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1                   Adsorption of proteins on TiO<sub>2</sub> particles influences their  
2                                   aggregation and cell penetration

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12

13            **Abstract**

14            TiO<sub>2</sub> nanoparticles known as E171 are one controversial food additive due to its potential toxicity.  
15            In this work, the main hypothesis is that the proteins adsorbed on the TiO<sub>2</sub> nanoparticles prevent  
16            their aggregation and favor the cell penetration. To do so, the TiO<sub>2</sub> nanoparticles were coated  
17            with gelatin and β-lactoglobulin (two proteins commonly found in food) to reach interfacial  
18            concentrations about 0.25 mg/mg and 0.32 mg/mg, respectively. The measurement of NP size  
19            showed that the protein coating improve the colloidal stability of TiO<sub>2</sub> nanoparticles. The FTIR  
20            analysis suggests that the β-lactoglobulin structure is modified after adsorption. Then, the  
21            penetration of TiO<sub>2</sub> penetration inside human intestinal epithelial cells was shown and quantify  
22            by using confocal Raman microscopy. Finally, the promoting role of the protein coating on the cell  
23            penetration was demonstrated for both the gelatin and β-lactoglobulin.

24

25           **1. Introduction**

26       Nanoparticles (NP) are commonly used in food as texturing and flavor agents (McCracken et al.,  
27       2016). One widely used additive is the food grade TiO<sub>2</sub> also identified as E171, which is found in  
28       various food products (including candies and chocolate bars) as well as toothpaste. In 2012, a  
29       simulation of TiO<sub>2</sub> consumption emphasizes that children below 10 years old are the most exposed  
30       individuals (Weir et al., 2012). TiO<sub>2</sub> is also used for packaging and, in that case, food can be  
31       contaminated by the packing (Lin et al., 2014). The question about the innocuousness of TiO<sub>2</sub>  
32       nanoparticles (TiO<sub>2</sub> NP) is not new since in 2011, Skocaj et al. addressed the question: "Titanium  
33       dioxide in our everyday life; is it safe?" (Skocaj et al., 2011). Closer than 10 years after, the impact  
34       of food grade TiO<sub>2</sub> on health is not totally elucidated and there is a need for interdisciplinary  
35       approaches to understand organ/NP interaction (Chaudhry et al., 2008).

36       Numerous effects of TiO<sub>2</sub> have been reported on different organs such as liver, brain, intestines  
37       or spleen (Jovanović, 2015). The TiO<sub>2</sub> NPs are known as inducers of inflammation in several organs  
38       such as kidney (Gui et al., 2011). In addition, genotoxicity has been observed for numerous  
39       structures of TiO<sub>2</sub> NPs (including anatase and rutile) with different exposure modes (ingestion,  
40       inhalation...) (Shi et al., 2013). The TiO<sub>2</sub> NPs induce oxidative damages to DNA. The E171 toxicity  
41       on intestinal cells is moderated but sufficient to suggest a role in tumor growth in the colon of  
42       mice (Dorier et al., 2017; Proquin et al., 2018). Moreover, E171 is suspected to worsen existing  
43       intestinal disease (Urrutia-Ortega et al., 2016). Indeed, the sedimentation of TiO<sub>2</sub> could induce a  
44       loss of intestinal microvilli (Faust et al., 2014) and a slight deregulation of the fatty acids profiles  
45       in the intestine. However, the TiO<sub>2</sub> NPs do not seem to impact the bacteria gas production  
46       (Dudefoi et al., 2017). Finally, in cells, TiO<sub>2</sub> NPs were found to disrupt the structure of lysosomes  
47       and to damage the mitochondria (Zhang et al., 2018) and could have synergy action with food  
48       molecules such as glucose (Wang et al., 2013).

49 The relevant properties which explain NP toxicity are their solubility, shape, surface charge, size  
50 distribution and structure (McCracken et al., 2016). Cell penetration was reported after dermal  
51 exposure and in hairless skin, TiO<sub>2</sub> NPs were found in epidermis cells (Wu et al., 2009). Basically,  
52 TiO<sub>2</sub> NPs can penetrate in cells as shown for MCF-7 and TERT cells using confocal Raman  
53 spectroscopy (Salehi et al., 2014). Recently, debate around the use of TiO<sub>2</sub> in food was revived. In  
54 2017, Bettini et al. reported that after 100 days, the TiO<sub>2</sub> nanoparticles initiate preneoplastic  
55 lesions and promote colon inflammation. This suggested a potential role of TiO<sub>2</sub> NPs in  
56 autoimmune disease and colorectal cancers (Bettini et al., 2017). Following this publication, the  
57 French Agency for Food, Environmental and Occupational Health and Safety has recommended  
58 the limitation the TiO<sub>2</sub> as food additive as emergency measure regarding the lack of knowledge  
59 about the real toxicity (Anses, 2019).

60 As the other nanoparticles, TiO<sub>2</sub> tends to aggregate in solution. This is enhanced by the presence  
61 of salt in the media (Allouni et al., 2009). The ability of proteins to prevent nanoparticle  
62 aggregation is well known (Lepoitevin et al., 2015). For noble metals, such property is modulated  
63 to design colorimetric sensors (Sabela et al., 2017). The coating of TiO<sub>2</sub> NPs by human serum  
64 albumin, bovine serum albumin or fetal bovine serum prevents the aggregation process (Allouni  
65 et al., 2009; Yusoff et al., 2018). In their review, McCracken et al. hypothesized that NP coating by  
66 protein could be an important factor of their toxicity (McCracken et al., 2016).

67 Protein adsorption on material is a complex and fascinating phenomena whose mechanism is not  
68 fully understood after several decades of investigation. However, the structural classification of  
69 proteins could be a way to predict their behavior on a flat surface (Coglitore et al., 2019).  
70 Numerous parameters affect protein adsorption on nanoparticles such as pH, concentration,  
71 temperature, salt concentration (Bhakta et al., 2015) and additional molecules such as polyphenol  
72 (Coglitore et al., 2018). Hard proteins such as lysozyme keep their structure after adsorption on

73 solid interface (Balme et al., 2013). Although considered as soft protein, BSA does not unfold on  
74 food grade E171 (Kim & Doudrick, 2019). The adsorption of blood proteins on anatase TiO<sub>2</sub> NPs  
75 affects the agglomerate size and surface charge. Additionally, this altered the electrostatic binding  
76 affinity with fibroblasts (Allouni et al., 2015). Because proteins can strongly interact with TiO<sub>2</sub> NPs,  
77 we addressed the question about the consequences of such interaction. Typically, under salted  
78 media, NP aggregation reduces the ability of the NP to penetrate inside the cell. Food proteins,  
79 by preventing this aggregation process, could therefore enhance the penetration of TiO<sub>2</sub> NPs in  
80 the cell. In order to verify this hypothesis, we selected two proteins *i.e.* β-lactoglobulin and gelatin.  
81 β-lactoglobulin is mainly composed of β-sheets which provide a high internal energy. In presence  
82 of silica or hydrophobic surface, the conformational changes of β-lactoglobulin are small  
83 (Wahlgren & Arnebrant, 1990). When β-lactoglobulin is adsorbed on clay, only weak structural  
84 modification occur (Assifaoui et al., 2014). The second protein is gelatin, which is widely used in  
85 candies mixed with E171. Gelatin interacts with TiO<sub>2</sub> NPs and such interaction has been used to  
86 pattern TiO<sub>2</sub> porous microspheres (Liu et al., 2015). However, the main application of gelatin TiO<sub>2</sub>  
87 composite is food packaging (He et al., 2016).

88 Mostly, intracellular imaging methods, such as electron microscopy, cryoelectron microscopy, and  
89 immunofluorescence microscopy, due to the fixation, freezing and use of dyes or biomarkers are  
90 destructive and harmful to the cells. As Raman spectroscopy does not need chemical fixation,  
91 markers, or genetic modifications, it is considered as a non-invasive imaging method (Klein et al.,  
92 2012; Salehi et al., 2013). Raman spectroscopy is based on inelastic scattering of photons from  
93 the incident wavelength. The monochromatic light interacting with the sample, produces a  
94 Raman spectrum, composing different bands related to the vibrational frequencies of different  
95 functional groups. Subsequently every molecule has an exclusive fingerprint or Raman spectrum.  
96 The high spatial resolution and possibility of imaging in aqueous environment make Raman

97 microscopy an ideal tool for life and fixed imaging of a single cell (Salehi et al., 2014). The adequate  
98 data analysis according to the vibrational spectra of different cellular organelles, anticancer drugs  
99 or nanoparticles provides alternative to existing methods for cell imaging under normal  
100 physiological conditions (Gulka et al., 2020; Salehi et al., 2018). The unique spectral signature of  
101 nanoparticles enable tracing coated/ non-coated TiO<sub>2</sub> cellular penetration.

102 As previously mentioned, the toxicity of TiO<sub>2</sub> was demonstrated on different type cells in vitro as  
103 well as in-vivo. In these cases, numerous external parameters can influence the cell penetration  
104 of the nanoparticle including the protein adsorption. This work aims to verify for the first time the  
105 hypothesis of the role the protein coating on the TiO<sub>2</sub> nanoparticle cell penetration. Indeed, this  
106 question is motivated to elucidate the first step mechanism of the tiO<sub>2</sub> toxicity. To do so, gelatin  
107 and β-lactoglobulin adsorption on TiO<sub>2</sub> rutile was first investigated to determine their surface  
108 concentration and eventual structural modifications. Then, we evaluated the impact of each of  
109 the two proteins on TiO<sub>2</sub> NPs aggregation. Finally, the penetration of NPs coated or not with  
110 proteins was investigated by confocal Raman microscopy.

111

## 112 **2. Materials and methods**

### 113 **a. Preparation of NPs**

114 The titanium dioxide (TiO<sub>2</sub>) powder (99% rutile) (SIGMA) was suspended in distilled water at a  
115 concentration of 1 mg/ml. The solution was then sonicated for 1 hour at 60°C (ref sonicator: ELMA  
116 Ultrasonic Cleaner S100H) to break up any interactions that may have formed between the  
117 different particles. Gelatin (L3908) and β-lactoglobulin (G6144) from Sigma were dissolved in  
118 distilled water at a concentration of 1 mg/ml and sonicated. Three solutions were prepared at a  
119 final concentration of 0.75 mg/ml of TiO<sub>2</sub> NPs: NPs alone, NPs with gelatin and NPs with β-  
120 lactoglobulin. In these solutions, the ratio was about 3 TiO<sub>2</sub> NPs to 1 protein. In order to promote

121 the adsorption of these proteins on NPs as effectively as possible, the solutions were placed on  
122 an agitator for one hour at room temperature and then stored at 4°C. A sonication of 5 minutes  
123 at room temperature for coated rated NPs and one hour at 60°C for unrated uncoated NPs was  
124 applied before each use.

#### 125 **b. Characterization of protein adsorption**

126 The interfacial concentration of protein on TiO<sub>2</sub> NPs was obtained by the method of supernatant  
127 depletion previously reported (Lepoitevin et al., 2014). Briefly, after incubation, the mixtures  
128 containing TiO<sub>2</sub> NPs and proteins were centrifuged (15 min at 15000 rpm). Protein concentration  
129 in the supernatant was determined by UV-Vis absorption (Jasco) at 290 nm. The structural  
130 modifications of adsorbed protein were characterized by ATR-FTIR (Nexus) under D<sub>2</sub>O. The  
131 measurement of NP size was performed by Dynamic Light Scattering (DLS) (Nanophox) using a  
132 laser light of 632 nm. For each condition or type of NPs, the measurement was carried out in  
133 triplicate. Several parameters such as medium (water, DMEM/F12 medium and RPMI medium)  
134 condition and time (from day 0 to day 5) were modified between different measurements. The  
135 data analysis was done using QuickFit software.

#### 136 **c. Cell culture**

137 HT29 human colon adenocarcinoma cells were grown in DMEM/F12 medium (M1) or RPMI  
138 medium (M2) supplemented with 10% FCS, 100U/ml penicillin, 100 mg/ml streptomycin and  
139 100 mg/ml sodium pyruvate. For Raman experiments, 3x10<sup>5</sup> HT-29 cells were cultivated for 24 h  
140 onto polished and disinfected calcium fluoride CaF<sub>2</sub> (Crystran Ltd, Dorset, UK) substrates in 35 mm  
141 Petri dishes. After cell adherence to the substrate, the cells were incubated with dissolved  
142 nanoparticle solutions at 2 µg/mL concentration for one hour. The cells on the CaF<sub>2</sub> substrates  
143 were then fixed with 2% PFA (paraformaldehyde) after thorough rinsing with PBS. Cells were kept  
144 in PBS and transferred directly for Raman measurements. Before carrying out measurements of

145 TiO<sub>2</sub> on the cells, several reference spectra of the different NP solutions were carried out for 4  
146 conditions, TiO<sub>2</sub> powder and the three solutions of TiO<sub>2</sub> coated or not by proteins (NP/NPG/NPL).

147 **d. Analysis of cell penetration by Raman microscopy**

148 ***Raman data acquisition***

149 To collect the Raman spectra, Witec Confocal Raman Microscope System alpha 300R (Witec Inc.,  
150 Ulm, Germany) was used. The excitation light in confocal Raman microscopy was generated via a  
151 frequency-doubled Nd:YAG lasers (New-port, Irvine, CA, USA) at a wavelength of 532 nm. A 60×  
152 NIKON water immersion objective with a numerical aperture of 1.0 and a working distance of  
153 2.8 mm (Nikon, Tokyo, Japan) was used to focus the laser beam onto the cells. The laser power  
154 after the objective was 15 mW but finally, lower power was absorbed by cells in PBS. The  
155 scattered radiation goes through an edge filter to the electron multiplying charge coupled device  
156 camera EMCCD (DU 970N-BV353, Andor, Hartford, USA). The EMCCD chip size was 1600 × 200  
157 pixels, the camera controller of a 16-bit A/D converter operated at 2.5 MHz. The acquisition time  
158 of a single spectrum was set to 0.5 sec. A zone of 150 × 150 pixels (spatial unit) per image was  
159 recorded, giving 22500 spectra for one image. Data acquisition and analysis was performed using  
160 Image Plus 2.08 Witec software. A spatial resolution of 300 nm and a depth resolution of 1µm  
161 were measured for the system. Considering the limit of Raman microscope (300 nm), the  
162 measurements of particle sizes on each image (processed and analyzed using Witec software)  
163 were done by zooming on particles and using line size measurement tool.

164 ***Raman data analysis***

165 Two data analysis methods have been applied. The first method presented integrated Raman  
166 intensities in specific regions in particular C-H stretching mode. The lipid-protein distribution in  
167 the cells is shown by the integrated Raman intensities of C-H stretching mode (2800–3000cm<sup>-1</sup>)  
168 using Image Plus software, Witec. Consequently, a map of the region regarding these integrated



169 intensities was provided. A false color image of CH Raman peak contains bright yellow hues for  
170 the highest intensities and dark orange hues for the lowest integrated intensities. K-mean cluster  
171 analysis (KMCA) as the second data analysis method separated data into k-mutually clusters.  
172 KMCA was done using the Witec Project Plus (Ulm, Germany) software.

### 173 **e. Statistical analysis**

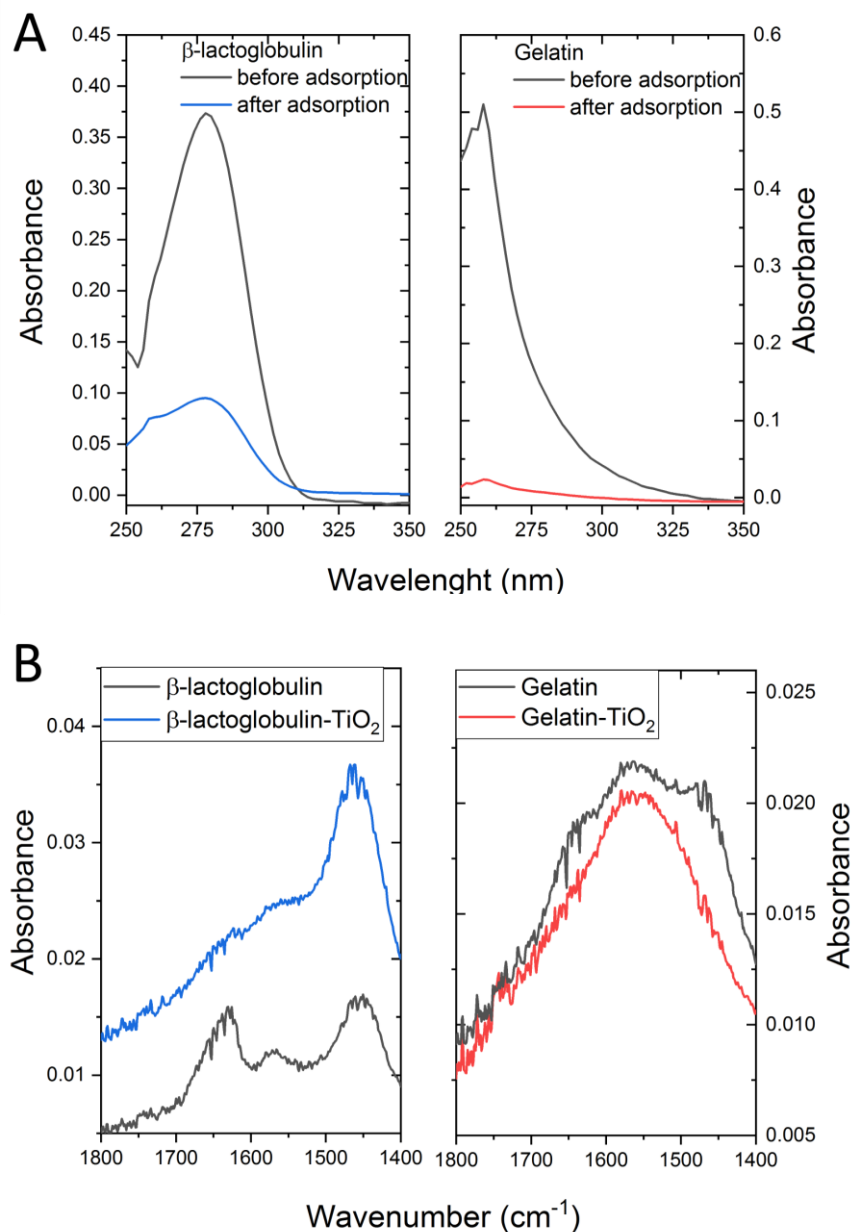
174 All experiments were conducted independently at least three times. Results were expressed as  
175 the mean  $\pm$  standard error of the mean (S.E.M). Statistical comparisons were performed with one-  
176 way ANOVA or t-test as indicated. A probability level (p value) of 0.05 was chosen for statistical  
177 significance.

178

## 179 **3. Results and discussion**

### 180 **a. Protein adsorption on TiO<sub>2</sub> NPs**

181 Prior to the investigation of the protein impact on colloidal stability and cell penetration, we  
182 characterized protein loading on TiO<sub>2</sub> NPs. Indeed, the protein/NP interaction is extremely  
183 dependent on the intrinsic properties of the material and on the media. Here, we have used water  
184 in order to prevent salt induced aggregation of TiO<sub>2</sub> NPs and thus to optimize the accessible  
185 surface of NPs for protein adsorption.



186

187 **Figure 1: (A) Absorbance spectra of  $\beta$ -lactoglobulin (left) and gelatin (right) before (black line)**

188 **and after (red line) contact with  $\text{TiO}_2$  NPs at a concentration of 0.75 mg/ml for 1 hour. (B) FTIR**

189 **spectra of  $\beta$ -lactoglobulin (left) and gelatin (right) in  $\text{D}_2\text{O}$  (black line) and loaded on  $\text{TiO}_2$  NPs**

190 **(blue or red line)**

191 The interfacial concentration of protein was determined from the depletion of supernatant

192 method. Figure 1A shows the absorbance spectra for  $\beta$ -lactoglobulin and gelatin before and after

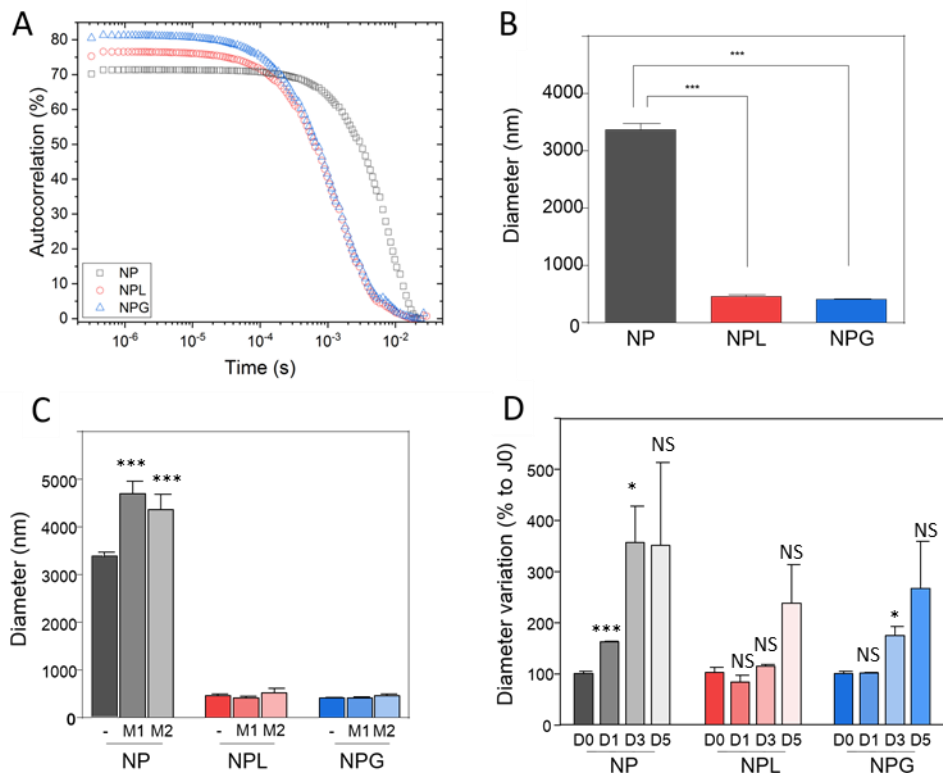
193 adsorption on TiO<sub>2</sub> NPs. We can observe a large decrease of protein concentration in solution  
194 upon adsorption. The interfacial concentrations for β-lactoglobulin and gelatin were found about  
195 0.25 mg/mg and 0.32 mg/mg of TiO<sub>2</sub> NPs, respectively.

196 We attempted to obtain information about the structural modifications of proteins induced by  
197 their adsorption on TiO<sub>2</sub> NPs. In Figure 1B are plotted the FTIR spectra recorded under D<sub>2</sub>O. For  
198 the β-lactoglobulin, the band in 1640 cm<sup>-1</sup> was assigned to the amide I. In D<sub>2</sub>O, the amides II gave  
199 two bands in 1560 cm<sup>-1</sup> and 1455 cm<sup>-1</sup> relative to N-H and N-D bonds, respectively. This  
200 emphasizes the accessibility to the solvent of the amide II moieties to exchange H by D. After  
201 loading on TiO<sub>2</sub> NPs, the 3 bands were still present. Unfortunately, the absorption of Ti-OD bond  
202 about 1590 made a deep structural analysis impossible. However, we observed an increase of the  
203 band in 1455 cm<sup>-1</sup> that could be assigned to the conversion of N-H to N-D. This suggested that the  
204 amides I bound with H were more exposed to the solvent after loading on TiO<sub>2</sub> NPs due to  
205 structural modifications.

206 For gelatin in D<sub>2</sub>O, the 3 bands were also present but more difficult to distinguish because of the  
207 low order degree of the protein. However, we noticed that the two bands of amides I were large  
208 showing the existence of N-H and N-D bonds. After loading, the band about 1690 cm<sup>-1</sup> masked  
209 the other ones. Conversely to β-lactoglobulin, the adsorption of gelatin did not seem to promote  
210 the exposition of the amides II to the solution.

#### 211 **b. Colloidal stability of TiO<sub>2</sub> NPs with and without protein loading**

212 As previously mentioned, the colloidal stability of TiO<sub>2</sub> NPs could strongly be affected by protein  
213 adsorption. Importantly, this is also a key factor for cell penetration. Thus, we then investigated  
214 by dynamic light scattering the TiO<sub>2</sub> NP size, in water and in two cell culture media.



215

216 **Figure 2: Nanoparticle size by diffusion light scattering (A) Autocorrelation function of TiO<sub>2</sub> NPs**  
 217 **without (black) and with protein (in red, β-lactoglobulin and in blue, gelatin). The sizes of TiO<sub>2</sub>**  
 218 **NPs are obtained in water (B) and culture media (DMEM/F12 medium (M1) or RPMI medium**  
 219 **(M2)) (C). (D) Size evolution of the small component as a function of time (From Day 0 to Day**  
 220 **5). TiO<sub>2</sub> NPs alone is noted NP, TiO<sub>2</sub> NPs with gelatin is noted NPG and TiO<sub>2</sub>-NPs with β-**  
 221 **lactoglobulin is noted NPL. Statistical significance is shown as p-value from one-way Anova test**  
 222 **(panels B and C) or t-test in panel D (\*: p<0.05; \*\*\*: p<0.001).**

223

224 In Figure 2A, the autocorrelation functions for TiO<sub>2</sub> NPs in water, with and without proteins are  
 225 reported. We observed a shift toward short times when TiO<sub>2</sub> NPs were coated with proteins. At  
 226 first approximation, the autocorrelation functions were fitted with only one component to obtain  
 227 an average size distribution. For uncoated TiO<sub>2</sub> NPs, the distribution was centered around 3.2 μm

228 while after protein coating it was about 400 nm (Figure 2B). This is due to a rapid aggregation of  
229 TiO<sub>2</sub> NPs after sonication. Here, we can note that both gelatin and β-lactoglobulin prevented the  
230 fast aggregation phenomena of TiO<sub>2</sub> NPs.

231 In general, NP aggregation is favored by salt addition. Thus, similar analyses were performed using  
232 two different media used for cell culture (Figure 2C). This experiment was also motivated because  
233 the role of protein loading cannot be predicted since it is dependent on the protein and the  
234 eventual binding with small molecules, as shown for gold NP (Coglitore et al., 2018; Lepoitevin et  
235 al., 2015) or clay mineral (Trigueiro et al., 2018).

236 As expected, the size of raw TiO<sub>2</sub> NPs increased due to salt-induced aggregation. Conversely,  
237 coating with proteins appeared to prevent the aggregation process. These results agree with  
238 previous investigations showing that the extracellular polymeric substances from *Bacillus subtilis*  
239 improve the colloidal stability of TiO<sub>2</sub> NPs (Di Lin et al., 2017; Di Lin et al., 2016). Gelatin has a low  
240 internal energy structure and thus can optimize its conformation around the TiO<sub>2</sub> NPs. Conversely,  
241 β-lactoglobulin is mainly composed of β-sheet, has a high internal energy and is positively charged  
242 at pH 7.4. However, the FTIR revealed structural modifications, which probably allowed the  
243 interaction between the protein and TiO<sub>2</sub> NPs. This means that the electrostatic interaction  
244 between the TiO<sub>2</sub> NPs and β-lactoglobulin are not shielded by salt addition thus explaining the  
245 colloidal stability.

246 TiO<sub>2</sub> NPs penetrate inside cells only if their size are hundreds nanometer scale. The low accuracy  
247 of the autocorrelation fit with only one component suggested a polydispersity of the samples. To  
248 further analysis the size of TiO<sub>2</sub> NPs, we separated two populations from the autocorrelation  
249 curves. A first population with a diameter below 100 nm was assigned to non-aggregated  
250 particles. The second population micrometer scale corresponded to aggregates. The relative  
251 weight of each population showed that non-aggregated particles were the main one (upper than

252 99%), for the TiO<sub>2</sub> NPs coated with proteins. We then focused on this population more prone to  
253 penetrate inside the cells. The stability of these small TiO<sub>2</sub> NPs as a function of time was  
254 investigated for duration up to 5 days (Figure 2D). Without protein coating, TiO<sub>2</sub> NPs were prone  
255 to aggregate with time as monitored by the diameter variation as a function of the time. Typically,  
256 at D0 the mean size of the TiO<sub>2</sub>-NP was about 100 nm and reached 480 nm after D5. This is not  
257 surprising since TiO<sub>2</sub> NPs are not stable in solution. With gelatin, the TiO<sub>2</sub> NPs size was stable until  
258 D1 around 91 nm. Then, the diameter variation increased to reach 250 % at D5. Finally, β-  
259 lactoglobulin offered the most effective protection against aggregation since the size of NPL  
260 particles remained constant during the 3-days.

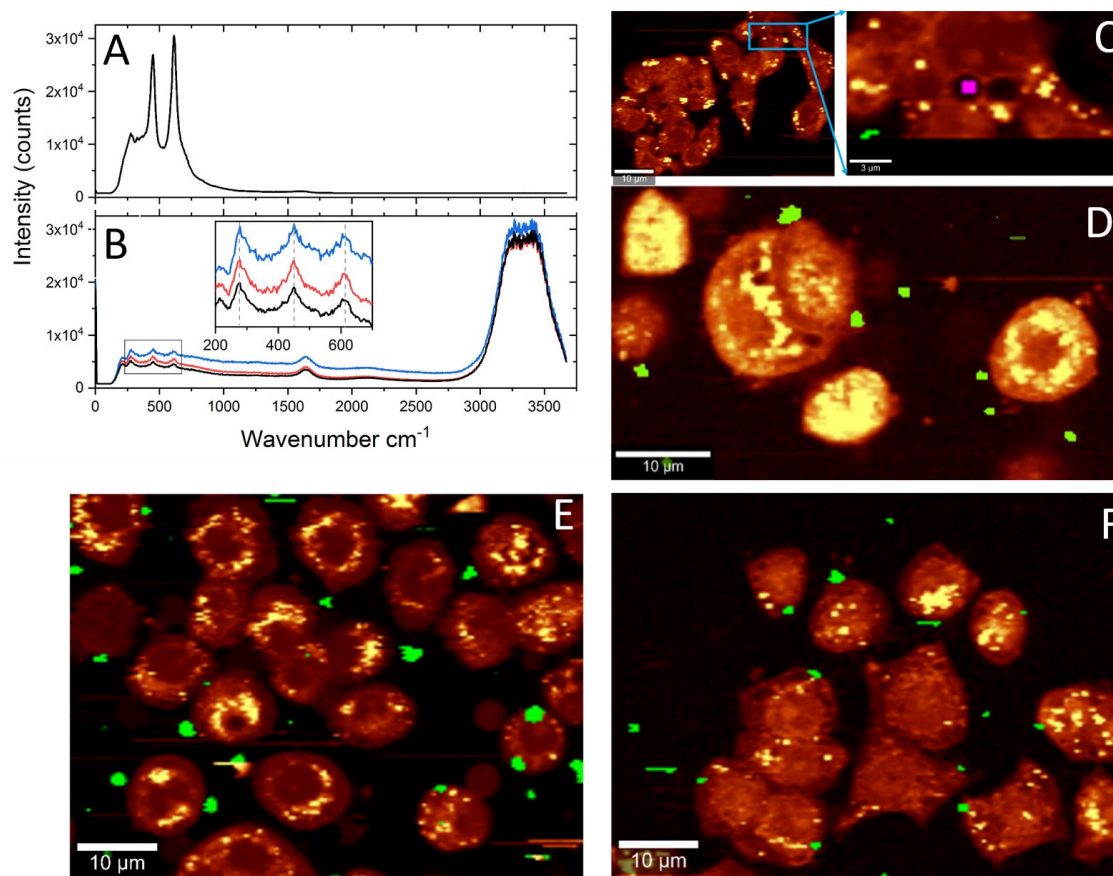
261 At this stage, we demonstrated that protein coating prevented the TiO<sub>2</sub> NPs aggregation in water  
262 as well as in salt media. On the other hand, the colloidal stability was also improved during the  
263 time especially for the β-lactoglobulin.

#### 264 **c. Effect of proteins on cell penetration of TiO<sub>2</sub> NPs**

265 To find out the role of protein, intracellular penetration of TiO<sub>2</sub> NPs was investigated using the  
266 human intestinal epithelial cell lines HT-29, cultured in DMEM-F12 medium. The characterization  
267 of TiO<sub>2</sub> NPs and their location were obtained using confocal Raman microscopy.

268 The Raman spectrum of TiO<sub>2</sub> powder showed three significant peaks at 275 cm<sup>-1</sup>, 450 cm<sup>-1</sup> and  
269 610 cm<sup>-1</sup> (Figure 3A). Coated/non-coated TiO<sub>2</sub> NPs in solution, showed the same peak position  
270 confirming the integrity of TiO<sub>2</sub> NPs. The low intensities of TiO<sub>2</sub> NPs coated with proteins (blue and  
271 red line for NPG and NPL, respectively) peaks were due to the coating, which avoids particles  
272 accumulation and lower Raman signal. In addition, powder signal was higher as compared to the  
273 NP in solution, as the number of particles under the laser spot was lower for the floating particles  
274 in solution.

275



276

277 **Figure 3: Raman spectra of (A) TiO<sub>2</sub> powder, (B) TiO<sub>2</sub>-NPs (black line), TiO<sub>2</sub>-NPs with gelatin (blue**  
 278 **line) and TiO<sub>2</sub>-NPs with β-lactoglobulin (red line), the inset is a zoom of 200 cm<sup>-1</sup> – 700 cm<sup>-1</sup>**  
 279 **region. (C) Integration over Raman intensities of CH region, HT-29 cells incubated with TiO<sub>2</sub>**  
 280 **NPGs and selected areas to focus on the TiO<sub>2</sub> presence inside the cell in pink color and outside**  
 281 **in green. The image (D), (E) and (F) were obtained in presence of TiO<sub>2</sub>-NPs, TiO<sub>2</sub>-NPs with gelatin**  
 282 **and TiO<sub>2</sub>-NPs with β-lactoglobulin respectively.**

283

284 Figure 3(B-E) illustrates the image acquisition of cells in presence of TiO<sub>2</sub> NPs with and without  
 285 protein coating. Raman images present the biomolecules and two clusters in pink (intracellular  
 286 penetrated particles) and green (extracellular particles). The intensity of CH bonds is plotted in  
 287 yellow hues have the maximum intensity (for proteins or lipids) and dark hues have zero intensity

288 of CH bonds (background and out of cells). After 1 hour incubation of HT-29 cells with raw  
289 nanoparticles, TiO<sub>2</sub> NPs aggregates were mainly present outside the cell or were in interaction  
290 with the cell membrane. Conversely, TiO<sub>2</sub> NPs coated with proteins were found inside the cells  
291 suggesting that protein coating favored cell penetration. We also noticed that TiO<sub>2</sub> NPs coated  
292 with proteins were more abundant as compared to the particles without coating.

293 Further analysis reported on Figure 4 shows the percentage of total penetration of  
294 coated/noncoated NPs. The quantification was obtained from a total number of 296 cells from all  
295 the Raman images that were analyzed. The average size of TiO<sub>2</sub> NPs inside the cells showed that  
296 the raw TiO<sub>2</sub> NPs were larger (1.4 μm) than the coated TiO<sub>2</sub> NPs (1 μm and 0.8 μm for gelatin and  
297 β-lactoglobulin, respectively). The spatial resolution of Raman microscopy is 300 nm and thus a  
298 discrepancy between the DLS is not surprising. Indeed, the Raman microscopy provides  
299 information on several nanoparticles that can count one by one as soon as their size is larger than  
300 the resolution while the DLS provides a means measurement of the nanoparticle size that can be  
301 distorted in the case of heterogeneous samples. However, the tendency between confocal Raman  
302 microscopy results and DLS measurements were in good agreement.

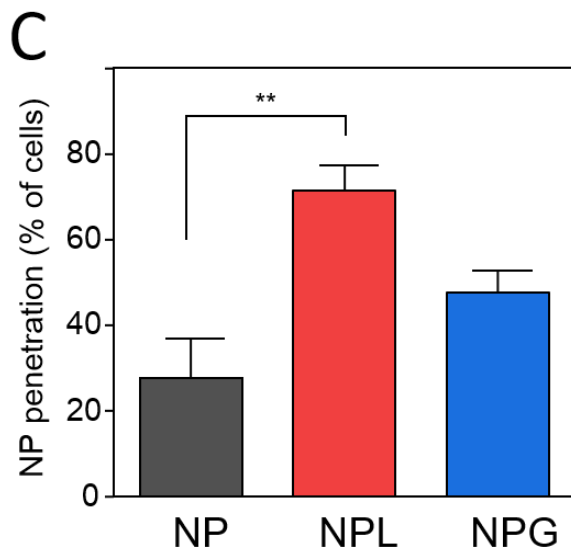
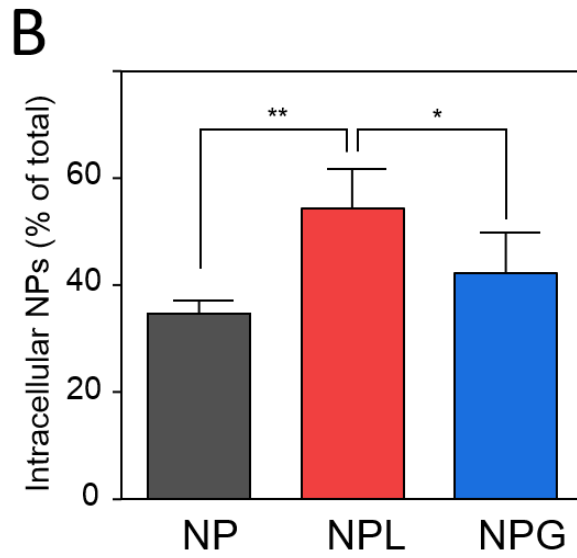
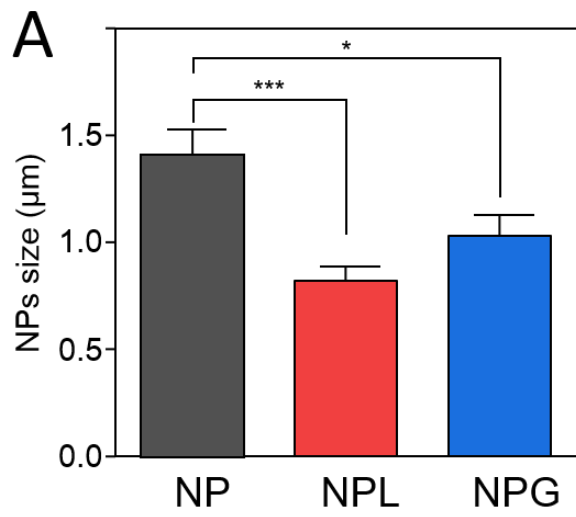
303 We also quantified the percentage of NPs that penetrated inside the cells (Figure 4B and C). For  
304 the raw TiO<sub>2</sub> NPs, 35% of particles were located inside HT-29 cells. This ratio increased up to 42%  
305 and 55% when the TiO<sub>2</sub> NPs were coated with gelatin or β-lactoglobulin, respectively. Such  
306 enhancement of cell penetration induced by protein coating was significant and probably linked,  
307 at least in part, to the difference in size of the three types of NPs. We also evaluated the  
308 distribution of TiO<sub>2</sub> NPs in the cell population. Without proteins, the TiO<sub>2</sub> NPs penetrated only 28%  
309 of the cells. The coating with protein favored the dissemination of TiO<sub>2</sub> NPs since 48% and 70% of  
310 cells showed internalized TiO<sub>2</sub> NPs after coating with gelatin and β-lactoglobulin, respectively. The



311 difference between the two proteins can be assigned to their ability to prevent the TiO<sub>2</sub> NPs  
312 aggregation.

313 As demonstrated, confocal Raman microscopy with high spatial resolution allowed us to analyze  
314 and trace coated/non-coated TiO<sub>2</sub> nanoparticles. This label-free method using the spectral  
315 fingerprint of nanoparticles permitted the monitoring of their intracellular penetration. The  
316 results obviously present the significant difference between coated particles agglomeration and  
317 penetration. Non-coated TiO<sub>2</sub> are more prone to aggregation and therefore their intracellular  
318 penetration was lower. Conversely, protein adsorption prevented the TiO<sub>2</sub> NPs aggregation and  
319 thus favored cell penetration.

320



322 **Figure 4: Characterization of TiO<sub>2</sub> NP penetration inside HT-29 cells (A) NP size, (B) ratio of**  
323 **intracellular TiO<sub>2</sub> NPs and (C) percentage of cells containing TiO<sub>2</sub> NPs. TiO<sub>2</sub> NPs alone is noted**  
324 **NP, TiO<sub>2</sub> NPs with gelatin is noted NPG, and TiO<sub>2</sub> NPs with β-lactoglobulin is noted NPL.**  
325 **Statistical significance is shown as p-value from one-way Anova test (\*: p<0.05; \*\*: p<**  
326 **0.01;\*\*\*: p<0.001).**

327

#### 328 **4. Conclusion**

329 To sum up, we investigated the impact of protein adsorption on cell penetration of TiO<sub>2</sub> NPs. Our  
330 results showed that both β-lactoglobulin and gelatin were loaded on TiO<sub>2</sub> NPs. The FTIR suggested  
331 that β-lactoglobulin (but not gelatin) adsorption induced a structural modification. The salt-  
332 induced aggregation and the colloidal stability were improved by protein adsorption. The cell  
333 penetration investigated by confocal Raman microscopy revealed that the β-lactoglobulin favored  
334 more efficiently than gelatin TiO<sub>2</sub> NP cell penetration. Moreover, our results showed a correlation  
335 between the ability of proteins to prevent NP aggregation and to facilitate cell penetration.

336 Overall, this work proves that proteins present in food have a significant impact on TiO<sub>2</sub> NP  
337 penetration in human intestinal epithelial cells. This tends to reinforce the hypothesis that NP  
338 coating by proteins could be an important factor explaining their toxicity. Thus, in further  
339 investigations aiming to understand the toxicity of TiO<sub>2</sub> NP, the role of food proteins should be  
340 taken into account.

341

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