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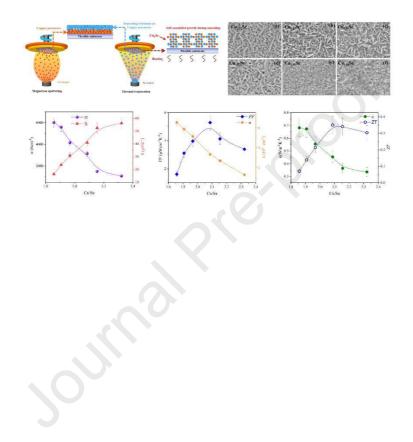
## **Credit Author Statement**

Xiao-lan Huang: Methodology, Data curation, Investigation. Dong-wei Ao: Investigation, Writing- Original draft preparation. Tian-bao Chen: Visualization, Investigation. Yue-xing Chen: Formal analysis. Fu Li: Resources. Shuo Chen: Validation. Guang-xing Liang: Formal analysis, Resources. Xiang-hua Zhang: Writing- Reviewing and Editing. Zhuang-hao Zheng: Conceptualization, Supervision, Writing - Review & Editing. Ping Fan: Supervision.

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# **Graphical Abstract**

The well-crystallized Cu<sub>2</sub>Se thermoelectric thin films have been synthesized by an efficient layer-by-layer combination reaction method, and the high power factor of 5.3  $\mu$ Wcm<sup>-1</sup>K<sup>-2</sup> and *ZT* value of 0.35 with low thermal conductivity are achieved at room-temperature by optimizing the composition.



# High-performance copper selenide thermoelectric thin films for flexible thermoelectric application

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**Abstract:** Flexible thermoelectric (TE) materials have wide application in fabricating portable/wearable devices due to the advantages of being bendable, small-size and light-weight. Inorganic based flexible thin films have attracted much attention due to their high TE performance. So far, developing high TE properties and environment friendly flexible thin films for practical applications are still a considerable challenge. Copper selenide (Cu<sub>2</sub>Se) is a non-toxic and low-cost material, providing relatively safe TE modules for thin film devices. In this work, Cu<sub>2</sub>Se based thin films with single crystal phase have been fabricated successfully at flexible substrate by an efficient layer-by-layer combination reaction method. By optimizing the composition, high power factor of 5.3  $\mu$ Wcm<sup>-1</sup>K<sup>-2</sup> and *ZT* value of 0.35 with low thermal conductivity are achieved at room-temperature. By bending tests, it is revealing that the prepared thin films possess well flexibility and the designed flexible device

displays stable out-put power that demonstrate the potential of inorganic thermoelectric materials to be mountable on flexible/wearable substrates for energy harvesting and management devices.

Keywords: Thermoelectric materials; Thin film; Cu<sub>2</sub>Se; Flexible

## **1. Introduction**

Thermoelectric conversion is a solid-state, secure, convenient and environment friendly energy conversion technology allowing the direct conversion of heat into electricity and vice versa, which has broad applications in cooling, power generation, and waste heat recovery.<sup>[1-4]</sup> The TE performance of thermoelectric materials is evaluated by the dimensionless figure of merit (ZT),  $ZT = S^2 \sigma T \kappa^{-1}$ , where S is the Seebeck coefficient,  $\sigma$  is the electrical conductivity,  $\kappa$  is the thermal conductivity, and T is the absolute temperature. Besides,  $S^2\sigma$  is called as power factor (*PF*), which is used to estimate the electronic properties.<sup>[5]</sup> Recently, the explosive growth of portable and wearable microelectronics has stimulated the development of flexible thermoelectric materials promising for the applications in wearable devices as the self-power generator from the temperature difference between the skin and the ambient environment.<sup>[6]</sup> Therefore, fabricating high performance flexible TE materials have received great attention and significant work has been reported in the recent years.<sup>[7-12]</sup> Due to the high performance of inorganic materials, research on depositing inorganic materials on flexible substrates such as fibers, organics, and carbon nanotubes have been widely reported.<sup>[13-17]</sup> Until now, most of the flexible thin films and devices were fabricated based on the classic Te-based materials because of their

excellent TE properties. <sup>[18-19]</sup> However, Te-based materials remain a significant challenge for large-scale commercialization due to the presence of less earth-abundant, toxic, and costly elements. Thus, developing low-cost and nontoxic TE thin films with acceptable performance to substitute Te-based materials is highly desired.

Recently, transition metal selenium compound semiconductor materials <sup>[20-24]</sup>, especially the copper selenide  $(Cu_2Se)^{[25]}$ , have attracted great attention for thermoelectrics. Nontoxic Cu<sub>2</sub>Se has excellent thermoelectric properties that a relatively large ZT value over 2.0 has already been achieved  $^{[26]}$ , making it extremely promising as a potential alternative for the traditional Te-based materials. Thus, fabricating high-performance Cu<sub>2</sub>Se thin films have also been widely studied and various methods, including solution <sup>[27-28]</sup>, electrodeposition <sup>[29]</sup>, pulsed laser deposition <sup>[30]</sup>, ion beam sputtering deposition <sup>[31]</sup>, reactive evaporation <sup>[32]</sup>, galvanic deposition <sup>[33]</sup>, and pulsed/magnetron sputtering reactive <sup>[34]</sup>, have been used. For instance, Lin et al. used solution method to prepare Cu<sub>2</sub>Se based thin film, resulting maximum *PF* of 4.6  $\mu$ Wcm<sup>-1</sup>K<sup>-2</sup> on the flexible polyimide substrate, respectively.<sup>[27]</sup> Scimeca et al. optimized the carrier concentration of Cu<sub>2</sub>Se based thin films to significantly improve their TE properties through a soaking process in Cu<sup>+</sup> ion solution.<sup>[28]</sup> However, the reported Cu<sub>2</sub>Se thin films to date usually exhibit much lower TE properties than that of the bulk Cu<sub>2</sub>Se.

Magnetron sputtering and thermal evaporation methods are commonly used to prepare thin films, especially for substances which are extremely susceptible to oxidation. Based on the advantages of these two technologies, it is expected to produce

high-performance Cu<sub>2</sub>Se thin films by using magnetron or evaporation method. However, some unavoidable issues limit the use of these two technologies, i.e. Se has much high saturated vapor pressure and cannot be efficiently controllable during the sputtering process; Cu is difficult to evaporate because of its high boiling point. Therefore, there are only a few reports about Cu<sub>2</sub>Se thin films prepared by magnetron or evaporation. In this work, an efficient layer-by-layer combination reaction method by integrating the magnetron sputtering and thermal evaporation technology is developed to prepare flexible Cu<sub>2</sub>Se thin films. The effect of composition content on the microstructure and thermoelectric properties of the flexible Cu<sub>2</sub>Se thin films were systematically investigated.

## 2. Experimental Section

The process of the Cu<sub>2</sub>Se based thin film fabrication is represented by schematic diagram in Fig. 1. At first, Cu precursor layer was sputtered by using magnetron sputtering deposition method and the flexible Kapton polyimide film (200HN) are used as the substrate which can be used at the temperatures up to 200 °C. The sputtering power, background pressure and working pressure were kept at 15 W,  $8\times10^{-4}$  Pa and 1.6 Pa, respectively. The Cu layer is controlled to be 100 nm with the sputtering rate of 1 Ås<sup>-1</sup>. After completing the Cu deposition, the Se layer was continued to deposit on the Cu precursor layer by using thermal evaporation deposition method at another deposition system. The working current of the evaporation source was 60 A with the background pressure of  $5.0\times10^{-4}$  Pa. The average deposition rate of Se is ~ 0.8 Ås<sup>-1</sup>, which was controlled using a quartz

microbalance, and the thickness of Se layer is varied from 70 nm to 120 nm. Finally, all the thin films with different Se contents were annealed at 200  $\Box$  for one hour using a heating system in a glove box under nitrogen atmosphere.

Crystal structure of the thin films are analyzed by X-ray diffraction (XRD, Ultima-iv) with the 2 $\theta$  angle range from 10° to 70°, using CuK<sub>a</sub> radiation (under operation requirements of 40 mA and 40 kV). Surface topography of the Cu<sub>2</sub>Se based thin film are investigated using a scanning electron microscope (SEM, Supra 55) with an electron beam accelerated of 5 kV. Composition is investigated by using an energy dispersive X-ray spectroscope (EDS, Bruker Quantax 200) equipped with the SEM. Transmission electron microscopy (TEM, JEM-3200FS) is used to measure the micro-structure of the thin films. The chemical valence was characterized by conducting X-ray photoelectron spectroscopy (XPS, Escalab 250Xi) with a monochrome Al  $K_{\alpha}$  X-ray source of 1486.6 eV. Before test, samples were plasma etched for 30 seconds to remove impurities on the surface of the thin film. The total thickness of the thin films was measured by a Dektak3 ST surface-profile measurement system and shown in Table S1 (Supporting information). The mobility and carrier density are investigated by Van der Pauw Hall measurement (HL5500PC, Nanometrics). In-plane electrical conductivity and Seebeck coefficient are measured by the Seebeck-conductivity coefficient measurement system (SBA458, Netzsch). Thermal conductivity at room temperature is determined by a transient hot-wire theory method (TC3000, Xiatech). The current-voltage (I-V) curves of the thermoelectric device of thin films are measured by a Keithley 2400.

#### 3. Results and analysis

The composition content of the Cu-Se thin films was analyzed by EDS as shown in Table 1. The Cu: Se atomic ratio is close to 2:1 when the Se layer thickness is 100 nm, and show slightly Cu-excess and Se-excess when the Se thickness is less than and higher than 100 nm, respectively. For better discussion, the prepared Cu-Se thin films are named as Cu<sub>2.32</sub>Se, Cu<sub>2.15</sub>Se, Cu<sub>2.09</sub>Se, Cu<sub>1.97</sub>Se, Cu<sub>1.91</sub>Se and Cu<sub>1.86</sub>Se, corresponding to the Cu atomic ratio in the thin films. Figure 2 shows the phase structural of the thin films with different Cu content measured by XRD. After comparing to the XRD pattern (PDF#37-1187) of the powder Cu<sub>2</sub>Se, it confirms that all the diffraction peaks of the samples belong to the cubic structure of Cu<sub>2</sub>Se without impurity peaks, suggesting that the samples have well-crystallinity with single Cu<sub>2</sub>Se phase. Moreover, according to the Debye–Sheerer equation <sup>[35]</sup>, the estimated grain size of the thin films (Fig. S1, supporting information) calculated using the XRD data is around 25 nm, indicating the nano-crystallinity of the samples. The binding states of Cu and Se in the  $Cu_{2.09}$ Se and  $Cu_{1.97}$ Se samples are investigated by XPS. As shown in Fig. 3(a), the core level spectrum of  $Cu_{2.09}$ Se sample reveals that two strong peaks are obtained at ~933 eV of Cu  $2p_{3/2}$  and ~953 eV of Cu  $2p_{1/2}$ , which are well matched with the spin-orbit phenomena of compound state of Cu (I). Similar observation for Cu of Cu<sub>1.97</sub>Se sample is obtained. Thus, the results suggest that both the Cu-excess and Cu-deficiency samples have stable Cu<sup>1+</sup> state <sup>[36]</sup>. A broad peak ranging from 53-57 eV is observed from the samples as shown in Fig. 3(b). It can be identified into two symmetric peaks to Se  $3d_{5/2}$  and Se  $3d_{3/2}$  respectively located at 54.5 eV and 55.7

eV, which is the characteristic shape of compound Se (–II) in a consistent bonding environment.<sup>[36]</sup> Additionally, no high-valence copper or selenium related to oxidation state can be observed from the XPS patterns, which can be attribute to the fact that the thin film is prepared in a vacuum and heat treated is also carried in a low-oxygen environment. Combining the XRD and EDS results, it can be concluded that the prepared Cu<sub>2</sub>Se thin films have few defects and well-crystallinity, which is beneficial to obtain high TE performance.

The surface morphology of the samples is shown in Fig. 4. Nano-sheet-liked particles are observed on the surface of the Cu-excess samples as shown in Fig. 4 (a), (b) and (c), indicating the flake structure of the thin films. The length of the nano-particles decreases with the decrease of Cu content, while the width increases. Interestingly, compact particles without obvious shape structure are obtained from the Cu<sub>1.91</sub>Se and  $Cu_{1.86}$ Se samples. As shown from the XRD results that the peak intensity increases and some addition peaks are observed after change the composition ratio, indicating the change of thin film growth structure. Thus, such difference of morphology is anticipated due to the different stacked growth of the grains at different composition ratio. High-resolution TEM (HRTEM) analysis of the sample are performed to further investigate the microstructure of the thin films. Figure 5(a) displays the energy dispersive X-ray spectroscope mapping images of the Cu<sub>1.97</sub>Se sample and indicates that the Cu and Se have uniform distribution. Figure 5(b) shows the high-resolution TEM image and the measured lattice spacing is 3.42 Å, which correspond to the (060) plane of the  $Cu_2Se$ . The  $Cu_{2,09}Se$  sample also displays uniform distribution of the Cu

and Se as shown in Fig. 5(c). Figure 5(d) displays that crystalline nanoparticles stack compactly with each other, and the measured lattice spacing is 3.42 Å, which correspond to the (060) plane of the Cu<sub>2</sub>Se, suggesting the well-crystallinity of the sample. Interestingly, no Cu related crystal clusters are observed from the mapping image which is agreed with the XPS and XRD results. In general case, the ex-excess Cu might as the liquid-liked state and exists in the crystal grain's <sup>[37]</sup> which can also be seen in the Fig. 5(d).

The room-temperature electrical conductivity  $\sigma$  and Seebeck coefficient S are shown in Fig. 6(a). With the increase of Cu content, the  $\sigma$  decreases, which is attributed to the decrease of carrier concentration as shown in Fig. 6(b). Conversely, the mobility (Table S2, supporting information) increases when the Cu content increases, that benefits to obtain high S coefficient, and the maximum value can be achieved to be over  $55\mu VK^{-1}$ . Power factor of the samples are calculated and shown in Fig. 6(b). It increases firstly, and reaches to a maximum value of 5.3  $\mu$ Wcm<sup>-1</sup>K<sup>-2</sup> from the Cu<sub>2.09</sub>Se sample, then decreases when the Cu content continued to raise. Nevertheless, the achieved maximum power factor can be compared to one of the highest values of flexible Cu<sub>2</sub>Se thin films prepared by other method. <sup>[38]</sup> The in-plane thermal conductivity of samples is shown in Fig. 6(c) and displays the decreasing trend with the increasing Cu content. Interestingly, the excess-Cu samples have low thermal conductivity  $\kappa$ . Based on the micro-structure results as mentioned above, it can be attributed as the nano-structure of the thin films, that scatter the mid- and long- length phonons more efficiently. The estimated ZT in Fig. 6(c) demonstrates that the

Cu-excess samples have higher values than that of the Se-excess samples due to the higher *S* and lower  $\kappa$ . The maximum *ZT* of 0.35 is achieved from the Cu<sub>2.09</sub>Se sample which is close to the highest value at room-temperature of the bulk Cu<sub>2</sub>Se materials, <sup>[39-41]</sup> indicating the effectiveness of this innovative technique for preparing high-performance flexible Cu<sub>2</sub>Se TE thin film.

The mechanical property of the flexible thin film is measured by a bending resistance system of which the schematic diagram and optical image are shown in Fig. 7(a) and (b). The relative resistance change ( $\Delta R/R_0 = (R'-R_0)/R_0$ ) was applied to quantitatively investigate the mechanical stability of the thin films, where  $R_0$ , R' and  $\Delta R$  were the original resistance, bending resistance and difference between R' and  $R_0$ , respectively.<sup>[42]</sup> The influence of bending radius (3.2 mm, 3.8 mm, 4.1 mm, 4.8 mm, 5.3 mm, 5.8 mm, 6.2 mm, 6.5 mm) as well as the bending cycles (200, 400, 600, 800) on the  $\Delta R/R_0$  of the flexible thin films are shown in Fig. 7(c). It can be seen that the  $\Delta R/R_0$  is slightly increased with the increase of bending radius and bending cycles. The maximum difference of  $\Delta R/R_0$  is about 3 % when the bending cycles increased to be 800 with the radius of 6.5 mm, which is close to the value reported from the literatures <sup>[43]</sup>, suggesting that the thin film has well flexible properties.

In order to show the potential energy-harvesting applications of the present copper selenide thin film for wearable electronic devices, a simple TE device is fabricated. The structure diagram and optical image of the fabricated device are shown in Fig. 7(d) and (e). The device consists of 10 pieces of the thin film and all legs were connected in series by painting silver paste. Figure 7(f) shows the current-voltage (*I-V*)

curves and current-power (*I-P*) curves of the device at different temperature difference ( $\Delta T = T_{hot} - T_{cold}$ ). As shown in Fig. 7(f), the open circuit voltage ( $U_{oc}$ ) was 0.96 mV, 1.09 mV, 2.01 mV, and 3.84 mV corresponding to  $\Delta T = 23 \Box$ , 28 $\Box$ , 33 $\Box$  and 38 $\Box$ , respectively. The measured  $U_{oc}$ s are basically consistent with the results that were calculated using the Seebeck effect.

The output power *P* is calculated by the equation as follows:<sup>[44]</sup>

$$P = I^{2} \mathbf{R}_{load} = \left(\frac{U_{oc}}{\mathbf{R}_{in} + \mathbf{R}_{load}}\right)^{2} \mathbf{R}_{load}$$
(1)

Where *I*,  $R_{load}$  and  $R_{in}$  are output current, the external resistance and the internal resistance respectively. In addition, the power density (*P*<sub>D</sub>) is calculated by the following equation:<sup>[44]</sup>

$$P_D = \frac{P_{max}}{N \cdot A} = \frac{P_{max}}{N \cdot w \cdot t}$$
(2)

Where *N*, *A*, *w*, and *t* is the number of the legs, the cross-sectional area of the leg, the width of the leg, the thickness of the thin film, respectively. As shown in Fig. 3(f), the output power has the highest value 3.31 nW when the output current was 1.84  $\mu$ A under the temperature difference of 38  $\Box$ . When  $\Delta$ T= 23  $\Box$ , 28  $\Box$ , 33  $\Box$  and 38  $\Box$ , the calculated maximum *P*<sub>D</sub> was 56.14 mWm<sup>-2</sup>, 75.43 mWm<sup>-2</sup>, 226.32 mWm<sup>-2</sup> and 580.70 mWm<sup>-2</sup>, respectively. The maximum *P*<sub>D</sub> is comparable with that of some previous reports <sup>[45-47]</sup>.

## 4. Conclusions

In this work, flexible Cu<sub>2</sub>Se thin films with various atomic ratios were prepared by using a simple layer-by-layer combination reaction method. The self-assembled

growth of single  $\alpha$ -phase Cu<sub>2</sub>Se thin films is successfully realized after annealing. Although the electrical conductivity decreases with an increase of Cu content but a significant increase in the Seebeck coefficient is observed due to the decrease of carrier concentration. It is observed that the Cu-excess thin films exhibit low thermal conductivity, which is attributed to the nano-plate-liked structure and ultimately high *ZT* of 0.35 with well flexibility Cu<sub>2</sub>Se thin film is achieved at room temperature. Additionally, the prepared thin film exhibits good flexibility with about 97 % of the original electrical conductivity retention after 800 bending cycles and the fabricated device has stable output power with the maximum power density of 580.70 mWm<sup>-2</sup>. Therefore, the competitive performance of the devices reported in this work has the potential for portable/wearable electronic applications in the field of thermoelectric devices.

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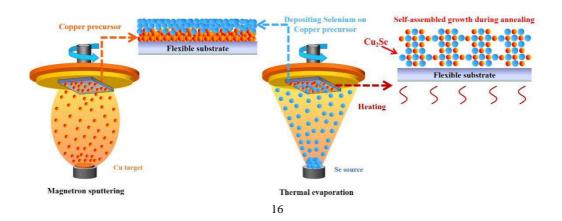
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Table 1 Composition content of the Cu-Se thin film as function of Se layer thickness

Se layer thickness (nm)	70	80	90	100	110	120
Cu (at.%)	69.9	68.3	67.6	66.3	65.6	65.0
Se (at.%)	30.1	31.7	32.4	33.7	34.4	35.0
Cu : Se	2.32:1	2.15:1	2.09:1	1.97:1	1.91:1	1.86:1



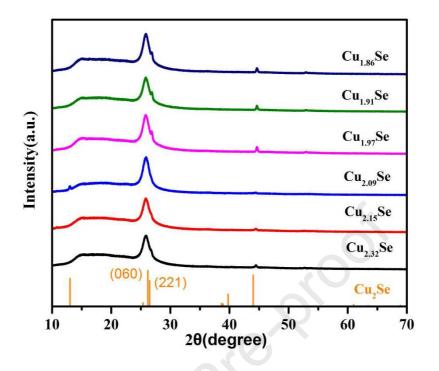


Fig. 1. The schematic diagram of fabricating the Cu<sub>2</sub>Se thin film

Fig. 2 XRD data of flexible Cu<sub>2</sub>Se thin films with different composition ratios

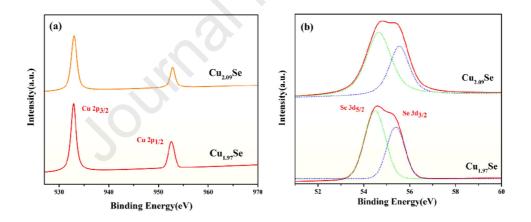


Fig. 3 XPS data of flexible  $Cu_2Se$  thin films with a composition ratio of the 1.97:1 and 2.09:1, (a) Cu; (b) Se

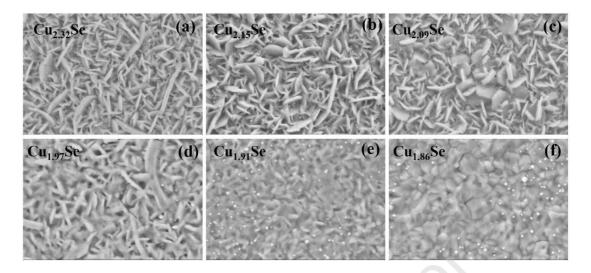


Fig. 4 SEM images of flexible Cu<sub>2</sub>Se thin films with different composition ratio (a) Cu<sub>2.32</sub>Se; (b) Cu<sub>2.15</sub>Se; (c) Cu<sub>2.09</sub>Se; (d) Cu<sub>1.97</sub>Se; (e) Cu<sub>1.91</sub>Se; (f) Cu<sub>1.86</sub>Se

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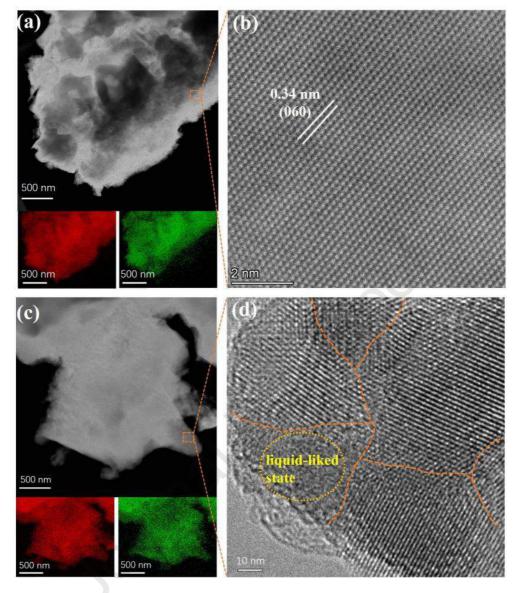


Fig. 5 (a) TEM characterization and EDS elemental mappings of the  $Cu_{1.97}$ Se thin film. (b) Lattice fringes in (a) performed by HRTEM. (c) TEM characterization and EDS elemental mappings of the  $Cu_{2.09}$ Se thin film. (d) Lattice fringes in (c) performed by HRTEM.

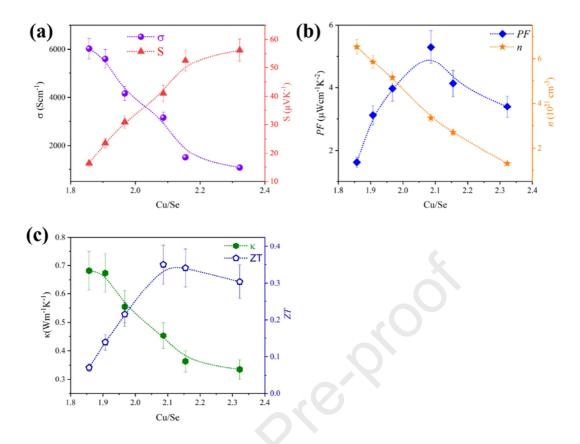


Fig. 6 The thermoelectric properties of flexible  $Cu_2Se$  thin films as function of atomic ratios (a) electrical conductivity and Seebeck coefficient; (b) Power factor and carrier concentration; (c) thermal conductivity and *ZT* 

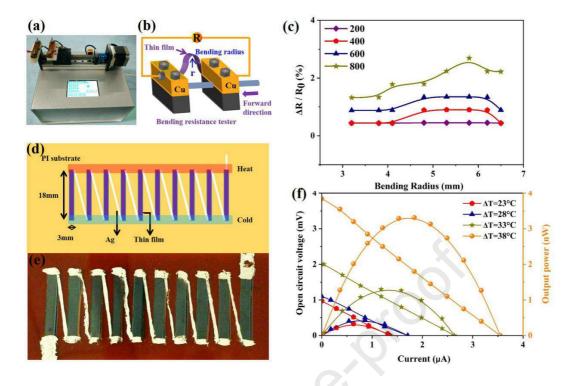


Fig. 7. (a) Optical image and (b) the schematic diagram of the bending system. (c) the influence of bending cycle and bending radius on the resistance difference of the  $Cu_{2.09}Se$  thin film. (d) the structure diagram and (e) optical image of the fabricated device. (f) open voltage and output power as function of current of the device.

# Highlights

- layer-by-layer combination reaction method for preparing high-performance Cu<sub>2</sub>Se thin films.
- (2) The self-assembled growth of Cu<sub>2</sub>Se with single crystal phase were successfully fabricated.
- (3) High power factor of 5.3  $\mu$ Wcm<sup>-1</sup>K<sup>-2</sup> and *ZT* value of 0.35 are achieved at room-temperature.
- (4) Thin film possesses well flexibility and the designed flexible device displays stable out-put power.

# **Declaration of interest statement**

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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