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# Electronic structure, elastic and optical properties of Bi2Te3/Sb2Te3

## thermoelectric composites in the periodic-superlattice thin films

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**Abstract**: The outstanding p-type Sb<sub>2</sub>Te<sub>3</sub>-based materials such as the Bi<sub>2</sub>Te<sub>3</sub>/Sb<sub>2</sub>Te<sub>3</sub> thermoelectric thin films with consideration of anisotropy and synergetic optimization of thermoelectric performance are promising used in practical applications through improving the power factor and reducing the thermal conductivity. Electronic structure, heat capacity and homogenized elastic properties of anisotropic Bi<sub>2</sub>Te<sub>3</sub> and Sb<sub>2</sub>Te<sub>3</sub> crystals are investigated, and then modeled to calculate elastic properties of the composite Bi<sub>2</sub>Te<sub>3</sub>/Sb<sub>2</sub>Te<sub>3</sub> thin films. Debye temperature, thermal conductivity and *ZT* value are especially anaylsed and found that the Sb-5p/Bi-6p and Te-5p at Debye temperatures of 179.29 K and 178.20 K, with the energy gap of 0.207 eV and 0.256 eV in Bi<sub>2</sub>Te<sub>3</sub> and Sb<sub>2</sub>Te<sub>3</sub>, which contribute a multiscale modeling scheme from periodic single crystal film to multilayer composite Bi<sub>2</sub>Te<sub>3</sub>/Sb<sub>2</sub>Te<sub>3</sub> thin films with higher ZT value, resulting in the significant reduction of band gap and the thermal conductivity due to its superlattice multi-layer structure.

**Keywords**: Composite Bi<sub>2</sub>Te<sub>3</sub>/Sb<sub>2</sub>Te<sub>3</sub> structure; Debye temperatures; Homogenized elastic properties; Thermoelectrics;

#### **1. Introduction**

Based on Seebeck effect and Peltier effect, thermoelectric (TE) materials to realize the conversion between thermal energy and electrical energy are milestone materials in the thermal power generation, solid state refrigeration, wireless sensing, smart device and high-performance optoelectronic device [1-6]. The bismuth telluride (Bi<sub>2</sub>Te<sub>3</sub>) and antimony telluride (Sb<sub>2</sub>Te<sub>3</sub>) in high-quality Bi<sub>2</sub>Te<sub>3</sub>/Sb<sub>2</sub>Te<sub>3</sub>-based TE materials are obligatory in photodiodes, sensors, thermoelectric generators, thermal generator or solid-state cooling devices [7-10]. The performance of thermoelectric materials is generally measured by a dimensionless value (ZT) from the formulas:  $ZT = S^2 \cdot \sigma \cdot T/\kappa$ , where, S is the Seebeck coefficient,  $\sigma$  is the conductivity, T is the absolute temperature,  $S^2 \cdot \sigma$  is power factor (*PF*),  $\kappa$  is the thermal conductivity  $(\kappa = \kappa_e + \kappa_l)$ . The  $\kappa$  is consisted of the carrier thermal conductivity  $\kappa_e$  and lattice thermal conductivity  $\kappa_l$ . which can be improved thermoelectric mechanical between and performances in polycarbonate/single-walled carbon nanotube composite films [10]. The potential inorganic semiconductor material is still Bi<sub>2</sub>Te<sub>3</sub>/Sb<sub>2</sub>Te<sub>3</sub> compounds [7, 8]. Except for the Bi<sub>2</sub>Te<sub>3</sub> and Sb<sub>2</sub>Te<sub>3</sub> strong topological insulators,  $(Bi_2Te_3)_x(Bi_2Se_3)_{1-x}$  and  $Sb_2$   $(Te_{1-x}Se_x)_3$  with higher high ZT value from 1.0-2.0 are discovered as the important milestones [9-13], with the research methods like DFT method [12, 14], heat pulse method [13], X-ray powder diffraction [14, 15] etc.

In the present case of thermoelectric compounds of  $Sb_2Te_3$  and  $Bi_2Te_3$ , few reports of theory is available on electronic structure, thermoelectric or mechanical properties [9-15,16]. The  $Bi_2Te_3$  alloy is reported by using molecular epitaxy to prepare  $Bi_2Te_3$  superlattice materials [17] and the thermoelectric compounds later are extended to Bi-Se-Te ternary alloys [18, 19]. For  $Bi_2Te_3$  alloy, the electronic structure has been described <sup>[20]</sup> and the mechanical properties coupling the obtained elastic constant  $C_{ij}$  are especially investigated from the first-principle calculation [20, 21]. The p-type Bi<sub>2</sub>Te<sub>3</sub>-Sb<sub>2</sub>Te<sub>3</sub> single crystal synthesed in experiment by the moving heating method, with a PF value upper to  $5.3 \times 10^{-3}$  W/(K<sup>2</sup>·m) [4], which is 67.9 % higher than that of Bi<sub>2</sub>Te<sub>3</sub> alloy. Sb<sub>2</sub>Te<sub>3</sub>-based films can be prepared by thermal evaporation [4, 7], sputtering [18], electrochemical deposition [5], molecular beam epitaxy, pulsed laser deposition (PLD) [6, 13], MBE deposition [23] and chemical vapor deposition deposition (CVD) techniques [24], with the ZT region of 0.40-1.75 and the PF region of 1.32-4.30 [4-7, 13, 18], separately. Among the thermoelectric Sb<sub>2</sub>Te<sub>3</sub>-based thin films materials studied at present, Bi-Te semiconductor with a ZT about 1 and the conversion efficiency less than 10.0 %, which is far from satisfactory [1-3]. In particular, micro thermoelectric devices based on thermoelectric Sb<sub>2</sub>Te<sub>3</sub>-based films especially for the composite  $(Bi_2Te_3)_x(Sb_2Te_3)_{1-x}$  films greatly depends on these two layered hexahedral structure with  $R\overline{S}_m$  space group and their synergetic optimization combination [24-27]. Therefore, Sb atom is added to replace Bi in Bi<sub>2</sub>Te<sub>3</sub> to form solid solution on the basis of the Bi-Te binary alloy, so as to change the carrier type and density of the p-type Sb<sub>2</sub>Te<sub>3</sub> semiconductor (band-gap  $E_e < 0.3 \text{ eV}$ ) [28] and improve the doping properties of p-type Sb<sub>2</sub>Te<sub>3</sub>-based thin films. It is therefore necessary to study the mechanical stability, Debye temperature, and band gap of Sb<sub>2</sub>Te<sub>3</sub>, Bi<sub>2</sub>Te<sub>3</sub> and composite Sb<sub>2</sub>Te<sub>3</sub>/Bi<sub>2</sub>Te<sub>3</sub> superlattice structure, separately.

By reviewing the references, due to their highly complex band structures and broadband wavelengths, some basic physical properties of Bi<sub>2</sub>Te<sub>3</sub> and Sb<sub>2</sub>Te<sub>3</sub> of TE layered semiconductor materials still remain unexplored, especially for understanding their strong optical absorbance and the narrow direct band gap of Bi<sub>2</sub>Te<sub>3</sub>/Sb<sub>2</sub>Te<sub>3</sub>-based TE materials. However, the detailed thermodynamic parameters, Debye temperature, heat capacity and the homogenized elastic properties of anistropic Bi<sub>2</sub>Te<sub>3</sub>, Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub>-Sb<sub>2</sub>Te<sub>3</sub> crystal structures remain scarce and far way from obtaining an agreement. Through the systematical research on the based ground elastic constants  $C_{ij}$ , elastic moduli, Debye temperature, Young's modulus and Poisson ratios of hexagonal Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> and Sb<sub>2</sub>Te<sub>3</sub> and Sb<sub>2</sub>Te<sub>3</sub> this work provides a theoretical basis to investigate the surface adsorption and doping on Bi<sub>2</sub>Te<sub>3</sub> and Sb<sub>2</sub>Te<sub>3</sub> thin film of (Bi<sub>2</sub>Te<sub>3</sub>)<sub>x</sub>(Bi<sub>2</sub>Se<sub>3</sub>)<sub>1-x</sub>, Sb<sub>2</sub>(Te<sub>1-x</sub>Se<sub>x</sub>)<sub>3</sub>, Bi<sub>2</sub>Te<sub>2-7</sub>Se<sub>0.3</sub> and Bi<sub>0.46</sub>Sb<sub>1.54</sub>Te<sub>3</sub> with higher *ZT* value or composite ply-type -Bi<sub>2</sub>Te<sub>3</sub>-Sb<sub>2</sub>Te<sub>3</sub>-Bi<sub>2</sub>Te<sub>3</sub>-thin films, aiming to the exploitation of the multiscale modeling and physical properties in calculation of the meaningful p-type Sb<sub>2</sub>Te<sub>3</sub>-based thin film materials. Due to the anisotropy of single crystal, the simulation of thin film materials is more complicated. By optimizing the band structure of nano-composite structures, the density of states near Fermi energy levels and phonon scattering can be increased, and the synergistic optimization of thermoelectric properties with higher *ZT* value can be fatherly realized in future.

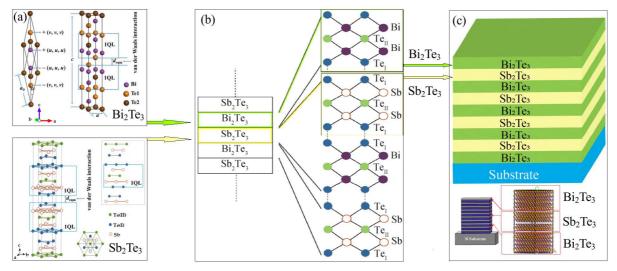
#### 2. Modeling and computational approach

### 2.1 Modeling of Sb<sub>2</sub>Te<sub>3</sub>, Bi<sub>2</sub>Te<sub>3</sub> and ply-type -Bi<sub>2</sub>Te<sub>3</sub>-Sb<sub>2</sub>Te<sub>3</sub>- structures

Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> belongs to the hexagonal  $R\bar{3}m$  coordinate system ( $\alpha=\beta=90^{\circ}$ ,  $\gamma=120^{\circ}$ , No.166) [3, 12, 27-29] with a periodic layered structure unit consists of five atomic layers along the C-axis. The order arranges alternately in a cycle of Te(I)-Bi/Sb-Te(II)-Bi/Sb-Te(I), where Te(I) is connected with Bi/Sb by covalent bond and ionic bond, and Te(II) is connected with Bi/Sb by covalent bond, adjacent Te(I) and Te(I) are connected by the van der Waals bond [27-29]. The interaction between Te and Te of two adjacent crystal plane atomic layers is relatively small. Modeling of Sb<sub>2</sub>Te<sub>3</sub>, Bi<sub>2</sub>Te<sub>3</sub> and ply-type -Bi<sub>2</sub>Te<sub>3</sub>-Sb<sub>2</sub>Te<sub>3</sub>-structures are shown in Fig.1.

From Fig.1 (a), the primitive cells of Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> and contain three Layer Quintuple Layer (QL) elements, each of which consists of two Sb/Bi atomic layers and three Te atomic layers. Lattice parameters *a* and *c* of Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> are 4.269Å, 30.465Å and 4.384Å, 30.488Å, which are close to references[15, 25, 27-29] of Sb<sub>2</sub>Te<sub>3</sub> and references[9, 15, 26-28] of Bi<sub>2</sub>Te<sub>3</sub>. The volumes of Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> are

480.82Å<sup>3</sup> and 507.46 Å<sup>3</sup>, which are close to references [15, 29] of Sb<sub>2</sub>Te<sub>3</sub> and reference [15] of Bi<sub>2</sub>Te<sub>3</sub>. Based on the crystals of Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub>, a vacuum layer of approximately 15Å was constructed to obtain the surface lattice structure, and a silicon substrate length about 6Å was added to the calculation, resulting in a surface structure of 1 to 6 units of QL layer, with the periodic superlattice model of composite ply-type -Bi<sub>2</sub>Te<sub>3</sub>-Sb<sub>2</sub>Te<sub>3</sub>- structures is established in Fig.1(b), in order to keep consistent with the super lattice thin film model of composite ply-type -Bi<sub>2</sub>Te<sub>3</sub>-Sb<sub>2</sub>Te<sub>3</sub>- thin film structures in Fig.1(c) of high-thermoelectric performance considering size effect at nanoscale [30-32].



**Fig.1** Modeling of Sb<sub>2</sub>Te<sub>3</sub>, Bi<sub>2</sub>Te<sub>3</sub> and ply-type  $-Bi_2Te_3-Sb_2Te_3-$  structures: (a) Crystal structure of Bi<sub>2</sub>Te<sub>3</sub> and Sb<sub>2</sub>Te<sub>3</sub>; (b) Synthesis of Bi<sub>2</sub>Te<sub>3</sub>/Sb<sub>2</sub>Te<sub>3</sub> superlattice; (c) Schematic diagram of the crystal structure of the Bi<sub>2</sub>Te<sub>3</sub>/Sb<sub>2</sub>Te<sub>3</sub> superlattice.

#### 2.2 Computational approach

The Perdew-Burke-Ernzerhof (PBE) and mBJ (modified Becke-Johnson potential) functions by generlized gradient approximation (GGA) were adopted as the exchange-correlation potential [12, 29]. The ultra soft pseudopotential was used to deal with the interaction between real ion and valence electron, and plane wave basis was set to describe the wave function of the system electron. The  $4 \times 4 \times 4$  k-point mesh for Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> was used in the self-consistent field. The plane-wave basis set cut off energy *E*<sub>cut</sub> was 300 eV. The mechanical properties and Debye temperature with considering experimental parameters of Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> structures [33-35, 36] were calculated accordingly.

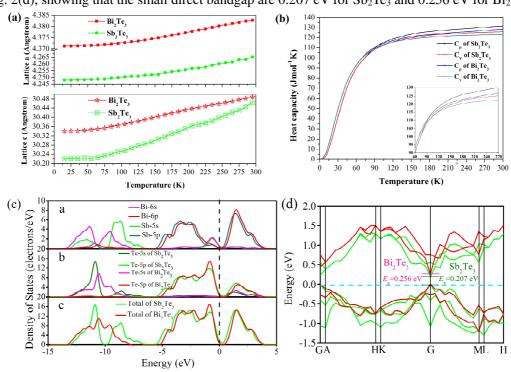
#### 3. Results and discussion

#### 3.1 Lattice parameters, electronic structure and density of state (DOS)

Variation of lattice parameters under various temperatures, electronic band structures and DOS of Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> thermoelectric compounds are shown in Fig.2.

From Fig.2 (a), with the increase of temperature, the lattice parameters increased. It can be seen from Fig.2 (b) that the  $C_V$  value gradually reaches to the Dulong–petit limit  $3n \cdot N_A \cdot K_B$ , which is close to the other reference [9, 11]. Unlike the temperature, the pressure has little effect on the  $C_p$  and  $C_V$  [45]. From Fig. 2 (c), the calculated total DOS display the place that the conduction band and the low lying valence band contributed by s and p orbital states. The DOS of the Fermi level is slightly greater than zero, indicating the weak metallic properties. The total DOS of Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> in the range of 15 ~ -6 eV and 15 ~ -7 eV are mainly composed of Sb-5s/Bi-6s and Te-5s states, while in the range of -6 ~ 0 eV and -7 ~ 0 eV composed of Sb-5p/Bi-6p and Te-5p. Above the Fermi level, the total DOS is mainly consists of Sb-5p/Bi-6p, Te-5s

and Te-5p states. Since, Sb and Bi with p-electron bind in an ionic fashion with the neighbouring Te-p orbitals in Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> molecules. For the conduction band of Bi<sub>2</sub>eT<sub>3</sub>, the contribution of Bi-6p state is larger, and for the valence band, the contribution of Te(I)-p state is larger. The Te(II)-5p has a smaller contribution to the conduction band and valence band near the Fermi level. The electronic structures and DOS calculated by PBE-GGA along the direction of high symmetry points in the Brillion zone are shown in Fig. 2(d), showing that the small direct bandgap are 0.207 eV for Sb<sub>2</sub>Te<sub>3</sub> and 0.256 eV for Bi<sub>2</sub>Te<sub>3</sub>.



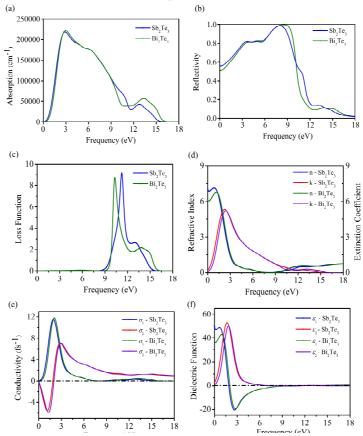
**Fig.2** Lattice parameters, heat capacity, electronic band structures and DOS of Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> thermoelectric compounds: (a) Lattice parameters of Bi<sub>2</sub>Te<sub>3</sub> and Sb<sub>2</sub>Te<sub>3</sub>; (b) Heat capacity of Bi<sub>2</sub>Te<sub>3</sub>/Sb<sub>2</sub>Te<sub>3</sub>; (c) Electronic band structures of Bi<sub>2</sub>Te<sub>3</sub>/Sb<sub>2</sub>Te<sub>3</sub>; (d) DOS of Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub>.

#### 3.2 Optical properties comparison of single Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> crystals

The optical constants such as absorption, reflectivity, energy loss function refractive index, conductivity, dielectric function, are shown in Fig. 3.

The optical absorption in Fig. 3 (a) that the peak of the absorption region is at 3.0 eV. In the range of photon energy greater than 18.0 eV, the absorption spectrum of Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> tends to zero, indicating that there is no absorption peak. The main reflection at sharp peak in Fig. 3 (b) occurs at 8.5 eV and 9.0 eV, and the reflectivity of the material is extremely high. The crystal has a higher resistivity in the energy region of 3.0-10.0 eV and 3.0-9.0 eV. Besides, the maximum loss function peak of Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> appears at 11.5 eV and 10.5 eV in Fig. 3 (c). The refractive index  $n(\omega)$  of Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> at 0 eV is calculated to be 7.0 and 6.0, and their main peak appears both at 1.5 eV, but the corresponding refractive index is about 7.5 and 6.8, separately. Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> have both a light refraction characteristic of 0~1.5 eV far greater than their absorption characteristic. The static value n(0) at 0ev for Sb<sub>2</sub>Te<sub>3</sub> is 7.13 and 6.18 for Bi<sub>2</sub>Te<sub>3</sub>. From the optical conductivity in Fig. 3 (e), there is an obvious peak at 2.0 eV with the value of 11fs<sup>-1</sup>. There is also a weak peak at 13.0 eV. From Fig. 3(d) to Fig. 3(f), the spectral characteristics  $n(\omega)$  are identical to the dielectric function  $\varepsilon_1(\omega)$ . In Fig. 3(f), for the Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> at 0 eV, the real part of the dielectric function ( $\varepsilon_1$ ) corresponds to a static dielectric constant of about 48.0 and 37.0. The imaginary part of the dielectric function has a peak (2.5 eV and 2.0 eV), and  $\varepsilon_2$  is about 56.0 and 50.0. These differences can be attributed to the common transition of Sb-5s/Bi-6p and Te-5s between the conduction band and the

valence band. The static dielectric constant  $\varepsilon_1(0)$  is parallel to the static refractive index n(0).



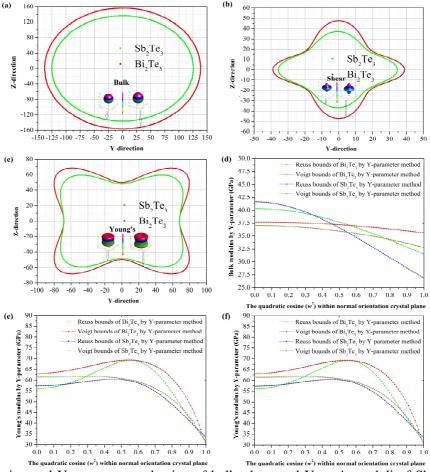
**Fig. 3** Calculated optical properties of  $Sb_2Te_3$  and  $Bi_2Te_3$  from 0 to 18 eV. (a) Absorption coefficient (b) optical reflectivity (c) electron energy loss function (d) refractive index and extinction coefficient (e) optical conductivity and (f) dielectric function constant

As the exact band gaps of  $Sb_2Te_3$  and  $Bi_2Te_3$  remain many controversies [46], it is necessary to study the optical properties, anisotropy and precise band gap, due to the little difference between the experimental Seebeck coefficient *S* of  $Bi_{2-x}Sb_xTe_3$  and the calculated *S* of  $Sb_2Te_3$  [46], to better understand the mechanical properties of the  $Sb_2Te_3/Bi_2Te_3$  superlattice thin films considering thin flim thickness.

#### 3.3 Anistropy of single crystal and homogenization of polycrystals

The calculated elastic constants  $C_{ij}$  of Sb<sub>2</sub>Te<sub>3</sub> ( $C_{11}$ =70.52GPa,  $C_{12}$ =15.38GPa,  $C_{13}$ =25.13GPa,  $C_{33}$ =48.16GPa,  $C_{44}$ =22.87GPa) and Bi<sub>2</sub>Te<sub>3</sub> ( $C_{11}$ =72.11GPa,  $C_{12}$ =12.69GPa,  $C_{13}$ =28.25GPa,  $C_{33}$ =50.35GPa,  $C_{44}$ =28.78GPa) are close to other Sb<sub>2</sub>Te<sub>3</sub> reference [29] and Bi<sub>2</sub>Te<sub>3</sub> references [9, 11, 25], which are all satisfied with the Born's stability restriction relations [29, 37, 38], indicating that the modeled crystals are stable in mechanical mechanics. Based on the definition of Reuss and Voigt bounds, Y-parameter estimation method containing hexagonal crystal surface normal [49] is used for the comparative study the homogenization of polycrystals, exhibiting the anisotropy of elastic moduli in function of the orientation of the crystal plane. Surface construction of bulk, shear and Young's moduli of Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> are in Fig. 4.

Based on the elastic constants  $C_{ij}$ , the Debye temperature [29, 38] and homogenized elastic moduli can be estimated based on the Voigt-Reuss-Hill estimation [40]. The Bulk modulus *B*, shear modulus *G* and Young's modulus *E* are 35.39GPa, 22.12GPa and 54.91GPa for Sb<sub>2</sub>Te<sub>3</sub>, in comparison of 36.94GPa, 59.75GPa and 24.28GPa for Bi<sub>2</sub>Te<sub>3</sub>, which are close to reference [29] of Sb<sub>2</sub>Te<sub>3</sub> and reference [25] of Bi<sub>2</sub>Te<sub>3</sub>. Young's modulus in *x/y*-direction ( $E_{x/y}$ ) and that in *z*-direction ( $E_z$ ) are 57.32GPa and 33.46GPa for Sb<sub>2</sub>Te<sub>3</sub>. The  $E_{x/y}$  and  $E_z$  of Bi<sub>2</sub>Te<sub>3</sub> are 56.08GPa and 31.53GPa, which is close to references [9, 25]. The B/G ratios of Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> are 1.521 and 1.600, less than the critical value 1.75 proposed by Pug's criterion, showing that toughness is relatively small [38]. The higher *E* is, the more covalence enhances and the less ductile character is [47, 48]. Besides, Bi<sub>2</sub>Te<sub>3</sub> is more covalent than Sb<sub>2</sub>Te<sub>3</sub> and shows less ductile character than Sb<sub>2</sub>Te<sub>3</sub> by Pugh's ratios. From Zener's anisotropy factor, Bi<sub>2</sub>Te<sub>3</sub> is less isotropic than Sb<sub>2</sub>Te<sub>3</sub> and may bring in cracks or microstructure defects [38]. Therefore, Poisson's ratio of 0.24 for Sb<sub>2</sub>Te<sub>3</sub> is more resistant than 0.23 for Bi<sub>2</sub>Te<sub>3</sub>, with a certain character of fracture resistance [37, 38].



**Fig. 4** Construction and Y-parameter evaluation of bulk, shear and Young's moduli of  $Sb_2Te_3$  and  $Bi_2Te_3$  polycrystals: (a) Bulk modulus construction; (b) Shear modulus construction; (c) Young's modulus construction; (d) Shear modulus by Y-parameter estimation; (e) Shear modulus by Y-parameter estimation; (f) Young's modulus by Y-parameter estimation.

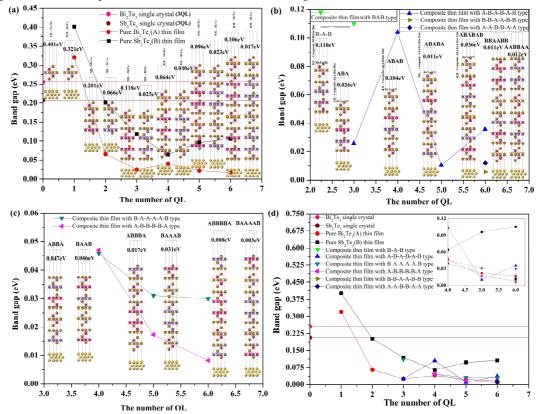
Relationship between elastic moduli and plane orientations for Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> by Reuss and Voigt bounds in Fig. 4(a)-(c) shows that the bulk, shear and Young's moduli are anisotropic. Young's modulus of Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> has a significant deviation in shape from the sphere, indicating a great anisotropy. Fig. 4 (a)-(c) point out the three-dimension projection curve, and there is little change on the surface of bulk modulus curves, with the outer curve of Bi<sub>2</sub>Te<sub>3</sub> and inner curve of Sb<sub>2</sub>Te<sub>3</sub>, which is in accord with the A<sup>U</sup> value of Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> (are greater than zero and deviates to zero), indicating that Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> for Bi<sub>2</sub>Te<sub>3</sub> and Sb<sub>2</sub>Te<sub>3</sub>. Moreover, the comparison of Young's modulus and shear modulus of Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> are very close to each other.

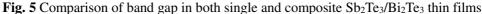
Furthermore, the general anisotropy index  $(A^U)$ , compression and shear anisotropy percentages  $(A_B$  and  $A_G)$  correspond to elastic anisotropy [37, 38]. It should be noted that Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub>has special

chemical bond and obvious anisotropy in lattice space distribution, which leads to the anisotropy of mechanical and thermoelectric properties. Anisotropy plays important role in Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> films [42, 43], especially in p-type Sb<sub>2</sub>Te<sub>3</sub> and n-type Bi<sub>2</sub>Te<sub>3</sub> thin films [44].

#### 3.4 Band gap comparison of both single and composite thin films

By constructing superlattice of different structures, different surface states can be obtained. On the basis of studying the factors that can change the surface states, the surface states can be adjusted artificially. The superlattice containing  $Sb_2Te_3/Bi_2Te_3$  film structures are constructed and the energy bands are found to be dependent on the superlattice film structure, shown in Fig.5.





From Fig. 5 that the surface states of the topological insulator films with a superlattice structure can be changed significantly when the crystal unit cell only contains 3QL or 4 QL. As more quintuple layers in the unit cell, the configuration has less impact on the surface state. As in Fig. 5(a), the band gap of pure-phase films decreases to a relatively narrow level with the increase of the layer thickness. From Fig. 5(b)-(c), when the crystal cell contains 3 QL to 4 QL, the influence of the structure of the film on the surface state is obvious. In the case of 3 QL, the surface state gap of the material reaches the room temperature scale if the heavier Bi atomic layer is placed in the outer layer. If the lighter Sb atom layer is placed in the outer layer of the 3QL superlattice, not only the surface state gap is 4 times of the room temperature scale (0.026 eV), but also the time reversal symmetry is destroyed. In the case of 4 QL, the surface state gap of superlattice thin films grown alternately is larger than that of other materials. From Fig. 5(d), when the crystal cell contains 5 QL to 6 QL, the surface energy band properties of the superlattice films on the surface energy gap is not significant due to the scattering phonons with the thermal conductivity reduced [50, 51].

#### 3.5 Discussion of thermoelectric properties

The calculated  $C_V$  of Bi<sub>2</sub>Te<sub>3</sub> and Sb<sub>2</sub>Te<sub>3</sub> are 123.04 J·mol<sup>-1</sup>·K and 128.25 J·mol<sup>-1</sup>·K, while the calculated C<sub>p</sub> of Bi<sub>2</sub>Te<sub>3</sub> and Sb<sub>2</sub>Te<sub>3</sub> are 125.85 J·mol<sup>-1</sup>·K and 131.57 J·mol<sup>-1</sup>·K. The calculated B of Bi<sub>2</sub>Te<sub>3</sub> and Sb<sub>2</sub>Te<sub>3</sub> are 35.39 GPa and 36.94 GPa. The compressional velocity  $v_i$ , the sound velocity  $v_s$  and the average sound velocity  $v_m$  are separately calculated, which are close to the reference [29] of Sb<sub>2</sub>Te<sub>3</sub> and the reference [25] of Bi<sub>2</sub>Te<sub>3</sub>. From the comparison of references [14, 15, 46] of Sb<sub>2</sub>Te<sub>3</sub> and references [25, 45] of Bi<sub>2</sub>Te<sub>3</sub>, Debye temperature  $\theta_D$  of Sb<sub>2</sub>Te<sub>3</sub> is higher than the Bi<sub>2</sub>Te<sub>3</sub>. Furthermore, the orientation dependent Grüneisen parameter  $\gamma$  has been extracted by Barron [52] and Cahill [53], namely,  $\gamma_{a/c} = [\alpha_{a/c}(C_{11}$ + C<sub>12</sub>) +  $\alpha_{ac}C_{13}$ ]/C<sub>V</sub> along the a/c direction. Based on the  $\gamma$  by the Poisson ratio [54] and the Barron-Cahill model,  $\gamma$  of Bi<sub>2</sub>Te<sub>3</sub> and Sb<sub>2</sub>Te<sub>3</sub> are 1.50 and 1.71. It indicates that the thermal expansion coefficient  $\alpha$  rapidly increase from starting and gradually achieve a constant value. For instance, by the equation of  $\alpha = 2\alpha/l + \alpha_{\perp}$ , the  $\alpha$  of Bi<sub>2</sub>Te<sub>3</sub> and Sb<sub>2</sub>Te<sub>3</sub> reaches 4.698×10<sup>-5</sup> K<sup>-1</sup> ( $\alpha/l$ =1.294×10<sup>-5</sup> K<sup>-1</sup>,  $\alpha_{\perp}$ =2.110×10<sup>-5</sup> K<sup>-1</sup>) and 6.935×10<sup>-5</sup> K<sup>-1</sup> ( $\alpha$ //=1.852×10<sup>-5</sup> K<sup>-1</sup>,  $\alpha_{\perp}$ =3.231×10<sup>-5</sup> K<sup>-1</sup>) at 275 K, being consistent with the Bi<sub>2</sub>Te<sub>3</sub> value of  $4.60 \times 10^{-5}$  K<sup>-1</sup> as reported by Feng [9] and the calculated Sb<sub>2</sub>Te<sub>3</sub> value of  $7.10 \times 10^{-5}$  $K^{-1}$  as reported by Bessas [15]. Bi<sub>2</sub>Te<sub>3</sub> and Sb<sub>2</sub>Te<sub>3</sub> have the average atomic mass of 168.29 amu and 124.64 amu [15], with isothermal compressibility K<sub>T</sub> of 2.67 Mbar<sup>-1</sup> and 3.30 Mbar<sup>-1</sup>[55]. The calculated lattice thermal conductivities for Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> are 2.38 W·m<sup>-1</sup>K<sup>-1</sup> and 1.56 W·m<sup>-1</sup>K<sup>-1</sup> at 275 K, compared with the  $\kappa_{xx}$  of 1.90 W m<sup>-1</sup>K<sup>-1</sup> and 1.20 W m<sup>-1</sup>K<sup>-1</sup> and  $\kappa_{zz}$  of 1.27 W m<sup>-1</sup>K<sup>-1</sup> and 0.40 W m<sup>-1</sup>K<sup>-1</sup> for Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> at 300 K [46]. By the equation of  $ZT = S^2 \cdot T/\rho \cdot \kappa_T$  [17], the calculated ZT of 0.81 and 0.61 for Sb<sub>2</sub>Te<sub>3</sub> and Bi<sub>2</sub>Te<sub>3</sub> at 275 K. K<sub>L</sub> for all the 60-Å-period structures, no matter the superlattice structure of 30Å/30Å or 10Å/50Å or 20Å/40Å, is about 0.25 W m<sup>-1</sup>K<sup>-1</sup> [17]. Similarly, the calculated total thermal conductivities of the Sb<sub>2</sub>Te<sub>3</sub>/Bi<sub>2</sub>Te<sub>3</sub> superlattice structures with 2QL and 6QL are 0.33 W·m<sup>-1</sup>K<sup>-1</sup> and 0.46 W·m<sup>-1</sup>K<sup>-1</sup>, compared to the total thermal conductivity is 0.37 W·m<sup>-1</sup>·K<sup>-1</sup> using the same electrical resistivity  $\rho_{\perp}$  value of ~1.23×10<sup>-3</sup>  $\Omega$ ·cm (along c axis) and ~1.04×10<sup>-3</sup>  $\Omega$ ·cm (along a-b axis)[17] and the same  $S^2T$  value by the Weidemann-Franz law for determining the ZT value. As the maximum Seebeck coefficient of Bi<sub>2</sub>Te<sub>3</sub>/Sb<sub>2</sub>Te<sub>3</sub> superlattice is 260.00  $\mu$ V/K [17], using the electrical conductivity  $\sigma$  of 1230 S cm<sup>-1</sup>[56], the ZT value of the superlattice structuresis greatly improved, with the decrease of thermal conductivity from 0.46 W m<sup>-1</sup>K<sup>-1</sup> to 0.33 W m<sup>-1</sup>K<sup>-1</sup>. The ZT for Sb<sub>2</sub>Te<sub>3</sub>/Bi<sub>2</sub>Te<sub>3</sub> superlattice structures with 2QL and 6QL are 2.69 and 1.93 at 275K, which is close to that of 2.40 at 300K of p-type 10Å/50Å Bi2Te3/Sb2Te3 superlattices [17].

#### 4. Conclusion

The electronic structure shows that Bi<sub>2</sub>Te<sub>3</sub> (0.256 eV) has the bigger band gap than Sb<sub>2</sub>Te<sub>3</sub> (0.207 eV) because of its larger mass (800.80 g/mol). Thus, energy gap can be enlarged with doping heavy elements in compounds for practical applications. For the composite Bi<sub>2</sub>Te<sub>3</sub>/Sb<sub>2</sub>Te<sub>3</sub> structure film, the surface states of the topological insulator films with a superlattice structure are affected by the substrates when the unit cell contains 3 QL, it indicates that we may adjust the surface states by different substrates. With the increase of QL thickness more than 5, because the system energy gap has become stable, the influence of changing the structure of the superlattice film on the energy gap is no longer obvious. The AB-type periodic composite Bi<sub>2</sub>Te<sub>3</sub>/Sb<sub>2</sub>Te<sub>3</sub> film will lead to a large number of point defects in Sb<sub>2</sub>Te<sub>3</sub>, thus strongly scattering phonons, and the superlattice can effectively scatter phonons due to its special superlattice structure, resulting in the thermal conductivity significantly reduced so as to increase the ZT value of the superlattice structures.

#### **CRediT** authorship contribution statement

**Jia Fu**: Validation, Formal analysis, Resources, Writing - original draft, Visualization, Supervision, Project administration. **Jiaxuan Huang**: Methodology, Data curation. **Fabrice Bernard**: Conceptualization.

#### **Declaration of competing interest**

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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