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## 1 Two temperate corals are tolerant to low pH regardless of

## 2 previous exposure to natural CO<sub>2</sub> vents

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#### 24 Abstract

Ocean acidification is perceived to be a major threat for many calcifying organisms, including 25 scleractinian corals. Here we investigate (1) whether past exposure to low pH environments 26 27 associated with CO<sub>2</sub> vents could increase corals tolerance to low pH and (2) whether zooxanthellate corals are more tolerant to low pH than azooxanthellate corals. To test these 28 hypotheses, two Mediterranean colonial corals Cladocora caespitosa (zooxanthellate) and 29 Astroides calvcularis (azooxanthellate) were collected from CO<sub>2</sub> vents and reference sites and 30 incubated in the laboratory under present-day (pH on the total scale, pH<sub>T</sub> 8.07) and low pH 31 32 conditions ( $pH_T$  7.70). Rates of net calcification, dark respiration and photosynthesis were monitored during a six-month experiment. Monthly net calcification was assessed every 27 to 33 34 35 d using the buoyant weight technique, whereas light and dark net calcification was 35 estimated using the alkalinity anomaly technique during 1 h incubations. Neither species showed any change in net calcification rates, respiration, and photosynthesis regardless of 36 37 their environmental history, pH treatment and trophic strategy. Our results indicate that C. 38 caespitosa and A. calycularis could tolerate future ocean acidification conditions for at least 6 months. These results will aid in predicting species' future responses to ocean acidification, 39 and thus improve the management and conservation of Mediterranean corals. 40

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#### 48 Introduction

Ocean acidification describes the shift in carbonate chemistry caused by the uptake of 49 anthropogenic CO<sub>2</sub> by the ocean (Caldeira and Wickett 2003, Gruber et al. 2019). It causes an 50 51 increase in dissolved inorganic carbon and bicarbonate ion concentrations, along with a concurrent decrease in pH, carbonate ion concentration and its associated saturation state ( $\Omega$ ; 52 Orr et al. 2005). Calcifying species are perceived to be particularly vulnerable to ocean 53 54 acidification (Kroeker et al. 2013a). For example, corals are particularly sensitive to changes in carbonate chemistry as their skeleton is made of aragonite, a metastable form of calcium 55 56 carbonate that is less stable than calcite (Erez et al. 2011, Foster and Clode 2016). As a result, coral calcification is expected to decrease as seawater acidity increases (Gattuso et al. 1999). 57 Coral calcification decreases by on average 22% under pH values expected by the end of the 58 59 century under the high-CO<sub>2</sub> emissions scenario (RCP 8.5 from AR5 of ICPP; Chan and Connolly 2013). However, a range of contradictory species- and experience-specific 60 responses have been described (e.g. Kornder et al. 2018), with some species such as massive 61 62 Porites spp. being tolerant to ocean acidification (Fabricius et al. 2011). This resistance to ocean acidification was also described in several temperate and cold-water corals (e.g. 63 Rodolfo-Metalpa et al. 2010). Such tolerance is seen in many studied parameters such as net 64 calcification rate, respiration, photosynthetic physiology, calcifying fluid pH (McCulloch et 65 66 al. 2012a, Wang et al. 2020). According to Varnerin et al. (2020) even juveniles from the 67 temperate Oculina arbuscula were only slightly affected by ocean acidification.

A central question when assessing the future of corals in a high-CO<sub>2</sub> world is whether they
have the capacity to acclimatize and further adapt to ocean acidification. This question
remains unanswered as corals are generally long-lived organisms for which it is extremely
challenging to study acclimatization and adaptation processes over several generations.
Acclimatization refers to reversible phenotypic changes limited by genotype, occurring under

naturally varying conditions by a physiological readjustment of the organism's tolerance 73 74 levels (Edmunds and Gates 2008). In contrast, adaptation refers to changes in the genetic composition of a population by natural selection, in which traits will be passed on to the next 75 generation (Savolainen et al. 2013). In a recent study, Cornwall et al. (2020) conducted a one-76 year experiment showing that coralline algae have the capacity to acclimatize to low pH after 77 seven generations of exposure. Replicating such a study on corals is virtually impossible as 78 they are long-lived organisms which only reach sexual maturity after several years. CO<sub>2</sub> vents 79 therefore offer a unique opportunity to investigate the response of corals to ocean acidification 80 over long periods of time. In such systems, volcanic CO<sub>2</sub> bubbles from the seafloor which 81 82 acidifies the surrounding seawater and creates an environment, with carbonate chemistry 83 conditions that mimic those expected in the future (Hall-Spencer et al., 2008, Camp et al. 2018, González-Delgado and Hernández 2018). By investigating individuals, species, and 84 85 communities along transects radiating from the sources of CO<sub>2</sub>, it is possible to substitute time for space. It must be pointed out, however, that CO<sub>2</sub> vents are imperfect windows into the 86 future because they generally do not mimic other climate-related changes such as ocean 87 warming. Furthermore, it is not possible to disentangle any changes in the mean versus 88 variability of seawater pH (Hall-Spencer et al. 2008). Overall, these systems exhibit 89 90 significant decreases in trophic complexity and biodiversity, as well as a major decrease in the abundance of calcifying organisms (Kroeker et al. 2013b, Linares et al. 2015, Teixidó et al. 91 2018). Corals living in the vicinity of CO<sub>2</sub> vents have been exposed to low and high 92 93 variability pH conditions for an extended period of time (years to decades). They are therefore ideal models to investigate acclimatization and adaptation to ocean acidification (Fantazzini et 94 al. 2015, Foo et al. 2018). Populations or individuals living in these variable and more 95 extreme environments are hypothesized to be more tolerant to suboptimal environmental 96 conditions such as acidification (e.g. Cornwall et al. 2018). 97

CO<sub>2</sub> vents have been used for transplant experiments between acidified and nearby reference 98 99 sites with ambient pH and no vent activity. For example, a decline in calcification was described in the zooxanthellate corals Cladocora caespitosa and Balanophyllia europaea 100 transplanted to a CO<sub>2</sub> vent site in Ischia (Rodolfo-Metalpa et al. 2011). The skeletal growth of 101 102 *Porites astreoides* living in a natural pH gradient showed no acclimatization to ocean acidification despite their life-long exposure to low pH (Crook et al. 2013). Few laboratory 103 104 studies have focused on temperate scleractinian corals from natural CO<sub>2</sub> vents despite the relevance of studying the tolerance of temperate coral populations to ocean acidification 105 (Trotter et al. 2011, Teixidó et al. 2020). Such studies are important for understanding and 106 107 elucidating how these species will respond to the suboptimal pH conditions projected for the 108 end of this century.

109 The symbiosis between scleractinian corals and endosymbiotic algae, commonly referred to as zooxanthellae, has been studied thoroughly (Davy et al. 2012). Photosynthetic products 110 from zooxanthellae contribute significantly to the energy budget of the scleractinian host 111 112 (Allemand et al. 2011, Muller-Parker et al. 2015) and can play an important role in calcification (Erez et al. 2011, Inoue et al. 2018). However, not all corals harbor 113 zooxanthellae (azooxanthellate corals). Gibbin et al. (2014) demonstrated that zooxanthellae 114 regulate intracellular pH under acidified conditions, whereas azooxanthellate corals can suffer 115 from reduced intracellular pH which can lead to intracellular acidosis. It can thus be expected 116 117 that photosynthesis may enhance calcification by (1) providing energy for this energetically costly process and (2) by increasing pH at the site of calcification (Allemand et al. 2011). In a 118 transplant experiment in the CO<sub>2</sub> vents of Panarea, Prada et al. (2017) showed that the 119 120 calcification of two azooxanthellate corals Leptopsammia pruvoti and Astroides calycularis decreased at low pH, whereas calcification of the zooxanthellate coral B. europaea was 121 unaffected. A similar result was observed in pH-controlled aquaria by Ohki et al. (2013) as 122

polyps of Acropora digitifera with symbionts showed higher calcification rates than polyps 123 124 without symbionts under low pH, suggesting that symbionts confer tolerance to ocean acidification. However, in a study led by McCulloch et al. (2012b), azooxanthellate cold-125 water corals presented higher intracellular pH regulation than shallow-water zooxanthellate 126 127 corals under acidification. As it is not truly known whether or not symbionts promote the maintenance of elevated pH at the site of calcification, it is interesting to compare 128 129 zooxanthellate and azooxanthellate corals as they may display contrasting responses to ocean acidification. 130

131 The present study focuses on two Mediterranean long-lived corals of key relevance for conservation, the zooxanthellate species C. caespitosa (Linnaeus, 1767) and the 132 azooxanthellate species A. calycularis (Pallas, 1766). Cladocora caespitosa is a unique 133 134 zooxanthellate reef-building coral that provides a valuable ecosystem service in the Mediterranean Sea which is lacking large bioconstructions (Kersting and Linares 2012). 135 Astroides calvcularis is commonly found in the southwestern Mediterranean Sea and can be 136 137 highly abundant in low light, shallow rocky habitats (Zibrowius 1995). These two scleractinian corals naturally occur at two newly discovered CO<sub>2</sub> vent systems along the coast 138 of Ischia, Italy. In situ data obtained by Teixidó et al. (2020) showed variable responses in 139 skeletal structure and growth patterns of A. calycularis, with colonies from the CO<sub>2</sub> vent site 140 characterized by smaller size, fewer polyps, and less porous, denser skeletons. Thus, these 141 142 two scleractinian coral species grown under elevated pCO<sub>2</sub> environments are great model systems for assessing tolerance to acidification. Here, two hypotheses are tested: (1) past 143 exposure to low pH conditions confers tolerance to ocean acidification and (2) the coral-algal 144 145 symbiosis modulates this tolerance. To test these hypotheses, corals from natural CO<sub>2</sub> vent systems and reference sites were exposed to present-day (pH on the total scale,  $pH_T \sim 8.07$ ) 146

and low pH conditions (pH<sub>T</sub>  $\sim$ 7.70) during a six-month long laboratory experiment, in which calcification, respiration and photosynthesis rates were assessed.

#### 149 Material and Methods

150 <u>Study sites</u>

We studied populations of the corals Cladocora caespitosa and Astroides calycularis that 151 naturally occur at two distinct CO<sub>2</sub> vent sites with low pH, and two reference sites with 152 ambient pH and no vent activity at the island of Ischia, Italy (Fig. S1 Supplementary 153 materials). Cladocora caespitosa occurs near a CO<sub>2</sub> vent site adjacent to rocky reefs at about 154 10 m depth (Chiane del Lume, hereafter Vent 1), whereas A. calycularis occurs in another 155 CO<sub>2</sub> vent site between 1 - 2 m depth (Grotta del Mago, a semi-submerged cave hereafter Vent 156 157 2, where CO<sub>2</sub> vents occur at 5 m depth). Vent gas compositions at Vent 1 and Vent 2 were predominantly CO<sub>2</sub> (mean  $\pm$  SE: 93.9  $\pm$  0.6 for Vent 1, 93.4  $\pm$  1.0 for Vent 2), with 158 undetectable levels of hydrogen sulfide (< 0.0002 %), and did not elevate the temperature (for 159 more information on gas measurements, see Teixidó et al. 2020). The reference sites without 160 any visible vent activity and exhibiting ambient pH were located: i) approximately 100 m 161 west from Vent 1 for C.caespitosa (hereafter Ambient 1, 10 m depth) and ii) 1.5 km away 162 from Vent 2 for A. calycularis (San Pancrazio, an overhang, hereafter Ambient 2, 1-2 m 163 164 depth) (Fig. S1 Supplementary materials).

#### 165 <u>Coral field surveys</u>

166 The *C. caespitosa* abundance was quantified using 12 quadrats of 50 cm x 50 cm, which were

167 randomly placed along the rocky reef at 10 m depth at the two sites (n = 24 quadrats in total).

168 Abundance was assessed by counting the number of colonies found in each quadrat. In

addition, size of colonies was determined by measuring the major colony diameter with a

plastic ruler (with an error of  $\pm 1$  cm) (Peirano et al. 2001). Major colony diameter was 170 171 selected as the best colony size descriptor (Kersting 2012). The A. calycularis abundance was quantified using 12 photo-quadrats (25 x 25 cm) between 1 and 2 m depth in the two other 172 sites (n = 24 quadrats in total, Vent 2, Ambient 2). Size frequency-distribution was calculated 173 by counting the number of polyps of each colony and each colony was then pooled into one of 174 five size classes (I: 1-5 polyps: II: 6-10 polyps; III: 11-15 polyps; IV: 16-20 polyps; V: > 20175 176 polyps). The survey's depth for *C. caespitosa* and *A. calycularis* corresponded to the depth range at which colonies were sampled for the experiment (see below). Abundance and size-177 frequency data for A. calycularis have been described previously in Teixidó et al. (2020). 178

### 179 In situ $pH_T$ and carbonate chemistry associated with $CO_2$ vents and ambient pH sites

SeaFET<sup>TM</sup> Ocean pH sensors (Satlantic) were deployed to quantify variation in pH at the two 180 CO<sub>2</sub> vent and the two ambient pH sites. Measurements were taken every 15 minutes. One sensor 181 was deployed at the Vent 1 site from May 7 to 21, 2019, and one sensor was deployed at the 182 183 Ambient 1 site during the same period. At the semi-submerged cave (Vent 2), sensors were deployed in May-June 2019 (before summer) and in September 2018 (after summer). Dates of 184 deployment for Vent 2 were from September 8 to September 24, 2018 and from May 30 to June 185 186 18, 2019 at Vent 2. One sensor was deployed in the reference area Ambient 2 during the same period. Before deployment, the SeaFETs were calibrated with ambient pH water in the 187 aquarium facilities at the Center Villa Dohrn (Ischia, Italy). The mean offset between 188 calibration samples and calibrated SeaFET pH was  $\pm 0.002$  units, indicating high quality pH 189 dataset. Discrete water samples were collected using Niskin bottles to measure the carbonate 190 system parameters during the pH sensor deployment. Samples for total alkalinity (A<sub>T</sub>) were 191 collected using standard operating protocols. The HCl (0.1 M) titrant solution was calibrated 192 against certified reference materials distributed by A.G. Dickson (CRM, Batches #153, #171, 193 and #177). Precision of the A<sub>T</sub> measurements of CRMs was < 2 and  $< 1 \mu$ mol kg<sup>-1</sup> from nominal 194

values. A<sub>T</sub> and pH<sub>T</sub> were used to determine the remaining carbonate system parameters at in
situ temperature and depth of each sampling period in the R package seacarb v3.2.12. For full
details of pH sensors deployment, calibration, and total alkalinity standard operating protocols,
see Teixidó et al. (2020).

**199** Coral collection

A total of 81 colonies of C. caespitosa and A. calycularis were randomly sampled for the 200 aquarium experiment. Fourteen (Vent 1) and sixteen (Ambient 1) colonies of C. caespitosa 201 202 were collected at 10 m depth; twenty-seven (Vent 2) and fourteen (Ambient 2) colonies of A. 203 calycularis were collected from 1 to 2 m depth. A portion of approximately 3 x 3 cm was 204 gently dislodged from each colony using a hammer and chisel. They were placed in coolers 205 filled with seawater changed regularly and maintained in Ischia for two days prior to 206 transportation to the aquarium facilities of the Laboratoire d'Océanographie de Villefranche, France. The duration of the transfer was less than 10 h; colonies were wrapped in seawater-207 208 soaked paper tissues. The colonies were cleaned, tagged with a number stuck under it using 209 water-resistant epoxy glue, and maintained at an ambient  $pH_T$  of ~8.05 during 2 weeks.

#### 210 Experimental set-up and treatments

211 Colonies of *C.caespitosa* and *A. calycularis* were maintained in two constant pH treatments

during six months (202 days), one "present-day" treatment with a pHT =  $8.08 \pm 0.01$  (mean  $\pm$ 

SE, n = 348, total number of weekly pHT measures in each 12 present-day experimental tank)

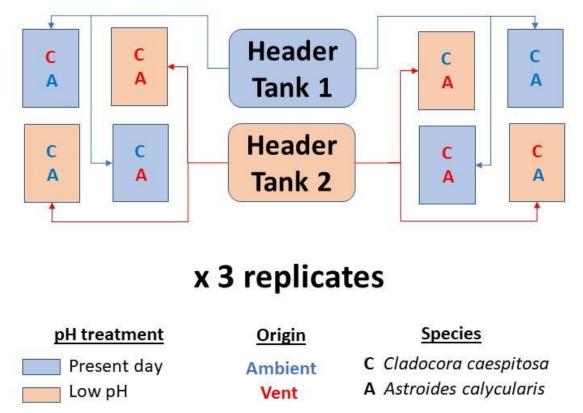
and one "low pH" treatment with a pHT =  $7.72 \pm 0.01$  (mean  $\pm$  SE, n = 348, total number of

- 215 weekly pHT measures in each 12 low pH experimental tank) corresponding to the pH values
- expected by the end of the century under the RCP 8.5 CO<sub>2</sub> emissions scenario. More
- specifically, 14 colonies from Vent 1 and 16 colonies from Ambient 1 of *C. caespitosa* and 27
- from Vent 2 and 14 from Ambient 2 of A.calycularis were used for this experiment. The vent-

low pH condition was used to test the acclimation or long-term tolerance of the colonies to 219 220 low pH in the lab, after a life-time of exposure to in situ low pH. The ambient-low pH condition was used to test the short-term tolerance to low pH of colonies that have not been 221 exposed to low pH before. The vent-present day condition tested the response to ambient pH 222 of colonies exposed in the long-term to in situ low pH. The ambient-present day condition 223 was the control. There were six independent 5 L experimental tanks for each of the four 224 225 conditions (origin x pH treatment), for a total of 24 experimental tanks. Each experimental tank contained one to three colonies of each species, depending on their size, to get 226 approximately the same total amount of polyps per tank (Tab. S1 Supplementary materials). 227 228 Seawater pumped from Villefranche Bay at 5 m depth was continuously flowing into six 25 L header tanks, used as water storage and to maintain gravity pressure for water flow into the 229 experimental tanks. Three header tanks were maintained at  $pH_T \sim 8.07$  (present-day 230 231 treatment) and the other three at  $pH_T \sim 7.7$  (low pH treatment). pH was controlled in the header tanks using pH controllers (APEX, Neptune Systems) which regulated the delivery of 232 pure CO<sub>2</sub>. Each header tank gravity-fed four experimental four-liter tanks at a rate of 100 ml 233 min<sup>-1</sup> (Fig. 1). 234

Light was provided by 89 W LED light bars (Aquaristik, Aqualumix). Irradiance gradually 235 increased from 0 at 06:00 to a maximum irradiance of 180 µmol photons m<sup>-2</sup> s<sup>-1</sup> between 236 11:00 and 14:00, and gradually decreased to 0 at 18:00. According to the attenuation 237 238 coefficients of PAR in Villefranche Bay calculated in Martin and Gattuso (2009), the maximum irradiance levels used in the study (180  $\mu$ mol photons m<sup>-2</sup> s<sup>-1</sup>) is within the 239 irradiance range observed at 10 m, where *Cladocora caespitosa* can be found in the 240 Mediterranean Sea (154 to 449 µmol photons m<sup>-2</sup> s<sup>-1</sup>). As A. calycularis were collected in a 241 semi-submerged cave and an overhang, dark plastic bags covered one side of the experimental 242 tanks to shade colonies from direct light. The 24 experimental tanks were kept in 6 water 243

baths under controlled temperature following Ischia's natural changes (Fig. S2 Supplementary
materials, APEX, Neptune Systems). Submersible pumps (NEWA) provided water motion in
each experimental tank. Three times a week, each tank was provided with a 30 mL solution of
freshly hatched brine shrimps (*Artemia* sp.) for feeding.



- Figure 1. Experimental set-up used to test the effects of origin (ambient vs vent site) and
- pH treatment (present day,  $pH_T = 8.07 \pm 0.01$  and low pH,  $pH_T = 7.72 \pm 0.01$ ) on the
- coral *C. caespitosa* and *A. calycularis*. The experimental set-up was repeated 3 times,
- resulting in 6 header tanks, and 24 experimental tanks in which corals were randomlyassigned.
- 253 <u>Health status of experimental colonies</u>
- 254 With pictures at the start of the experiment, each colony has been classified into two
- categories: colonies with loss of coenosarc (tissue between polyps) and colonies with tissue
- connecting their polyps. The number of dead polyps per colony (complete disappearance of
- tissue in the calice) was counted by comparing the pictures of the same colony at the start and
- the end of the experiment.

#### 259 <u>Carbonate chemistry in experimental tanks</u>

pH in the header and experimental tanks was measured weekly using a handheld pH-meter 260 (826 pH mobile, Metrohm) calibrated with a TRIS buffer (batch #T33 provided by A. 261 262 Dickson, Scripps Institution of Oceanography, USA) before each set of measurements. Total alkalinity  $(A_T)$  of each experimental and header tank was measured weekly during the first 263 four weeks in order to ascertain whether conditions were altered by metabolic activity. 264 Subsequently, and for the remainder of the experiment,  $A_{\rm T}$  was measured every month in four 265 randomly selected tanks and one header.  $A_{\rm T}$  was determined by potentiometric titration using 266 267 a Metrohm 888 Titrando following the method of (Dickson et al. 2007). Titrations of certified reference material (Batch #186) provided by A. Dickson were used to assess the accuracy of 268 the measurements and were within 6.5  $\mu$ mol kg<sup>-1</sup> of the reference value. Ischia's temperatures 269 270 are between 0.87 to 2.67°C higher than in the Bay of Villefranche during the studied period. Water pumped on the Bay of Villefranche was heated to be as close as possible to Ischia's 271 temperatures. Salinity at Ischia and in the Bay of Villefranche-sur-Mer are similar. Salinity 272 273 data were gathered from the salinity recorded weekly in the Bay of Villefranche by the Service d'Observation Rade de Villefranche, SO-Rade, of the Observatoire Océanologique 274 and the Service d'Observation en Milieu Littoral, SOMLIT/CNRS-INSU. pH<sub>T</sub>, temperature, A<sub>T</sub> 275 and salinity were used to calculate the other carbonate chemistry parameters using the R 276 277 package seacarb (Gattuso et al. 2020) (Tab. 1, Tab. S2 Supplementary materials).

#### 278 Six-month and monthly calcification

Calcification rate was assessed using the buoyant weight technique (Davies 1989). Weighing
was done every 27 to 35 days. Changes in wet weight were converted to dry weight using the
following equation:

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$$dry \, weight = \frac{wet \, weight}{(1 - \frac{water \, density}{aragonite \, density})}$$

with an aragonite density of 2.93 g cm<sup>3</sup>. Six-month calcification rates corresponding to the 283 284 entire experiment duration were determined as the change in dry weight between the first and the last weighing, normalized by the corals living tissue surface at the end of the experiment 285 and the number of days of the experiment (202 days). Regular rates of calcification over time 286 were normalized to the surface of the corals and to the number of days between weighing 287 (~30 d). Surfaces were determined by the aluminum foil technique for *C. caespitosa* (Marsh 288 289 1970) because these 3D colonies were impossible to analyze with 2D pictures. A. calycularis surfaces were measured by photographic analysis with the software ImageJ for (IMAGEJ, 290 291 NIH US Department of Health and Human Services) as they had flat shape. Surface 292 measurements were completed at the start and end of the experiment. Marginal changes in surface area were observed between the beginning and end of the experiment. 293 294 Dark respiration, gross photosynthesis and short-term calcification 295 Dark respiration, net photosynthesis and short-term calcification were determined during less than 1 h incubations in 500 ml transparent perspex chambers, placed in a temperature-296 297 controlled water bath to maintain a constant temperature. Incubations were carried out a 298 month after the start of the experiment (hereafter "start") and a month before the end of the experiment (hereafter "end") at the same temperature of 20°C. Magnetic stirrers were used to 299 300 provide water motion in the incubation chambers. Each colony was transferred to an individual chamber filled with water from the experimental tanks from which the corals were 301 302 taken for the incubation. A blank incubation with no coral colony and filled with water from the same header tank was performed for each four experimental tanks. Incubations lasted 30 303 to 60 min depending on the magnitude of the metabolic activity. To measure oxygen uptake 304

during respiration, colonies were maintained in darkness by covering the chamber with an 305 306 opaque plastic bag and incubations started after a 10 min acclimation period. Production of oxygen by photosynthesis was only measured on colonies of C. caespitosa under an 307 irradiance of 180 µmol photons m<sup>2</sup> s<sup>-1</sup>. Oxygen saturation within the chambers was measured 308 every 5 s using a fiber optic oxygen sensor (PreSens, OXY-4 mini), which was calibrated at 309 0% saturation in an oxygen-free solution created by saturating seawater with sodium sulfite, 310 311 and at 100% saturation in an oxygen-saturated seawater obtained by bubbling ambient air during 5 min. The difference in oxygen saturation between the beginning and end of the 312 incubation was converted to O<sub>2</sub> mg L<sup>-1</sup> using the gas\_O2sat function of the marelac R package 313 314 (Soetaert et al. 2020). Oxygen consumption was then normalized by coral surface area, incubation time and chamber volume to obtain the rates of respiration (negative in dark 315 conditions) and net photosynthesis (oxygen production in excess of the respiratory demand 316 317 under light conditions). The rate of gross photosynthesis was calculated by subtracting the respiration rate in dark to the rate of net photosynthesis, with the assumption that respiration 318 rate is the same under light and dark conditions. 319

Water samples were collected at the beginning and end of the incubation for each colony for

321 calcification determination using the alkalinity anomaly technique (Smith and Key 1975).

322 This method relies on the stoichiometric relationship between total alkalinity and

calcification, where the precipitation of one mole of  $CaCO_3$  lowers the total alkalinity by two moles. Alkalinity anomalies were normalized by the coral surface area, incubation time, and chamber volume to calculate light and dark calcification rates. The alkalinity anomaly method was only applied to the zooxanthellate *C. caespitosa*.

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Biological surveys: A generalized linear mixed model was used to test for differences in 330 abundance (poisson family) as a function of site (fixed factor, 2 levels) and quadrat as a 331 332 random effect. Chi-square contingency tables were used to compare the size-frequency distributions between sites. Laboratory experiment: The assumptions of normality and 333 equality of variance were evaluated through graphical analyses of residuals using QQ plot 334 functions in R software. Mean values of the colonies in the experimental tank were used as a 335 replicate for six-month calcification. Each incubation was treated as a statistical replica 336 337 because they were done in individual chambers on single individuals. The effect of header tanks was first tested with an ANOVA and since its significance was negligible (p > 0.05), it 338 was not considered in subsequent analyses. A two way-ANOVA was then used to test for 339 340 differences in net calcification caused by the origin (vent or ambient pH sites) and the pH treatments (present-day or low pH) for both C. caespitosa and A. calycularis. For net 341 calcification as a function of time, a three-way ANOVA was applied, with origin, pH 342 treatment, and time as fixed factors. Tukey's post hoc tests were conducted when significant 343 differences were detected. Three-way repeated measure ANOVA was used for dark 344 respiration, gross photosynthesis and short-term calcification since the same colonies of C. 345 caespitosa and A. calycularis were used at the start and end of the incubations. Results have 346 347 been reported as means  $\pm$  standard error of the mean (SE) hereafter, unless otherwise noted.

348 **Results** 

### 349 In situ environmental conditions and populations of corals

350 Water carbonate chemistry and in situ monitoring of seawater  $pH_T$  (pH on the total scale) at

- 351 the CO<sub>2</sub> vent and reference sites with ambient pH showed mean  $pH_T = 7.91$  at Vent 1,  $pH_T =$
- 352 7.91 at Vent 2,  $pH_T = 8.05$  at Ambient 1, and  $pH_T = 7.97$  at Ambient 2, respectively (Tab. 1).

- 353 The field surveys showed significant lower abundance of *C. caespitosa* at Vent 1 compared to
- Ambient 1 (1.8  $\pm$  0.16 colonies versus 3.3  $\pm$  0.35, Z = 2.2, p < 0.02, Fig. S3 Supplementary
- materials). No large colonies of more than 21 cm were found at Vent 1 (Fig. S1,
- 356 Supplementary materials). This reduction of size was also observed for *A.calycularis* at Vent
- 2 (92 % had up to 10 polyps *versus* 72 % in Ambient 2), which differed significantly between
- 358 sites ( $\chi^2 = 76.7$ , *p*<0.0001). Abundance of *A. calycularis* was lower in Vent 2 (67.1 ± 10.3 at
- 359 Vent 2 *versus*  $80.3 \pm 14.5$  at Ambient 2, Z = 2.65, p < 0.08).

361 TABLE 1. Measured and estimated seawater physiochemical parameters at the CO<sub>2</sub> vent sites, reference areas with ambient pH and both pH treatment

in the experimental tanks for salinity (S), temperature (T), total alkalinity (AT), dissolved inorganic carbon ( $C_T$ ), pH<sub>T</sub>, *p*CO<sub>2</sub>, calcite ( $\Omega c$ ) and aragonite ( $\Omega a$ ) saturation. Values are means  $\pm$  SD with 25<sup>th</sup> and 75<sup>th</sup> percentiles. Calculated concentrations of C<sub>T</sub>, *p*CO<sub>2</sub>,  $\Omega c$  and  $\Omega a$  are shown. 1: Parameters measured from discrete water samples; 2: parameters measured with in situ sensors. pH conditions: vent system (low pH); ambient pH; ambient pH experimental treatment (present day); acidified pH experimental treatment (low pH).

Local name	pH conditions	S	T (°C)	$A_T(\mu mol \ kg^{-1})$	C <sub>T</sub> (µmol kg <sup>-1</sup> )	$pH_{\mathrm{T}}$	pCO <sub>2</sub> (µatm)	Ωc	Ωa
	Low pH	$37.9^1 \pm 0$	$16.8^2 \pm 0.4$	$2532^1 \pm 16$	$2324\pm47$	7.91 <sup>2</sup>	$626 \pm 311$	$3.69\pm0.61$	$2.39\pm0.39$
Vent1		(37.9, 37.9),	(16.6, 17.1),	(2522, 2545),	(2292, 2341),	(7.89, 7.99),	(497, 643),	(3.43, 4.13),	(2.22, 2.67),
		n=3	n=1326	n=25	n=1326	n=1326	n=1326	n=1326	n=1326
Ambient 1	Ambient pH	$37.3^1 \pm 0$	$17.3^{2} \pm 0.4$	$2618^1\pm15$	$2338\pm20$	8.05 <sup>2</sup>	$448 \pm 42$	$4.80\pm0.30$	$3.11\pm0.19$
		(37.2, 37.3),	(17.0, 17.6),	(2607, 2633),	(2324, 2350),	(8.03, 8.07),	(417, 471),	(4.61, 5.01),	(2.98, 3.24),
		n=3	n=1331	n=14	n=1331	n=1331	n=1331	n=1331	n=1331
Vent 2	Low pH	$37.3^1\pm0.2$	$25.9^2\pm0.2$	$2564^1\pm7$	$2542\pm79$	7.65 <sup>2</sup>	$2905 \pm 1664$	$1.68\pm0.59$	$1.11\pm0.39$
		(37.2, 37.5),	(25.8, 26.0),	(2561, 2566),	(2477, 2585),	(7.58, 7.90),	(1724, 3438),	(1.21, 2.21),	(0.80, 1.47),
(Sept 2018)		n=9	n=1530	n=9	n=1530	n=1530	n=1530	n=1530	n=1530
XX + 0	Low pH	$37.8^1 \pm 0$	$21.8^2\pm2.1$	$2541^{1}\pm20$	$2352\pm89$	$7.74^{2}$	$983\pm868$	$3.56 \pm 1.05$	$2.33\pm0.69$
Vent 2		(37.8, 37.8),	(19.8, 23.8),	(2533, 2550),	(2289, 2389),	(7.74, 7.93),	(590, 978),	(2.96, 4.39),	(1.94, 2.88),
(June 2019)		n=7	n=1841	n=7	n=1841	n=1841	n=1841	n=1841	n=1841
Ambient 2	Ambient pH	$37.9^1\pm0$	$26.4^2 \pm 1$	$2642^1\pm17$	$2324 \pm 21$	$7.97^{2}$	$556 \pm 57$	$5.53\pm0.32$	$3.67\pm0.22$
		(37.9, 37.9),	(25.9, 27.0),	(2629, 2659),	(2310, 2338),	(7.94, 8.00),	(513, 597),	(5.31, 5.75),	(3.53, 3.82),
		n=7	n=1691	n=17	n=1691	n=1691	n=1691	n=1691	n=1691
Experimental tanks		$37.8\pm0.5$	$22.6 \pm 1.9$	$2542\pm54$	$2206\pm63$	$8.06\pm0.06$	$415\pm70$	$5.65\pm0.69$	$3.71\pm0.46$
	Present day	(37.3, 38.3),	(20.7, 24.5),	(2488, 2596),	(2143, 2269),	(8.00, 8.12),	(345, 485),	(4.96, 6.34),	(3.25, 4.17),
		n=25	n=347	n=347	n=347	n=347	n=347	n=347	n=347
Experimental tanks		$37.8\pm0.5$	22.7±1.9	$2542\pm54$	$2389\pm65$	$7.73\pm0.09$	$1048\pm237$	$2.99\pm0.56$	$1.97\pm0.37$
	Low pH	(37.3, 38.3),	(20.8, 24.6),	(2488, 2596),	(2324, 2454),	(7.64, 7.82),	(811, 1285),	(2.43,3.55),	(1.60, 2.34),
		n=25	n=348	n=348	n=348	n=348	n=348	n=348	n=348

#### Experimental conditions 366

373

Weekly temperature in the tanks varied according to natural changes in the summer 367 temperature profile at 10 m depth in Ischia, from  $19.08 \pm 0.06$  °C at the end of May 2019 368 (mean  $\pm$  SE, n = 24) to 21.51  $\pm$  0.07 °C at the beginning of December (mean  $\pm$  SE, n = 24) 369 with a maximum value of  $25.17 \pm 0.15$  °C in August (mean  $\pm$  SE, n = 24, Fig. S2 370 Supplementary materials). The "present-day" treatment was maintained at a mean pH<sub>T</sub> of 8.06 371  $\pm$  0.06 (n = 347) whereas "low pH" treatment had a pH<sub>T</sub> of 7.73  $\pm$  0.09 (n = 348). Salinity was 372  $37.8 \pm 0.5 (n = 25).$ 

Colonies of Astroides calycularis had physical differences depending on their site of origin. 374 375 85% (23/27 colonies) of vent and 14% (2/14 colonies) of ambient colonies did not present 376 coenosarc (i.e. tissue between polyps). Furthermore, by comparing photos from the beginning and end of the experiment, it was observed that none of the A. calycularis colonies from 377 ambient suffered polyp death, while three vent colonies showed 2/5, 1/7 and 4/9 dead polyps 378 379 (number of dead polyps / total number of polyps in the colony). No polyp loss was observed in Cladocora caespitosa. 380

Six-month, monthly and short-term calcification rates 381

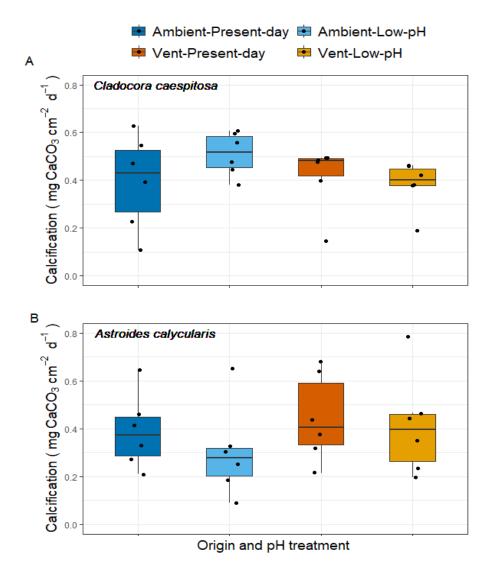
The rates of six-month calcification of C. caespitosa and A. calycularis were not affected by 382

the origin and the pH treatment (Fig. 2; Tab. S3 Supplementary materials; Two-way ANOVA, 383

 $F_{1,20}$ = 1.778, P = 0.2 for *C*. *caespitosa*,  $F_{1,20}$ = 0.13, P = 0.72 for *A*. *calycularis*). 384

- No statistically significant effect of origin and pH treatment on calcification rate was found in 385
- 386 C. caespitosa over time (Fig. 3A;  $F_{1,110}$ = 1.919, P = 0.17). However, different calcification
- rates with time were found for A. calycularis (Fig. 3B,  $F_{1,112}$ = 26.2, P < 0.001), with 387
- significantly higher calcification measured during the first month of the experiment  $(1.0 \pm 0.2)$ 388

- mg CaCO<sub>3</sub> cm<sup>-2</sup> d<sup>-1</sup>). The mean monthly calcification rate was  $0.2 \pm 0.1$  mg CaCO<sub>3</sub> cm<sup>-2</sup> d<sup>-1</sup>
- 390 during the following months of experiment.



#### 391

Figure 2. Six-month calcification rates of *C. caespistosa* and *A. calycularis* collected from vent and ambient sites exposed to the two pH treatments (pHT =  $8.08 \pm 0.01$ , present

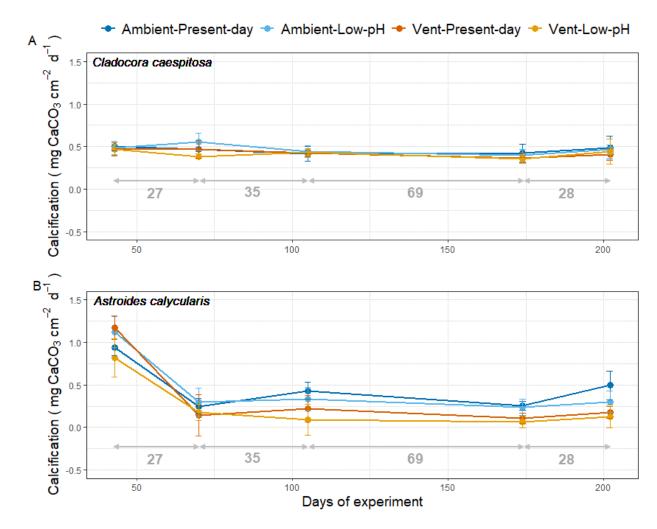
day;  $pHT = 7.72 \pm 0.01$ , low pH) during the entire duration of the experiment (202 days).

Panel A) shows *Cladocora caespitosa* and panel B) shows *Astroides calycularis*. Dots

represent the mean calcification rates of colonies per experimental tank and boxes represent

the median and the 25 and 75% quartiles. The color of the box indicates the origin and the

398 treatment. n = 6 per pH treatment and origin.



399

Figure 3. Calcification rates of *C. caespistosa* and *A. calycularis* from vent and ambient sites under the two pH treatments ( $pH_T = 8.08 \pm 0.01$ , present day;  $pH_T = 7.72 \pm 0.01$ , low pH) over time (every 27 to 35 days). Values are means  $\pm$  SE, Panel A) *Cladocora caespitosa* and B) *Astroides calycularis*. The color of the dots and lines indicates the origin and the treatment. We had a technical problem with day 139 weighing which is therefore not represented here. Numbers below arrows between dots represent the number of days between measurements. n = 6 per pH treatment and origin.

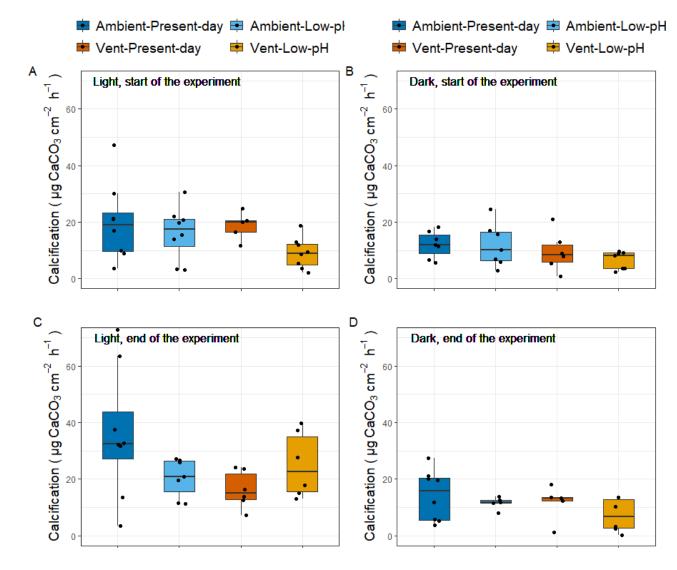
407 Short-term calcification of *C. caespitosa* in the dark and in the light was not significantly

different at the start (Fig. 4A and 4B;  $10 \pm 1.1$  and  $15.65 \pm 1.85 \ \mu g \ CaCO_3 \ cm^{-2} \ h^{-1}$ ,

409 respectively) and the end of the experiment (Fig. 4C and 4D;  $11.42 \pm 1.40$  and  $25.21 \pm 3.03$ 

410  $\mu$ g CaCO<sub>3</sub> cm<sup>-2</sup> h<sup>-1</sup>, respectively). Colonies light and dark calcification were also not affected

411 by origin and pH treatment (Fig. 4;  $F_{1,40}$ = 0.07, P = 0.795).





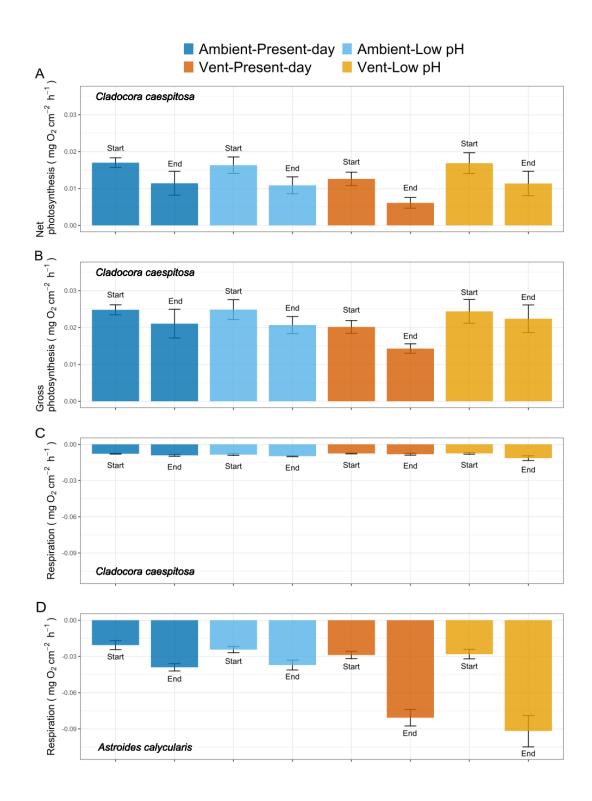
413Figure 4. Short-term calcification rates for *Cladocora caespitosa* from vent and ambient414sites under the two pH treatments ( $pH_T = 8.08 \pm 0.01$ , present day;  $pH_T = 7.72 \pm 0.01$ , low415pH) in the light and dark at the start ( $27^{th}$  of June 2019) and end of the experiment ( $24^{th}$ 416of October 2019). Values are means  $\pm$  SE. Rates were calculated using alkalinity anomaly417method following short term incubations (< 1 h) at 20°C. Panel A) and C) under light</td>418conditions, B) and D) under dark conditions. Dots represent the calcification rates of the coral419colonies and boxes represent median and 25 and 75% quartiles. The color of the box indicates

- 420 the origin and the treatment. n = 5 to 8 per pH treatment and origin.
- 421

### 422 Dark respiration and gross and net photosynthesis

- 423 Net photosynthesis of *C. caespitosa* was not affected by origin and pH treatment (Fig. 5A;
- 424  $F_{1,40}= 2.36, P = 0.134$  and  $F_{1,40}= 1.24, P = 0.272$ , respectively), but was different between the

- beginning and the end of the experiment (Fig. 5A;  $F_{1,40}$ = 13.14, P < 0.001). Gross
- 426 photosynthesis of *C. caespitosa* did not exhibit any statistically significant difference between
- 427 origin, pH treatment and time (Fig. 5B;  $F_{1,24}$ = 0.17, P = 0.683). Mean gross photosynthesis
- 428 for all pH treatments was similar at the start and end of the experiment ( $0.024 \pm 0.001$  vs
- 429  $0.020 \pm 0.002$  mg O<sub>2</sub> cm<sup>-2</sup> h<sup>-1</sup>, n = 30). As light and dark respiration may differ, gross
- 430 photosynthesis measurements are uncertain, and most likely underestimated.
- 431 Dark respiration rate of *C. caespitosa* was not significantly different between origin and pH
- 432 treatment (Fig. 5C,  $F_{1,24}$ = 0.008, P = 0.93 and  $F_{1,24}$ = 3.09, P = 0.091, respectively), but was
- 433 significantly different between the beginning and end of the experiment ( $F_{1,24}$ = 9.45, P =
- 434 0.005). In June, mean dark respiration rates for all conditions (origin and pH treatment) was 435  $-0.0079 \pm 0.0003 \text{ mg O}_2 \text{ cm}^{-2} \text{ h}^{-1} (n = 30)$  while in November it was  $-0.0097 \pm 0.0006 \text{ mg O}_2$ 436  $\text{cm}^{-2} \text{ h}^{-1} (n = 28)$ , which represents a 23% of increase in dark respiration.
- At the start of the experiment, the dark respiration of A. calycularis did not differ between 437 origin or pH treatment (Fig. 5D,  $F_{1,37}$ = 0.32, P = 0.58). At the end of the experiment (Fig. 438 5D), a significant difference between site of origin was observed ( $F_{1,36}$ = 18.5, P < 0.0001), 439 with corals from the ambient site having a mean dark respiration of  $-0.038 \pm 0.003$  mg O<sub>2</sub> cm<sup>-</sup> 440  $^{2}$  h<sup>-1</sup> (n = 13) whilst colonies from the CO<sub>2</sub> vent site had a rate of -0.087 ± 0.008 mg O<sub>2</sub> cm<sup>-2</sup> h<sup>-1</sup> 441  $^{1}$  (n = 27). The overall (all treatments) mean dark respiration rate were also different at the 442 start and at the end of the experiment, with a lower mean dark respiration at the start (-0.026  $\pm$ 443  $0.002 \text{ vs} - 0.071 \pm 0.007 \text{ mg } \text{O}_2 \text{ cm}^{-2} \text{ h}^{-1}$ ;  $F_{1,36} = 52.3, P < 0.0001$ ). 444



446Figure 5. Net and gross photosynthesis and dark respiration rates of *C. caespitosa* and *A.*447*calycularis* from the vent and ambient sites under the two pH treatments ( $pH_T = 8.08 \pm$ 4480.01, present day;  $pH_T = 7.72 \pm 0.01$ , low pH) at the start (27th of June 2019) and end of the449**experiment (24th of October 2019).** Values are means  $\pm$  SE, Panel A) *Cladocora* 

*caespitosa*'s net photosynthesis, B) *Cladocora caespitosa*'s gross photosynthesis, C)

*Cladocora caespitosa*'s dark respiration. D) *Astroides calycularis*' dark respiration. The color

452 of the bars indicates the origin and the treatment. n = 6 to 15 per pH treatment and origin.

#### 453 **Discussion**

During this six-month laboratory experiment, we tested whether colonies of the 454 zooxanthellate coral *Cladocora caespitosa* and the azooxanthellate coral *Astroides calycularis* 455 growing in naturally acidified waters were more tolerant to low pH than colonies from 456 ambient pH sites or acclimated. In contrast to our initial hypothesis, calcification rates and 457 458 metabolic responses did not change for any coral species no matter the site of origin and pH treatment. Therefore, our results suggest that both species, independently of environmental 459 history and trophic strategy, can tolerate low pH conditions for six months, under controlled 460 conditions in aquaria. In contrast, in situ surveys revealed lower abundance and smaller sizes 461 for both corals at CO<sub>2</sub> vent sites compared with ambient sites. 462

The hypothesis that corals from vent sites would be less impacted by low pH stems from 463 reports that organisms experiencing highly variable or low pH in their natural habitat may be 464 465 more tolerant to ocean acidification (Cornwall et al. 2018). Corals living in suboptimal environmental conditions may possess physiological and potentially genetic mechanisms that 466 467 confer tolerance to such environmental conditions (Camp et al 2018, Teixidó et al 2020). For 468 example, microhabitats experiencing periodic temperature extremes are potential habitats for 469 heat tolerant corals (Schoepf et al. 2015), yet these habitats can exhibit a mosaic of populations, harboring species previously reported with high to low thermal tolerance 470 471 (Ainsworth et al. 2016). Such physiological mechanisms and intraspecific variability may exist in C. caespitosa and A. calycularis which highlights the importance of studying the 472 influence of suboptimal environmental conditions on tolerance to stressors. 473

474 Previous research has shown a decrease in calcification when corals were transplanted from

ambient pH sites to CO<sub>2</sub> vents (Rodolfo-Metalpa et al. 2011; Prada et al. 2017). However,

476 previous studies on temperate/cold-water coral including *C. caespitosa* and *A. calycularis* 

477 have also shown a greater tolerance to ocean acidification than most tropical corals (see Tab.

S5 Supplementary materials). Calcification rates reported in the present study agree with 478 479 earlier studies. For C. caespitosa, Rodolfo-Metalpa et al. (2010) observed a yearly calcification rate of 0.51 mg CaCO<sub>3</sub> cm<sup>-2</sup> d<sup>-1</sup>, which is similar to what we measured during 480 this six-month experiment  $(0.43 \pm 0.03 \text{ mg CaCO}_3 \text{ cm}^{-2} \text{ d}^{-1})$ . The calcification rates of A. 481 *calycularis* measured here are also of the same order of magnitude to those reported by 482 Movilla et al. (2016)  $(0.39 \pm 0.04 \text{ and } 0.47 \pm 0.06 \text{ mg CaCO}_3 \text{ cm}^{-2} \text{ d}^{-1}$ , respectively). In 483 contrast to temperate and cold-water corals, tropical corals tend to be more affected by ocean 484 acidification, as they exhibit a mean decrease in calcification of 10-25% when exposed to the 485 pCO<sub>2</sub> expected by the end of the century under the high-CO<sub>2</sub> emissions scenario (Chan and 486 487 Connolly 2013; Cornwall et al. 2021). Most tropical corals investigated so far are fast calcifiers, growing in warm waters, therefore as seawater pH declines, they may be unable to 488 maintain these energetically costly fast rates of calcification (Comeau et al. 2014). In contrast, 489 490 temperate and cold-water corals are slower calcifiers with lower energy requirements for mineralization, thus making them more likely to maintain constant calcification rates at lower 491 pH (Rodolfo-Metalpa et al. 2010, Comeau et al. 2014). Ries et al. (2010), also suggested that 492 the effect of acidification on the calcification of temperate corals, such as Oculina arbuscula, 493 494 may manifest at a lower pH threshold compared to tropical corals. The capacity of C. 495 caespitosa and A. calycularis to control the carbonate chemistry at their site of calcification could explain this tolerance to low pH. For example, using the geochemical proxy  $\delta^{11}$ B it was 496 shown that the temperate coral Balanophyllia europaea at the CO<sub>2</sub> vents of Panarea Island 497 (Italy) can maintain constant calcifying fluid pH when exposed to pH ranging from 8.07 to 498 7.74 (Wall et al. 2019). This confirms the capacity of some corals to maintain internal pH 499 500 homeostasis and utilize this as a way of calcifying under reduced seawater pH (Comeau et al. 2019). 501

From this study no conclusions can be drawn as to whether colonies from the vent sites are 502 503 acclimated to low pH, as colonies from both vent sites and the ambient sites were not affected by low pH in the laboratory. However, controlling the transport of calcium and inorganic 504 carbon to the calcifying medium and producing the skeletal organic matrix are all energy-505 506 demanding processes (Allemand et al. 2011, Wall et al. 2019), therefore the maintenance of calcification rates under these treatments may have come at a cost. For example, Wall et al. 507 (2019) suggest that the inactivation of some calicoblastic cells could compensate for the 508 509 energy required to regulate internal pH and maintain calcification rates under low seawater pH, but at the cost of increased skeletal porosity. This is supported by a study that showed a 510 511 constant growth rate of the temperate coral *B. europaea* in acidified conditions at the expense 512 of skeletal density (Fantazzini et al. 2015), which in return increases the sensitivity to bioerosion and susceptibility to damage (Crook et al. 2013). Skeletal porosity was not 513 514 measured in the present study. However, increased sensitivity to bioerosion can also be linked to the tissue coverture of the skeleton. Rodolfo-Metalpa et al. (2010) observed that following 515 transplantation to low pH conditions, skeleton dissolution was observed in Cladocora 516 caespitosa which had exposed skeleton, and not in Balanophyllia europaea which had less 517 518 exposed skeleton. In our study, we observed that most of the A. calycularis colonies collected 519 at the CO<sub>2</sub> vent sites did not possess coenosarc (i.e. coral tissue between polyps) and exhibited 520 polyp death during our experiment. These phenomena were also observed in natural colonies of A. calycularis (Teixidó et al. 2020) and the corals Pocillopora damicornis and Oculina 521 522 patagonica when incubated at low pH (Kvitt et al. 2015). The increased sensitivity to bioerosion of A. calycularis and C. caespitosa with exposed skeleton could explain in part the 523 field observation. 524

Field observations confirmed the difference between the population of the CO<sub>2</sub> vent and
ambient sites, as both corals were less abundant and had smaller colonies at the vent sites

(Fig. S3 Supplementary materials, Teixidó et al. 2020). These contrasting results between 527 528 field and laboratory observations, could be explained by differences in pH variability, which is much greater in the field with short periods of very low pH. Alternatively, other 529 environmental parameters such as food availability or biological interactions (e.g., 530 competition with algae) may contribute to these differences. pH variability at the CO<sub>2</sub> vent 531 sites can be very high, as pH<sub>T</sub> can range from 6.70 to 8.13 in a day (Foo et al. 2018, Teixidó et 532 al. 2020), whereas  $pH_T$  was very stable during our pH-controlled laboratory experiment (pH<sub>T</sub> 533 =  $7.72 \pm 0.01$  for low pH treatment). It may be that no differences in calcification and 534 photosynthesis were observed in the laboratory, due to the fact that both corals are slow-535 536 growing and long-lived species that have had a lifetime to acclimatize to their environmental 537 pH. Therefore, the six-month time scale of the experiment may not have been enough to trigger any visible variation. However, even if our results did not show any statistical 538 539 difference between treatments during the laboratory experiment, different trends in calcification between conditions can be observed for A. calycularis. The colonies from 540 ambient-low pH condition seem to have lower calcification rate than the control colonies of 541 ambient-present day condition (Fig. 2), suggesting a possible impact of low pH on 542 543 calcification in the short term. The same trend is observed with the colonies from the vent 544 site, as the vent-low pH presents reduced calcification rate compared to the vent-present day 545 condition. Moreover, calcification was similar in the vent-low pH and the control ambientpresent day conditions. This trend suggests a possible acclimation to low pH in corals 546 547 growing at the CO<sub>2</sub> vent site

Zooxanthellate and azooxanthellate corals were both used in the present study to assess the
role that photosynthesis could play in mitigating the effects of ocean acidification on
calcification. In zooxanthellate corals, light-enhanced calcification describes the relationship
between calcification, photosynthesis and irradiance (Gattuso et al. 1999; Allemand et al.

2011). Four main mechanisms explaining light-enhanced calcification were proposed: 552 553 photosynthesis could (1) supply energy to the animal through the translocation of photosynthates, (2) increase pH at the site of calcification by using  $CO_2$ , (3) remove 554 phosphate, which is poison to carbonate crystallization, and (4) facilitate the production of the 555 organic matrix which serves as a scaffold in which CaCO<sub>3</sub> is precipitated (Allemand et al. 556 557 2011). Here, short-term calcification of C. caespitosa was higher in light than in the dark, with a mean light:dark calcification ratio of 2.38 (See Tab. S4 Supplementary materials), 558 which is close to the mean ratio of 3.0 obtained by Gattuso et al. (1999). However, neither 559 light nor dark calcification rates differed with experimental pH conditions, thus we cannot 560 561 conclude that photosynthesis did provide better tolerance to ocean acidification. Furthermore, 562 no difference in the response of calcification to ocean acidification was observed between the zooxanthellate (C. caespitosa) and azooxanthellate species (A. calvcularis). The data from the 563 564 in situ survey at Ischia did not highlight a difference between the zooxanthellate and azooxanthellate corals either, as the size and abundance of colonies from CO<sub>2</sub> vent sites were 565 lower for both species. Contrastingly, Prada et al. (2017) reported that the zooxanthellate 566 coral *B. europaea* was unaffected by low pH conditions, whereas the azooxanthellate corals 567 Leptopsammia pruvoti and A. calycularis showed a negative effect of pH on calcification. 568 569 However, the study of Prada et al. (2017) was conducted in situ, on corals transplanted next to 570 CO<sub>2</sub> vents in Panarea Island, whereas our study is based on a laboratory experiment and field observations. Feeding could also explain this difference as it can mitigate the negative effects 571 572 of environmental stressors on coral physiology (Fox et al. 2018). In our experiment, corals were fed freshly hatched Artemia nauplii three-times a week therefore the demand of energy 573 574 required to maintain calcification under low pH was likely sustained by heterotrophy (Edmunds 2011). Kornder et al. (2018) showed that in studies where corals were fed, the 575 negative impacts of ocean acidification on corals were mitigated. Therefore, the fact that 576

577 Movilla et al. (2012) observed a decrease in calcification at low pH in *C. caespitosa* may be 578 explained by the fact that the colonies were only fed once per week.

No difference in photosynthesis was observed between site of origin or pH treatment. This 579 580 aligns with the observation that there is a limited impact of environmentally relevant decreases in pH on photosynthesis (Comeau et al. 2017), and that low pH rarely causes 581 bleaching (Anthony et al. 2008). Moreover, Mediterranean corals harbor Clade B 582 Symbiodinium, that have growth rates, photosynthetic capacities and respiration rates that are 583 unaffected by acidification (Brading et al. 2011). Respiration was also not affected by the pH 584 585 treatment in C. caespitosa, confirming previous observations for this species (Rodolfo-Metalpa et al. 2010), and other coral species (Comeau et al. 2017). However, colonies of A. 586 *calycularis* from the vent site exhibited a higher rate of respiration at the end of the 587 588 experiment (November) for both low and present-day pH treatments. This result contrasts with studies that show a decrease in respiration at lower pH (e.g. Comeau et al. 2014), or no 589 impact on respiration (Rodolfo-Metalpa et al. 2010; Comeau et al. 2017). However, an 590 591 increase in respiration could be due to the increased energetic demand associated with maintaining calcification rates at lower pH (Comeau et al. 2017). For example, a decrease in 592 seawater pH<sub>T</sub> of 0.7 units requires an additional 1-2 kJ of energy mol<sup>-1</sup> of CaCO<sub>3</sub> 593 594 precipitated (McCulloch et al. 2012a). For zooxanthellate corals, this energy is less than 1% of the energy provided by photosynthesis, however, for azooxanthellate corals such as A. 595 596 *calycularis*, respiration is the only way to up-regulate this energy demand. Moreover, the increase of respiration from colonies of vent sites of A. calycularis could be partly due to the 597 598 smaller size of colonies at the vent site than at the ambient site (Edmunds and Burgess 2016). Although the corals studied here showed tolerance in the laboratory to pH conditions expected 599 by 2100, it is a fragile equilibrium that can be disrupted by the inclusion of changes in other 600

601	environmental variables such as temperature that could lead to a greater impact on
602	physiological processes and survival, as observed in the in situ survey and predicted by the
603	negative trend on the calcification of A. calycularis. For example, marine heatwaves can lead
604	to mass mortality of C. caespitosa and A. calycularis (Rodolfo-Metalpa et al. 2005) and are
605	projected to be more frequent and severe in the future (Galli et al. 2017, Smale et al. 2019).
606	Moreover, the impact of ocean acidification on reproduction as well as larval and juvenile
607	survival, which can limit the persistence and dispersion of a species remains to be
608	investigated. The present study contributes to our knowledge of Cladocora caespitosa and
609	Astroides calycularis' ability to survive under future ocean acidification conditions. Such
610	information will improve our ability to manage and conserve the biodiversity of
611	Mediterranean calcifying species.
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#### 825 Author contributions

826 S.C., N.T., B.M, and J.-P.G designed the study. N.T., S.C., A.M. and B.M. were involved with

fieldwork. C.C, B.M., T.G., S.C., and N.T performed the experiments. C.C., N.T. and A.M.

analyzed the data. C.C. wrote the first draft of the manuscript which was then finalized by all

829 co-authors.