

## Expression of Rat Aspartyl-tRNA Synthetase in Saccharomyces cerevisiae

Fabrice Agou, Jean-Pierre Waller, Marc Mirande

### ▶ To cite this version:

Fabrice Agou, Jean-Pierre Waller, Marc Mirande. Expression of Rat Aspartyl-tRNA Synthetase in Saccharomyces cerevisiae. Journal of Biological Chemistry, 1996, 271 (46), pp.29295-29303. 10.1074/jbc.271.46.29295 . hal-03276099

## HAL Id: hal-03276099 https://hal.science/hal-03276099

Submitted on 1 Jul 2021

**HAL** is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers. L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.



Distributed under a Creative Commons Attribution 4.0 International License

# Expression of Rat Aspartyl-tRNA Synthetase in Saccharomyces cerevisiae

ROLE OF THE  $\rm NH_2\text{-}TERMINAL$  POLYPEPTIDE EXTENSION ON ENZYME ACTIVITY AND STABILITY\*

(Received for publication, March 12, 1996, and in revised form, August 13, 1996)

#### Fabrice Agou<sup>‡</sup>, Jean-Pierre Waller, and Marc Mirande<sup>§</sup>

From the Laboratoire d'Enzymologie et Biochimie Structurales, CNRS, Gif sur Yvette, France

Cytoplasmic aspartyl-tRNA synthetase from mammals is one of the components of a multienzyme complex comprising nine synthetase activities. The presence of an amino-terminal extension composed of about 40 residues is a characteristic of the eukaryotic enzyme. We report here the expression in the yeast Saccharomyces cerevisiae of a native form of rat aspartyl-tRNA synthetase and of two truncated derivatives lacking 20 or 36 amino acid residues from their amino-terminal polypeptide extension. The three recombinant enzyme species were purified to homogeneity. They behave as  $\alpha_2$  dimens and display catalytic parameters in the tRNA aminoacylation reaction identical to those determined for the native, complex-associated form of aspartyl-tRNA synthetase isolated from rat liver. Because the dimer dissociation constant of rat AspRS is much higher than that of its bacterial and yeast counterparts, we could establish a direct correlation between dissociation of the dimer and inactivation of the enzyme. Our results clearly show that the monomer is devoid of amino acid activation and tRNA aminoacylation activities, indicating that dimerization is essential to confer an active conformation on the catalytic site. The two NH<sub>2</sub>-terminal truncated derivatives were fully active, but proved to be more unstable than the recombinant native enzyme, suggesting that the polypeptide extension fulfills structural rather than catalytic requirements.

Aminoacyl-tRNA synthetases are essential enzymes that catalyze a key step during the process of protein biosynthesis: activation of a specific amino acid and the charging of the cognate tRNA (1). On the basis of the comparison of the primary structures of the 20 enzymes from *Escherichia coli*, it was shown that this family is composed of two phylogenetically distinct protein groups harboring mutually exclusive sets of conserved sequence motifs (2). Accordingly, the topological arrangement of the active site domain, as well as the mode of tRNA:enzyme recognition, are essentially different for these two enzyme classes (3). Class I enzymes are characterized by the two consensus sequences HIGH and KMSKS, which are arranged within a structural fold of the Rossmann type: a six-stranded parallel  $\beta$ -sheet. On the other hand, the structural

This is an Open Access article under the CC BY license.

base for the catalytic domain of class II enzymes is a sixstranded anti-parallel  $\beta$ -sheet that supports three conserved sequence motifs.

Crystallographic studies designate motifs 2 and 3 of class II aminoacyl-tRNA synthetases as essential pieces of their active site (4-9). This was confirmed by site-directed mutagenesis of presumed key amino acids within these conserved sequences (4, 10-12). Motif 1 was originally described as a prominent structural element building the subunit interface of class II enzymes that form  $\alpha_2$  dimers. Because the corresponding residues belong to a long  $\alpha$ -helix and a  $\beta$ -strand located at the dimeric interface, it was first believed that motif 1 is a key element of oligomerization. However, recent data suggest that this motif is predominantly involved in stabilizing the conformation of the active sites of the two monomers and participates in intersubunit communication (9, 13, 14). Furthermore, a canonical motif 1 was also identified in two monomeric class II enzymes: Ala-tRNA synthetase from yeast, insect, or human (15) and Phe-tRNA synthetase from yeast mitochondria (16). Therefore, the conserved structural motif that builds the dimeric interface of class II synthetases is likely to be indispensable, but not sufficient for oligomerization.

Aspartyl-tRNA synthetase (AspRS)<sup>1</sup> is one of the well known class II enzymes (4, 17). The primary structures of 12 AspRS enzymes, from archaebacteria (Pyrococcus sp.), eubacteria (E. coli, Haemophilus influenzae, Thermus thermophilus, Mycobacterium leprae, Mycoplasma genitalium, Mycoplasma capricolum, mitochondria from Saccharomyces cerevisiae) and eukaryotes (Caenorhabditis elegans, cytoplasma from S. cerevisiae, Rattus norvegicus, Homo sapiens) have been determined through gene cloning, and the 3-D structures of a prokaryotic (T. thermophilus) and an eukaryotic (S. cerevisiae) AspRS resolved. One particular feature that distinguishes mammalian AspRS from all other known AspRS species concerns its assembly within a multienzyme complex comprising eight other aminoacyl-tRNA synthetases: glutamyl-, prolyl-, isoleucyl-, leucyl-, methionyl, glutaminyl-, lysyl-, and arginyltRNA synthetase (18). Rat (19) and human (20) AspRS display an NH<sub>2</sub>-terminal polypeptide extension that could fold into an amphiphilic  $\alpha$ -helix (21). This extension is dispensable for the catalytic activity of the enzyme (22, 23), but it was shown in vivo that the association of AspRS to the complex is abolished by the removal of 34 residues from the NH2-terminal extremity of the protein (22).

To investigate *in vitro* the functional properties of rat AspRS and of genetically engineered mutant enzymes, large amounts of purified proteins are required. The availability of the cloned

<sup>\*</sup> This work was supported in part by grants from CNRS, the Fondation pour la Recherche Médicale, the Association pour la Recherche sur le Cancer, and the Agence Nationale de Recherches sur le SIDA. The costs of publication of this article were defrayed in part by the payment of page charges. This article must therefore be hereby marked "*advertisement*" in accordance with 18 U.S.C. Section 1734 solely to indicate this fact.

<sup>‡</sup> Fellow of the Ligue Nationale contre le Cancer.

<sup>§</sup> To whom correspondence should be addressed. Tel.: 33-1-69-82-35-05; Fax: 33-1-69-82-31-29; E-mail: marc.mirande@lebs.cnrs-gif.fr.

<sup>&</sup>lt;sup>1</sup> The abbreviations used are: AspRS, aspartyl-tRNA synthetase; rAspRS, recombinant native AspRS; Cx-AspRS, complex-associated AspRS; BSA, bovine serum albumin; DTE, dithioerythritol; AMPPCP, adenosine 5'- $(\beta,\gamma$ -methylene)triphosphate.

cDNA provided the opportunity to construct an overproducing strain, to isolate rat AspRS species in a cellular background precluding its association with the other components of the multisynthetase complex, to analyze the catalytic activity as well as the stability of rat AspRS species isolated as free enzymes. In this paper, taking advantage of the high dimer dissociation constant of rat AspRS as compared with its bacterial and yeast counterparts, we also demonstrate that dimerization of a class II synthetase is absolutely required for amino acid activation as well as tRNA aminoacylation.

#### EXPERIMENTAL PROCEDURES

Strains and Plasmids—M13mp9, the yeast shuttle vectors pYeDP1/ 8-10 and pYeDP60 (Ref. 24; gift from D. Pompon, Centre de Génétique Moléculaire, Gif sur Yvette, France), and their derivatives were propagated in *E. coli* JM101Tr ( $\Delta(lac-pro)$  sup*E* thi recA56 srl-300::Tn10 F'(traD36 proAB lacI<sup>q</sup> lacZ $\Delta$ M15)). The haploid *S. cerevisiae* strain W303-1B (MAT $\alpha$  leu2 his3 trp1 ura3 ade2 can<sup>R</sup> cyr<sup>+</sup>) was provided by D. Pompon.

General Recombinant DNA Techniques—Restriction endonucleases and DNA modifying enzymes were purchased from Boehringer Mannheim or Perkin-Elmer. Radionucleotides were from Amersham. General recombinant DNA manipulations were carried out according to standard procedures (25). Nucleotide sequences were determined by the dideoxynucleotide chain termination method (26). All constructions were confirmed by DNA sequencing. Polymerase chain reaction experiments were performed with 6.6 ng of plasmid DNA, 100 pmol of oligonucleotide primers, and 2.5 units of Taq DNA polymerase, in a Perkin-Elmer thermocycler.

Enzymatic Assays-Initial rates of tRNA aminoacylation were measured at 25 °C, unless otherwise stated, in 0.1 ml of 20 mM imidazole-HCl buffer (pH 7.5), 150 mM KCl, 0.5 mM DTE (1, 4-dithioerythritol), 5 mM MgCl<sub>2</sub>, 3 mM ATP, 60 µM <sup>14</sup>C-labeled aspartic acid (Amersham, 50 Ci/mol), and saturating amounts of partially purified beef liver tRNA. The incubation mixture contained catalytic amounts of enzymes appropriately diluted in 10 mM Tris-HCl (pH 7.5), 10 mM β-mercaptoethanol, 20% glycerol, containing bovine serum albumin at 4 mg/ml. One unit of AspRS activity is the amount of enzyme producing 1 nmol of aspartyltRNA/min, at 25 °C. Specific activities of purified enzymes were calculated with protein concentrations determined by using absorption coefficients of 0.41, 0.38, and 0.39  $A_{280}~{\rm units \cdot mg^{-1} \cdot cm^2}$  for rAspRS (the recombinant native AspRS) AspRS- $\Delta$ N36 (rAspRS with a deletion of 36 NH2-terminal residues) and AspRS-ΔN20 (deletion of 20 NH2-terminal residues), respectively. For the determination of  $K_m$  values for tRNA,  $tRNA^{Asp}$  concentrations of 0.022–6.6  $\mu$ M were used, as determined from the aspartyl-acceptance of the tRNA preparation (148 pmol of [14C]aspartic acid/ $A_{260}$  unit). Michaelis parameters were obtained by nonlinear regression of the theoretical Michaelis equation to the experimental curve using the KaleidaGraph 3.0.4 software (Abelbeck Software).

The isotopic [<sup>32</sup>P]PP<sub>i</sub>–ATP exchange reaction was assayed at 25 °C, unless otherwise stated, in 0.1 ml of 20 mM imidazole-HCl buffer (pH 7.5), 0.1 mM EDTA, 2.5 mM  $\beta$ -mercaptoethanol, 10 mM MgCl<sub>2</sub>, 2 mM ATP, 2 mM aspartic acid, and 2 mM [<sup>32</sup>P]PP<sub>i</sub> (Amersham, 0.4 Ci/mol). Reaction was initiated by the addition of an appropriate enzyme dilution. One unit of AspRS activity in the exchange reaction is the amount of enzyme producing 1 nmol of ATP/min, at 25 °C.

Construction of the Expression Vector pYeDP60/DRS for Overproduction of Rat AspRS in S. cerevisiae—An EcoRI-EcoRI cDNA fragment containing the complete coding sequence of the rat AspRS cDNA (19) was cloned into the EcoRI site of pYeDP1/8-10 (24) to give pYeDP1/ DRS. This plasmid has the yeast 2-µm origin of replication and the URA3-selectable marker but displays a poor mitotic stability. A high stability vector, pYeDP60, was derived from pYeDP1/8-2 by insertion of a yeast DNA fragment containing the ADE2 gene (24). pYeDP1/8-2 is identical to pYeDP1/8-10 except for the presence of a polylinker in the opposite orientation. To insert the rat AspRS cDNA into pYeDP60, we took advantage of the high efficiency of homologous recombination in yeast. The yeast strain W303-1B was cotransformed with a ClaI fragment from pYeDP1/DRS containing the GAL10-CYC1 promoter, the rat AspRS cDNA, and part of the phosphoglycerate kinase terminator, and with pYeDP60 digested with BamHI and EcoRI, located in the polylinker region between the GAL10-CYC1 promoter and the phosphoglycerate kinase terminator. Ura<sup>+</sup> and Ade<sup>+</sup> transformants resulted from recombination events between homologous regions of the two fragments.

Construction of pYeDP60/DRS-ΔN36—The BamHI-EcoRI fragment

from M13mp9/DRS- $\Delta$ N (22) was introduced into pYeDP60 digested with *Bam*HI and *Eco*RI to give pYeDP60/DRS- $\Delta$ N36 $\Delta$ Met. The coding region of rat AspRS started at residue Ser<sup>37</sup>. The NH<sub>2</sub>-terminal sequence MSLDP was provided by insertion of the two complementary oligonucleotides 5'-GATCATAATGTCTCTG-3' and 3'-TATTACA-GAGACCTAG-5' into the unique *Bam*HI site of pYeDP60/DRS- $\Delta$ N36 $\Delta$ Met to give pYeDP60/DRS- $\Delta$ N36.

Construction of pYeDP60/DRS- $\Delta N20$ —We produced a DNA fragment by amplification of M13mp9/DRS1 between the oligonucleotides 5'-GGGATCCTTCTCGAGTATGATACAATCACAAG-3' (F01) and DRS666IC (22). This corresponds to DRS1 cDNA over positions 101-666 and allows the insertion of a XhoI site in the coding sequence of rat AspRS. The amplified fragment was digested with BamHI and HpaI and ligated into M13mp9/DRS1 digested with BamHI and HpaI to give M13mp9/DRSAAN. The linkers L02, made of 5'-GATCCATAATGC-CCAGCGCCAACGCCCT-3' and 3'-GTATTACGGGTCGCGGTTGCGG-GAGATC-5', and L04, made of 5'-CTAGAAAGGGTGTC-3' and 3'-TT-TCCCACAGAGCT-5', were inserted into M13mp9/DRSΔΔN digested with BamHI and XhoI. The resulting plasmid M13mp9/DRSA11-30 was digested with BamHI and EcoRI and the DNA insert cloned into pYeDP60 digested by BamHI and EcoRI.

Expression of Rat AspRS in S. cerevisiae-The yeast cell W303-1B was transformed by the lithium chloride method (27). Ura<sup>+</sup> and Ade<sup>+</sup> transformants were selected on plates containing 0.7% yeast nitrogen base (Difco), 0.1% casamino acids (Difco), 5% D-glucose, and 20 µg/ml L-tryptophan. Expression of rat AspRS was induced by growth on plates containing 2% D-galactose instead of glucose. Cells were grown at 28 °C in 50 ml of YPGal (0.5% yeast extract (Difco), 0.5% Bacto-peptone (Difco), 2% D-galactose) to a cell density corresponding to  $A_{600} \sim$  1–2. All subsequent steps were performed at 4 °C. Cells were harvested; washed twice with 100 mM Tris-HCl (pH 8.0), 10 mM MgCl<sub>2</sub>, 1 mM EDTA, 1 mM DTE; resuspended in 0.3 ml of 25 mM potassium phosphate (pH 7.5), 10% glycerol, 1 mM DTE, 1 mM EDTA containing 1 mM diisopropyl fluorophosphate, 12  $\mu$ g/ml pepstatin A, and 6  $\mu$ g/ml leupeptin (Protein Research Foundation, Osaka, Japan); and broken with glass beads (200  $\mu$ l, 0.45–0.50 mm) by vigorous shaking (0.5 min, 10 times). After centrifugation for 15 min at  $10,000 \times g$ , the supernatant was assayed for AspRS activity by the aminoacylation of tRNA. Protein concentration was determined according to Ref. 28.

Purification of rAspRS from a S. cerevisiae-overproducing Strain— Yeast cells W303-1B:pYeDP60/DRS that express the native recombinant AspRS were grown at 28 °C in 3 liters of YPGal to an absorbance of  $A_{600} \sim 8$ . Cells were washed, resuspended in 40 ml of extraction buffer containing the protease inhibitors, as described above, and lysed in an Eaton press at 5000 p.s.i. The lysate was diluted 2.5-fold with extraction buffer supplemented with 1 mM diisopropyl fluorophosphate, and centrifuged at 10,000 × g for 30 min to remove cell debris.

All chromatographic steps were performed at 4 °C. The suspension was applied to a S Sepharose FF column (Pharmacia,  $2.5 \times 9.4$  cm) equilibrated in 25 mM potassium phosphate (pH 7.5), 10% glycerol, 1 mM EDTA, 1 mM DTE and eluted with a 1500-ml linear gradient of potassium phosphate from 25 to 300 mM (pH 7.5) containing the same additives. Fractions containing rat AspRS (eluted at 75 mM) were fractionated further on a Q Sepharose FF column (Pharmacia,  $1.1 \times 10.5$ cm) equilibrated in 75 mM potassium phosphate, pH 7.5, 10% glycerol, 1 mm DTE and eluted with a 300-ml linear gradient of potassium phosphate from 75 to 300 mM (pH 7.5). Fractions containing AspRS activity (eluted at 110 mm) were pooled, diluted 5-fold with 10% glycerol and 1 mm DTE, and applied to a  $1.1 \times 8.5$ -cm column of tRNA-Sepharose (29) equilibrated in 25 mm potassium phosphate (pH 7.5), 10% glycerol, 1 mM DTE. The column was eluted with a 200-ml linear gradient of potassium phosphate from 25 mM to 200 mM (pH 7.5). Fractions containing AspRS activity (eluted at 70 mM) were pooled; concentrated by vacuum dialysis; dialyzed against 25 mm potassium phosphate (pH 7.5), 1 mM DTE, 50% glycerol; and stored at -20 °C. The homogeneity of the enzyme preparation was checked by SDS-polyacrylamide gel electrophoresis (30) and by NH2-terminal amino acid sequence analysis. Automated Edman degradations were performed by J.-P. Le Caer (Laboratoire de Physiologie Nerveuse, Gif sur Yvette, France) using an Applied Biosystems model 470A sequencer.

Purification of the NH<sub>2</sub>-terminal Truncated Forms of Rat AspRS Produced in Yeast—The purification of the mutant enzymes was conducted as described for rAspRS, except that all buffers contained 20% glycerol to prevent enzyme inactivation. The NH<sub>2</sub>-terminal deletions modify the global net charge of the protein: -1 for AspRS- $\Delta$ N36 and +2for AspRS- $\Delta$ N20. Accordingly, they were eluted on S Sepharose FF, Q Sepharose FF, and tRNA-Sepharose at phosphate concentrations of 70 and 135 mM, 125 and 100 mM, and 60 and 105 mM, respectively, for AspRS- $\Delta$ N36 and AspRS- $\Delta$ N20.

Analytical Gel Filtration—The apparent native molecular mass of the AspRS species was determined at room temperature by gel filtration on a Superose 12 HR 10/30 (Pharmacia) column equilibrated in 50 mM potassium phosphate (pH 7.5), 10 mM  $\beta$ -mercaptoethanol, containing 5 or 20% glycerol, and developed at a flow rate of 0.2 ml·min<sup>-1</sup>. All samples (100  $\mu$ g of protein) were loaded in 0.2 ml. The calibration curve was established by using bovine serum albumin, thyroglobulin, and soybean trypsin inhibitor as marker proteins. For a particular protein, its elution was described in term of the corresponding  $K_{\rm av}$  value.  $K_{\rm av} = (V_{\rm e} - V_0)/(V_{\rm t} - V_0)$ , where  $V_{\rm e}$  is the elution volume of the particular molecule,  $V_0$  the void volume of the column, and  $V_{\rm t}$  the total bed volume.  $V_0$  and  $V_{\rm t}$  were determined with thyroglobulin (670 kDa) and DTE (154 Da), respectively.

Analytical Ultracentrifugation—Sedimentation equilibrium experiments were performed in a Beckman Optima XL-A analytical ultracentrifuge, using an An-60 Ti rotor and a double-sector cell of 12 mm path length. The solution column was 0.1-0.2 cm; data were collected with a spacing of 0.001 cm. Equilibrium was verified from the superimposition of duplicate scans recorded at 4-h intervals. The experimental data were first fitted to a model for a single homogeneous species following the equation below.

$$c(r) = c(r_{\rm ref}) \exp\{[M(1 - \bar{v}\rho)\omega^2/2RT](r^2 - r_{\rm ref}^2)\}$$
 (Eq. 1)

c(r) is the concentration of the protein at radial position  $r, c(r_{\rm ref})$  is the concentration of the protein at an arbitrary reference distance  $r_{\rm ref}$  M is the molecular weight,  $\bar{\nu}$  the partial specific volume of the solute,  $\rho$  is the density of the solvent,  $\omega$  is the angular velocity, and R and T are the molar gas constant and the absolute temperature, respectively. The partial specific volume was calculated from the amino acid composition of the protein sample, and the density of the buffers was determined from published tables (31). Where the experimentally determined M did not converge on the true mass of the samples (56,960, 54,650, and 53,510 for rAspRS, AspRS-\DeltaN20, and AspRS-\DeltaN36, respectively), the data were fitted to a monomer-i-mer equilibrium, according to Equation 2, where  $c_i$  are the concentrations of the i-mer.

$$c(r) = \sum_{i} c_{i}(r_{\rm ref}) \exp\{[iM(1 - \bar{v}\rho)\omega^{2}/2RT](r^{2} - r_{\rm ref}^{2})\}$$
(Eq. 2)

In all cases, since inclusion of a second virial coefficient gave no improvement of the fit, an ideal component system was considered.

#### RESULTS

*Expression of Rat AspRS in Yeast*—We first tried to express and purify rat AspRS from a bacterial overproducing strain. Whereas aminoacylation activities could be readily measured in crude extracts from *E. coli* transformed with various expression vectors (19), rat AspRS proved to be very unstable during subsequent chromatographic steps, leading to a poorly active enzyme preparation. Expression of heterologous genes in the yeast *S. cerevisiae* has proved to be useful as an alternative approach (32). In order to obtain high yields of recombinant proteins, we used a multicopy shuttle vector with a high mitotic stability. The pYeDP60 vector propagates in yeast grown in rich medium, allowing to obtain a high cell density (24). Foreign proteins are expressed under the control of the hybrid GAL10-CYC1 promoter, which is activated in galactose-containing media and repressed in the presence of glucose.

The cDNA encoding rat AspRS was introduced into pYeDP60 and the recombinant plasmid used to transform the yeast strain W303-1B grown in the presence of glucose. To test for the level of expression and the putative toxicity of native rat AspRS in yeast, transformed cells were grown in rich medium containing galactose (YPGal). As shown in Fig. 1, a polypeptide with an apparent molecular mass of 57 kDa, identical to that of the AspRS component of the multisynthetase complex, is specifically detected in a crude extract from yeast cells carrying the recombinant plasmid pYeDP60/DRS. By Western blot analysis, only this polypeptide is detected with antibodies raised against the AspRS component from the sheep liver complex (data not shown). In addition, AspRS activity was assayed in

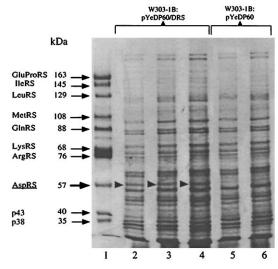


FIG. 1. Expression of rat AspRS in yeast. Crude extract from yeast cell W303-1B transformed with pYeDP60 (*lanes 5* and 6) or with pYeDP60/DRS (*lanes 2-4*) were analyzed by SDS-PAGE with a 8% get, by loading 20  $\mu$ g (*lanes 2* and 5), 40  $\mu$ g (*lanes 3* and 6), or 60  $\mu$ g (*lane 4*) of total protein. The polypeptide specifically expressed in strain W303-1B:pYeDP60/DRS is indicated by an *arrowhead*. The recombinant protein accounts for approximately 3% of cell proteins. The purified multisynthetase complex from rat is shown at *left* (*lane 1*). The identity and molecular mass of the polypeptide components of the complex are marked, and the AspRS component is *underlined*. The gel was stained with Coomassie Blue.

the corresponding crude extracts by the aminoacylation of beef tRNA. In these conditions, W303-1B transformed with pYeDP60/DRS overproduced AspRS activity 80-fold, as compared with the endogenous yeast AspRS activity detected with the control strain. Since the mock and overproducing strains grew at approximately the same rate, no deleterious effect resulted from the expression of rat AspRS in *S. cerevisiae*.

Purification of Rat AspRS Produced in Yeast—The recombinant AspRS (rAspRS) was purified to homogeneity from a culture of W303-1B:pYeDP60/DRS grown in the presence of galactose as the carbon source. Three chromatographic steps on S Sepharose FF, Q Sepharose FF, and tRNA-Sepharose were used as described under "Experimental Procedures." At each step, rAspRS was eluted as a single, symmetrical peak. This purification procedure yields 15 mg of homogeneous rAspRS/32 g of cell pellet, with a global recovery of 38% of AspRS activity (Table I). The purified rAspRS has an apparent molecular mass of 57 kDa identical to that of the AspRS component of the multisynthetase complex from rat and was judged to be at least 95% homogeneous according to SDS-PAGE analysis (Fig. 2, lanes 2 and 5). Its specific activity of 380 units/mg indicates that the rat enzyme produced in yeast is fully active. Indeed, in the same assay conditions, AspRS activity from the purified complex from rat is of 28 units/mg of complex (33), that is 350 units/mg of the AspRS component, taking into account the presence of 1 mol of AspRS ( $\alpha_2$ , 114 kDa)/mol of complex (1,450 kDa). To assess the integrity and the homogeneity of the enzyme preparation, 25 cycles of automatic Edman degradations were performed. A single amino acid sequence was obtained, corresponding to residues 2-26 of the cDNA-encoded protein. The removal of the initiating methionine when the second amino acid is a proline is in accordance with the rules of post-translational modification in yeast (34).

Expression and Purification of  $NH_2$ -terminal Truncated Derivatives of Rat AspRS—Two derivatives of rat AspRS were constructed in pYeDP60, expressed in yeast, and purified to homogeneity. The first of these derivatives, AspRS- $\Delta$ N36, previously expressed in cultured Chinese hamster ovary cells, was

Purification step	Protein	Total activity <sup>a</sup>	Specific activity	Yield	Relative purification factor
	mg	units	units/mg	%	
Purification of rAspRS					
Crude extract	1350	14,850	11	100	1
S-Sepharose FF	47	10,998	234	74	21
Q-Sepharose FF	18	6120	340	41	31
tRNA-Sepharose	15	5700	380	38	35
Purification of AspRS- $\Delta$ N36					
Crude extract	1633	19,000	11	100	1
S-Sepharose FF	62	15,760	250	83	23
Q-Sepharose FF	28	10,363	366	55	34
tRNA-Sepharose	11	4795	410	25	38
Purification of AspRS-ΔN20					
Crude extract	1190	11,892	10	100	1
S-Sepharose FF		8710		73	
Q-Sepharose FF		5850		49	
tRNA-Sepharose	13	5150	395	43	39

TABLE I Purification of native AspRS and of  $NH_2$ -terminal truncated derivatives

<sup>a</sup> One unit of activity corresponds to the formation of 1 nmol of aminoacyl-tRNA/min at 25 °C in the standard assay conditions.

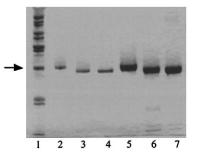


FIG. 2. SDS-PAGE analysis of the purified AspRS species produced in yeast. The purified enzyme preparations (0.2  $\mu$ g of protein in *lanes* 2–4; 2  $\mu$ g of protein in *lanes* 5–7) rAspRS (57 kDa; *lanes* 2 and 5), AspRS- $\Delta$ N20 (54.6 kDa; *lanes* 3 and 6), and AspRS- $\Delta$ N36 (53.5 kDa; *lanes* 4 and 7) were analyzed on a 10% polyacrylamide gel. The rat multisynthetase complex (*lane* 1) was used as size marker; the AspRS component is indicated by an *arrow*.

already described (22). Another NH<sub>2</sub>-terminal truncated form of rat AspRS was constructed. The AspRS- $\Delta$ N20 derivative has an amino acid sequence identical to that of the native enzyme, except for a 20-amino acid deletion from Gln<sup>11</sup> to Gly<sup>30</sup>, corresponding to the precise removal of a putative amphipathic  $\alpha$ -helix (20). The level of expression of AspRS- $\Delta$ N20 and AspRS-ΔN36 is similar to that of rAspRS, as judged by Western blotting and activity measurement. The truncated AspRS derivatives were purified as described for rAspRS, except that all buffers were supplemented with 20% glycerol instead of 10%. The use of higher glycerol concentrations was found to be required to preserve AspRS activity of the mutant enzymes, which are very sensitive to dissociation, as shown below. The two truncated enzymes are recovered with yields comparable with that obtained for the native enzyme, and the purified preparations display similar specific activities (Table I). This result already demonstrates that the NH<sub>2</sub>-terminal extension is dispensable for enzyme activity. The native and truncated forms rAspRS, AspRS-\DeltaN20, and AspRS-ΔN36 display electrophoretic mobilities in accordance with their calculated masses of 57, 54.6, and 53.5 kDa, respectively (Fig. 2).

Quaternary Structure of AspRS and Dissociation Equilibrium—The molecular masses of the rat AspRS species produced in yeast were determined by gel filtration on Superose 12 (Fig. 3). The molecular mass standards included the monomer (67 kDa) and dimer (134 kDa) of bovine serum albumin and soybean trypsin inhibitor (20 kDa). An apparent native molecular mass of 100 kDa was deduced for rAspRS, when analyzed in the presence of either 5% or 20% glycerol in the equilibration buffer. This value reasonably compares with a theoretical mass

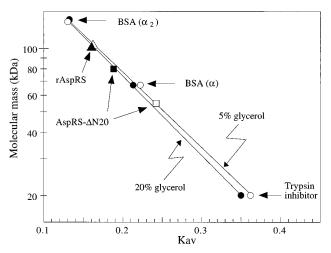


FIG. 3. Superose 12 chromatography of rAspRS and AspRS-  $\Delta$ N20. Gel filtration was conducted as described under "Experimental Procedures," in the presence of 20% ( $\bigcirc$ ,  $\blacksquare$ ,  $\triangle$ ) or 5% ( $\bigcirc$ ,  $\square$ ,  $\triangle$ ) glycerol in the equilibration buffer. The apparent, native molecular masses of rAspRS ( $\triangle$ ,  $\triangle$ ) and AspRS- $\Delta$ N20 ( $\blacksquare$ ,  $\square$ ) were estimated to 100 kDa ( $\blacktriangle$ ) and 80 kDa ( $\blacksquare$ ) in the presence of 20% glycerol or 100 kDa ( $\triangle$ ) and 55 kDa ( $\square$ ) with 5% glycerol.

of 114 kDa for a dimer. The behavior of the truncated species AspRS- $\Delta$ N20 and AspRS- $\Delta$ N36 is very sensitive to the glycerol concentration of the buffer (Fig. 3). The molecular masses of 80 kDa and 85 kDa observed for the  $\Delta$ N20 and  $\Delta$ N36 species, respectively, in the presence of 20% glycerol, indicate a tendency to dissociation of the dimers. In the presence of 5% glycerol, molecular masses of 55 and 57 kDa are determined and no AspRS activity can be recovered. These results suggested that the truncated forms of AspRS are dimers that readily dissociate and inactivate in the presence of low glycerol concentrations.

The stability and dissociation of the various AspRS species were examined by sedimentation equilibrium analysis. The oligomeric structure of the enzyme preparations was determined under different experimental conditions (Table II, Fig. 4). The influence of the glycerol on the oligomeric structure of AspRS was first analyzed. The native rAspRS behaves as a dimer in 50 mM Tris-HCl, pH 7.5, containing either 20% or 5% glycerol. By contrast, AspRS- $\Delta$ N36 has a tendency to dissociate, even in the presence of 20% glycerol, and forms large aggregates at 20 °C. These high molecular weight species are mainly due to the instability of the monomer. Lowering the temperature to 4 °C stabilizes the monomer, thus preventing its irreversible aggregation. 
 TABLE II

 Summary of sedimentation equilibrium experiments with recombinant native and truncated forms of rat AspRS

$\operatorname{Buffer}^a$	Temperature	Detection wavelength	Initial protein concentration	Speed	Monomer-oligomer equilibrium $^b$	
					Distribution of molecular species	Deduced $M_{\rm r}$ of the monomer
	°C	nm	mg/ml	rpm	%	
rAspRS						
50 mM Tris-HCl, 20% glycerol	20	280	0.5	13,000	100 $\alpha_2$	56,000
50 mM Tris-HCl, 5% glycerol	$20^{c}$	280	0.5	13,000	$100 \alpha_2$	57,000
50 mm Tris-HCl, 200 mm KCl	20	280	1.0	15,000	$15 \alpha, 85 \alpha_2$	55,000
,	$20^{c}$	280	0.2	15,000	35 $\alpha$ , 65 $\alpha_2$	60,000
25 mM potassium phosphate, 200 mM KCl	4	280	0.3	15,000	80 $\alpha$ , 5 $\alpha_2$ , 15 $\alpha_4$	56,000
	4	220	0.03	15,000	100 $\alpha$	60,000
$AspRS-\Delta N36$						
50 mM Tris-HCl, 20% glycerol	4	280	0.6	15,000	60 $\alpha$ , 40 $\alpha_2$	52,000
,	4	280	0.2	15,000	80 $\alpha$ , 20 $\alpha_2$	55,000
	20	280	0.2	15,000	$\alpha_n$	
50 mM Tris-HCl, 200 mM KCl	20	280	0.2	15,000	$\alpha_n$	
	4	220	0.05	15,000	$\alpha_n$	
$AspRS-\Delta N20$						
50 mM Tris-HCl, 200 mM KCl	20	280	0.2	15,000	$\alpha_n$	
	$4^c$	220	0.05	15,000	100 <sup>°</sup> α	52,000
AspRS from S. cerevisiae						
50 mм Tris-HCl, 500 mм KCl	4	280	0.5	15,000	100 $\alpha_2$	63,000

<sup>*a*</sup> All buffers are adjusted to pH 7.5 and contain 1 mM dithioerythritol.

<sup>b</sup> The data were analyzed as described under "Experimental Procedures" by using Equation 1 or 2. The values that best fit experimental data are indicated, taking into account the presence of various  $\alpha_i$  species in solution.  $\alpha_n$  indicates that heterogeneous high molecular weight aggregates were formed.

<sup>c</sup> Profiles of protein distribution are shown in Fig. 4.

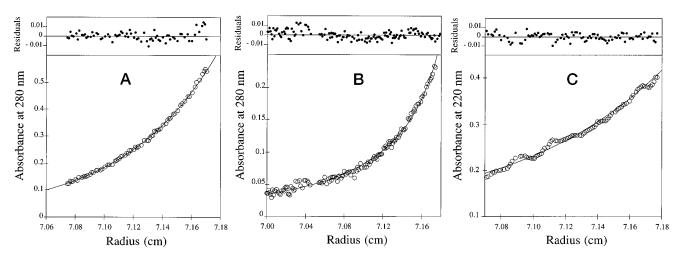


FIG. 4. Sedimentation equilibrium distribution of rAspRS and AspRS- $\Delta$ N20. Relative concentration of rAspRS (*A* and *B*) or AspRS- $\Delta$ N20 (*C*), measured by absorbance at 280 or 220 nm, is plotted as a function of radial distance after sedimentation in conditions of Table II. *A*, rAspRS at 0.5 mg<sup>m</sup>l<sup>-1</sup> in 50 mM Tris-HCl pH 7.5 and 5% glycerol, at 20 °C; *B*, rAspRS at 0.2 mg<sup>m</sup>l<sup>-1</sup> in 50 mM Tris-HCl, pH 7.5, and 200 mM KCl, at 20 °C; *C*, AspRS- $\Delta$ N20 at 0.05 mg<sup>m</sup>l<sup>-1</sup> in 50 mM Tris-HCl, pH 7.5, and 200 mM KCl, at 20 °C; *C*, AspRS- $\Delta$ N20 at 0.05 mg<sup>m</sup>l<sup>-1</sup> in 50 mM Tris-HCl, pH 7.5, and 200 mM KCl, at 4 °C. Experimental values (*open circles*) were fitted (*curves*) as described under "Experimental Procedures": residuals are indicated above each fit. The best fits were calculated for monodisperse solutes of 114 kDa (*A*) and 52 kDa (*C*), or a monomer-dimer equilibrium (*B*) with the mass of the monomer equal to 60 kDa and a dimer dissociation constant of 0.5 × 10<sup>-6</sup> M.

The effect of chaotropic salts on the stability of AspRS dimers was examined (Table II). The addition of KCl to 0.2 M induced partial dissociation of rAspRS and caused AspRS- $\Delta$ N36 and AspRS- $\Delta$ N20 to aggregate. In the case of AspRS- $\Delta$ N20, no aggregation occurs when sedimentation equilibrium analysis is conducted at 4 °C and at low protein concentration, and an homogeneous monomer species is observed (Fig. 4). The AspRS- $\Delta$ N36 derivative is more unstable, generating large aggregates in these conditions.

When the distribution of material along the column could be satisfactorily fitted to a model of monomer-dimer equilibrium, dissociation constants of  $0.5 \times 10^{-6}$  M and  $5 \times 10^{-6}$  M were deduced for rAspRS in 200 mM KCl at 20 °C and AspRS- $\Delta$ N36 in 20% glycerol at 4 °C, respectively. Following sedimentation analysis of AspRS from *S. cerevisiae* in the presence of up to

500 mM KCl, at 4 °C, the experimental distribution was fitted to a monodisperse solute of 125.5 kDa. This value precisely correlates with the calculated mass of a dimeric species (126.8 kDa). No reversible thermodynamic equilibrium is observed, indicating that the dimer dissociation constant of yeast AspRS is at least 10-fold lower than that of the rat enzyme.

The sedimentation equilibrium data indicate that rat AspRS can be readily dissociated by lowering glycerol concentration, temperature, or protein concentration or by increasing the concentration of chaotropic salts, conditions that do not significantly affect the dimeric yeast enzyme. The NH<sub>2</sub>-terminal truncated derivatives show a more pronounced tendency to dissociation. Whereas their dimer dissociation constants at 20 °C are lower than at 4 °C, the truncated monomer species prove to be more unstable at 20 °C than at 4 °C, leading to

#### Functional Analysis of Rat Aspartyl-tRNA Synthetase

TABLE III

Kinetic constants of native rat AspRS and of  $NH_2$ -terminal truncated derivatives in the  $tRNA^{Asp}$  aminoacylation reaction

Kinetic constant <sup>a</sup>	$AspRS-Cx^b$	rAspRS	$AspRS-\Delta N36$	$AspRS-\Delta N20$
$\frac{K_{M} \text{ tRNA}^{\text{Asp}} (\mu \text{M})}{k_{\text{cat}} (\text{s}^{-1})}$	$\begin{array}{c} 0.50 \pm 0.05 \\ 1.0 \pm 0.1 \end{array}$	$\begin{array}{c} 0.50 \pm 0.05 \\ 1.1 \pm 0.1 \end{array}$	$0.6 \pm 0.1 \\ 1.1 \pm 0.1$	$\begin{array}{c} 0.6 \pm 0.1 \ 1.1 \pm 0.1 \end{array}$

<sup>*a*</sup> The  $K_M$  values refer to aspartic acid acceptor tRNA.

<sup>b</sup> Values determined for the AspRS component of the purified multisynthetase complex from rat liver. The  $k_{cat}$  value is corrected for the presence of 1 mol of AspRS ( $\alpha_2$ , 114 kDa)/mol of complex (1450 kDa).

aggregation and irreversible inactivation, as shown below.

Dissociation of AspRS and Inactivation—Initial rates of rAspRS, AspRS- $\Delta$ N36, and AspRS- $\Delta$ N20 in the tRNA<sup>Asp</sup> aminoacylation reaction were determined in conditions where these enzymes behave as stable dimers, in the presence of 20% glycerol and 4 mg/ml BSA in the assay mixture (Table III). The corresponding Michaelis parameters were compared with those of the native, natural enzyme, the AspRS component from the multisynthetase complex purified from rat liver (33). The  $K_m$  value of each AspRS produced in yeast for tRNA<sup>Asp</sup> (0.5–0.6  $\mu$ M) is identical to that of the complex-associated AspRS from rat (Table III). Similarly, the measured  $k_{\rm cat}$  values are indistinguishable. Therefore, association of AspRS within the complex has no major influence on its catalytic activity.

Whereas the bacterial and yeast AspRS are very stable dimers, the monomer-dimer ratio can be easily modified in the case of the rat enzyme, by using mild experimental conditions (Table II). Glycerol stabilizes the dimeric form (Table II) as well as the enzymatic activity (Fig. 5). Decreasing glycerol concentration leads to rapid inactivation of AspRS- $\Delta$ N36 and, to a lesser extent, of AspRS- $\Delta$ N20 (Fig. 5). The presence of up to 20% glycerol is required to keep AspRS- $\Delta$ N36 or AspRS- $\Delta$ N20 fully active when incubated at 0.01 mg·ml<sup>-1</sup> (~0.1  $\mu$ M). In the presence of 5% or 0.5% glycerol, the half-life time of AspRS- $\Delta$ N36 is of 12 and 3 min, respectively. The full-length rAspRS is much less sensitive to the removal of glycerol in the incubation mixture. Its dissociation and inactivation requires the addition of chaotropic salts.

Upon incubation of rAspRS at 0.5 mg·ml<sup>-1</sup> (4.4  $\mu$ M) in the presence of 0.2 M KCl, dissociation occurs (Table II). The enzymatic activity of the sample decreases with a half-life time of about 80 min (Fig. 6). The residual activity is measured at 0 °C after dilution in the assay mixture containing 20% glycerol, to stabilize the remaining fraction of dimeric enzyme. In the case of rAspRS, when the assay was performed at 25 °C, a significant reactivation of AspRS was observed during the 10-min incubation in the assay mixture. When assayed at 0 °C, AspRS activity remains constant during this 10-min interval, and thus reflects the ratio of active and inactive forms in the sample after incubation in 0.2 M KCl at the times indicated. As shown in Fig. 6, the inactivation rate is dependent of the initial enzyme concentration. At protein concentrations of 0.44 and 0.06  $\mu$ M,  $t_{1/2}$  of rAspRS inactivation are of about 37 and 22 min, respectively. Amino acid activation and aminoacylation activities are lost at the same rate.

When the multisynthetase complex was incubated in the presence of 0.2 M KCl, its AspRS component (Cx-AspRS) inactivated, albeit less rapidly than rAspRS. Upon incubation of the complex at protein concentrations of 0.3 or 0.03  $\mu$ M in conditions described in the legend of Fig. 6, the  $t_{1/2}$  of Cx-AspRS inactivation was of about 40 min. Because Cx-AspRS remains tightly associated to the complex, the inactivation rate of Cx-AspRS was independent of the initial protein concentration in the incubation mixture.

When rAspRS is incubated at 20 °C, without KCl and in the presence of 5% glycerol, the enzyme is a dimer (Table II) and remains fully active even if incubated at low concentration (0.005 mg·ml<sup>-1</sup>, ~0.04  $\mu$ M) (Fig. 6). In the same conditions,

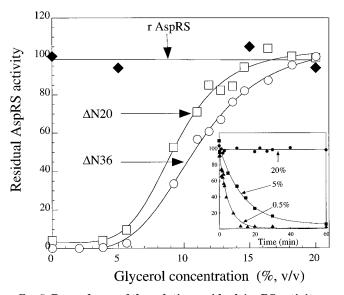


FIG. 5. Dependence of the relative residual AspRS activity on the glycerol concentration. The residual activities of rAspRS ( $\blacklozenge$ ), AspRS- $\Delta$ N20 ( $\square$ ), or AspRS- $\Delta$ N36 ( $\bigcirc$ ) incubated at 0.01 mg/ml (0.1  $\mu$ M) for 30 min, at 4 °C, in 100 mM Tris-HCl, pH 7.5, 10 mM  $\beta$ -mercaptoethanol, containing BSA at 4 mg/ml and variable concentrations of glycerol, was measured in the tRNA<sup>Asp</sup> aminoacylation reaction immediately after dilution in the assay mixture containing 20% glycerol. The relative residual activity refers to the activity determined for each enzyme in the presence of 20% glycerol, prior to incubation in the buffer containing lower glycerol concentrations. *Inset*, time course of AspRS- $\Delta$ N36 inactivation after incubation as above, at 20 °C, in the presence of 20%, 5%, or 0.5% glycerol, as indicated.

AspRS-ΔN36 or AspRS-ΔN20 dissociates (Fig. 3 and Table II) and their enzymatic activity decreases rapidly, with a  $t_{1/2}$  of inactivation of about 7 min (Fig. 6). The rate of enzyme inactivation decreased with increasing protein concentration, to give a  $t_{1/2}$  of inactivation of ~20 min at 0.5 mg·ml<sup>-1</sup> (4.5  $\mu$ M). The substrates ATP or tRNA protect the enzyme against inactivation. In the presence of ATP (Fig. 6) or tRNA<sup>Asp</sup> in the incubation mixture, also containing 10 mM MgCl<sub>2</sub>, AspRS- $\Delta$ N36 remains fully active, even when incubated at low concentration ( $\sim 0.04 \ \mu\text{M}$ ). A similar stabilizing effect was observed for rAspRS and AspRS-ΔN20. Addition of GTP instead of ATP has no stabilizing effect. The substrate ATP stabilizes the dimeric form of the enzyme. Upon chromatography of AspRS- $\Delta$ N20 on a Superose 12 column, as described in Fig. 3, in a buffer containing 5% glycerol, the enzyme behaves as a dimer or as a monomer, respectively, with or without the addition of 3 mm ATP and 10 mM MgCl<sub>2</sub> in the equilibration buffer.

The concentration-dependent inactivation rate observed above for the free, native or  $NH_2$ -terminal truncated forms of AspRS, cannot be explained if considering a simple monomerdimer equilibrium, as shown in Reaction 1.

$$\begin{array}{c} \begin{array}{c} k_1 \\ D \rightleftharpoons 2M \\ k_{-1} \end{array} \\ \end{array} \\ \begin{array}{c} \text{Reaction 1} \end{array}$$

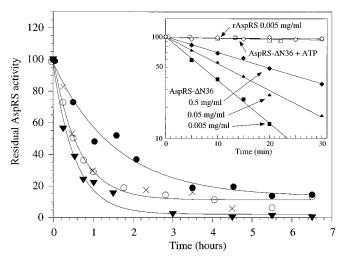


FIG. 6. Concentration dependence of rAspRS and AspRS- $\Delta$ N36 inactivation. The residual activity of rAspRS incubated at 0.5 mg/ml (4.4  $\mu$ M;  $\bigcirc$ ), 0.05 mg/ml (0.44  $\mu$ M;  $\bigcirc$ , ×), or 0.007 mg/ml (0.06  $\mu$ M;  $\checkmark$ ) in 25 mM potassium phosphate, pH 7.5, 200 mM KCl, 10 mM DTE, at 0 °C, was measured in the tRNA<sup>Asp</sup> aminoacylation ( $\odot$ ,  $\bigcirc$ ) or [<sup>32</sup>P]PP<sub>i</sub>–ATP exchange (×) reactions. At the time intervals indicated, the dimermonomer equilibrium was frozen by dilution at 0 °C in the assay mixture containing 20% glycerol and the assay was performed at 0 °C. The curves are simulated time courses using the rate constants given in the text. *Inset*, the residual aminoacylation activity of rAspRS (0.005 mg/ml;  $\diamond$ ), or of AspRS- $\Delta$ N36 incubated at 0.5 ( $\diamond$ ), 0.05 ( $\blacktriangle$ ), or 0.005 ( $\blacksquare$ ,  $\Box$ ) mg/ml in 100 mM Tris-HCl, pH 7.5, 10 mM  $\beta$ -mercaptoethanol, containing BSA at 4 mg/ml and 5% glycerol, with ( $\Box$ ) or without ( $\blacksquare$ ) addition of 10 mM MgCl<sub>2</sub> and 3 mM ATP, at 20 °C, was determined as described above.

Assuming that the dimer is the only active enzyme fraction, the time course of rAspRS inactivation (Fig. 6) can be satisfactorily fitted to the scheme shown by Reaction 2.

$$\begin{array}{ccc} k_1 & k_2 \\ D \rightleftharpoons k_{-1} & 2M \rightleftharpoons 2M^1 \\ k_{-2} \end{array}$$
REACTION 2

M and  $M^R$  refer to monomeric species with or without the ability to dimerize, by using rate constants of  $k_1 = 0.1 \text{ min}^{-1}$ ,  $k_{-1} = 0.09 \ \mu \text{M}^{-1} \cdot \text{min}^{-1}$ ,  $k_2 = 0.05 \text{ min}^{-1}$ ,  $k_{-2} = 0.0015 \text{ min}^{-1}$ . Concerning the NH<sub>2</sub>-terminal truncated AspRS species, the finding that large aggregates form upon incubation in dissociating buffers (Table II) suggests that the monomer is unstable and undergoes conformational changes that favor its aggregation. Accordingly, the simplest mechanism that can account for the experimental data is that shown in Reaction 3.

$$\begin{array}{c} \begin{array}{c} k_1 \\ D \rightleftharpoons 2M \rightleftharpoons k_2 \\ k_{-1} \end{array} & 2M^R \rightarrow 2M^I \rightarrow M^n \\ \end{array}$$
BEACTION 3

M, M<sup>R</sup> and M<sup>I</sup> refer to the native, folded monomer (M) or to monomeric forms with reversible (M<sup>R</sup>) or irreversible (M<sup>I</sup>) conformational changes. M<sup>I</sup> has a tendency to aggregate, leading to multimeric species (M<sup>n</sup>). This mechanism of dissociation/inactivation of AspRS was tested in conditions where no aggregation occurs. Assuming that only D exhibits enzymatic activity, the time course of AspRS- $\Delta$ N20 inactivation (Fig. 7) can be satisfactorily fitted to this scheme by using rate constants of  $k_1 = 0.07 \text{ min}^{-1}$ ,  $k_{-1} = 0.39 \ \mu\text{M}^{-1} \cdot \text{min}^{-1}$ ,  $k_2 = 1.2 \ \text{min}^{-1}$ ,  $k_{-2} = 1 \ \text{min}^{-1}$ , and  $k_3 = 10 \ \text{min}^{-1}$ . The rate-limiting step corresponds to dissociation of the enzyme. The removal of the NH<sub>2</sub>-terminal extension generates a very unstable M<sup>R</sup> species, leading to a monomer M<sup>I</sup>, which is irreversibly inactivated. In the case of

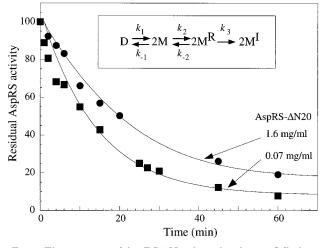


FIG. 7. Time course of AspRS- $\Delta$ N20 inactivation and fitting of experimental data. The residual activity of AspRS- $\Delta$ N20 incubated at 1.6 mg/ml (~15  $\mu$ M; •) or 0.07 mg/ml (~0.6  $\mu$ M; •) in 100 mM Tris-HCl, pH 7.5, 10 mM  $\beta$ -mercaptoethanol, containing BSA at 4 mg/ml and 5% glycerol, at 4 °C, was measured in the tRNA<sup>Asp</sup> aminoacylation reaction after dilution in the assay mixture containing 20% glycerol. The KIN-SIM program (35) was used to fit experimental data to the mechanism shown in the *inset*. The curves are simulated time courses using the rate constants given in the text.

rAspRS, the rate constant  $k_3$  is much lower and the dissociation/inactivation of the enzyme could be readily reversed, as shown below. The calculated theoretical curves shown in Figs. 6 and 7 are in good agreement with the experimental values. Therefore, the concentration dependence of AspRS inactivation is satisfactorily described if assuming that the only active enzyme species is the dimer. Consequently, reactivation of AspRS, requiring a process of bimolecular assembly, should conform to a second-order mechanism.

Dimerization of AspRS and Restoration of Enzyme Activity-Following incubation of rAspRS in an inactivation buffer containing 150 mM KCl, the time course of aspartylation was measured after dilution of at least 40-fold into the assay mixture. When the assay was performed at 25 °C, the amount of Asp-tRNA<sup>Asp</sup> formed did not increase linearly with the reaction time (Fig. 8A,  $\Box$ ). This behavior contrasts with that observed when rAspRS is not incubated in the dissociation buffer prior to the assay (Fig. 8A,  $\blacksquare$ ). The enhancement of the rate of Asp-tRNA<sup>Asp</sup> formation with the reaction time can be best explained if considering that reassociation of rAspRS occurs during the assay, and that the bimolecular reaction  $2M \rightarrow D$  is the rate-limiting step of the reactivation pathway. Accordingly, the initial rates of Asp-tRNA<sup>Asp</sup> formation increase by raising enzyme concentration in the assay (Fig. 8, inset). When the assay was performed at 0 °C (Fig. 8B), in conditions that do not assist the dimerization process, the increase of the reaction rate with time and the concentration dependence of the initial velocity curves, were no longer observed.

Such an enhancement of the aminoacylation rate with the reaction time was also detected when AspRS activity was assayed after incubation of the multisynthetase complex in the inactivation buffer (Fig. 8A,  $\bigcirc$ ). It seems very unlikely that this effect could be due to a general structural perturbation of the complex. Indeed, formation of Ile-tRNA<sup>Ile</sup>, catalyzed by the IleRS component, a monomeric enzyme (36), varied linearly with the reaction time (Fig. 8A,  $\diamond$ ). As in the case of rAspRS, a linear time-course of Asp-tRNA<sup>Asp</sup> formation was observed when the reaction is conducted at 0 °C (Fig. 8*B*); no reactivation/reassociation of Cx-AspRS and Cx-AspRS are very similar, indicating that in both cases, enzyme inactivation/reactivation

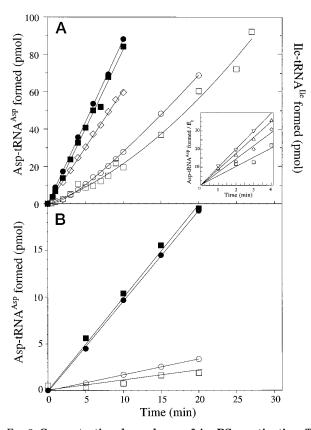


FIG. 8. Concentration dependence of AspRS reactivation. The purified enzyme preparations, rAspRS at 0.009 mg/ml (80 nM) and the rat multisynthetase complex at 0.1 mg/ml (~70 nM; 1 mol of the dimeric Cx-AspRS and of the monomeric Cx-IleRS components per mol of complex), were incubated at 4 °C, for 18 h, in a stabilizing buffer containing 50 mM potassium phosphate, pH 7.5, 2 mM DTE, 20% glycerol, and BSA at 4 mg/ml (closed symbols), or in a dissociating buffer containing 50 mM Tris-HCl, pH 7.5, 150 mM KCl, 2 mM DTE, and BSA at 4 mg/ml (open symbols). A, the time course of the tRNA<sup>Asp</sup> ( $\blacksquare$ ,  $\Box$ ,  $\bullet$ ,  $\bigcirc$ ) or tRNA<sup>Ile</sup> ( $\diamond$ ) aminoacylation reaction catalyzed by the residual enzyme fractions of rAspRS ( $\blacksquare$ ,  $\Box$ ), Cx-AspRS ( $\bullet$ ,  $\bigcirc$ ), and Cx-IleRS ( $\diamond$ ) was determined at 25 °C, after dilution in 100 mM Tris-HCl pH 7.5, 2 mM DTE, 4 mg/ml BSA. Enzyme concentrations in the assay are of about 2 nm. Inset, dependence of the initial rates of Asp-tRNA<sup>Asp</sup> formation on rAspRS concentration in the assay mixture. rAspRS was incubated in the dissociation buffer as described above, and the formation of Asp-tRNA<sup>Asp</sup> was followed in the presence of various rAspRS concentrations in the assay mixture:  $1(\Box)$ ,  $2(\diamond)$ ,  $4(\triangle)$ , or  $8(\bigtriangledown)$  nM. Results are expressed in moles of Asp-tRNA<sup>Asp</sup> formed/mol of enzyme ( $E_t$ ) as a function of time. B, the time course of the tRNA<sup>Asp</sup> aminoacylation reaction catalyzed by the residual rAspRS ( $\blacksquare$ ,  $\Box$ ) or Cx-AspRS ( $\bigcirc$ ,  $\bigcirc$ ) enzyme fractions was determined at 0 °C after appropriate dilution in the assay mixture containing 10% glycerol. Enzyme concentrations in the assay are of about 4 nm.

correlates with dissociation/reassociation of a dimeric AspRS. However, the initial rate of Asp-tRNA<sup>Asp</sup> formation, catalyzed at 25 °C or 0 °C by Cx-AspRS previously incubated in dissociating conditions, increases linearly with the enzyme concentration (in the range of 2–60 nm). As already mentioned above, Cx-AspRS is not released from the complex after incubation in the presence of 0.2 m KCl. Therefore, association of two AspRS monomers, which remain bound to the complex, is an intramolecular process, and reactivation follows an apparent first-order mechanism independent of enzyme concentration.

#### DISCUSSION

The  $NH_2$ -terminal extension of rat AspRS, as well as its association within a multienzyme complex, are dispensable *in vitro* for the catalytic activity, as assessed by the determination of the Michaelis parameters for the complex-associated, the free native recombinant, and the free  $NH_2$ -terminal truncated forms, which proved to be indistinguishable. This study provides additional evidence for the concept that eukaryotic synthetases possess additional autonomous domains not found in their prokaryotic counterparts, and endowed with a functional role unrelated to catalysis. The facultative character of those extra-domains was reported for other eukaryotic AspRS (23, 37), as well as for other synthetases from yeast or mammals (38–43).

The ability of rat AspRS to dissociate under nondenaturing conditions provided a suitable tool to study the structure/function relationships for an AspRS species. The dissociation of rat AspRS and the concomitant loss of both amino acid activation and tRNA aminoacylation activities establish for the first time that dimerization of AspRS, a representative of class IIb synthetases, is absolutely required for its catalytic activity. Our results are in agreement with those previously reported for a class IIa synthetase, Pro-tRNA synthetase from E. coli, where dimerization has also been shown to be required for activity (44). Therefore, oligomerization seems to be a key element for enzyme activity catalyzed by class II aminoacyl-tRNA synthetases with an  $\alpha_2$ -type structure, although the two active sites of AspRS simultaneously bind ATP or the analog AMP-PCP (4). In addition, two tRNA<sup>Asp</sup> molecules are bound per dimeric enzyme, each tRNA molecule involving a single monomer (45). This situation contrasts with that observed for a class IIa dimeric enzyme, Pro-RNA synthetase from T. thermophilus (46), where the variable arm and the T $\Psi$ C loop of a tRNA molecule bind to one monomer, whereas aminoacylation of that tRNA molecule is performed into the active site of the other monomer. In that particular case, structural data provide a rational explanation for the absolute requirement of dimerization for activity.

Monomeric species of bacterial and yeast AspRS have not been obtained so far in nondenaturing conditions (37, 47), and a strict relationship between activity and the  $\alpha_2$  conformation could not be established. However, experimental evidence supporting the structural importance of dimerization of yeast AspRS on the conformation of its active site has been reported (13). Mutation of the conserved  $Pro^{273}$  residue of motif 1, located at the dimeric interface, impaired the catalytic activity of the enzyme but did not abolish dimerization. The catalytic capacity of a mutant subunit was found to be restored into heterodimers formed by association of a mutant subunit with a wild-type subunit. The effect of dimerization was interpreted in terms of a structural constraint that helps to fold the two active sites into an active conformation, in a cooperative manner. Accordingly, structural data reveal that the two active sites are closely related, through a network of hydrogen bonds and salt bridges involving the side chains of several residues from motifs 1, 2, and 3 of both subunits. Although these residues are located at the dimeric interface, especially those from motif 1, several observations suggest that they do not have a pre-eminent role in the stabilization of the  $\alpha_2$  dimers. (i) Well conserved motifs 1, 2, and 3 have been unambiguously identified in two class II enzymes, eukaryotic Ala-tRNA synthetase and veast mitochondrial Phe-tRNA synthetase, which are functional monomers (15, 16). (ii) A mutation in motif 1 lowered the activity of the enzyme but did not affect its dimerization capacity (13). (iii) In this study, we have shown that rat AspRS displays a dimer dissociation constant at least 1 order of magnitude above that of the yeast enzyme, although their motifs 1 are very similar (17). The NH<sub>2</sub>-terminal truncated derivatives are very unstable dimers. It is therefore very likely that the NH<sub>2</sub>-terminal extension of the eukaryotic enzymes plays a major role in the stabilization of active dimers. These extensions are absent from the prokaryotic AspRS. In that case, the stability of the dimeric interface is contributed by an additional COOH-terminal domain (17). The possibility to easily monitor the dimer/monomer equilibrium provided the opportunity to correlate the activity of AspRS and its oligomeric state and led to the identification of an intimate connection between the formation of the dimer and the emergence of a productive 4156-4167

conformation of the active site. (i) Dissociation of the dimer correlates with the loss of amino acid activation and tRNA aminoacylation activities. (ii) In the presence of ATP, the inactivation rate is much slower and the two subunits remain associated. These results suggest that the productive conformation of the active site requires structural rearrangements that are induced by dimerization, and correlatively that the presence of ATP in the active site stabilizes its productive conformation and therefore stabilizes the dimeric conformation of the enzyme. Accordingly, one would expect that the two active sites are functionally related. The finding that a LystRNA synthetase mutant affected in its dimeric interface displays cooperativity toward lysine binding strongly supports this hypothesis (14).

The crystal structure of T. thermophilus and S. cerevisiae AspRS provided a detailed description of the dimeric interface (17). In addition to residues from the three conserved motifs of class II synthetases, which are highly conserved in all known AspRS species, interchain interactions involve four major contact areas: the long helix H3 and strand A1 from the beginning of the COOH-terminal catalytic domain, a four-stranded antiparallel  $\beta$ -sheet involving strands I1 and I2 from the two monomers, helix H2b from the hinge region of one monomer that interacts with helices H4 and H11 from the catalytic domain of the other subunit, and the NH<sub>2</sub>-terminal anticodon-binding domain of one monomer and the catalytic domain of the other. Because of the close homology between the various AspRS species in these regions, a rational explanation for the lesser stability of the mammalian enzyme awaits accurate structural investigations. The availability of large amounts of purified rat AspRS opens the way for crystallization studies. However, one may already speculate that the NH2-terminal polypeptide extension that characterizes all the eukaryotic AspRS species, whose structural arrangement could not be defined in the crystal structure of the yeast enzyme, might be implicated in the stabilization of the  $\alpha_2$  dimer, through direct contacts with the other subunit or through the stabilization of the  $\rm NH_2\text{-}terminal$ anticodon binding domain of one subunit into a conformation that favors its interaction with the catalytic domain of the other subunit.

Acknowledgments-We are indebted to J. Rossier and J.-P. Le Caer for the protein sequence analyses, to G. Batelier for performing the analytical ultracentrifuge experiments, to D. Moras for the gift of a sample of yeast AspRS, and to D. Pompon for the pYeDP60 vector. We gratefully acknowledge the excellent technical assistance of M.-T. Latreille during part of this work.

#### REFERENCES

- 1. Schimmel, P., and Söll, D. (1979) Annu, Rev. Biochem. 48, 601-648
- 2. Eriani, G., Delarue, M., Poch, O., Gangloff, J., and Moras, D. (1990) Nature 347.203-206
- 3. Cavarelli, J., and Moras, D. (1993) FASEB J. 7, 79-86

- 4. Cavarelli, J., Eriani, G., Rees, B., Ruff, M., Boeglin, M., Mitschler, A., Martin, F., Gangloff, J., Thierry, J.-C., and Moras, D. (1994) EMBO J. 13, 327-337
- 5. Belrhali, H., Yaremchuk, A., Tukalo, M., Larsen, K., Berthet-Colominas, C., Leberman, R., Beijer, B., Sproat, B., Als-Nielsen, J., Grübel, G., Legrand, J.-F., Lehmann, M., and Cusack, S. (1994) Science 263, 1432-1436
- 6. Onesti, S., Miller, A. D., and Brick, P. (1995) Structure (London) 3, 163-176 7. Mosyak, L., Reshetnikova, L., Goldgur, Y., Delarue, M., and Saffro, M. G.
- (1995) Nat. Struct. Biol. 2, 537-547 8. Arnez, J. G., Harris, D. C., Mitschler, A., Rees, B., Francklyn, C. S., and Moras,
- D. (1995) EMBO J. 14, 4143-4155 9. Logan, D. T., Mazauric, M.-H., Kern, D., and Moras, D. (1995) EMBO J. 14,
- 10. Madern, D., Anselme, J., and Härtlein, M. (1992) FEBS Lett. 299, 85-89
- 11. Davis, M. W., Buechter, D. D., and Schimmel, P. (1994) Biochemistry 33,
- 9904-9911
- 12. Lu, Y., and Hill, K. A. W. (1994) J. Biol. Chem. 269, 12137-12141
- Eriani, G., Cavarelli, J., Martin, F., Dirheimer, G., Moras, D., and Gangloff, J. (1993) Proc. Natl. Acad. Sci. U. S. A. 90, 10816-10820
- 14. Commans, S., Blanquet, S., and Plateau, P. (1995) Biochemistry 34, 8180-8189
- 15. Ripmaster, T. L., Shiba, K., and Schimmel, P. (1995) Proc. Natl. Acad. Sci. U. S. A. 92, 4932–4936
- 16. Sanni, A., Walter, P., Boulanger, Y., Ebel, J.-P., and Fasiolo, F. (1991) Proc. Natl. Acad. Sci. U. S. A. 88, 8387–8391
- Delarue, M., Poterszman, A., Nikonov, S., Garber, M., Moras, D., and Thierry, J. C. (1994) *EMBO J.* 13, 3219–3229
- 18. Mirande, M. (1991) Prog. Nucleic Acid Res. Mol. Biol. 40, 95-142
- 19. Mirande, M., and Waller, J.-P. (1989) J. Biol. Chem. 264, 842-847 20. Jacobo-Molina, A., Peterson, R., and Yang, D. C. H. (1989) J. Biol. Chem. 264, 16608 - 16612
- 21. Reed, V. S., and Yang, D. C. H. (1994) J. Biol. Chem. 269, 32937-32941
- 22. Mirande, M., Lazard, M., Martinez, R., and Latreille, M.-T. (1992) Eur. J. Biochem. 203, 459-466
- Escalante, C., and Yang, D. C. H. (1993) J. Biol. Chem. 268, 6014-6023 23.
- 24. Urban, P., Cullin, C., and Pompon, D. (1990) Biochimie (Paris) 72, 463-472
- 25. Sambrook, J., Fritsch, E. F., and Maniatis, T. (1989) Molecular Cloning: A Laboratory Manual, 2nd Ed., Cold Spring Harbor Laboratory, Cold Spring Harbor, NY
- 26. Sanger, F., Nicklen, S., and Coulson, A. R. (1977) Proc. Natl. Acad. Sci. U. S. A. 74. 5463-5467
- 27. Ito, N., Fukuda, Y., Murata, K., and Kimura, A. (1993) J. Bacteriol. 153, 163 - 168
- 28. Gornall, A. G., Bardawill, C. J., and David, M. M. (1949) J. Biol. Chem. 177, 751-766
- 29. Kellermann, O., Brevet, A., Tonetti, H., and Waller, J.-P. (1979) Eur. J. Biochem. 99, 541–550
- Laemmli, U. K. (1970) Nature 227, 680-685
   Laue, T. M., Shah, B. D., Ridgeway, T. M., and Pelletier, S. L. (1992) in Analytical Ultracentrifugation in Biochemistry and Polymer Science (Harding, S. E., Rowe, A. J., and Horton, J. C., eds) pp. 90-125, Royal Society of Chemistry, Cambridge, United Kingdom
- 32. Romanos, M. A., Scorer, C. A., and Clare, J. J. (1992) Yeast 8, 423-488
- 33. Cirakoglu, B., and Waller, J.-P. (1985) Biochim. Biophys. Acta 829, 173-179
- 34. Huang, S., Elliott, R. C., Liu, P.-S., Koduri, R. K., Weickmann, J. L., Lee, J.-H., Blair, L. C., Ghosh-Dastidar, P., Bradshaw, R. A., Bryan, K. M., Einarson, B., Kendall, R. L., Kolacz, K. H., and Saito, K. (1987) Biochemistry 26, 8242-8246
- 35. Barshop, B. A., Wrenn, R. F., and Frieden, C. (1983) Anal. Biochem. 130, 134 - 145
- 36. Lazard, M., Mirande, M., and Waller, J.-P. (1985) Biochemistry 24, 5099-5106 37. Eriani, G., Prevost, G., Kern, D., Vincendon, P., Dirheimer, G., and Gangloff,
- J. (1991) Eur. J. Biochem. 200, 337-343
- 38. Cirakoglu, B., and Waller, J.-P. (1985) Eur. J. Biochem. 151, 101-110 39. Vellekamp, G., Sihag, R. K., and Deutscher, M. P. (1985) J. Biol. Chem. 260,
- 9843-9847 40. Walter, P., Weygand-Durasevic, I., Sanni, A., Ebel, J.-P., and Fasiolo, F. (1989) J. Biol. Chem. 264, 17126-17130
- 41. Martinez, R., and Mirande, M. (1992) Eur. J. Biochem. 207, 1-11
- 42. Ludmerer, S. W., Wright, D. J., and Schimmel, P. (1993) J. Biol. Chem. 268, 5519-5523
- 43. Bec, G., Kerjan, P., and Waller, J.-P. (1994) J. Biol. Chem. 269, 2086-2092
- 44. Lee, M.-L., and Muench, K. H. (1969) J. Biol. Chem. 244, 223-230
- 45. Cavarelli, J., Rees, B., Ruff, M., Thierry, J.-C., and Moras, D. (1993) Nature **362,** 181–184
- 46. Biou, V., Yaremchuk, A., Tukalo, M., and Cusack, S. (1994) Science 263, 1404 - 1410
- 47. Eriani, G., Dirheimer, G., and Gangloff, J. (1990) Nucleic Acids Res. 18, 7109-7118