



HAL
open science

Bethe–Salpeter Study of the Optical Absorption of trans and cis Azobenzene-Functionalized Metal–Organic Frameworks Using Molecular and Periodic Models

Aseem Rajan Kshirsagar, Claudio Attaccalite, Xavier Blase, Jing Li, Roberta Poloni

► To cite this version:

Aseem Rajan Kshirsagar, Claudio Attaccalite, Xavier Blase, Jing Li, Roberta Poloni. Bethe–Salpeter Study of the Optical Absorption of trans and cis Azobenzene-Functionalized Metal–Organic Frameworks Using Molecular and Periodic Models. *Journal of Physical Chemistry C*, 2021, 125 (13), pp.7401-7412. 10.1021/acs.jpcc.1c00367 . hal-03359630

HAL Id: hal-03359630

<https://hal.science/hal-03359630>

Submitted on 30 Sep 2021

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

Bethe-Salpeter Study of the Optical Absorption of *trans* and *cis* Azobenzene-Functionalized Metal-Organic Frameworks using Molecular and Periodic Models

Aseem Rajan Kshirsagar,[†] Claudio Attaccalite,[‡] Xavier Blase,[¶] Jing Li,[§] and Roberta Poloni^{*,†}

[†]Univ. Grenoble Alpes, CNRS, Grenoble-INP, SIMaP, 38000 Grenoble, France

[‡]CNRS/Aix-Marseille Université, Centre Interdisciplinaire de Nanoscience de Marseille UMR 7325, Campus de Luminy, 13288 Marseille Cedex 9, France

[¶]Univ. Grenoble Alpes, CNRS, Institut Néel, 38042 Grenoble, France

[§]Univ. Grenoble Alpes, CEA, Leti, 38000 Grenoble, France

Received January 15, 2021; E-mail: roberta.poloni@grenoble-inp.fr

Abstract: The optical absorption spectra of the azobenzene-functionalized metal-organic framework, PCN-123, are calculated in *cis* and *trans* configurations using the Bethe-Salpeter equation (BSE) formalism and the *GW* approximation using periodic and non-periodic models. In the visible, near-UV and mid-UV region the optical excitations in the MOF are associated with the azobenzene functionalities and this results in spectral features similar to the case of the gas phase azobenzene and the azo-functionalized ligand. The most noticeable difference is the significantly more intense S_1 band for *cis* in the MOF as compared to the free molecules which points to a faster and more complete *cis*→*trans* isomerization in the framework, with strong implications for the design of MOFs with high photoconversion efficiencies. Consistent with these findings, all the molecular models employed to represent the MOF, including the smallest, are found to yield a reasonable description of the low energy optical spectra (between 2 and 5 eV) of the periodic framework, with the exception of the stronger S_1 band of *cis* in the MOF, a feature that we attribute to a limitation of the fragment model to correctly represent the wavefunction of the extended framework.

Introduction

Due to their controllable chemical structure and porosity, metal-organic frameworks (MOFs) have attracted much attention during the past 20 years for a wide range of applications such as chemical sensing,¹ gas storage,² gas separation,³ catalysis,⁴ electronics,⁵ optoelectronics,^{6,7} and energy storage devices.⁸ The high internal surface area, pore volume, and selective affinity^{9,10} for CO₂ make MOFs among the best candidate materials for CO₂ capture.^{11–15} The main issue associated with carbon capture technologies is that the adsorption and desorption cycles, performed for example by using temperature swing adsorption, are associated with large energy penalties and therefore high operating costs.¹⁶

A fascinating strategy proposed to perform more energy-efficient adsorption/desorption cycles is the use of light-responsive frameworks. In these MOFs, the desorption of certain gases can be obtained using molecular photo-switches. Specifically, the use of light excitation, preferably in the visible range, rather than temperature for example, can enable a more energy efficient separation process, because the use of thermal energy can be fully or partially

by-passed in favour of filtered sunlight, thus potentially reducing the cost of the separation/capture process.

The first work showing a reversible change in gas uptake upon light irradiation was reported in 2012 by Park et al. for PCN-123 MOF.¹⁷ By functionalizing the ligands of MOF-5¹⁸ with azobenzene moieties, the authors reported a significant change in CO₂ uptake upon UV-Vis irradiation possibly as a consequence of the photoisomerization of the azobenzenes anchored to the MOF scaffold. In the following years, a few experimental studies confirmed that the change in the gas adsorption properties in other photoresponsive azobenzene-functionalized MOFs is indeed related to different fractions of *cis* and *trans* isomers within the MOF.^{19–23} In the case of PCN-123, the microscopic mechanism behind the reversible adsorption and desorption of CO₂ was later identified by some of us to be the blocking and unblocking of the metal node, i.e. the strongest adsorption site for CO₂, by *cis* and *trans* configurations, respectively.²⁴ Larger photoisomerization yields were predicted to result in larger differences in gas uptake²⁴ between adsorption and desorption thus contributing to enhance the efficiency of the process. As such, designing MOFs with more efficient photoswitching properties can be considered as one of the central focuses in this field.

In the case of solvated azobenzene molecules, large photo-switching yields are obtained by exploiting the selective absorption of light by *cis* or *trans* at specific wavelengths.^{25–28} Typically, *cis* → *trans* photo-switching is achieved upon irradiation with visible light ($\lambda \sim 420$ nm) by exciting the S_1 peak (i.e. the $n \rightarrow \pi^*$ transition) where *cis* has a larger molar absorption than *trans*, and *viceversa*, the *trans* → *cis* photo-switching is performed by exciting with UV light ($\lambda \sim 325$ nm) the intense S_2 ($\pi \rightarrow \pi^*$) of *trans*.²⁶ In the case of MOFs, a few experimental studies have shown that the main spectral features of functionalized photoresponsive MOFs resemble the ligand case in the UV-Vis region and that photoisomerization does occur in the MOF.^{19,20,23,29,30} However, due to the lack of computational studies on the subject, the nature of the excitations in the MOF and their similarity with the molecular case have not been established yet. This represents the first open question that we want to address in this work. By studying to which extent the optical properties of the functionalizing gas phase molecule or the ligand resemble the periodic MOF we may validate the use of small MOF fragments for future studies. Besides, the resemblance of the character of the excitations between the MOF and the molecular counterpart (azobenzene here) has strong implications on the mechanism of isomerization

and thus the quantum yield of isomerization.

Despite MOFs being crystalline solids, their electronic bands often feature small or negligible dispersion.³¹ This has motivated several authors to employ molecular fragment models to study the electronic and optical properties of MOF crystals.^{32–40} Using fragments of MOFs, if justified, can be advantageous as it permits to employ accurate wavefunction methods such as coupled-cluster, configuration-interaction, or multi-configurational methods with reasonable computational expense. The second aspect that we want to address is the accuracy of the fragment approach to study the optical absorption properties of this class of materials and this is done here for the paradigmatic case of PCN-123. We employ the Bethe-Salpeter equation (BSE) formalism^{41–50} to describe the two-body electron-hole bound state and the *GW* approach^{44,47,48,51–53} to compute quasiparticle energies. Compared with the more standard time-dependent density-functional theory⁵⁴ (TD-DFT), the BSE/*GW* formalism exhibits a few advantages. The *GW* eigenvalues used for constructing BSE Hamiltonian are more accurate than DFT Kohn-Sham (KS) eigenvalues. Then, the non-local screened Coulomb interaction, W , provides an accurate description of the non-local Coulomb interaction between the electron and hole, of importance for charge-transfer or Rydberg excitations.⁵⁵ The latter of the two is possibly the reason behind the success of BSE in accurately describing excitations of different types.^{56–58}

First, we assess the validity of a molecular model by employing three different choices of fragments with a progressively increasing number of metal nodes and azobenzene functional groups, and we compare the results with the corresponding fully-functionalized periodic MOF. A comparison is performed at the DFT level using Perdew-Burke-Ernzerhof (PBE) parametrized semilocal exchange-correlation (XC) functional,⁵⁹ at the *GW* quasiparticle level, and *in fine* using BSE to study the optical spectra and the character of the excitons. We find that the BSE/*GW* spectra of *cis* and *trans* PCN-123 closely resemble those of the free azobenzene molecule. The BSE/*GW*/PBE results show that all fragments, including small ones, yield optical spectra in good agreement with the periodic result between 2 and 5 eV ($\sim 250 - 600$ nm), i.e. the region where absorption is dominated by the azobenzene functionalities. Fragment models are used also to compute the *cis* and *trans* spectra using BSE/*GW*/PBEh, where the PBEh is the PBE global hybrid functional and the amount of exact exchange (EXX) in PBEh⁶⁰ is tuned have a DFT starting point fulfilling the ionization potential (IP) condition.⁶¹ This choice is motivated by our recent study⁶² where we showed that this allows to achieve a good quantitative agreement with experimental S_1 for a series of solvated azobenzene derivatives. Besides providing the spectral region where the largest selective absorption of each isomer occurs, the calculated spectra point to the possibility of a more efficient *cis* \rightarrow *trans* back isomerization in the MOF due to the significantly more intense S_1 (i.e. $n \rightarrow \pi^*$) of *cis* azobenzene in the framework, as compared to gas phase of solvated azobenzene.

Computational methods

The *GW* method is adopted to correct the KS eigenvalues by using a self-consistent procedure on the eigenvalues^{63,64} described below. The KS eigenvalues are corrected pertur-

batively, by replacing the exchange-correlation contribution to the KS eigenvalues by the expectation value of *GW* self-energy operator $\Sigma(r, r', E)$ onto the KS eigenstates, as follows:

$$\epsilon_n^{GW} = \epsilon_n^{KS} + \langle \phi_n^{KS} | \Sigma^{GW}(\epsilon_n^{GW}) - V^{XC} | \phi_n^{KS} \rangle, \quad (1)$$

Here, the subscript n corresponds to n^{th} KS state, the superscripts KS and *GW* denote the level of theory, $|\phi_n^{KS}\rangle$ is the n^{th} KS eigenstate, and V^{XC} is the DFT XC potential. For the first iteration, G_0W_0 , the $\Sigma(r, r', E)$ is computed using the input DFT eigenstates $\epsilon_n^{KS}, \phi_n^{KS}$. In the subsequent iteration, the $\epsilon_n^{G_0W_0}$ replace ϵ_n^{KS} and so forth until the quasiparticle gap is converged within 0.1 eV to yield the so-called *evGW* quasiparticle energies. Only a limited number of frontier eigenvalues are corrected explicitly at *GW* level and the rest are corrected using the scissor operator. See below for more details.

The *evGW* energies and the KS eigenvectors $\epsilon_n^{evGW}, \phi_n^{KS}$ are then used to build the BSE Hamiltonian. Within the BSE formalism, the optical excitations are computed by solving a linear algebra problem similar to the Casida equation of TD-DFT.⁵⁴ The BSE eigenvalue problem can be written compactly as:

$$\begin{bmatrix} R & C \\ -C^* & -R^* \end{bmatrix} \begin{bmatrix} X_\mu \\ Y_\mu \end{bmatrix} = \Omega_\mu \begin{bmatrix} X_\mu \\ Y_\mu \end{bmatrix} \quad (2)$$

where R (R^*) represents the resonant coupling between electron-hole excitations (de-excitations), whereas, the off-diagonal blocks C and C^* describe the non-resonant coupling between excitations and de-excitations. X_μ and Y_μ are the excitation and de-excitation components, respectively, of the two-body electron-hole (*eh*) eigenstate $\Psi_\mu(r_e, r_h)$ expressed in the transition basis $[\phi_v(r_h)\phi_c(r_e)]$ and μ is the index of excitation, with ϕ_v and ϕ_c representing the KS valence and conduction states. Thus Ω_μ are the eigenvalues of BSE, i.e. the excitation energies, and the corresponding eigenvectors $\Psi_\mu(r_e, r_h)$ are written as:

$$\Psi_\mu(r_e, r_h) = \sum_{vc} [X_\mu^{vc} \phi_v(r_h)\phi_c(r_e) + Y_\mu^{vc} \phi_c(r_h)\phi_v(r_e)] \quad (3)$$

The eigenvectors yield the probability amplitude of finding the electron and hole at r_e and r_h , respectively. To compare the character of the exciton between the different approaches, we compute and plot for each relevant excitation, μ , the hole-averaged electron density and charge-averaged hole density as follows:

$$\rho_\mu(r_e) = \int dr_h \Psi_\mu(r_e, r_h) \Psi_\mu^*(r_e, r_h) \quad (4)$$

$$\rho_\mu(r_h) = \int dr_e \Psi_\mu(r_e, r_h) \Psi_\mu^*(r_e, r_h) \quad (5)$$

Periodic calculations

The DFT calculations are performed using Quantum ESPRESSO.^{65,66} Optimized norm-conserving Vanderbilt pseudopotentials obtained from PSEUDO-DOJO vault are used to model the interaction between ionic core and the valence electrons.^{67,68} The geometry of PCN-123 MOF was optimized using the rev-vdW-DF2 functional⁶⁹ by taking the global minimum configuration for *trans* and *cis* found in our previous work.²⁴ Specifically, these geometries are obtained by performing a full geometrical optimization using

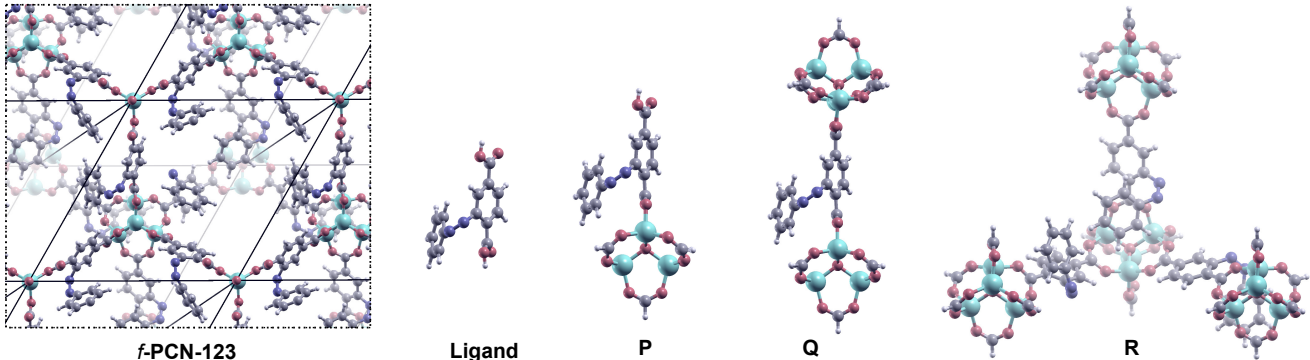


Figure 1. Illustrations of the unit cell of the PCN-123 MOF, the PCN-123 ligand, and fragments P, Q and R, in *cis* configuration. Colour code as follows, C: gray, N: blue, H: white, Zn: cyan, O: red. The periodic MOF is shown by aligning the [111] direction perpendicular to the plane; the black lines outline the boundaries of primitive unit cells of the PCN-123 crystal.

the primitive cell of PCN-123 which contains six ligands and a total of 178 atoms, including the azobenzene functionalities. In *cis*, all molecular functionalizations were optimized in the *cis* 2 configuration discussed in Ref. 24, i.e. with each azobenzene ring blocking one metal node. The DFT eigenvalues and eigenstates for excited state calculations are obtained using PBE XC functional.⁵⁹ Kinetic energy cut-off of 80 Ry and 320 Ry is used for truncation of basis set representing Kohn-Sham orbitals and Kohn-Sham charge density respectively. Reciprocal space is sampled using Γ point only which is found to be sufficient in absence of band dispersion.

The BSE/*GW* calculations are performed using YAMBO.^{70,71} To avoid escalation of computational expenses, the frequency dependence of the dielectric matrix is approximated using Godby-Needs plasmon-pole model.⁷² For the calculation of dielectric screening, a total 3000 KS states are used including the 380 valence states. Out of all the KS states, 128 valence states and 128 conduction states are corrected explicitly by *GW* and used to build the BSE Hamiltonian. To reduce the cost, only the resonant coupling between the electron and hole is considered for the BSE. For the same reason, the sizes of the plane-waves basis used for different steps in BSE and *GW* are truncated at lower kinetic energy cut-offs than the cut-off used for DFT. Details of these truncation parameters and their convergences are provided in Tab. S1 of the supplementary information (SI). Bruneval-Gonze terminator is used to speed-up the convergence of correlation part of the *GW* self-energy with respect to sum over the number of unoccupied states.⁷³

Fragment calculations

The geometries of the fragments of PCN-123 are constructed by cleaving parts of the fully functionalized (i.e. every ligand is functionalized) and fully optimized PCN-123 MOF²⁴ by terminating unsaturated bonds with H atoms. Because we focus on the low energy excitations which are possibly associated with the azobenzene functionalization, the fragments are constructed so as to keep the chemical connectivities of the ligand intact. In our earlier work, the structure of the azobenzene functionalization in PCN-123 was found to be sensitive to the MOF milieu due to steric factors and strong van der Waals interactions.²⁴ To not deviate from the structure of periodic PCN-123, no further geometrical optimiza-

tion is performed on the fragment except for the relaxation of the coordinates of the added H terminations.

Three different fragment models were adopted and these are shown in Fig. 1. The smallest, named P, includes only one metal node and an azobenzene-functionalized ligand; fragment Q includes two complete metal nodes and again one functionalized ligand; fragment R (176 atoms), includes four metal nodes and three functionalized ligands. An illustration of the fully functionalized periodic PCN-123, together with the ligand also used for comparison in this work, are also reported in Fig. 1.

The DFT calculations of the fragments are performed using NWChem.⁷⁴ The correlation-consistent basis sets aug-cc-pVDZ^{75,76} are used for azobenzene, the PCN-123 ligand, and fragments P and Q. A cheaper minimally-augmented maug-cc-pVDZ⁷⁷ basis set is used for the larger fragment R. LANL2DZ effective core potential (ECP) and the corresponding basis set is used for Zn in fragment R.^{78,79} The use of ECP and the smaller basis set for fragment R causes a blue shift by up to ~ 0.2 eV in the BSE spectra (see Fig. S1). For fragment Q and R, both PBE and PBEh with 40% EXX are used to compute the DFT ground state. Such a percentage of EXX in the PBEh global hybrid functional has been tuned to reproduce the *GW* quasiparticle ionization potential with the DFT highest-occupied molecular orbital (HOMO) eigenvalue.

The BSE/*GW* calculations for fragments are performed using FIESTA.^{57,63,80} Unlike for the periodic case, the frequency dependence of dielectric screening is not approximated by a plasmon-pole function. The contour deformation technique and the Coulomb-fitting resolution of identity (RI-V) with the universal Weigend Coulomb fitting auxiliary basis⁸¹ are used for integration over frequencies to calculate the *GW* self-energy $\Sigma(r, r', E)$. The full BSE Hamiltonian is considered in this case. The number of eigenvalues corrected at *GW* level and the number of states used to build the BSE Hamiltonian vary for each fragment and they are tabulated in Tab. S2.

To deal with the electrostatic polarization of the fragments due to the MOF environment in the ground state, a conductor like screening model⁸² (COSMO) as implemented in NWChem is used. The MOF environment is thus approximated using the optical dielectric constant of the periodic PCN-123 in the zero-frequency limit ($\epsilon_{opt} \sim 1.76$) obtained using the random-phase approximation in YAMBO.

To account for the screening by the environment at

the GW and BSE level, a QM/MM (Quantum Mechanics/Molecular Mechanics) approach is used by employing discrete polarizable model.^{83,84} The MM part is modeled as a fixed cubic lattice of polarizable pseudo-atoms which are assigned the atomic polarizability computed from the zero-frequency optical dielectric constant (ϵ_{opt} or square of the refractive index) of periodic PCN-123 by using the Clausius-Mossotti formula. The reaction-field matrix which describes the screening by the MM part is computed using the MESCAL code.⁸⁵ This screening from the environment is added to the screened Coulomb potential W by solving the following set of equations:

$$\tilde{v} = v + v\chi^{MM}v \quad (6)$$

$$W = \tilde{v} + \tilde{v}\chi_0W \quad (7)$$

where χ_0 is the independent-particle susceptibility, v is the bare Coulomb potential which is dressed (i.e. \tilde{v}) by the environment reaction field,^{83,84} and thus W is the screened Coulomb potential incorporating the screening from the MM part. The GW energies are then corrected by using the ΔCOHSEX correction⁸³ (i.e. Coulomb-Hole screened exchange) which is essentially the difference between the COHSEX self-energy calculated with and without the MM screening. Practically, first the $evGW$ eigenvalues are calculated in gas phase and then the ΔCOHSEX correction is added. Here, $evGW$ refers to G_3W_3 and G_2W_2 for PBE and PBEh DFT starting points, respectively. At the first GW step, only the HOMO and LUMO are GW corrected, and the scissor operator is applied. In the subsequent steps, more states are explicitly corrected to yield quasiparticle values converged within 0.1 eV similar to YAMBO. The environmental effect are then included as follows:

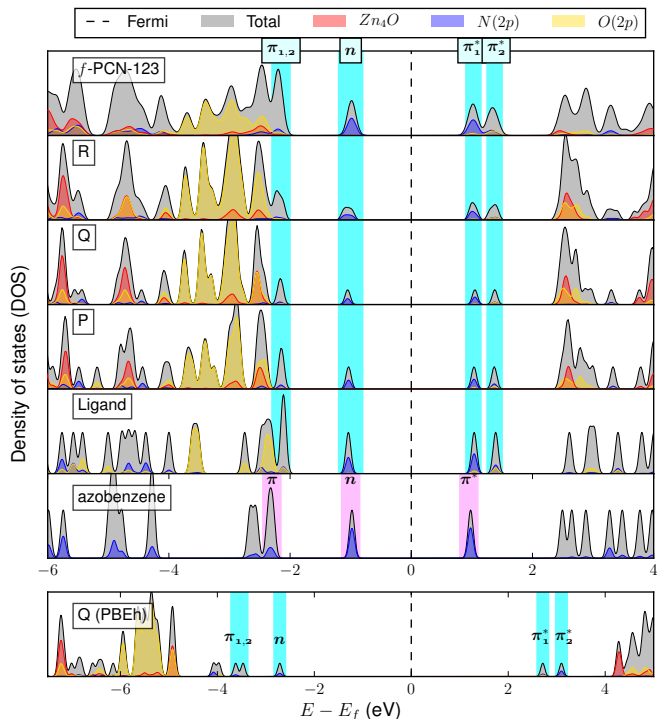
$$\begin{aligned} \epsilon_n^{GW_e} &\approx \epsilon_n^{GW} + \Delta\epsilon_n^{\text{COHSEX}} \\ &= \epsilon_n^{GW} + \langle \phi_n^{KS} | \Sigma^{\text{COHSEX/MM}} - \Sigma^{\text{COHSEX}} | \phi_n^{KS} \rangle \end{aligned} \quad (8)$$

A detailed description and the accuracy of use of QM/MM model in conjunction with GW and BSE has been discussed in literature.^{62,83,84,86,87} The screened Coulomb potential including environmental effects is then used for the BSE Hamiltonian.^{62,87}

Results and Discussion

Kohn-Sham ground state

The density of states (DOS) computed using PBE for the three fragment models of PCN-123, the gas phase azobenzene and the periodic PCN-123, is shown in Fig. 2. We only report the *cis* case, for clarity, the *trans* case being similar is reported in Fig. S3. The total DOS (gray) and the projected-DOS (pDOS) are shown in each case. For the latter we plot the DOS projected onto the 2p-states of N (blue), the 2p-states of the O atoms connecting the metal-node and the ligand (yellow), and the states associated with the Zn_4O node (red). The magenta strips in the case of azobenzene highlight the π , n , π^* states which are involved in the excitations found in the UV-Vis region of the absorption spectra and leading to photoisomerization,²⁶ as discussed in the introduction. For the PCN-123 ligand, the frontier orbitals retain a character analogous to the π , n and π^* of *cis* and *trans* azobenzene (see Fig. S2 of SI) with some mixing of



(a)

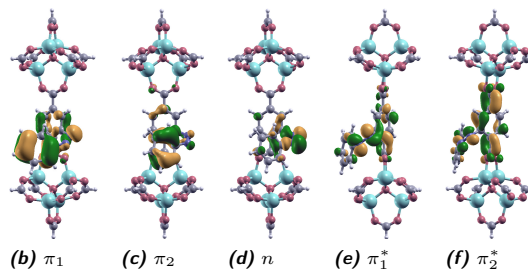


Figure 2. (a) pDOS of different models of *cis* PCN-123 calculated with PBE. The upper panel represents the pDOS of the fully-functionalized periodic system, i.e. *f*-PCN-123 while (P), (Q) and (R) represent fragments with a progressive increase in size as illustrated in Fig. 1. A 0.1 eV broadening is applied in every case. For fragment Q the PBEh (40% exact exchange) is also plotted for comparison. The DOS and pDOS are presented with different scales for different models. (b) and (c) π , (d) n , (e) π_1^* and (f) π_2^* KS eigenstates of *cis* fragment Q represented by plotting isosurface of $|\phi_n|^2$ with green and yellow depicting opposite signs of the original wavefunction $\pm\phi_n$. The midpoint of the fundamental gap is taken as the Fermi level.

p -states of O atoms into π and π^* . Two anti-bonding π states close in composition and energy, i.e. π_1^* and π_2^* , arise in all cases in place of the single π^* of azobenzene and are highlighted in cyan in the PDOS. The breaking of symmetry between the two benzene rings upon attachment of the carboxylic acid or groups in one of the two phenyl rings lead to two different π^* states which localize predominantly on the N atoms and the benzenedicarboxylic acid (see lower panel of Fig. 2).

The electronic structure of the three fragments, in the vicinity of the band gap, maintains features similar to the periodic MOF in terms of energy and composition (see cyan strips in Fig. 2). The wavefunction corresponding to these states is plotted in the lower panel of Fig. 2 for *cis* fragment

Q. A similar behavior is found for all MOF models. For comparison, these states are also plotted for azobenzene and for the ligand in Fig. S2. Even for the ligand, the DOS resembles the MOF case for the KS states near the gap; below -2.5 eV and above 2.5 eV this agreement vanishes due to the presence of the states associated with the Zn₄O metal node. Moving away from the gap, the DOS of the three fragments P, Q and R which include single or multiple metal-oxide nodes show a reasonable resemblance with the *f*-PCN-123 case, for both occupied and unoccupied levels. A similar behavior is found for *trans f*-PCN-123 (see Fig. S3).

The DOS of *cis* fragment Q calculated using PBEh with 40% exact exchange is also shown in Fig. 2 for comparison (see Fig. S4 for *trans*). As expected, the PBE band gap of 2.1 eV for *cis* and 1.9 eV for *trans* increases to 5.5 and 5.2 eV, respectively, using PBEh, as expected from the EXX-tuning procedure that matches the DFT KS IP onto the *evGW* level.

Because we do not impose any symmetry constraint during the geometrical optimization, the steric hindrance of the pendant groups inside the MOF results in different configurations of the six azobenzenes within the unit cell and in the twisting and bending of the BDC (benzenedicarboxylate) ligands.²⁴ Such geometrical distortions and configurational disorder mimic to some extent the structural disorder of the pendant groups found experimentally in PCN-123¹⁷ and lead to the broadening of the DOS and the BSE spectra (*vide infra*).

The *GW* correction: fragment versus periodic

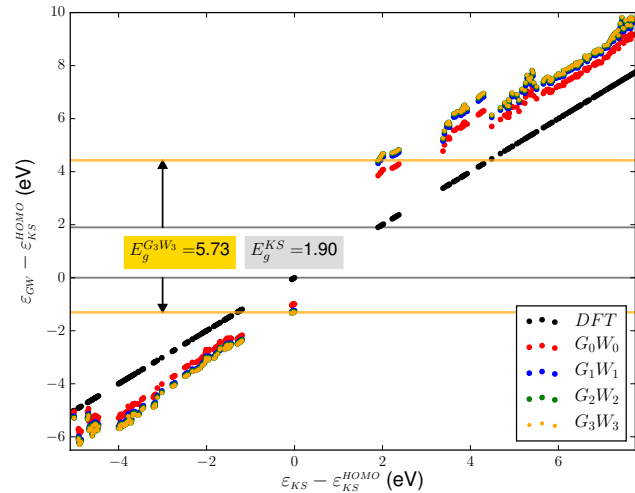


Figure 3. *GW* quasiparticle energies (in eV) of *cis f*-PCN-123 computed at four *GW* iterations.

The *GW* quasiparticle energies computed at four *GW* iterations using YAMBO for the *cis f*-PCN-123 are plotted in Fig. 3. Only the states that are explicitly corrected are shown in the figure. We find that three *GW* iterations (G_2W_2) are sufficient to reach a convergence with criterion of $\Delta E_g < 0.1$ eV (with E_g =band gap) with respect to the previous step and the *evGW* band gap is 5.73 eV (G_3W_3 value), almost 4 eV larger than the PBE gap (1.9 eV). Assuming a similar behaviour of *GW* convergence for *trans f*-PCN-123,

three iterations of *GW* are performed and deemed sufficient. A similar trend is found with the fundamental gap opening from 1.68 eV (PBE) to 5.31 eV (G_2W_2) (see Fig. S5). Henceforth, the term *evGW* refers to G_2W_2 for YAMBO calculations.

The *GW* correction was also computed for all the fragment models and for the PCN-123 ligand. The final *evGW* quasiparticle gaps are reported in Tab. 1. In case of non-periodic systems, the *evGW* eigenvalues are reported after applying the Δ COHSEX correction accounting for environmental screening as explained in the previous section. The Δ COHSEX procedure closes the band gap by about ~ 0.7 - 0.9 eV with respect to the gas phase *evGW* value (see Fig. S9). This arises from polarization (screening) effects that stabilize holes and electrons. The *evGW* quasiparticle energies for the ligand and fragment models plotted against the KS eigenvalues are reported in Fig. S8.

The comparison of the *evGW* quasiparticle corrections computed for the periodic MOF, the three fragment choices, and the ligand, are reported in Fig. S6. Overall, the correction shows the same qualitative behaviour. We find a good quantitative agreement among the non-periodic models, but we note some discrepancy between the non-periodic and the periodic models. The periodic YAMBO calculations yield smaller *evGW* corrections compared to the non-periodic FIESTA calculations (see Fig. S7). Thus despite the PBE gap being essentially the same for the periodic and non-periodic case (within 0.2 eV), the periodic *evGW* band gap is ≈ 1 eV smaller ($\sim 0.8 - 1.1$ eV) compared to the non-periodic case (see Tab. 1). This difference may be attributed to the different computational approaches such as the use of pseudopotentials and the plasmon-pole approximation, in YAMBO, and the use of incomplete bases in FIESTA, as shown and discussed in detail in Ref. 88. Also, on the fragment side, the possible discrepancies on the long-range contribution to the screened Coulomb potential W , as described by the QM/MM Δ COHSEX scheme, may affect the fragment *GW* calculations. As shown below, the much better agreement of optical BSE excitations, for which the impact of a change in the environmental screening is known to be much smaller, points in this direction. Finally, the fragment approximation may result in somehow different KS eigenstates. This can be inferred by looking at the independent-particle spectra computed for the periodic model and for fragment Q using the *epsilon.x* utility of Quantum ESPRESSO (see Fig. S11) for both calculations. Specifically, the ratio between the first two excitations is significantly different in the two cases, with the first excitation being significantly less intense (compared to the second) in the fragment as compared to the periodic MOF.

For non-periodic models, the *evGW* eigenvalues are calculated also with PBEh with 40% exact exchange as starting point (see Tab. 1). The hybrid functional used here fulfils the ionization potential theorem such that the DFT HOMO eigenvalue matches with the *evGW* ionization potential (within 0.2 eV). In our earlier work,⁶² we showed that the use of such IP-tuned hybrid functionals to describe the Kohn-Sham ground state yields very accurate BSE/*GW* excitation energies of crucial $n \rightarrow \pi^*$ transition for a range of azobenzene derivatives. The DFT band gap calculated here with PBEh is significantly larger (i.e. > 3 eV) (see Fig. 2), as expected, compared to the PBE one, bringing it much closer to the *evGW* band gap. This is further shown in Fig. S10 where the *evGW* correction is plotted against the

Table 1. DFT and *evGW* gaps for periodic and non-periodic models of PCN-123 calculated with PBE and PBEh (see text). The *evGW* gap for the non-periodic calculations are computed by employing an embedding scheme to account for environmental effects at both the ground state and GW level, using ΔCOHSEX for the latter.

Model	PBE		<i>evGW</i> /PBE		PBEh		<i>evGW</i> /PBEh		code
	<i>cis</i>	<i>trans</i>	<i>cis</i>	<i>trans</i>	<i>cis</i>	<i>trans</i>	<i>cis</i>	<i>trans</i>	
Ligand	2.08	1.94	6.59	6.44	5.39	5.24	6.71	6.53	FIESTA
P	2.07	1.88	6.38	6.23	5.38	5.19	6.51	6.36	FIESTA
Q	2.10	1.90	6.52	6.33	5.45	5.20	6.68	6.50	FIESTA
R	1.94	1.88	6.52	6.43	5.28	5.30	6.62	6.65	FIESTA
<i>f</i>-PCN-123	1.90	1.68	5.73	5.31	-	-	-	-	YAMBO

KS energies for the frontier states of fragment P, Q and R: the *evGW* correction is ~ 1 eV when using the PBEh starting point compared to ~ 3 eV for PBE. It is interesting to note that, regardless of the XC functional used for the DFT starting point, the *evGW* quasiparticle energies converge to close values (see Fig. S12) for both *trans* and *cis*, for either the fragments and the ligand, thus confirming the robustness of *GW* method.^{89–92}

The BSE optical spectra: fragment versus periodic

The BSE/*evGW*/PBE spectra of *cis* and *trans* configurations of *f*-PCN-123 and the fragment models are plotted in Fig. 4. Despite the fact that the difference in the *evGW* band gap between the periodic and non-periodic calculations reaches up to 0.8 eV for *cis* and 1.1 eV for *trans* (see Tab. 1), the energy of the S_1 excitation (i.e. $n \rightarrow \pi_1^*$) agrees within $\sim 0.1 - 0.3$ eV for both *cis* and *trans*. As mentioned above, this is possibly due to the much reduced impact of environmental screening onto the neutral optical excitations, for which solvatochromic shift are known to be much smaller than the shift in electronic energy levels. As a matter of fact, errors in the screened Coulomb potential enters with opposite sign in the quasiparticle gap than in the electron-hole interaction, leading to well-known cancellation of errors at the BSE level. Thus, large differences in the quasiparticle gap may reduce in the corresponding optical gap.⁹³

For *cis* configuration, the first optical band corresponds to the HOMO \rightarrow LUMO ($n \rightarrow \pi_1^*$) transition, similar to the case of gas phase or solvated azobenzene. The second optical band (yellow in Fig. 4), the S_2 , corresponds to the $n \rightarrow \pi_2^*$ excitation which is absent in azobenzene. The third (sky-blue) optical band, the S_3 , is due to the $\pi \rightarrow \pi_1^*$ excitation. The S_4 , in red, contains mixed $\pi \rightarrow \pi_1^*/\pi \rightarrow \pi_2^*$ excitations. From the fourth optical band onwards, the one-to-one correspondence among excitons of different models fades away, mostly owing to the increasing contribution of O atoms of the ZnO_4 noded and to the more delocalized character of the exciton.

For *trans*, the S_1 band (gray) is due to the $n \rightarrow \pi_1^*$ excitation, whereas the S_2 (yellow) is dominated by $\pi \rightarrow \pi_1^*$ along with some contribution from $n \rightarrow \pi_2^*$ and $\pi \rightarrow \pi_2^*$, specifically for fragment R and the periodic system. The character and energy of S_1 and S_2 are similar to the azobenzene molecule. It is important to stress that for fragment R, the energy of S_1 and S_2 for both *cis* and *trans* is blueshifted compared to the smaller PCN-123 fragments possibly due to the use of effective-core-potentials and a smaller basis, as discussed in the computational section. This is shown

in Fig. S1 where the effect of these approximations on the calculation of the BSE spectrum is reported for a smaller fragment.

To illustrate the similarities among the different models, the electron-averaged hole density (in yellow) and the hole-averaged electron density (in green) of the $n \rightarrow \pi_1^*$ and $\pi \rightarrow \pi_1^*$ transitions are plotted, respectively, for *cis* and *trans*, in Fig. 5 for the ligand, fragments Q and R, and the periodic MOF. For clarity, we neglect fragment P which exhibits the same character of the exciton. We plot the excitonic density for these two transitions, since these correspond to the most intense bands in the low energy range which are usually employed for excitation in the UV-Vis range to induce large photoisomerization yields. Excitation at the maxima of these two bands will result in efficient *trans*-to-*cis* photoisomerization when the S_2 (of *trans*) is excited, and viceversa, when the S_1 (of *cis*) is excited.^{27,28} The excitonic densities of the fragment model are reported in Figs. S14-S18 (for *cis*) and Figs. S20-S24 (for *trans*). For the $n \rightarrow \pi_1^*$ and $\pi \rightarrow \pi_1^*$ excitations, a good correspondence is found among the different MOF models. The character of the excitation is fairly maintained, however, the exciton may delocalize when models with multiple ligands are employed. In general, as the energy of the exciton increases the character becomes more delocalized (see Figs. S18 and S24). The localization on multiple ligands is found also for certain excitons in the S_1 band for both *cis* (see Figs. S18 and S17) and *trans* (see Figs. S24 and S23). Furthermore, excitons localized on different ligands can have a different character on each ligand. For example, the exciton number 23 in *trans f*-PCN-123 plotted in Fig. S24 consists of both n and π holes on different ligands and, similarly, the electron has both π_1^* and π_2^* character on different ligands.

Overall, the non-periodic models reproduce fairly well the optical absorption of the periodic MOF up to 4.5 eV, consistent with excitons being localized on the azobenzene-functionalized ligands in this spectral region, while at higher energies the absorbance of the Zn_4O nodes becomes significant. See for example, the excitons of the S_4 for *cis f*-PCN-123 (Fig. S18), and the S_3 and S_4 for *trans f*-PCN-123 (Fig. S24). This implies that excitations in the visible, near-UV and mid-UV spectral region can be used to target azobenzene excitation thus possibly resulting in isomerization provided that the excited state dynamics is similar to the molecular case and that steric hindrance is not a limiting factor. This is unlike many other azobenzene based photo-switching systems where use of UV light for azobenzene isomerization is very restricted due to absorption in that range by other chemical entities.⁹⁴

To better visualize the spectral regions with large selective absorption, we plot the BSE/*evGW*/PBE spectra of

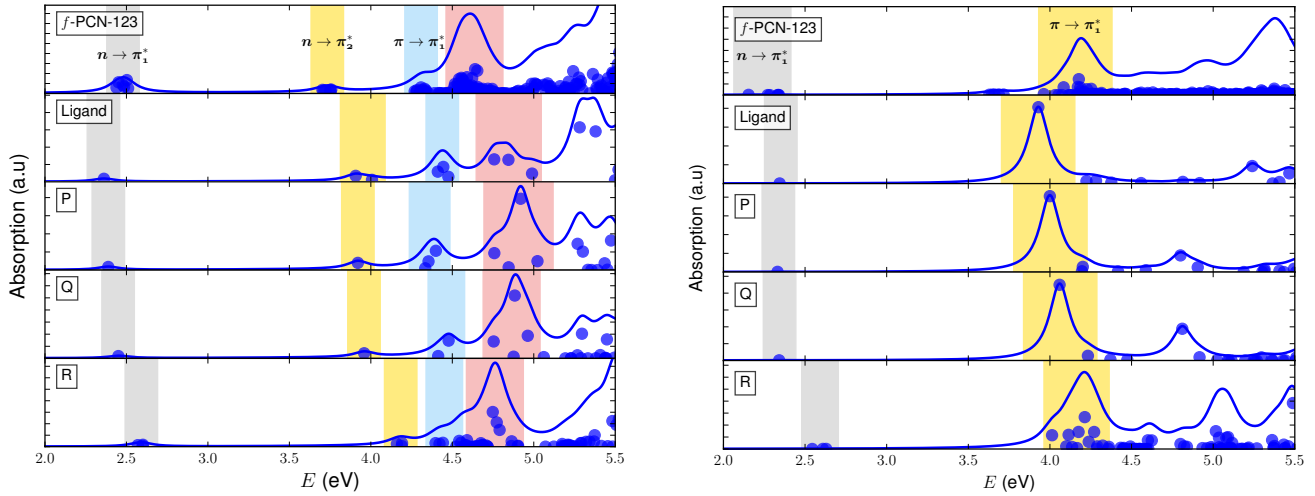


Figure 4. BSE/*evGW* spectra of different models of *cis* (a) and *trans* (b) PCN-123 obtained by multiplying the oscillator strength by a Lorentzian of 0.1 eV broadening (HWHM). The color strips denote the optical bands involving excitations with similar character (see text).

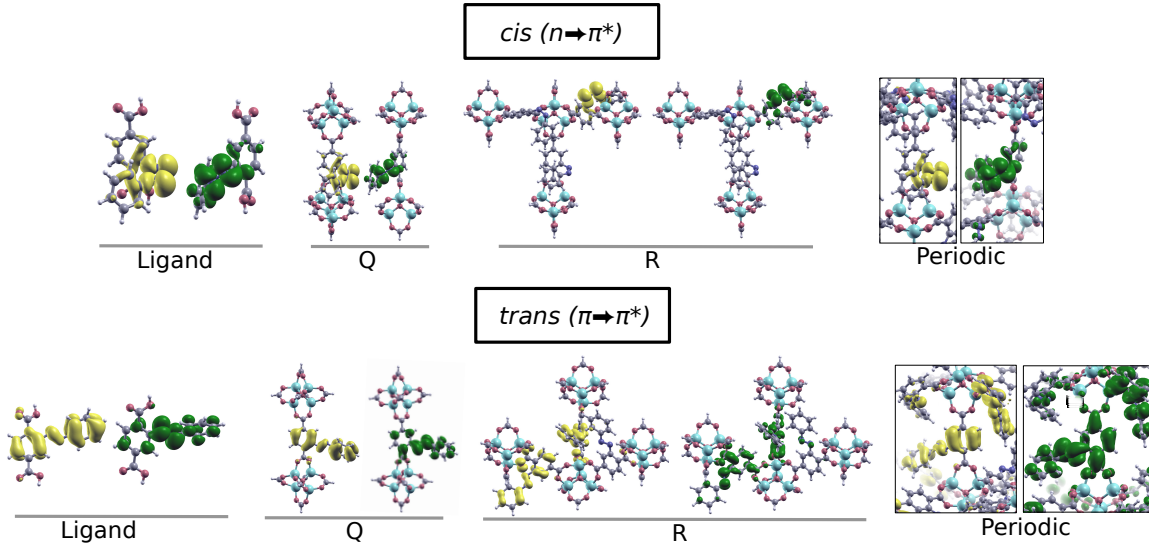


Figure 5. Excitonic densities of electron and hole as defined in eq. 4 and 5. The densities plotted in top row correspond to $n \rightarrow \pi_1^*$ (*cis*) and densities in bottom row correspond to $\pi \rightarrow \pi_1^*$ (*trans*). The hole and electron densities are plotted in yellow and green colors, respectively.

trans and *cis* in the same figure, for azobenzene, the PCN-123 ligand (in gas phase geometry), fragment R and the *f*-PCN-123 (see Fig. 6). For fragment R we plot also the BSE/*evGW*/PBEh spectra.

Moving from azobenzene to fragment R and finally to *f*-PCN-123, the most intense peak in both *trans* (i.e. $\pi \rightarrow \pi_1^*$) and *cis* (i.e. $n \rightarrow \pi_1^*$) undergoes a blueshift by ~ 0.5 eV and ~ 0.2 , respectively. It is important to note that the $n \rightarrow \pi_1^*$ optical band of *cis* in the periodic calculations exhibits a significantly higher intensity compared to any other fragment model. The visual inspection of the $n \rightarrow \pi_1^*$ exciton in the different MOF models discussed above and reported in the upper panel of Fig. 5 allow to establish a good correspondence between the periodic and the fragment models. However, the spatial localization and the symmetry of the wavefunction may differ quantitatively thus affecting directly the

value of the oscillator strength. As discussed above, this is confirmed by the independent-particle spectra computed for both the periodic and the fragment Q. The two spectra have similar features and the biggest difference is the significantly more intense first excitation in the case of *f*-PCN-123 as shown by the plotted ratio between the two cases in the lower panel of Fig. S11. Thus, we attribute this discrepancy in the intensity of the first excitation to a limitation of the fragment approach in describing the correct KS periodic eigenstates.

The excitation energies and the oscillator strengths of $n \rightarrow \pi^*$ excitation of azobenzene calculated with BSE/*GW* was shown to be very sensitive to the choice of DFT functional⁶² owing to the change in the KS states. On the other hand, the position of $\pi \rightarrow \pi^*$ optical band is comparatively resilient to change in the DFT functional. This occurs also

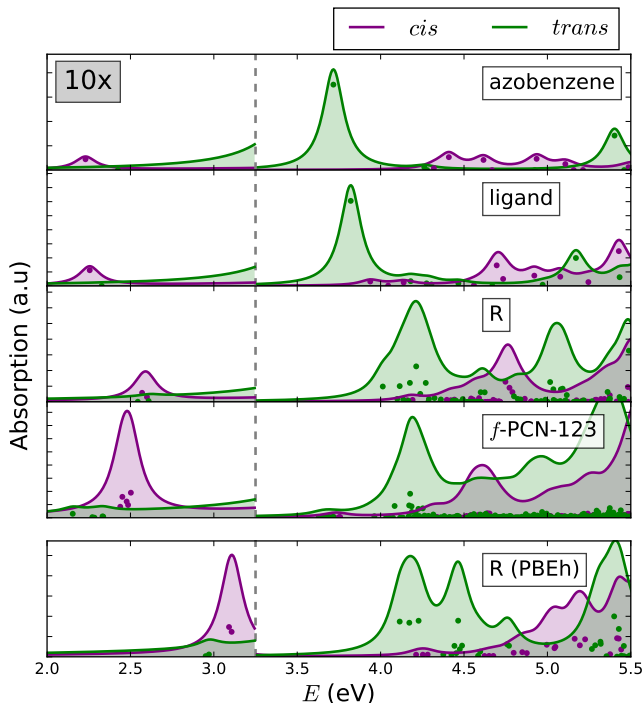


Figure 6. BSE/*evGW*/PBE spectra of azobenzene, PCN-123 ligand, fragment R, and *f*-PCN-123. All spectra in the low energy region are multiplied by 10 for visual clarity. The lower panel show the BSE/*evGW*/PBEh (40% EXX) spectra of fragment R.

in this case: we report the comparison of the BSE/*evGW* spectra of gas phase azobenzene computed with PBE and with the IP-tuned PBEh (60% EXX) in Fig. S25. The experimentally-extracted values of the azobenzene S_1 energy (*cis*: 2.92 eV, *trans*: 2.82 eV) are in excellent agreement with the BSE S_1 excitation energies computed using the IP-tuned PBEh (*cis*: 3.00 eV, *trans*: 2.83 eV).^{25,27,62} A non-negligible deviation from experiment is instead found when using the PBE starting point (*cis*: 2.23 eV, *trans*: 2.43 eV.)

In light of these results, to attempt a quantitative description of the optical properties of this material we compute the BSE spectra of fragment R with PBEh using 40 % of exact exchange which tunes the Kohn-Sham IP to the *evGW* IP. The BSE spectra of either isomers is shown in the lower panel of Fig. 6. When PBEh is used, the energy of $n \rightarrow \pi_1^*$ blueshifts by ~ 0.5 eV for both *cis* and *trans*. The $\pi \rightarrow \pi_1^*$ excitation of *trans* does not vary appreciably in energy. The oscillator strength of $n \rightarrow \pi_1^*$ increases for *cis* from 0.012 to 0.033 (~ 2.7 times) upon change in the functional. The wide and intense band found for *cis* in the range of 4.5-5.0 eV is blueshifted to energies higher than 5 eV making *trans* the dominant absorber between 4 and 5 eV. In case of *trans*, the use of hybrid functional separates the $\pi \rightarrow \pi_1^*$ excitations from $n \rightarrow \pi_2^*$ and $\pi \rightarrow \pi_2^*$ excitations resulting in two distinct optical bands instead of a single one as in the case of PBE. Like for the PBE case, the BSE/*evGW*/PBEh spectra show a small blueshift of the S_1 of *cis* and the S_2 of *trans* in the PCN-123 MOF as compared to gas phase azobenzene (in Fig. S25). The S_1 of *cis* is at 3.0 eV for azobenzene (see Fig. S25) and approx. 3.1 eV for PCN-123 (see Fig. 6) and the S_2 of *trans* at 3.9 eV for azobenzene and 4.1-4.5 eV for PCN-123.

Finally, we briefly comment on the effect of the geometry

of the azo-molecules on the optical absorption properties, in particular when fragment models are employed. Because of the structural disorder within the MOF and because the geometry of the fragments is taken by carving one (fragment Q) or multiple (fragment R) ligands from the periodic MOF containing six geometrically different ligands, it can happen that by taking only one of these, such as in fragment Q, the relative intensity between *trans* and *cis* could change. This is indeed the case when we compute the BSE/*evGW*/PBEh (40% EXX) spectra of *trans* and *cis* using a fragment Q taken by carving the specific ligand that exhibits the largest distortion compared to the other five ligands (and compared to gas phase). This specific choice would yield spectra with the $n \rightarrow \pi_1^*$ excitation in *trans* being more intense than in *cis* (see Fig. S26). Some caution is advised in this regard to avoid a biased result.

Conclusion

To summarize, our study of the optical absorption spectra of *cis* and *trans* PCN-123 MOF, performed by employing the BSE/*evGW* approach allows to shed light for the first time on the nature of the electronic excitations of an azobenzene-functionalized MOF. Similarities and differences among the optical absorption of the fully functionalized MOF, the azobenzene molecule and the MOF azobenzene-functionalized ligands are shown and discussed. On the computational point of view the accuracy of the fragment model is assessed by adopting a QM/MM approach where the environmental effects are considered both at the DFT level and at the *GW* and BSE level. In the low energy region, up to 4.5 eV, a satisfactory description of the optical properties of the periodic material can be obtained already by using the smallest fragment (fragment P) which consists of one metal node and a single functionalized ligand. The agreement systematically improves as we increase the size of the fragment by adding an increasing number of metal nodes and azo-functionalized ligands. In this region the spectral features of the MOF resemble closely the case of gas phase azobenzene, and the ligand. This is consistent with recent experimental studies of azobenzene functionalized IR-MOF-74,^{29,30} ortho-fluoro-azobenzene functionalized SURMOF¹⁹ and core-shell MOF²³ showing a good similarity between the optical absorption of the MOF and the ligand case.²⁰ One important difference between the azobenzene or ligand molecule with respect to the MOF is the stronger S_1 feature in *cis* predicted in the latter. This suggests a faster and more complete *cis* \rightarrow *trans* isomerization in the MOF upon excitation of the S_1 band, compared to the molecular case^{28,38} for which indeed slow and incomplete back-isomerization is observed. We believe this results has important implications for the design of photoswitches-based MOFs.

In conclusion, in terms of spectral features and character of the excitations, the periodic MOF resembles the case of free azobenzene and the azobenzene-functionalized ligand of the MOF, with the most intense bands slightly blueshifted in the MOF compared to the azobenzene molecule. Our calculations place the largest selective absorption by *cis* at around 3.2 eV and by *trans* at ~ 4.2 eV. The most important difference between the molecular and periodic models is the ratio between the oscillator strength of *cis* and *trans* for the S_1 , which is found to be significantly larger in the periodic MOF compared to any fragment model, including the free

azobenzene molecule as discussed above. This is attributed to the limitation of non-periodic models to provide a quantitative description of the electronic structure of the periodic system at the DFT level. As the excitation energy increases the fragment model becomes less adequate to describe the MOF due to the more delocalized character of the exciton and to the increasing contribution from the metal nodes.

We hope this study will help accelerate the design of photoactive frameworks with good photoisomerization efficiencies and enhanced capture performances.

Acknowledgement This work was funded by the project ANR-15-CE06-0003-01 funded by the French National Agency for Research. Calculations were performed using resources granted by GENCI under the CINES grant number A0020907211. Additionally, the froggy platform of the CIMENT infrastructure, which is supported by the Rhone-Alpes region (GRANT CPER07_13 CIRA) and the Equip@Meso project was employed for the calculations.

Supporting Information Available: Computational details; DOS for *trans*; DFT KS states; *evGW* corrections for all models used; excitonic densities for all the MOF models used, for azobenzene and ligand; XYZ info available This material is available free of charge via the Internet at <http://pubs.acs.org/>.

References

- Gamonal, A.; Sun, C.; Mariano, A. L.; Fernandez-Bartolome, E.; Guerrero-SanVicente, E.; Vlasisavljevich, B.; Castells-Gil, J.; Marti-Gastald, C.; Poloni, R.; Wanemacher, R.; Cabanillas-Gonzalez, J.; Costa, J. S. Divergent Adsorption-Dependent Luminescence of Amino-Functionalized Lanthanide Metal-Organic Frameworks for Highly Sensitive NO₂ Sensors. *The Journal of Physical Chemistry Letters* **2020**, *11*, 3362–3368.
- Chen, Z.; Li, P.; Anderson, R.; Wang, X.; Zhang, X.; Robison, L.; Redfern, L. R.; Moribe, S.; Islamoglu, T.; Gómez-Gualdrón, D. A.; Yildirim, T.; Stoddart, J. F.; Farha, O. K. Balancing volumetric and gravimetric uptake in highly porous materials for clean energy. *Science* **2020**, *368*, 297–303.
- Qian, Q.; Asinger, P. A.; Lee, M. J.; Han, G.; Rodriguez, K. M.; Lin, S.; Benedetti, F. M.; Wu, A. X.; Chi, W. S.; Smith, Z. P. MOF-Based Membranes for Gas Separations. *Chemical Reviews* **2020**, *120*, 8161–8266.
- Yang, D.; Gates, B. C. Catalysis by Metal Organic Frameworks: Perspective and Suggestions for Future Research. *ACS Catalysis* **2019**, *9*, 1779–1798.
- Usman, M.; Mendiratta, S.; Lu, K.-L. Metal-Organic Frameworks: New Interlayer Dielectric Materials. *ChemElectroChem* **2015**, *2*, 786–788.
- Resines-Urien, E.; Burzurí, E.; Fernandez-Bartolome, E.; García García-Tuñón, M. Á.; de la Presa, P.; Poloni, R.; Teat, S. J.; Costa, J. S. A switchable iron-based coordination polymer toward reversible acetonitrile electro-optical readout. *Chem. Sci.* **2019**, *10*, 6612–6616.
- He, H.; Cui, Y.; Li, H.; Shao, K.; Chen, B.; Qian, G. Controllable broadband multicolour single-mode polarized laser in a dye-assembled homoepitaxial MOF microcrystal. *Light: Science & Applications* **2020**, *9*.
- Baumann, A. E.; Burns, D. A.; Liu, B.; Thoi, V. S. Metal-organic framework functionalization and design strategies for advanced electrochemical energy storage devices. *Communications Chemistry* **2019**, *2*.
- Poloni, R.; Lee, K.; Berger, R. F.; Smit, B.; Neaton, J. B. Understanding Trends in CO₂ Adsorption in Metal-Organic Frameworks with Open-Metal Sites. *The Journal of Physical Chemistry Letters* **2014**, *5*, 861–865, PMID: 26274079.
- Poloni, R.; Kim, J. Thermodynamics of gas adsorption in MOFs using Ab Initio calculations. *International Journal of Quantum Chemistry* **2016**, *116*, 569–572.
- Vaidhyanathan, R.; Iremonger, S. S.; Shimizu, G. K. H.; Boyd, P. G.; Alavi, S.; Woo, T. K. Direct Observation and Quantification of CO₂ Binding Within an Amine-Functionalized Nanoporous Solid. *Science* **2010**, *330*, 650–653.
- McDonald, T. M. et al. Cooperative insertion of CO₂ in diamine-appended metal-organic frameworks. *Nature* **2015**, *519*, 303–308.
- Li, H.; Wang, K.; Sun, Y.; Lollar, C. T.; Li, J.; Zhou, H.-C. Recent advances in gas storage and separation using metal-organic frameworks. *Materials Today* **2018**, *21*, 108–121.
- Ding, M.; Flaig, R. W.; Jiang, H.-L.; Yaghi, O. M. Carbon capture and conversion using metal-organic frameworks and MOF-based materials. *Chem. Soc. Rev.* **2019**, *48*, 2783–2828.
- Hu, Z.; Wang, Y.; Shah, B. B.; Zhao, D. CO₂ Capture in Metal-Organic Framework Adsorbents: An Engineering Perspective. *Advanced Sustainable Systems* **2019**, *3*, 1800080.
- Huck, J. M.; Lin, L.-C.; Berger, A. H.; Shahrak, M. N.; Martin, R. L.; Bhowan, A. S.; Haranczyk, M.; Reuter, K.; Smit, B. Evaluating different classes of porous materials for carbon capture. *Energy Environ. Sci.* **2014**, *7*, 4132–4146.
- Park, J.; Yuan, D.; Pham, K. T.; Li, J.-R.; Yakovenko, A.; Zhou, H.-C. Reversible alteration of CO₂ adsorption upon photochemical or thermal treatment in a metal-organic framework. *J. Am. Chem. Soc.* **2011**, *134*, 99–102.
- Rosi, N. L.; Eckert, J.; Eddaoudi, M.; Vodak, D. T.; Kim, J.; O’Keeffe, M.; Yaghi, O. M. Hydrogen Storage in Microporous Metal-Organic Frameworks. *Science* **2003**, *300*, 1127–1129.
- Wang, Z.; Knebel, A.; Grosjean, S.; Wagner, D.; Bräse, S.; Wöll, C.; Caro, J.; Heinke, L. Tunable Molecular Separation by Nanoporous Membranes. *Nat. Commun.* **2016**, *7*, 13872.
- Müller, K.; Knebel, A.; Zhao, F.; Bléger, D.; Caro, J.; Heinke, L. Switching Thin Films of Azobenzene-Containing Metal-Organic Frameworks with Visible Light. *Chem. Eur. J.* **2017**, *23*, 5434–5438.
- Kanj, A. B.; BÅijrck, J.; Grosjean, S.; BrÅdse, S.; Heinke, L. Switching the enantioselectivity of nanoporous host materials by light. *Chem. Commun.* **2019**, *55*, 8776–8779.
- Jiang, Y.; Tan, P.; Qi, S.-C.; Liu, X.-Q.; Yan, J.-H.; Fan, F.; Sun, L.-B. Metal-Organic Frameworks with Target-Specific Active Sites Switched by Photoresponsive Motifs: Efficient Adsorbents for Tailorable CO₂ Capture. *Angewandte Chemie International Edition* **2019**, *58*, 6600–6604.
- Mutruc, D.; Goulet-Hanssens, A.; Fairman, S.; Wahl, S.; Zimathies, A.; Knie, C.; Hecht, S. Modulating Guest Uptake in Core-Shell MOFs with Visible Light. *Angewandte Chemie International Edition* **2019**, *58*, 12862–12867.
- Yang, C.-T.; Kshirsagar, A. R.; Eddin, A. C.; Lin, L.-C.; Poloni, R. Tuning Gas Adsorption by Metal Node Blocking in Photoresponsive Metal-Organic Frameworks. *Chem. Eur. J.* **2018**, *24*, 15167–15172.
- Andersson, J.-Å.; Petterson, R.; Tegnér, L. Flash photolysis experiments in the vapour phase at elevated temperatures I: spectra of azobenzene and the kinetics of its thermal cis-trans isomerization. *Journal of Photochemistry* **1982**, *20*, 17–32.
- Bandara, H. M. D.; Burdette, S. C. Photoisomerization in Different Classes of Azobenzene. *Chem. Soc. Rev.* **2012**, *41*, 1809–1825.
- VetrÁková, L.; Ladányi, V.; Al Anshori, J.; Dvořák, P.; Wirz, J.; Heger, D. The absorption spectrum of cis-azobenzene. *Photochemical & Photobiological Sciences* **2017**, *16*, 1749–1756.
- Ladányi, V.; Dvořák, P.; Anshori, J. A.; VetrÁková, L.; Wirz, J.; Heger, D. Azobenzene photoisomerization quantum yields in methanol redetermined. *Photochemical & Photobiological Sciences* **2017**, *16*, 1757–1761.
- Brown, J. W.; Henderson, B. L.; Kiesz, M. D.; Whalley, A. C.; Morris, W.; Grunder, S.; Deng, H.; Furukawa, H.; Zink, J. I.; Stoddart, J. F.; Yaghi, O. M. Photophysical pore control in an azobenzene-containing metal-organic framework. *Chemical Science* **2013**, *4*, 2858.
- Wang, Z.; Grosjean, S.; Bräse, S.; Heinke, L. Photoswitchable Adsorption in Metal-Organic Frameworks Based on Polar Guest-Host Interactions. *ChemPhysChem* **2015**, *16*, 3779–3783.
- Mancuso, J. L.; Mroz, A. M.; Le, K. N.; Hendon, C. H. Electronic Structure Modeling of Metal-Organic Frameworks. *Chemical Reviews* **2020**, *120*, 8641–8715.
- Gu, Z.-G.; Heinke, L.; Wöll, C.; Neumann, T.; Wenzel, W.; Li, Q.; Fink, K.; Gordan, O. D.; Zahn, D. R. Experimental and theoretical investigations of the electronic band structure of metal-organic frameworks of HKUST-1 type. *Appl. Phys. Lett.* **2015**, *107*, 102 1.
- Fumanal, M.; Corminboeuf, C.; Smit, B.; Tavernelli, I. Optical absorption properties of metal-organic frameworks: solid state versus molecular perspective. *Physical Chemistry Chemical Physics* **2020**, *22*, 19512–19521.
- Vermoordele, F.; Vandichel, M.; Van de Voorde, B.; Ameloot, R.; Waroquier, M.; Van Speybroeck, V.; De Vos, D. E. Electronic Effects of Linker Substitution on Lewis Acid Catalysis with Metal-Organic Frameworks. *Angewandte Chemie International Edition* **2012**, *51*, 4887–4890.
- Ji, M.; Lan, X.; Han, Z.; Hao, C.; Qiu, J. Luminescent Properties of Metal-Organic Framework MOF-5: Relativistic Time-Dependent Density Functional Theory Investigations. *Inorganic Chemistry* **2012**, *51*, 12389–12394.
- Conti, I.; Garavelli, M.; Orlandi, G. The Different Photoisomerization Efficiency of Azobenzene in the Lowest $n\pi^*$ and $\pi\pi^*$ Singlets: The Role of a Phantom State. *J. Am. Chem. Soc.* **2008**, *130*, 5216–5230.
- Fliegl, H.; Köhn, A.; Hättig, C.; Ahlrichs, R. Ab Initio Calculation of the Vibrational and Electronic Spectra of trans- and cis-Azobenzene. *J. Am. Chem. Soc.* **2003**, *125*, 9821–9827.
- Zimmerman, G.; Chow, L.-Y.; Paik, U.-J. The Photochemical Isomerization of Azobenzene. *J. Am. Chem. Soc.* **1958**, *80*, 3528–3531.
- Duarte, L.; Fausto, R.; Reva, I. Structural and spectroscopic

- characterization of E- and Z-isomers of azobenzene. *Phys. Chem. Chem. Phys.* **2014**, *16*, 16919–16930.
- (40) Crecca, C. R.; Roitberg, A. E. Theoretical study of the isomerization mechanism of azobenzene and disubstituted azobenzene derivatives. *J. Phys. Chem. A* **2006**, *110*, 8188–8203.
- (41) Salpeter, E. E.; Bethe, H. A. A Relativistic Equation for Bound-State Problems. *Physical Review* **1951**, *84*, 1232–1242.
- (42) Albrecht, S.; Reining, L.; Sole, R. D.; Onida, G. Ab Initio Calculation of Excitonic Effects in the Optical Spectra of Semiconductors. *Physical Review Letters* **1998**, *80*, 4510–4513.
- (43) Strinati, G. Application of the Green's functions method to the study of the optical properties of semiconductors. *La Rivista del Nuovo Cimento* **1988**, *11*, 1–86.
- (44) Martin, R. M.; Reining, L.; Ceperley, D. M. *Interacting Electrons*; Cambridge University Press, 2016.
- (45) Rohlfing, M.; Louie, S. G. Electron-Hole Excitations in Semiconductors and Insulators. *Physical Review Letters* **1998**, *81*, 2312–2315.
- (46) Benedict, L. X.; Shirley, E. L.; Bohn, R. B. Optical Absorption of Insulators and the Electron-Hole Interaction: An Ab Initio Calculation. *Physical Review Letters* **1998**, *80*, 4514–4517.
- (47) Onida, G.; Reining, L.; Rubio, A. Electronic excitations: density-functional versus many-body Green's-function approaches. *Reviews of Modern Physics* **2002**, *74*, 601–659.
- (48) Palumbo, M.; Hogan, C.; Sottile, F.; BagalÀa, P.; Rubio, A. Ab initio electronic and optical spectra of free-base porphyrins: The role of electronic correlation. *The Journal of Chemical Physics* **2009**, *131*, 084102.
- (49) Ping, Y.; Rocca, D.; Galli, G. Electronic excitations in light absorbers for photoelectrochemical energy conversion: first principles calculations based on many body perturbation theory. *Chemical Society Reviews* **2013**, *42*, 2437.
- (50) Blase, X.; Duchemin, I.; Jacquemin, D. The Bethe-Salpeter Equation in Chemistry: Relations with TD-DFT, Applications and Challenges. *Chem. Soc. Rev.* **2018**, *47*, 1022–1043.
- (51) Hedin, L. New Method for Calculating the One-Particle Green's Function with Application to the Electron-Gas Problem. *Physical Review* **1965**, *139*, A796–A823.
- (52) Strinati, G.; Mattausch, H. J.; Hanke, W. Dynamical Correlation Effects on the Quasiparticle Bloch States of a Covalent Crystal. *Physical Review Letters* **1980**, *45*, 290–294.
- (53) Hybertsen, M. S.; Louie, S. G. Electron correlation in semiconductors and insulators: Band gaps and quasiparticle energies. *Physical Review B* **1986**, *34*, 5390–5413.
- (54) Casida, M. E. Time-dependent density-functional theory for molecules and molecular solids. *Journal of Molecular Structure: THEOCHEM* **2009**, *914*, 3–18.
- (55) Faber, C.; Boulanger, P.; Attaccalite, C.; Duchemin, I.; Blase, X. Excited states properties of organic molecules: from density functional theory to the GW and Bethe-Salpeter Green's function formalisms. *Phil. Trans. R. Soc. A* **2014**, *372*, 20130271.
- (56) Jacquemin, D.; Duchemin, I.; Blase, X. Is the Bethe-Salpeter Formalism Accurate for Excitation Energies? Comparisons with TD-DFT, CASPT2, and EOM-CCSD. *The Journal of Physical Chemistry Letters* **2017**, *8*, 1524–1529.
- (57) Jacquemin, D.; Duchemin, I.; Blase, X. Benchmarking the Bethe-Salpeter Formalism on a Standard Organic Molecular Set. *Journal of Chemical Theory and Computation* **2015**, *11*, 3290–3304.
- (58) Bruneval, F.; Hamed, S. M.; Neaton, J. B. A systematic benchmark of the ab initio Bethe-Salpeter equation approach for low-lying optical excitations of small organic molecules. *The Journal of Chemical Physics* **2015**, *142*, 244101.
- (59) Perdew, J. P.; Burke, K.; Ernzerhof, M. Generalized Gradient Approximation Made Simple. *Phys. Rev. Lett.* **1996**, *77*, 3865–3868.
- (60) Perdew, J. P.; Ernzerhof, M.; Burke, K. Rationale for mixing exact exchange with density functional approximations. *The Journal of Chemical Physics* **1996**, *105*, 9982–9985.
- (61) Perdew, J. P.; Levy, M. Comment on "Significance of the Highest Occupied Kohn-Sham Eigenvalue". *Phys. Rev. B* **1997**, *56*, 16021–16028.
- (62) Kshirsagar, A. R.; D'Avino, G.; Blase, X.; Li, J.; Poloni, R. Accurate Prediction of the S1 Excitation Energy in Solvated Azobenzene Derivatives via Embedded Orbital-Tuned Bethe-Salpeter Calculations. *Journal of Chemical Theory and Computation* **2020**, *16*, 2021–2027.
- (63) Blase, X.; Attaccalite, C.; Olevano, V. First-Principles GW calculations for fullerenes, porphyrins, phthalocyanine, and other Molecules of Interest for Organic Photovoltaic Applications. *Phys. Rev. B* **2011**, *83*, 115103.
- (64) Kaplan, F.; Harding, M. E.; Seiler, C.; Weigend, F.; Evers, F.; van Setten, M. J. Quasi-Particle Self-Consistent GW for Molecules. *J. Chem. Theory Comput.* **2016**, *12*, 2528–2541.
- (65) Giannozzi, P. et al. QUANTUM ESPRESSO: a modular and open-source software project for quantum simulations of materials. *Journal of Physics: Condensed Matter* **2009**, *21*, 395502.
- (66) Giannozzi, P. et al. Advanced capabilities for materials modelling with Quantum ESPRESSO. *Journal of Physics: Condensed Matter* **2017**, *29*, 465901.
- (67) van Setten, M.; Giantomassi, M.; Bousquet, E.; Verstraete, M.; Hamann, D.; Gonze, X.; Rignanese, G.-M. The PseudoDojo: Training and grading a 85 element optimized norm-conserving pseudopotential table. *Computer Physics Communications* **2018**, *226*, 39–54.
- (68) Hamann, D. R. Optimized norm-conserving Vanderbilt pseudopotentials. *Physical Review B* **2013**, *88*.
- (69) Hamada, I. van der Waals density functional made accurate. *Physical Review B* **2014**, *89*.
- (70) Marini, A.; Hogan, C.; Grüning, M.; Varsano, D. yambo: An ab initio tool for excited state calculations. *Computer Physics Communications* **2009**, *180*, 1392–1403.
- (71) Sangalli, D. et al. Many-body perturbation theory calculations using the yambo code. *Journal of Physics: Condensed Matter* **2019**, *31*, 325902.
- (72) Oshlies, A.; Godby, R. W.; Needs, R. J. GWself-energy calculations of carrier-induced band-gap narrowing in-type silicon. *Physical Review B* **1995**, *51*, 1527–1535.
- (73) Bruneval, F.; Gonze, X. Accurate GWself-energies in a plane-wave basis using only a few empty states: Towards large systems. *Physical Review B* **2008**, *78*.
- (74) Aprà, E. et al. NWChem: Past, present, and future. *The Journal of Chemical Physics* **2020**, *152*, 184102.
- (75) Dunning Jr, T. H. Gaussian Basis Sets for Use in Correlated Molecular Calculations. I. The Atoms Boron through Neon and Hydrogen. *J. Chem. Phys.* **1989**, *90*, 1007–1023.
- (76) Woon, D. E.; Dunning Jr, T. H. Gaussian basis sets for use in correlated molecular calculations. III. The atoms aluminum through argon. *J. Chem. Phys.* **1993**, *98*, 1358–1371.
- (77) Papajak, E.; Leverentz, H. R.; Zheng, J.; Truhlar, D. G. Efficient Diffuse Basis Sets: cc-pVxZ+ and maug-cc-pVxZ. *Journal of Chemical Theory and Computation* **2009**, *5*, 1197–1202.
- (78) Roy, L. E.; Hay, P. J.; Martin, R. L. Revised basis sets for the LANL effective core potentials. *J. Chem. Theory Comput.* **2008**, *4*, 1029–1031.
- (79) Hay, P. J.; Wadt, W. R. Ab initio effective core potentials for molecular calculations. Potentials for the transition metal atoms Sc to Hg. *The Journal of Chemical Physics* **1985**, *82*, 270–283.
- (80) Blase, X.; Attaccalite, C. Charge-Transfer Excitations in Molecular Donor-Acceptor Complexes within the Many-Body Bethe-Salpeter Approach. *Appl. Phys. Lett.* **2011**, *99*, 171909.
- (81) Weigend, F. Accurate Coulomb-fitting basis sets for H to Rn. *Physical Chemistry Chemical Physics* **2006**, *8*, 1057.
- (82) York, D. M.; Karplus, M. A Smooth Solvation Potential based on the Conductor-Like Screening Model. *J. Phys. Chem. A* **1999**, *103*, 11060–11079.
- (83) Duchemin, I.; Jacquemin, D.; Blase, X. Combining the GW formalism with the polarizable continuum model: A state-specific non-equilibrium approach. *J. Chem. Phys.* **2016**, *144*, 164106.
- (84) Li, J.; D'Avino, G.; Duchemin, I.; Beljonne, D.; Blase, X. Combining the Many-Body GW Formalism with Classical Polarizable Models: Insights on the Electronic Structure of Molecular Solids. *J. Phys. Chem. Lett.* **2016**, *7*, 2814–2820.
- (85) D'Avino, G.; Muccioli, L.; Zannoni, C.; Beljonne, D.; Soos, Z. G. Electronic Polarization in Organic Crystals: A Comparative Study of Induced Dipoles and Intramolecular Charge Redistribution Schemes. *J. Chem. Theory Comput.* **2014**, *10*, 4959–4971.
- (86) Li, J.; D'Avino, G.; Duchemin, I.; Beljonne, D.; Blase, X. Accurate description of charged excitations in molecular solids from embedded many-body perturbation theory. *Phys. Rev. B* **2018**, *97*, 035108.
- (87) Duchemin, I.; Guido, C. A.; Jacquemin, D.; Blase, X. The Bethe-Salpeter formalism with polarisable continuum embedding: reconciling linear-response and state-specific features. *Chem. Sci.* **2018**, *9*, 4430–4443.
- (88) van Setten, M. J.; Caruso, F.; Sharifzadeh, S.; Ren, X.; Scheffler, M.; Liu, F.; Lischner, J.; Lin, L.; Deslippe, J. R.; Louie, S. G.; Yang, C.; Weigend, F.; Neaton, J. B.; Evers, F.; Rinke, P. GW100: Benchmarking G0W0 for Molecular Systems. *Journal of Chemical Theory and Computation* **2015**, *11*, 5665–5687, PMID: 26642984.
- (89) Golze, D.; Dvorak, M.; Rinke, P. The GW Compendium: A Practical Guide to Theoretical Photoemission Spectroscopy. *Frontiers in Chemistry* **2019**, *7*.
- (90) Rangel, T.; Hamed, S. M.; Bruneval, F.; Neaton, J. B. Evaluating the GW Approximation with CCSD(T) for Charged Excitations Across the Oligoacenes. *Journal of Chemical Theory and Computation* **2016**, *12*, 2834–2842.
- (91) Marom, N.; Caruso, F.; Ren, X.; Hofmann, O. T.; Körzdörfer, T.; Chelikowsky, J. R.; Rubio, A.; Scheffler, M.; Rinke, P. Benchmark of GW methods for azobenzenes. *Physical Review B* **2012**, *86*.
- (92) Bruneval, F.; Marques, M. A. L. Benchmarking the Starting Points of the GW Approximation for Molecules. *Journal of Chemical Theory and Computation* **2012**, *9*, 324–329.
- (93) Faber, C. Electronic, excitonic and polaronic properties of organic systems within the many-body GW and Bethe-Salpeter formalisms: towards organic photovoltaics. Theses, Université de Grenoble, 2014.
- (94) Beharry, A. A.; Sadovski, O.; Woolley, G. A. Azobenzene Photoswitching without Ultraviolet Light. *Journal of the American Chemical Society* **2011**, *133*, 19684–19687.