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**A 10,000-year record of trace metal and metalloid (Cu, Hg, Sb, Pb)
deposition in a western Alpine lake (Lake Robert, France):
deciphering local and regional mining contamination**

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Keywords: Mercury; Antimony; Lead isotopes; Copper; Alpine Lake; Holocene sediment
Record; Bronze Age Pollution

1 **Abstract**

2 Concentrations of trace metals and metalloids (Cu, Hg, Sb, Pb), major
3 elements, stable Pb and C isotope ratios, total organic carbon (TOC), C/N atomic
4 ratios, were analyzed in two sediment cores encompassing 10,000 years of
5 sedimentation sampled in Lake Robert (French Alps).

6 The results showed that the establishment of a soil cover and vegetation during
7 the Holocene climatic optimum increased Cu and Hg deposition in sediments.
8 Results also enabled reconstruction of the history and local vs regional origins of
9 pollution. During the Bronze Age (1800-1000 BC), Cu mining activities increased Hg
10 and Sb accumulation rates (AR) by a factor of 7 (Hg) and 9 (Sb) compared to pre-
11 Bronze Age values. The rise in Hg AR was equivalent to that of the industrial era
12 highlighting a major local Hg point source not recorded in regional archives. During
13 the Iron Age, Pb pollution was attributed to Sicilian ores whereas during the Roman
14 Empire Pb pollution had the isotopic signature of Spanish Pb ores. During the
15 Medieval period (ca. AD 1200), Pb and Hg pollution was due to the exploitation of the
16 local Ag-Pb mines. Finally, during the industrial era, metal accumulations rates were
17 90 (Pb), 5 (Hg) and 20 (Sb) times higher than those in the pre-Bronze Age. The
18 isotopic composition of Pb reflected mixed sources including industrial Pb and
19 gasoline Pb. Synchronously, the increase in Sb and Hg pollution is attributed to
20 global and regional atmospheric pollution.

21

22

1 **1 Introduction**

2

3 In their pioneering study of Pb levels in Arctic ice cores, Murozumi et al., (1969),
4 showed that the Northern Hemisphere has been affected by Pb pollution from Roman
5 to modern times. Since that study, long term (pre)-historical metal pollution (mainly
6 Pb, Hg and rarely Cu or other metals) has been identified in Europe in different
7 environmental archives including mires (García-Alix et al., 2013; Küttner et al., 2014;
8 Kylander et al., 2005; Martínez-Cortizas et al., 1999, 1997; Martínez Cortizas et al.,
9 2016; Monna et al., 2011, 2004; Pontevedra-Pombal et al., 2013; De Vleeschouwer
10 et al., 2009; Küttner et al., 2014; Brännvall et al., 1999, 1997; Höppner et al., 2005;
11 Krachler et al., 2008; Roos-Barracough and Shotyk, 2003; Shotyk et al., 2005,
12 2004), Posidonia mats (Serrano et al., 2011), estuary sediments (Leblanc et al.,
13 2000), lagoon sediments (Elbaz-Poulichet et al., 2011; Manteca et al., 2017), lake
14 sediments (Chapron et al., 2007; Cooke et al., 2011 and references therein; Renberg
15 et al., 2002, 1994; Thevenon et al., 2011; Schütze et al., 2018) and glacier ice
16 (Preunkert et al., 2019). Long term metal pollution has also been brought to light
17 through the study of lake sediment records in the Andes (Cooke et al., 2011;
18 Guédron et al., 2019) and China (Lee et al., 2008). The main difference between
19 trace metal and metalloid (TMM) records is their respective residence time in the
20 atmosphere, which ranges from approximately 6 months for Hg (Pirrone and Mason,
21 2009) to a few days or weeks for Pb, Sb and Cu (Krachler et al., 2005; Pacyna, 1987;
22 Sturges and Barrie, 1987), which can result in their dispersal throughout the
23 hemisphere to thousands of kilometers depending on the climatic context. The
24 intensity and location of anthropogenic activities also has a major influence on the

1 magnitude of TMM records in environmental archives, the highest signals were
2 recorded at the scale of the Northern Hemisphere during the industrial era.

3 In Europe, mining was already widespread in prehistoric and historic times
4 (Bindler et al., 2009) especially in the Alps where numerous relatively small copper
5 and polymetallic copper bearing deposits have been exploited since the late
6 Neolithic/Early Bronze Age in the Italian (Artioli et al., 2015), Austrian (Höppner et al.,
7 2005) and French (Bourgarit et al., 2008; Carozza, 2010) Alps. Despite the fact most
8 Alpine lake sediments can be used to reconstruct ancient metal pollution (Cooke and
9 Bindler, 2015), there are only few long term records of TMM in these lakes. Yet, as
10 pointed out by (Bindler et al., 2011), such studies are important to establish natural
11 sediment reference conditions for metals and the legacy of long range and local
12 pollution of lakes in Europe. Guyard et al., (2007) reported peak Pb and Cu
13 concentrations during the Roman period (AD 115-330) and Chalcolithic (3770-3870
14 BP) in sediment layers in Lake Bramant in the western French Alps that reflecting
15 both local and distant mining activities. Later in the same area, Garçon et al., (2012)
16 showed Pb and Ag accumulation peaks in the Lake Blanc d'Huez sediments during
17 the Medieval period, which these authors attributed to exploitation of the Pb-Ag vein
18 in the lake catchment. Thevenon et al., (2011) investigated variations in
19 concentrations of Cu, Cr, Zn, Pb, Hg in Lake Lucerne and Meidsee in the northern
20 Alps in sediment layers encompassing the last 14,000 years. Increased Hg and Pb
21 enrichment compared with crustal values and Pb isotopes provided evidence for the
22 impact of human activities on the lake sediment chemistry during the Chalcolithic
23 (e.g. 3800 BC). Anthropogenic trace metals increased during the Iron Age-Roman
24 period (ca 800 BC to AD 500), the Late Middle Ages (ca AD 1400), after AD 1600
25 and from the industrial revolution (AD 1850) on, the industrial revolution being

1 responsible for an exponential increase in metal accumulation in the lakes.

2 The present study focused on Lake Robert in the western French Alps. It was
3 designed to assess the baseline concentrations and accumulation rates of TMM (Cu,
4 Hg, Sb and Pb) and to decipher the impact of local and regional mining on a remote
5 high-altitude lake.

6 TMM concentrations and stable Pb isotopes were determined in the layers of
7 two sediment cores covering the entire Holocene. The cores were correlated and
8 dated using ^{14}C .

9 TMM deposition in lakes is controlled not only by anthropogenic inputs but also
10 by natural factors such as the influx and the sources of organic (OM)/mineral matter
11 (e.g. Schütze et al., 2018) that are dependent on climate-induced environmental
12 changes. In order to better understand variations in lake mineral sedimentation, dry
13 bulk density (DBD), major elements (Ca, Mg, Al, Ti, Si) and total organic carbon
14 (TOC) were determined. In addition, the total organic carbon to nitrogen (C/N) atomic
15 ratio and the C isotopic signature ($\delta^{13}\text{C}$) were used to differentiate the source of
16 organic matter in the lake sediments. Indeed, land-derived OM has a higher C/N ratio
17 than 14–20 and negative $\delta^{13}\text{C}$ signatures (-24 to -29‰), while phytoplankton and
18 aquatic macrophytes have lower C/N ratios (between 4 and 10) and less negative
19 $\delta^{13}\text{C}$ values [-12 to -17 ‰, (Meyers and Teranes, 2001)]. Both the variations of the
20 C/N can thus be used to decipher the autochthonous (in-lake production) and
21 allochthonous (catchment-derived) organic matter (Thevenon et al., 2012).

22 Water oxygenation is also an important factor controlling metal retention in lake
23 sediments. For example, anoxic conditions in the hypolimnion enhance Cu
24 scavenging by iron sulfides, whereas Cu can be released into solution during aerobic
25 degradation of OM when hypolimnetic oxygenation occurs (e.g. Achterberg et al.,

1 1997). In oxic water and sediments, Sb is adsorbed/coprecipitated to Fe-Mn
2 oxy(hydro)xides (Belzile et al., 2001). Sb can be released into solution during the
3 reduction of these oxides. In anoxic conditions, Sb sorbs to strongly insoluble Fe
4 sulfides (Chen et al., 2003). Contrary to Cu and Sb, Pb and Hg are generally not
5 subject to diagenetic mobility_after deposition (Feyte et al., 2012; Gallon et al., 2004;
6 Rydberg et al., 2008).

7 Because Mn is more rapidly reduced compared to Fe under anoxic conditions
8 leading to preferential release of Mn (Davison, 1993), higher Fe/Mn ratios indicate
9 lower O₂ concentrations in the water column, whereas lower ratios suggest higher O₂
10 levels as shown in many lake studies (Brüchmann and Negendank, 2004; Koinig et
11 al., 2003; Loizeau et al., 2001; Naeher et al., 2013; Sabatier et al., 2017). In this
12 study, the Fe/Mn ratio was used as a proxy of water oxygenation at the time of
13 sediment deposition.

14

15 **2 Study site**

16

17 The name Lake Robert in fact refers to a set of four endorheic lakes formed
18 after the last glacial maximum retreat in the French Alps at an altitude of 1998 m asl
19 (Fig.1). The lake is frozen from approximately mid-November to mid-May.
20 Fluctuations in the level of the lake caused by seasonal rainfall and snow melt mean
21 the two westernmost lakes are connected for part of the year. The total surface area
22 of the lake is about 0.11 km² and water depth ranges from around 4 m in the south-
23 western lake to 12 m in the north-western lakes in April. The Lake Robert watershed
24 is small (1 km²) and is mainly composed of serpentized pyroxenites, peridotites,
25 uralitized gabbros and euphotides, post Wurmian till, moraine and coarse scree.

1 There are no permanent streams and very limited local lithogenic inputs are delivered
2 to the lake sediments.

3 Old mines are found in the vicinity of the lakes including Challanches mine
4 located 6 km from the lake at 2000 m asl. (Fig.1). From the 18th to the beginning of
5 the 20th century the ore was mainly exploited for its richness in Ag but other TMM
6 were also present. In particular, the presence of numerous Sb containing minerals
7 such as stibnite (Sb_2S_3), cinnabar (HgS) and As, Cu and Ag bearing minerals (i.e.,
8 stibarsen (SbAs), dyscrasite (Ag_3Sb) and $\text{Cu}_6[\text{Cu}_4(\text{Fe,Zn})_2]\text{Sb}_4\text{S}_{13}$) was reported by
9 (Clavel, 1963; Lacroix, 1893). Old mines have been reported in the Rousses Massif
10 high-altitude (2700-3000 m asl.). The exploitation of Brandes argentiferous Pb mine
11 in this Massif (Fig.1) in the Middle Ages, from AD 1150–1200 to AD 1339, is well
12 documented in written archives (Bailly-Maitre, 1996). Several older mines that
13 exploited quartz veins mainly containing chalcopyrite or more rarely copper arsenides
14 and/or antimonides and of superficial formations of copper carbonates in the Bronze
15 Age (2200-1700 BC) have recently been discovered in the vicinity [Fig.1 (Bailly-
16 Maitre and Gonon, 2008; Moulin et al., 2010)].

17

18 **3 Methods**

19

20 **3.1 Sampling and sample preparation**

21 Two gravity cores ROB15-P5 (115 cm) and ROB15-P6 (110 cm) were collected
22 in the south-western lake (E 5° 54.823'; N 45° 07.968') using an UWITEC gravity
23 corer equipped with a hammer (Fig.1). The two cores were located less than 5 m
24 apart. A hole was drilled in the ice cover to recover sample cores from the sediment
25 at the bottom of the 12-m deep lake in April 2015. In the laboratory, the cores were

1 split, photographed, and logged in detail, i.e. all the physical sedimentary structures
2 and the vertical succession of facies were recorded. High resolution XRF analysis of
3 Al, Si, Ca, Ti, Mn, Fe, Cu and Pb was performed on the two cores using core
4 scanner. In addition, total organic carbon (TOC) concentrations, ^{13}C , ^{14}C , dry bulk
5 density (DBD), C/N and Hg were determined in core ROB15-P5 and Ca, DBD, Mg,
6 Al, Ti, Si, Fe, Mn, Cu, Sb, Pb and Pb isotopes in discrete samples from core ROB15-
7 P6. The later were chosen to include the major Pb and Cu peaks identified by XRF
8 analysis as well as baseline levels.

9

10 **3.2 Analysis of ^{14}C**

11 Accelerator mass spectrometers (AMS) were used to measure ^{14}C at the
12 ARTEMIS *Laboratoire de Mesure* ^{14}C (LMC14) at the CEA (Atomic Energy
13 Commission) at Saclay and Poznan Radiocarbon Laboratory. Organic macro-
14 remains were extracted by sieving the sediments. The IntCal13 calibration curve
15 (Reimer et al., 2013) was used for ^{14}C age calibration. The age/depth model was
16 generated using the software-package “clam” (version 3.0.2 R Development Core
17 Team, 2011) (Blaauw, 2010).

18 As ^{14}C analysis was performed only on the ROB15-P5 core, the ages calculated
19 for this core were correlated the ROB15-P6 core using XRF high resolution data for
20 both cores. The depths in the ROB16-P6 were corrected using the same correlation.

21

22 **3.3 High resolution XRF geochemical analysis**

23 Variations in concentrations of Al, Si, Ca, Ti, Mn, Fe, Cu and Pb were
24 determined by high resolution XRF analysis on the surfaces of the split sediment
25 cores at 1-mm intervals, using a non-destructive Avaatech core-scanner (see

1 Supplementary Information (SI)).

2

3 **3.4 Major elements, dry bulk density (DBD)**

4 Core ROB15-P5 was sampled at 5-mm intervals and dried at 60 °C for 4 days
5 to obtain its DBD. Major elements (Si, Al, Ca, Ti, Fe, Mn) were determined in 31
6 samples of core ROB15-P6 at the SARM in Nancy (France) using ICP-OES after
7 LiBO₂ fusion as described by (Carignan et al., 2001). The variations obtained using
8 XRF geochemical analysis and ICP-OES are in good agreement (SI Fig. S1).

9

10 **3.5 Total organic carbon (TOC), nitrogen (N) analysis**

11 Total carbon (TC), TOC content were measured in discrete samples from core
12 ROB15-P5 using a Cavity Ring-Down Spectrometer (Picarro, Inc®) coupled with
13 Combustion Module (Costech, Inc®) (CM-CRDS) using analytical methods
14 previously reported in (Guédron et al., 2019), calibration and sample preparation
15 according to (Balslev-Clausen et al., 2013; Paul et al., 2007). More details are given
16 in SI.

17 Total nitrogen (N) concentrations were obtained by combustion in an automatic
18 C-N analyzer (Elementar VarioMax) on the same samples.

19

20 **3.6 Cu, Sb, Pb and Pb isotope analysis**

21 High resolution XRF analysis only allows semi quantitative determination of the
22 concentrations of some elements. In order to precisely determine Cu, Sb, Pb
23 concentrations, and to extend the set of elements studied 31 samples were selected
24 in core ROB15-P6. These samples were chosen to better represent the variations
25 observed in the XRF profile. Their Cu, Sb, Pb concentrations and Pb isotopic

1 composition were determined with the Thermo Scientific® iCAP Q ICP-MS (using the
2 kinetic energy discrimination mode and He as collision gas on the AETE-ISO, OSU
3 OREME platform of the University of Montpellier, France (for more details, see SI).

4 For Cu and Pb, as shown in SI Fig. S2, the variations obtained using XRF high
5 resolution analysis and ICP-MS are in good agreement.

6

7 **3.7 Hg analysis**

8 Total Hg concentrations in core ROB15-P5 were determined by atomic
9 absorption spectrophotometry after dry mineralization and gold amalgamation using
10 an automatic mercury analyzer (Altec, Model AMA 254) with a relative precision of \pm
11 10% (Guedron et al., 2009). The analytical quality was insured by analyzing replicate
12 samples (the measurement error was always $< 6\%$), by analyzing certified reference
13 materials (MESS-3 = $90.3 \pm 2.8 \text{ ng g}^{-1}$, $n=12$, 2.3% accuracy). The detection limit
14 ($3SD_{\text{blk}}$) was $0.010 \mu\text{g g}^{-1}$.

15

16 **3.8 Enrichment Factors (EF) and TMM Accumulation Rates (AR_T)**

17 TMM concentrations can be influenced by the relative proportions of mineral
18 matter derived from the bedrock and of OM either autochthonous (in lake production)
19 or allochthonous (terrestrial). In order to minimize a possible influence, TMM
20 concentrations (SI Fig. S2) were normalized to Ti, which is a conservative element
21 and a good tracer of minerogenic inputs in lakes of glacial origin (Bajard et al., 2016)
22 and expressed as enrichment factors (EF) relative to mean background (bkg) values
23 calculated as follows:

24 $XEf = (X/Ti)_{\text{sample}} / (X/Ti)_{\text{bkg}}$, where $X = \text{Cu, Sb, Hg}$. The bkg values are those
25 observed below 94 cm depth in core ROB15-P6 for Cu, Sb, Pb and 73.5 cm depth in

1 core ROB15-P5 for Hg.

2 TMM AR_T was calculated as follows:

3 $XAR_T=(SR)(DBD)([X])$ where $[X]=Cu, Sb, Pb, Hg$ concentrations, SR
4 =sedimentation rate, DBD=dry bulk density.

5

6 **4 Results**

7

8 **4.1 Sedimentology**

9 Based on the description of the sediment and TOC content, the ROB15-P5 core
10 was subdivided into four units (Fig.2):

11 Unit I (0-6 cm) is composed of olive to grey clayey silt with relatively low OM
12 content (mean TOC= 6%);

13 Unit II (6-78 cm) is composed of olive silty clay, darker in color between 35 and
14 60 cm and lighter between 60 and 78 cm with high OM content (mean TOC=8.7%).
15 The variations in color were linked to OM content, which was higher (mean
16 TOC=12%) in the darker facies.

17 Unit III (78-101 cm) is composed of grey clay to light orange grey clay at the
18 base with submillimetric laminations in the lower part of this unit. The laminations
19 were formed by light grey clay to grey silt alternation with a very low (0.8%) TOC
20 content.

21 Unit IV (101-115 cm) is composed of a mixture of pebble, gravel and sand with
22 few fine-grained sediments. The contact between Unit III and Unit IV is very abrupt
23 with the presence of a large (5 cm) polished pebble with grooves. Some pebbles are
24 also angular, and their nature is mainly related to the rock present in the watershed.

25 The ROB15-P6 core has a similar succession of sedimentary Units. Unit I (0-6.5

1 cm), Unit II (6.5-82 cm) showed the same variations in color, and Unit III (82-110 cm)
2 was also present, whereas Unit IV was absent in this core. Based on this
3 sedimentary description and XRF data, the two cores are precisely correlated.

4

5 **4.2 Chronostratigraphy**

6 A total of nine radiocarbon analyses were performed on plant macrofossils
7 collected in the top 69 cm (Unit I and Unit II) of the sediment core (Table 1). Unit III
8 and IV were devoid of plant macrofossil remains, hence ^{14}C dating was not possible.
9 Sample SacA42479, at 55 cm, dated to 4900 BP, appeared to be slightly too old, the
10 explanation being that it was probably temporarily stored in the watershed or
11 reworked. Even though this date is not very old compared to the long-term trend, if
12 we retained it, the age model would include short term variations in this part of the
13 core, whereas no variations were observed in other proxies and the calculated fluxes
14 would also be impacted by this age. For these reasons, the age of sample
15 SacA42479 was not included in our age/depth model. We then calculated an
16 age/depth relationship using smooth spline interpolation in the Rcode package
17 “Clam” version 2.2 (Blaauw, 2010). The first 69 cm of the sediment sequence from
18 ROB15-P5 core covered the last 9.98 kyr BP (Fig.2). The mean sedimentation rates
19 estimated from this sediment record at 0.11 mm yr^{-1} varied between 0.28 and 0.03
20 mm yr^{-1} . The higher sedimentation rate corresponds to the upper part of the core. A
21 major change in the sedimentation rate occurred around 2125 ± 85 BP at 35 cm
22 corresponding to the change in color in Unit II. Above this depth, the mean
23 sedimentation rate was around 0.19 mm yr^{-1} and below this depth, around 0.053 mm
24 yr^{-1} (Fig. 2). In the ROB15-P5 core, sediment accumulation rates varied between
25 $2.10^{-7} \text{ g m}^{-2} \text{ yr}^{-1}$ and $1.310^{-8} \text{ g m}^{-2} \text{ yr}^{-1}$ and presented the same variations as the

1 sedimentation rates, except in the older part of the core due to higher density of the
2 sediment. If we extrapolate the age/depth model to the base of Unit II we obtain an
3 age of $11,090 \pm 440$ BC. The sedimentation rate in the lower part of Unit II was
4 probably higher than that in relation to more terrigenous inputs (Fig.3), so the base of
5 this unit probably corresponds to the beginning of the Holocene and Unit III to the
6 late glacial period. During this period - at least at the beginning - a small glacier was
7 probably present in the watershed as suggested by the clay to silt lamination at the
8 base of the Unit. The sedimentation rate in Unit III was probably higher than during
9 the Holocene, as evidenced by more terrigenous content. Even though we cannot
10 precisely date the transition between Unit III and Unit IV, Unit IV probably
11 corresponds to moraine deposits before the glacial retreat in this part of the Alps and
12 the lake formation.

13 Even though the stratigraphy made it possible to distinguish four Units, TMM
14 analysis was performed in Unit I, II and III only when the lake was present. For core
15 ROB15-P6, the two deepest sediment levels of Unit III could not be dated due to
16 absence of plant macrofossils and were consequently attributed arbitrary ages.

17 The chronology in the upper part of Unit I is not well constrained as we did not
18 measure short-lived radionuclides, but we assume that the base of Unit I
19 corresponds to the beginning of the construction of ski resorts in the watershed in the
20 middle of the 20th century.

21

22 **4.3 Major elements and TOC**

23 The sediment cores contained significant proportions of minerogenic elements
24 as indicated by the mean average concentrations of Al ($4.6 \pm 0.6\%$), Fe ($2.6 \pm 0.6\%$), Ti
25 and Mn (in the thousand and hundred $\mu\text{g g}^{-1}$ range, respectively). As expected in a

1 catchment devoid of carbonated rocks, the Ca content was low ($0.6\pm 0.1\%$) but higher
2 Mg concentrations ($4.0\pm 1.1\%$) were encountered. As shown in Fig. 3 and in SI Fig.
3 S1, concentrations of Al, Ti, Fe and Mn were high in Unit III (prior to 10,800 BC). In
4 Unit II, the concentrations then decreased down to 49 cm depth (around 2400 BC)
5 and except for Mn and Ca, peak concentrations were generally identified at 38 cm
6 depth (around 720 BC). Two other peak concentrations of all elements were
7 observed at 15.5 cm depth (around AD 1340) in Unit 2 and at 4.5 cm depth
8 concentrations were maximum at 4.5 cm depth (around AD 1890) in Unit 1.

9 In Unit III, the Fe/Mn ratio was relatively high ($0.81\leq \text{Fe/Mn}\leq 0.97$) compared to
10 that observed at the basis of Unit II between 54 and 65 cm depth (between 3130 and
11 6020 BC) (around 0.64). The ratio then increased substantially to reach a maximum
12 of 0.78 at 37 cm depth (around 500 BC). From 34 cm depth (AD 20) to the upper
13 core level, Fe/Mn ratios varied in a small range between 0.66 and 0.71

14 With a mean concentration of $24\pm 2\%$, silica was the main component of
15 sediment cores. In Unit III, concentrations were relatively high, ranging from 24% to
16 26%. In Unit II, the concentrations decreased to 20% at 47 and 52 cm depth (2815
17 and 2080 BC) (Fig. 3), then concentrations increased and relatively high values (up
18 to 27%) were measured at depths of 9 and 28.5 cm (AD 645 and 1656). Unit I
19 generally displayed lower concentrations (23% to 24%) than those observed in the
20 top sediment levels of Unit II.

21 In addition to silica, OM is also an important component of particulate matter.
22 TOC was relatively low (0.8%) in Unit III. The concentrations then increased, and a
23 broad peak with values reaching 15.2% was observed at 49 cm depth (around 2400
24 BC) (Fig.3). Subsequently, a general decreasing trend was observed with a minimum
25 value of 5% at the base of Unit I (AD 1850).

1 The C/N values varied in a narrow range (9.3-15.6), with an average value of
2 12.3 ± 1.9). In Unit III and at the base of Unit II, the C/N values were around 9.6. Like
3 TOC, C/N values then increased to a maximum of 16 at 61.5 cm depth (around 4680
4 BC) and remained above 14 until 46 cm depth (around 1800 BC). The ratios then
5 decreased towards the surface reaching a minimum value of 9.3 at 17.5 cm depth
6 (around AD 1270). In Unit I the C/N ratios ranged from 9.9 to 12.

7 $\delta^{13}\text{C}$ values also remained within a narrow range ($-27.9 \pm 1.3 \text{ ‰}$). The highest
8 values up to 25.5 were observed in Unit II, after which values decreased again to
9 reach a minimum (-29.7‰) at 27 cm depth (around 5770 BC). Values tended then to
10 increase slightly but remained below 27‰. In Unit I higher values (26.5-26.9‰) were
11 observed.

12

13 **4.4 TMM concentrations and total accumulation rates (AR_T)**

14 CuEF (Fig. 4) increased regularly to reach maximum (3.9) at 52 cm depth
15 (around 2815 BC) followed by a decreasing trend with a small shoulder at 45 cm
16 depth (around 1790 BC). A minimum value (0.76) was reached at the top of Unit II. In
17 Unit I no enrichment of Cu was observed and CuEF values remained around 1.
18 Comparatively high CuAR_T (up to $690 \mu\text{g m}^{-2} \text{yr}^{-1}$) were observed in Unit III (Fig. 4). In
19 Unit II, a large but lower peak was observed between 42 and 62 cm depth (1350 to
20 5000 BC), whereas the highest accumulation rates (up to $1743 \mu\text{g m}^{-2} \text{yr}^{-1}$) were
21 recorded in Unit I.

22 From 49 cm to 65 cm depth (from around 2370 to 6020 BC), SbEF ranged from
23 0.39 to 0.59. (Fig. 4). From 49 cm depth, values increased strongly to reach a
24 maximum of 3.6 4 at 42 cm depth (around 1350 BC), after which EF decreased with
25 three comparatively smaller peaks at 36.1 cm depth (320 BC) (SbEF=1.97). From

1 this depth up to the top of the core SbEF did not indicate Sb enrichment compared to
2 background values.

3 The variations in SbAR_T generally resembled those of SbEF (Fig. 4) except in
4 Unit I. In Unit III, SbAR_T was low (mean= $3.4 \pm 0.5 \mu\text{g m}^{-2} \text{yr}^{-1}$, displaying minor
5 variations between 49 and 76 cm depth (2370 and 9600 cal BC). As concentration,
6 SbAR_T displayed a peak ($26.3 \mu\text{g m}^{-2} \text{yr}^{-1}$, 7 times higher than the mean natural flux)
7 at 42 cm depth (around 1350 BC). In the two upper levels of Unit I corresponding to
8 the 20th century AD, high flux values (56 and $29 \mu\text{g m}^{-2} \text{yr}^{-1}$, 16 and 8 times the mean
9 natural flux) were observed.

10 HgEF gently increase from 73.5 cm depth (around 8900 cal BC) up to 57.5 cm
11 depth (around 3730 BC) where HgEF values reached a plateau around 12 (Fig. 4).
12 From 47.5 cm depth (around 2015 BC) HgEF increased sharply to reach maximum
13 EF values (54) at 46 cm depth (1800 BC). HgEF then decreased but remained high
14 ($\text{HgEF} > 4.43$). In the uppermost sediment level dated to AD 1997, HgEF was 7.91,
15 which is 7 times lower than the maximum observed in Unit II. Between 67 and 76 cm
16 depth (from around 10,600 to 6350 BC) HgAR_T varied narrowly (mean= $0.23 \pm 0.06 \mu\text{g}$
17 $\text{m}^{-2} \text{yr}^{-1}$) (Fig.4). Variations in HgAR_T resembled those of HgEF below 144 cm depth
18 (around 500 cal BC) with a sharp maximum at $6.67 \mu\text{g m}^{-2} \text{yr}^{-1}$ at 46 cm depth
19 (around 1800 BC). This flux is comparable to the present-day flux. Above this depth,
20 two additional small peaks appeared at 36 cm depth (320 BC) ($\text{HgAR}_T = 2 \mu\text{g m}^{-2} \text{yr}^{-1}$)
21 and between 17 and 24 cm depth (AD 935 to 1300) ($\text{HgAR}_T = 3.90 \mu\text{g m}^{-2} \text{yr}^{-1}$). From
22 the top of Unit II (8.5 cm depth) and up to the surface a marked increase in fluxes
23 was also observed with values reaching $6 \mu\text{g m}^{-2} \text{yr}^{-1}$ at 5.5 cm depth. The value at
24 the surface, $5.1 \mu\text{g m}^{-2} \text{yr}^{-1}$, was slightly lower than that observed at 46 cm depth but
25 in the same order of magnitude.

1 From 65 cm depth (around 6020 BC) until 49 cm depth (around 2370 BC),
2 PbEF (Fig. 4) varied little with values around 1 implying no Pb enrichment relative to
3 background values. From 49 cm depth (around 2370 BC) PbEF started to increase.
4 Four peaks were observed one at 47 cm depth (2080 BC) (EF=3.6), one at 36 cm
5 depth (320 BC) (EF=20.7), another at 19 cm depth (AD 1215) (EF=8.9) and finally at
6 the surface of the core (EF=8.9).

7 The variations in $PbAR_T$ paralleled those of PbEF ($r^2=0.83$, except in Unit I
8 (Fig.4). From the bottom of the core to 49 cm depth (2370 BC), $PbAR_T$ varied
9 smoothly between 11 and 21 $\mu\text{g m}^{-2} \text{yr}^{-1}$. The highest flux (294 $\mu\text{g m}^{-2} \text{yr}^{-1}$) was
10 observed at 36 cm depth (around 320 BC). The flux in the top of the core was
11 comparatively lower (119 $\mu\text{g m}^{-2} \text{yr}^{-1}$).

12

13 **4.5 Lead isotopes**

14 In Unit II and III until 65 cm depth (around 6020 BC), $^{206}\text{Pb}/^{207}\text{Pb}$ ratios (Fig. 5)
15 were high (mean \pm 1s=1.202 \pm 0.003). Between 52 and 62 cm depth (5000 to 2815 BC)
16 slightly lower $^{206}\text{Pb}/^{207}\text{Pb}$ values (1.190 \pm 0.004) were recorded. Then, although highly
17 variable, $^{206}\text{Pb}/^{207}\text{Pb}$ ratios showed a decreasing trend and their distribution mirrored
18 that of PbEF. Low $^{206}\text{Pb}/^{207}\text{Pb}$ ratios ($^{206}\text{Pb}/^{207}\text{Pb}=1.1881$) were observed at 47 cm
19 depth (around 2080 BC), between 35 and 37 cm depth (140 and 500 BC)
20 ($^{206}\text{Pb}/^{207}\text{Pb}=1.178$), and between 7 and 25 cm depth (AD 880 to 1790)
21 ($1.179 \leq ^{206}\text{Pb}/^{207}\text{Pb} \leq 1.182$). In Unit I, $^{206}\text{Pb}/^{207}\text{Pb}$ ratios decreased markedly to reach
22 1.158 in the uppermost sediment level.

23

24 **5. Discussion**

25

26 **5.1 Variations in sedimentary inputs and biogeochemical conditions**

1 Prior to around 8000 BC, sedimentation rates, concentrations of terrigenous
2 elements were high whereas TOC concentration were low (Fig.3). During this period
3 (end of the Late Glacial Period), cold conditions favored erosion of bedrock probably
4 by the small glacier in the watershed. During the Holocene climatic optimum, from
5 the end of the Late Glacial Period (around 10,800 BC), TOC concentrations
6 increased to reach a maximum around 2400 BC (Fig.3). Simultaneously, the C/N
7 ratio increased from 9.5, a value which is in the range of that of phytoplankton and/or
8 aquatic macrophyte, to 15, characteristic of land derived OM. In parallel, $\delta^{13}\text{C}$ values
9 increased to $-27.9 \pm 1.3 \text{ ‰}$ (Fig. 3) which is also typical of terrestrial OM (Thevenon
10 et al., 2012), thus the variations in TOC reflect a constant increase in inputs of soil
11 derived OM. The higher temperatures during this period probably favored, the
12 development of a soil and of vegetation as also described in other alpine lakes during
13 the same period (Bajard et al., 2017, 2016; Mourier et al., 2010). This vegetation has
14 stabilized the soils, thereby limiting erosion as shown by the low sedimentation rates
15 and the low concentrations of minerogenic elements (Si, Al, Ti) (Fig. 3). From 2400
16 BC, the TOC decreased as also observed in other high Alpine Lakes (Giguet-Covex
17 et al., 2011) and the C/N ratios tended toward those of phytoplankton and/or aquatic
18 macrophytes. Concentrations of minerogenic elements increased in parallel
19 indicating increased inputs of mineral matter. This corresponds to the onset of the
20 Holocene Neoglacial period in which glaciers advanced in the Mont Blanc massif (Le
21 Roy et al., 2015) and the Ecrins-Pelvoux massifs (Le Roy et al., 2017) in phase with
22 a regional increase in mineral matter inputs and/or erosion, as recorded in Lake
23 Bramant (Guyard et al., 2007), Lake Anterne (Giguet-Covex et al., 2011), Lake Petit
24 (Brisset et al., 2013), Lake Bourget (Arnaud et al., 2012), Lake la Thuile (Bajard et
25 al., 2016) and Lake Savine (Sabatier et al., 2017).

1 From the Iron Age on, sedimentation rates increased due to human activity,
2 which according to (Arnaud et al., 2016) was responsible for general erosion at local
3 scale in the Northern French Alps from this date on. Increasing pastoral activities in
4 the catchment plus deforestation led to a continuous increase in the
5 sedimentation rate until around AD 1340 when Black Death pandemic ravaged
6 Europe. After a period of stagnation erosion resumed around AD 1500 and increased
7 exponentially during the two last centuries. The most recent episode which is marked
8 by both an increase in the sedimentation rates and in Ti or Al concentrations in the
9 shallowest sediment levels (Fig. 3) is probably linked to human activities:
10 industrialization of the valley and the building of roads and railways lines in the 19th
11 century AD that probably increased the supply of atmospheric dust and of the ski
12 resort in the Lake Robert watershed in the second half of the 20th century AD.

13 Other studies have evidenced fluctuations in lake water oxygenation in alpine
14 lakes during the same period. Considering bottom water oxygenation, higher values
15 of the Fe/Mn ratio at the end of the Late Glacial Period and Early Holocene (from
16 prior to 10,800 BC to 8400 cal BC) and during the neoglacial period from around
17 2400 BC with a maximum in 500 BC, could reflect episodes of oxygen deficiency in
18 lake bottom waters during the cold periods as also inferred in the neighboring Lake la
19 Thuile (Bajard et al., 2016). However, the absence of a correlation between the
20 Fe/Mn ratios and the TMM concentrations suggests that such redox changes have
21 had no influence or only negligible influence on retention of the TMM studied here.

22

23 **5.2 Local history of TMM pollution**

24

25 *Paleolithic-Mesolithic-Neolithic-Chalcolithic (Prior to 10,800-2800 BC)*

1 During this period, PbEF indicated no enrichments and prior to 5000 BC, the Pb
2 isotopes signature (Fig. 5) resembled that of moraine samples reported by (Garçon
3 et al., 2012). At the end of the Mesolithic (from around 5000 BC), PbEF did not
4 change but the $^{206}\text{Pb}/^{207}\text{Pb}$ isotope ratios decreased. This decrease, which was
5 concomitant with a slight decrease in PbAR_T, reflects a change in the source of Pb
6 inputs to the lake. The most likely explanation is an increase in the contribution of the
7 atmospheric Pb source and a comparative decrease in the contribution of the source
8 of lithogenic Pb linked to the erosion decline in the watershed at the onset of the
9 Holocene climatic optimum due soil stabilization by vegetation (Bajard et al., 2016).

10 CuEF and CuAR_T increased from 10,800 BC on, which is before the emergence
11 of Cu metallurgy in Europe (Carozza, 2010) and reached a maximum around 2800
12 BC. The parallel increase in Cu(EF) and TOC (Fig. 6) suggests that Cu deposition in
13 lake sediment was enhanced by the influx of OM into the lake during the Holocene
14 climatic optimum. This is not surprising, given the strong affinity of Cu for
15 complexation with organic ligands (Irving and Williams, 1953) which largely controls
16 its biogeochemical cycling (Benedetti et al., 1996; Hering and Morel, 1988;
17 Kinniburgh et al., 1999).

18 Mercury enrichment factors also increased slowly and Hg concentrations were
19 strongly correlated with TOC (Fig. 6) emphasizing the fact that during this pristine
20 warm climate period, the in-lake accumulation of Hg was mostly driven by soil
21 erosion (Schütze et al., 2018). This is consistent with the strong affinity of Hg for TOC
22 (Lindberg and Harriss, 1974; Skyllberg et al., 2000), and in particular for soil OM
23 known as a great sink for atmospheric Hg through vegetation uptake (Andersson,
24 1979; Demers et al., 2013; Guédron et al., 2013). This suggests that both
25 atmospherically derived Hg and dissolved organic carbon bound Hg that leached

1 from the watershed were likely scavenged onto the OM and/or recycled onto
2 sulfidized OM during early diagenesis in the lake sediment (Feyte et al., 2012).

3

4 *Bronze Age (2800 -1000 BC)*

5 The first significant peak of PbEF occurred around 2080 BC while the maximum
6 HgEF values between around 1800 and 1600 BC were 4 to 5 times higher than those
7 observed between 4000 and 2000 BC (Fig. 4). SbEF values were also high around
8 1350 BC during the Bronze Age. As already mentioned, the study area has
9 numerous metal ores. In the Bronze Age, chalcopyrite rich veins were exploited in
10 the Rousses massif from 2200 BC until 1700 BC (Bailly-Maitre and Gonon, 2008;
11 Carozza, 2010). These mines are located 15 km from Lake Robert as the crow flies.
12 The occurrence of grey Cu containing Sb in the form of bournonite and tetrahedrite
13 has already been reported in ores exploited in the Bronze Age (Chermette and
14 Termier, 1970). Analysis of a chalcopyrite sample from the ores exploited in the
15 Bronze Age had high ($115 \mu\text{g g}^{-1}$) Hg content. This ore was exploited by firesetting
16 (Bailly-Maitre and Gonon, 2008) which probably favored the release of Hg and Sb
17 into the atmosphere. Although CuEF generally decreased in the Bronze Age
18 following the decline of primary production, the peak shoulders in the Cu distribution
19 and especially in CuAR_T observed around 1800 BC (Fig. 4), could also reflect mining
20 pollution. The Pb isotopic composition could be interpreted as the reflection of two
21 end-members: a natural background and that of Rio Tinto mines that were already
22 exploited at that time (Leblanc et al., 2000). However, the Hg and Sb pollution and
23 the similarity between the Pb isotopic ratios in lake sediments and those of the
24 bournonite and the galena of the Cu-Pb-Ag vein samples from the Rousses Massif
25 (Garçon et al., 2012) suggest that early Cu mining in the Rousses Massif was

1 responsible for most of the Pb, and of all the Hg and Sb pollution of Lake Robert
2 during the Bronze Age. Only slight increases in HgAR have been reported in the high
3 altitude alpine Lake Meidsee during the Bronze Age (Thevenon et al., 2011).
4 Consequently, the marked rise in HgAR_T, which is the equivalent of that of the
5 Industrial era (Fig. 4), likely results from rapid recycling of Hg emitted in the vicinity of
6 this mining area.

7

8 *Iron Age-Roman Empire (800 BC -AD 500 AD)*

9 The main Pb pollution event occurred between ca 500 BC and AD 20 with a
10 maximum in ca 320 BC (Iron Age). According to the Pb isotopic composition, the
11 pollution did not originate from local ores. One possible provenance of the pollution is
12 Sardinia (Baronia region) (Fig. 5).

13 Later, after AD 20 during the Roman Empire, although data are scarce, the EF
14 Pb indicated low Pb pollution and the Pb composition reflected either local Pb or a
15 mixture of natural and Iberian Pb. Archeological data plead for the second
16 hypothesis. Indeed, there are no archeological proofs for an exploitation of the
17 Bronze Age mines during the Roman Empire (Bailly-Maître, 2014).

18 After the major pollution event in the Bronze Age, EF and AR_T of Sb and Hg
19 remained relatively low. In agreement with Pb isotope ratios, these results also plead
20 for the abandonment of the Bronze Age mines in favor of distant mines during the
21 Iron Age and the Roman Empire.

22

23 *Medieval -Modern-Industrial periods (AD 800-1980)*

24 During the medieval period, PbEF and PbAR_T reached a peak around AD 1215.
25 Broad peaks of HgAR_T, SbAR_T and SbEF were also visible from 900 to AD 1340.

1 The Pb isotope signatures are consistent with pollution caused by exploitation of the
2 local Brandes Ag-Pb mine in Rousses Massif that was intensely exploited at that time
3 (Bailly-Maître, 1996; Chermette and Termier, 1970). High concentrations of Hg and
4 Sb were measured in the galena found in the soils of a workshop (Hg=3.6 $\mu\text{g g}^{-1}$;
5 Sb=1986 $\mu\text{g g}^{-1}$) and in the mine tailings (Hg=2.1 $\mu\text{g g}^{-1}$; Sb=1022 $\mu\text{g g}^{-1}$) (Table 2).
6 Pollution declined after AD 1340 because the mine closed in AD 1330 due to climate
7 cooling, the exhaustion of some ores but also due to the Black Death pandemic in
8 Europe (Bailly-Maître, 1996).

9 After the medieval period, the isotopic Pb composition evolved from that of the
10 local Pb ores in AD 1790 to that of French industrial Pb or of leaded gasoline in 1976
11 (Elbaz-Poulichet et al., 1986) (Fig. 4).

12

13 **5.3 Deciphering local and regional mining contamination**

14

15 This study highlights increased atmospheric TMM pollution in Lake Robert in
16 prehistoric and historical times: the Bronze Age (2000-800 BC), the Iron Age and
17 Roman Empire (800 BC- AD 500), the medieval period (AD 500-1500), the modern
18 and industrial periods (AD 1500-present).

19 Pb pollution in the Bronze Age has been also reported in Spain (Martínez
20 Cortizas et al., 2016) but in that case, pollution, which was generally low and was not
21 recorded in all European archives was generally linked to small local exploitations.
22 The Pb isotope ratios in Lake Robert reveal mixed sources in the form of local natural
23 Pb from moraines and Pb from local minerals found in the Cu ores that were
24 exploited at that time. The high level of Hg and Sb pollution observed in Lake Robert
25 has not been observed in other regional archives (Mont Blanc ice, Etang de la
26 Gruère mire). The source of pollution was thus probably only local and although the

1 AR_T increased by a factor of 7 (Hg) and 9 (Sb) compared to pre-Bronze Age values,
2 the emissions were too small to affect distant locations.

3 Pb pollution is ubiquitous in the records of three other periods in southwestern
4 or western Europe in the Swiss Jura (Shotyk et al., 1998), in Swiss Wallis (Thevenon
5 et al., 2011), in a French Pyrenean mire (Hansson et al., 2017), in a northwestern
6 Spanish mire (Martínez Cortizas et al., 2016-Fig. 7), in a coastal Mediterranean
7 lagoon (Elbaz-Poulichet et al., 2011), in Atlantic shelf sediments (Mil-Homens et al.,
8 2017) in Mont Blanc (Preunkert et al., 2019) and Greenland ice (Hong et al., 1994;
9 McConnell et al., 2018- Fig. 7; Rosman et al., 1997). In the Iron Age and Roman
10 Empire, this pollution has mostly been attributed to atmospheric transport from the
11 Iberian Peninsula where the Rio Tinto and Murcia mines were intensely exploited by
12 the Phoenicians, the Carthaginians and subsequently by the Romans. During the
13 Roman period, between 50 BC and AD 500, it has been estimated that Iberian mines
14 contributed 60% of European production, British mines 10%, the remainder
15 originated from comparatively smaller mines in Sardinia, Italy, Gaul, the Carpathians
16 and the Balkans (Rosman et al., 1997). During the Iron Age before 50 BC, the
17 isotopic composition in Lake Robert resembles that of Sardinian Pb (Baronia). Later
18 during the Roman period, Pb isotope composition could reflect a mixture of local
19 background Pb and Iberian Pb. Light Hg pollution was also detected in the Iron Age-
20 Roman Empire in the archive studied here as well as in several archives in the Alps
21 (Meidsee sediments Thevenon et al., 2011-Fig. 7), in the Jura (Etang de la Guère
22 mire, Roos-Barraclough and Shotyk, 2003- Fig. 7), in the Black Forest (Glaswaldsee
23 sediments, Schütze et al., 2018- Fig. 7), in Spain (Penido Vello mire, (Martinez-
24 Cortizas et al., 1999). This pollution is generally attributed to the mining of the
25 Almaden Hg ore (i.e., cinnabar) but local mining and/or metallurgy cannot be

1 excluded. Regarding Sb, observations in mires in southwestern Europe (Pyrenean,
2 Hansson et al., 2017; Swiss Jura, (Shotyk et al., 1996) and in the Arctic (Krachler et
3 al., 2008) and in Mont Blanc ice (Preunkert et al., 2019) also produced evidence for
4 pollution in the Iron Age-Roman period that was attributed to mining in the Iberian
5 Peninsula.

6 Pb pollution in the Middle Ages is also reported in all European and Arctic
7 archives. At that time, many local Ag-Pb ores were exploited throughout Europe
8 primarily to make coins. Pb isotope ratios in Lake Robert mainly reflect pollution
9 caused by local mines. Hg pollution also occurred in the Middle Ages, but recent
10 studies suggest that deforestation and burning of biomass could also have played an
11 important role (Elbaz-Poulichet et al., 2011; Enrico et al., 2017).

12 In the modern period, Pb, Hg and Sb AR_T increased in Lake Robert to reach
13 maximum values of $2723 \mu\text{g m}^{-2} \text{yr}^{-1}$ (Pb) and $56 \mu\text{g m}^{-2} \text{yr}^{-1}$ (Sb) in AD 1976, and
14 $5.14 \mu\text{g m}^{-2} \text{yr}^{-1}$ (Hg) in AD 1997. These values are 90 (Pb), 5 (Hg) and 20 (Sb) times
15 higher than those of the pre-Bronze Age epoch.

16 This increase reflects both local and global TMM emissions. At the end of the
17 20th century, Pb emissions and fallout were dominated by leaded antiknock
18 compounds in gasoline, the use of which was forbidden in 2000 in Europe. At global
19 scale, most Hg atmospheric pollution originated from fossil-fuel fired power plants
20 (Pirrone et al., 2010). In the vicinity of Lake Robert, several local/regional sources of
21 Hg including coal mining and electrometallurgy were identified by Guédron et al.,
22 (2016) which could also contribute to the observed pollution. Increasing
23 concentrations, enrichments and accumulation rates of Sb have been observed in
24 mires in Europe (Shotyk et al., 2005, 2004), Alpine ice (Van de Velde et al., 1999)
25 and in polar ice in the Canadian Arctic (Krachler et al., 2008, 2005) since the

1 industrial revolution, illustrating the dramatic increase in global emissions of
2 atmospheric Sb linked to the numerous uses of Sb (flame retardant, brake pads,
3 additive to plastic) and also the mining of sulfide ores. In Lake Robert, the increase in
4 $SbAR_T$ could be also due to the construction of the ski resort in the watershed that
5 increased the AR of the sediment by remobilizing sediments/soils polluted by past
6 mining activity. Similar remobilization processes of past pollution has been reported
7 for pesticides (Sabatier et al., 2014) or P content (Bajard et al., 2018) in other lakes
8 in the French Alps.

9

10

11 **6. Conclusion**

12

13 This multiproxy study in Lake Robert in the French Alps enabled the
14 reconstruction of the history and origins (local vs regional) of TMM pollution over the
15 last 10,000 years.

16 The variations in TMM enabled identification of four periods of pollution: the
17 Bronze Age (1800-1000 BC), the Early Iron age and Roman Empire (800BC- AD
18 500), the Middle Ages (AD 500-1500), the modern and the industrial period (AD1500-
19 2000).

20 One striking feature is the notable Hg and Sb pollution in the Bronze Age linked
21 to early Cu mining in the Rousses Massif. During that period the Hg flux resembled
22 the flux today. The significant increase in pollution in recent (20th century AD)
23 samples particularly in the case of Sb, is partly related to the remobilization of
24 polluted soil particles during the construction of the ski resorts.

25 This study confirmed the importance of atmospheric Pb pollution particularly

1 during the Iron Age-Roman Empire, with a potential source in Sardinia during the Iron
2 Age and in Iberia during the Roman Empire.

3 Last but not least it also shows that climate-induced environmental changes in a
4 high altitude lake catchment can strongly affect Cu deposition and to a lesser extent
5 that of Hg. During the Holocene climatic optimum (3350-2100 BC), the warmer
6 climate conditions were favorable for soil development and hence a vegetation cover
7 leading to an influx of OM and correlatively a substantial increase in Cu and Hg flux
8 to the lake due to their strong affinity for MO.

9

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21

22 **Data availability**

23 Supplementary Information (SI) and datasets related to this article can be found at:
24 Elbaz-Poulichet, Guedron, Develle, Freydier, Perrot, Rossi, Piot, Delpoux, Sabatier,
25 (2019), "Lake Robert data Elbaz-Poulichet et al., 2019 QSR", Mendeley Data, v1

1 <http://dx.doi.org/10.17632/sswzys96rd.1>

2

lab_ID	Depth (mm)	Age (BP)	Calibrated age ranges at 95% (cal BP)
SacA42475	120	320 ± 30	305 - 465
Poz-77372	180	830 ± 30	687 - 789
SacA42476	250	1150 ± 30	978 - 1170
Poz-77373	338	1920 ± 35	1740 - 1948
SacA42477	390	2660 ± 30	2742 - 2844
SacA42478	470	3595 ± 30	3836 - 3976
SacA42479	550	4900 ± 30*	5588 – 5707*
Poz-77374	630	6040 ± 40	6786 - 6995
SacA42480	690	8200 ± 45	9024 - 9286

1

2 Table 1: ¹⁴C dates for Lake Robert (ROB15-P5). * rejected date.

3

4

Sample ID	Ore type	Location	Cu (µg/g)	Sb (µg/g)	THg (µg/g)	Pb (µg/g)
LBV1	Bournonite (PbCuSbS ₃)	Lake Blanc Huez	N.A.	N.A.	202	N.A.
LBV2	Malachite (Cu ₂ CO ₃ (OH) ₂)	Lake Blanc Huez	N.A.	N.A.	30	N.A.
LBV3	Galena (PbS)	Lake Blanc Huez	N.A.	N.A.	2.4	N.A.
LBV7	Barite (BaSO ₄)	Lake Blanc Huez	N.A.	N.A.	0.44	N.A.
Halde Soil	-	Brandes Mine	5311	1986	3.6	17224
Halde tailing	-	Brandes Mine	1545	1022	2.1	26351
Halde Galena	Galena (PbS)	Brandes Mine	232	691	N.A.	41046
OSUG-00248	Chalcopyrite	Demoiselles Mine	N.A.	N.A.	112	N.A.

1

2 Table 2 : Cu, Sb, Hg and Pb elemental content of minerals and mining soils or
3 tailings of the Grandes Rousses massif. Samples LBV1, LBV2, LBV3 and LBV7 are
4 from the Lake Blanc Huez (N 45° 6' 52", E 6° 6' 21") with their Pb isotopic
5 composition published in (Garçon et al., 2012), soil, mine tailing and Galena from the
6 Brandes mine (Huez, N 45° 5' 7.2", E 6° 5' 6.5") where provided by G. Sarret and
7 chalcopyrite of the Demoiselles mine (Vaujany, N 45°10' 44", E 6° 7' 52" , reference
8 OSUG-00248) was provided by the Mineralogical Collection of the Grenoble
9 Observatory of Earth Sciences (OSUG).

10

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- 6 Thevenon, F., Guédron, S., Chiaradia, M., Loizeau, J.-L., Poté, J., 2011. (Pre-)
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- 9 Van de Velde, K., Ferrari, C., Barbante, C., Moret, I., Bellomi, T., Hong, S., Boutron,
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11 antimony in high altitude alpine firn and ice. *Environ. Sci. Technol.* 33, 3495–
12 3501.
- 13

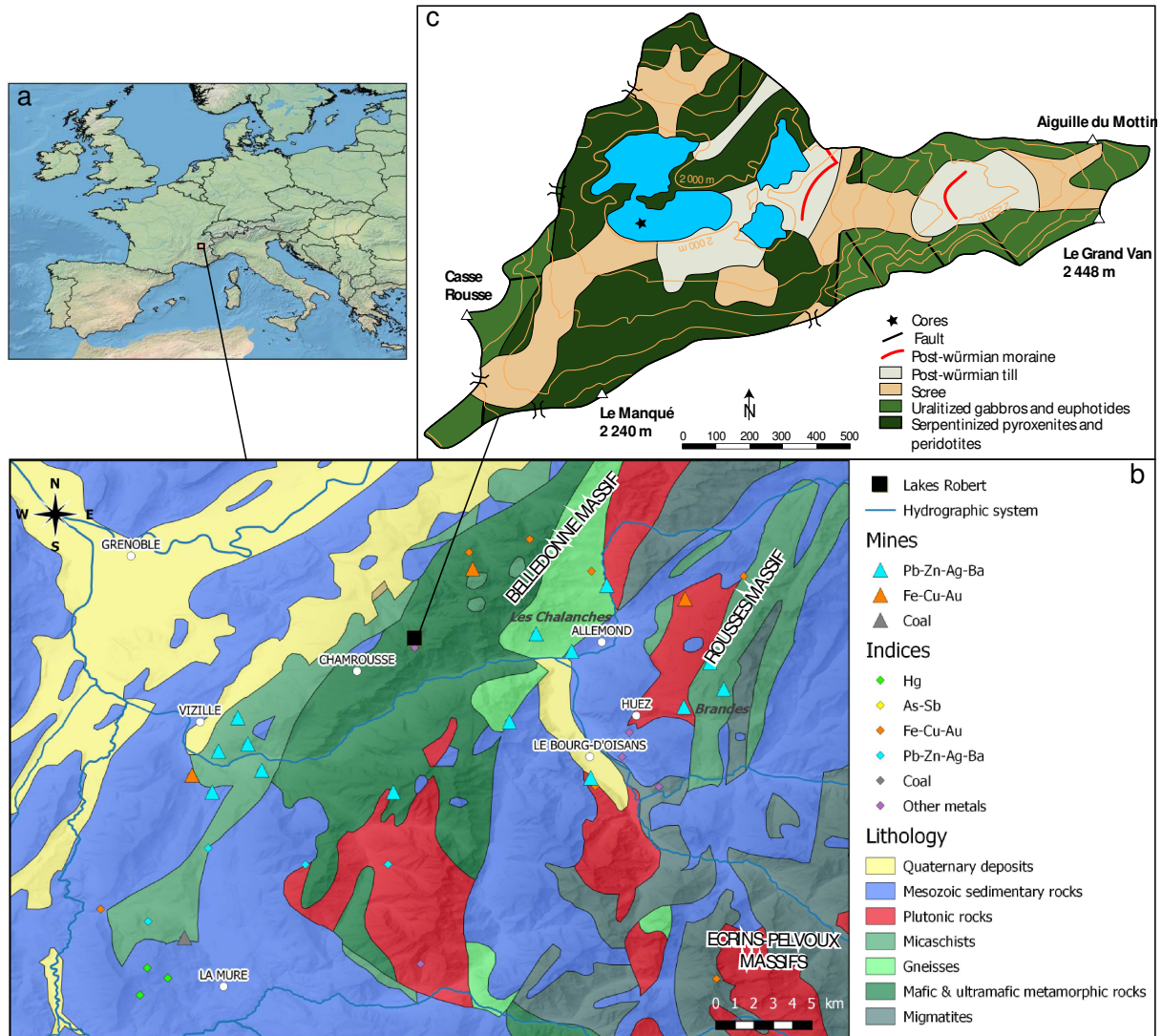


Fig. 1: Map showing the location of Lake Robert in the French Alps (a). Main geological units, the locations of recognized metal and metalloid indices and the locations of the main mining sites are shown in (b). The location of the coring site and the main lithological units of its drainage basin are shown in (c).

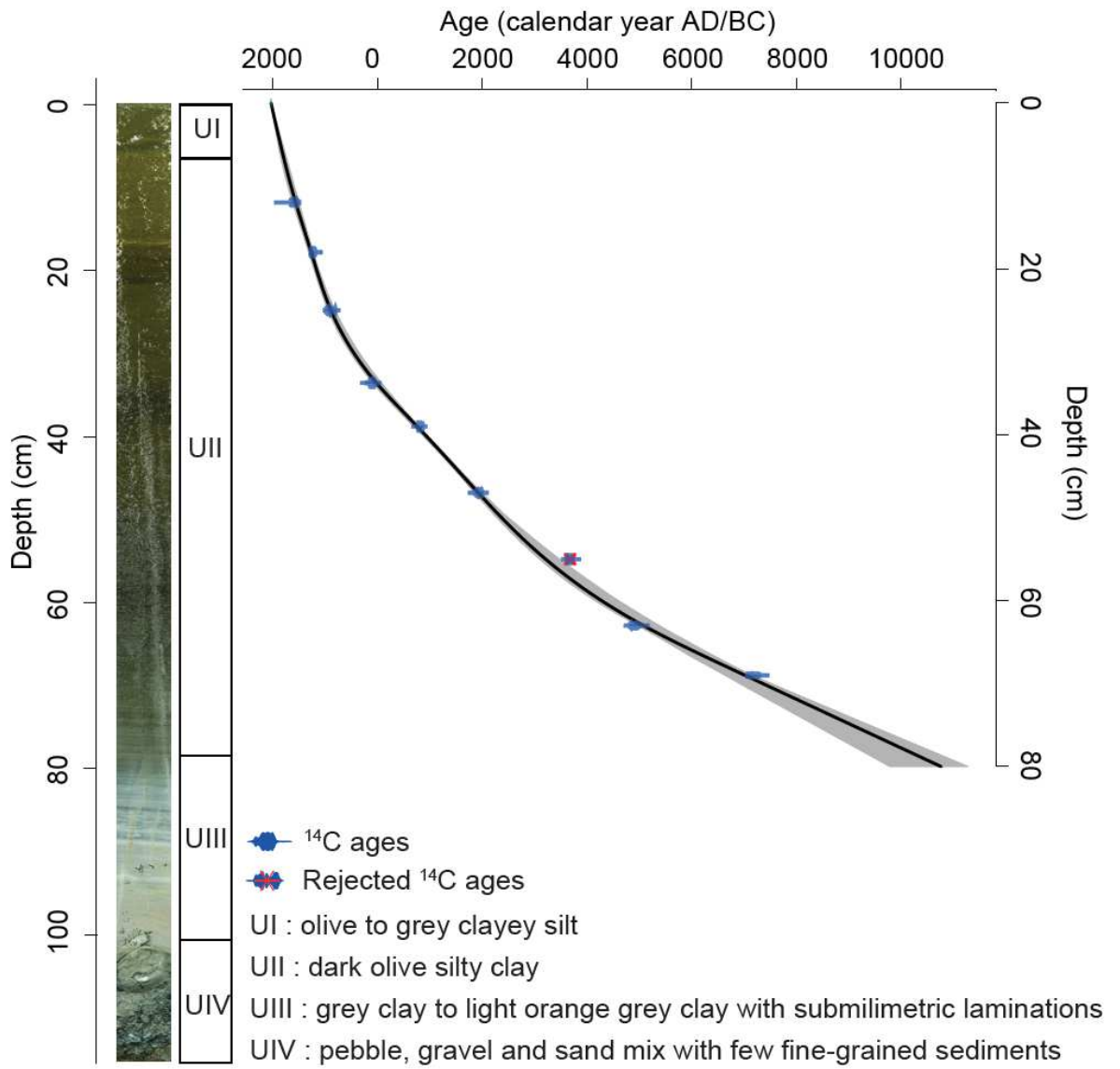


Fig. 2: Age-depth model and lithostratigraphic description of core ROB15-P5

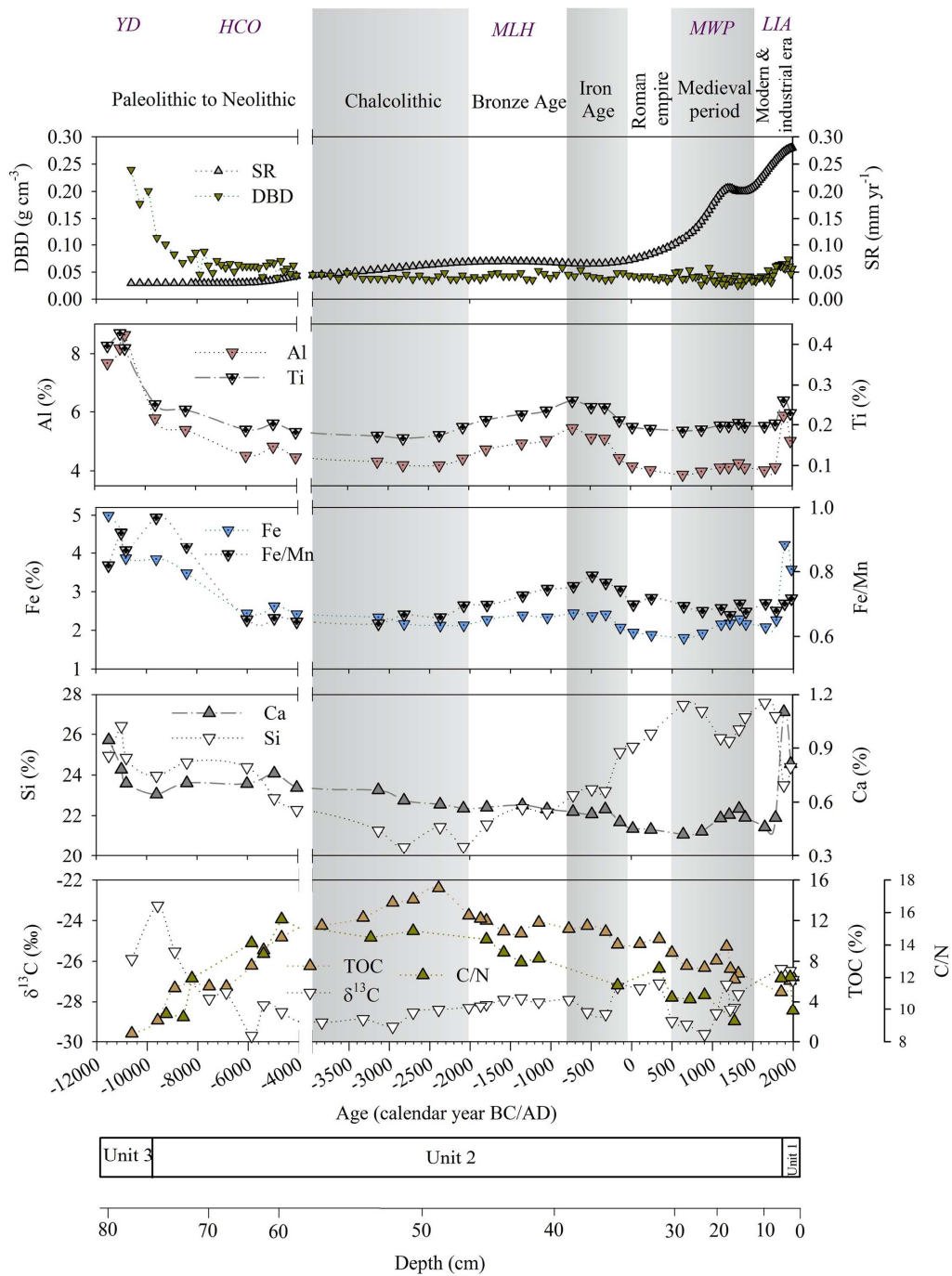


Fig. 3: Dry bulk density (DBD) sedimentation rate (SR), Al, Ti Fe, Mn, Fe/Mn, SiO₂, as a function of age (calendar year AD/BC) in core ROB15-P6. Sediment levels older than 10,800 BC could not be dated and were assigned arbitrary dates.

(YD: Younger Dryas; HCO: Holocene climatic optimum; MLH: middle to late Holocene ; MWP: medieval warm period; LIA: little ice age)

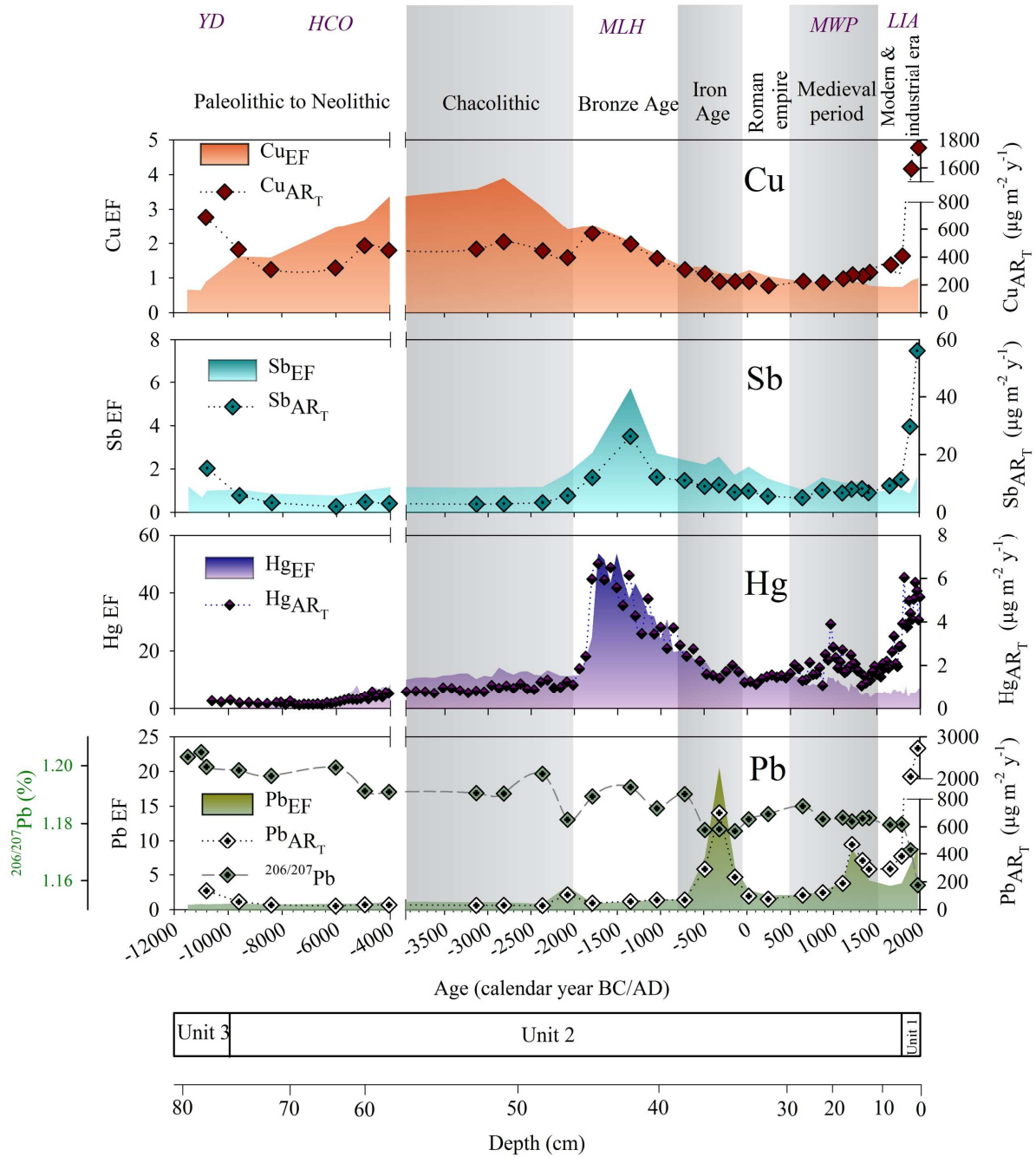


Fig. 4: Enrichment factors (EF) relative to background concentration normalized to Ti, total accumulation rates (AR_T) for Cu, Sb, Hg and Pb, ²⁰⁶Pb/²⁰⁷Pb vs calibrated age (AC/DC) in Lake Robert. Sediment levels older than 10,800 cal BC could not be dated and were assigned arbitrary dates.

See text for the calculation of EF.

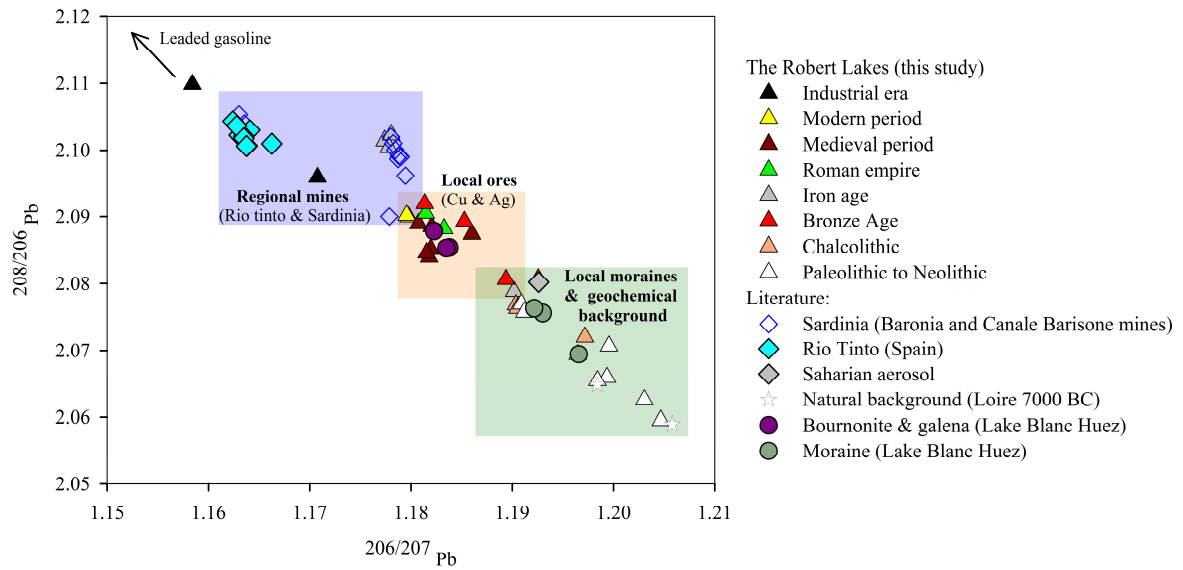


Fig. 5: Plot of $^{206}\text{Pb}/^{207}\text{Pb}$ ratios vs $^{208}\text{Pb}/^{206}\text{Pb}$ ratios in Lake Robert and in potential natural and anthropogenic sources. Sources of literature data: Sardinia and Rio Tinto (Oxford archaeological lead isotope database (OXALID) (Stos-Gale and Gale, 2009); Saharan aerosol (Alfonso et al., 2001); natural background (Elbaz-Poulichet et al., 1984); Bournonite, Galena and Moraine (Garçon et al., 2012)

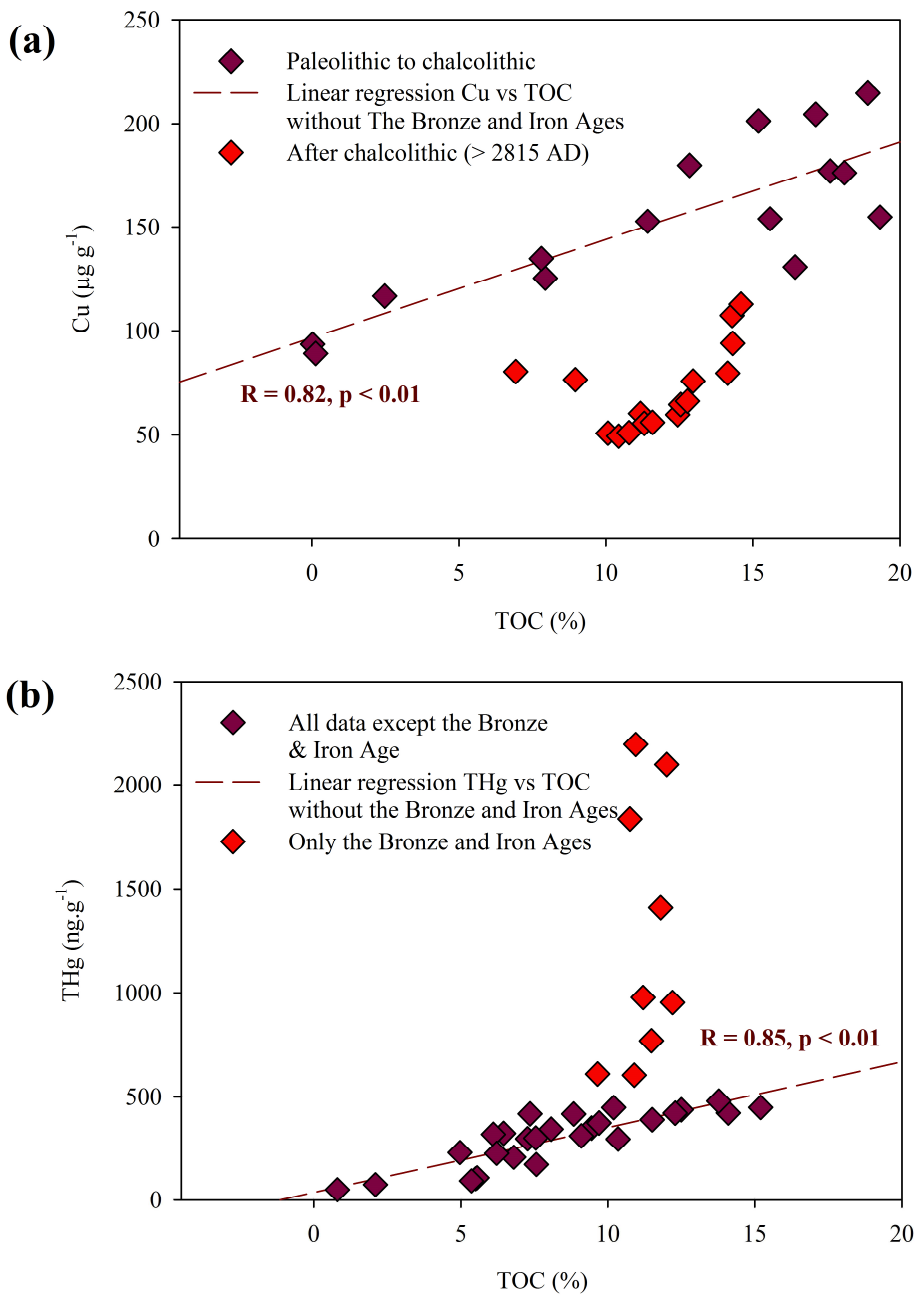
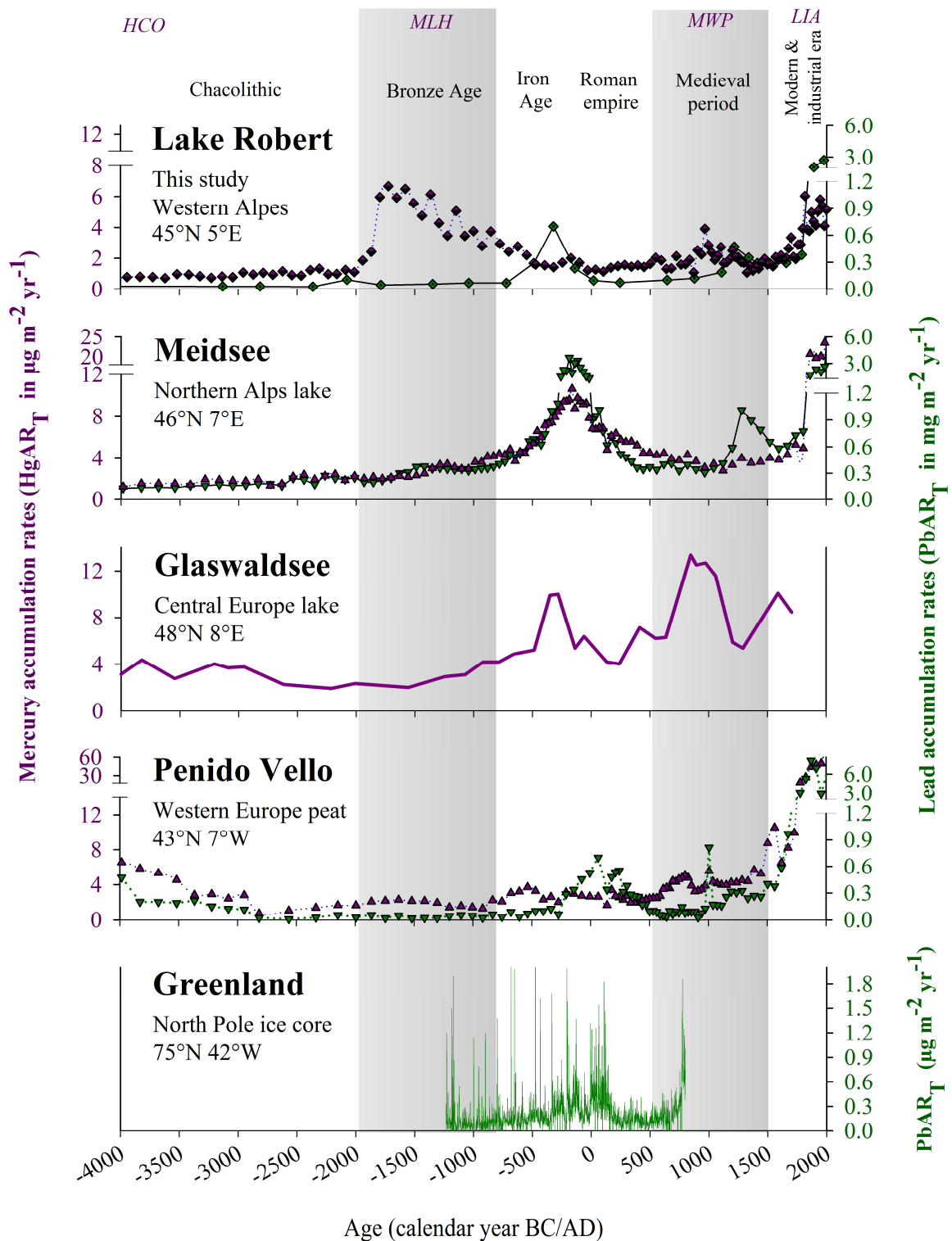


Fig.6: Cu (a) and Total Hg (b) vs TOC



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Fig. 7: Pb and Hg AR_T in lake Robert record compared to selected European records: lake Meidsee (Thevenon et al., 2011), Lake Glaswaldsee (Schütze et al., 2018), Penido Vello (Martinez-Cortizas et al., 1999, 1997) and to Greenland Ice record (McConnell et al., 2018)