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Towards the design, synthesis and preliminary biological evaluation of potential corono and clipcarbenes: novel *bis*-imidazolium-*bis*-heterocycle macrocyclic ligands

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ABSTRACT

The synthesis of novel macrocyclic *N-bis*-imidazolium-*bis*-heterocyclic carbenes (NHCs) with rigid and constrained structures to improve the stability of their transition metal coordination complexes in order to obtain efficient photosensitizers is reported. Two pathways have been developed for the preparation of coronocarbenes (L1-L3) containing bipyridine or bithiazole groups. Flexible hemi-macrocycles with potential as clipcarbenes (L1"-L3") are also described. The crystal structures of ligands L1 and L3 are detailed and the preliminary biological evaluation of L1-L3 is presented.

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Introduction

Since the first crystal structure reported by Arduengo and coworkers in 1991 [1], N-heterocyclic carbenes (NHCs) and their metal complexes have garnered extensive research attention with subsequent broadening of their applications [2]. Most of these supramolecular architectures have been developed for their photoactivable properties as potential photosensitizers (PS) in several fields such as dynamic phototherapy [3], dye-sensitized photovoltaic cells [4], [5] and catalysis [6]. However, only a few studies have focused on the preparation of structures combining both a macrocycle and an imidazolium, and on their metal coordination complexes. Among them, macrocyclic tetra-NHC ligands have been reported with structures similar to porphyrins, showing better electronic transfers to the transition metal due to the presence of the NHC units [7], [8]. Smaller macrocycles containing azolium rings have also been described. In this case, the NHCs embedded in the macrocyclic ring allow synergistic functions and have notably been applied in organometallic catalysis [9].

Moreover, *bis*-heterocyclic ligands have great importance in coordination chemistry, as illustrated by the various symmetric and asymmetric *bis*-heterocyclic cryptands and coronands described in the literature [10]. These compounds and their

complexes display notable photochemical properties (e.g. luminescence, antenna effect) due to their strong chelating donor heteroatom coupled with a delocalized π electron system [11].

Our aim was therefore to combine NHCs and *bis*-heterocyclic entities, and thus their associated photophysical properties, into new organized supramolecular structures: *N-bis*-heterocyclic carbene macrocycles.

Hence, we describe two pathways for the synthesis of a novel family of macrocyclic imidazolium salts covalently bearing two imidazolium moieties - as NHC precursors - and two *bis*-heterocyclic units (bipyridine or bithiazole). Our expectation for such structures is to obtain good photophysical properties due to the σ -donor capacity of the imidazolylidene units and the π -conjugated system represented by the *bis*-heterocycle which could absorb light through its π - π * transitions. These macrocyclic carbenes should be able to ensure the absorption of photons in the visible range. This way, they will have a wider spectral absorbance window, leading to a better photoactivation efficiency as expected from high-performing photosensitizers [12].

We also report the synthesis of new flexible *N-bis*-imidazole-*bis*-heterocyclic (**L1'-L3'**) and *N-bis*-imidazolium-*bis*-heterocyclic (**L1''-L3''**) ligands, obtained during the preparation of the abovementioned macrocycles. 1D and 2D NMR spectroscopy and mass

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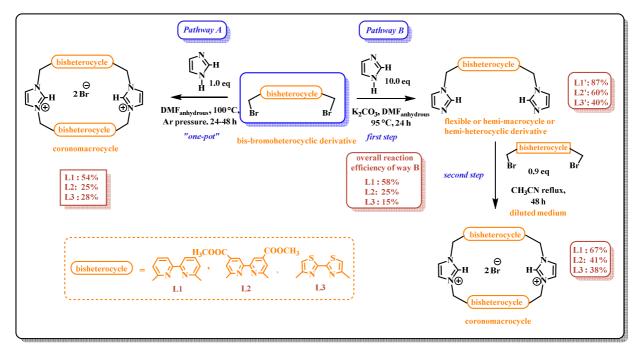
spectrometry characterization of each new compound are provided in the ESI. In addition, the molecular structures of two of the macrocyclic salts, *N-bis*-imidazolium-*bis*-bipyridine L1 and *N-bis*-imidazolium-*bis*-bithiazole L3, obtained by single-crystal X-ray diffraction are described.

In parallel, preliminary biological evaluation of macrocycles L1-L3 was conducted. The results obtained from the cytotoxicity and anti-bacterial studies are presented herein and the biological procedures are provided in the ESI.

Results and Discussion

Synthesis of homoleptic *bis*-imidazolium-*bis*-heterocyclic macrocycles

The preparation of *bis*-imidazolium-*bis*-heterocycle macrocycles requires the prior synthesis of *bis*-(bromomethyl)-*bis*-heterocycles [10], [11], [13] as substrates. Therefore, macrocycles (**L1-L3**) possessing bipyridine, bipyridine *bis*-ester and bithiazole units were prepared *via* two synthetic pathways for cyclization (Scheme 1).

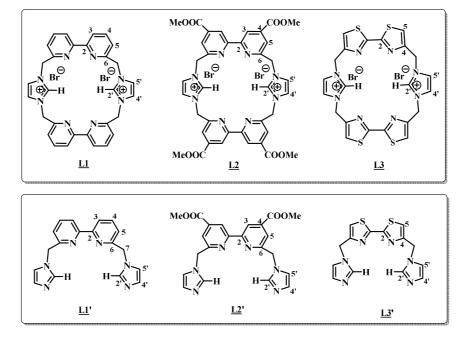


Scheme 1. Synthetic pathways towards hemi- and macro-bis-heterocycles.

The first synthetic pathway (pathway A) is a one-pot cyclization using a stoichiometric mixture of imidazole and the dibromide derivative, in dimethylformamide (DMF) as solvent. The reaction is performed in a Schlenk type reactor over a period of 24 to 48 hours at 100 °C under argon pressure. The *bis*-imidazolium-*bis*-heterocyclic-bromide macrocycles **L1-L3** are obtained with reasonable yields.

The second synthetic pathway (pathway B) is a two-step macrocyclization procedure. The first step consists of synthesizing the hemi-heterocyclic derivatives (**L1'-L3'**) using the method described by Benoît and co-workers [14] with a 10 equivalent excess of imidazole for 1 equivalent of the dibromide derivative in a minimum of solvent (anhydrous DMF) in the

presence of 4 equivalents of K_2CO_3 at 95 °C for 24 hours. Simple extraction allows the hemi-heterocycles (**L1'-L3'**) to be isolated with good yields. The second step consists of the cyclization process and is carried out under high dilution conditions in order to favour the intramolecular reaction. The reaction is performed using 1 equivalent of the hemi-heterocyclic derivative (**L1'-L3'**) to 0.9 equivalents of the corresponding *bis*-bromoheterocyclic derivative. Simple extraction with water and dichloromethane allows removal of the unreacted dibromide derivative. The expected *bis*-imidazolium-*bis*-heterocycle bromide salt **L1-L3** is then dissolved in the aqueous phase and freeze-dried. **L1-L3** are obtained pure without any further purification with good yields.



Scheme 2. Representation of the homoleptic bis-imidazolium-bis-heterocyclic macrocycles L1- L3 (top)

and flexible *bis*-imidazole-*bis*-heterocycles L1', L2' and L3' (bottom).

The NMR 1D (1 H and 13 C) and 2D (HMBC, HSQC) spectroscopy and mass spectrometry of L1-L3 and L1'-L3' are detailed in the ESI.

The two pathways A and B described in Scheme 1, show the successful synthesis of three homoleptic bis-heterocyclic macrocycles **L1-L3**. Pathway A affords the desired macrocycles in only one step under pressure and high temperature. The reaction time is faster (between 24 to 48 hours) compared to pathway B (24 hours for the first step and 48 hours for the second step). The main difference between the two pathways lies in the purification step. In pathway A, purification on silica chromatography is necessary. Due to their high polarity, purification of the *bis*-imidazolium-*bis*-heterocycle cationic macrocycle requires a polar acetone/water/KNO₃ (saturated) eluent. However, only a fraction of the column even if the eluent is very polar. Reverse phase column purification did not result in any improvement.

In contrast, in the first step of pathway B, extraction using a mixture of dichloromethane/water is sufficient to recover the hemi-macrocycle (L1'-L3') from the organic phase. The excess imidazole remains in the aqueous phase. During the second step

of pathway B, extraction with the same solvents (dichloromethane/water) is also sufficient to recover the ligand (**L1-L3**) from the aqueous phase without the need for further purification, which simplifies the procedure.

X-Ray crystallography

Pure crystals of ligands L1 and L3, obtained by slow evaporation in a $CH_2Cl_2/MeOH$ solvent mixture, were analysed by singlecrystal X-ray diffraction. Crystallization attempts by slow evaporation were also made with ligand L2 but failed to provide suitable crystals. Details regarding data collection and processing are provided in the ESI.

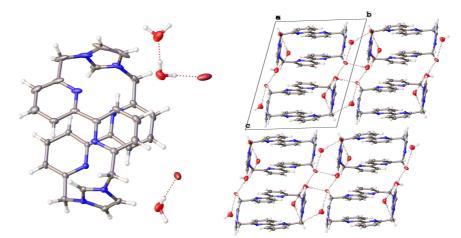


Figure 1. X-Ray crystallographic structure of ligand L1.

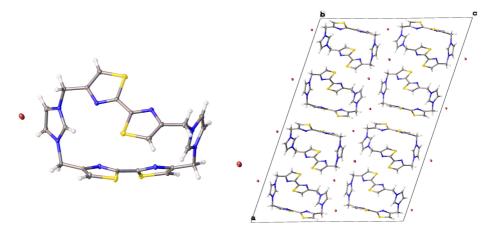
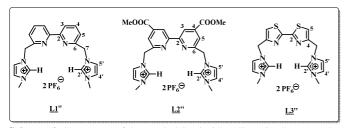


Figure 2. X-Ray crystallographic structure of ligand L3.

Synthesis of methyl-bis-imidazolium-bis-heterocycles

In order to obtain new NHC pincer ligands (or flexible compounds) from *bis*-imidazole-*bis*-heterocycles L1'-L3', it is necessary to quarternarize the nitrogen of the imidazole ring. The preparation methyl bis-imidazolium-bis-heterocycle of hexafluorophosphates L1"-L3" consisted of a conventional methylation reaction using methyl iodide in DMF at 90 $^\circ\mathrm{C}$ overnight. After cooling to room temperature, the reaction mixture is precipitated as a hexafluorophosphate salt using a saturated KPF₆ solution. The obtained yields range from 83-90% and their structures are represented in Scheme 4. Kuhn and coworkers [18] also described the synthesis of the L1" hexafluorophosphate ligand in 80% yield by a direct synthesis route using 1 equivalent of 6,6'-bis-(bromomethyl)-2,2'bipyridine in the presence of 5 equivalents of 1-methylimidazole.



Scheme 4. Structures of the methyl *bis*-imidazolium-*bis*-heterocycles hexafluorophosphate L1"-L3".

NMR (1H and 2D) spectroscopy and mass spectrometry were used to determine the structures of the new pincer ligands L1", L2" and L3" and can be viewed in the ESI.

Biological evaluation

Preliminary biological evaluation of the macrocycles (L1-L3) showed encouraging results. All three macrocycles display varying degrees of cytotoxicity when tested at 100 µg/ mL on MRC-5 cells, expressed as the cytotoxic concentration (CC). Macrocycle L1 is the most cytotoxic compound, followed by L2, whereas L3 is the least cytotoxic against the MRC-5 cells, with cell viabilities of 89%, 95% and 98%, respectively. Macrocycles L1-L3 display varying antimicrobial activity as measured by their minimal inhibitory concentrations (MIC) (Table 1) [19]. Only L1 shows noteworthy antibacterial activity (MIC ≤ 256 µg/mL) and is the most active against Gram-positive bacteria Staphylococcus aureus and Enterococcus faecalis (MIC = 64 μ g/mL). Furthermore, L1 shows antibacterial activity against the Gram-negative bacteria Pseudomonas aeruginosa and Klebsiella pneumoniae, where both pathogens have the same degree of susceptibility (MIC = 256 μ g/ mL). In contrast, L1 is slightly more active against the Gram-negative pathogen, Escherichia *coli* (MIC = $128 \mu g/mL$). This observation is compelling, in terms of the resistance of Gram-negative bacteria, due to the presence of lipopolysaccharides in their outer membrane. Our results show that L1 has a selectivity index (SI) value (SI = CC/MIC) greater than 1 for two of the screened pathogens Staphylococcus aureus and Enterococcus faecalis (Table 1).

Table 1. Antimicrobial activity (minimum inhibitory concentration, MIC) and selectivity index of L1, L2, L3 for five pathogenic bacteria. The cytotoxic concentration (CC) used for calculation of the SI is 100 µg/ mL, allowing the survival of more than 50% of the host cells. The values in bold are considered as noteworthy (MIC $\leq 256 \mu$ g/mL, SI > 1). Sa = Staphylococcus aureus, Ef = Enterococcus faecalis, Pa = Pseudomonas aeruginosa, Ec = Enterobacter cloacae; Ec = Escherichia coli; Kp = Klebsiella pneumoniae.

Compound	MIC (µg/ mL)					SI (CC/MIC)				
	Sa	Ef	Pa	Ec	Кр	Sa	Ef	Pa	Ec	Кр
L1	64	64	256	128	256	1.56	1.56	0.39	0.78	0.39
L2	≥256	≥256	≥256	≥256	≥256	≤ 0.39	≤ 0.39	≤ 0.39	≤ 0.39	≤ 0.39
L3	≥256	≥256	≥256	≥256	≥256	≤ 0.39	≤ 0.39	≤ 0.39	≤ 0.39	≤ 0.39

A selectivity index value greater than 1 indicates that the compound is more toxic to the pathogenic bacteria than to human cells. The greater the selectivity index value, the safer the compound is for the host.

Conclusion

We have described the synthesis of novel homoleptic bisimidazolium-bis-heterocycle macrocycles with bipyridine or bithiazole units (L1-L3), as well as interesting reaction intermediates: bis-imidazole-bis-heterocyclic compounds (L1'-L3'), and methyl-bis-imidazolium heterocyclic derivatives (L1"-L3"). The targeted applications of these compounds would be in research, most notably biomedical for antimicrobial photodynamic therapy (aPDT). It is also important to note that these new macrocycles are soluble in water. They have also shown promising bioactivities (cytotoxicity and anti-bacterial activity). Complexation studies via the formation of their carbenes in the cases of ligand L and L" with transition metals is currently under way.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

Acknowledgments

This manuscript is dedicated to Pr. Alain Marsura

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