

# Shallow urban aquifers under hyper-recharge equatorial conditions and strong anthropogenic constrains. Implications in terms of groundwater resources potential and integrated water resources management strategies

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1	Shallow urban aquifers under hyper-recharge equatorial conditions and
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26

#### 27 Abstract

28 Humid equatorial regions are recognized as the least documented in term of hydrogeological 29 functioning of aquifers despite the fact that they house a lot of developing countries and that 30 groundwater is often the main water resource. Regarding this aspect, a study was conducted in 31 sub-Saharan Africa, focusing on the Mio-Pliocene aquifer in Douala megacity (Cameroon) 32 which is the rainiest city in West-Africa (about 4000 mm/year) with one of the greatest 33 demographic growth rate of the African continent. Firstly, groundwater recharge rate has been 34 calculated through water balance and Water Table Fluctuation methods. Results show that the 35 aquifer is characterized by a high recharge of 600-760 mm/year. Then infiltration process and 36 groundwater flow conditions have been examined by combining hydrogeological and isotopic 37 methods. Rainwater infiltrated is recycled in the vadose zone through plants roots 38 transpiration and groundwater flows with a Darcy velocity of 5 m/day. From the recharge area 39 to the estuary, the mineralization increases controlled by anthropogenic activities and water-40 rocks interactions which are amplified by the residence time and accelerated by the hot and 41 humid climate of Douala. The paper ends with the determination of natural background levels 42 (NBLs) and threshold values (TV) of chemical components in groundwater to assess the 43 contamination for different flow paths. This multi-proxy study and the establishment of NBLs 44 and TV can be beneficial to improve groundwater resources management. Moreover, the 45 conceptual model provided in this study could be used as a reference for porous aquifers 46 submitted to high rainfall amount.

47

48 Keywords: urban hydrogeology; groundwater recharge, stable isotopes; geochemistry,
49 integrated water resources management; threshold values.

#### 1. Introduction

While many works have been published on recharge, functioning and management of aquifers 51 52 in arid or semi-arid areas (e.g., Sami 1992; Allison et al. 1994, Scanlon et al. 2006, Rahmani 53 et al. 2017), there is still very little knowledge about groundwater functioning in humid 54 tropical regions and papers on humid tropics hydrogeology remain rare; mainly focusing on 55 water quality and on the economic and environmental significance of this resource (Foster 56 1993; Foster et al. 2018; Nlend et al. 2018). Fosberg et al. (1961) physically identify these 57 regions by: (i) a monthly temperature equaling or exceeding  $20^{\circ}$ C for at least eight months of 58 the year; (ii) a vapor pressure and a relative humidity average of  $\sim 20$  millibars and 65% 59 respectively, for at least 6 months of the year; (iii) an annual rainfall exceeding 1500 mm and; 60 (*iv*) the rain falling all year with at least six months having precipitation  $\geq$ 75mm/month. They 61 represent approximately 22% of the earth's surface (UNESCO 1990, 2000; State of the 62 tropics 2014) and house a lot of developing countries which present the highest demographic 63 growth rates and whose population is expected to represent 50% of humanity by 2050 (State 64 of the tropics 2014). Wohl et al. (2012) reported then that humid tropics are dynamic 65 environments where freshwater is under pressure from population growth, land use and 66 climate change. In addition to their importance for the water supply, these hydro-systems are 67 a key factor in the preservation of humid tropical ecosystems (animal-plant-soil-water). because of the intimate relationship between surface and groundwater and the frequently 68 69 shallow groundwater table with abundant humid equatorial vegetation in such environments 70 (Foster & Chilton 1993). Regarding all these aspects, there is an urgent need to increase the 71 knowledge in aquifers functioning in humid tropical regions and peculiarly in sub-Saharan 72 African areas which represent 30% of the humid tropics (45% in latin America and 25% in 73 Asia) -according to State of the tropics (2014) -; and is often regarded as the world's fastest 74 urbanizing region (Saghir & Santoro 2018).

75 In this objective, the Douala megacity (Cameroon, Western Central Africa) is a perfect 76 example of a tropical area with approximately 4000 mm/yr, 230 rainy days/yr, a mean air 77 temperature of 27°C (Tchiadeu & Olinga 2012) and a rapid urbanization process due to a 78 demographic boom that has begun in the early 80's (Nlend et al. 2018). Douala is the 79 economic capital of Cameroon and the most dynamic town in the Central Africa region by 80 ensuring, for example, the transit of goods from the rest of the world to landlocked countries 81 like the Central African Republic and Chad). The demographic dynamism of Douala is 82 expressed by a large movement of people from the countryside to the coastal cities due to 83 economic reasons. Indeed, this metropolis, the largest city in Cameroon with a big harbour, is highly developed with varied industries (textiles, agro-food, chemicals, glassworks, cement, 84 85 wood, fertilizers, etc.). In this large city, 49% of the drinking water is produced from sub-86 surface by the Cameroon Water utilities (Camwater) and the Mio-Pliocene (MP) sands aquifer 87 represents the most exploited resource. This latter is tapped both formally by Camwater and 88 informally by many industries and people for domestic uses and drinking water supply 89 purposes (Mafany 1999). However, the anthropogenic fingerprint at Douala has already 90 showed harmful effects on local water resources as it experienced cholera epidemics in 2004 91 (Guevart et al. 2006) and 2012 (Ketchemen-Tandia & Banton 2012), making access to clean 92 water a major priority for populations.

Regarding these important issues, this article follows the previous studies which have mainly dwell long on groundwater quality (Takem et al. 2010; Kopa et al. 2012; Ngo Boum et al. 2015; Fantong et al. 2016; Tatou et al. 2017; Ketchemen-Tandia et al. 2017 and Wirmvem et al. 2017) but the hydrogeological functioning of the MP aquifer is still very little known. Therefore, this research is a first attempt to conceptualize the functioning (recharge, infiltration, groundwater flow organization and acquisition of mineralization) of the main groundwater resource of the city which is the MP aquifer in a context of high rainfall regime 100 by using an innovating combination of hydrogeological, isotopic and chemical methods 101 coming both from historical data collection and new investigations over the study area. Beside 102 the hydrogeological and water quality analysis, the paper aims at determining the natural 103 background levels (NBLs) and proposes threshold values (TVs) of groundwater chemical 104 components of the region in order to propose management tools able to provide guidance for 105 the future exploitation of groundwater resources. Such type of investigation can be used as a 106 reference for urban porous aquifers in humid tropical areas and in the establishment of 107 improved integrated water management strategies in Africa.

108

#### 109 2. Framework of the study region

#### 110 **2.1. Location and topography**

111 The Douala region  $(03^{\circ}45'N - 04^{\circ}20'N, 09^{\circ}34'E - 10^{\circ}00'E)$  is a coastal site (~ 33 km to the 112 ocean) located approximately 70 km southwest to the Cameroon Mount (~ 4100 masl), 113 covering an area of more than 300 km<sup>2</sup> on the Gulf of Guinea, in the Republic of Cameroon 114 (Fig. 1).

116

115

## Figure 1

- 117 The relief is little contrasted with altitudes varying between ~3 masl on the estuary zone to 118 about 70 masl in the eastern side of the region (Figure 2).
- 119

#### Figure 2

#### 120 2.2. Climate

121 The climate of Douala is characterized by strong marine influences (Olivry 1986). The annual 122 rainfall regime is unimodal with a long rainy season from March to November and a very short "dry season" in December–February (Fig. 3). Considering the 65 years (1951-2016) of
rainfall recorded at Douala meteorological station (04°01'N, 09°42'E), the annual average is
3854.3 mm, the maximum and minimum monthly values are respectively measured in August
(742.4 mm) and December (34 mm). These very rainy conditions throughout the year make
Douala to be considered as a singular hyper-rainy city.

128

### Figure 3

The peculiar climate of Douala is also linked to the relative uniformity of air temperature and very high atmospheric humidity conditions. The mean annual temperature is 27°C; lowest and highest temperatures are observed respectively in August (25.4 °C) and February (28.6 °C), giving low annual thermal amplitude of 3.2 °C (Fig. 3). The relative humidity (records of 1971-2009) ranges between 79% and 90% during the rainy season (March to November) and between 77% and 81% during the dry season (December to February), with an annual mean of 85%.

#### 136 **2.3. Hydrography**

137 The wet conditions described above are reflected at the surface by a very dense hydrographic 138 network (Fig. 2) confirming the observations of Braune & Xu (2009) in this part of Africa. 139 Apart from the swamps that cover lowland areas, two main watercourses drain the study area: 140 the Wouri and the Dibamba Rivers which present many tributaries (Fig. 1 & 2). Both the 141 Wouri and the Dibamba dendritic drainage basins discharge into the Atlantic Ocean where 142 tides entering the estuaries make them vulnerable to seawater intrusion (Onguene et al. 2015). 143 The hydrological regime of these rivers is similar to that of rainfall. Low flow are registered 144 from November to December (discharge  $<200 \text{ m}^3/\text{s}$ ) and then increase slowly up to June and 145 abruptly up to September when it reaches a maximum value of 900 m<sup>3</sup>/s (Olivry 1986). At

this moment, the soil often becomes saturated and the streams overflow their channels causingfloods favoured by the low altitude of the area. Then, from October discharge is decreasing.

#### 148 **2.4. Soils and vegetation**

Douala has two types of soils: ferralitic and hydromorphic (Segalen 1967; Zogning 1987; Ndome 2010). In both cases, the authors identified a large proportion of fine and coarse sands (45 to 80%), and clays (10 to 50%). The proportions of kaolinite (50-60%) are the most important while goethite (35-42%) is significant and gibbsite (2-10%) is negligible (Ngeutchoua 1996). Ferralitic and hydromorphic soils occupy respectively 80% and 20% of the study region. The latter is mainly observed in alluvial plains and valley bottoms.

155 The vegetation is that of low altitude rainforest with large mangroves along the coast, 156 evergreen forest and some plantations (banana, corn, sugar cane, fruits, etc). Riparian 157 vegetation is observed near the valleys and depression areas (Din Ndongo 2001).

158

#### 159 **2.5.** Geology and hydrogeology: emphasis on Mio-Pliocene formations

160 The Douala region is constructed on a 5 km-thick sequence of unconsolidated to semi-161 consolidated sedimentary rocks going from Pleistocene to Aptian (Fig. 4; Regnoult 1986). 162 These sediments are associated to tectono-sedimentary dynamics which prevailed during the 163 opening of the South Atlantic and the formation of West African passive margin (Moulin 164 2003; Helm 2009). Continental in the eastern part of the basin, sediments progressively move 165 westward to marginal sea, continental shelf and deep seabed environments. They can be 166 divided into seven formations from the bottom to the top (Fig. 4): i) Mundeck's 167 conglomerates and sandstones (Apto-Albian), *ii*) Logbadjeck's conglomerates, sandstones and 168 limestones (Turonian-Santonian), iii) Logbaba's sands, sandstones and clays (Campanian-169 Maastrichtian), iv) Nkapa's limestone, sandstone, marl and shales (Paleocene-Eocene), v) Souellaba's clays and sands (Oligocene-Lower Miocene), *vi*) Matanda's clays, sands and
gravels (Upper Miocene-Pliocene) and *vii*) Wouri layers (Quaternary) consisting of clays,
sands and gravels (Dumort 1968; Njike Ngaha 1984; Regnoult 1986; Nguene et al. 1992;
SNH/UD 2005; Manga 2008; Njoh et al. 2014; Njoh & Petters 2014; Njike Ngaha et al.
2014).

175

#### Figure 4

176

The Mio-Pliocene aquifer is represented by Souellaba and Matanda formations, overlying the
Eo-Oligocene and covered in some places by the Quaternary sediments 20 to 70 m thick
(Feumba 2015). It consists in a multi-layered aquifer separated from more superficial and
deeper aquifers by discontinuous clay layers. Its thickness globally increases NE to SW (Fig.
4). Although many lithological lateral variations, the global structure of the MP aquifer from
the top to the bottom can be described as follows (Table 1) by using well logs and previous
works on hydrogeology (Bourgeois 1977 and Ketchemen-Tandia 2011):

- 184 (i) the Pliocene is clayey, clayey-sandy and sandy-clayey. Its depth generally does not
  185 exceed 40 m;
- 186 (ii) the Upper Miocene consisting of coarse grained sands with gravel intercalations.
- 187 Sandy sequences, slightly silty and calcareous, appear well developed at the NW

188 of Douala where their thickness exceeds 200 m (Bourgeois 1977);

- 189 (iii) the Lower Miocene formation made up of cross-bedded sands and marked by190 some passages of lignite and limestone.
- 191
- 192
- 193
- 194

Table 1

196

Separated by a clay layer, the difference between Mio-Pliocene and Quaternary formations
lies in the lithology; Quaternary sands are finer than those of the Mio-Pliocene.
The lateral variations observed in lithology of the MP aquifer are also reflected in
hydrodynamic settings (Table 2). High values of standard deviation (Table 2) in
transmissivity and hydraulic conductivity suggest an important anisotropy of the MP aquifer.
Table 2

#### **3. Methods of investigation**

#### 205 **3.1. Inventory of sampling points**

206 Wells logs and results from hydraulic testing, when available, have been compiled from 207 previous projects in the study area. Indeed, priority has been given to wells having this basic 208 information and which are still used currently by local populations. In addition to this 209 inventory, a field campaign was conducted to target springs in the study area. Finally, a total 210 of 58 sampling points, including 43 wells (26 boreholes and 17 hand-dug wells) and 15 211 springs on the MP formation have been selected according to their distribution over the 212 Douala region. In general, hand-dug wells tap water at shallow levels (<18 m) while 213 boreholes will correspond to water from more important depth (maximum = 90 m). Boreholes 214 are often equipped with foot or hand pumps or directly connected to a faucet. The only 215 disadvantage of this engineering structure in Douala is that they are most of the time screened 216 at several depths. For some wells, this can complicate the understanding of groundwater flow 217 and hydrochemical processes that take place within the different levels. Springs encountered 218 in the study area are located at the bottom of steep slopes, bordering important marshy areas.

220

#### 221 **3.2.** Field measurements, sampling and analytical procedures

222 Groundwater sampling and in situ measurements (water levels and physicochemical parameters) were achieved in February 2017. Springs have been sampled directly at their 223 224 outflows. Wells and boreholes have been sampled by pumping, using the in-situ equipment. 225 Assuming the fact that wells in Douala are daily used for domestic supply, it was considered 226 during the campaign that there is no stagnant water within wells and that their purge is already 227 achieved. The depth of groundwater in the well relative to the soil surface *i.e* water table (h) 228 was measured by using a potentiometric probe. Knowing the altitude (Z) of the sampling 229 point, the water table was converted in potentiometric level (H) through the formula H = Z - h230 and by considering the height of the well edge. Physicochemical parameters (pH, electrical 231 conductivity, water temperature and dissolved oxygen) were measured using a WTW 232 2FD46G multi-parameter. Measurements have been validated only after all of the in situ 233 parameters reached the stability criteria (Rey et al. 2017) for three consecutive measurements 234 separated by 5-minutes intervals. Alkalinity measurements were performed using a digital titrator (HACH). Samples for major ions analysis (Cl<sup>-</sup>, NO<sub>3</sub><sup>-</sup>, SO<sub>4</sub><sup>2-</sup>, HCO<sub>3</sub><sup>-</sup>, Na<sup>+</sup>, K<sup>+</sup>, Mg<sup>2+</sup> and 235 Ca<sup>2+</sup>) were collected in two 30 mL polyethylene bottles after filtration through 0.45 µm 236 237 nitrocellulose membranes. Samples for cations analysis were acidified with ultrapure HNO<sub>3</sub><sup>-</sup> 238 at 70%. Bottles were filled to the top with no headspace and kept at 4°C until analyses. 239 Chemical analyses were conducted at the Hydrogeology Department of the University of 240 Corsica, France by ionic chromatography using a Dionex ICS 1100 chromatograph. The 241 quality of the analysis was checked by calculating the ionic balance error: analyses were 242 rejected if the ionic balance error was greater than 5%.

Samples for stable isotopes of the water molecule were stored in 20 mL amber glasses bottles.All care was taken to avoid evaporation by taking water directly in the well. As for chemistry,

bottles were filled to the top with no headspace and samples were kept cool until being transferred to the laboratory where measurements were performed at the Hydrogeology Department of the University of Corsica, France using a liquid–water stable isotope analyzer DLT-100 (Los Gatos Research) according to the analytical scheme recommended by the IAEA (Penna et al. 2010). Values are reported in per mil units (‰) compared to Vienna Standard Mean Ocean Water standard (VSMOW). The quality of the isotopic analysis was checked using a standard deviation up to 1‰ for  $\delta^2$ H and up to 0.1‰ for  $\delta^{18}$ O.

252

#### **4. Results and discussion**

#### 254 **4.1. Estimation of Mio-Pliocene aquifer recharge**

In this paper, two methods were applied: a water balance and a physical method on saturated zone (Water Table Fluctuations, WTF). These approaches were chosen given the availability of required data. The choice of these methods was driven by the goal to obtain an average rate of recharge for the aquifer. Moreover, WTF and water balance methods are known to be well applied both in humid and semi-arid contexts requiring low data and at relative low costs (Scanlon et al. 2002). The first one relying on groundwater data and the second one on climate data.

262

#### 263 Water balance calculation

Recharge is usually calculated as the remainder when losses identified in the form of runoff and evaporation, have been deducted from precipitation (see equations 1 and 2). The term "evaporation" here envelops transpiration.

$$267 \quad P = R + Ea + \Delta SWS + Ieff \tag{1}$$

268 Ieff = P - (R + Ea + 
$$\Delta$$
SWS) (2)

269 P is the monthly rainfall, Ea the real evapotranspiration, R the surface and sub-surface 270 drainage, and  $\Delta$ SWS the storage variation in the unsaturated zone. The rainfall used in the 271 balance is given by the measurements at the Douala meteorological station. Real 272 evapotranspiration (Ea), described in detail by Allen et al (1991), was estimated from the 273 corrected potential evapotranspiration (PETcor) and average Kc values per crop (Allen et al. 274 1998). PETcor was estimated by correction of the potential evapotranspiration calculated by 275 Thornthwaite's equation, as proposed by Medeiros et al. (2012). All these annual average 276 parameters can be considered as homogeneous spatially regarding the topography of the study 277 area.

Therefore, taking into account the potential evapotranspiration, actual evapotranspiration and rainfall amount recorded month by month the yearlong at Douala; the Figure 5 shows that the MP aquifer is potentially recharged from June to November when P > PET and when the soil water deficit is filled. From December to February, P<PET and PET > Ea, then the recharge of soil reserve is null and the groundwater table is down. Indeed, the potential recharge follows the rainfall regime.

284

#### Figure 5

This potential recharge of 2556.5 mm corresponds to effective infiltration plus runoff. However, for the study area, no experimentally measured surface drainage data were available for the analyzed period. The data used in this paper are from Emvoutou (2018) who used remote sensing and GIS approach by considering: land use/land cover, drainage density, topography and slopes for the 2000-2004 period. The average value obtained is estimated at 46.4% of annual rainfall amount. Thus, the infiltration can be estimated at 767 mm/year i.e approximately 20 % of rainfall amount.

#### 293 WTF approach

Based on the premise that rises in groundwater levels in shallow aquifers are due to recharge
water arriving at the water table and going immediately into storage (Healy & Cook 2002),
the water table fluctuation method can be applied here for estimating groundwater recharge.
The recharge rate can be obtained by applying the following equation:

298 Ieff = Sy 
$$\frac{\Delta h}{\Delta t}$$
 (3)

Eq. (3) assumes that water that reaches the saturated zone is immediately stored and the other components of the subsoil water balance (underground evapotranspiration, base flow, inflow and outflow) are zero during the period of recharge.

302 "Sy" is the specific yield, approximately equal to the storage coefficient. It is the volume of
303 water released from a unit volume of saturated aquifer material (Maréchal et al. 2006). "h" is
304 the water table height and "t" is time.

In this paper, empirical value of Sy provided by Johnson (1967) has been used, while values of "h" come from the monthly monitoring carried out by Feumba et al (2011) in Douala. They reported that the lowest depth to water (2 m) is observed in August while the highest one is in December (9.8 m). Therefore, the fluctuation of water table is ranging between 2.00 and 9.80 m thus giving  $\Delta h = 7.8$  m.

By considering the nature of soils (and the upper levels of MP) mainly made of sandy clay materials, and if it supposed that the studied aquifer material is homogenous vertically and laterally, the specific yield of 7% defined by Johnson (1967) can be used here to determine the aquifer recharge according to the equation 3.

By applying the formula presented in equation 3, recharge is then calculated from January 2010 to December 2010 thanks to the monitoring carried out by Feumba et al (2011). An average value of 546 mm has been obtained. Since the annual rainfall amount recorded in 2010 is 3585.2 mm, the recharge coefficient ( $I_{eff}/P$ ) is approximately equal to 15%. 318 The small gap between the two calculated values from water balance and WTF can be 319 explained by:

(*i*) the evaluation of water level at the period of one year which is not sufficient. (*ii*) the
application of WTF method at a monthly scale can underestimate the recharge. For instance,
if there is a small amount of recharge that is less than drainage from the water table, then the
water table will continue to drop but at a slower rate.

324 In summary, we observe that the aquifer recharge period corresponds to six successive 325 months. By taking into account the replenishment of soils reserves, we can conclude that the 326 aquifer is recharged approximately all the yearlong. This indicates that the studied aquifer 327 presents excellent quantitative potential, especially for water supply. The recharge rate of 328 15% - 20% (578.2 – 732.3 mm/year) of annual rainfall amount is lower than that highlighted 329 by model analysis (Mohan et al 2018) -where the annual recharge rate is estimated at more 330 than 1 m-, but similar to that obtained in other similar humid tropical areas such in Ghana and 331 Benin where there is also ferralitic soils (Obuobie et al. 2012; Kotchoni et al. 2019). In the 332 next section we will identify the recharge areas and discuss about the infiltration process and 333 groundwater flow within the aquifer.

#### 334 4.2. Location of recharge areas, infiltration process and analysis of groundwater flow

#### 335 Location of recharge areas

The potentiometric map of the MP aquifer is presented on Fig. 6. We identify on this map, two piezometric domes in Ndogpassi – Nyalla (SE) and Kotto (NE) corresponding to most elevated points (66 and 50 m.a.s.l respectively) which can be considered as focused recharge areas of the MP aquifer at Douala. Indeed, the piezometric structure of the aquifer is in good agreement with the topography (Fig. 2) of the region as described in section 3.1. This feature is a characteristic of unconfined aquifer.

#### 344 Infiltration process

345 Water stable isotopes (Fig. 7) show that rainwater infiltrates the soil without undergoing 346 evaporation phenomenon. Following Barnes & Allison (1988) and Lu et al (2017) 347 observations in wet soils, the zone in which significant evaporation occurs is narrow or 348 inexistent. This is in agreement with the hyper-humidity conditions characterizing the study region. However, MP groundwater are ranging from -5.1% to -3.1%  $\delta^{18}$ O and -25.7% to -349 9%  $\delta^2$ H with mean values of -3.5  $\delta^{18}$ O and -13  $\delta^2$ H while the weighted mean value of rainfall 350 corresponds to -2.8%  $\delta^{18}$ O and -12.2%  $\delta^{2}$ H. This is particularity well displayed on Fig. 6 351 352 where groundwaters plot above the Douala and Global MWL (Meteoric Water Line). Such 353 result suggests a complexity in infiltration mechanisms.

354

355

#### Figure 7

356

This difference between isotopic composition of rainwater and groundwater at Douala, can be
further examined through the analyses of the deuterium excess ("d-excess"; Eq. (4);
Dansgaard 1964).

$$360 \quad \mathbf{d} = \delta^2 \mathbf{H} - 8\delta^{18} \mathbf{O} \tag{4}$$

The d-excess weighted mean in Douala precipitation (2006–2018) is 10.4 % $_{o}$ , close to dexcess of the GMWL (d-excess = 10% $_{o}$ ; Fig. 7 & 8) thus reflecting an oceanic origin of the water vapor (Craig 1961). In aquifer, d-excess<sub>mean</sub> = 14.8% $_{o}$ . Approximatively, the same value is observed both in shallower and deep water. D-excess value of MP groundwater is significantly higher than the one recorded in the recharge water (Fig. 8). However, it is similar to that recorded in rainfall at Bangui (13.5% $_{o}$ ) and Kisangani (15% $_{o}$ ; IAEA/WMO 2019) in the Congo basin (Fig. 8) where there is an influence of continental recycling (Ketchemen et al. 368 2011; Risi et al. 2013; Nlend et al. 2020). Could this process occur within the vadose zone 369 through the transpiration of plants roots? According to Blavoux (1978) and Barnes & Allison 370 (1988), transpiration both removes some enriched water and causes a water flux at depth 371 which is greater than the evaporative flux at the soil surface. Indeed plants transpiration 372 favors root water transport from deeper to shallower soil layers, leading to isotopic depletion 373 (Lu et al. 2017). Therefore, the d-excess-shift observed between recharge water and 374 groundwater in Douala can be attributed to plants transpiration within the unsaturated zone. 375 Soil water use by plants can be dominant when the soil water reserves are null (December-376 February) and even during the replenishment from March to May (Fig. 5). Barbeta and Peñuelas (2017) reported that this plant groundwater uptake represents on average 49 % of 377 378 transpired water in dry seasons and 28 % in wet seasons worldwide. This hypothesis is 379 supported by the common occurrence of deep roots which can exceed respectively 3.5 m and 380 1 m for woody and herbaceous species of the study region. In addition, according to Barbeta 381 and Peñuelas (2017), as Douala is a flat area with landscape mostly represented by plain and 382 riparian zones, the contribution of groundwater to transpiration must be particularly 383 significant. The absence of significant difference between isotopic compositions of shallower 384 and deep water (respectively hand dug-wells and boreholes, Fig. 7) may illustrate the 385 influence of transpiration on shallow groundwater stable isotopes. The changes from 386 rainwater to aquifer are likely a result of the lateral roots extracting water from the subsoil and 387 redistribution to the superficial layer, The same observation was done by Lu et al (2017) 388 thanks to experimentation employing customized transparent chambers coupled with a laser-389 based isotope analyzer measuring continuously near surface variations in the stable isotopic 390 composition of evaporation (E,  $\delta_E$ ), transpiration (T,  $\delta_T$ ) and ET ( $\delta_{ET}$ ) to partition the total 391 water flux.

392 Fig.7 shows also the existence of paleo-groundwater within the MP multilayer aquifer system. Indeed the sample of Yabea exhibits a strongly depleted signature (-5.1%  $\delta^{18}$ O) compared to 393 394 the other points. It is a borehole of 95 m deep with a water table at 54 m. According to 395 Ketchemen-Tandia (2011) this groundwater resource from the most confined part of the 396 aquifer, exhibiting depleted isotope values, has zero tritium content; while tritium contents 397 recorded in springs and hand-dug wells vary from 2.1 to 5.9 TU, in good agreement with the 398 regional rainfall input (GNIP data at the Ndjamena station; IAEA/WMO 2018). This indicates 399 that the groundwater of Yabea is, for sure, originating from a recharge having occurred prior 400 to the atmospheric bomb tests (i.e. before 1953; Fontes 1983) and therefore is more than 70 401 years old and probably corresponding to a palaeorecharge episodes dating a few millennials 402 ago considering the amplitude of the isotopic depletion (Jirakova et al., 2011) as already 403 described in the Western Africa (Huneau et al., 2011). Therefore the studied aquifer gathers 404 water from recent infiltration and paleo-recharge.

405

406

#### Figure 8

407

408 Analysis of groundwater flow

Water reaching the saturated zone globally moves from the eastern side of the study regiontoward the discharge area in the Wouri estuary zone (Fig. 6).

However, the variable groundwater flow directions and the variability of the distance between the potentiometric surface contours (Fig. 6) characterize a non-uniform flow regime. Potentiometric surface contours become more and more distant from the recharge area, indicating a dominance of permeable formations towards the estuary in agreement with geological/hydrogeological description. Moreover, the general shape of the potentiometric surface is convex, showing a radial aquifer with divergent streamlines contrary to semi-arid 417 and arid regions where convergent streamlines towards the water course axis are generally418 observed (e.g., Ouali et al. 1999; Lienou et al. 1999).

419 Hydraulic gradients (i) from upstream to downstream, calculated along the transect line C-D 420 (Fig. 6) from hand-dug wells; vary between 3.5% and 2.1% with a mean of 2.3% 421 characteristic of unconfined aquifer conditions. Thanks to this mean value of *i*, and assuming a porosity (n) of 0.07 for MP sands and an average hydraulic conductivity (K) of  $1.5 \times 10^{-3}$ , a 422 423 calculation of groundwater velocity (V = k \* i/n) has been performed. V was estimated to 424 4.9x10<sup>-5</sup> m/s or 5 m/day. Therefore, it would take 70 years for water to travel from the 425 upstream zone (at PK 13) to the estuary area (Fig. 6). This Darcy velocity is very important 426 compared to that generally observed in porous media (Piccinini et al. 2016). It may be 427 attributed firstly to the high hydraulic conductivity of the aquifer and secondly to the high 428 rainfall recorded at Douala which leads to shallow groundwater table and therefore contribute 429 somehow to high values of hydraulic gradient as observed.

430 In the discharge area, the MP aquifer can develop interactions with the Wouri and Dibamba 431 Rivers (Fig. 6). It can be suspected that when there are superimposed, a natural upward water 432 flux from the MP to the Quaternary aquifers should occur. Moreover, in this discharge zone, 433 groundwater can also be influenced by ocean tides. However, regarding the geographical 434 position of the aquifers (Fig. 6), Quaternary sands seem to be more vulnerable to saline 435 intrusions than the Mio-Pliocene formation. Finally, it is worthwhile noting that groundwater 436 discharge also occurs at numerous springs in the study area due to very local topographic 437 control.

In summary, the keys messages in this section are: (i) rainwater reaches the saturated zone without undergoing evaporation; (ii) the hyper-rainy conditions contribute directly or indirectly to water recycling within the vadose zone and to maintain the groundwater at a shallow level even during the dry season.

In the next section we will investigate which factor controls the water chemistry along theflow and especially if there is any influence of the regional hyper-recharge rate.

#### 444 **4.3. Water chemistry**

#### 445 *Main features*

446 As shown in Table 3, groundwater temperatures (median values of 28.6°C, 28.2°C and 447 27.9°C respectively for boreholes, hand-dug wells and springs) are close to that of air 448 temperatures in February (26.9°C to 30.5°C with a mean of 28.6°C) when the field 449 measurements were carried out. It is clear that temperatures increase slightly with the depth as 450 water from boreholes present the highest values in term of maximum, mean and median 451 (Table 3). This can probably be associated to the geothermal gradient. The pH values show a 452 wide range from strong acidic conditions to slightly alkaline (Table 3) thus reflecting various 453 inputs into the aquifer system. However, when considering mean and/or median values (Table 454 3), the acidic character of MP water is obvious. The electrical conductivity (EC) in boreholes, 455 hand-dug wells and springs range from 13.6 to 1632 µS/cm, 47 to 1048 µS/cm, and 111.8 to 456 375 µS/cm respectively (Table 3). As for pH values, this tends to reflect various influences on 457 MP groundwater system. Nevertheless, the median EC in boreholes is significantly lower than those in hand-dug wells and springs. Therefore, the deeper the water flows, the more the 458 459 groundwater mineralization decreases. It is already possible to hypothesize that the shallower 460 aquifer system is the most affected by various inputs entering the system and peculiarly by 461 anthropogenic factors such as: industrial effluents, water from pit latrines and leaking of solid 462 waste.

- 463
- 464

465 Dissolved oxygen (DO) concentrations show wide range of values from 2.3 mg/l to 7.8 mg/l, 466 2.2 mg/l to 7 mg/l and 2 mg/l to 7.8 mg/l for boreholes, hand-dug wells and springs 467 respectively (Table 2). This heterogeneity in DO values can be related to microbial activities 468 within the different flow paths of the Mio-Pliocene (Keesari et al. 2015). On the other hand, 469 the fact that most boreholes are screened at different depths could affect the redox conditions 470 within the system.

471 Concerning the major ions chemistry, NO<sub>3</sub><sup>-</sup> is the dominant anion in boreholes and springs 472 accounting respectively for 74% and 64% of the total major ions; after which the concentrations of anions in groundwater types follow the order Cl<sup>-></sup>  $SO_4^{2->} HCO_3^{-}$ . In water 473 from hand-dug wells there is no dominant anion; Cl<sup>-</sup>, SO<sub>4</sub><sup>2-</sup> and NO<sub>3</sub><sup>-</sup> present approximately 474 475 the same percentage (around 33%). In contrast, the concentrations in major cations are dominated for each sample by Na<sup>+</sup> followed by Ca<sup>2+</sup>, K<sup>+</sup> and Mg<sup>2+</sup>. This order of abundance 476 477 of ions is in agreement with the hydrochemical types observed on the Piper diagram (Fig. 9) 478 with a clear abundance of Na-K-Cl water types for the different flow paths. However, it is 479 worthwhile noting that the borehole sample of Yabea (the most depleted in stable isotopes; 480 Fig. 7) is the only one presenting a Ca-Mg-HCO<sub>3</sub> water type. This water-type reflects a lack 481 of anthropogenic influence on this water and a predominance of geological natural processes on groundwater chemistry. 482

- 483
- 484

#### Figure 9

485

486 Evolution of major ions chemistry along a flow-path

487 The evolution of concentrations in major ions along the E-F transect (Fig. 6) is displayed on488 Figure 10. Only boreholes samples are considered on this flow-path since they appear to be

489 less influenced by human activities. It concerns the settlements of Nkolbong, Parizo, 490 Ndogpassi, KM5, Mboppi and Essengue located respectively at 16, 14, 12, 6, 4 and 2 km 491 from the sea. EC increases along the flow-path, from 32 to 1632  $\mu$ S/cm. The lowest 492 concentrations characterize the recharge area (Nkolbong and Prizo, Fig. 6) while the highest is 493 observed close to the discharge area (Mboppi and Essengue).

- 494
- 495

Figure 10

496

Concentrations in major ions globally increase along the flow. Cl<sup>-</sup>, HCO<sub>3</sub><sup>-</sup>, Mg<sup>2+</sup>, Na<sup>+</sup> and K<sup>+</sup> 497 present the same behavior along the flow-path while Ca<sup>2+</sup> and SO<sub>4</sub><sup>2-</sup> also evolve in parallel, 498 499 decreasing abruptly close to the estuary involving specific control on groundwater chemistry 500 in the discharge area. The case of  $NO_3^{-1}$  is more peculiar. Concentrations in this element vary significantly along the flow path. Lowest values are found both in the recharge zone and near 501 502 the estuary. Moreover,  $NO_3^-$  in boreholes, hand-dug-wells and springs do not exhibit any 503 relationship with DO reflecting strong spatial variations of redox conditions. As many sub-504 Saharan African metropolis, the high level concentrations in nitrates can be linked to local pollution effects showing probably an input of latrine effluents (Lapworth et al. 2017). 505

506 Following these two previous sections describing the main features and the evolution of 507 groundwater chemistry along the flow, the next section will aim at determining the 508 hydrochemical processes involved.

509

#### 510 Processes controlling water chemistry

511 Generally, the soluble ions concentrations in groundwater are partly deriving from the 512 precipitation input, evaporation, dilution, water rocks interactions, residence time and 513 anthropogenic factors. All the natural processes can be investigated through Gibbs diagram

514	which is helpful to discriminate geochemical evolutions in groundwater. A Gibbs diagram has
515	been drawn for boreholes groundwater on Fig.11. It shows that groundwater has been affected
516	largely by water-rock interaction (Figure 11) even if three groups can be separated: $(i)$ the
517	group of precipitation dominance characterized by low mineralization (13.6 < EC < 19 and
518	3.2 < TDS < 11.5) suggesting a recent infiltration of rainwater in the aquifer system, ( <i>ii</i> ) the
519	group of water-rock reactions with samples characterized by a freshening process due to
520	important recharge conditions and (iii) the group of water-rocks reactions with rock
521	dominance represented only by the sample of Yabea. This sample presents a high value of
522	TDS (EC = 431 $\mu$ S/cm) with low concentration in Na <sup>+</sup> . The Gibbs diagram confirms that
523	water from this borehole may be an old groundwater in the deep part of the aquifer system.
524	Indeed, a significant time of water-rocks interactions would explain the mineralization as well
525	as lower content in Na <sup>+</sup> may refer to a groundwater reservoir protected from anthropogenic
526	influences.
527	
528	Figure 11
529	
530	In summary, it is clear that the water residence time is a control factor of groundwater
531	chemistry. In this study, the long residence time of groundwater leads to a $Ca - HCO_3$ water
532	type with high TDS (see Fig. 9) and no evidence of human pollution.
533	
534	However, details must be provided concerning the water-rock interaction process highlighted
535	on the Gibbs diagram (Fig. 11). It is a relatively broad classification that may include many
536	kinds of processes such as silicate weathering, carbonate dissolution, and evaporate
537	dissolution. In the present case, the fact that 69% and 88% of samples have respectively a

538 molar ratio higher than 2 and 1 - respectively for  $Ca^{2+}/Mg^{2+}$  and  $Ca^{2+}/Na^{+}$  ratios - indicates a

predominance of silicate weathering process in groundwater chemistry (Lee & Strickland 539 540 1988; Lakshmanan et al. 2003; Rajmohan & Elango 2004). The lithology of the strata in the 541 study area, consisting of sands and clays and the mineralogy of the aquifer, supports the 542 above argument. This silicate weathering through hydrolysis reactions is favoured at Douala 543 by the relative high temperatures and low thermal amplitude according to Segalen (1965) who 544 showed that hydrolysis reactions are more intense and faster in humid tropical regions. 545 Thanks to the constantly high temperature, the presence of free H<sup>+</sup> ions in the superficial 546 water allows a chemical decomposition of the rock minerals.

547 It is worthwhile noting that the sample of Yabea identified on the Gibbs diagram in the field 548 of rock dominance is affected by carbonate dissolution. This assertion is supported by the 549 saturation indexes (SI) corresponding to -0.99, -1.1 and -1.5 respectively for calcite, aragonite 550 and dolomite while for the rest of boreholes SI<sub>calcite</sub>, SI<sub>aragonite</sub> and SI<sub>dolomite</sub> vary respectively 551 from -2.3 to -4.9, -2.5 to -5.1 and -3.5 to -9.7. Indeed, since previous works on geology 552 mentioned the presence of carbonate layers in the lower Pliocene, this reinforces the 553 hypothesis that this sample corresponds to a borehole tapping a very deep part of the MP 554 aquifer and therefore relatively old water. The dissolution phenomenon may also be enhanced 555 by the hyper-rainy character of Douala which enables significant drainage in promoting the 556 solubilisation of elements (Helgeson et al. 1984). Thus, the peculiar climate of Douala is a 557 control factor of water-rocks interactions accelerating the reactions of hydrolysis and 558 dissolution.

559 Besides, the water-rocks interaction reinforced by the water residence time and the hot and 560 humid climate of Douala, another considerable control factor of groundwater chemistry is the 561 anthropization phenomenon.

562 To check the influence of this phenomenon on the Mio-Pliocene aquifer of Douala, we 563 compiled historical data on nitrates concentrations measured in 2011 (Fantong et al. 2016),

2014 (Emvoutou 2018), 2015 (Wirmvem et al. 2017) and 2017 (this study) during the dry 564 565 season at Douala. Nitrates generally well reflect the impact of human activities on the water 566 resource. Thus, regarding the urbanization trend in Douala, the high increase in population 567 and inhabitants density (Nlend et al. 2018), we expect that NO<sub>3</sub><sup>-</sup> content in groundwater 568 would increase overtime. For the 2011-2017 period, maximum values show a linear variation, 569 a positive trend: from 24.0 mg/l to 245.6 mg/l. Median values evolve from 6.0 to 29.3 mg/l 570 (Fig. 12). These latter values are significantly above the global natural levels of  $NO_3^-$  in 571 groundwater (Appelo & Postma 1993). The progressive increase of nitrates in groundwater is 572 a worrying feature revealing the strong anthropogenic constrains on groundwater resource and 573 therefore the water quality decline.

- 574
- 575

#### Figure 12

576

577 Further, the influence of cation exchange was investigated. Like Emvoutou et al (2018), we 578 found a lack of cation exchange which is evidence that the MP aquifer is not impacted by 579 seawater intrusion. Indeed, according to Martínez & Bocanegra (2002, when sea water 580 intrudes a fresh water aquifer, cation exchange takes place and sodium is taken-up by the exchanger, while Ca<sup>2+</sup> is released into groundwater. Therefore, the increase in water 581 582 mineralization, chloride and sodium near the estuary (Fig. 9) are not linked to saline intrusion. 583 The Cl<sup>-</sup>/HCO<sub>3</sub><sup>-</sup> ratio confirms this fact; with 1.8 molar ratio (Cl<sup>-</sup> = 94.4 mg/l) for the sample of 584 Essengue while Wirmvem et al (2017) reported that surface water at the estuary has a Cl<sup>-</sup> 585 /HCO<sub>3</sub><sup>-</sup> molar ratio of 100 with approximately 5000 mg/l of Cl<sup>-</sup>.

In summary, the water-rocks interactions processes can then be considered as amplified by the residence time and the typical climate conditions of Douala (hot and humid). In addition, the distribution of  $NO_3^-$  in boreholes, hand dug-wells and springs (Table 3) and the historical evolution of this ion in groundwater suggest a strong impact of human activities on the aquifer system. Further, as it has been highlighted that seawater does not intrude freshwater, high levels of  $Cl^{-}$ ,  $SO_4^{2-}$  and  $Na^+$  would also correspond to consequences of human activities. In the next section, we will evaluate the anthropogenic part in water chemistry by determining the natural background and threshold values for major elements.

594

#### 595 **4.4.** Determination of natural background levels (NBL) and threshold values (TV)

596 Background level of a substance in groundwater is defined as the concentration of a given 597 element, species or chemical substance present in solution, which is derived from natural 598 geological, biological or atmospheric sources (Edmunds et al. 2003). However, there are 599 various methods to identify NBL: hydrochemical simulation of solution in aquifers (Sellerino 600 2014), pre-selection (Coetsiers et al. 2009), component separation, probability plot and other 601 statistical methods (e.g., Sellerino 2014; Zhang et al. 2017). Among these set of methods, the 602 preselection is the simplest to approach the natural groundwater composition of an aquifer when no national methodology exists for the derivation of NBL. In addition, this method is 603 604 the more suitable for non-expert users and all the stakeholders involved in water management. 605 The pre-selection methodology here consists in the removing samples from unknown depth, 606 samples close to the estuary (where saline intrusion could be suspected or tide influence), 607 groundwater analyses with nitrates concentrations larger than 10 mg/l, so as to remove 608 polluted samples.

Then the NBL will be helpful to determine the threshold values for groundwater. According to Last et al (2014) TV represent a groundwater quality standard that is based on interactions with aquatic and terrestrial ecosystems, interference with legitimate uses or functions of groundwater, and hydro-geological characteristics including background levels. In this study

- 613 it was proposed to define the TV following two different cases based on the ratio between the
- 614 natural background level and a relevant reference value (Muller et al. 2006):
- 615 Case 1: NBL  $\leq$  REF: TV = (NBL+REF)/2
- 616 Case 2: NBL > REF: TV = NBL
- REF is an appropriate reference value, defined by the World Health Organization (WHO) orby any other legislation.
- NBL and TV calculated are shown in Table 4, resulting from the application of the selected
  methodology to the dataset. The values for NBL are presented as minimum and maximum.
  The TV was calculated by using the maximum NBL values.
- 622 Table 4 shows that the concentrations in chemical elements are mostly above the NBL in 623 shallower water from hand-dug wells and springs than in deep water from boreholes. This can 624 be explained by the fact that shallow groundwater is more sensible to anthropogenic 625 influences. However, samples presenting HCO<sub>3</sub><sup>-</sup> contents above the NBL are mostly observed 626 in boreholes. These samples correspond to the deepest groundwater. Since, this element is 627 only related to natural processes it is therefore clear that the deeper the groundwater flows, the 628 more the groundwater mineralization increases due to water-rocks interactions. Groundwater 629 depth should be considered in identifying geochemical processes. Likewise, as expected, the 630 majority of samples which show chemical contents above the TV are from hand-dug wells. 631 However, it is worthwhile noting, as it can be seen from Table 4, that 73% and 46% of 632 samples from boreholes have a  $NO_3^-$  above the defined NBL and TV respectively.
- This observation implies that the aquifer pollution is probably not limited to shallower flow
  path. Thus water from boreholes is also vulnerable to nitrate pollution. This latter appears as a
  serious concern in term of qualitative potential of groundwater.
- Finally, the pristine natural background value of nitrate obtained in this study for MP sands(6.6 mg/l) is very close to that obtained by Zhang et al (2017) in North China plain on gravel

638	sediments. The nitrate TV (28.3 mg/l) is also similar to the one found by Coetsiers et al
639	(2009) on fluvial deposits in Flanders, Belgium. NBL and TV values obtained in this study
640	also correspond to that provided by Appelo & Postma (1993). This tends to validate the
641	accuracy of the results for the Douala region and the methodology used.
642	
643	

# Table 4

#### 645 5. Final discussion: proposition of a conceptual scheme of MP aquifer functioning

The present study brings out the detailed hydrogeological features characterizing an aquifer in hyper-humid region. The approach used involves hydro-climatic data, measures of potentiometric levels, stable isotopes and hydrochemistry. It is the first time that such kind of study is achieved in the Central Africa tropical humid region. The results obtained allows for the design of the hydrogeological conceptual model of the region including the structure and functioning of the aquifer system (Figure 13).

652

653

#### Figure 13

654 It has been shown that the hyper-rainy conditions lead to hyper-recharge of the aquifer 655 approximately all the yearlong. This water then infiltrates the soil without undergoing 656 evaporation. Mio-Pliocene groundwater flows towards the estuary supplying the main river of 657 the region and the superficial Quaternary aquifer. Two flow paths have been identified: the 658 shallower one is homogeneous in term of isotopes indicating a mixing between different flow 659 paths probably due multi-level screened drillings, but heterogeneous in term of major ions 660 chemistry thus reflecting various influences on this upper part of the aquifer. The more the 661 water flows deeply, the more the isotope and chemical content decreases while homogenizing. 662 Regarding the lateral flow, the mineralization generally increases from the recharge zone to 663 the estuary. One key result is that water from the deepest part of the aquifer seems to have a relatively long residence time. The most depleted value in  $\delta^{18}$ O (~ -5%), while the weighted 664 665 mean rainfall is of -2.8  $\delta^{18}$ O-, the low tritium contents (0 TU) and the very low content of 666  $NO_3^-$  (<10 mg/l) recorded on the sample of Yabea suggest the presence of old water (over 70 667 years old) in the aquifer system. Thus, Mio-Pliocene aquifer gathers water recharged recently 668 and older ones. Thanks to the estimation of NBL and TV it can be estimated that deep flow

669 path is reached at approximately 50 meters to the soil. However, as some boreholes have 670 chemical contents above the NBL or TV, it is obvious that deeper levels of the aquifer already 671 show a slight influence of anthropogenic activities on the chemistry. While NBL and TV 672 derivation is an important task for European Union (EU) Member States, it is the first time 673 that such investigation was conducted in Sub-Saharan Africa. In the implementation of 674 Integrated Water Resources Management (IWRM), well defined NBL and TV helped the EU 675 Member states to correctly assess the chemical status of ground-water bodies and to 676 implement the Groundwater Directive (Wendland et al. 2008; Preziosi et al. 2010). Indeed 677 NBL and thresholds limits constitute a basis for clean-up remediation and adequate water legislation. 678

679 From the conceptual scheme obtained, in term of groundwater potential, the important 680 recharge of the Mio-Pliocene aquifer (linked to the hyper-humid climate) reflects the good 681 quantitative potential of the resource. This hyper-recharge also affects the groundwater 682 quality by maintaining the groundwater at a shallow level even during the dry season. Thus, 683 groundwater resource would be easily affected by contaminants at the surface. High-intensity 684 rainfall events pose a risk to shallow and poorly protected groundwater sources. This is 685 particularly an issue for unimproved dug wells and springs. Shallow groundwater levels also 686 pose a significant risk to shallow water sources in urban settings due to reduced attenuation in 687 the unsaturated zone when groundwater intersects with the base of pit latrines and sewer 688 networks (Lapworth et al. 2017). Another feature resulting indirectly from aquifer hyper-689 recharge is the high groundwater velocity which can foster the rapid transfer of pollutants 690 within the aquifer. Thus in the light of this work, it is urgent to implement pragmatic water 691 management strategies. The groundwater recharge zones identified on the potentiometric map 692 (Fig 6) must be protected. Any activity presenting a risk of contamination of water should be 693 prohibited and managers should promote the establishment of protection perimeters around

the tapping points. The knowledge of lateral flow velocity could help to design security 694 695 distances between pollution sources and groundwater pumping points. In addition, for 696 security and sustainability of groundwater use, a legal framework must now be established for 697 groundwater abstraction especially for boreholes construction to control more efficiently the 698 exploitation of the deeper part of the aquifer since it gathers relatively old and good quality 699 water. Indeed, the paleo-groundwater in the Mio-Pliocene must be considered as a strategic 700 resource since it may be non-renewable. Therefore, the abstraction of this type of water must 701 be planned and only reserved to public authorities. Furthermore, multilevel screened wells 702 must be prohibited to avoid mixing between different flow paths. In addition, stakeholder 703 participation through local committees for environment protection should be in charge of the 704 maintenance of landfills in settlements and/or watersheds. This is essential to limit the 705 influence of human activities on the aquifer system through a better land use management. 706 Furthermore, since surface waters are intimately related to groundwater, the continuous 707 degradation of this latter would impact on the long-term the socio-ecosystem of the local 708 river.

709 Finally, beyond the critical evaluation of NBL and TV determination method, this 710 research aims at increasing consciousness in the water management authorities 711 that groundwater status assessment should be conceived as a scientific evaluation 712 procedure and not as a technical/administrative exercise to be performed using any set 713 of available data. As such, it should be supported by adequate knowledge of 714 hydrogeological setting, hydrochemical processes occurring in groundwater bodies, and 715 human activities acting overland to be progressively verified.

#### 716 **6.** Conclusion

717 The results of this work may be extended to many porous aquifers in hyper-humid tropics.718 Indeed, this study by highlighting specific behaviors (e.g., high recharge rate, water recycling

719 within the vadose zone through plants transpiration, high groundwater velocity, acceleration 720 of hydrolysis and dissolution reactions thanks to the hot and humid climate) of aquifers in 721 hyper-humid areas, is a contribution to a better understanding of tropical hydrology. The 722 water table fluctuation method has proven to be effective for estimating groundwater recharge 723 in a shallow water table aquifer of humid area. The estimate is consistent with that found in 724 the literature and not far to that obtained by the application of water balance method. In spite 725 of low thermal amplitude characterizing the study region, the use of water stable isotopes 726 appears as a very powerful tool to study processes affecting the transfer of rainwater to the 727 vadose zone. Stable isotopes also shown that the studied aquifer submitted to hyper-recharge 728 gathers both recent infiltrated water and paleogroundwater. While in the past they have been 729 implemented only in arid and temperate climates, this study constitutes a new testimony of 730 the possibilities offered by isotope techniques in hydrogeological studies in humid countries.

731 Substantial information on groundwater recharge, groundwater flow conditions, groundwater 732 chemistry added to the determination of NBLs and TV will be helpful to elaborate future 733 groundwater management strategies, especially in developing countries of Africa. Indeed, the 734 limited waste management and high intrinsic vulnerability lead to urban aquifer (as Mio-735 Pliocene at Douala) highly impacted by chemical loading posing a clear health risk to users. 736 NBLs defined are essential when considering deterioration of groundwater quality, and in 737 addition these values can be a basis when restoring groundwater quality. Threshold values 738 should be incorporated in formulating new water quality standards for groundwater in 739 Cameroon, and more generally in sub-Saharan Africa. More generally, this approach provides 740 useful information for the establishment of a new water supply scheme with respect to the 741 climate specificity and the urban context.

#### 742 Authorship contribution statement

B. Nlend: Conceptualization, Investigation, Methodology, Resources, Writing - original and
revised manuscripts. H. Celle-Jeanton: Conceptualization, Methodology, Resources,
Supervision, Writing - review & editing. F. Huneau: Conceptualization, Investigation,
samples analyses, Supervision, Writing- review & editing. E. Garel: samples analyses,
Methodology, Writing - review & editing. S. Boum-Nkot: Conceptualization, Writing review & editing. J. Etame: Methodology & supervision.

749

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**Table 4** Estimated NBLs\* and TV\*\* of parameters for the MP aquifer at Douala. Reference(Ref) used are those defined by WHO (2011).

Age	Lithology	Thickness	Hydrogeological characteristic
Pliocene	Clay, sandy-clay, clayey-sands	0 - 40 m	Aquiclude/aquitard/aquifer
Lower Miocene	Cross-bedded sands with passages of lignite and limestone		
Upper Miocene	Coarse grained sand (slightly silty and calcareous) Gravel	>220 m	Aquifer

**Table 1** Hydro-stratigraphy of the Mio-Pliocene aquifer.

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	Mean	Minimum	Maximum	Standard deviation
Transmissivity (m²/s)	5.6 x 10 <sup>-3</sup>	4.6 x 10 <sup>-4</sup>	6.8 x 10 <sup>-2</sup>	6.1 x 10 <sup>-3</sup>
Hydraulic conductivity (m/s)	1.5×10 <sup>-3</sup>	2.6 x 10 <sup>-5</sup> m/s	0.2 m/s	0.03

		Groundwater from boreholes						Groundwater from hand-dug wells				Groundwater from springs (n = 10)				
		(n = 26)						(n = 18)								
Parameter	Unit	Min	Max	Mean	Med	SD	Min	Max	Mean	Med	SD	Min	Max	Mean	Med	SD
pН		3	8.5	4.7	4.4	0.9	3.2	7.6	5.7	5.8	0.8	3.3	7.6	4.9	4.8	1.1
Temp.	°C	27.4	31.5	28.9	28.6	1	27.8	29	28.3	28.2	0.4	27.2	29.8	28	27.9	0.7
EC	μS/m	13.6	1632	200.7	105.8	318.9	47	1048	428.4	399.5	268.8	111.8	375	255.3	274	85
DO	mg/l	2.3	7.8	6	6.5	1.4	2.2	7	4.5	4.4	1.4	2	7.8	4.3	4.3	1.8
K <sup>+</sup>	mg/l	0.1	19.3	4.7	2.9	5.4	1	41	12.4	10.6	10.3	2.3	11.7	6.8	6.7	3.6
Na <sup>+</sup>	mg/l	0.0	52.9	15.8	13.1	14.5	4.7	86.9	36.6	37.6	23.5	9.7	60.8	28	24.7	13.7
Ca <sup>2+</sup>	mg/l	0.2	32.9	5.4	3	7.4	1.8	81.1	24.7	22.9	21.4	1.4	21.3	7.7	6	6.4
Mg <sup>2+</sup>	mg/l	0.1	21.4	2.3	0.8	4.8	0.3	6.7	2	1.3	1.7	0.5	1.7	1	0.9	0.3
Cl-	mg/l	1.6	94.4	17.1	7.4	20.9	4.3	131.1	50.4	45.9	38.8	6.8	72.1	40	30	19.4
SO4 <sup>2-</sup>	mg/l	0.5	44.7	7.2	1.6	11.9	0.8	90.5	35	43.8	26	0.7	33.4	9.3	4.9	10.3
HCO <sub>3</sub> -	mg/l	0	137	9.4	0	28.8	0	222.2	28.9	0.4	68.9	0	0	0	0	0
NO <sub>3</sub> -	mg/l	1.1	145.5	35	26.1	36.1	7.6	245.6	56.7	43.6	56.1	28.2	93.4	61.6	61	21.6

Table 3 Descriptive statistics of chemical parameters. Concentrations of major ions are expressed in mg/L.

Parameters	Ref	NI	BLs	TV	Number of	samples above the	NBL (%)	Number of samples above the TV (%)			
		Min	Max		Boreholes	Hand-dug wells	Springs	Boreholes	Hand-dug wells	Springs	
EC (µS/cm)	250	13.6	41.3	145.7	73.1	100	100	38.5	77.8	90	
K <sup>+</sup> (mg/l)	100	0.1	0.5	50.3	76.9	100	100	0	0	0	
Na <sup>+</sup> (mg/l)	200	0.04	6.7	103.4	65.4	88.9	100	0	0	0	
$Ca^{2+}(mg/l)$	75	0.2	1.1	38.1	65.4	100	100	0	15.4	0	
Mg <sup>2+</sup> (mg/l)	30	0.1	0.8	15.4	61.5	83.3	80	0	0	0	
Cl <sup>-</sup> (mg/l)	250	1.6	4	127	69.2	100	100	0	5.6	0	
$SO_4^{2-}(mg/l)$	250	0.5	1.5	125.8	30.8	88.9	80	0	0	0	
HCO <sub>3</sub> <sup>-</sup> (mg/l)	200	0	9	104.5	15.4	11	0	3.9	5.6	0	
NO <sub>3</sub> <sup>-</sup> (mg/l)	50	1.1	6.6	28.3	26.9	100	100	46.1	66.7	100	

Table 4 Estimated NBLs\* and TV\*\* of parameters for the MP aquifer at Douala. Reference (Ref) used are those defined by WHO (2011).

\*NBLs = Natural background levels, \*\*TV = Threshold values

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Graphical abstract

