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## ARTICLE

# Sol-Gel Biotemplating Route with Cellulose Nanocrystals to Design Photocatalyst Boosting the Hydrogen Generation

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Light harvesting capability and charge carriers lifetime play critical roles in determining the photoefficiency of a photocatalyst. Herein, a one-pot method is proposed to design mesostructured TiO<sub>2</sub> materials by taking advantage of the ability of cellulose nanocrystals (CNC) to self-assemble into chiral nematic structures during solvent evaporation. After the xerogel formation, the as-obtained CNC/TiO<sub>2</sub> hybrid films exhibit a chiral nematic structure and tunable Bragg peak reflection, generating lamellar TiO<sub>2</sub> mesostructure after the biotemplate removal by calcination. More prominently, this straightforward method can be extended to couple TiO<sub>2</sub> with other metallic nanoparticles, improving the light-harvesting and charge carriers separation of these photocatalysts, in particular for boosting hydrogen generation. This foolproof approach opens new doors for the development of nanostructured materials for solar energy conversion and catalysis.

## 1. Introduction

Known as the bottleneck for the solar energy conversion efficiency, the capability of light harvesting and the separation of photogenerated charge carriers have attracted great interest.<sup>1</sup> Structuring photoactive materials into photonic crystals or hierarchical structures are promising approaches for improving the efficiency of photoactive materials. A photonic crystal can be used either as a reflector element separated to the photoactive layer for light accumulation,<sup>2-4</sup> or as a photoactive absorber to slow down the velocity of light.<sup>5,6</sup> Hierarchical structures, like in green leaves, can improve light scattering for optimal light management toward an efficient conversion of photons into chemical energy.<sup>7</sup> In this context, structuring inorganic materials at the mesoscale using self-biotemplate assembly process appears as a sustainable method for the synthesis of functional material. Notably, cellulose nanocrystals (CNC, as a biotemplate) have the specificity of forming a chiral nematic (CN) structure, which can be replicated into inorganic materials, a structure never achieved before with classical aqueous sol-gel approaches using surfactants as templates.<sup>8,9</sup> Freestanding silica films with CN structure were obtained by the polycondensation of tetraethyl orthosilane or methyl triethoxysilane precursors dissolved in CNC aqueous

solution.<sup>10,11</sup> This breakthrough has generated promises for the development of functional materials and the design of novel photonic crystals. However, the method is restricted only to mesoporous silica or organosilica films, which limits the application to the sensing and separation of chiral molecules.<sup>12,13</sup> This limitation stems from the high sensitivity of the other precursors (metal chloride or alkoxides) to moisture leading to their fast hydrolysis and condensation, disorganizing the stability of chiral nematic structure of CNC into isotropic phase. Acid catalysis in sol-gel process is one way for stabilizing titania precursors in an aqueous solution containing CNC by promoting slow polycondensation but with the expense of limited hydrolysis.<sup>14-16</sup> These infructuous attempts turn to colorless mesoporous TiO<sub>2</sub> film where the CN structures of CNC are completely lost. The CN organization of CNC is very sensitive to the solvent polarity, ionic strength and pH of the aqueous solution.<sup>17</sup> Other alternatives, such as hard templating or impregnation methods, have been proposed to transfer the helical structure into titanium dioxide films.<sup>18-20</sup> Despite their efficiency and versatility, all these methods require multistep impregnation/drying to get the final material. Recently, a successful self-assembly of CNC/TiO<sub>2</sub> was proposed.<sup>21</sup> The procedure requires complex steps to dissolve TiO<sub>2</sub> nanoparticles and use sodium titanate precursor (unsuitable for photocatalysis or dye sensitized solar cells) to elaborate a water-soluble peroxotitanate in CNC aqueous solution. The remaining sodium ion is known to have a detrimental effect on the photoefficiency of TiO<sub>2</sub>, as it acts as a recombination center for the photogenerated charge carriers.<sup>22,23</sup> Clearly, it is challenging and highly desirable to develop an effective yet simple one-pot sol-gel self-biotemplating approach that can be extended to TiO<sub>2</sub> and other functional metal oxides.

Here, we propose a straightforward one-pot method enabling both the structural design of TiO<sub>2</sub> film and the coupling

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with metallic nanoparticles to improve the photon conversion efficiency. The one-pot self-biotemplating sol-gel approach enables the synthesis of CNC/TiO<sub>2</sub> and CNC/TiO<sub>2</sub>-Metal (loaded metal oxide) hybrid films with tunable Bragg peak reflection. A stable mixture of titanium diisopropoxide bis(acetylacetonate) (TAA) and an aqueous solution of CNC were used for the elaboration of the films. After removing the CNC biotemplate, the final TiO<sub>2</sub> films exhibit a birefringent lamellar structure producing an increase of the photogenerated charge carriers lifetime. The assessment of charge carriers lifetime and the photocatalytic activity for hydrogen production were found to be significantly improved, compared to commercial TiO<sub>2</sub>-P25 (employed here as an internal standard). To validate the reliability and universality of our method, the procedure was extended to TiO<sub>2</sub> loaded with different metallic nanoparticles (Cu, Ni, Bi, V) to increase the charge carriers separation, boosting the photocatalytic efficiency for hydrogen generation.

## 2. Experimental section

### 2.1 Synthesis of mesoporous lamellar TiO<sub>2</sub> films

A CNC aqueous suspension (10 mL, 4.0 wt.%, pH = 6.3) was sonicated for 1 h. Different volumes of TAA solutions (Table S1†) were added in 2 mL ethanol stirred for 10 min, then the ethanolic TAA solutions were mixed with sonicated CNC suspension. After a slight shaking and stirring for 1 h giving homogeneous bright yellow mixtures, hereafter, poured into 55 mm diameter polystyrene Petri dishes and dried under room temperature for about 48 h to obtain iridescent CNC/TiO<sub>2</sub> hybrid films. The resulting hybrid films were then calcined in air at a heating rate of 1 °C /min to 500 °C for 2 h and naturally cooled to room temperature yielding white mesoporous lamellar TiO<sub>2</sub> films.

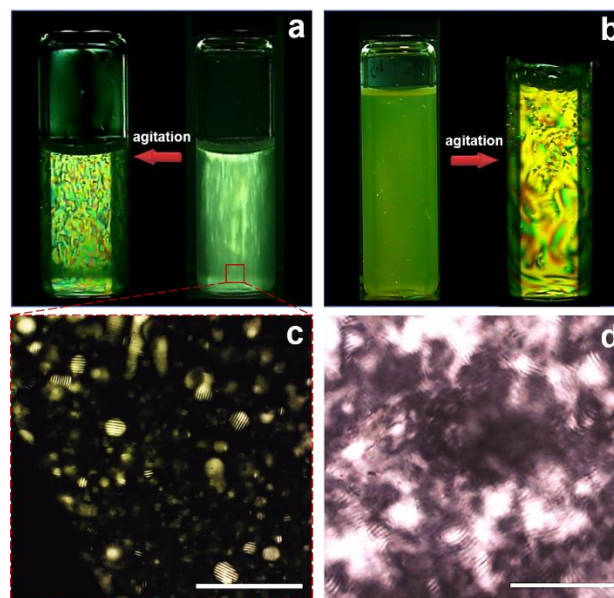
### 2.2 Synthesis of metals loaded lamellar TiO<sub>2</sub> films

A same CNC aqueous suspension was sonicated for 1 h. 1 mg of copper(II) acetate and nickel(II) acetylacetonate were added to 1 mL ethanol (bismuth(III) nitrate and vanadium(V) oxide were dissolved in deionized water) sonicated until clear metal oxide solutions were obtained. 0.2 mL TAA solution was added in 1 mL ethanol stirred for 10 min, then mixed with metal oxide solutions and stirred for another 10 min. Followed by mixing with sonicated CNC suspension and stirred for 1 h, the mixtures were poured into polystyrene Petri dishes to form the hybrid films. After the same calcination procedure such as-prepared lamellar TiO<sub>2</sub>, obtaining loaded metal oxide films.

## 3. Results and Discussion

The ester sulfated surface of CNC (CNC-OSO<sub>3</sub>H) ensures the stability of the nanorods by electrostatic repulsion in aqueous suspensions. The ability of the negatively charged CNC nanorods to form CN phases relies on several parameters, such as the concentration of sulfate groups, the volume fraction of CNC and the pH of the medium.<sup>24,25</sup> 4 wt. % CNC is the critical concentration corresponding to the onset of isotropic to chiral nematic phase transition.<sup>26</sup> The cross-polarized optical microscopy (POM) image of this solution is

shown in Fig. 1a. At this concentration, the suspension is weakly birefringent and tactoids (liquid crystalline droplets) with different sizes and shapes dispersed in a continuous isotropic phase have been observed (Fig. 1c). Both the CNC and hybrid TAA/CNC suspensions exhibit non-persistent strong shear-induced polychromatic birefringence between crossed polarizers, related to pretransitional effects (Fig. 1a and b).



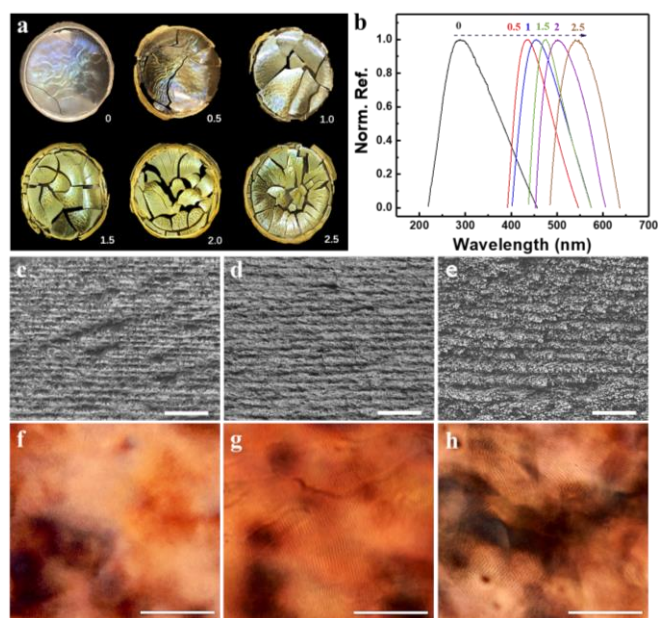
**Fig. 1** Photographs of (a) pure CNC and (b) TAA/CNC (weight ratio=0.5) suspensions observed between crossed polarizers before and during agitation (shear-induced polychromatic birefringence). Typical POM images of (c) CNC and (d) TAA/CNC suspensions during slow evaporation at room temperature (scale bar = 50 μm).

CNC/TiO<sub>2</sub> hybrid mesostructured films are prepared by evaporation induced self-assembly (EISA) of the solution containing the ethanolic solution of the Ti(acac)<sub>2</sub>OiPr<sub>2</sub> precursor (TAA) and CNC suspension (4 wt.%). The mixture is highly miscible after agitation, and a stable water-based TiO<sub>2</sub>-precursor solution (yellow and transparent) can easily be obtained (Fig. 1b). However, the solution does not show any birefringence, indicating isotropic phase transition due to the dilution induced by the addition of the precursor. In the TAA precursor, acetylacetonate acts as a chelating agent, which limits the precursors' reactivity towards hydrolysis and polycondensation, making the solution stable enough during the EISA process. The pH of the hybrid solution varies from 6.26 to 6.04 when the TAA ratio increases from 0.5 to 2.5, corresponding to the initial pH of the CNC suspension (pH=6.3). Droplets evaporation of the TAA/CNC hybrid suspensions observe using POM are presented in Fig. 1d and S1.† The formation of typical fingerprint textures clearly indicates the presence of a CN structure during the formation of CNC/TiO<sub>2</sub> hybrid films. This feature is confirmed by performing time-lapse photography of a TAA/CNC suspension during EISA process, showing the formation of an iridescent film with time due to the local ordering of CNC in CN structure (Fig. S2†). Hence, our straightforward one-pot method enables to elaborate CNC/TiO<sub>2</sub> photonic hybrid films with tunable colors as well as a variable maximum of Bragg peak reflection (from the UV to the entire visible range). Although the CNC films are flexible with limited cracks, the photonic hybrid films are likely to crack in fragments whose size is getting smaller as the concentration

of the TAA precursor increases. This is related to the water evaporation during EISA, which induces a shrinkage through the polycondensation of the titania precursor, generating capillary stress gradients within the films. The iridescent color rising from the films under natural light indicates the formation of the periodic structure reflecting a selective wavelength range. The color of the films turns from blue, green to yellow, in agreement with the red-shifted maximum wavelength of the Bragg reflection (Fig. 2a). The maximum wavelength reflected by the hybrid photonic films ( $\lambda_{max}$ ) is related to the effective refractive index " $n_{eff}$ ", the pitch " $P$ " and the angle of the film axis to the emitted light following the equation:<sup>27</sup>

$$\lambda_{max} = n_{eff} \cdot P \cdot \sin(\theta) \quad (1)$$

Increasing the TAA/CNC weight ratio confirms the red-shift of the Bragg peak reflection as evidenced by UV-vis-NIR spectroscopy (Fig. 2b). Since the refractive indexes of the pure titania xerogel and CNC remain unchanged ( $n = 1.9^{28}$  and  $1.51^{29}$ , respectively), the red-shift would probably rise from the increase of the pitch " $P$ ".

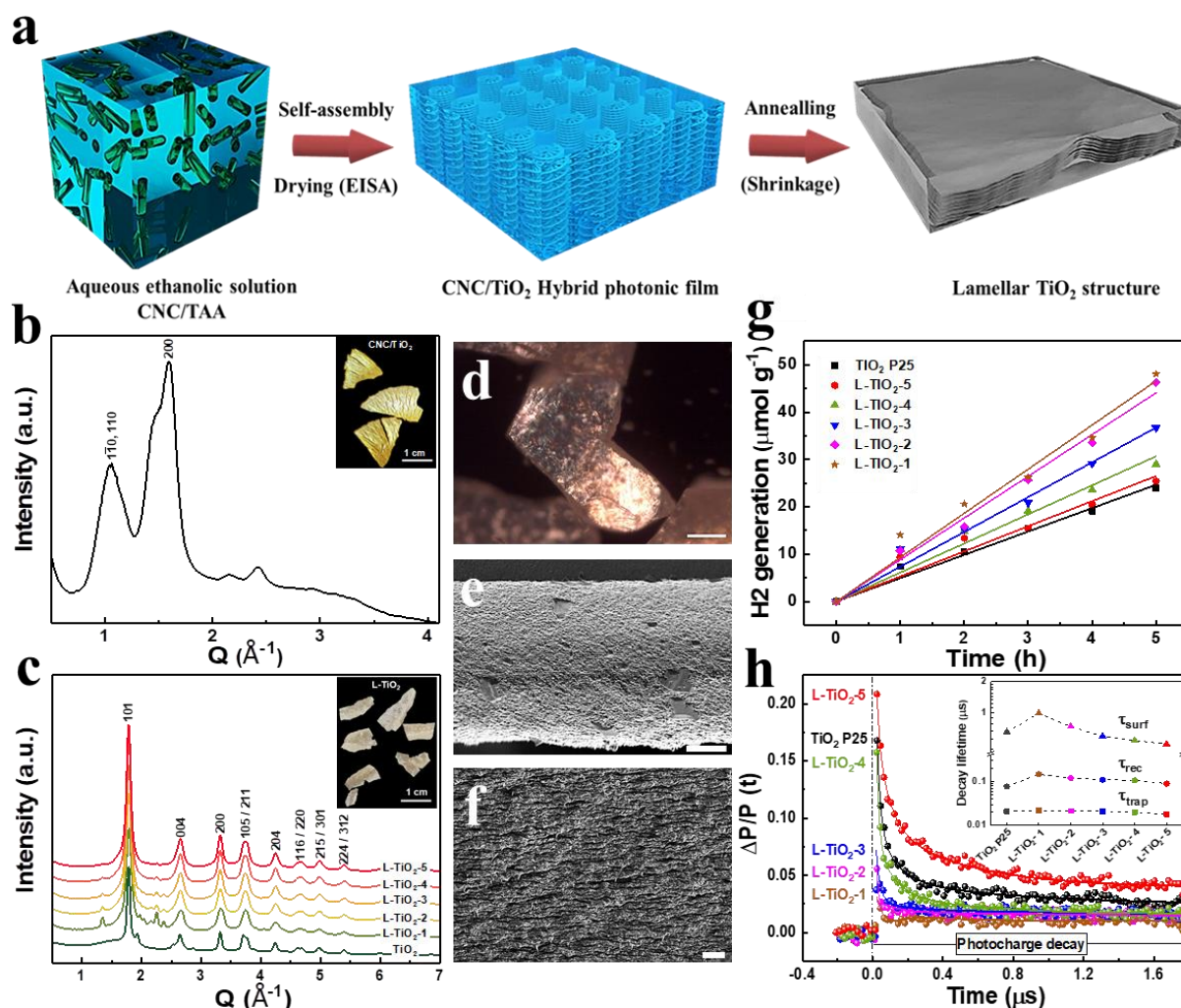


**Fig. 2** (a) Photographs and (b) UV-Vis spectra of CNC/TiO<sub>2</sub> hybrid films. SEM images of CNC/TiO<sub>2</sub> hybrid films: (c) weight ratio=0.5, (d) weight ratio=1.5, (e) weight ratio=2.5 (scale bar = 1 μm). POM images of CNC/TiO<sub>2</sub> hybrid films: (f) weight ratio=0.5, (g) weight ratio=1.5, (h) weight ratio=2.5 (scale bar = 50 μm).

The resulting CN structures in the CNC/TiO<sub>2</sub> hybrid films are further evidenced by the SEM imaging of the film cross-sections (Fig. 2c-e, Fig. S3a and b†). A twisted structure, periodically repeated along the film axis, is observed whatever the composition of the film. This is a typical feature of the chiral nematic organization, which the distance between two layers correspond to the half-helical pitch " $P/2$ " of the CN structure.<sup>30</sup> Values of half-helical pitch estimated from these SEM images are found to increase with the titanium loading, confirming the effect of the TAA/CNC ratio on the observed  $\lambda_{max}$  red-shift (Table S1†). Indeed, as suggested above, increasing of the  $P$  value, likely due to the increasing of titania wall thickness

(increase of the TAA concentration), induces a red-shift of the maximum Bragg peak reflection, in agreement with the UV-visible spectroscopic analysis (Fig. 2b). The photonic films have been further characterized using POM (Fig. 2f-h, Fig. S3c and d†). All the films show a strong birefringence but a random distribution of the dark and bright regions, suggesting a different orientation of the helical axis towards the up-coming light (random orientation of the liquid crystal domains). It should be noted the typical fingerprint textures are observed in POM images, which definitively confirm the preservation of the chiral nematic ordering in such as-prepared hybrid photonic films regardless of the titanium loading.

The inorganic TiO<sub>2</sub> films are obtained after removal of the CNC template by annealing the samples at 500 °C under air following the procedure, described in Fig. 3a. Calcination can effectively remove the CNC from the hybrid films, as evidenced by FTIR spectroscopy (only the band corresponding to Ti-O bond is observed, Fig. S4†) producing mesoporous TiO<sub>2</sub> films as well as ensuring the crystallisation of TiO<sub>2</sub> into its photoactive anatase phase. Nitrogen physisorption measurements confirm the mesoporous structure of the calcined films. Isotherms are reversible and the type IV characteristic hysteresis loop (obvious for higher TAA concentration) suggests interconnected pores with a cylindrical shape (Fig. S5a†). A well-defined progressive uptake measure at low relative pressure ( $p/p_0=0.05-0.40$ ) is characteristic of micropores, followed by a sudden increase due to the filling of the pore in the mesoporous range. The BET specific surface area reaches 64.5 - 72.6 m<sup>2</sup> g<sup>-1</sup> and the pore volume ranges from 0.35 to 0.14 cm<sup>3</sup> g<sup>-1</sup>. The narrower pore size distribution implies identical pore for all the films, with an average pore diameter of 10 nm, a value in agreement with the CNCs' width (Table S2 and Fig. S5b†).<sup>14</sup> Wide-angle X-ray scattering (WAXS) patterns performed on hybrid photonic films before calcination display several peaks at  $Q \sim 1.1 \text{ \AA}^{-1}$  [(1 $\bar{1}$ 0), (110)] and  $1.6 \text{ \AA}^{-1}$  (200), characteristics of crystalline form of CNC (Fig. 3b).<sup>31</sup> After calcination, these peaks are no longer present in the calcined films but instead, we can observe the occurrence of broad diffraction peaks corresponding to nanosized TiO<sub>2</sub> crystallites with anatase form (Fig. 3c). The calcined films observe by POM present in Fig. 3d show remarkable birefringent textures implying a preserved anisotropic structure (Fig. S6†). In agreement with this statement, SEM images of the cross-sections of TiO<sub>2</sub> films reveal a transfer of the alignment layers to mesostructured lamellar TiO<sub>2</sub> (referenced as L-TiO<sub>2</sub> hereafter) with long-range order (Fig. 3e and f). The layers parallel to the substrate are aligned predominantly with highly anisotropic structural features and the thickness of the layer increase as the TAA loading. The replica of the CN structure could not be confirmed from the SEM images, probably due to the crystal growth and sintering of the TiO<sub>2</sub>, or the movement of atoms occurring during the calcination step lead to the change of the chiral nematic structure to 1D lamellar one. These results evidence the change of the CN structure in CNC/TiO<sub>2</sub> hybrid films to inorganic TiO<sub>2</sub> films with a lamellar mesostructure.



**Fig. 3** (a) Schematic illustration of the fabrication of a lamellar  $\text{TiO}_2$  structure ( $\text{L-TiO}_2$ ) after removal of the biotemplate by calcination. WAXS pattern of (b)  $\text{CNC}/\text{TiO}_2$  hybrid film and (c)  $\text{L-TiO}_2$  films. (d) POM image of  $\text{L-TiO}_2$ -5 film showing strong birefringence (scale bar = 100  $\mu\text{m}$ ). SEM images of (e)  $\text{L-TiO}_2$ -2 and (f)  $\text{L-TiO}_2$ -5 films (scale bar = 1  $\mu\text{m}$ ). (g) Hydrogen generation and (h) TRMC signals of  $\text{L-TiO}_2$  films.

The photocatalytic activity of the periodic long-range  $\text{L-TiO}_2$  films is evaluated for hydrogen generation. The  $\text{L-TiO}_2$  films show higher photoefficiency compared to standard  $\text{TiO}_2$ -P25 whatever the TAA/CNC ratio of the films (Fig. 3g). The  $\text{H}_2$  generation rate of  $\text{L-TiO}_2$ -1 is 287.2  $\mu\text{mol h}^{-1} \text{g}^{-1}$ , as nearly twice as high as  $\text{TiO}_2$ -P25 (153.1  $\mu\text{mol h}^{-1} \text{g}^{-1}$ ). This result indicates that structuring the  $\text{TiO}_2$  material in a periodic long-range lamellar structure can considerably improve its photoefficiency for hydrogen generation. Interestingly, the TAA/CNC ratio of the starting material is inversely related to the photocatalytic efficiency of hydrogen generation. Increasing the TAA/CNC ratio leads to higher photogenerated charge carriers density, which is attributed to the increase in wall thickness of the layers.

Consideration should be given to the interaction of light with these structured photocatalysts to understand deeply the relationship underlying between these results. Indeed, the light reflection in periodically structured material is reported to be an efficient strategy to improve the photon-to-charge carriers production.<sup>18</sup> For instance, the photon velocity is slowing down significantly at the photonic band gap edges (blue and red) in a photonic structure, increasing the absorbance light factor of  $\text{TiO}_2$ .<sup>32</sup> Bioinspired porous  $\text{TiO}_2$  material with a replica of chloroplast structure obtained from plant leaves shows enhanced light harvesting capability.<sup>7, 33</sup>

The electronic properties of variable  $\text{L-TiO}_2$  films are assessed by using time-resolved microwave conductivity

(TRMC). The TRMC is a non-destructive and a contactless technique that enables to track the dynamics, the density and the lifetime of the photogenerated charge carriers on such periodic long-range lamellar mesostructured TiO<sub>2</sub> films.<sup>34,35</sup> After illumination, the TRMC signals rise following ultrafast charge separation process (<200 fs),<sup>36</sup> which is mainly due to mobile electrons at the surface of TiO<sub>2</sub>, since it is usually assumed that holes with a limited mobility remain on the bulk. The time-dependent change of microwave power transmission  $\Delta P(t)/P$  is proportional to the photoconductivity  $\Delta\sigma(t)$  variation and therefore proportional to the density of mobile charge carriers, which can be expressed as:<sup>37,38</sup>

$$\Delta P(t)/P(t) = A \cdot \Delta\sigma(t) = A \cdot \sum \Delta n_i(t) \mu_i \quad (2)$$

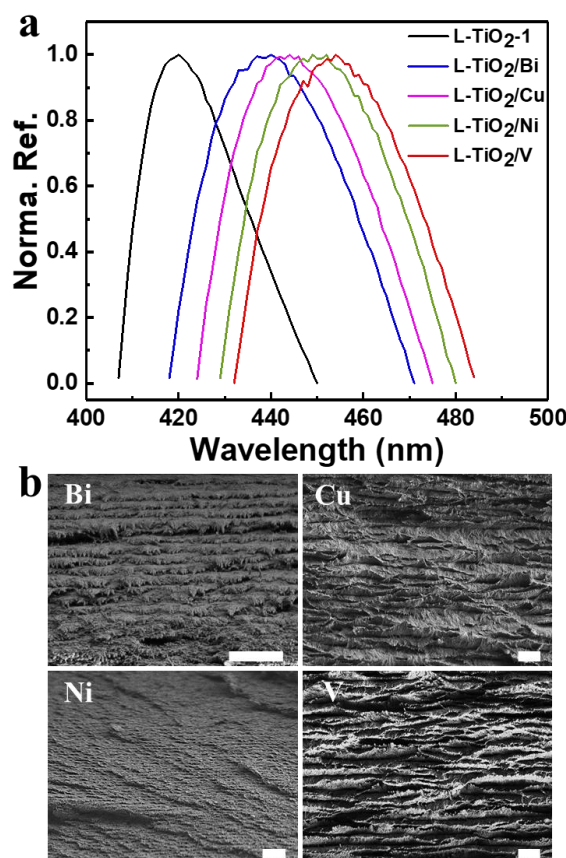
where  $\Delta n_i(t)$  defines the density of photogenerated charge carriers "i" at known time "t", and  $\mu_i$  defines their mobility. In the present study, all TRMC measurements are recorded under 360 nm UV illumination, and with a light energy density received by the samples set at 1.4 mJ cm<sup>-2</sup>. According to the TRMC results, the photogenerated charge carriers associated with the TRMC signal gradually raises as the TAA/CNC ratio increases (Fig. 3h). Assuming an equal number of absorbed photons for all samples, we surmise that the improvement in the density of the photogenerated charge carriers may be related to the light scattering in the long-range lamellar TiO<sub>2</sub> structure. As depicted in Fig. 3e and f, the lamellar structure could be defined as alternating TiO<sub>2</sub> layers separated by voids, having a variable refractive index. The thickness of the TiO<sub>2</sub> layers increases with the ratio of TAA/CNC, which improves the confinement time of photons as well as their reflection. Therefore, the light propagation in such lamellar mesostructure improves the light harvesting and consequently the population of photogenerated charge carriers. TRMC results show an increase in the signal amplitude (density of charge carriers e<sup>-</sup>/h<sup>+</sup>) with the TAA/CNC ratio, corresponding to an inverse correlation of H<sub>2</sub> production efficiency. For example, L-TiO<sub>2</sub>-1 displays twice the hydrogen production efficiency of L-TiO<sub>2</sub>-5, while the population of photogenerated charge carriers are ten times less than L-TiO<sub>2</sub>-5. This result suggests that the density of charge carriers is probably not the limiting parameter for the H<sub>2</sub> generation.

Thus, we turn to investigate the TRMC signals' decay, which typically occurs after the e<sup>-</sup>/h<sup>+</sup> production during the 8 ns laser pulse, giving useful informations about the photocharge lifetime (Fig. 3h). The decay of the TRMC signals of photocharges can be divided into three decay stages: charge trapping ( $\tau_{trap}$ ), charge recombination ( $\tau_{rec}$ ) and surface reactions ( $\tau_{surf}$ ). The capture time of electrons and holes occurs in less than 30 ps, and it is accompanied by rapid recombination.<sup>39,40</sup> Both of the trapping and recombination phenomena are considered as the main processes that can explain the TRMC signals' decay.<sup>41,42</sup> The time constant of electrons trapping and recombination can be quantitatively estimated from the signal decay using the following equation:<sup>37</sup>

$$\Delta P(t)/P(t) = \sum A_n \cdot \exp(-t/\tau_n) \quad (3)$$

where  $A_n$  and  $\tau_n$  are proportionality and time constants of each of the three decay processes, respectively. The

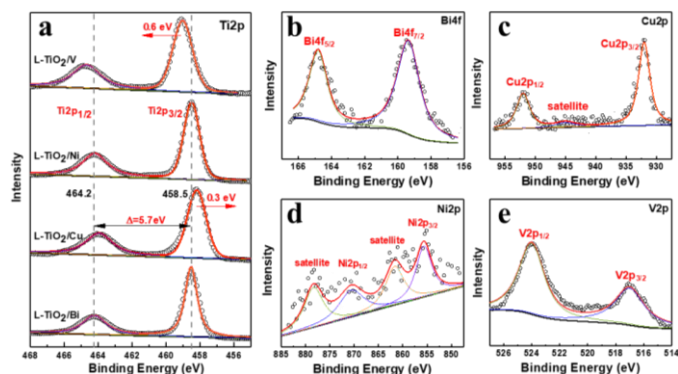
photocharge lifetime presents in the inset of Fig. 3h is found to be correlated with the wall thickness in the lamellar structure. The low charge carrier number produced at the surface of the L-TiO<sub>2</sub>-1 sample has the longest lifetime, which could be translated by low trapping and recombination yield. This trend evidences that L-TiO<sub>2</sub>-1 synthesized at a lower TAA/CNC ratio shows weak electron trapping capability, thereby enhancing the proportion of active charges for H<sub>2</sub> generation. The interfacial charge recombination is likely to occur in the boundaries (interfaces) of the long-range layers with thicker walls where defects could be localized.



**Fig. 4** (a) UV-Vis spectra and (b) SEM images of CNC/loaded metal oxide hybrid films (scale bar = 1 μm).

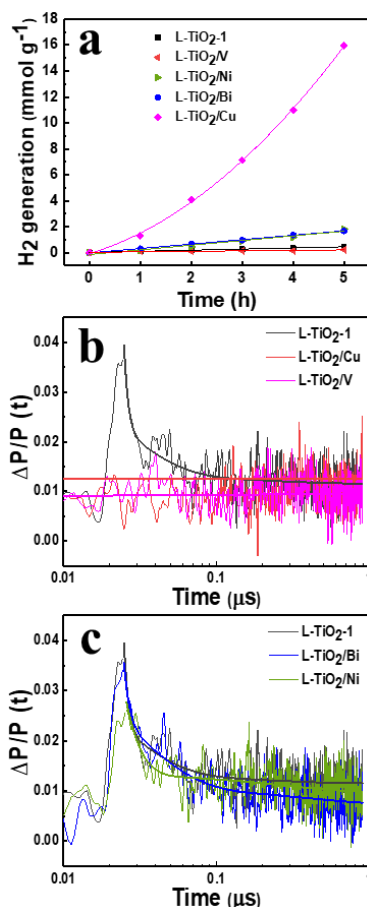
Owing to the interesting photoactivity of L-TiO<sub>2</sub>-1 for hydrogen generation, several coupling schemes (Cu, Ni, Bi, V) have been performed to improve the photoefficiency yield even further. Metal loaded to TiO<sub>2</sub> can trap electrons, which impedes the recombination of e<sup>-</sup>/h<sup>+</sup> resulting in a significantly enhance of photocatalytic activity.<sup>43,44</sup> The solubility of the metals precursors combine to the one-pot self-assembly method enables to elaborate biotemplated modified metal oxides. The photonic colors rising from the light reflection at the surface of hybrid films (CNC/loaded metal oxides) do not show a notable variation (Fig. S7<sup>†</sup>), which is consistent with the slight red-shift in Bragg peak reflection (Fig. 4a). SEM images evidence of the remaining of the twisted periodic structure, as already observed for the CNC/TiO<sub>2</sub> hybrid films (Fig. 4b). This shows for the first

time that the proposed one-pot self-assembly method could be extended to the elaboration of metallic nanoparticles loaded to metal oxide ( $\text{TiO}_2$ ) with mesostructured architecture.



**Fig. 5** (a) XPS spectra of  $\text{Ti}2p$  of different loaded metal oxides. XPS spectra of (b)  $\text{L-TiO}_2/\text{Bi}$ , (c)  $\text{L-TiO}_2/\text{Cu}$ , (d)  $\text{L-TiO}_2/\text{Ni}$  and (e)  $\text{L-TiO}_2/\text{V}$ .

The compositions of the metals loaded lamellar  $\text{TiO}_2$  films are confirmed by energy-dispersive X-ray spectroscopy (EDS) analysis (Fig. S8<sup>†</sup>) and the surface of these materials is further analyzed by X-ray photoelectron spectroscopy (XPS). The general XPS spectra of the loaded films show detection of carbon and oxygen elements (Fig. S9a and b<sup>†</sup>). The films show  $\text{Ti}2p$  peak deconvoluted into two curves, which can be assigned to the spin-orbit components  $2p_{3/2}$  and  $2p_{1/2}$ , respectively. The separations between the  $\text{Ti}2p_{3/2}$  and  $\text{Ti}2p_{1/2}$  peaks are measured to be 5.7 eV for all the films, in agreement with the binding energy (BE) separation observed for anatase  $\text{TiO}_2$  (Fig. 5a).<sup>45,46</sup> XPS spectra of the target metallic nanoparticles are presented in Fig. 5 b-e. In the case of bismuth loaded with  $\text{L-TiO}_2$ -1, symmetric peaks of  $\text{Bi}4f$  state are observed. The  $\text{Bi}4f_{7/2}$  and  $\text{Bi}4f_{5/2}$  asymmetric bands are both resolved into two bands, at 159 eV and 165 eV, respectively (Fig. 5 b).<sup>47</sup> This indicates the existence of amorphous oxide species in a small fraction mixed with the  $\text{Bi}^{3+}$  oxide state in  $\text{Bi}_2\text{O}_3$ .<sup>48</sup> The XPS spectra of  $\text{L-TiO}_2/\text{Cu}$  shows characteristic peaks of  $\text{Cu}2p_{3/2}$  and  $\text{Cu}2p_{1/2}$  levels are respective at the BE 932.5 eV and 952.1 eV, corresponding to  $\text{Cu(I)}$  species in cuprous oxide.<sup>49,50</sup> The weak satellite observed at 945 eV also demonstrated the presence of  $\text{Cu(I)}$  (Fig. 5 c). In the case of Ni loaded  $\text{L-TiO}_2/\text{Ni}$ , two peaks at BE of 873.7 eV and 855.1 eV correspond to  $\text{Ni}2p_{1/2}$  and  $\text{Ni}2p_{3/2}$ , respectively, and their corresponding satellites at 880 eV and 861 eV (Fig. 5 d). These peaks are assigned to  $\text{Ni}^{2+}$  in  $\text{Ni(OH)}_2$  (derived from the oxidation of  $\text{NiO}$ ).<sup>52</sup> As seen in Fig. 5 e, the BE of  $\text{V}2p_{3/2}$ ,  $\text{V}2p_{1/2}$  and  $\text{O}1s$  are locked at 517.4, 524.8 and 530 eV, and respectively, agree well with  $\text{V}^{5+}$  and  $\text{O}^{2-}$  in  $\text{V}_2\text{O}_5$ .<sup>52,53</sup> The absorption of the lamellar  $\text{TiO}_2$  films can be extended to visible range by metal loading, benefiting the photocatalytic efficiency (Fig. S10<sup>†</sup>).



**Fig. 6** (a) Hydrogen generation of metals loaded lamellar  $\text{TiO}_2$  films. TRMC signals of (b)  $\text{L-TiO}_2/\text{Cu}$  and  $\text{L-TiO}_2/\text{V}$ , (c)  $\text{L-TiO}_2/\text{Bi}$  and  $\text{L-TiO}_2/\text{Ni}$ .

The photocatalytic performances of metal loaded to  $\text{L-TiO}_2$ -1 show variable photoefficiencies (Fig. 6a). For instance,  $\text{L-TiO}_2/\text{Cu}$  exhibits the highest  $\text{H}_2$  generation rate,  $7.88 \text{ mmol h}^{-1} \text{ g}^{-1}$ , which is 40-times higher than solely  $\text{L-TiO}_2$ -1. The  $\text{H}_2$  generation rates using Ni and Bi are,  $1.0 \text{ mmol h}^{-1} \text{ g}^{-1}$  and  $0.9 \text{ mmol h}^{-1} \text{ g}^{-1}$ , respectively, twice that obtained from  $\text{L-TiO}_2$ -1. However, when V is involved, the photocatalytic activity deteriorates, implying an inactivation of the  $\text{L-TiO}_2$ . According to the TRMC investigations of these photocatalysts, the  $\text{L-TiO}_2$ -1 loaded with either Cu or V shows no signal (Fig. 6b). In the first case, Cu acts as an electrons collector for better charge carriers separation and accelerates the reaction rate of hydrogen reduction, while  $\text{V}^{5+}$  state is considered as a recombination center that inhibits the  $\text{H}_2$  production.<sup>54,55</sup> The position of the  $\text{Cu}_2\text{O}$  conduction band makes the thermodynamical transfer of electrons, resulting in the special charge carriers separation.<sup>56-58</sup> The recombination of  $e^-/h^+$  in  $\text{TiO}_2$  is prevented by efficient inter-particle charge transfer leading to improve the hydrogen generation photoefficiency of  $\text{L-TiO}_2/\text{Cu}$ .<sup>59,60</sup> Furthermore, the BE of  $\text{Ti}2p$  shifts to lower energy (0.3 eV) in  $\text{L-TiO}_2/\text{Cu}$ , suggesting partial surface reduction through the formation of oxygen vacancies and lower Ti oxidation state ( $\text{Ti}^{3+}$ ), which was reported to boost the photocatalytic activity of  $\text{TiO}_2$ .<sup>61,62</sup> Conversely,  $\text{L-TiO}_2/\text{V}$  exhibits a slightly higher BE of  $\text{Ti}2p$  (0.6 eV) forming a higher average oxidation state of Ti ions (Fig. 5a).<sup>63</sup>

NiO, as well as Bi<sub>2</sub>O<sub>3</sub> (2.8eV<sup>64</sup>), can form a p-n junction with TiO<sub>2</sub>, generating an internal electric field, thereby giving similar inhibition to the e<sup>-</sup>/h<sup>+</sup> recombination.<sup>65</sup> A similar conclusion can be drawn from the TRMC signal (Fig. 6c), where the increase of the decay rate evidence efficient electrons capture.

## Conclusions

In summary, we report the first demonstration of a simple and reliable one-pot sol-gel self-assembly method for the design of a layered photocatalyst in controlled architecture. The co-assembly of water-soluble titanium diisopropoxide bis(acetylacetonate) precursor and the lyotropic cellulose nanocrystal dispersion allows the formation of CNC/titania hybrid material with a chiral nematic structure. Successfully removal of biotemplate generates a lamellar mesostructure whose birefringence is rising from its anisotropic structure. This kind of long-range ordered lamellar structure can improve the capability of light harvesting and extension of the charge carriers lifetime. The straightforward self-assembly method can be easily applied to the coupling of TiO<sub>2</sub> with different metallic nanoparticles to further enhance its photoefficiency, resulting in higher hydrogen generation. This work offers an easy way to design lamellar mesostructured photocatalyst with the ability to extend charge carriers lifetime, which opens a new avenue for photocatalytic and solar energy conversion.

## Conflicts of interest

There are no conflicts to declare.

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