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# Selective extraction of neutral lipids and pigments from *Nannochloropsis salina* and *Nannochloropsis maritima* using supercritical CO<sub>2</sub> extraction: Effects of process parameters and pre-treatment

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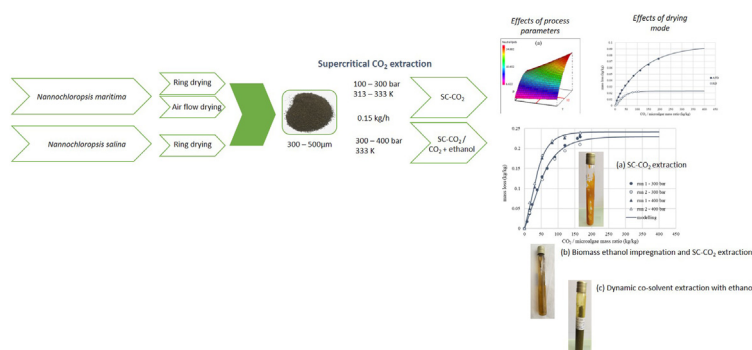
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## HIGHLIGHTS

- The effects of process parameters on neutral lipids and pigments extraction from *Nannochloropsis sp* were investigated.
- Extraction kinetics were compared between air flow dried and ring dried biomass.
- RSM was used to investigate the effects of pressure and temperature on the recovery of neutral lipids and pigments.
- Sovová's mathematical model was applied to better understand the mechanism of extraction kinetics.

## GRAPHICAL ABSTRACT



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## ABSTRACT

Supercritical CO<sub>2</sub> extraction experiments were conducted to investigate the effects of pretreatment and process parameters on neutral lipids, chlorophylls and carotenoids recovery on two species of *Nannochloropsis*. For *Nannochloropsis maritima*, a factorial experimental design was performed (P: [100–300] bar, T: [313–333] K). The highest extraction yields were obtained at the highest pressures and temperatures. Two drying modes, ring drying and air flow drying, were compared. Although total extraction yield and extraction kinetics were observed to be greater using air flow dried microalgae, extracts from this drying method resulted in partial degradation of glycerides in free fatty acids. Ring dried extracts maintained the same neutral lipid composition as the initial biomass. Based on these results, ring dried *Nannochloropsis salina* was extracted using supercritical CO<sub>2</sub> at 333 K and both 300–400 bar. Extraction curves were modelled using the Sovová's mathematical model.

**Abbreviations:** AFD, Air flow drying / Air flow dried; Chl a, Chlorophyll a; DM, Dry matter; EPA, Eicosapentaenoic acid; FD, Freeze drying / Freeze Dried; FFA, Free fatty acids; HPTLC, High performance thin layer chromatography; RD, Ring drying / Ring Dried; RSM, Response surface methodology; SC-CO<sub>2</sub>, Supercritical CO<sub>2</sub>; SEM, Scanning electron microscopy; TAG, Triacyl glycerides; TLC, Thin layer chromatography.

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## Nomenclature

AARD	Average absolute relative deviation (%)
$a_s$	Specific area between the regions of intact and broken cells ( $m^{-1}$ )
$C_u$	Solute content in the untreated solid ( $kg_{\text{solute}} / kg_{\text{solid}}$ )
$d_p$	Microalgae particle diameter (m)
$e$	Mathematical model extraction yield ( $kg_{\text{extract}} / kg_{\text{insoluble solid}}$ )
$E$	Amount extracted (kg)
$k_s$	Solid-phase mass transfer coefficient (m/s)
$n$	Number of experimental points
$N$	Solid charge in the extractor (kg)
$N_m$	Charge of insoluble solid (kg)
$P$	Pressure (bar)
$q$	Relative amount of the passed solvent ( $kg_{\text{solvent}} / kg_{\text{insoluble solid}}$ )
$\dot{Q}$	CO <sub>2</sub> flow rate ( $kg / s$ )
$r$	Grinding efficiency or fraction of broken cells
$t$	Extraction time (s)
$T$	Temperature (K)
$x_u$	Concentration in the untreated solid ( $kg_{\text{solute}} / kg_{\text{insoluble solid}}$ )
$y_s$	Solute solubility ( $kg_{\text{solute}} / kg_{\text{solvent}}$ )
<i>Greek letters</i>	
$\gamma$	Solvent to matrix ratio in the bed ( $kg_{\text{solvent}} / kg_{\text{insoluble solid}}$ )
$\varepsilon$	Bed void fraction

## Introduction

Microalgal lipids and antioxidants are promising raw materials for the nutraceutical, food, energy, cosmetic and pharmaceutical industries [1]. Microalgal lipid composition varies between species and is also affected by growth conditions including medium composition, temperature, light intensity, and aeration rate [2]. Among microalgae species, *Nannochloropsis* species is very promising for several industrial applications and is considered a source of commercially valuable pigments [3–5]. Under ideal conditions it contains 25 up to 45% lipids that can be used as feedstock for biofuel production and the nutraceutical industry as both eicosapentaenoic acid (EPA) and other long chain triacylglycerides (TAGs) [1,2,6]. For these reasons, *Nannochloropsis* sp. were selected for this study.

Microalgal lipid extraction is typically performed using toxic and non-selective (polar and neutral lipids are both solubilized) organic solvents such as n-hexane. If industrially relevant fractions of the extracted oil are desired, additional separation steps are required. A widely reported [7–10] alternative to organic solvents is supercritical CO<sub>2</sub> (SC–CO<sub>2</sub>). There are several advantages of using SC–CO<sub>2</sub> including its selectivity of neutral lipids, its gaseous state at ambient pressure making it simple to remove residual solvent, and its recyclability. Algal SC–CO<sub>2</sub> extraction efficiency primarily depends on pre-treatment conditions. The first pretreatment step for an efficient extraction is dewatering (solar drying, air flow drying, freeze drying, microwave, ...) followed by the increase of the cracking of specific surface area by mechanical methods (bead milling, ultra-sonication, ...). Non-mechanical methods like chemical cell disruption or enzymatic cell hydrolysis can be applied for helping the release of solutes of interest and their further recovery during extraction. Extraction efficiency also depends on free solute accessibility and internal diffusion (choice of process parameters and/or the use of co-solvent) [7,10–14].

A summary of recent studies dealing with the SC–CO<sub>2</sub> extraction of neutral lipids and valuable compounds from *Nannochloropsis* species [4,11,15–25] are reported in Table 1. Although a detailed and accurate comparison between these studies is difficult due to the range of different *Nannochloropsis* strains, growing conditions, pre-treatments, and extraction conditions (CO<sub>2</sub> flow rate, particle size diameter, initial water content, etc.). Some interesting tendencies can be highlighted. Studies dealing with the pretreatment effects on the extraction kinetics and efficiency [11,17,25] showed that air flow drying (AFD) was the most suitable drying mode leading to high extraction kinetics of neutral lipids in comparison to freeze drying (FD). Indeed, FD preserves the integrity of the microalga cell membranes and thus induces more limitation for diffusion leading to slower extraction kinetics. Adding distilled water to dried *N. oculata* to moisture contents up to 20 wt% seems to have no influence on the oil extraction yields [11]. Extracts contained both oil and water. Another pretreatment recently investigated suggests that cell wall damage of marine *Nannochloropsis* sp. caused by an osmotic gradient can have a positive effect on SC–CO<sub>2</sub> extraction kinetics with ethanol as co-solvent [25]. Performing one to three consecutive washing steps with distilled water created an osmotic pressure imbalance across the cell wall prior to FD. The authors showed that the pretreatment doubled the total extraction yield while increasing the concentration of acylglycerides and free fatty acids. Nevertheless, hydrolysis of glycerides was observed probably due to the pretreatment washing procedure.

Studies reporting the effects of process parameters [4,15,16,18,21,22] showed that when FD was performed, the highest extraction kinetics were obtained for pressures higher than 300 bar and temperatures close to 323 K. The highest extracted amount of chlorophyll *a* (Chl *a*) and total carotenoids with pure SC–CO<sub>2</sub> was obtained at 400 bar and 60 °C. The highest carotenoids/Chl *a* concentration ratio was obtained at 200 bar and 333 K [16]. The yield of extracted pigments depends on the operating conditions (influencing the SC–CO<sub>2</sub> density and the pigment vapor pressure) and on the nature of the other extracted compounds that can play the role of co-solvents [18].

When the effects of co-solvents were investigated [4,19,22–25], it was possible to observe that ethanol was the most appropriate co-solvent when pigments (chlorophyll, carotenoids, ...) were targeted. The yield of extraction depended strongly on the SC–CO<sub>2</sub> density and the amount of ethanol introduced [4]. When acetone was used as co-solvent, the extraction yield and lipidic extract composition weren't significantly enhanced compared to the extraction with pure SC–CO<sub>2</sub> at 320 bar and 313 K [19]. Furthermore, no significant difference was found in the fatty acids profile. Azeotropic mixture (Hexane/Ethanol) helped to extract a high amount of neutral and polar lipids but also EPA at 340 bar and 353 K when microwave pretreatment was applied to the dried raw biomass to enhance the cell disruption [24].

A review of the studies reported in Table 1 offers insights but also identifies gaps in the consistency of reported information (particle size diameter, initial water content, description of drying method, ...). These missing data are of prime importance for comparison purposes and scale up studies. Few studies report pretreatment conditions such as drying treatment mode and effects of high water content. Finally, the extraction of pigments was mainly investigated using a co-solvent. Additional SC–CO<sub>2</sub> investigations are needed to fill these gaps so industrial scale-up of this technology can be realized.

The goal of this study is to deepen the studies on the SC–CO<sub>2</sub> extraction of valuable compounds (neutral lipids, chlorophyll and total carotenoids) from *Nannochloropsis* species by investigating the effects of operating parameters (pressure, temperature), pretreatment (drying mode and effects of high water content) with pure SC–CO<sub>2</sub> and ethanol as co-solvent. Two species of

**Table 1**Main recent studies related to the SC-CO<sub>2</sub> extraction of valuable compounds from *Nannochloropsis* species.

Extraction with pure SC-CO <sub>2</sub>								
Microalgae	pre-treatments / particle size	P (bar)	T (K)	CO <sub>2</sub> flow rate	Extraction time (h)	Major extracted compounds	Best yield / goal	References
<i>Nannochloropsis</i> sp.	FD, d <sub>p</sub> =370 μm	400, 500, 700	313, 328	10 kg/h	6	C16:0, C16:1:n-7, EPA	Highest extraction yield: 25 % at 700 bar and 328 K / Investigate the effects of the operating conditions on extraction kinetics, yield and fatty acid composition.	[15]
<i>Nannochloropsis gaditana</i>	FD	100, 200, 300, 400, 500	313, 323, 333	0.012 kg/h	3	Chl a, carotenoids	Highest extraction yield at 400 bar and 333 K / The aim was to ascertain the influence of pressure and temperature on SC-CO <sub>2</sub> extraction of Chl a and carotenoids.	[16]
<i>Nannochloropsis</i> sp.	Oven drying at 343 K Grinding	125, 200, 300	313, 333	0.021, 0.037 kg/h	-	Neutral lipids	Best extraction yield: 33 % at 300 bar, 313 K and 0.037 kg/h / The aim was to use the biomass and apply biorefinery approach for the production of oil, high added-value compounds and biohydrogen.	[4]
<i>Nannochloropsis oculata</i>	AFD at 308 K, FD water content 4-5 wt% and 20 wt% d <sub>p</sub> < 160 μm, d <sub>p</sub> = 315 - 1000 μm	400	333	0.5 kg/h	-	C16:0, C16:1ω7, C18:1ω9	Extraction yield: 12 % for AFD and d <sub>p</sub> < 160 μm / Investigate the effects of high water-content, drying mode and particle size on extract composition and extraction kinetics	[11]
<i>Nannochloropsis oculata</i>	AFD at 308 K, FD water content 4-5 wt% and 20 wt% d <sub>p</sub> < 160 μm, d <sub>p</sub> = 315 - 1000 μm nitrogen limitation culture	400	333	0.4, 0.5 kg/h	-	C16:0, C16:1ω7, C18:1ω9, C20:5ω3	Extraction yield: 30 % for AFD, nitrogen limitation culture and d <sub>p</sub> = 160-315 μm / Investigate the effects of pre-treatment on extraction kinetics and modelling.	[17]
<i>Nannochloropsis gaditana</i>	d <sub>p</sub> = 616 μm, water content of 30 wt%	317, 377, 381, 425, 464, 479, 533, 543	309, 313, 323, 333, 337		1.2	Carotenoids, tocopherol, C16:0, C16:1, C18:1, EPA	Best extraction conditions for carotenoids: 533 bar and 337 K / Study of the effects of temperature and CO <sub>2</sub> density on the content of carotenoids and tocopherols.	[18]
<i>Nannochloropsis salina</i>	Drying at atmospheric pressure and temperature (298 K) for 16 h, Water content of 4.12 wt%, d <sub>p</sub> = 500-1000 μm	320	313	1.5 kg/h	-	Palmitoleic acid, Palmitic acid, Oleic acid, EPA	Extraction yield: 6.25 % / Investigate the extraction of lipids and valuable compounds from microalgae.	[19]
<i>Nannochloropsis oculata</i>	Washed with ammonium formate to remove salt, dried in a dehydrator at 318 K during 40h	300	313	9 mL/min	2	C16:0, C16:1, EPA, total carotenoids	Extraction yield: 15.6 % / Comparison of fatty acid composition, total carotenoids and antioxidants with sub-critical n-butane extraction.	[20]

Table 1 (Continued)

Extraction with pure SC-CO <sub>2</sub>									
Microalgae	pre-treatments / particle size		P (bar)	T (K)	CO <sub>2</sub> flow rate	Extraction time (h)	Major extracted compounds	Best yield / goal	References
<i>Nannochloropsis oculata</i>	FD, d <sub>p</sub> = 200 μm		250, 450, 750	323	1.5 kg/h	2	Neutral lipids, glycolipids, phospholipids	Extraction yield: 20 % at 750 and 450 bar / Investigate the selective extraction of neutral lipids from microalgae.	[21]
Extraction with SC-CO <sub>2</sub> + co solvent									
Microalgae	pre-treatments / particle size	co-solvent	P (bar)	T (K)	CO <sub>2</sub> flow rate	Extraction time (h)	Major compounds	Yield and or goal	References
<i>Nannochloropsis gaditana</i>	FD	Ethanol 5 mol%	200, 300, 400, 500	313, 323, 333	0.012 kg/h	3	Carotenoids and chlorophylls	Highest yield of carotenoids and chlorophylls at 500 bar and 333 K / The aim was to ascertain the influence of both pressure and temperature when using ethanol as co-solvent.	[22]
<i>Nannochloropsis</i> sp.	Oven drying at 343 K, grinding	Ethanol 5, 10 wt%	125, 200, 300	313, 333	0.021, 0.037 kg/h	–		Best extraction yield: 44 % at 300 bar, 313 K and 0.021 kg/h / The aim was to use the biomass and apply biorefinery approach for the production of oil, high added-value compounds and biohydrogen.	[4]
<i>Nannochloropsis salina</i>	Oven drying at 310 K for 2 days, water content 16 wt%, d <sub>p</sub> < 500 μm	Ethanol 5 wt%	300	318	0.4 kg/h	1.5	C16:0, C16:1cisω7, EPA	Extraction yield: 30.4 % / Increase the recovery of omega-3 and alpha linoleic acid in the extract.	[23]
<i>Nannochloropsis salina</i>	Drying at atmospheric pressure and temperature (298 K) for 16 h, water content of 4.12 wt% d <sub>p</sub> = 500-1000 μm	Acetone 3.2 wt%	320	313	1.5 kg/h	1.5	Palmitoleic acid, Palmitic acid, Oleic acid, EPA	Extraction yield: 6.38 % / Extraction of lipids and valuable compounds from microalgae.	[19]
<i>Nannochloropsis salina</i>	Classical microwave oven drying Water content of 6 wt%	Hexane / ethanol 4, 8 and 12 wt%	200, 270 and 340	313, 333, 353	6 kg/h	1	EPA, C16:1, C18:1, C16:0	Best extraction yield: 31.37 % at 340 bar, 353 K at solvent to solid ratio of 12 wt% / Investigation of the SC-CO <sub>2</sub> extraction of lipids from algal biomass using azeotropic co-solvents.	[24]
<i>Nannochloropsis</i> sp.	Washing step with MiliQ water to remove salts, FD, Grinding	Ethanol 15% v/v	230 and 380	308, 333	2 mL/min	6	Free fatty acids, neutral lipids, phospholipids	Extraction yield: 23.1 % at 230 bar, 333 K / Create osmotic gradient to damage algae cells for a better extraction yield	[25]

*Nannochloropsis* were considered: *Nannochloropsis maritima* (*N. maritima*) and *Nannochloropsis salina* (*N. salina*). To our knowledge, no SC-CO<sub>2</sub> extraction experiments have been reported on *N. maritima*. The effects of extraction process parameters using pure SC-CO<sub>2</sub> (AFD pretreatment) on *N. maritima* using Response Surface Methodology (RSM) was investigated. To complete the studies reporting effects of drying mode on microalgae [10,11,17], the effects of a new drying mode was investigated with ring drying (RD) as a viable pretreatment option. A ring dryer is an industrial-scale, pneumatic system that was developed to increase versatility of flash drying technology. RD is described in detail in the following references [26–28]. It is used in a number of industries and operates by exposing a slurry to superheated air in a circular form factor that facilitates controlled drying conditions and prevents overheating of the final product. Drying time is significantly shorter than AFD (a few seconds versus several hours) and thus limits the risk of lipid degradation. To our knowledge, this drying mode has not been reported for SC-CO<sub>2</sub> extraction in the literature. The SC-CO<sub>2</sub> extraction of valuable compounds from RD *N. salina* was investigated with pure CO<sub>2</sub> and with ethanol as co-solvent. A new method to disrupt cell walls by impregnating dried microalgae with ethanol was performed. The effect of a high-water content of the biomass on the extraction kinetics and extracted compounds was also investigated. Experimental extraction curves were modelled using Sovová's broken and intact cell (BIC) mathematical model to better understand the mechanism of extraction kinetics.

## Materials and methods

### Raw materials and chemicals

*Nannochloropsis* samples were provided by Cellana LLC (Hawaii, USA). The algae cultures were grown in outdoor conditions, harvested, and thickened to 22% solids with a Westfalia disk stack centrifuge. *N. maritima* samples were provided both as ring dried (RD) material and as wet paste material. *N. salina* samples were provided as RD material. The ring dryer used in this study operated at 353 K and the average retention time was less than 30 s. The wet paste was air flow dried (AFD) in a circulating air oven (classical plant dryer) SFP 6 type (R.E.U.S France) which was set to the optimum conditions of 308 K for 20 h according to a previous study [10]. The final water content of the resulting samples was about 13 wt% and about 3.4 wt% for AFD and RD biomass respectively. The dried biomass was ground into powder and sized into homogenous particle size ranges with a vibratory sieve shaker at a mean particle size of 300–500 μm. Algae powder was stored at 277 K prior to experiments to avoid degradation of the compounds. SC-CO<sub>2</sub> extraction experiments were carried out with carbon dioxide of 99.7% purity supplied by Air Liquide Méditerranée (France). Absolute ethanol with a purity of 99.8%, used as co-solvent, was supplied by Fisher (France).

### Experimental setup

The process diagram of the experimental setup is given in

Fig. 1 Extractions were performed using laboratory scale equipment provided by SEPAREX (Champigneulle, France). The CO<sub>2</sub> high pressure pump (3) used for this study was a high-pressure liquid CO<sub>2</sub> pump Jasco PU-4386 (Jasco France). A high-pressure solvent pump (10), GILSON 307 Pump with standard head pump of 5SC type (Gilson Inc., USA), was connected to the extraction pilot in order to clean the pipes and the micrometric valve (V3) at the end of each experiment or for co-solvent extractions. All experiments were carried out in a 10 cm<sup>3</sup> extraction vessel (6) filled to a vol-

ume of 80%, which corresponded to a sample mass of 5 g for each experiment.

Considering the small charges of biomass in the extraction vessel, the amount of extracted neutral lipids *E* was calculated considering the mass losses of the extraction autoclave. Weight losses were recorded prior to extraction and 30 min after extraction to allow liberation of residual CO<sub>2</sub>. Extract samples were recovered in glass vessels which were subsequently topped with nitrogen and stored at 255 K until compositional analysis. The yield was expressed as the mass ratio of the sample mass loss (*E*) to the mass introduced initially in the extraction autoclave (*N*). This yield was assimilated to neutral lipids yield since it is well known that SC-CO<sub>2</sub> solubilizes non polar-lipids.

Extraction experiments with pure CO<sub>2</sub> were conducted with valves V4 and V5 closed. The CO<sub>2</sub> flow and pressure were controlled with micrometric valve V3. The CO<sub>2</sub> flowrate was measured thanks to a flowmeter located at the end of the extraction line (8). When the extraction experiments were completed, pipes and micrometric valve were washed with absolute ethanol by closing valve V2 and opening valve V4. The apparatus was then dried with a flow of gaseous CO<sub>2</sub> during a few minutes (Valves V2, V3 and V5 open, valve V4 closed).

For experiments with co-solvent, absolute ethanol was added at system pressure to SC-CO<sub>2</sub> via a union tee (1/8") prior to the extraction vessel (blue line in

Fig. 1). Only a qualitative evaluation of the extraction is made by analyzing the extract composition. Co-solvent experiments were conducted for 3 h and the extracts were analyzed for composition as described below.

### Experiments performed on *N. Maritima*

Extraction experiments performed on dried *N. maritima* were conducted without co-solvent at a constant CO<sub>2</sub> flow rate of 0.15 kg/h. The effects of pressure and temperature on the yield of extracted neutral lipids, on Chl a and total carotenoids concentration in the extracts at a CO<sub>2</sub> / microalgae mass ratio of 180 kg/kg were investigated using Response Surface Methodology (RSM). Three levels were considered for the pressure and the temperature (-1, 0, 1 respectively for pressures of 100, 200 and 300 bar and temperatures of 313, 323 and 333 K). The degree of significance of each factor (p-value) was given by NemrodW software and was discussed. A classical plan composed of 9 experiments was considered. Each response *Y* (neutral lipids yield, chlorophyll *a* concentration and total carotenoids concentration) was modelled using a second-order polynomial model given in Eq. 1.

$$Y = b_0 + b_1P + b_2T + b_{11}P^2 + b_{22}T^2 + b_{12}.T.P \quad (1)$$

Where *b*<sub>0</sub>, *b*<sub>1</sub>, *b*<sub>2</sub>, *b*<sub>11</sub>, *b*<sub>22</sub>, *b*<sub>12</sub> are the coefficient of the response surface equation calculated by NemrodW software (LPRAI, Marseille, France).

The effect of drying mode on extraction kinetics was investigated at 100 bar and 313 K.

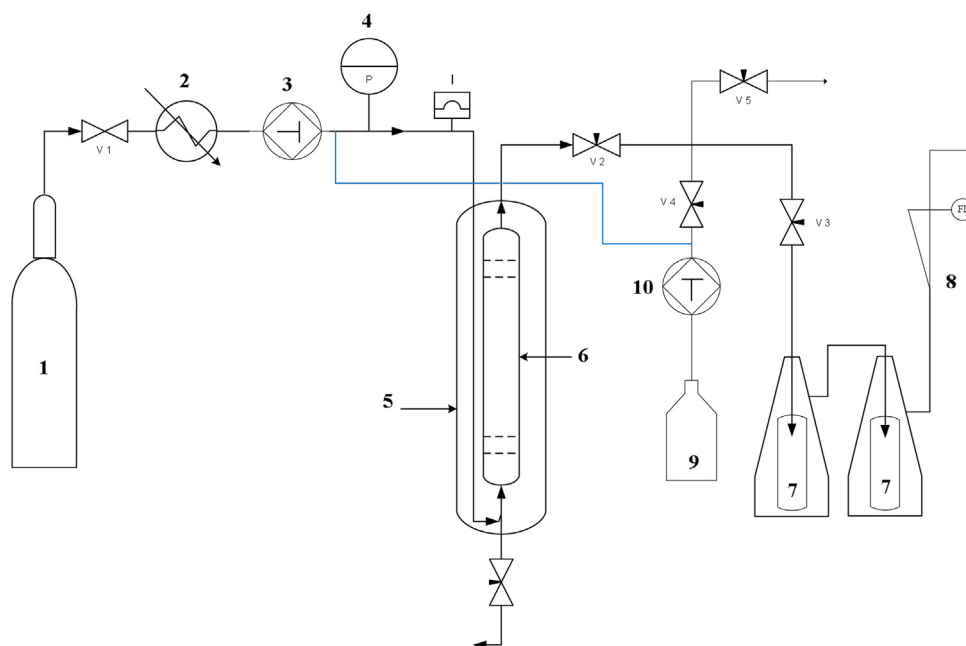
All experimental conditions were summarized in Table 2.

### Experiments performed on *N. Salina*

SC-CO<sub>2</sub> extraction experiments on RD *N. salina* were performed at pressures of 300 and 400 bar, a temperature of 333 K and a CO<sub>2</sub> flow rate of 0.15 kg/h with pure CO<sub>2</sub>, ethanol as co-solvent and with biomass at high water content. The operating conditions were chosen by building on the previous reported study [22]. All experiments performed on RD *N. salina* were reported in Table 2.

Two kinds of experiments were performed using ethanol as co-solvent. The first experiment consisted in impregnating the ground and sieved microalgae with absolute ethanol for 24 h. The biomass





**Fig. 1.** Process diagram of the experimental setup: (1) CO<sub>2</sub> tank, (2) cooler, (3) CO<sub>2</sub> high pressure liquid pump, (4) pressure gauge, (5) extraction apparatus heater, (6) extraction vessel, (7) glass vessel, (8) flowmeter, (9) ethanol tank, (10) Gilson high pressure solvent pump.

**Table 2**  
Summary of experimental conditions.

Strain	Drying mode	Initial water content (wt %)	Extraction conditions				
			Co-solvent	100 bar	200 bar	300 bar	400 bar
<i>N. maritima</i>	AFD	13	-	313 K	313 K	313 K	-
			-	323 K	323 K	323 K	-
			-	333 K	333 K	333 K	-
<i>N. salina</i>	RD	3.4	-	-	-	-	
		3.4	-	-	333 K	333 K	
		23.9	-	-	-	-	
		3.4	Ethanol <sup>a</sup>	-	-	333 K	-

<sup>a</sup> Two ethanol co-solvent conditions: maceration (24 h) and in-line with SC-CO<sub>2</sub> (4.3% wt/wt).

was then introduced in the extraction vessel to perform SC-CO<sub>2</sub> extraction. The ethanol:biomass mass ratio was 2:5 (wt/wt). In the second experiment, the co-solvent was pumped by a high-pressure pump (Gilson 307) and mixed with the pressurized CO<sub>2</sub> flow before entering the extraction vessel.

To deepen the study about the effects of water on extraction kinetics and on extract composition, distilled water was added to RD *N. salina* samples. An amount of 5 g of ground and sieved powder of microalgae was introduced in a sealed vessel filled with distilled water until the desired water content was reached. The water content was increased from 3.4 wt% to 23.9 wt%.

### Analytical methods

#### Neutral and polar lipids

Extracts were analyzed for determining lipidic profiles as follows: 1–5 mg samples were diluted in a solution of chloroform:methanol (2:1) to a concentration of 1 mg/mL and stored at 253 K until analysis. High Performance Thin Layer Chromatography (HPTLC) analysis of the lipids content was performed by ATS4 spray deposition on a Merck 60 silica gel followed by ADC2 development at 70 mm. Samples were dissolved in a n-hexane:diethyl ether:acetic acid (17:3:0.2) solution for neutral lipid analysis and in a acetone:toluene:water (91:30:8) solution for polar lipids. Revelation was carried out by immersion in a copper sulfate/orthophosphoric acid solution at 443 K for 20 min. Spectral

analysis was performed on a TLC scanner 3 at 500 nm. Monitoring and data treatment were completed using Wincat software.

Gas Chromatography Mass Spectrometry (GC-MS) analysis of the composition in Fatty Acid Methyl Esters (FAME) was carried out using an Agilent GC equipped with a ZB-WAX (0.25 m × 30 m) column, H<sub>2</sub> as a carrier gas - split mode on two detectors. Prior to analysis, a trans-methylation was carried out in a solution of methanol / sulfuric acid 5% (1.5 h; 358 K) followed by extraction with n-hexane. Quantification was completed using Flame Ionization Detection (FID) and mass spectroscopy. Calibration was performed with standards ranging from C14 to C24.

#### Chlorophyll a and total carotenoids

Total carotenoids and chl a concentration was carried out by measuring the absorbance of extract samples diluted in pure acetone using a Jenway 6715 UV/vis spectrophotometer. The correlations proposed by Lichtenthaler et al. [29], given in Eq. 2 to 4, were used in this study. Before absorbance measurements of extract, a blank measure was made with pure acetone.

$$C_a(\mu\text{g/mL}) = 11.24A_{661.6} - 2.04A_{644.8} \quad (2)$$

Where C<sub>a</sub> is the concentration of chlorophyll a (Chl a), A<sub>661.6</sub> the absorbance (arbitrary unit) at the wavelength of 661.6 nm

and  $A_{644.8}$  is the absorbance (arbitrary unit) at the wavelength of 644.8 nm.

$$C_b(\mu\text{g/mL}) = 20.13A_{644.8} - 4.19A_{661.6} \quad (3)$$

Where  $C_b$  is the concentration of chlorophyll *b* (Chl *b*).

Finally, the concentration in carotenoids is estimated using the following expression:

$$C_{\text{totalcarotenoids}(x+c)}(\mu\text{g/mL}) = (1000A_{470} - 1.9C_a - 63.14C_b) / 214 \quad (4)$$

The diluted solutions composed of the extracts in acetone must be fully transparent as a turbid solution leads to a higher absorption between 400 and 800 nm. This higher absorption would give an overestimation of the pigment level especially for Chl *b* and total carotenoids. Hence, turbidity must be checked by measuring the absorbance at 750 nm ( $A_{750}$ ). For a fully transparent extract,  $A_{750}$  should equal zero since Chl *a* and *b* and carotenoids do not absorb in this region. Analysis showed that the studied strains do not contain Chl *b* which is in accordance with studies reporting that *Nannochloropsis* contains mainly Chl *a* and carotenoids (violaxanthin) [3,4]. The concentrations of each pigment (Eqs. 2 – 4) were expressed for a solvent volume of 5 mL [30]. In this study, the results were expressed in  $\mu\text{g}$  of pigment / g of dry microalga ( $N$ ).

#### Scanning electron microscopy (SEM)

Samples of *N. maritima* and *salina* were observed with a Scanning Electron Microscope. The visualizations were made on a Hitachi TM3000 Tabletop Microscope (15 kV). The sample powders of microalgae were deposited on an adhesive pastille and metallized with a thin layer of Au-Pd to allow better sample conduction and visualization. The metallization time did not exceed 30 s.

#### Modelling of the extraction curves

The extraction curves were modelled using Sovová's mathematical model for supercritical fluid extraction of natural products [31]. This model takes into account the structure of the biological material studied and accounts for the sudden reduction in extraction rate after the first extraction period that is observed during SC-CO<sub>2</sub> extractions from seeds. This model was already applied to describe the extraction curves obtained from microalgae [10,17,23]. Four types of extraction curves can be found: A, B, C or D. In the decision on extraction type one can refer to the shape of the extraction curves as shown in Fig. 2. If the first part of the extraction curve consists of one straight section, the type is A or D. If it consists of two straight sections, it is of type B or C.

In this study, the first part of the extraction consists of one straight section. Hence, the type can be A or D. The slope of the first part of the curve is related to solute solubility for type A while it is lower than solute solubility for type D.

Experiments were performed to estimate the solute solubility in SC-CO<sub>2</sub> at the operating conditions for *N. maritima* and *N. salina* based on the methodology proposed by Perrut et al. [32]. Extracted solute was spread on glass beads (diameter of about 200  $\mu\text{m}$ ) and SC-CO<sub>2</sub> extraction was performed at the same operating condition with a CO<sub>2</sub> flow rate of 0.02 kg/h for 80 min to ensure fluid phase equilibrium in the system. It appears that the slope of the extraction curve at initial time was related to the solute solubility whatever the drying mode. These observations lead to the selection of type A.

The flow pattern was assumed to be plug flow. The bed void fraction  $\varepsilon$  of *N. maritima* is 0.379 and 0.497 for AFD and RD biomass respectively and 0.295 for RD *N. salina*. The methodology for the determination of  $\varepsilon$  and plug flow assumption was already described in details in previous work [17]. The type A extraction curve can be

described by considering Eq. 5 and 6 (the model parameters are described in the nomenclature):

#### First extraction period

$$e = qy_s \text{ for } 0 \leq q \leq q_1 \quad (5)$$

Second extraction period

$$e = x_u [1 - C_1 \exp(-C_2 q)] \text{ for } q > q_1 \quad (6)$$

With:

$$r = 1 - C_1 \exp\left(\frac{-C_2 q_1}{2}\right) \quad (7)$$

$$k_s a_s = (1 - r)(1 - \varepsilon) \frac{\dot{Q} C_2}{N_m} \quad (8)$$

$$e = \frac{E}{N_m} \quad (9)$$

$$q = \frac{\dot{Q} t}{N_m} \quad (10)$$

$$N_m = (1 - C_u N) \quad (11)$$

$$C_u = \frac{x_u}{1 + x_u} \quad (12)$$

For the first part of the extraction curve ( $0 \leq q \leq q_1$ , where  $q_1$  represents the end of the first extraction period), the slope  $y_s$  value, which represents the extracted solute solubility, was chosen according to the solubility determination previously described. The second extraction period ( $q > q_1$ ) is described by adjusting constant parameters  $C_1$  and  $C_2$  on experimental data. Estimations of parameters  $k_s a_s$  and  $r$  can then be obtained by considering Eq. 7 to 12. The adjustable parameters  $C_1$  and  $C_2$  were calculated by minimizing the sum of least squares between the experimental and calculated values of  $e$ . The average absolute relative deviation (AARD) given in Eq. 13, was used to evaluate the efficiency of the model.

$$\text{AARD}(\%) = \frac{100}{n} \sum \left| \frac{\text{Experimental yield} - \text{Calculated yield}}{\text{Experimental yield}} \right| \quad (13)$$

The modelled extraction curves were expressed as the variation of the yield (mass ratio of the sample mass loss to the mass introduced initially in the extraction autoclave) as function of the CO<sub>2</sub> / microalgae mass ratio ( $\dot{Q} \times t / N$ ).

## Results and discussions

### Extraction of valuable compounds from AFD *N. Maritima*

#### Effects of process parameters

Prior to SC-CO<sub>2</sub> extraction, a repeatability test (Fig. 3) was performed on AFD biomass at 300 bar and 333 K. The average deviation between the two sets of experiments is about 5.9% which is satisfactory. The shape of the curve shows a nearly complete extraction with a yield of 14.5% using a CO<sub>2</sub>/microalgae mass ratio of 180 kg/kg. The modelled curve confirms this observation, the increase in the extracted amount of solute is extremely low from 180 kg/kg up to 400 kg/kg.

The effects of operating conditions on the neutral lipid yield (Fig. 4 (a)), Chl *a* (Fig. 4 (b)) and total carotenoid (Fig. 4 (c)) concentration at a CO<sub>2</sub>/microalgae mass ratio of 180 kg/kg have been reported in Fig. 4.

The polynomial coefficients of Eq. 4 are reported in Table 3. The degree of significance of each factor (pressure, temperature) is represented in Table 3 by its p-value in %. When a factor has a p-value lower than 5%, it influences the process in a statistically significant way.



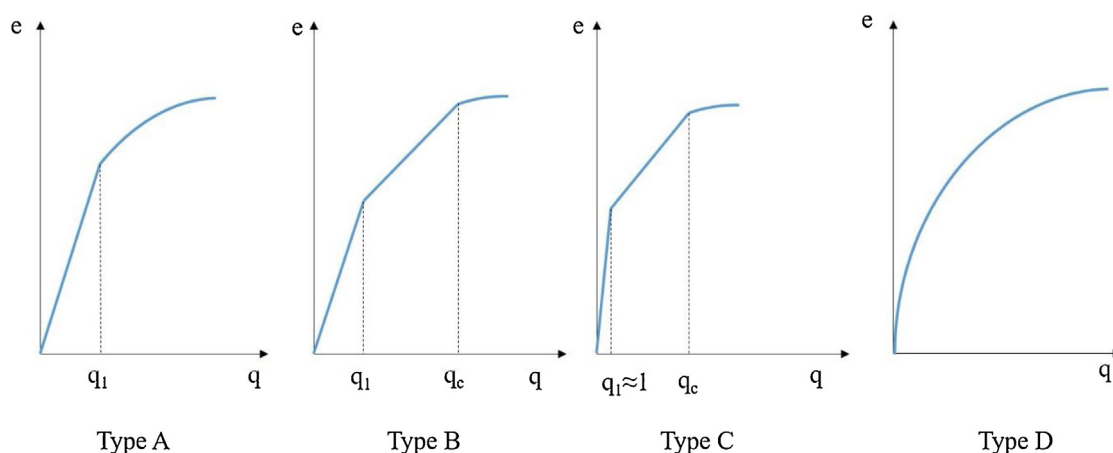


Fig. 2. Aspects of the type of extraction curves according to Sovová's mathematical model.

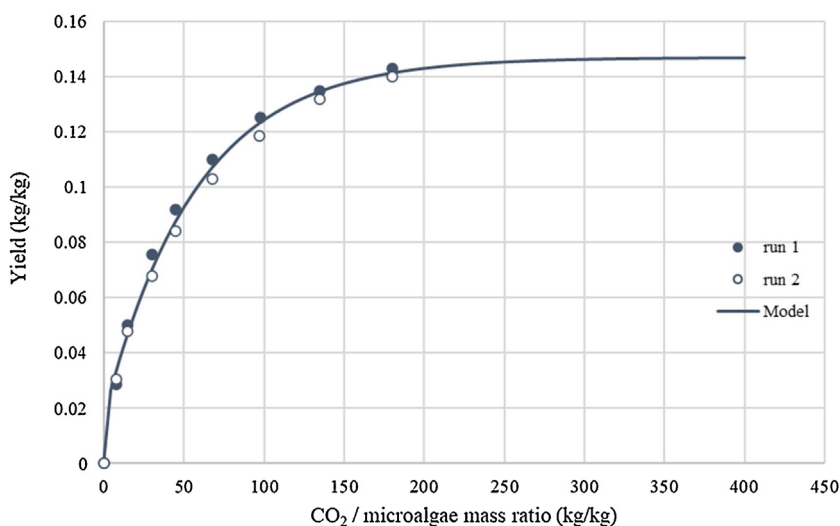


Fig. 3. SC-CO<sub>2</sub> extraction curves, repeatability test on AFD *N. maritima* at 300 bar and 333 K.

Table 3

Regression coefficients of the polynomial RSM equation and analysis of variance for neutral lipids yield, Chl a and total carotenoids concentration.

Coefficient (factor)	Neutral lipids yield (%)		Chl a (μg/g)		Total carotenoids (μg/g)	
	Coefficients values	p-value (%)	Coefficients values	p-value (%)	Coefficients values	p-value (%)
b <sub>0</sub>	11.41	-	0.321	-	0.35	-
b <sub>1</sub> (P)	2.465	0.72*	0.362	9.2	0.302	2.18 <sup>a</sup>
b <sub>2</sub> (T)	1.562	2.53 <sup>a</sup>	0.005	97.5	0.055	48.1
b <sub>11</sub> (PP)	-1.87	6.4	0.288	34.3	0.105	44.2
b <sub>22</sub> (TT)	0.052	94.1	-0.312	31.2	-0.165	25.9
b <sub>12</sub> (PT)	1.055	10.6	0.007	97	0.083	39.9
R <sup>2</sup>	0.96	-	0.74	-	0.89	-

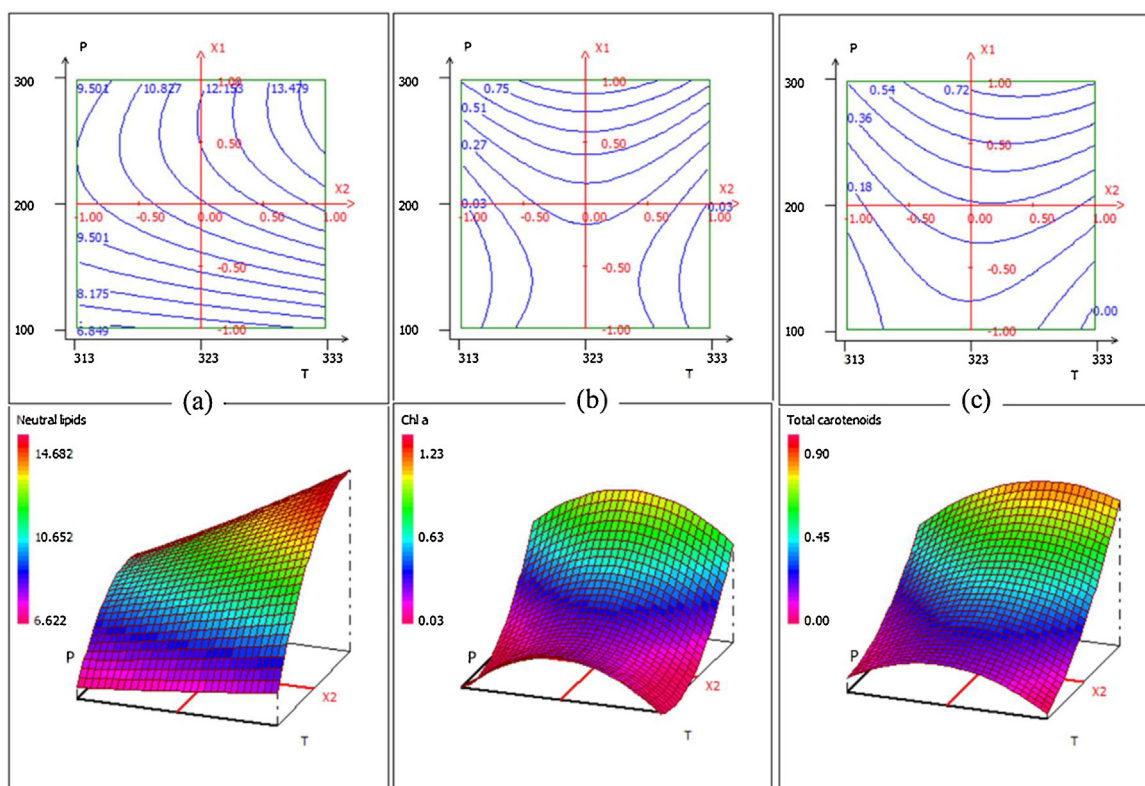
<sup>a</sup> Significant factor.

Regardless of temperature and drying process, at 100 bar the recovery of Chl a and total carotenoids in the extracts was very low (between zero and 0.07 μg/g). This result was expected as carotenoids have low solubility in SC-CO<sub>2</sub> at these experimental conditions [33]. A similar result was obtained by Macias Sanchez et al. [16].

Pressure and temperature were reported to significantly influence the amount of extracted neutral lipids (Table 3 and Fig. 4 (a)). Nevertheless, pressure had a more significant effect than temperature. Indeed, increasing the pressure at a set temperature significantly increased the amount of extracted neutral lipids. Increasing both pressure and temperature led to an increase of the

extracted amount of neutral lipids. This is due to a higher solubility of the extracted compounds in the SC-CO<sub>2</sub>. The highest neutral lipids yield (14.5%) was found at 300 bar and 333 K. The retrograde solubility behavior was not observed at these operating conditions.

In the experimental regime below 200 bar, no significant effects of pressure and temperature were found for the recovery of Chl a while pressure was found to be a significant factor for the recovery of total carotenoids (Table 3). Above 200 bar, increasing pressure at a given temperature increases extract concentration of both pigments. A similar tendency was reported by Macias-Sanchez et al. [16] on *N. gaditana*. The optimal experimental operating condi-



**Fig. 4.** *N. maritima* 2D and 3D surface plot of (a) neutral lipids; (b) Chl a; (c) total carotenoids as function of pressure and temperature. Axes labels are identical in all three plots.

tions for the recovery of pigments from AFD *N. maritima* were found to be at 300 bar and 323 K (1.39  $\mu\text{g/g}$  and 0.95  $\mu\text{g/g}$  for Chl a and carotenoids respectively). Macias-Sanchez et al. reported, at 300 bar, a better recovery of pigments at 333 K. The difference may be due to drying method leading to different final cell structures.

Macias-Sanchez et al. [16] reported that the highest amount of Chl a and carotenoids were recovered at 400 bar and 333 K. An extraction experiment was carried out at these experimental conditions with a  $\text{CO}_2$  flow rate of 0.15 kg/h. At a  $\text{CO}_2$ /microalgae mass ratio of 180 kg/kg, the neutral lipids yield was about 15.5 wt% and the concentration of Chl a and total carotenoids were 1.07 and 0.83  $\mu\text{g/g}$  respectively. At 333 K, increasing the pressure led to increasing the recovery of neutral lipids and the concentration of pigments in the extracts. In this study, the best operating conditions for the extraction of both Chl a and total carotenoids were 300 bar and 323 K at a  $\text{CO}_2$  flow rate of 0.15 kg/h..

Considering the results reported in this study and by Macias-Sanchez et al. [16], when pigments are targeted, pressures lower than 200 bar should be avoided due to the low solubility of pigments in  $\text{SC-CO}_2$ .

#### Neutral lipids composition

Lipid analyses to determine FFA and TAG contents (lipid classes) were performed on the  $\text{SC-CO}_2$  extracts for pressures ranging from 100 to 300 bar and temperatures ranging from 313 to 333 K. Extracts were esterified and the fatty acid methyl esters (FAMES) were identified (Analyses could not be performed for the experiments conducted at 100 bar, 323 and 333 K due to a low amount of collected oil (extracts were primarily composed of volatile compounds)).

Table 4). TAG and FFA amounts are described as the mass of the constituents relative to the extract mass (wt%). Analyses could not

**Table 4**

Neutral lipid content in extracted oil from *N. maritima*.

Experiment	FFA (wt%)	TAG (wt%)	TAG + FFA (wt%)	C20:5 (wt%) <sup>a</sup>
100 bar, 313 K	39.8	5.1	44.9	7.1
200 bar, 313 K	28.7	31.6	60.3	8.3
200 bar, 323 K	30.7	28.1	58.8	7.7
200 bar, 333 K	28.9	31.9	60.8	7.9
300 bar, 313 K	32.1	21.3	53.4	8.5
300 bar, 323 K	20	36.1	56.1	8.3
300 bar, 333 K	27.6	33.9	61.5	7.5

<sup>a</sup> Total lipids.

be performed for the experiments conducted at 100 bar, 323 and 333 K due to a low amount of collected oil (extracts were primarily composed of volatile compounds).

A non-negligible amount of EPA (C20:5), a high added value  $\omega$ -3 acid was found. The amount of EPA varied between 7.1 and 8.5 wt%. An amount of the same order of magnitude (8.06 wt%) was reported by Millao et al. [18] at 593 bar and 337 K on FD *N. gaditana*.

Other FAMES including C15:0, C16:1, C18:0, C18:1, C18:2, C20:0 and C20:4, were found to be at constant value respectively 0.4%, 20%, 1.5%, 0.5%, 1.3%, 0.3% and 2.6% regardless of the extraction conditions. Palmitoleic acid (C16:1) was found to be the main extracted FAME from *N.maritima*. Similar results were reported in the literature [4,15,19,20]. Based on these studies it is well-known that the amount of extracted FAMES should be the same when Soxhlet extraction is performed.

The overall results reported in Table 4 suggest a degradation of TAG to FFA, a similar observation was reported by Crampon et al. [11] and Mouahid et al. [17]. This degradation was perhaps due to the remaining water content after the AFD process. The main issue is that a higher drying period would lead to a degradation of valuable compounds [10]. The highest amount of TAG + FFA (61.5 wt%)

was found to be recovered at 300 bar and 333 K and the lowest (44.9 wt%DM) at 100 bar and 313 K.

#### Effect of drying mode on SC–CO<sub>2</sub> extraction kinetics

The effect of the drying mode on the SC–CO<sub>2</sub> extraction kinetics from *N. maritima* was studied at 100 bar and 313 K. The extraction curves are reported in Fig. 5 and clearly show a higher extraction kinetic when *N. maritima* is AFD.

The mass losses from AFD biomass averaged about 55% higher than those from RD biomass. In an effort to understand how the drying mode affected the structure of the microalgae, RD and AFD *N. maritima* were evaluated by SEM (Fig. 6). SEM images suggest that RD produced fewer fine cracks in the particulates' surface than AFD. The thick cell wall structure of *Nannochloropsis* sp. is known to be difficult to disrupt for extraction of intracellular products [34,35]. It seems that the high temperature and short residence time of RD (about 353 K; less than 30 s) did not cause sufficient alteration of cell walls of the biomass. Considering that the initial extraction curve is related to solubility, it appears that when RD was applied, a less solute was accessible to SC–CO<sub>2</sub>. Additionally, the SC–CO<sub>2</sub> extracted solute composition varied depending on the drying mode. Quantities of FFA and TAG in wet *N. maritima* paste, in RD biomass (pre-extraction), and AFD biomass (pre-extraction) were determined. The TAG/(TAG + FFA) ratios were about 83%, 85%, and 65% for the algal paste, RD biomass, and AFD biomass, respectively.

This suggests that the amounts of TAG and FFA were not significantly modified during RD compared to AFD where higher water content (~13 wt%) and longer processing time were associated with degradation of TAG to FFA by hydrolysis. A similar effect due to water was observed by Elst et al. [25]. Depending on the drying mode, the cell walls of microalgae were impacted in different ways inducing a different accessible solute composition to SC–CO<sub>2</sub>. It seems that cells are more intact when RD is applied compared to AFD. In addition, the intensity and duration of AFD appears to have caused a degradation of glycerides to FFA and modify the initial solute composition as shown in other studies [10,11]. The extraction kinetics were lower when RD was performed but, the extracted oil quality is better in comparison with extracted oil from AFD biomass. Since glycerides were the solute of interest in this study, RD seems to be more appropriate than AFD.

#### Experiments performed on RD *N. Salina*

Considering that RD did not lead to a degradation of microalgae compounds (section 4.1.3), it was chosen to perform SC–CO<sub>2</sub> extraction on RD *N. salina*. The SC–CO<sub>2</sub> extraction experiments were performed at 300 and 400 bar, at a temperature of 333 K, and a CO<sub>2</sub> flow rate of 0.15 kg/h. The effects of co-solvent and high-water content were studied.

#### Extraction of valuable compounds from *N. Salina* with pure SC–CO<sub>2</sub>

The SC–CO<sub>2</sub> extraction curves obtained from *N. salina* are presented in Fig. 7. Two set of experiments were performed for each operating condition to check the repeatability. The average deviations between the two sets of experiments were comprised between 2.2% and 3.9% which is very satisfactory. In Fig. 7, the shapes of the modelled SC–CO<sub>2</sub> extraction curves indicate that the extraction was near completion at a CO<sub>2</sub>/microalgae mass ratio of 180 kg/kg (yield of 23% at 300 bar and 24% at 400 bar). Considering that only neutral lipids are extracted by SC–CO<sub>2</sub> and the extraction is almost complete, it can be considered that the strain of *N. salina* presented in this study contains about 24 wt% of neutral lipids.

Pigment analysis (Chl a and carotenoids) are reported in Table 5. The total carotenoids concentration was consistently higher than Chl a concentration. Increasing the pressure from 300 to 400 bar

**Table 5**

Chl a and total carotenoids concentration in *N. salina* extracts for a CO<sub>2</sub> / dry microalgae mass ratio of 180 kg/kg.

Experiment	Concentration (µg/g)	
	Chl a	Total carotenoids
300 bar	0.11	0.58
300 bar ethanol impregnation	0.34	0.66
400 bar	0.12	0.78
400 bar – 23.9 wt% water	0.27	0.49

induced an increase in the total carotenoids' concentration in the SC–CO<sub>2</sub> extracts (about 29%), while the concentration of Chl a was more or less the same. Increasing the pressure seems to have the effect only on the recovery of total carotenoids. The observations were slightly different than those reported by Macias-Sanchez et al. [16] who reported an increase of both Chl a and total carotenoids from FD *Nannochloropsis gaditana*. This difference can be explained by an accessible solute of different composition caused by the drying mode.

#### Extraction of valuable compounds from *N. Salina* with ethanol as co-solvent

Two experiments were performed at 300 bar and 333 K with ethanol as co-solvent. In the first extraction, the biomass was impregnated with ethanol (ethanol:biomass = 2:5 wt/wt). Then the system was kept in a closed vessel under controlled humidity to allow the maceration of the biomass in ethanol for a duration of about 24 h. The second experiment was performed using a mixture of SC–CO<sub>2</sub> with ethanol as co-solvent. The mass ratio CO<sub>2</sub>/ethanol was 4.3% (wt/wt). Difference in color between the extracts recovered at 300 bar and 333 K with and without co-solvent has been observed. The different colored extracts, orange without any co-solvent, maroon for maceration and green for SC–CO<sub>2</sub> ethanol with ethanol as co-solvent, indicate that they did not have the same composition.

SEM was performed on RD *N. salina* (Fig. 8) for the extraction conditions studied. The effect of the maceration process on cell disruption is shown by comparing Fig. 8 (a) and (c). It can be seen that the particulate structure of the biomass was significantly disrupted and mainly allowed the extraction of both Chl a and carotenoids (Table 5). Using an ethanol co-solvent increased the recovery of both Chl a and total carotenoids by 67.6 and 12% respectively compared to pure SC–CO<sub>2</sub> (Fig. 8 (b) and (d)). The extract color indicates that the particulate structure of the biomass was disrupted more intensively after co-solvent extraction and allowed increased extraction of Chl a.

#### Effects of high-water content

A SC–CO<sub>2</sub> extraction experiment was performed at 400 bar and 333 K on a sample containing a water content of 23.9 wt%. The results obtained were compared to those obtained with extraction experiment performed with sample of water content of about 3.4 wt%. The extraction kinetics were reported in Fig. 9.

In previous works [10,11] when *Dunaliella salina* paste biomass was submitted to different degrees of air flow drying (water content from 5 to 23 wt%) [10] and when AFD *N. oculata* biomass contained moisture up to 20 wt% [11], an increase of the global extraction kinetics was reported. Water was not reported to act as barrier for diffusion for *N. oculata* and was reported to act as cosolvent for the extraction of pigments from *Dunaliella salina*. In the present study, the extraction kinetics are higher with lower water content. When extraction was performed on biomass with water content of 23.9 wt%, the shape of the extraction curve differed from that usually obtained (Figure 9). The analysis of the extract shows that when water was added, the concentration of Chl a was approximately two times higher than RD *N. salina* while the concentration of total

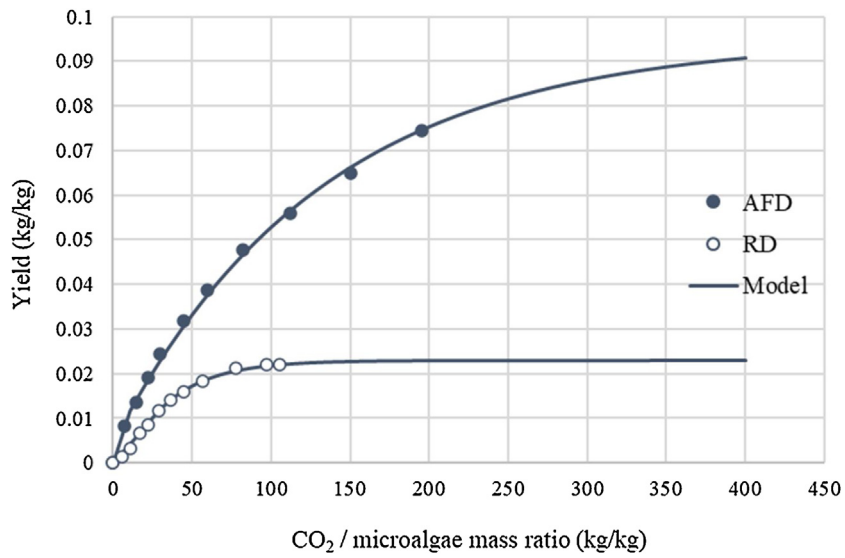


Fig. 5. Effects of drying mode on extraction kinetics at 100 bar, 313 K and 0.15 kg/h.

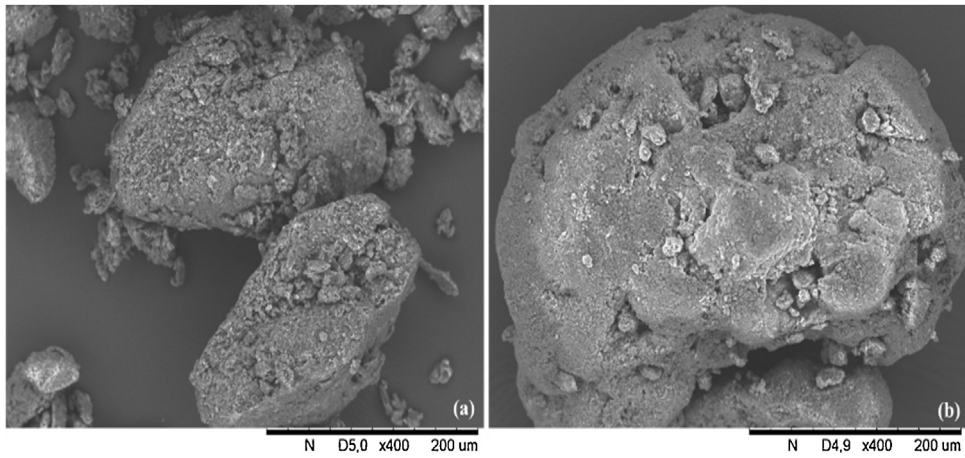


Fig. 6. Pre-extraction SEM images of *N. maritima* (a) AFD biomass; (b) RD biomass.

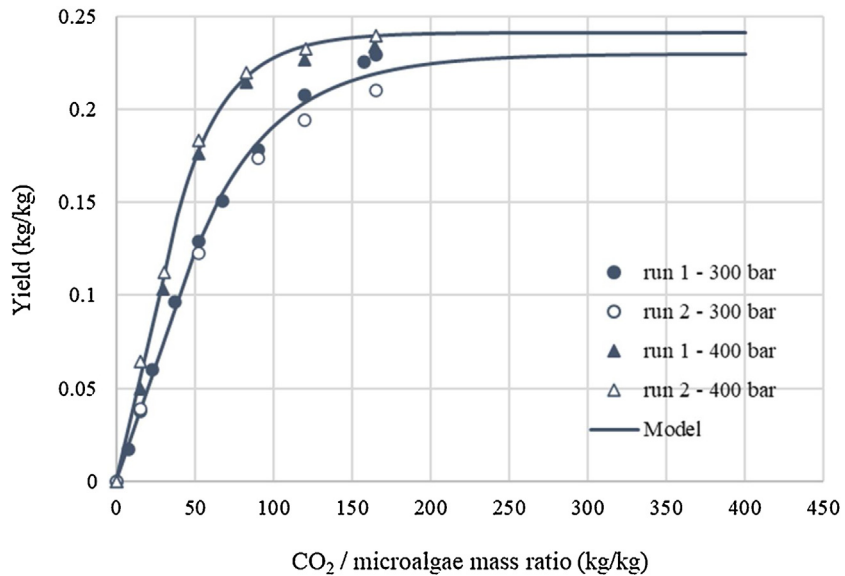
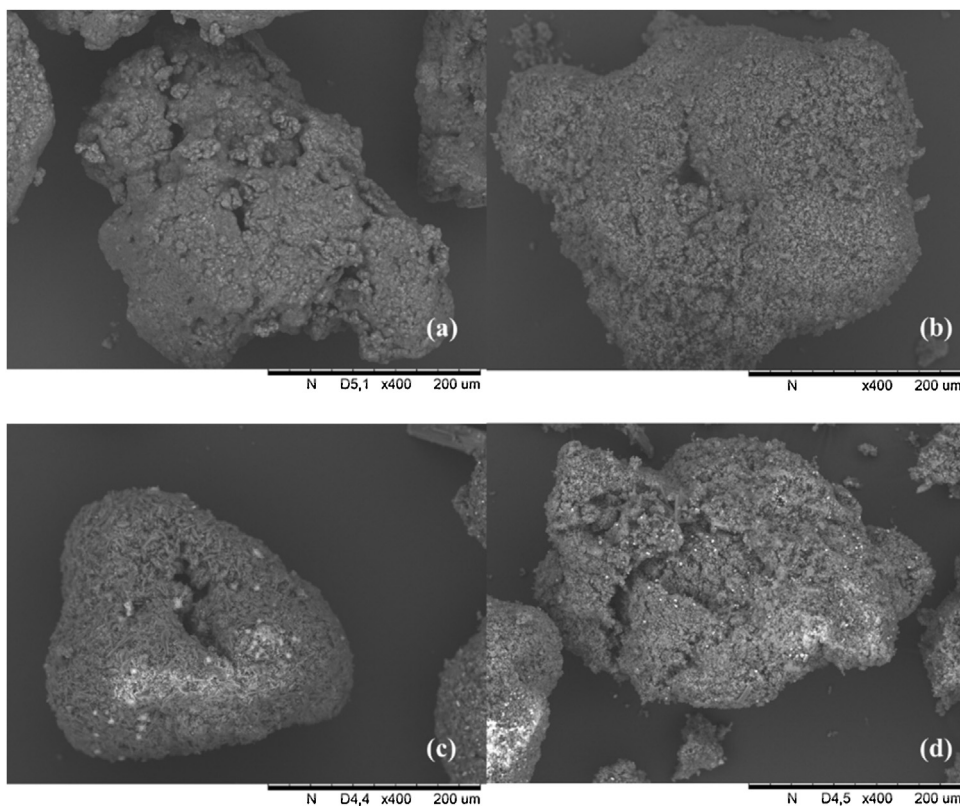
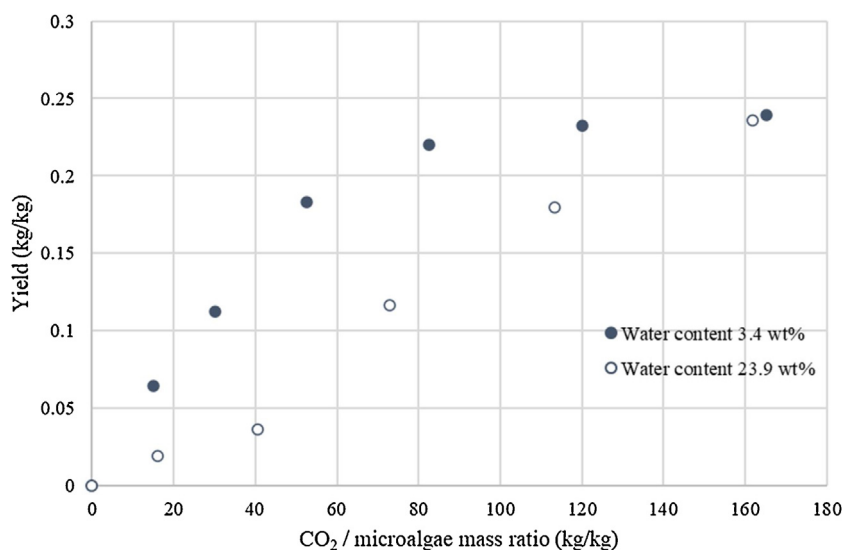


Fig. 7. Repeatability on RD *N. salina* at 333 K, 300 bar and 400 bar.





**Fig. 8.** SEM images of RD *N.salina* (a) before SC-CO<sub>2</sub> extraction, (b) after pure SC-CO<sub>2</sub> extraction, (c) after co-solvent extraction, (d) after maceration in ethanol.



**Fig. 9.** Effects of water content at 400 bar and 333 K on RD *N. salina*.

carotenoids was 42% lower (Table 5) indicating that water may act as co-solvent for Chl a. Depending on the drying method, the cell structure of the dried biomass is different. RD likely causes the difference in the interaction between water and *N. salina* biomass structure versus the results reported by Crampon et al on *N. oculata* [11]. When water was added to the RD biomass, a water layer may have been formed acting as a barrier to the diffusion of SC-CO<sub>2</sub>. During the extraction process, this additional water was evacuated with CO<sub>2</sub> flow – leaving no water in either the extracts or the residues. This was demonstrated by characterizing the water content in the algal residue after extraction. As water has very low

solubility in SC-CO<sub>2</sub> [36], it can be assumed that at a set extraction period, the amount of water on the biomass was low enough to be solubilized by SC-CO<sub>2</sub> and to act as co-solvent for the recovery of Chl a.

#### Modelling discussions

The calculated model parameters for *N. maritima* and *N. salina* were reported in Table 6. As discussed in section 4.1.3, the drying mode impacts the microalgae cell walls in different ways inducing a different composition and amount of accessible free solutes

**Table 6**  
Sovová's mathematical model parameters for *N. maritima* and *N. salina*.

Strain	Operating conditions	$y_s \times 10^3$	$C_u$	$C_1$	$C_2$	$k_{s,a_s} \times 10^5$	$r$	$q_1$	AARD
<i>N. maritima</i>	300 bar - 333 K (AFD)	5.920	0.147	0.886	0.0149	7.695	0.148	5.265	2.80
	100 bar - 313 K (AFD)	1.093	0.095	0.951	0.0069	3.651	0.075	8.287	2.31
	100 bar - 313 K (RD)	0.400	0.023	1.248	0.0316	12.921	0.047	17.057	9.50
<i>N. salina</i>	300 bar - 333 K (RD)	2.461	0.230	1.269	0.015	8.833	0.251	68.134	1.33
	400 bar - 333 K (RD)	3.600	0.241	1.455	0.025	17.072	0.107	39.498	1.62

to SC-CO<sub>2</sub>. The amount and the mixture composition are then different when RD and AFD are applied. At 100 bar and 313 K, the comparison between the two drying modes showed that the extracted mixture had a higher solubility in SC-CO<sub>2</sub> when AFD was applied.

When RD was applied the representative part of the solubility mixture on the extraction curve is longer according to the estimated values of  $q_1$  which delay the appearance of the second part of the extraction curve (diffusion part). Since the solid particles of dried and grinded samples have similar size and shapes, the values of  $k_{s,a_s}$  parameter can be assimilated to the values of the mass transfer coefficient between the broken cell and the solvent  $k_s$ .

The extractions performed on *N. maritima* showed that the mass transfer coefficient is higher when RD was applied due to a longer period for solubility mixture in SC-CO<sub>2</sub> (high value of  $q_1$ ). Considering that solutes compositions are different according to drying mode, it is difficult to give an accurate conclusion. Nevertheless, a general tendency can be identified. Compared to AFD, when RD was applied it can be assumed that the part of its easily accessible solute to SC-CO<sub>2</sub> is high. When this solute mixture is extracted, the transfer between the broken cells and the solvent is faster (higher value of  $k_{s,a_s}$ ). Hence, the choice of drying mode may depend on the targeted compounds in the solute composition. The focus on the extraction kinetics can lead to unwanted compounds and/or to a degradation of the extracted compounds.

The calculated parameters (Table 6) are of the same order of magnitude than those reported by Mouahid et al. on *N. oculata* [17] and *Dunaliella salina* [10]. In the first study, the authors assumes that the extracted solute was the same for both AFD and FD while in the second study that FD induce high transfer limitation due to diffusion (extraction of type D) due to the shape of the curves. The reported experimental tendencies were the same than the one observed in this work: a higher extraction kinetics when AFD was performed with a shorter solubility period. The total amount of accessible solutes is lower when RD or FD is performed, nevertheless, transfer parameters (according to Sovová's mathematical model) are good and no degradation of the extracted compounds was observed. It is worth noting that the additional information reported in the work give a new perspective:

- the choice of extraction type according to the shape of the extraction curve as first approximation is a correct approach, nevertheless in some specific cases it can lead to mistakes. It is then advised to perform additional analysis like solubility estimation and/or extracted solute analysis.
- The drying method should be chosen according to targeted compounds and not according to the highest extraction kinetics. RD and FD seem to be more appropriate to avoid.

## Conclusion

Supercritical CO<sub>2</sub> extractions were performed on *N. maritima* and *N. salina* provided by Cellana LLC. The effects of drying pre-treatment and co-solvent on extraction efficiency were presented. No degradation of TAG to FFA was observe when RD was applied contrary to AFD where lipid degradation was observed. When pure

SC-CO<sub>2</sub> extraction was applied at 300 bar and 323 K it was possible to extract high amount of Chl a and total carotenoids. The amount of C16:1 and EPA remain constant with varying extraction conditions. Maceration in ethanol enhanced the extraction of total carotenoids by disrupting the surface of the algae particles before extraction. It is worth noting that the water-content of 23.9 wt% in the biomass leads to strong diffusion but had a co-solvent effect when extracting Chl a, but ultimately resulted in a lower recovery of total carotenoids. Depending on the drying process, the SC-CO<sub>2</sub> extracted solute mixture has a different composition leading to different extraction kinetics. This may be due to the drying time duration and intensity which can lead to a degradation of compounds and/or to non-significant alteration of the biomass cell walls. The first part of the extraction curve is related to solute solubility whatever the drying mode. Hence, the extraction type was of type A according to Sovová's mathematical model for all extraction curves.

## Author Contribution

The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript. These authors contributed equally.

## Declaration of Competing Interest

We declare that we do not have any conflicts of interest to declare.

## Acknowledgment

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