

# Efficient degradation of the organic UV filter benzophenone-3 by Sphingomonas wittichii strain BP14P isolated from WWTP sludge

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1	Efficient degradation of the organic UV filter benzophenone-3 by Sphingomonas
2	wittichii strain BP14P isolated from WWTP sludge
3	
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14	
15	
16	Abstract
17	Benzophenone-3 (BP3) is a widely used organic UV filter present in many
18	environmental compartments. One way BP3 is released into the environment is
19	through effluents from wastewater treatment plants (WWTPs). These plants are
20	possible sources for degradation activity as WWTP sludge can potentially degrade
21	BP3. Our goal was to identify the BP3 degrading microorganism(s) in WWTP sludge
22	and to investigate whether the degradation was co-metabolic. Initial WWTP sludge
23	microcosms spiked with BP3 showed 100% degradation after 20 days. Multiple

24	transfers of these microcosms, while maintaining a strong selective pressure for
25	BP3 degradation capabilities, resulted in the dominance of one bacterial strain. This
26	strain, Sphingomonas wittichii BP14P was subsequently isolated and shown to
27	degrade BP3 in a growth dependent manner. Strain BP14P utilized BP3 as the sole
28	energy and carbon source and completely degraded BP3 after 7 days in minimal
29	media. We tested the capability of BP14P to degrade nine other UV filters, but the
30	degradation ability seemed to be restricted to BP3. However, whether this
31	specificity is due to the lack of degradation genes, cellular transport or low
32	bioavailability of the other UV filters remained unclear. The efficient degradation of
33	BP3 by a group of bacteria well known for their potential for xenobiotic degradation
34	is an important step forward for a complete risk assessment of the long-term
35	environmental impact of BP3.
36	
37	Keywords: Sunscreens, BP3 degradation, growth-dependent, Sphingomonas
38	wittichii
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	wittichii 1. Introduction
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39 40 41 42	<ul> <li><b>1. Introduction</b></li> <li>Benzophenone-3 (BP3) is an organic UV filter commonly used in sunscreen</li> <li>formulations and other cosmetic products. Due to its widespread use, BP3 is now</li> </ul>
<ol> <li>39</li> <li>40</li> <li>41</li> <li>42</li> <li>43</li> </ol>	1. Introduction Benzophenone-3 (BP3) is an organic UV filter commonly used in sunscreen formulations and other cosmetic products. Due to its widespread use, BP3 is now found in many environmental compartments such as freshwater and seawater,

47	has been demonstrated that BP3 can be absorbed through the skin but the effects of
48	elevated systemic BP3 concentrations in humans are still unclear (Suh et al., 2020).
49	BP3 has been shown to be toxic in several species used for toxicological testing
50	including fish, corals and algae (see recent review and references therein (Lozano et
51	al., 2020)). Thus, the toxicity of BP3 combined with the possibility of
52	bioaccumulation indicates that BP3 might pose an ecological risk (Díaz-Cruz et al.,
53	2019).
54	
55	Wastewater treatment plants (WWTPs) are a major source of BP3 to the
56	environment, in addition to recreational water activities and industrial wastewater
57	discharges (Mao et al., 2018). Indeed, BP3 has been found in influents and effluents
58	of WWTPs (Ramos et al., 2016; Mao et al., 2018) in relatively high concentrations.
59	Often, the concentrations of BP3 are lower in the effluents of WWTPs than the
60	influents. This difference can be attributed to sorption to activated sludge and/or
61	biodegradation.
62	
63	Biodegradation of BP3 has been shown to occur in different systems. Liu and
64	collegues observed degradation of BP3 in activated sludge microcosms (Liu et al.,

65 2012) and in microcosms using aquifer materials as an innoculum (Liu et al., 2013).

66 The data from this latter study was used as input in a model to better understand

67 the fate of UV filters, including BP3, in aquifers (Rodríguez-Escales and Sanchez-Vila,

68 2020). Among others, this model took into account sorption processses, both to

69 sediments and biomass, but also assumed that the UV-filters were degraded in a co-

70 metabolic manner with the oxidation of sediment organic matter as the main71 process.

72

73 The microorganisms responsible for the degradation in the studies described above 74 were not identified (Liu et al., 2012; Liu et al., 2013). So far the only bacterial strain 75 shown to be capable of BP3 degradation is *Methyliphilus* sp. strain FP-6, which 76 degraded BP3 co-metabolically when grown with several other carbon substrates 77 (Jin et al., 2019). The white root fungus *Trametes versicolor* can also degrade BP3 78 (Badia-Fabregat et al., 2012; Gago-Ferrero et al., 2012) but it is not known if this 79 degradation was a co-metabolic process since dried and sterilized sludge was used 80 as solid support in these experiments. 81 82 The objective of our study was to investigate the potential degradation of BP3 in 83 microcosms with WWTP sludge, to determine whether the BP3 degradation is 84 indeed co-metabolic and whether any bacterial species exist that can completely 85 mineralize BP3. 86 87 88 2. Materials and methods. 89 90 2.1 Chemicals 91 The UV filters used for biodegradation studies, 2-hydroxy-4-methoxybenzophenone 92 (BP3) and methylene bis-benzotriazolyl tetramethylbutylphenol (MBBT) were

93	purchased from Sigma-Aldrich (Lyon, France), while 2-ethylhexyl salicylate (ES),
94	homosalate (HS), butyl methoxydibenzoylmethane (BM), octocrylene (OC),
95	diethylamino hydroxybenzoyl hexyl benzoate (DHHB), diethylhexyl butamido
96	triazone (DBT), and ethylhexyl triazone (ET) were kindly provided to us by Pierre
97	Fabre Pharmaceuticals. Analytical standards for the above-mentioned UV filters
98	were purchased from Sigma-Aldrich (Saint-Quentin-Fallavier, France). Analytical-
99	grade dichloromethane (DCM), 1-methyl-2-pyrrolidinone (NMP), methanol, formic
100	acid (98 %) and molecular grade formaldehyde were obtained from Sigma-Aldrich.
101	Pure water was obtained from an Elga Purelab Flex System (Veolia LabWater STI,
102	Antony, France). Glassware was cleaned with DCM and calcinated at 450 $^{\circ}$ C for 2
103	hours to remove all traces of organic matter.
104	

## 105 2.2 Sampling of sludge, initial culture and enrichment culture setup

106 The sludge was sampled from the Argelès-sur-mer (France) wastewater treatment

107 plant (WWTP) on August 24th, 2018, from the bottom of a clarifying tank. The

108 sludge was collected in glass mason jars and stored at 4 °C until utilized for

109 degradation experiments. The sludge contained 2765 (+/- 310) mg/L total

suspended solids (TSS) and the initial BP3 concentration was below the detection

111 limit for our analysis (4.6 ng/g, see Fagervold et al. (2019)). For the initial

112 degradation experiments, 100 mL sludge was mixed with 500 mL freshwater

113 mineral media (OECD301) in a 1-L glass bottle (live). BP3 was added in methanol (2

114 mg/mL) to a final concentration of 10 µg/ml. A sterile control (SC) was prepared by

autoclaving twice (48 hours apart) a 1-L bottle containing sludge and media at 120

°C for 20 minutes, followed by the addition of sodium azide (1 %) and then BP3. The two bottles (one live and one SC) were incubated at 25 °C in the dark and were systematically shaken and opened inside a fume hood twice per week to allow aeration.

- 120
- 121 An enrichment culture was initiated with inoculum from the initial flask that

122 showed degradation. The final theoretical BP3 concentration was about 10  $\mu$ g/mL

123 for Enrichment 1 (the first transfer), but this concentration was increased to a

124 theoretical concentration of  $100 \,\mu\text{g/mL}$  for Enrichment 2 and subsequent

enrichments. To obtain a concentration of  $100 \ \mu g/mL$ , 2 mL of a solution of BP3 in

126 acetone (2.5 mg/mL) was added to 100-mL Erlenmeyer flasks containing 2 g of inert

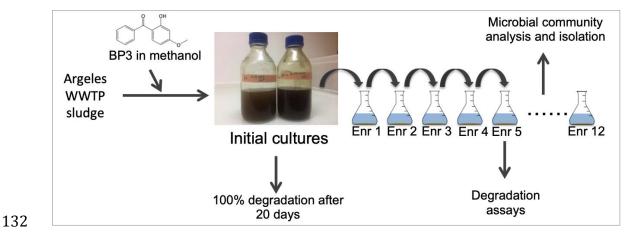
sand. The acetone was left to evaporate overnight. Mineral media (50 mL, as above)

128 was added and the flasks were autoclaved. Each month, the cultures were

129 transferred (1% vol/vol) into new flasks with sand and BP3 (see Figure 1 for

130 diagram of the experimental setup). The cultures were incubated at 25 °C in the

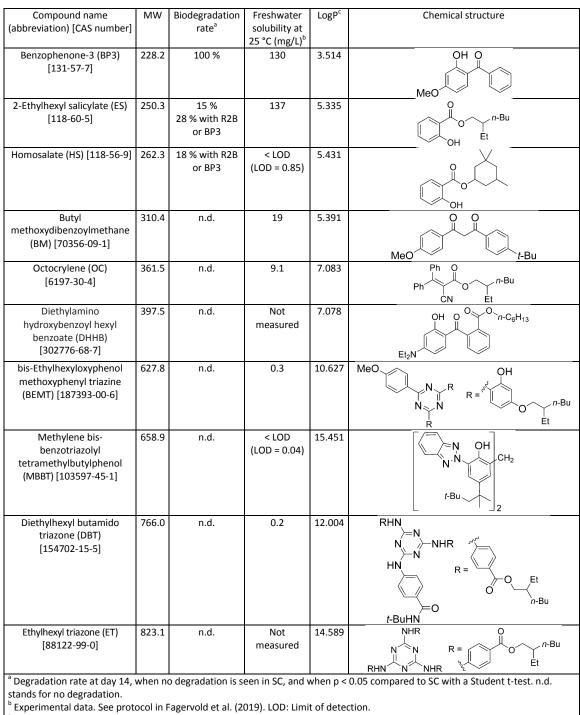
131 dark on a platform shaker at 100 rpm.



133 **Fig. 1.** Schematic of the experimental design.

# 134 **Table 1**

- 135 Selected chemical properties and structure of the UV filters used in this study
- 136 (ordered by increasing molecular weight). Note the abbreviations that are used
- 137 throughout.
- 138



<sup>c</sup> According to Reaxys database (https://www.reaxys.com/, accessed on June 9, 2020)

## 141 *2.3 Biodegradation assays*

142 Biodegradation assays were conducted using 15-mL glass tubes with Teflon lined 143 caps (Pyrex, Analytic lab, France). The tubes were set up as for the enrichment 144 cultures, but with 0.2 g sand, 3 mL media, and care was taken so that for each batch 145 of degradation assays, the exact same amount of BP3 (around 100 µg/mL) was 146 added to the tubes. Three tubes were sacrificed for each time-point. The tubes were 147 incubated at 25 °C after inoculation. For the enrichment cultures, the inoculum for 148 the assay was 150 µL of supernatant after rigorous mixing of the enrichment 149 culture. Inoculum preparation for the biodegradation assay of strain BP14P was 150 done as follows: The strain was grown in 100 % R2B broth (Thermo Fisher 151 Scientific, Illkirch-Graffenstaden, France) for about 40 h until the culture had 152 reached an optical density at 600nm of approximately 1.0. The cell suspension was 153 centrifuged at 4200 g for 10 min and subsequently washed 2 times with minimal 154 media before a final resuspension in minimal media (4% v/v). The cultures were 155 incubated at 25 °C in the dark on a rotary shaker at 100 rpm. 156 157 2.4 BP3 extraction 158 Two different protocols were used for the extraction of BP3, one for the initial

159 biodegradation experiments and another for the biodegradation assays for

160 enrichment cultures and isolated strains.

161

162 For the initial cultures containing WWTP sludge, 5 mL slurry was taken in duplicate

163 from each flask for each timepoint, both live and SC. The slurry was centrifuged (950

164 x g for 20 min) to separate the aqueous and solid phases. The aqueous and solid 165 phases were then extracted and analyzed separately. The supernatant (aqueous 166 phase) was transferred to a new tube and subjected to liquid/liquid extraction with 167 acidified (0.1 % formic acid) DCM. This was achieved by adding 5 mL acidified DCM 168 to the aqueous sample, mixing by vortexing and shaking overnight at 25 °C in the 169 dark. The DCM fraction was decanted off and dehydrated with Na<sub>2</sub>SO<sub>4</sub>. The DCM was 170 then evaporated using the HT-4X system (Genevac, Biopharma Technologies, 171 Diemoz, France) and the residue was dissolved in 1 mL of NMP/water (8:2, v/v) 172 before analysis.

173

174 The solid phase was extracted as described by Fagervold and coworkers (Fagervold

et al., 2019) with modifications. Briefly, the pellet was frozen at –80 °C, lyophilized

and extracted twice with 2 ml acidified (0.1% formic acid) DCM/methanol mix (1:1,

177 vol/vol) and sonicated for 10 min. The supernatants were combined, evaporated

178 and re-dissolved in NMP/water (8:2, v/v) before HPLC analysis.

179

For the biodegradation assays, the whole tube containing sand and liquid media was
sacrificed for extraction at each timepoint. DCM (3 mL) containing an exact known
amount of MBBT (internal standard) was added to each tube. The tubes were mixed
by vortexing twice for 10 seconds, 1 h apart and shaken overnight at 100 rpm at 25
°C in the dark. After a brief sonication, 1 mL of the DCM phase was transferred to a
HPLC vial and diluted 20X in DCM before HPLC analysis.

186

187 2.5 HPLC analysis

188 The UV filters were analyzed using an Ultimate 3000<sup>™</sup> HPLC system, equipped with 189 a DAD detector (Thermo Fisher Scientific) and a Phenomenex Kinetex Biphenyl 2.6 190  $\mu$ m, 150 × 4.6 mm column as previously described (Fagervold et al., 2019) with 191 some minor modifications. The injection volume was 5  $\mu$ l and the samples were 192 diluted in 100% DCM. We expanded the analysis described in Fagervold et al. 193 (2019) to include all targeted UV-filters. The calibration curves and retention times 194 of all the different UV filters (Table 1) were determined as described by Fagervold 195 and coworkers (Fagervold et al., 2019). The recovery rate was above 85 % for the 196 extraction of BP3 in the initial sludge microcosms. The recovery rate for the 197 degradation assays were variable for the different UV filters, ranging from 63 % for 198 ES to 93 % for OC. However, the extraction efficiency was repeatable for each UV 199 filter over time, which is the most important factor for the detection of degradation 200 activity.

201

202 For determination of possible BP3 degradation products, we also performed a

203 UHPLC-MS/MS analyses as described by Stien and coworkers (Stien et al., 2019).

204 The data were acquired using the Trace Finder<sup>TM</sup> 3.1 software and were processed

205 with FreeStyle<sup>™</sup> and Compound Discoverer<sup>™</sup> (all from Thermo Fisher Scientific).

206

207 2.6 Microbial community analysis of enrichment cultures

208 DNA was extracted using a Maxwell® 16 LEV Blood DNA Kit (Promega,

209 Charbonnières-les-Bains, France) with the Maxwell® 16 MDx Instrument (AS3000;

210 Promega) following the manufacturer's instructions and with an initial lysing step. 211 Briefly, 2 mL of supernatant was transferred to a microcentrifuge tube and 212 centrifuged at 14 500 x g for 10 minutes. The pellet was resuspended in 300 µL 213 MilliQ water and 300 mL of the lysing buffer from the Maxwell® 16 LEV Blood DNA 214 Kit was added to "lysing matrix B beads" (MP Biochemicals, Illkirch, France) and 215 homogenized for 20 s at 6 m/s on a FastPrep-24<sup>™</sup> 5G Instrument (MP biochemicals). After extraction, the DNA was stored at –80 °C until further analysis. 216 217 The composition and diversity of the bacterial communities in enrichment cultures 218 were determined by Automated Ribosomal Intergenic Spacer Analysis (ARISA) and 219 by Illumina sequencing (RTL genomics, Texas, USA). The ARISA was performed as 220 described by Fisher et al. (Fisher MM and Triplett EW, 1999) with minor 221 modifications. Briefly, for the PCR we utilized the intergenic spacer primers 1406F 222 and 23SRY (Fisher MM and Triplett EW, 1999) and the KAPA2G Fast HotStart 223 ReadyMix (Roche, Analytic Lab, St Mathieu de Treviers, France) with the following 224 PCR cycling parameters: 94 °C for 3 minutes, 30 cycles of 94 °C for 15 s, 55 °C for 15 225 s and 72 °C for 30 s, then a final elongation of 72 °C for 1 min. The products were 226 purified, denatured and injected into a 16 capillary Applied Biosystems Sequencer 227 3130XL (Thermo Fisher Scientific) together with the internal standard 228 MapMarker® X-Rhodamine Labeled 50-1000 bp (Bioventures Inc., Tennessee, USA) 229 for the determination of peak lengths. For the analysis, individual fragment lengths 230 (within a 0.1 bp "error range") were considered as separate Operational Taxonomic 231 Units (OTUs) and the relative abundance of each OTU was calculated by dividing the 232 area of the peak by the total area of the chromatogram after 200 bp. Furthermore,

233 we only took into account peaks contributing to more than 1 % of the relative peak 234 area. Concerning the Illumina sequencing analysis, the DNA was sent to RTL 235 genomics where paired end, 2×300 bp MiSeq Sequencing was performed using 236 standard conditions of the sequencing provider. We opted for dual indexing and 237 used the bacterial primers 341F (CCTACGGGNGGCWGCAG) and 806R 238 (GACTACHVGGGTATCTAATCC) (Herlemann et al., 2011). The sequences were 239 processed using Qiime2 (Bolyen et al., 2019). Briefly, the sequences were quality 240 filtered, stitched, denoised and clustered using dada2. Representative sequences 241 were classified using the Naive Bayes classifier and the Silva 132 database as a 242 reference. Our two samples (Enr 11 and Enr 12) yielded about 41000 and 47000 243 sequences, of which 39000 and 44000, respectively, belonged to a single OTU. There 244 were a total of 37 OTUs in the two samples, but most were present at very low 245 relative abundances.

246

247 2.7 Strain isolation and sequencing

248 The supernatant of an actively degrading enrichment culture transferred 7 times 249 (Enrichment 7, or Enr 7) was diluted 100 fold and spread on R2B agar and 250 incubated for 3 days at 25 °C in the dark. Several colony morphologies were 251 observed and each morphotype was re-streaked on R2B agar to ensure purity. 252 These strains were tested for BP3 degradation activity and one positive strain, 253 BP14P, was further characterized. DNA was extracted as described above and 254 sequenced on the same sequencer used for ARISA. Here, we used universal bacterial primers 27F and 1492R in the first PCR and then the internal primers 907R, 804F 255

and S8 for the sequencing reactions realized by the dideoxy reaction Sanger

257 sequencing using the BigDye<sup>™</sup> Terminator v3.1 Cycle Sequencing Kit (Thermo

258 Fischer) and the manufacturer's protocol.

259

260 2.8 Flow Cytometry

261 Strain BP14P cells were counted throughout the degradation assay. The glass tubes 262 were vortexed and 1 mL of cell suspension was transferred to a microcentrifuge 263 tube before adding glutaraldehyde (1% v/v final concentration). The tubes were 264 incubated at 4 °C in the dark for 15 minutes before flash freezing with liquid N<sub>2</sub>. The 265 samples were kept at -80 °C until analysis. Samples were defrosted at room 266 temperature and incubated with the nucleic acid stain SYBRGreen I (Molecular 267 Probes) for 15 min at room temperature in the dark before enumeration (Marie et 268 al., 1997) using a FACSCanto flow cytometer (BD-Biosciences) equipped with optics 269 fiber emitted light (407, 488 and 633 nm). Fluorescent 1.002 µm beads 270 (Polysciences Inc., Europe) were added to each sample as an internal standard to 271 normalize cell properties and to compare cell populations. Accurate analyzed 272 volumes and subsequent estimations of cell concentrations were calculated using 273 Becton-Dickinson TrucountTM beads. Total bacterial cells (TBA) were enumerated 274 according to the variations of light scatter properties (relative cell size) and green 275 fluorescence related to nucleic acid content. 276

277

#### 278 3. Results and discussion

280 3.1 Degradation of BP3 in initial sludge microcosms and enrichment cultures 281 To detect BP3 degradation, the concentration of BP3 in the amended flask 282 containing active sludge was compared with the amended flask containing sterilized 283 sludge. The theoretical added total BP3 concentration was 10 µg/mL slurry and the 284 slurry contained approximately 2 mg/mL dry matter. We analyzed the aqueous and 285 the solid phase separately and it was clear that some BP3 partitioned onto the solid 286 phase in the slurry (Figure 2). This fraction was bioavailable since we could not 287 detect any BP3 in either the solid or the aqueous phase after 20 days of incubation. 288 The concentration in the SC remained approximately the same after 20 days, 289 showing that this is not an artifact due to abiotic sorption processes. Thus, we 290 observed a complete degradation of BP3 in live cultures compared to the SC after 20 291 days.

292

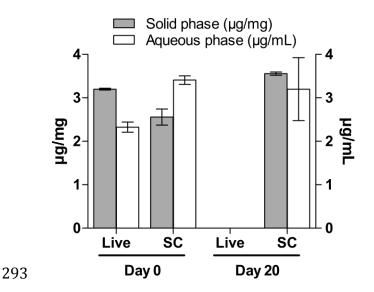


Fig. 2. BP3 concentrations in initial microcosms with WWTP sludge. BP3

295 concentrations in the aqueous phase ( $\mu$ g/mL, white bars) and in the solid phase ( $\mu$ 

296 g/mg, grey bars) at day 0 and after 20 days of incubation. Live, active degrading297 slurries and SC, sterile control.

298

299	The live culture actively degrading BP3 was transferred into an Erlenmeyer flask
300	containing 10 $\mu$ g/mL BP3 (Enrichment 1, Enr 1). After 30 days, the Enr 1 culture
301	was transferred into a flask containing a higher BP3 concentration (100 $\mu g/mL)$
302	(see figure 1), Enr 2. We added sand to the flasks with the hypothesis that it may
303	provide a solid support for the microorganisms. The cultures were transferred
304	every 30 days to further enrich BP3 degrading microorganisms. To maintain a
305	selective pressure on the microbial community, the only available carbon source for
306	the microorganisms was BP3. We could clearly see growth in the Erlenmeyer flask
307	based on the turbidity of the medium.
308	
309	After 5 consecutive transfers, we verified the degradation activity of the Enr 5

310 enrichment culture by performing biodegradation assays in separate tubes. After 7

311 days of incubation, we could only detect trace amounts BP3 in the live

312 biodegradation assays, while the SC showed no decrease compared to day-0

313 samples. This was also observed in another biodegradation assay from Enr 9 (after

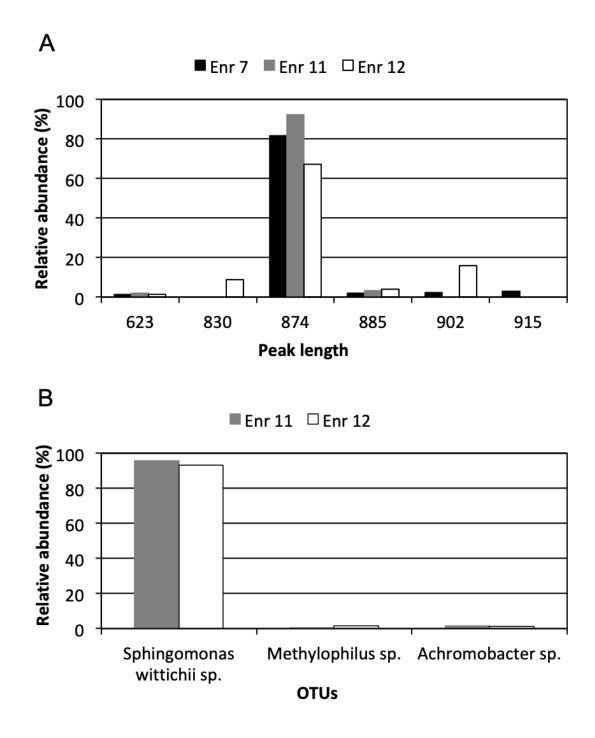
314 9 transfers) after 5 days. Thus, it was clear that these enriched cultures degrade BP3

in a relatively fast manner.

316

317 3.2 Microbial communities in enrichment cultures

318



**Fig. 3.** Analysis of the microbial community diversity in BP3 degrading enrichment

- 321 cultures. (A) ARISA analysis of enrichment cultures "Enr 7" (black bars), "Enr 11"
- 322 (grey bars) and "Enr 12" (white bars) where each peak is one "OTU" and the relative
- 323 abundance is based upon area of each peak. (B) Relative abundance of different

OTUs based upon Illumina sequencing in "Enr 11" (grey bars) and "Enr 12" (whitebars).

326

327 In order to estimate the diversity and stability of the enrichment cultures degrading 328 BP3, we performed ARISA of several enrichments and confirmed these results with 329 Illumina sequencing. We extracted DNA of cultures at day-4 after transfer and this 330 was done for different transfers (Enr 7, 11 and 12). Briefly, ARISA is based upon the 331 concept that different bacteria have different sized intergenic spacers between the 332 16S rRNA gene and the 23S rRNA gene on their chromosome. An ARISA profile will 333 give an approximation of the microbial diversity of a sample, with one peak 334 representing one "OTU". From the ARISA we could see 5 different main peaks 335 (Figure 3A) and it was clear that the enrichment cultures were dominated by one 336 specific bacterium corresponding to a peak length of 874 bp. This was confirmed 337 with the results from the Illumina sequencing, where we detected only 3 major 338 OTUs, with a clear dominance of an OTU identified as Sphingomonas wittichii with a 339 relative abundance of about 95 % in both Enr 11 and 12 (Figure 3B). We hypothesized that this strain was responsible for the degradation of BP3 in these 340 341 enrichment cultures.

342

343 3.3 Isolation and identification of the BP3 mineralizing strain BP14P

344 The most predominant strain in the enrichment cultures, BP14P, was isolated and

its degradation capability tested in minimal media (OECD301) and in minimal media

- amended with 10 % R2B media. This was done to investigate whether strain BP14P
- 347 would degrade BP3 when other carbon sources were present.

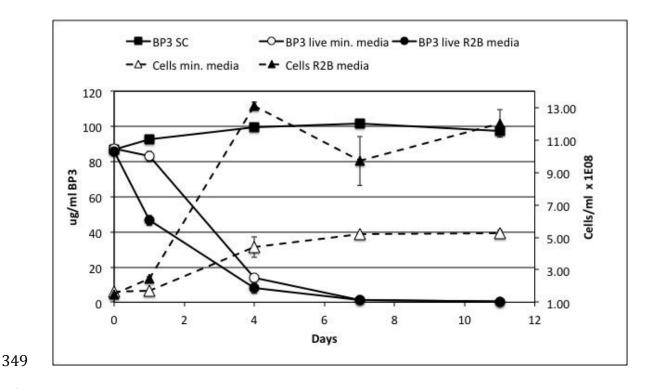


Fig. 4. Degradation of BP3 by *Sphingomonas wittichiii* sp. strain BP14P coupled to
growth. BP3 concentrations in sterile controls (black), BP3 concentrations in live
cultures with minimal media (open circles), BP3 concentrations in live cultures with
minimal media and 10 % R2B media (black circles), cells per ml in cultures with
minimal media (open triangles) and cells per ml in cultures with minimal media and
10% R2B media (black triangles).

359 Strain BP14P degraded BP3 relatively fast as we could detect an almost 50% 360 degradation after one day in cultures amended with R2B media, and 90% 361 degradation after 4 days in both cultures with R2B but also in cultures with minimal 362 media (Figure 4). This degradation was clearly linked to growth since the 363 concentration of cells increased with time in cultures where the only carbon source 364 was BP3. We hypothesize that this strain completely mineralized BP3 and used BP3 365 as both carbon and energy source. To the best of our knowledge, this is the first 366 report showing a growth dependent degradation of BP3 in a pure strain. 367 368 A BLAST comparison of the 16S rRNA gene sequence of strain BP14P (1150 bp) 369 showed that it was 100 % identical to *Sphingomonas wittichii* DC-6, isolated from 370 activated sludge (KC410868) and only one base pair different (1149/1150) to the 371 widely studied strain S. wittichii RW1 (NR\_074268.1). DC-6 can mineralize the 372 herbicides alachlor, acetochlor, and butachlor (Chen et al., 2013) through N-373 dealkylation activity by a Rieske non-heme iron oxygenase system (Chen et al.,

374 2014). S. wittichii RW1 was first isolated from an urban river sample (Wittich et al.,

375 1992) and is able to degrade a plethora of aromatic compounds, like chlorinated

376 dibenzo-*p*-dioxins (DDs) and dibenzofurans (DFs) and to grow on DD and DF as sole

377 carbon and energy sources (Wittich et al., 1992). The first step in the degradation of

378 DD and DF is carried out by a dioxygenase and interestingly many of the genes

379 responsible for degradation of aromatic compounds are present on megaplasmids in

both strains Sphingomonas wittichii RW1 and DC-6 (Armengaud et al., 1998; Cheng

et al., 2019; Miller et al., 2010). Indeed, this feature seems to be common in

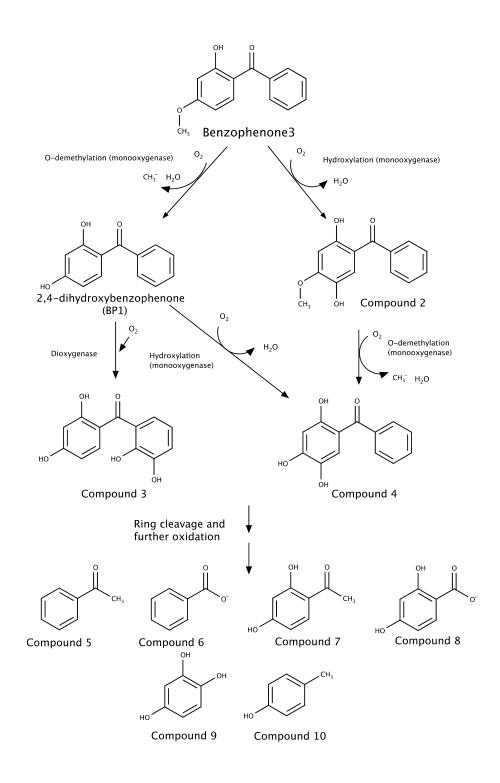
382 *Sphingomonas* that degrade xenobiotics (Basta et al., 2004).

383

# 384 *3.4 Putative degradation pathway of BP3*

385 During the HPLC-UV analysis, we could clearly detect an UV absorbing peak at an 386 earlier retention time than BP3 in some samples. To identify this peak and other 387 putative degradation products or intermediates, we analyzed several samples using 388 LC-MS/MS and molecular networks (MN) constructed by means of the Global 389 Natural Products Social Molecular Networking (GNPS) platform (Wang et al., 2016) 390 (https://gnps.ucsd.edu/). GNPS is a MS/MS data community sharing and curation 391 platform. Its MN tool organizes all MS/MS spectra recorded in complex mixtures of 392 compounds by families of analogous compounds, each grouped in one MN. This 393 enables the user to highlight possible degradation products in samples without any 394 presupposition of the chemical structure. In the present MN analysis, the number of 395 similar fragments was set to 2, and the minimum cosine for linking two parent ions 396 was set to 0.7. With these parameters, only one compound clustered with BP3, with 397 a parent mass at m/z 215.070 ([M+H]<sup>+</sup>). This compound was identified as BP1, a 398 BP3 demethylation product (Figure 5). BP1 was detected in high concentrations in 399 day 1 samples of the degradation assay with S. wittichii BP14P when R2B was 400 added. At this timepoint, BP3 was already about 50% degraded. This peak was 401 detected only at trace levels in the following timepoints, meaning that BP1 was 402 rapidly degraded. This initial demethylation reaction has been previously reported 403 to be a probable degradation step in activated sludge microcosms (Liu et al., 2012)

- 404 and as one probable pathway for BP3 degradation by *Methylophilus* sp. strain FP-6
- 405 (Jin et al., 2019).



**Fig. 5.** Putative degradation pathway of BP3.

410 We investigated other possible degradation pathways by submitting the BP3 411 structure to an online pathway prediction tool (Gao et al., 2010) and then searching 412 for particular exact masses of the predicted products  $(m/z \text{ of corresponding } [M+H]^+$ 413 ions) in all the LC-MS profiles of our samples. Based upon the prediction, another 414 possible first step is a hydroxylation of the benzene ring (Compound 2 in Figure 5). 415 An ion corresponding to this compound was detected in some of our samples, as 416 was compound **4**. However, we did not detect compound **3** in any of our samples. 417 Based upon these results we hypothesize that the main degradation pathway 418 involves an initial demethylation of BP3 to BP1, followed by a hydroxylation to give 419 **4**. However, an initial hydroxylation of BP3 yielding compound **2** and a following 420 demethylation to result in compound **4** might be a minor pathway. 421 422 Jin and colleagues identified other compounds in *Methylophilus* sp. strain FP-6 423 cultures degrading BP3 (Jin et al., 2019). The presence of these compounds, not

depicted in Figure 5, suggested a degradation pathway involving hydroxylation of
BP3 and ring cleavage, but no demethylation. We did not find any evidence for this
pathway in *S. wittichii* BP14P. This might be due to the different nature (co-

427 metabolism versus growth linked) of the degradation process of the two different428 strains in question.

429

430 Compounds **5-9** are examples of possible products after cleavage based upon the

431 prediction system. However, these low molecular weight compounds could not be

432 detected by our LC system that analyzed molecules between 133 Da and 2000 Da.

433 Compound **10** was detected by Liu et al. (2012), but was not predicted by the434 prediction tool.

435

#### 436 3.5 UV filter degradation capabilities of strain BP14P are limited

437 We tested the capability of *S. wittichii* BP14P to degrade other UV filters, listed in

438 Table 1. Indeed, Miller et al. (2010) suggest that one of the reasons why *S. wittichii* 

RW1 is so successful at degrading substrates with low water solubility might be

440 because of the abundance of TonB-dependent receptors, which can act as

transporters of aromatic compounds. Further, because the expression of some

442 dioxygenases (like dioxygenases dxnA1A2) have shown to be substrate dependent

443 (Armengaud et al., 1998), we also added BP3 to parallel assays to investigate

444 whether the putative expression of genes involved in degradation of BP3 can

stimulate the degradation of other aromatic compounds. Accordingly, we tested the

446 UV filters listed in Table 1 alone in biodegradation assays, as well as with extra

447 carbon (10% R2B) and with BP3.

448

We incubated *S. wittichii* sp. BP14P with the 9 different UV filters for 14 days. BP3 was over 95% degraded in all the assays where it was added. However, we detected very little degradation of the other UV filters (Table 1). The only other UV filter that was partially degraded with no other carbon sources present was ES (15 % after 14 days). *S. wittichii* BP14P degraded ES slightly more when BP3 and other carbon sources were present, namely 28 % after 14 days. Finally, HS was 18 % degraded after 14 days when BP3 and other carbon sources were present.

To summarize, *S. wittichii* BP14P did not degrade other UV filters to a great extent.
This specificity could be due to limited bioavailability of the other UV filters, a
limitation of putative transporters, or potentially, a limitation in the genes involved
in degradation.

461

462 3.6 What drives the specificity of BP3 degradation by strain BP14P?

463 Other researchers have observed that BP3 can be biodegraded. Liu et al. (2012)

464 found that BP3 was completely degraded after 42 days of incubation in both aerobic

and anaerobic microcosms also using inoculum from a WWTP. This suggests that

466 several groups of bacteria could be possible degraders, but this was not investigated

467 by Liu and collegues. *Methylophilus* sp. strain FP-6 was shown to co-metabolically

468 degrade BP3, degrading 15 % of BP3 after 8 days when no other carbon source was

469 provided but degrading BP3 to a higher extent when other carbon sources were

470 present (Jin et al., 2019). Based on the data provided, we believe that *Methylophilus* 

471 sp. strain FP-6 degraded BP3 co-metabolically. Strikingly, strain BP14P was very

472 specifically enriched in the cultures through many transfers, suggesting there were

473 no other bacteria present in the sludge that could degrade BP3 in a growth

474 dependent manner or in an equally efficient manner as strain BP14P.

475

476 Even though the 16S rRNA gene of strain BP14P was 99.99% and 100% identical to

477 S. wittichii RW1 and DC-6, respectively, the functional similarity between these

478 strains remains to be investigated through a more polyphasic approach, including

479 genome sequencing. Interestingly, S. wittichii RW1 has a very specific dioxygenase 480 activity that enables it to degrade DD and DF (Armengaud et al., 1998) and thus we 481 hypothesize that S. wittichii BP14P degrades BP3 through the activity of at least one 482 specific mono- or dioxygenase responsible for the initial demethylation (see figure 483 4) or another limiting step in the degradation pathway. We base this hypothesis on 484 the fact that the mono- or dioxygenase activity needed for BP3 degradation did not 485 seem to be widely distributed as no other bacteria were enriched in these cultures. 486 However, whether it uses a homologue of the *S. wittichii* RW1 enzyme, remains to be 487 investigated.

488

489 *S. wittichii* strains seem to be especially suited to degrade xenobiotics. For example

490 *S. wittichii* RW1 has extreme redundancy in its use of aromatic compound

491 metabolism (Coronado et al., 2012). This group of microorganisms generally

492 contains megaplasmids containing degradation genes and horizontal gene transfer

seems to be very common. Importantly, it is well known that *S. wittichii* RW1 can

494 survive and retain its degradation capability in soil when amended for

bioremediation purposes (Megharaj et al., 1997). All this makes it particularly

496 fascinating to enlarge the degradation repertoire of this group to include UV filters

497 and BP3.

498

### 499 **4. Conclusions and perspectives**

500 The extensive use of BP3 has resulted in ubiquitous BP3 contamination in various

501 environments, including in WWTPs and in many aquatic ecosystems (Fagervold et

502 al, 2019). For a complete environmental risk assessment of BP3, it is important to 503 understand its fate and all the processes the molecule is subjected to. Degradation, 504 though either co-metabolism or mineralization is an important factor in this 505 assessment. We show here that BP3 can be completely degraded in a few days in 506 WWTPs by *S. wittichii* BP14P. Since BP3 can be utilized by a bacterium as an energy 507 source it then ensues that BP3 has the potential to be degraded in other 508 environmental compartments than WWTP sludge. As a consequence, the long-term 509 accumulation of BP3 in the environment could be less likely than previously 510 assumed. However, once released into marine ecosystems, it is as yet unknown 511 whether BP3 can be degraded. This question should be the subject of future studies, 512 as well as performing similar degradation testing on other sunscreen UV filters, with 513 a priority on the most toxic UV filters including octinoxate, octocrylene and 514 homosalate

515

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- 531

# 532 Figure captions

533 Figure 1. Schematic of the experimental design.

534

535 Figure 2. BP3 concentrations in initial microcosms with WWTP sludge. BP3

536 concentrations in the aqueous phase ( $\mu$ g/mL, white bars) and in the solid phase ( $\mu$ 

537 g/mg, grey bars) at day 0 and after 20 days of incubation. Live, active degrading

538 slurries and SC, sterile control.

539

540 Figure 3. Analysis of the microbial community diversity in BP3 degrading

541 enrichment cultures. (A) ARISA analysis of enrichment cultures Enr 7 (black bars),

542 Enr 11 (grey bars) and Enr 12 (white bars) where each peak is one OTU and the

- 543 relative abundance is based upon area of each peak. (B) Relative abundance of
- 544 different OTUs based upon Illumina sequencing in Enr 11 (grey bars) and Enr 12

545 (white bars).

547	Figure 4. Degradation of BP3 by Sphingomonas wittichiii sp. strain BP14P (left axis)
548	and growth followed by flow cytometry (right axis). BP3 concentrations in sterile
549	controls (SC, black square), BP3 concentrations in live cultures with minimal media
550	(open circles), BP3 concentrations in live cultures with minimal media
551	supplemented with 10 % R2B media (black circles), cell concentrations of cultures
552	with minimal media (open triangles) or minimal media supplemented with $10\%$
553	R2B media (black triangles).
554	
555	
556	Figure 5. Putative degradation pathway of BP3. Hypothesized degradation pathway
557	of BP3 based upon the online prediction tool "EAWAG-BBD Pathway Prediction
558	System"( <u>http://eawag-bbd.ethz.ch/predict/</u> )
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