

Supporting Information

Circularly Polarized Luminescence of Eu(III) Compounds with chiral BINOL-derived Bisphosphate Ligands

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{[Dy(hfac)₃((S)-L³)]₃}_n: 32.8 mg (0.04 mmol) of [Dy(hfac)₃(H₂O)₂] were added to a solution of 30.3 mg (0.04 mmol) of (S)-L³ or (R)-L³ in 5 mL of CH₂Cl₂. After 15 minutes of stirring, 30 mL of *n*-hexane were layered. After several days, the solution was slowly evaporated leading to a colourless and microcrystalline solid. 27.6 mg, 44 % yield for {[Dy(hfac)₃((S)-L³)]₃}_n. Anal. calcd (%) for C₁₇₇H₁₂₉Dy₃F₅₄O₄₂P₆: C 45.90, H 2.79; found: C 45.82, H 2.94. Representative I.R. (KBr, range 3200–400 cm⁻¹): 3079 (w), 2937 (w), 2863 (w), 1654 (s), 1591 (w), 1555 (m), 1527 (m), 1491 (s), 1256 (s), 1201 (s), 1147 (s), 1102 (m), 1075 (m), 1027 (m), 1018 (m), 992 (m), 971 (w), 962 (w), 808 (w), 798 (m), 778 (w), 755 (m), 741 (w), 687 (w), 661 (m), 586 (m), 529 (w), 507 (w) and 466 (w) cm⁻¹. Slow evaporation from a solution in

CH₂Cl₂/*n*-hexane led to single crystals suitable for structural X-ray studies. $[\alpha]_D^{25} = -6.5$ for $\{[\text{Dy}(\text{hfac})_3((S)\text{-L}^3)]_3\}$ (*c* = 1.0, CHCl₃).

It crystallizes in the P2₁2₁2₁ (N°19) chiral orthorhombic space group (Table S2) and its asymmetric unit is composed of three (S)-L³ ligands and three Dy(hfac)₃ moieties. The Dy(hfac)₃ moieties are linked by the (S)-L³ ligands through the P=O groups to form a one-dimensional structure (Figure S1). The Dy(III) ion is surrounded by eight oxygen atoms coming from three hfac⁻ anions and two P=O groups. The arrangement of the ligands leads to distorted square antiprism (D_{4d} symmetry) coordination polyhedron for the three crystallographically independent Dy(III) centres. The deviation from the ideal symmetries are determined by SHAPE¹ analysis and depicted in Table S3. From such analysis, it appears that the Dy1 and Dy3 adopt an almost regular square antiprism while the coordination sphere of Dy2 is close to a D_{2d} symmetry (triangular dodecahedron). The average Dy-O bond length is (2.34(6) Å) and the Dy-O_{hfac} are slightly longer than the Dy-O_{P=O} bond lengths. The P=O-Dy-O_{P=O} angles range from 139.9° to 152.8°. The angle between the planes formed by the binaphthyl groups are found close in the three (S)-L³ ligands of the asymmetric unit with values ranging from 75.0(3)° to 86.0(1)°. The intra-chain Dy-Dy distances are found equal to 12.528(2) Å (Dy1-Dy2), 12.339(2) Å (Dy2-Dy3) and 12.662(3) Å (Dy3-Dy1). The crystal packing (Figure S2) highlights both F...F and H...F contacts to assume the cohesion of the crystal but no π-π stacking has been found. The shortest inter-chain Dy-Dy distances are found equal to 11.447(2) Å (Dy1-Dy1) so slightly shorter than the intra-chain distances.

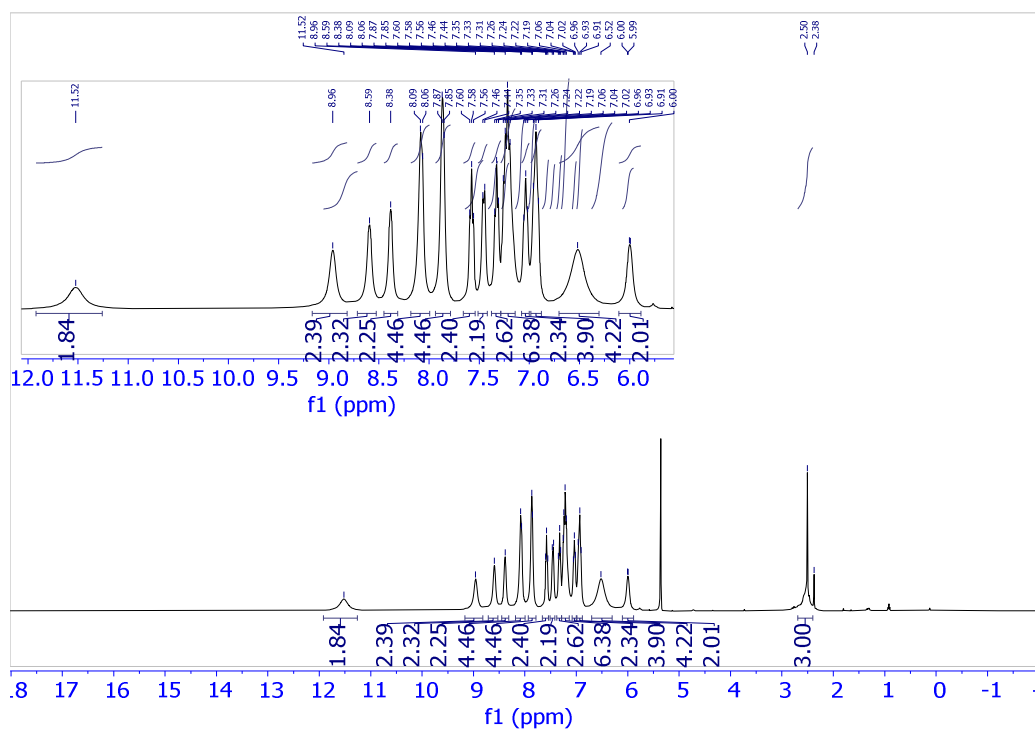


Figure S1. ^1H NMR spectrum of [(*S,S,S*)/(*R,R,R*)-2] in CD_2Cl_2 at 298 K (400 MHz). Inset: expanded region from 5.56 to 12.06 ppm.

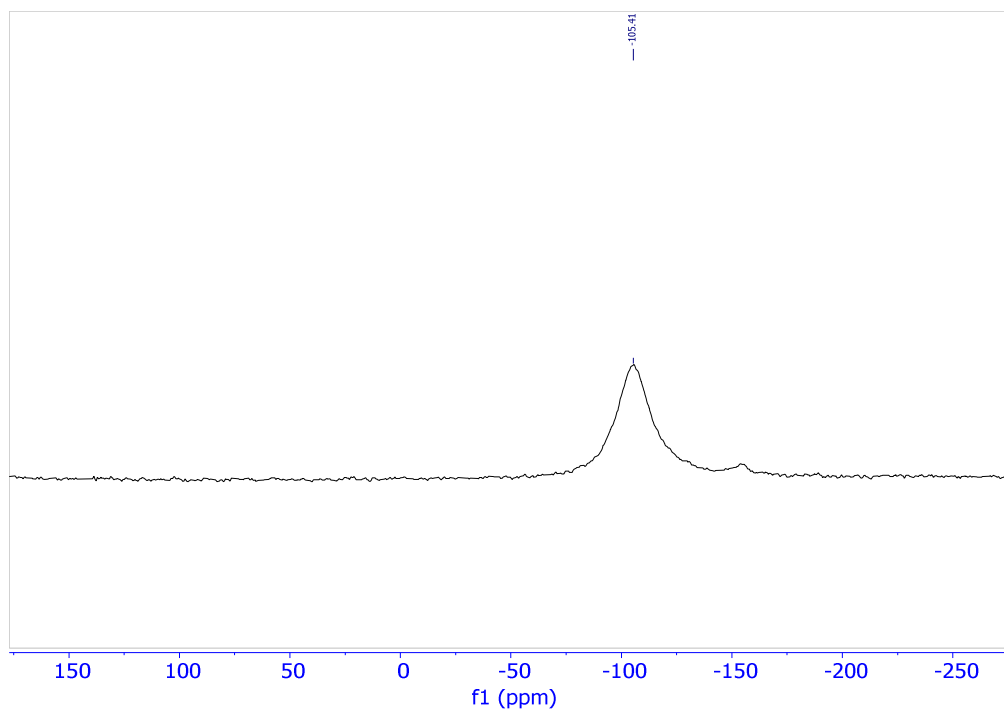


Figure S2. ^{31}P spectrum of [(*S,S,S*)/(*R,R,R*)-2] in CD_2Cl_2 at 298 K (162 MHz).

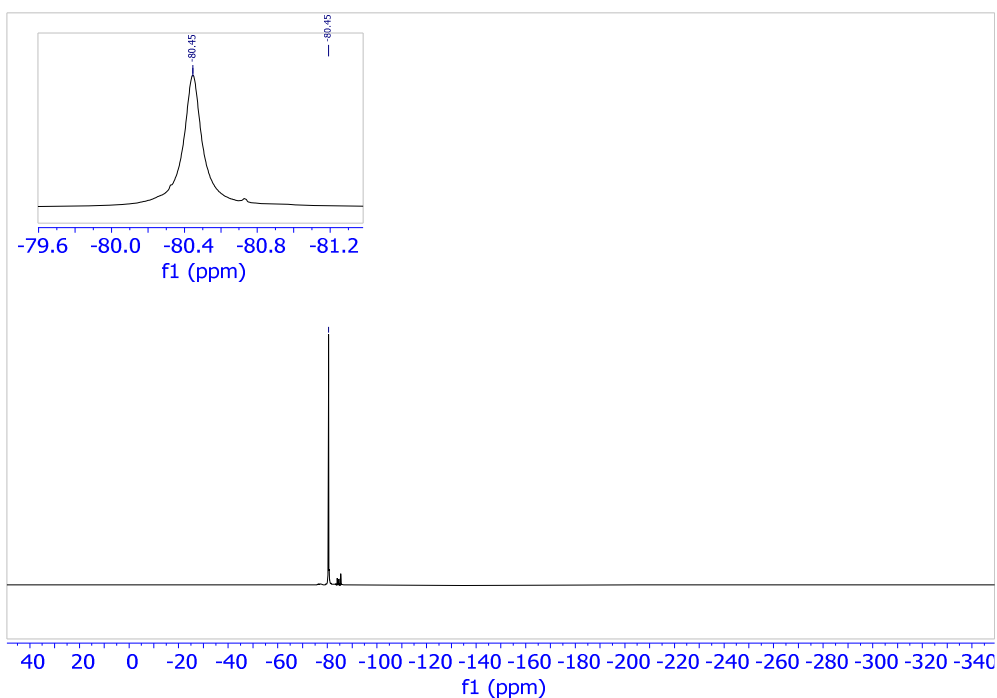


Figure S3. ^{19}F spectrum of [(*S,S,S*)/(*R,R,R*)-2] in CD_2Cl_2 at 298 K (376 MHz). Inset: expanded region from -81.3 to -79.6 ppm.

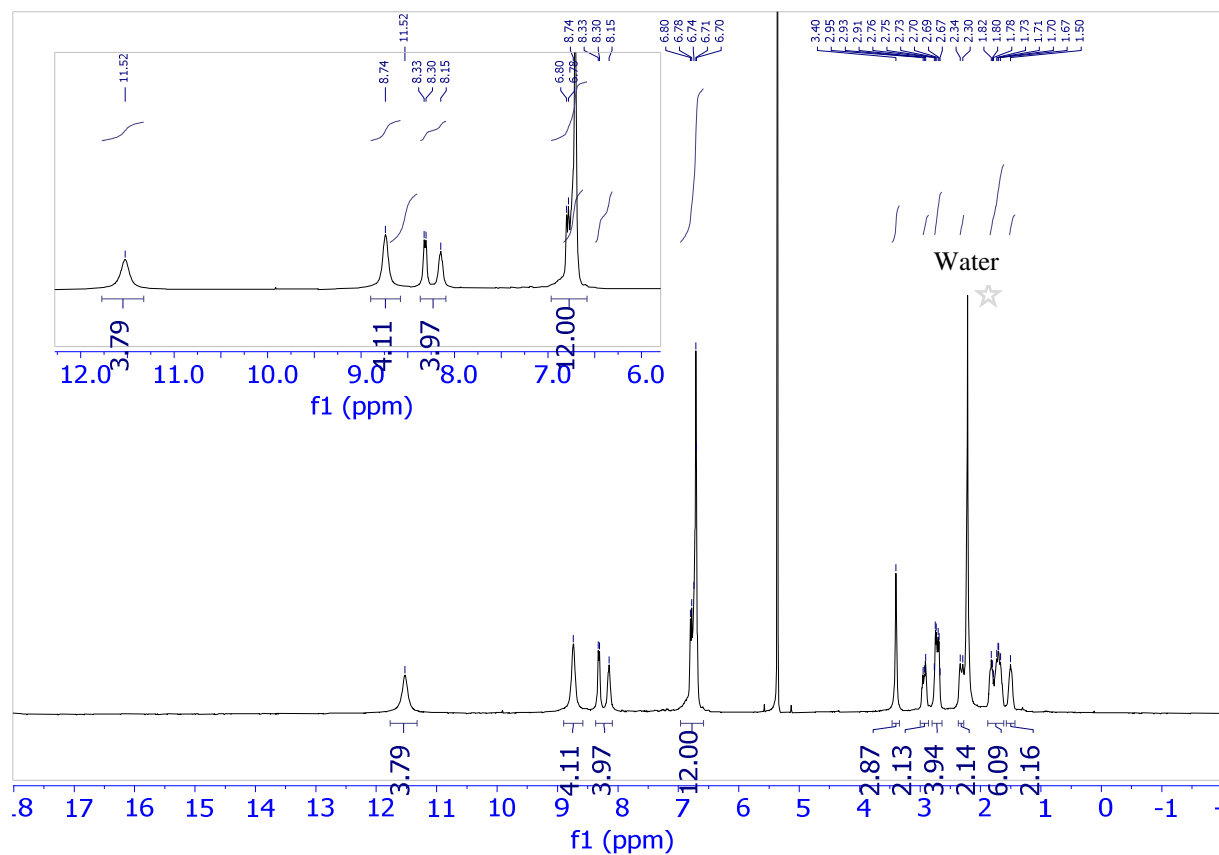


Figure S4. ^1H NMR spectrum of [(*S*)/(*R*)-3] in CD_2Cl_2 at 298 K (400 MHz). Inset: expanded region from 5.80 to 12.22 ppm.

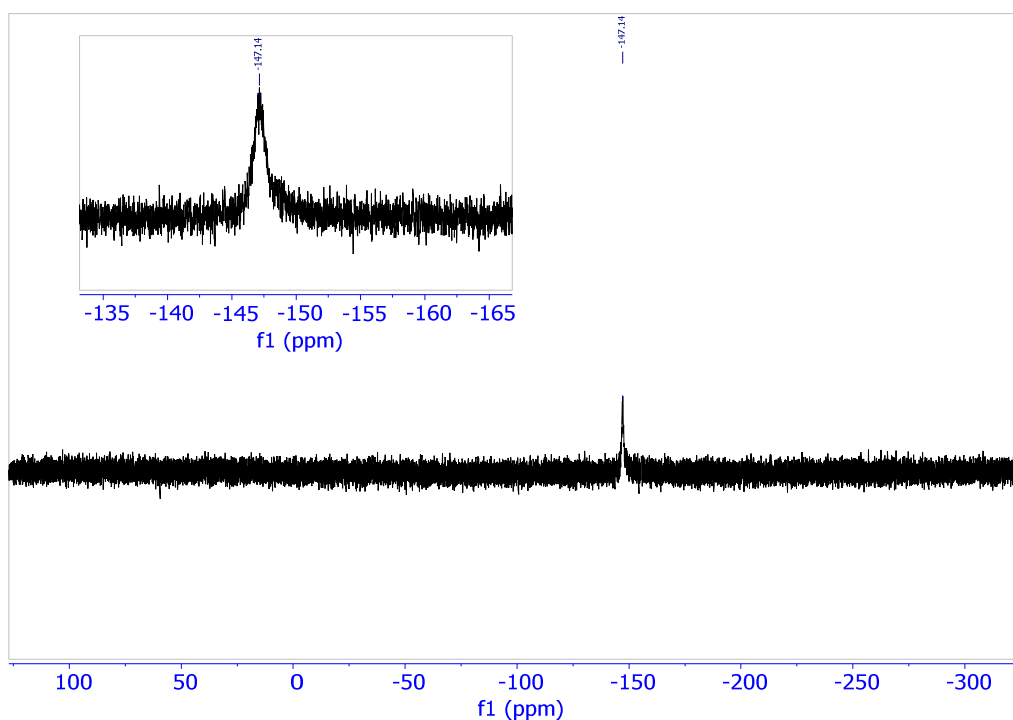


Figure S5. ^{31}P spectrum of [(*S*)/(*R*)-**3**] in CD_2Cl_2 at 298 K (162 MHz). Inset: expanded region from -166 to -135 ppm.

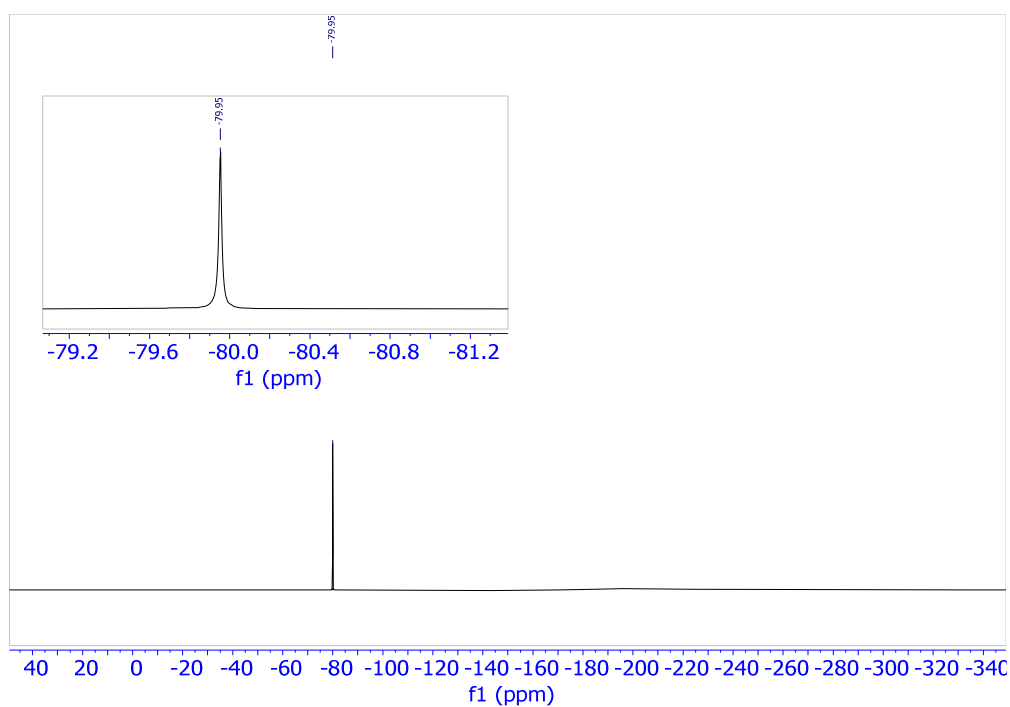


Figure S6. ^{19}F spectrum of [(*S*)/(*R*)-**3**] in CD_2Cl_2 at 298 K (376 MHz). Inset: expanded region from -81.4 to -79.1 ppm.

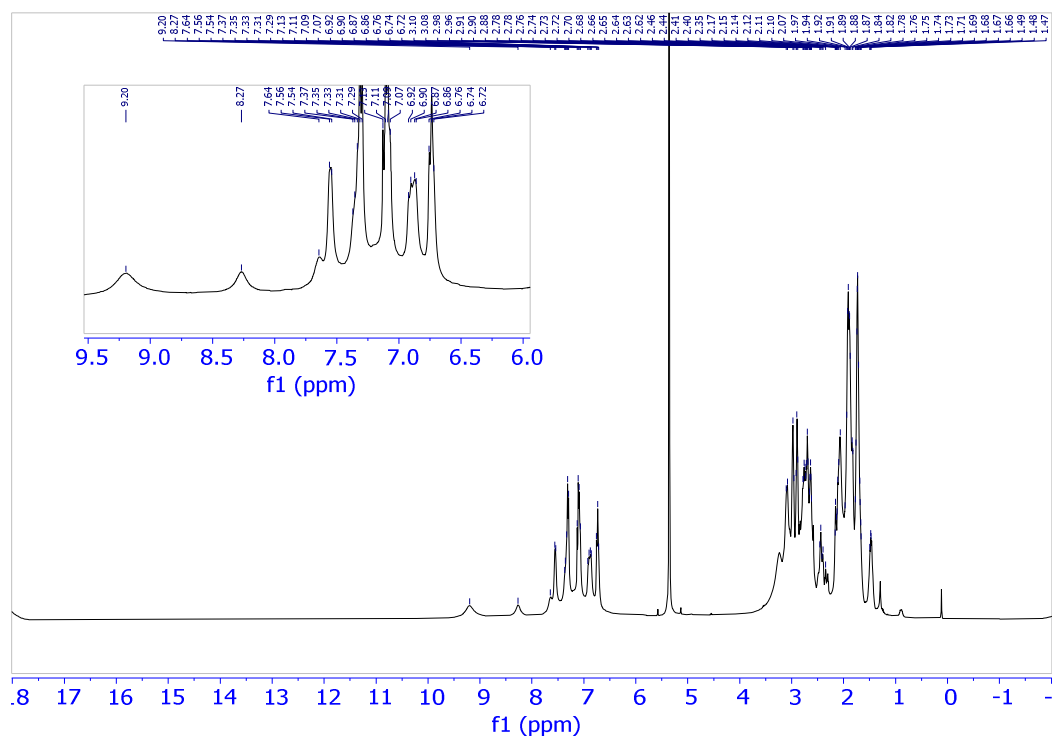


Figure S7. ¹H NMR spectrum of [(*S,S,S*)/(*R,R,R*)-4] in CD₂Cl₂ at 298 K (400 MHz). Inset: expanded region from 5.98 to 9.52 ppm.

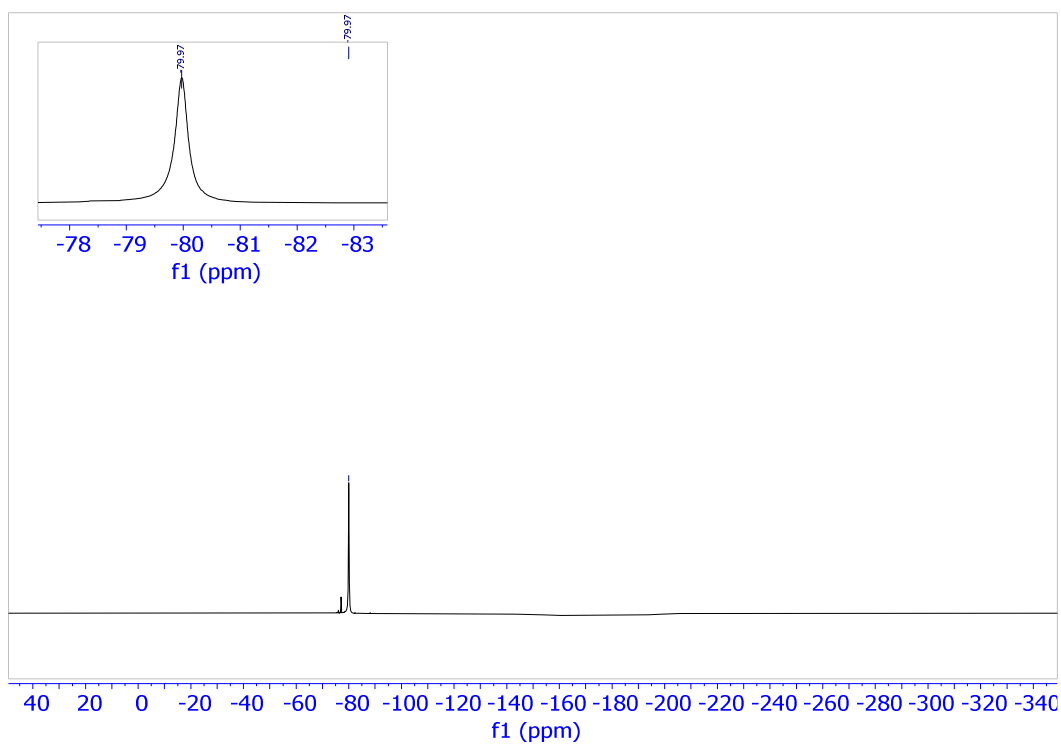


Figure S8. ¹⁹F spectrum of [(*S,S,S*)/(*R,R,R*)-4] in CD₂Cl₂ at 298 K (376 MHz). Inset: expanded region from -83.5 to -77.4 ppm.

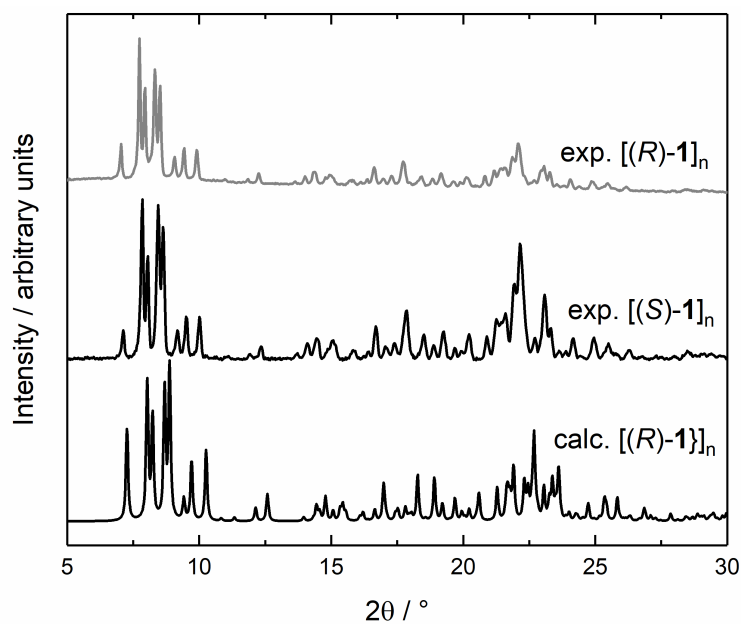


Figure S9. Superposition of experimental powder X-ray diffraction patterns from $[(S)\text{-}1]_n$ and $[(R)\text{-}1]_n$ measured at 300 K and simulated from $[(R)\text{-}1]_n$ single-crystal data obtained at 150 K².

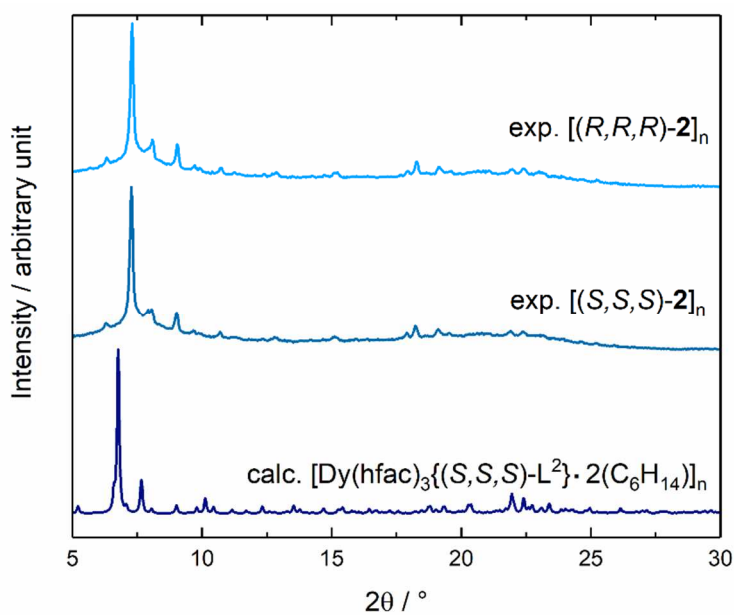


Figure S10. Superposition of experimental powder X-ray diffraction patterns from $[(S,S,S)\text{-}2]_n$ and $[(R,R,R)\text{-}2]_n$ measured at 300 K and simulated from $\{[\text{Dy}(\text{hfac})_3\{(\text{S,S,S})\text{-L}^2\}] \cdot 2(\text{C}_6\text{H}_{14})\}_n$ single-crystal data obtained at 150 K².

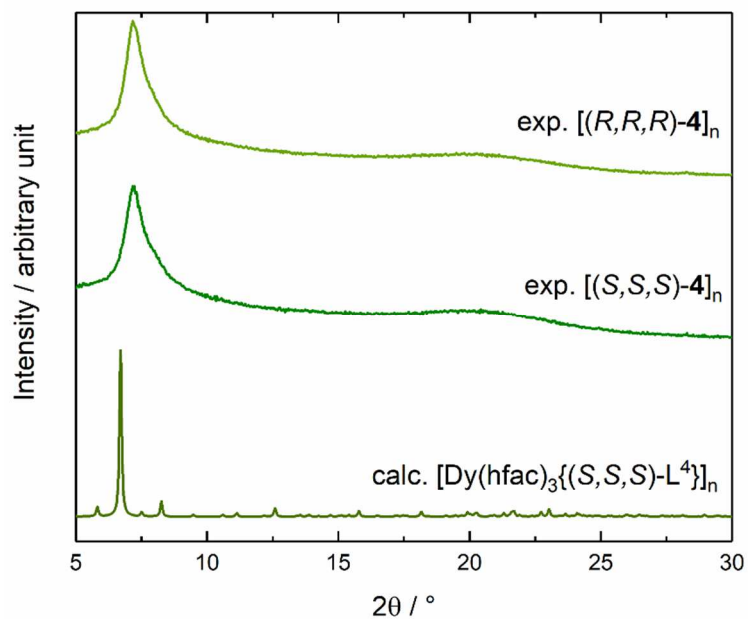


Figure S11. Superposition of experimental powder X-ray diffraction patterns from [(*S,S,S*)-**4**]_n and [(*R,R,R*)-**4**]_n measured at 300 K and simulated from {[Dy(hfac)₃((*S,S,S*)-L⁴)]_n single-crystal data obtained at 150 K³.

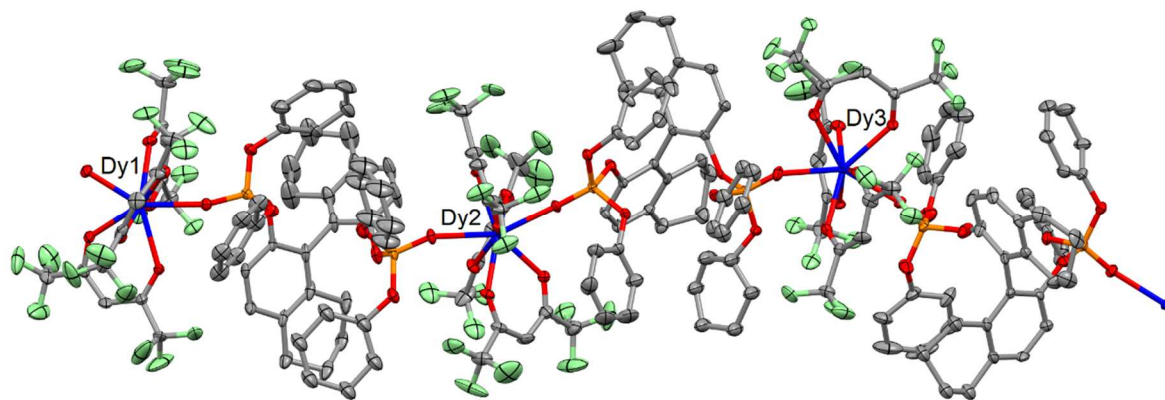


Figure S12. ORTEP view of the asymmetric unit for {[Dy(hfac)₃((*S*)-L³)]_n. Thermal ellipsoids are drawn at 30% probability. Hydrogen atoms are omitted for clarity.

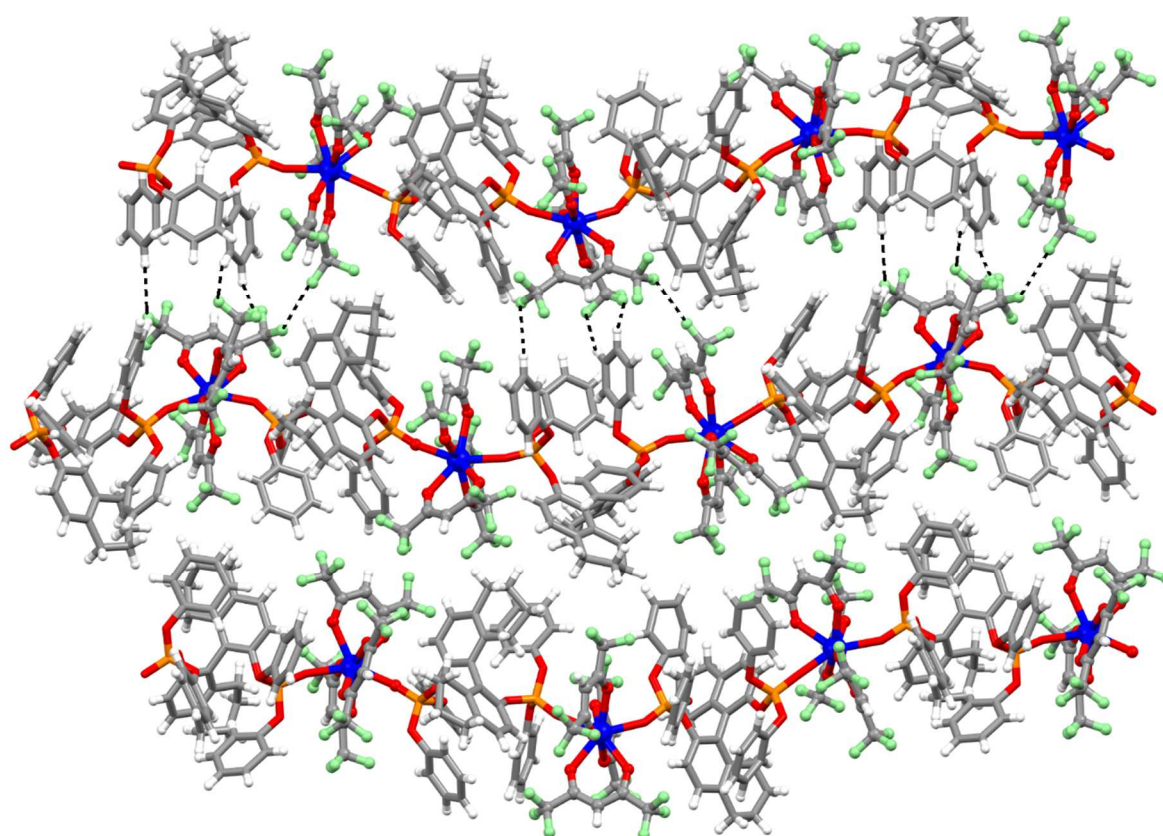


Figure S13. Crystal packing of $\{[\text{Dy}(\text{hfac})_3((S)\text{-L}^3)]\}_n$ highlighting the $\text{F}\cdots\text{F}$ and $\text{H}\cdots\text{F}$ contacts.

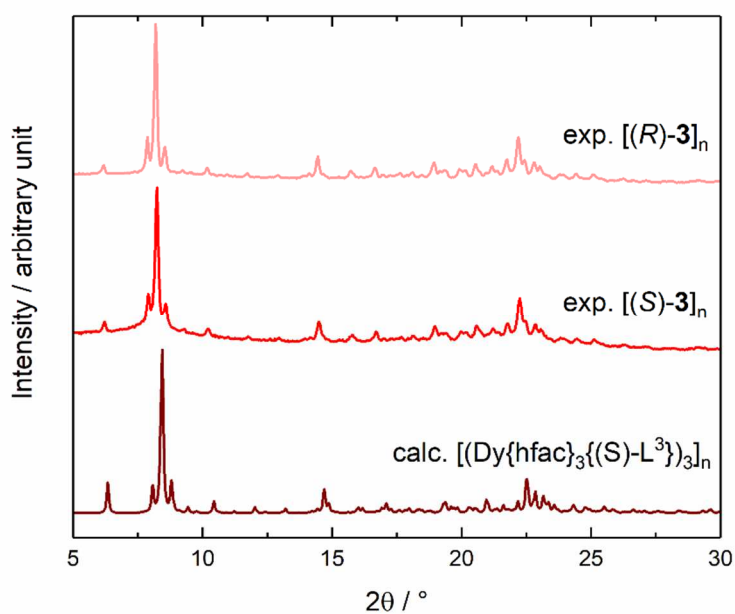


Figure S14. Superposition of experimental powder X-ray diffraction patterns from $[(S)\text{-3}]_n$ and $[(R)\text{-3}]_n$ measured at 300 K and simulated from $\{[\text{Dy}(\text{hfac})_3((S)\text{-L}^3)]_3\}_n$ single-crystal data obtained at 150 K³.

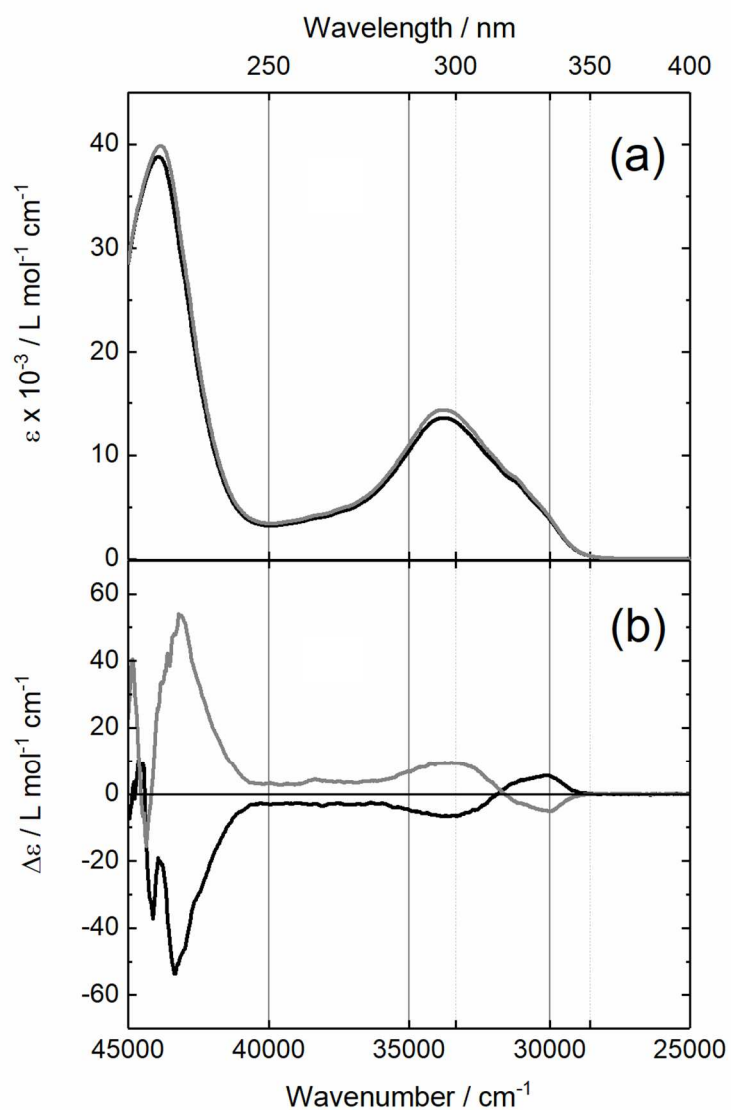


Figure S15. (a) UV-visible absorption and (b) ECD spectra in CH_2Cl_2 solution at $C = 1 \cdot 10^{-5}$ mol L^{-1} and room temperature for [(S)-1] (gray line) and [(R)-1] (black line).

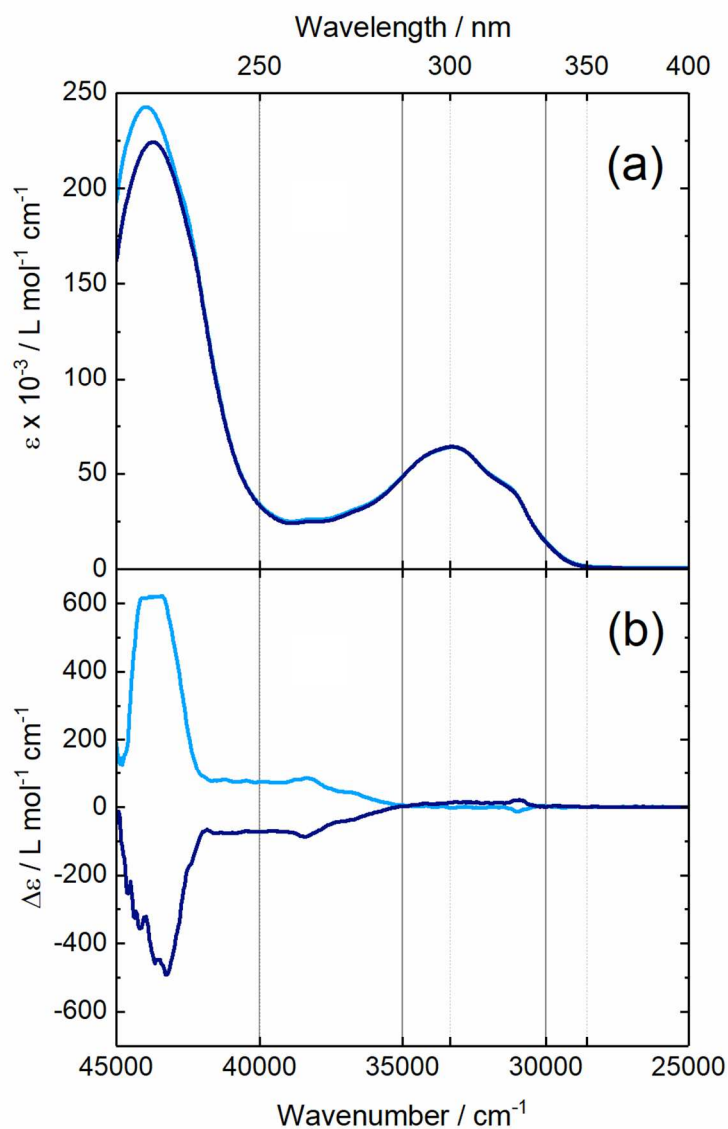


Figure S16. (a) UV-visible absorption and (b) ECD spectra in CH_2Cl_2 solution at $C = 1 \cdot 10^{-5} \text{ mol L}^{-1}$ and room temperature for $[(S,S,S)\text{-}2]^*$ (light blue line) and $[(R,R,R)\text{-}2]$ (dark blue line).
 * The ECD signal at low energy is slightly saturated for this compound.

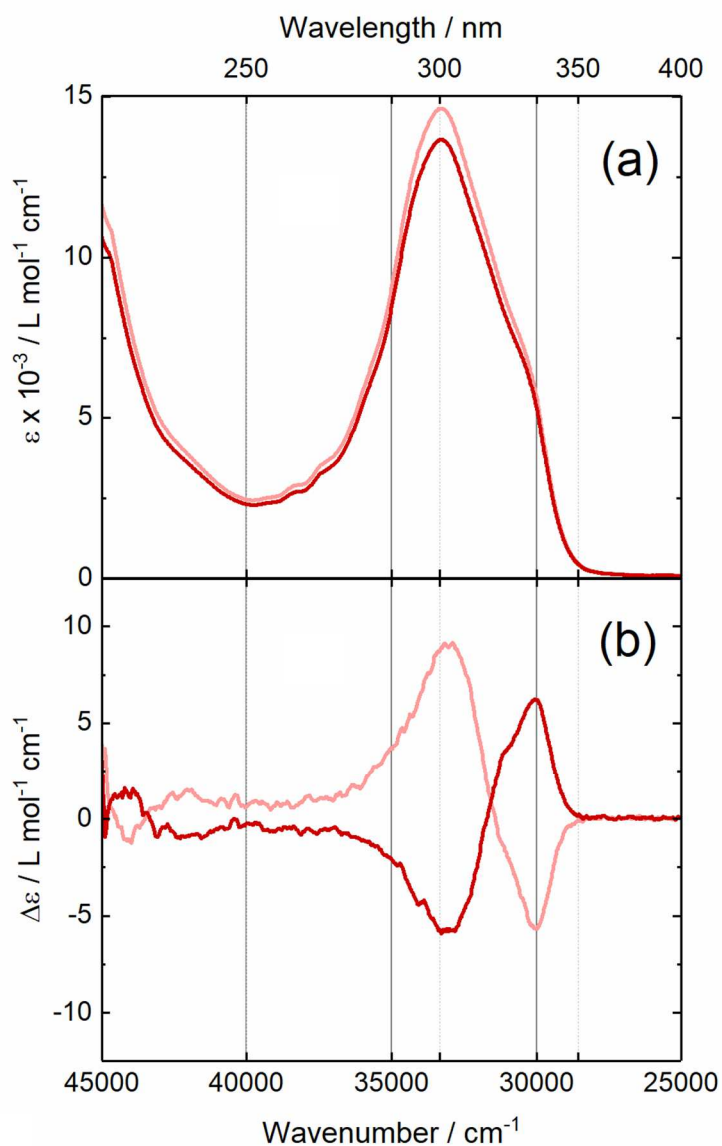


Figure S17. (a) UV-visible absorption and (b) ECD spectra in CH₂Cl₂ solution at C = 1·10⁻⁴ mol L⁻¹ and room temperature for [(*S*)-**3**] (light red line) and [(*R*)-**3**] (dark red line).

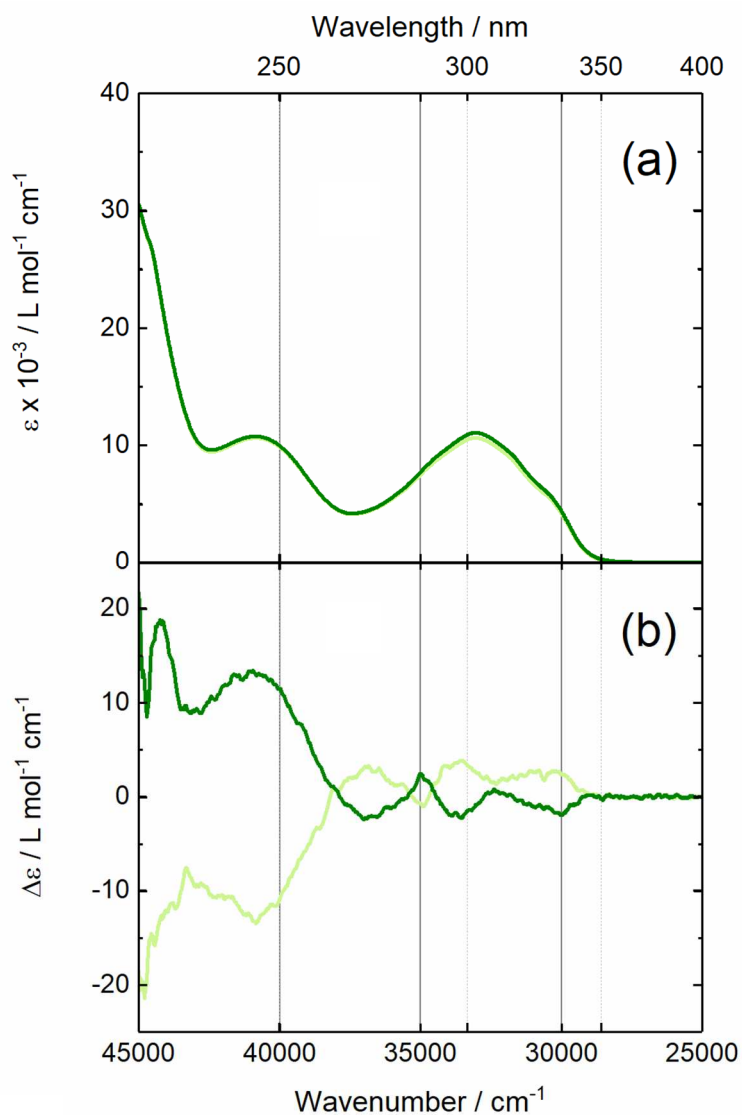


Figure S18. (a) UV-visible absorption and (b) ECD spectra in CH_2Cl_2 solution at $C = 5 \cdot 10^{-5}$ mol L^{-1} and room temperature for [(*S,S,S*)-**4**] (light green line) and [(*R,R,R*)-**4**] (dark green line).

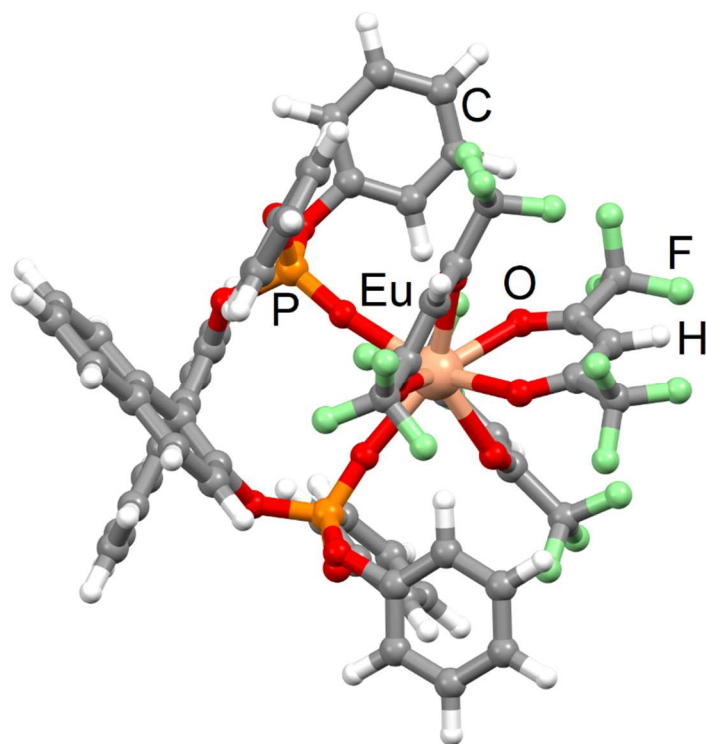


Figure S19. DFT optimized structure of [(*S*)-1] expected in CH₂Cl₂ solution.

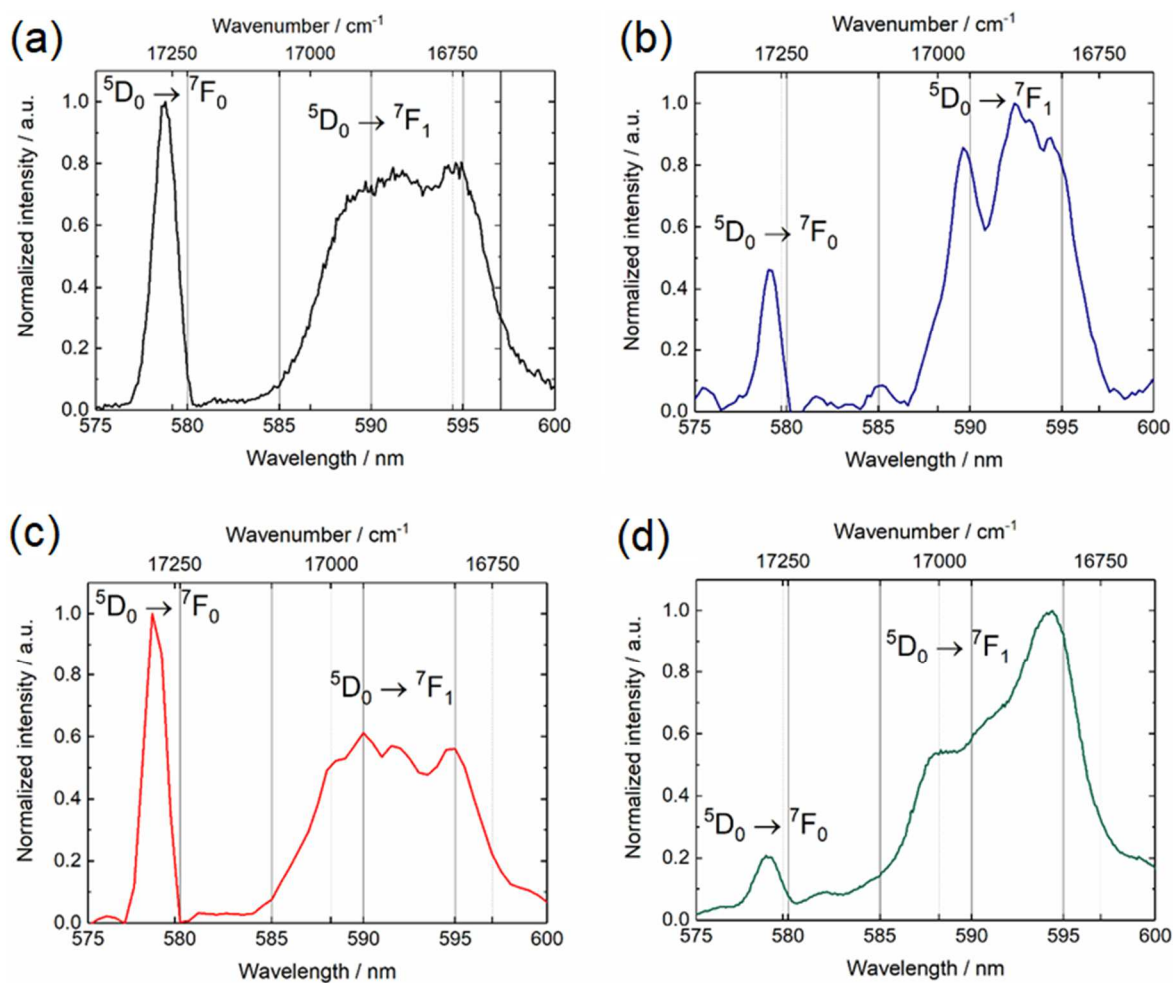


Figure S20. Emission spectra of the $^5D_0 \rightarrow ^7F_J$ ($J = 0, 1$) transitions measured in CH_2Cl_2 solution under 335 nm (29851 cm^{-1}) irradiation at room temperature for $[(S/R)\text{-1}]$ (a), $[(S,S,S/R,R,R)\text{-2}]$ (b), $[(S/R)\text{-3}]$ (c) and $[(S,S,S/R,R,R)\text{-4}]$ (d).

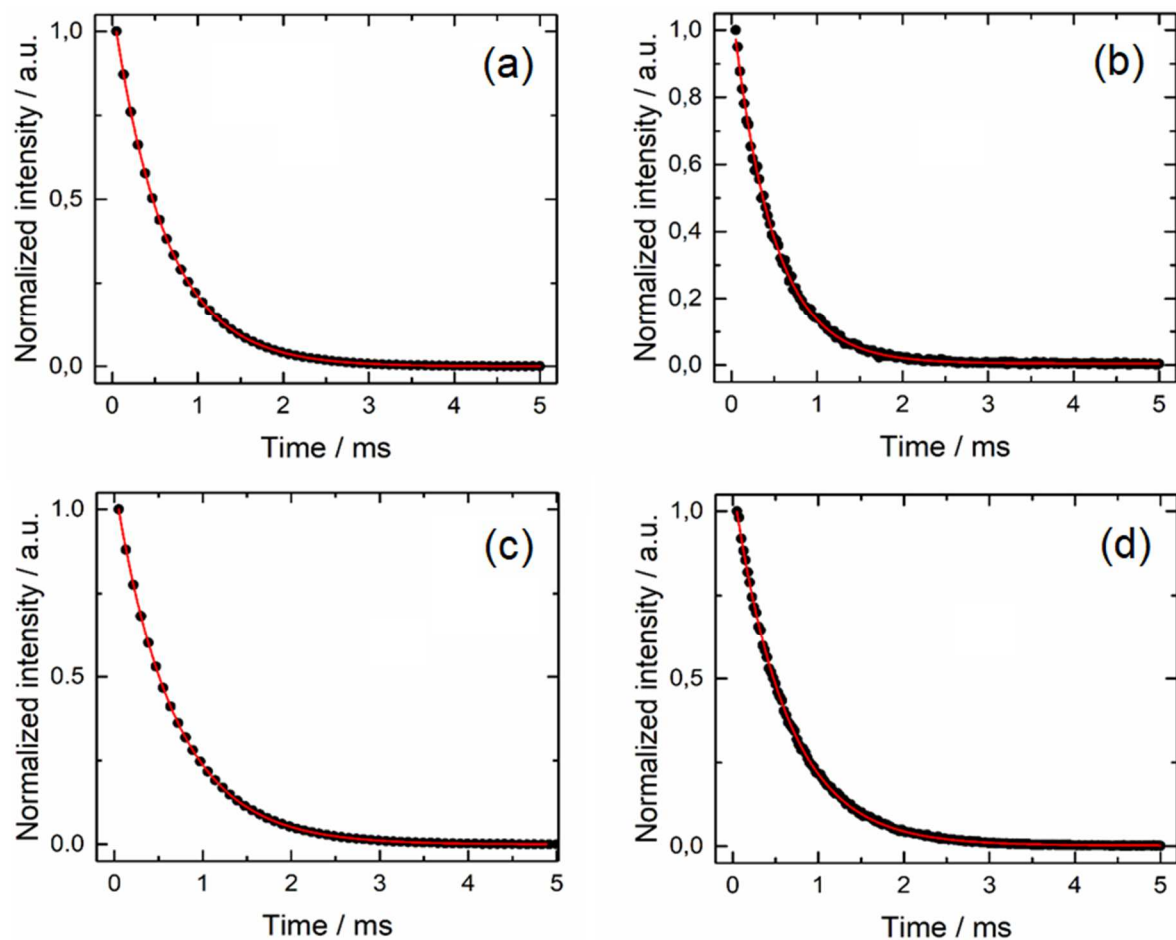


Figure S21. Emission lifetime decays (black dots) and mono-exponential fit curves (red line) for [(*S/R*)-1] (a), [(*S,S,S/R,R,R*)-2] (b), [(*S/R*)-3] (c) and [(*S,S,S/R,R,R*)-4] (d) in CH₂Cl₂ solution excited at 330 nm (30303 cm⁻¹).

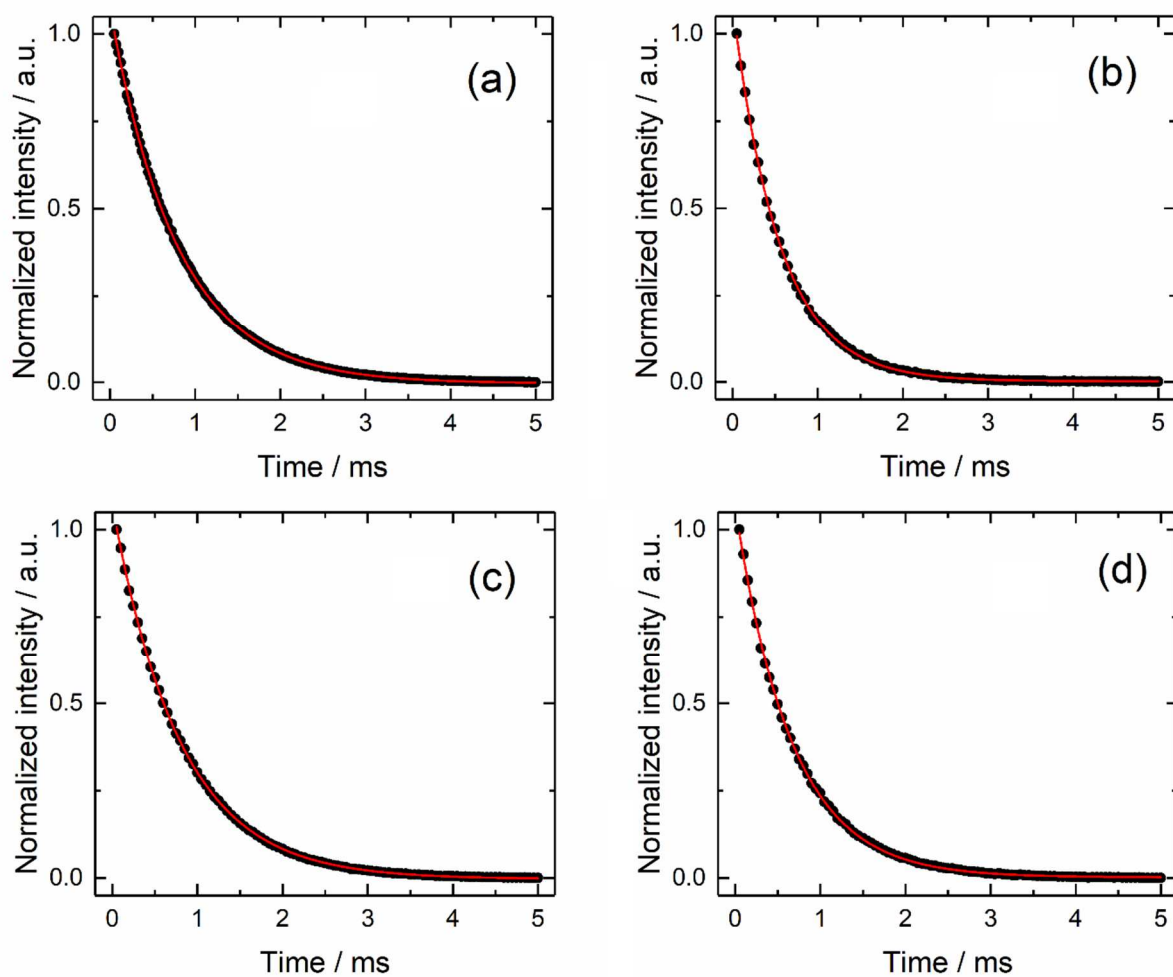


Figure S22. Emission lifetime decays (black dots) and mono-exponential fit curves (red line) for [(*S/R*)-1]_n (a), [(*S,S,S/R,R,R*)-2]_n (b), [(*S/R*)-3]_n (c) and [(*S,S,S/R,R,R*)-4]_n (d) in solid-state excited at 330 nm (30303 cm⁻¹).

Table S1. Cell parameters for $[(S,S,S)\text{-4}]_n$

Compound	$\{[\text{Eu}(\text{hfac})_3((S,S,S)\text{-L}^4)]\}_n$ $[(S,S,S)\text{-4}]_n$
Formula	$\text{C}_{75}\text{H}_{63}\text{EuF}_{18}\text{O}_{14}\text{P}_2$
M / g.mol ⁻¹	1533
Crystal system	Orthorhombic
Cell parameters	a = 18.853(25) Å
	b = 18.853 Å
	c = 26.512(30) Å
Volume / Å ³	9422.9(227)
T / K	150(2)

Table S2. X-ray crystallographic data for $\{[\text{Dy}(\text{hfac})_3((S)\text{-L}^3)]_3\}_n$.

Compound	$\{[\text{Dy}(\text{hfac})_3((S)\text{-L}^3)]_3\}_n$
Formula	$\text{C}_{177}\text{H}_{129}\text{Dy}_3\text{F}_{54}\text{O}_{42}\text{P}_6$
M / g.mol ⁻¹	4627.11
Crystal system	Orthorhombic
Space group	$\text{P2}_1\text{2}_1\text{2}_1$ (N°19)
Cell parameters	a = 11.9074(5) Å
	b = 36.1296(16) Å
	c = 43.7830(18) Å
Volume / Å ³	18835.9(14)
Z	4
T / K	150(2)
2θ range / °	$4.20 \leq 2\theta \leq 55.17$
$\rho_{\text{calc}} / \text{g.cm}^{-3}$	1.632
μ / mm^{-1}	1.357
Number of reflections	177762
Independent reflections	43311
R _{int}	0.0364
$\text{Fo}^2 > 2\sigma(\text{Fo})^2$	39888
Number of variables	2376
R ₁ , ωR ₂	0.0546, 0.1198

Table S3. SHAPE analysis of the coordination polyhedra around the lanthanide in the polymeric compound $\{[\text{Dy}(\text{hfac})_3((S)\text{-L}^3)]_3\}_n$.

Compounds	Metal	CShM _{SAPR-8} (square antiprism D _{4d})	CShM _{BTPR-8} (biaugmented trigonal prism C _{2v})	CShM _{TDD-8} (triangular dodecahedron D _{2d})
$\{[\text{Dy}(\text{hfac})_3((S)\text{-L}^3)]_3\}_n$	Dy1	0.464	2.745	1.987
	Dy2	1.133	1.923	1.174
	Dy3	0.306	1.623	1.573

Table S4. Selected dissymmetry factors g_{lum} for [(S/R)-1] in CH₂Cl₂ solution at room temperature.

[(S)-1]		[(R)-1]	
wavelength / nm	g_{lum}	wavelength / nm	g_{lum}
587.4	6.4E-03	587.4	-5.2E-03
591.4	4.6E-02	591.4	-4.7E-02
596.0	2.7E-02	596.0	-2.7E-02
611.4	-1.2E-03	611.5	1.7E-03
612.8	-7.9E-04	612.9	7.9E-04
614.5	2.1E-03	614.2	-1.6E-03
619.3	-2.6E-03	619.2	2.1E-03
624.1	-7.4E-03	624.0	6.9E-03
649.5	-2.0E-02	649.5	1.9E-02
652.3	1.6E-02	652.3	-1.5E-02
655.3	6.9E-03	655.3	-5.5E-03
690.1	2.2E-02	690.1	-1.9E-02
696.3	-1.4E-02	696.3	1.7E-02
700.6	7.1E-03	700.6	-6.5E-03
705.4	1.2E-02	705.4	-4.8E-03

Table S5. Selected dissymmetry factors g_{lum} for [(S,S,S/R,R,R)-2] in CH₂Cl₂ solution at room temperature.

[(S,S,S)-2]		[(R,R,R)-2]	
wavelength / nm	g_{lum}	wavelength / nm	g_{lum}
587.0	6.3E-02	587.0	-6.3E-02
594.9	1.2E-01	594.9	-1.1E-01
613.1	-7.1E-03	613.1	7.9E-03
616.1	-1.9E-03	616.1	1.7E-03
619.7	4.4E-03	619.7	-3.0E-03
622.2	-5.4E-03	622.8	8.1E-03
650.0	-1.8E-02	650.0	1.9E-02
651.9	2.5E-02	651.9	-1.9E-02
655.5	5.9E-03	655.5	-7.5E-03
692.0	-9.0E-03	692.0	2.8E-03
696.6	5.1E-03	696.6	-8.5E-03
700.2	-4.5E-03	700.2	4.1E-03

Table S6. Selected dissymmetry factors g_{lum} for [(S/R)-3] in CH₂Cl₂ solution at room temperature.

[(S)-3]		[(R)-3]	
wavelength / nm	g_{lum}	wavelength / nm	g_{lum}
590.0	-5.4E-03	590.0	4.7E-03
592.0	2.8E-02	592.2	-2.8E-02
594.0	-8.8E-03	594.0	9.3E-03
596.4	1.8E-02	596.4	-1.8E-02
611.6	-7.4E-04	611.8	6.4E-04
614.8	1.7E-03	615.0	-2.3E-03
620.4	-2.7E-03	620.6	2.5E-03
624.8	-7.6E-03	625.0	6.8E-03
650.0	-1.4E-02	650.6	1.9E-02
654.0	9.4E-03	654.0	-6.8E-03
691.0	1.6E-02	691.2	-1.8E-02
695.5	-1.4E-02	697.4	1.6E-02
701.0	4.6E-03	701.4	-6.5E-03

Table S7. Selected dissymmetry factors g_{lum} for [(S,S,S/R,R,R)-4] in CH₂Cl₂ solution at room temperature.

[(S,S,S)-4]		[(R,R,R)-4]	
wavelength / nm	g_{lum}	wavelength / nm	g_{lum}
587.0	2.3E-02	587.0	-2.0E-02
589.7	1.8E-02	589.7	-2.2E-02
594.3	5.3E-02	594.3	-5.3E-02
611.9	-2.7E-03	611.9	3.2E-03
650.9	-7.0E-03	650.9	1.2E-02
652.5	1.6E-02	652.5	-1.4E-02
655.5	7.1E-03	655.5	-2.0E-03
690.5	1.5E-04	690.5	4.3E-03
695.9	7.5E-03	695.9	-8.2E-03
699.0	-3.5E-04	699.0	5.1E-03

Table S8. Experimental lifetime (τ_{obs}) for the europium compounds in solid-state excited at 330 nm (30303 cm⁻¹).

Compounds	τ_{obs} (ms)
	(S) or (S,S,S)
[1]	0.78
[2]	0.55
[3]	0.80
[4]	0.66

Table S9. Selected dissymmetry factors g_{lum} for [(S/R)-1]_n in solid-state at room temperature.

[(S)-1] _n		[(R)-1] _n	
wavelength / nm	g_{lum}	wavelength / nm	g_{lum}
587.4	5.7E-03	587.4	-5.7E-03
594.7	2.6E-02	594.7	-2.2E-02
613.4	-4.9E-03	613.4	4.6E-03
617.1	-1.4E-03	617.1	1.1E-03
621.3	5.1E-03	621.3	-4.7E-03
649.5	-1.9E-02	649.5	2.1E-02
651.3	2.9E-03	651.3	-2.9E-03
652.9	-8.5E-03	652.9	1.3E-02
655.1	7.0E-03	655.1	-5.3E-03
690.9	4.3E-03	690.9	-3.6E-03
695.7	-4.3E-03	695.7	4.1E-03
698.7	5.1E-03	698.7	-5.0E-03
701.4	-6.7E-04	701.4	5.3E-03

Table S10. Selected dissymmetry factors g_{lum} for [(S,S,S/R,R,R)-2]_n in solid-state at room temperature.

[(S,S,S)-2] _n		[(R,R,R)-2] _n	
wavelength / nm	g_{lum}	wavelength / nm	g_{lum}
587.2	-5.5E-03	586.4	6.4E-03
595.3	-1.3E-02	594.6	1.8E-02
610.5	1.7E-03	613.6	-3.4E-03
612.8	2.1E-03	612.8	-1.8E-03
615.8	1.2E-03	615.3	-1.8E-03
621.5	-2.3E-03	621.3	3.6E-03
648.9	7.5E-03	649.4	-1.6E-02

Table S11. Selected dissymmetry factors g_{lum} for [(S/R)-3]_n in solid-state at room temperature.

[(S)-3] _n		[(R)-3] _n	
wavelength / nm	g_{lum}	wavelength / nm	g_{lum}
587.2	-3.8E-03	587.5	3.5E-03
591.8	2.8E-02	591.5	-2.8E-02
595.4	-1.6E-02	595.4	1.3E-02
612.5	8.9E-04	612.5	-8.9E-04
614.0	-1.4E-03	613.8	1.1E-03
616.8	1.3E-03	617.0	-8.5E-04
621.5	-5.8E-04	621.0	5.6E-04
623.0	1.5E-03	623.3	-9.0E-04
649.6	7.6E-03	649.6	-7.9E-03
651.5	-4.7E-03	651.5	4.6E-03

Table S12. Selected dissymmetry factors g_{lum} for [(S,S,S/R,R,R)-4]_n in solid-state at room temperature.

[(S,S,S)-4] _n		[(R,R,R)-4] _n	
wavelength / nm	g_{lum}	wavelength / nm	g_{lum}
587.0	-1.5E-02	587.0	1.3E-02
590.8	9.5E-03	590.8	-9.1E-03
594.7	-2.5E-02	594.7	2.3E-02
612.9	4.9E-03	612.9	-2.9E-03
617.5	3.1E-03	617.5	-1.7E-03
621.5	-4.1E-03	621.5	4.6E-03
649.3	1.7E-02	649.3	-1.3E-02
651.5	-4.2E-03	651.5	2.9E-03
655.5	2.4E-03	655.5	-2.6E-04

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