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Probing ²⁹Si-¹⁷O connectivities and proximities by solid-state NMR

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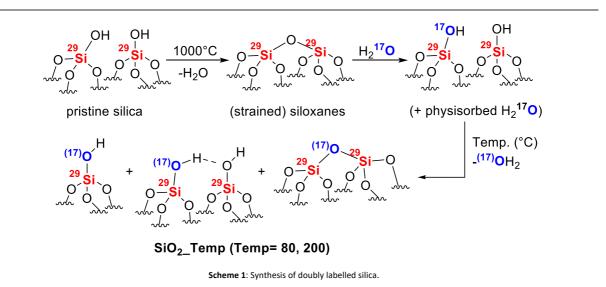
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Abstract:

The measurement of dipolar and *J*- couplings between ²⁹Si and ¹⁷O isotopes is challenging owing to (i) the low abundance of both isotopes and (ii) their close Larmor frequencies, which only differ by 19%. These issues are circumvented here by the use of isotopic enrichment and dedicated triple-resonance magic-angle spinning NMR probe. The surface of ²⁹Si-enriched silica was labelled with ¹⁷O isotope and heated at 80 and 200 °C. ²⁹Si-¹⁷O connectivities and proximities were probed using two-dimensional (2D) through-bond and through-space heteronuclear multiple-quantum coherences (*J*- and *D*-HMQC) experiments between ¹⁷O and ²⁹Si nuclei. The simulation of the build-up of the *J*- and *D*-HMQC signals allowed the first experimental measurement of *J*- and dipolar coupling constants between ¹⁷O and ²⁹Si nuclei. These HMQC experiments allow distinguishing two distinct siloxane (SiOSi) oxygen sites: (i) those covalently bonded to Q³ and Q⁴ groups, having a hydroxyl group as a second neighbour and (ii) those covalently bonded to two Q⁴ groups. The measured *J*- and dipolar coupling constants of siloxane ¹⁷O nucleus with Q⁴ ²⁹Si nuclei differ from those with Q³ ²⁹Si nuclei. These results indicate that the ²⁹Si-¹⁷O one-bond *J*-coupling and Si-O bond length depend on the second neighbours of the Si atoms.

Introduction:

The design and understanding of catalytic materials critically depend on efficient spectroscopic methods, in order to determine structure-activity relationships. Among the available characterisation techniques, solid-state NMR holds a prominent position, as it provides structural information down to the molecular level, even for sites lacking long-range positional order¹⁻³. Solid-state NMR spectroscopy allows probing the local environment of a nucleus, not only by measuring its chemical shift and its quadrupolar coupling, but also by observing connectivities and proximities through J- and dipolar couplings^{4, 5}. When relating this to catalytic materials, this powerful approach allows to draw a reliable picture on the nature and topology of the sites relevant to specific reactivity. More specifically, in the case of silica, a class of material that is used on a huge scale both as catalyst or as catalytic support, NMR techniques have been proved to be essential for the precise characterization of their surface and of the active sites within SiO₂-supported catalysts^{1, 6, 7}. Moreover, the silica exhibits different local surface environments, as a result from water elimination stemming from silanol condensation (Scheme 1). Whereas ¹H, ²⁹Si and ¹⁷O NMR and first-principles calculations are commonly used to probe the local structure of the silica⁸⁻¹³, ¹⁷O-²⁹Si heteronuclear correlation and the measurement of J- and dipolar couplings between these isotopes remain a great challenge. The interest in its overcoming is substantial: such correlation would provide bond and/or distance information mandatory for the precise description of the silica structure, which would be a first step for the refined understanding of silica-supported catalysts.



J-couplings provide unique information about the chemical bonds. They are widely used for the structural characterization of molecules in liquids^{14, 15}, since the splitting due this interaction is often directly observed on the 1D spectra. The *J*-coupling splitting can also be directly observed on the spectra of highly mobile solids, such as plastic crystals^{16, 17}. Conversely, in rigid solids, the *J*-coupling splittings are often masked by anisotropic interactions, such as dipolar interactions, chemical shift anisotropy or quadrupolar interaction for nuclear spin $I \ge 1$, as well as the distribution of isotropic chemical shifts due

to atomic-level disorders. Nevertheless, two-dimensional (2D) NMR experiments have been developed to observe and measure *J*-coupling in solids, including:

(i) homonuclear *J*-coupling between spin-1/2 nuclei (${}^{31}P{}^{-31}P^{18}$, ${}^{29}Si{}^{-29}Si{}^{-13}C{}^{-13}C^{20}$ and ${}^{15}N{}^{-15}N^{21}$) or between quadrupolar nuclei (${}^{11}B{}^{-11}B^{22}$ and ${}^{71}Ga{}^{-71}Ga{}^{23}$),

(ii) heteronuclear *J*-couplings between spin-1/2 nuclei (³¹P-²⁹Si²⁴, ³¹P-¹¹³Cd²⁴, ³¹P-⁷⁷Se²⁴, ¹H-¹³C²⁵) or between spin-1/2 and half-integer quadrupolar isotopes (²⁷Al-³¹P²⁶, ²⁷Al-²⁹Si²⁶, ³¹P-⁷¹Ga²⁷, ⁷⁷Se-⁷¹Ga²⁸, ¹³C- ¹⁷O, ¹⁵N-¹⁷O²⁹, ¹H-¹⁷O³⁰, ²⁷Al-¹⁷O³¹).

Nevertheless, to the best of our knowledge, ¹⁷O-²⁹Si *J*- and dipolar couplings have never been measured. One-bond ¹⁷O-²⁹Si ¹*J*-coupling constants have so far only been calculated using density functional theory (DFT) in SiP₂O₇ polymorphs³². The only example of 2D ¹⁷O-²⁹Si correlation spectrum reported in the literature is a 2D ¹⁷O-detected through-space heteronuclear multiple-quantum correlation (*D*-HMQC) spectrum of doubly ¹⁷O and ²⁹Si-enriched germanosilicate zeolites, for which coherence transfer relies on ¹⁷O-²⁹Si dipolar coupling³³. The main limitation to measure ¹⁷O-²⁹Si couplings is the low natural abundance (*NA*) of these isotopes: *NA*(²⁹Si) = 4.68%, and *NA*(¹⁷O) = 0.037%. Furthermore, as the ¹⁷O and ²⁹Si Larmor frequencies only differ by 19%, a specific triple-resonance magicangle spinning (MAS) NMR probe is required, for which the X and Y channels can be tuned to these close resonance frequencies.

We show here how the connectivities and proximities between ¹⁷O and ²⁹Si nuclei can be probed by measuring ¹⁷O-²⁹Si *J*- and dipolar couplings. We also report the first 2D ¹⁷O-²⁹Si through-bond HMQC (*J*-HMQC) spectrum. The experimental data are obtained on a ²⁹Si-enriched silica with surface enriched in ¹⁷O.

Experimental Section

Sample preparation.

The samples were prepared in an argon-filled glove box, and materials were handled under inert atmosphere using standard Schlenk techniques. Samples of ²⁹Si-labelled silica (with ²⁹Si enrichment level of 100%) and ¹⁷O-enriched water (with ¹⁷O enrichment level of 70%) were purchased from Cortecnet. The ²⁹Si-labelled (0.5 g) silica was treated at 1000 °C under high vacuum (10^{-5} Torr) for 15 h, using a quartz tube reactor inserted in a tube furnace while connected to a high vacuum line. ¹⁷O-enriched water (0.1 g, ¹⁷O content 70%) was introduced in the reactor and left to react at room temperature for 8 h³⁴. The material was then submitted to heat treatment under high vacuum for 15 h at the desired temperature (80 or 200°C), using the same experimental setup as for the first annealing step. The corresponding samples are denoted SiO₂₋₂₀₀ hereafter.

NMR experiments

1D Hahn echo and 2D MQ-MAS NMR experiments were recorded at 18.8 T using a double-resonance (HX) 3.2 mm MAS NMR probe, spinning at 20 kHz at the NMR facility of the Advanced Characterization Platform of the Chevreul Institute located in Lille. Other NMR experiments were recorded at 20 T using a triple-resonance (HXY) 3.2 mm MAS NMR probe, spinning at 20 kHz at CEMHTI located in Orléans. The

1D ¹H, ¹⁷O and ²⁹Si NMR spectra of Figs. **1** were acquired using Hahn-echo experiments for ¹H and ¹⁷O isotopes and one-pulse experiment for ²⁹Si nucleus. The radiofrequency (RF) nutation frequencies were equal to 90 and 45 kHz for ¹H and ²⁹Si respectively. We used a central transition (CT)-selective pulse lasting 10 µs and a RF amplitude of 8 kHz for the ¹⁷O NMR experiment. The 1D ¹H, ²⁹Si and ¹⁷O NMR spectra of Figs. 1 result from averaging 32, 368 and 1024 transients, respectively, with recycle delays of 1, 5 and 0.5 s. 2D ¹⁷O triple-quantum (3Q)-MAS NMR spectrum of SiO₂₋₂₀₀ was acquired using the threepulse z-filter pulse sequence³⁵. The 2D 3QMAS spectrum was sheared with the xfshear program included in the TOPSPIN software. We used excitation and reconversion pulses lasting t_p = 3.9 and 1.3 μ s, respectively, with $v_1 = 64$ kHz and a CT selective $\pi/2$ last pulse lasting 15.2 μ s with $v_1 = 5$ kHz. The T_2' time constant for the ¹⁷O transverse losses, which cannot be refocused by a π pulse, was measured using Hahn echo experiment, in which the CT-selective π pulse with $v_1 = 8$ kHz was bracketed by two rotorsynchronized delays. The Hahn echo spectra result from averaging 2560 transients with a recycle delay τ_{RD} = 0.5 s. We recorded ¹⁷O{²⁹Si} J and D-HMQC experiments enhanced by double-frequency sweep (DFS³⁶) and guadrupolar Carr-Purcell Meiboom-Gill (QCPMG)³⁷ as well as 2D double-guantum (2Q)filtered DFS-D-HMQC-QCPMG spectra. For those experiments, we used equal defocusing and refocusing delays. Before the acquisition of J-HMQC data, the magic angle was carefully adjusted by maximizing the intensity of the spinning sidebands in the ²H spectrum of deuterated hexamethylbenzene (HMB) in order to minimize the residual ²⁹Si-¹⁷O dipolar couplings. This leads to an estimated accuracy of the magic angle of 0.01°^{30, 38}. The DFS pulse lasted 2 ms and used $v_1 = 8$ kHz. During the DFS pulse, the frequencies of the RF spikelets were linearly swept in a symmetric manner from 800 to 100 kHz with respect to the CT. The RF field strength of the central transition selective pulses on ¹⁷O channel is 8 kHz, while the RF nutation frequency of the ²⁹Si $\pi/2$ pulse is 48 kHz. A continuous wave ¹H decoupling with $v_1 = 100$ kHz was applied during the whole pulse sequence. For D-HMQC experiment, $SR4_1^2$ recoupling with $v_1 = 40$ kHz was applied on the ²⁹Si channel. During QCPMG acquisition, the delay between successive CT-selective π pulses was 1.4 ms and 50 echoes were acquired in every transient. Note that no enhancement of Si-17OH signal was detected when applying ¹H decoupling during the QCPMG scheme, indicating that the T_2' constant is not limited by the ¹H-¹⁷O dipolar couplings but is governed by the modulation of the ¹⁷O quadrupolar interaction due to librational motion of the silanol groups³⁹. We also acquired 1D $^{1}H\rightarrow^{17}O$ through-space refocused INEPT (D-RINEPT) spectrum of SiO₂₋₂₀₀ at $v_R = 18$ kHz. For this experiment, the RF-field of the pulses, which do not belong to the recoupling scheme, was equal to 100 kHz on ¹H channel and 8 kHz on ¹⁷O channel. The ¹H-¹⁷O dipolar couplings were reintroduced by applying SR4²₁ recoupling with v_1 = 36 kHz on ¹H channel. ¹H and ²⁹Si isotropic chemical shifts were referenced to the resonance of neat tetramethylsilane (TMS) at δ_{iso} = 0 ppm, whereas the ¹⁷O isotropic chemical shifts were referenced to water at δ_{iso} = 0 ppm. Additional experimental details are given in the figure captions.

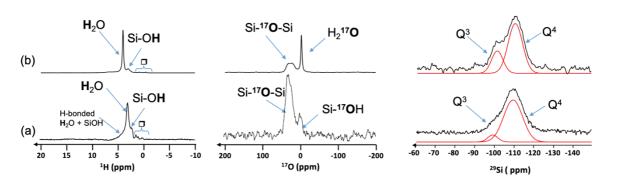
Numerical simulations

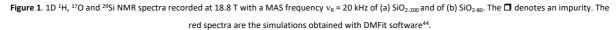
Simulations were performed with SIMPSON software⁴⁰. The powder average was calculated using 1512 different orientations between the molecular and rotor frames: 168 { α_{MR} , β_{MR} } × 9 γ_{MR} angles. The { α_{MR} ,

 β_{MR} Euler angles were selected according to the REPULSION algorithm⁴¹, while the γ_{MR} angle was equally stepped from 0 to 360°. The simulations were carried out for an isolated ²⁹Si-¹⁷O²⁹Si spin system. The ¹⁷O quadrupolar coupling constant and the asymmetry parameter of the electric field gradient were fixed to 5.4 MHz and 0.32, respectively, which correspond the best-fit NMR parameters to simulate ¹⁷O signal of siloxane group in MQMAS spectrum (Fig. S1). A ¹⁷O CSA of 50 ppm^{42, 43}, *J* anisotropy and ²⁹Si-²⁹Si *J*- and dipolar coupling were considered. The two ²⁹Si nuclei were located in the XZ plane of the principal axis system of the ¹⁷O electric field gradient (efg) tensor. As explained below, the Si-O-Si angle and the Si-O distances were varied. We simulated the ¹⁷O{²⁹Si} *D*-HMQC experiments. All pulses were applied on resonance. Other simulation parameters were identical to the reported experimental values.

Results and discussion

Building on our previous experience with selectively surface-enriched silica,³⁴ this sample was synthesized, using commercial ²⁹Si-labelled SiO₂ and ¹⁷O-enriched H₂O (see experimental section). In a first stage, the heating of the silica under vacuum at 1000 °C during 15 h provided a highly hygroscopic material, which was then reacted with ¹⁷O-labelled water, yielding Si-O-Si and Si-OH centres. Further treatment of the resulting material at 80 or 200 °C temperature leads to the samples of interest SiO₂₋₈₀ and SiO₂₋₂₀₀, respectively (Scheme **1**), which are enriched in ²⁹Si in the whole particle volume and in ¹⁷O near or on the surface. These treatment temperatures were chosen **1**) to avoid extensive dehydroxylation, which would deplete ¹⁷O content and 2) to limit the extent of ¹⁷O oxygen migration into the bulk of the material so that only surface sites are enriched with ¹⁷O.³⁴ Furthermore, these intermediate temperatures allow the removal of (most of) physisorbed H₂O, while avoiding a complete condensation of silanols (Si-OH) into siloxane moieties (Si-O-Si).





The 1D ¹H, ¹⁷O and ²⁹Si NMR spectra of SiO₂₋₂₀₀ are displayed in Fig. 1a. The ¹H NMR spectrum exhibits characteristic signals at 3.5 and 2.1 ppm, assigned to physisorbed water and isolated Si-OH groups, respectively as well as a deshielded shoulder ascribed to hydrogen-bonded SiOH^{45, 46}. The ²⁹Si spectrum consists of two overlapping signals of Si(OSi)_n(OH)_{4-n} with n = 3 and 4 (denoted Q³ and Q⁴)⁴⁷. The relative amount obtained with the deconvolution of these sites is 15% for the Q³ and 85% for the Q⁴. The ¹⁷O

spectrum displays a broad signal centered at 28 ppm corresponding to Si-¹⁷O-Si moieties as well as a lower intensity peak at 0 ppm ascribed to Si-¹⁷OH ^{34, 48}. We also recorded 2D ¹⁷O 3Q-MAS spectrum (see Fig. **S1**) where two signals are clearly separated. The signal with $\delta_{iso} = 40$ ppm, $C_Q = 5.4$ MHz and $\eta_Q = 0.32$ is assigned to siloxane Si-¹⁷O-Si environment, while the signal resonating at 25 ppm is attributed to an impurity. However, its relative amount is approximately equal to 5%. The lack of silanol Si-¹⁷O-H environment in the 2D 3Q-MAS spectrum stems most probably from its small C_Q value⁴⁹ and low relative amount compared to the Si-¹⁷O-Si site³⁴. The 1D ¹H, ¹⁷O and ²⁹Si NMR spectra of SiO₂₋₈₀ are displayed in Fig. 1b. The ¹H and ¹⁷O NMR spectra of this sample exhibit a more intense signal of physisorbed water, in agreement with the lower temperature of the thermal treatment, whereas its ²⁹Si NMR spectrum is similar to that of SiO₂₋₂₀₀, with a greater relative amount of Q³ (26% of Q³ and 74% of Q⁴).

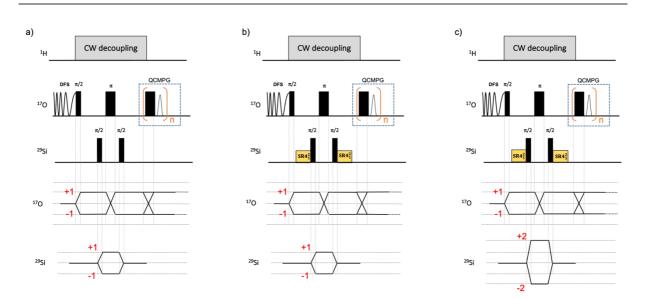


Figure 2. ¹⁷O-(²⁹Si} DFS-HMQC-QCPMG pulse sequences using coherence transfers via ¹⁷O-²⁹Si (a) *J*- or (b,c) dipolar coupling. The coherence transfer pathways for ¹⁷O and ²⁹Si channels are shown at the bottom of the figure. For the *D*-HMQC sequences, we selected via phase cycling either (b) single-quantum (SQ) and (c) 2Q ²⁹Si coherences during the indirect evolution period, t_1 . The pulse programs of pulse sequences (b) and (c) are given in the Appendix 2 of the Supporting Information.

To gain further insights into the connectivities and proximities between ¹⁷O and ²⁹Si isotopes, we recorded 2D ¹⁷O-²⁹Si heteronuclear correlation. ¹⁷O excitation was employed since it is more sensitive than ²⁹Si excitation because ¹⁷O longitudinal relaxation times are much shorter than those of ²⁹Si nuclei $T_1(^{17}O) \ll T_1(^{29}Si)^{50-52}$. Besides, ¹⁷O detection was preferred since in the investigated sample, only the surface sites are enriched in ¹⁷O and a significant fraction of ²⁹Si nuclei are not bonded to ¹⁷O and can produce t_1 -noise in 2D ¹⁷O-²⁹Si heteronuclear correlation spectra using ²⁹Si detection. Therefore, ¹⁷O and ²⁹Si signals were correlated using *J* and *D*-HMQC spectra with ¹⁷O excitation and the indirect detection of ²⁹Si nuclei (¹⁷O{²⁹Si}). Furthermore, the signal was enhanced using the DFS³⁶, which increases the polarization of the CT by manipulating the population of the satellite transitions (ST). In addition, the T_2' time constant is equal to 65 ms for siloxane ¹⁷O nuclei. Hence, their signal was enhanced by using the

QCPMG detection, which allowed the acquisition of 50 echoes in every transient. The corresponding sequence is displayed in Fig. **2**. 2D ¹⁷O{²⁹Si} DFS-*J* and *D*-HMQC-QCPMG spectra of SiO₂₋₂₀₀ are shown in Figs. **3**. The *D*-HMQC-QCPMG spectrum exhibits two cross peaks in the Si-O-Si region attributed to the ¹⁷O nuclei close to ²⁹Si(Q³) and ²⁹Si(Q⁴) environments.⁴⁷ The lack of Si-¹⁷OH signal stems from (1) the small amount of Si-¹⁷OH as compared to the Si-¹⁷O-Si groups and (2) their short T_2' time constant equal to 9 ms, which reduces the efficiency of the coherence transfer and the signal enhancement by QCPMG scheme.

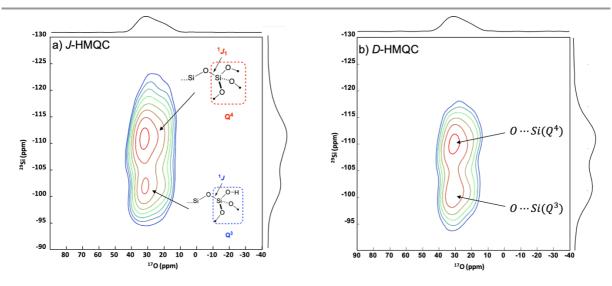


Figure 3. 2D ¹⁷O{²⁹Si} DFS-(a) *J*- and (b) *D*-HMQC-QCPMG NMR spectrum of SiO₂₋₂₀₀ recorded at 20 T with $v_R = 20$ kHz (a) The *J*-HMQC spectrum was recorded with $\tau = 10$ ms, 14848 transients for each of 24 t_2 increments with a recycle delay $\tau_{RD} = 0.5$ s, leading to an experimental time of 50 h (b) The *D*-HMQC spectrum was recorded with $\tau = 3$ ms ms (recoupling time), 5120 transients for each of 22 t_1 increments with a recycle delay $\tau_{RD} = 0.5$ s, leading to an experimental time of 16 h The projections along ²⁹Si and ¹⁷O dimensions are also shown.

The ¹⁷O{²⁹Si} J-HMQC-QCPMG spectrum shown in Fig. **3a** exhibits the same cross peaks as the *D*-HMQC spectrum (Fig. **3b**). These cross-peaks represent the first experimental observation of ¹⁷O-²⁹Si *J*-couplings. For the sake of comparison, ¹⁷O-{²⁹Si} *J*- and *D*-HMQC spectra of SiO₂₋₈₀ were also recorded (see Fig. **S3**) and also display O-Si(Q³) and O-Si(Q⁴) cross-peaks, and no cross-peak for Si-OH. The lack of cross-peak at $\delta(^{17}O) = 0$ ppm is consistent with its assignment to water (Fig. **1b**).

In order to measure ¹⁷O-²⁹Si *J*-couplings, we recorded 1D ¹⁷O{²⁹Si} DFS-*J*-HMQC-QCPMG spectra with various defocusing and refocusing delays, τ (see Fig. **S2**). During these delays, the single-quantum coherences associated to the ¹⁷O CT evolve under the ¹⁷O-²⁹Si *J*-couplings. However, these delays are not affected by the dipolar-quadrupolar cross-term since the Hamiltonian of this term is proportional to $[3I_z^2 - I(I + 1)\mathbf{1}_I]S_z$, ^{53 54} which commutes with the operators associated to the ¹⁷O CT. In the expression of the above product of spin operators, I = 5/2 is the spin value of ¹⁷O, I_z and S_z denote the operators associated to the z-component of the spin angular momentum of ¹⁷O and ²⁹Si nuclei, respectively, and $\mathbf{1}_I$ is the identity matrix of size 2*I*+1. Furthermore, the magic angle was adjusted with an accuracy better than 0.05°, which results in residual ¹⁷O-²⁹Si dipolar couplings smaller than ±1 Hz.

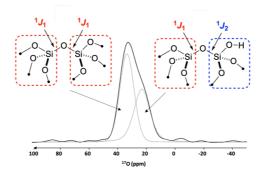


Fig. 4. 1D ¹⁷O-{²⁹Si} DFS-J-HMQC-QCPMG NMR spectra of SiO₂₋₂₀₀ recorded with τ = 9 ms. Experimental parameters are given in the caption of the Figure 5.

As seen in Figs. **S2** and **4**, 1D ¹⁷O{²⁹Si} DFS-J-HMQC-QCPMG spectra exhibit a peak at 34 ppm as well as a shoulder at 22 ppm. These two signals are barely resolved in the 3Q-MAS spectrum of Fig. **S1**. However, the build-up curves at these two shifts differ, as seen in Fig. **5b** and **c**. The maximal intensity is reached at $\tau = 7$ ms for the deshielded site, instead of 9 ms for the shielded signal. These distinct buildup curves suggest the presence of at least two distinct ¹⁷O sites subject to different ¹⁷O-²⁹Si *J*-couplings. Furthermore, the intensity of the deshielded signal is always higher than that of the shielded one (Fig. **S2**). Based on this difference in intensity, the deshielded and shielded sites were assigned to symmetrical (Si(Q⁴)-O-Si(Q⁴)) and asymmetrical (Si(Q³)-O-Si(Q⁴)) siloxane environments, respectively (see Fig. **5a**), since the amount of Q³ sites near the sulface of SiO₂₋₂₀₀, the contribution of Si(Q³)-O-Si(Q³) sites to the *J*-HMQC build-up curves has been neglected. For the symmetrical environment (scheme **A** in Fig. **5a**), onebond ¹⁷O-²⁹Si *J*-couplings were assumed to be equal and the transfer efficiency of ¹⁷O-{²⁹Si} *J*-HMQC experiment is proportional to

$$2A[\sin^2(\pi J_1 \tau) \cos^2(\pi J_1 \tau)] \left[\exp\left(-\frac{2\tau}{T_2 \prime}\right) \right]$$
(eq. 1).

where J_1 denotes the ¹*J*-coupling constant of the symmetrical ¹⁷O site bonded to two ²⁹Si(Q⁴) nuclei and *A* is a pre-factor related to the signal amplitude. For the asymmetrical siloxane (scheme **B** in Fig. **5a**), the transfer efficiency of ¹⁷O{²⁹Si} *J*-HMQC experiment is proportional to

$$A[\sin^{2}(\pi J_{1}\tau)\cos^{2}(\pi J_{2}\tau) + \sin^{2}(\pi J_{2}\tau)\cos^{2}(\pi J_{1}\tau)]\left[\exp\left(-\frac{2\tau}{T_{2}\tau}\right)\right]$$
(eq. 2).

where J_1 and J_2 denote the ¹*J*-coupling constants of the asymmetrical ¹⁷O site with ²⁹Si(Q⁴) and ²⁹Si(Q³) nuclei, respectively. In order to constrain the simulation of the build-up curves for the asymmetrical ¹⁷O site, the ¹*J*-coupling constants between ¹⁷O site and ²⁹Si(Q⁴) nuclei were assumed to be equal in both symmetrical and asymmetrical sites. Furthermore, for the build-up curves simulation of both symmetrical and asymmetrical siloxane ¹⁷O sites, we used $T_2' = 65$ ms, measured using Hahn-echo experiment.

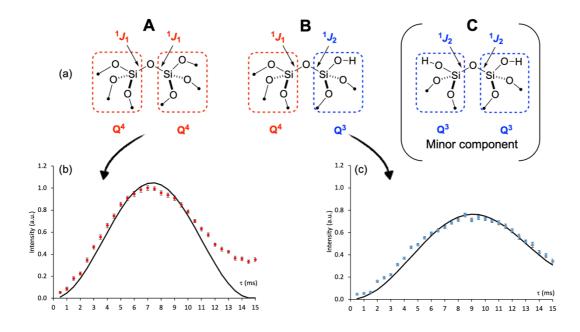


Figure 5 (a) Structure of local siloxane environments. (b,c) Build-up curves of ${}^{17}O{}^{29}Si}$ DFS-*J*-HMQC-QCPMG signals at (b) 34 and (c) 22 ppm assigned to siloxane sites. The red points and the blue squares correspond to the experimental data, whereas the black lines are the simulated curves. The build-up curves were obtained by averaging 10624 transients for each of 30 t_1 increments with a recycle delay $\tau_{RD} = 0.5$ s, leading to an experimental time of 45 h. The red and the blue points correspond to the experimental data, whereas the black lines are the simulated curves. The errors bars in the experimental curves subsume the contributions of the noise standard deviation and the error due to the nonlinear fit with DMFIT⁴⁶.

The experimental *J*-HMQC build-up curve shown in Fig. **5b** was fitted to Eq. **1**. The best-fit was obtained for $J_1={}^{1}J_{(O-Si(Q4))} = 34$ Hz. Using this J_1 value, the experimental *J*-HMQC build-up curve in Fig. **5c** was fitted to Eq. **2** and the best-fit $J_2={}^{1}J_{(O-Si(Q3))}$ parameter was 23 Hz, which is lower than ${}^{1}J_{(O-Si(Q4))} = 34$ Hz (Table 1). The significant difference between ${}^{1}J$ -coupling constants with Q³ and Q⁴ sites shows the influence of the second neighbours of 29 Si nuclei on the strength of the ${}^{1}J$ -coupling with siloxane 17 O nuclei. This effect might result from difference between Si(Q⁴)-O-Si(Q⁴) and Si(Q³)-O-Si(Q⁴) angles. The deviation between experimental data points and the simulated curves in Fig. 5b for long τ delay probably stems from a distribution of the *J*-coupling constant owing to the amorphous nature of the investigated silica. Note that the best-fit 17 O- 29 Si ${}^{1}J$ -coupling constants are comparable to the 13 C- 17 O ${}^{1}J$ -coupling constants measured using heteronuclear spin-echo 29 or a *J*-resolved experiment 55 , which ranges from 21 to 37 Hz.

Table 1. Best-fit ¹⁷O-²⁹Si ¹*J*-coupling constants in Hz and *A* pre-factor determined from the fit of experimental build-curves shown in Fig. 5b and c to Eqs. 1 and 2, respectively.

	Symmetrical O	Asymmetrical O
¹ <i>J</i> _{(O-Si(Q4))}	J_1 = 34 ± 2 Hz	$J_1 = 34 \pm 2$ Hz
¹ J _{(O-Si(Q3))}		$J_2 = 23 \pm 2$ Hz
A	2.1 ± 0.2	1.4 ± 0.2

Similarly, we recorded the build-up curves of 1D ¹⁷O-{²⁹Si} DFS-D-HMQC-QCPMG experiment. The corresponding build-up curves for symmetrical and asymmetrical siloxane sites resonating at 34 and 22 ppm are displayed in Fig. S5. In order to analyse these build-up curves, we simulated using SIMPSON software⁴⁰ the build-up curve of ¹⁷O-{²⁹Si} *D*-HMQC-QCPMG scheme for {²⁹Si-¹⁷O-²⁹Si} spin systems. The two Si⁻⁻O distances were assumed to be equal for the symmetrical ¹⁷O site but to differ for the asymmetrical ¹⁷O site. For the symmetrical ¹⁷O site, the Si^mO distances were varied from 1.562 to 1.628 Å (corresponding to a dipolar coupling constant of -750 and -850 Hz respectively), whereas Si-0-Si was varied from 130° to 170° (Fig. S6 and S7). The best agreement between the simulated and experimental build-up curves was obtained for $d_{0...Si(04)} = 1.628$ Å and 140° < Si-0 –Si < 150°. Such value is in line with published data⁵⁶. For the asymmetrical siloxane environment, one distance was fixed to $d_{0...Si(Q4)} = 1.628$ Å. The best agreement was found for an angle of Si -0-Si = 144° and a distance $d_{0...Si(Q3)} = 1.562 \text{ Å}$ which is significantly shorter than the $d_{0...Si(Q4)}$ distance, measured here or reported in the literature^{56, 57}. The small deviations between experimental and simulated build-up curves might stem from (i) a slight distribution of ¹⁷O-²⁹Si dipolar constant owing to the amorphous nature of the investigated silica, but also (ii) the dipolar coupling of ¹⁷O nucleus with nearby non-covalently bonded ²⁹Si nuclei which is approximately equal to 75 Hz or the effect of ²⁹Si-²⁹Si *J*- or dipolar coupling constants, which are approximately equal to 12 and 150 Hz ^{58, 59}.

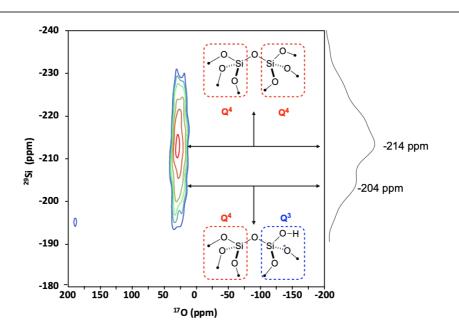


Figure 6. 2D ¹⁷O{²⁹Si} 2Q-filtered DFS-*D*-HMQC-QCPMG NMR spectrum of SiO₂₋₂₀₀ recorded at 20 T with $v_R = 20$ kHz. The projection along ²⁹Si 2Q dimension is also shown. The spectrum results from averaging 8192 and the recycle delay is 0.5 s. The recoupling time is 2 ms leading to an experimental time of 25 h.

We also carried out 2D ¹⁷O-{²⁹Si} DFS-*D*-HMQC-QCPMG experiment of SiO₂₋₂₀₀, for which the doublequantum (2Q) coherences of ²⁹Si nuclei were selected during the indirect evolution period, t_1 , using phase cycling (see Fig. **2c**). Such experiment allows selecting the signal of ¹⁷O nuclei near two ²⁹Si nuclei⁶⁰. As seen in Fig. **6**, ¹⁷O signal correlate with 2Q ²⁹Si shifts of coherences at -214 and -204 ppm, ascribed to ${}^{29}Si(Q^4)-{}^{17}O-{}^{29}Si(Q^4)$ and ${}^{29}Si(Q^4)-{}^{17}O-{}^{29}Si(Q^3)$ moieties, respectively. This result confirms the presence of these two environments near the silica surface as well as the negligible amount of ${}^{29}Si(Q^3)-{}^{17}O-{}^{29}Si(Q^3)$ sites, which would resonate near –194 ppm.

Conclusion

We have probed connectivities and proximities between ²⁹Si and ¹⁷O nuclei in silica sample enriched in ²⁹Si and ¹⁷O isotopes near the surface using ¹⁷O{²⁹Si} *J*- and *D*-HMQC experiments at high-magnetic field. These experiments have allowed the observation of two distinct siloxane oxygen atoms, those linking two Q⁴ sites and those linking Q³ and Q⁴ sites. However, the signal of silanol groups has not been detected because of their short *T*'₂ time constant. Furthermore, we have measured for the first time ¹⁷O-²⁹Si ¹*J*-coupling constants for siloxane groups. We have found that the ¹⁷O-²⁹Si ¹*J*-coupling constants are smaller for Q⁴ sites than Q³ ones and hence, that the second neighbour of ²⁹Si nuclei, H or Si atoms, affects the amplitude of ¹⁷O-²⁹Si ¹*J*-coupling for siloxane ¹⁷O nuclei. We have also shown that the simulation of the build-up of ¹⁷O{²⁹Si *D*-HMQC signals allows estimating ²⁹Si-¹⁷O bond length and Si-O -Si angle. The access to these elusive coupling constants opens new prospects for the characterization of silica surfaces, silica-supported catalysts and silicate-containing materials.

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Conflicts of interest

There are no conflicts to declare.

Electronic Supplementary Information

¹⁷O-{²⁹Si} DFS-HMQC-CPMG *J* and *D*-mediated NMR spectra of SiO₂₋₈₀, 1D experiments on SiO₂₋₂₀₀, SIMPSON input program, details of the pulse sequences.

Notes and references

- 1. D. Grekov, T. Vancompernolle, M. Taoufik, L. Delevoye and R. M. Gauvin, *Chem. Soc. Rev.*, 2018, **47**, 2572-2590.
- 2. H. Nagashima, C. Martineau-Corcos, G. Tricot, J. Trebosc, F. Pourpoint, J. P. Amoureux and O. Lafon, *Ann. Rep. NMR Spectrosc., Vol 94*, 2018, **94**, 113-185.
- 3. C. Coperet, W. C. Liao, C. P. Gordon and T. C. Ong, J. Am. Chem. Soc., 2017, 139, 10588-10596.
- 4. S. E. Ashbrook, J. M. Griffin and K. E. Johnston, Annu. Rev. Anal. Chem., Vol 11, 2018, 11, 485-508.
- 5. C. Bonhomme, C. Gervais and D. Laurencin, *Progr. Nucl. Magn. Reson. Spectrosc.*, 2014, **77**, 1-48.
- 6. D. Grekov, Y. Bouhoute, I. Del Rosal, L. Maron, M. Taoufik, R. M. Gauvin and L. Delevoye, *Phys. Chem. Chem. Phys.*, 2016, **18**, 28157-28163.
- 7. J. Pelletier, J. Espinas, N. Vu, S. Norsic, A. Baudouin, L. Delevoye, J. Trébosc, E. Le Roux, C. Santini, J. M. Basset, R. M. Gauvin and M. Taoufik, *Chem. Commun.*, 2011, **47**, 2979-2981.
- A. Jystad and M. Caricato, *Chem. Mater.*, 2018, **30**, 7813-7822.
- 9. F. A. Perras, K. C. Boteju, Slowing, II, A. D. Sadow and M. Pruski, *Chem. Commun.*, 2018, **54**, 3472-3475.
- 10. F. Mauri, A. Pasquarello, B. G. Pfrommer, Y. G. Yoon and S. G. Louie, *Phys. Rev. B*, 2000, **62**, R4786-R4789.
- 11. T. Charpentier, P. Kroll and F. Mauri, J. Phys. Chem. C, 2009, **113**, 7917-7929.
- 12. N. M. Trease, T. M. Clark, P. J. Grandinetti, J. F. Stebbins and S. Sen, J. Chem. Phys., 2017, 146.
- 13. D. J. Srivastava, J. H. Baltisberger, I. Florian, F. Fayon, R. A. Shakhovoy, M. Deschamps, N. Sadiki and P. J. Grandinetti, *Phys. Rev. B*, 2018, **98**.
- 14. J. Keeler, Understanding NMR Spectroscopy, 2nd Edition, Wiley, 2010.
- 15. S. Berger, S. Braun and H. O. Kalinowski, NMR Spectroscopy of the Non-Metallic Elements, Wiley, 1997.
- 16. T. Terao, H. Miura and A. Saika, J. Chem. Phys., 1981, **75**, 1573-1574.
- 17. K. W. Zilm and D. M. Grant, J. Magn. Reson., 1982, 48, 524-526.
- 18. T. Allman, J. Magn. Reson., 1989, **83**, 637-642.
- 19. C. A. Fyfe, H. Gies and Y. Feng, J. Chem. Soc.-Chem. Commun., 1989, 1240-1242.
- 20. A. Lesage, C. Auger, S. Caldarelli and L. Emsley, J. Am. Chem. Soc., 1997, 119, 7867-7868.
- 21. S. P. Brown, M. Perez-Torralba, D. Sanz, R. M. Claramunt and L. Emsley, *J. Am. Chem. Soc.*, 2002, **124**, 1152-1153.
- 22. F. A. Perras and D. L. Bryce, J. Am. Chem. Soc., 2013, 135, 12596-12599.
- 23. F. A. Perras and D. L. Bryce, J. Phys. Chem. Lett., 2014, 5, 4049-4054.
- 24. D. Franke, C. Hudalla and H. Eckert, *Solid State Nucl Magn Reson*, 1992, **1**, 33-40.
- 25. A. Lesage, D. Sakellariou, S. Steuernagel and L. Emsley, J. Am. Chem. Soc., 1998, **120**, 13194-13201.
- 26. C. A. Fyfe, K. C. Wongmoon, Y. Huang and H. Grondey, J. Am. Chem. Soc., 1995, **117**, 10397-10398.
- 27. M. Deschamps, F. Fayon, V. Montouillout and D. Massiot, *Chem. Commun.*, 2006, 1924-1925.
- 28. H. Nagashima, J. Trebosc, L. Calvez, F. Pourpoint, F. Mear, O. Lafon and J. P. Amoureux, *J. Magn. Reson.*, 2017, **282**, 71-82.
- 29. I. Hung, A. C. Uldry, J. Becker-Baldus, A. L. Webber, A. Wong, M. E. Smith, S. A. Joyce, J. R. Yates, C. J. Pickard, R. Dupree and S. P. Brown, *J. Am. Chem. Soc.*, 2009, **131**, 1820-1834.
- 30. S. L. Carnahan, B. J. Lampkin, P. Naik, M. P. Hanrahan, Slowing, II, B. VanVeller, G. Wu and A. J. Rossini, *J. Am. Chem. Soc.*, 2019, **141**, 441-450.
- 31. D. Iuga, C. Morais, Z. H. Gan, D. R. Neuville, L. Cormier and D. Massiot, *J. Am. Chem. Soc.*, 2005, **127**, 11540-11541.
- C. Bonhomme, C. Gervais, C. Coelho, F. Pourpoint, T. Azais, L. Bonhomme-Coury, F. Babonneau, G. Jacob, M. Ferrari, D. Canet, J. R. Yates, C. J. Pickard, S. A. Joyce, F. Mauri and D. Massiot, *Magn. Reson. Chem.*, 2010, 48, S86-S102.
- 33. G. P. M. Bignami, D. M. Dawson, V. R. Seymour, P. S. Wheatley, R. E. Morris and S. E. Ashbrook, *J. Am. Chem. Soc.*, 2017, **139**, 5140-5148.
- 34. N. Merle, J. Trébosc, A. Baudouin, I. Del Rosal, L. Maron, K. Szeto, M. Genelot, A. Mortreux, M. Taoufik, L. Delevoye and R. M. Gauvin, *J. Am. Chem. Soc.*, 2012, **134**, 9263-9275.
- 35. J. P. Amoureux, C. Fernandez and S. Steuernagel, *J. Magn. Reson.*, 1996, **123**, 116-118.
- 36. D. luga and A. P. M. Kentgens, J. Magn. Reson., 2002, **158**, 65-72.
- 37. F. H. Larsen, H. J. Jakobsen, P. D. Ellis and N. C. Nielsen, J. Phys. Chem. A, 1997, 101, 8597-8606.
- 38. S. Antonijevic and G. Bodenhausen, *Angew. Chem.-Int. Ed*, 2005, **44**, 2935-2938.
- 39. T. Kobayashi, J. A. DiVerdi and G. E. Maciel, J. Phys. Chem. C, 2008, **112**, 4315-4326.
- 40. M. Bak, J. T. Rasmussen and N. C. Nielsen, J. Magn. Reson., 2000, 147, 296-330.
- 41. M. Bak and N. C. Nielsen, J. Magn. Reson., 1997, **125**, 132-139.
- 42. T. Charpentier, S. Ispas, M. Profeta, F. Mauri and C. J. Pickard, *Journal of Physical Chemistry B*, 2004, **108**, 4147-4161.
- 43. F. H. Larsen and I. Farnan, *Chem. Phys. Lett.*, 2002, **357**, 403-408.
- 44. D. Massiot, F. Fayon, M. Capron, I. King, S. Le Calve, B. Alonso, J. O. Durand, B. Bujoli, Z. H. Gan and G. Hoatson, *Magn. Reson. Chem.*, 2002, **40**, 70-76.
- 45. H. N. Kim and S. K. Lee, *Geochim. Cosmochim. Acta*, 2013, **120**, 39-64.

- 46. C. E. Bronnimann, R. C. Zeigler and G. E. Maciel, J. Am. Chem. Soc., 1988, **110**, 2023-2026.
- 47. E. Lippmaa, M. Magi, A. Samoson, G. Engelhardt and A. R. Grimmer, *J. Am. Chem. Soc.*, 1980, **102**, 4889-4893.
- 48. T. H. Walter, G. L. Turner and E. Oldfield, J. Magn. Reson., 1988, 76, 106-120.
- 49. C. Fernandez, J. P. Amoureux, J. M. Chezeau, L. Delmotte and H. Kessler, *Microporous Mater.*, 1996, **6**, 331-340.
- 50. J. Trébosc, B. Hu, J. P. Amoureux and Z. Gan, J. Magn. Reson., 2007, 186, 220-227.
- 51. G. Tricot, O. Lafon, J. Trebosc, L. Delevoye, F. Mear, L. Montagne and J. P. Amoureux, *Phys. Chem. Chem. Phys.*, 2011, **13**, 16786-16794.
- 52. G. Tricot, J. Trébosc, F. Pourpoint, R. Gauvin and L. Delevoye, in *Ann. Rep. Nucl. Magn. Reson. Spectrosc., Vol 81*, ed. G. A. Webb, 2014, vol. 81, pp. 145-184.
- 53. S. Wi and L. Frydman, J. Chem. Phys., 2000, **112**, 3248-3261.
- 54. F. A. Perras and D. L. Bryce, J. Magn. Reson., 2014, 242, 23-32.
- 55. G. J. Rees, S. P. Day, K. E. Barnsley, D. Iuga, J. R. Yates, J. D. Wallis and J. V. Hanna, *Phys. Chem. Chem. Phys.*, 2020, **22**, 3400-3413.
- 56. T. M. Clark and P. J. Grandinetti, J. Phys.: Condens. Matter, 2003, 15, S2387-S2395.
- 57. T. M. Clark and P. J. Grandinetti, J. Non-Cryst. Solids, 2000, 265, 75-82.
- 58. R. Challoner and C. A. McDowell, J. Magn. Reson., 1992, **98**, 123-133.
- 59. D. J. Srivastava, P. Florian, J. H. Baltisberger and P. J. Grandinetti, *Phys. Chem. Chem. Phys.*, 2018, **20**, 562-571.
- 60. C. Martineau, F. Fayon, C. Legein, J. Y. Buzare, G. Silly and D. Massiot, *Chem. Commun.*, 2007, 2720-2722.