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Conformational effect on the almost free internal rotation in 4-hexyn-3-ol studied by microwave spectroscopy and quantum chemistry

Konrad Eibl,^{1,2} Wolfgang Stahl,¹ Isabelle Kleiner,² and Ha Vinh Lam Nguyen^{2,a)} ¹Institute of Physical Chemistry, RWTH Aachen University, Landoltweg 2, D-52074 Aachen, Germany ²Laboratoire Interuniversitaire des Systèmes Atmosphériques (LISA), CNRS UMR 7583, Université Paris-Est Créteil, Université Paris Diderot, Institut Pierre Simon Laplace, 61 Avenue du Général de Gaulle, F-94010 Créteil, France

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The microwave spectrum of 4-hexyn-3-ol, $CH_3-C\equiv C-CH(OH)-CH_2CH_3$, was recorded in the frequency range of 2–26.5 GHz by molecular jet Fourier transform microwave spectroscopy. The conformational analysis based on quantum chemical calculations yielded nine conformers exhibiting C₁ symmetry, of which three could be assigned in the experimental spectrum. The propynyl methyl group **CH₃-C≡C** – experiences internal rotation with a very low barrier due to the presence of the cylindrically symmetric $-C\equiv C$ – group serving as a spacer to the rest of the molecule, which is 7.161 012(7) cm⁻¹, 4.236 5(26) cm⁻¹, and 7.901 6(39) cm⁻¹ for the three assigned conformers, respectively. The spectrum was analyzed with the program *XIAM* using the combined axis method and the program *BELGI-C₁* using the rho axis method and a very flexible Hamiltonian which yields fits with root-mean-square deviations within the measurement accuracy. *Published by AIP Publishing*. https://doi.org/10.1063/1.5044542

I. INTRODUCTION

Fourier transform microwave (FTMW) spectroscopy has become the most important method to investigate the large amplitude motions of a molecule, where internal rotation is one of the main fields of interest. Due to higher resolution and more accurately determined molecular parameters in comparison to, for example, fluorescence or electronic spectroscopy, many barriers to internal rotation have been determined by this method.^{1–3} The barrier height varies in a wide range, but could be classified in three main classes. The first class comprises molecules with high barriers (>600 cm^{-1}) with ethyl chloride as a typical example (1200 cm^{-1}) .⁴ The second class contains molecules where the barrier height is intermediate $(200-600 \text{ cm}^{-1})$. There are many examples on molecules possessing one or more internal rotors with barrier heights in this range, such as o-methylanisole, 2,5-dimethylfuran, and methyl methacrylate.^{5–7} The torsional splittings are typically up to several hundreds of MHz, and the microwave spectrum can be modeled with sufficient accuracy by considering the V_3 potential and the position of the internal rotor in the molecule. Finally, molecules in the last class undergo internal rotations of low barrier heights ($<200 \text{ cm}^{-1}$),^{8–10} with a sub-class of very low barriers $(<30 \text{ cm}^{-1})$.^{11–13} The rotational spectra of such molecules are hard to assign and to model because of large torsional splittings up to several GHz and because high order terms in the Hamiltonian are often required to reproduce the spectra to experimental accuracy. A very low barrier to internal rotation might arise by separating the methyl rotor from the rest of the molecule by the acetylenic group with cylindrical symmetry. Only a very limited number of molecules of the type CH₃—C \equiv C—**R** have been studied by microwave spectroscopy, among them are dimethylacetylene- d_3 (**R** = CD₃),¹⁴ 1,1,1-trifluoro-2-butyne (**R** = CF₃),¹⁵ methylsilylacetylene (**R** = SiH₃),¹⁶ tetrolyl fluoride (**R** = COF),¹⁷ 2-butynoic acid (**R** = COOH),¹⁸ 1-chloro-2-butyne (**R** = CH₂Cl),¹⁹ 2-butynol (**R** = CH₂OH),²⁰ and 3-pentyn-1-ol (**R** = CH₂CH₂OH),²¹ which is the most recent investigation. The three-fold barrier to internal rotation of the CH₃–C \equiv C— methyl group (called the propynyl methyl group) is lower than 10 cm⁻¹ in all cases.

This motivated the authors to measure the microwave spectrum of 4-hexyn-3-ol ($\mathbf{R} = CH(OH) - CH_2CH_3$) to find out if the length of the alkyl chain and the position of the hydroxyl group affect the barrier to the internal rotation of the propynyl methyl group. Moreover, in all previous studies, only one conformer was present in the experimental spectrum. In the case of 4-hexyn-3-ol, the alkyl chain is sufficiently long to support conformational isomerism. Currently, no rule exists to predict whether the conformations of a molecule affect the barrier height of the methyl rotor. In some cases, such as *m*-methylanisole,²² *m*-methylbenzaldehyde,²³ or N,N-diethylacetamide,²⁴ significant differences in V_3 potentials between the rotational conformers were observed, while in some other cases like *n*-butyl acetate²⁵ and methyl butyrate,²⁶ they remain almost unchanged in different conformers. The title molecule is ideally suited to study the conformational effect on very low barrier heights, which is observed for the first time in microwave spectroscopy.

^a)Author to whom correspondence should be addressed: lam.nguyen@lisa. u-pec.fr.



FIG. 1. The energetically most favorable conformer of 4-hexyn-3-ol (conformer I). The dot represents the position of the center of mass. The hydrogen atoms 11 and 16 are located behind the carbon atoms 4 and 6, respectively.

II. THEORETICAL

A. Quantum chemical calculations

Because the rotations of the methyl groups about the triple bond do not create new conformations, the conformational landscape of 4-hexyn-3-ol is fully defined by the dihedral angles $\varphi_1 = \angle(C_3, C_4, C_5, C_6)$ and $\varphi_2 = \angle(C_3, C_4, O_7, H_{17})$, corresponding to rotations about the C_4 - C_5 (ethyl group) and the C_4 - O_7 bond (OH group), respectively. The atom numbers are given in Fig. 1. For a conformational analysis, we calculated a two-dimensional potential energy surface (2D-PES) depending on φ_1 and φ_2 by varying both angles in a grid of 10°, while all other geometry parameters were optimized at the MP2/6-311++G(d,p) level of theory using the Gaussian09 program package.²⁷ The potential energies were parameterized with a 2D Fourier expansion. The PES was drawn as a contour plot illustrated in Fig. 2 calculated from the Fourier coefficients, which are available in Table S-1 of the supplementary material. The PES exhibits nine minima within an energy range of 12 kJ mol⁻¹, associated with nine combinations from three possible staggered orientations of the ethyl ($\varphi_1 = \pm 60^\circ$ or 180°) and the hydroxyl group



FIG. 2. The potential energy surface of 4-hexyn-3-ol calculated at the MP2/6-311++G(d,p) level of theory depending on the dihedral angles $\varphi_1 = \angle(C_3, C_4, C_5, C_6)$ and $\varphi_2 = \angle(C_3, C_4, O_7, H_{17})$. The numbers in the color code indicate the energy (in percent) relative to the energetic minimum $E_{min} = -308.988\,974$ hartree (0%) and the energetic maximum $E_{max} = -308.974\,156$ hartree (100%).

 $(\varphi_2 = \pm 60^\circ \text{ or } 180^\circ)$ relative to the rest of the molecule. Re-optimizations under full geometry relaxation and frequency calculations confirmed that all conformations on the PES are true minima. The nine conformers possess C₁ symmetry; each of which thus has an enantiomer with the same potential energy, and the mirror plane contains the C_1, C_2, C_3 , C₄, O₇ atoms. The rotational constants, dipole moment components, and energies relative to that of the most stable conformer are summarized in Table I. The energies corrected for the zero point vibrations of conformers V-IX are 3-10 kJ mol⁻¹ higher than that of conformer I. From our experiences, they are not present in the jet-cooled spectrum where the rotational temperature is very low (about 2 K), and therefore, only conformers I-IV are considered in the spectral analysis. The geometries of these four conformers are illustrated in Fig. 3. Their Cartesian coordinates are available in Table S-2 of the supplementary material.

As mentioned in the Introduction, we expect that the propynyl methyl group of 4-hexyn-3-ol undergoes internal rotation with a very low barrier height, similar to the

TABLE I. The energies ΔE (in kJ mol⁻¹) relative to the lowest energy conformer **I** with its absolute energy of $E = -308.988\,999$ hartree, zero point corrected energies ΔE_{ZPE} (in kJ mol⁻¹) relative to conformer **I** with its absolute zero point energy of $E_{ZPE} = -308.841\,360$ hartree, V_3 potential of the propynyl methyl group (in cm⁻¹), rotational constants (in GHz), and dipole moment components (in Debye based on the nuclear coordinates given in Table S-2 of the supplementary material) of the nine conformers of 4-hexyn-3-ol calculated at the MP2/6-311++G(d,p) level of theory.

Conf.	ΔE	ΔE_{ZPE}	V_3	Α	В	С	μ_a	μ_b	μ_c
I	0	0	8.8	3.5840	1.5370	1.1341	1.87	-0.79	0.85
II	0.33	0.68	1.8	4.2144	1.4300	1.2152	-2.05	0.04	0.71
III	1.31	1.31	1.8	3.5350	1.5450	1.1361	1.40	1.67	-1.03
IV	1.85	1.85	7.6	6.4839	1.1408	1.0136	1.88	-0.45	0.85
V	2.77	2.93	7.4	4.1383	1.4308	1.2123	1.46	1.91	-0.13
VI	3.36	3.03	5.4	6.2847	1.1453	1.0142	-1.47	1.54	-0.89
VII	8.87	7.32	4.9	3.5659	1.5465	1.1367	-0.83	1.64	0.33
VIII	9.64	8.12	6.6	4.1482	1.4454	1.2198	-0.44	0.61	1.50
IX	11.90	10.50	7.3	6.2828	1.1510	1.0166	0.58	-1.19	1.15



FIG. 3. (a) Geometries of conformers **I**, **II**, and **IV** of 4-hexyn-3-ol viewed along the C_5-C_4 axis. (b) View along the O_7-C_4 axis, showing possible staggered orientations of the OH group which distinguish conformers **I**, **III**, and **VII**.

case of 3-pentyn-1-ol²¹ and other molecules of the type $CH_3-C\equiv C-R$. As a consequence, very large A-E torsional splittings of all rotational lines were assumed to be present in the microwave spectrum. The barrier height was calculated at the MP2/6-311++G(d,p) level of theory for all conformers by varying the dihedral angle $\alpha = (C_5, C_4, C_1, H_8)$ in a grid of 10°, while all other parameters were optimized. The results are also given in Table I.

III. MICROWAVE SPECTROSCOPY

A. Measurements

The 4-hexyn-3-ol sample with a stated purity of 95% was purchased from Alfa Aesar, Karlsruhe, Germany and used without further purification. The rotational spectrum was observed with a molecular jet FTMW spectrometer operating in the 2–26.5 GHz range.²⁸ A few drops of 4-hexyn-3-ol were placed on a 5 cm piece of a pipe cleaner and inserted into a stainless steel tube. Helium at a pressure of 150-200 kPa was flown over the sample, and the helium-substance mixture was expanded into the cavity. For assignment purposes, a broadband scan was recorded in the range of 10.0-15.5 GHz at a step width of 250 kHz, indicating the positions of the lines in the spectrum. A portion of the scan is illustrated in Fig. 4. All lines were remeasured afterwards at higher resolution. The average of the line widths is about 20 kHz, corresponding to a measurement accuracy of approximately 2 kHz.



FIG. 4. A portion of the broadband scan between 10 600 and 10 700 MHz. All signals are marked with their assigned quantum numbers, conformer, and symmetry species.

B. Spectral analysis and fits

Quantum chemical calculations have shown that conformer I is the most stable one. Therefore, we focused at the beginning on the spectrum of this conformer because its rotational transitions should be the strongest which might simplify the assignment.

The internal rotation effect was first neglected, and conformer I of 4-hexyn-3-ol was treated as a rigid rotor. Using the rotational constants and dipole moment components given in Table I, we calculated a theoretical microwave spectrum with the program $XIAM^{29}$ in its rigid-rotor mode for a comparison with the measured one. The intense R branch a-type lines with J = 3-6 are within the broadband scan, and their frequencies were very well predicted. The assignment was thus straightforward on the basis of their characteristic pattern and the predicted frequencies, yielding the B and C rotational constants. Afterwards, c-type transitions were also assigned, which fixed the A rotational constant. This enabled us to predict the whole spectrum of the pseudo rigid rotor with sufficient accuracy to find more A species lines in the frequency range 2-26.5 GHz. At this stage, 93 A species lines could be assigned and fitted with the program XIAM using the three rotational constants and the quartic centrifugal distortion constants to an rms deviation of 1.2 kHz, which is within the measurement accuracy.

Using the same approach, the pseudo rigid rotor spectra of conformer **II** and conformer **IV** with 68 and 47 A species lines could be assigned and fitted to rms deviations of 1.2 kHz and 0.9 kHz, respectively. Despite many attempts, it was not possible to assign any lines of conformer **III**. The results of the fits for conformers **I**, **II**, and **IV** are given as Fit XIAM A in Table **II**.

In the next step, we considered the internal rotation of the propynyl methyl group and predicted a one-top spectrum for conformers I, II, and IV using the rotational constants from the rigid rotor fit and the V_3 potential as well as the angles between the internal rotor axis and the principal axes from ab initio calculations. First of all, the A species lines were refitted in a one-top fit with the internal rotation parameters in order to eliminate the deviations resulting from treating the molecule as a pseudo-rigid rotor. Using the adjusted parameters, the positions of the E species lines of conformer I were calculated and compared with the observed frequencies. For the *R* branch *a*-type lines in the broadband scan mentioned in Sec. III A, the predicted E species frequencies were close to the experimental ones. Therefore, the E species of these lines could be easily assigned, yielding the V_3 potential as well as the angles between the internal rotor axis and the principal axes. In other branches, deviations between the predicted and observed frequencies ranged from a few hundred MHz up to one GHz. Finally, the E species lines of c-type transitions were also identified. The A-E splittings varied from about 10 MHz ($7_{52} \leftarrow 6_{51}$ transition) up to about 7 GHz ($2_{20} \leftarrow 1_{10}$ transition). Assigning well-predicted lines with consistent A-E splittings within a branch first and re-fitting the molecular parameters before continuing with transitions in other branches was the key to the successful assignment of all E species lines. Finally, a global fit with 93 A and 54 E species transitions was achieved using

			Ι			П			IV	
Par. ^a	Unit	XIAM A	XIAM A/E	BELGI	XIAM A	XIAM A/E	BELGI	XIAM A	XIAM A/E	BELGI
A	MHz	3687.13751(20)	3609.669 6(74)	3609.361 63(98)	4325.601 29(58)	4221.126(16)	4186.30(69)	6800.466 81(31)	6536.239(23)	6536.347(20)
В	MHz	1529.707 405(68)	1529.0634(34)	1528.80747(81)	1431.081456(96)	1430.382 2(42)	1431.862(73)	1145.170 25(11)	1145.0039(17)	1145.003 04(80)
C	MHz	1131.302 989(52)	1131.183 5(25)	1130.92667(90)	1217.672117(95)	1217.432 6(34)	1213.635(35)	1017.91452(12)	1017.8088(15)	1017.807 59(16)
D_J	kHz	0.2848(67)	0.2848^{b}	:	0.29588(71)	0.29588^{b}	÷	0.073 45(32)	0.073 45 ^b	:
D_{JK}	kHz	3.1492(36)	3.1492 ^b	:	1.7006(49)	$1.700 6^{b}$:	0.4600(73)	0.4600^{b}	:
D_K	kHz	-0.758(16)	-0.758 ^b	:	3.50(12)	3.50^{b}	:	:	:	:
d_1	kHz	0.09554(30)	0.09554 ^b	:	0.07091(30)	0.07091 ^b	÷	0.013 38(16)	0.013 38 ^b	:
d_2	kHz	2.503 2(98)	2.503 2 ^b	:	1.859(35)	1.859 ^b	÷	0.759(51)	0.759 ^b	:
$D_{p\hat{1}2J}$	kHz	:	÷	:	:	-12.8(19)	÷	:	-5.63(65)	:
F	GHz	:	163.556 079 ^c	163.556079 ^d	:	164.087 50 ^c	164.087 50 ^d	:	166.591 221 ^c	166.591 221 ^d
V_3	cm ⁻¹	:	7.161012(7)	7.21596(14)	:	4.236 5(26)	3.788 1(86)	:	7.9016(39)	7.8817(57)
θ	÷	:	0.021 983 553 ^e	0.021994959(28)	:	0.025 390 898 ^e	0.0253310(11)	:	0.040 094 337 ^e	$0.040\ 085\ 5(19)$
$\angle(\mathbf{i}, a)$	deg	:	14.183 1(6)	14.186009(37)	÷	16.675 5(19)	15.066(22)	:	11.2074(18)	11.204 92(46)
$\angle({\rm i},b)$	deg	:	77.7582(9)	77.755 596(41)	÷	76.338 6(23)	78.225(26)	:	81.6805(29)	81.685 60(59)
(i, c)	deg	÷	82.947 3(10)	82.9460602(11)	÷	80.621 0(22)	80.7355(21)	:	82.543 8(29)	82.54172(52)
N_A/N_E^{f}		93/0	93/54	93/54	68/0	68/43	68/43	47/0	47/40	47/40
0 <mark>8</mark>	kHz	1.2	181.9	1.8	1.2	186.6	1.3	0.9	71.8	1.3
^a Standard	error in pare	otheses in the unit of the l	ast digit. Watson's S reduct	tion and the I' representation	n were used.					

^bFixed to the XIAM A values.

^c Derived parameter from F_0 , which is not fitted due to high correlation with V_3 , but fixed to 160 GHz, a value often found for methyl groups.

^dNot fitted. Fixed to the XIAM A/E value.

^e Derived parameter from F_0 .

^fNumber of fitted A and E species lines. ^grms deviation of the fit.

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TABLE III. Molecular parameters in the rho axis system obtained using the BELGI-C1 code.

Par. ^a	Unit	Operator ^b	I ^c	Пc	IV ^c
A	MHz	\mathbf{P}_{a}^{2}	3587.814 54(98)	4165.12(69)	6530.415(20)
В	MHz	$\mathbf{P}_{\mathrm{b}}^{2}$	1546.655 83(81)	1446.580(33)	1148.65973(61)
С	MHz	\mathbf{P}_{c}^{2}	1134.625 48(89)	1220.092(36)	1020.082 68(15)
D_{ab}	MHz	$\{\mathbf{P}_{a}, \mathbf{P}_{b}\}$	192.733 96(66)	203.65(45)	141.597(11)
Daci	MHz	$\{\mathbf{P}_a, \mathbf{P}_c\}$	93.57901(11)	-133.82163(24)	110.422 02(72)
Δ_J	kHz	$-\mathbf{P}^4$	0.2890(13)	0.290 86(74)	0.073 21(46)
Δ_K	kHz	$-\mathbf{P}_{a}^{4}$	-1.063(23)	6.68(15)	
Δ_{JK}	kHz	$-\mathbf{P}^2\mathbf{P}_a^2$	3.252(56)	-1.073(50)	0.479 5(61)
δ_J	kHz	$-2\mathbf{P}^2(\mathbf{P}_{\rm b}^2 - \mathbf{P}_{\rm c}^2)$	0.096 66(58)	0.069 51(35)	0.012 81(24)
δ_K	kHz	$-\{\mathbf{P}_{a}^{2},(\mathbf{P}_{b}^{2}-\mathbf{P}_{c}^{2})\}$	2.408(15)	0.512(34)	0.814(76)
V_3	cm ⁻¹	$0.5(1 - \cos(3\alpha))$	7.215 96(14)	3.788 1(86)	7.8817(57)
F	GHz	\mathbf{p}_{α}^{2}	163.556079 ^d	164.087 50 ^d	166.591 221 ^d
ρ		$\mathbf{P}_{a}\mathbf{p}_{\alpha}$	0.021 994 959(28)	0.025 331 0(11)	0.040 085 5(19)
d_{ab}	GHz	$\{\mathbf{P}_{a}, \mathbf{P}_{b}\}(1 - \cos(3\alpha))$		0.037 20(48)	
Δ_{ab}	MHz	$\{\mathbf{P}_{a},\mathbf{P}_{b}\}\mathbf{p}_{\alpha}^{2}$			0.0903(98)
δ_{ab}	MHz	$\{\mathbf{P}_a^2, \mathbf{P}_b\}\mathbf{p}_{\alpha}$	-0.009 98(30)	-0.11666(55)	
D_{abJ}	kHz	$\mathbf{P}^2{\{\mathbf{P}_a,\mathbf{P}_b\}}$	-0.403 2(85)		
D_{abK}	kHz	$\{\mathbf{P}_a^3, \mathbf{P}_b\}$		7.59(24)	
F_V	MHz	$\mathbf{P}^2(1 - \cos(3\alpha))$	0.278 46(99)	4.0529(37)	
G_V	MHz	$\mathbf{P}^2 \mathbf{p}_{\alpha}^2$			-0.01080(26)
L_V	MHz	$\mathbf{P}^{2}\mathbf{P}_{a}\mathbf{p}_{\alpha}$	0.013 87(58)		0.005 262(15)
c_{I}	MHz	$2\mathbf{p}_{\alpha}^2(\mathbf{P}_{b}^2-\mathbf{P}_{c}^2)$		-0.04617(49)	
С4	kHz	$\{\mathbf{P}_{a}, (\mathbf{P}_{b}^{2} - \mathbf{P}_{c}^{2})\}\mathbf{p}_{\alpha}$	2.944(45)		
k5	GHz	$\mathbf{P}_{a}^{2}(1-\cos(3\alpha))$		0.026 92(73)	
N_A/N_E^{e}			93/54	68/43	47/40
σ	kHz		1.8	1.3	1.3

^aAll constants refer to a rho axis system; therefore, the inertia tensor is not diagonal and the constants cannot be directly compared to those of a principal axis system. \mathbf{P}_a , \mathbf{P}_b , \mathbf{P}_c are the components of the overall rotation angular momentum, and \mathbf{p}_{α} is the angular momentum of the internal rotor rotating about the internal rotor axis by an angle α . {**A**, **B**} is the anti-commutator **AB** + **BA**. ^bThe product of the parameter and operator from a given row yields the term actually used in the vibration–rotation–torsion Hamiltonian, except for *F*, ρ , and *A*, which occur in the Hamiltonian in the form $F(\mathbf{p}_{\alpha} - \rho \mathbf{P}_a)^2 + A\mathbf{P}_a^2$.

^c Values of the parameters from the present fit. Statistical uncertainties are shown as one standard uncertainty in the last digit. ^d Fixed value. See footnote d of Table II.

^eNumber of fitted A and E species lines.

the program *XIAM* with an rms deviation of 181.9 kHz, which is much larger than the measurement accuracy of 2 kHz. The E species transitions of conformers **II** and **IV** were assigned in the same manner. The global fits containing 68 A and 43 E species transitions for conformer **II** as well as 47 A and 40 E species transitions for conformer **IV** possessed the rms deviations of 186.6 kHz and 71.8 kHz, respectively. All fitted parameters are given as Fit XIAM A/E in Table **II**. A list of fitted frequencies along with their residuals is available in Table S-3 of the supplementary material.

Since the fits with *XIAM* could not achieve the measurement accuracy, the same datasets were fitted using the *BELGI*- $C_1 \operatorname{code}^{30}$ which uses a two-step diagonalization procedure performing in the rho axis system. The torsional Hamiltonian $H_T = F(\mathbf{p}_{\alpha} - \rho \mathbf{P}_a)^2 + V(\alpha)$ is diagonalized in the basis set $\exp(i[3k + \sigma_1]\alpha)|K\rangle^{31}$ in the first step, where the value of the integer k ranges from $-10 \operatorname{to} 10$. σ_1 is either 0 for the A species or ± 1 for the E species, and $|K\rangle$ is the symmetric top eigenfunction. The other terms are the potential function $V(\alpha)$ of the internal rotation angular momentum \mathbf{p}_{α} conjugated to the torsional angle α , the coupling constant between the internal rotation and the global rotation ρ , and the *a* component of the total

rotational angular momentum P_a . The Hamiltonian matrix of the size $(2k + 1) \times (2k + 1) = 21 \times 21$ is diagonalized for each *K* value, and we obtain the torsional eigenvalues and eigenvectors. In the second step, we use a basis set consisting of the products of the torsional eigenfunctions, obtained from the first step diagonalization and the symmetric top wavefunctions $|JK\rangle$ to diagonalize the rest of the terms of the Hamiltonian, i.e., H_R , H_{cd} , and H_{int} allowed for C₁ symmetry. The nine lowest torsional states v_t are taken to create a matrix of the size $(2J + 1) \cdot 9 \times (2J + 1) \cdot 9$ associated with the second step.

By floating 17, 18, and 14 parameters, respectively, the rms deviations of the fits of conformers **I**, **II**, and **IV** decrease to 1.8 kHz, 1.3 kHz, and 1.3 kHz, respectively. The *BELGI-C*₁ parameters in the rho axis system are available in Table III, and the parameters that could be transferred to the principal axis system are also in Table III.

IV. RESULTS AND DISCUSSION

Using the *XIAM* code, the spectra of conformers **I**, **II**, and **IV** of 4-hexyn-3-ol could be reproduced to rms deviations of 181.9, 186.6, and 71.8 kHz, respectively, which are

much higher than the measurement accuracy of 2 kHz. This is frequently observed in the *XIAM* fits of molecules with low barriers to internal rotation.^{33–35} If only A and E species transitions with low K_a values ($K_a \le 1$) are included in the fits, the rms deviations decrease to measurement accuracy. Because the deviations are large, it is not possible to determine the quartic centrifugal distortion constants in all XIAM A/E fits. Therefore, they were fixed at the values obtained from the XIAM A fits.

By adding higher order parameters using $BELGI-C_1$, the rms deviations of the fits for conformers **I**, **II**, and **IV** are 1.8 kHz, 1.3 kHz, and 1.3 kHz, respectively, which are within the measurement accuracy. This leads to the conclusion that the available set of parameters in *XIAM* is not able to accurately describe the influence of the very low hindered internal rotor at higher K_a quantum numbers for 4-hexyn-3-ol.

The experimental results are directly compared to the constants obtained from *ab initio* calculations given in Table I, and the experimental sets of rotational constants can be unambiguously attributed to the structures of conformers I, II, and IV obtained from quantum chemistry. The calculated *B* and *C* constants are in almost perfect agreement with the experimental values, whereby the deviations are less than 10 MHz for both conformers. Larger deviations of up to 50 MHz are observed for the *A* rotational constants.

There are two important factors which influence the barrier to internal rotation: the electronic effect and the sterical hindrance. The effect of sterical hindrance in different conformers on the barrier height has been recently observed for two conformers of *m*-methyl anisole, where the more sterically hindered cis-conformer has a higher barrier than the transconformer.²² Assuming that the electronic effect is the same in the three observed conformers I, II, and IV of 4-hexyn-3ol, the main impact for the differences between the three-fold potentials given in Table II is the sterical effect. From the calculated geometries illustrated in Fig. 3, the sterical hindrance is the lowest in conformer IV because the methyl group in the ethyl moiety is furthest from the propynyl methyl group, while it is similar for conformers I and II. Therefore, chemical intuition would expect that the barrier to internal rotation of conformer IV is the lowest and those of conformers I and II are similar. However, the calculated and experimental values both show that the lowest barrier is found for conformer II and similar barriers are observed for conformers I and IV. Our current conclusion is that although the barrier height is very low ($<10 \text{ cm}^{-1}$), the conformation still significantly affects the barrier height. Because 4-hexyn-3-ol is the only example for this class of molecule showing a conformational effect on the barrier height, it is not yet possible to state any quantitative conclusion. This requires further investigations on similar systems.

No lines belonging to conformer **III** could be assigned in the experimental spectrum, though quantum chemistry calculated this conformer to be more stable than conformer **IV**. As conformer **III** only differs from the most stable conformer **I** by the position of the hydrogen atom in the hydroxyl group, the rotational constants of conformers **I** and **II** are very similar and consequently their rigid rotor spectra should be similar as well. Sample rigid-rotor calculations show that the *R* branch $K_a = 0$ and $\Delta K_c = +1$ transition frequencies of conformers I and III differ by only about 25 kHz. This kind of splittings is not observed in the experimental spectrum, and we conclude that conformer III is not present under our measurement conditions. The energy required for conformational relaxation from conformer III to conformer I is predicted to be only about 400 cm⁻¹, leading to a high possibility for the conversion of conformer III to conformer I during the expansion into the cavity. After the three conformers were assigned, only some very weak lines remain in the broadband scan, implying that ¹³C isotopologues or complexes are not observable for 4-hexyn-3-ol under our measurement conditions.

SUPPLEMENTARY MATERIAL

See supplementary material for the Fourier coefficients of the potential energy surface, Cartesian coordinates, and frequency lists.

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- ¹H. V. L. Nguyen and W. Stahl, ChemPhysChem **12**, 1900 (2011).
- ²A. K. Huff, R. B. Mackenzie, C. J. Smith, and K. R. Leopold, J. Phys. Chem. A **121**, 5659 (2017).
- ³N. L. Owen, J. Mol. Struct. 6, 37 (1970).
- ⁴W. Stahl, H. Dreizler, and M. Hayashi, Z. Naturforsch., A **38**, 1010 (1983).
- ⁵L. Ferres, H. Mouhib, W. Stahl, and H. V. L. Nguyen, ChemPhysChem 18, 1855 (2017).
- ⁶V. Van, J. Bruckhuisen, W. Stahl, V. Ilyushin, and H. V. L. Nguyen, J. Mol. Spectrosc. **343**, 121 (2018).
- ⁷S. Herbers, D. Wachsmuth, D. A. Obenchain, and J.-U. Grabow, J. Mol. Spectrosc. **343**, 96 (2018).
- ⁸L. Ferres, W. Stahl, I. Kleiner, and H. V. L. Nguyen, J. Mol. Spectrosc. 343, 44 (2018).
- ⁹R. Kannengießer, M. J. Lach, W. Stahl, and H. V. L. Nguyen, ChemPhysChem 16, 1906 (2015).
- ¹⁰H. V. L. Nguyen and W. Stahl, J. Mol. Spectrosc. **264**, 120 (2010).
- ¹¹A. Roucou, I. Kleiner, M. Gouber, S. Bteich, G. Mouret, R. Bocquet, F. Hindle, W. L. Meerts, and A. Cuisset, ChemPhysChem 19, 1056 (2018).
- ¹²V. V. Ilyushin, E. A. Alekseev, S. F. Dyubko, I. Kleiner, and J. T. Hougen, J. Mol. Spectrosc. 227, 115 (2004).
- ¹³P. H. Turner, M. J. Corkill, and A. P. Cox, J. Phys. Chem. 83, 1473–1482 (1979).
- ¹⁴J. Nakagawa, M. Hayashi, Y. Endo, S. Saito, and E. Hirota, J. Chem. Phys. 80, 5922 (1984).
- ¹⁵S. Blanco, M. E. Sanz, A. Lesarri, J. C. López, and J. L. Alonso, Chem. Phys. Lett. **397**, 379 (2004).
- ¹⁶J. Nakagawa, K. Yamada, M. Bestera, and G. Winnerwisser, J. Mol. Spectrosc. **110**, 74 (1985).
- ¹⁷K. D. Hensel and M. C. L. Gerry, J. Chem. Soc., Faraday Trans. **90**, 3023 (1994).
- ¹⁸ V. Ilyushin, R. Rizzato, L. Evangelisti, G. Feng, A. Maris, S. Melandri, and W. Caminati, J. Mol. Spectrosc. **267**, 186 (2011).
- ¹⁹V. M. Stolwijk and B. P. van Eijck, J. Mol. Spectrosc. **124**, 92 (1987).
- ²⁰R. Subramanian, S. E. Novick, and R. K. Bohn, J. Mol. Spectrosc. **222**, 57 (2003).
- ²¹K. Eibl, R. Kannengießer, W. Stahl, H. V. L. Nguyen, and I. Kleiner, Mol. Phys. **114**, 3483 (2016).
- ²²L. Ferres, W. Stahl, and H. V. L. Nguyen, J. Chem. Phys. **148**, 124304 (2018).
- ²³A. J. Shirar, D. S. Wilcox, K. M. Hotopp, G. L. Storck, I. Kleiner, and B. C. Dian, J. Phys. Chem. A **114**, 12187 (2010).

- ²⁴R. Kannengießer, S. Klahm, H. V. L. Nguyen, A. Lüchow, and W. Stahl, J. Chem. Phys. **141**, 204308 (2014).
- ²⁵T. Attig, L. W. Sutikdja, R. Kannengießer, I. Kleiner, and W. Stahl, J. Mol. Spectrosc. 284-285, 8 (2013).
- ²⁶A. O. Hernandez-Castillo, C. Abeysekera, B. M. Hays, I. Kleiner, H. V. L. Nguyen, and T. S. Zwier, J. Mol. Spectrosc. **337**, 51 (2017).
- ²⁷M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino,
 - G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda,
 - J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai,
 - T. Vreven, J. A. Montgomery, Jr., J. E. Peralta, F. Ogliaro, M. Bearpark,
- J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, R. Kobayashi,
- J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar,
- J. Tomasi, M. Cossi, N. Rega, J. M. Millam, M. Klene, J. E. Knox, J.
- B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski,

R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, O. Farkas, J. B. Foresman, J. V. Ortiz, J. Cioslowski, and D. J. Fox, GAUSSIAN 09, Revision A.02, Gaussian, Inc., Wallingford, CT, 2009.

- ²⁸J.-U. Grabow, W. Stahl, and H. Dreizler, Rev. Sci. Instrum. **67**, 4072 (1996).
- ²⁹H. Hartwig and H. Dreizler, Z. Naturforsch., A **51**, 923 (1996).
- ³⁰I. Kleiner and J. T. Hougen, J. Chem. Phys. **119**, 5505 (2003).
- ³¹E. Herbst, J. K. Messer, F. C. De Lucia, and P. Helminger, J. Mol. Spectrosc. 108, 42 (1984).
- ³²D. Jelisavac, D. C. Cortés-Gómez, H. V. L. Nguyen, L. W. Sutikdja, W. Stahl, and I. Kleiner, J. Mol. Spectrosc. 257, 111 (2009).
- ³³H. V. L. Nguyen, H. Mouhib, W. Stahl, and I. Kleiner, Mol. Phys. **108**, 763 (2010).
- ³⁴H. V. L. Nguyen, A. Jabri, V. Van, and W. Stahl, J. Phys. Chem. A 118, 12130 (2014).
- ³⁵A. Jabri, V. Van, H. V. L. Nguyen, W. Stahl, and I. Kleiner, ChemPhysChem 17, 2660 (2016).