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**Geochemical characteristics of rare earth elements in sediments of
the North China Plain: implication for sedimentation process**

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Abstract: Rare earth elements (REE) are powerful tracers for our understanding of sedimentary provenance and depositional processes. The present study investigated geochemical characteristics of REE in 226 sediment samples collected from piedmont, central and littoral plains of the North China Plain. Results showed that average total REE concentrations exhibited a decreasing trend as piedmont > central > littoral, and 82%, 70% and 67% samples had total REE concentrations higher than the value of Upper Continental Crust (UCC). The variation in REE concentrations was associated with sedimentary provenance and sediment physicochemical properties. Ternary diagrams of sandstone classification and $\text{Al}_2\text{O}_3\text{-CaO+Na}_2\text{O-K}_2\text{O}$ reflected that sediments originated from greywackes and they have been experienced an incipient to intermediate chemical weathering with piedmont sediments to a higher degree. Average Chemical Index of Alteration values are 68, 51 and 55 in piedmont, central and littoral sediments, respectively. Sediment North American Shale Composite (NASC)-normalized patterns were characterized by light REE (LREE) enrichment over heavy REE (HREE) and coherent negative Eu anomalies (Eu/Eu^* ranging between 0.57 and 1.00). Value of $(\text{La/Yb})_{\text{NASC}}$ ranged from 0.86 to 1.15 (average 1.23), and from 1.03 to 1.49, and from 0.91 to 1.49 in three zones, indicating REE fractionation from piedmont to littoral occurred over sedimentation processes. A positive correlation between $(\text{La/Yb})_{\text{NASC}}$ and total REE concentrations was observed in piedmont and central sediment samples, implying that LREE were preferentially enriched over HREE with an accumulation of REE. While this correlation was not found in littoral sediment samples, where a general decreasing trend along depth for

43 REE concentrations existed. Results of present study shows implications for using
44 REE as a tracer of sedimentary provenance from a basin scale.

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46 Keywords: Lanthanides; Geochemistry; Sedimentation; Fractionation; Critical Zone

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1. Introduction

Rare earth elements are the lanthanides with atomic numbers ranging from 57 (La, lanthanum) to 71 (Lu, lutetium) (Henderson, 1984) as well as yttrium (Y, atomic number 39) and scandium (Sc, atomic number 21; Tyler, 2004). The geochemical signatures of rare earth elements and yttrium (thereafter denoted REE) in sediments and sedimentary rocks are powerful tracers in delineating the processes that govern them. These include magmatism, chemical weathering, diagenesis, erosion and water-rock interactions (McLennan, 1989). As a result, the concentrations and fractionations of REE in sediments and soils have received considerable attention in the past few decades.

Numerous investigations have been devoted to studying geochemical behaviors and mobility of REE in different sediments and soils, including lake, river and marine sediments (Pourret and Tuduri, 2017; Xu et al., 2018, Wang et al., 2018; Consani et al., 2020), mining surroundings (Hu et al., 2004; Wang and Liang, 2015), wetlands (Davranche et al., 2011; Cheng et al., 2012), agricultural areas (Silva et al., 2015; Han et al., 2017), alluvial deposits (Pédrot et al., 2015; Xie et al., 2014), unconsolidated clay sediments (Guo et al., 2010; Sá Paye et al., 2016) and sedimentary rocks (McLennan, 2001; Šmuc et al., 2012). It's recognized that REE concentrations and distributions in sediments depend on sedimentary provenance and sediment's physicochemical properties. Significant correlations between REE concentrations and contents of clay and metals (i.e., Fe, Mn, Al) were observed by Mihajlovic and Rinklebe (2018); while pH value and cation exchange capacity were shown not

important factors in their study. Grain size impacted REE concentrations in sediment as well. This is indicated by the results of Kimoto et al. (2006), showing that decreasing grain size generally increased REE concentrations due to that clay minerals could host REE (Laveuf and Cornu, 2009). More specific, the impact of clay, silty and sand fractions on solid REE was recently investigated in different soil profiles (Mihajlovic and Rinklebe, 2018). Presence of Fe/Mn oxides was another factor influencing REE in sediments/soils, and more REE can be accumulated by amorphous Fe/Mn oxides as compared to crystalline ones (Yan et al., 1999; Compton et al., 2003).

Constituents of REE in sediments and soils for a large part result from natural processes. A global distribution pattern of REE in the soils/sediments was recently evaluated by Adeel et al. (2019), which was achieved by spatially visualized the distribution of the REE concentrations in soils/sediments derived from all over the world. Results obtained from the study showed that total REE concentrations of agriculture soils/sediments ranged from 83 to 9840 mg/kg and concentrations of light REE (LREE) were higher than heavy REE (HREE). This indicates that, in addition to the natural processes, anthropogenic inputs contribute to the REE concentrations in sediments/soils in some individual regions. Indeed, fertilization with REE-bearing fertilizers and infiltration of residuals was suggested to be the main mechanism of agriculture-sourced REE entering into soils and sediments (Hu et al., 2004). However, mechanisms for REE mobilization in different soils/sediments and the potential risks of long-term exposure to REE-abundant environment are not adequately understood.

Average REE concentration in China's soil was established to be 177 mg/kg (Liang et al., 2005), being comparable to the value (189 mg/kg) in the earth's crust (Wei et al. 1991). China is one of the countries that early widely applied REE in various sectors (Pang et al., 2002; Hu et al., 2006), such as agriculture, the REE have been used as a prompter for crops since 1990s. This has led to high REE pollution levels (i.e. La >80; Nd >85; Sm >45) (Adeel et al., 2019; and references therein). Indeed, sediment and soil contaminations caused by elevated REE have been reported in various regions across China, including in Poyang lake (Wang et al., 2018), Baotou city (Li et al., 2010), Tibet plateau (Wu et al., 2018), Jiaozuo Bay (Zhang and Gao, 2015), Sanjiang Plain (Cheng et al., 2012) and Liaodong Bay (Lin et al., 2013). The North China Plain (NCP) is one of the largest sedimentary basin in Asia (Xing et al., 2013), and is provincial politics, economy and culture center of China. The plain is characterized by large population (> 0.35 billion) and highly-developed agricultural industry, which plays an important role in the nation's grain manufacture (Chen and Ni, 1987; Kendy et al., 2004). The densely-populated and industrialized areas were shown to be accompanied by increased discharge of REE into environment (i.e., Gd and Sm) (Kulaksız and Bau, 2013; Hatje et al., 2016), due to REE applications in industry, medicine as well as agriculture. Studies concerning REE concentrations in sediments of the NCP have been performed (Wei et al., 2010; Liu et al., 2016; Zhang and Gao, 2015; Liu et al., 2019a, b), but a full of investigation of mobility, distributions and behaviors of REE in sediments is still lacking. This would hinder a systematical evaluation of REE levels as well as the possible anthropogenic impact,

which is vital for establishing a baseline of distribution and background of REE concentration in the contaminated sediments and for administrating regulation guidelines (Alfaro et al., 2018; Consani et al., 2020). Therefore, the objectives of this study are to: (i) investigate REE concentration and fractionation characteristics in sediments collected from three zones; (ii) characterize the relationships between REE distribution and sediment properties; (iii) provide dataset concerning REE geochemistry for future comparative studies in the region.

2. Material and methods

2.1 Study area

The North China Plain (NCP) is located in the eastern China with longitude of 112°30'~119°30'E, and latitude of 34°46'~40°25'N. The terrains are mostly plains with two mountainous regions (Yanshan and Tanghang mountains) bounding to the north and south, and an oceanic region (Bohai Bay) to the east (Chen et al., 1999), respectively. The plain has a total area of approximately 31×10^4 km² and population of about 0.35 billion inhabitants. The NCP has a semi-arid humid climate with a temperature range of -23 °C~42.7 °C (in history). Average annual rainfall ranges from 500 mm to 600 mm, mostly (>74%) dominating in summer. Average annual evaporation is 1.5 to 3 times of annual rainfall.

The NCP is one of largest sedimentary basins in Asia (Xing et al., 2013). According to Chen et al. (1999), the NCP's geological strata are comprised of four stratigraphic areas: Yinshan-Nuruerhu, Yanshan, Taihangshan and Hebei, with ages

135 ranging from the oldest Archaeozoic to the newest Holocene. Formations in study
136 area have been declining since the Cenozoic due to neotectonics. Geographically, the
137 plain generally decreases from the north to the east with a slope of 0.3‰~0.6‰, and
138 thus it's divided into three zones: piedmont alluvial-proluvial plain, central
139 alluvial-lacustrine plain and littoral plain.

140 Surface sediments in the NCP mainly are Cenozoic deposits with ages from
141 Paleogene to Holocene. Lower Tertiary stratum is composed of red mudstone with red
142 and gray sandstone being interbedded. Upper Tertiary strata are characterized by
143 consolidated to semi-consolidated alluvial-diluvial deposits, including siltstone,
144 mudstone, sandstone and pebbly sandstone. Quaternary sediments are divided into
145 four groups from top to bottom: Holocene (Q₄), Upper (Q_{3m}), Middle (Q_{2c}) and
146 Lower (Q_{1n}) Pleistocene (Chen, 1999). The Holocene sediments group has a
147 thickness of 5 m to 40 m, and mainly consists of yellow and gray sandy loam, fine
148 sand and clay being interbedded with sandy gravel. The Upper Pleistocene is alluvial,
149 diluvial and eolian losses and yellow loam. The Middle Pleistocene strata are
150 widely distributed in intermountain basin and foothills, and consist of red and brown
151 clay, loam and sandy loam, and red mud pebble layers. The Lower Pleistocene
152 sediments group has a buried depth of 350 m to 500 m with a thickness of 80 m to
153 100m, and it's mainly loam, sand loam and lenticular sand layers.

154 Piedmont alluvial-proluvial sediments are suggested to derive from Taihang and
155 Yanshan mountains by weathering, transport and sedimentation process. Central
156 alluvial-lacustrine sediments are deposited from materials carried by Yellow River

and Zhangweinan River running from SW to NE since middle Pleistocene (Shao et al., 1989). The littoral plain forms with interactions between terrene and ocean, and sedimentary deposits come from terrestrial rivers draining into Bohai Bay eventually.

2.2 Sediment sampling

Sediment samples were taken from three boreholes (Y09, S30 and H02) drilled in three different zones (Fig. 1) of the NCP in the July of 2013. The core sediments were sampled with an interval generally ranging between 1 m and 2 m. Slurries and muds attached on the surface of the core sediments were scrapped off to avoid possible cross contaminations. The samples were parceled with silver paper and were stored in clean plastic bags which were immediately filled with pure N₂ gas for further preservation. All the sealed core sediments were placed in a shading environment with temperature below 4 °C during sampling. Finally, the samples were transported to laboratory and preserved in a freezer before analysis.

2.3 Sediment analysis

All samples used for major and minor constituents analyses were dried under room temperature in the lab, and were ground to grain size <74 µm with an agate mortar for analysis. The X-ray fluorescence (XRF) (ARL Advant X) technique was used for measurement of various oxides (SiO₂, Fe₂O₃, Al₂O₃, MgO, CaO, Na₂O, K₂O, MnO, P₂O₅ and TiO₂) following the national standard method (GB/T 14506). Sediment samples were prepared with a glass flux sheet method. Specifically, 0.7 g (±0.001) sample was firstly placed in a 25 mL porcelain crucible; followed by

addition of 5.2 g (± 0.001) anhydrous lithium tetraborate, 0.4 g (± 0.001) lithium fluoride and 0.3 g (± 0.01) ammonium nitrate, the mixture was evenly stirred. Then the mixture was transferred to a platinum (95%)-gold (5%) alloyed crucible for fusing 10 to 15 mins under temperature 1150 to 1250 °C. Finally, fusant was made into glass beads using a fusion machine. All glass beads were preserved in a clean and dry environment before analysis. Reference materials GSS (shown below) were prepared with the same method for subsequent calibration of XRF. The sample intensity (I_s) was calculated by the difference between peak (I_p) and background (I_b) values, as shown with equation (2-1).

$$I_s = I_p - I_b \quad (2-1)$$

Trace elements including REE were quantified by employing ICP-MS (AG 7500) in the digested samples. The samples were digested following the method previously described in Liu et al. (2016). Briefly, 9 mL concentrated HCl (37% v/v), 2 mL HNO₃ (65% v/v) and 9 mL HF (40% v/v) were used for digesting 0.5 g dried samples for 24 h under 108 °C condition. Then the mixture was dried and dissolved again with purified HNO₃ (2% v/v). The obtained solutions were used for trace element analysis. To check the stability of the equipment system and the analytical accuracy, reference sample including GSS1, GSS2, GSS3 and GSS5, obtained from the Center of China National Standard Reference Material, were ran simultaneously. The recommended and test values of REE and metal oxides in standard references were shown in supplementary materials (Table S1). Furthermore, the parallel sample was performed every ten samples for further monitoring of the analytical precisions. The results show

that elements mostly had accuracies (RSD, $\delta 1$) ranging between 2% and 5%. The standard materials had accuracy generally better than 3%, except for P₂O₅, which had an accuracy better than 8% in some samples. The analytical precisions of REE in all samples were generally better than 5%.

2.4 REE normalization

Rare earth element concentrations in all sediment samples were normalized with average REE concentrations of the North American Shale Composite (NASC). The corresponding normalization REE values were taken from Taylor and McLennan (1985). Fractionation between LREE and HREE was quantified using fractionation measures [(La/Lu)_{NASC} and (Nd/Yb)_{NASC}]. Anomalies including Ce (Ce/Ce*) and Eu (Eu/Eu*) were calculated by extrapolation using normalized values of neighboring REE. The corresponding formulas were shown with equations (2-2) to (2-4).

$$\text{Ce/Ce}^* = \text{Ce}_{\text{NASC}} / (\text{La}_{\text{NASC}} \times \text{Pr}_{\text{NASC}})^{0.5} \quad (2-2)$$

$$\text{Eu/Eu}^* = \text{Eu}_{\text{NASC}} / (\text{Sm}_{\text{NASC}} \times \text{Gd}_{\text{NASC}})^{0.5} \quad (2-3)$$

$$\text{Gd/Gd}^* = \text{Gd}_{\text{NASC}} / (\text{Eu}_{\text{NASC}} \times \text{Tb}_{\text{NASC}})^{0.5} \quad (2-4)$$

where the superscript * indicates the geogenic background.

3. Results and discussion

3.1 Sediment geochemistry and provenance

Sediment lithology in three boreholes are mainly clay, silt and silty clay (Fig. 2).

Fine and course sands were generally interbedded with silt and silty clay at different

sampling depths. This is better observed in the central (S30) and littoral (H02) borehole bar graphs (Figs. 2b and c). In piedmont borehole, a gravel layer occurred at depth between 38 m and 61 m (Fig. 2a). Gravel was not found in boreholes S30 and HH02. Overall, fine grains prevailed in central and littoral sediment samples. Colors of the fine grains changed in three zones as well. Pale yellow and pale brownish-yellow clay and silty were mainly distributed in piedmont sediments. For central borehole sediments, the color changed from pale brownish-yellow to brown and tan vertically. For littoral sediments, taupe and tawny silty and silty clay were mainly in upper and middle parts, and pale brownish-yellow and pale brown in lower part (Fig. 2c).

The component of sediment samples was dominated by SiO_2 (Table 1). For piedmont sediments (borehole Y09), contents (wt%) of SiO_2 ranged from 55.55 to 78.36 with an average value of 66.77. Central (borehole S30) and littoral (borehole HH02) sediments had SiO_2 contents (wt%) lower than those of piedmont, being in the ranges of 35.71 to 71.56 (average 56.62) and 32.04 to 77.03 (average 59.51), respectively. Aluminum oxide (Al_2O_3) had the highest contents among the determined metal oxides, with ranges of 6.37 wt% to 14.14 wt%, 7.42 wt% to 13.82 wt% and 7.32 wt% to 15.95 wt% and average values of 10.47 wt%, 10.95 wt% and 12.18 wt% in piedmont, central and littoral sediments, respectively. Iron oxide (Fe_2O_3) was the second abundant metal oxides, with a range from 1.86 wt% to 9.86 wt% and an average of 5.46 wt% in piedmont sediments. The average content (wt%) of Fe_2O_3 in central and littoral sediments was 4.45 wt% and 4.06 wt%, respectively, exhibiting

slightly lower than that of piedmont sediments. Other metal oxides including Na_2O , K_2O , CaO , MgO , P_2O_5 and TiO_2 showed relatively lower content in most of the measured sediment samples as compared to SiO_2 and Al_2O_3 (Table 1). Average value (wt%) of loss on ignition (LOI) was 3.88, 8.56 and 8.02 in piedmont, central and littoral zones with ranges of 1.13 to 7.36, 3.12 to 19.91 and 2.5 to 17.56, respectively.

Changes of sediment lithology were related to the sedimentary provenances and processes (McLennan, 2001). Piedmont sediments were thought to be derived from piedmont regions (Taihang and Yanshan Mountains), where coarse-grained clastic rocks were widely distributed (Zhang et al., 2018). Occurrences of central sediments were attributed to fluviation of Yellow river and rivers springing in Taihang Mountain (Zhang et al., 2018). This would account for sediments being deposited alternately with clay, silty clay and fine sand (Fig. 2). Terrigenous clastics deposition and the impact of ocean led to the formation of littoral sedimentation. Weathering and erosion of the bedrocks around the piedmont plain and subsequent transport and sedimentation contributed to the NCP sediments. Ternary diagram of sandstone classification (Blatt et al., 1980) showed that the majority of sediment samples were located in the left area, and three central samples in the right area, indicating that sediments originally mostly were greywacke (Fig. 3a). The dominant SiO_2 contents and fine-grained components reflected that the sedimentary materials had been undergone intensive denudations and a long-distance transport before sink. Indeed, the ternary diagram of Al_2O_3 - CaO + Na_2O - K_2O (A-CN-K) showed that the weathering trend of the sediment samples were mostly along the axis of $(\text{CaO}+\text{Na}_2\text{O})$ (Fig. 3b).

This probably indicates that sediment samples from the NCP were in an early weathering stage where minerals such as plagioclase was incongruently decomposed in association with leaching of Na and Ca generating products of illite, kaolinite and smectite (Nesbitt et al., 1980). The offset to the uppermost of K₂O axis was attributed to the addition of terrigenous detrital in the later sedimentary stage (Cox et al., 1995).

To quantify the weathering intensity in different samples, the chemical index of alteration (CIA) values (initially was valid for granitic rocks) were calculated following the method proposed by Nesbitt and Young (1982) [$CIA = (Al_2O_3 / (Al_2O_3 + K_2O + Na_2O + CaO^*)) \times 100$; $CaO^* = [CaO - (10/3 \times P_2O_5)]$ (Panahi et al., 2000)].

Chemical index of alteration can be an index reflecting sample in a weathering trend from fresh status to intensely-altered status. Granites and granodiorites usually have CIA values ranging from 45 to 50, and fresh basalts have values between 30 and 45. Muscovite has a value of 75, and illite, montmorillonites and beidellites give values from 75 to 85 (Nesbitt and Young, 1982). Results showed that CIA value of piedmont sediment samples ranged from 55 to 78 with an average value of 68, and central and littoral samples ranged from 37 to 70 (average 51), and from 43 to 70 (average 55), respectively. This indicates that sediment samples were incipiently (CIA: 50 to 60) to intermediately (CIA: 60 to 80) chemically weathered, according to the classification standards suggested by Fedo et al. (1995). It's also observed that piedmont sediment samples generally had higher CIA values as compared to those of central and littoral sediments, which means a stronger weathering may have occurred in the piedmont sediment samples. These differences probably were associated with the

hydrogeological conditions (i.e. groundwater residence time, outcrop lithology and aquifer replenishment condition). The results were consistent with piedmont samples being located in the upper part of the A-CN-K diagram, while central and littoral sediment samples below the piedmont samples (Fig. 3b). Similar studies have been performed to investigate the impact of chemical weathering on sedimentary processes based on sediment geochemical compositions (LaMaskin et al., 2008; Dostal and Keppie, 2009; Xie et al., 2014). Conclusions have been made that sediment geochemistry could serve as a good proxy for understanding the origins and processes of sedimentation, as shown in the investigation.

3.2 REE distribution characteristics

3.2.1 REE concentrations

Total REE (Σ REE) concentrations were shown in Table 1. Piedmont sediment (borehole Y09) Σ REE ranged from 190 mg/kg to 354 mg/kg with an average value of 227 mg/kg. Sediments from central (borehole S30) and littoral (borehole H02) zones had Σ REE concentrations ranging from 89 mg/kg to 291 mg/kg (average 183mg/kg) and from 87.90 mg/kg to 238 mg/kg (average 177 mg/kg), respectively. Generally, sediment Σ REE concentrations showed a decreasing trend (piedmont > central > littoral) along groundwater flow direction. The average Σ REE concentrations in littoral sediment samples from study area were comparable to the values of Upper Continental Crust (UCC) (169 mg/kg) (McLennan, 2001) and Chinese soil (177 mg/kg) (Wei et al., 1991). While average Σ REE concentrations of littoral and central

sediments were higher than the values of UCC and soil in China. All measured average REE concentrations were within the levels (166 mg/kg to 222 mg/kg) reported in similar researches performed in the same study area (Lan et al., 2016). Others showed that core sediments' Σ REE concentrations were slightly higher than the value of surface sediment (Liu et al., 2019b).

Sediment Σ REE concentrations varied along depth. In piedmont borehole, Σ REE were relatively constant within 11 m below sea level. A small decrease was observed at depth from 11 m to 16 m, before Σ REE concentrations increased to >300 mg/kg at 25 m. After a new decrease until about 63 m, Σ REE concentrations fluctuated for the remaining sampling depths (Fig. 2a). Σ REE concentrations in central sediments were highly variable along borehole, although a decreasing trend occurred from land surface to 30 m. Vertical changes of Σ REE concentrations were mainly related to sediment physicochemical properties. Sediment with different grain sizes, such as clay, silt and silty clay, had different capacities in hosting REE. The major minerals in the aquifer sediments of study area included quartz, K-feldspar, hornblende and calcite accounting for 5%-14% in the piedmont (Chen and Ni, 1987). Clay minerals (i.e. kaolinite, illite and montmorillonite) were more commonly distributed in the central and littoral plains as compared to the piedmont plain (Chen et al., 2005). In piedmont where fine sand prevailed (i.e. from 38 m to 62 m), Σ REE concentrations progressively decreased (Fig. 2a). Low Σ REE concentrations of central sediments were found at depth of approximately 30 m, 65 m to 68 m, 107 m, 151 m to 154 m, where fine sands were sandwiched with silt and silty clay (Fig. 2b). Silt sand and fine

sand more commonly occurred at deeper depth, leading to a decreasing trend of Σ REE concentrations (Fig. 2c). This demonstrates that presence of clay was one of factors controlling Σ REE concentrations. REE were thought to be hosted by clay component containing in the sediments (Yang et al., 2002). The disparities of clay content along depth for a large part could account for REE variations. This was further supported by the vertical variations of Si/Al and Ti/Al values, which were proxies for grain size classification (i.e. clay, silt, and silty clay) (Fig. S2). Statistical analyses showed that Σ REE concentrations were negatively correlated to SiO_2 and were significantly positively correlated to Fe_2O_3 and Al_2O_3 (except for littoral sediment samples) (Table 2), reflecting that REE mainly adsorbed onto/incorporated into Fe and Al -containing oxides/clay minerals while repelled SiO_2 . Moreover, trends of $\text{SiO}_2/\text{Al}_2\text{O}_3$ values were generally consistent with those of CIA, suggesting that silicate weathering like feldspar played a role in this process (Fig. 2). In other words, in addition to clay content, Fe and Al-bearing oxides could be another factor influencing REE concentrations in investigated sediment samples. Metal oxides containing Fe, Mn and Al provides important binding sites for sorption of REE (Ohta and Kawabe, 2000; Pourret and Davranche, 2013; Liu et al., 2017; Mihajlovic et al., 2019). These findings are in line with the results documented by Wang et al. (2014), showing that fine fractions with abundant clay minerals and Fe/Al/Mn-containing oxides tended to accumulate REE, while coarse fractions with less clay minerals and more quartz led to a dilution of REE. However, further information on mineralogical compositions of sediment samples is needed for better constraining the impact of mineralogy on REE

concentrations from a quantitative perspective. In all, sediment REE concentrations were essentially regulated by their genetic properties and were revealed by their vertical variations.

3.2.2 REE fractionation patterns

All sediment samples had relatively flat NASC-normalized REE patterns with a gentle enrichment of LREE over HREE (Fig. 4). The degree of LREE enrichment was quantified by $(La/Yb)_{NASC}$ values, which ranged between 0.86 and 1.15 with an average of 1.23 in piedmont sediments. Central and littoral sediment samples had $(La/Yb)_{NASC}$ ranges of 1.03 to 1.14 (average 1.21) and 0.91 to 1.49 (average 1.09), respectively (Table 1). Sediments mostly showed negative Ce anomalies with Ce/Ce^* ranging between 0.69 and 1.17 (average 0.91), and between 0.85 and 0.98 (average 0.90), and between 0.80 and 0.98 (average 0.89) in piedmont, central and littoral plains, respectively. Negative Eu anomalies commonly occurred in investigated sediments with Eu/Eu^* ranges of 0.82 to 1.07 (average 0.91), 0.83 to 1.31 (average 0.95), and 0.86 to 1.34 (average 0.95) in three zones, respectively. Correlation analysis indicated that ΣREE concentrations generally increased with an increase of $(La/Yb)_{NASC}$ in piedmont and central sediments, while this trend was not observed in littoral sediment samples (Fig. 5a). These results reflected REE fractionation occurred with the accumulation processes. The correlation between $(La/Yb)_{NASC}$ and CIA values showed that the larger $(La/Yb)_{NASC}$ values occurred in samples containing higher REE concentrations and generally having higher CIA values (Fig. 5b). Thus, LREE enrichment could result from the chemical weathering of silicate in crustal

materials leading to an increase in LREE concentrations in sediments (Oliveira et al., 2003; Lin et al., 2013). Larger CIA values were obtained in piedmont and central zones as compared to the values in littoral zones (discussed above). The different domain of littoral sediment samples from their piedmont and central counterparts may be related to impact of ocean and/or human activities as well. However, more studies are needed to verify this speculation.

Sediment samples mostly had negative Eu anomalies (Fig. 5d). Average Eu/Eu^* anomalies were quite similar in three different zones. This indicated that sedimentary processes, which evolved weathering of bedrocks around plains and migration and deposition in the flat plains, did not fractionate Eu to a larger extent with respect to its neighboring Sm and Gd. Early investigations demonstrated that Eu compositions and distributions in continental crust were fundamentally controlled by shallow, intracrustal differentiation evolving Eu-bearing minerals (i.e. plagioclase) (McLennan, 2001), which was shown to possess 4% to 23% (wt) in piedmont sediments (Liu et al., 2016). However, Eu/Eu^* values were negatively correlated to ΣREE concentrations (Fig. 5d), which suggested that evolution of Eu anomalies was related to changes of redox conditions as well during sedimentation. Preferential mobilization of Eu(II) over other trivalent REE leads to a segregation of Eu from lanthanide series under reducing conditions (Lee et al., 2003), and hence higher ΣREE concentrations occur in lower- Eu/Eu^* samples, as shown in Figure 5d. This mechanism has been proposed for using Eu anomalies as an indicator for investigating REE origins as well as mobility of redox-sensitive metal(oid) (Guo et al., 2010; Davranche et al., 2011).

Another possibility resulting in negative correlation between Eu anomalies and Σ REE concentrations would be related to minerals contained in sediments. Such as plagioclase, which tends to preferentially accumulate Eu(II) over trivalent REE by means of isomorphism with Ca^{2+} , Na^{+} and K^{+} (Banks et al., 1999).

Cerium anomalies (Ce/Ce^{*}) were relatively constant in central and littoral sediments despite the variable Σ REE concentrations, while they were distributed scattered as a function of Σ REE concentrations in piedmont sediment samples (Fig. 5c). The fluctuation of redox conditions could account for this phenomenon due to fact that Ce is readily to be precipitated as insoluble Ce(IV) via oxidative scavenging, and when conditions evolve into reducing, the precipitated Ce will be released into solution by reductive dissolutions (Bau and Koschinsky, 2009; Steinmann and Stille, 2008). Cerium cycle was believed to be biologically mediated (Moffett, 1990; Tanaka et al., 2010) and evolved metal (i.e. Fe, Mn) oxides (Yu et al., 2017). Results of our recent study showed that Ce precipitation from oxic aqueous solution was more likely to occur in the piedmont aquifers where low-Fe/Mn-concentration groundwater prevailed (Liu et al., 2019a), while no pronounced positive Ce anomalies were commonly observed in the piedmont aquifer sediments. Thus, both positive and negative Ce anomalies, and no Ce anomalies can be found in solid phases in nature, as the results shown in present study.

3.3 Implication for REE signatures

Statistical analyses showed that individual REE concentrations were significantly correlated with correlation coefficient (R^2) all better than 0.8. These results, together

415 with the findings that all investigated sediment samples had coherent normalized
416 patterns, indicated that sediments had similar sources/inputs of REE as well as
417 common geochemistry. The remarkable uniformity of sedimentary REE patterns, with
418 absolute concentration ranging from La to Lu, slight LREE enrichment and common
419 negative Eu anomalies, implied that REE were migrated and transported as an integral
420 whole during chemical weathering and sedimentation processes including erosion,
421 sedimentary sorting and diagenesis. The sediment paralleled NASC-normalized REE
422 patterns suggested that REE information recorded in sediments could serve as an ideal
423 tool for studying crustal evolution and sedimentary provenance. The higher REE
424 abundances in sediment than the values of UCC reflected that an enrichment of REE
425 occurred during the process of sedimentation. Since sedimentary REE uniformity was
426 thought to result from effective mixing of various provenance components from the
427 upper crust (McLennan, 2001) and they were transported primarily in particulates, the
428 sedimentary origins and secondary mobilization would be the plausible cause for
429 higher REE abundance. It should be noted that, in addition to natural processes, REE
430 can be accumulated in sediments and soils by various ways including sewage
431 discharges (Verplanck et al., 2005; Hatje et al., 2016), agricultural activities (Alfaro et
432 al., 2018) and atmospheric depositions (Tyler, 2004). Positive REE anomalies will
433 show up in REE normalized patterns when individual elemental concentration is
434 greater than its background value, as for Gd anomalies that have been reported
435 worldwide (Ebrahimi and Barbieri, 2019). However, the sediments cannot suffer from
436 anthropogenic pollutions in this study since no pronounced positive REE anomalies

were observed in their normalized patterns. In summary, sedimentary REE patterns conveys important information regarding the composition of the continental crust and anthropogenic impact.

4. Conclusions

226 sediment samples were collected from three different zones (i.e. piedmont, central and littoral) of the north China plain (NCP) to investigate the concentrations and fractionations of rare earth element (REE). Sediment geochemistry indicated that sediments mostly were derived from greywacke, as suggested by triangular diagram of sandstone classification. All sediment samples had been undergone incipient to intermediate weathering with average CIA of 68, 51 and 55 in three zones, respectively. Total REE concentrations ranged from 190.24 to 353.87 mg/kg (average 228.62 mg/kg), and from 88.85 to 290.80 mg/kg (average 182.88 mg/kg), and from 87.90 to 237.61 mg/kg (average 176.69 mg/kg) in piedmont, central and littoral plains, exhibiting a decreasing trend as piedmont > central > littoral. Sediment physicochemical properties such as mineral composites and metal oxide abundance controlled REE concentrations, as observed along depth in three sediment sampling profiles. All sediment samples had coherent NASC-normalized REE patterns that were characterized by light REE (LREE) enrichment over heavy REE (HREE) with $(La/Yb)_{NASC}$ ranging between 0.86 and 1.49. Negative Ce and Eu anomalies generally occurred in the investigated sediments with Ce/Ce* and Eu/Eu* ranges of 0.69 to 1.17 and 0.82 to 1.34, respectively. The present study provides important dataset for future studies in this region.

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Figure and Table captions

Table 1 Concentrations of REE and major oxides and REE fractionation indices in the sediment samples from three zones

Table 2 Pearson's correlations between REE concentrations and major oxides, CIA and LOI in the sediment samples from three zones (Y09 (118 m): piedmont, n = 50; S30 (184 m): central, n=163; H02 (110 m): littoral, n = 63; "***" indicates $p < 0.01$, "*" $p < 0.05$)

Figure 1 Study area and sediments sampling locations (boreholes)

Figure 2 Changes of sediment lithology and REE parameters along depth: (a) piedmont borehole Y09, (b) central borehole S30, (c) littoral borehole H02.

Figure 3 Ternary diagram of sandstone classification and Al_2O_3 -CaO+Na₂O-K₂O (A-CN-K)

Figure 4 NASC normalized REE patterns and absolute REE concentrations all sediment samples (The line indicates a trend of normalized REE concentrations. The box-whisker shows distributions of absolute REE concentrations)

Figure 5 Total REE concentrations as a function of $(La/Yb)_{NASC}$ (a), Ce/Ce^* (c) and Eu/Eu^* (d); a relationship between CIA and $(La/Yb)_{NASC}$ (b).

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Table 1

Component	Unit	piedmont			central			littoral		
		Y09			S30			H02		
		Min.	Avg.	Max.	Min.	Avg.	Max.	Min.	Avg.	Max.
La	mg/kg	19.54	40.93	63.81	16.59	33.16	54.53	16.80	32.05	43.99
Ce	mg/kg	39.75	82.99	135.94	31.99	66.34	108.59	31.81	63.09	87.26
Pr	mg/kg	4.60	9.71	14.98	3.87	7.72	12.38	3.87	7.39	10.01
Nd	mg/kg	16.50	35.71	55.46	13.96	28.41	45.28	13.70	27.28	36.75
Sm	mg/kg	3.12	6.81	10.43	2.66	5.56	8.73	2.58	5.40	7.30
Eu	mg/kg	0.70	1.31	1.91	0.72	1.12	1.60	0.72	1.09	1.50
Gd	mg/kg	2.73	6.09	9.67	2.32	4.99	7.72	2.22	4.87	6.97
Tb	mg/kg	0.45	0.97	1.52	0.34	0.78	1.24	0.34	0.78	1.10
Dy	mg/kg	2.49	5.41	8.35	1.97	4.34	6.75	1.97	4.32	6.03
Ho	mg/kg	0.55	1.11	1.70	0.40	0.88	1.35	0.41	0.88	1.22
Er	mg/kg	1.53	3.10	4.42	1.11	2.46	3.72	1.08	2.47	3.40
Tm	mg/kg	0.26	0.53	0.75	0.19	0.42	0.64	0.20	0.43	0.59
Yb	mg/kg	1.59	3.20	4.47	1.24	2.64	3.93	1.18	2.62	3.50
Lu	mg/kg	0.27	0.53	0.73	0.20	0.44	0.66	0.21	0.44	0.60
Y	mg/kg	14.60	30.20	44.94	10.53	23.61	36.11	10.80	23.59	34.05
SiO ₂	wt%	55.55	66.77	78.36	35.71	56.62	71.56	32.04	59.51	77.03
Fe ₂ O ₃	wt%	1.86	5.46	9.86	2.00	4.45	6.51	1.83	4.06	6.61
Al ₂ O ₃	wt%	6.37	10.47	14.14	7.42	10.95	13.82	7.32	12.18	15.96
MgO	wt%	0.27	1.38	2.41	1.01	2.34	3.70	0.61	2.14	5.27
CaO	wt%	0.40	1.25	5.45	0.97	7.01	19.23	1.30	6.12	18.63
Na ₂ O	wt%	0.15	1.15	2.28	0.66	1.65	3.31	0.52	3.38	46.29
K ₂ O	wt%	1.89	2.58	3.37	1.55	2.20	2.82	1.57	2.26	2.89
MnO	wt%	0.02	0.07	0.30	0.03	0.07	0.15	0.04	0.07	0.16
P ₂ O ₅	wt%	0.06	0.1	0.69	0.13	0.25	0.36	0.04	0.12	0.17
TiO ₂	wt%	0.26	0.69	0.99	0.44	0.93	1.54	0.51	0.96	1.22
LOI	wt%	1.13	3.88	7.36	3.12	8.56	19.91	2.5	8.02	17.56
ΣREE	mg/kg	109	229	354	89	183	291	88	177	238
LREE	mg/kg	84	178	279	70	142	231	7	136	183
HREE	mg/kg	9.93	20.94	31.59	7.77	16.96	26.00	7.61	16.81	23.39
LREE/HREE	-	6.93	8.43	10.83	7.51	8.36	9.69	7.33	8.15	9.80
Ce/Ce*	-	0.69	0.91	1.17	0.85	0.90	0.98	0.80	0.89	0.98
Eu/Eu*	-	0.82	0.91	1.07	0.83	0.95	1.31	0.86	0.95	1.34
Gd/Gd*	-	0.95	1.06	1.15	0.93	1.05	1.12	0.87	1.04	1.10
(Y/Ho) _{NASC}	-	1.09	1.13	1.22	1.04	1.11	1.17	1.06	1.11	1.19
(La/Yb) _{NASC}	-	0.86	1.23	1.15	1.03	1.21	1.49	0.91	1.09	1.49

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Table 2

	borehole	REE	SiO ₂	Fe ₂ O ₃	Al ₂ O ₃	MgO	CaO	Na ₂ O	K ₂ O	LOI	CIA
REE	Y09	1									
	S30	1									
	H02	1									
SiO ₂	Y09	-0.79**	1								
	S30	-0.58**	1								
	H02	-0.67**	1								
Fe ₂ O ₃	Y09	0.72**	-0.92**	1							
	S30	0.90**	-0.56**	1							
	H02	0.89**	-0.67**	1							
Al ₂ O ₃	Y09	0.67**	-0.77**	0.85**	1						
	S30	0.51**	0.249**	0.60**	1						
	H02	0.19	0.430**	0.30*	1						
MgO	Y09	0.55**	-0.77**	0.79**	0.55**	1					
	S30	0.66**	-0.61**	0.79**	0.31**	1					
	H02	0.71**	-0.84**	0.71**	-0.32**	1					
CaO	Y09	-0.06	-0.11	0.05	-0.342*	0.44**	1				
	S30	0.12	-0.82**	0.04	-0.72**	0.21*	1				
	H02	0.15	-0.58**	0.06	-0.40**	0.12	1				
Na ₂ O	Y09	-0.74**	0.78**	-0.70**	-0.79**	-0.31*	0.31*	1			

	S30	-0.81**	0.79**	-0.75**	-0.24*	-0.57**	-0.44**	1			
	H02	0.19	-0.51**	0.13	-0.76**	0.68**	-0.07	1			
K ₂ O	Y09	0.13	-0.09	0.001	-0.125	-0.19	-0.40**	-0.18	1		
	S30	0.17	0.26**	0.36**	0.58**	0.17	-0.61**	0.09	1		
	H02	0.17	-0.003	0.46**	0.49**	0.08	-0.39**	-0.19	1		
LOI	Y09	0.50**	-0.84**	0.60**	0.46**	0.44**	0.03	-0.52**	-0.003	1	
	S30	0.21*	0.91**	0.16	0.64**	0.23*	0.83**	0.52**	1.74	1	
	H02	0.27*	0.85**	0.25*	0.47**	0.42**	0.62**	-0.04	-0.03	1	
CIA	Y09	0.51**	-0.75**	0.82**	0.54**	0.44**	-0.003	-0.66**	-0.04**	0.56**	1
	S30	0.26	0.06	0.36**	0.02	0.09	-0.52**	-0.06	0.34**	0.002	1
	H02	0.35**	0.03	0.48**	0.25*	0.26*	-0.29*	-0.01	0.21*	-0.01	1

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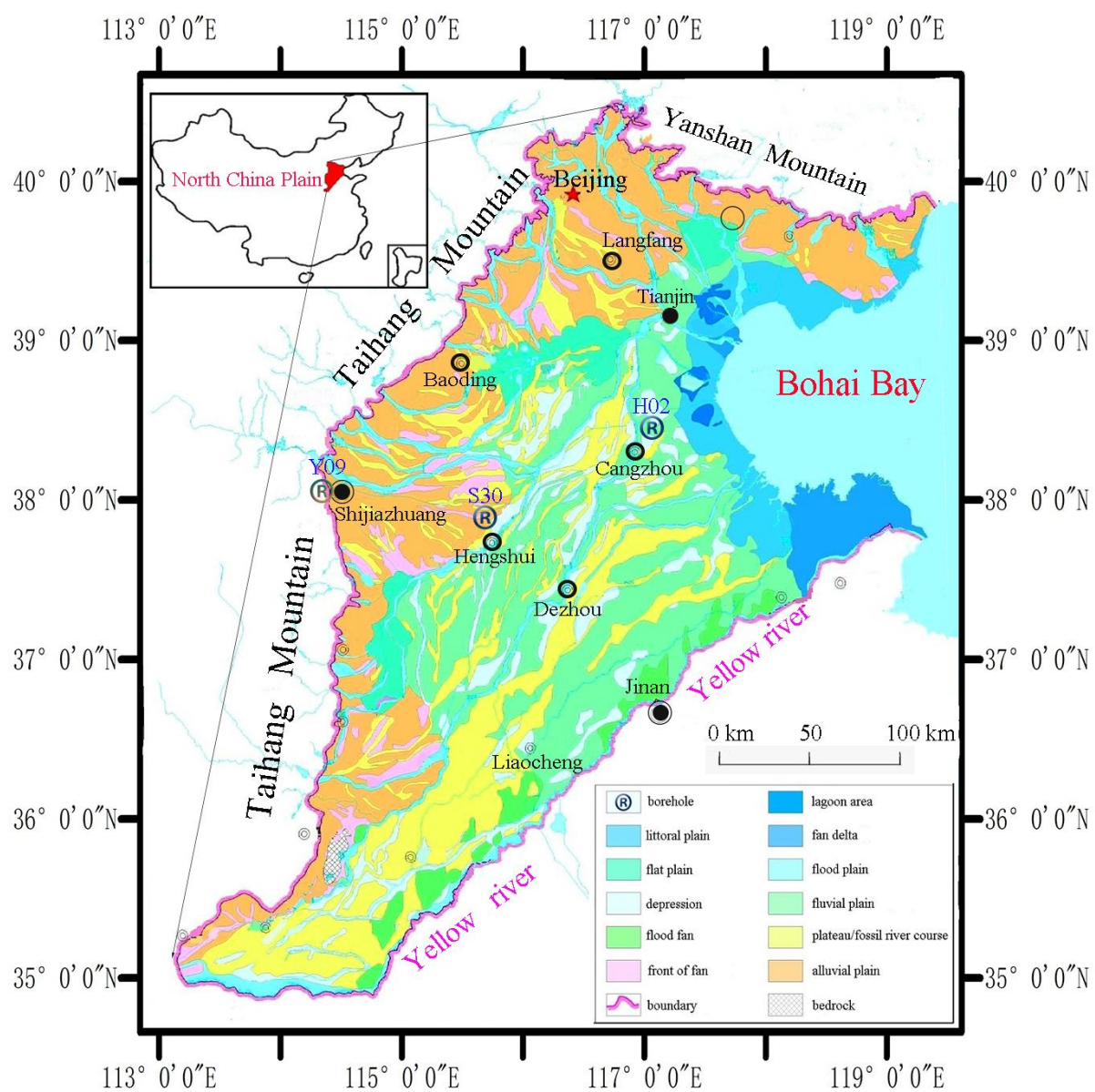


Fig. 1

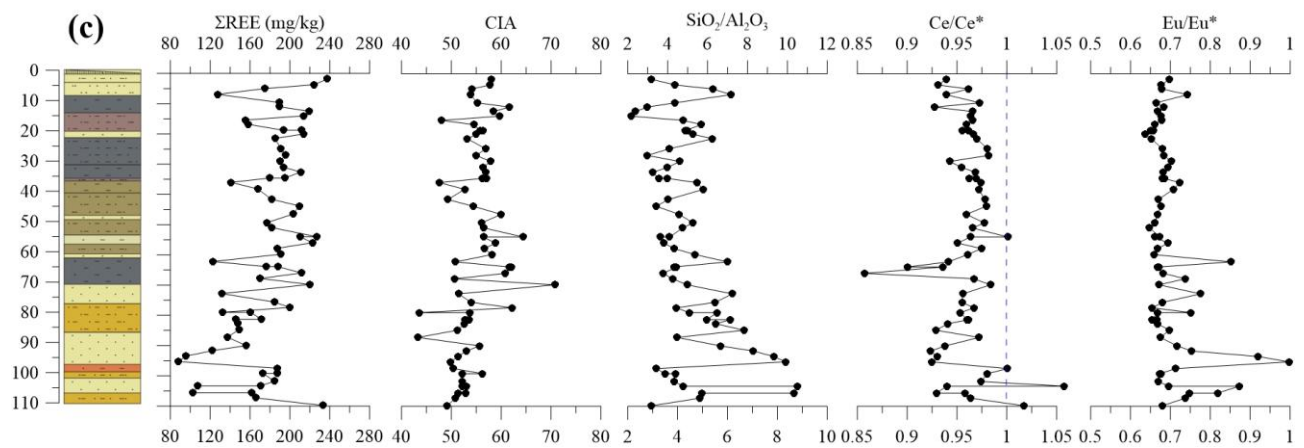
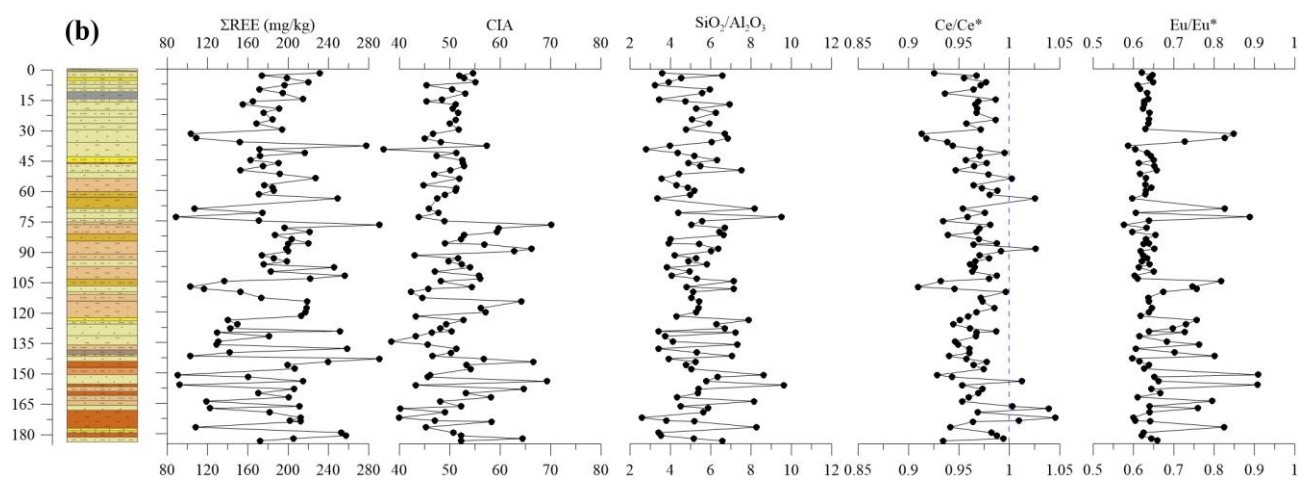
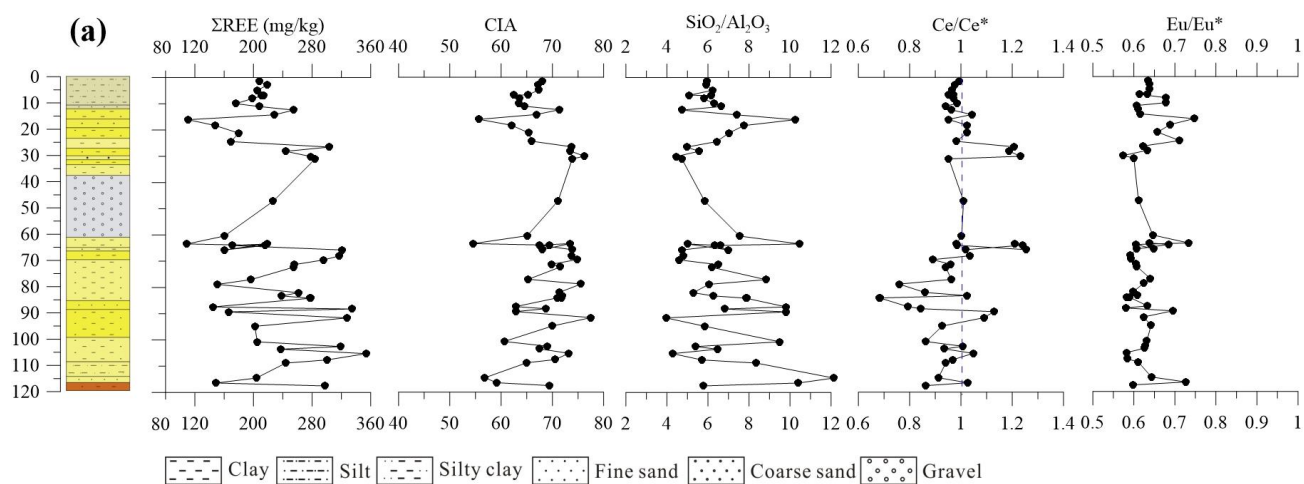


Fig. 2

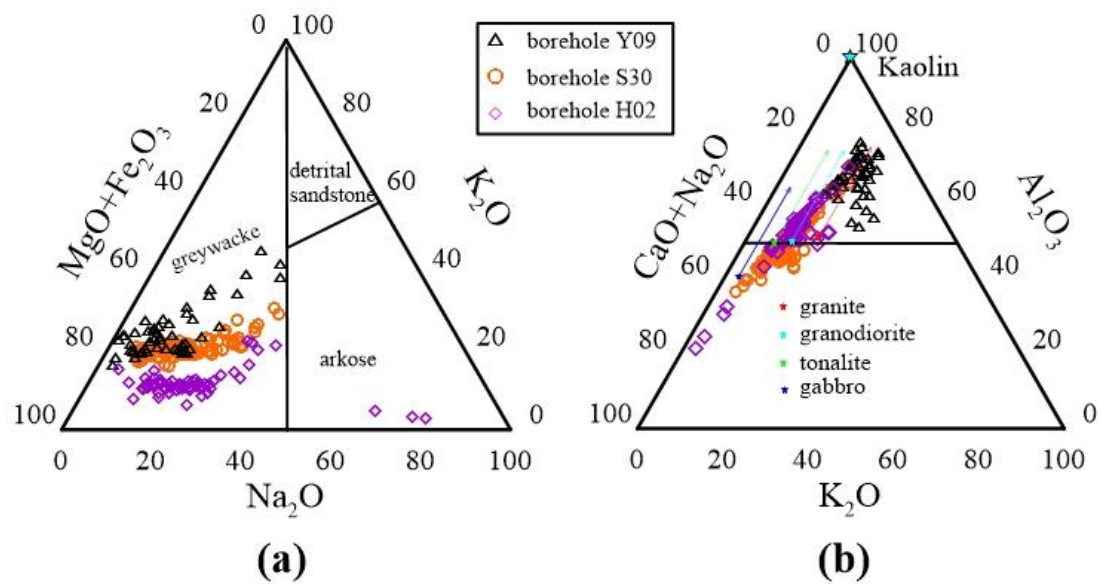


Fig. 3

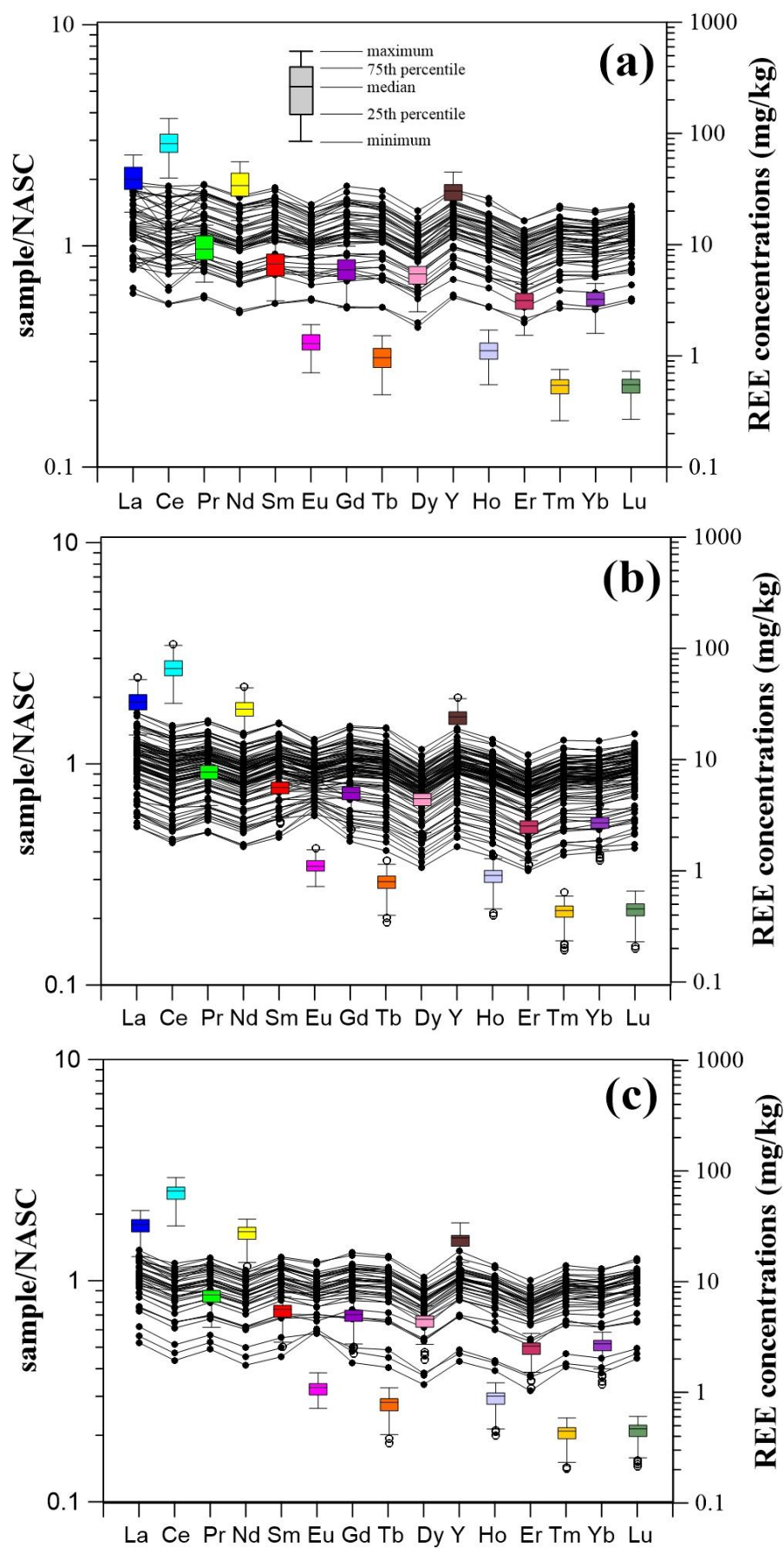
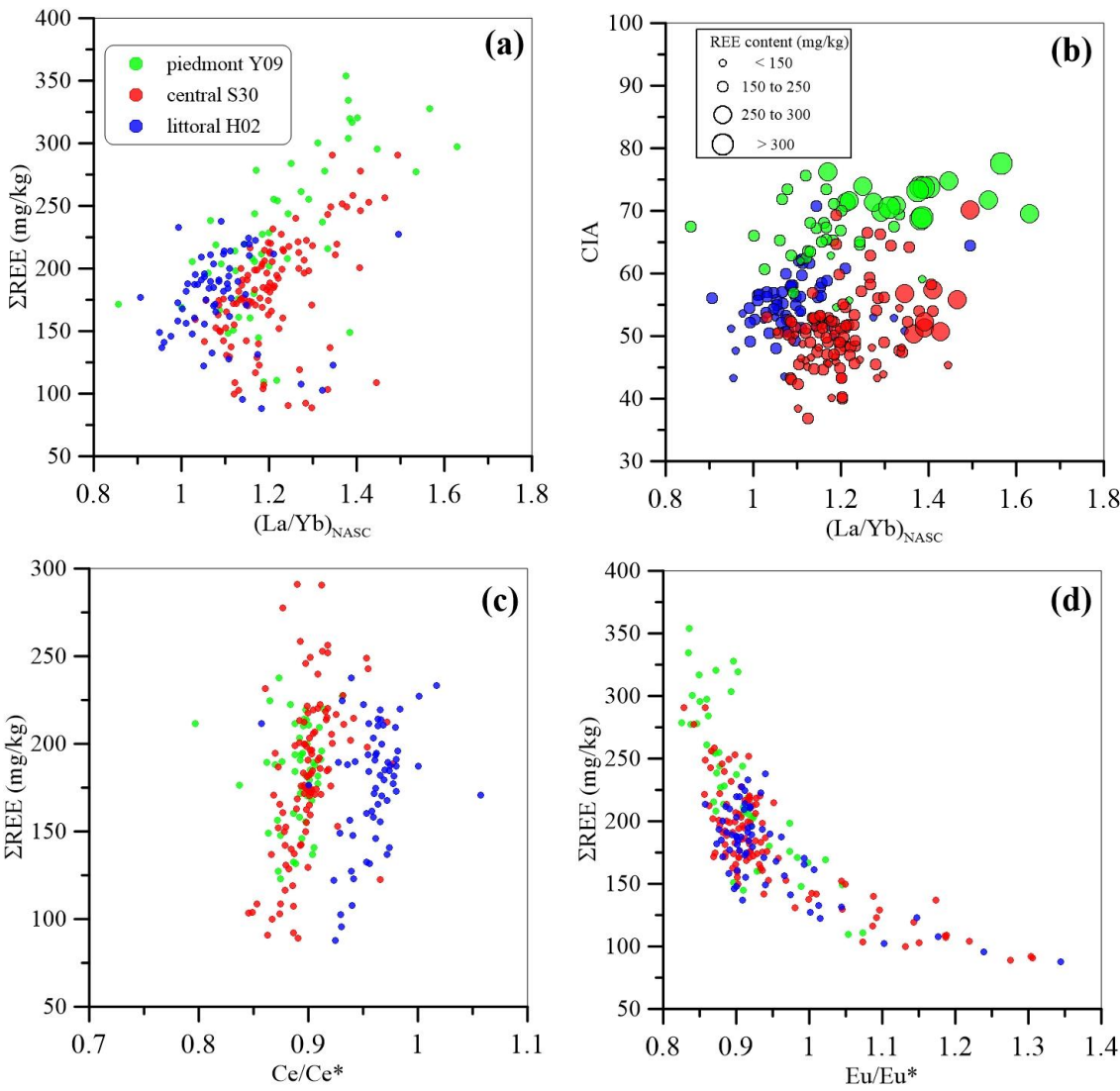


Fig. 4

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Fig. 5