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Limit of the buoyancy ratio in Boussinesq approximation for double-diffusive convection in binary mixture

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ABSTRACT

This paper deals with a mathematical and numerical investigation of double-diffusive natural convective heat and mass transfer in a cavity filled with Newtonian fluid with significant density and mass diffusivity changes. In such a situation, the assumption of the Boussinesq approximation is not justified, and an appropriate model based on a set of Low Mach Number equations is used. The active parts of two vertical walls of the cavity are maintained at fixed but different temperatures and concentrations, while the other two walls, as well as inactive areas of the sidewalls, are considered to be adiabatic and impermeable to mass transfer. The coupled momentum, energy, and solute transfer equations in binary mixtures of ideal gases are solved through a global iterative procedure based on the finite volume methods in the context of the low Mach number approximation. The study includes the effect of the buoyancy ratio N with the aim to find its application limit in the Boussinesq conditions. The results show that if we use the Boussinesq approximation to study double-diffusive convection, the value of parameter N must be between -6 and 27.

t

time, s

NOMENCLATURE

C c _p D g H	concentration, kg/kg specific heat binary mass diffusion coefficient, m ² /s gravitational acceleration, m/s ² height of the enclosure, m	T U, V u, v W X, Y	temperature, K dimensionless velocity-components velocity-components, <i>m/s</i> mass fraction dimensionless coordinates
k Le = α/D	thermal conductivity, W/m K Lewis number	GREEK SYMBO	LS
$M M^{*} M^{*}$ $n N^{*} N^{*$	molecular weight, kg/kmol molecular weight ratio, $M^* = M_1/M_2$ unit outward normal to surface <i>dS</i> Buoyancy ratio convective Nusselt number dynamic pressure, Pa mean thermodynamic pressure, Pa Prandtl number Rayleigh number local Sherwood number	$lpha=k/ ho.c_p$ eta $arepsilon_T$ $arepsilon_T$ $arepsilon_m$ $ u$ $\Pi=(p-ar p+ ho_0gy)$	thermal diffusivity thermal expansion coefficient, K^{-1} non-Boussinesq thermal parameter $\varepsilon_T = \Delta T/2T_0$ non-Boussinesq solutal param- eter $\varepsilon_m = \frac{\Delta C_1(1-M^*)}{M^* + C_{10}(1-M^*)}$ kinematic viscosity, m^2/s $)/\rho_0(\alpha_0/H)^2$ dimensionless dynamic pressure

μ	dynamic viscosity, kg m ⁻¹ s ⁻¹
ρ	fluid density, kg m $^{-3}$
τ	dimensionless time

SUBSCRIPTS

avg	average value
BS	Boussinesq model
с	cold
CP	constant properties
h	hot
max	maximum value
mid	midplane
min	minimum value
0	reference state
1	for gas
2	for air
*	dimensionless parameters

I. INTRODUCTION

Compressible low-Mach number flows of double diffusive convection appear frequently in nature and industrial processes such as atmospheric and oceanographic flows, combustion, vapor deposition, and natural gas storage. The compressible low-Mach number flow $(Ma \ll 1)$ is defined as such when pressure variations are small, but temperature/concentration (and hence density) gradients are large.

Natural convection in enclosures filled with binary mixtures due to combined thermal and solutal body forces has been investigated extensively both experimentally and numerically and has received increasing attention. Over the past decades, a considerable body of research has thus been conducted in this field, both experimentally and numerically.^{1–5}

The greatest portion of the more pertinent binary gas literature has focused on analytical and numerical investigations of natural convection due to horizontal temperature and concentration gradients. $^{6-12}$

With the exception of some works using the non-Boussinesq model,^{13–17} it appears that the Boussinesq approximation was used in both cases of aiding and opposing buoyancy forces, which states that variations of density in the equations of motion can safely be ignored everywhere except in its association with the external force. The approximation is well justified in the case of incompressible fluids.

Ferziger and Peric¹⁸ assert that the use of the Boussinesq approximation in the Navier–Stokes equations introduces errors of the order of 1% if the temperature differences are less than 2 °C for water. These errors are led to augment for more important difference of temperature and can lead to qualitatively wrong resolutions.

The effect of buoyancy ratio on the flow structure was investigated numerically for a binary mixture gas in a rectangular enclosure by many authors.^{19–23}

In most of the works studying double-diffusive convection, a wide range of the buoyancy ratio N was considered. The question that arises is whether the buoyancy ratio N has a limit when using the Boussinesq approximation. Such a ratio N denotes the relative strengths of the thermal and solutal buoyancy forces, and it is based on the concentration and temperature differences between left and right vertical walls. In this study, we will try to answer this question by

taking into account the variation of the thermophysical properties of the studied fluid.

When the fluids are compressible, the equations governing the system become quite complicated. In such situations, all assumptions used to justify the Boussinesq approximation fail and a different modeling approach is required, one that accounts for realistic nonlinear fluid property variations. The Low Mach Number approximation (Ma \ll 1) suggested by Paolucci²⁴ was developed in our previous works for natural convection; natural convection coupled with surface radiation and combined natural convection-volumetric radiation.^{25–28} This approach is used in this study to solve the momentum, heat, and solute equations, which lead to take into account fluid flows under large temperature/concentration gradients and variable transport coefficients.

The purpose of the present paper is twofold: the first is to develop a mathematical model describing the double diffusive convection under non-Boussinesq conditions and the other is to investigate the limit of the buoyancy ratio of the double-diffusive convection in vertical enclosures under the Boussinesq approximation.

II. MATHEMATICAL FORMULATION

The aim of this section is to provide the governing equations to be solved by the low-Mach number solver for the double-diffusive convection. The low-Mach number approximation is carried out, enabling the full low-Mach number system of equations to be written.

The geometry investigated as shown in Fig. 1 is an enclosure filled with a gas mixture in which natural convection is driven by thermal and solutal gradients. The left wall is kept at high temperature (T_h) and high concentration (C_{1h}) , the right wall at low temperature (T_c) and low concentration (C_{1c}) and the bottom and top walls are adiabatic and impermeable. No-slip conditions are applied on all boundaries. The kinematic viscosity, the thermal diffusivity, and the concentration diffusivity are determined by the properties of this binary gas mixture.

A. Governing equations

Since we are interested in the case of large thermal and solutal gradients, the low-Mach-number equations are adopted^{24,27} in order to describe such a flow.



FIG. 1. The flow configuration and coordinate system.

Stemming from the compressible Navier–Stokes equations, the Low Mach Number (LMN) approximation constitutes an important numerical problem for low speed compressible flows and has the advantage that it presents the same mathematical structure as the incompressible Navier–Stokes equations.

In this approximation, and as shown in Ref. 24, the total pressure is divided into two components: a mean thermodynamic pressure $\bar{p}(t)$ which is spatially uniform and depends only on time and a dynamic pressure p_{dyn} . Since for Low Mach number (Ma \ll 1), the thermodynamic pressure is very high compared to the dynamic pressure, and this decomposition leads to elimination of the acoustic waves while large variations of density with temperature and concentration are allowed.

The conservation equations of mass, momentum, energy, and species for laminar double-diffusive convection in the horizontal cavity can be expressed as follows:

$$\frac{\partial \rho}{\partial t} + \nabla . \rho V = 0, \tag{1}$$

$$\rho\left(\frac{\partial \boldsymbol{V}}{\partial t} + \boldsymbol{V}\nabla\boldsymbol{V}\right) = \rho \boldsymbol{g} - \nabla p_{dyn} + \nabla .\mu \nabla \boldsymbol{V} + \frac{1}{3} \nabla \mu \nabla .\boldsymbol{V}, \quad (2)$$

$$\rho c_p \left(\frac{\partial T}{\partial t} + \mathbf{V} \nabla T \right) = \left(\nabla . k \nabla T \right) + \frac{d\bar{p}}{dt}, \tag{3}$$

$$\rho\left(\frac{\partial C}{\partial t} + V\nabla C\right) = (\nabla . \rho D\nabla C),\tag{4}$$

V is the velocity vector of components (u, v).

The system of conservation equations is completed by the ideal gas law used in order to determine the mixture density field ρ for a binary mixture of gases with molecular weights M_1 and M_2 ,

$$\bar{p} = \sum_{i=1}^{N} \bar{p}_i = \frac{nRT}{\Omega} = \frac{RT}{\Omega} \sum_{i=1}^{N} n_i = \frac{RT}{\Omega} \sum_{i=1}^{N} \frac{m_i}{M_i},$$
(5)

 M_i : molecular weight of the ith component. m_i : mass of component (i). n_i : number of moles of component (i). With: m = n.M total mass and $\rho = \frac{m}{\Omega}$ mix density.

We therefore have

$$\bar{p} = \frac{RT}{\Omega} \sum_{i=1}^{N} \frac{m_i}{M_i} = \frac{\rho RT}{m} \sum_{i=1}^{N} \frac{m_i}{M_i},$$

$$\bar{p} = \frac{RT}{\Omega} \sum_{i=1}^{N} \frac{m_i}{M_i} = \rho RT \sum_{i=1}^{N} \frac{\left(\frac{m_i}{m}\right)}{M_i} = \rho RT \sum_{i=1}^{N} \frac{C_i}{M_i},$$
(6)

 C_i is the concentration, mass fraction.

Since we only have two constituents (air and an other gas), we will have:

$$\bar{p} = \rho RT \sum_{i=1}^{N} \frac{C_i}{M_i} = \rho RT \left(\frac{C_1}{M_1} + \frac{C_2}{M_2} \right) = \rho \frac{R}{M_1} T \left(C_1 + \frac{M_1}{M_2} C_2 \right)$$
$$= \rho \frac{R}{M_1} T \left(C_1 + \frac{M_1}{M_2} (1 - C_1) \right).$$

Therefore,

$$\bar{p} = \rho R^* T \big(C_1 (1 - M^*) + M^* \big). \tag{7}$$

with

$$R^* = \frac{R}{M_1}$$
 and $M^* = \frac{M_1}{M_2}$.

It is also assumed that the dynamic viscosity is, depending on the case studied, taken either constant $\mu(T) = \mu(T_0) = \mu_0$ or given by the Sutherland law²⁹

$$\frac{\mu(T)}{\mu_0} = \left(\frac{T}{T_0}\right)^{3/2} \frac{T_0 + S_\mu}{T + S_\mu},\tag{8}$$

where $S_{\mu} = 110.5$ K, $\mu_0 = 1.68 \times 10^{-5}$ kg m⁻¹ s⁻¹.³⁰ Thermal conductivity is given by

$$k(T) = \mu(T) \frac{c_{p0}k_0}{\nu_0}.$$
(9)

More generally, the diffusion coefficient of gases into the air can be approximated by the following formula:³¹

$$D = D_0 \frac{p_0}{\bar{p}} \left(\frac{T}{T_0}\right)^{1.685}.$$
 (10)

The non-slip boundary conditions are imposed over the walls, so the boundary and initial conditions can be expressed as

$$u = v = T = C_1 = 0 \text{ at } t = 0,$$

$$u = v = 0, T = T_h, C_1 = C_{1h} \text{ for } 0 \le y \le H \text{ and } x = 0,$$

$$u = v = 0, T = T_c, C_1 = C_{1c} \text{ for } 0 \le y \le H \text{ and } x = H,$$

$$u = v = 0, \frac{\partial T}{\partial y} = 0, \frac{\partial C_1}{\partial y} = 0 \text{ for } 0 \le x \le H \text{ and } y = 0,$$

$$u = v = 0, \frac{\partial T}{\partial y} = 0, \frac{\partial C_1}{\partial y} = 0 \text{ for } 0 \le x \le H \text{ and } y = H.$$

B. Additional equations

In the numerical resolution of Eqs. (1)–(5), the problem that arises is that there are more unknowns than equations (at 2D, there are seven unknowns). This comes from the decomposition of the pressure, which introduces the thermodynamic pressure \bar{p} as an additional unknown. In order to close the problem, we need an additional equation representing the initial state, that of the conservation of the mass.

1. Calculation of \bar{p}

In an open cavity, the thermodynamic pressure is the ambient pressure

$$\bar{p}(t) = p_0, \forall t. \tag{11}$$

In a closed cavity, it is determined by the conservation of the total mass

$$\int_{\Omega} \rho d\Omega = M_0 = \int_{\Omega} \rho_0 d\Omega , \ \forall t.$$
 (12)

 Ω is the volume of the system and M_0 is defined as the initial mass of the system.

Using Eq. (7), we get

$$\int_{\Omega} \frac{\bar{p}}{R^* T(C_1(1-M^*)+M^*)} d\Omega = \int_{\Omega} \frac{p_0}{R^* T_0(C_{10}(1-M^*)+M^*)} d\Omega.$$
(13)

Therefore,

$$\bar{p} = p_0 \frac{\int_{\Omega} \frac{1}{T_0(M^* + C_{10}(1 - M^*))} d\Omega}{\int_{\Omega} \frac{1}{T(M^* + C_1(1 - M^*))} d\Omega}.$$
(14)

2. Calculation of $\frac{d\bar{P}}{dt}$

The second additional equation is given by the calculation of the term $d\bar{p}/dt$. The mass conservation equation, associated with the energy equation and the equation of state, allows writing the divergence of the velocity. According to Eq. (7), we have

$$\frac{d\rho}{dt} = \frac{1}{R^*T(M^* + C_1(1 - M^*))} \frac{d\bar{p}}{dt} - \frac{\bar{p}}{R^*T^2(M^* + C_1(1 - M^*))} \frac{dT}{dt} - \frac{\bar{p}(1 - M^*)}{R^*T(M^* + C_1(1 - M^*))^2} \frac{dC_1}{dt}.$$
(15)

From equations of continuity, energy, and concentration, we get

$$\nabla . \boldsymbol{V} = -\frac{1}{\rho} \frac{d\rho}{dt},\tag{16}$$

$$\frac{dT}{dt} = \frac{1}{\rho c_p} \nabla .k \nabla T + \frac{1}{\rho c_p} \frac{d\bar{p}}{dt},$$
(17)

$$\frac{dC_1}{dt} = \frac{1}{\rho} \nabla . \rho D \nabla C_1, \qquad (18)$$

with $c_p = \frac{\gamma R^*}{\gamma - 1}$, So,

$$\nabla \cdot \mathbf{V} = -\frac{1}{\rho R^* T (M^* + C_1 (1 - M^*))} \frac{d\bar{p}}{dt} + \frac{\bar{p}}{\rho R^* T^2 (M^* + C_1 (1 - M^*))} \frac{dT}{dt} + \frac{\bar{p} (1 - M^*)}{\rho R^* T (M^* + C_1 (1 - M^*))^2} \frac{dC_1}{dt}.$$
(19)

Taking into account the equation of state given by (7), Eq. $\left(19\right)$ becomes then:

$$\nabla \cdot \boldsymbol{V} = \frac{1}{\bar{p}} \left[\left(\frac{\gamma - 1}{\gamma} (M^* + C_1 (1 - M^*)) - 1 \right) \frac{d\bar{p}}{dt} + \frac{\gamma - 1}{\gamma} \times (M^* + C_1 (1 - M^*)) \nabla \cdot k \nabla T + R^* T (1 - M^*) \nabla \cdot \rho D \nabla C_1 \right].$$
(20)

For impermeable wall conditions,

$$\int_{\Omega^*} \nabla . V d\Omega = \int_S V . \mathbf{n} . dS = 0.$$
⁽²¹⁾

So,

$$\int_{\Omega} \nabla \cdot \boldsymbol{V} d\Omega = \int_{\Omega} \frac{1}{\bar{p}} \left[\left(\frac{\gamma - 1}{\gamma} (M^* + C_1 (1 - M^*)) - 1 \right) \frac{d\bar{p}}{dt} + \frac{\gamma - 1}{\gamma} \times (M^* + C_1 (1 - M^*)) \nabla \cdot k \nabla T + R^* T (1 - M^*) \nabla \cdot \rho D \nabla C_1 \right]$$
(22)

 \bar{p} depends only on time. The application of the divergence theorem gives

$$\frac{d\bar{p}}{dt} = \frac{1}{\int_{\Omega} d\Omega - \int_{\Omega} \frac{\gamma - 1}{\gamma} (M^* + C_1(1 - M^*)) d\Omega} \times \left[\frac{\gamma - 1}{\gamma} (M^* + C_1(1 - M^*)) \int_{S} k \frac{\partial T}{\partial n} dS + \frac{c^2}{\gamma} (1 - M^*) \right] \times \left[\int_{S} \rho D \frac{\partial C_1}{\partial n} dS \right],$$
(23)

where the speed of sound in an ideal gas *c* is given by: $c = \sqrt{\gamma R^* T}$.

C. Dimensionless form of equations

The following parameters are used to put the equations in dimensionless form:

$$\tau = \frac{\alpha t}{H^2}, X = \frac{x}{H}, Y = \frac{y}{H}, U = \frac{uH}{\alpha}, V = \frac{vH}{\alpha},$$
$$\Pi = (p - \bar{p} + \rho_0 g y) / \rho_0 (\alpha_0 / H)^2, \ \bar{P} = \frac{\bar{p}}{p_0}, \ \theta = \frac{T - T_0}{\Delta T},$$
$$W = \frac{C_1 - C_{10}}{\Delta C_1} \text{ or } \Delta T = T_h - T_c, \ T_0 = \frac{T_h + T_c}{2} \text{ and}$$
$$\Delta C_1 = C_{1h} - C_{1c}, \ C_{10} = \frac{C_{1h} + C_{1c}}{2}.$$

The thermo-physical properties (density, dynamic viscosity, thermal conductivity, thermal diffusivity, and mass diffusivity) are scaled, respectively, by $\rho_0, \mu_0, k_0, \alpha_0, D_0$ where the subscript 0 denotes values at the reference temperature T_0 . The influence of the temperature on the specific heat is assumed to be negligible so that $c_p/c_{p0} = 1$.²⁴

The system of equations is expressed in non-dimensional conservative form^{27} as

$$\frac{\partial \rho^*}{\partial \tau} + \frac{\partial \rho^* U}{\partial X} + \frac{\partial \rho^* V}{\partial Y} = 0, \qquad (24)$$

$$\rho^* \left(\frac{\partial U}{\partial \tau} + U \frac{\partial U}{\partial X} + V \frac{\partial U}{\partial Y} \right) = -\frac{\partial \Pi}{\partial X} + \Pr \left\{ \frac{\partial}{\partial X} \left(\mu^* \frac{\partial U}{\partial X} \right) + \frac{\partial}{\partial Y} \left(\mu^* \frac{\partial U}{\partial Y} \right) + \frac{1}{3} \nabla \mu^* \nabla . V^* \right\},$$
(25)

$$\rho^* \left(\frac{\partial V}{\partial \tau} + U \frac{\partial V}{\partial X} + V \frac{\partial V}{\partial Y} \right) = -\frac{\partial \Pi}{\partial Y} - Ra \Pr \frac{\rho^* - 1}{2\varepsilon_T} + \Pr \left\{ \frac{\partial}{\partial X} \left(\mu^* \frac{\partial V}{\partial X} \right) + \frac{\partial}{\partial Y} \left(\mu^* \frac{\partial V}{\partial Y} \right) + \frac{1}{3} \nabla \mu^* \nabla . V^* \right\},$$
(26)

$$\rho^* \left(\frac{\partial \theta}{\partial \tau} + U \frac{\partial \theta}{\partial X} + V \frac{\partial \theta}{\partial Y} \right) = \frac{\partial}{\partial X} \left(k^* \frac{\partial \theta}{\partial X} \right) + \frac{\partial}{\partial Y} \left(k^* \frac{\partial \theta}{\partial Y} \right) + \frac{\gamma - 1}{2 \varepsilon_T \gamma} \frac{d\bar{P}}{d\tau}, \qquad (27)$$

$$\rho^* \left(\frac{\partial W}{\partial t} + U \frac{\partial W}{\partial X} + V \frac{\partial W}{\partial Y} \right) = \frac{1}{Le} \left(\frac{\partial}{\partial X} \left(\rho^* D^* \frac{\partial W}{\partial X} \right) + \frac{\partial}{\partial Y} \left(\rho^* D^* \frac{\partial W}{\partial Y} \right) \right),$$
(28)

$$\bar{p} = \frac{\int_{\Omega} d\Omega^*}{\int_{\Omega^*} \frac{1}{(2\varepsilon_T \theta + 1)(1 + \varepsilon_m W)} d\Omega^*},$$
(29)

$$\rho^* = \frac{\bar{p}}{(1 + 2\varepsilon_T \theta)(1 + \varepsilon_m W)},$$
(30)
1

$$\frac{d\bar{P}}{d\tau} = \frac{1}{\int_{\Omega^*} \frac{d\Omega^*}{M^* + C_{10}(1 - M^*)} - \int_{\Omega^*} \frac{\gamma - 1}{\gamma} (1 + \varepsilon_m W) d\Omega^*} \times \left[2\varepsilon_T (1 + \varepsilon_m W) \int_{S^*} k^* \frac{\partial \theta}{\partial n} dS^* + \frac{\varepsilon_m}{Le} \int_{S^*} \rho^* D^* \frac{\partial W}{\partial n} dS^* \right],$$
(31)



FIG. 2. Evolution of (a) *N* and (b) ε_m as a function of $M^*(N)$.

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TABLE I.	Air benchmark	test	cases. ^{33,34}

	Heuveline ³³	Vierendeels ³⁴	Presen	Present work					
Test case 1 (Ra = 10^6 , $\varepsilon_h = 0.6$, and constant properties)									
Nu(h)	8.859 778	8.85 978	8.85 847						
Nu(c)	8.85 978	8.85 978	8.85 847						
\bar{p}/p_0	0.85 634	0.856 340	0.85 584						
Test case 2 ($Ra = 10^6$, a	$x_b = 0.6$, and Sutherland law)								
Nu(h)	8.6889	8.6866	8.70 078	8.68 451					
Nu(c)	8.6831	8.6866	8.70 057	8.68 316					
\bar{p}/p_0	0.9249	0.924 489	0.92 498	0.92 425					
Type of Mesh	400 000	2048×2048	300 imes 300	202 imes 202					

TABLE II. Buoyancy ratio parameter as a function of M^* and ε_m
--

Ν	5	2	1	0.5	0.1	0	-0.1	-0.5	-1	-2	-5
M^*	0.524	0.744	0.855	0.922	0.984	1	1.017	1.09	1.196	1.476	4.333
ε_m	0.167	0.0667	0.0333	0.0167	0.00333	0	-0.0033	-0.0167	-0.0333	-0.0667	-0.167

M^*	1.00167	1.017	1.0344	1.09	1.1515	1.1734	1.196	1.3226	4.3333
ε _m	-0.0003	-0.0033	-0.0067	-0.0167	0.0267	-0.03	-0.033	-0.05	-0.167
Ν	-0.01	-0.1	-0.2	-0.5	-0.8	-0.9	-1	-1.5	-5.0
Nu(Sh) present study	16.465	16.049	15.55	13.69	10.67	8.81	2.88	13.69	23.86
35	16.4	16.0	15.5	13.6	10.6	8.8		13.6	23.7
20	16.3	15.9	15.4	13.5	10.5	8.6		13.5	13.6
36	16.7	16.0	15.3	13.6	10.6	8.8		13.5	23.7

TABLE III. Average Nusselt and Sherwood numbers along the hot wall, Le = 1, Pr = 0.71, and $Ra = 10^7$.

where $\varepsilon_m = \frac{\Delta C_1(1-M^*)}{M^*+C_{10}(1-M^*)}$ and $\varepsilon_T = \frac{\Delta T}{2T_0}$. Note that

$$M^* \to 0, \ \varepsilon_m \to \frac{\Delta C_1}{C_{10}},$$
 (32)

$$M^* \to \infty, \ \varepsilon_m \to \frac{\Delta C_1}{C_{10} - 1}.$$
 (33)

 $M^{*}
ightarrow 0, \, arepsilon_{m}
ightarrow 2,$ (34)

$$M^* \to \infty, \ \varepsilon_m \to \frac{\Delta C_1}{\frac{\Delta C_1}{2} - 1}.$$
 (35)

The corresponding boundary conditions are

$$U = V = 0, \ \theta = \theta_c = 0.5, \ W = W_c = 0.5 \ \text{at } X = 0, \ 0 \le Y \le 1,$$

$$U = V = 0, \ \theta = \theta_h = -0.5, \ W = W_h = -0.5 \ \text{at} \ X = 1, \ 0 \le Y \le 1$$



FIG. 3. Distributions of local Nusselt numbers on the hot wall in (a) present study and (b)³⁵ at $Ra = 10^7$ and Le = 1.

$$U = V = 0, -\frac{\partial\theta}{\partial Y} = 0, -\frac{\partial W}{\partial Y} = 0 \text{ at } Y = 0, \ 0 \le X \le 1,$$
$$U = V = 0, \ \frac{\partial\theta}{\partial Y} = 0, \ \frac{\partial W}{\partial Y} = 0, \text{ at } Y = 1, \ 0 \le X \le 1.$$

The transport coefficients $\mu^*(\theta)$ and $k^*(\theta)$ are given, respectively, by

$$\mu^* = (2\varepsilon_T \theta + 1)^{3/2} \frac{(1 + S_\mu/T_0)}{2\varepsilon_T \theta + 1 + S_\mu/T_0},$$
(36)

$$k^* = \frac{\mu^*}{\Pr}.$$
 (37)

The mass diffusion coefficient is given by

$$D^* = \frac{1}{\bar{P}} (2\varepsilon_T \theta + 1)^{1.685}.$$
 (38)



FIG. 4. Distributions of local Sherwood numbers on the hot wall in (a) present study and (b)³⁵ at $Ra = 10^7$ and N = -0.1.

To characterize heat and mass transfer of the double-diffusive convection in the cavity, the local Nusselt number Nu and Sherwood number *Sh* on the heated vertical surface are defined by

$$Nu = -\frac{\partial \theta}{\partial X}\Big|_{X=0},\tag{39}$$

$$Sh = -\rho^* D^* \frac{\partial W}{\partial X} \bigg|_{X=0}.$$
 (40)

The average convective Nusselt number and Sherwood number are calculated by integrating the temperature and concentration gradient over the vertical wall as

$$Nu_{c_{avg}} = \int_0^1 Nu(X) dX,$$
(41)

$$Sh_{avg} = \int_0^1 Sh(X) dX. \tag{42}$$

D. Boussinesq approximation recovery

Boussinesq's approximation is characterized by a small difference in temperature and concentration ($\varepsilon_T \rightarrow 0$, $\varepsilon_m \rightarrow 0$), a zero velocity divergence ($\nabla . \vec{V} = 0$), and thermodynamic pressure tending toward unity ($\bar{P} = 1$).

If we hold these hypotheses, Eq. (39) becomes

$$\rho^* = \frac{1}{(1 + 2\varepsilon_T \theta)(1 + \varepsilon_m W)} \bigg|_{\substack{\varepsilon_T \to 0 \\ \varepsilon_m \to 0}}.$$
(43)

The term $\frac{\rho^*-1}{2\varepsilon_T}$ in Eq. (35) becomes

$$\frac{\rho^* - 1}{2\varepsilon_T} = -\frac{\varepsilon_m W}{2\varepsilon_T} - \theta - \varepsilon_m \theta W.$$
(44)

So, for $(\varepsilon_T \rightarrow 0, \varepsilon_m \rightarrow 0)$



FIG. 5. Near-wall refinement for grid using exponential distribution.

FIG. 6. Rate of change of (a) ρ^* and (b) D^* as a function of M^* (N) at $Ra = 10^6$ and Le = 1.

$$\frac{\rho^* - 1}{2\varepsilon_T} \to -\frac{\varepsilon_m W}{2\varepsilon_T} - \theta, \tag{45}$$

where $\frac{\varepsilon_m}{2\varepsilon_T}$ tend toward a value *N* (positive or negative, depending on the sign of ε_m) and which represents buoyancy ratio $N = \frac{\beta_1 \Delta C_1}{\beta_1 \Delta T} = \frac{\varepsilon_m}{2\varepsilon_T}$, and where

$$\beta_s = \frac{(1-M^*)}{M^* + C_{10}(1-M^*)}$$
 and $\beta_t = \frac{1}{T_0}$.

 ∂

We get

$$\frac{\partial U}{\partial X} + \frac{\partial V}{\partial Y} = 0, \tag{46}$$

$$\frac{\partial U}{\partial \tau} + U \frac{\partial U}{\partial X} + V \frac{\partial U}{\partial Y} = -\frac{\partial P}{\partial X} + \Pr \nabla U, \qquad (47)$$

$$\frac{\partial V}{\partial \tau} + U \frac{\partial V}{\partial X} + V \frac{\partial V}{\partial Y} = -\frac{\partial P}{\partial Y} + \Pr \nabla V + Ra \Pr(\theta + NW), \quad (48)$$

$$\frac{\partial\theta}{\partial\tau} + U\frac{\partial\theta}{\partial X} + V\frac{\partial\theta}{\partial Y} = \nabla\theta, \tag{49}$$

$$\frac{\partial W}{\partial t} + U \frac{\partial W}{\partial X} + V \frac{\partial W}{\partial Y} = \frac{1}{Le} \Delta W.$$
(50)

III. NUMERICAL SIMULATION

The numerical solution of the governing differential equations for the velocity, pressure, and temperature fields is obtained by using a finite volume technique. A power scheme was also used in approximating advection-diffusion terms. The SIMPLER algorithm whose details can be found in Patankar³² with a staggered grid is employed to solve the coupling between pressure and velocity. The discretization equations were solved by the Gauss-Seidel method. The iteration method used in this program is a line-by-line procedure, which is a combination of the direct method and the resulting Tri Diagonal Matrix Algorithm (TDMA).

IV. VALIDATION

In order to check on the accuracy of the numerical technique employed for the solution of the problem considered in the present study, we have first validated our numerical code to ensure that it delivers correct results. The code has been verified by comparing our results with those available in the literature.^{20,3} Results for the problem of the pure natural convection for a large temperature gradients and for the problem of the double-diffusive convection for a low temperature and solutal concentration gradients are compared with those obtained by 33,34 and, 24 respectively. For the case of thermal convection for large

FIG. 7. Variation of the thermodynamic pressure \overline{P} as a function of M^* (N) at $Ra = 10^{6}$ and Le = 1.

FIG. 8. Maximum and minimum (a) horizontal and (b) vertical velocities as a function of $M^*(N)$ at $Ra = 10^6$ and Le = 1.

temperature gradients, calculations were carried out for air initially at temperature $T_0 = 600$ K and for $Ra = 10^6$ and $\varepsilon_T = 0.6$. The test cases include problems with constant and variable properties. Data presented in Table I, including average Nusselt numbers at the active walls and the pressure ratio \bar{p}/p_0 , show a good agreement with results of Heuveline³³ and Vierendeels.³⁴

For double-diffusive convection, we used specified values of the molecular weight ratio M^* , which correspond to values of buoyancy ratio N widely used in the literature. We also used weak thermal and solutal gradients to obtain quantitative bounds for the validity of the Boussinesq approximation.

Table II shows the values of buoyancy ratio $N = \frac{\varepsilon_m}{2\varepsilon_T}$ corresponding to the different values of molecular weight ratio M^* and non-Boussinesq solutal parameter ε_m .

The quantities N and ε_m are inversely proportional to M^* and following a hyperbolic trajectory (Fig. 2), so that

$$M^* \to 0, \ \varepsilon_m \to \frac{\Delta C}{C_0} = 2 \text{ or}; \ N = \frac{\varepsilon_m}{2\varepsilon_T} = 60,$$

 $M^* \to \infty, \ \varepsilon_m \to \frac{\Delta C}{C_0 - 1} = -0.222 \text{ or}; \ N = \frac{\varepsilon_m}{2\varepsilon_T} = -6.66.$

Table III shows the values of average Nusselt and Sherwood numbers along the hot wall for Le = 1, Pr = 0.71, and $Ra = 10^7$. Figures 3 and 4 show the distributions of the local Nusselt (*Nu*) and Sherwood (*Sh*) numbers on the hot wall for over a range for buoyancy ratio *N* and for Le = 1, Pr = 0.71, and $Ra = 10^7$.

It can be seen from the comparison that the solution found by the low Mach number approximation approaches the Boussinesq

FIG. 9. Variation of the maximum and minimum (a) density and (b) mass diffusivity as a function of M* (N) at Ra = 10⁶ and Le = 1.

solution, the present work is in a very good agreement with the previous works. Note that for the case N = -1 ($M^* = 1.196$) a slight deviation between the results is observed, which means that the non-Boussinesq effect begins to affect the flow. In this special case, the thermal and solutal buoyancy forces are canceled. Thereby, the convection motion disappears, and the heat and mass transfers are induced only by diffusion.

V. RESULTS AND DISCUSSION

This section is devoted to find a limit value for the buoyancy ratioN in the Boussinesq approximation for natural convection flows evolving inside a close and differentially heated cavity, which is filled with an air-gas mixture, both for assisting and opposing flows depending on the molecular weight ratio parameter M^* .

The grid independence test is carried out to examine the dependence of numerical accuracy on grid sizes. The distribution of grids in the near walls was done using an exponential bunching methodology chosen in such a way that there is a greater mesh density near the wall as shown in Fig. 5. It turns out that the solution becomes independent of grid size at 202×202 ; this is a best trade-off between the accuracy and the computation time

Therefore, Prandtl (Pr), Rayleigh (*Ra*), Lewis (*Le*) numbers, non-Boussinesq thermal parameter ε_T , and concentration gradient (ΔC) are fixed to Pr = 0.71, *Ra* = 10⁶, *Le* = 1, ε_T = 0.017(ΔT = 10 K), and ΔC = 0.2 (20%, C_0 = 0.1), respectively. The controlled variables are the molecular weight ratio $0 \le M^* \le 20$, which corresponds to non-Boussinesq solutal parameter ε_m , $-0.21 \le \varepsilon_m \le 1.07$ and to buoyancy ratio *N*, $-6.3 \le N \le 32$. The thermophysical properties are estimated at a reference temperature of 300 K.

It should be noted that for Boussinesq conditions, the solutions remain symmetric, and consequently we always have the following results: $\rho_{\max}^* \approx \rho_{\min}^*$, $D_{\max}^* \approx D_{\min}^*$, $U_{\max} \approx U_{\min}$, $V_{\max} \approx V_{\min}$, $\bar{P} \approx 1$, $Nu \approx Sh$ for Le = 1, and also isotherms merge with isoconcentrations.

Depending on the molecular weight ratio, M^* ($M^* \succ 1$, $M^* \prec 1$), the definition of ε_m implies negative values of ε_m when the molecular weight of the gas which diffuses into the cavity is higher than that of the inert gas ($M^* \succ 1$, opposing flows) and vice versa ($M^* \prec 1$, aiding flows). The solutal body force can either augment or oppose to the thermal body force.

The value Le = 1 was used for the Lewis number to yield a similar distribution of the Nusselt and Sherwood numbers, as well as for the distribution of isovalues of temperature and concentration, this is only true in the Boussinesq approximation.

It is considered that the Boussinesq approximation is valid as long as the temperature/solutal difference in the fluid does not modify the physical properties by more than 10% compared to their values at the reference temperature T_0^{37}

The results are presented in terms of streamlines, isotherms, local and average Nusselt *and* Sherwood numbers, and cross section of temperature and velocity profiles. To check the symmetry of the solution when the Boussinesq approximation is achieved, the results are also presented, as a function of molecular weight ratio M^* , in terms of rates of change of density $\Delta \rho^*$ and mass diffusivity ΔD^* , maximum and minimum horizontal ($U_{\text{max}}, U_{\text{min}}$) and vertical ($V_{\text{max}}, V_{\text{min}}$) velocities, maximum and minimum density ρ^* and mass diffusivity D^* , and thermodynamic pressure \bar{P} .

FIG. 10. Variation of the average Nusselt and Sherwood numbers at the heated surface as a function of $M^*(N)$ at $Ra = 10^6$ and Le = 1.

The deviation from the Boussinesq approximation is investigated according to the value of M^* in the range $0 \le M^* \le 20(-6.3 \le N \le 32)$ for a mass fraction at the walls $\Delta C = 20\%$. Whatever their values, the thermal conductivity and kinematic viscosity always remain constant and equal to their values given by Sutherland's law. The thermophysical properties which change by changing the parameter M^* are the density and the mass diffusivity.

According to Figs. 6(a) and 6(b), it can be seen that the effect of M^* on the variation in mass diffusivity is greater than that on density.

Figures 6(a) and 6(b) show the evolution of the change rate in percentage of the density and of the mass diffusivity compared to their values in the reference state. It can be seen that the effect of M^* on the variation in mass diffusivity is greater than that on density. The results show that for values of $M^* \leq 0.105$ ($N \geq 27$), the rate of variation of the density exceeds 10%, while the rate of change of mass diffusivity

FIG. 11. Local Nusselt and Sherwood numbers on the hot wall $M^*(N)$ at $Ra = 10^6$ and Le = 1.

FIG. 12. (a) Streamlines, (b) temperature, and (c) concentration contours as a function of $M^*(N)$ at $Ra = 10^6$ and Le = 1.

exceeds 10% for values of $M^* \leq 0.178(N \geq 18)$. For this range of variation of M^* , Fig. 7 shows a significant decrease in the thermodynamic pressure, varying up to 17% from its reference value due to a loss of a large number of molecules between the initial and steady-state, while it is almost constant and tends toward unity if $M^* \geq 0.5$. This indicates that in the context of the Boussinesq approximation, the ratio M^* must not be less than 0.105 corresponding to N = 27. For values of M^* less than 0.105 (i.e., for very light gases) compressibility has a considerable effect on the flow and the Boussinesq approximation is no longer valid and the low Mach number model must be used to account for the effects of compressibility.

For $M^* \leq 0.105$, a small decrease in M^* leads to a large increase in velocities (Fig. 8) and a large decrease in thermodynamic pressure because the number of molecules within the cavity becomes much small. On the other hand, the pressure does not vary much for $M^* \geq$ 0.35 ($N \leq 9.4$). A very important variation is also observed on the density and the mass diffusivity [Figs. 9(a) and 9(b)], compared to their reference values.

The local and average Nusselt and Sherwood numbers (Figs. 10 and 11) do not present a large deviation [2.66% for $M^* = 0.08$ (N = 32)].

When the Boussinesq approximation is invoked, the temperature [Fig. 12(b)] and concentration [Fig. 12(c)] fields merge, it is difficult to delineate between isotherms and isoconcentration, the energy and mass fluxes at the walls are the same (Fig. 10). This symmetry is also found on the streamlines [Fig. 12(a)] plotted for very weak compressible flows. The deviation of the thermodynamic pressure from its value in the Boussinesq approximation ($\bar{P} = 1$) is small (Fig. 7).

For values of M^* close to 1, results show that the steady-state Nu does not differ much from the Boussinesq value (Nu = 8.82, for the case of pure natural convection).²⁷ The variation in density [Figs. 6(a) and 9(a)] and mass diffusivity [Figs. 6(b) and 9(b)] with respect to their reference values is negligible. The predictions of incompressible and compressible models did not differ significantly, and the solutions remain symmetric. These results are clearly seen in Figs. 12–15 for iso-contours of streamlines, temperature and concentration, temperature at mid-height cavity, vertical velocity at the horizontal cross section,

FIG. 13. Cross section of the temperature at mid-height cavity at Y = 0.5 for $Ra = 10^6$ and Le = 1.

FIG. 14. Vertical velocity at the horizontal cross section at Y = 0.5 for $Ra = 10^6$ and Le = 1.

and horizontal velocity at the vertical cross section, respectively. It can be seen that the temperature (Fig. 13) and vertical velocity (Fig. 14) profiles are centro-symmetric with zero values at the cavity center. This symmetry is also found on the horizontal velocity (Fig. 15).

At this limit of M^* , $M^* \ge 0.105$, the variation in density and mass diffusivity [Figs. 9(a) and 9(b)] with respect to their reference values is no longer considerable. The maximum and minimum values of the horizontal and vertical velocities [Figs. 8(a) and 8(b)] approach increasingly indicating symmetry in their profiles.

For low M^* , higher temperature and concentration gradients are present in the region close to the lower part of the hot wall and top part of the cold wall as shown in Figs. 12(a) and 12(b) ($M^* = 0.2$ and 0.08).

For a considerable M^* , higher temperature and concentration gradients are observed near the top part of the hot wall and near the lower part of the cold wall [Figs. 12(a) and 12(b) for $M^* \ge 4.33$], which indicates higher heat and mass transfer by convection and diffusion, the velocity gradients become steeper at both vertical sides of the

FIG. 15. Horizontal velocity at the vertical cross section at X = 0.5 for $Ra = 10^6$ and Le = 1.

enclosure [Figs. 9(a) and 9(b) for $M^* \ge 4.33$ and $M^* \le 0.2$]. These higher gradients in temperature and concentration cannot be observed if we use the Boussinesq model.

For very low values of M^* , $M^* \ll 0.105$, The symmetry seen in the previous case is broken due to the non-linear density and mass diffusivity variations whose role is gradually strengthened as the flow becomes more compressible.

On the other hand, for values of $M^* \leq 0.178$ ($N \geq 18.95$), the rate of change of mass diffusivity exceeds 10%, [Fig. 6(b)], at which the thermodynamic pressure deviates by 8%, (Fig. 7), compared to its value considered in the Boussinesq approximation ($\bar{P} = 1$),the symmetrical property of the solution is lost by 8%.

VI. CONCLUSIONS

Mathematical and numerical modeling of double-diffusive natural convection in an enclosure region is considered. The mathematical model has been formulated using the low Mach number formulation.

Numerical solution of the governing differential equations for the velocity, pressure, temperature, and concentration fields is obtained by using a finite volume technique. A power scheme was also used in approximating advection–diffusion terms. The SIMPLER algorithm with a staggered grid is employed to solve the coupling between pressure and velocity.

For large values of M^* , the low Mach number approximation approaches the Boussinesq Model for incompressible flows. Therefore, the solution schemes are usually based on numerical methods developed for incompressible flows.

Validation tests on the model have shown that the numerical simulation results in Boussniesq approximation are in good agreement compared with the numerical data available in literature.

Results show that the Boussinesq approximation is not sufficient to simulate double-diffusive natural convective flow when the buoyancy ratio exceeds the values between -6 and 27 and the flow becomes more compressible.

The changes in the thermal and dynamic fields introduced by the non-Boussinesq formulation are considerably larger than those for the aiding case.

If we take into account the variation of the mass diffusivity, it is preferable to use values of N, which must not exceed N = 19.

Outside this range of *N*, all assumptions used to justify the Boussinesq approximation fail and a different modeling approach is required, one that accounts for realistic nonlinear fluid property variations.

It is interesting to note that there are a number of avenues which still remain to be explored or which have opened up during this work; we can mention the extension to the 3D modeling problem.

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DATA AVAILABILITY

The data that support the findings of this study are available within the article.

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