Supporting Information

A methodology for emulsion design based on emulsion science and expert knowledge. Part 2: An Application in the Cosmetics Sector

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Selection of ingredients – case study

Ingredient type		Ingredient	Concentrat	Reason	Warnings/re
			ion		commendati
	1				ons
Oily phase 20%	Emollients/ oils	Caprylic/Capric Triglyceride Isopropyl myristate	6 to 10%	According to ingredient database presented in the first article, these oils give a light sensation and a low oily after-feeling. Other oils have similar properties (within the database from the previous articles it is possible to find more). The availability criterion was also considered for a final decision. They are added in high proportions to ensure the desired final sensation	Proportions can be change during experimentati on to have a low oily sensation in the final product
	Oil - natural	Shea butter almond oil	Low concentratio n <2%	Natural oils are used in low concentrations to avoid rapid product oxidation (as preventing in warnings from Table 5). Almond oil is selected because it has a low oily sensation as expected from product specifications. Shea butter is selected due to its high moisturizing effect (also a requirement) and the product claim of naturalness	
	Fragrance	Lemongrass oil	<0.1%	Fragrance was consulted with a group of customers. Calendula oil is also part of the product fragrance	Attention should be paid to the low resistance of fragrances to high temperature
Surfactants	Surfactant system	System 1: Main surfactant- Polysorbate 80	2.5-3.5%	As presenting in a warning from Table 5, a surfactant system with a relatively high HLB is required to	Except for sucrose palmitate, all surfactants

Table S1. Explanation - Ingredients for a facial moisturizing cream containing calendula oil - case study

		+Sorbitan		produce stable O/W	are soluble in
		stearate		emulsion. Considering this	the oily phase
		+Glyceryl		and the ingredient database	• •
		monostearate		presented in the first article	
				on the methodology, three	
		System 2:		surfactant systems are	
		Main surfactant-		selected:	
		Sucrose		- Two surfactant systems	
		palmitate		based on polyethoxylated	
		+Cetearvl		(POE) compounds which	
		glucoside		have a high HLB (over 10)	
		+Glycervl		and are relatively	
		monostearate		inexpensive. Because of	
				availability, a system with	
		System 3:		polysorbate 80 as main	
		Main surfactant-		surfactant and another	
		POE 100		system with POE 100	
		stearate		stearate as main surfactant	
		+Sorbitan		are selected. Similar	
		stearate		systems have been used in	
		+Glycervl		similar conditions for	
		monostearate		cosmetic creams with	
				positive results (Iwata and	
				Shimada, 2013).	
				- A third surfactant system	
				based on sucrose palmitate	
				which is biobased and has a	
				high HLB is selected. This	
				system must be prepared at	
				neutral pH and once the	
				emulsion is formed, pH can	
				be decreased 1 .	
_	Rheological	Cetearyl alcohol	2-5% 3%	These two oily compounds	Soluble in the
	modifier/	and cetyl	2070,070	can be used to increase	oily phase
	Thickener	nalmitate		product viscosity up to the	ony phase
	Thiekener	panniac		expected viscosity	
				according to the design	
				problem The percentage of	
				use is to be tuned	
				experimentally to achieved	
				the desired rheology and	
				texture 2	
Δαμερικ	Rheological	Xanthan gum	0.1_0.2%	Xanthan rum is selected	Soluble in
nhase	modifier		0.1-0.270	from our database because	water
Phase	(second)			it has a neudonlastic	water,
	(second)			habayior (as avposted from	
		l		benavior (as expected from	l

	Acid	citric or lactic	Until pH of	Figure 4), it works well at acid pH, and it is naturally derived. It is used at low concentrations 0.1-0.2% because of its high efficiency ³	Dispersible in glycols as glycerin
	A ciu	acid	(4.5-5.7)	in cosmetic	emulsificatio n to adjust the final pH value of the product
	Glycol	Glycerin	3%	It is a highly availability and low-cost humectant. It is added to perform the function of hydration. A percentage of 3 was defined based on similar formulations ³	Water soluble
	Preservative	Benzyl alcohol and dehydroacetic acid	0.6%	According to our cosmetic preservatives database, for an acid pH it is possible to use preservatives based on phenoxyethanol, benzyl alcohol, benzoic acid, potassium benzoate and parabens. Parabens must be avoided, because they have a negative public image and are contrary to the specification of naturalness. Considering these arguments, and because of its availability, a preservative system based on benzyl alcohol and dehydroacetic acid is used. Concentration is selected by following supplier recommendations.	Soluble in polar solvents such as glycerin, Decrease pH of the emulsion
	Antioxidant	Calendula oil Acetate tocopherol		Calendula essential oil is selected because of design specifications. Acetate tocopherol is also added to boom antioxidant action (to achieve antioxidant expected activity)	Low resistance to high temperature

References

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