

Supporting Information

From Langmuir Blodgett to Grafted Films

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10 pages, 13 figures, 2 tables.

Image of the modified Langmuir trough

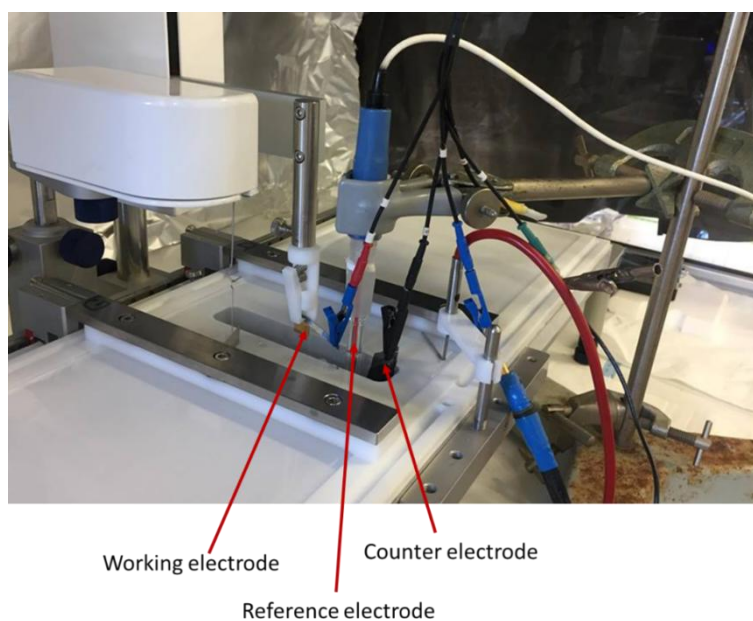


Fig. SI-1: image of the modified Langmuir trough for LB-eG process.

Electrografting of ODA on gold

Electrochemical oxidation of ODA on gold.

We have measured a voltammogram of ODA in different solvents. In ACN (Fig. SI-2) the adsorption of ODA blocks the electrode and reduces the background current.

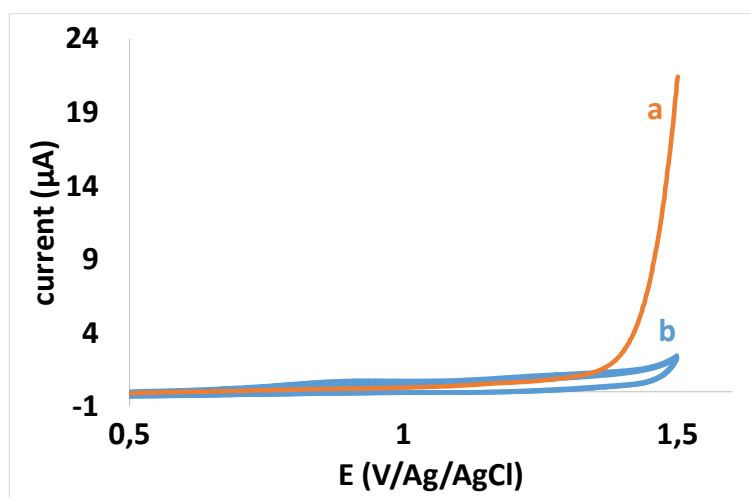


Fig. SI-2: Cyclic voltammogram on an *Au* electrode a) in ACN + 0.1 M NBu_4BF_4 and b) after addition of a nominal concentration $c = 1$ mM of ODA, scan rate $v = 0.1 \text{ Vs}^{-1}$.

ODA is more soluble in DMF; in DMF + 0.1M NBu_4BF_4 , the oxidation peak of Au is observed at $E_p = 1.0 \text{ V}/(\text{Ag}/\text{AgCl})$; and the oxidation wave of ODA is observed as a broad peak at $E_p(\text{ODA}) = 1.36 \text{ V}/\text{SCE}$, the same potential as butylamine (Fig. SI-3).

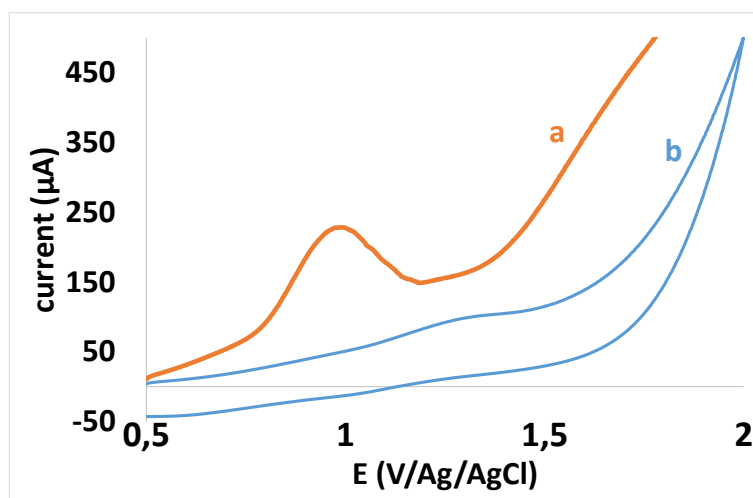


Fig. SI-3: Cyclic voltammogram on an Au electrode a) in DMF + 0.1 M NBu_4BF_4 and b) after addition of ODA ($c = 1 \text{ mM}$). Scan rate $\nu = 0.1 \text{ Vs}^{-1}$

We also recorded the background in the same pH 9 solution used for the formation of the Langmuir film. The onset of the background is about + 0.8 V/SCE (Fig. SI-4).

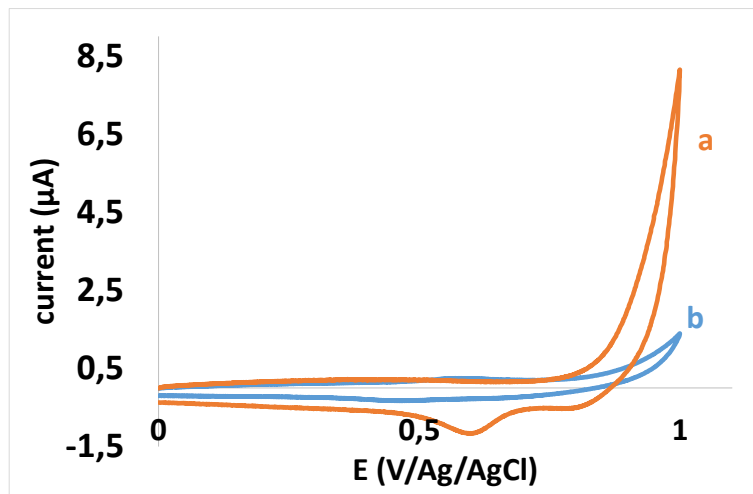


Fig. SI-4: Cyclic voltammogram on an Au electrode a) in a pH 9 solution and b) after addition of ODA ($c = 1 \text{ mM}$). Scan rate $\nu = 0.1 \text{ Vs}^{-1}$.

Water contact angle on $\text{Au-ODA}_{\text{LB-G}}$.

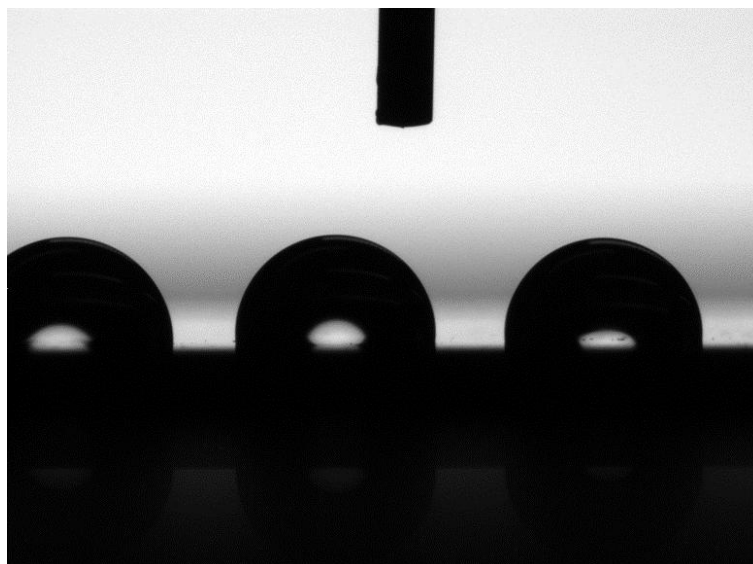


Fig. SI-5: The water contact angle of $Au\text{-ODA}_{\text{LB-G}}$.

The water contact angle of $Au\text{-ODA}_{\text{LB-G}}$ is $101.3^\circ \pm 0.5^\circ$.

Infrared Spectrum of $Au\text{-ODA}_{\text{LB}}$.

The stability of $Au\text{-ODA}_{\text{LB}}$ film was tested by ultrasonication for 3 min. As shown by the IR spectra in Fig. SI-6, the ODA_{LB} film is mainly desorbed under such conditions. The monolayer dewetting was also confirmed by AFM (data not shown).

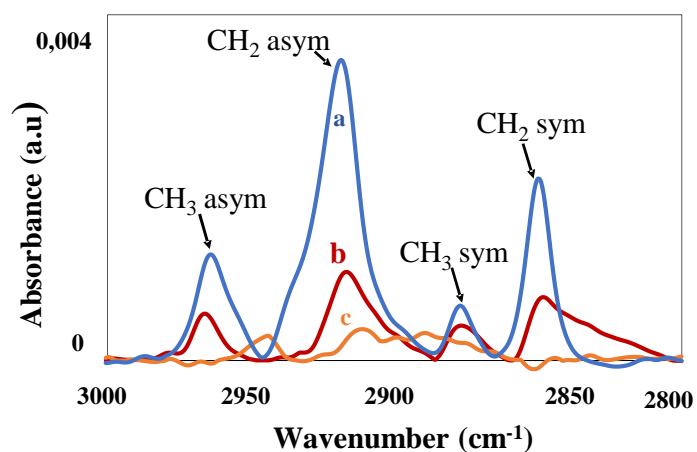
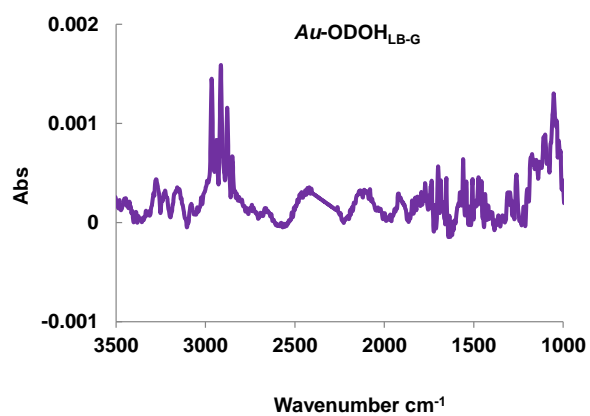
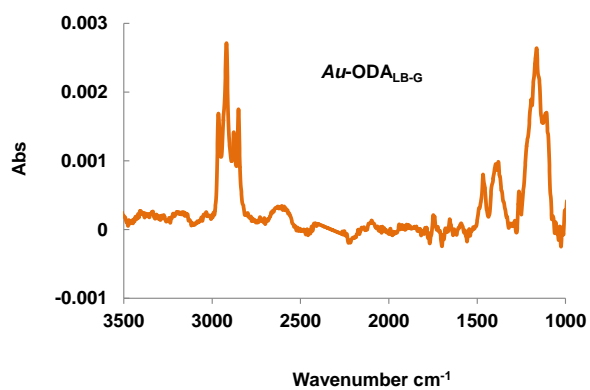
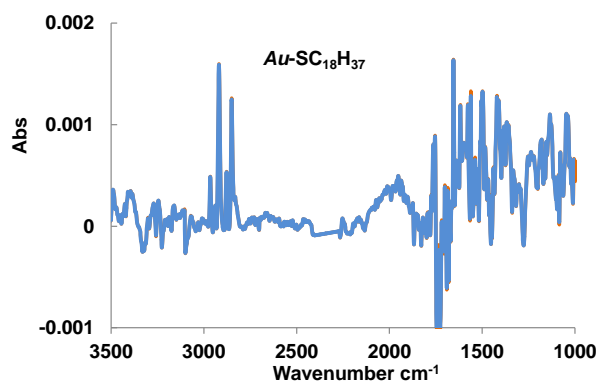
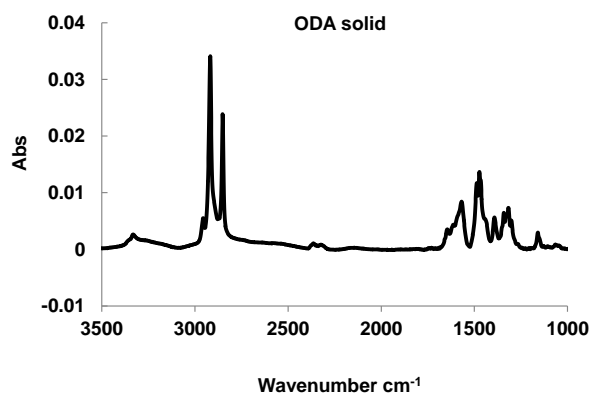


Fig. SI-6: IRRAS spectra of a) $Au\text{-SC}_{18}\text{H}_{37}$, b) $Au\text{-ODA}_{\text{LB}}$, c) $Au\text{-ODA}_{\text{LB}}$ after ultrasonication.

Infrared Spectra



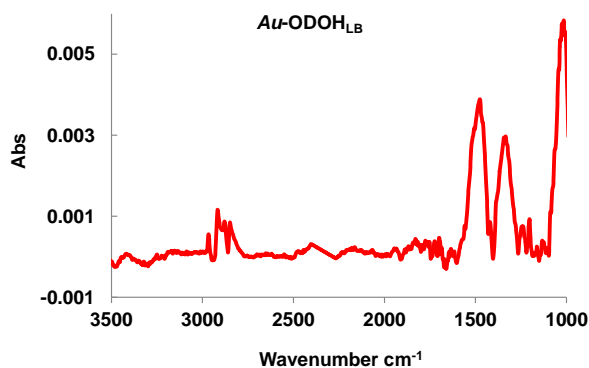


Figure SI-7: Infrared spectra of ODA solid, $Au-SC_{18}H_{37}$, $Au-ODA_{LB-G}$, $Au-ODOH_{LB-G}$ and ODA_{LB} .

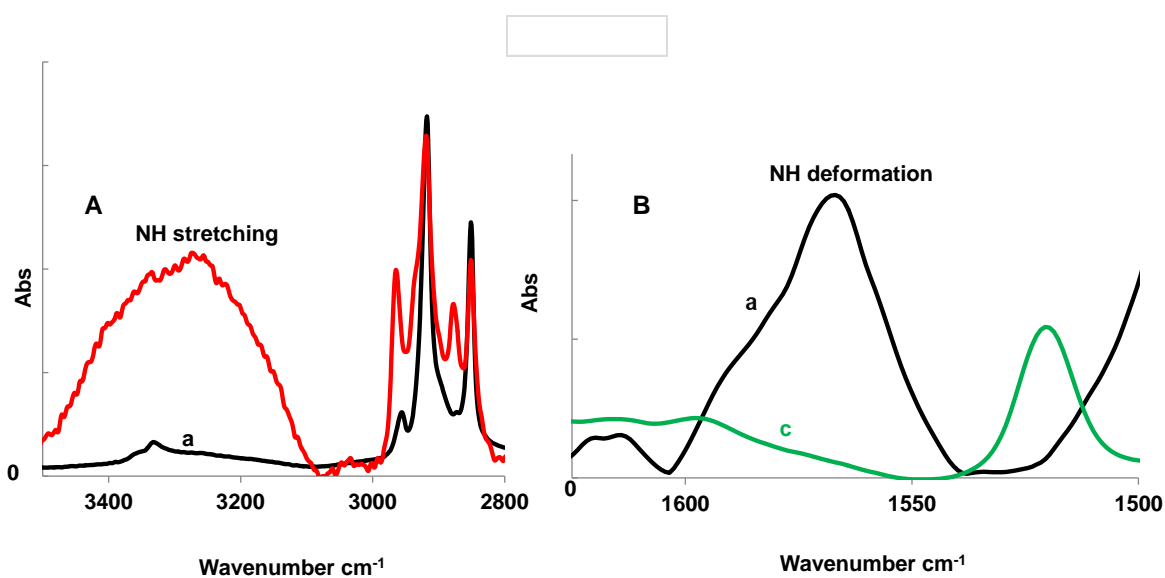


Figure SI-8: A) NH stretching and B) deformations vibrations of a) solid ODA, b) $Au-ODA_{LB-G}$, c) $Au-ODA_{LB}$. Normalized spectra, arbitrary absorbance units.

XPS of $Au-ODA_{LB-G}$.

Table SI-1 and the spectra (Fig. SI-8) indicate the presence of Au4f in agreement with a thin film, of the nitrogen of ODA with C1s/N1s ~ 18 as expected from the chemical formula and of a contamination by O1s.

Table SI-1. XPS of $Au-ODA_{LB-G}$.

Atom	Binding Energy (eV)	Atomic proportions %
Au4f	83.9	9.5
C1s	285.2	70.8
N1s	400.4	4.0

O1s	532.6	15.7
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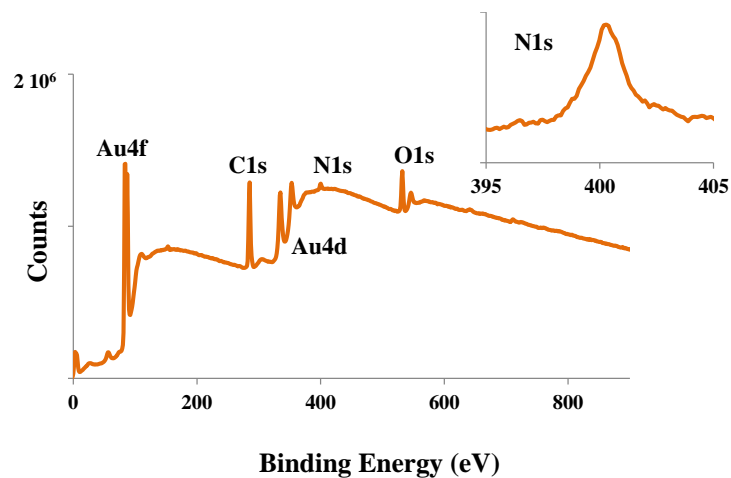


Fig. SI-9: XPS spectrum of $Au-ODA_{LB-G}$.

Electrografting of ODOH on gold

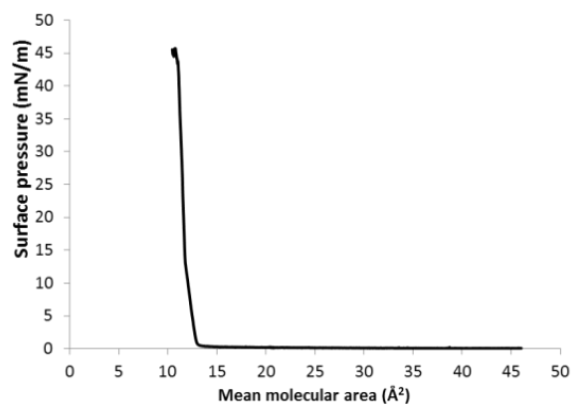


Fig. SI-10: (a) Compression isotherm of ODOH on pH 2 on aqueous solution at 21°C prior to grafting on Au.

Water Contact Angle and thickness of Au-ODOH_G.

The water contact angle $95.7^\circ \pm 1.4^\circ$ and the thickness of the film $1.51 \text{ nm} \pm 0.4 \text{ nm}$ are similar to that obtained with ODA.

XPS of Au-ODOH_{LB-G}.

Table SI-2 and the spectrum (Fig. SI-10) indicate the presence of Au4f in agreement with a thin film. Presence of the oxygen of ODOH_G was detected on the surface but also some oxygen and nitrogen contaminations.

Table SI-2. XPS of Au-ODOH_{LB-G}.

Atom	Binding Energy (eV)	Atomic proportions %
Au4f	83.9	33.8
C1s	285.2	51.6
N1s	400.4	1.0
O1s	532.6	13.6

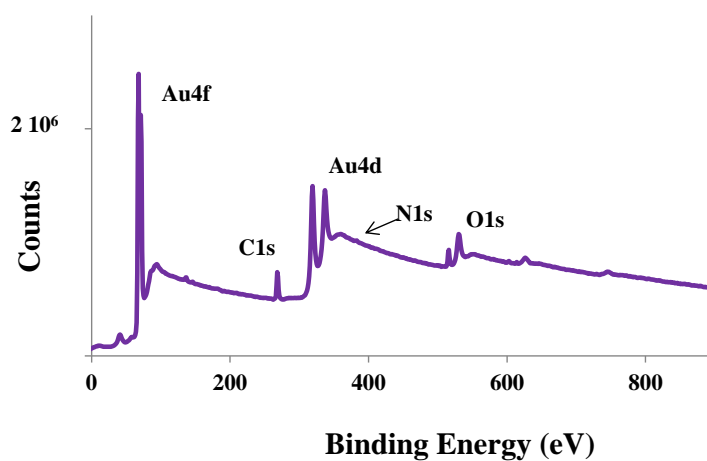


Fig. SI-11: XPS spectrum of Au-ODOH_{LB-G}.

AFM images of Au-ODOH_{LB-G}.

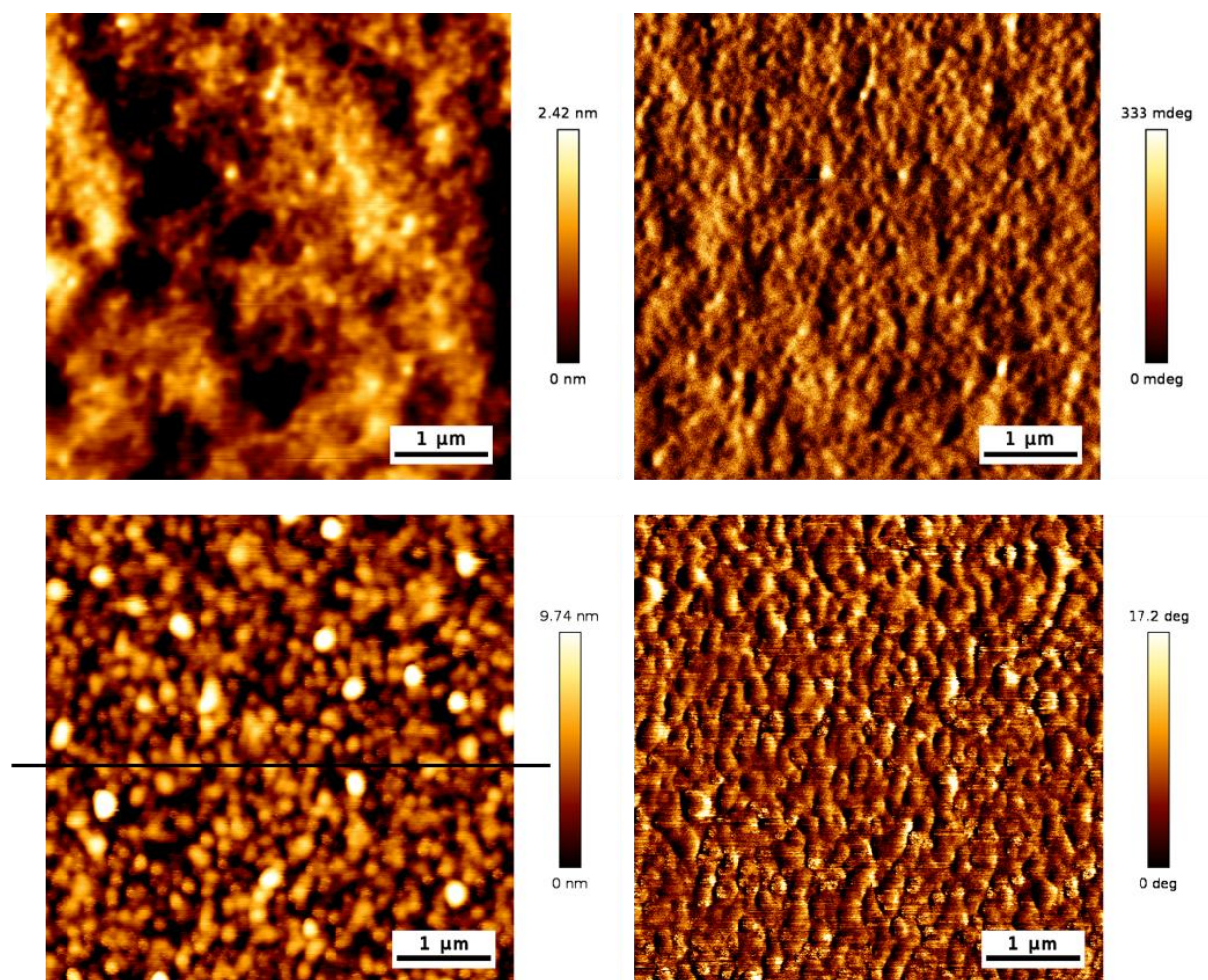


Fig. SI-12: AFM images of *Au-ODOH_{LB-G}*.

Fig. SI-11 presents AFM height (left) and phase (right) images of (upper images) bare *Au* surface before LB-eG process of ODOH and of (lower images) *Au-ODOH_{LB-G}* after ODOH LB-eG process on *Au* surface. The surface appears very rough after transfer: numerous small round shaped aggregates with 3 to 5 nm heights cover the gold surface. Among them, some are higher, up to 10 nm height.

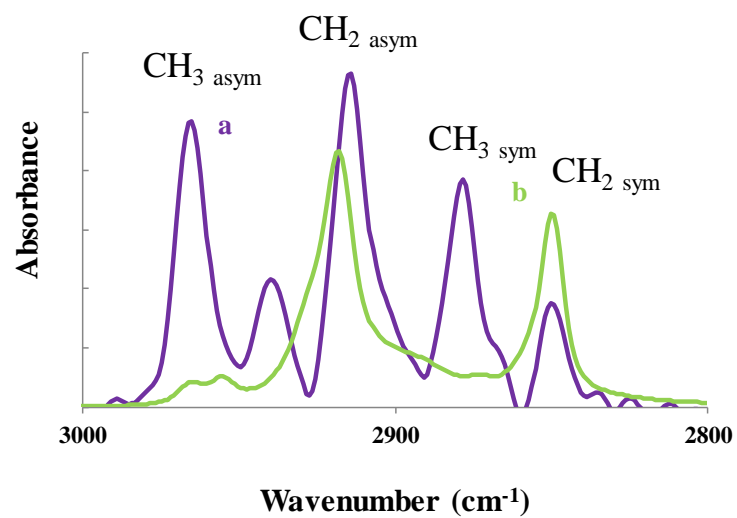


Fig. SI-13: IRRAS spectra of a) $Au\text{-ODOH}_{\text{LB-G}}$, b) $Au\text{-ODOH}$ solid. Normalized spectra, arbitrary absorbance units.