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MICROCHIP COCRYSTALLIZATION WITH SUPERCRITICAL CO₂

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GAS

compressed CO₂

What is a cocrystal ?

In the pharmaceutical field, a cocrystal is a solid with a single crystalline phase composed of two entities : the active pharmaceutical ingredient (API) and the coformer (**COF**).

Cocrystals are produced in other to enhance drugs physicochemical properties by incorporation of the coformer into their crystalline structure, without altering therapeutic effects.

Supercritical CO₂ as an anti-solvent

ANY VISUAL OBSERVATIONS FOR TIME-RELATED EVENT ANALYSIS

Anti-solvent effect

The system {API+COF} is preliminarily dissolved into a solvent miscible with CO_2 , then the solubility of the solution is reduced by adding supercritical CO_2 .



Two main scCO₂ precipitation methods Gasous Anti-Solvent (GAS) :

scCO₂ is added gradually into the solution {solvent+API+COF}

Supercritical Anti-Solvent (SAS) :

The solution {solvent+API+COF} is injected into a continuous flow of scCO₂



[C. Neurohr, PhD Thesis, 2015]

SAS

 CO_2

Solution

A METHODOLOGY

Microfluidic approaches

Silicium-Pyrex on-chip cocrystallization

Microfluidic platforms are great tools for cocrystal investigations :

- Fast screening capability
- High reproducibility

Introduction

Control on hydrodynamics

Silicium-Pyrex microchip fabrication



UV exposure Photoresist

Silicon wafer

with oxidation layer

OBJECTIVES

to time-related events Access (kinetics / diffusion / induction time) by studying nucleation and growth of cocrystals in microfluidic platforms thanks to in-situ analysis and observations Control crystal phase structure •

- Improved heat and mass transfers
- In-situ on-chip analysis under high pressure -



[S.Morais, PhD Thesis, 2016]

Strategies for microchip design & Preliminary results

(Modulated property)



Solvent-CO₂ thermodynamic data are essential for the development of microchips





Aim



For both design, nucleation will occur with diffusion of scCO₂ into the « microreactor »

Immobilization of solution into wells



conditions immobilize Monophasic B solution {solvent+API+COF} in wells and fill the main channel with scCO₂

CONCLUSION

 $T_i T_i + \Delta T$ Preliminary results : Working conditions with design (B): P = 80 bar / T = 35°C Solvent = acetone API = NaproxenNeedles of naproxen COF = nothingInduction time < 3minutes

Combination of microfluidic tools and knowledge on cocrystallization in comressed CO₂ led to have information on precipitation processes like composition, diffusion and induction time by using *in-situ* analysis technologies.

