OTHQ-GSH (23). The photoreaction of OTA in the presence of GSH or NAC was used for the generation of OTB-GSH or OTB-NAC, while the corresponding photoreaction of OTHQ with GSH or NAC was used for the preparation of OTHQ-GSH or OTHQ-NAC. Purifications were performed using an Agilent 1200 series HPLC equipped with an autosampler, autocollector, diode array detector (monitored at 258 nm and 340 nm) and fluorescence detector [excitation (λ_{ext}) = 340 nm; emission (λ_{em}) = 465 nm]. Separations were carried out using an Agilent C18 column (150 mm x 4.6 mm, particle size 5 µm) with the following mobile phase: 0.1 % formic acid in H₂O:0.1 % formic acid in acetonitrile (70:30) to (25:75) in 17 min by a linear gradient at a flow rate of 0.75 mL min⁻¹. The GSH- and NAC-conjugate peaks were combined and lyophilised to dryness using a Labconco FreeZone^{4.5} freeze-dry system. The authentic conjugate samples were checked for purity through HPLC analysis (>96 % based on integration of the HPLC trace) and were characterised by MS. Mass spectra were obtained using an Agilent 1100 LC-MSD with a single quadrupole detector with electrospray negative ionisation (ESI-). The OTB-NAC conjugate was also characterised by nuclear magnetic resonance (NMR) spectroscopy on a Bruker Avance 600 MHz NMR spectrometer equipped with a HCN pulsed-field gradient cryoprobe. Data collection was carried out at room temperature in ²H₆-dimethyl sulfoxide, DMSO d_6 (99.9 %) in a 2 mm NMR tube. The ¹H NMR spectrum was acquired using 1024 scans and a sweep width of 10 ppm. The two-dimensional ¹H-¹H correlation (COSY) spectrum was collected with 2048 complex points in t_2 , 512 points in t_1 , and 64 transients per t, increment. Data were multiplied with optimised phase-shifted squared sinebell apodization functions with zero filling to 1024 points in the t_1 dimension. Spectra were referenced to the DMSO-d₅H peak at 2.50 ppm and were processed using the Topspin 2.1 software (Bruker BioSpin Ltd., Milton, Canada).

Rat tissue samples

Dark Agouti rat lung, liver, and kidney samples were available in the laboratory at Toulouse, France and were obtained from a rat study performed by CIT (International Centre of Toxicology, Evreux) in compliance with the Principles of Good Laboratory Practice as described in: (i) OECD Principles on Good Laboratory Practice (revised in 1997), ENV/MC/ CHEM (98) 17 and (ii) Commission Directive 1999/11/EC of 8th March 1999 adapting to technical progress the Principles of Good Laboratory Practice as specified in the Council Directive 87/18/EEC on the harmonisation of laws, regulations and administrative provisions relating to the application of the Principles of Good Laboratory Practice and the verification of their applications for tests on chemical substances (OJ No. L 77 of 23rd March 1999). The rat study was modelled after the protocol described in depth in the paper by Castegnaro et al. (27). In the current study, 7-week-old Dark Agouti rats were fed wheat contaminated with various concentrations of OTA over a 28-day period. The rats (25 males and 25 females) were housed in individual cages (43 cm x 21.5 cm x 20 cm) and kept in environmentally controlled conditions (ventilation, 22°C, 12 h dark/ light cycles). The wheat provided by the French miller (Paulic Minotiers) was of a common variety used for making bread. All rats had free access to noncontaminated wheat over a 7-day acclimation period. This was followed by exposure of the rats to OTAcontaminated wheat. One batch of 8 kg was coded as control. Four batches, 8 kg each, were artificially contaminated with OTA, respectively at: 2.5 µg kg⁻¹, 7 μ g kg⁻¹, 40 μ g kg⁻¹, and 100 μ g kg⁻¹. To uniformly incorporate OTA into the wheat, OTA was dissolved in ethanol and introduced into the wheat (in the presence of 1 % water) as a fine spray over a 15-min period. Each batch of 8 kg wheat was distributed into 53 plastic bags, each containing 150 g of wheat. The plastic bags were stored at -20 °C in closed plastic containers and were allowed to thaw at room temperature 1 day prior to use. The OTA content of the wheat samples was confirmed using HPLC with fluorescence detection (29). On completion of the treatment period, the rats were euthanised and the organs were weighed and frozen at - 80 °C until use.

OTA and metabolite extraction

OTA and its metabolites were extracted from the organ tissue samples using a procedure similar to the

one outlined by Molinié et al. (29) for the extraction of OTA from cereals and modified by Pfohl-Leszkowicz et al. (33). Tissue samples (1 g) were homogenised in the presence of 20 mL 0.1 mol L^{-1} MgCl₂ and 0.05 mol L^{-1} HCl (pH 1.5). The aqueous acidic solution was then extracted three times with CHCl₂ (20 mL, 10 mL, 10 mL). Following each CHCl₂ treatment, the samples were centrifuged (10 min, \sim 3,000 x g, 4 °C) in order to separate the CHCl₂ component from the acidic aqueous phase. The combined CHCl, extracts were then treated with 40 mL sodium bicarbonate and the solution was shaken for ~ 10 min. The bicarbonate solution was collected and acidified to pH 1.5 with HCl and then extracted three times $(3 \times 20 \text{ mL})$ with CHCl₂. The combined CHCl, extracts were dried under vacuum and the resulting residue was dissolved in 1 mL methanol, filtered, and concentrated under nitrogen to a final volume of 200 μ L.

OTA and OTA metabolite analysis

Reverse-phase HPLC analysis of OTA and its metabolites were carried out using a Gilson 811B dynamic HPLC pump. OTA was analysed using a Nucleosil 100-3 C18 column (150 mm x 4.6 mm, particle size 3 mm) under isocratic elution (methanol, 600 mL; acetonitrile, 600 mL; water, 800 mL; sodium acetate, 0.68 g; glacial acetic acid, 28 mL) with detection performed with a programmable fluorimeter GTI spectrovision (λ_{ext} =340 nm; λ_{em} =465 nm). The limit of detection (LOD) was 0.05 ng g⁻¹; the limit of quantification (LOQ) was 0.2 ng g⁻¹ (34). The OTA metabolites were separated using a Prontosil C18 column (250 mm x 4 mm, particle size 3μ m) with solvent A: methanol:acetonitrile:6.5 mmol L⁻¹ ammonium formate (200:200:600) adjusted to pH 3 with formic acid; and solvent B: methanol: acetonitrile:6.5 mmol L⁻¹ ammonium formate (350:350:300) adjusted to pH 3 with formic acid, using gradient elution, as outlined previously (34, 35). Detection was also performed using the fluorimeter GTI spectrovision (λ_{ext} =340 nm; λ_{em} =465 nm) that permitted a LOD of ~ 0.05 ng g⁻¹ for OTA.

RESULTS

Generation of GSH and NAC conjugates

The photoreaction of OTA or OTHQ in the presence of 15 molar equivalent GSH or NAC was

utilised to generate the GSH- and NAC-conjugates of OTA. As outlined previously (23), photoirradiation (15 min) of OTA (1 mmol L^{-1}) at 350 nm in the presence of excess GSH generates OTB-GSH (Figure 1) as the major product. The conjugate has a phenolic absorption at $l_{max} = 331$ nm and a parent ion at m/z673. In the negative ESI spectrum a major fragment ion is observed at m/z 400, which is 32 mass units heavier than OTB (m/z 368) and stems from b-elimination of the attached benzoquinol-SH (23). The photoreaction also generates OTHQ-GSH (Figure 1) as a minor product (23) that forms by nucleophilic attachment of GSH to the quinone electrophile OTQ (30). This conjugate is the sole species formed from the photoreaction of OTHQ (1 mmol L⁻¹) with excess GSH. It has been fully characterised previously using NMR spectroscopy and possesses a phenolic absorption at $l_{max} = 352$ nm and a parent ion at m/z 689 with a prominent fragment ion at m/z 416 from b-elimination of benzoquinol-SH (23, 30).

Figure 2 shows the ESI- spectra of the corresponding NAC-conjugates of OTA (OTB-NAC) and OTHQ (OTHQ-NAC). The OTB-NAC conjugate had a phenolic absorption at 331 nm and a parent ion at m/z 529 with a major fragment at m/z 400 (Figure 2A), as noted for fragmentation of OTB-GSH. The phenolic absorption of OTHQ-NAC occurred at 360 nm and the conjugate had a parent ion at m/z 545 with a major fragment at m/z 416 (Figure 2B). Spectral data for OTA, OTHQ and the corresponding GSH- and NAC-conjugates are given in Table 1.

The OTB-NAC conjugate was also characterised by NMR spectroscopy. Figure 3 shows the twodimensional ¹H-¹H correlation (COSY) spectrum of



Figure 2 Negative ionisation electrospray mass spectra of (A) OTB-NAC and (B) OTHQ-NAC.

the conjugate recorded in DMSO-d₆ at room temperature. The ¹H chemical shifts of the conjugate are given in Table 2. The Table also includes ¹H chemical shift data for OTHQ, which was also recorded in DMSO-d₆ (16), and OTA, which was recorded in CDCl₃ by Dais et al. (36). The NMR data together with the MS data provided an unambiguous assignment of the OTB-NAC structure, as ¹H signals for the attached NAC moiety could be determined and H-6 of the OTB component was observed at ~ 8 ppm, as expected from the ¹H NMR data for OTHQ (16) and OTA (36).

species	RT	λ_{\max}^{d}	parent ion m/z	fragment ions m/z
OTA^b	13.3	333	402	358, 314
$OTHQ^b$	10.9, 11.0	350	384	340, 296
$OTB-GSH^b$	6.8	331	673	672, 544, 400
OTHQ-GSH ^b	6.2	352	689	688, 560, 416
OTB-NAC	9.5	331	529	400
OTHQ-NAC	8.0, 8.2	360	545	416

Table 1 Spectral data for photoreactions of OTA and OTHQ with GSH and NAC^a

^a Photoreactions of ochratoxins (1 mmol L^{-1}) in the presence of 15 molar equivalent GSH or NAC were carried out in 50 mmol L^{-1} phosphate buffer pH 7.4 for 15 min with irradiation at 350 nm using a Rayonet Chamber Reactor, Model RPR-200.

^b Data taken from reference 23.

^c Retention time in minutes using an Agilent C18 (150 mm x 4.6 mm, 5 mm particle size) column with a flow rate of 0.75 mL min⁻¹ using the following mobile phase: 0.1 % formic acid in $H_2O:0.1$ % formic acid in ACN (70:30) to (25:75) in 17 min by a linear gradient.

^{*d*} *Phenolic ring absorbance in nm.*



Figure 3 ¹H-¹H COSY spectrum of OTB-NAC recorded at 600 MHz in DMSO-d₆.



Figure 4 Change in liver (A, C) and kidney (B, D) weight of male (A, B) and female (C, D) Dark Agouti rats following varying levels of OTA exposure for 28 days.

Rat studies

From the rat study it was noted that the weight gain of both male and female Dark Agouti rats was unaffected by the level of OTA ingestion (2.5 μ g kg⁻¹, 40 μ g kg⁻¹, and 100 μ g kg⁻¹) following the 28-day period. Figure 4 shows the corresponding changes in liver and kidney weight for males and females. A significant decrease (p<0.01, Student t-test) in male (A) and female (C) liver weight was observed for all rats receiving OTA, regardless of concentration, compared to the control (C in the x-axis). This was not the case for the kidney weight (B, D). For both males (B) and females (D), the highest dose (100 μ g kg⁻¹) of OTA did not significantly affect the kidney weight compared to the control, even though using Wilcoxon rank test a significant decrease was observed with lower doses.

Table 2 Proton NMR data for OTB-NAC, OTHQ and OTA.

proton	δ ¹ H OTB-NAC ^a	δ ¹ H OTHQ ^b	δ ¹ Η ΟΤΑ ^c
Н-3	4.39	4.29	4.69
H-4a	3.35	3.12	3.22
H-4b	2.55	2.66	2.79
H-6	8.06	7.72	8.36
H-13	4.43	4.29	4.96
H-14a	3.12	3.12	3.30
H-14b	3.08	3.12	3.15
H-16-20 (phenyl)	7.19 to 7.23	7.29	7.13 to 7.30
H-21 (CH ₃)	1.34	1.43	1.53
H-23a	3.06	_	_
H-23b	2.81	_	_
H-24	3.91	_	_
H-25 (NH)	7.53	_	_
H-27 (CH ₃)	1.75	_	_

 $^{a\ l}H$ NMR chemical shifts in ppm recorded at 600 MHz in DMSO-d_{s}

^b Chemical shift data taken from reference 16, recorded at 300 MHz in DMSO-d_c.

^c Chemical shift data taken from reference 36, recorded at 400 MHz in CDCl₃.

Figure 5 shows the distribution of OTA in the lung, liver, and kidney of male (A) and female (B) Dark Agouti rats following varying levels of OTA exposure for the 28-day period. The levels of OTA (ng g⁻¹) presented in Figure 5 were obtained following OTA extraction from the tissues and performing HPLC analysis with fluorescence detection (excitation 340 nm, emission 465 nm), using an authentic sample of OTA for comparison for an LOD of 0.05 ng g⁻¹ and an LOQ of 0.2 ng g⁻¹. The relative levels of OTA in the tissues of both sexes were similar to the highest levels being found in the lung and liver. The OTA amounts in different tissues increased with OTA dose.



Figure 5 Distribution of OTA in tissues of male (A) and female (B) Dark Agouti rats following varying levels of OTA exposure for 28 days.

For rats receiving the highest dose (100 μ g kg⁻¹) of OTA, tissue samples were also analysed for metabolite formation using authentic samples of OTa, OTB-GSH, OTHQ-GSH, OTB-NAC, and OTHQ-NAC for comparison. For these studies, OTA and its metabolites extracted from four liver and four kidney samples from each sex were combined for HPLC analysis with fluorescence detection (excitation 340 nm, emission 465 nm) after gradient separation. Figure 6 shows a representative HPLC chromatogram depicting the relative levels of OTA, OT α , OTB-GSH, and OTHQ-GSH from male (A) and female (B) liver

samples, while the corresponding HPLC chromatograms from the analysis of the male (A) and female (B) kidney samples are shown in Figure 7. In both tissue samples peaks corresponding to the NAC-conjugates were not observed. However, peaks that co-migrated with OT α and the GSH-conjugates were found. Compared to the peak for OTA, levels of metabolite formation were greater for males (A) than females (B) in both liver (Figure 6) and kidney (Figure 7). The most striking difference between kidney and liver in both sexes was the high degree of OTB-GSH formation in the kidney relative to OT α and OTHQ-GSH. The liver of both sexes generated higher levels of OTHQ-GSH, and relatively high levels of OT α were also detected in male rats (Figure 6A).



Figure 6 HPLC chromatograms with fluorescence detection (excitation 340 nm, emission, 465 nm) showing relative levels of OTA, OTα, OTB-GSH and OTHQ-GSH from male (A) and female (B) livers (4 from each sex) following treatment with 100 µg kg⁻ⁱ OTA for 28 days.

DISCUSSION

The results of our studies highlight the potential utility of GSH- and NAC-conjugates of OTA and OTHQ as biomarkers for OTA exposure. The photoreaction of OTA in the presence of excess GSH (23) or NAC can be used to readily generate authentic samples of OTB-GSH, OTHQ-GSH (Figure 1), or OTB-NAC and OTHQ-NAC (Figure 2). The OTHQ-GSH conjugate stems from the attachment of GSH to the quinone (OTQ) electrophile, which is produced by cytochrome P450 metabolism of OTA (16, 20). The OTB-GSH or OTB-NAC conjugates are generated



Figure 7 HPLC chromatograms with fluorescence detection (excitation 340 nm, emission, 465 nm) showing relative levels of OTA, OTα, OTB-GSH and OTHQ-GSH from male (A) and female (B) kidneys (4 from each sex) following treatment with 100 μg kg⁻¹ OTA for 28 days.

photochemically from displacement of the C5-Cl atom of OTA by the sulphur atom of GSH or NAC. The electrophilic intermediate in this reaction also reacts with dG to yield the OTB-dG adduct (21), which has been detected by LC-MS in the kidney of male rats fed a carcinogenic dose of OTA (22). Thus, OTB-GSH and OTHQ-GSH, along with the corresponding NACconjugates, were considered as appropriate biomarkers for OTA exposure. The conjugates stem from the same electrophiles that are deemed important for OTAmediated DNA adduction (14), which is now expected to play a key role in the mutagenicity and carcinogenicity of OTA (12). However, unlike the DNA adducts generated by OTA, the GSH- and NAC-conjugates of OTA are stable to acid and can be extracted from biological tissues using strategies developed for the parent toxin (27, 29, 37). This suggested that the conjugates would be easier to detect than the corresponding DNA adducts and would still serve as an indication that the biological target (DNA) had been exposed to OTA-mediated electrophiles.

To test the potential for conjugate formation by OTA, Dark Agouti rat liver and kidney samples were analysed for levels of OTA, $OT\alpha$, OTB-GSH, OTHQ-GSH, OTB-NAC, and OTHQ-NAC, using HPLC with fluorescence detection (34, 35). The rat organ samples were available from a study in which 7-week-old Dark Agouti male and female rats were fed various concentrations of OTA in wheat for a period of 28

days; similar to the protocol previously reported (27). Given that male rats are more susceptible than females to OTA-mediated carcinogenesis (7, 38), it was anticipated that the conjugate formation analysis would provide new insights into sex and organ differences in OTA-mediated metabolism, which might help explain the susceptibility of male rat kidney to OTA (39, 40).

Initially, the impact of OTA on liver and kidney weight (Figure 4) and the distribution of OTA in the lung, liver, and kidney of male and female rats (Figure 5) were analysed. The distribution study (Figure 5) showed no significant difference between males and females regarding OTA concentrations, which is in agreement with a recent study carried out by Vettorazzi et al. (41) on the distribution of OTA in male and female F344 rats. These studies suggest that other factors, such as differences in metabolism (40, 42), could explain the higher sensitivity of male rats to OTA carcinogenicity.

The HPLC data presented in Figures 6 and 7 demonstrate that differences in OTA metabolism between male and female rats and between the liver (Figure 6) and the kidney (Figure 7) do exist. In both liver and kidney the extent of OTA metabolism was greater in male than female rats. In the kidney, both sexes generate OTB-GSH as the principle metabolite. This was not the case in the liver where male rats generated appreciable quantities of $OT\alpha$ and OTHQ-GSH in addition to OTB-GSH (Figure 6A). The female rats also generated low levels of all three metabolites (Figure 6B). These results provide insight into the susceptibility of male rat kidney to OTA carcinogenesis. First, male rats generate higher levels of GSHconjugates than female rats, suggesting a greater level of OTA bioactivation, which is required for DNA adduction (13, 14) and probably in vivo mutagenicity (12). Second, much higher levels of OTa are detected in the liver (Figure 6A) compared to the kidney (Figure 7A) and formation of $OT\alpha$ is a detoxification pathway for OTA (24-26), suggesting greater sensitivity of the kidney to OTA. At present, the GSH-conjugates of OTA in rat liver and kidney were characterised by HPLC with fluorescence detection using authentic samples for comparison. Future studies will focus on the use of LC-MS/MS strategies to unambiguously confirm the presence of the GSH-conjugates in the rat tissue samples. The GSH- and NAC-conjugates reported in this work will also be used as standards for the analysis of urine samples taken from animals and humans exposed to OTA.

CONCLUSION

We demonstrated that exposure of male and female rats to OTA results in the production of GSHconjugates that have been detected using HPLC with fluorescence detection. These results suggest that OTA undergoes metabolism in rat liver and kidney to generate electrophiles that react covalently with GSH. These data add additional support to the hypothesis that differences in OTA metabolism between male and female rats provide a rationale for the sensitivity of male rat kidney to OTA carcinogenesis.

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Sažetak

GLUTATIONSKI KONJUGATI OKRATOKSINA A KAO BIOMARKERI IZLOŽENOSTI

U ovom je ispitivanju korištena fotoreaktivnost kancerogenog mikotoksina okratoksina A (OTA) kako bi se stvorili izvorni uzorci reduciranih glutationskih (GSH) i *N*-acetilcisteinskih (NAC) konjugata osnovnog toksina. Ovi konjugati, uz netoksični OT α , koji se stvara hidrolizom amidne veze OTA putem karboksipeptidaze A, upotrijebljeni su kao biomarkeri za ispitivanje metabolizma OTA u jetri i bubregu ženki i mužjaka štakora soja Dark Agouti. Mužjaci su se pokazali podložnijima stvaranju bubrežnih tumora uzrokovanih OTA toksinom od ženki. Utvrdili smo da se raspodjela OTA u bubrezima ženki i mužjaka značajno ne razlikuje. Međutim mužjaci su imali intenzivniji metabolizam OTA nego ženke. U jetri su utvrđene mnogo više razine OT α u usporedbi s bubregom, a rezultati upućuju na to da je stvaranje OT α detoksifikacijski put za OTA. Zaključujemo da bi se veća osjetljivost mužjaka štakora na toksičnost OTA mogla pripisati spolno uvjetovanim razlikama u njegovu metabolizmu.

KLJUČNE RIJEČI: bioaktivacija, glutation, kancerogenost, metabolizam, mutagenost, N-acetilcistein, okratoksin A, stvaranje adukata u molekuli DNA

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