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Air-broadened N₂O line-shape parameters and their temperature dependences by requantized classical molecular dynamics simulations

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Abstract

In this paper, we used requantized Classical Molecular Dynamics Simulations (rCMDS), corrected by the room temperature air-broadening coefficients measured in the v₃ band of N₂O, to predict line-shape parameters including the broadening coefficient, its speed dependence component, the Dicke narrowing and the line-mixing coefficients, associated with various lineshape models for air-broadened N₂O absorption lines. For that, we first computed the autocorrelation functions of the N_2O dipole moment, responsible for the absorption transitions. The Fourier-Laplace transform of these functions directly yields the absorption spectra. Calculations were made for three temperatures 200, 250 and 296 K and 1 atm of N₂O/air mixtures and for a large range of values of the ratio between the Doppler and Lorentzian widths. The obtained spectra were then fitted with the quadratic speed-dependent Voigt profile including the first-order line mixing using a multi-spectrum fitting procedure. Comparisons with values determined from room temperature high-precision measurements in the v_3 band of N_2O by using the same profile [Loos et al, J. Quant. Spectrosc. Rad. Transf., 151, 2015, 300-309] show that our retrieved line broadening coefficients are overestimated by about 4%. The difference between the measured and predicted line broadening coefficients was used to empirically correct the rCMDS auto-correlation functions and thus the corresponding absorption spectra. These latter were then fitted with the Voigt, the speed-dependent Voigt, and the speed-dependent Nelkin-Ghatak profiles, all of them being associated with the first-order line-mixing approximation, providing the corresponding lineshape parameters for lines up to $J'' \leq 60$. The temperature dependences of various line-shape

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parameters were also deduced. The results show a very good agreement with available experimental data for all the considered parameters.

1. Introduction

Nitrous oxide (N_2O) is a long-lived atmospheric trace gas with a lifetime of about 150 years, playing an important role in radiative transfer and atmospheric chemistry [1-3]. N₂O is estimated to contribute for about 6% to the global warming effect and it is predicted to be the most important ozone-depleting emission in the twenty-first century [1,3,4]. The concentration of N₂O in the Earth's atmosphere has been (and still is) monitored by both ground-based and spaceborne instruments (e.g. [5-7]). Precise spectroscopic parameters of N₂O lines for wide ranges of pressure and temperature are needed to accurately retrieve atmospheric N₂O concentrations. Parameters characterizing refined effects on the line shape, beyond the usual collision-induced line broadening and shifting coefficients of the Voigt profile, were shown to have an important role for highaccuracy retrieval (e.g. [8,9]). For N₂O, most of the previous laboratory spectroscopic studies used the Voigt profile to model the line shapes. The most studied non-Voigt effect for N₂O is collisional line mixing in the Q branch of some perpendicular bands (e.g. [10-13]). In [14], line-mixing and speed-dependence effects were taken into account in the analysis of air-broadened N₂O spectra in the v₃ band, measured at room temperature with a Fourier Transform Spectrometer. This band was then measured and analyzed in [15] with the same line-shape model, the obtained line parameters being in good agreement with those of [14]. In [16] spectroscopic parameters for twenty N₂broadened N₂O lines in the 3v₃ band were retrieved from spectra measured with a comb-assisted cavity ring-down spectrometer using the speed-dependent Voigt profile. More recently, Odintsova et al [17] used several line-shape models including the speed-dependent Voigt and speeddependent Nelkin-Ghatak, to analyze eight self-broadened N₂O lines in the $3v_1+2v_2$ band. Further efforts are needed to provide complete sets of N₂O line-shape parameters and their temperature dependences for applications and populating spectroscopic databases.

While line-shape parameters are usually obtained from fits of laboratory measured spectra, they can also be predicted by theoretical calculations. Quantum scattering calculations were performed in order to compute the line shapes and predict line-shape parameters for H_2 , HD perturbed by He [18,19] and CO in Ar [20]. Because of their high computational cost, such calculations were mainly applied to simple systems. Thanks to the increasing computer power and

in conjunction with the availability of more accurate full-dimensional potential energy surfaces, this method will be probably applied to more complex molecular systems in the future. On the other hand, line-shape parameters for various molecular systems such as CO₂ perturbed by N₂ and O₂ [21,22], CO in various perturbers [23], O₂ in air [24] were successfully predicted using requantized classical molecular dynamic simulations (rCMDS). The principle of these calculations is very simple: based on the classical time evolution of a system containing a large number of molecules, one can compute the time dependent auto-correlation function of the dipole moment, responsible for the transition. The Fourier-Laplace transform of the latter directly yields the corresponding absorption spectrum. Using an adapted requantization procedure, one can deduce the absorption spectrum with the corresponding rotational structure. The simulated spectrum can then be fit with phenomenological line-shape models yielding the corresponding line-shape parameters for N₂-[21] and O₂-broadened [22] CO₂ lines, for instance.

In this work, rCMDS are used to obtain line-shape parameters and their temperature dependence for air-broadened N₂O lines. The intermolecular potential used and the computational method are described in Sec. 2. Since the rCMDS-predicted line broadening coefficients can differ from measured values by several percent, a correction procedure, as proposed for CO in [23], was applied, using a few experimental information from [14]. This correction procedure is presented in Sec. 3. The resulting rCMDS spectra were then analyzed using the usual Voigt, the speed-dependent Voigt as well as the speed-dependent Nelkin-Ghatak profiles, all including the first order line-mixing. This analysis procedure is described in Sec. 3 and the obtained line-shape parameters are presented and compared with existing experimental and theoretical data in Sec. 4. Conclusions and perspectives are given in Sec.5.

2. Calculations procedure and potential energy surface used

Classical molecular dynamics simulations (CMDS) were performed at 1 atm and three temperatures 200, 250 and 296 K for N₂O-air mixtures with 50% of air, which contains 80% of N₂ and 20% of O₂. The principle of these calculations can be found in [25] and some of our previous papers [21,26]. Here, only interactions between N₂O and N₂, N₂O and O₂ were authorized in the calculations. This is equivalent to simulating a system in which N₂O molecules are infinitely diluted in 0.5 atm of air. The molecules are considered as rigid rotors. A site-site functional form

with Coulombic and atom-atom contributions was used to represent the interaction potentials between N₂O and N₂, and between N₂O and O₂ molecules. The molecular geometries, the charges and sites were taken from [27] for N₂O and from [28] for N₂ and O₂ molecules. The atom-atom contributions were calculated as 12-6 Lennard-Jones potentials whose parameters were determined using the Lorentz-Berthelot combination rules [29] from parameters for N₂O-N₂O [27], N₂-N₂ and O₂-O₂ [28] interactions.

For each considered temperature, simulations were made for 3.10⁸ active molecules, which were distributed in thirty thousand boxes with periodic boundary conditions; each of them contains 2.10⁴ molecules. The size of each box is determined using the perfect gas law, from the number of molecules and the considered temperature and pressure. At the initial time, the center-of mass (c.o.m) position of each molecule was randomly chosen with the constraint that molecules should be separated from each other by distances of at least 9 Å, ensuring that no situation involving pairs of unphysically strongly interacting molecules is generated. This condition imposes the use of a temporization time which was 20 ps in the calculations. Translational and angular speeds verifying the Maxwell-Boltzmann distribution were initialized while random oriented c.o.m. velocity vector and molecular axis were attributed. The parameters (c.o.m. position, velocity, angular momentum, and molecular orientation) describing the classical state of each molecule were then computed for each time step up to about 1000 ps.

A requantization procedure, proposed in [26] for CO₂, was used here to requantify the classical rotation of the N₂O molecules. Specifically, for a molecule of rotational angular speed ω , we find the corresponding integer *J* for which $\hbar J/I$ (where *I* is the moment of inertia) is the closest to ω . This corresponds to matching the classical rotational frequency with the quantum position of the P branch lines. Once *J* is found, ω is requantized by changing it to $\omega = \hbar J/I$ while the angular orientation is kept unchanged. The auto-correlation function of the dipole moment, assumed along the molecular axis (as it is the case for the asymmetric and symmetric stretching absorption bands of N₂O) was then calculated during the requantized CMDS (rCMDS). The Doppler effect associated to the translational motion was taken into account in this auto-correlation function using Eq. (2) of [26].

For each considered temperature, eight values of the Doppler width Γ_D were used in the calculations, providing a broad range of values of the ratio Γ_L/Γ_D (from about 0.07 to 7), Γ_L being the Lorentz width. This enables to cover a wide range of situations, from the near purely Doppler

to collision-dominant regimes. Since all the effects of vibrational motion were disregarded in our rCMDS, from the point of view of collisional effects on the line shape, this corresponds to calculations for eight values of pressure with a unique Doppler width. The area-normalized absorption coefficient was obtained as the Fourier-Laplace transform of the auto-correlation function of the dipole moment. The spectra were thus calculated without the use of any adjustable parameter. Note that they do not show any pressure shift since our rCMDS do not take into account the dephasing of the dipole, which is associated with the fact that the effects of intermolecular interactions for molecules in the upper and lower states of the optical transitions are different.

3. Correcting the rCMDS and analysis procedure

As shown in [21,22,24], line-shape parameters predicted by rCMDS can be of several percent different from accurate measured values. Refined line-shape parameters such as the speed dependence of the broadening and the Dicke narrowing coefficient have a small impact on the line profile so that differences of several percent are completely negligible contrary to the line-broadening coefficients for which these differences can lead to large change on the calculated line shape. A correction is thus necessary. In a previous work [23], we demonstrated that the entire line-shape parameters set for a line can be obtained by correcting the rCMDS results using a value of the line-broadening coefficient measured at a single given pressure. Specifically, for each line, the difference between the rCMDS-predicted and the measured line broadenings is used to correct the auto-correlation functions calculated by rCMDS [23].

In this work, the measured values of [14] are retained for this correction procedure. In [14], air-broadened line-shape parameters for lines with $J'' \leq 40$ (J'' being the rotational quantum number of the lower level) in the v₃ band of N₂O were retrieved from FTS spectra measured at room temperature for a wide range of pressure (from 100 to 1000 mbar). The speed-dependent Voigt (sdV) profile including the first-order line mixing (LM) was used to fit the measured line shapes. To correct the rCMDS, we first fit the rCMDS-calculated spectra with the same line-shape model using a multi-spectrum fitting procedure. In the fits, the Doppler widths were fixed to values used in the simulations. For each line, the area and a linear base line were adjusted individually for each Doppler width while the other parameters were constrained to be the same for all spectra. The pressure-shift and its speed dependence were fixed to zero since no pressure shift effect is predicted by our rCMDS. Figure 1 shows a comparison between the line broadening coefficients,

 γ_0 , predicted by rCMDS (in black) and the measured values of [14]. As can be observed, rCMDSpredicted values are larger from 2 to 6% when compared to the experimental coefficients. Note that values for lines with J'' up to 60 were predicted by rCMDS while measurements were limited to J'' \leq 40. The measured values were thus extrapolated for J'' up to 60 using a Padé approximant, the corresponding values being also shown in Fig. 1. Measured values for J'' \leq 40 and extrapolated values for 40 < J'' \leq 60 were then used to correct the rCMDS results. Similarly to what was done in [23], for each transition, we multiply the computed dipole auto-correlation functions by $\exp(-2\pi cP\Delta\gamma t)$ in which $\Delta\gamma = \gamma_0^{meas} - \gamma_0^{rCMDS}$, c being the speed of light in vacuum and P the pressure of the calculation. The Fourier-Laplace transform of these corrected auto-correlations functions is again computed providing the corrected rCMDS-calculated spectra for the considered transition. Note that the correction procedure is applied to the calculations performed at room temperature only, the temperature dependences of the line-shape parameters being deduced from non-corrected rCMDS spectra.

The three line-shape models: Voigt (V), speed-dependent Voigt (sdV) [30] and speeddependent Nelkin-Ghatak (sdNG) profiles [31,32] were retained to analyze the corrected rCMDScalculated spectra. The quadratic speed dependence model [33,34] is used to represent the speed dependence of the line broadening in the sdV and sdNG profiles. In all cases, line-mixing effects were also taken into account through the use of the first-order approximation [35]. Our results show that the sdNG and sdV profiles lead to very similar fit residuals, both in much better agreement with the simulated spectra than the Voigt profile (see Fig. 2 for instance).

The temperature dependences of the different line-shape parameters were obtained directly from fits of rCMDS spectra without any corrections. They were modeled using the usual power law, i.e. $A(T) = A(T_0) \left(\frac{T_0}{T}\right)^{n_A}$ with $T_0 = 296$ K, A(T) being either the line broadening coefficient, the speed dependence component, the Dicke narrowing or the line-mixing parameter. From the values of A(T) obtained from fits of the (non-corrected) rCMDS-calculated spectra at 200, 250 and 296 K, the temperature exponents n_A were obtained for each line. All the obtained parameters and their temperature dependences are listed in Table 1, and presented in the next section. The uncertainties of the rCMDS-deduced values are not reported here since it is almost impossible to accurately determine them. Indeed, this would require the determination of those due to the used intermolecular potential, to the numerical approaches used in the rCMDS, and to the retrieval of the parameters searched for in the fits.



Fig.1: Top: Comparison between line broadening coefficients deduced from fits of rCMDScalculated spectra with the sdV and first-order LM model (black circles) and values of [14] (blue squares) obtained from fits of FTS measured spectra with the same line-shape model. Bottom: the ratio of the rCMDS-predicted broadening coefficients and measured values. m is the rotational quantum number in the lower level of the lines, m = -J'' in the P branch.



Fig. 2: (Top panel) Example of spectra calculated from the corrected rCMDS auto-correlation functions for the P(18) line of N_2O broadened by air at T=296 K and $P_{air} = 0.5$ atm and for three values of the Doppler width $\Gamma_D = 0.0046$, 0.0382 and 0.0764 cm⁻¹ leading to three values of $\Gamma_L/\Gamma_D = 8.49$, 1.02 and 0.51, respectively. (Bottom) The corresponding residuals obtained from multi-fitting these calculated spectra using the V, sdV and sdNG profiles, LM effect was also taken into account.

4. Line-shapes parameters and their temperature dependences

3.1. Air-broadening coefficients

The room temperature air-broadening coefficients, γ_0 , obtained from multi-fitting the rCMDS spectra, calculated and corrected as described previously, with the V, sdV, sdNG profiles together with the first-order line mixing, are plotted in Fig. 3. Because of the limited signal-to-noise ratio of the calculated spectra (from 500 to 3000, depending on the line and on the Doppler width used), we could not determine reliable line-shape parameters for lines with |m| greater than 60. Comparisons between the coefficients obtained with the three profiles show that the values obtained with the V profile are, on average, 4% smaller than those obtained with the sdV and sdNG models. To the best of our knowledge, the speed-dependence of the line width for air-broadened

N₂O lines was investigated in [14] and [15] only, both for the ν_3 band. Since our rCMDS are corrected using the experimental values of [14], our rCMDS-deduced values for γ_0 using the sdV profile are almost identical to those of [14] (see Table 1 of the present paper and Table 3 of [14]). Note that the latter were retained in the last version of the HITRAN database [36]. The results of [15] are plotted in Fig. 3, showing a very good agreement with an average difference of about 1% between our sdV values and those of [15].



Fig.3: Room temperature air-broadening coefficients retrieved from the corrected-rCMDS calculations (PS – present study) with the Voigt profile (upper panel) and with the sdV and sdNG profiles (lower panel) versus |m|. Measured values from [37-40] obtained with the Voigt profile and those of [15] with the sdV model are also plotted for comparison.

Our Voigt values are compared with those of [37-39] (Fig. 3), obtained from fits of measured spectra in the v_3 band using the same profile. Since the vibrational dependence of the line broadening for N₂O is likely negligible, the measured data of [40], obtained with the Voigt profile for the $3v_3$ band are also plotted in Fig. 3 for comparison. For |m| < 40, a very good

agreement is observed for all the considered datasets with a standard deviation smaller than 2%, except for that of [40]. The differences are slightly larger for |m| > 40. As mentioned in [37], [38] and [39], the uncertainties of the measured values were estimated to be about 4%, 2% and 3%, respectively, numbers which are of the same order as their differences with our rCMDS-deduced values. For instance, the average difference between our values and those of [38] is 2% only.

The obtained temperature exponents for γ_0 are shown in Fig. 4 together with the experimental values of [37,39] and the constant value (i.e. 0.75) adopted in HITRAN2016 [36]. As can be observed, the calculated values obtained with the three considered line shapes are very close to each other. For small values of |m|, the predicted results are in good agreement with those of [39]. Both show a rather linear rotational dependence of n_{γ_0} while it is a sinusoidal variation in [37]. At high |m|, the predicted values of n_{γ_0} are smaller than the experimental ones by about 0.1 (which is, as estimated in [39], the uncertainty of [37]).



Fig. 4: The temperature dependence exponent, n_{γ_0} , of the collisional line broadenings, determined in this work and comparison with the experimental values of [37], [39] and the constant value (0.75) adopted in HITRAN2016 [36].

3.2. The speed dependence of the line width

Room temperature values of the speed dependence coefficients, γ_2 , retrieved from the corrected-rCMDS spectra using the sdV and sdNG profiles are listed in Table 1 and plotted in Fig. 5 (the corresponding ratio γ_2/γ_0 being plotted in the lower panel) where they are compared with those of [14] and [15]. In [14], a value of γ_2 for each line was retrieved while in [15] a constant value of 0.0895 for γ_2/γ_0 was used for all the lines. An excellent agreement between our values and those of [14] is observed for transitions with |m| < 20 while, for higher rotational quantum numbers, the discrepancy becomes larger. Due to lower signal to noise ratios for lines with |m| < 110 and |m| > 45 and to the numerical correlations between the various line-shape parameters (between γ_2 and β for instance), larger dispersion is observed for these lines when using the sdNG profile. Note that within the quadratic approximation for the speed-dependent broadening and the power law for the temperature dependence of the line broadening coefficient, γ_2/γ_0 can be predicted by using Eq. (13) of [41]. The HITRAN values of the temperature exponent n_{γ_0} (constant and equal to 0.75) were thus used to deduce the corresponding ratio. However, as shown by Fig. 5, this prediction largely underestimates both experimental [14,15] and our rCMDS-deduced values. Better agreement is obtained when using the average value of n_{γ_0} predicted by rCMDS (about 0.7).



Fig. 5: Room temperature rCMDS-deduced values of the speed-dependent coefficient, γ_2 (top panel), and the γ_2/γ_0 ratio (bottom panel), for air-broadened N₂O, associated with the sdV (black squares) and sdNG (blue diamonds) profiles and comparison with experimental values of [14] (green crosses), [15] (red line) as well as values calculated using Eq. (13) of [41] and the temperature exponent (0.75) from HITRAN2016 [36] (dashed pink line) or the mean value (0.7) of our predicted n_{γ_0} (dashed olive line).

The predicted temperature dependence exponents of the speed dependence coefficients are plotted in Fig. 6. The large dispersion of n_{γ_2} is due to the relatively weak influence of γ_2 on the line shape and the limited signal to noise ratio of the calculated spectra. Since no rotational dependence can be observed, we recommend to use the average value of 0.51 for n_{γ_2} .



Fig. 6: Temperature exponents of the speed dependent coefficient of the line width

3.3. The Dicke narrowing parameter

The obtained Dicke narrowing coefficients, β , and their temperature dependences are presented in Fig. 7. To the best of our knowledge, there are no experimental nor other theoretical determination of this parameter for N₂O in air using the sdNG model. Large and random rotational variations of β and of its temperature dependence exponent are obtained, likely due to the weak influence of this parameter on the line shapes. We recommend the use of their averaged values: $\beta = 3.72 \times 10^{-3}$ cm⁻¹/atm and $n_{\beta} = 1$.



Fig. 7: The Dicke narrowing parameter, β , obtained from fits of the corrected-rCMDS calculated spectra with the sdNG profile, and its temperature dependence exponent n_{β} . The averaged values $\beta = 3.72 \times 10^{-3} \text{ cm}^{-1}/\text{atm}$ and $n_{\beta} = 1$ are also plotted.

3.4. The first-order line-mixing parameter

The first-order line-mixing coefficients, ζ , and their temperature dependence exponents n_{ζ} are plotted in Figs. 8 and 9, respectively. As can be observed, ζ is independent of the line profile used in the fits of the rCMDS-calculated spectra. The values of ζ obtained at room temperature are compared with the measured values of [14,15] as well as with those modeled by an empirical law in [15], showing good agreements. The absolute values of ζ , deduced from rCMDS are slightly larger than the experimental ones, especially at very low and high values of |m|. To the best of our knowledge, there is no measured nor previous calculated result for the temperature dependence exponents of ζ for air-broadened N₂O lines. However, the evolution of n_{ζ} with the rotational quantum number is very similar to that obtained from experiments and calculations for CO_2 in N_2 [21,42] and in air [22].



Fig. 8: First-order line-mixing coefficients deduced from spectra calculated using roomtemperature corrected-rCMDS and comparison with the measured values of [14,15] and those calculated from an EPG (Exponential Power Gap) model [15].



Fig. 9: The rCMDS-deduced temperature dependence exponent of the first-order line-mixing coefficient.

4. Conclusion

Line-shape parameter including the broadening coefficient, its speed dependent component, the Dicke narrowing and the line-mixing coefficients for air-broadened N₂O lines were predicted using requantized Classical Molecular Dynamics Simulations (rCMDS) corrected using room temperature broadening coefficients measured in the v_3 band. Except for the pressure-induced line shift, the entire line-shape parameter sets, corresponding to the Voigt, the speed-dependent Voigt and the speed-dependent Nelkin-Ghatak profiles including first-order line mixing are provided. The obtained line-shape parameters (except for the first order line-mixing coefficient which is branch-dependent) are valid for both the *P* and *R* branches. Since the vibrational dependence of the line-broadening coefficients of N₂O, as for CO₂, is likely small, these data can be used for other vibrational bands for which no accurate and consistent experimental or theoretical

data are available. Considering the larger dispersion of the results obtained with the sdNG profile (see Figs. 5, 6 for instance) and the fact that fit residuals obtained with this profile are not significantly better than those obtained with the sdV, we recommend the use of the sdV and the associated line-shape parameters for N₂O in air. Comparisons between our predicted parameters and available experimental data show good agreement. These data are certainly useful to address the precision and accuracy requirements of application as well as for populating spectroscopic databases.

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Line			sdV				,	V		sdNG					
	γ_0	n_{γ_0}	γ_2	ζ	n_{ζ}	γ ₀	n_{γ_0}	ζ	n_{ζ}	γ_0	n_{γ_0}	γ_2	ζ	n_{ζ}	
P(1)	100.80	0.82	9.77	-32.56	0.99	97.11	0.660	-29.89	0.81	99.73	1.22	5.46	-32.40	1.16	
P(2)	95.95	0.71	9.32	-26.62	0.69	92.40	0.714	-25.14	0.76	95.57	0.71	7.99	-26.59	0.69	
P(3)	93.77	0.71	10.34	-17.31	0.73	89.68	0.718	-16.83	0.78	91.94	0.70	1.23	-17.18	0.72	
P(4)	92.09	0.70	9.92	-16.52	0.48	88.20	0.700	-15.79	0.52	91.21	0.69	6.81	-16.48	0.48	
P(5)	90.58	0.70	9.52	-13.90	0.93	86.86	0.709	-13.34	0.97	90.23	0.69	8.37	-13.89	0.93	
P(6)	89.22	0.69	9.79	-11.45	0.60	85.34	0.694	-10.98	0.62	88.56	0.68	7.60	-11.43	0.60	
P(7)	88.12	0.69	9.49	-10.01	0.56	84.37	0.696	-9.61	0.59	88.10	0.68	9.42	-10.01	0.55	
P(8)	87.06	0.69	9.39	-10.03	0.71	83.35	0.693	-9.73	0.74	86.79	0.67	8.53	-10.02	0.70	
P(9)	85.43	0.68	9.70	-6.41	0.37	81.54	0.689	-6.14	0.40	84.58	0.67	6.80	-6.41	0.36	
P(10)	84.95	0.69	9.82	-5.87	0.94	80.99	0.696	-5.62	0.97	84.21	0.69	7.41	-5.86	0.94	
P(11)	83.62	0.68	9.35	-4.02	0.52	79.88	0.694	-3.86	0.55	83.24	0.69	8.15	-4.02	0.52	
P(12)	82.80	0.69	9.37	-2.75	0.45	79.03	0.697	-2.62	0.46	82.33	0.68	7.86	-2.76	0.45	
P(13)	82.04	0.68	9.02	-3.40	0.63	78.44	0.696	-3.28	0.63	81.70	0.68	7.93	-3.40	0.62	
P(14)	81.04	0.68	9.09	-1.46	-1.91	77.40	0.690	-1.38	-1.96	80.37	0.68	6.83	-1.46	-1.91	
P(15)	80.14	0.68	8.95	0.56	-0.87	76.55	0.689	0.60	-1.01	79.53	0.69	6.92	0.56	-1.71	
P(16)	79.37	0.69	8.91	0.74	0.89	75.79	0.694	0.78	0.55	78.97	0.68	7.63	0.74	0.82	
P(17)	78.82	0.69	9.02	2.05	4.01	75.17	0.698	1.94	3.94	78.39	0.68	7.68	2.04	4.0	
P(18)	78.11	0.68	8.68	1.48	1.81	74.62	0.694	1.46	1.98	77.65	0.68	7.19	1.48	1.82	
P(19)	77.43	0.68	8.92	2.13	3.19	73.81	0.692	2.11	3.26	76.94	0.68	7.36	2.12	3.19	
P(20)	76.61	0.68	8.57	2.36	2.19	73.16	0.694	2.31	2.20	76.24	0.68	7.37	2.35	2.20	
P(21)	76.12	0.69	8.52	3.61	1.80	72.68	0.700	3.55	1.86	75.74	0.68	7.29	3.61	1.80	
P(22)	75.65	0.68	8.43	4.70	1.68	72.26	0.698	4.58	1.66	75.09	0.69	6.52	4.70	1.68	
P(23)	75.45	0.69	8.28	4.63	1.19	72.12	0.700	4.49	1.24	75.18	0.69	7.43	4.63	1.19	
P(24)	75.05	0.69	8.07	5.88	1.65	71.83	0.700	5.77	1.68	74.71	0.68	6.93	5.87	1.64	
P(25)	74.60	0.68	8.18	6.08	1.14	71.32	0.694	5.86	1.18	74.21	0.68	6.87	6.08	1.14	
P(26)	73.81	0.69	8.05	6.56	1.79	70.58	0.707	6.32	1.86	73.41	0.70	6.69	6.56	1.79	
P(27)	73.45	0.69	8.20	7.07	0.92	70.13	0.705	6.84	0.97	72.97	0.69	6.59	7.07	0.92	

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P(28)	73.05	0.69	7.93	7.43	1.09	69.87	0.71	7.22	1.12	72.65	0.69	6.59	7.43	1.09
P(29)	72.83	0.69	7.83	6.86	1.38	69.69	0.71	6.64	1.41	72.41	0.69	6.41	6.85	1.37
P(30)	72.51	0.70	7.70	8.73	1.28	69.44	0.71	8.52	1.34	72.01	0.69	5.90	8.73	1.29
P(31)	72.30	0.71	7.75	8.11	1.16	69.20	0.72	7.90	1.19	71.83	0.69	6.09	8.12	1.16
P(32)	72.10	0.69	7.49	7.73	0.66	69.12	0.71	7.51	0.70	71.89	0.70	6.76	7.72	0.66
P(33)	71.96	0.70	7.93	9.36	1.37	68.76	0.72	9.12	0.14	71.34	0.69	5.74	9.34	1.36
P(34)	71.67	0.69	7.74	9.17	1.16	68.56	0.71	8.95	1.18	71.20	0.69	6.08	9.15	1.16
P(35)	71.42	0.70	7.38	9.55	0.50	68.49	0.72	9.25	0.54	71.08	0.70	6.20	9.55	0.28
P(36)	71.30	0.70	7.30	10.01	1.12	68.41	0.71	9.69	1.17	70.73	0.69	5.17	10.01	1.11
P(37)	71.00	0.70	7.57	10.27	0.94	67.97	0.71	10.00	1.01	70.43	0.70	5.49	10.26	0.94
P(38)	71.00	0.70	7.63	10.17	-0.06	67.94	0.71	9.69	-0.02	70.43	0.70	5.55	10.17	-0.07
P(39)	70.76	0.71	6.97	10.54	0.11	68.02	0.74	10.16	0.16	70.33	0.71	5.34	10.53	0.10
P(40)	70.77	0.71	7.24	11.21	0.37	67.90	0.73	10.78	0.44	70.14	0.72	4.75	11.18	0.37
P(41)	70.05	0.69	6.91	12.69	0.85	67.34	0.72	12.21	0.91	69.80	0.71	5.99	12.69	0.86
P(42)	69.86	0.71	6.88	8.77	0.76	67.15	0.73	8.49	0.81	69.61	0.71	5.96	8.77	0.76
P(43)	69.66	0.73	7.19	11.42	1.21	66.80	0.75	10.99	1.21	69.18	0.71	5.44	11.40	1.21
P(44)	69.49	0.71	6.70	13.98	0.20	66.72	0.73	13.48	0.33	69.00	0.72	5.14	13.96	0.20
P(45)	69.23	0.72	6.63	8.53	0.40	66.65	0.72	8.10	0.47	69.11	0.71	6.18	8.53	0.38
P(46)	69.14	0.75	6.61	13.44	3.47	66.56	0.76	13.03	2.86	69.20	0.76	6.80	13.44	2.76
P(47)	68.89	0.69	6.57	11.05	-1.20	66.31	0.69	10.69	-0.90	68.83	0.67	6.37	11.05	-0.91
P(48)	69.10	0.71	7.11	8.62	0.57	66.28	0.74	8.26	0.62	68.05	0.74	1.97	8.60	0.62
P(49)	68.77	0.73	5.98	16.22	1.81	66.48	0.76	15.72	1.97	69.61	0.74	8.60	16.26	1.95
P(50)	68.29	0.72	6.78	14.88	1.63	65.61	0.74	14.46	1.28	67.74	0.68	4.62	14.84	1.17
P(51)	68.32	0.66	5.33	11.74	-1.12	66.33	0.73	11.26	-2.09	68.92	0.81	7.44	11.71	-2.17
P(52)	67.94	0.73	6.33	14.33	1.27	65.48	0.72	13.62	1.52	67.51	0.77	4.54	14.33	1.65
P(53)	68.08	0.77	6.62	11.94	1.91	65.46	0.73	11.50	2.49	67.84	0.71	5.75	11.93	2.16
P(54)	67.86	0.69	7.43	10.40	-0.46	64.86	0.69	10.20	-0.33	66.79	0.62	2.96	10.28	-0.41
P(55)	67.51	0.67	5.58	16.64	1.77	65.40	0.71	16.06	1.59	67.38	0.77	5.04	16.64	1.65
P(56)	67.59	0.78	6.32	12.32	-1.25	65.12	0.77	12.05	-0.73	67.12	0.72	4.38	12.32	-0.64

P(57)	66.84	0.83	8.09	17.52	-3.11	63.50	0.77	16.56		65.61	0.90	-5.44	17.25	-8.50
P(58)	67.30	0.40	6.63	16.02	4.18	64.69	0.60	15.68	5.42	66.33	0.64	1.62	15.85	5.24
P(59)	66.42		7.59	10.72		63.31		9.85						

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Table 1: Air-broadened line-shape parameters for N₂O lines, deduced from fits of rCMDS-calculated spectra with the V, the sdV and the sdNG profile, all of them including the first-order line mixing. γ_0 , γ_2 and ζ are respectively the broadening coefficient, the speed dependence coefficient of the line broadening (both in 10⁻³ cm⁻¹/atm) and the first-order line-mixing coefficient (in 10⁻³ atm⁻¹) at room temperature (296K). n_{γ_0} and n_{ζ} are the temperature dependence exponents of the line broadening and mixing coefficients, respectively. The Dicke narrowing coefficient, β , and its temperature dependence exponent, n_{β} , (for the sdNG profile) as well as the temperature dependence exponent of γ_2 , n_{γ_2} , are independent of the rotational quantum number; $\beta = 3.72 \ 10^{-3} \ \text{cm}^{-1}/\text{atm}$, $n_{\beta} = 1$ and $n_{\gamma_2}=0.51$ (see text).

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