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#### Prediction of water retention properties of Syrian clayey soils 1

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#### 15 ABSTRACT

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16 Studies on clayey soils developed in temperate areas have shown that their water retention properties 17 are related to both the clay content and the specific pore volume of the clay, the latter being related to the hydric history of the soil. Our objective was to discuss the validity of these results for clayey soils 18 19 developed in semi-arid areas. Samples were collected in soils located in Syria. Physico-chemical 20 properties were determined. Water content was measured at field capacity and for water potentials 21 ranging from -10 to -15000 hPa. X-ray diffraction analyses were performed on the clay fraction to identify the clay. Results showed that the clays have both a high cation exchange capacity (0.707 to 22 23 0.891 mmol+ per g of clay) and a high external specific surface area (112 and 178 m<sup>2</sup> per g of clay). 24 These values are consistent with the X-ray diffraction results which showed the presence of a high proportion of smectite in most horizons and secondarily of varying proportions of illite and chlorite; 25 26 kaolinite, while present, was not abundant. Results also showed that the amount of water retained by the clay according to the water potential was closely related to the specific pore volume of the clay at 27 field capacity. Regression equations established by using both the data published earlier and those of 28 29 this study enabled us to predict the water retention properties of clayey soils for a larger range of clay 30 mineralogy and climatic environments including semi-arid environments than previously discussed in 31 the literature.

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Keywords: Clay content, Bulk density, Cation exchange capacity, Specific surface area, X-ray
 diffraction, Pedotransfer function.

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#### 36 Introduction

Clayey soils are often intensively cultivated in semi-arid and arid areas, particularly when irrigation is 37 available (e.g., Ethan & Umar 2001; Mason et al., 2015). Their physical properties are closely related 38 39 to the clay content and clay characteristics such as their mineralogical nature, the size of the elementary clay particles and the nature of the exchangeable cations (Tessier & Pédro 1987; Quirk 40 1994; Wilson 1999; Churchman & Velde 2019). Whatever the soil texture and more markedly in semi-41 42 arid and arid areas, it is important to ascertain the water retention properties of soils in order to manage 43 the water supply and consequently crop growth appropriately (Gaiser et al., 2000). To accomplish this, 44 pedotransfer functions that enable prediction of the water retention properties of soils have therefore 45 been developed (e.g., Bouma & Van Lanen 1987; Wösten et al., 1999; Al Majou et al., 2008a & 2008b; Ostovari et al., 2015; Kalumba et al., 2020; Singh et al., 2020; Xu Xu et al., 2021), with several 46 47 studies focusing on soils of semi-arid and arid areas (Tóth et al., 2012; Khlosi et al., 2012 & 2016; Al Majou et al., 2018; De Paepe et al., 2018; Santra et al., 2018; Dharumarajan et al., 2019; Soleimani et 48

49 al., 2020). At the same time, some studies aimed at a better understanding of the water retention properties of soils according to the characteristics of their clay minerals but the validity of the results 50 51 for soils of semi-arid and arid areas remains unclear (Gaiser et al., 2000; Moret-Fernandez et al., 2013). For clayey soils developed on sedimentary materials, Bruand and Tessier (2000) showed that 52 relationships between water retention properties and clay particles characteristics can be established. 53 The clay particles, thanks to their fine size ( $< 2\mu m$ ), play a privileged role in water retention and soil 54 55 structuring but also in the retention and bioavailability of chemical elements essential for plants (Bergaya & Lagaly 2013; Churchman & Velde 2019). Depending on the characteristics of clay 56 particles, i.e. both their size and shape, the nature of the cations in the interlayer space and on the 57 58 external surfaces, the clay fabric, which corresponds to both the volume of clay particles and the associated pore volume resulting from their assemblage, can vary highly from one soil to another, with 59 significant consequences for the water retention properties of the soil (Tessier et al., 1992; Bruand & 60 Tessier, 2000; Boivin et al., 2004). However, most studies on the water retention properties of clayey 61 soils were developed for clayey soils in temperate areas and their validity for clayey soils in other 62 63 climate areas is still under discussion. D'Angelo et al. (2014) studied the high sensitivity to drought of Chinese red clay soils developed on clayey sediments in humid subtropical areas. They showed that it 64 was related to a very dense clay fabric which resulted from a high degree of consolidation, leading to a 65 66 small porosity able to retain water available for plants. Other clayey soils in tropical and subtropical 67 areas exhibit a strong micro-granular structure and their water retention properties are closely related 68 to both inter- and intra-microgranular porosity (Balbino et al., 2004; Tawornpruek et al., 2005; Reatto 69 et al., 2007). In this study, the objective was to analyze the water retention properties of clayey soils 70 developed in semi-arid areas of Syria and to show how results recorded in earlier work on soils located 71 in temperate areas (Bruand & Zimmer, 1992; Bruand & Tessier, 2000; Al Majou et al., 2008a) can be 72 applied to the soils studied. We discuss how both the clay content and pore volume developed by the 73 clay fabric, the latter being related to clay mineralogy and size of the elementary clay particles as well as the hydric history, can be used to predict the water retention properties of the clayey soils studied. 74

#### 76 Materials and methods

#### 77 The soils studied

The soils studied are located in semi-arid areas of Syria (Rigot 2006) between latitudes 32 and 41° N 78 and longitudes 35 to 42° E. A set of 10 samples (clay content > 0.3 g g<sup>-1</sup>) including 4 horizons A and 6 79 horizons B were collected in 5 clavey soils located in the vicinity of the towns of Jabah (JB 1 and 80 81 JB 2), Al-Suwayda (SW 1), Al-Tha'Ala (TH 1) and Al-Qamishli (QM 1) (Table 1). The soils 82 developed on clayey surface formations including coarse elements of basic extrusive rocks or 83 calcareous alluvium parent materials (Table 1). They are Cambisols (JB 1 and JB 2), Vertisols (SW 1 and TH 1) and a Clacisol (QM 1) according to the IUSS Working Group WRB (2015) (Tableau 1) and 84 all Inceptisols according to the USDA Soil taxonomy (Soil Survey Staff, 2010) (Table 1). They are 85 representative of the major Syrian clayey soils (Land Classification and Soil Survey of the Syrian 86 87 Arab Republic, 1982) in two Syrian agro-climatic areas as defined by Khlosi et al. (2012).

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#### 89 Physico-chemical and hydric analyses

90 Undisturbed roughly parallelepiped samples  $500-800 \text{ cm}^3$  in volume were collected in winter when the 91 soil was near to field capacity and as a consequence near maximum swelling. The samples were stored 92 at  $4-5^{\circ}$ C to reduce biological activity and in sealed plastic containers to avoid water loss.

Part of the samples was air-dried and manually ground in a mortar to < 2 mm material. Particlesize distribution was measured using the pipette method after pretreatment with hydrogen peroxide and sodium hexametaphosphate (Robert & Tessier 1974). The CaCO<sub>3</sub> content (*CaCO<sub>3</sub>*, in g per g of oven-dried soil) was measured according to Dupuis (1969). Cation exchange capacity (*CEC*, in mmol<sub>+</sub> per g of oven-dried soil) and exchangeable cations (in mmol<sub>+</sub> per g of oven-dried soil) were measured using ammonium acetate buffered at pH 7.0, and organic carbon content (*OC*, in g per g of oven-dried soil) by oxidation using an excess amount of potassium bichromate in sulphuric acid controlled at 100 135°C (Baize, 2018). The N<sub>2</sub>-BET external specific surface area (*SSA*, in m<sup>2</sup> per g of oven-dried soil)
101 was measured for the < 2 mm fraction dried at 105 °C (Fripiat et al., 1971).</li>

102 Centimetric clods (5 to 10 cm<sup>3</sup>) were separated by hand from the stored decametric samples. 103 Their water content at field capacity ( $W_{fc}$ , in g of water per g of oven dried soil) and bulk density ( $D_b$ ) 104 were measured using the kerosene method (Monnier et al., 1973). The gravimetric water content (W, 105 in g of water per g of oven-dried soil) at water potentials, h, of -10, -33, -100, -330, -1000, -3300 106 and -15000 hPa was measured using a pressure membrane or pressure plate apparatus. The clods were 107 placed on a paste made of < 2 µm particles of kaolinite to establish continuity of water between the 108 clods and the membrane or the porous plate of the apparatus (Bruand et al., 1996).

The mineralogical composition of the  $< 2 \mu m$  material of the horizons was determined by X-ray 109 diffraction (XRD) using the air dried < 2 mm material. The clay fraction was collected using the 110 sedimentation method at 20°C after mechanical dispersion and saturated with Mg<sup>++</sup> according to 111 Robert and Tessier (1974). Oriented clay deposits were prepared on glass slides and studied using an 112 X-ray diffractometer equipped with a Si(Li) solid detector to filter the  $Cu_{k\alpha}$  radiation ( $\lambda Cu_{k\alpha}$  = 113 1.5418 Å) of a standard European type X-ray tube (40 kV, 40 mA). The XRD patterns were collected 114 from 2 to 24° 20 per step of 0.05° 20, on successively air-dried at 20°C, glycoled, heated to 150 and 115 then 550°C oriented clay deposits (Robert & Tessier 1974; Bruand & Prost 1986). 116

117

#### 118 Criteria used to evaluate the performance of the regression equations

119 The performance of the regression equations established to predict the gravimetric water of the 120 horizons was assessed using the mean error (ME), the standard deviation (SD) and the root mean 121 square error (RMSE):

122 
$$ME = \frac{1}{l} \sum_{j=1}^{l} (W_{p,i,j} - W_{m,i,j})$$
(1)

123 
$$SD = \left\{\frac{1}{l}\sum_{j=1}^{l} \left[ \left(W_{p,i,j} - W_{m,i,j}\right) - ME \right]^2 \right\}^{1/2}$$
(2)

124 
$$RMSE = \left\{\frac{1}{l}\sum_{j=1}^{l} \left(W_{p,i,j} - W_{m,i,j}\right)^2\right\}^{1/2}$$
(3)

where  $W_{p,i,j}$  is the predicted gravimetric water content of the horizon at matric potential *i* for the horizon *j*,  $W_{m,i,j}$  is the measured gravimetric water content of the horizon at matric potential *i* for the horizon *j*, and *l* is the number of horizons studied.

In the same way, *ME*, *SD* and *RMSE* were calculated to evaluate the performance of the equationsused to predict the volumetric water content of the horizons as follows:

130 
$$ME = \frac{1}{l} \sum_{j=1}^{l} \left( \theta_{p,i,j} - \theta_{m,i,j} \right)$$
(4)

131 
$$SD = \left\{ \frac{1}{l} \sum_{j=1}^{l} \left[ \left( \theta_{p,i,j} - \theta_{m,i,j} \right) - ME \right]^2 \right\}^{1/2}$$
(5)

132 
$$RMSE = \left\{\frac{1}{l}\sum_{j=1}^{l} \left(\theta_{p,i,j} - \theta_{m,i,j}\right)^{2}\right\}^{1/2}$$
(6)

where  $\theta_{p,i,j}$  is the predicted volumetric water content of the horizon at matric potential *i* for the horizon *j*,  $\theta_{m,i,j}$  is the measured volumetric water content of the horizon at matric potential *i* for the horizon *j*, and *l* is the number of horizons studied.

136 The values of *ME* and *SD* provide information on the estimation bias and precision, respectively. The sign of ME indicates whether the prediction overestimated (positive) or underestimated (negative) 137 the value predicted. As for RMSE, it varies according to both the overall prediction bias and overall 138 prediction precision. The calculation of ME gives equal weight to all errors whereas RMSE assigns 139 more weight to larger errors than to smaller ones. The greater the difference between ME and RMSE, 140 the more large errors there are. Thus, the best prediction performance is recorded when ME is as close 141 142 as possible to 0 and SD and RMSE are as small as possible, indicating the most unbiased and precise 143 prediction.

#### 145 **Results**

#### 146 Physico-chemical characteristics

The main characteristics of the studied samples are given in Table 2. The horizons studied showed 147 very low or low CaCO<sub>3</sub> content ( $\leq 0.06$  g g<sup>-1</sup>) except for the horizons QML 1–1 and 1–2 for which the 148 CaCO<sub>3</sub> content was 0.30 and 0.31 g  $g^{-1}$ , respectively. The organic carbon content ranged from 0.002 to 149  $0.008 \text{ g g}^{-1}$ , the highest organic carbon content being recorded for the horizons A. The clay content 150 (CC, in g per g of oven-dried soil at  $105^{\circ}$ C) ranged from 0.43 to 0.69 g g<sup>-1</sup> between the different 151 horizons studied but did not vary highly with depth in each soil (Table 2). The cation exchange 152 capacity (CEC, in mmol<sub>+</sub> per g of oven-dried soil at 105°C) ranged from 0.304 to 0.492 mmol<sub>+</sub>  $g^{-1}$ . 153 The external specific surface area (SSA, in m<sup>2</sup> per g of oven-dried soil at 105°C) ranged from 61 to 154 91 m<sup>2</sup> g<sup>-1</sup>. 155

156

#### 157 *Clay mineralogy*

The XRD patterns of the clay extracted from the horizons studied showed (Figure 1) the presence ofdifferent types of clays (Robert & Tessier 1974; Bruand & Prost 1986; Moore & Reynold 1997):

160 - 1:1 clays which are kaolinite as shown by the peak at 0.72 nm which is similar at 20°C, 150°C
 161 and after treatment with ethylene glycol. The absence of a peak at 0.72 nm after heating to
 162 550°C confirmed the presence of kaolinite which is dehydroxylated at such a temperature;

non-swelling 2:1 clays which are illite as shown by the peak at 1.12–1.02 nm at 20°C which
did not change after treatment with ethylene glycol. After heating at 550°C, this peak shifted
to 1.01 nm, indicating the closing of the magnesian interlamellar spaces;

swelling 2:1 clays which are smectite as shown by the shift of the wide peak at 1.44–1.57 nm at 20°C to 1.72–1.77 nm after treatment with ethylene glycol because of the swelling of the interlamellar space due to intercalation of the ethylene glycol molecules. After heating at 550°C, all the peaks recorded for higher values than 1.01 nm disappeared, indicating the closing of the magnesian interlamellar spaces due to the loss of the ethylene glycol molecules;

171 – 2:1/1 clays which are chlorites as shown by the peak at 1.42–1.47 nm similar at 20°C and after
 172 treatment with ethylene glycol. The remaining peak at 1.42 nm recorded after heating at
 173 550°C is indeed consistent with the presence of chlorites which are not dehydroxylated at such
 174 a temperature.

A semi-quantitative evaluation of the content in the different types of clays identified is given in Table3.

177

#### 178 Water retention properties

The water retention properties showed a wide variety of water content between -10 and -15000 hPa matric water potential (Table 4). The smallest and highest water contents at field capacity were 0.262 and 0.507 g g<sup>-1</sup>, respectively. This high variation in water content at field capacity was consistent with the water content recorded at -10 hPa which ranged from 0.328 to 0.540 g g<sup>-1</sup> and at -15000 hPa which ranged from 0.207 to 0.385 g g<sup>-1</sup>.

184

### 185 **Discussion**

#### 186 Physico-chemical characteristics and clay mineralogy

As the soil horizons contained little organic carbon (Table 2), which is consistent with the soil organic
matter dynamics in semi-arid areas (Mrabet, 2011; Schellekens et al., 2013), the contribution of

organic matter to the cation exchange capacity was considered as being negligible compared with the cation exchange capacity of the clay. Indeed, if we assume that the cation exchange capacity of the organic matter of the soil studied is equal to 4 mmol<sub>+</sub> g<sup>-1</sup> of organic carbon, which is a value among the highest values recorded in the literature (Schnitzer, 1978; Tate & Theng, 1980; Stevenson, 1982), the contribution of organic matter to the *CEC* would range from 0.008 to 0.032 mmol<sub>+</sub> g<sup>-1</sup> of oven dried soil for the horizons studied. Thus, the cation exchange capacity of the clay (*CEC<sub>cl</sub>*, in mmol<sub>+</sub> per g of oven-dried clay at 105°C) was calculated as follows:

$$CEC_{cl} = (CEC/CC) \tag{7}$$

with *CC* the clay content. Results showed that  $CEC_{cl}$  ranged from 0.707 to 0.891 mmol<sub>+</sub> g<sup>-1</sup> (Table 5), indicating that the clay fraction was mainly composed of swelling 2:1 clays (Robert & Tessier 1974; Wilson, 1999; Churchman & Velde 2019). These values of  $CEC_{cl}$  are consistent with the X-ray diffraction data (Figure 1) which show the predominance of 2/1 clay minerals with smectite content ranging from not abundant to abundant depending on the horizon (Table 3).

As the calculated surface area developed by quartz spheres with a diameter > 2  $\mu$ m was < 1.1 m<sup>2</sup>  $g^{-1}$ , the contribution of both the silt and sand particles can be assumed to range from 0.2 to 0.6 m<sup>2</sup> g<sup>-1</sup> for the soil studied and therefore negligible compared to the *SSA* measured (Table 2). Thus, the external specific surface area (*SSA*) of the clay fraction was calculated (*SSA<sub>cl</sub>*, m<sup>2</sup> per g of oven-dried clay at 105°C) as follows:

$$SSA_{cl} = (SSA/CC) \tag{8}$$

The results showed that  $SSA_{cl}$  ranged from 112 to 178 m<sup>2</sup> g<sup>-1</sup> (Table 5) which are high external specific surface areas for clays, indicating a very small size of the elementary clay particles in the horizons studied (Bergaya & Lagaly 2013; Churchman & Velde 2019).

Compared to the 37 clayey soils studied by Bruand and Tessier (2000), the  $CEC_{cl}$  recorded for the Syrian clayey soils studied were higher. The  $SSA_{cl}$  recorded in this study were also higher than those recorded by Bruand and Tessier (2000) except for the horizons JB 1 and JB 2 for which the  $SSA_{cl}$  were

among the highest recorded by these authors. To extend the range of both the  $CEC_{cl}$  and  $SSA_{cl}$ 214 discussed, the relationship between  $CEC_{cl}$  and  $SSA_{cl}$  was established using both the data published by 215 216 Bruand and Tessier (2000) and those recorded in this study (Figure 2). The closeness of the relationship ( $R^2 = 0.78$ , n = 47) is consistent with the results recorded by Bruand and Zimmer (1992) 217 and Bigorre et al. (2000), indicating that a high proportion of  $CEC_{cl}$  variance was explained by  $SSA_{cl}$ 218 for the soils studied and consequently by the size of the elementary clay particles (Figure 2). Thus, the 219 220 cation exchange capacity of the elementary clay particles appears to be mainly related to the negative charges present on the external surfaces of these particles (Robert et al., 1991; Bergaya & Lagaly 221 222 2013). The high distance to the regression line recorded for the horizons TH 1-2, TH 1-3, OM 1-1 and QM 1-2 indicates that the interfoliar space of the elementary clay particles, which  $SSA_{cl}$  does not 223 account for, contributes more than for the other horizons to their cation exchange capacity (Figure 2, 224 Table 5) (Bergaya & Lagaly 2013; Churchman & Velde 2019). 225

226 Comparison of the XRD patterns with the values of CEC<sub>cl</sub> and SSA<sub>cl</sub> recorded showed the presence of a small amount of smectite for JB 2-1, JB 2-2 and JB 2-3 when CEC<sub>cl</sub> and SSA<sub>cl</sub> were 227 among the highest values recorded (Figure 1, Tables 3 and 5). The low intensity of the peak between 228 229 1.44 and 1.57 nm at 20°C and between 1.72 and 1.77 nm after treatment with ethylene glycol 230 (Figure 1) is in all likelihood related to the weak organization along the crystallographic plane *ab* of 231 the elementary clay particles in the clay deposit on the glass slide because of the very small size of the elementary clay particles as indicated by SSA<sub>cl</sub> (Table 5) (Tessier & Pédro 1987). The proportion of 232 smectite in JB 2-1, JB 2-2 and JB 2-3 is therefore probably higher than that indicated by the XRD 233 patterns (Figure 1) and reported in Table 3. 234

235

#### 236 Pore volume developed by the assemblage of elementary clay particles

Assuming that water was mainly retained by the clay phase whatever the potential, as earlier shown by Bruand and Zimmer (1992), Bruand and Tessier (2000) and D'Angelo et al. (2014), we considered that for the centimetric clods at field capacity the volume of cracks and biopores was negligible compared with the pore volume developed by the assemblage of the elementary clay particles. Thus, the specific volume of clods at the field conditions  $(V_{fc})$  is related to the specific volume of the solid phase  $(V_s, \text{ in cm}^3 \text{ per g of oven-dried soil})$  and to the specific pore volume  $(V_p, \text{ in cm}^3 \text{ per g of oven$  $dried soil})$  as follows:

$$V_{fc} = V_s + V_p \tag{9}$$

245  $V_p$  was calculated for each horizon using  $V_s = 0.377$  cm<sup>3</sup> g<sup>-1</sup> which corresponds to a particle density of 246 2.65 and  $V_{fc}$  which corresponds to the reciprocal of  $D_b$  measured on centimetric clods (Table 2), 247 giving:

248 
$$V_p = (1/D_b) - V_s$$
 (10)

The values of  $V_p$  recorded ranged from 0.671 to 0.840 cm<sup>3</sup> g<sup>-1</sup>. The specific pore volume of the clay ( $V_{p,cl}$ , in cm<sup>3</sup> per g of oven-dried clay) was calculated as follows:

$$V_{p,cl} = V_p/CC \tag{11}$$

The quantity  $V_{p,cl}$ , which can be considered as a quantitative expression of the clay fabric, ranged from 252 0.612 to 0.722 cm<sup>3</sup> g<sup>-1</sup> (Table 5). This is a much smaller range of variation than that recorded for the 253 soils studied by Bruand and Tessier (2000) ( $0.337 \le V_{p,cl} \le 1.484 \text{ cm}^3 \text{ g}^{-1}$ ). However, it is interesting to 254 255 note that although the values of CEC<sub>cl</sub> recorded for the Syrian clayey horizons studied are higher than 256 those recorded for the clayey soils studied by Bruand and Tessier (2000), 6 soils of those studied by the latter showed higher  $V_{p,cl}$  than those recorded in this study, thus illustrating that  $V_{p,cl}$  varies 257 according to the clay mineralogy but also according to the hydric history of the soil (Tessier and 258 259 Pédro, 1984 and 1987). In the same way, even if  $V_{p,cl}$  could theoretically be assumed to be closely related to the size of the elementary clay particles, the results recorded showed the lack of a close 260 relationship between  $V_{p,cl}$  and  $SSA_{cl}$ , again because of the hydric history of the soil (Tessier and Pédro, 261 262 1984 and 1987). This is the case of the Syrian clayey soils studied with a very poor relationship between  $V_{p,cl}$  and  $SSA_{cl}$  ( $\mathbb{R}^2 = 0.24$ ). 263

#### 265 Water retention properties of the clay

Since the difference in water retention properties between the horizons results from the variation in both the clay content and the water retention properties of the clay, the latter were calculated by correcting the gravimetric water content of the horizon (W) for clay content (CC) at different water potentials as follows:

$$W_{cl} = W/CC \tag{12}$$

The quantity  $W_{cl}$  can be considered as the amount of water retained by the pore volume related to the clay fabric ( $W_{cl}$ , in g of water per g of oven-dried clay). Regression equations were established between  $W_{cl}$  and  $V_{p,cl}$  with the data published by Bruand and Tessier (2000) and then with both the data published by these authors and those recorded in this study for Syrian clayey soils (Table 6).

275 Although the range of variation of  $CEC_{cl}$  and  $SSA_{cl}$  was markedly different between, on the one hand, the dataset published by Bruand and Tessier (2000) with CEC<sub>cl</sub> ranging from 0.227 to 0.666 276  $\text{mmol}_+\text{g}^{-1}$  and  $SSA_{cl}$  ranging from 53 to 139 m<sup>2</sup> g<sup>-1</sup> and, on the other hand, the Syrian dataset with 277  $CEC_{cl}$  ranging from 0.707 to 0.891 mmol<sub>+</sub> g<sup>-1</sup> and  $SSA_{cl}$  from 112 to 178 m<sup>2</sup> g<sup>-1</sup> (Table 5), the 278 regression equations established by Bruand and Tessier (2000) (Table 6) were tested with the Syrian 279 dataset to be able to discuss the expected improvement in the prediction quality when the regression 280 equations are established with both datasets (Table 7). Their performance was evaluated using ME, SD 281 and *RMSE*. Results showed that *ME* ranged from  $-0.065 \text{ g g}^{-1}$  at -10 hPa water potential to 0.047 g g<sup>-1</sup> 282 at -1000 hPa water potential (mean =  $-0.008 \text{ g g}^{-1}$ ), SD from 0.042 g g<sup>-1</sup> at -15000 hPa water potential 283 to 0.069 g g<sup>-1</sup> at -330 hPa water potential (mean = 0.057 g g<sup>-1</sup>) and *RMSE* from 0.051 g g<sup>-1</sup> at -3300 284 hPa water potential to 0.091 g  $g^{-1}$  at -10 hPa water potential (mean = 0.068 g  $g^{-1}$ ) (Table 7). 285

Regression equations were established between  $W_{cl}$  and  $V_{p,cl}$  using both the data recorded by Bruand and Tessier (2000) and the data recorded in this study (Table 6). They were tested successively on the two datasets by recalculating them each time without taking into account the horizon on which 289 they were tested. The number of horizons used to establish the regression equations was 46, one being left out when calculating each regression equation. When tested on the dataset published by Bruand 290 and Tessier (2000), results showed that ME ranged from  $-0.011 \text{ g g}^{-1}$  at -330 hPa and -1000 hPa291 water potential to 0.014 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.001 g g<sup>-1</sup>), SD from 0.035 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.001 g g<sup>-1</sup>), SD from 0.035 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.001 g g<sup>-1</sup>), SD from 0.035 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.001 g g<sup>-1</sup>), SD from 0.035 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.001 g g<sup>-1</sup>), SD from 0.035 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.001 g g<sup>-1</sup>), SD from 0.035 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.001 g g<sup>-1</sup>), SD from 0.035 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.001 g g<sup>-1</sup>), SD from 0.035 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.001 g g<sup>-1</sup>), SD from 0.035 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.001 g g<sup>-1</sup>), SD from 0.035 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.001 g g<sup>-1</sup>), SD from 0.035 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.001 g g<sup>-1</sup>), SD from 0.035 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.001 g g<sup>-1</sup>), SD from 0.035 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.001 g g<sup>-1</sup>), SD from 0.035 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.001 g g<sup>-1</sup>), SD from 0.035 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.001 g g<sup>-1</sup>), SD from 0.035 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.001 g g<sup>-1</sup>), SD from 0.035 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.001 g g<sup>-1</sup>), SD from 0.035 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.001 g g<sup>-1</sup>), SD from 0.035 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.001 g g<sup>-1</sup>), SD from 0.035 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.001 g g<sup>-1</sup>), SD from 0.035 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.001 g g<sup>-1</sup>), SD from 0.035 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.001 g g<sup>-1</sup>), SD from 0.035 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.001 g g<sup>-1</sup>), SD from 0.035 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.001 g g<sup>-1</sup>), SD from 0.035 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.001 g g<sup>-1</sup>), SD from 0.035 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.001 g g<sup>-1</sup>), SD from 0 292 3300 hPa water potential to 0.043 g g<sup>-1</sup> at -10 hPa water potential (mean = 0.039 g g<sup>-1</sup>) and RMSE 293 from 0.035 g g<sup>-1</sup> at -3300 hPa water potential to 0.045 g g<sup>-1</sup> at -10 hPa water potential according 294  $(\text{mean} = 0.040 \text{ g g}^{-1})$  (Table 7). When tested on the Syrian clayey soils, results showed a lower 295 prediction performance than when tested on the French dataset, with ME ranging from -0.052 g g<sup>-1</sup> at 296 -10 hPa water potential to 0.037 g g<sup>-1</sup> at -1000 hPa water potential (mean = -0.006 g g<sup>-1</sup>), SD from 297 0.043 at -15000 hPa water potential to 0.070 g  $g^{-1}$  at -330 hPa water potential (mean = 0.058 g  $g^{-1}$ ) and 298 *RMSE* from 0.050 g g<sup>-1</sup> at -3300 hPa water potential to 0.083 g g<sup>-1</sup> at -10 hPa water potential (mean = 299  $0.065 \text{ g g}^{-1}$  (Table 7). Comparison of these values with those recorded when the regression equations 300 were established with the data of Bruand and Tessier (2000) showed that the range of bias for water 301 potential ranging from -10 to -15000 hPa was reduced by 20%, with no variation in the precision 302 303 when the regression equations were established with both the datasets (Table 7). The overall prediction quality was slightly increased, as indicated by the *RMSE* values which were reduced by 304 20% when the regression equations were established using both datasets (Table 7). 305

Thus, when clayey soils with clays developing a high CEC and a hydric history acquired in a semi-arid climate are taken into account, this slightly improves the validity of using  $V_{p,cl}$  to predict the water retention properties of the clay of horizons belonging to soils developed under a larger range of climatic environments than those already studied in the literature.

310

#### 311 Water retention curve of the horizons

312 The regression equations established between  $W_{cl}$  and  $V_{p,cl}$  make it possible to predict the water 313 content at the scale of the horizon since:

314 
$$W_i = CC \times \left[a_i + \left(b_i \times V_{p,cl}\right)\right]$$
(13)

with  $W_i$ , the gravimetric water content in g of water per g of oven-dried soil at the water potential *i*, *CC*, the clay content of the soil in g of clay per g of oven-dried soil,  $a_i$  and  $b_i$ , the coefficients of the regression equations at the water potential *i* (Table 6), and  $V_{p,cl}$ , the specific pore volume of the clay in  $cm^3$  per g of oven-dried clay for the soil at field capacity. The predicted volumetric water content at water potential *i* ( $\theta_i$ , in cm<sup>3</sup> of water per cm<sup>3</sup> of soil) can be calculated as follows:

$$\theta_i = D_b \times W_i. \tag{14}$$

321 Then, by using equations (7), (8) and (11):

322 
$$\theta_i = D_b \times CC \times \left\{ a_i + \left[ b_i \times \left( (1/D_b) - V_s \right) / CC \right] \right\}$$
(15)

323 which becomes:

324 
$$\theta_i = D_b \times CC \times \{a_i + [b_i \times ((1/D_b) - 0.377)/CC]\}$$
(16)

using  $V_s = 0.377 \text{ cm}^3 \text{g}^{-1}$  which corresponds to a particle density of 2.65. The  $\theta_i$  calculated at the different water potentials and for the different horizons were compared to the  $\theta_i$  measured. Results showed *ME* ranging from -0.028 cm<sup>3</sup> cm<sup>-3</sup> at -33 hPa water potential to 0.025 cm<sup>3</sup> cm<sup>-3</sup> at -1000 hPa water potential (mean = -0.005 cm<sup>3</sup> cm<sup>-3</sup>), *SD* from 0.036 cm<sup>3</sup> cm<sup>-3</sup> at -15000 hPa water potential to 0.054 cm<sup>3</sup> cm<sup>-3</sup> at -10 and -33 hPa water potential (mean = 0.046 cm<sup>3</sup> cm<sup>-3</sup>) and *RMSE* from 0.039 cm<sup>3</sup> cm<sup>-3</sup> at -3300 hPa water potential to 0.059 cm<sup>3</sup> cm<sup>-3</sup> at -33 hPa water potential (mean = 0.048 cm<sup>3</sup> cm<sup>-3</sup>) (Table 8).

These results can be compared with those recorded earlier with soils located in semi-arid areas. Tombul et al. (2004) applied early pedotransfer functions established using multiple linear regression to 126 horizons originating from soils developed in a semi-arid area in Turkey. They compared the measured and predicted volumetric water contents at -330 and -15000 hPa and recorded a mean *RMSE* of 0.072 and 0.030 cm<sup>3</sup> cm<sup>-3</sup>, respectively. Mosaddeghi and Mahboubi (2011) studied the water

retention properties of 63 soils located in a semi-arid area of western Iran. They developed 337 pedotransfer functions using multiple linear regression and the comparison of the measured and 338 339 predicted volumetric water content at 11 water potentials between -10 and -15000 hPa led to RMSE ranging from 0.034 to 0.044 cm<sup>3</sup> cm<sup>-3</sup> for the 63 topsoils and from 0.034 to 0.047 cm<sup>3</sup> cm<sup>-3</sup> for the 340 subsoils. More recently, Dharumarajan et al. (2019) established pedotransfer functions using also 341 multiple linear regression for soils developed in a semi-arid area of India. They studied 740 horizons 342 343 originating from 149 soils with varying textures. The RMSE recorded when they compared the measured and predicted volumetric water content at -330 and -15000 hPa was 0.046 and 0.032 344  $cm^3 cm^{-3}$ , respectively. 345

Using multiple linear regression, artificial neural networks and support vector machines, Khlosi et al. (2016) discussed the performance of pedotransfer functions established with Syrian soils which were mainly loam, clay loam and clay soils. They applied their pedotransfer functions to 18 horizons originating from Syrian soils and recorded values of *RMSE* higher than those recorded in this study with RMSE ranging from 0.065 to 0.089 cm<sup>3</sup> cm<sup>-3</sup> (mean = 0.080 cm<sup>3</sup> cm<sup>-3</sup>) with multiple linear regressions, from 0.055 to 0.075 cm<sup>3</sup> cm<sup>-3</sup> (mean = 0.067 cm<sup>3</sup> cm<sup>-3</sup>) with artificial neural networks and from 0.048 to 0.064 cm<sup>3</sup> cm<sup>-3</sup> (mean = 0.058 cm<sup>3</sup> cm<sup>-3</sup>) with support vector machines.

353 As the size of the datasets used in these different studies was highly variable and larger than in our 354 study, the comparison with our results and between the different studies remains limited to the observation that the RMSE values are of the same order of magnitude. In our study, we used only two 355 soil characteristics, the clay content and the bulk density when the soil was close to field capacity, 356 357 when the above studies have used much more soil characteristics. Our results indicate for soils of semi-arid areas the interest of developing pedotransfer functions which take variables closely related 358 to the water retention mechanisms in soils more into account. The values of  $\theta$  calculated with the 359 proposed equations at the different water potentials can be used to fit a model of water retention curve 360 which is usually required to predict plant-available water and more generally, to run plant-soil-water 361 models. Future work will aim at enlarging both the range of textures and of climatic conditions under 362

363 which the clayey soils developed, with the aim of studying the way in which their water retention 364 properties are related to the hydric history.

365

#### 366 Conclusion

367 Our objective was to analyze the water retention properties of clayey soils developed in semi-arid areas of Syria and to discuss how these properties could be explained by the characteristics of the clay. 368 369 Results showed that the clay of the soils studied has both a higher cation exchange capacity and a 370 higher external specific surface area than those recorded for clayey soils studied in temperate areas and used as a reference in this study. The cation exchange capacity and external specific surface area 371 values recorded are consistent with the clay mineralogy, as shown by the X-ray diffraction data which 372 revealed the presence in most horizons of a high proportion of smectite and secondarily a varying 373 proportion of illite and chlorite, while kaolinite was present but not abundant. Results also showed that 374 375 the amount of water retained by the clay at water potentials ranging from -10 to -15000 hPa was 376 closely related to the specific pore volume developed by the elementary clay particles at field capacity. 377 Thus, regression equations established by using both the data earlier published and those recorded in 378 this study were proposed to predict the water retention properties of clayey soils over a larger range of 379 clay mineralogies and climatic environments than previously reported in the literature. The specific pore volume as well as the clay content were found to be the most important predictors of water 380 381 retention properties, the former corresponding to the pore volume developed by the elementary clay 382 particles which varies according to the clay mineralogy and hydric history of the soil. Finally, from a more applied point of view, our results showed the significance of the bulk density measured in 383 conditions close to the field capacity to predict the water retention properties of clayey soils. It should 384 385 be more systematically measured during soil studies.

386

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#### 547 List of Figures

548

549

studied under room conditions (20 °C), after saturation with ethylene-glycol (EG), after heating at 150
°C and 550 °C.
Figure 2. Relation between the cation exchange capacity of the clay (*CEC<sub>cl</sub>*) and the specific surface
area of the clay (*SSA<sub>cl</sub>*) for both the clayey soils studied by Bruand and Tessier (2000) (o) and the
Syrian clayey soils of this study (•).

Figure 1. X-ray diagrams of oriented deposits of the clay material extracted from the clayey horizons

558	List of tables
559	
560	Table 1. Main characteristics of the soils studied.
561	
562	Table 2. Main characteristics of the horizons studied.
563	
564	<b>Table 3</b> . Semi-quantitative evaluation of clay minerals identified by DRX.
565	
566	<b>Table 4</b> . Water content at field conditions $(W_{fc}, g g^{-1})$ and at different water potentials $(W_h, g g^{-1})$ .
567	
568	Table 5. Characteristics calculated for the clay of the Syrian soils studied.
569	
570	<b>Table 6</b> . Regression equations between the water content of clay $(W_{cl})$ at different water potentials and
571	the clay fabric $(V_{p, cl})$ calculated with the data published in tables 1, 2 and 3 by Bruand and Tessier
572	(2000).
573	
574	Table 7. Mean error of prediction (ME), standard deviation (SD) and root mean square error (RMSE)
575	recorded with the regression equations established between the gravimetric water content of clay ( $W_{cl}$ ,
576	in g g <sup>-1</sup> of oven dried clay) at different water potentials and the pore volume developed by the
577	elementary clay particles ( $V_{p, cl}$ , in cm <sup>3</sup> g <sup>-1</sup> of oven dried clay) at field capacity.
578	
579	<b>Table 8.</b> Mean error of prediction (ME), standard deviation (SD) and root mean square error (RMSE)
580	recorded with the regression equations established between the volumetric water content of clay ( $\theta_h$ , in
581	$cm^3 cm^{-3}$ ) at different water potentials h and both the clay content and the reciprocal of bulk density
582	measured in conditions close to field capacity with both the data published by Bruand and Tessier
583	(2000) and those recorded in this study when applied to the horizons of this study.
584	



Figure 1. X-ray diagrams of oriented deposits of the clay material extracted from the clayey horizons
studied under room conditions (20 °C), after saturation with ethylene-glycol (EG), after heating at 150
°C and 550 °C.



**Figure 2.** Relation between the cation exchange capacity of the clay ( $CEC_{cl}$ ) and the specific surface area of the clay ( $SSA_{cl}$ ) for both the clayey soils studied by Bruand and Tessier (2000) (o) and the Syrian clayey soils of this study (•).

**Table 1**. Main characteristics of the soils studied.

Soil location	Coordinates UTM	Soil type Soil Survey Staff (2010) / IUSS Working Group WRB (2015)	Parent material	Land use	Agro-climatic zone
Jabbah (JB 1) Jabbah (JB 2) Al-Suwayda (SW 1) Al-Tha'Ala (TH 1) Al-Qamishli (QM 1)	33.1623N - 35.9368E 33.1520N - 35.9243E 32.7118N - 36.5494E 32.7115N - 36.4860E 41.0324N - 41.1303E	Typic Haploxerept / Hypereutric Cambisol Typic Haploxerept / Hypereutric Cambisol Vertic Chromoxerept / Chromic Vertisol Vertic Xerochept / Chromic Vertisol Calcic Xerocrept / Vertic Calcisol	Clayey surface formations including coarse elements of basic extrusive rocks Calcareous alluvium materials	Meadow Meadow Annual crops Olive trees Annual crops	1 1 2 2 1

605 Agro-climatic zones according to Khlosi et al. (2012): zone 1 is characterized by a mean annual precipitation >300 mm in 2 out of 3 seasons and shrubland, degraded forest, rock outcrops and permanent crops; zone 2 is characterized by a mean annual precipitation of 250-300 mm 2 out of 3 seasons, permanent crops and rainfed field crops with irrigated crops.

Soil Sample	Horizon	Iorizon Depth cm	rizon Depth cm	Texture	$D_b$	Partic	le size distri g.g <sup>-1</sup>	bution	CaCO <sub>3</sub> g.g <sup>-1</sup>	OC g.g <sup>-1</sup>	CEC mmol <sub>+</sub> .g <sup>-1</sup>		Exchangea mmo	ble cations	5	SSA m <sup>2</sup> .g <sup>-1</sup>
					Clay <2µm	Silt 2-	Sand 50-				Ca <sup>2+</sup>	Mg <sup>2+</sup>	Na <sup>+</sup>	$\mathbf{K}^+$		
ID 1 1	٨	5 25	Б	1 / 9	0.42	<u> </u>	2000µm	<0.01	0.008	0.204	0.190	0.086	0.010	0.010	65.0	
JB 1-1 JB 2-1	A	5-25 5-25	F	1.48	0.45	0.34	0.23	<u>≤0.01</u> ≤0.01	0.008	0.304	0.180	0.080	0.019	0.010	03.9 74.7	
JB 2-2	В	30-60	F	1.44	0.46	0.30	0.24	< 0.01	0.004	0.375	0.208	0.113	0.016	0.008	78.9	
JB 2-3	В	60-80	F	1.41	0.46	0.31	0.23	0.02	0.002	0.410	0.209	0.116	0.010	0.007	81.7	
SW 1-1	А	0-25	F	1.26	0.58	0.22	0.20	0.04	0.004	0.442	0.306	0.097	0.008	0.009	83.0	
SW 1-2	В	30-70	VF	1.28	0.65	0.16	0.19	0.05	0.002	0.478	0.333	0.093	0.010	0.007	91.3	
TH 1-2	В	25-50	VF	1.28	0.66	0.15	0.19	0.06	0.003	0.475	0.294	0.119	0.016	0.011	74.6	
TH 1-3	В	50-75	VF	1.25	0.69	0.11	0.20	0.06	0.002	0.492	0.317	0.082	0.024	0.008	77.6	
QM 1-1	А	0-25	F	1.32	0.53	0.25	0.22	0.30	0.006	0.395	0.290	0.082	0.010	0.005	60.8	
QM 1-2	В	25-50	F	1.34	0.57	0.24	0.19	0.31	0.004	0.410	0.320	0.089	0.006	0.003	67.4	
Mean	_	_	_	1.36	0.55	0.24	0.21	0.087	0.004	0.413	0.266	0.098	0.014	0.008	75.6	
SD	_	_	_	0.09	0.09	0.08	0.03	0.111	0.002	0.058	0.059	0.014	0.006	0.002	8.6	
Min	_	_	_	1.25	0.43	0.11	0.19	0.01	0.002	0.305	0.180	0.082	0.006	0.003	60.8	
Max	-	_	_	1.49	0.69	0.34	0.26	0.31	0.008	0.492	0.333	0.119	0.024	0.011	91.3	

Table 2. Main characteristics of the horizons studied. 611

612 613 Texture according to the FAO triangle (FAO, 1990), F: Fine, VF: Very Fine; D<sub>b</sub>, bulk density; CaCO<sub>3</sub>, calcium carbonate content; OC: organic carbon content; CEC, cation exchange capacity; SD,

standard deviation.

# 616 617 **Table 3**. Semi-quantitative evaluation of clay minerals identified by DRX.

Soil		Clay r	ninerals					
Sample	Kaolinite	Illite	Smectite	Chlorite				
JB 1-1	+	+	++	-				
JB 2-1	+	+	+	-				
JB 2-2	+	+	+	-				
JB 2-3	+	+	+	_				
SW 1-1	+	_	+++	+				
SW 1-2	+	_	+++	+				
TH 1-2	+	_	+++	+				
TH 1-3	+	_	+++	+				
QM 1-1	+	++	++	++				
QM 1-2	+	++	++	++				
+++ abundant, ++ 1	+++ abundant, ++ moderately abundant, + not abundant, - not identified.							

Soil	$\mathbf{W}_{fc}$	Water retained at a water potential of hPa (g $g^{-1}$ )							
sample	-	W-10	<i>W</i> -33	W-100	W-330	W-1000	W-3300	W-15000	
JB 1-1	0.318	0.360	0.343	0.303	0.272	0.261	0.242	0.222	
JB 2-1	0.262	0.331	0.301	0.284	0.254	0.241	0.236	0.207	
JB 2-2	0.283	0.328	0.320	0.296	0.261	0.248	0.244	0.224	
JB 2-3	0.292	0.342	0.330	0.317	0.265	0.251	0.245	0.224	
SW 1-1	0.371	0.445	0.439	0.433	0.352	0.320	0.316	0.279	
SW 1-2	0.375	0.447	0.424	0.408	0.340	0.338	0.322	0.279	
TH 1-2	0.472	0.534	0.508	0.468	0.431	0.311	0.302	0.270	
TH 1-3	0.507	0.540	0.532	0.483	0.451	0.423	0.414	0.385	
QM 1-1	0.342	0.402	0.397	0.342	0.299	0.287	0.265	0.249	
QM 1-2	0.344	0.382	0.357	0.341	0.302	0.290	0.264	0.255	
Mean	0.357	0.411	0.395	0.368	0.323	0.297	0.285	0.259	
SD	0.075	0.075	0.076	0.070	0.067	0.052	0.053	0.048	
Min	0.262	0.328	0.301	0.284	0.254	0.241	0.236	0.207	
Max	0.507	0.540	0.532	0.483	0.451	0.423	0.414	0.385	
CD standard da									

**Table 4.** Water content at field conditions  $(W_{fc}, g g^{-1})$  and at different water potentials  $(W_h, g g^{-1})$ .

*SD*, standard deviation.

628	Table 5. Characteristics calculated for the clay of the
629	Syrian soils studied.

Soil	$CEC_{cl}$	$SSA_{cl}$	$V_{p, cl}$	
Sample	$mmol_+.g^{-1}$	$m^{2}.g^{-1}$	$cm^{3}.g^{-1}$	
JB 1-1	0.707	153	0.695	
JB 2-1	0.767	166	0.654	
JB 2-2	0.815	172	0.690	
JB 2-3	0.891	178	0.722	
SW 1-1	0.762	143	0.718	
SW 1-2	0.735	141	0.622	
TH 1-2	0.720	113	0.613	
TH 1-3	0.712	112	0.612	
QM 1-1	0.745	115	0.718	
QM 1-2	0.719	118	0.648	
Mean	0.757	141	0.669	
SD	0.054	24	0.042	
Min	0.707	112	0.612	
Max	0.891	178	0.722	

 $\overline{CEC_{cl}}$ , cation exchange capacity of the clay;  $SSA_{cl}$ , specific surface area of the clay;  $V_{p,cl}$ , specific pore volume of the clay

631 632

**639Table 6.** Regression equations between the water content of clay  $(W_{cl})$  at different water potentials and the<br/>clay fabric  $(V_{p, cl})$  calculated with the data published in tables 1, 2 and 3 by Bruand and Tessier (2000).<br/>Water potentialWater potentialRegression equations calculated:

hPa	Regression equations calculated.									
	with the data publ	ished		with both the data published by	Bruand and	I Tessier				
	by Bruand and Tessie	er (2000)		(2000) and those recorded	l in this stud	ју				
	(n=37)			(n=10)		-				
		R²	Ν		R <sup>2</sup>	N				
-10	$W_{cl} = -0.0184 + 1.0516 V_{p, cl}$	0.96	37	$W_{cl} = -0.0107 + 1.0615 V_{p, cl}$	0.92	47				
-33	$W_{cl} = -0.0150 + 1.0327 V_{p, cl}$	0.96	37	$W_{cl} = -0.0087 + 1.0378 V_{p, cl}$	0.93	47				
-100	$W_{cl} = -0.0310 + 1.026 V_{p, cl}$	0.96	37	$W_{cl} = -0.0244 + 1.0201 V_{p, cl}$	0.94	47				
-330	$W_{cl} = -0.0231 + 0.9779 V_{p, cl}$	0.96	37	$W_{cl} = -0.0135 + 0.9474 V_{p, cl}$	0.92	47				
-1000	$W_{cl} = -0.0027 + 0.8865 V_{p, cl}$	0.96	37	$W_{cl} = +0.0030 + 0.8613 V_{p, cl}$	0.92	47				
-3300	$W_{cl} = +0.0484 + 0.7387 V_{p, cl}$	0.95	37	$W_{cl} = +0.0531 + 0.7239 V_{p, cl}$	0.91	47				
-15000	$W_{cl} = +0.1462 + 0.4269 V_{p, cl}$	0.84	37	$W_{cl} = +0.1479 + 0.4388 V_{p, cl}$	0.77	47				

**Table 7.** Mean error of prediction (*ME*), standard deviation (*SD*) and root mean square error (*RMSE*) 651 recorded with the regression equations established between the gravimetric water content of clay ( $W_{cl}$ , in 652 g g<sup>-1</sup> of oven dried clay) at different water potentials and the pore volume developed by the elementary 653 clay particles ( $V_{p, cl}$ , in cm<sup>3</sup> g<sup>-1</sup> of oven dried clay) at field capacity.

Water	<i>ME</i> , <i>SD</i> and <i>RSME</i> recorded with the regression equations established:										
potential	with th	e data publ	ished by	with both the data published by Bruand and Tessier (2000)							
hPa	Bruand	and Tessie	er (2000)	a	nd those re	ecorded in	this study wh	en applied t	o:		
	wl	nen applied	l to:								
		(n=37)				(r	n=46)				
	the hor	rizons of th	is study	the ho	orizons stu	died	the ho	rizons of this	s study		
				by Bruand	and Tessi	er (2000)					
		(n=10)			(n=37)			(n=10)			
	ME	SD	RMSE	ME	SD	RMSE	ME	SD	RMSE		
-10	-0.065	0.067	0.091	+0.014	0.043	0.045	-0.052	0.068	0.083		
-33	-0.044	0.066	0.077	+0.010	0.040	0.040	-0.035	0.067	0.073		
-100	-0.014	0.058	0.056	+0.002	0.040	0.039	-0.011	0.059	0.057		
-330	+0.044	0.069	0.079	-0.011	0.041	0.042	+0.034	0.070	0.075		
-1000	+0.047	0.050	0.067	-0.011	0.037	0.038	+0.037	0.051	0.061		
-3300	+0.021	0.049	0.051	-0.005	0.035	0.035	+0.017	0.050	0.050		
-15000	-0.043	0.042	0.058	+0.011	0.040	0.041	-0.034	0.043	0.053		
Mean	-0.008	0.057	0.068	0.001	0.039	0.040	-0.006	0.058	0.065		
Min	-0.065	0.042	0.051	-0.011	0.035	0.035	-0.052	0.043	0.050		
Max	+0.047	0.069	0.091	+0.014	0.043	0.045	+0.037	0.070	0.083		

661	Table 8. Mean error of prediction (ME), standard deviation (SD) and root mean square
662	error (RMSE) recorded with the regression equations established between the
663	volumetric water content of clay ( $\theta_h$ , in cm <sup>3</sup> cm <sup>-3</sup> ) at different water potentials h and
664	both the clay content and the reciprocal of bulk density measured in conditions close
665	to field capacity with both the data published by Bruand and Tessier (2000) and those
666	recorded in this study when applied to the horizons of this study.

Water potential	n=10		
(hPa)			
	$\frac{ME}{(\text{cm}^3/\text{cm}^3)}$	$\frac{SD}{(\text{cm}^3/\text{cm}^3)}$	$\frac{RMSE}{(cm^3/cm^3)}$
-33	-0.028	0.054	0.059
-100	-0.011	0.046	0.045
-330	+0.021	0.053	0.054
-1000	+0.025	0.039	0.044
-3300	+0.010	0.039	0.039
-15000	-0.025	0.036	0.042
Mean	-0.005	0.046	0.048
Min	-0.028	0.036	0.039
Max	+0.025	0.054	0.059